

UNIVERSITI PUTRA MALAYSIA

PHYSICAL, STRUCTURAL AND MECHANICAL PROPERTIES OF WOLLASTONITE-BASED FOAM GLASS CERAMICS USING EGGSHELL AND SODA-LIME-SILICA GLASS WASTE

DUR IFFA BINTI SAPARUDDIN

FS 2021 14



PHYSICAL, STRUCTURAL AND MECHANICAL PROPERTIES OF WOLLASTONITE-BASED FOAM GLASS CERAMICS USING EGGSHELL AND SODA-LIME-SILICA GLASS WASTE



By

DUR IFFA BINTI SAPARUDDIN

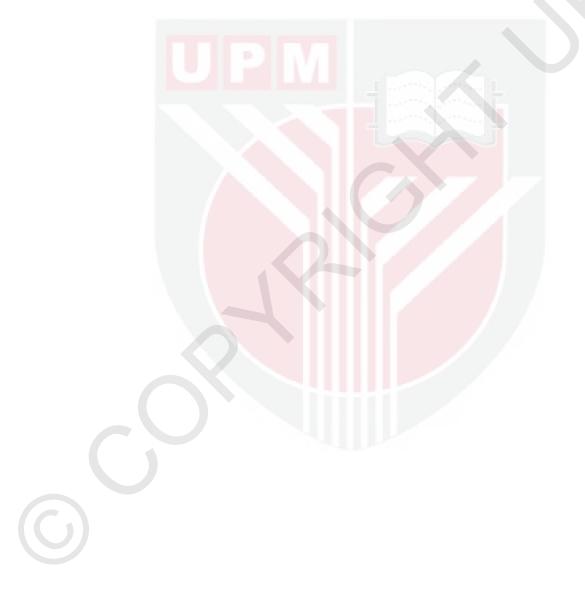
Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirements for the Degree of Master of Science

December 2020

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs, and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



DEDICATIONS

To my beloved parents, Who have made me what I am today

> To siblings and family For their love and care

To all of my lovely friends For cheering up my day

And most importantly to all of my lecturers For guiding me with a massive knowledge throughout this journey

Thank you all

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Master of Science

PHYSICAL, STRUCTURAL AND MECHANICAL PROPERTIES OF WOLLASTONITE-BASED FOAM GLASS CERAMICS USING EGGSHELL AND SODA-LIME-SILICA GLASS WASTE

By

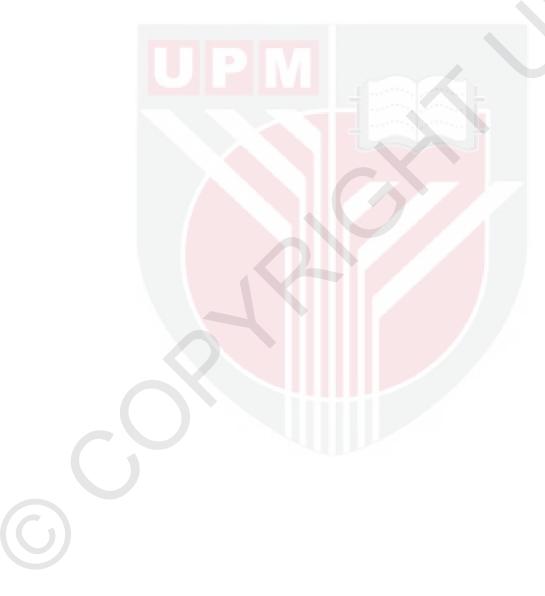
DUR IFFA BINTI SAPARUDDIN

Chairman Faculty : Science

Foam glass-ceramics represents a highly valuable solution for lightweight fill material, which has a low density, incombustible, and good in mechanical strength. In this work, foam glass-ceramics were prepared by the solid-state conventional method, using soda-lime-silica (SLS) glass waste as a matrix glass and eggshell (ES) as a foaming agent. The purpose of this study is to synthesize and study the changes that occur in physical, structural, and mechanical properties of foam glass-ceramic derived from waste materials by varying the ES content, sintering temperature, and sintering duration. Both of the raw materials were milled and sieved to the particle size of 45 µm. The powders then were homogenized with the empirical formula, $[ES]_x[SLS]_{100-x}$ where x = 1, 3, 6, and 9 wt.%. and uniaxially pressed with 5 MPa. The obtained compact powders were heat-treated at 700, 800, and 900 °C for 30, 60, and 120 min with the constant heating rate at 10 °C/min. The foam-glass ceramic samples were then analyzed based on physical properties; measured by density, linear expansion, and porosity. The structural properties of the foam glass-ceramic samples were determined by X-Ray Diffraction (XRD), Field Emission Scanning Electron Microscopy (FESEM), and Fourier Transform Infrared Spectroscopy (FTIR). Then, the mechanical properties of the sample were measured using Universal Testing Machine (UTM). The optimum parameter of the foam glassceramics was obtained at 800 °C for 60 min with the substitution of 3 to 6 wt.% ES content. In fact, the optimum temperature was supported by the thermogravimetric analyzer (TGA) which indicates that the mass loss of CaCO₃ in ES occurs at the temperature of 800 °C. The samples prepared this way to provide a minimum bulk density of 0.326 - 0.421 g/cm³ with the maximum porosity of 87.2 - 83.16 %. Furthermore, XRD analysis revealed that the formation and growth of cristobalite (SiO₂) and wollastonite (CaSiO₃) crystal phases after the heat-treatment process at 800 °C. Meanwhile, the FTIR analysis indicates the strong intensity of Ca-O bond at 620 to 650 cm⁻¹ of wavenumber associate with decreasing the intensity peak of



C–O after the heat-treatment process at 800 °C. FESEM analysis showed that the pores are larger and abundant distributed with the size of the diameter's pore is 861 – 1200 μ m in the samples with 3 to 6 wt.% of ES content. The results of compressive strength for the sample with 3 and 6 wt.% ES content increases from 0.04 and 0.42 MPa, respectively, depending on the crystalline phase content. Therefore, 6 wt.% ES is an appropriate ratio for fabricating foam glass-ceramics with a low density and high compressive strength. The preparation of foam glass-ceramics using solid wastes provide a promising way to prepare ceramic aggregate for the construction application as well as giving benefit to the economy and environment issue.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

SIFAT FIZIKAL, STRUKTURAL DAN MEKANIKAL WOLLASTONITE-BERASASKAN KACA SERAMIK BERLIANG MENGGUNAKAN SISA KULIT TELUR DAN KACA-SODA-KAPUR-SILIKA

