



UNIVERSITI PUTRA MALAYSIA

**CARBONATED ALKALI-ACTIVATED OLIVINE WITH GLASS FIBER FOR
SOIL STABILIZATION**

WISAM DHEY AB KHALAF

FK 2019 146



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By

WISAM DHEYAB KHALAF

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia
In Fulfilment of the Requirements for the Degree of Doctor of Philosophy**

November 2018

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of requirement for the degree of Doctor of Philosophy

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November 2018

Chairman : Professor Bujang Kim Huat, PhD
Faculty : Engineering

Soil stabilization is a universal approach commonly used in counter-balancing of soil ground under structures. This method of soil improvement utilizes binders such as cement and/or lime to enhance the mechanical properties of soil for construction purposes. However, the production of these binders has been known to increase the levels of carbon dioxide (CO₂) in the environment. Therefore, in an attempt to stabilize soil conditions, the search for sustainable materials which are essentially harmless to surrounding soils when treated and at the same time are cost-efficient, is justified. Olivine, with a chemical composition of [(Mg,Fe)₂SiO₄] can be considered as a sustainable material which has the natural capability of capturing CO₂ in the environment and creating carbonated minerals. The high amount of magnesium oxide (MgO) and aluminum oxide (Al₂O₃), as well as an adequate amount of silicon dioxide (SiO₂) form the chemical composition of olivine, making olivine a good choice for use in soil improvement activities in terms of its pozzolanic reaction and hydration. The present study was undertaken to emphasize some problems on the utilization of olivine as a newly proposed sustainable material for soil improvement programs. The study highlights the applicability of glass fiber with an alkali activated soil-olivine mixture, with and without carbonation, which helps in determining the Unconfined Compressive Strength (UCS), Indirect Tensile Strength (ITS), and the Flexural Strength (FS). In the study, Fourier Transform Infrared Spectroscopy (FTIR), Scanning Electron Microscopy (SEM), Energy-Dispersive X-ray spectroscopy (EDX), and X-ray Diffraction (XRD) analyses were also executed on pure soil and alkali activated soil-olivine mixtures with and without carbonation. The first stage was to analyze the performance of the preliminary investigation in order to evaluate the effectiveness of olivine on some basic geotechnical characteristics of silty clay soil. The compaction test and the Unconfined Compression Strength (UCS) were used as a practical indicator to investigate the strength development. According to the test findings, utilizing 30% olivine resulted

in a sharp increase in the compaction and the UCS of the samples, in the same curing time.

In the second stage of this study, carbonated alkaline activation of soil+30% olivine, was adopted as a viable technique to evaluate binder formation due to CO₂ pressure change. In simpler terms, the binder formation is generally a synthetic alkali aluminosilicate which is produced from the reaction of a solid aluminosilicate with pre-designed concentrated aqueous alkaline solutes. After this, pressurized CO₂ is injected into it form the new binder (MgCO₃/CaCO₃). Based on the obtained UCS values at exposure pressure of up to 300 kPa, for a 7 day exposure period, using alkali-activated olivine, it was found that the peak strength of soil+30%olivine was increased by up to 55 times compared to that of host soil. Regarding exposure period, it was found that based on UCS results at an exposure period of up to 7 days, using alkali activated olivine, the peak strength of soil+30%olivine was increased by up to 55 times compared to that of host soil.

The third stage was to identify the effect of the alkali agent molarity on the strength development. The rules of alkali agent (NaOH) molarity in binder formation were examined (with and without carbonation). In accordance to UCS values, 10 M of NaOH after 7 days of exposure and 300 kPa CO₂ pressure, increased peak strengths by up to 55 times compared to that of host soil and 5 times to that of alkali activated samples (without carbonation).

In the fourth and last stage, besides the shear strength development, in order to increase the tensile strength and ductility of soil+30%olivine, the combined effect of fibre inclusion and alkaline activation (with and without carbonation) was described and reported. In this stage, along with the 30% olivine in presence of high alkali solutes, mineral glass fibers were used as a strong reinforcement inclusion. Besides the UCS test, indirect tensile strength and flexural strength tests were carried out at pre-designed curing regems. The test results indicated that the inclusion of glass fibers within alkali-activated soil+30%olivine caused a further increase in the peak stress and tensile strength, and a decrease in the loss of post-peak strength. The results show that the incorporation of carbonation in alkali activated soil+30%olivine+3%glass fiber, increased the peak strength by up to 1.2 times to that of a mixture without glass fiber.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

KAJIAN OLIVINE TERAKTIF-ALKALI BERKABONAT DENGAN GENTIAN KACA UNTUK PENSTABILAN TANAH

Oleh

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Penstabilan tanah adalah satu pendekatan sejagat yang digunakan secara meluas bagi penyeimbangan tanah dibawah struktur. Kaedah penambahbaikan struktur tanah ini menggunakan pengikat seperti simen dan/atau kapur bagi meningkatkan ciri mekanikal tanah untuk tujuan pembinaan. Walaubagaimana pun, pengeluaran bahan pengikat tersebut diketahui meningkatkan paras karbon dioksida (CO₂) di persekitaran. Dengan itu, dalam usaha menstabilkan keadaan tanah, pencarian bahan mampan yang tidak berbahaya kepada tanah sekitar, dan pada masa yang sama, mempunyai kos yang cekap, adalah wajar. Olivine, dengan komposisi kimia [(Mg,Fe)₂SiO₄] dianggap sebagai bahan mampan yang mempunyai keupayaan semulajadi dalam memerangkap CO₂ dari persekitaran dan menjana mineral berkarbonat. Kandungan magnesium oksida (MgO) dan silikon oksida yang tinggi dan jumlah silikon dioksida (SiO₂) yang mencukupi dalam komposisi kimia olivine, menjadikan olivine sebagai satu pilihan untuk digunakan dalam penambahbaikan tanah dari segi reaksi pozzolanik dan penghidratan. Kajian ini dilaksanakan untuk memberi penekanan keatas beberapa masalah terhadap penggunaan olivine sebagai satu bahan yang mampan untuk program penambahbaikan tanah. Kajian ini juga menonjolkan kebolegunaan gentian kaca dengan tanah yang dirawat dengan olivine, menentukan kekuatan mampatan yang tiada had (UCS), kekuatan tegangan tidak langsung (ITS) dan kekuatan lenturan (FS). Dalam kajian ini, analisis Fourier Transform Infrared Spectroscopy (FTIR), Scanning Electron Microscopy (SEM), Energy-dispersive X-ray Spectroscopy (EDX), dan X-ray Diffraction (XRD) analisis juga dilaksanakan keatas tanah tulen dan campuran tanah-olivine yang diktifkan alkali dengan dan tanpa pengkarbonan. Objektif pertama daripada empat objektif dalam kajian ini adalah untuk meneliti kesan olivine keatas kekuatan mekanikal tanah asal, mengenalpasti ciri tanah dan olivine dan menentukan peratusan tanah/olivine yang berkesan melalui kekuatan mekanikal. Objektif kedua kajian adalah untuk mengkaji mekanisme asas bagi matriks tanah-olivine dalam kewujudan

NaOH di bawah tekanan gas CO₂ dan dalam jangkamasa yang berbeza, menentukan kesan karbonasi olivine di bawah tekanan gas CO₂ yang berbeza dan kesan karbonasi olivine didalam jangkamasa pemulihan CO₂ yang berbeza ke atas ciri mekanikal tanah. Objektif ketiga kajian ialah untuk mengenalpasti mekanisma asas matriks tanah-olivine dalam kewujudan NaOH yang berbeza molariti dan gas CO₂. Fokus kajian adalah untuk menentukan kesan pengkarbonan olivine dalam molariti NaOH yang rendah ke atas ciri mekanikal tanah. Objektif keempat dan terakhir adalah untuk menganggarkan nilai peratusan gentian yang ditambah ke dalam tanah, olivine dan NaOH dan kesan pengkarbonan ke atas interaksi antara gel geopolimerik dan gentian kaca. Dalam fasa pertama kajian, 30% olivine meningkatkan UCS sebanyak 12 kali ganda berbanding tanah tanpa rawatan. Keputusan struktur mikro menunjukkan bahawa olivine meningkatkan ciri kejuruteraan tanah, oleh itu, kestabilan tanah. Dalam fasa kedua, kesan pelbagai tekanan CO₂ dan jangkamasa pemulihan telah dikaji untuk memahami kesan-kesan tersebut ke atas campuran tanah-olivine yang diaktifkan alkali. Keputusan UCS menunjukkan peningkatan ketara dalam kekuatan mampatan dengan peningkatan tekanan daripada 200 ke 300 kPa. Kadar peningkatan adalah masing-masing 1.9% dan 1.04% antara 330 dan 400 kPa. Keputusan UCS bagi pelbagai jangkamasa pemulihan menunjukkan peningkatan dalam kekuatan mampatan selepas pemulihan selama 7 hari. Analisis SEM pada 300 kPa CO₂ menunjukkan pengurangan tekanan menyebabkan pemberhentian tanah akibat pengkarbonan dan penghidratan MgO untuk mengeluarkan MgCO₃ dan Mg(OH)₂. Peningkatan yang sama dan ketara dalam ketumpatan direkodkan pada SEM selepas 7 hari pemulihan. Analisis XRD mengesahkan kiasan ini. Di fasa 3, keputusan UCS menunjukkan 10M NaOH merekodkan nilai tertinggi 5035 kPa bagi sampel berkarbonat, iaitu 4.7 kali ganda lebih tinggi daripada sampel tanpa pengkarbonan. Dalam analisis SEM, kajian menunjukkan perpecahan olivine daripada NaOH bagi mendapatkan struktur tanah yang sama. Analisis EDX menunjukkan bahawa sementara nisbah Si/Al and Na/Al meningkat, kekuatan mampatan tanah turut meningkat. Analisis FTIR menunjukkan puncak Si-O, Si-O-Al, Al-O, H-O-H, -OH, dan C-O masing-masing membuktikan fungsi olivine dan NaOH sebagai pengikat pengaktif.

Di fasa yang keempat, keputusan ujian UCS, ITS dan FS menunjukkan pengkarbonan tanah-olivine yang diaktifkan alkali, sebagai pengaktif, merekodkan kekuatan kemampatan yang lebih tinggi berbanding tanpa pengkarbonan. Apabila dibandingkan dengan kerja sebelum ini, kajian ini mendapati bahawa lebih banyak kandungan tanah liat di dalam tanah, maka lebih banyak olivine diperlukan. Seterusnya, lebih banyak olivine diperlukan, lebih tinggi tekanan CO₂ diperlukan untuk pengkarbonan sepenuhnya. Lapan dan sepuluh molar NaOH memberi UCS yang ketara bagi campuran tanah-olivine dalam kewujudan CO₂, sementara 10 and 12 M NaOH adalah lebih ketara dalam sampel yang diaktifkan alkali. Perlakuan CSOF 3% adalah ketara yang menunjukkan keberkesanan peningkatan pengkarbonan dalam meningkatkan interaksi antara gel yang terbentuk dan yang diperkuatkan.

