

# UNIVERSITI PUTRA MALAYSIA

# EFFECTS OF NON-STOICHIOMETRY ON MAGNETIC PROPERTIES AND MICROSTRUCTURE OF Nio.3Zno.7Fe2±x04 AND Mgo.sZno.s Fe2±x04

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DOCTOR OF PHILOSOPHY UNIVERSITI PUTRA MALAYSIA



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By

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## EFFECTS OF NON-STOICHIOMETRY ON MAGNETIC PROPERTIES AND MICROSTRUCTURE OF Ni<sub>0.3</sub>Zn<sub>0.7</sub>Fe<sub>2±X</sub>O<sub>4</sub> AND Mg<sub>0.5</sub>Zn<sub>0.5</sub> Fe<sub>2±X</sub>O<sub>4</sub>

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Various non-stoichiometric compositions of Ni-Zn ferrites and Mg-Zn ferrites were investigated. The samples were prepared by the conventional sintering method while being subjected to an air atmospheric condition. The measurement of magnetic properties such as magnetic permeability, magnetic loss, Curie temperature, magnetic flux density and microstructure were performed to understand the magnetic properties of samples prepared by systematic compositional changes. X-ray diffraction results indicate that the samples are in good crystalline form. Curie temperature variation can be explained on the basis of Neel's two sub-lattices model and could be due to distribution of magnetic ions between two sub-lattices. The dependence of magnetic permeability with temperature shows that the trends exhibited by all the samples are similar. With increase in temperature, permeability increases gradually and then shows sudden drop at the Curie point because the anisotropy constant decreases more rapidly than the saturation magnetization. The dependence of permeability on iron oxide content obeys Globus model. The change of permeability in the cation-deficient region is caused



by the positive contribution of the anisotropy constant to the total anisotropy during conversion of  $Fe^{3+}$  ions to  $Fe^{2+}$  ions. However, for the anion-deficient region, the variation in permeability is mainly accounted to be due to microstructural changes.

The loss factor and the quality factor were also calculated and this study has revealed that the iron oxide concentration is a determining factor for high quality ferrites. Finally, this study shows that high permeability values of Ni-Zn and Mg-Zn ferrites occur in the cation-deficient region with x = 0.002-0.006 weight percent.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

## KESAN TAK STOIKIOMETRI TERHADAP SIFAT MAGNET DAN MIKROSTRUKTUR Ni<sub>0.3</sub>Zn<sub>0.7</sub>Fe<sub>2±x</sub>O<sub>4</sub> DAN Mg<sub>0.5</sub>Zn<sub>0.5</sub> Fe<sub>2±x</sub>O<sub>4</sub>

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Berbagai komposisi tak-stoikiometri Ni-Zn ferit dan Mg-Zn ferit telah dikaji. Sampelsampel disediakan dengan kaedah pensinteran lazim dan didedahkan kepada keadaan atmosfera udara. Sifat magnet yang diukur seperti ketelapan magnet, kehilangan magnet, suhu Curie, ketumpatan magnet tepu dan mikrostruktur telah dikaji untuk memahami ciri-ciri magnet bagi sampel yang disediakan melalui perubahan komposisi yang sistematik. Pembelauan sinar-X menunjukkan sampel berada dalam bentuk hablur yang baik. Perubahan suhu Curie boleh diterangkan berasaskan kepada model dua subkekisi Neel dan kepada taburan ion-ion bermagnet pada kedua-dua tapak. Kebergantungan ketelapan magnet terhadap suhu menunjukkan lengkung yang dipamerkan untuk semua sampel adalah sama. Dengan pertambahan suhu, ketelapan meningkat dan menunjukkan penurunan yang mendadak pada titik Curie disebabkan oleh pemalar ketakisotropan mengurang secara mendadak berbanding dengan



PERPUSTAKAAN SULTAN ABDUL SAMAD ketelapan magnet dengan iron mematuhi model Globus. Perubahan ketelapan magasta pada kawasan kurang-kation adalah disebabkan oleh sumbangan pemalar ketakisotropan positif terhdap jumlah ketakisotropan semasa pertukaran ion Fe<sup>3+</sup> kepada Fe<sup>2+</sup>. Bagaimanapun untuk kawasan kurang-anion, perubahan ketelapan terutamanya disebabkan oleh perubahan mikrostruktur.

