



**UNIVERSITI PUTRA MALAYSIA**

**PHOSPHORUS FORMS AND ADSORPTION CAPACITY IN  
JAMBU SOIL SERIES**

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FACULTY OF AGRICULTURE  
UNIVERSITY PUTRA MALAYSIA

SERDANG, SELANGOR

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PHOSPHORUS FORMS AND ADSORPTION CAPACITY IN JAMBU SOIL

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By

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A project report submitted to Faculty of Agriculture,  
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In Fullfillment of PRT 4999 (Final Year Project)  
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FACULTY OF AGRICULTURE  
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## CERTIFICATION

This project report entitled Phosphorus Forms and Adsorption Capacity in Jambu Soil Series is prepared by SYAFFAF A'LIAH BT ZAINAL and submitted to the Faculty of Agriculture in partial fulfillment of the requirement of PRT 4999 (Final Year Project) for the award of degree of Bachelor of Agriculture Science.

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## ABSTRACT

Phosphorus is often the most limiting plant nutrient, especially in tropical soils. Total P content in Malaysian sandy soil is hundred times lower than the other mineral soils. A study was conducted to study the forms of phosphorus and phosphorus adsorption capacity in the different horizons of Jambu soil series. Sandy soil of Jambu soil series were collected from Merang, Terengganu by Roslan et al.(2010) at different depths which were 0-15cm, 15-30cm, 30-45cm and 45-60cm. The soils were determined for the total P, available P, organic and inorganic P, and P adsorption capacity. The amount of P varies within the soil depths. Total phosphorus content in 0-15cm, 15-30cm, 30-45cm and 45-60cm depth were 6.42 mg/kg, 6.51mg/kg, 5.5 mg/kg and 5.08 mg/kg, respectively. The available P content increased with increasing soil depths which were 1.99 mg/kg, 2.35 mg/kg, 2.51mg/kg and 3.31 mg/kg. The amount of organic P and inorganic P depends on the amount of total P in the soil. The amount of inorganic P was higher than the amount of organic P. Based on adsorption coefficient ( $K_f$  value) obtained from Freundlich adsorption isotherm, the highest phosphorus adsorption was at depth 0-15cm, followed by 15-30cm, 30-45cm and 45-60cm.

## ABSTRAK

*Fosforus merupakan salah satu nutrient yang terhad kepada pokok terutamanya di negara tropika. Tanah pasir di Malaysia selalunya mengandungi fosforus seratus kali lebih rendah berbanding tanah mineral yang lain. Objektif kajian ini adalah untuk mengkaji bentuk fosforus dan kapasiti penjerapan fosforus pada horizon yang berbeza bagi tanah pasir Siri Jambu. Sampel tanah pasir iaitu Siri Jambu diambil daripada Merang, Terengganu oleh Roslan et al.(2010) pada kedalaman yang berbeza iaitu 0-15sm, 15-30sm, 30-45sm dan 45-60sm. Jumlah P di dalam tanah, P tersedia, organik P dan bukan organik P, dan kadar penyerapan P oleh tanah dikaji bagi setiap kedalaman tanah. Kandungan P adalah berbeza bagi setiap kedalaman. Jumlah kandungan P pada kedalaman 0-15sm, 15-30sm, 30-45sm dan 45-60sm ialah 6.42 mg/kg, 6.51 mg/kg, 5.5 mg/kg dan 5.08 mg/kg. Kandungan fosforus tersedia meningkat dengan peningkatan kedalaman tanah iaitu 1.99 mg/kg, 2.35mg/kg, 2.51 mg/kg dan 3.31 mg/kg. Jumlah organik P dan bukan organik P bergantung kepada jumlah keseluruhan P. Kandungan bukan organik P lebih tinggi daripada kandungan organik P. Berdasarkan isoterma penjerapan Freundlich, penjerapan P adalah paling tinggi pada kedalaman 0-15sm dan diikuti oleh kedalaman 15-30sm, 30-45sm dan 45-60sm.*

## CHAPTER 1

### INTRODUCTION

BRIS soil is very common in Kelantan-Terengganu Plains. BRIS soil in Malaysia comprises about 0.5% of the total land area which is approximately 160,000 ha. BRIS soil comprises more than 95% of sand up to 150cm of the soil profile. The topsoil is dominated by coarse sand fraction while the subsoil is dominated by very fine sand (Roslan *et al.* 2010). BRIS soil is not economically suitable for planting crops due to their inherent poor fertility. Fertility of BRIS soil is inherently low due to low cation exchange capacity, CEC and water holding capacity, low pH (Roslan *et al.* 2010), low nutrient retention capacity and carbon content, and high aluminium toxicity and Phosphorus fixation.

