



UNIVERSITI PUTRA MALAYSIA

**ELASTIC, OPTICAL AND THERMAL PROPERTIES OF ZINC-LEAD
TELLURITE GLASS SYSTEMS**

SALAH HASSAN ALMOKHTAR ALAZOUMI

FS 2018 95



**ELASTIC, OPTICAL AND THERMAL PROPERTIES OF ZINC-LEAD
TELLURITE GLASS SYSTEMS**

By

SALAH HASSAN ALMOKHTAR ALAZOUMI

**Thesis Submitted to the School of Graduate Studies, Universiti Putra
Malaysia, in Fulfilment of the Requirements for the Degree of
Doctor of Philosophy**

August 2018

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs, and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



DEDICATION

This work is dedicated to my late parents Hassan and Iftaimah. May Allah forgive their shortcoming and wrongs.



Abstract of a thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

ELASTIC, OPTICAL AND THERMAL PROPERTIES OF ZINC-LEAD TELLURITE GLASS SYSTEMS

By

SALAH HASSAN ALMOKHTAR ALAZOUMI

August 2018

Chairman : Professor Sidek Abdul Aziz, PhD
Faculty : Science

This thesis presents the comprehensive study on elastic, optical and thermal properties of tellurite glasses, binary lead tellurite (PbO-TeO_2) and ternary zinc lead tellurite (PbO-ZnO-TeO_2) glass systems prepared using the conventional melt-quenching technique. The physical, structural, elastic, optical and thermal properties of the glasses were investigated using ultrasonic technique, X-ray diffraction (XRD), scanning electron microscopy and energy dispersive X-ray spectroscopy (SEM/EDX), differential scanning calorimetry (DSC), UV-Visible spectroscopy (UV-Vis) and Fourier transform infrared (FTIR) spectroscopy.

Density (ρ), molar volume (V_m), oxygen packing density (OPD), refractive index, band gap energy (E_g), average cross-link density (n_c), the number of bonds per unit volume (n_b) and Poisson's ratio (σ) values were calculated from the characterizations data obtained for the interpretation of physical, structural and optical properties of the glass system under study. The density measurements were carried out under ambient condition using densitometer (Model: MD- 300S Densimeter) which works on the basis of Archimedes principle. Ultrasonic pulse-echo technique carried out at room temperature, and resonating frequency of 5 MHz was used for the determination of the ultrasonic velocities (longitudinal and shear). Elastic (longitudinal, shear, bulk and Young's) moduli, Poisson's ratio, Debye temperature, and a softening temperature of all glass samples have been calculated using the measured ultrasonic velocities and densities. Experimental values of elastic moduli were compared with calculated values of elastic moduli using bond compression model, Makishima-Mackenzie model, and Rocherulle's model.

The profiles of the EDX analysis showed the presence of all the mentioned elements in the prepared glass system. It is observed from the result obtained that the use of alumina crucible induces a partial dissolution of Al_2O_3 in the melt that modifies the original composition. The prepared glass samples were homogeneous, lime green color and became more transparent as PbO increases. The XRD pattern confirmed the amorphous nature of the two glass systems. The FTIR spectra for the binary glass system showed a decrease in non-bridging oxygen (NBO) concentration with PbO introduction, which later then increased as more PbO was added to the system. Addition of more ZnO in the ternary system showed a pattern of increasing and decreasing NBO concentration in the glass network.

An addition of PbO content in the binary leads to the increase in the density, while molar volume decreased in the tellurite glass system. The density of ternary glass system decrease, and molar volume decreased with the increase of ZnO concentration in the (PbO - ZnO - TeO_2) glass system. The elastic moduli of the glasses were found to be composition dependent; the elastic moduli values decreased (from $L = 63.41$ to 53.55 GPa, $G = 21.78$ to 16.42 GPa, $K = 34.37$ to 31.65 GPa, $E = 53.94$ to 41.99 GPa) for binary tellurite system with increase of PbO concentration and increased (from $L = 47.09$ to 50.61 GPa, $G = 21.78$ to 16.42 GPa, $K = 27.01$ to 28.01 GPa, $E = 38.09$ to 42.30 GPa) in the case of ternary system increase in ZnO concentration. With the increase in the PbO in the binary system, the microhardness and Debye temperature decreased as the elastic moduli, while the Poisson ratio increased. In the case of the ternary system, increase in the ZnO content increased the microhardness and Debye temperature and decreased the Poisson ratio. The refractive index of the TeO_2 - PbO glasses fall in the range of 2.39 to 2.50 for increasing mole fraction of PbO content. The refractive indices for ternary glass system show a varying between 2.504 – 2.580. Thermal analysis of the glasses was realized regarding glass transition temperature (T_g), crystallization temperature (T_c), and glass stability against crystallization (ΔT). The values of T_g , T_c , ΔT and T_m fall in the range of 380 to 311 °C, 379 to 425 °C, 45 to 67 and 469 to 534 °C respectively for the binary glass system and 291 to 300 °C, 389 to 459 °C, 89 – 166 °C and 534 490 °C respectively for the ternary glass system.

The two glass systems were observed to be amorphous, with good mechanical strength (on the bases of their elastic properties), thermal stability as well as high refractive index. The high refractive index value index as well as high optical transparency (In the case of ZnO incorporated) for the two series of glasses gives the ternary glass system good potential for optical fiber applications.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

SIFAT ELASTIK, OPTIK DAN TERMA BAGI SISTEM KACA ZINK- PLUMBUM TELLURITE

Oleh

SALAH HASSAN ALMOKHTAR ALAZOUMI

Ogos 2018

Pengerusi : Profesor Sidek Abdul Aziz, PhD
Fakulti : Sains

Tesis ini menerangkan tentang kajian secara komprehensif terhadap sifat elastik, optikal dan terma system kaca tellurite, plumbum tellurite binari (PbO-TeO_2) dan zink plumbum tellurite ternari (PbO-ZnO-TeO_2) menggunakan teknik lindapan leburan konvensional. Sifat fizikal, struktural, elastik, optikal dan terma bagi kaca ini di kaji dengan menggunakan teknik ultrasonik, pembelauan sinar-X (XRD), mikroskop transmisi elektron dan spektroskopi serakan tenaga sinar-X (SEM/EDX), kalorimetri pengimbasan pembezaan (DSC), spektroskopi sinar ultra ungu-cahaya Nampak (UV-Vis) dan spektroskopi infra merah pembezaan Fourier (FTIR).