Faktor kehilangan dan faktor kualiti juga dikira dan kajian ini telah menunjukkan kepekatan oksida ferum adalah faktor penentu untuk kualiti ferit yang baik. Akhir sekali, kajian ini menunjukkan nilai ketelapan tinggi Ni-Zn dan Mg-Zn ferit terhasil pada kawasan kurang kation dengan x = 0.002-0.006 peratus berat.



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# TABLE OF CONTENTS

ABSTRACT	ii
ABSTRAK	iv
ACKNOWLEDGEMENTS	vi
APPROVAL	vii
DECLARATION	ix
LIST OF TABLES	xiii
LIST OF FIGURES	xiv
LIST OF SYMBOLS AND ABBREVIATIONS	xvii

## CHAPTER

1

INTRODUCTION	1
General	1
Non-stoichiometry Overview	5
Ni-Zn Ferrites	6
Mg-Zn Ferrites	6
Statement of Objectives	8
Main Objective	9
Scope of Study	9
Future Prospects of Ferrite	10
Basic Research	10
Development of New Ferrites	11
Improvement in Manufacturing Technology	12
New Applications	13

# 2

# LITERATURE REVIEW

Chemical Consideration	15
Effects of Non-Stoichiometry on	
Magnetic Properties	15
Effects of Iron Content on Magnetic Properties	17
Effect of iron Oxide Concentration on	
Microstructure	23
Microstructural Consideration	25
Grain Size Effects on Permeability	25
Effects of Porosity on permeability	28
Grain Boundary Consideration	29



THEORY	31
Introduction	31
Chemistry of Spinel Ferrites	32
Ferrimagnetism	36
Weiss Domain	39
Properties Influencing Magnetic Behavior	42
Initial Permeability	42
Control of Magnetic Anisotropy and	
Magnetostriction	46
Electrical Resistivity	49
Magnetic Losses	53
Ferrite Microstructure	56
Grain Growth	57
Grain Growth in the presence of Pores,	
Inclusion or Solid Solution Impurities	59
Crystal Imperfections	61
Types of Point Defects	62
Vacancies	62
Interstitial	63
Impurity Atoms	63
Schottky Defect	63
Frenkel Defect	64
Non-Stochiometric Compound	65
Effects of Impurities on Non-Stochiometric	
Compound	70
Diffusion in Ceramic Oxides	70
Vacancy Diffusion	71
Interstitial Diffusion	73

5
5
6
6
7
8
8
9
9
0
1
1
3
4
4
5
5



Curie Temperature	86
Resistivity	86
Density	87
Hysteresis Loop	87
Grain-Size and Pore size Measurement	89
X-ray Diffraction Measurement	89
Shrinkage	89
Error Estimate	90

<b>RESULTS AND DISCUSSION</b>	91
Introduction	91
Lattice Constant	92
Microstructure and Grain Growth	98
Magnetic Permeability	110
Hysteresis Loop	116
Magnetic Flux Density	121
Curie Temperature	127
Sintered Density	133
Resistivity	139
Relative Loss Factor (RLF)	146

6	CONCLUSION	153
	Summary of Main Results	154
	General Conclusion	157
	Suggestion	158

BIBLIOGRAPHY	159
APPENDICES	172
<b>BIODATA OF THE AUTHOR</b>	177
LIST OF PBLICATIONS	177



## LIST OF TABLES

Table	Page
1.1: Summary of Ni-Zn ferrites applications	6
1.2: Summary of Mg-Zn ferrite applications	7
2.1: Magnetostriction of some ferrites	18
3.1: Cation distribution of spinel ferrites	34
3.2: Radii of metal ions involved in spinel ferrite	35
3.3: Radii of tetrahedral and octahedral sites in some ferrites	35
3.4: Magnetic moment of some simple ferrites	37
4.1: The error estimate for characteristic of Ni-Zn and Mg-Zn ferrites	90
5.1: Data of average grain size and shrinkage of $Ni_{0.3}Zn_{0.7}Fe_{2\pm X}O_4$	101
5.2: Data of average grain size and shrinkage of $Mg_{0.5}Zn_{0.5}Fe_{2\pm X}O_4$	105
5.3: Hysteresis parameters of $Ni_{0.3}Zn_{0.7}Fe_{2\pm X}O_4$	118
5.4: Hysteresis parameters of Mg <sub>0.5</sub> Zn <sub>0.5</sub> Fe <sub>2±X</sub> O <sub>4</sub>	120



