

UNIVERSITI PUTRA MALAYSIA

REMOVAL OF β-CAROTENE FROM PALM OIL MILL EFFLUENT BY MOLECULAR IMPRINTED POLYMER BASED ON CYCLODEXTRIN- CO-VINYLCARBAZOLE AND CYCLODEXTRIN-CO-HYDROXYETHYL METHACRYLATE

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WARDA MOHAMED ALI ALTOGBIA

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Master of Science

REMOVAL OF β-CAROTENE FROM PALM OIL MILL EFFLUENT BY MOLECULAR IMPRINTED POLYMER BASED ON CYCLODEXTRIN-CO-VINYLCARBAZOLE AND CYCLODEXTRIN-CO-HYDROXYETHYL METHACRYLATE

By

WARDA MOHAMED ALI ALTOGBIA April 2019 Chairman : Professor Nor Azah Yusof, PhD Faculty : Science

Palm oil mill effluent (POME) the most significant pollutant in the form of wastewater, if not managed properly, it could have negative effects on the environment include the emission of biogas and water pollution which comes from discharging the brownish tick POME to the water bodies. Discharge of dark brownish colored of POME into water bodies will prevent the growth of aqua organism by reducing the penetration of sunlight and affect the photosynthetic activity. Such concern provides incentive to remove color from POME. In recent years, molecular imprinted polymers (MIP) attracted wide attention in treatment of wastewater due to their capability for molecular recognition, easiness of preparation, stability and cost-effective production. In this research, MIPs were synthesized by bulk polymerization method using β cyclodextrin (β -CD) as monomer, 9-vinylcarbazole (9VC) and 2-hydroxylethyl methacrylate (HEMA) as co monomers, Toluene2,4-diisocyanate (TDI) as crosslinker, N,N-Dimethylformamide (DMF) as porogen solvent and Benzoyl peroxide (BPO) as initiator. The temperature used during polymerization process was 70°C. The obtained MIP was crushed, sieved and washed to remove the template. Increase the ratio of monomer to cross-linker lead to increase the yield weight of the MIPs. The optimal ratio of monomer to cross-linker was 1:5:58 and 1:17:58 for MIP-β-CD-9VC and MIP-B-CD-HEMA respectively. FTIR result showed that MIP (Molecular Imprinted Polymer) and NIP (Non-Imprinted Polymer) have similar characteristic peak with different peaks intensity, indicating the similarity in the backbone structure of polymerization. TGA result displayed high thermal stability with final decomposition at 320°C and 307°C for MIP-β-CD-9VC and MIP-β-CD-HEMA respectively. FESEM shows that MIP has rough surface while NIP possess densely packed particles with smooth surface. From BET result, the surface area for MIP-β-CD-9VC is higher than NIP-β-CD-9VC, while the surface area for MIP-β-CD-HEMA is smaller compared with NIP- β -CD-HEMA. The pH study shows that sorption of β - carotene increased with decreasing the pH of POME and the maximum sorption capacities achieved at pH 2 were 10 μ g/g and 8 μ g/g for MIP- β -CD-9VC and MIP- β -CD-HEMA, respectively. Also, the sorption of β -carotene increased with increasing the dosage of MIP, and the maximum sorption achieved by using 500 mg of MIP. Both MIP- β -CD-9VC and MIP- β -CD-HEMA followed Freundlich isotherm with R²=0.9688 and R²=0.9817 respectively. Besides that, the sorption of β -carotene by MIPs is fast within few minutes and both MIP- β -CD-9VC and MIP- β -CD HEMA followed the pseudo second order with R²=0.9999. The reusability study for MIP- β -CD-9VC and MIP- β -CD-HEMA were tested five times and no significant loss in sorption capacity. However, MIP- β -CD-HEMA showed higher reusability compared with MIP- β -CD-9VC.

A new type of molecular imprinted polymer as a sorbents for removal of β carotene from palm oil mill effluent (POME) was successfully synthesized. The synthesized MIP- β -CD-9VC and MIP- β -CD-HEMA showed several characteristic such as high thermal stability, fast sorption kinetic and good reusability. Thus, MIP- β -CD-9VC and MIP- β -CD-HEMA can be potentially used as a sorbents for the removal of β -carotene from palm oil mill effluent (POME). Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

PENYINGKIRAN BETA-KAROTEN DARI EFLUEN KILANG MINYAK KELAPA SAWIT OLEH POLIMER TERCETAK MOLEKUL BERDASARKAN CYCLODEXTRIN-CO-VINYLCARBAZOLE DAN CYCLODEXTRIN-CO-HYDROXYETHYL METHACRYLATE

Oleh

WARDA MOHAMED ALI ALTOGBIA April 2019

Pengerusi: Profesor Nor Azah Yusof, PhDFakulti: Science

Efluen kilang minyak kelapa sawit (POME) adalah bahan pencemar yang paling signifikan dalam bentuk air sisa, dan jika ia tidak diuruskan dengan baik, ia boleh memberi kesan negatif terhadap alam sekitar, termasuk pelepasan biogas dan pencemaran air yang terhasil oleh pengeluaran POME tebal berwarna coklat ke dalam air. Kebimbangan sebegitu memberikan insentif kepada usaha untuk menjalankan rawatan terhadap POME. Pelepasan warna gelap POME ke dalam badan airakan menghalang pertumbuhan organisma akuatik dengan megurangkan penembusan cahaya matahari dan menjejaskan aktiviti fotosintesis. Dalam tahun-tahun kebelakangan ini, polimer tercetak molekul (MIP) telah menarik perhatian luas dalam rawatan air sisa disebabkan oleh keupayaannya untuk mengenal pasti molekul, penyediaannya yang mudah, kestabilannya dan keberkesanan kos pengeluarannya. Dalam kajian ini, MIP disintesis dengan kaedah pempolimeran pukal dengan menggunakan β-cyclodextrin (β-CD) sebagai monomer, 9-vinylcarbazole (9VC) dan 2-hydroxylethyl methacrylate (HEMA) sebagai monomer bersama, Toluene 2,4diisocyanate (TDI) sebagai penghubung silang, N,N-Dimethylformamide (DMF) sebagai pelarut porogen dan Benzoyl peroksida (BPO) sebagai pemula. Suhu setinggi 70°C digunakan semasa proses pempolimeran. MIP yang diperolehi dihancurkan, disaring dan dibasuh untuk mengeluarkan templat tersebut. Peningkatan nisbah monomer kepada penghubung silang telah menyebabkan kenaikan dalam berat hasil MIP. Nisbah optimum monomer kepada penghubung silang adalah 1:5:58 dan 1:17:58 untuk MIP-B-CD-9VC dan MIP-B-CD-HEMA masing-masing. Keputusan FTIR menunjukkan bahawa oleh kerana MIP (polimer tercetak molekul) dan NIP (polimer tidak dicetak) mempunyai ciri puncak yang sama walaupun keamatan puncak mereka berbeza, maka ini menunjukkan bahawa struktur tulang belakang pempolimeran mereka adalah sama. Keputusan TGA menunjukkan kestabilan haba yang tinggi dengan penguraian akhir pada 320°C dan 307°C untuk MIP-β-CD-9VC dan MIP-β-