P is often the most limiting plant nutrient, especially in tropical soils. Mahdi *et al.*, (2011) stated that total P content in Malaysian sandy soil is hundred times lower than the other mineral soil which is only  $<100\mu\text{ P/g}$  ( $\sim 200\text{kg/ha}$ ). Phosphorus is important for plant because Phosphorus is involve in many plant processes which are energy transfer reactions, development of reproductive structures, protein synthesis, crop maturity and root growth. Based on Benton (1998), P in plant tissue is around 0.15 to 1.00%.

The orthophosphate,  $\text{H}_2\text{PO}_4^-$  and  $\text{HPO}_4^{2-}$  are the primary forms of Phosphorus taken up by plants (Hinsinger, 2001). Phosphorus is not available for plants when phosphate ions reacts with soluble Fe and Al in the soil forming insoluble P. This mechanism is called P fixation. According to Jones (1982), not all of the phosphorus content in soil are available for plant uptake because most of the phosphorus is in the form that is difficult to utilize. Brady and Weil (2002) stated that the sufficient available P in soil solution for optimum growth in most plants is about 0.2 mg/L.

Organic P and inorganic P are the two main forms of phosphorus in the soils and the important P source for optimum plant growth. The organic P is usually found in humus and other organic materials. Inorganic P is found in various combinations with other elements which most of the form are not soluble in water. Although organic P and inorganic P are important sources for plant growth, their availability are controlled by soil characteristics and environment conditions.

Thus, the study was undertaken with the following objectives:

- i. To study the forms of Phosphorus ( Total P, Available P, Organic P and Inorganic P) in Jambu soil series.
- ii. To study the Phosphorus adsorption capacity in the different horizon of Jambu soil series.

## REFERENCES

- Abdu, N (2006). Soil-Phosphorus Extraction Methodologies: A Review. *African Journal of Agricultural Research* Vol.1 (5), pp. 159-161.
- Ahmed, M. F., Kennedy, I. R., Choudhury, A. T. M. A., Kecskés, M. L., and Deaker, R. (2008). Phosphorus Adsorption in Some Australian Soils and Influence of Bacteria on the Desorption of Phosphorus. *Communications in Soil Science and Plant Analysis*, 39:9-10, 1269-1294.
- Brady, N. C. and Weil, R. R. (2002). *The Nature and Properties of Soils*, Thirteenth Edition. Upper Saddle River, New Jersey: Prentice Hall.
- De Datta, S. K., Biswas, T. K., and Charoenchamratcheep, C. (1990). Phosphorus Requirement and Management for Lowland Rice pages 307-324 in *Phosphorus Requirements for Sustainable Agriculture in Asia and Oceania: Proceedings of a Symposium 1989*. International Rice Research Institute.
- Frederick, R. T., and Louis M. T. (1993). *Soils and Soil Fertility* :Fifth Edition. Oxford University Press. pp 215-235.
- Gregory, T., L. Karns Chelsey and Ken D. Shimizu. (2005). A critical examination of the use of Freundlich isotherm in characterizing molecularly imprinted polymers (MIPS). *Analytica Chimica Acta.*, 528: 107-13.
- Hinsinger, P. (2001). Bioavailability of soil inorganic P in the rhizosphere as affected by root-induced chemical changes: a review. *Plant and Soil* 237: 173-195.
- Imai, N., Kitayama, K., and Titin, J. (2010). Distribution of Phosphorus in an Above-To-Below Ground Profile in an Bornean Tropical Rain Forest. *Journal of Tropical Ecology* 26:627-636.