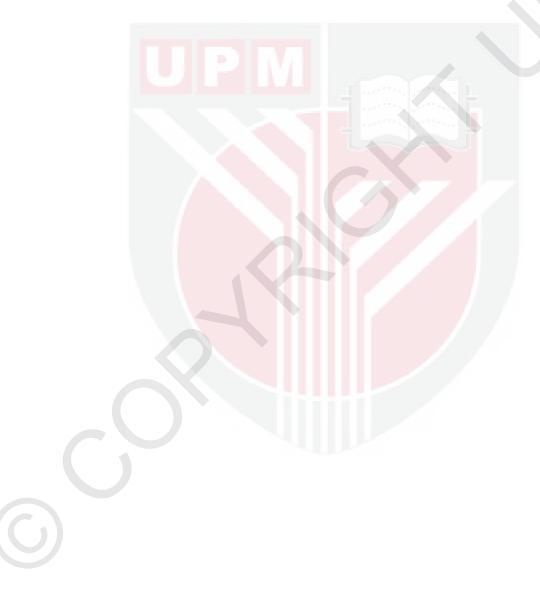
Oleh

DUR IFFA BINTI SAPARUDDIN

Disember 2020

Pengerusi : Mohd Hafiz Mohd Zaid, PhD Fakulti : Sains

Seramik kaca berongga merupakan penyelesaian penting untuk bahan pengisi ringan, yang mempunyai ketumpatan rendah, tidak mudah terbakar dan kekuatan mekanik. Dalam tesis ini, seramik kaca berongga disediakan dengan kaedah konvensional keadaan pepejal, menggunakan kaca soda-kapur-silika (SLS) sebagai kaca matriks dan kulit telur (ES) sebagai agen pembuih. Tujuan kajian ini adalah untuk mensintesis dan mengkaji sifat fizikal, struktur, dan mekanikal kaca-seramik dari bahan buangan dengan mengubah kandungan ES, suhu sintering, dan tempoh pensinteran. Kedua-dua serbuk bahan mentah itu digiling dan diayak dengan saiz zarah 45 µm. Serbuk kemudian dihomogenkan dengan formula empirik, $[ES]_x[SLS]_{100-x}$ di mana x = 1, 3, 6 dan 9 wt.% komposisi berat dan ditekan dengan 5 MPa. Serbuk padat yang diperolehi dirawat panas pada suhu 700, 800, dan 900 °C selama 30, 60, dan 120 minit dengan kadar pemanasan pada 10 °C/min. Sampel seramik kaca berongga dianalisis berdasarkan sifat fizikal; diukur dengan ketumpatan, pengembangan linier, dan keliangan. Sifat struktur sampel busakeramik kaca ditentukan oleh XRD, FESEM dan FTIR. Kemudian, sifat mekanik sampel diukur menggunakan UTM. Parameter optimum untuk menyediakan seramik kaca berongga diperoleh pada suhu 800 °C selama 60 minit dengan 3 hingga 6% berat ES. Pemerhatian suhu optimum disokong oleh TGA yang menunjukkan bahawa kehilangan jisim CaCO3 di ES berlaku pada suhu 800 °C. Sampel disediakan dengan cara ini untuk memberikan ketumpatan pukal minimum 0.326 – 0.421 g/cm³ dengan keliangan maksimum 87.2 - 83.16%. Selanjutnya, analisis XRD mendedahkan bahawa pembentukan dan pertumbuhan fasa kristal cristobalite dan wollastonite setelah proses perlakuan panas pada suhu 800 °C. Sementara itu, analisis FTIR menunjukkan intensiti kuat ikatan Ca-O pada 620 hingga 650 cm⁻¹ dari bilangan gelombang dikaitkan dengan penurunan puncak intensiti C-O setelah proses rawatan panas pada 800 °C. Analisis FESEM menunjukkan bahawa pori-pori lebih besar dan banyak didistribusikan dengan ukuran pori diameter adalah 861 - 1200 µm pada sampel 3 hingga 6 wt.% peratusan ES. Hasil kekuatan mampatan untuk sampel dengan kandungan ES 3 dan 6 wt.% meningkat masing-masing dari 0.04 dan 0.42 MPa, bergantung pada kandungan fasa kristal. Oleh itu, kandungan ES pada 6 wt.% adalah nisbah yang sesuai untuk pembuatan seramik kaca berongga dengan ketumpatan rendah dan kekuatan mampatan yang tinggi. Penyediaan seramik kaca busa menggunakan sisa pepejal memberikan cara yang menjanjikan untuk menyiapkan agregat seramik untuk aplikasi pembinaan serta memberi manfaat kepada masalah ekonomi dan persekitaran.



ACKNOWLEDGEMENTS

Alhamdulillah, praise to Allah, Lord of the world because give me an opportunity in terms of strength, patience, courage, and provide me with good health and intellectual ability to capable finish this research within the given time. The thesis title physical, structural, and mechanical properties of wollastonite-based foam glass ceramics using eggshell and soda-lime-silica glass waste able to complete successfully. I take this opportunity to express my gratitude to the people who have been instrumental in the successful completion of this thesis.

Firstly, I would like to express a lot of thank goodness to my supervisor, Dr. Mohd Hafiz Mohd Zaid, and my co-supervisor, Prof. Dr. Sidek Hj. Ab Aziz with Assoc. Prof. Dr. Khamirul Amin Matori, the important person during my study who gave a lot of encouragement, guidance, and support during the completion of this research. Special appreciation also goes to my beloved parents and family members for their unflagging love and support throughout my life. Higher appreciation also to my members from Ceramic Ultrasonic Laboratory (CURL) for their opinions and suggestions incomplete this dissertation. Apart from that, I also acknowledge the financial support from Jabatan Perkhidmatan Awam (JPA). This dissertation would have been impossible without them.

Last but not least, I offer my regards and blessings to all of those who supported me in any aspect during the completion of the project, and may Allah S.W.T. repay all your kindness.

This thesis was submitted to the Senate of the Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Mohd Hafiz bin Mohd Zaid, PhD

Senior Lecturer Faculty of Science Universiti Putra Malaysia (Chairman)

Sidek bin Hj. Ab Aziz, PhD Professor

Faculty of Science Universiti Putra Malaysia (Member)

Khamirul Amin bin Matori, PhD

Associate Professor Faculty of Science Universiti Putra Malaysia (Member)

ZALILAH MOHD SHARIFF, PhD Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date: 11 March 2021

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software

Signature:				
Name and Matrie	c No: Dur Iffa	binti Sapar	uddin,	GS52657

7

Date:

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) were adhered to.

Signature: Name of Chairman	
of Supervisory Committee:	Dr. Mohd Hafiz bin Mohd Zaid
	PM
Signature:	
Name of Member of Supervisory	
Committee:	Professor Dr. Sidek bin Hj. Ab Aziz
Signature:	
Name of Member	
of Supervisory	
Committee:	Associate Professor Dr. Khamirul Amin bin Matori

TABLE OF CONTENTS

			Page
ABSTR	RACT		i
ABSTR	AK		iii
ACKN	OWL	EDGEMENTS	V
APPRO)VAL		vi
DECLA	ARAT	ION	viii
LIST O	F TA	BLES	xiii
LIST O)F FI	GURES	xiv
LIST C	F AB	BREVIATIONS AND SYMBOLS	xviii
CHAP	ΓER		
1	TNIT	DODUCTION	1
1		RODUCTION Declargeour d of the study	1
	1.1	Background of the study	1
	1.2	Problem Statement	3
	1.3	Objectives	4
	1.4		4
	1.5	Outline of the thesis	4
2	LIT	ERATURE REVIEW	5
	2.1	Introduction	5
	2.2	Glass	5
		2.2.1 Glass formation	5
		2.2.2 Glass properties	6
		2.2.3 Type of glass waste	7
	2.3	Glass-ceramics	8
		2.3.1 Cristobalite	9
		2.3.2 Wollastonite	10
	2.4	Foam glass-ceramics	10
		2.4.1 Mechanism production of foam glass-ceram	
		2.4.2 Physical study	13
		2.4.3 Structural study	14
		2.4.4 Mechanical study	15
		2.4.5 Application of foam glass-ceramics	16
	2.5	Factor contribute to the properties of foam glass-cer	ramics 16
		2.5.1 Particles size	17
		2.5.2 Types of foaming agent	17
		2.5.3 Sintering temperature	18
		2.5.5 Sintering temperature	10

3

MET	THODOLOGY	
3.1	Introduction	
3.2	Materials preparation	
	3.2.1 SLS glass powder	
	3.2.2 ES powder	