CD-HEMA masing-masing. FESEM menunjukkan bahawa MIP mempunyai permukaan yang kasar manakala NIP mempunyai zarah-zarah yang disusun padat dengan permukaan yang licin. Hasil BET menunjukkan bahawa MIP-β-CD-9VC mempunyai luas permukaan yang lebih tinggi daripada NIP-β-CD-9VC, sementara luas permukaan MIP-B-CD-HEMA adalah lebih kecil dibandingkan dengan NIP-B-CD-HEMA. Kapasiti penyerapan optimum semasa kajian pH telah dicapai pada pH 2 untuk kedua-dua MIP-\beta-CD-9VC dan MIP-β-CD-HEMA. Juga, penyerapan βkaroten telah meningkat dengan kenaikan dos MIP, dan penyerapan maksimum yang dicapai dengan menggunakan 500 mg MIP. Kedua-dua MIP-B-CD-9VC dan MIP-B-CD-HEMA telah mengikut isotherm Freundlich dengan $R^2 = 0.9688$ dan $R^2 = 0.9817$ masing-masing. Selain itu, penyerapan ß-karoten oleh MIP adalah cepat dan hanva memakan masa beberapa minit sahaja dan kedua-dua MIP-B-CD-9VC dan MIP-B-CD HEMA telah mengikut urutan kedua pseudokod dengan R²=0.9999. Kajian kebolehgunaan semula untuk MIP-B-CD-9VC dan MIP-B-CD-HEMA telah diuji sebanyak lima kali dan tiada kehilangan kapasiti penyerapan yang ketara berlaku. Walau bagaimanapun, MIP-β-CD-HEMA menunjukkan kebolehgunaan semula yang lebih tinggi berbanding dengan MIP-β-CD-9VC.

Sejenis polimer yang dicetak dengan molekul baru sebagai penyerap untuk penyingkiran β -karotene daripada efluen kilang miyak sawit (POME) telah berjaya disintesis. MIP- β -CD-9VC yang disintesis dan MIP- β -CD-HEMA menunjukan beberapa ciri-ciri seperti kestabilan haba yang tinggi, kinetic penjerapan pantas dan boleh digunkan semula. Oleh itu, MIP- β -CD-9VC dan MIP- β -CD-HEMA berpotensi digunakan sebagai sorben untuk penyingkiran β -karotene daripada efluen kilang miyak sawit (POME).

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This thesis was submitted to the Senate of the Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

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C

LIST OF ABBREVIATIONS

	9VC	9-Vinylcarbazole
	ATR	Attenuated Total Reflectance
	BET	Brunaue Emmett Teller
	BPO	Benzol peroxide
	DMF	Dimethyl formamide
	DTG	Differential Thermogravimetric Analysis
	FESEM	Field Emission Scanning Electron Microscopy
	FTIR	Fourier Transform Infrared
	HEMA	2-Hydroxyethyl Methacrylate
	MIP	Molecular Imprinted Polymer
	MIP-β-CD	Molecular Imprinted Polymer-
	MIP-β-CD-9VC	Molecular Imprinted Polymer-β-Cyclodextrin-9-Vinylcarbazole
	MIP-β-CD-HEMA	Molecular Imprinted Polymer- β-Cyclodextrin-2-Hydroxyethyl Methacrylate
	NIP	Non- Imprinted Polymer
	NIP-β-CD	Non-Imprinted Polymer- β-Cyclodextrin
NIP-β-CD-9VC		Non-Imprinted Polymer- β-Cyclodextrin-9-Vinylcarbazole
	NIP-β-CD-HEMA	Non-Imprinted Polymer-β-Cyclodextrin-2-Hydroxyethyle Methacrylate
	POME	Palm Oil Mill Effluent
	TDI	Tolylene-2,4-diisocyanate
	TGA	Thermal Gravimetric Analysis
	β-CD	β-Cyclodextrin

CHAPTER 1

INTRODUCTION

1.1 Background of the research

The Malaysian palm oil industry is growing rapidly and has become a very important agriculture-based industry. The country is currently one of the largest producers and exporters of palm oil. However, the wet process of palm oil milling consumes a large amount of process water. The production of one tonne of crude palm oil requires approximately 5–7.5 tonnes of water, and more than 50% of the water will end up as palm oil mill effluent (POME) Wu et al. (2007).

POME is acidic and has a pH of between 4 and 5, Chemical Oxygen Demand (COD) of 40,000 -100,000 mg/L, and Biological Oxygen Demand (BOD) of 25,000-60,000 mg/L. The color of the effluent is due to the presence of organic matters such as lignin, phenolics and carotenes (Bello et al. 2013; Ahmed et al. 2015). Palm oil mill effluent (POME) with very dark color will not meet the discharge standard since it has been identified as a major contributor to the pollution load in ground water.