Ketumpatan (ρ), isipadu molar (V_m), isipadu kepadatan oksigen (OPD), indeks refraktif, tenaga jurang jalur (E_g), ketumpatan purata taut-silang (n_c), bilangan ikatan per unit isipadu (n_b) dan nilai nisbah Poisson (σ) dikira dari data yang diperolehi dari sifat fizikal, struktural dan optikal bagi system gelas yang dikaji. Pengukuran ketumpatan diperolehi menggunakan densitometer (Model: MD- 300S Densimeter) berdasarkan kepada prinsip Archimedes. Teknik gema-denyut ultrasonic dijalankan pada suhu bilik menggunakan frekuensi beresonan 5 MHz untuk menentukan halaju ultrasonik (longitud dan ricih). Elastik (longitud, ricih, pukal dan Young) moduli, nisbah Poisson, suhu Debye dan suhu pelembutan bagi semua sampel kaca dikira menggunakan halaju dan ketumpatan ultrasonik. Nilai eksperimen bagi modulus elastik dibandingkan dengan modulus elastik menggunakan model pemampatan ikatan, model Makishima-Mackenzie dan Model Rocherulle.

Analisis EDX menunjukkan kehadiran semua elemen yang digunakan untuk penghasilan sistem kaca. Hasil dari pemerhatian yang menggunakan krusibel alumina telah mengaruh pelarutan sebahagian dari Al_2O_3 dalam leburan yang telah

mengubah komposisi asal. Sampel kaca adalah homogen, berwarna hijau muda dan menjadi semakin lut-sinar apabila PbO meningkat. Graf XRD mengesahkan sifat amorfus bagi sistem dua kaca. Spectrum FTIR bagi sistem kaca binari menunjukkan pengurangan oksigen bukan penitian (NBO) dengan kehadiran PbO, yang akhirnya meningkat apabila semakin banyak PbO ditambah kedalam sistem. Penambahan lebih banyak ZnO dalam sistem ternari menunjukkan pola penambahan dan pengurangan NBO dalam rangkaian kaca.

Penambahan kandungan PbO ke dalam plumbum binari meningkatkan ketumpatan dan mengurangkan isipadu molar sistem kaca tellurite. Ketumpatan sistem kaca ternari menurun sementara isipadu molar mengurang dengan penambahan ZnO dalam sistem kaca (PbO-ZnO-TeO₂). Modulus elastik bagi kaca didapati bergantung kepada komposisi; nilai modulus elastik berkurang (julat L = 63.41 ke 53.55 GPa, G = 21.78 ke 16.42 GPa, K = 34.37 ke 31.65 GPa, E = 53.94 ke 41.99 GPa) bagi sistem kaca tellurite binari dengan meningkatnya kepekatan PbO dan meningkat (julat L = 47.09 ke 50.61 GPa, G = 21.78 ke 16.42 GPa, K = 27.01 ke 28.01 GPa, E = 38.09 ke 42.30 GPa) dalam sistem ternari meningkat dalam kepekatan ZnO. Dengan peningkatkan PbO dalam sistem binari, mikrokekeraan dan suhu Debye menurun seperti modulus elastik dan nisbah Poisson meningkat. Dalam sistem ternari, peningkatan kadungan ZnO meningkatkan mikrokekeraan dan suhu Debye dan menurunkan nisbah Poisson. Analisis terma kaca dapat mengenalpasti suhu peralihan kaca (T_g), suhu penghabluran (T_c), dan kestabilan kaca terhadap penghabluran (ΔT). Indeks refraktif bagi kaca TeO₂-PbO didapati meningkat bagi pecahan mol PbO. Indeks refraktif bagi sistem kaca ternari adalah diantara 2.504 – 2.580. Indeks biasan bagi kaca TeO₂-PbO berada dalam julat antara 2.39 ke 2.50 apabila pecahan mol PbO meningkat. Manakala indeks biasan bagi sistem kaca ternari menunjukkan nilai berbeza antara 2.504 – 2.580. Analisis terma bagi kaca menunjukkan nilai suhu peralihan kaca (T_g), suhu penghabluran (T_c), dan kestabilan kaca melawan penghabluran (ΔT). Nilai T_g , T_c , ΔT and T_m berada dalam julat 380 ke 311 °C, 379 ke 425 °C, 45 ke 67 °C dan 469 to 534 °C bagi sistem kaca binari dan 291 ke 300 °C, 389 ke 459 °C, 89 ke 166 °C dan 534 ke 490 °C bagi sistem kaca ternari.

Pemerhatian ke atas kedua-dua sistem kaca tersebut menunjukkan ia adalah amorfus, dengan kekuatan mekanikal (berdasarkan sifat elastik), kestabilan terma yang baik dan juga indek biasan yang tinggi. Nilai indek biasan yang tinggi dan juga kelutsinaran optikal yang tinggi (dalam kes penambahan ZnO) bagi dua siri kaca menunjukkan sistem kaca ternari mempunyai potensi yang baik bagi kegunaan fiber optik.

ACKNOWLEDGEMENTS

My sincere appreciations go to my wife, children, family, and friends for their continued support of the success of the program. I would like to extend my profound to my supervisor, Prof. Sidek B Hj Ab Aziz and other team members; Prof. Halimah Bt Mohamed Kamari and Dr. Mohd Hafiz Bin Mohd Zaid for their encouragements and unending support during the whole journey. Your critical observations and contributions have made me more hardworking and successful in my PhD journey.



This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Sidek Bin Ab Aziz, PhD
Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Halimah Binti Mohamed Kamari, PhD
Professor
Faculty of Science
Universiti Putra Malaysia
(Member)

Mohd Hafiz Bin Mohd Zaid, PhD
Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Member)

ROBIAH BINTI YUNUS, PhD
Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software

Signature: _____

Date: _____

Name and Matric No: Salah Hassan Almokhtar Alazoumi, GS 43560

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) were adhered to.

Signature: _____

Name of Chairman
of Supervisory
Committee:

Professor Dr. Sidek Bin Ab Aziz

Signature: _____

Name of Member
of Supervisory
Committee:

Professor Dr. Halimah Binti Mohamed Kamari

Signature: _____

Name of Member
of Supervisory
Committee:

Dr. Mohd Hafiz Bin Mohd Zaid

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiv
LIST OF FIGURES	xviii
LIST OF ABBREVIATIONS AND SYMBOLS	xxii
CHAPTER	
1 INTRODUCTION	1
1.1 Research background	1
1.2 Tellurite glass application	2
1.3 Problem statement	3
1.4 Research objectives	4
1.5 Scope of the study	4
1.6 Importance of the study	5
1.7 Thesis Organization	6
2 LITERATURE REVIEW	7
2.1 Introduction	7
2.2 Nature of glass	7
2.2.1 Glass formation	7
2.3 Tellurite Glass	9
2.3.1 Structure of Tellurite Glass	9
2.3.2 Properties and Applications of Tellurite Glass	9
2.4 Ultrasonic Studies and Elastic Properties	11
2.4.1 Binary tellurite glass	12
2.4.2 Ternary tellurite glass	14
2.5 Optical Studies of Tellurite Glass	16
2.6 Thermal Studies of Tellurite Glass	17
2.7 Summary of review of related literature	18
3 METHODOLOGY	21
3.1 Introduction	21
3.2 Preparation of glass samples	21
3.2.1 Weighing	23
3.2.2 Melt quenching technique	23
3.2.3 Cutting, polishing, and grinding	23
3.3 Characterization and measurement	23
3.3.1 Density and Molar volume	23