22

Х

	3.3	Foam				24	
		3.3.1	Weighir	ng, mixing a	ind milling process		26
		3.3.2	Pelleting	g process			26
		3.3.3	Heat-tre	atment proc	ess		26
	3.4	Sampl	e characte	rization			27
		3.4.1		terials chara	acterization		27
			3.4.1.1	Waveleng	th Dispersive	X-ray	
				-	nce (WDXRF)		27
			3.4.1.2		avimetric analysis (TGA)	28
		3.4.2		-	characterization	1011)	29
		5.1.2			neasurement		29
			3.4.2.2		ity measurement		29
			3.4.2.3 Linear expansion				30
			3.4.2.4	-	Traction (XRD)		31
			3.4.2.5	Fourier	Transform	Infrared	51
			5.4.2.5		opy (FTIR)	lillaleu	32
			2126	-		Electron	32
			3.4.2.6		mission Scanning	Electron	32
			3.4.2.7		py (FESEM)		32 33
			5.4.2.7	Universal	Testing Machine (U) I IVI)	33
4	RESU	JLTS &	DISCUS	SION			34
	4.1	Introdu	action				34
	4.2	Precur	sor materi	als analysis			34
		4.2.1	WDXR	F analysis			34
		4.2.2	TGA an	alysis			35
	4.3	Physic	al analysi	s of foam gl	lass-ceramic		35
		4.3.1		nsity analys			36
			4.3.1.1	Different	content of ES and	d sintering	
				duration		-	36
			4.3.1.2	Different	temperature and	sintering	
				duration	1	U	39
		4.3.2	Bulk der		ty relationship		43
		4.3.3			expansion relations	hip	44
	4.4				glass-ceramics	1	45
		4.4.1	XRD an		6		45
				~	content of ES		45
			4.4.1.2		sintering temperatur	·e	53
			4.4.1.3		sintering duration		53
		4.4.2	FTIR an				54
		1. 1.2	4.4.2.1	2	content of ES		55
			4.4.2.2		sintering temperatur	e	64
			4.4.2.3		sintering duration	C	65
		4.4.3			sintering duration		67
		4.4.3 FESEM analysis 4.4.3.1 Different content of ES					67
		4.4.3.2 Different sintering temperature					68
			4.4.3.2		sintering duration	U	69
	4.5	Maaha			n glass-ceramics		70
	4.3			-	0		70
		4.5.1	Compre	ssive streng	ui allarysis		70

			Different content of ES and sintering duration Different sintering temperature and sintering duration	71 74
5	5.1	Introduction Conclusion	RECOMMENDATIONS	79 79 79 79 80
BIOD	ERENC DATA C			82 90 91



G

LIST OF TABLES

Table		Page
2.1	IR absorption band and band assignment of CaSiO ₃	14
2.2	Compressive strength value from previous research	15
2.3	Properties of commercial foam glass-ceramics from different manufacturers	16
3.1	Chemical composition of eggshell and soda-lime silica glass powder	25
4.1	Chemical composition of SLS glass and ES	34

LIST OF FIGURES

ŀ	Figure		Page
1	1.1	Three main types of foam glass-ceramics form	2
2	2.1	The volume-temperature of glass transform to liquid	6
2	2.2	Glass-ceramic process from a bulk glass via volume crystallization or from glass powder via sinter-crystallization	9
2	2.3	Schematic diagram of initial bubbles formation during sintering of a glass/carbon powder mixture	13
3	3.1	SLS glass cullet from waste bottles	21
3	3.2	Flow chart of SLS glass powder preparation	22
3	3.3	Eggshell waste	23
3	3.4	Flow chart of ES powder preparation	24
3	3.5	Flow chart of sample preparation	25
3	3.6	Diagram for the heat-treatment process	26
3	3.7	Schematic diagram of the XRF spectrometer	28
3	3.8	Diagram of the coupling thermobalance to the gas analyzer	29
3	3.9	The four-point of the diameter of the samples	31
3	3.10	(a): Diameter of pellet before heat treatment (b) Example of the pellet after heat treatment	31
3	3.11	Parallel cross section of samples	33
4	4.1	Weight loss curve of the ES as a function of temperature	35
4	4.2	Bulk density sintered at 700 °C for different content of ES and sintering duration	37
	4.3	Bulk density sintered at 800 $^{\circ}\mathrm{C}$ for different content of ES and sintering duration	38
4	4.4	Bulk density sintered at 900 $^{\circ}\mathrm{C}$ for different content of ES and sintering duration	39
4	4.5	Bulk density from 1 wt.% ES for different sintering temperature and sintering duration	40

4.6	Bulk density from 3 wt.% ES for different sintering temperature and sintering duration	41
4.7	Bulk density from 6 wt.% ES for different sintering temperature and sintering duration	42
4.8	Bulk density from 9 wt.% ES for different sintering temperature and sintering duration	43
4.9	Bulk density-porosity of foam glass-ceramics at different content of ES for 800 °C of temperature	44
4.10	Bulk density-linear expansion of foam glass-ceramics at different sintering temperatures for 6 wt.% ES	45
4.11	XRD pattern of foam glass-ceramic sample sintered at 700 °C for 30 min	46
4.12	XRD pattern of foam glass-ceramic sample sintered at 700 °C for 60 min	47
4.13	XRD pattern of foam glass-ceramic sample sintered at 700 °C for 120 min	47
4.14	XRD pattern of foam glass-ceramic sample sintered at 800 °C for 30 min	49
4.15	XRD pattern of foam glass-ceramic sample sintered at 800 °C for 60 min	49
4.16	XRD pattern of foam glass-ceramic sample sintered at 800 °C for 120 min	50
4.17	XRD pattern of foam glass-ceramic sample sintered at 900 °C for 30 min	51
4.18	XRD pattern of foam glass-ceramic sample sintered at 900 °C for 60 min	52
4.19	XRD pattern of foam glass-ceramic sample sintered at 900 °C for 120 min	52
4.20	XRD pattern at different sintering temperatures from 6 wt.% ES	53
4.21	XRD pattern at different sintering duration from 6 wt.% ES	54
4.22	IR absorption of foam glass-ceramic sample sintered at 700 $^{\circ}$ C for 30 min	55
4.23	IR absorption of foam glass-ceramic sample sintered at 700 $^{\circ}$ C for 60 min	56

4.24	IR absorption of foam glass-ceramic sample sintered at 700 °C for 120 min	57
4.25	IR absorption of foam glass-ceramic sample sintered at 800 °C for 30 min	58
4.26	IR absorption of foam glass-ceramic sample sintered at 800 °C for 60 min	59
4.27	IR absorption of foam glass-ceramic sample sintered at 800 °C for 120 min	60
4.28	IR absorption of foam glass-ceramic sample sintered at 900 °C for 30 min	61
4.29	IR absorption of foam glass-ceramic sample sintered at 900 °C for 60 min	62
4.30	IR absorption of foam glass-ceramic sample sintered at 900 °C for 120 min	63
4.31	IR absorption for 6 wt.% ES at different temperature	65
4.32	IR absorption for 6 wt.% ES at different sintering duration	66
4.33	Microstructure of foam glass-ceramics at 800 °C for 60 min, (a) 1 wt.% (b) 3 wt.% (c) 6 wt.% (d) 9 wt.% of ES	68
4.34	Microstructure of foam glass-ceramics from 6 wt.% ES at different sintering temperature for 60 min, (a) 700 (b) 800 (c) 900 °C	69
4.35	Microstructure of foam glass-ceramics from 6 wt.% ES at different sintering duration at 800 °C, (a) 30 (b) 60 (c) 120 min	70
4.36	Stress-strain curve analysis of foam glass-ceramics	71
4.37	Compressive strength at 700 °C of temperature for different content of ES and sintering duration	72
4.38	Compressive strength at 800 °C of temperature for different content of ES and sintering duration	73
4.39	Compressive strength at 900 °C of temperature for different content of ES and sintering duration	74
4.40	Compressive strength from 1 wt.% ES for different sintering temperature and sintering duration	75
4.41	Compressive strength from 3 wt.% ES for different sintering temperature and sintering duration	76

- 4.42 Compressive strength from 6 wt.% ES for different sintering temperature and sintering duration
- 4.43 Compressive strength from 9 wt.% ES for different sintering temperature and sintering duration