The brownish color of the effluent has raised the public concern regarding its impact on water body. The color pigment present in the effluent has caused colorization of the pipelines in the industry; this reduces possibility of reclamation. As the world continues to find ways to deal with the worsening global water crisis, reclamation of wastewater so that it can be reused in industrial processes would undoubtedly provide an alternative water resource in dealing with problems of water shortage Tan et al. (2017).

Among the technologies which can be used to remove color from POME are advanced oxidation process, membrane filtration, ozonation, ion exchange, and adsorption Zahrim et al. (2014). Among these methods, adsorption is by far the most versatile and frequently used due to its low initial cost, simplicity of design, ease of operation, and insensitivity to toxic. There are number of studies have been conducted to investigate the ability of several adsorbents to remove color from POME, such as activated carbon silica particles and agricultural waste Neoh et al. (2014). However, the ability of these adsorbents to achieve the intended objectives is limited by their selectivity and specificity properties. Furthermore, there has been no report on the reusability of the adsorbents. In the recent past, the usage of molecular imprinted polymers to treat wastewater has been attracting a lot of attention.

Molecularly imprinted polymers (MIPs) are cross-linked polymeric materials that exhibits high selectivity and great affinity for a particular target molecule (template) present during the synthesis process (Asman et al. 2016; Dai et al. 2011). Additionally, MIPs have higher physical strength, robustness, resistance to high pressure and temperature, and inertness against various chemicals (organic solvents, acids, bases, and metal ions). The cost for producing MIPs is low and their life span can be as long as several years at room temperature Cheong et al. (2013). MIPs can be synthesized from a wide range of substances Poma et al. (2010)

The imprinting process occurs in the interaction between a functional monomer and a selected template in a specific solvent which acts as the porogen in the presence of a cross-linker. The template molecule are removed subsequent to polymerization through extensive washing steps to disrupt the interactions between the template and the monomers, thus leaving cavities of binding sites that are complementary to the template in size, shape and position of the functional group. The cavities thus formed can function as selective binding sites for the template molecule (Figueiredo et al. 2016). The resulting MIP can be used in growing number of applications, such as chromatographic separations, solid-phase extraction, and catalysis Xu et al. (2015).

The properties, physical appearance, morphology and the performance of the MIP are determined by reaction conditions such as formulation of MIP's reaction mixture, including choice of cross-linker, functional monomer, porogenic solvent, reaction temperature, and time Cheong et al. (2013).

MIPs can be classified based on whether they are fabricated using the covalent, noncovalent, or semi-covalent method. In the non-covalent imprinting method, which is the most frequently used method; a template is mixed with an appropriate functional monomer, a suitable porogenic solvent, a cross-linker agent and a polymerization initiator. Specific binding sites are formed via self- assembly of the template and the functional monomer, which is capable of forming a fairly stable complex via dipole interaction, hydrophilic interaction, van der Waals forces, hydrogen bonding, ion pair, etc. Following polymerization the template is removed by extracting the polymer with appropriate solvent Kyzas et al. (2014).

1.2 Problem Statement

Palm oil industries have been making significant contribution to the economic growth of South East Asian countries, hence increasing the standard of living of the population in these countries. Unfortunately, it has also been identified as the major sector discharging the largest pollution load in the form of Palm Oil Mills Effluent (POME) Tabassum et al. (2015). POME is considered as one of the most important source of environmental pollution in Malaysia due to the large volume of POME produced by the palm oil mills, along with high concentration of chemical oxygen demand (COD) (65,000 mg/L) and biochemical oxygen demand (BOD) (25,000 mg/L) (Parthasarathy et al. 2016). Among the concerns related to the increasing production of palm oil in Southeast Asia are water pollution, emission of greenhouse gases, and land conversion Ibrahim et al. (2010).



Color remains as one of the major constituents of the discharged effluent, and should be removed for environmental and esthetic reasons. The color of POME not only diminishes the esthetic value of the water bodies, it also interferes with the penetration of light into water, hence causing disturbances to the aquatic ecosystem Zahrim et al. (2017). The discharge of brownish wastewater into water bodies could have adverse effect on aquatic lives since the color filters the light which passes through the water. This will result in reduced photosynthesis activity and less dissolved oxygen. To prevent this impact, color should be removed from POME prior to being discharged into the environment Ratpukdi et al. (2012).

Various methods are available in industries for removal of color from POME such as ozonation, membrane filtration and adsorption using different types of adsorbates. However, there are several disadvantages of these methods for examples ozonation is considered as expensive technique for treatment of POME. Other than that, the major drawback of membrane filtration is fouling. So far, adsorption is an essential process for treatment of POME using commercial activated carbon or waste materials. However, most of these adsorbents are high cost, low reuse, tedious post-processing. Besides, all of these absorbents are applied for non-specific absorption, exhibiting poor selectivity. For these reasons, adsorbents with lower cost, higher reusability and selectivity still essential to be further investigates.

The aim of this research is to prepare molecular imprinted polymer which can be used for selective removal of β -carotene from Palm Oil Mill Effluent (POME). In this research, the MIPs are synthesized using the bulk polymerization method. The parameters studied are pH, dosage studies, sorption kinetics, sorption isotherms and reusability. Due to the versatility and high level of selectivity, and the recognition which can be achieved, the potential of using MIPs as sorbents to remove color (β carotene) from POME is very promising.

1.3 Objectives of the research

The goal of this research is to remove β -carotene from Palm Oil Mill Effluent using Molecular Imprinted Polymer (MIP). The following specific objectives were designed to achieve this goal;

- To synthesize molecular imprinted polymer (MIP) by bulk polymerization method using β-cyclodextrin (β-CD) as functional monomer, 9-Vinylcarbazole (9VC) and 2-Hydroxyethyle Methacrylate (HEMA) as co monomers and Tolylene-2, 4-diisocyanate (TDI) as cross linker.
- 2- To characterize the synthesized MIP by using Fourier-transform infrared spectroscopy (FTIR), Field emission scanning electron microscopy (FESEM), thermal gravimetric analysis (TGA), and Brunaue Emmett Teller (BET).
- 3- To study the binding capacity of β -carotene toward MIP based on pH, dosage, isotherm, kinetic and reusability.