3.3.1.1	Density measurement	23
3.3.1.2	Computation of molar volume	24
3.3.2	Structure of Tellurite Glass	24
3.3.2.1	X-ray diffraction (XRD) spectroscopy and Bragg's Law	25
3.3.2.2	Fourier transform infrared (FTIR) spectroscopy	26
3.3.3	Ultrasonic measurement	26
3.3.3.1	Ultrasonic velocity	28
3.3.3.2	Elastic properties	28
3.3.3.2.1	Elastic Modulus	28
3.3.3.2.2	Acoustic Debye and Softening Temperature	29
3.3.4	Theoretical models	30
3.3.4.1	Bond compression model	30
3.3.4.2	Makishima and Mackenzie model	33
3.3.4.3	Rocherulle model	34
3.3.5	UV-Visible spectroscopy	35
3.3.6	Energy Band Gap	36
3.3.7	Refractive Index, Molar Refractive Index, and Metallization Criterion	37
3.3.8	Dielectric Constant, Linear Dielectric Susceptibility, Reflection Loss and Transmission Coefficient	38
3.3.9	Polarizability	39
3.3.10	Other Parameters	40
3.3.10.1	Oxygen Packing Density (OPD)	40
3.3.10.2	Lead/Zinc Ion Concentration (N)	40
3.3.10.3	Lead/Zinc Inter-ionic Distance	40
3.3.10.4	Polaron Radius (R_p)	40
3.3.10.5	Optical Basicity	41
3.3.11	Thermal analysis	41
3.3.12	Differential scanning calorimetry (DSC)	41
3.3.13	Glass crystallization (T_c), Transition temperature (T_g), and Thermal Stability (ΔT)	42
3.3.14	Error Analysis of Measurements	43
4	RESULTS AND DISCUSSION	44
4.1	Introduction	44
4.2	X-ray Diffraction (XRD) and (SEM/EDX)	44
4.3	Binary PbO-TeO ₂ glass	49
4.3.1	Density, molar volume, and oxygen packing density (OPD)	50
4.3.2	FTIR Spectral Analysis of Lead Tellurite Glass	51
4.3.3	Elastic Properties of Lead Tellurite Glass	53
4.3.3.1	Elastic Modulus and Ultrasonic Velocities	53
4.3.3.2	Microhardness (H), Poisson's ratio and other parameters	57
4.3.3.3	Theoretical Elasticity Models	60

	4.3.3.3.1	Bond compression model	60
	4.3.3.3.2	Makishima–Mackenzie model	62
	4.3.3.3.3	Rocherulle model	64
	4.3.3.3.4	Consistency of the experimental and theoretical elastic modulus and Poisson’s ratio	65
4.3.4		Optical Properties of Lead Tellurite Glass	69
	4.3.4.1	UV-Visible analysis	69
	4.3.4.2	Optical band gap and Urbach’s energy	70
	4.3.4.3	Refractive index, Molar refraction, and Molar polarizability	74
	4.3.4.4	Molar Electronic Polarizability, Electronic Polarizability, and Oxide Ion Polarizability	77
	4.3.4.5	Metallization criterion, Dielectric constant, Reflection loss, and Optical transmission coefficient	78
	4.3.4.6	Ion Concentration, Polaron Radius, Inter-ionic distance, Field Strength of Pb^{2+} ion Yield, Linear Dielectric Susceptibility, and Optical Basicity	82
	4.3.5	Thermal Properties of Lead Tellurite Glass	84
4.4		Ternary $PbO-TeO_2-ZnO$ glass	87
	4.4.1	Density, molar volume, and oxygen packing density	87
	4.4.2	FTIR Spectra Investigation of Zinc Lead Tellurite Glass	89
	4.4.3	Elastic Properties of Zinc Lead Tellurite Glass	90
	4.4.3.1	Ultrasonic Velocity and Elastic Modulus	90
	4.4.3.2	Poisson’s ratio, Microhardness (H), and other Parameters	93
	4.4.3.3	Theoretical Elasticity Models	96
	4.4.3.3.1	Bond compression model	96
	4.4.3.3.2	Makishima–Mackenzie model	99
	4.4.3.3.3	Rocherulle model	101
	4.4.3.3.4	Agreement between experimental and theoretical elastic moduli and Poisson’s ratio	102
	4.4.4	Optical Properties of Zinc Lead Tellurite Glasses	105
	4.4.4.1	UV-Visible analysis	105
	4.4.4.2	Optical band gap and Urbach’s energy	106
	4.4.4.3	Refractive index, molar refraction, and molar polarizability	111
	4.4.4.4	Molar Electronic Polarizability, Electronic Polarizability, and Oxide Ion Polarizability	113
	4.4.4.5	Metallization criterion, Dielectric constant, Reflection loss, and Optical transmission coefficient	114

4.4.4.6	Zinc Ion Concentration (N), Polaron radius (R_p), Inter-ionic distance (R_i), Field Strength of Zn^{2+} ions Yield (F), Linear Dielectric Susceptibility (χ), and Optical Basicity	118
4.4.5	Thermal Properties of Zinc Lead Tellurite Glass	120
5	CONCLUSION AND RECOMMENDATIONS	123
5.1	Conclusion	123
5.2	Recommendations for Future Studies	125
	REFERENCES	126
	APPENDICES	148
	BIODATA OF STUDENT	151
	LIST OF PUBLICATIONS	152

LIST OF TABLES

Table	Page
3.1 The composition of prepared samples	21
4.1 Experimental values of density (ρ), molar volume (V_m) and oxygen packing density (OPD) of binary $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glass system	50
4.2 Assignment of infrared transmission bands for $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glasses	53
4.3 TeO_4 concentration of prepared glass sample with a different concentration of lead oxide concentration	53
4.4 The Mean ultrasonic, shear, longitudinal velocities of $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	54
4.5 Experimentally obtained values for moduli of elasticity, E to G ratio and the Poison's ratio (σ) $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	56
4.6 Softening temperature (T_s), microhardness (H), Debye temperature (θ_D), fractal bond connectivity (d), thermal expansion coefficient (α_P) and acoustic impedance (Z) for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	60
4.7 Crosslink density per cation (n_c), coordination number (n_f), cation-anion bond length r (nm) and oxide's stretching force constant F (Nm^{-1}) for the oxides; TeO_2 and PbO (Lambson et al., 1984; El-Mallawany, 1990)	60
4.8 The network bonds number per unit volume (n_b), bond compression bulk modulus (K_{bc}), ratio (K_{bc}/K_e), average cross-link density (\bar{n}_c), average force constant (F), the average ring diameter (l) of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glasses	61
4.9 Unit volume dissociation energy and packing factor of TeO_2 and PbO oxides	62
4.10 Unit volume dissociation energy (G_t), molecular weight (M_i), packing factor (C_t), packing density (V_t), elastic moduli (E_m , G_m , K_m) and Poison's ratio (σ_m) according to Makishima-Mackenzie model for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	64
4.11 Packing density (C_t), elastic moduli (E_r , G_r , K_r) and the Poisson's ratio (σ_r) based on Rocherulle model (Rocherulle et al., 1989) for $(\text{PbO})_x (\text{TeO}_2)$	64

