



UNIVERSITI PUTRA MALAYSIA

**OPTICAL DETECTION SYSTEM FOR FREE FATTY ACIDS IN CRUDE
PALM OIL BASED ON N-BROMOSUCCINAMIDE CATALYST**

NORSYAMIMI BINTI CHE SULAMAN

ITMA 2019 11



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PALM OIL BASED ON N-BROMOSUCCINAMIDE CATALYST**

By

NORSYAMIMI BINTI CHE SULAMAN

**Thesis Submitted to the School of Graduate Studies, Universiti Putra
Malaysia, in Fulfilment of the Requirements for the Degree of
Master of Science**

January 2019

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Master of Science

OPTICAL DETECTION SYSTEM FOR FREE FATTY ACIDS IN CRUDE PALM OIL BASED ON N-BROMOSUCCINAMIDE CATALYST

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January 2019

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Free fatty acids (FFAs) content in crude palm oil (CPO) is one of parameter that dictated the price of palm oil product. In this study, an alternative optical technique system on detection of FFAs in CPO based on chemical pathway using base catalyst was investigated in order to enhance the current method that practices in industrial with more advantageous in term of using time saving, greener practices and environmentally friendly. FFAs in CPO involved a direct conversion of FFA to form fatty hydroxamic acid (FHA) with hydroxylamine in order to increase its detection prior to coloured metal complexes formation. Coloured metal complexes are an indicator in optical detection using UV-Visible (UV-Vis) spectrophotometer analysis. However, the colour changes of the FHA-V(V) complexation was observe by naked eye before characterize using the UV-Vis spectrophotometer for intensity of the complex. Before this, there was no previous study that study the detection of the FFA in CPO via chemical pathway. A rapid response time of 5 min reaction with one third of mass reactant reduce from initial mass was achieved in conversion of FFAs to FHAs in presence of N-Bromosuccinamide (NBS) as base catalyst and Vanadium (V) (V(V)) as metal ion were used. The reaction was carried out under optimize temperature at 70 °C and 150 rpm shaking. The absorptivity of FHA complexes was detected at 430 nm (range 420 nm to 440 nm) using UV-Vis spectrophotometer. From studies, it can be concluded that detection of the FFA in CPO was successful. The result from develop method was validated with standard titration method by Malaysian Palm Oil Board (MPOB) using spiked CPO sample. Optimization studies were investigated in order to achieve an optimum condition for each parameter for best monitoring system. A linear calibration plot correlation coefficients (R^2) is 0.9902 which able to detect down to 0.7813 % of FFA in CPO. A good correlation was obtained between proposed method and MPOB standard titration method ($R=0.9157$). Paired t-test was used to determine the significant difference between these two methods. According to paired t-test

calculation, there is no significant difference between these two methods at 95% confident level (calculated $t=1.099$, tabulated $t=2.776$).



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sebagai memenuhi keperluan untuk ijazah Master Sains

