



UNIVERSITI PUTRA MALAYSIA

***PRODUCTION OF BIOSURFACTANT FROM BIODIESEL SIDE-STREAM
GLYCERINE BY *Pseudomonas aeruginosa* RS6***

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UNIVERSITI PUTRA MALAYSIA
BERILMU BERBAKTI

**PRODUCTION OF BIOSURFACTANT FROM BIODIESEL SIDE-STREAM
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By

SHOBANAH MENON A/P BASKARAN

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfilment of the Requirements for the Degree of Master of Science**

November 2020

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Master of Science

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November 2020

Chair : Assoc. Prof. Mohd Rafein Zakaria, PhD
Faculty : Biotechnology and Biomolecular Sciences

Biosurfactant is an amphipathic compound produced by microorganisms as a secondary metabolite. Since biosurfactants are biologically produced compounds that have been shown to be less harmful to the environment, biodegradable, and safe as opposed to conventional synthetic surfactants because they can be produced from renewable substrates. Biosurfactants showed an efficient and successful application in numerous industries such as in agriculture (biopesticide), medical (antimicrobial agent) and washing industry (cleaning agent). Waste glycerine is a by-product of the biodiesel industry, and the current glycerol market is unable to accommodate the excess amounts produced; as a result, new markets for refined glycerol must be developed. This study aims to obtain the optimized fermentation condition for the production of rhamnolipids biosurfactant by *Pseudomonas aeruginosa* RS6 using waste glycerine as a substrate. In this study, the effect of temperature, initial pH medium, waste glycerine concentrations, nitrogen sources and nitrogen concentrations on the production of rhamnolipid biosurfactant was studied. The rhamnolipid produced was determined by high-performance liquid chromatography (HPLC) method, emulsification index, and surface tension measurement. Rhamnolipids produced under the optimized fermentation conditions were further characterized using liquid chromatography-mass spectrometry (LC-MS) and Fourier transform infrared spectrometry (FT-IR) and evaluated on the antimicrobial activity and toxicity of rhamnolipids produced by strain RS6 using selected pathogens to see the potential of using rhamnolipids in agricultural industry. The optimized conditions for the production of rhamnolipids was obtained when *P. aeruginosa* RS6 was grown in a basal salt medium at initial pH 6.5 supplemented with 1.0% waste glycerine (v/v) and 0.2 M sodium nitrate at 35°C. Moreover, the fermentation was carried out in a 2 L stirred tank bioreactor to see the difference in rhamnolipids production compared to in shake flask. About 2.72 g/L and 3.0 g/L of rhamnolipids were obtained from shake flask and stirred tank bioreactor, respectively. In the crude extract of biosurfactant obtained, FT-IR confirmed the presence of rhamnolipids and

LC-MS confirmed the crude consists of both mono- and di-rhamnolipids. Rhamnolipids produced can inhibit the growth of both Gram-positive (*Bacillus pumilus*) and Gram-negative (*Pantoea stewartii*, *Pantoea ananatis* and *Erwinia mallotivora*) plant pathogens (ranging from 37% to 77% of inhibition) and showed low toxicity on fish embryonic toxicity test (91.67% of zebrafish survivals). Results obtained showed that waste glycerine from biodiesel industry could be used as a renewable carbon source for the production of rhamnolipids by *P. aeruginosa* RS6. The rhamnolipids produce can be applied in the agricultural industry as an antimicrobial agent to combat the pathogenic microorganisms causing diseases to plants.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

