



UNIVERSITI PUTRA MALAYSIA

***OPTIMIZATION OF PROCESS PARAMETERS IN PREPARING  
NANOEMULSION CONTAINING KOJIC MONOOLEATE USING  
RESPONSE SURFACE METHODOLOGY AND EVALUATION OF  
TYROSINASE INHIBITION THROUGH IN VITRO AND IN SILICO  
METHODS***

MUHAMMAD AZIMUDDIN ROSELAN

FS 2021 20



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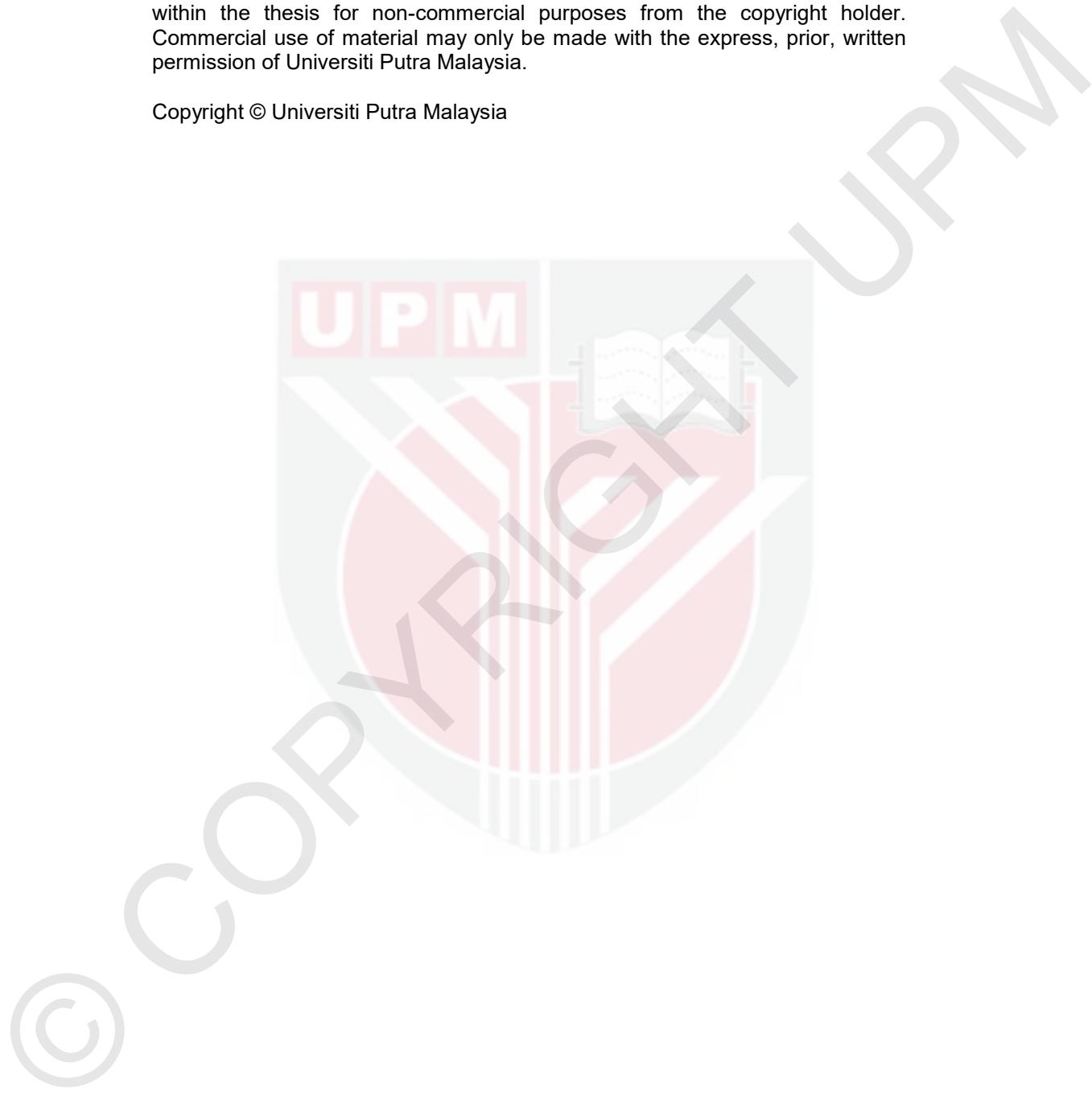
MUHAMMAD AZIMUDDIN ROSELAN