ACKNOWLEDGEMENTS

Firstly, I would like to offer my gratitude to Allah “the Lord of the Worlds”, for imparting me with the pleasure of facilitating the completion of this research. I would also like to sincerely express my gratitude to my beloved parents, my siblings, and my wife, for their love, support, and help.

I would like to express my thanks to my supervisors, Prof. Bujang Kim Huat, Dr. Haslinda Nahazanan, Prof. Mohd Saleh B Jaafar and Dr. Afshin Asadi for their guidance, supervision, patience, encouragement and support during the long journey of the finalization and completion of this thesis.

I would also like to humbly appreciate the financial support of the Universiti Putra Malaysia and Fundamental Research Grant Scheme (FRGS/1/2015/TK01/UPM/01/2) entitled —Sustainable Soil Stabilization by Olivine and its Mechanisms, funded by the Ministry of Higher Education in Malaysia (Project ID 93474-135837), for this research.

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TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiii
LIST OF FIGURES	xiv
LIST OF ABBREVIATIONS	xvii
CHAPTER	
1 INTRODUCTION	1
1.1 Introduction	1
1.2 Problem Statement	4
1.3 Objectives	4
1.4 Organization of this Thesis	5
2 LITERATURE REVIEW	6
2.1 Soil Stabilization	6
2.1.1 Cement and Lime Technique for Soil Stabilization	7
2.1.2 By-Product Usage in Stabilization of the Soil	9
2.1.3 Stabilization of Soil by Magnesium Oxide	12
2.1.4 Olivine Usage for Soil Stabilization	14
2.2 The Materials Designed for Soil Reinforcement	17
2.3 Alkaline Activation	22
2.3.1 The Source Binder Role in Alkaline Activation	24
2.3.2 Using Alkaline Activation Technique in Soil-Stabilization	28
2.4 Carbonated Alkaline Activation	29
2.5 Summary	36
3 METHODOLOGY	40
3.1 Introduction	40
3.2 Materials	43
3.2.1 Olivine	43
3.2.2 Soil	45
3.2.3 Alkaline Activator (NaOH)	48
3.2.4 Glass Fibers	48
3.3 Mechanical Tests	49
3.3.1 Compaction Test	49
3.3.2 Unconfined Compressive Strength	49
3.3.3 Indirect Tensile Strength	56
3.3.4 Flexural Strength	56
3.4 Microstructural Tests	57
3.4.1 Scanning Electron Microscopy (SEM/EDX)	57

3.4.2	Fourier-Transform Infrared Spectroscopy (FTIR)	57
3.4.3	X-ray Diffraction (XRD)	58
4	RESULTS AND DISCUSSIONS	59
4.1	Introduction	59
4.2	Effect of Olivine on the Mechanical Strength of Silty Clay Soil	59
4.2.1	Effect of Olivine on Soil Compaction Test	59
4.2.2	Unconfined Compressive Strength of Olivine Treated Soil	61
4.2.3	Microstructure Analysis	63
4.3	Effect of CO ₂ Exposure Period and Pressure	66
4.4	NaOH Molarity Effect on Strength Development	69
4.4.1	Unconfined Compressive Strength of Alkaline Activated samples (without carbonation)	69
4.4.2	Unconfined Compressive Strength of Alkaline Activated samples (with carbonation)	71
4.4.3	Microstructure Analysis	74
4.4.3.1	SEM Characterization of Alkaline Activation Samples	74
4.4.3.2	SEM Characterization of Carbonated Alkaline Activation Samples	75
4.4.3.3	FTIR Characterization of Alkaline Activation Samples	76
4.4.3.4	FTIR Characterization of Carbonated Alkaline Activation Samples	78
4.4.3.5	XRD Characterization of Alkaline Activated Soil-Olivine Samples with and without Carbonation	79
4.5	Glass Fiber in Alkali Activated Mixture	83
4.5.1	Unconfined Compressive Strength of Glass Fibre in Alkali-activated Samples (without Carbonation)	83
4.5.2	Unconfined Compressive Strength of Glass Fibre in Alkali-activated Samples (with Carbonation)	87
4.5.3	Effect of Fibre Inclusion on Toughness of Treated Samples	89
4.5.4	Effect of Fibre Inclusion on Indirect Tensile Strength of Alkali Activated Samples (Without and With Carbonation)	91
4.5.5	Effect of Fibre Inclusion on Flexural Strength of Alkali Activated Samples (Without and With Carbonation)	93
4.5.6	Microstructural Analysis	95
4.5.6.1	SEM/EDX Analysis	95
4.5.6.2	Fourier Transform Infrared Spectroscopy	101

5	SUMMARY, GENERAL CONCLUSIONS RECOMMENDATIONS FOR FUTURE RESEARCH	AND	103
5.1	Summary		103
5.2	General Conclusions		104
5.3	Recommendations for Future Study		106
	REFERENCES		107
	BIODATA OF STUDENT		127
	LIST OF PUBLICATIONS		128



LIST OF TABLES

Table		Page
2.1	The Literature- Review of the by-products materials usage in the stabilization of the soil	11
2.2	The Olivine Insignificant Chemistry	16
2.3	Olivine Physical Properties	16
2.4	Soil Reinforcement Summary Which Display the Performance of the Natural in Addition to the Fibers Products	19
2.5	Soil Reinforcement Research Summary Displaying the Derformance of the Human Fibers Manufacturing	20
2.6	Chemical Composition of Some Chosen Pure Minerals and Rocks and Their Possibility Carbon Dioxide Capturing Capacity	31
3.1	Physicochemical Properties of Olivine	44
3.2	Physical characteristics of soil	45
3.3	Chemical analysis of soil and glass fiber	46
3.4	Physical properties of glass fiber	48
3.5	Research design of samples in first stage	49
3.6	Notation and description of carbonated alkaline activated olivine treated soil	51
3.7	Identification and characterization of alkaline activated soil-olivine mixture with and without carbonation.	52
3.8	Mixture attributions of different series of test samples.	55

LIST OF FIGURES

Figure		Page
1.1	World CO ₂ Emissions from Fuel Combustion By Sectors.	2
2.1	Olivine Crystal Skelton	15
2.2	World Wide Olivine Distribution	16
2.3	Poly (Sialates) Body Skelton	23
2.4	The Particle Size Effect on Increasing Surface Area	33
3.1	Flow Chart of Methodology	42
3.2	Particle Size Distribution of Olivine Before and After Milling	43
3.3	SEM Image of Olivine	44
3.4	XRD Pattern for Olivine Showing Characteristic Diffraction Planes	45
3.5	Particle Size Distribution of Soil	46
3.6	SEM Image of Untreated Soil	47
3.7	XRD of Untreated Soil	47
3.8	Glass Fibers	48
3.9	Triaxial Setup for Soil Sample Carbonation Used in This Research	50
3.10	Schematic Diagram for Carbonating Alkaline Activated Soil-Olivine Samples	53
4.1	Effect of the Addition of Olivine on Dry Density and Water Content.	60
4.2	Comparison of Dry Density and Water Content in Different Percentages of Olivine.	60
4.3	The Unconfined Compressive Strength of Natural Soil and (10, 20, 30 and 40%) Olivine Treated Soil after 7 Day Curing Time.	62
4.4	Comparison Unconfined Compression Strength and Toughness in Different Percentages of Olivine.	62
4.5	SEM Image of 30% Olivine Treated Soil (ASO30) With 20 μ m Zoom.	64

4.6	SEM Image of 30% Olivine Treated Soil (ASO30) With 100 μ m Zoom.	65
4.7	The Unconfined Compressive Strength of Carbonated 30% of Olivine Treated Soil, at 200,300 and 400kPa Carbon Dioxide Pressure for Period of 7 Days of Carbonation	67
4.8	The Unconfined Compressive Strength of Carbonated 30% of Olivine Treated Soil, at 1,2,4,7 And 9 Days Curing Period of Carbonation.	68
4.9	(a) The Unconfined Compression Stress of Sodium Hydroxide Treated Soil-Olivine Mixture With 4, 6, 8, 10 And 12M. (b) UCS-Toughness Comparison of Those Samples. (c) Stress-Strain Behaviour of Soil-Olivine in Presence of 10M NaOH after 90 Day Curing Time	70
4.10	(a) The UCS of Carbonated Sodium Hydroxide Treated Soil-Olivine Mixture With 4, 6, 8, 10 And 12M. (b) UCS-Toughness Comparison of Those Samples. (c) Stress-Strain Behaviour of Soil-Olivine in Presence of 10M NaOH After 7 Day Curing Time and 300kPa CO ₂ Pressure.	73
4.11	The SEM Image of Alkaline Activated Soil Olivine Mixture for: (a) 8M, And (b) 10M of NaOH; And After 90 Day Curing Time.	75
4.12	The SEM Image of Carbonated Alkaline Activated Soil Olivine Mixture for: (a) 8M, And (b) 10M of NaOH; And After 7 Day Curing Time and 300kPa CO ₂ Pressure.	76
4.13	FTIR of Alkaline Activated Soil-Olivine With 8, 10, and 12M after 90 Day Curing Period.	77
4.14	FTIR of Carbonated Alkaline Activated Soil-Olivine with 6, 8, 10, and 12M after 7 Day Curing Period and 300kPa CO ₂ Pressure.	79
4.15	XRD of Alkaline Activated Soil-Olivine Samples After 90 Days Curing Time for: (a) 8, (b) 10, And (c) 12M of NaOH (B: Btucite, Mg: Magnesium, M: Mullite, S: Serpentine, SS: Sodium Silicate, Q: Quartz).	81
4.16	The XRD of Carbonated Alkaline Activated Soil-Olivine Samples for: (a) 6, (b) 8, (c) 10, and (d) 12M of NaOH with 300kPa CO ₂ Pressure and 7 Days Curing Period.	83
4.17	Stress-Strain Behaviour with UCS, Fiber Content and Toughness Relationship for Reinforced Alkali Activated Soil-Olivine Samples after (a, b) 7 days, (c, d) 28 Days, and (e, f) 90 Days Curing.	86