77

78



LIST OF ABBREVIATIONS AND SYMBOLS

ES	Eggshell
SLS	Soda-lime silicate
CaCO ₃	Calcium carbonate
SiO ₂	Silicon dioxide
wt.%	Weight percentage
XRF	X-ray fluorescence
TGA	Thermogravimetric analysis
XRD	X-ray diffraction
FTIR	Fourier transform infrared
FESEM	Field emission scanning electron microscopy
UTM	Universal testing machine
Na ₂ O	Sodium oxide
K ₂ O	Potassium oxide
CaO	Calcium oxide
MgO	Magnesium oxide
Na ₂ CO ₃	Sodium carbonate
MgCO ₃	Magnesium carbonate
Al ₂ O ₃	Aluminium oxide
CaSiO ₃	Calcium silicate (Wollastonite)
PVA	Polyvinyl alcohol
μm	micrometer
ρ _b	Bulk density
ρt	True density
3	Porosity
Di	Initial diameter
Df	Final diameter
n	integer
λ	wavelength
d	Distance between atomic layers in the crystal

- δ Stress/pressure applied on a materialε Strain
- JCPDS Joint Committee on Powder Diffraction Standards



CHAPTER 1

INTRODUCTION

1.1 Background of the study

Nowadays, municipal solid wastes are abundant in the landfill sites. In 2012, the World Bank estimates waste generation in world cities is around 1.3 billion tonnes/year and expected to reach approximately around 2.2 billion tonnes by 2025 (Hoornweg & Bhada-Tata, 2012). In Malaysia, 80% of solid waste categorized as recyclable waste is disposed of at landfills. However, these wastes are unable to recycle completely due to the limited source of separation (Moh & Manaf, 2014). Therefore, these kinds of recyclable waste materials such as paper, plastics, and glass having the greatest potential to be converted from waste to new valuable products. Recently, wastes have provided value-added opportunities and cost-effective for use as raw materials and give products of wealth technology. Several previous studies use waste as a matrix that has a similar chemical composition such as glass bottles, rice husk (Fernandes et al., 2019), fly ash (Guo et al., 2014), red mud (Liu et al., 2019) and mineral wool (Ji et al., 2019). The type of waste glass such as cathode ray tube panel, SLS glass, lab-ware glass, and printed circuit board.

There is a different type of glass waste, having its chemical composition which contributes to potential applications. As an example, soda-lime-silica (SLS) glass is a conventional glass type used daily, hence it has been a significant amount of urban wastes. The reuse of SLS glass is still limited due to the societal and technological factors, though its production is non-hazardous (Pontikes et al., 2007). SLS glass is the type of waste glass produced from crushed glass, bottles, containers, and lighting products manufactured by melting calcium carbonate (CaCO₃), silicon dioxide (SiO₂), sodium carbonate (Na₂CO₃), and other minor constituents (Silva et al., 2017). The 'lime' given because of the presence of additives of calcium oxide (CaO), and sodium oxide (Na₂O) to reduce the softening point from 1600 to 700 °C for the preparation of this type of glass. The softening point temperature of SLS glass is slightly low compared to other silicate glass which promotes less energy usage during the sintering process.

Foam glass-ceramic is a porous material, lightweight and characterized as a heterophase system (Sasmal et al., 2015). It is useful and can be a candidate to be used in thermal insulation materials such as ceramic materials in the furnace, insulating for roofs, ceilings, and walls (Hurley, 2003; Souza et al., 2017). Producers of foam glass-ceramics from Europe and North America currently use a high percentage of processed post-consumer glass in their products. There are three types of foam glass-ceramics as shown in Figure 1.1. From Figure 1.1, loose foam glass-ceramics then broken into loose foam glass-ceramics aggregate. Next, blocks and shapes of foam glass-ceramics are produced by a batch process. It also continuous production of blocks and shapes in molds that are then cut and shaped to a specific size. The last

type of foam glass-ceramic is in the form of pelletization which is a continuous production of spherical pellets of foam glass-ceramics that can apply in the manufacture of blocks, panels and slabs. The significance of this foam type material for the industrial where it can minimize heat loss or heat gain (Aditya et al., 2017).



Figure 1.1 : Three main types of foam glass-ceramics form (Source: Hurley, 2003)

In the glass-ceramic foam research, the porous structure formed inside softened glass makes it advantageous to construction applications. There is a high production cost in fabricating commercial foam glass-ceramics nowadays because of the superior physical and mechanical properties. Hence, there is limited usage of this material in building and industry because of the high-priced. Besides, current insulator (polymeric materials) has high flammability and poor sound absorption property compared to foam glass (Ji et al., 2019). Moreover, the production of commercial foam glass-ceramics from Pittsburgh Corning Company required a high sintering temperature at 1000 °C. The properties of final product foam glass-ceramics are highly depending on the size and distribution of the foaming agent, and sintering temperature, duration as well as heating rate. The type of porous could be closed porosity and open porosity depending on the several parameters that affect the viscosity of the system. The closed porosity is mainly applicable to heat-insulating materials. Meanwhile, open porosity is used for soundproof material. Open porosity usually formed in porous glass-ceramic using a decomposing type of foaming agent and in glass that able to crystallizes (König et al., 2020).

In this research work, low-cost foam glass-ceramics using waste materials (SLS glass as a glass matrix and eggshells (ES) as a foaming agent) as a precursor has been studied which having a low bulk density, homogenous close porosity, and desired mechanical strength. A series of $[ES]_x[SLS]_{100-x}$ where x = 1, 3, 6, and 9 wt.% are prepared from solid-state conventional method to fabricate foam glass-ceramics. The properties of starting materials such as SLS glass and ES have been characterized by X-ray fluorescence (XRF) and thermogravimetric analyzer (TGA), respectively. Meanwhile, the characteristic of foam glass-ceramics has been characterized to study the effect of ES content, sintering temperature, and sintering duration. The structural and mechanical properties of foam glass-ceramics have been studied by X-ray diffraction (XRD), Fourier transforms infrared spectroscopy (FTIR), Field emission scanning electron microscopy (FESEM), and compressive strength by Universal testing machine (UTM). In view of the above fact, the aim of this study is the fabrication and characterization of foam glass-ceramics derived from waste materials, SLS glass, and ES waste as potential materials used in the field of construction for concrete aggregate.

1.2 Problem Statement

Nowadays, there is an abundant of solid wastes accumulate in landfill sites. The food waste encounter 49.3% among other type of solid waste as the worldwide consume it in daily life (Moh & Manaf, 2014). Generally, the food waste would give a negative impact to the environment, consequently led to the solid waste pollution. In this research, the foam glass-ceramics is fabricated derived from waste materials which are SLS glass and ES wastes in order to solve environment issue. Furthermore, there are limited reports and systematic study regarding to the physical, structural and mechanical properties of foam glass-ceramics derived from waste materials. ES is used as a foaming agent because it contains 95% of CaCO₃ and these carbonate is decomposed to CO₂ gas at 800 °C of temperature (Fernandes et al., 2013).

In recent years, foam glass-ceramics have attracted great attention due to its properties and a wide range of applications. Foam glass-ceramics exhibit excellent properties such as long-lifetime, reusability, water- and chemical-resistant, and dimensionally stable. Foam glass-ceramics is a valuable product that provides the cellular structure inside softened glass particles. The porous structure contributes to the lightweight of the product and it is easily brittle. Therefore, the strength of the foam glass-ceramics needs to be considered to enhance the stability of the porous structure during the installation process. Nowadays, there is an insufficient of information related to the mechanical properties of foam glass-ceramics because the foam glass-ceramics have much attention on heat insulation, sound absorption, non-flammable, and moisture-resistant (Guo et al., 2020).

For that reason, a comprehensive study of the crystallization, properties, and effect of heat-treatment on foam glass-ceramics derived from SLS glass and ES as a foaming agent is carried out to improve the physical, structural and mechanical properties of foam glass-ceramics. The transformation of the glass to glass-ceramics

3

has become an integral part to improve the quality and properties of the final products. The results of this research are expected to find potential applications for the aggregate in construction building blocks.