**Thesis Submitted to the School of Graduate Studies, Universiti  
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Master of Science**

**December 2020**

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in  
fulfilment of the requirement for the degree of Master of Science

**OPTIMIZATION OF PROCESS PARAMETERS IN PREPARING  
NANOEMULSION CONTAINING KOJIC MONOOLEATE USING RESPONSE  
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INHIBITION THROUGH *IN VITRO* AND *IN SILICO* METHODS**

By

**MUHAMMAD AZIMUDDIN ROSELAN**

**December 2020**

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**Faculty : Science**

Kojic monooleate (KMO), synthesized from the esterification of kojic acid (KA) and oleic acid, has shown a better depigmenting effect than KA. Previously, our research group had successfully loaded KMO into a nanoemulsion system. The composition of each ingredient was then optimized using mixture experimental design (MED) that resulted in a droplet size of 110.01 nm. Although the nanoemulsion was in nano-sized range, they found that the nanoemulsion only stable at the temperature of 4 and 25°C, and unstable at the temperature of 45°C. Thus, this study aims to further optimize the process parameters in producing the nanoemulsion containing KMO using response surface methodology (RSM). The effects of time of high shear (5-20 min), speed of high shear (4500-6500 rpm), and speed of stirrer (200-300 rpm) were investigated. The optimized process parameters in producing the nanoemulsion containing KMO with nano-sized range were 8.04 min (time of high shear), 4905.41 rpm (speed of high shear) and 271.82 rpm (speed of stirrer). The optimized nanoemulsion containing KMO showed good agreement between predicted droplet size (103.71 nm) and actual droplet size ( $103.97 \pm 0.13$  nm) with residual standard error (RSE) value less than 2.0%. An analysis of variance (ANOVA) showed that the fitness of the quadratic polynomial fit the experimental data with large *F*-values (148.79) and small *p*-values (*p*<0.0001) and an insignificant lack-of-fit.

The physicochemical characterization showed that the optimized nanoemulsified KMO was in the nanosize range ( $103.97 \pm 0.13$  nm) and had a zeta potential of  $-45.4 \pm 0.05$  mV and a polydispersity index (PDI) of  $0.312 \pm 0.14$ , indicating that the nanoemulsion produced was stable, and classified as monodispersed. The pH and conductivity of the optimized nanoemulsion ( $3.98 \pm 0.05$  mS/cm) signifying the oil-in-water (O/W) nanoemulsion characteristic. A morphology study revealed that the oil droplets in the optimized nanoemulsion containing KMO were spherical in shape, without any aggregation. In addition, the rheological behavior of the optimized nanoemulsion revealed that the nanoemulsion exhibited shear thinning and non-Newtonian behavior. The optimized nanoemulsion containing KMO remained stable under a centrifugation test and storage stability at different storage temperatures of 4, 25 and 45 °C over 90 days.

The MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide) assay performed on the normal 3T3 cell lines showed that the nanoemulsion containing KMO was not cytotoxic with  $IC_{50}$  (concentrations of sample required to inhibit the cell viability by 50%) more than 500 µg/mL ( $IC_{50} > 500$  µg/mL). The tyrosinase inhibitory assay revealed that the nanoemulsion containing KMO inhibited tyrosinase activity with the  $IC_{50}$  value of 68.20 µg/mL, compared to the positive control (KA) with the  $IC_{50}$  value of 124.28 µM. The permeation study revealed that  $45.94 \pm 0.03\%$  of KMO was released from the nanoemulsion and able to permeate the cellulose acetate membrane after 8 h of study time. The total KMO permeated across the membrane per unit area after 8 h of study time was  $14355.21$  µg.cm $^{-2}$ , with the flux (J) of  $1757.1$  µg.cm $^{-2}.h^{-1}$  and permeation coefficient ( $K_p$ ) value of  $0.09$  cm.h $^{-1}$ . The kinetic mechanism analysis revealed that the permeation data was most fitted with the zeroth-order model.

*In silico* molecular docking revealed that the binding energy for the KMO against mushroom tyrosinase (PDB ID: 2Y9X) is -5.70 kcal/mol, stronger than KA with the binding energy of -4.01 kcal/mol. The interaction of KMO on mushroom tyrosinase is via hydrophobic interaction involving His61, His85, Glu256, His259, Asn260, His263, Phe264, Met280, Gly281, Ser282, Val283, and Ala286 residues. These results predicted that KMO may inhibit mushroom tyrosinase. In conclusion, the nanoemulsion containing KMO with nano-sized range, good stability and physicochemical properties with potent tyrosinase inhibitor properties was successfully optimized in this study.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia  
sebagai memenuhi keperluan untuk ijazah Master Sains

**PENGOPTIMUMAN PARAMETER PROSES DALAM PENYEDIAAN  
NANOEMULSI MENGANDUNGI KOJIK MONOOLEATE MENGGUNAKAN  
Kaedah Gerak Balas Permukaan dan Penilaian Perencat  
Tirosina Menggunakan Kaedah *in Vitro* dan *in Silico***

Oleh

**MUHAMMAD AZIMUDDIN ROSELAN**

**Disember 2020**

**Pengerusi : Siti Efliza Ashari, PhD**  
**Fakulti : Sains**

Kojik monooleate (KMO), yang disintesis melalui proses esterifikasi asid kojik (KA) dan asid oleik telah menunjukkan kesan dipigmentasi yang lebih baik berbanding KA. Terdahulu, kumpulan penyelidikan kami telah berjaya menghasilkan nanoemulsi yang mengandungi KMO. Komposisi optimum bagi setiap bahan nanoemulsi kemudiannya telah disiasat menggunakan reka bentuk eksperimental campuran (MED). Saiz zarah nanoemulsi pada komposisi optimum adalah 110.01 nm. Walaupun nanoemulsi yang dihasilkan adalah bersaiz nano, namun begitu nanoemulsi tersebut hanya stabil pada suhu 4 dan 25°C, dan tidak stabil pada suhu 45°C. Justeru, kajian ini bertujuan untuk mengenal pasti proses parameter yang optimum dalam penghasilan nanoemulsi yang mengandungi KMO menggunakan kaedah gerak balas permukaan (RSM). Kesan masa penyebati berkuasa tinggi (5-20 min), kelajuan penyebati berkuasa tinggi (4500-6500 rpm) dan kelajuan penyebati (200-300 rpm) telah diselidik. Parameter proses optimum dalam penghasilan nanoemulsi yang mengandungi KMO bersaiz nano adalah 8.04 min (masa penyebati berkuasa tinggi), 4905.41 rpm (kelajuan penyebati berkuasa tinggi) dan 271.82 rpm (kelajuan penyebati). Saiz zarah anggaran (103.71 nm) adalah bersesuaian dengan saiz zarah sebenar ( $103.97 \pm 0.13$  nm) dengan nilai baki (RSE) kurang daripada 2.0%. Analisis varians (ANOVA) menunjukkan bahawa model kuadratik adalah sesuai dengan data eksperimental dengan nilai-*F* yang besar (148.79) dan nilai-*p* yang kecil ( $p<0.0001$ ).