LIST	OF	FIG	URES
------	----	-----	------

Figure	Page
2.1: Variation of dependencies of the permeability of some Mn ferrites with composition from Guillaud (1957)	19
2.2a: Permeability of Mn-Zn ferrites as a function of $Fe_2O_3$	20
2.2b: Variation of magnetostriction of the same Mn-Zn ferrites as a function of Fe <sub>2</sub> O <sub>3</sub> content	21
2.2c: Variation of magnetostriction of Mn-Zn ferrites with Guillaud (1957)	21
2.3: Permeability of a Mn-Zn ferrite as a function of grain size in microns. Guillaud	26
2.4: Permeability of a Ni-Zn ferrite as a function of grain size in microns. Guillaud	27
3.1: Spinel structure of ferrites	33
3.2: Ferrimagnetism; a) anti-parallel allignment of magnetic moments of two sub-lattices b) the net magnetic moment of the lattice	36
3.3: Super-exchange interaction by mediated oxygen ion	38
3.4: Magnetic domain grain growth and rotation as a demagnetized ferromagnetic material is magnetized to saturation by an applied magnetic field	40
3.5: Variation of permeability and crystal anisotropy as a function of Fe <sub>2</sub> O <sub>3</sub> . Guillaud (1960)	45
3.6: Variation of permeability and magnetostriction as a function of Fe <sub>2</sub> O <sub>3</sub> . Guillaud (1960)	45
3.7: Variation of permeability and crystal anisotropy as a function of Fe <sub>2</sub> O <sub>3</sub> . Guillaud (1960)	46
3.8: Temperature dependence of the resistivity for Ni-Zn and Mn-Zn ferrites	50
3.9: Dependence of the resistivity of the ferrite $Ni_{0.3}Zn_{0.7}$ on the iron content	52
<ol> <li>Typical μ', μ" – frequency specterum show magnetic resonance, domain wall relaxation and diffusion relaxation</li> </ol>	56

3.11:	Schematic drawing of polycrystalline speciment with grain growth mechanism. The boundaries move toward the centre curvature	
	(arrows). As a results, the small grains eventually disappear	59
3.12:	Two-dimensional representations of vacancy and interstitial	62
3.13:	Model of lattice disorder in ionic compound a) Frenkel defect and b) schottky defect	64
3.14:	A model of lattice disorder in the cation sub-lattice (extra positive charge)	68
3.15:	A model of lattice disorder in the cation sub-lattice. The circle with two plus sighns stands for an electron hole	<b>69</b>
3.16:	Vacancy mechanism of diffusion	72
3.17:	Relaxation mechanism of diffusion	73
3.18:	Interstitial mechanism of diffusion	73
3.19:	Interstitial mechanism of diffusion. The collinear variant	74
3.20:	Interstitial mechanism of diffusion. The non-collinear variant	74
4.1: ]	Schematic flow chart of the compositional design technique For Ni-Zn and Mg-Zn	77
4.2:	Process steps for microstructural analysis for Ni-Zn and Mg-Zn Ferrites	83
4.3: ] 1	Measurement of Lp and Q factor using HP 4195A and HP 4284A precision LCR meter	84
4.4:	An example of direction of current passing through a toroidal ferrite sample	86
4.5:	Density measurement by Archimedes principle/method	87
4.6:	Shows the connection of the sample to the Walker Scientific AMH-20	88
4.7:	The circuit configuration for hysteresis measurement	88
5.1:'	The XRD patterns of Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2±X</sub> O <sub>4</sub> sintered at 1300 °C.	92
5.2:	The variation of lattice constant for $Ni_{0.3}Zn_{0.7}Fe_{2+X}O_4$ .	93
5.3:	The XRD patterns of Mg <sub>0.5</sub> Zn <sub>0.5</sub> Fe <sub>2±x</sub> O <sub>4</sub> sintered at 1300 °C	96
5.4:	The variation of lattice constant for $Mg_{0.5}Zn_{0.5}Fe_{2+X}O_4$ .	97
5.5:	Micrograph of stochiometric composition of Nio $_{3}Zn_{0.7}Fe_{2+x}O_{4}$ (x = 0)	99