REFERENCES

- Ahmad, A.L., & Chan, C.Y. (2009). Sustainability of Palm Oil Industries: An Innovative Treatment via Membrane Technology. J. Appl. Sci, 9(17), 3074-3079.
- Ahmed, Yunus, Zahira Yaakob, Parul Akhtar, and Kamaruzzaman Sopian. 2015. "Production of Biogas and Performance Evaluation of Existing Treatment Processes in Palm Oil Mill Effluent (POME)." Renewable and Sustainable Energy Reviews 42(November 2014):1260–78.
- Ahmed, M. A., Abdelbar, N. M., and Mohamed, A. A. (2017). Molecular imprinted chitosan-TiO₂ nanocomposite for the selective removal of Rose Bengal from wastewater. *International Journal of Biological Macromolecules*, 107 (A), 1046-1053.
- Ain, N., Khairi, S., Yusof, N. A., & Abdullah, A. H. (2015). Removal of Toxic Mercury from Petroleum Oil by Newly Synthesized Molecularly-Imprinted Polymer, (Ii), 10562–10577.
- Al-Degs, Y. S., Abu-Surrah, A. S., & Ibrahim, K. A. (2009). Preparation of highly selective solid-phase extractants for Cibacron reactive dyes using molecularly imprinted polymers. *Analytical and Bioanalytical Chemistry*, 393(3), 1055– 1062.
- Alkhatib, M. F., Mamun, A. A., Akbar, I. (2015). Application of response surface methodology (RSM) for optimization of color removal from POME by granular activated carbon. *International Journal of Environmental Science and Technology*, 12 (4), 1295–1302.
- Amata, N. A. A., Tan, Y. H., Lau, W. J., Lai, G. S., Ong, C. S., Mokhtar, N. M., Sani, N. A. A., Ismail, A. F., Goh, K. S., and Chong, O. (2015). Tackling colour issue of anaerobically-treated palm oil mill effluent using membrane technology. *Journal of Water Process Engineering*, 8, 221-226.
- Andersson, L. I., & Mosbach, K. (1990). Enantiomeric resolution on molecularly imprinted polymers prepared with only non-covalent and non-ionic interactions. J. Chromatogr. 516, 313–322.
- Arshady, R., & Mosbach, K. (1981). Synthesis of substrate-selective polymers by host-guest polymerization. *Makromol. Chem*, 182, 687–692.
- Asman, Saliza, Sharifah Mohamad, and Norazilawati Muhamad Sarih. 2016. "Study of the Morphology and the Adsorption Behavior of Molecularly Imprinted Polymers Prepared by Reversible Addition-Fragmentation Chain Transfer (RAFT) Polymerization Process Based on Two Functionalized β-Cyclodextrin as Monomers." *Journal of Molecular Liquids*. 214:59–69.

- Auriola, M., Youssef, F. M., Rajeshwar, D. T., Craig, D. A., and Rao, Y. S. (2006). Endocrine disrupting compounds removal from wastewater, a new challenge. *Process Biochemistry*, 41 (3), 525-539.
- Batlokwa, B. S., Mokgadi, J., Nyokong, T., and Torto, N. (2011). Optimal template removal from molecularly imprinted polymers by pressurized hot water extraction. *Chromatographia*, 5(73), 589-593.
- Beltran, A., Marcé, R. M., Cormack, P. A. G., and Borrull, F. (2010). Synthetic approaches to parabens molecularly imprinted polymers and their applications to the solid-phase extraction of river water samples. *Anal. Chim. Acta*, 677, 72–78.
- Bello, M. M. et al. 2013. "POME Is Treated for Removal of Color from Biologically Treated POME in Fixed Bed Column: Applying Wavelet Neural Network (WNN)." Journal of Hazardous Materials 262:106–13. Retrieved
- Beyazit, S., Tse Sum Bui, B., Haupt, K., and Gonzato, C. (2016). Molecularly imprinted polymer nanomaterials and nanocomposites by controlled/living radical polymerization. *Prog. Polym. Sci.* 62, 1–21.
- Bhagyashree, T., Balasubramanian, S., Yassine, O., Patrick, D., Rajeshwar, D. T., and Gerardo, B. (2016). Review on fate and mechanism of removal of pharmaceutical pollutants from wastewater using biological approach. *Bioresource Technology*, 224, 1-12.
- Boulanouar, S., Mezzache, S., Combès, A., and Pichon, V. (2018). Molecularly imprinted polymers for the determination of organophosphorus pesticides in complex samples. *Talanta*, 176, 465–478.
- Bow, Y., Sutriyono, E., Nasir, S., and Iskandar, I. (2017). Preparation of molecularly imprinted polymers simazine as material potentiometric sensor. *MATEC Web Conf*, 101.
- Branger, C., Meouche, W., Margaillan, A. (2013). Recent advances on ion-imprinted polymers. *React. Funct. Polym*, 73, 859–875.
- Camilla, F. S., Keyller, B. B., and Clebio, S. d. N. (2017). Rational design of a molecularly imprinted polymer for dinotefuran: theoretical and experimental studies aimed at the development of an efficient adsorbent for microextraction by packed sorbent. *Analyst*, 143, 141-149.
- Chanida, R. & Poonsuk, P. (2011). Decolorization and Phenol Removal of Anaerobic Palm Oil Mill Effluent by Phanerochaete chrysosporium. *ATCC 24725*. TIChE International Conference. Thailand (2011).
- Chen, L. (2011). Recent advances in molecular imprinting technology : Current status, challenges and highlighted applications. *Chem Soc Rev*, (38), 797–805.