1.3 Objectives

- 1. To synthesis and optimize the fabrication of foam glass-ceramic derived from SLS glass and ES waste using solid-state sintering method.
- 2. To study the effect of difference ES content on the physical, structural, and mechanical properties of foam glass-ceramics.
- 3. To determine the effect of heat-treatment temperature and duration on the physical, structural, and mechanical properties of foam glass-ceramics.

1.4 Significant of the study

Nowadays, there are a huge number of solid wastes are placed in the landfill. The amount of this solid waste continues to increase which is more than 30,000 tons in Peninsular Malaysia (Jereme et al., 2015). In fact, there is a lack of landfill sites to dispose the solid wastes. Therefore, it is an essential part to encourage the reuse the wastes such as SLS glass from glass bottles and the ES food waste to the wealth product. Hence, it can minimize the waste dump issue. In this research, the foam glass-ceramic is fabricated by mixing the SLS glass and ES as a foaming agent. It's useful as thermal insulation materials and acoustic appearance, especially to the construction industry (De Moraes et al., 2019). Besides, it can be applied to the various area of industry depend on the pore structure either the isolated pores or communicating pores. Therefore, the effect of the porosity of foam glass-ceramics is determined in this study by varying the different content of ES, different sintering temperature, and duration.

1.5 Outline of the thesis

In this study, Chapter 1 will give introduction of foam glass-ceramics from SLS glass and ES, the problem statements, the objectives, and also the importance of this study. The theory of glass, glass-ceramic and previous works including past and current has been carried out by other researchers are covered in Chapter 2. The apparatus, methodology and characterization of the precursor glass and foam glass-ceramics are explained in Chapter 3. The results concerning the effect of ES addition to the SLS glass, progression of heat treatment temperatures and duration towards the physical, structural and mechanical properties of foam glass-ceramics are analyzed and discussed in Chapter 4. Finally, the conclusion and suggestion for future works are state in Chapter 5.

REFERENCES

- Aditya, L., Mahliaa, T. M., Rismanchi, B., Nge, H. M., Hasan, M. H., Metselaar, H. S., Muraza, O., & Aditiya, H. B. (2017). A review on insulation materials for energy conservation in buildings. *Renewable and Sustainable Energy Reviews*, 73, 1352-1365.
- Agathopoulos, S., Tulyaganov, D. U., Ventura, J. M., Kannan, S., Saranti, A., Karakassides, M. A., & Ferreira, J. M. (2006). Structural analysis and devitrification of glasses based on the CaO-MgO-SiO₂ system with B₂O₃, Na₂O, CaF₂ and P₂O₅ additives. *Journal of Non-crystalline Solids*, 352(4), 322-328.
- Alim, D. M. (2009). Production and Characterization of Foam Glass from Container Glass Waste. (Unpublished doctoral dissertation). The American University, Cairo.
- Almasri, K. A., Aziz, S. A., Matori, K. A., & Zaid, M. H. M. (2017). Effect of sintering temperature on physical, structural and optical properties of wollastonite based glass-ceramic derived from waste soda lime silica glasses. *Results in Physics*, 7, 2242-2247.
- Arcaro, S., Maia, B. G., Souza, M. T., Cecsoneto, F. R., Granados, L., & Novaes de Oliveira, A. P. (2016). Thermal insulating foams produced from glass waste and banana leaves. *Materials Research*, 19(5), 1064-1069.
- Atilla, Y., Guden, M., & Tasdemirci, A. (2013). Foam glass processing using a polishing glass powder residue. *Ceramics International*, 39, 5869-5877.
- Bateni, N. H., Hamidon, M. N. & Matori, K. A. (2014). Effect of soda-lime-silica glass addition on the physical properties of ceramic obtained from white rice husk ash. *Journal of the Ceramic Society of Japan*, 122(2), 161-165.
- Beckhoff, B., Kanngießer, B., Langhoff, N., Wedell, R., & Wolff, H. (2006). Handbook of practical X-ray fluorescence analysis. Heidelberg: Springer.
- Bento, A. C., Kubaski, E. T., Sequinel, T., Pianaro, S. A., Varela, J. A., & Tebcherani, S. M. (2013). Glass foam of macroporosity using glass waste and sodium hydroxide as the foaming agent. *Ceramics International*, 39(3), 2423-2430.
- Bernardo, E., & Albertini, F. (2006). Glass foams from dismantled cathode ray tubes, *Ceramics International*, 32(6), 603-608.
- Bernardo, E., Pontikes, Y., & Angelopoulos, G. (2013). Optimisation of low temperature sinter crystallisation of waste derived glass. *Advances in Applied Ceramics*, 111, 472-479.
- Bernardo, E., Scarinci, G., Bertuzzi, P., Ercole, P., & Ramon, L. (2010). Recycling of waste glasses into partially crystallized glass foams. *Journal of Porous Materials*, 17, 359-365.

Brouwer, P. (2010). Theory of XRF. Almelo: PANalytical.

- Casasola, R., Rincon, J. M., & Romero, M. (2012). Glass-ceramic glazes for ceramic tiles: a review. *Journal of Materials Science*, 47(2), 553-582.
- Chang, S. Y., Wang, C. & Sun, C. C. (2019). Relationship between hydrate stability and accuracy of true density measured by helium pycnometry. *International Journal of Pharmaceutics*, 567, 1-8.
- Chatterjee, A., Khobragade, P., Mishra, S., & Naik, J. (2017). Advanced microemulsion synthesis and characterization of wollastonite (CaSiO₃)/polystyrene one-dimensional nanorods withcore-shell structures. *Particuology*, 30, 118-128.
- Chatterjee, N. D., Johannes, W., & Leistner, H. (1984). The system CaO-Al₂O₃-SiO₂-H₂O: new phase equilibria data, some calculated phase relations, and their petrological applications. *Contributions to Minerology and Petrology*, 88(1-2), 1-13.
- Chen, Z., Wang, H., Ji, R., Liu, L., Cheeseman, C., & Wang, X. (2019). Reuse of mineral wool waste and recycled glass in ceramic foams. *Ceramics International*, 45(12), 15057-15064.
- Da Costa, F. P., Morais, C. R., & Rodrigues, A. M. (2020). Sustainable glass-ceramic foams manufactured from waste glass bottles and bentonite. *Ceramics International*, 46(11), 17957-17961.
- De Moraes, E. G., Sangiacomo, L., Stochero, N. P., Arcaro, S., Barbosa, L. R., Lenzi, A., Siligardi, C., & de Oliveira, A. P. (2019). Innovative thermal and acoustic insulation foam by using recycled ceramic shell and expandable styrofoam (EPS) wastes. *Waste Management*, 89, 336-344.
- Effendy, N., Wahab, Z. A., Aziz, S. A., Matori, K. A., Zaid, M. H.M, & Rashid, S. A. (2017). Characterization and optical properties of erbium oxide doped ZnO-SLS glass for potential optical and optoelectronic materials. *Material Express*, 7(1), 59-65.
- Farnan, I., Grandinetti, P. J., Baltisberger, J. H., Stebbins, J. F., Werner, U., Eastman, M. A., & Pines, A. (1992). Quantification of the disorder in networkmodified silicate glasses. *Nature*, 358(6381), 31-35.
- Egerton, R.F. (2006). Physical principles of electron microscopy. Alberta: Springer.
- Fernandes, F. A., Arcaro, S., Junior, E. F., Serra, J. C., & Bergmann, C. P. (2019). Glass foams produced from soda-lime glass waste and rice husk ash applied as partial substitutes for concrete aggregates. *Process Safety and Environmental Protection*, 128, 77-84.
- Fernandes, H. R., Andreola, F., Barbieri, L., Lancellotti, I., Pascual, M. J., & Ferreira, J. M. (2013). The use of egg shells to produce cathode ray tube (CRT) glass foams. *Ceramics International*, 39, 9071-9078.
- Fernandes, H. R., Tulyaganov, D. U., & Ferreira, J. M. (2009a). Preparation and characterization of foams from sheet glass and fly ash using carbonates as foaming agents. *Ceramics International*, 35(1), 229-235.