SISTEM PENGESANAN ASID LEMAK BEBAS DALAM MINYAK SAWIT MENTAH BERDASARKAN PEMANGKIN N-BROMOSUCCINAMIDE

Oleh

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Kandungan asid lemak bebas (FFAs) dalam minyak sawit mentah (CPO) adalah salah satu parameter yang menentukan harga produk minyak sawit. Dalam kajian ini, satu sistem teknik optik alternatif untuk mengesan FFAs dalam CPO berdasarkan laluan kimia menggunakan pemangkin alkali telah disiasat untuk meningkatkan kaedah semasa yang diamalkan dalam industri dengan lebih berfaedah dari segi menggunakan penjimatan masa, amalan hijau dan mesra alam. FFAs dalam CPO melibatkan penukaran langsung FFA untuk membentuk asid hydroxamic lemak (FHA) dengan hidrokslamin untuk meningkatkan pengesanan sebelum pembentukan kompleks logam berwarna. Kompleks logam berwarna adalah penunjuk dalam pengesanan optik menggunakan analisis spektrofotometer UV-Visible (UV-Vis). Walau bagaimanapun, perubahan warna kompleks FHA-V (V) diamati dengan mata kasar sebelum mencirikan menggunakan spektrofotometer UV-Vis untuk pengesanan kompleks. Sebelum ini, tiada lagi kajian yang mengkaji pengesanan FFA dalam CPO melalui laluan kimia. Kajian ini telah hanya memerlukan masa yang singkat iaitu selama 5 dengan menggunakan satu pertiga berat jisim dalam penukaran FFAs kepada FHAs dengan kehadiran N-Bromosuccinamide (NBS) sebagai pemangkin asas dan Vanadium (V) (V (V)) sebagai ion logam digunakan. Reaksi itu dilakukan di bawah suhu mengoptimumkan di 70 ° C dan 150 rpm goncangan. Penyerapan kompleks FHA dikesan pada 430 nm (jarak 420 nm hingga 440 nm) menggunakan spektrofotometer UV-Vis. Dari kajian, dapat disimpulkan bahawa pengesanan FFA di CPO berjaya. Hasil dari kaedah pembangunan telah disahkan dengan kaedah titrasi oleh Lembaga Minyak Sawit Malaysia (MPOB) menggunakan sampel CPO tambahan. Kajian pengoptimuman telah diselidiki untuk mencapai keadaan optimum bagi setiap parameter untuk sistem pemantauan yang terbaik. Koefisien korelasi plot penentukuran linear (R²) adalah 0.9902 yang dapat mengesan turun kepada 0.7813% FFA dalam CPO. Satu korelasi yang baik diperolehi antara kaedah

yang dicadangkan dan kaedah titrasi piawai MPOB ($R = 0.9157$). Ujian t-pasangan digunakan untuk menentukan perbezaan yang ketara antara kedua-dua kaedah ini. Berdasarkan pengiraan t-test berpasangan, tidak terdapat perbezaan yang signifikan antara kedua-dua kaedah pada tahap yakin 95% (dikira $t = 1.099$, ditala $t = 2.776$).



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Best Regards
Norsyamimi Che Sulaman

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TABLE OF CONTENTS

| | Page |
|--|-----------|
| ABSTRACT | i |
| ABSTRAK | iii |
| ACKNOWLEDGEMENTS | v |
| APPROVAL | vi |
| DECLARATION | viii |
| LIST OF TABLES | xiii |
| LIST OF FIGURES | xiv |
| LIST OF SCHEMES | xvi |
| LIST OF ABBREVIATIONS | xvii |
| CHAPTER | |
| 1 INTRODUCTION | 1 |
| 1.1 Background | 1 |
| 1.2 Problem Statement | 2 |
| 1.3 Justification | 3 |
| 1.4 Objective of Study | 4 |
| 1.4.1 General Objective | 4 |
| 1.4.2 Specific Objectives | 4 |
| 1.5 Scope of Work | 4 |
| 2 LITERATURE REVIEW | 5 |
| 2.1 Palm Oil and Its Composition | 5 |
| 2.2 Free Fatty Acids (FFA) | 6 |
| 2.3 Developed Technique Monitoring FFA Detection | 7 |
| 2.3.1 Titrimetric method | 8 |
| 2.3.2 Fourier-Transform Infrared (FTIR) Spectroscopy | 9 |
| 2.3.3 Near-Infrared Reflectance (NIR) Spectroscopy | 12 |
| 2.3.4 Gas Chromatography (GC) | 13 |
| 2.3.5 High-Performance Liquid Chromatography (HPLC) | 15 |
| 2.3.6 Flow Injection Analysis (FIA) | 17 |
| 2.4 Optical Chemical Sensor | 20 |
| 2.5 Fatty Hydroxamic Acid | 22 |
| 2.5.1 Synthesis of Fatty Hydroxamic Acid (FHA) | 22 |
| 3 METHODOLOGY | 25 |
| 3.1 Materials and Reagents | 25 |
| 3.2 Instrumentation | 26 |
| 3.2.1 Nanodrop2000/2000C Spectrophotometer (Thermo Scientific) | 26 |
| 3.2.2 Water bath Shaker (Memmert) | 26 |
| 3.3 Preparation of Reagents | 27 |
| 3.3.1 Preparation of Base Catalysts | 27 |