**PENGHASILAN BIOSURFAKTAN DARIPADA GLISERIN ALIRAN SISI
BIODIESEL OLEH *Pseudomonas aeruginosa* RS6**

Oleh

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Biosurfaktan adalah sebatian amfipatik yang dihasilkan oleh mikroorganisma sebagai metabolit sekunder. Oleh kerana sebatian biosurfaktan dihasilkan secara biologi yang telah terbukti kurang berbahaya kepada alam sekitar, terbiodegradasi, dan selamat berbanding dengan surfaktan sintetik konvensional kerana ia boleh dihasilkan dari substrat yang boleh diperbaharui. Biosurfaktan menunjukkan aplikasi yang cekap dan berjaya dalam pelbagai industri seperti dalam industri pertanian (biopestisid), perubatan (agen antimikrob) dan basuh (agen pembersih). Sisa gliserin adalah bahan buangan dari industri biodiesel, dan pasaran gliserol semasa tidak dapat menampung jumlah lebihan yang dihasilkan; maka, pasaran baru untuk gliserol tulen mesti dibangunkan. Kajian ini bertujuan untuk memperoleh keadaan fermentasi yang optimum bagi penghasilan biosurfaktan rhamnolipid oleh *Pseudomonas aeruginosa* RS6 menggunakan sisa gliserin sebagai substrat. Dalam kajian ini, kesan suhu, medium pH awal, kepekatan sisa glyserin, sumber nitrogen dan kepekatan nitrogen ke atas penghasilan biosurfaktan rhamnolipid telah dikaji. Rhamnolipid yang dihasilkan ditentukan dengan kaedah kromatografi cecair prestasi tinggi (HPLC), indeks emulsifikasi dan pengukuran ketegangan permukaan. Rhamnolipid yang dihasilkan dalam keadaan fermentasi yang optimum kemudian dicirikan dengan menggunakan kromatografi cecair spektrometri jisim (LC-MS) dan spektrometri infra-merah transformasi Fourier (FT-IR), dan dinilai untuk aktiviti antimikrobial dan ketoksikan rhamnolipid yang dihasilkan oleh strain RS6 menggunakan organisma terpilih untuk melihat potensi penggunaan rhamnolipid dalam industri pertanian. Keadaan optimum untuk penghasilan rhamnolipid diperolehi ketika *P. aeruginosa* RS6 dihidupkan di dalam medium garam asas pada pH awal media 6.5 dengan penambahan 1.0% sisa gliserin (v/v) dan 0.2 M natrium nitrat pada 35°C. Selain itu, fermentasi dilakukan dalam bioreaktor tangki teraduk 2 L untuk melihat perbezaan penghasilan rhamnolipid berbanding dengan dalam kelalang goncang. Kira-kira 2.72 g/L dan 3.0 g/L rhamnolipid diperolehi daripada

kelalang goncang dan bioreaktor tangki teraduk, masing-masing. Dalam ekstrak kasar biosurfaktan yang diperoleh, FT-IR mengesahkan adanya rhamnolipid dan LC-MS pula mengesahkan ekstrak kasar tersebut terdiri daripada mono- dan di-rhamnolipid. Rhamnolipid yang dihasilkan dapat merencatkan pertumbuhan kedua-dua patogen tumbuhan Gram-positif (*Bacillus pumilus*) dan Gram-negatif (*Pantoea stewartii*, *Pantoea ananatis* and *Erwinia mallotivora*) (antara 37% hingga 77% perencatan) dan menunjukkan ketoksikan yang rendah pada ujian ketoksikan embrio ikan (91.67% zebrafish hidup). Keputusan yang diperolehi menunjukkan bahawa sisa gliserin dari industri biodiesel boleh digunakan sebagai sumber karbon yang boleh diperbaharui untuk penghasilan rhamnolipid oleh *P. aeruginosa* RS6. Rhamnolipid yang dihasilkan boleh digunakan dalam industri pertanian sebagai agen antimikrob untuk memerangi mikroorganisma patogenik yang menyebabkan penyakit kepada tumbuhan.

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I certify that a Thesis Examination Committee has met on **19 November 2020** to conduct the final examination of Shobanah Menon Baskaran on her thesis entitled "**Production of Biosurfactant from Biodiesel Side-Stream Glycerine by *Pseudomonas aeruginosa* RS6**" in accordance with the Universities and University College Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15th March 1998. The Committee recommends that the student be awarded the Master of Science.