Pencirian fizikokimia menunjukkan bahawa nanoemulsi yang mengandungi KMO adalah di dalam lingkungan saiz nano ( $103.97 \pm 0.13$  nm) dan mempunyai potensi zeta  $-45.4 \pm 0.05$  mV dan indeks polidispersiti (PDI)  $0.312 \pm 0.14$ . Ini menunjukkan bahawa nanoemulsi yang dihasilkan adalah stabil, dan ia diklasifikasikan sebagai monodispersi. pH nanoemulsi yang dihasilkan adalah  $5.75 \pm 0.02$ , menjadikan ia sesuai dengan pH kulit manusia dan nilai konduktivitinya yang tinggi ( $3.98 \pm 0.05$  mS/cm) menunjukkan ciri-ciri nanoemulsi minyak-dalam-air. Kajian morfologi menunjukkan bahawa titisan minyak di dalam nanoemulsi yang mengandungi KMO adalah berbentuk sfera, tanpa sebarang pengagregatan di dalam sistem. Ujian reologi telah mempamerkan bahawa nanoemulsi yang mengandungi KMO mempunyai sifat rincih penipisan dan non-Newtonian. Nanoemulsi yang mengandungi KMO kekal stabil apabila diuji dengan daya emparan dan semasa penyimpanan di suhu 4, 25 dan  $45^{\circ}\text{C}$  selama 90 hari.

Assay MTT (3-(4,5-dimetilthiazol-2-il)-2,5-difenil-tetrazolium bromida) yang dilakukan ke atas sel normal 3T3 menunjukkan bahawa nanoemulsi mengandungi KMO yang dihasilkan adalah tidak sitotoksik, dengan  $\text{IC}_{50}$  (kepekatan sampel yang diperlukan untuk membunuh sel sebanyak 50%) melebihi  $500 \mu\text{g/mL}$  ( $\text{IC}_{50} > 500 \mu\text{g/mL}$ ). Keputusan perencatan tirosina cendawan menunjukkan bahawa nanoemulsi yang mengandungi KMO merencat tirosina dengan nilai  $\text{IC}_{50}$   $68.20 \mu\text{g/mL}$ , berbanding kawalan positif (KA) dengan nilai  $\text{IC}_{50}$   $124.28 \mu\text{M}$ . Kajian resapan secara *in vitro* menunjukkan bahawa sebanyak  $45.94 \pm 0.03\%$  KMO berjaya dilepaskan dari sistem nanoemulsi dan meresap merentasi membran asetat selulosa selepas 8 jam. Jumlah KMO yang berjaya meresap merentasi membran per unit keluasan selepas 8 jam adalah  $14355.21 \mu\text{g.cm}^{-2}$ , dengan nilai flux ( $J$ )  $1757.1 \mu\text{g.cm}^{-2}.\text{j}^{-1}$  dan nilai pemalar resapan ( $K_p$ )  $0.09 \text{ cm.j}^{-1}$ . Analisis mekanisma kinetik menunjukkan bahawa data resapan adalah paling bersesuaian dengan model order-sifar.

Penyatuan molekul secara *in silico* menunjukkan bahawa tenaga pengikatan antara KMO dan enzim tirosina cendawan (ID PDB: 2Y9X) adalah  $-5.70 \text{ kcal/mol}$ , lebih kuat berbanding KA dengan tenaga pengikatan  $-4.01 \text{ kcal/mol}$ . Interaksi antara KMO dan enzim tirosina cendawan adalah melalui interaksi hidrofobik yang melibatkan residue His61, His85, Glu256, His259, Asn260, His263, Phe264, Met280, Gly281, Ser282, Val283, dan Ala286. Keputusan ini menunjukkan bahawa KMO mungkin merencat enzim tirosina cendawan. Secara kesimpulannya, nanoemulsi mengandungi KMO yang bersaiz nano, mempunyai tahap kestabilan dan ciri fizikokimia yang bagus dengan sifat perencatan tirosina yang cemerlang telah berjaya dioptimumkan.

