

## **UNIVERSITI PUTRA MALAYSIA**

# SUPERCONDUCTING PROPERTIES OF SOL-GEL DERIVED Bi(Pb)-Sr-Ca-Cu-O SYSTEM

SALEH AYED AHMAD AL-KHAWALDEH

FSAS 1998 18



## SUPERCONDUCTING PROPERTIES OF SOL-GEL DERIVED Bi(Pb)-Sr-Ca-Cu-O SYSTEM

## SALEH AYED AHMAD AL-KHAWALDEH

MASTER OF SCIENCE UNIVERSITI PUTRA MALAYSIA

1998



## SUPERCONDUCTING PROPERTIES OF SOL-GEL DERIVED Bi (Pb)-Sr-Ca-Cu-O SYSTEM

## By

#### SALEH AYED AHMAD AL-KHAWALDEH

Thesis Submitted in Fulfilment of the Requirements for the Degree of Master of Science in the Faculty of Science and Environmental Studies
Universiti Putra Malaysia

May 1998



رُبُرُ وَ الْمُ الْمُ الْمُ الْمُ الْمُ الْمُ اللّهِ اللّهِ اللّهُ إِنَّ اللّهُ الللللّهُ الللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ الللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللللّهُ اللّهُ الللللّهُ اللللّهُ اللللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّهُ اللّه

إلى من تطيب النهس بلقائهم بعد عناء الغربة ، ، ،
إلى أبيى، عثلي الاعلى في العطاء والتضدية ، ، ،
إلى أميى، خلجة فوءا حيى وقرة عيني التي لا تناو ، ، ،
إلى أحرتي و أخواتي الذين بحبه تفنى المعموم و اليهم أكون ، ، ،
إلى إخرتي و أخواتي الذين بحبه تفنى المعموم و اليهم أكون ، ، ،
إلى الاهل والعشيرة . . . الفنو والانتماء . . .

اليكو جميعاً أشد الإعتذار ... و بإممكو أقدو هذا العمل المتراضع ...



#### **ACKNOWLEDGEMENTS**

Firstly, I am very grateful to ALLAH for giving me the strength and patience to complete this research. I would thank not just the few that inspired me, but also those among who have accepted my mistakes and shortcoming. I am specially grateful to those who have taken time for my research and thesis with their busy schedule to follow my research at various stages of my study.

In this acknowledgement, I am specially indebted to my chairman supervisor Assoc. Prof. Dr. Abdul Halim Shaari, for his advice, comments, suggestions, help and invaluable guidance throughout my research. Also I wish to express my acknowledgement to my co-supervisor Dr. Abdul- Majeed Azad (Abu Ahmed) for his fruitful discussion, continued encouragement, help and guidance. I am also pleased to thank my co-supervisor Dr. Jamil Suradi for his constant support all the time throughout my research work.

I would like to thank Assoc. Prof. Dr. Abdul Halim shaari for providing a full scholarship during my study through its IRPA program supported by the Ministry of Science and Technology of Malaysia.



I appreciate the assistant given by Department of Physics (UPM); the head of department, Assoc. Prof. Dr. Wan Mahmood, the academic staffs, secretaries and the technicians, especially Mr. Razak Harun. My special thanks also goes to my friends in the superconducting and magnetic materials groups, I would like in particular to thank S. B. Mohamed and Azhan Hashim for their numerous contributions and continued discussions in my research progress.

My special thanks go to Assoc. Prof. Dr. Karen Badri (Department of Chemistry, UPM) and Prof. Dr. Radha Krishnan (Institute of Higher Learning, UM) for their unfailing co-operation. I would like also to thank all the graduate and undergraduate students in materials science groups for their friendly co-operation.

I am grateful to Ms. Zurina (XRD Unit, UM), Ms. Azilah (SEM Unit, UPM), Mr. Abdul Wahab (FT-IR Unit UPM) and Mr. Kamal (TGA, DTA Unit, UPM) for their guidance on using the instruments. And last but not least, to Mr. Sri Jegan (Chemistry Department, UPM) for his support and co-operation.