- Fernandes, H. R., Tulyaganov, D. U., & Ferreira, J. M. (2009b). Production and characterisation of glass ceramic foams from recycled wasted materials. *Advances in Applied Ceramics*, 108(1), 9-13.
- Francis, A. A., & Rahman, M. K. (2014). Structure characterization and optimazation of process parameters on compressive properties of glass-based foam composites. *Environmental Progress and Sustainable Energy*, 33(3), 800-807.
- Gabbott, P. (2008). Principles and applications of thermal analysis. Oxford: Blacwell Publishing.
- Gibson, W. M., Chen, Z. W., & Li, D. (2008). High definition X-ray fluorescence: principles and techniques. *X-Ray Optics and Instrumentation*, 1-10.
- Gong, Y., Dongol, R., Yatongchai, C., Wren, A., Sundaram, S., & Mellott, N. (2016). Recycling of waste amber glass and porcine bone into fast sintered and high strength glass foams. *Journal of Cleaner Production*, 112, 4534-4539.
- Graça, M. P. F., da Silva, M. F., Sombra, A. S. B., & Valente, M. A. (2007). Electric and dielectric properties of a SiO₂-Na₂O-Nb₂O₅ glass subject to a controlled heat-treatment process. *Physica B: Condensed Matter*, 396(1), 62-69.
- Grundy, A. N., Jung, I. H., Pelton, A. D., & Decterov, S. A. (2008). A model to calculate the viscosity of silicate melts. *International Journal of Materials Research*, 99(11), 1195-1209.
- Grushko, I. S., & Maslakov, M. P. (2018). Crystalline phase formation in a foam glass matrix and its effect on material performance. *Glass and Ceramics*, 75(11-12), 465-470.
- Guo, H. W., Gong, Y. X., & Gao, S. Y. (2020). Preparation of high strength foam glass-ceramics from waste cathode ray tube. *Materials Letters*, 64(8), 997-999.
- Guo, Y., Zhang, Y., Huang, H., & Hu, P. (2014). Effect of heat treatment process on the preparation of foamed glass ceramic from red mud and fly ash. *Applied Mechanics and Materials*, 670-671, 201-204.
- Gupta, T. K., & Jean, J. H. (1994). Origin of cristobalite formation during sintering of a binary mixture of borosilicate glass and high silica glass. *Journal of Materials Research*, 9(4), 1000-1005.
- Hanpongpun, W., Jiemsirilers, S., & Thavorniti, P. (2007). Effects of clear and amber cullet on physical and mechanical properties of glass-ceramics containing zinc hydrometallurgy waste. *Journal of Solid Mechanics and Materials Engineering*, 1, 1305-1312.
- Hoornweg, D., & Bhada-Tata, P. (2012). A global review of solid waste management. Washington: World Bank.
- Hu, A. M., Li, M., Dali, D. M., & Liang, K. M. (2005). Crystallization and properties of a spodumene-willemite glass ceramic. *Thermochimica Acta*, 437(1), 110-113.

- Hurley, J. (2003). *A UK markey survey for foam glass*. Banbury: The Waste and Resources Action Programme.
- Ji, R., Zheng, Y., Zou, Z., Chen, Z., Wei, S., Jin, X., & Zhang, M. (2019). Utilization of mineral wool waste and waste glass for synthesis of foam glass at low temperature. *Construction and Building Materials*, 215, 623-632.
- Jusoh, W. N., Matori, K. A., Zaid, M. H. M., Zainuddin, N., Khiri, M. Z., Rahman, N. A., Jalil. R., & Kul, E. (2019). Effect of sintering temperature on physical and structural properties of Alumino-Silicate-Fluoride glass ceramics fabricated from clam shell and soda lime silicate glass. *Results in Physics*, 12, 1909-1914.
- Kang, H., & Schoenung, J. (2005). Electronic waste recycling: A review of U.S. infrastructure and technology option. *Resources Conservation Recycling*, 45, 368-400.
- Kang, S. J. L. (2004). Sintering: densification, grain growth and microstructure. Butterworth-Heinemann.
- Karamanov, A., & Pelino, M. (2008). Induced crystallization porosity and properties of sintereds diopside and wollastonite glass-ceramics. *Journal of the European Ceramic Society*, 28(3), 555-562.
- Khan, R. I., & Ashraf, W. (2019). Effects of ground wollastonite on cement hydration kinetics and strong development. *Construction and Building Materials*, 218, 150-161.
- Kohara, S., Suzuya, K., Takeuchi, K., Loong, C. K., Grimsditch, M., Weber, J. K. R., Tangeman J. A., & Key, T. S. (2004). Glass formation at the limit of insufficient network formers. *Science*, 303(5664), 1649-1652.
- König, J., Lopez-Gil, A., Cimavilla-Roman, P., Rodriguez-Perez, M. A., Petersen, R. R., Østergaard, M. B., Iversen, N., Yue, Y., & Spreitzer, M. (2020). Synthesis and properties of open- and closed-porous foamed glass with a low density. *Construction and Building Materials*, 247, 118574-118583.
- König, J., Petersen, R. R., & Yue, Y. (2014). Influence of the glass-calcium carbonate mixture's characteristics on the foaming process and the properties of the foam glass. *Journal of the European Ceramic Society*, 34(6), 1591-1598.
- König, J., Petersen, R. R., & Yue, Y. (2015). Fabrication of highly insulating foam glass made from CRT panel glass. *Ceramics International*, 41(8), 9793-9800.
- König,, J., Petersen, R. R., Yue, Y., & Surorov, D. (2017). Gas-releasing reactios in foam-glass formation using carbon and M_xnO_y as the foaming agents. *Ceramics International*, 43(5), 4638-4646.
- Li, Y., Cheng, X. D., Gong, L. L., Feng, J. J., Cao, W., Zhang, R. F., & Zhang, H. P. (2015). Fabrication and characterization of anorthite foam ceramics having low thermal conductivity. *Journal of European Ceramic Society*, 35(1), 267-275.