4.12	Experimental elastic moduli (E_e , G_e , K_e), bond compression model (E_{bc} , G_{bc} , K_{bc}), Rocherulle model (E_r , G_r , K_r) and Makishima and Mackenzie model (E_m , G_m , K_m) for Young's, shear and bulk modulus, respectively and Poisson's ratio (σ_e , σ_{bc} , σ_r , σ_m) of binary $(PbO)_x (TeO_2)_{1-x}$ glass system	68
4.13	Direct band gap, Indirect band gap, and Urbach energy of $(PbO)_x (TeO_2)_{(1-x)}$ glass system	70
4.14	Refractive index, Molar refraction, and Molar polarizability of $(PbO)_x (TeO_2)_{(1-x)}$ glass system	76
4.15	Molar Electronic Polarizability, Electronic Polarizability, Refractive Index Based Oxide Ion Polarizability and Band Gap Based Oxide Ion Polarizability for $PbO)_x (TeO_2)_{(1-x)}$ glass system	77
4.16	Reflection loss, Optical transmission coefficient, Metallization criterion, and Dielectric constant of $(PbO)_x (TeO_2)_{(1-x)}$ glass, $x = 0-30$ PbO mole %	80
4.17	Pb^{2+} Ion Concentration, Polaron Radius, Inter-ionic distance, Field Strength of Pb^{2+} ions Yield, Linear Dielectric Susceptibility and Optical Basicity of $(PbO)_x (TeO_2)_{(1-x)}$, glass system	83
4.18	The glass transition temperature T_g , crystallization temperatures T_c , melting temperatures T_m , glass stability range ΔT and Hruby's parameter K_g of $(PbO)_x (TeO_2)_{(1-x)}$ glass, $x = 0-30$ PbO mole %	86
4.19	Experimental values of density (ρ), molar volume (V_m) and oxygen packing density (OPD) of ternary $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system	87
4.20	Experimental values of longitudinal (V_L), shear (V_S) and Mean (U_m) ultrasonic wave velocity of ternary $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system	91
4.21	Experimental values of elastic moduli, Poisson's ratio (σ) and (E/G) ratio of ternary $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system	92
4.22	Microhardness (H), Debye temperature (θ_D), softening temperature (T_s), Thermal expansion coefficient (α_p) fractal bond connectivity (d) and acoustic impedance (Z) of ternary $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system	94

4.23	Coordination number (n_f), cation–anion bond length r (nm), crosslink density per cation (n_c) and Stretching force constant of the oxide $F(N.m^{-1})$ of (TeO_2) , (PbO) and (ZnO) oxides (El-Mallawany, 1998)	96
4.24	The number of network bonds per unit volume (n_b), the average crosslink density (\bar{n}_c), the experimentally determined bulk modulus (K_e), the bond compression bulk modulus (K_{bc}), the ratio (K_{bc}/K_e), the average force constant (\bar{F}), the average ring diameter (l) of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glasses	97
4.25	Packing factor and dissociation energy per unit volume of (TeO_2) , (PbO) and (ZnO) oxides	100
4.26	Molecular weight (M_i), Dissociation energy per unit volume (G_t), packing density (V_t), and elastic moduli (E_m , G_m , K_m) calculated Poisson's ratio (σ_m) according to Makishima–Mackenzie mode of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glasses	101
4.27	Packing density (C_t), Calculated elastic moduli (E_r , G_r , K_r) and Poisson's ratio (σ_r) for $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system from the model of Rocherulle et al. model	101
4.28	Experimental elastic moduli (E_e , G_e , K_e), bond compression model (E_{bc} , G_{bc} , K_{bc}), Rocherulle model (E_r , G_r , K_r) and Makishima and Mackenzie model (E_m , G_m , K_m) for Young's, shear and bulk modulus, respectively and Poisson's ratio (σ_e , σ_{bc} , σ_r , σ_m) of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glass system	103
4.29	Direct band gap, Indirect band gap and Urbach energy of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	107
4.30	Refractive index, Molar refraction and Molar polarizability of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	112
4.31	Molar Electronic Polarizability, Electronic Polarizability, Refractive Index Based Oxide Ion Polarizability and Band Gap Based Oxide Ion Polarizability for $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$, glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	114
4.32	Reflection loss, Optical transmission coefficient, Metallization criterion, and Dielectric constant of $[ZnO]_x [(TeO_2)_{0.7}-PbO]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	116

- 4.33 Zinc Ion Concentration (N), Polaron radius (R_p), Inter-ionic distance (R_i), Field Strength of Zn^{2+} ions Yield (F), Linear Dielectric Susceptibility (χ) and Optical Basicity for $[ZnO]_x [(TeO_2)_{0.7}PbO]_{0.3}]_{1-x}$ glasses 118
- 4.34 The glass transition temperature T_g , crystallization temperatures T_c , melting temperatures T_m , glass stability range ΔT and Hruby's parameter K_g of $[ZnO]_x [(TeO_2)_{0.7}PbO]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol% 122



LIST OF FIGURES

Figure	Page
2.1 The volume-temperature diagram for a glass forming liquid	8
3.1 Schematic diagram summarizes the glass making process shows the steps of the glass samples preparation process	22
3.2 The DSC thermogram	42
4.1 XRD patterns of $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glasses	45
4.2 XRD patterns of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses	45
4.3 (a1-a2) SEM images, (b1-b2) EDX pattern of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glasses, $x = 0, 0.30 \text{ mol}\%$	48
4.4 (a) SEM, (b) EDX pattern of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses, $x = 0, 15 \text{ mol}\%$	49
4.5 Samples of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glasses, $x = 0, 0.10, 0.15, 0.20, 0.25, 0.30 \text{ mol}\%$	49
4.6 Density (ρ) and molar volume (V_m) of the $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glasses	50
4.7 Oxygen packing density of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glass, $x = 0-0.30 \text{ PbO mol}\%$	51
4.8 The FTIR spectra of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glass system	52
4.9 TeO_4 concentration of $(\text{PbO})_x (\text{TeO}_2)_{1-x}$ glass system	52
4.10 Ultrasonic velocities (Longitudinal (V_L) and shear (V_S)) of $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glasses	54
4.11 Elastic moduli longitudinal (L), shear (G), bulk modulus (K) and Young's (E) vs. the mole percentage of (PbO) for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ tellurite glasses	55
4.12 Poisson's and (E/G) ratio variation against PbO mol% for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	57