- Chen, L. (2014). Recent Advances in Molecularly Imprinted Polymers in Food Analysis. *Journal of applied polymer*, 131(8), 1–27.
- Cheong, Won Jo. 2013. "Recent Applications of Molecular Imprinted Polymers for Enantio-Selective Recognition." *Talanta* 106(March): 45–59.
- Chrzanowska, Anna M, Anna Poliwoda, and Piotr P Wieczorek. 2015. "Characterization of Particle Morphology of Biochanin A Molecularly Imprinted Polymers and Their Properties as a Potential Sorbent for Solid-Phase Extraction." *Materials Science & Engineering C* 49: 793–98.
- Cristi, A., Dias, B., & Figueiredo, E. C. (2008). Molecularly imprinted polymer as a solid phase extractor in flow analysis, *Talanta*, 76 (2008), 988–996.
- Dai, C. M., Geissen, S. U., and Zhang, Y. L. (2011) Selective removal of diclofenac from contaminated water using molecularly imprinted polymer microspheres. *Environ Pollut*, 159, 1660–1666.
- Dai, Chao Meng, Xue Fei Zhou, Ya Lei Zhang, Shu Guang Liu, and Juan Zhang. 2011. "Synthesis by Precipitation Polymerization of Molecularly Imprinted Polymer for the Selective Extraction of Diclofenac from Water Samples." Journal of Hazardous Materials.
- Dai, C. M., Zhang, J., Zhang, and Y. L. (2013a). Application of molecularly imprinted polymers to selective removal of clofibric acid fromwater. *PloS one*, 8 (10), 1– 8.
- Dai, C. M., Zhang, J., Zhang, and Y. L. (2013b). Removal of carbamazepine and clofibric acid from water using double templates-molecularly imprinted polymers. *Environ Sci Pollut R*, 20 (8), 5492–5501.
- El-taliawy, H., Ekblad, M., Nilsson, F., Hagman, M., Paxeus, N., Jönsson, K., Cimbritz, M., Jansen, J. I. C., and Bester, K. (2017). Ozonation efficiency in removing organic micro pollutants from wastewater with respect to hydraulic loading rates and different wastewaters. *Chemical Engineering Journal*, 325, 310-321.
- Eng, Zihui M., Jinfang W. Ang, Liangmo Z. Hou, Qinghai W. Ang, and Daoqian Z. Hu. 1999. "High Performance Cocktail Functional Monomer for Making Molecule Imprinting Polymer.
- Figueiredo, L., Erny G. L., Santos, L., and Alves, A. (2016). Applications of molecularly imprinted polymers to the analysis and removal of personal care products: A review. *Talanta*, 1(146), 754-65.
- Filipa, A., & Lobo, F. (2015). Synthesis and Characterization of Molecularly Imprinted Polymer Particles (MIPs) for Biomedical Applications, (October).

- Foo, K. Y., and Hameed, B. H. (2009). Utilization of biodiesel waste as a renewable resource for activated carbon: application to environmental problems. *Renew* Sust Energ Rev, 13 (9), 2495–2504.
- Giuseppe Vasapollo *, Roberta Del Sole, Lucia Mergola, Maria Rosaria Lazzoi, Anna Scardino, S. S. and G. M. (n.d.). review paper for mip preparation.
- Gottschalk, C., Libra, J. A., and Saupe, A. (2009). Ozonation of water and waste water: A practical guide to understanding ozone and its applications. John Wiley & Sons, New York.
- Guoyuan, Y., Hong, T., Jun, L., Changsong, Z., Jiali, L., Yuanyou, Y., Jijun, Y., and Ning, L. (2017). A novel ion-imprinted polymer induced by the glycylglycine modified metal-organic framework for the selective removal of Co(II) from aqueous solutions. *Chemical Engineering Journal*, 333, 280-288.
- Hashim, S. N. N. S., Boysen, R. I., Schwarz, L. J., Danylec, B., Hearn, M. T. W. (2014). A comparison of covalent and non-covalent imprinting strategies for the synthesis of stigmasterol imprinted polymers. J. Chromatogr. A, 1359, 35– 43.
- He, C., Long, Y., Pan, J., Li, K., and Liu, F. (2017). Application of molecularly imprinted polymers to solid-phase extraction of analytes from real samples. J Biochem Biophys Methods, 70 (2), 133-150.
- Hongyuan, Y. & Kyung, H. R. (2006). Characteristic and Synthetic Approach of Molecularly Imprinted Polymer. Int. J. Mol. Sci, 7, 5, 155-178.
- Huang, S. H., and Chen, D. H. (2009). Rapid removal of heavy metal cations and anions from aqueous solutions by an amino-functionalized magnetic nano-adsorbent. *J Hazard Mater*, 163 (1), 174–179.
- Ibrahim, M.Hakimi. 2010. "Review of Current Palm Oil Mill Effluent (POME) Treatment Methods : Vermicomposting as a Sustainable Practice." (January).
- Jing, T., Wang, J., and Liu, M. (2014) Highly effective removal of 2, 4-dinitrophenolic from surface water and wastewater samples using hydrophilic molecularly imprinted polymers. *Environ Sci Pollut R*, 21 (2), 1153–1162.
- Kirsch, N., & Whitcombe, M.J. (2005). *The Semi-Covalent Approach*, in: M. Yan, O. Ramström (Eds.), Mol. Imprinted Mater. Sci. Technol., Marcel Dekker, New York, pp. 93–122.
- Klein, J. U., Whitcombe, M. J., Mulholland, F., Vulfson, E.N. (1999). Template-Mediated Synthesis of a Polymeric Receptor Specific to Amino Acid Sequences. Angew. Chem. Int. Ed, 38, 2057–2060.
- Krupadam, R. J., Khan, M. S., and Wate, S. R. (2010). Removal of probable human carcinogenic polycyclic aromatic hydrocarbons from contaminated water using molecularly imprinted polymer. *Water Res*, 44 (3), 681–688.