- Liu, T., Lin, C., Liu, P., Liu, J., Li, C., Han, L., Zhou, X., Yang, Q., & Lu, A. (2019). Preparation and characterization of partially vitrified ceramic material. *Journal of Non-Crystalline Solids*, 505, 92-101.
- Liu, T., Zhang, J., Wu, J., Liu, J., Li, C., Ning, T., Luo, Z., Zhou, X., Tang, Q., & Lu, A. (2019). The utilization of electrical insulators waste and red mud for fabrication of partially vitrified ceramic materials with high porosity and high strength. *Journal of Cleaner Production*, 223, 790-800.
- Lu, J., Lu, Z., Peng, C., Li, X., & Jiang, H. (2014). Influence of particle size on sinterability, crytallisation kinetics and flexural strength of wollastonite glass ceramics from waste glass and fly ash. *Materials Chemistry and Physics*, 148(1-2), 449-456.
- Lunip, A., Kanagesan, S., Aziz, S., & Rao, B. (2016). Physical Properties of foam glass ceramics prepared by cathode ray tube panel glass and clam shell. *International Journal of Science, Engineering and Technology Research*, 5(7), 2344-2352.
- Manevich, V. E. & Subbotin, K. Y. (2008). Foam glass and problems of energy conservation. *Glass and Ceramics*, 65(3-4), 105-108.
- Ma, Q., Wang, Q., Luo, L., & Fan, C. (Eds.). (2018). Proceedings from ICMSE '18: 6th Annual International Conference on Material Science and Engineering. Suzhou, Jiangsu: IOP Publishing Ltd.
- Moh, Y. C. & Manaf, L. A. (2014). Overview of household solid waste recycling policy status and challenges in Malaysia. *Resources, Conservation and Recycling*, 82, 50-61.
- Mohanalakshmi, V., Adhithiyan, V. M., Kumar, R. J., & Preethi, L. A. (2016). Geotechnical properties of soil stabilized with wollastonite. *International Journal of Engineering Research*, 5(3), 703-706.
- Monich, P. R., Romero, A. R., Hollen, D., & Bernardo, E. (2018). Porous glassceramics from alkali activation and sinter-crystallization of mixtures of waste glass and residues from plasma processing of municipal of solid waste. *Journal of Cleaner Production*, 188, 871-878.
- Mugoni, C., Montorsi, M., Siligardi, C., Andreola, F., Lancellotti, I., Bernardo, E., & Barbieri, L. (2015). Design of glass foams with low environmental impact. *Ceramics International*, 41, 3400-3408.
- Müller, R., Zanotto, E. D., & Fokin, V. M. (2000). Surface crystallization of silicate glasses: nucleation sites and kinetics. *Journal of Non-Crystalline Solids*, 274(1), 208-231.
- Østergaard, M. B., Petersen, R. R., König, J., Johra, H., & Yue, Y. (2017). Influence of foaming agents on solid thermal conductivity of foam glasses prepared from CRT panel glass. *Journal of Non-Crystalline Solids*, 465, 59-64.

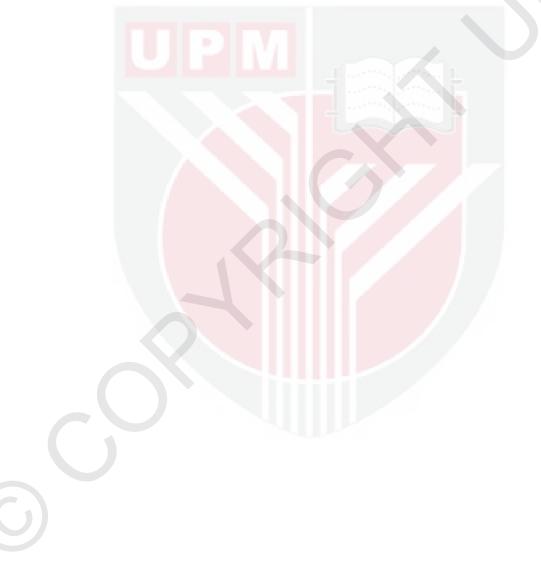
- Owoeye, S. S., Matthew, G. O., Ovienmhanda, F. O., & Tunmilayo, S. O. (2020). Preparation and characterization of foam glass from waste container glasses and water glass for application in thermal insulations. *Ceramics International*, 46, 11770-11775.
- Paligari, L., Dapiaggi, M., Pavese, A., & Francescon, F. (2013). A kinetic study of the quartz-cristobalite phase transition. *Journal of the European Ceramic Society*, 33, 3403-3410.
- Pedone, A. (2009). Properties calculations of silica-based glasses by atomistic simulations techniques: a review. *The Journal of Physical Chemistry C*, 113(49), 20773-20784.
- Petersen, R. R., König, J., & Yue, Y. (2017). The viscosity window of the silicate glass foam production. *Journal of Non-Crystalline Solids*, 456, 49-54.
- Petersen, R. R., König, J., & Yue, Y. (2015). The mechanism of foaming and thermal conductivity of glasses foamed with MnO₂. *Journal of Non-Crystalline Solids*, 425, 74-82.
- Pontikes, Y., Esposito, L., Tucci, A., & Angelopoulos, G. N. (2007). Thermal behaviour of clays for traditional ceramics with soda-lime-silica waste glass admixture. *Journal of the European Ceramic Society*, 27(2-3), 1657-1663.
- Prado, M. O., Fredericci, C., & Zanotto, E. D. (2003). Isothermal sintering with concurrent crystallization of polydispersed soda-lime-silica glass beads. *Journal of Non-Crystalline Solids*, 331(1-3), 145-156.
- Rawlings, R. D., Wu, J. P., & Boccaccini, A. R. (2006). Glass-ceramics: Their production from wastes-A Review. *Journal Material Sciences*, 41(3), 733-761.
- Reindl, J. (2003). Reuse or recycling of glass into other than glass bottles. Madison: US Environmental Protection Agency.
- Richet, P., Robie, R. A., & Hemingway, B. S. (1991). Thermodynamic properties of wollastonite, pseudowollastonite and CaSiO₃ glass and liquid. *European Journal of Minerology*, 3(3), 475-484.
- Saparuddin, D. I., Zaid, M. H. M., Sidek, S. A., & Matori, K. A. (2020). Reused of eggshell waste and recycled glass in the fabrication porous glass-ceramics. *Applied Sciences*, 10, 5404.
- Sasmal, N., Garai, M., & Kamarkar, B. (2015). Preparation and characterization of novel foamed porous glass-ceramics. *Material Characterization*, 103, 90-100.
- Scarinci, G., Brusatin, G., & Bernardo, E. (2005). *Cellular Ceramics: Structure, manufacturing, properties and applications.* Weinheim: Wiley-VCH.
- Shelby, J. E. (2005). *Introduction to glass science and technology*. New York: Royal Society of Chemistry.

- Shuai, C., Sun, H., Gao, C., Feng, P., Guo, W., Yang, W., Xu, H, Li, Q., Yang, Y., & Peng, S. (2018). Fabricating the nanostructured surfaces of CaSiO₃ scaffolds. *Applied Surface Science*, 455, 1150-1160.
- Silva, R., Brito, J., Lye, C., & Dhir, R. (2017). The role of glass waste in the production of ceramic-based products and other applications: A review. *Journal of Cleaner Production*, 167, 346-364.
- Singh, A. K. (2013). A comparative study on optical properties of Se-Zn-In and Se-Zn-Te-In chalcogenide glasses. *Optik-International Journal for Light and Electron Optics*, 124(15), 2187-2190.
- Sinton, C. W., & LaCourse, W. C. (2001). Experimental survey of the chemical durability of commercial soda-lime-silicate glasses. *Materials Research Bulletin*, 36(13), 2471-2479.
- Siriphannon, P., Kameshima, Y., Yasumori, A., Okada, K., & Hayashi, S. (2002). Formation of hydroxyapatite on CaSiO₃ powders in simulated body fluid. *Journal of the European Ceramic Society*, 22, 511-520.
- Souza, M. T., Maia, B. G., Texeira, L. B., Oliveira, K. G., Teixeira, A. H., & Novas de Oliveira, A. P. (2017). Glass foams produced from glass bottles and eggshell wastes. *Process Safety and Environmental Protection*, 111, 60-64.
- Spiridonov, Y. A., & Orlova, L. A. (2003). Problems of foam glass production. *Glass* and *Ceramics*, 60(9-10), 313-314.
- Steiner, A.C. (2006). Foam glass production from vitrified municipal fly ashes. (Unpublished doctoral dissertation). Eindhoven University of Technology, Netherlands.
- Taha, M. A., Youness, R. A., & Zawrah, M. F. (2020). Phase composition, sinterability and bioactivity of amorphous nano-CaO-SiO₂-CuO powder synthesized by sol-gel technique. *Ceramics International*, 46(15), 24462-24471.
- Taurino, R., Lancellotti, I., Barbieri, L., & Leonelli, C. (2014). Glass-ceramic foams from borosilicate glass waste. *International Journal of Applied Glass Science*, 5(2), 136-145.
- Titus, D., Samuel, E. J. J., Roopan, S. M. (2019). *Green synthesis, characterization* and applications of nanoparticles. Amsterdam: Elsevier.
- Tulyaganov, D., Fernandes, H., Agathopoulos, S., & Ferreira, J. (2006). Preparation and characterization of high compressive strength foams from sheet glass. *Porous Materials*, 13(2), 133-139.
- Turnbull, D., & Cohen, M. H. (1961). Free-volume model of the amorphous phase: glass transition. *The Journal of Chemical Physics*, 34(1), 120-125.
- Vancea, C., & Lazău, I. (2013). Glass foam from window panes and bottle glass wastes. *Journal of Chemistry*, 12(7), 804-811.