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LIST OF ABBREVIATIONS

%	Percentage
±	Plus-minus
°C	Degree Celsius
β	Beta
µg/mL	Microgram per millilitre
µL	Microliter
µm	Micrometre
ACP	Acyl carrier protein
<i>B. pumilus</i>	<i>Bacillus pumilus</i>
BSM	Basal salt medium
C/N	Carbon-to-nitrogen
CaCl ₂	Calcium chloride
cm ⁻¹	Per cubic centimetre
CMC	Critical micelle concentration
CoCl ₂ · 6 H ₂ O	Cobalt (II) chloride hexahydrate
CuSO ₄ · 5 H ₂ O	Copper sulfate pentahydrate
DMSO	Dimethyl sulfoxide
dTDPL-rhamnose	Deoxy thymidine diphosphate L-rhamnose
<i>E. coli</i>	<i>Escherichia coli</i>
<i>E. mallotivora</i>	<i>Erwinia mallotivora</i>
E ₂₄	Emulsification index after 24 h
E ₄₈	Emulsification index after 48 h
E ₇₂	Emulsification index after 72 h
ESI	Electrospray ionization
FAME	Fatty acid methyl ester
FeCl ₃ · 6 H ₂ O	Ferric chloride hexahydrate
FT-IR	Fourier transform infrared
g/L	Gram per litre
GC-FID	Gas chromatography with flame ionization detector
h	Hour
HAA	Hydroxyalkanoyloxy alkanooate
HCl	Hydrochloric acid
hpe	Hour post-exposure
hpf	Hour post-fertilization
HPLC	High-performance liquid chromatography
HPLC-UV	High-Performance Liquid Chromatography-Ultraviolet
HSD	Honestly Significant Difference
K ₂ HPO ₄	Dipotassium hydrogen phosphate
KCl	Potassium chloride
KH ₂ PO ₄	Potassium dihydrogen phosphate
L	Litre
LC	Lethal concentration
LC-MS	Liquid chromatography-mass spectrometry
M	Molarity
m/z	Mass to charge
m ³	Cubic metre

MEL	Mannosylerythritol-lipids
MEOR	Microbial enhanced oil recovery
mg	Milligram
mg/mL	Milligram per millilitre
MgSO ₄	Magnesium sulphate
min	Minute
mL	Millilitre
mL/min	Millilitre per minute
mm	Millilitre
mN/m	Millinewton per meter
MnSO ₄ · H ₂ O	Manganese sulfate monohydrate
NA	Nutrient agar
Na ₃ C ₆ H ₅ O ₇ · 2H ₂ O	Sodium citrate dihydrate
NaCl	Sodium chloride
NaNO ₃	Sodium nitrate
NaOH	Sodium hydroxide
NB	Nutrient broth
NTWG	Non-treated waste glycerine
OD	Optical density
<i>P. aeruginosa</i>	<i>Pseudomonas aeruginosa</i>
<i>P. ananatis</i>	<i>Pantoea ananatis</i>
<i>P. stewartii</i>	<i>Pantoea stewartii</i>
RhIA	3-hydroxyacyl-ACP:3- hydroxyacyl-ACP O-3hydroxyl-acyl-transferase
RhIB	Rhamnosyltransferase I
RhIC	Rhamnosyltransferase II
rpm	Rotation per minute
SDS	Sodium dodecyl sulphate
sp.	Species
SPE	Solid-phase extraction
TLC	Thin layer chromatography
TNTC	Too numerous to count
TWG	Treated waste glycerine
UK	United Kingdom
UPM	Universiti Putra Malaysia
USA	United States of America
v/v	Volume per volume
vvm	Volume of air per volume of liquid per minute
wt	Weight
ZnSO ₄ · 7H ₂ O	Zinc sulfate heptahydrate

CHAPTER 1

INTRODUCTION

The word surfactant was derived from the phrase ‘surface active agents’ and defined the behaviour of amphiphilic molecules at the interfaces between different phases, gas, liquid, and solid (Marchant & Banat, 2012). The surface-active compounds self-aggregate in both polar and non-polar solvent systems and are therefore used in countless industrial and consumer products as emulsifiers, dispersants, wetting agents, and foaming agents (Bhadani *et al.*, 2020). Surfactants are essential components for most consumer products, such as detergents, fabric softeners, shampoos, cosmetics, body lotions, and personal care products. They are also commonly used in papermaking industry, dyeing and textile manufacturing, food and beverage processing, pigments, paints and coatings industries, and so forth (Kumar & Tyagi, 2014). In the pharmaceutical industries, surfactants are an essential component for the solubilization of hydrophobic drugs in aqueous media, as components of emulsions, surfactant self-assembly vehicles for oral and transdermal drug delivery, and as agents for enhancing drug absorption and penetration (Pandey *et al.*, 2014). In the agricultural industries, the agrochemical formulations include the usage of surfactant, which is not only important for its preparation and long-term physical stability maintenance, but also important for improving the biological efficiency of agrochemicals, enhancing the foliar absorption of herbicides, growth boosters and defoliant (Castro *et al.*, 2014).