4.13	The fractal bond connectivity (d) and microhardness (H) variation with PbO mol% in $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	58
4.14	Variation of softening and Debye temperature variation with PbO mol% in $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	59
4.15	The variation of the K_{bc}/K_e ratio on the ring diameter of the $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	62
4.16	The packing density variation with PbO mol% for $(\text{PbO})_x$	63
4.17	Packing density variation with PbO mol% in binary $(\text{PbO})_x$	65
4.18	a,b,c : Calculated elastic moduli of $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, $x = 0- 0.30$ mol%	67
4.19	UV absorption spectra $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glasses with $x= 0,0.10,0.15,0.20,0.25,0.30$ mol%	69
4.20	a-f: The optical band gap of $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	73
4.21	Energy gap values for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$ glass system	74
4.22	Refractive index of $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$ glass system	75
4.23	Molar refraction R_m and molecular polarizability α_m for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	76
4.24	Variation of the Electronic Polarizability and Molar Electronic Polarizability against the PbO mol% for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glasses	78
4.25	Metallization criterion M and dielectric constant ϵ for $(\text{PbO})_x (\text{TeO}_2)_{(1-x)}$ glass system	79
4.26	Reflection loss R_L and optical transmission T for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, glass system	81
4.27	Dielectric and Optical Dielectric Constants for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, glass system	81
4.28	Polaron radius for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, glass system	82
4.29	Variation of Linear Dielectric Susceptibility and Optical Basicity with PbO mol% for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, glasses	83
4.30	The DSC Pattern for $(\text{PbO})_x(\text{TeO}_2)_{(1-x)}$, glass system with 0% PbO	85

4.31	Variation of transition temperature of glass with different contents of PbO	86
4.32	Density (ρ) and molar volume (V_m) of the $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ glasses	88
4.33	Oxygen packing density of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^x$, $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	89
4.34	FTIR spectra of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	90
4.35	Ultrasonic velocities (Longitudinal (V_L) and shear (V_S)) of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ glasses	91
4.36	Elastic moduli longitudinal (L), shear (G), bulk modulus (K) and Young's (E) vs. the mole percentage of (ZnO) for $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ tellurite glasses	92
4.37	Variation of Poisson's and (E/G) ratio for glass samples with different mol % of (ZnO)	93
4.38	Variation of Debye and softening temperature for glass samples with different mol % of ZnO	95
4.39	Microhardness (H) and fractal bond connectivity (d) of ternary $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ glass system	96
4.40	Variation of stretching force constant with ZnO (mol%) content	98
4.41	Variation of the ratio (K_{bc}/K_e) of the glass system $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ with the mol% of the ZnO	98
4.42	Dependence of the ratio (K_{bc}/K_e) on the ring diameter of the glass system $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$	99
4.43	Dependence of the packing density of ternary glass system $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ on the percentage of the modified ZnO	100
4.44	Dependence of the packing density of ternary glass system $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ on the percentage of the modified ZnO	102
4.45	Dependence of bulk modulus (K_e, K_m, K_{bc} , and K_r) on (ZnO) (mol %) concentration	104
4.46	Dependence of shear modulus (G_e, G_m, G_{bc} and G_r) on (ZnO) (mol %) concentration in $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}^{1-x}$ glass	104

4.47	Dependence of Young's modulus (E_e , E_m , E_{bc} and E_r) on (ZnO) (mol %) concentration in $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glass	105
4.48	UV absorption spectra $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	106
4.49	a-e: The optical band gap of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses	110
4.50	Energy gap values for $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	110
4.51	Refractive index of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	111
4.52	Molar refraction, and Molar polarizability of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$,	112
4.53	Variations of the Molar Electronic Polarizability and Electronic Polarizability with ZnO mol% for $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$, glasses with $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	113
4.54	Metallization criterion, and Dielectric constant of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$, $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	115
4.55	Optical Transmission Coefficient of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$,	116
4.56	Reflection Loss of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$, $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	117
4.57	Dielectric (ϵ) and Optical Dielectric Constants (ϵ_{opt}) of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$, $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	117
4.58	Polaron Radius of $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$, $x = 0.15, 0.17, 0.20, 0.22, 0.25$ mol%	119
4.59	Variations of the Linear Dielectric Susceptibility and Optical Basicity with ZnO Mol% for $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ glasses	119
4.60	The DSC Pattern of the $[\text{ZnO}]_x [(\text{TeO}_2)_{0.7}\text{-PbO}]_{0.3}]_{1-x}$ Glass System	120
4.61	Variation of transition temperature of glass with different contents of ZnO	121

LIST OF ABBREVIATIONS AND SYMBOLS

XRD	X-Ray diffraction
FTIR	Fourier transform infrared
UV-Vis	Ultraviolet-Visible
SEM	Scanning electron microscopy
EDX	Energy dispersive X-ray spectroscopy
OPD	Oxygen packing density
BO	Bridging oxygen non
NBO	Non-bridging oxygen
DSC	Differential scanning calorimetry
DTA	Differential thermal analysis
IR	Infrared
TeO ₂	Tellurium oxide
PbO	Lead oxide
ZnO	Zinc oxide
Al ₂ O ₃	Aluminum oxide
P	Density
V _m	Molar volume
V _L	Longitudinal velocity
V _s	Shear velocity
U _m	Mean ultrasonic velocity
E	Young's modulus
K	Bulk modulus
L	Longitudinal modulus
G	Shear modulus

θ_D	Debye temperature
D	Fractal bond connectivity
Z	Acoustic impedance
α_P	Thermal expansion coefficient
H	Microhardness
Σ	poisson's ratio
T_g	Glass transition temperature
T_s	Softening temperature
T_C	Crystallization temperature
ΔT	Glass stability
T_m	Melting temperature
K_g	Hruby's parameter
K_B	Boltzmann's constant
h	Plank's constant
\bar{F}	Average stretching force constant
n_b	Number of network bond per unit volume
n_r	Number of network bonds per glass formula unit
F	Calculated stretching force constant
n_c	Number of cross-link per cation in oxide
N_c	Number of cation per glass formula unit
η	Total number of cations per glass formula unit
L	External ring diameter
V_t	Packing density
G_t	Dissociation energy
λ	Wavelength
A	Absorption

E_{opt}	Energy band gap
t	Sample thickness
n	Refractive index
R_m	Molar refractive index
M	Metallization criterion
ϵ	Dielectric constant
ϵ_{op}	Optical dielectric constant
R_L	Reflection loss
T	Transmission coefficient
α_m	Molar polarizability
R_p	Polaron radius