5.6: Typical optical micrographs of $Ni_{0.3}Zn_{0.7}Fe_{2+X}O_4$ with excess iron with $x = a$ )0.002, b)0.004, c)0.006, d)0.008 and e)0.010	100
5.7: Typical optical micrographs of Ni Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2-x</sub> O <sub>4</sub> with deficit iron with x = a)0.002, b)0.004, c)0.006, d)0.008 and e)0.010	103
5.8: Micrograph of stochiometric composition of $Mg_{0.5}Zn_{0.5}Fe_{2\pm X}O_4$	106
5.9: Typical optical micrographs of $Mg_{0.5}Zn_{0.5}Fe_{2+X}O_4$ with excess iron with x = a)0.002, b)0.004, c)0.006, d)0.008 and e)0.010	107
5.10: Typical optical micrographs of Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2-X</sub> O <sub>4</sub> with deficit iron with x = a)-0.002, b)-0.004, c)-0.006, d)-0.008 and e)-0.010	109
5.11: The variation of magnetic permeability with iron-oxide content	110
5.12: The variation of magnetic permeability for Mg-Zn ferrites	114
5.13: Hysteresis loop for Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2+X</sub> O <sub>4</sub>	117
5.14: Hysteresis loop for Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2-x</sub> O <sub>4</sub>	117
5.15: Hysteresis loop for Mg <sub>0.5</sub> Zn <sub>0.5</sub> Fe <sub>2+x</sub> O <sub>4</sub>	119
5.16: Hysteresis loop for Mg <sub>0.5</sub> Zn <sub>0.5</sub> Fe <sub>2+X</sub> O <sub>4</sub>	120
5.17: The variation of magnetic flux density for $Ni_{0.3}Zn_{0.7}Fe_{2\pm X}O_4$ .	121
5.18: The variation of magnetic flux density for $Mg_{0.5}Zn_{0.5}Fe_{2\pm X}O_4$	124
5.19: Variation of inductance with temperature for $Ni_{0.3}Zn_{0.7}Fe_{2\pm X}O_4$ .	127
5.20: The variation of Curie temperature for for Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2+x</sub> O <sub>4</sub>	129
5.21: Variation of inductance with temperature for $Mg_{0.5}Zn_{0.5}Fe_{2\pm X}O_4$	131
5.22: The variation of Curie temperature for $Mg_{0.5}Zn_{0.5}Fe_{2.2}O_4$	132
5.23: The variation of sintered density of $Ni_{0.3}Zn_{0.7}Fe_{2+X}O_4$	133
5.24: The variation of sintered density of $Mg_{0.5}Zn_{0.5}Fe_{2+x}O_4$	137
5.25: Variation of resistivity of Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2+X</sub> O <sub>4</sub>	139
5.26: Variation of resistivity of $Mg_{0.5}Fe_{24}xO_4$	143
5.27: Variation of relative loss factor (RLF) of Ni <sub>0.3</sub> Zn <sub>0.7</sub> Fe <sub>2±X</sub> O <sub>4</sub>	146
5.28: Variation of relative loss factor (RLF) of Mg <sub>0</sub> 5Zn <sub>0</sub> 5Fe3+xO <sub>4</sub>	150



## LIST OF SYMBOLS AND ABBREVIATIONS

- A cross sectional area
- H applied field
- H<sub>c</sub> coercive force
- $\mu_B$  Bohr magneton
- T<sub>c</sub> Curie temperature
- ρ resistivity
- f frequency
- $\mu$ ' real of permeability or magnetic loss
- $\mu$ " imaginary part of permeability
- B induction
- B<sub>s</sub> saturated induction
- B<sub>r</sub> remanence induction
- L inductance
- D<sub>o</sub> outer diameter
- D<sub>i</sub> inner diameter
- Tan  $\delta$  loss tangent
- N number of wire turns
- PVA polyvinyl alcohol
- RLF relative loss factor
- $\sigma$  internal stress
- T temperature
- t thickness
- K<sub>1</sub> first anisotropy constant.