- Kyzas, G. Z., Lazaridis, N. K., and Bikiaris, D. N. (2013). Optimization of chitosan and β-cyclodextrin molecularly imprinted polymer synthesis for dye adsorption. *Carbohydr Polymer*, 2 (91), 198-208.
- Lai, E. P. C., Maleki, Z. D, and Wu, S. Y. (2010). Characterization of molecularly imprinted and nonimprinted polymer submicron particles specifically tailored for removal of trace 17β - estradiol in water treatment. *J Appl Polym Sci*, 116 (3), 1499–1508.
- Li, Y., Li, X., Li, Y. Q. (2009). Selective removal of 2, 4-dichlorophenol from contaminated water using non-covalent imprinted microspheres. *Environ Pollut*, 157 (6), 1879–1885.
- Limkhuansuwan, V., and Chaiprasert, P. (2010). Decolorization of molasses melanoidins and palm oil mill effluent phenolic compounds by fermentative lactic acid bacteria. *Journal of Environmental Sciences*, 22, 1209-1217.
- Luo, J., Jiang, S., Liu, X. (2014). Electrochemical sensor for bovine hemoglobin based on a novel graphene-molecular imprinted polymers composite as recognition element. *Sens. Actuators B*, 203, 782–789.
- Luo, X., Zhana, Y., Huanga, Y., Yanga, L., Tua, X., and Luo, S. (2011). Removal of water-soluble acid dyes from water environment using a novel magnetic molecularly imprinted polymer. *Journal of Hazardous Materials*, 187 (2011), 274–282.
- Madikizela, L. M., & Chimuka, L. (2016). Synthesis, adsorption and selectivity studies of a polymer imprinted with naproxen, ibuprofen and diclofenac. *Journal of Environmental Chemical Engineering*, 4(4), 4029–4037.
- Madikizela, L. M., Tavengwa, N. T., and Chimuka, L. (2018). Applications of molecularly imprinted polymers for solid-phase extraction of non-steroidal anti-inflammatory drugs and analgesics from environmental waters and biological samples. *Journal of Pharmaceutical and Biomedical Analysis*, 147 (5), 624-633.
- Maohong, Z., Hairen, Y., and Xiaowei, Z. (2014). Isolation and Characterization of a Novel Heterotrophic Nitrifying and Aerobic Denitrifying Bacterium Pseudomonas stutzeri KTB for Bioremediation of Wastewater. *Biotechnology and Bioprocess Engineering*, 19, 231-238.
- Maria, S., & Beena, M. (2017). Carbon Nanotube Based Ion Imprinted Polymer as Electrochemical Sensor and Sorbent for Zn(II) ion from Paint Industry Wastewater. *International Journal of Polymer Analysis and Characterization*, 23, 18-28.
- Mariana, D., Prabal, S., Ecevit, Y., Katrin, M., Thomas, L., Simon, E. (2017). Molecularly imprinted polymers synthesized via template immobilization on fumed silica nanoparticles for the enrichment of phosphopeptides. *Mol Recognit*; 2677, 1-10.

- Meng, Z., Chen, W., and Mulchandani, A. (2005). Removal of estrogenic pollutants from contaminated water using molecularly imprinted polymers. *Environ Sci Technol*, 39 (22), 8958–8962.
- Moawed, E. A., & Radwan, A.M. (2017). Application of acid modified polyurethane foam surface for detection and removing of organochlorine pesticides from wastewater. *Journal of Chromatography B*, 1045, 95-102.
- Mohammed Rafie Rushdy. (2013). Decolorisation of Biologically Treated Palm Oil Mill Effluent (POME) Using Adsorption Technique. *International Refereed Journal of Engineering and Science*, 2 (10), 01-11.
- Mohammed, R. R. (2013). Decolorisation of biologically treated palm oil mill effluent (POME) using adsorption technique. *IRJES*, 2, 1-11.
- Mohammed, R. R., & Chong, M.F. (2014). Treatment and decolorization of biologically treated Palm Oil Mill Effluent (POME) using banana peel as novel biosorbent. J. Environ. Manage, 132, 237–249
- Mohammed, R.R. (2013). Decolorisation of Biologically Treated Palm Oil Mill Effluent (POME) Using Adsorption Technique. *IRJES*, 2(10), 1–11
- Mohammed, R.R., & Chong, M. F. (2014). Treatment and decolorization of biologically treated Palm Oil Mill Effluent (POME) using banana peel as novel biosorbent. *J. Environ. Manage*, 132, 237-249.
- Mustapa, A.N., Manan, Z. A., Mohd Azizi, C.Y., Setianto, W. B., and Mohd Omar, A. K. (2011). Extraction of β-carotenes from palm oil mesocarp using subcritical R134a. *Food Chemistry*, 125 (1), 262-267.
- Nabila, S. A., Hanan, E., Hanan, S. I., Walid, E. H., and Medhat, A. I. (2013). A novel structure for removal of pollutants from wastewater. *Spectrochimica Acta Part A*, 121, 216-223.
- Nadia, M. C., Peter, W., Sophie, F., Lee, W., Eva, F., and Grégorio, C. (2017). Waterinsoluble β-cyclodextrin–epichlorohydrin polymers for removal of pollutants from aqueous solutions by sorption processes using batch studies: A review of inclusion mechanisms. *Progress in Polymer Science*, 78, 1-23.
- Ndunda, E. N., & Mizaikoff, B. (2016). Molecularly imprinted polymers for the analysis and removal of polychlorinated aromatic compounds in the environment: a review. *Analyst*, 141(11), 3141-56.
- Neoh, Chin Hong, Chi Yong Lam, Chi Kim Lim, Adibah Yahya, and Zaharah Ibrahim. 2014. "Decolorization of Palm Oil Mill Effluent Using Growing Cultures of Curvularia Clavata." Environmental Science and Pollution Research 21(6):4397–4408.