- Vitale-Brovarone, C., Novajra, G., Milanese, D., Lousteau, J., & Knowles, J. C. (2011). Novel phosphate glasses with different amounts of TiO₂ for biomedical applications: dissolution tests and proof of concept of fibre drawing. *Materials Science and Engineering: C*, 31(2), 434-442.
- Wang, H., Chen, Z., Ji, R., Liu, L., & Wang, X. (2018). Integrated utilization of high alumina fly ash for synthesis of foam glass ceramic. *Ceramics International*, 44(12), 13681-13688.
- Wang, H., Feng, K., & Sun, Q. (2018). Effect of calcium carbonate on the preparation of glass ceramic foams from water-quenched titanium-bearing blast furnace slag and waste glass. *Advances in Applied Ceramics*, 1-8.
- Wei, H., Thompson, R. B., Park, C. B., & Chen, P. (2010). Surface tension of high density polyethylene (HDPE) in supercritical nitrogen: Effect of polymer crystallization. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 354, 347-352.
- Xi, C., Zheng, F., Xu, J., Yang, W., Peng, Y., Li, Y., Li, P., Zhen, Q., Bashir, S., & Liu, J. L. (2018). Preparation of glass-ceramic foams using extracted titanium tailing and glass waste as raw materials. *Construction and Building Materials*, 190, 896-909.
- Xu, B., Liang, K. M., Cao, J. W., & Li, Y. H. (2010). Preparation of foam glass ceramics from phosphorus slag. Advanced Materials Research, 105-106, 600-603.
- Yoon, S. D., Lee, J. U., Lee, J. H., Yun, Y. H., & Yoon, W. J. (2013). Characterization of wollastonite glass-ceramics made from waste glass and coal fly ash. *Journal of Materials Science and Technology*, 29(2), 149-153.
- Zaid, M. H. M., Matori, K. A., Aziz, S. A., Zakaria, A., & Ghazali, M. M. (2012). Effect of ZnO on the Physical Properties and Optical Band Gap of Soda Lime Silicate Glass. *International Journal of Molecular Sciences*, 13, 7550-7558.
- Zanotto, E. D. & Prado, M. O. (2001). Isothermal sintering with concurrent crystallisation of monodispersed and polydispersed glass particles. *Physics and Chemistry of Glasses*, 42(3), 191-198.
- Zhang, Q., He, F., Shu, H., Qiao, Y., Mei, S., Jin, M., & Xie, J. (2016). Preparation of high strength glass ceramic foams from waste cathode ray tube and germanium tailings. *Construction and Building Materials*, 111, 105-110.
- Zhu, M., Ji, R., Li, Z., Wang, H., Liu, L., & Zhang, Z. (2016). Preparation of glass ceramic foams for thermal insulation applications from coal fly ash and waste glass. *Construction and Building Materials*, 112, 398-405.
- Zhu, W., Chen, J., Hao, C., & Zhang, J. (2014). Microstructure and strength of Al₂O₃/Al₂O₃ joints bonded with ZnO-Al₂O₃-B₂O₃-SiO₂ glass-ceramic. *Journal of Materials Science & Technology*, 30(9), 944-948.

BIODATA OF STUDENT

Dur Iffa binti Saparuddin was born on 6th August 1993 in Kuantan, Pahang, Malaysia. She received her primary education at Sekolah Kebangsaan Indera Mahkota Utama and continued her secondary education at Sekolah Menengah Kebangsaan Sultanah Hajjah Kalsom, Kuantan, Pahang. After receiving SPM results, she furthered her study at Universiti Teknologi Pahang (UITM) Pahang for the Diploma in Science before continuing her undergraduate studies in Bachelor of Science (Hons.) Physics for 3 years, at the same university and graduated on April 2018. In the present, she further her study in Master Science in the field of Materials Science at Universiti Putra Malaysia under supervision by Dr. Mohd Hafiz Mohd Zaid.



LIST OF PUBLICATIONS

Journal

- Saparuddin, D. I., Hisham, N. A. N., Sidek, H. A. A., Matori, K. A., Honda, S., Iwamoto, Y., Zaid, M. H. M. (2020). Effect of sintering temperature on the crystal growth, microstructure and mechanical strength of foam glassceramic from waste materials. Journal of Materials Research and Technology, 9(3), 5640–5647.
- Saparuddin, D.I., Zaid, M. H. M., Sidek, H. A. A., Matori, K. A. (2020). Reuse of eggshell waste and recycled glass in the fabrication porous glass-ceramics, Applied Sciences, 10, 5404.
- Hisham, N. A. N., Zaid, M. H. M., Saparuddin, D. I., Sidek, H. A. A., Muhammad, F. D., Honda, S., Iwamoto, Y. (2020). Crystal growth and mechanical properties of porous glass-ceramics derived from waste soda-lime-silica glass and clam shells. Journal of Materials Research and Technology, 9(4), 9295–9298.
- Jaafar, S. H., Zaid, M. H. M., Matori, K. A., Sabri, M. G. M., Shofri, M. F. S. M. Hisham, N. A. N., Saparudin, D. I. Effect of sintering temperatures and foaming agent content to the physical and structural properties of wollastonite based foam glass-ceramics, Science of Sintering, 52(3), 1-13.

Seminar and Conference

Hisham, N. A. N., Saparuddin, D. I., Zaid, M. H. M., Aziz, S. A., Muhammad, F. D., Shofri, M. F. S. M., Jaafar, S. H. (2019). "Physical and structural properties of foam glass-ceramics from waste materials", as poster presenter at Material Technology Challenge (MTC 2019), 27th March 2019, UPM Serdang, Selangor.



UNIVERSITI PUTRA MALAYSIA

STATUS CONFIRMATION FOR THESIS / PROJECT REPORT AND COPYRIGHT

ACADEMIC SESSION : Second Semester 2020/2021

TITLE OF THESIS / PROJECT REPORT :

PHYSICAL, STRUCTURAL AND MECHANICAL PROPERTIES OF WOLLASTONITE-BASED FOAM GLASS CERAMICS USING EGGSHELL AND SODA-LIME-SILICA GLASS WASTE

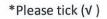
NAME OF STUDENT: DUR IFFA BINTI SAPARUDDIN

I acknowledge that the copyright and other intellectual property in the thesis/project report belonged to Universiti Putra Malaysia and I agree to allow this thesis/project report to be placed at the library under the following terms:

1. This thesis/project report is the property of Universiti Putra Malaysia.

- 2. The library of Universiti Putra Malaysia has the right to make copies for educational purposes only.
- 3. The library of Universiti Putra Malaysia is allowed to make copies of this thesis for academic exchange.

I declare that this thesis is classified as :





CONFIDENTIAL



RESTRICTED



OPEN ACCESS

Act 1972). (Contains restricted information as specified by the

(Contain confidential information under Official Secret

I agree that my thesis/project report to be published as hard copy or online open access.

organization/institution where research was done).

This thesis is submitted for :

PATENT

(Signature of Student) New IC No/ Passport No.:

Embargo from		until	
	(date)		(date)

Approved by:

(Signature of Chairman of Supervisory Committee) Name:

Date :

Date :

[Note : If the thesis is CONFIDENTIAL or RESTRICTED, please attach with the letter from the organization/institution with period and reasons for confidentially or restricted.]