| | | |
|----------|--|-----------|
| 3.3.1.1 | Sodium Hydroxide (NaOH) (1% w/v) | 27 |
| 3.3.1.2 | N-Bromosuccinimide (NBS) (1% w/v) | 27 |
| 3.3.2 | Preparation of Metals Ion Solution | 27 |
| 3.3.2.1 | Vanadium (V) Oxide Solution (0.1 M V ₂ O ₅) | 27 |
| 3.3.2.2 | Copper (II) Chloride Solution (0.1M CuCl ₂) | 27 |
| 3.3.2.3 | Iron (III) Chloride Solution (0.1M FeCl ₃) | 28 |
| 3.3.3 | Preparation of Standard Potassium Hydroxide (KOH) | 28 |
| 3.3.3.1 | Standardized of Potassium Hydroxide | 28 |
| 3.4 | Synthesis of Fatty Hydroxamic Acid (FHA) and Complexation Reaction | 28 |
| 3.4.1 | Synthesis of Fatty Hydroxamic Acid (FHA) | 28 |
| 3.4.2 | Complexation Reaction | 29 |
| 3.5 | Optimization of FHA Synthesis | 29 |
| 3.5.1 | Effect of Organic Solvent | 30 |
| 3.5.2 | Effect of Mass Reactant Reduction | 30 |
| 3.5.3 | Effect of Varying Base Catalyst | 31 |
| 3.5.4 | Effect of Reaction Time | 31 |
| 3.5.5 | Effect of Reaction Temperature | 32 |
| 3.6 | Selection of Metal Ion for Complexation | 32 |
| 3.7 | Standardization of FFA Standard Solution with MPOB Official Method | 33 |
| 3.7.1 | Preparation of Spiked Crude Palm Oil Stock Solution | 33 |
| 3.7.2 | Preparation of Palmitic Acid Stock Solution (100 a.d) | 33 |
| 3.8 | Method for Calibration and Validation based on Developed Method and MPOB Standard Method | 34 |
| 3.8.1 | Titrimetry Method (MPOB Standard Method) | 34 |
| 3.8.2 | Developed Method | 34 |
| 4 | RESULT AND DISCUSSION | 35 |
| 4.1 | FFA Detection by FHA Complexation Based on Chemical Pathway Utilizing Optical Technique | 35 |
| 4.2 | Synthesis Parameter | 38 |
| 4.2.1 | Effect of Organic Solvent | 38 |
| 4.2.2 | Effect of Mass Reactant Reduction | 39 |
| 4.2.3 | Effect of Varying Base Catalysts | 40 |
| 4.2.4 | Effect of Reaction Time | 42 |
| 4.2.5 | Effect of Reaction Temperature | 44 |
| 4.2.6 | Effect of Metal Ions | 46 |
| 4.2.7 | Optimization of parameter analysis | 53 |
| 4.2.8 | Reproducibility | 53 |
| 4.3 | Standardization of Palm Oil Sample Using Standard Titration Method | 54 |
| 4.3.1 | Calibration Plot | 55 |

| | | |
|----------|--|-----------|
| 4.3.2 | Validation Study for the Proposed Method | 56 |
| 5 | CONCLUSION | 58 |
| 5.1 | Recommendations | 59 |
| | REFERENCE | 60 |
| | APPENDICES | 68 |
| | BIODATA OF STUDENT | 73 |
| | PUBLICATION | 74 |