4.18	(a) Stress-Strain Behaviour, and (b) UCS, Fiber Content and Toughness Relationship, for Reinforced Carbonated Alkali Activated Soil-Olivine Samples after 7 Days Curing Period.	88
4.19	Tensile Load–Deflection Relationship in Selected Alkaline Activated Samples.	92
4.20	Tensile Load–Deflection Relationship in Selected Carbonated Alkaline Activated Samples.	92
4.21	Tensile Failure Morphology of Alkaline Activation Samples for Indirect Tensile strength Test: (a) Without Reinforcement Inclusion, And (b) With Reinforcement Inclusion Sample.	93
4.22	Flexural Response of Alkali Activated Test Samples After 90 Days Curing.	94
4.23	Flexural Response of Carbonated Alkali Activated Test Samples after 7 Days Curing.	95
4.24	SEM Image of Glass Fibre Surface.	96
4.25	SEM Image of Glass Fibre Inclusion in Conjunction with Alkali	97
4.26	SEM Image of Glass Fibre Inclusion in Conjunction With Carbonated Alkali Activation After 7 Days Curing.	98
4.27	SEM Image of Glass Fibre Inclusion in Conjunction with Alkali Activated Olivine.	99
4.28	SEM Image of Glass Fibre Inclusion in Conjunction with Carbonated Alkali Activated Olivine.	100
4.29	FTIR of Reinforced Alkali Activated (ASOF5%) Sample.	101
4.30	FTIR of Carbonated Alkali Activated Samples with (CSOF _{3%}) and without (CSO ₁₀) Reinforcement.	102

LIST OF ABBREVIATIONS

ASTM	American Society for Testing and Materials
GHG	Greenhouse gases
CSH	Calicum Silicate Hydrates
CAH	Calicum Aluminate Hydrates
XRD	X-ray diffractograms
SEM	Scanning Electron Microscopy
EDX	Energy-Dispersive X-ray spectroscopy
GGBS	Ground Granulated Blast furnace Slag
CKD	Cement Kiln Dust
POFA	Palm Oil Full Ash
CASH	Calicum Aluminate Silicate Hydrates
OWC	Optimum Water Content
MDD	Maximum Dry Density
UCS	Unconfined Compression Strength
LL	Liquid Limit
PL	Plastic Limit
PI	Plasticity Index
S	Soil
SO ₁₀	Soil mixed with 10% Olivine
SO ₂₀	Soil mixed with 20% Olivine
SO ₃₀	Soil mixed with 30% Olivine
SO ₄₀	Soil mixed with 40% Olivine
ASO ₄	Alkali activated Soil Olivine mixture with 4M of NaOH
ASO ₆	Alkali activated Soil Olivine mixture with 6M of NaOH
ASO ₈	Alkali activated Soil Olivine mixture with 8M of NaOH
ASO ₁₀	Alkali activated Soil Olivine mixture with 10M of NaOH
ASO ₁₂	Alkali activated Soil Olivine mixture with 12M of NaOH
CSO ₄	Carbonated alkali activated Soil Olivine mixture with 4M of NaOH

CSO ₆	Carbonated alkali activated Soil Olivine mixture with 6M of NaOH
CSO ₈	Carbonated alkali activated Soil Olivine mixture with 8M of NaOH
CSO ₁₀	Carbonated alkali activated Soil Olivine mixture with 10M of NaOH
CSO ₁₂	Carbonated alkali activated Soil Olivine mixture with 12M of NaOH
MSH	Magnesium Silicate Hydrate



CHAPTER 1

INTRODUCTION

1.1 Introduction

The stabilization process of soil always plays a significant role in the civil engineering field, in that it designates reliability of each consequent which can be established. By its own nature, soil is a completely adjustable material, yet also complex in its own way. Soil materials could offer great prospects that in civil engineering, and hence, they are being widely employed for the dexterous usage, easy worldwide availability and low costs of implementation. However, the most commonly occurring soil materials are not exactly suitable for construction or engineering purposes. Several materials are available for application to different kinds of soils that depending on their characteristics, modify its composition and properties. The general rules indicate that several soils do not possess sufficient strength. Therefore, utilizing certain materials by adding them to soils, may result in adjustment and fortification of the soils themselves. Concrete is a broadly utilized material, used for the purpose of soil stabilization.

The annual growing average in the utilization for coal, oil, and natural gas, was assessed to be about 1.4% also 1.6% and 2.4%, respectively. The greenhouse gasses (GHG) emissions which are predicted to intensify even further as anthropogenic activities are on the rise. Estimations indicate that by 2030, the emissions would be beyond 70%. It is predicted that the energy associated with CO₂ expanse could be more than 38 Billion at a worldwide circle in 2030, which could lead to universal climate modifications (Ke Mcneil Price and Khanna, 2013).

It was predicted that the needs and usage of numerous fuels would increase by 1.7% each year, until 2030 (Biol, 2002). An early sample of study in climate change estimated concentration levels of CO₂ to be between 430 ppm and 530 ppm by the year 2100. To keep up with those predictions, it would require both, long term and medium term cuts of CO₂ emissions cumulatively, and other GHG restrictions. To bring the concentration levels of CO₂ from 430 ppm to 480 ppm instead of 530 ppm, would need a reduction in the discharge of GHG from 70% to 40% from 2010 through 2050 (Edenhifer et al, 2014).

Figure 1.1 presents an overview of global CO₂ emission in 2015, which highlights that two sectors, namely Transport and Electricity & Heat, produced nearly two-thirds of global CO₂ emissions. . Other sectors include energy industries other than electricity, residential industry and others such as commerical fishing, forestry and agriculture.

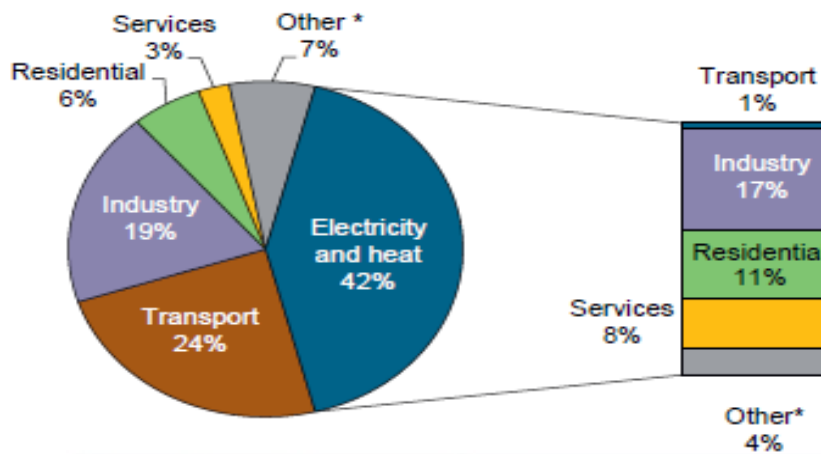


Figure 1.1: World CO₂ Emissions from Fuel Combustion By Sectors.

(Source: IEA, 2015)

Recent studies have demonstrated the usage of some substances, such as MgO, as better substitutions of cement for soil stabilization (Ahnreg and Johanssin, 2003; Jegandan, Al-Tabbaa, Lisk and Osman, 2010). The optimum rate of MgO, as the examination showed, could produce properties similar to that of pure cement. Furthermore, mixing soil with some waste substances in such combinations has proven to yield excellent results.

The American Society for Testing and Materials (ASTM) states that certain waste materials which are also pozzolanic in nature, could help in the process of soil stabilization. The utilization of MgO as an additive in the soil stabilization process has many advantages including, increasing the soil strength, decreasing the amount of cement used in soil, and capturing CO₂ in the soil, atmosphere, and rain. The utilization of CO₂ sequestration techniques offer a promising technology in the field of CO₂ fixation, by helping in the, fixing of CO₂ as a mineral (mineralization). This can be attributed to the minerals which can capture CO₂ and produce stable carbonates, resulting in permanent CO₂ sequestration. The process of chemical weathering by alkaline earth minerals include a process called mineral carbonation, which helps in capturing CO₂ (Kwon, 2011).

In the potential CO₂ capturing methods the usage of olivine (Mg-Fe)₂ SiO₄ is seen as a copious mineral technique. It was the employment of Mg and iron which led to the production of the Olivine matter. The outcome was mainly discovered in the mafic alongside the ultramafic igneous on rocks from the olivine production process, where other metamorphic rocks were typically founded (Jeas, 2011).

Earth olivine products were very rare to find, as they occur only in mantle xenoliths, which are the confident Basalt type in Dunite, and Peridotite massifs and phenocrysts, which are formed as a result of volcanic procedures (Schuiling, 2001).

In view of the environmental points, these byproducts and natural resources ought to be considered for the soil stabilization process. For a complete control of climate change olivine was considered to be the main contender for its superstation of the CO₂, the burning of 1L to relief CO₂ would require less than 1L of olivine typically, 1M of olivine burns 2M of CO₂ (Schuiling, 2001).

The above described CO₂ outcome from olivine happens as a result of mineral stabilization of the produced carbonates. In accordance with ASTM D5370 olivine is considered equal to a pozzolanic material, due to the enormous quantity of Al₂O₃, Fe₂O₃ also SiO₂, which could hypothetically be used to could advance soil products. Additionally, using MgO in a similar ratio as olivine creates a high possibility of carbonation of soil through the hydration process, giving it extra strength.

The worldwide distribution of olivine production as indicated in the Malaysian Geological Survey, is mostly concentrated along specific volcanic rocks, a great sum of which are present on the Andesite Dacite Basalt, which is located in the Mountains of Tawau in Sabah which is deliberated to be a key olivine foundation site (Tahir Musta and Rahim, 2010).

The first part of this research presents study of the olivine on the stabilization of soil and its benefits on engineering characteristics. A study is also done on key properties of olivine by analyzing some structures which have been treated by olivine using micro techniques. The second part of this research studies the carbonation effects of olivine on the alkaline stability of clayey soil which has been treated with olivine at different CO₂ pressures and at different curing times. The third stage of this research shows the function of the alkaline activator, sodium hydroxide (NaOH), at range of molarities on clayey soil which has been treated with olivine. The last part of this study explores the effect of carbonation in increasing the interaction between glass fibers (as the reinforcement member) and alkaline activated soil which has been treated by olivine.

In this research, the investigation used the CO₂ sequestration technique to stabilize the olivine soil products. Although in Malaysia, these soil products can still be detracted as unstable soils. Additionally, the research which added the alkaline activation by incorporating sodium hydroxide (NaOH) alongside olivine in order to increase the sequestration of CO₂, also ended up breaking and destroying the chemical bonding of SiO₂, for the stabilization process of soil. Similarly MgO, through a carbonation and hydration method of the olivine and MgO mixture, released SiO₂ pozzolanic materials.

The impact of this research drives from:

1. Decreasing the environment cost of constructing buildings.
2. Decreasing the time required to complete the construction activities.
3. Storage of CO₂ from atmosphere into a stable chemical form (MgCO₃/CaCO₃).
4. Using sustainable materials (olivine/glass fibers) for soil stabilization.
5. Strengthening silty clay soil.
6. Increasing the ductility of alkali activated material (with and without carbonation).