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This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

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|                  |  |
|------------------|--|
| ANOVA            | Analysis of variance   |
| CCD              | Central composite design   |
| CCM              | Curcumin   |
| DLS              | Dynamic light scattering   |
| DMSO             | Dimethyl sulfoxide   |
| DTX              | Docetaxel  |
| IC <sub>50</sub> | Concentrations of sample that inhibit activity of enzyme or cells by 50% |
| KA               | Kojic acid   |
| KMO              | Kojic monooleate   |
| L-DOPA           | 3,4-dihydroxyphenyl L-alanine  |
| MED              | Mixture experimental design  |
| MTT              | 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide            |
| OFAT             | One-factor-at-a-time   |
| O/W              | Oil-in-water   |
| PBS              | Phosphate buffer solution  |
| PDI              | Polydispersity index   |
| R <sup>2</sup>   | Coefficient of determination   |
| RSE              | Residual standard error  |
| RSM              | Response surface methodology   |
| TEM              | Transmission electron microscopy   |
| UV               | Ultraviolet radiation  |
| W/O              | Water-in-oil   |

## CHAPTER 1

### INTRODUCTION

Skin hyperpigmentation disease occurs when there is an abnormal production of melanin synthesized in the melanocyte. This can be caused by an excess of melanin levels in the skin through several factors such as ultraviolet (UV) radiation, radicals and hormones (Kanlayavattanakul & Lourith, 2018). Melanin is a skin pigment, derived from tyrosine during the melanogenesis process, which determines the color of mammalian skin, eyes and hair (Huang *et al.*, 2008). The melanogenesis process is catalyzed by an enzyme known as tyrosinase. Due to its role in melanogenesis, tyrosinase has been considered as an important target for the treatment of hyperpigmentation diseases in the cosmeceuticals applications (Ha *et al.*, 2005).

Kojic acid (KA) is commonly used in the cosmetic industry as an antityrosinase, as it can inhibit tyrosinase through copper chelating. However, KA is less stable as it loses its effectiveness as a skin whitening agent when being exposed to sun or air. To overcome this limitation, derivatives of KA such as kojic acid monooleate (KMO) have been synthesized, through the esterification process of KA and oleic acid.

Previously, Syed Azhar *et al.* (2018) had successfully loaded KMO into a nanoemulsion system. The composition of each ingredients was then optimized using mixture experimental design (MED) that resulted in a droplet size of 110.01 nm. Although the nanoemulsion was in nano-sized range, they found that the nanoemulsion only stable at the temperature of 4 and 25°C, and unstable at the temperature of 45°C.

According to Yuan *et al.* (2008), different compositions and processing parameters in the production of the nanoemulsion will result in different physicochemical properties of the nanoemulsion. Thus, in order to obtain a KMO nanoemulsion with nano-sized range and good stability, the process parameters in the production of the nanoemulsion need to be further optimized. Thus, this issue is the main subject in this study to maintain the required droplet size and improved the stability of the nanoemulsion by optimizing the process parameters using a multivariate statistical approach, response surface methodology (RSM).

Furthermore, cytotoxicity and the efficacy of the nanoemulsion will also serve as the main issue to be investigated. The cytotoxic activity of the nanoemulsion was evaluated on the normal mouse fibroblast cell line (3T3). Meanwhile, the efficacy of the nanoemulsion was investigated through antityrosinase assay and permeation study.

Apart from that, the molecular interactions between KMO with its target protein, tyrosinase was predicted using molecular docking simulation. Molecular docking is a computational method that predicts the favored orientation of a ligand (KMO) to its target protein (tyrosinase) and calculates the binding energy, binding mode as well as binding residues of a protein-ligand complex. In other words, molecular docking was performed to understand the interaction between a ligand with its target protein.