LIST OF TABLES

| Table | | Page |
|-------|--|------|
| 2.1 | Previous study of FTIR analysis | 10 |
| 2.2 | Previous studies of detection of FFA using NIR analysis | 12 |
| 2.3 | Summarize of different types of detector used in GC analysis | 14 |
| 2.4 | Comparative advantageous and disadvantageous of method in detection of FFA | 19 |
| 3.1 | Reagents were used in this research | 25 |
| 3.2 | A series of a standard solution of spiked for working calibration curve | 33 |
| 4.1 | Mass reduction of reactant by mass fraction | 40 |
| 4.2 | The summarize of optimized parameters and selection of metal for FFA detection | 53 |
| 4.3 | Reproducibility Study for FHA complexes | 54 |
| 4.4 | Percentage of FFA in CPO by standard titration method | 54 |
| 4.5 | The percentage of FFA by varying concentration of spiked CPO | 55 |
| 4.6 | The percentage of FFA from Developed method | 55 |
| 4.7 | Paired t-test analysis for the comparative study of the proposed method | 57 |

LIST OF FIGURES

| Figure | | Page |
|--------|--|------|
| 2.1 | A cross-section of oil palm fruit. A fruit comprises of fleshly part or pericarp (exocarp, mesocarp and endocarp) and endosperm | 5 |
| 2.2 | Block diagram of FFA detection methods in oils and fats | 8 |
| 2.3 | Basic principle stage in the operation of sensor | 20 |
| 2.4 | General arrangement of spectroscopic measurements: A – light reflection, B– light refraction, C – light absorption, D – light emission | 21 |
| 2.5 | General structure of hydroxamic acid | 22 |
| 3.1 | Flow chart diagram on a developed method for detection of FFA | 29 |
| 4.1 | Comparison of spectra between CPO (3% FFA), FHA,V(V) (0.1M) and FHA-V(V) complex after reaction of FHA with V(V) at equal volume (200 μ L). Reaction conditions: reaction time 45 min, temperature 70 $^{\circ}$ C, agitation 150 rpm and using NBS catalyst | 37 |
| 4.2 | A preliminary studies on selection of organic solvent for dissolving CPO in hexane and isopropanol | 39 |
| 4.3 | The coloured FHA Vanadium complexes using NaOH and NBS catalyst with reaction condition as follows; 45 min reaction time, 70 $^{\circ}$ C reaction temperature and 150 rpm agitation | 41 |
| 4.4 | Effect of different base catalyst on FHA Vanadium complexes with reaction condition as follows; 45 min, NBS catalyst, 70 $^{\circ}$ C reaction temperature and 150 rpm agitation | 42 |
| 4.5 | Effect of reaction time on FHA Vanadium complexes for time intervals 5, 10, 15, 30 and 45 min with reaction condition as follows; 1% NBS catalyst, 70 $^{\circ}$ C reaction temperature and 150 rpm agitation | 43 |
| 4.6 | Blue coloured metal complexes after reaction with V(V) with condition as follows 1% NBS catalyst, 70 $^{\circ}$ C reaction temperature and 150 rpm agitation | 44 |

| | | |
|------|---|----|
| 4.7 | A) FHA Vanadium complexes after 30 s shaking (B) FHA Vanadium complexes after being left overnight | 45 |
| 4.8 | Effect of reaction temperature by varying from 40 °C to 90 °C with reaction condition as follow; 1% NBS catalyst, 5 min reaction time and 150 rpm agitation time | 46 |
| 4.9 | Tautomerism of hydroxamic acid which is keto (A) and enol (B) coordination | 46 |
| 4.10 | The possible structure of the hydroxamic acid metal complex | 47 |
| 4.11 | Metal ion solution dilute in HCl. From left: (a) FeCl ₃ (Fe ³⁺), (b) CuCl ₂ (Cu ²⁺), and (c) V ₂ O ₅ (V ⁵⁺) | 48 |
| 4.12 | The colour changes of complexation of FHA by different metal ions. (A) FHA reacted with metal ions solution within 30 s shaking (B) FHA-metal ions complex with two layers formation (C) FHA-metal complex after separation | 50 |
| 4.13 | Effect of metal ions by varying metal ion using V(V), Cu(II) and Fe(III) with reaction condition as follow; 1% NBS catalyst, 5 min reaction time and 150 rpm agitation time | 51 |
| 4.14 | The complex structure of FHAs with (a) Fe(III) (Haron <i>et al.</i> , (2012); Moghaddam <i>et al.</i> , (2014)), (b) Cu(II) (Haron <i>et al.</i> , (2012 ^a); Moghaddam <i>et al.</i> , (2014)) and (c) V(V) (Dess <i>et al.</i> , 1992) ions solution | 52 |
| 4.15 | Calibration curve based on derivative spectrophotometric under an optimum condition at a wavelength of 402 nm | 56 |
| 4.16 | The correlation curve between a developed method with MPOB standard method for FFA detection in CPO | 57 |