1.2 Problem Statement

Unstable soils are a geotechnical dilemma in Malaysia. Lime and/or cement were considered the most largely used stabilizers. However, these materials pose a few environmental problems resulting from CO₂ emission in the outside atmosphere; hence, there is a need to use materials which contain fewer damaging side effects on the natural environmental stabilization of soil, and furthermore decrease the content of waste materials on the surface of the earth. Moreover, some of the lime/cement-based products are also considered to be dangerous and harmful for soil surroundings.

Furthermore, the use of cementitious binders in soil stabilization results in low flexural strength and poor tensile as well as a brittle behaviour (Sukontasukkul and Jamsawang 2012; Correia, Oliveira, and Custódio 2015). Moreover, the stabilized soil tends to fail under tension, due to its brittleness (Sukontasukkul and Jamsawang 2012; Correia et al. 2015).

Therefore, this study introduces olivine and carbonated glass fibers as a contemporary sustainable materials, according to the range of the molarities of “NaOH”, for soil stabilization through CO₂ sequestration.

1.3 Objectives

Finding out the efficiency of olivine as a maintainable material is the main objective of this study, by examining and investigating the complete and distinct alkaline instigation procedure designed for the “soil stabilization”. Accordingly, the subsequent objectives which will be specifically addressed in the course of this study are:

1. To evaluate the effect of olivine on the mechanical strength of silty clay soil.

2. To assess the underlying mechanisms of soil-olivine matrix in the presence of NaOH with different CO₂ gas pressures in different exposure times.
3. To assess the underlying mechanisms of soil-olivine matrix in the presence of NaOH with a range of molarities with and without carbonation.
4. To evaluate the effect of carbonation on the strength and underlying mechanism of soil-olivine-glass fiber mixture in the presence of an alkaline activator.

1.4 Organization of this Thesis

This thesis will be organized by chapters; Chapter two will present a literature review on soil stabilization with the usage MgO and its by-products alongside the customary binders.

Furthermore, this chapter will also cover the key principles in the olivine mineral carbonation processes. The examination of alkaline activation on the maintenance of soils will be also mentioned in this chapter.

The third chapter will address in further detail the different effects of the olivine percentages on few engineering properties, and will also present a microstructure analysis of soil treated by olivine before carbonation in addition to alkaline activation process.

The third chapter also describes the techniques used in the research for soil stabilization. This chapter further describes the materials used, the classification tests as well as the chemical and physical, unconfined compressive strength, indirect tensile strength, flexural strength and micro-structural tests.

The fourth chapter gives an overview of the results of the testing programme, along with the analysis and discussion of these test results, aided with drown curves.

The fifth chapter presents a brief summary of the research methods and the results obtained, followed by a description on the full conclusion of this study and additional recommendations purposed for future studies.

REFERENCES

- Abdul Rahim, Azizli, Man, Rahmiati, & Nuruddin. (2014). Effect of Sodium Hydroxide Concentration on the Mechanical property of Non Sodium Silicate Fly Ash Based Geopolymer. *Journal of Applied Sciences*, 14(23), 3381–3384.
- Åhnberg, & Johansson. (2003). Stabilising effects of different binders in some Swedish soils. *Proceedings of the ICE - Ground Improvement*, 7(1), 9–23.
- AIST-Kansai. (2002). Crystal Structure Gallery.
- Alavéz-Ramírez, Montes-García, Martínez-Reyes, Altamirano-Juárez, & Gochi-Ponce. (2012). The use of sugarcane bagasse ash and lime to improve the durability and mechanical properties of compacted soil blocks. *Construction and Building Materials*, 34, 296–305.
- Alhassan. (2008). Potentials of Rice Husk Ash for Soil Stabilization. *AU Journal of Technology*, 11(4), 246–250.
- Ali, F. H., Adnan, & Choy. (1992). Use of rice husk ash to enhance lime treatment of soil. *Canadian Geotechnical Journal*, 29(5), 843–852.
- Ali, M., & Mullick. (1998). Volume stabilisation of high MgO cement: effect of curing conditions and fly ash addition. *Cement and Concrete Research*, 28(11), 1585–1594.
- Ali Ates. (2016). Mechanical properties of sandy soils reinforced with cement and randomly distributed glass fibers (GRC). *Composites Part B: Engineering*, 96, 295-304. DOI: 10.1016/j.compositesb.2016.04.049.
- Altun, & Yılmaz. (2002). Study on steel furnace slags with high MgO as additive in Portland cement. *Cement and Concrete Research*, 32, 1247–1249.
- Amaral, Oliveira, Salomão, Frollini, & Pandolfelli. (2010). Temperature and common-ion effect on magnesium oxide (MgO) hydration. *Ceramics International*, 36(3), 1047–1054.
- Andersson, Johansson, & Axelsson. (2002). Stabilization of Organic Soils by Cement and Pozzolanic Reactions. Swedish Deep Stabilization Research Center. C/O Swedish Geotechnical Institute.
- Andreani, Luquot, Gouze, Godard, Hoisé, & Gibert. (2009). Experimental study of carbon sequestration reactions controlled by the percolation of CO₂-rich brine through peridotites. *Environmental Science & Technology*, 43(4), 1226–31.

- Arulrajah, Disfani, Maghoolpilehrood, Horpibulsuk, Udonchai, Imteaz, & Du. (2015). Engineering and environmental properties of foamed recycled glass as a lightweight engineering material. *Journal of Cleaner Production*, 94, 369–375.
- Bakharev, Sanjayan, & Cheng. (1999). Alkali activation of Australian slag cements. *Cement and Concrete Research*, 29(1), 113–120.
- Bakri, Kamarudin, Norazian, Ruzaidi, & Zarina. (2011). Microstructure Studies on the Effect of the Alkaline Activators Ratio in Preparation of Fly Ash-Based Geopolymer. *IPCBEE*, 10, 13–17.
- Barker, Rogers, & Boardman. (2006). Physio-Chemical Changes in Clay Caused by Ion Migration from Lime Piles. *Journal of Materials in Civil Engineering*, 18(2), 182–189.
- Basha, Hashim, Mahmud, & Muntohar. (2005). Stabilization of residual soil with rice husk ash and cement. *Construction and Building Materials*, 19(6), 448–453.
- Bernal, Mejía de Gutiérrez, & Provis. (2012). Engineering and durability properties of concretes based on alkali-activated granulated blast furnace slag/metakaolin blends. *Construction and Building Materials*, 33, 99–108.
- Birle, Gibbs, Moore, & Smith. (1968). Crystal structures of natural olivines. *The American Mineralogist*, 53.
- Birol. (2002). *World energy outlook*. IEA. Paris.
- Blencoe, & Palmer. (2004). Carbonation of metal silicates for long-term CO₂ sequestration. US Paten. USA.
- Bondar, Lynsdale, Milestone, Hassani, & Ramezaniapour. (2011). Effect of type, form, and dosage of activators on strength of alkali-activated natural pozzolans. *Cement and Concrete Composites*, 33(2), 251–260.
- Brantley. (2008). Kinetics of mineral dissolution. In *Kinetics of water-rock interaction* (pp. 151–210). Springer New York.
- Brew, D. R M, & Glasser. (2005). Synthesis and characterisation of magnesium silicate hydrate gels. *Cement and Concrete Research*, 35(1), 85–98.
- Brew, D. R. M., & MacKenzie. (2007). Geopolymer synthesis using silica fume and sodium aluminate. *Journal of Materials Science*, 42(11), 3990–3993.
- British Standard. (2003a). British Standard Methods of test for Soils for civil engineering purposes. Part 2: Classification tests. BS 1377-2: 1990. In *British Standard* (pp. 1–49).

- British Standard. (2003b). British Standard Methods of test for Soils for civil engineering purposes. Part 4: Compaction-related tests. BS 1377-4: 1990. In British Standard (pp. 1–53).
- British Standard. (2003c). British Standard Methods of test for Soils for civil engineering purposes. Part 7: Shear strength tests (total stress). BS 1377-7:1990.
- Brooks. (2009). Soil stabilization with fly ash and rice husk ash. *International Journal of Research and Reviews in Applied Sciences*, 1(3), 209–217.
- Buchwald, Hilbig, & Kaps. (2007). Alkali-activated metakaolin-slag blends—performance and structure in dependence of their composition. *Journal of Materials Science*, 42(9), 3024–3032.
- Cai, Du, Liu, & Singh. (2015). Physical Properties, Electrical Resistivity and Strength Characteristics of Carbonated Silty Soil Admixed with Reactive Magnesia. *Canadian Geotechnical Journal*, 52(999), 1–15.
- Cai, Y., Shi, B., Gao, W., Chen, F.J., and Tang, C.s. (2006). Experimental study on engineering properties of fibre-lime treated soils. *Chinese Journal of Geotechnical Engineering*, 28(10): 1283-1287.
- Celaya, Veisi, Nazarian, & Puppala. (2011). Accelerated Design Process of Lime-stabilized Clays. *Geo-Frontiers, ASCE*, 4468–4478.
- Chauhan, M.S., Mittal, S., and Mohanty, B. (2008). Performance evaluation of silty sand subgrade reinforced with fly ash and fibre. *Geotextiles and Geomembranes*, 26(5): 429-435.
- Chen, Y., & Brantley. (2000). Dissolution of forsteritic olivine at 65 C and 2 < pH < 5. *Chemical Geology*, 165(3-4), 267–281.
- Chen, Z.-Y., O'Connor, & Gerdemann. (2006). Chemistry of aqueous mineral carbonation for carbon sequestration and explanation of experimental results. *Environmental Progress*, 25(2), 161–166.
- Chew, Kamruzzaman, & Lee. (2004). Physicochemical and Engineering Behavior of Cement Treated Clays. *Journal of Geotechnical and Geoenvironmental Engineering*, 130(7), 696–706.
- Chiang, Ferraro, Fratini, Ridi, Yeh, Jeng, ... Baglioni. (2014). Multiscale structure of calcium- and magnesium-silicate-hydrate gels. *Journal of Materials Chemistry A*, 2(32), 12991–12998.
- Clark, J. (1998). Future of automotive body materials: steel, aluminum and polymer composites. Hoogovens Technology, Massachusetts Institute of Technology, Cambridge, MA.