## **CHAPTER 1**

## INTRODUCTION

## General

Ferrites are ceramic ferromagnetic materials generally gray or black in color containing oxygen and at least two kinds of metal ions, one of which is usually Fe<sup>3+</sup>. They are among the most widely used materials, in many low cost, high performance electronic devices since 19<sup>th</sup> century (Ishino, 1987). Commercially, ferrites are classified into three important classes due to specific crystal structures:

 soft ferrites, 2) garnet structure and 3) hard ferrites with hexagonal structure (Callister, 1990; Levinson, 1988). However, this study only consider the soft ferrites with spinel structure.

Soft ferrites particularly Nickel-Zinc and Manganese-Zinc ferrites are of great interest in high frequency applications including power line filters, local area networks transformers, filter inductors and electromagnetic interference suppressors which are challenging the ferrite industry to produce high quality cores capable of meeting increasing demands. Their usage in this field has been growing unabatedly for several decades (Zhu, 2000) since the appearance of the first commercial ferrite products in about 1945 (Wolfarth, 1980). Recently, Mg-Zn ferrites have become important to industry, because of their applications in intermediate frequency transformers and antenna cores. They are replacing Mn-Zn ferrites in these area besides offering the advantage of easy synthesis (Bhosale et al., 1997).

The main motivation for the uses of these materials are high permeability and low magnetic loss (Znidarsic, 1996, Nomura, 1995) which can only be achieved by carefully controlling the ferrite microstructure and grain boundary chemistry in addition to the chemical compositions. The preparation of high permeability ferrite is a complicated task because there are strongly dependent on several factors such as chemical composition, chemical purity, homogeneity, microstructure (grain size/pore size), stoichiometry and so on (Kang, 2000; Park, 2001; Patil, 1998). Oxygen potential in sintering atmosphere is also well known to have a great influence on magnetic and electrical properties. That is to say, it governs the non-stoichiometry of spinel phase, which can be responsible for the performance of materials (Otsuki and Yamada, 1995).

Ferrite performance is not determined by the high value of initial permeability alone; a low loss value, represented by quality factors, relative loss factors and power loss, is also important. Moreover, high saturation flux density, high-fired density and frequency characteristics are necessary in order to achieve high performance with low cost magnetic materials.

The difference in properties and performance of ferrites as compared with most other magnetic materials is due to the fact that the ferrites are oxide materials rather than metal. Ferromagnetism is derived from the unpaired electron spins in only a few metal atoms, these being iron, cobalt, nickel, manganese and rare earth elements. It is

not surprising that the highest magnetic moments and therefore the highest saturation magnetizations are to be found in metals. The oxides, on the other hand suffer from a dilution effect of the large oxygen ions in the crystal lattice. In addition the net magnetic moment resulting from ferromagnetic alignment of a atomic spins is reduced because a different, less efficient type of exchange mechanism is operative. The oxygen ions do serve a useful purpose, however since they insulate the metal ions and therefore greatly increase the resistivity. These properties make the ferrites especially useful at higher frequencies.

The progress of soft ferrites was started since 1936 using Cu-Zn ferrite that was used for antenna and intermediate frequency transformer. But due to some problems its production was discontinued about 1970. They were gradually replaced by Ni-Cu-Zn ferrite. Presently Ni-Cu-Zn ferrite has many application including noise filter, rotary transformer and multi-layer chip component. The advantages of these chip devices are the excellent magnetic shielding and capability of miniaturization (Hsu, 1995).

Presently, Mn-Zn ferrite acted as the mainstream industrial product of core materials. The main core characteristics are core losses, which contribute to the major part of the total electric loss (Znidarsic, 1996). The core loss of Mn-Zn ferrite has been remarkably decreased year by year as the result of the technical achievements made by many workers (Akashi, 1961; Stinjtjes, 1989).

The Ni-Zn ferrites cores exhibit volume resistivity, moderate temperature stability and high Q factors for the 500 kHz to 100 MHz frequency range. They are well suited for low power, high inductance resonant circuits. Their low loss on frequency



higher than 1 MHz made these ferrites suitable for low flux density applications. The Mn-Mg ferrites with rectangular hysteresis loop properties and usually used for memory and switching cores in digital computers and one of the fastest growing and practically applicable ferrites.