LIST OF SCHEMES

| Scheme | Page |
|---|-------------|
| 1.1 Principle of detection of free fatty acids (FFAs) content in crude palm oil (CPO) | 3 |
| 2.1 Chemical reaction equation of Blatt procedure | 22 |
| 2.2 The reaction equation of triacylglyceride with hydroxylamine | 23 |
| 4.1 Basic principle of detection of FFAs in CPO using metal ions based on optical detection | 36 |
| 4.2 Propose mechanism of reaction between FFA in CPO with hydroxylamine | 38 |

LIST OF ABBREVIATIONS

| | |
|--------|---|
| AOCS | American Oil Chemist Society |
| BPO | Bleached Palm Oil |
| CPO | Crude Palm Oil |
| DAG | Diacylglyceride |
| DBPO | Distilled Bleach Palm Oil |
| DG | Diglyceride |
| ECD | Electron Capture Detector |
| FFA | Free Fatty Acid |
| FHA | Fatty Hydroxamic Acid |
| FID | Flame Ionization Detector |
| FTIR | Fourier Transform Infrared |
| GC | Gas Chromatography |
| GC-FID | Gas Chromatography-Flame Ionisation Detection |
| GCMS | Gas Chromatography Mass Spectroscopy |
| GLC | Gas Liquid Chromatography |
| HCl | Hydrochloric Acid |
| HPLC | High Performance Liquid Chromatography |
| HRMS | High Resolution Mass Spectrometry |
| KOH | Potassium Hydroxide |
| MAG | Monoacylglyceride |
| MG | Monoglyceride |
| MIR | Mid-Infrared Region |

| | | |
|----------------|--|--------|
| MLR | Multiple Linear Regression | |
| MPOB | Malaysian Palm Oil Board | |
| MS | Mass Spectrometry | |
| NaOH | Sodium Hydroxide | |
| NBS | N-Bromosuccinamide | |
| NIR | Near-Infrared Region | |
| NPD | Nitrogen Phosphorus Detector | |
| PCR | Principal Component Regression | |
| PID | Photo Ionisation Detector | |
| PKO | Palm Kernel Oil | |
| PLS | Partial Least Square | |
| R ² | Correlation Coefficient | |
| RBD | Refluxed Bleached Deodorize | |
| RBPO | Refluxed Bleach Palm Oil | |
| RSD | Relative Standard Deviation | |
| SE | Standard Error | |
| SEC | Standard Error of Calibration | |
| TAG | Triacylglyceride | |
| TG | Triglyceride | |
| TLC | Thin Layer Chromatography | |
| TOTOX | Total Oxidation Value | |
| UHPLC | ULTRA HIGH-PERFORMANCE CHROMATOGRAPHY | LIQUID |
| UV-Vis | UV-Visible | |

CHAPTER 1

INTRODUCTION

1.1 Background

Palm oil industry was mainly focusing on such as cultivating of fruit bunches, processing fresh fruit bunches (FFB) in mills for CPO and PKO, producing refined palm oil (RPO) from CPO, and fractionating palm oil to obtain liquid olein and solid stearin fraction as well as oleochemical products (Choo, 2014). In September 2010, Malaysian government was introduced the Economic Transformation Program (ETP) under the 12 National Key Economic Area (NKEA) to drive the nation's economy. The ETP is a comprehensive effort that outlines a 10-year economic roadmap to reinforce Malaysia toward becoming a high-income nation by 2020. The palm oil industry was one of the industries that been affected positively by this implementation of ETP. The palm oil sector NKEA is aimed at changing the traditional approach by improving upstream productivity, increasing downstream expansion and sustainable development of the oil palm industry (Choo, 2014).