- Consoli, N.C., de Moraes, R.R., and Festugato, L. (2013). Variables controlling strength of fibre-reinforced cemented soils. *Proceedings of the ICE-Ground Improvement*, 166(4): 221-232.
- Consoli, Lopes, Prietto, Festugato, & Cruz. (2011). Variables Controlling Stiffness and Strength of Lime-Stabilized Soils. *Journal of Geotechnical and Geoenvironmental Engineering*, 137(6), 628–632.
- Consoli, Rosa, Cruz, & Rosa. (2011). Water content, porosity and cement content as parameters controlling strength of artificially cemented silty soil. *Engineering Geology*, 122(3-4), 328–333.
- Criado, Palomo, & Fernandezjimenez. (2005). Alkali activation of fly ashes. Part 1: Effect of curing conditions on the carbonation of the reaction products. *Fuel*, 84(16), 2048–2054.
- Cristelo, Glendinning, Fernandes, & Pinto. (2012a). Effect of calcium content on soil stabilisation with alkaline activation. *Construction and Building Materials*, 29, 167–174.
- Cristelo, Glendinning, Fernandes, & Pinto. (2013). Effects of alkaline-activated fly ash and Portland cement on soft soil stabilisation. *Acta Geotechnica*, 8(4), 395–405.
- Cristelo, Glendinning, Miranda, Oliveira, & Silva. (2012). Soil stabilisation using alkaline activation of fly ash for self compacting rammed earth construction. *Construction and Building Materials*, 36, 727–735.
- Cristelo, Glendinning, & Pinto. (2011). Deep soft soil improvement by alkaline activation. *Proceedings of the ICE - Ground Improvement*, 164(1), 1–10.
- Cwirzen, & Habermehl-Cwirzen. (2012). Effects of Reactive Magnesia on Microstructure and Frost Durability of Portland Cement Based Binders. *Journal of Materials in Civil Engineering*, 25(12), 1941–1950.
- Dash, & Hussain. (2012). Lime stabilization of soils: reappraisal. *Journal of Materials in Civil Engineering*, 24(6), 707–714.
- Daval, Sissmann, Menguy, Saldi, Guyot, Martinez, ... Hellmann. (2011). Influence of amorphous silica layer formation on the dissolution rate of olivine at 90°C and elevated pCO₂. *Chemical Geology*, 284(1-2), 193–209.
- Davidovits, J. (1991). Geopolymers. *Journal of Thermal Analysis*, 37(8), 1633–1656.
- Davidovits, Joseph, & Sawyer. (1985). *Early high-strength mineral polymer*. USA.

- Davidovits, J. (1988). Geopolymer chemistry and properties. Proceeding of 1st European Conference on Soft Mineralogy (Geopolymere '88), Compiègne, Paris.
- De Silva, Bucea, & Sirivivatnanon. (2009). Chemical, microstructural and strength development of calcium and magnesium carbonate binders. *Cement and Concrete Research*, 39(5), 460–465.
- De Aza, P., Guitian, F., and De Aza, S. (1994). Bioactivity of wollastonite ceramics: in vitro evaluation. *Scripta metallurgica et materialia*, 31(8): 1001-1005.
- Degirmenci, Okucu, & Turabi. (2007). Application of phosphogypsum in soil stabilization. *Building and Environment*, 42(9), 3393–3398.
- Dermatas, & Meng. (2003). Utilization of fly ash for stabilization/solidification of heavy metal contaminated soils. *Engineering Geology*, 70(3-4), 377–394.
- Dey, V., Kachala, R., Bonakdar, A., and Mobasher, B. (2015). Mechanical properties of micro and sub-micron wollastonite fibers in cementitious composites. *Construction and Building Materials*, 82: 351-359.
- Ding, L., Ma, L., Luo, H., Yu, M., and Wu, X. (2011). Wavelet Analysis for tunneling-induced ground settlement based on a stochastic model. *Tunnelling and Underground Space Technology*, 26(5): 619-628.
- Du, Bo, Jin, & Liu. (2015). Durability of reactive magnesia-activated slag-stabilized low plasticity clay subjected to drying–wetting cycle. *European Journal of Environmental and Civil Engineering*, 1–16.
- Du, Zhang, & Y. (2011). Investigation of Strength and California Bearing Ratio Properties of Natural Soils Treated by Calcium Carbide Residue. *Geo-Frontiers*, 1237–1244.
- Dufaud, Martinez, & Shilobreeva. (2009). Experimental study of Mg-rich silicates carbonation at 400 and 500 °C and 1 kbar. *Chemical Geology*, 265(1-2), 79–87.
- Dunsmore. (1992). A geological perspective on global warming and the possibility of carbon dioxide removal as calcium carbonate mineral. *Energy Conversion and Management*, 33(5), 565–572.
- Duxson, P., Fernández-Jiménez, Provis, Lukey, Palomo, & Deventer. (2007). Geopolymer technology: the current state of the art. *Journal of Materials Science*, 42(9), 2917–2933.
- Duxson, P., Mallicoat, Lukey, Kriven, & van Deventer. (2007). The effect of alkali and Si/Al ratio on the development of mechanical properties of metakaolin-based geopolymers. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 292(1), 8–20.

- Duxson, Peter, & Provis. (2008). Designing precursors for geopolymer cements. *Journal of the American Ceramic Society*, 91(12), 3864–3869.
- Duxson, Peter, Provis, Lukey, Mallicoate, Kriven, & Van Deventer. (2005). Understanding the relationship between geopolymer composition, microstructure and mechanical properties. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 269(1-3), 47–58.
- Dwivedi, & Jain. (2014). Fly ash – waste management and overview: A Review. *Recent Research in Science and Technology*, 6(1), 30–35.
- Edenhofer, Pichs-Madruga, Sokona, Kadner, Minx, & Brunner. (2014). Mitigation of climate change. Summary for Policymakers. IPCC.
- Edil, Acosta, & Benson. (2006). Stabilizing Soft Fine-Grained Soils with Fly Ash. *Journal of Materials in Civil Engineering*, 18(2), 283–294.
- Estabragh, A., Namdar, P., and Javadi, A. (2012). Behavior of cement-stabilized clay reinforced with nylon fiber. *Geosynthetics International*, 19(1): 85-92.
- Farouk, & Shahien. (2013). Ground improvement using soil–cement columns: Experimental investigation. *Alexandria Engineering Journal*, 52(4), 733–740.
- Fasihnikoutalab, M. H., Asadi A., Unluer C., Huat, B. B., Ball R. J., Pourakbar, S. (2017). Utilization of Alkali-Activated Olivine in Soil Stabilization and the Effect of Carbonation on Unconfined Compressive Strength and Microstructure. *Journal of Materials in Civil Engineering* 29(6) DOI: 10.1061/(ASCE)MT.1943-5533.0001833.
- Fasihnikoutalab, M. H., Asadi A., Huat, B. B., Ball R. J., Pourakbar, S., Singh P. (2016). Utilisation of carbonating olivine for sustainable soil stabilisation. *Environmental Geotechnics* 4(3): 184-198. doi.org/10.1680/jenge.15.00018.
- Fatahi, B., Khabbaz, H., and Fatahi, B. (2012). Mechanical characteristics of soft clay treated with fibre and cement. *Geosynthetics International*, 19(3): 252-262.
- Fawer, Concannon, & Rieber. (1999). Life cycle inventories for the production of sodium silicates. *The International Journal of Life Cycle Assessment*, 4(4), 207–212.
- Fernández-Jiménez, & Palomo. (2005). Composition and microstructure of alkali activated fly ash binder: Effect of the activator. *Cement and Concrete Research*, 35(10), 1984–1992.
- Friedlingstein, Andrew, Rogelj, Peters, Canadell, Knutti, ... Le Quere. (2014). Persistent growth of CO₂ emissions and implications for reaching climate targets. *Nature Geosci*, 7(10), 709–715.

- Garcia, Beaumont, Perfetti, Rouchon, Blanchet, Oger, ... Haeseler. (2010). Experiments and geochemical modelling of CO₂ sequestration by olivine: Potential, quantification. *Applied Geochemistry*, 25(9), 1383–1396.
- García, Chimenos, Fernández, Miralles, Segarra, & Espiell. (2004). Low-grade MgO used to stabilize heavy metals in highly contaminated soils. *Chemosphere*, 56(5), 481–91.
- Gartner. (2004). Industrially interesting approaches to —low-CO₂ cements. *Cement and Concrete Research*, 34(9), 1489–1498.
- Gerdemann, & Dahlin. (2003). Carbon dioxide sequestration by aqueous mineral carbonation of magnesium silicate minerals. Albany Research Center.
- Gjorv, & Sakai. (2003). Concrete technology for a sustainable development in the 21st century. New York, E & FN Spon Press.
- Gleize, Müller, & Roman. (2003). Microstructural investigation of a silica fume–cement–lime mortar. *Cement and Concrete Composites*, 25(2), 171–175.
- Glukhovskiy, Rostovskaja, & Rumyna. (1980). High strength slag-alkaline cements. In *Proceedings of the 7th International Congress on the Chemistry of Cement*, Paris.
- Gullu, H., and Hazirbaba, K. (2010). Unconfined compressive strength and postfreeze– thaw behavior of fine-grained soils treated with geofiber and synthetic fluid. *Cold regions science and technology*, 62(2): 142-150.
- Guo, Shi, & Dick. (2010). Compressive strength and microstructural characteristics of class C fly ash geopolymer. *Cement and Concrete Composites*, 32(2), 142–147.
- Habert, d'Espinoze de Lacaillerie, & Roussel. (2011). An environmental evaluation of geopolymer based concrete production: reviewing current research trends. *Journal of Cleaner Production*, 19(11), 1229–1238.
- Hänchen, M., Prigiobbe, & Storti. (2006). Dissolution kinetics of fosteritic olivine at 90–150° C including effects of the presence of CO₂. *Geochimica et Cosmochimica Acta*, 70(17), 4403–4416.
- Hänchen, Markus, Krevor, Mazzotti, & Lackner. (2007). Validation of a population balance model for olivine dissolution. *Chemical Engineering Science*, 62(22), 6412–6422.
- Hangx, & Spiers. (2009). Coastal spreading of olivine to control atmospheric CO₂ concentrations: A critical analysis of viability. *International Journal of Greenhouse Gas Control*, 3(6), 757–767.

- Harichane, Ghrici, & Kenai. (2011). Effect of the combination of lime and natural pozzolana on the compaction and strength of soft clayey soils: a preliminary study. *Environmental Earth Sciences*, 66(8), 2197–2205.
- Harrison. (2003). New cements based on the addition of reactive magnesia to Portland cement with or without added pozzolan, 1–11.
- Harrison. (2008). Reactive magnesium oxide cements. US Patent 7,347,896.
- Haug, T. A., Kleiv, & Munz. (2010). Investigating dissolution of mechanically activated olivine for carbonation purposes. *Applied Geochemistry*, 25(10), 1547–1563.
- Haug, T. a., Munz, & Kleiv. (2011). Importance of dissolution and precipitation kinetics for mineral carbonation. *Energy Procedia*, 4, 5029–5036.
- Higgins. (2005). Soil stabilisation with ground granulated blastfurnace slag. UK Cementitious Slag Makers Association (CSMA), 1–15.
- Horpibulsuk, S, Rachan, & Raksachon. (2009). Role of fly ash on strength and microstructure development in blended cement stabilized silty clay. *Soils and Foundations*, 49(1), 85–98.
- Horpibulsuk, Suksun. (2012). Strength and Microstructure of Cement Stabilized Clay. INTECH.
- Horpibulsuk, Suksun, Phetchuay, & Chinkulkijniwat. (2012). Soil Stabilization by Calcium Carbide Residue and Fly Ash. *Journal of Materials in Civil Engineering*, 24(2), 184–193.
- Horpibulsuk, Suksun, Rachan, Chinkulkijniwat, Raksachon, & Suddeepong. (2010). Analysis of strength development in cement-stabilized silty clay from microstructural considerations. *Construction and Building Materials*, 24(10), 2011–2021.
- Horpibulsuk, Suksun, Rachan, & Suddeepong. (2011). Assessment of strength development in blended cement admixed Bangkok clay. *Construction and Building Materials*, 25(4), 1521–1531.
- Hossain, K. M. a, & Mol. (2011). Some engineering properties of stabilized clayey soils incorporating natural pozzolans and industrial wastes. *Construction and Building Materials*, 25(8), 3495–3501.
- Hossain, M., Narioka, & Sakai. (2006). Effect of ordinary portland-cement on properties of clayey soil in Mie prefecture. *Journal of the Japanese Society of Soil Physics*, 103, 31–38.
- Huijgen, & Comans. (2003). Carbon dioxide sequestration by mineral carbonation. Netherlands.