- Norrlöw, O., Glad, M., and Mosbach, K. (1984). Acrylic polymer preparations containing recognition sites obtained by imprinting with substrates. *J. Chromatogr. A.* 299, 29–41.
- Oguntimein, G. B. (2015). Biosorption of dye from textile wastewater effluent onto alkali treated dried sunflower seed hull and design of a batch adsorber. Journal of Environmental Chemical Engineering, 3(4), 2647–2661.
- Parthasarathy, Shridharan, Rafie Rushdy Mohammed, Chong Mei Fong, Rachel L. Gomes, and Sivakumar Manickam. 2016. "A Novel Hybrid Approach of Activated Carbon and Ultrasound Cavitation for the Intensification of Palm Oil Mill Effluent (POME) Polishing." Journal of Cleaner Production 112:1218–26.
- Poma, Alessandro, Anthony P. F. Turner, and Sergey A. Piletsky. 2010. "Advances in the Manufacture of MIP Nanoparticles." Trends in Biotechnology 28(12):629– 37. Retrieved
- Pratiwi R, Megantara S, Rahayu D, Pitaloka I, Hasanah AN. Comparison of Bulk and Precipitation Polymerization Method of Synthesis Molecular Imprinted Solid Phase Extraction for Atenolol using Methacrylic Acid. J Young Pharm, 2019;11(1):12-16.
- Rafatullah, M., Sulaiman, O., and Hashim, R. (2010). Adsorption of methylene blue on low-cost adsorbents: a review. *J Hazard Mater*, 177 (1), 70–80.
- Ramstrbm, Olof, Lars I. Andersson, and K. Mosbach. 1993. "Recognition Sites Incorporating Both Pyridinyl and Carboxy Functionalities Prepared by Molecular Imprinting." 7562–64.
- Ramstron, O., & Yan, M. (2005). *Molecularly Imprinted Materials*. Chapter 1 "Molecular imprinting- an introduction," Marcel Dekker, New York, pp. 1-12.
- Rane, J., Adhikar, P., and Bakal, R. L. (2015). Molecular Imprinting: An Emerging Technology. Asian Journal of Pharmaceutical Technology & Innovation, 03 (11), 75 – 91.
- Ratpukdi, Thunyalux. 2012. "Decolorization of Anaerobically Treated Palm Oil Mill Wastewater Using Combined Coagulation and Vacuum Ultraviolet-Hydrogen Peroxide." International Journal of Chemical Engineering and Applications 3(5):333–36.
- Reungoat, J., Macova, M., Escher, B. I., Carswell, S., Mueller, J. F., and Keller, J. (2010). Removal of micropollutants and reduction of biological activity in a full scale reclamation plant using ozonation and activated carbon filtration. *Water Research*, 44 (2), 625-637.
- Rutz, J. K., Borges, C. D., Zambiazi, R. C., Michele, M. C., Luiza, S. K., Caciano, P.Z. N. (2017). Microencapsulation of palm oil by complex coacervation for application in food systems. *Food Chemistry*, 220, 59–66.

- Saeed, M, O., Azizlia, K., Mohamed, H, I, and Mohammed J. K. B. (2015). Application of CCD in RSM to obtain optimize treatment of POME using Fenton oxidation process. *Journal of Water Process Engineering*, 8, 7-16.
- Samarth, B. N., Kamble, V., Mahanwar, A. P., Rane., A. V., and Abitha V. K. (2015). A historical perspective and the development of molecular imprinting polymer- A review. *Chemistry International*, 1(4), 202-210.
- Sarafraz-Yazdi, A., & Razavi, H. (2015). Application of molecularly-imprinted polymers in solid-phase microextraction techniques. *TrAC Trends in Analytical Chemistry*, 73(81), 65-74.
- Singh, Minni. 2016. "Beta Carotene Beta Cyclodextrin Inclusion Complex : Towards Enhanced Aqueous Solubility. *journal of global Biosciences*,5, 3665–3675.
- Sogut, E. G., & Caliskan, N. (2017). Isotherm and kinetic studies of Pb (II) adsorption on raw and modified Diatomite by using non-liner regression method METHOD, (April).
- Sun, Xiaoli et al. 2014. "Highly Selective Dummy Molecularly Imprinted Polymer as a Solid-Phase Extraction Sorbent for Five Bisphenols in Tap and River Water." *Journal of Chromatography A*, 1343, 33–41.
- Surikumaran, Hemavathy, Sharifah Mohamad, and Norazilawati Muhamad Sarih. 2014. "Molecular Imprinted Polymer of Methacrylic Acid Functionalised β-Cyclodextrin for Selective Removal of 2,4–diclorophenol": *Int.J.Mol.Sci* 2014,15, 6111-6136.
- Tabassum, Salma, Yejian Zhang, and Zhenjia Zhang. 2015. "An Integrated Method for Palm Oil Mill Effluent (POME) Treatment for Achieving Zero Liquid Discharge - A Pilot Study." Journal of Cleaner Production 95:148–55. Retrieved
- Tamayo, F. G., Turiel, E., and Martín-Esteban, A. (2007). Molecularly imprinted polymers for solid-phase extraction and solid-phase microextraction: Recent developments and future trends. *Journal of Chromatography A*,1152, 32–40.
- Tan, Y. H., Goh, P. S., Ismail, A. F., Ng, B. C., and Lai, G. S. (2017). Decolourization of aerobically treated palm oil mill effluent (AT-POME) using polyvinylidene fluoride (PVDF) ultrafiltration membrane incorporated with coupled zinc-iron oxide nanoparticles. *Chemical Engineering Journal*, 308, 359-369.
- Tang, Y., Lan, J., Gao, X., Liu, X., Zhang, D., Wei, L., et al. (2016). Determination of clenbuterol in pork and potable water samples by molecularly imprinted polymer through the use of covalent imprinting method. *Food Chem*, 190, 952–959.
- Ternes, T., and Gunten, U. V. (2010). Emerging Contaminants in water: Occurrence, fate, removal and assessment in the water cycle (from wastewater to drinking water). *Water Research*, 44 (2), 351–668.