My special gratitude goes to Al al- Bayt University (AABU) and Prof. Dr. Mohammad Adnan Al-Bakheet (The President) for allowing me to proceed for higher education. Also not forgetting, the academic staffs in the Physics Department and friends (AABU) for their moral support and co-operation.

I would like to express my acknowledgement to those who supported me since my arrival to Malaysia; special thanks go to Moh'd Smadi, Eid Ziod, Mohammad Saleh, Dr. Iman Baklisy and Hamed Jasim for their help, unfailing progress and support, also I cannot forget my brothers, Faraj Ahmad Abu Alaiwi and Rafa Elayyan al-Qutaish who shared me the out home dreams. Also I would like to convey my special thanks to my friend, khaleel Al- Hasan and all the Jordanian friends inside and outside UPM for their moral encouragement.

My special thanks go to my friend Isam and his sister Hanan Qudsieh for their numerous contributions and encouragement during my research work. I have also to thank Abdul- Majeed family for their kindness and moral support during my stay in Malaysia. My thanks also go to all the Arabian, Malaysian and other international friends for all their kindness and continued encouragement.



I am specially indebted to my friend Hasan Al- Omoush for keeping me in touch with my family in Jordan during one year, special thanks for his noble deeds that I will never forget it.

Finally, I would like to express my most sincere, warmest gratitude to my father, mother, grandmothers, brothers, sisters, uncles, aunts, cousins, nephews and nieces for their praying, love, generous moral and financial support during my study. Not forgetting all my best friends and relatives in "Bala'ama" my beloved town and Zarka where I was born.

Only Allah can repay to those who have helped me
"Jazakum Allah Kol Khyer"

Amin



## TABLES OF CONTENTS

		Page
ACKNO	WLEDGEMENTS	iii
LIST OF	TABLES	x
LIST OF	PLATES	xii
LIST OF	FIGURES	xiii
ABBRE	VIATIONS AND KEY WORDS	xviii
ABSTRA	СТ	xxi
ABSTRA	<b>NK</b>	xxiii
СНАРТ	'ER	
I	INTRODUCTION  General Introduction  The Main Features of Superconductivity  Pb-BSCCO System  Crystal Structure  Net-Shaping and Application  The Problem Encountered in BSCCO System  Objective of The Thesis  Thesis Synopsis	1 2 8 8 10 15
II	Ceneral History of Superconducting Materials  BSCCO System Prepared by Sol-gel Technique Other Techniques for Preparing BSCCO System  Effect of Pb Doped in Oxide Technique Theory of Superconductivity  Meissner Effect London Theory Non-local Generalization of London Theory Ginzburg- Landau Theory The Electron- lattice Interaction BCS Theory	. 19 . 21 . 25 . 26 . 29 . 31 . 33 . 34 . 35



III	SYNTHESIS AND CHARACTERISATION	
	TECHNIQUES OF BSCCO SUPERCONDUCTORS	42
	General Introduction	42
	Chemical Reactions In Sol-Gel Process	43
	Salient Feature of The Technique	46
	Other Techniques	49
	Precipitation and Co-precipitation Technique	49
	Aerosol Technique	50
	Characterisations	51
	Electrical Resistance	52
	Magnetic Susceptibility	54
	X-ray Diffraction	59
	Scanning Electron Microscopy (SEM)	61
	Thermogravimetric Analyser (TGA)	63
	Infrared Spectroscopy	64
	1 1 3	
IV	SOL-GEL SYNTHESIS OF Bi(Pb)-Sr-Ca-Cu-O	66
	Overview	66
	Chemical Calculations of The Mixed Metals Acetates	67
	Preparation of the Gel	69
	Heat Treatment of the Gel: Study of Gel-To-	
	Crystalline Transition	73
	Results and Discussion	75
	Thermal Analysis of the Heated Gel	75
	Infrared Spectrum Analysis of the Heated Gel	76
	X- ray Diffraction Analysis of the Heated Gel	78
	SEM Observation of the Heated Gel	79
V	INFLUENCE OF HEAT TREATMENT ON SOL-GEL	
	DERIVED Bi(Pb)-Sr-Ca-Cu-O OXIDES	80
	Introduction	80
	Experimental Procedure	81
	Electrical Resistance	83
	The Influence of Sintering Time	83
	The Influence of Sintering Temperature	87
	AC- Susceptibility Measurements	91
	The Influence of Sintering Time	91
	The Influence of Sintering Temperature	95
	Microstructure Investigation	98
	X-ray Diffraction Analysis	101
	The Influence of Sintering Time	101
	The Influence of Sintering Temperature	103