CHAPTER 1

INTRODUCTION

1.1 Research background

The physical and chemical properties of glass constituents have a strong influence on the elastic and structural properties of a glass system (Arulmozhi and Sheelarani, 2011). Since the strength of non-crystalline solid materials is always proportional to their elastic modulus, the strength of the materials can be estimated indirectly by studying their elastic properties (Zaid et al., 2011). The elastic constant of glassy materials provides information about a material's structure since elastic constants, interatomic forces, and potentials are directly related to electric constants (Zhou et al., 2014; Sidek et al., 2013; Bootjomchai et al., 2014; El-Aal and Afifi, 2009). The microstructural feature, network former, and behaviour of a modifier are influenced by acoustical parameters (Halimah et al., 2010; Sidek et al., 2014; El-Moneim, 2016). Of late, ultrasonic technique has become a very important tool in the evaluation of the acoustical parameters of glasses, such as elastic modulus, microhardness, Poisson's ratio, ultrasonic attenuation coefficient, and Debye temperature (Abd El-Moneim, 2014; Abd El-Moneim, 2012). Bridge et al. proposed that the longitudinal and shear velocities of ultrasonic waves and glass elastic constants can be evaluated using semi-empirical relations and theories (Bridge et al., 1983). Elastic modulus is computed using three structural parameters, i.e. number of bonds per unit volume, average stretching force constant, and average crosslink density per glass formula unit. Attempts have been made to estimate the Poisson ratio and elastic modulus of glasses from the unit volume dissociation energy and packing density of constituent chemical oxides of a glass network (Makishima and Mackenzie, 1973; Makishima and Mackenzie, 1975; Rocherulle et al., 1989; Abd El-Moneim and Alenezzy, 2016).

Researchers have been giving more attention to TeO_2 in comparison to other oxide glasses by virtue of their promising potential and quality in the high UV to IR transmission range, high refractive index, low phonon energy, high transmissivity of infrared radiation, and superior thermal stability and chemical durability (Wang et al., 2009; Stambouli et al., 2012). The prospect of using tellurite-based glasses in the fields of technology and science is very promising due to the excellent transparency of their visible and infrared regions, low melting temperature, high refractive index (~ 2), high dielectric constant, and high electrical conductivity (El-mallawany, 2002; Suehara et al., 1995; Villegas and Navarro, 2007; Kaur et al., 2010). Even though research in this field has been very successful, there are still some problems which need to be solved with regard to the strong up-conversion luminescence and low tensile strength of tellurite glasses. These problems have an adverse effect on the performance of tellurite glass devices in many practical applications (Halimah et al., 2010). Some of the identified drawbacks can be

eliminated by adding heavy metal oxides to the matrix of the glasses (Kaky et al., 2017). Researchers have been investigating the effects of incorporating heavy metal oxide such as PbO in tellurite glasses due to their ability to increase the density of tellurite glasses as well as improve their thermal and optical properties with regard to polarizability and thermal stability (Kaur et al., 2010). Tellurite glasses which have been incorporated with heavy metal oxides have shown promising properties in various optical applications. Studies have been done to understand the physical properties of various composition of tellurite glasses (Uma et al., 2018; Da Silva et al., 2018; Sun et al., 2018; Merzliakov et al., 2018; Rao et al., 2018; El-Mallawany et al., 2017; Moawad et al., 2004; Hager et al., 2011; Hager et al., 2010; El-Zaidia et al., 1985; Hussain et al., 2009; El-Mallawany, 2000; Mirgorodsky et al., 2012; Elkholy and El-Mallawany, 1995; Bouchaour et al., 2005).

The addition of ZnO to lead tellurite glass matrix have been shown to alter the physical and elastic properties of the glasses. Another area of study that is quite interesting is the use of multiple glass modifier, i.e. PbO and ZnO, to impart distinct properties to the network of tellurite glasses. Hence, the present work seeks to elucidate the structural and elastic properties of ternary zinc lead tellurite glasses modified with varying concentrations of ZnO, and to compare the values of theoretical elastic properties with the experimentally measured data of the investigated glasses.

1.2 Tellurite glass application

Various research have been carried out to explore the benefits of using tellurite glasses in different applications. Tellurite glasses have a high refractive index, low processing temperature, wide range of transmission (35 – 500 nm), high corrosion resistance, and good glass stability (El-Mallawany, 1998). They are also among the oxide glass formers which have the lowest vibrational energy (about 780 cm^{-1}). The high refractive index of tellurite glasses increase their local field correction at the rare-earth ion site, which in consequence improve their radiative transition rate (Lousteau et al., 2012).

When mixed with an appropriate amount of GeO_2 , tellurite glasses exhibits photo-induced phenomena which result in improved photo-structural of the refractive index (Pereira et al., 2017). This makes them a good solid electrolyte by virtue of the high ionic conductivity of the glasses. The addition GeO_2 to a tellurite glass network has been observed to improve its fiber-drawing ability, thereby making it a good optical fiber option in communication technology (Gouraud et al., 2015). When combined with Sb_2O_3 , tellurite glasses exhibits high polarizability, which in turn improve their refractive index. These glasses have a promising potential in various 3-D photonics applications, such as specific devices for nonlinear optics (Ultrafast power limiters and optical switches) (Pereira et al., 2016). Tellurite glasses doped with Er^{3+} , Nd^{3+} , Ho^{3+} or other rare earth ions have been found to be excellent materials for application

in high performance fiber amplifiers, fiber lasers, wavelength converting devices, and planar waveguide (Yousef, 2013; Linganna et al., 2016; Yanmin et al., 2015).

1.3 Problem statement

In the past several years many studies have been carried out in the fields of glass science and technology in the effort to find solutions or to improve the utilization of glass materials in domestic, scientific, industrial and technological applications (Dousti et al., 2015). Tellurite glasses are very important in many fields, such as in the mechanical, electrical, optical, and medical industries (Annapoorani et al., 2015; Umar et al., 2017). Researchers have investigated the physical, structural, elastic, thermal, mechanical, morphological, and optical properties of tellurite glasses, in addition to other properties, to find ways of fulfilling the specific requirements in various application (Gouraud et al., 2015).

Ma and co-authors fabricated $\text{Er}^{3+}/\text{Ho}^{3+}$ co-doped fluoro-tellurite glasses and investigated their utilization in laser applications (Ma et al., 2013). Another recent investigation incorporated rice husk in borotellurite glasses doped with erbium to determine the physical, structural and optical properties of the glasses for possible application in the erbium doped fiber amplifier (EDFA) technology (Umar et al., 2017). Many research which aim to improve the optical, elastic, and thermal characteristics of various tellurite glasses through the addition of different chemicals have been carried out. These include the work of Kaky et al., (2017) which explore the thermal, structural and optical absorption characteristics of tellurite rich glasses doped with heavy metal oxides. Awshah et al. (2017) studied the effect of Neodymium NPs on the elastic properties of zinc-tellurite glasses.