Since 1950s, ferrite materials have been used for many microwave devices such as magneto-static resonator, switches, turnable electro optics modulator, shifters etc. Presently, bulk ferrites and thick film ferrites yttrium iron garnet and Ni-Zn ferrites and Mn-Mg-Zn are most often used as microwave ferrites (Horvath, 2000).

In the preparation of microwave ferrite materials particular attention should be given to the purity of the raw materials, stoichiometry of the composition and the porosity as well as grain characteristics of the final product. A ferrite wave absorber (Ishino, 1987) has received much worldwide attention in response to an increasing demand of EMC. The material used for wave absorbers should have high permeability materials from frequency attenuation from 100 kHz to 1 MHz (Bruce, 1990).

Automotive electronic uses will also expand, for example in electrochemical valve opening and closing and direct fuel injection. From the standpoint of advanced Ceramics, magnetic materials should be developed with new manufacturing technologies such as synthesis by the super-lattice process and plasma jet spraying. Improvement in the conventional sintering process is also urgently required, such as casting ferrite, homogenous ferrite film, single crystal fiber and no pores ferrites (Sugimoto, 1999). However, the new challenge of ferrite technology should consider the cost as the most important factor to an increase in performance.



#### **Non-Stoichiometry Overview**

Soft magnetic materials (ferrites) are non-stoichiometric compound (Tanaka, 1978) where the term of the non-stoichiometry of spinel ferrites (Me, Fe)<sub>3-8</sub>O<sub>4</sub> (Me = Mg, Mn, MgMn and so forth) is due to a direct measure of average oxidation states of transition metal ions and point defect concentration therein (Kang et al., 1999). Consequently, it plays an important role in the magnetic, electrical and mechanical properties of ferrite materials such as permeability, magnetic loss, disaccomodation and microstructure (Gundlach, 1998). So, it is necessary to study the magnetic properties as a function of non-stoichiometry not only to the oxygen partial pressure but also to the Fe<sup>2+</sup> content in the bulk and the grain boundary (Inaba, 1997).

Non-stoichiometry effects on the ceramic properties especially for the ferrite material usually exist with two valence (or ionic) states for one of the ion types. For example, the iron oxide which can be present in both  $Fe^{2+}$  and  $Fe^{3+}$  states; the number of each of these ion types depends on the temperature and the ambient oxygen pressure. The formation of an  $Fe^{3+}$  ion disrupt the electro-neutrality of the crystal by introducing and excess +1 charge, which must be offset by some type of defect. This may be accomplished by the formation of one  $Fe^{2+}$  vacancy (or removal of two positive charges) for every two  $Fe^{3+}$  ions that are formed. The crystal is no longer stoichiometric because there is one more O ion than Fe ion; however the crystal remains electrically neutral. This phenomenon is fairly iron-oxide, and in fact, its chemical formula is often written as  $Fe_{1-x}$  (where x is some small and variable fraction substantially less than unity) to indicate a condition of non-stoichiometry with a deficiency of Fe (Callister, 1990).



## **Ni-Zn ferrites**

Ni-Zn ferrites were developed for a wide range of applications where high permeability and low loss were the main requirement. Ni-Zn ferrite is still one of the most important ferrites for such application and constitutes a substantial portion of present-day soft ferrite production. Ni-Zn ferrite has been extensively used as core materials for large number of devices and electrical components and its application is summarized in Table 1 (Hemeda 2001; Anil Kumar 1997; Wolfarth 1980; Seo, 1999).

Device	Device Function	Frequencies
Inductor Magnetic shielding Suppression bead Antenna rod Recording head	Frequency selection network Block unwanted signal (Cellular phone) EM receiver Information recording	1-100 MHz Up to 250 MHz Up to 15 MHz Up to 10 MHz

Table 1.1: Summary of Ni-Zn ferrite applications.

## **Mg-Zn ferrites**

Besides Ni-Zn ferrites, Mg-Zn ferrites also play a useful role in technological and magnetic application because of the favorable performance (Ahmed 2001, Rezlescu, 1998). In commercial practice, Mg-Zn ferrites used in lower requirement television yokes and fly back transformers because of the lower cost of Mg and because of the