VI	EFFECT OF Pb-DOPING IN SOL-GEL DERIVED	
	BSCCO SYSTEM	105
	Introduction	105
	Chemical Calculations	105
	Preparation of Powder	108
	Electrical Properties	110
	Ac-Susceptibility Measurements	112
	X-ray Diffraction Analysis	114
	Microstructure Observation	116
VII	SOL-GEL SYNTHESIS OF Bi(Pb)-Sr-Ca-Cu-O	
	SUPERCONDUCTORS USING METALS OXIDE	
	CARBONATE AND NITRATE PRECURSOR	118
	Introduction	118
	Experimental Procedure and Methodology	119
	Oxide-Carbonates System	
	Nitrates Precursor	120
	Chemical Calculations	121
	Result and Discussion	126
	Acetate Formation	126
	Thermal and Infrared Analysis of the Gel	130
	Electrical Measurements	132
	X-ray Diffraction	135
	Scanning Electron Microstructure (SEM)	
	Observation	137
VIII	CONCLUSIONS AND RECOMMENDATION	139
	Conclusions	139
	Suggestion for Future Research	143
BIBLIOG	BRAPHY	144
VITA	***************************************	148



## LIST OF TABLES

Table		Page
1.	Lattice parameter and T <sub>c</sub> values of BSCCO phases	9
2.	Atomic mass unit for the starting materials	67
3.	The molecular weight (m.w.) of the starting acetate metals in the system where $x=0.5$	68
4.	Weight of acetates to prepare 20 grams of sample	68
5.	Stoichiometric quantities of water required to hydrolyze the respective acetates	70
6.	Value of critical temperature in superconducting state	85
7.	Critical temperature in superconducting state sintered at various temperatures	88
8.	Summary of molecular weight of the starting acetates	106
9.	Molecular weight of starting acetate when $\underline{x=0.0}$	106
10.	Molecular weight of starting acetate when $\underline{x=0.4}$	106
11.	Molecular weight of starting acetate when $\underline{x=0.5}$	107
12.	Metals acetate required to prepare 20 (g) of sample with $\underline{x=0.0}$	107
13.	Metals acetate required to prepare 20 (g) of sample with $\underline{x=0.4}$	107
14.	Metals acetate required to prepare 20 (g) of sample with $\underline{x=0.5}$	108



15.	Lattice parameters of pure and Pb-doped samples	115
16.	Molecular weight (m.w) of oxide carbonate starting materials	122
17.	Molecular weight of nitrates starting materials	122
18.	The weight of each component of starting materials oxide-carbonate system	122
19.	The weight of each component of starting materials in nitrate precursors	123
20.	Acetic acid and water quantities that required to convert the oxide-carbonate system	124
21.	Acetic acid and water quantities that required to convert the nitrates precursor	125
22.	Experimental weight before and after the conversion reaction.	126
23.	The pH behaviour during the reaction time	127
24.	Carbonates and acetates' bonds as a result of IR	120



## LIST OF PLATES

Plates		Page
1.	SEM observation of the gel heated at 250° C	79
2.	Microstructure of fractured surface of pure sample	116
3.	Microstructure of fractured surface of Pb-doped sample (Pb 0.4)	
4.	Microstructure of fractured surface of Pb-doped sample (Pb 0.5)	117