The efficacy of the surfactant is defined by its ability to lower the surface tension, which is a measure of the surface free energy per unit area needed to bring the molecule from the bulk phase to the surface (Soberón-Chávez & Maier, 2011). The physicochemical characteristics that characterize a surfactant are its abilities to enhance the apparent water solubility of hydrophobic compounds, to create water-hydrocarbon emulsions, and to reduce surface tension (Desai & Banat, 1997). At present, commercial surfactants are synthesized from petrochemicals, animal fats, plants, and microorganisms (Akbari *et al.*, 2018). Surfactants were initially produced using renewable resources like fats and oil, while today, the majorities are of petrochemical origin (Rufino *et al.*, 2014). Since synthetic surfactant reduces worldwide non-renewable petrochemical resources, has low biodegradability and also can contribute to the environmental pollution, a green alternative which is microbial surfactant has been introduced as this surfactant is biologically produced. Besides, microbial surfactant also exhibits special characteristic which is constant effectiveness over a wide range of pH and temperature (Oliveira *et al.*, 2009).

Biosurfactants are produced by plants and animals as well as microorganisms. A wide range of research been conducted on surfactants produced by bacteria, yeast, and fungi (Soberón-Chávez & Maier, 2011). Biosurfactants are derived from

renewable resources and are low or non-toxic, biodegradable, exhibit excellent surface activity, possess high specificity, demonstrate efficacy under extreme conditions and can also be reused as opposed to conventional surfactants (Koglin *et al.*, 2010; Lima *et al.*, 2011; Xu *et al.*, 2011). Among biosurfactants, rhamnolipids, a class of glycolipid mainly known from bacteria of the *Pseudomonas* genus are mostly studied (Henkel *et al.*, 2012). Rhamnolipids have been shown to reduce the water surface tension from 72 to 28 mN/m, and the water-oil system interfacial tension from 43 to <1 mN/m (Kaskatepe & Yildiz, 2016). Nevertheless, the values are in the range of an effective and often used surfactant, sodium dodecyl sulfate (SDS). Antimicrobial properties are another additional feature of the rhamnolipids that makes them more attentive. There are several reports stated that the rhamnolipids produced have been demonstrated to be effective against Gram-positive bacteria, Gram-negative bacteria, fungi (Benincasa *et al.*, 2004; Haba *et al.*, 2003) and insect pest (Kim *et al.*, 2011). According to Chen *et al.* (2017), there is a dual mechanism of action for rhamnolipids where they have antimicrobial properties and also activate defensive responses against pathogens in plants. This dual property would lead to productivity improvements and the creation of new biopesticides.

Crude glycerine, also known as crude glycerol, is the by-product produced during biodiesel production. Crude glycerine is generally refined to pure glycerol, valuable raw material for industries, for example, food and beverages, pharmaceutical, cosmetics, tobacco, and textiles (Luo *et al.*, 2016). The amount of waste glycerine being produced is at a volumetric ratio of 1:10 to the amount of biodiesel production, which is for every 1 m³ of biodiesel, 0.1 m³ of crude glycerine is produced (Monteiro *et al.*, 2018). Crude glycerine purification is a tedious approach, and so the use of crude glycerine as it is as a source for any industrial product is a value-added approach (Garlapati *et al.*, 2016). Not only does the abundance of crude glycerine generated by the biodiesel industry affect the cost of the production of biodiesel, but it also critically creates an environmental problem (Anuar & Abdullah, 2016). As biodiesel production rises worldwide, glycerine generation will increase accordingly, creating a new issue which is the disposal of waste glycerine (Luo *et al.*, 2016; Quispe *et al.*, 2013). It can be not only expensive but also wasteful to dispose crude glycerine (He *et al.*, 2017). In this manner, it is necessary to find new uses for crude glycerine to improve the economic sustainability of the biodiesel industry and reduce the environmental effects of crude glycerine waste disposal. In recent years, intense research into the use and processing of crude glycerine is on the way with increased yields (Ethier *et al.*, 2011; Moon *et al.*, 2010; Sabourin-Provost & Hallenbeck, 2009). Many known microorganisms can use glycerol naturally as a sole source of carbon and energy. The low-cost glycerol can be used as a water-soluble substrate for the production of biosurfactants (Makkar *et al.*, 2011).