- Tom, Lou Ann, Nicole A Schneck, and Carla Walter. 2012. "Improving the Imprinting Effect by Optimizing Template: Monomer: Cross-Linker Ratios in a Molecularly Imprinted Polymer for Sulfadimethoxine." Journal of Chromatography B 909: 61–64.
- Turiel, E., Martin-esteban, A., Tamayo, F. G., Turiel, E., & Mart, A. (2007). Molecularly imprinted polymers for solid- phase extraction and solid-phase microextraction : Recent developments and future trends. J Chromatogr A, 1152 (1–2), 32–40.
- Wang, H., Yuan, X. Z., and Wu, Y. (2013). Graphene-based materials: fabrication, characterization and application for the decontamination of wastewater and wastegas and hydrogen storage/generation. *Adv Colloid Interfac*, 195, 19–40.
- Whitcombe, M. J., Kirsch, N., Nicholls, I. A. (2014). Molecular imprinting science and technology: A survey of the literature for the years 2004-2011. J. Mol. Recognit, 27, 297–401.
- Whitcombe, M.J., Rodriguez, M. E., Villar, P., Vulfson, E. N. (1995). A New Method for the Introduction of Recognition Site Functionality into Polymers Prepared by Molecular Imprinting: Synthesis and Characterization of Polymeric Receptors for Cholesterol. J. Am. Chem. Soc, 117, 7105–7111.
- Wu, T. Y., A. W. Mohammad, J. Jahim, and N. Anuar. 2007. "Palm Oil Mill Effluent (POME) Treatment and Bioresources Recovery Using Ultrafiltration Membrane: Effect of Pressure on ... Palm Oil Mill Effluent (POME) Treatment and Bioresources Recovery." (May 2017).
- Wu, N., Luo, Z., Ge, Y., Guo, P., Du, K., Tang, W., ... Fu, Q. (2016). A novel surface molecularly imprinted polymer as the solid-phase extraction adsorbent for the selective determination of ampicillin sodium in milk and. *J Pharm Anal*, 6 (3):157–164.
- Wulff, G., & Sarhan, A. (1972). The use of polymers with enzyme-analogous structures for the resolution of racemates. *Angew. Chem. Int. Ed. Engl*, 11, 341.
- Wulff, G., and Vietmeier, J. (1989). Enantioselective synthesis of amino acids using polymers possessing chiral cavities obtained by an imprinting procedure with template molecules. *Makromol. Chem.-Macromol. Chem. Phys*, 190, 1727–17345.
- Wulff, G., Sarhan, A., Zabrocki, K. (1973). Enzyme-analogue built polymers and their use for the resolution of racemates. *Tetrahedron Lett*, 14, 4329–4332.
- Xu, P., Zeng, G. M., and Huang, D. L. (2012). Use of iron oxide nanomaterials in wastewater treatment: a review. *Sci Total Environ*, 424, 1–10.
- Xu, Xuanwei et al. 2015. "A Molecularly Imprinted Polymer for the Selective Solid-Phase Extraction of Dimethomorph from Ginseng Samples." *Journal of Chromatography B* 988:182–86.

- Yahaya, S. M., & Seng, L. (2013). Palm oil mill effluent (pome) from Malaysia palm oil mills: waste or resource. *International Journal of Science, Environment and Technology*, 2 (6), 1138 – 1155.
- Yan, H., & Ho Row, K. (2006). Characteristic and Synthetic Approach of Molecularly Imprinted Polymer. *International Journal of Molecular Sciences*, 7(5), 155-178.
- Yao, J., Li, X., and Qin, W. (2008). Computational design and synthesis of molecular imprinted polymers with high selectivity for removal of aniline from contaminated water. *Analytica Chimica Acta*, 610 (2), 282-288.
- Ye, L., Weiss, R., and Mosbach, K: (2000). Synthesis and characterization of molecularly imprinted microspheres. *Macromolecules*, 33, 8239-8245.
- Yong, Y. S., Ivy, A. W. T., and Mohammad, O. A. (2017). Adsorption of colour, TSS and COD from palm oil mill effluent (POME) using acid-washed coconut shell activated carbon: Kinetic and mechanism studies. *MATEC Web of Conferences*, 87.
- Yousra, T., Ines, M., Rim, L., Asma, B. R., Amel, K., Hanene, C., Hadda, O., and Abdennaceur, H. (2017). Biofilms in bioremediation and wastewater treatment: characterization of bacterial community structure and diversity during seasons in municipal wastewater treatment process. *Environmental Science and Pollution Research*, 24, 3519–3530.
- Yusof, Nor Azah, Siti Khadijah Ab Rahman, Mohd Zobir Hussein, and Nor Azowa Ibrahim. 2013. "Preparation and Characterization of Molecularly Imprinted Polymer as SPE Sorbent for Melamine Isolation." *Polymers*. 2013, 5, 1215– 1228.
- Zahrim, A. Y., Z. D. Dexter, C. G. Joseph, and N. Hilal. 2017. "Effective Coagulation-Flocculation Treatment of Highly Polluted Palm Oil Mill Biogas Plant Wastewater Using Dual Coagulants: Decolourisation, Kinetics and Phytotoxicity Studies." Journal of Water Process Engineering 16:258–69.
- Zahrim, AY. 2014. "Palm Oil Mill Biogas Producing Process Effluent Treatment: A Short Review." *Journal of Applied Sciences* (November 2014).
- Zhang, J., Liu, D., Shi, Y., Sun, C., Niu, M., Wang, R., ... He, H. (2017). Determination of quinolones in wastewater by porous β-cyclodextrin polymer based solid-phase extraction coupled with HPLC. J Chromatogr B Analyt Technol Biomed Life Sci", 1068–1069 (October), 24–32.
- Zhang, M., Zeng, J., Wang, Y., and Chen, X. (2013). Developments and Trends of Molecularly Imprinted Solid-Phase Microextraction. *Journal of Chromatographic Science*, 51(7), 577–586.
- Zhijun, M., Qi, Z., Xingyuan, W., Changye, M., Liwei, S., Zhihao, G., and Liang, C. (2017). Fluoride ion adsorption from wastewater using magnesium(II),

aluminum(III) and titanium(IV) modified natural zeolite: kinetics, thermodynamics, and mechanistic aspects of adsorption. *Journal of Water Reuse and Desalination*, 2017.

Zhongbo, Z., and Hu, J. (2008). Selective removal of estrogenic compounds by molecular imprinted polymer (MIP). *Water Res*, 42 (15), 4101–4108.



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