The main objective of this research was to optimize nanoemulsion system containing KMO. In order to successfully achieve the main objective, the following specific objectives were carried out:

1. To optimize the process parameters in producing nanoemulsion containing KMO using response surface methodology (RSM).
2. To characterize the physicochemical properties of the optimized nanoemulsion containing KMO.
3. To evaluate the cytotoxicity and efficacy of the optimized nanoemulsion containing KMO; *in vitro* cytotoxicity against normal 3T3 cell line, antityrosinase activity, and permeation release.
4. To predict the binding mechanisms of KMO against mushroom tyrosinase (PDB ID: 2Y9X) through molecular docking simulation.

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Muhammad Azimuddin Roselan was born in Kota Bharu, Kelantan on 12<sup>th</sup> June 1996. He received his primary education at Sekolah Kebangsaan Islah, Kota Bharu, Kelantan. He continued his secondary study at Sekolah Menengah Kebangsaan Dato' Ahmad Maher, Kota Bharu, Kelantan. In 2014, he pursued his study at the Centre of Foundation Studies for Agricultural Sciences at Universiti Putra Malaysia (UPM), before continuing in Bachelor of Science (Honours) majoring in Chemistry for four years. Starting on September 2018, he enrolled his Master of Science program at the Department of Chemistry, Faculty of Science, UPM under the supervision of Dr. Siti Efliza Ashari.

## **LIST OF PUBLICATIONS**

### **Research Papers:**

Muhammad Azimuddin Roselan, Siti Efliza Ashari, Nur Hana Faujan, Siti Munirah Mohd Faudzi & Rosfarizan Mohamad (2020). An Improved Nanoemulsion Formulation Containing Kojic Monooleate: Optimization, Characterization and In Vitro Studies. *Molecules*, 25(11), 2616. (Published, Q2, IF 3.060)

Muhammad Azimuddin Roselan, Norzalina Zakaria, Sharifah Nurfadhlina Afifah Syed Azhar, Nur Hana Faujan & Siti Efliza Ashari (2020). A Preliminary Study: Molecular Docking, Optimization and Characterization of Kojic Monooleate Nanoemulsion for Cosmeceuticals Application. *Journal of Sustainability Science and Management*. (Accepted, Q4, IF 0.630)

Muhammad Azimuddin Roselan, Norzalina Zakaria, Nur Hana Faujan, Muhammad Alif Mohammad Latif, Siti Munirah Mohd Faudzi, Hazrina Ab Hadi & Siti Efliza Ashari (2020). In Vitro Cytotoxicity Assay, Mushroom Tyrosinase Inhibitory Activity and Franz Diffusion Cell Analysis of Nanodelivery System and In Silico Molecular Docking Study Against 2Y9X Target Enzyme with Kojic Monooleate. *Applied Biochemistry and Biotechnology*. (Submitted, Q3, IF 2.140)

### **Conferences and Exhibitions:**

Siti Efliza Ashari, Rosfarizan Mohamad, Nur Hana Faujan, Nur Farzana Izzati Mohd Jaslina, Muhammad Azimuddin Roselan, Sharifah Nurfadhlina Afifah Syed Azhar & Mohamad Ridzuan Yahya. Advanced Skin Technology: Upgrading into Whitemask Nanocosmeceutical Formula Incorporated with Palm-Based Kojic Monooleate in Thin Film System. International Conference and Exposition on Inventions by Institutions of Higher Learning 2019 (PECIPTA'19). 22-23 September 2019. (Silver medal)

Muhammad Azimuddin Roselan, Norzalina Zakaria, Sharifah Nurfadhlina Afifah Syed Azhar, Nur Hana Faujan & Siti Efliza Ashari. Three-Factors-Five-Levels Central Composite Design of Kojic Monooleate Nanoemulsion and Its Molecular Docking. 2<sup>nd</sup> Symposium on Multidisciplinary Science 2020 (V-SMS2020). 12 August 2020.



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