Several investigations have been done on tellurite glasses incorporated with varying amounts of either PbO or ZnO or both. Silva et al., (2001) studied the structural characteristics of a binary system lead tellurite. The glasses were analyzed using DSC, Raman spectroscopy, and XRD analysis. In 2010, Eraiah reported the results of a study on the optical properties of samarium doped lead-tellurite glasses. (Yousef et al., 2016) studied the optical and elastic characteristics of tellurite glasses, $75\text{TeO}_2\text{-}10\text{Nb}_2\text{O}_5\text{-}10\text{ZnO}\text{-}5\text{PbO}$, doped with Er^{3+} , while (Sidek et al., 2013) investigated the thermal properties of a binary zinc tellurite glasses. (Veeraiah, 1997) reported the results of a study of the elastic properties of $\text{ZnF}_2\text{-PbO-TeO}_2$ glass doped with varying amounts of rare earth. Tagiara and other partners examined the structural and thermal characteristics of a binary zinc tellurite glass (Tagiara et al., 2017). (Munoz et al., 2009) analyzed the characteristics of ternary tellurite glasses incorporated with tungsten and lead using XRD, FTIR, DTA and UV-V.

Plenty of attention has been given to heavy metal oxides, for instance PbO doping, in recent years by virtue of their effects on the structural (density), optical, mechanical and thermal characteristics of tellurite glasses. Not many research have

investigated the physical, structural, elastic, and optical characteristics of the binary lead tellurite (PbO-TeO_2) and ternary zinc lead tellurite (PbO-ZnO-TeO_2) glass systems, which means that there is a dearth of information on these types of glasses.

1.4 Research objectives

The present study aims to:

1. Examine how the incorporation of lead affect the structural, elastic, optical, and thermal characteristics of binary tellurite glass.
2. Examine how the incorporation of ZnO affect the structural, elastic, optical, and thermal characteristics of binary lead tellurite glass.
3. Elucidate the elastic modulus of tellurite glass system, compute the elastic modulus using the Makishima-Mackenzie, bond compression, and Rocherulle models, and correlate them with the experimental values.

1.5 Scope of the study

The scope of the study is as follow:

- 1) A series of precursor glass were prepared based on the stoichiometric equation of binary PbO-TeO_2 and ternary $\text{PbO-TeO}_2\text{-ZnO}$ glasses using the melt-quenching technique.
- 2) The physical, structural, elastic, optical, and thermal characteristics of the glasses were analyzed using ultrasonic technique, XRD, EDX/SEM, FTIR, DSC, and UV-Vis.
- 3) The elastic constants of the glass systems were predicted, and the microhardness, acoustic impedance, thermal expansion coefficient, softening temperature, and Debye temperature were computed by employing the experimental data.
- 4) The Makishima-Mackenzie theory, bond compression model, and Rocherulle model were used to compare the experimental values of the elastic modulus of the glasses with the theoretically determined values.
- 5) DSC spectroscopy was used to determine the glass transition (T_g) and crystallization (T_c) temperatures.
- 6) The coefficient of absorption, optical band gap energy, Urbach energy, refractive index, molar refractive index, and polarizability of the glasses were obtained using absorption spectra.

1.6 Importance of the study

Unlike conventional silicate, phosphate and borate glasses, tellurite glasses have higher electronic polarizability due to the high polarizability of the Te^{4+} ions and the lone electrons pair in the valence shell which is highly polarizable (Ersundu and Ersundu, 2016). TeO_2 based glasses with high transparency visible region have been reported to have high refractive index and are good candidates for EDFA, and laser and high speed optical switches (Pereira et al., 2016).

Many fields of science and technology, particularly the fields of optical fiber amplifiers, and bulk and waveguide lasers, are giving increasing attention to tellurite glasses doped with rare earth ions.

Rare earth doped tellurite glasses are currently used in infrared optical devices and visible optical devices; for instance, white light emitting diode (w-LED) are frequently used in automobile headlights and backlight (Annapurna et al., 2016).

The incorporation of heavy metal oxide, such as lead or bismuth, in glasses are the best alternative in the glass structure of rare earth doped hosts. Heavy metal oxide glasses are used to fabricate optical fibers in order to impart the fibres with very good environmental stability and lower optical loss in comparison to the use of fluorides. This makes them more suitable for use in short distance photonic applications (Poor et al., 2013).

The addition of ZnO to borotellurite glass network have been proven to accord the network with better stability as well as enhanced glass forming ability. Glasses containing PbO and ZnO have been shown to have a high third order nonlinear optical susceptibility and long infrared cut-off, making them suitable for application in photonics as components for infrared transmission (Kundu et al., 2014).

This work attempts to leverage the potentials of tellurium oxide, which has high polarizability, high refractive index, high dielectric constant, low phonon energy, good electrical conductivity, large third order non-linear susceptibility chemical durability and wide band infrared transmittance, and improve its properties by adding PbO and ZnO to accord the glasses with better stability and better glass forming ability. These glasses are expected to have very promising potential in optical fibers and photonic applications.

1.7 Thesis Organization

This thesis comprises five chapters, introduction, literature review, methodology, results and discussion, and conclusion. The thesis is structured as follows:

Chapter 1 gives a description of tellurite glass, i.e. PbO-TeO₂ glass, PbO-TeO₂-ZnO glass; it also presents the problem statements, objectives, scope and importance of the study. The theory of glass as well as previous and current investigation by other researchers are discussed in Chapter 2. Chapter 3 discusses the basic theory of glass formation, theories of elasticity, theories optics and thermal. This chapter also presents a detail description of sample preparation, design of the experiment, and measurement techniques. Chapter 4 presents the results, discussion and analysis of the glasses studied in this work. Finally, Chapter 5 presents the conclusion and suggestion for future research.

REFERENCES

- Abd El-Moneim, A. (2012). Correlation between acoustical and compositional parameters of borate and tellurite glasses. *Materials Chemistry and Physics*, 135(2–3), 653–657.
- Abd El-Moneim, A. (2014). Correlation between acoustical and structural parameters in some oxide glasses. *Journal of Non-Crystalline Solids*, 405, 141–147.
- Abd El-moneim, A. (2016). Correlation between acoustical and compositional properties of binary and multi- component oxide glasses Correlation between acoustical and compositional properties of binary and multi-component oxide glasses. *Materials Chemistry and Physics*, 173(April), 372–378.
- Abd El-Moneim, A. (2016). Phosphate-based glasses: Prediction of acoustical properties. *Physica B: Condensed Matter*, 487, 53–60.
- Abd El-Moneim, A., & D. Alenezy, M. (2016). Structural and acoustical properties of lead silicate glasses doped with alkali and alkaline earth oxides. *Phys. Chem. Glasses: Eur. J. Glass Sci. Technol. B*, 54 (5), 232–237.
- Afifi, H., & Marzouk, S. (2003). Ultrasonic velocity and elastic moduli of heavy metal tellurite glasses. *Materials Chemistry and Physics*, 80(2), 517–523.
- Afifi, H., Marzouk, S., & Abd el Aal, N. (2007). Ultrasonic characterization of heavy metal $\text{TeO}_2\text{-WO}_3\text{-PbO}$ glasses below room temperature. *Physica B: Condensed Matter*, 390(1–2), 65–70.
- Ahmmad, S. kareem, Samee, M. A., Edukondalu, A., & Rahman, S. (2012). Physical and optical properties of zinc arsenic tellurite glasses. *Results in Physics*, 2, 175–181.
- Al-Ani, S. K. J., Hogarth, C. A., & El-Malawany, R. A. (1985). A study of optical absorption in tellurite and tungsten-tellurite glasses. *Journal of Materials Science*, 20(2), 661–667.
- Annapoorani, K., Maheshvaran, K., Arunkumar, S., Murthy, N. S., & Marimuthu, K. (2015). Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy Structural and luminescence behavior of Er^{3+} ions doped Barium tellurofluoroborate glasses. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 135, 1090–1098.
- Annapurna, C. B., Mahamuda, S., Venkateswarlu, M., Swapna, K., Rao, A. S., & Prakash, G. V. (2016). Dy^{3+} ions doped single and mixed alkali fluoro tungsten tellurite glasses for LASER and white LED applications. *Optical Materials*, 62, 569–577.