Over decades of the existence of oil palm tree, palm oil industry continues to thrive to produce high-quality edible oil compared to other vegetable oils. Palm oil was first introduced as decorative crops that later regain a lot of benefits to consumer in terms of its nutrition (Basiron, 2007). Palm oil had a deep rich red colour which derived from its rich carotenoid contents, which also known as a pigment that often found in plants and animals. The appearance of palm oil was in viscous semi-solid and solid fat in temperate climate due to a saturated palmitic acid is a major component of its glycerides. Crude palm oil (CPO) and palm kernel oil (PKO) were a major type of palm oil production. Thus, high yield of oil extraction rate (OER) and CPO quality had become an interest to millers. However, there was many hurdles need to overcome to produce good quality of CPO.

A variety of physicochemical oil quality parameters were determined including free fatty acids (FFA), moisture, peroxide value, iodine value, and saponification number. However, FFA was considered as the most influential parameter that indicates a quality of CPO and other vegetable oil; its bad or good as well as its economic value indices. FFA was naturally released in palm oil which is caused by hydrolysis and can be increased by microbial activity (Che Man *et al.*, 1999). During the refining process, removal of FFA as a contaminant in edible oils was indeed, since high contents of FFA will lead to rancidity of the oil and will lower quality of flavour (Keurentjes *et al.*, 1991). According to standard specification of FFA content by Palm Oil Refiner Association of Malaysia, FFA content must not exceed 5% for crude palm oil (CPO) and not less than 0.1% of refined-bleached-

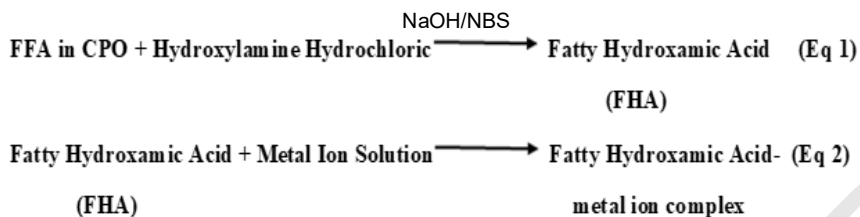
deodorised oil (RBDO) (Che Man *et al.*, 1999). The level content of FFA must be analysing first under this regulation for trading purpose.

The fundamental issues in producing high-quality palm oil were dictated by the high level content of FFA in palm oil. Various strategies and development in the detection of FFA had already established such as using spectrophotometric, amperometric, colorimetric, flow injection methods and others. However, those approaches often employ highly cost maintenance when involving high-performance instrumentation analysis even though in a shorter time, it performed a good analysis. Conventionally, FFA content was determined using manual titration method which carried out by titrating the palm oil sample against potassium hydroxide (KOH) by using phenolphthalein as an indicator (Ainie *et al.*, 2004). The limitations for this method include time-consuming, laborious usage, high solvent usage (isopropanol) as well as easily interrupted prone to human error. Technically, conventional method is needed for a skilful and competence employee in order to get an accurate result for end point determination which may lead to over titration due to the presence of orange colour of carotene compound (Saad *et al.*, 2007) . Thus, new development method in indicating FFA content had to be investigated using simpler way, greener and practical. Direct detection of FFA in vegetable oils especially palm oils based on sensor is not developed yet.