- Huijgen, & Comans. (2007). Carbon dioxide sequestration by mineral carbonation.
- IEA. (2012). CO₂ EMISSIONS FROM FUEL COMBUSTION IEA STATISTICS HIGHLIGHTS. IEA STATISTICS.
- Jambor, & Smith. (1964). Olivine composition determination with small-diameter X-ray powder cameras. *Mineral. Mag.*, (3), 730–741.
- Jegandan, Al-Tabbaa, Liska, & Osman. (2010). Sustainable binders for soil stabilisation. *Proceedings of the ICE - Ground Improvement*, 163(1), 53–61.
- Jesa. (2011). Olivine.
- Kalla, P., Misra, A., Gupta, R.C., Csetenyi, L., Gahlot, V., and Arora, A. (2013). Mechanical and durability studies on concrete containing wollastonite–fly ash combination. *Construction and Building Materials*, 40: 1142-1150.
- Kamon, & Nontananandh. (1991). Combining industrial wastes with lime for soil stabilization. *Journal of Geotechnical Engineering*, 117(1), 1–17.
- Kaniraj, & Havanagi. (2001). Behavior of cement-stabilized fiber-reinforced fly ash-soil mixtures. *Journal of Geotechnical and Geoenvironmental Engineering*, 127(7), 574–584.
- Kasama, Zen, & Iwataki. (2007). High-strengthening of cement-treated clay by mechanical dehydration. *Soils and Foundations*.
- Kavak, A., Bilgen, G., & Capar. (2011). Using Ground Granulated Blast Furnace Slag with Seawater as Soil Additives in Lime-Clay Stabilization. *Journal of ASTM International*, 8(7), 1–12.
- Kayar. (2011). All About Chemical Weathering.
- Kazemian, Huat, & Moayed. (2012). Undrained Shear Characteristics of Tropical Peat Reinforced with Cement Stabilized Soil Column. *Geotechnical and Geological Engineering*, 30(4), 753–759.
- Ke, Mcneil, Price, & Khanna. (2013). Estimation of CO₂ Emissions from China 's Cement Production : Methodologies and Uncertainties Methodologies and Uncertainties. *Energy Policy*, 57, 172–181.
- Kolias, Kasselouri-Rigopoulou, & Karahalios. (2005). Stabilisation of clayey soils with high calcium fly ash and cement. *Cement and Concrete Composites*, 27(2), 301–313.
- Komnitsas, & Zaharaki. (2007). Geopolymerisation: A review and prospects for the minerals industry. *Minerals Engineering*, 20(14), 1261–1277.

- Kumar, Walia, & Bajaj. (2007). Influence of Fly Ash, Lime, and Polyester Fibers on Compaction and Strength Properties of Expansive Soil. *Journal of Materials in Civil Engineering*, 19(3), 242–248.
- Kwon. (2011). Mineralization for carbon dioxide sequestration using olivine sorbent in the presence of water vapor. Georgia Institute of Technology.
- Kwon, Fan, DaCosta, & Russell. (2011). Factors affecting the direct mineralization of CO₂ with olivine. *Journal of Environmental Sciences*, 23(8), 1233–1239.
- Lee, S. L., & Karunaratne. (2007). Treatment of soft ground by Fibredrain and highenergy impact in highway embankment construction. *Proceedings of the ICE - Ground Improvement*, 11(4), 181–193.
- Lee, W. K. W., & Van Deventer. (2002a). The effect of ionic contaminants on the early-age properties of alkali-activated fly ash-based cements. *Cement and Concrete Research*, 32(4), 577–584.
- Lee, W. K. W., & Van Deventer. (2002b). The effects of inorganic salt contamination on the strength and durability of geopolymers. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 211(2-3), 115–126.
- Low, N.M., and Beaudoin, J. (1994a). Stability of portland cement-based binders reinforced with natural wollastonite micro-fibres. *Cement and Concrete Research*, 24(5): 874-884.
- Low, N.M., and Beaudoin, J.J. (1993). The effect of wollastonite micro-fibre aspect ratio on reinforcement of portland cement-based binders. *Cement and Concrete Research*, 23(6): 1467-1479.
- Li, C., Sun, & Li. (2010). A review: The comparison between alkali-activated slag (Si+Ca) and metakaolin (Si+Al) cements. *Cement and Concrete Research*, 40(9), 1341–1349.
- Li, L., Weidner, Raterron, Chen, & Vaughan. (2004). Stress measurements of deforming olivine at high pressure. *Physics of the Earth and Planetary Interiors*, 143-144, 357–367.
- Li, Z., & Chau. (2008). Reactivity and Function of Magnesium Oxide in Sorel Cement. *Journal of Materials in Civil Engineering*, 20(3), 239–244.
- Lime stabilisation practice. (2008). Astralia.
- Liska, & Al-Tabbaa. (2008). Performance of magnesia cements in pressed masonry units with natural aggregates: Production parameters optimisation. *Construction and Building Materials*, 22(8), 1789–1797.

- Liu, M. Y. J., Chua, Alengaram, & Jumaat. (2014). Utilization of Palm Oil Fuel Ash as Binder in Lightweight Oil Palm Shell Geopolymer Concrete. *Advances in Materials Science and Engineering*, 2014, 1–6.
- Liu, X., & Li. (2005). Effect of MgO on the composition and properties of alite-sulphoaluminate cement. *Cement and Concrete Research*, 35(9), 1685–1687.
- Liu, X., Li, & Zhang. (2002). Influence of MgO on the formation of Ca_3SiO_5 and $3\text{CaO}\cdot 3\text{Al}_2\text{O}_3\cdot \text{CaSO}_4$ minerals in alite – sulphoaluminate cement. *Cement and Concrete Research*, 32, 1125–1129.
- Liu, Z., Asce, Yu, & Asce. (2012). Multiscale Chemico-Thermo-Hydro-Mechanical Modeling of Early-Stage Hydration and Shrinkage of Cement Compounds. *Journal of Materials in Civil Engineering*, 24(6), 707–714.
- MacKenzie, Brew, Fletcher, Nicholson, Vagana, & Schmücker. (2006). Advances in Understanding the Synthesis Mechanisms of New Geopolymeric Materials. In *Novel Processing of Ceramics and Composites* (pp. 185–199). John Wiley & Sons, Inc.
- Maliakal, T., and Thiyyakkandi, S. (2013). Influence of randomly distributed coir fibers on shear strength of clay. *Geotechnical and Geological Engineering*, 31(2): 425-433.
- Mamlouk, M.S., Zaniewski, J.P., and Peng, X. (2006). *Materials for civil and construction engineers*. California: Addison-Wesley Press.
- Sargent, P., Hughes, P.N., Rouainia, M., and White, M.L. (2013). The use of alkali activated waste binders in enhancing the mechanical properties and durability of soft alluvial soils. *Engineering Geology*, 152(1): 96-108.
- Mazzotti. (2005). Mineral carbonation and industrial uses of carbon dioxide. In *Special Report on Carbon dioxide Capture and Storage* (pp. 320–338). Intergovernmental Panel on Climate Change.
- McLellan, Williams, Lay, van Riessen, & Corder. (2011). Costs and carbon emissions for geopolymer pastes in comparison to ordinary portland cement. *Journal of Cleaner Production*, 19(9-10), 1080–1090.
- Mijarsh, Megat Johari, & Ahmad. (2014). Synthesis of geopolymer from large amounts of treated palm oil fuel ash: Application of the Taguchi method in investigating the main parameters affecting compressive strength. *Construction and Building Materials*, 52, 473–481.
- Miller, & Azad. (2000). Influence of soil type on stabilization with cement kiln dust. *Construction and Building Materials*, 14(2), 89–97.
- Miura, Horpibulsuk, & Nagaraj. (2001). Engineering behavior of cement stabilized clay at high water content. *Soils and Foundations*, 41(5), 33–45.

- Mo, & Panesar. (2012). Effects of accelerated carbonation on the microstructure of Portland cement pastes containing reactive MgO. *Cement and Concrete Research*, 42(6), 769–777.
- Muhunthan, & Sariosseiri. (2008). Interpretation of geotechnical properties of cement treated soils. Washington.
- Munz, Kihle, Brandvoll, Machenbach, Carey, Haug, ... Eldrup. (2009). A continuous process for manufacture of magnesite and silica from olivine, CO₂ and H₂O. *Energy Procedia*, 1(1), 4891–4898.
- Naik. (2005). Sustainability of cement and concrete industries. Dundee, Scotland.
- Nematollahi, Sanjayan, & Shaikh. (2014). Comparative deflection hardening behavior of short fiber reinforced geopolymer composites. *Construction and Building Materials*, 70, 54–64.
- Ng, Voo, & Foster. (2012). Sustainability with Ultra-High Performance and Geopolymer Concrete Construction. In *Innovative Materials and Techniques in Concrete Construction* (pp. 81–100). Springer Science+Business Media.
- Ninov, Donchev, Lenchev, & Grancharov. (2007). Chemical stabilization of sandy-silty illite clay. *Journal of the University of Chemical Technology and Metallurgy*, 42(1), 67–72.
- Nontananandh. (2005). Scanning electron microscopic investigation of cement stabilized soil. In *Proceedings of the 10 th National Convention on Civil Engineering* (pp. 23–26). Pattaya, Thailand.
- O'Connor, W., Dahlin, & Nilsen. (2000). Carbon dioxide sequestration by direct mineral carbonation with carbonic acid. In *Proceedings of the 25th International Technical Conf. On Coal Utilization & Fuel Systems*, Coal Technology Assoc., Clear Water (pp. 1–15). United States: Coal Technology Association, Gaithersburg, MD.
- O'Connor, W., Dahlin, & Nilsen. (2001). Carbon dioxide sequestration by direct mineral carbonation: results from recent studies and current status. In *First National Conference on Carbon Sequestration* (pp. 1–10). Washington, DC: National Energy Technology Laboratory, U.S. Department of Energy.
- O'Connor, W. K., Dahlin, Rush, Gerdemann, & Penner. (2004). Energy and economic evaluation of ex situ aqueous mineral carbonation. Albany Research Center.
- Obuzor, Kinuthia, & Robinson. (2012). Soil stabilisation with lime-activated-GGBS—A mitigation to flooding effects on road structural layers/embankments constructed on floodplains. *Engineering Geology*, 151, 112–119.