- Indiati, R. (2000). Addition of phosphorus to soils with low to medium phosphorus retention capacities and effect on soil phosphorus sorption properties. *Communications in soil science and plant analysis*, 31:9-10, 1179-1194.
- Jones, U. S. (1982). *Fertilizers and Soil Fertility: Second Edition*. Reston Publishing Company, Inc.
- Khan, Q. U., Khan, M. J., Saif, U. R. and Ullah, S. (2010). Comparison of different models for phosphate adsorption in salt inherent soil series of Dera Ismail Khan. *Soil and Environ.* 29(1): 11-14.
- Kuo, S. and Lotse, E. G. (1974). Kinetics of phosphate adsorption and desorption by lake sediments. *Soil Sci. Soc. Am. Proc.*, 38:50-4.
- Lallajee, B. (1997). Phosphorus Fixation as Influenced by Soil Characteristics of Some Mauritius Soils. Food and Agricultural Research Council Mauritius.
- Law, W. M., and Tan, M. M. (1973). Chemical properties of some Peninsular Malaysian soil series. In: *Proceedings Conference of Chemistry and Fertility of Tropical Soils*, MSSS, Kuala Lumpur: 180-191.
- Madgoff, F. and Weil, R.R.(2004). *Soil Organic matter in sustainable agriculture*. CRC Press. Florida, USA.
- Mahdi, S. S., Hassan, G. I., Hussain, A., and Rasool, F. (2011). Phosphorus Availability Issue: Its Fixation and Role of Phosphate Solubilizing Bacteria in Phosphate Solubilization Research. *Journal of Agricultural Sciences* 2011, 2(1): 174-179.
- Olsen, S. R. and Watanabe F. S. (1957). A method to determine a phosphorus adsorption maximum of soils as measured by Langmuir Isotherm. *Soil Science Society of America Proceeding* 21: 144-149



- Paramanathan, S. (2000). Soil of Malaysia : Their Characteristics and Identification : Volume 1. Academy of Science Malaysia. pp 435-440.
- Potash & Phosphate Institute, PPI. (2005). International Soil Fertility Manual. USA.
- Roslan, I., Shamsudin, J., Fauziah, C.I., Anuar A.R.. (2010). Occurrence and properties of soils on sandy beach ridges in the Kelantan-Terengganu Plains, Peninsular Malaysia. *Journal Catena* 83: 55-63.
- Roslan, I., Shamsudin, J., Fauziah, C.I., Anuar A.R.. (2011). Fertility and Suitability of the Spodosols Formed on Sandy Beach Ridges Interspersed with Swales in the Kelantan-Terengganu Plains of Malaysia for Kenaf Production. *Malaysian Journal of Soil Science* Vol 15 : 1-24.
- Schulte, E. E. and Kelling, K. A. (1996). *Understanding Plant Nutrients: Soils and Applied Phosphorus*. Cooperative Extension Publications, University of Wisconsin.
- Shamshuddin, J., Fauziah, C. I., Anda, M. I., Kapok, J., and Shazana M. A. R. S. (2011). Using Ground Basalt and/or Organic Fertilizer to Enhance Productivity of Acid Soils in Malaysia for Crop Production. *Malaysian Journal of Soil Science*. Vol 15 : 127- 146.
- Sharpley, A. (2000). Phosphorus Availability. In Malcolm E. Summer (Ed.), *Handbook of Soil Science*. USA : (pp, D-18 – D-38). CRC Press.
- Singh, B. And Gilkes, R. J. (1991). Phosphorus sorption in relation to soil properties for the major soil types of south-west Australia. *Australian J. Soil Res.*, 29: 602-18.
- Soil Survey Staff, 2010. *Keys to soil taxonomy*. United States Department of Agriculture. Washington, DC.

Srinivasarao, C., Singh, R. N., Ganeshamurthy, A. N., Ghansham Singh & Masood Ali (2007). Fixation and Recovery of added Phosphorus and Potassium in different soil types of pulse-growing regions of India. *Communications in Soil Science and Plant Analysis*, 38:3-4, 449-460.

Usman, M. I., Edi, A. H. M., and Adzemi, M. A. (2013). Performances of BRIS Soil Genesis and Classification in Terengganu, Malaysia. *Journal of Biology, Agriculture and Healthcare*. Vol 3, No. 20.

Wiederholt, R. and Johnson, B. (2005). *Phosphorus Behaviour in the Environment*. North Dakota State University.

Yu, I. F. and Ho, S. B. (2009). Comparison of the Ignition Method and the Perchloric Acid Digestion Method for the Determination of Total Phosphorus in Agricultural Soils of Taiwan. *Communications in Soil Science and Plant Analysis*, Volume 40, Numbers 11-12, June 2009, pp 1953-1963(11).

Zaharah, A. R. (1979). Phosphate Adsorption by Some Malaysian Soils. *Pertanika* 2(2):73-83.

Zaharah, A. R. and Sharifuddin, H. A. H. (1979). Phosphorus forms, fixation and availability in Malaysian soils pages 13-24 in *Proceedings of a seminar on Chemistry and Fertility of Malaysian Soils* by Yaacob, O. Malaysian Soil Science Society.

Zaharah, A. R. Hawa, J., and Sharifuddin, H. A. H. (1985). Accumulation and Migration of Phosphate Applied as Rock Phosphate in Oil Palm Plantation. *Pertanika* 8(3), 317-321.