1.2 Problem Statement

During refining process, there are some parameter that need to remove in order to get a good quality of crude palm oil such as the amount of free fatty acid (FFA), phosphatides, odoriferous matter, water and impurities that meets industry standards (Junior *et al.*, 2012). Free fatty acid (FFAs) was the most importance factor that indicates the quality of the palm oil (Corley and Tinker, 2003). Thus, the amount of FFAs in palm oil need to be determined before distribute to the market. With all kinds of reasons and problems arise in the detection of FFA; an alternative method based on the chemical sensor using optical technique was designed using colourimetric detection. Chemical sensors were promising rapid detection, high specification, and sensitivity and selectivity, inexpensive, and easy for users as per requirements. Generally, the FFAs concentration in palm oil was based on the optical observation of the formation of the fatty hydroxamic acid-vanadium (V) (FHA-V(V)) complex, whereby the FFAs were converted to FHAs prior to complexation via aminolysis. Later, the complex characterized using a spectrophotometric method. The principle of the detection of FFAs content in CPO can be summarized as follows in **Scheme**

1.1:



Scheme 1.1 : Principle of detection of free fatty acids (FFAs) content in crude palm oil (CPO)

The equation 1 shows the aminolysis reaction involved in the conversion of FFAs to FHAs and the equation 2 show the complexation step between fatty hydroxamic acid with metal ions will form coloured FHA-metal complex. Rapid detection system was required in this study; thus, all parameter involves during aminolysis reaction were a crucial factor prior the development of the detection system. The parameter studies were included varying base catalyst, time reaction, temperature reaction and metal ions. By considering a sensitivity and selectivity of this chemical sensor, detection of FFA in palm oils is promising.

1.3 Justification

Titrimetric method was done over decades in determination of FFA in crude palm oil. This method was done by titrating the CPO in isopropanol against the KOH solution in presence of phenolphthalein as indicator. The disadvantage of this method was skilful and competence labour needed, high solvent usage and tedious procedure. This is because the CPO contain orange coloured carotene compound in which lead a difficulty in determine the end point of titration and may cause over titration. Basically, for normal titration which the colourless acid against the KOH solution resulted a pale pink coloured as the end point of the analysis. Thus, it is easy to determine the end point of titration. The crucial factor in detection of FFA in CPO by manual titrimetric was complicated due to difficulty to determine the end point of sample.

An alternative method was approach to detect the FFA in more rapid and simpler way to improve the current method on detection of FFA in CPO that had applied in industrial. This study involved aminolysis of the FFA in CPO with hydroxylamine hydrochloric acid convert FFA to FHA. FHAs was reacted with metal ion solution to form coloured FHA-metal complex. Then, coloured FHA-metal complex was observed by naked eyes and characterized using UV-Visible Spectrophotometer to determine the FFA concentration contain based on absorption intensity of FHA-metal complex.

1.4 Objective of Study

1.4.1 General Objective

The general objective of the study was to develop a colourimetric detection for free fatty acids (FFAs) in crude palm oil (CPO) samples utilizing base catalyst via optical technique.

1.4.2 Specific Objectives

The objectives of this research were:

1. To optimize the response time for the conversion of free fatty acids (FFAs) to fatty hydroxamic acids (FHAS) prior the colorimetric detection.
2. To investigate the effect of different base catalyst, effect of reaction temperature, effect of different metal ions and effect of mole ratio of reactant involved on conversion of free fatty acids (FFAs) to fatty hydroxamic acids (FHAs) to obtain the best monitoring.
3. To correlate the optimize parameters with MPOB standard titration method for validation to obtain the best monitoring system.

1.5 Scope of Work

The scope of study was focused to investigate free fatty acids content in crude palm oil (CPO) sample using optical technique utilizing base catalyst. The aim of study is to improve the response time in order to get a rapid analysis toward conversion of FFAs to fatty hydroxamic acids (FHA) in CPO instead of 24 hours reaction by previous synthesis. Other parameters were optimized in order to get a good conversion of FHA. Proposed method was validated with MPOB standard titrimetric method.

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PUBLICATION

Sulaman, N. C., Yusof, N. A., Abdullah, J., & Hajjan, R. (2017). A Novel Base Catalyzed Esterification Reaction for Spectrophotometric Determination of Free Fatty Acid in Crude Palm Oil. *Asian Journal of Chemistry*, 29(4).





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