Biosurfactants have already been used in the formulations of a few commercial products, mostly from far east in Asia, but there are still a few issues to overcome before they can be used more widely (Marchant & Banat, 2012). One of the problem is the low yield of biosurfactant being produced. The types and yields of biosurfactants produced are influenced by culture medium, carbon source, and the

growth conditions (pH, temperature, limiting nutrients, and trace elements) (Batista *et al.*, 2006). This study aims to determine the optimal environmental factors: temperature, initial pH medium, concentration of waste glycerine, nitrogen source and its concentration for maximum production of rhamnolipids biosurfactant by *Pseudomonas aeruginosa* RS6 with the fermentation of waste glycerine obtained from biodiesel production. The main objectives for this study are;

1. To obtain an optimum fermentation condition for the production of rhamnolipids in shake flask and stirred tank bioreactor by utilizing waste glycerine as a carbon source.
2. To determine the properties of rhamnolipids produced from *Pseudomonas aeruginosa* RS6.



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APPENDICES

Appendix A

Extraction of biosurfactant

- (a) Before vigorous shaking of acidified cell-free supernatant and ethyl acetate.
- (b) After two times of extraction process.

(a)



(b)



Appendix B

Crude extract

This is the crude extract obtained after concentrating using a rotary evaporator.



Appendix C

2 L bioreactor

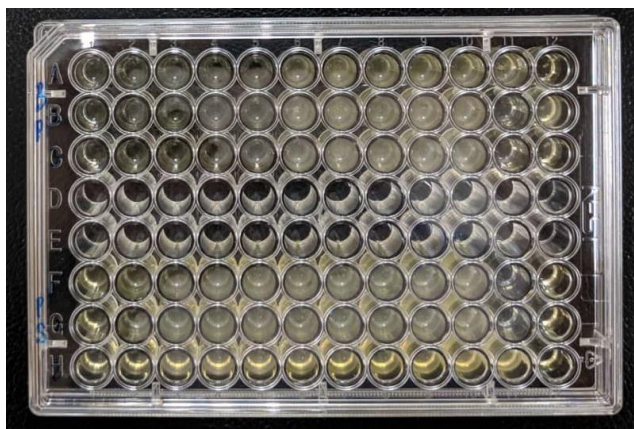
Fermentation of biosurfactant in a 2 L bioreactor with recycling accessory.



Appendix D

Antimicrobial activity using the microdilution method

This is the 96-well flat-bottom microdilution plates used to determine the antimicrobial activity of rhamnolipids against plant pathogens.



Appendix E

Results of HSD function of SAS (version 9)

This is the example of analysis output from the SAS for the parameter temperature.

The SAS System 16:25 Tuesday, July 27, 2020 1

Obs	trt	rep	RI	Ed	Ec
1	25	1	1.117	30.000	37.931
2	25	2	1.831	26.667	26.667
3	25	3	1.504	46.667	21.530
4	30	1	1.423	58.065	13.333
5	30	2	1.417	60.000	10.345
6	30	3	1.420	58.065	15.333
7	35	1	1.735	60.000	60.000
8	35	2	1.889	60.000	43.000
9	35	3	1.664	60.000	51.852
10	37	1	0.944	56.667	16.667
11	37	2	0.985	61.290	29.667
12	37	3	1.115	56.667	14.286