## LIST OF FIGURES

Figure		Page
1.	Abrupt change in resistivity and specific heat at T <sub>c</sub> occurs in superconducting materials	4
2.	Curvature of magnetic field lines round a superconducting sphere in constant applied field	4
3.	Variation of magnetization (M) as a function of magnetic field for (a) type-I superconductor and (b) type-II superconductor	6
4.	Type I and Type II superconductors as temperature-magnetic field dependent	7
5.	Vortex state and magnetic flux in type II Superconductors	7
6.	Crystal structures of $Bi_2Sr_2Ca_{n-1}Cu_{n+1}O$ 6+2n system with n= 0, 1, 2	9
7.	Ag-sheathed wire formed by mechanical rolling	12
8.	Superconductivity transition temperature records through the years	21
9.	Effect of zero field cooling (ZFC) and field cooling (FC) of a solid superconductor and perfect conductor	30
10.	Exponential decay of a constant applied magnetic field $B_z(x)$ inside the superconductor	33
11.	Relative scattering powers of atoms in the $Bi(Pb)_2Sr_2Ca_2Cu_{n+1}O_{6+2n}$ system with electron, x-ray or neutron irradiation	52
12.	Schematic four-Probe Resistance Device with temperature sensor used for measuring dc resistivity	53



13.	susceptometer	56
14.	Schematic diagram illustration of SEM JOEL system	62
15.	Wave number series for different molecular bonds	64
16.	The hydrolysis process of the starting solution	71
17.	The process of the gelation for hydrolyzed solution	72
18.	Preparation of gel powder	73
19.	The heat treatment and characterization of the gel samples	74
20.	TGA and DGA curves for heated -gel to 1000° C	76
21.	Infrared Spectrum of the Gel-heating at different temperature	77
22.	X-ray diffraction pattern of the gel-heating at different temperatures	78
23.	Presintering process flow-sheet of the dried gel powder	81
24.	The procedure of sintering in different time- temperature (t-T) profiles	82
25.	Resistance-temperature behaviour for samples sintered samples at different sintering duration at 845° C	83
26.	Differential resistance during the time sintering (Inset: Sintering time dependence of T(peak).)	84
27.	Variation of Variation of transition width ΔT with	86



28.	Residual factor $R_0$ (normalized) as a function of sintering time. (Inset: temperature dependence of normalized resistance)	86
29.	Resistance-temperature behaviour for the samples sintered at different temperatures for 150 h	87
30.	Variation of transition width ΔT with sintering temperature	89
31.	Variation of differential resistance peaks in samples sintered at different temperatures (inset: sintering time dependence of differential resistance peaks.)	89
32.	Residual factor $R_0$ (normalized) as a function of sintering temperature for 150 h. (Inset: temperature dependent of normalized resistance)	90
33.	Ac-susceptibility curves for sample sintered for different soak-times	92
34.	The bell-shaped curves of temperature variation of $\chi''$ for different sintering time	93
35. 36.	Coupling peaks, $T_p$ as a function of sintering time Differential of ac-susceptibility of $\chi'$ as a function of sintering time	93 94
37.	Ac-susceptibility measurements of the samples sintered at various temperature	95
38.	The bell-shaped curves of temperature variation of $\chi''$ for different sintering temperature	96
39.	Coupling peaks, T <sub>p</sub> as a function of sintering temperature	96
40.	Differential of ac-susceptibility of $\chi'$ as a function of sintering temperature	97
41.	Microstructure evolution in samples sintered for different soak-time at 845° C	99



42.	Microstructure evolution in samples sintered at various temperatures for 150 h	100
43.	XRD patterns of samples sintered for different soak- times at 845° C	102
44.	Change in volume fraction of the 2223 phase with soak-time	102
45.	X-ray patterns for sintered sample at various temperature	103
46.	Change in volume fraction of the 2223 phase with various temperature	104
47.	Flow-sheet for the synthesis of Pb-doped BSCCO Superconductor	109
48.	Variation of resistance in the pure and Pb-doped samples with temperatures	110
49.	Resistance measurements of the pure and Pb-doped samples	111
50.	Residual factor R <sub>0</sub> (normalized) of pure and Pb-doped samples. (inset: temperature dependent of normalized resistance)	112
51.	Ac-susceptibility measurements in pure and Pb-doped samples	113
52.	The bell shaped curves of temperature variation of $\chi''$ for pure and Pb -doped samples	114
53.	X-ray patterns analysis of the pure and Pb-doped samples	115
54.	Flow sheet for the preparation of gel from oxide and carbonate system	119
55.	Flow sheet for the preparation of gel from nitrate	120



