

## Investigation of structural and dielectric properties of subsolidus bismuth iron niobate pyrochlores

### ABSTRACT

Subsolidus  $\text{Bi}_{3.36}\text{Fe}_{2.08+x}\text{Nb}_{2.56-x}\text{O}_{14.56-x}$  (BFN) pyrochlores prepared by solid-state reaction at  $950^\circ\text{C}$  over 48 h showed a moderate solid solution range of  $-0.24 \leq x \leq 0.48$ . As to preserve the overall charge electroneutrality of the system, the doping mechanism was proposed to be a one-to-one substitution of  $\text{Nb}^{5+}$  by  $\text{Fe}^{3+}$  together with oxygen non-stoichiometry, i.e.  $\text{Nb}^{5+} \leftrightarrow \text{Fe}^{3+} - \text{O}^{2-}$ . BFN pyrochlores crystallized in a cubic symmetry, space group  $Fm\bar{3}m$  (No. 227),  $Z = 4$  and their refined lattice constants were found to be in the range  $10.5071(4)$ – $10.5107(7)$  Å. The microstructural analyses revealed their grain size range  $0.5$ – $8.4$   $\mu\text{m}$  and crystallite size range  $63$ – $78$  nm, respectively. These thermally stable BFN pyrochlores exhibited moderate high dielectric constants,  $\epsilon'$  in the range  $141$ – $150$  and dielectric losses,  $\tan \delta \sim 0.2$  at  $\sim 30^\circ\text{C}$  and 1 MHz.

**Keyword:** Pyrochlores; Solid-state method; Electrical properties; Niobate compounds