The ANOVA Procedure

Class Level Information

Class	Levels	Values
trt	4	25 30 35 37
rep	3	1 2 3

Number of observations 12

The ANOVA Procedure

Dependent Variable: RI

Source	DF	Sum of Squares > F	Mean Square	F Value	Pr
Model	3	0.85743333 0.0097	0.28581111	7.67	
Error	8	0.29791733	0.03723967		
Corrected Total		11	1.15535067		
R-Square		Coeff Var	Root MSE	RI Mean	
0.742141		13.58666	0.192976	1.420333	

Source	DF	Anova SS > F	Mean Square	F Value	Pr
trt	3	0.85743333 0.0097	0.28581111	7.67	

The ANOVA Procedure
Dependent Variable: Ed

Source	DF	Sum of Squares > F	Mean Square	F Value	Pr
Model	3	1358.778632 0.0013	452.926211	14.71	
Error	8	246.376829	30.797104		

Corrected Total 11 1605.155461

R-Square 0.846509
Coeff Var 10.50235
Root MSE 5.549514
Ed Mean 52.84067

Source	DF	Anova SS > F	Mean Square	F Value	Pr
trt	3	1358.778632 0.0013	452.926211	14.71	

The ANOVA Procedure
Dependent Variable: Ec

Source	DF	Sum of Squares > F	Mean Square	F Value	Pr
Model	3	2529.951160 0.0011	843.317053	15.51	
Error	8	435.019895	54.377487		

Corrected Total 11 2964.971054

R-Square 0.853280
Coeff Var 25.97958
Root MSE 7.374109
Ec Mean 28.38425

Source	DF	Anova SS > F	Mean Square	F Value	Pr
trt	3	2529.951160 0.0011	843.317053	15.51	

The ANOVA Procedure

Tukey's Studentized Range (HSD) Test for R1

NOTE: This test controls the Type I experimentwise error rate, but it generally has a higher Type II error rate than REGWQ.

Alpha	0.05
Error Degrees of Freedom	8
Error Mean Square	0.03724
Critical Value of Studentized Range	4.52880
Minimum Significant Difference	0.5046

Means with the same letter are not significantly different.

Tukey Grouping	Mean	N	trt
A	1.7627	3	35
	A		
B A	1.4840	3	25
	B A		
B A	1.4200	3	30
	B		
B	1.0147	3	37

The ANOVA Procedure

Tukey's Studentized Range (HSD) Test for Ed

NOTE: This test controls the Type I experimentwise error rate, but it generally has a higher Type II error rate than REGWQ.

Alpha	0.05
Error Degrees of Freedom	8
Error Mean Square	30.7971
Critical Value of Studentized Range	4.52880
Minimum Significant Difference	14.51

Means with the same letter are not significantly different.

Tukey Grouping	Mean	N	trt
A	60.000	3	35
	A		
A	58.710	3	30
	A		
A	58.208	3	37
B	34.445	3	25

The ANOVA Procedure

Tukey's Studentized Range (HSD) Test for Ec

NOTE: This test controls the Type I experimentwise error rate, but it generally has a higher Type II error rate than REGWQ.

Alpha	0.05
Error Degrees of Freedom	8
Error Mean Square	54.37749
Critical Value of Studentized Range	4.52880
Minimum Significant Difference	19.281

Means with the same letter are not significantly different.

Tukey Grouping	Mean	N	trt
A	51.617	3	35
B	28.709	3	25
B	20.207	3	37
B	13.004	3	30

BIODATA OF STUDENT

Shobanah Menon Baskaran was born on 10th of January 1995 at Hospital Kajang, Kajang, Selangor, Malaysia. Her educational journey started at Sekolah Kebangsaan Cina Yu Hua, Kajang from 2002 to 2007. She then furthered her secondary education at Sekolah Menengah Kebangsaan Jalan Empat, Bandar Baru Bangi, Selangor from 2008 to 2012. Upon receiving her Sijil Pelajaran Malaysia (SPM) result in 2013, she was offered to do Foundation in Agricultural Science in Universiti Putra Malaysia. The author graduated with a Bachelor of Science with Honours in Biotechnology from Universiti Putra Malaysia in 2018. Right after completing her bachelor's degree, she started her master's degree in September 2018 in Environmental Biotechnology under the supervision of Associate Professor Dr. Mohd Rafein Zakaria.



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