56.	Infrared spectra of the converted acetates	128
57.	TGA curve for the heated gel obtained from carbonates and oxide system	130
58.	Infrared spectra of the wet gel at the room temperature	131
59.	Temperature dependent of resistance of the sintered sample produced from carbonates and oxide system	132
60.	Temperature dependent of resistance of the sintered sample produced from nitrate precursors	133
61.	Differential resistance curve for the sintered samples synthesised from carbonate and oxide system	134
62.	Differential resistance curve for the sintered samples synthesised from nitrate precursors	134
63.	X-ray patterns according to the sintered samples at 845 and 855 ° C, produced from carbonate and oxide system	136
64.	X-ray patterns according to the sintered samples at 845 and 855 ° C, produced from nitrate precursors	136
65.	SEM photographs according to the sintered samples at 845 and 855 ° C, produced from oxide-carbonate system	137
66.	SEM photographs according to the sintered samples at 845 and 855° C, produced from Nitrates	120



#### ABBREVIATIONS AND KEY WORDS

T<sub>c</sub> Critical temperature

B<sub>c</sub>, B<sub>c1</sub>, B<sub>c2</sub> Critical magnetic field

BSCCO Bi-Sr-Ca-Cu-O system

GL theory Ginzburg-Landau theory

BCS theory Bardeen, Cooper and Schrieffer theory

LBCO La-Ba-Cu-O system

YBCO Y-Ba-Cu-O system

Y123 Family member in YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub>.

Bi(2201) Family member in  $Bi_2Sr_2Ca_nCu_{n+1}O_{6+2n}$ , n =0

Bi(2212) Family member in  $Bi_2Sr_2Ca_nCu_{n+1}O_{6+2n}$ , n = 1

Bi(2223) Family member in  $Bi_2Sr_2Ca_nCu_{n+1}O_{6+2n}$ , n=2

TBCCO Tl-Ba-Ca-Cu-O system

T(2223) Family member in  $Tl_2Ba_2Ca_nCu_{n+1}O_{6+2n}$ , n=2

HBCCO Hg-Ba-Ca-Cu-O system

TGA Thermogravimetric Analysis

DTA Differential Thermal Analysis

Calcination Heating process where the solid state reaction

occur

Sintering Heating process yielding for more compacting

of the sample grains and improve its properties



Acetate Precursor Metal acetate dissolved in suitable solvent

Citrate Precursor Metal citrate dissolved in suitable solvent

Oxide technique Solid state reaction method

 $G_{\rm s}$  Gibbs free energy per unit volume

 $G_n$  Free -energy density of the normal state

H Applied magnetic field

M Magnetization

 $\mu_0$  Permeability of free space

 $\xi$  Coherence length

 $\lambda_L$  Penetration depth

ε<sub>F</sub> Fermi energy

 $\theta_D$  Debye temperature

 $v_F$  Fermi velocity

k Boltzman constant

h Blanck constant

 $N_{\rm s}$  Superelectron density

 $N(\varepsilon)$  Density of state

 $\Delta_0$  The zero-temperature energy gap

C<sub>s</sub> Specific heat

λ Electron- phonon coupling constant

 $D_{ph}(\omega)$  Phonon density of the state



 $\alpha^2(\omega)$  The electron -phonon coupling strength

 $\mu^*$  Coulombic repulsion

a, b, c Lattice parameters

J<sub>c</sub> Critical current density

MRI Magnetic Resonance Image

SQUID Superconducting Quantum Interference Device

 $MO(R)_m$  Metals alkoxides

 $M(OH)_m$  Metal hydroxides

χ Susceptibility

AC Alternating Current

V Induced voltage

ζ Filling factor

V Sample volume

α Calibration coefficient

 $f, \omega$  Frequency and angular frequency

XRD X-ray diffraction

**d**<sub>hkl.</sub> Reciprocal d vector

hkl Miller indices

SEM Scanning Electron Microscope

IR Infrared

 $\overline{\mathbf{v}}$  Wave number



Abstract of thesis presented to the Senate of University Putra Malaysia in fulfilment of the requirements for the degree of Master of Science