- Okamoto, Ogasawara, Ogawa, & Tsuchiya. (2011). Progress of hydration reactions in olivine–H₂O and orthopyroxenite–H₂O systems at 250°C and vapor-saturated pressure. *Chemical Geology*, 289(3-4), 245–255.
- Okoro, Vogtman, & Yousif. (2011). Consolidation Characteristics of Soils Stabilized with Lime, Coal Combustion Product, and Plastic Waste. *Geo-Frontiers*, 1202–1209.
- Olajire. (2013). A review of mineral carbonation technology in sequestration of CO₂. *Journal of Petroleum Science and Engineering*, 109, 364–392.
- Olaniyan, & Olaoye. (2011). Soil Stabilization Techniques Using Sodium Hydroxide Additives. *International Journal of Civil & Environmental Engineering IJCEE-IJENS*, 11(06), 9–22.
- Olivine distribution resources. (2013).
- Olsson, Bovet, Makovicky, Bechgaard, Balogh, & Stipp. (2012). Olivine reactivity with CO₂ and H₂O on a microscale: Implications for carbon sequestration. *Geochimica et Cosmochimica Acta*, 77, 86–97.
- Pacheco-Torgal, Castro-Gomes, & Jalali. (2008a). Alkali-activated binders: A review Part 1. Historical background, terminology, reaction mechanisms and hydration products. *Construction and Building Materials*, 22(7), 1305–1314.
- Pacheco-Torgal, Castro-Gomes, & Jalali. (2008b). Alkali-activated binders: A review. Part 2. About materials and binders manufacture. *Construction and Building Materials*, 22(7), 1315–1322.
- Pakbaz, & Alipour. (2012). Influence of cement addition on the geotechnical properties of an Iranian clay. *Applied Clay Science*, 67-68, 1–4.
- Park, S.-S. (2011). Unconfined compressive strength and ductility of fiber-reinforced cemented sand. *Construction and Building Materials*, 25(2): 1134-1138.
- Palomo, Grutzeck, & Blanco. (1999). Alkali-activated fly ashes: a cement for the future. *Cement and Concrete Research*, 29, 1323–1329.
- Pedarla, Chittoori, Puppala, Hoyos, & Saride, Sireesh. (2010). Influence of Lime Dosage on Stabilization Effectiveness of Montmorillonite Dominant Clays. *Advances in Analysis, Modeling & Design*, 767–776.
- Petermann, Saeed, & Hammons. (2010). Alkali-Activated Geopolymers: A Literature Review. Panama City, FL.
- Petry, & Little. (2002). Review of Stabilization of Clays and Expansive Soils in Pavements and Lightly Loaded Structures—History, Practice, and Future. *Journal of Materials in Civil Engineering*, 14(6), 447–460.

- Pourakbar, Asadi, Huat, & Fasihnikoutalab. (2015). Stabilization of clayey soil using ultrafine palm oil fuel ash (POFA) and cement. *Transportation Geotechnics*, 3, 24–35.
- Prigiobbe, V., Hänchen, Costa, Baciocchi, & Mazzotti. (2009). Analysis of the effect of temperature, pH, CO₂ pressure and salinity on the olivine dissolution kinetics. *Energy Procedia*, 1(1), 4881–4884.
- Prigiobbe, V., Hänchen, Werner, Baciocchi, & Mazzotti. (2009). Mineral carbonation process for CO₂ sequestration. *Energy Procedia*, 1, 4885–4890.
- Prigiobbe, Valentina, & Mazzotti. (2011). Dissolution of olivine in the presence of oxalate, citrate, and CO₂ at 90°C and 120°C. *Chemical Engineering Science*, 66(24), 6544–6554.
- Provis. (2014). Geopolymers and other alkali activated materials: why, how, and what? *Materials and Structures*, 47, 11–25.
- Provis, & Van Deventer. (2009). *Geopolymers structure, processing, properties and industrial applications*. New Delhi: Woodhead publishing limited, Oxford Cambridge.
- Pincus, H., Maher, M., and Ho, Y. (1993). Behavior of fiber-reinforced cemented sand under static and cyclic loads. *Geotechnical and Geological Engineering*, 3(5): 110-127.
- Rahman. (1987). Effects of cement-rice husk ash mixtures on geotechnical properties of lateritic soils. *Soils and Foundations*, 27(2), 61–65.
- Rahmat, & Kinuthia. (2011). Effects of mellowing sulfate-bearing clay soil stabilized with wastepaper sludge ash for road construction. *Engineering Geology*, 117(3-4), 170–179.
- Rajamma, Labrincha, & Ferreira. (2012). Alkali activation of biomass fly ash–metakaolin blends. *Fuel*, 98, 265–271.
- Rajasekaran, & Narasimha Rao. (2000). Strength characteristics of lime-treated marine clay. *Proceedings of the ICE - Ground Improvement*, 4(3), 127–136.
- Rashad. (2013). Alkali-activated metakaolin: A short guide for civil Engineer – An overview. *Construction and Building Materials*, 41, 751–765.
- Reig, Soriano, Borrachero, Monzó, & Payá. (2014). Influence of the activator concentration and calcium hydroxide addition on the properties of alkali-activated porcelain stoneware. *Construction and Building Materials*, 63, 214–222.
- Roy, D. M., & Silsbee. (1991). *Alkali Activated Cementitious Materials: An Overview*. Materials Research Society, 245, 153–164.

- Roy, Della M. (1999). Alkali-activated cements Opportunities and challenges. *Cement and Concrete Research*, 29(2), 249–254.
- Ryu, Lee, Koh, & Chung. (2013). The mechanical properties of fly ash-based geopolymer concrete with alkaline activators. *Construction and Building Materials*, 47, 409–418.
- Sabat. (2012). Stabilization of expansive soil using waste ceramic dust. *Electronic Journal of Geotechnical Engineering*, 17, 3915–3926.
- Saldi, Daval, Morvan, & Knauss. (2013). The role of Fe and redox conditions in olivine carbonation rates: An experimental study of the rate limiting reactions at 90 and 150°C in open and closed systems. *Geochimica et Cosmochimica Acta*, 118, 157–183.
- Salih, Abang Ali, & Farzadnia. (2014). Characterization of mechanical and microstructural properties of palm oil fuel ash geopolymer cement paste. *Construction and Building Materials*, 65, 592–603.
- Sargent, Hughes, Rouainia, & White. (2013). The use of alkali activated waste binders in enhancing the mechanical properties and durability of soft alluvial soils. *Engineering Geology*, 152(1), 96–108.
- Sariosseiri, & Muhunthan. (2009). Effect of cement treatment on geotechnical properties of some Washington State soils. *Engineering Geology*, 104(1-2), 119–125.
- Sazonov, Kaiser, Heger, Meven, & Hutanu. (2009). Crystal and magnetic structure of Co_2SiO_4 olivine.
- Schuiling, R. (2001). Olivine, the miracle mineral. *Mineral. Journ. (Ukraine)*, 23(5/6), 81–83.
- Schuiling, R. D. (2013). Olivine: a supergreen fuel. *Energy, Sustainability and Society*, 3(1), 18.
- Schuiling, R., & Praagman. (2011). Olivine Hills: Mineral Water Against Climate Change. *Engineering Earth*, 1–6.
- Seifritz. (1990). CO_2 disposal by means of silicates. *Nature*, 345(6275), 486.
- Shaikh. (2013). Review of mechanical properties of short fibre reinforced geopolymer composites. *Construction and Building Materials*, 43, 37–49.
- Shand. (2006). *The chemistry and technology of magnesia*. Hoboken, New Jersey: John Wiley & Sons, Inc.
- Sipilä, Teir, & Zevenhoven. (2008). Carbon dioxide sequestration by mineral carbonation Literature review update 2005–2007. Åbo Akademi University.

- Sivakumar Babu, G., and Vasudevan, A. (2008). Strength and stiffness response of coir fiber-reinforced tropical soil. *Journal of materials in civil engineering*, 20(9): 571-577.
- Slaty, Khoury, Wastiels, & Rahier. (2013). Characterization of alkali activated kaolinitic clay. *Applied Clay Science*, 75-76, 120–125.
- Smith, & Stenstrom. (1965). Chemical analysis of olivines by the electron microprobe. *Mineralogical Magazine*.
- Song, Song, Lee, & Yang. (2014). Carbonation Characteristics of Alkali-Activated Blast-Furnace Slag Mortar. *Advances in Materials Science and Engineering*, 2014, 1–11.
- Sreekanth Chakradhar, R., Nagabhushana, B., Chandrappa, G., Ramesh, K., and Rao, J. (2006). Solution combustion derived nanocrystalline macroporous wollastonite ceramics. *Materials chemistry and Physics*, 95(1): 169-175.
- Sridharan, Bhaskar Raju, & Sivapullaiah. (2000). Role of amount and type of clay in the lime stabilization of soils. *Proceedings of the ICE - Ground Improvement*, 4(1), 37–45.
- Stavridakis. (2006). A Solution to the Problem of Predicting the Suitability of Silty–clayey Materials for Cement-stabilization. *Geotechnical and Geological Engineering*, 24(2), 379–398.
- Stevenson, & Sagoe-Crentsil. (2005). Relationships between composition, structure and strength of inorganic polymers : Part 2 Fly ash-derived inorganic polymers. *Journal of Materials Science*, 40(16), 4247–4259.
- Swanson, & Tatge. (1951). Standard X-ray diffraction patterns. *National Bureau of Standard Bull*, 4(4), 318–327.
- Tahir, MUSTA, & Rahim. (2010). Geological heritage features of Tawau volcanic sequence, Sabah. *Bulletin of the Geological Society of Malaysia*, 56, 79–85.
- Tang, C., Shi, B., Gao, W., Chen, F., and Cai, Y. (2007). Strength and mechanical behavior of short polypropylene fiber reinforced and cement stabilized clayey soil. *Geotextiles and Geomembranes*, 25(3): 194-202.
- TARGET MAP. (2012). Olivine distribution resources.
- Temuujin, Okada, & MacKenzie. (1998). Role of Water in the Mechanochemical Reactions of MgO–SiO₂ Systems. *Journal of Solid State Chemistry*, 138, 169–177.
- Temuujin, Williams, & van Riessen. (2009). Effect of mechanical activation of fly ash on the properties of geopolymer cured at ambient temperature. *Journal of Materials Processing Technology*, 209(12-13), 5276–5280.