## SUPERCONDUCTING PROPERTIES OF SOL-GEL DERIVED Bi {Pb}-Sr-Ca-Cu-O SYSTEM

By

## SALEH AYED AHMAD AL-KHAWALDEH

May 1998

Chairman: Assoc. Prof. Abdul Halim Shaari, Ph.D.

Faculty: Science and Environmental Studies

The Bi<sub>2-x</sub>Pb<sub>x</sub>Sr<sub>2</sub>Ca<sub>2</sub>Cu<sub>3</sub>O<sub>y</sub> (2223) superconductor, has been prepared by sol-gel technique, using metal acetate precursors. Room temperature hydrolysis followed by polycondensation and heating yielded a transparent blue gel. The key to successful gel formation was due to a firm control of the pH of the solution. It was required to maintain the pH at 5.5 throughout the sol to gelation process. The decomposition of the polyhydroxyl metal complex to amorphous gel was found to be completed at temperatures in the range of 200 to 250°C producing submicron size particles. The amorphous gel transformed into crystalline powder at 600°C to 700°C.



The effect of heat treatment as a function of sintering time and temperature have been also studied on  $Bi_{1.5}Pb_{0.5}Sr_2Ca_2Cu_3O_{i0+x}$  system. The 2223 phase was observed in samples sintered at 845° C for 48 h.

The effect of the Pb doping at the bismuth lattice site have been studied in the  $Bi_{2-x}Pb_xSr_2Ca_2Cu_3O_y$  ( x=0.0, 0.4, 0.5 ) samples. Two samples doped with 20 and 25 mole % of Pb in the bismuth site yielded single phase sample with  $T_c$  (R=0) above 102 K. Undoped sample had mixed phases ( 2212 and 2223 ). The study shows that the doped-sample with 25 mole % of Pb is the most suitable ratio that gives rise to the best superconducting properties.

New procedure adopted in this work succeeded in getting good quality gel from carbonates, oxides and nitrates as the starting materials. The study shows that high-purity single-phase superconductor with  $T_c$  (R=0) at 104 K was obtained from carbonate and oxide as starting materials. This value is comparable to that obtained from corresponding metal acetates.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia bagi memenuhi keperluan untuk ijazah Master Sains

SIFAT-SIFAT KESUPERKONDUKSIAN DISEDIAKAN MELALUI TEKNIK SOL-GEL SISTEM Bi (Pb)-Sr-Ca-Cu-O

#### Oleh

#### SALEH AYED AHMAD AL-KHAWALDEH

#### Mei 1998

Pengerusi: Prof Madya Abdul Halim Shaari, Ph.D.

Fakulti: Sains dan Pengajian Alam Sekitar

Superkonduktor Bi<sub>2-x</sub>Pb<sub>x</sub>Sr<sub>2</sub>Ca<sub>2</sub>Cu<sub>3</sub>O<sub>y</sub> (2223) telah disediakan melalui teknik sol-gel dengan menggunakan petunjuk logam asetat. Hidrolisis pada suhu bilik diikuti dengan polikondensasi dan pemanasan menghasilkan gel biru lutsinar. Kunci kejayaan pembentukan gel adalah pengawalan terhadap pH larutan. Ia diperlukan untuk mengekalkan pH pada 5.5 sepanjang proses penukaran sol ke gel. Penguraian polihidroksil logam komplek kepada gel amorfus didapati sempurna pada suhu antara julat 200° C hingga 250° C dan menghasilkan partikel bersaiz submikron. Gel amorfus bertukar kepada serbuk kristal pada suhu diantara 600° C hingga 700° C.