- The cement sustainability initiative. (2002). world Business Council for Sustainable Development. Florida.
- Thwe, M.M., and Liao, K. (2002). Effects of environmental aging on the mechanical properties of bamboo–glass fiber reinforced polymer matrix hybrid composites. *Composites Part A: Applied Science and Manufacturing*, 33(1): 43-52.
- Turner, & Collins. (2013). Carbon dioxide equivalent (CO₂-e) emissions: A comparison between geopolymer and OPC cement concrete. *Construction and Building Materials*, 43, 125–130.
- Uddin, Balasubramaniam, & Bergado. (1997). Engineering behaviour of cement-treated Bangkok soft clay. *Proceedings of the Institution of Civil Engineers - Geotechnical Engineering*, 28(1), 89–119.
- Unluer, & Al-Tabbaa. (2013). Impact of hydrated magnesium carbonate additives on the carbonation of reactive MgO cements. *Cement and Concrete Research*, 54, 87–97.
- van Deventer, Provis, Duxson, & Lukey. (2007). Reaction mechanisms in the geopolymeric conversion of inorganic waste to useful products. *Journal of Hazardous Materials*, 139(3), 506–13.
- Van Jaarsveld, J. G S, Van Deventer, & Lukey. (2003). The characterisation of source materials in fly ash-based geopolymers. *Materials Letters*, 57(7), 1272–1280.
- van Jaarsveld, J. G. ., van Deventer, & Lukey. (2002). The effect of composition and temperature on the properties of fly ash- and kaolinite-based geopolymers. *Chemical Engineering Journal*, 89(1-3), 63–73.
- Van Jaarsveld, J.G.S., Van Deventer, & Lorenzen. (1997). The potential use of geopolymeric materials to immobilise toxic metals: Part I. Theory and applications. *Minerals Engineering*, 10(7), 659–669.
- Vandeperre, Liska, & Al-Tabbaa. (2008a). Hydration and Mechanical Properties of Magnesia, Pulverized Fuel Ash, and Portland Cement Blends. *Journal of Materials in Civil Engineering*, 20(5), 375–383.
- Vandeperre, Liska, & Al-Tabbaa. (2008b). Microstructures of reactive magnesia cement blends. *Cement and Concrete Composites*, 30(8), 706–714.
- Veld, Roskam, & Enk. (2008). Desk study on the feasibility of CO₂ sequestration by mineral carbonation of olivine. Netherlands.
- Walker, & Pavía. (2010). Physical properties and reactivity of pozzolans, and their influence on the properties of lime–pozzolan pastes. *Materials and Structures*, 44(6), 1139–1150.

- Wang, H., Li, & Yan. (2005). Synthesis and mechanical properties of metakaolinite-based geopolymer. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 268(1-3), 1–6.
- Wang, S. D., & Scrivener. (2003). ²⁹Si and ²⁷Al NMR study of alkali-activated slag. *Cement and Concrete Research*, 33, 769–774.
- Weil, Dombrowsk, & Buchawald. (2009). Life-cycle analysis of geopolymers. *Geopolymers, structure, processing, properties and applications*. Woodhead Publishing Ltd, 5(3), 194–210.
- Weng, & Sago-Crentsil. (2007a). Dissolution processes, hydrolysis and condensation reactions during geopolymer synthesis: Part II-Low Si/Al ratio systems. *Advances In Geopolymer Science & Technology. Journal of Materials Science.*, 42(9), 3007–3014.
- Weng, & Sago-Crentsil. (2007b). Dissolution processes, hydrolysis and condensation reactions during geopolymer synthesis: Part I—Low Si/Al ratio systems. *Advances In Geopolymer Science & Technology. Journal of Materials Science*, 42(9), 2997–3006.
- Wild, & Tasong. (1999). Influence of ground granulated blastfurnace slag on the sulphate resistance of lime-stabilized kaolinite. *Magazine of Concrete Research*, 51(04), 247–254.
- Wilkinson, Haque, Kodikara, Adamson, & Christie. (2010). Improvement of Problematic Soils by Lime Slurry Pressure Injection: Case Study. *Journal of Geotechnical and Geoenvironmental Engineering*, 136(10), 1459–1468.
- William Carey, Peter C. Lichtner, Elias P. Rosen, Hans J. Ziock, and George D. Guthrie. (1995). *Geochemical Mechanisms of Serpentine and Olivine Carbonation*. Los Alamos National laboratory.
- Winnefeld, Leemann, Lucuk, Svoboda, & Neuroth. (2010). Assessment of phase formation in alkali activated low and high calcium fly ashes in building materials. *Construction and Building Materials*, 24(6), 1086–1093.
- Wogelius, RA, & Walther. (1991). Olivine dissolution at 25° C: Effects of pH, CO₂, and organic acids. *Geochimica et Cosmochimica Acta*, 55, 943–954.
- Wogelius, Roy a., & Walther. (1992). Olivine dissolution kinetics at near-surface conditions. *Chemical Geology*, 97(1-2), 101–112.
- Worrell, Price, Martin, Hendriks, & Meida. (2001). Carbon Dioxide Emissions from the Global Cement Industry. *Annual Review of Energy and the Environmen*, 21, 303–329.

- Xeidakis. (1996). Stabilization of swelling clays by Mg (OH) 2. Changes in clay properties after addition of Mg-hydroxide. *Engineering Geology*, 44(1-4), 107–120.
- Xu, & Van Deventer. (2000). The geopolymerisation of alumino-silicate minerals. *International Journal of Mineral Processing*, 59(3), 247–266.
- Yang, Song, Ashour, & Lee. (2008). Properties of cementless mortars activated by sodium silicate. *Construction and Building Materials*, 22(9), 1981–1989.
- Yi, Y. ., Liska, Unluer, & Al-Tabbaa. (2013). Initial investigation into the carbonation of MgO for soil stabilisation. In 18th International Conference on Soil Mechanics and Geotechnical Engineering, Paris (Vol. 5, pp. 2641–2644).
- Yi, Y., Li, Liu, & Asce. (2010). Alkali-Activated Ground-Granulated Blast Furnace Slag for Stabilization of Marine Soft Clay. *Journal of Materail in Civil Engineering*, 11(4), 246–250.
- Yi, Y., Liska, Akinyugha, Unluer, & Al-Tabbaa. (2013). Preliminary Laboratory-Scale Model Auger Installation and Testing of Carbonated Soil-MgO Columns. *Geotechnical Testing Journal*, 36(3), 1–10.
- Yi, Y., Liska, & Al-Tabbaa. (2014). Properties of Two Model Soils Stabilised with Different Blends and Contents of GGBS, MgO, Lime and PC. *Journal of Materials in Civil Engineering*, 26(2), 267–274.
- Yi, Y., Liska, Unluer, & Al-Tabbaa. (2013). Carbonating magnesia for soil stabilization. *Canadian Geotechnical Journal*, 50(8), 899–905.
- Yip, C. K., & Van Deventer. (2003). Microanalysis of calcium silicate hydrate gel formed within a geopolymeric binder. *Journal of Materials Science*, 38(18), 3851–3860.
- Yip, Christina K., Lukey, Provis, & van Deventer. (2008). Effect of calcium silicate sources on geopolymerisation. *Cement and Concrete Research*, 38, 554–564.
- Yip, Christina K., Provis, Lukey, & van Deventer. (2008). Carbonate mineral addition to metakaolin-based geopolymers. *Cement and Concrete Composites*, 30(10), 979–985.
- Yu, Y., Yin, J., and Zhong, Z. (2006). Shape effects in the Brazilian tensile strength test and a 3D FEM correction. *International journal of rock mechanics and mining sciences*, 43(4): 623-627.
- Yunsheng, Z., Wei, S., Zongjin, L., Xiangming, Z., and Chungkong, C. (2008). Impact properties of geopolymer based extrudates incorporated with fly ash and PVA short fiber. *Construction and Building Materials*, 22(3): 370-383.

- Yusuf, Megat Johari, Ahmad, & Maslehuddin. (2014). Evolution of alkaline activated ground blast furnace slag-ultrafine palm oil fuel ash based concrete. *Materials & Design*, 55, 387–393.
- Zevenhoven. (2002). Direct dry mineral carbonation for CO₂ emissions reduction in Finland. *Proc. of the 27th International Technical Conference on Coal Utilization & Fuel Systems*, 1–12.
- Zhang, M., El-Korchi, Zhang, Liang, & Tao. (2014). Synthesis factors affecting mechanical properties, microstructure, and chemical composition of red mud-fly ash based geopolymers. *Fuel*, 134, 315–325.
- Zhang, M., Guo, El-Korchi, Zhang, & Tao. (2013). Experimental feasibility study of geopolymer as the next-generation soil stabilizer. *Construction and Building Materials*, 47, 1468–1478.
- Zhang, S., Karato, Fitz Gerald, Faul, & Zhou. (2000). Simple shear deformation of olivine aggregates. *Tectonophysics*, 316(1-2), 133–152.
- Zhang, Z., Wang, Zhu, Reid, Provis, & Bullen. (2014). Using fly ash to partially substitute metakaolin in geopolymer synthesis. *Applied Clay Science*, 88-89, 194–201.
- ZHAO, & ZHAI. (2013). Leaching behavior mechanism of Mg₂SiO₄ in high NaOH content system. *The Chinese Journal of Nonferrous Metals*, 23(6), 1764–1768.
- Zhu, Ye, Liu, & Yang. (2013). Evaluation on Hydration Reactivity of Reactive Magnesium Oxide Prepared by Calcining Magnesite at Lower Temperatures. *Industrial & Engineering Chemistry Research*, 52(19), 6430–6437.

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My native language is Arabic. I'm married since 2002. I have four children.

LIST OF PUBLICATIONS

Wisam Dheyab, Afshin Asadi, Bujang B.K. Huat, Mohd Saleh Jaafar, and Lokmane Abdeldjouad (2018) "Soil Stabilized with Geopolymers for Low Cost and Environmentally Friendly Construction" GEOMATE conf. (accepted).

Lokmane Abdeldjouad, Afshin Asadi, Bujang B.K. Huat, Mohd Saleh Jaafar, and Wisam Dheyab (2018) "Effect of Curing Temperature on The Development of Hard Structure of Alkali-Activated Soil" GEOMATE conf. (accepted).

Wisam Dheyab, Afshin Asadi, Bujang Kim Huat, Haslinda N., Lokmane Abdeldjouad, and Ahmed Giuma Elkhebu (2018) "Application of Alkali-Activated Olivine Reinforced with Glass Fibers in Soil Stabilization" (in process).



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