- Arulmozhi, K. T., & Sheelarani, R. (2011). ANN modeling for the prediction of elastic moduli of ternary glass systems using physicochemical properties of the oxide components. *Journal of Non-Crystalline Solids*, 357(16–17), 3272–3277.
- Awshah, A. A. A., Kamari, H. M., Tim, C. K., Shah, N. M., Alazoumi, S. H., Aliyu, U. S., & Abd Azis, M. N. (2017). Effect of Neodymium Nanoparticles on Elastic Properties of Zinc-Tellurite Glass System. *Advances in Materials Science and Engineering*, 2017, 7.
- Azianty, S., & Yahya, A. K. (2013). Enhancement of elastic properties by WO₃ partial replacement of TeO₂ in ternary (80 - X)TeO₂-20PbO-xWO₃ glass system. *Journal of Non-Crystalline Solids*, 378, 234–240.
- Azianty, S., Yahya, A. K., & Halimah, M. K. (2012). Effects of Fe₂O₃ replacement of ZnO on elastic and structural properties of 80TeO₂-(20-x)ZnO-xFe₂O₃ tellurite glass system. *Journal of Non-Crystalline Solids*, 358(12–13), 1562–1568.
- Azlan, M. N., Halimah, M. K., Shafinas, S. Z., & Daud, W. M. (2014). Polarizability and optical basicity of Er³⁺ ions doped tellurite based glasses. *Chalcogenide Letters*, 11(7), 319–335.
- Azlan, M. N., Halimah, M. K., Shafinas, S. Z., & Daud, W. M. (2015). Electronic polarizability of zinc borotellurite glass system containing erbium nanoparticles. *Materials Express*, 5(3), 211–218.
- Azlan, M. N., Halimah, M. K., & Sidek, H. A. A. (2017). Linear and nonlinear optical properties of erbium doped zinc borotellurite glass system. *Journal of Luminescence*, 181, 400–406.
- Azlan, M. N., Halimah, M. K., Shafinas, S. Z., Daud, W. M., & Sidek, H. A. A. (2014). Influence of Erbium Concentration on Spectroscopic Properties of Tellurite. *Solid State Science and Technology*, 22(1-2), 148–156.
- Azuraida, A., Halimah, M. K., Sidek, A. A., Azurahaman, C. A. C., Iskandar, S. M., Ishak, M., & Nurazlin, A. (2015). Comparative studies of bismuth and barium boro-tellurite glass system: Structural and optical properties. *Chalcogenide Letters*, 12(10), 497–503.
- Bahari, H., Aziz, S. H. A., Kamari, H. M., Yunus, W. M. M., & Adikan, F. R. M. (2012). The effect of bismuth on the structure and mechanical properties of GeO₂-PbO-Bi₂O₃ ternary bulk glass system. *Journal of the Ceramic Society of Japan*, 120(1403), 280–285.
- Balda, R., Fernandez, J., Arriandiaga, M. A., & Fernandez-Navarro, J. M. (2007). Spectroscopy and frequency upconversion in Nd³⁺-doped TeO₂-TiO₂-Nb₂O₅ glass. *Journal of Physics: Condensed Matter*, 19(8), 086223–086234.

- Basavapoornima, C., Linganna, K., Kesavulu, C. R., Ju, S., Kim, B. H., Han, W., & Jayasankar, C. K. (2017). Spectroscopic and pump power dependent upconversion studies of Er^{3+} -doped lead phosphate glasses for photonic applications. *Journal of Alloys and Compounds*, 699, 959–968.
- Bellucci, D., Sola, A., Salvatori, R., Anesi, A., Chiarini, L., & Cannillo, V. (2017). Role of magnesium oxide and strontium oxide as modifiers in silicate-based bioactive glasses: Effects on thermal behaviour, mechanical properties and in-vitro bioactivity. *Materials Science & Engineering C*, 72, 566–575.
- Bernard, V. (2001). Absorption of UV– visible light. In *Molecular Fluorescence: Principles and Applications*, 8, 20–33.
- Berwal, N., Dhankhar, S., Sharma, P., Kundu, R. S., Punia, R., & Kishore, N. (2017). Physical, structural and optical characterization of silicate modified bismuth-borate-tellurite glasses. *Journal of Molecular Structure*, 1127, 636–644.
- Begum, A. N., & Rajendran, V. (2007). Structure investigation of TeO_2 -BaO glass employing ultrasonic study. *Materials Letters*, 61(11–12), 2143–2146.
- Bhatia, B., Meena, S. L., Parihar, V., & Poonia, M. (2015). Optical Basicity and Polarizability of Nd^{3+} -Doped Bismuth Borate Glasses. *New Journal of Glass and Ceramics*, 5, 44–52.
- Bogue, R., & Sladek, R. J. (1990). Elasticity and thermal expansivity of $(\text{AgI})(\text{AgPO}_3)$ glasses. *Physical Review B*, 42(8), 5280–5288.
- Bootjomchai, C., Laopaiboon, R., Pencharee, S., & Laopaiboon, J. (2014). Elastic moduli of borosilicate glasses doped with heavy metal oxides. *Journal of Non-Crystalline Solids*, 388, 37–45.
- Bouchaour, Z. K., Poulain, M., Belhadji, M., Hager, I., & El Mallawany, R. (2005). New oxyfluoroniobate glasses. *Journal of Non-Crystalline Solids*, 351(10), 818–825.
- Bounakhla, M., & Tahri, M. (2014). X-Ray Fluorescence Analytical Techniques. *National Center for Energy Sciences and Nuclear Techniques, Rabat, Morocco*.
- Bourhis, K., Massera, J., Petit, L., Ihalainen, H., Fargues, A., Cardinal, T., Hupab, L., Hupab, M., Dussauzee, M., Rodriguez, V., Boussard, C., Plédelf, C., Bureau, B., Roiland, C., Ferraris, M. (2015). Influence of P_2O_5 and Al_2O_3 content on the structure of erbium-doped borosilicate glasses and on their physical, thermal, optical and luminescence properties. *Materials Research Bulletin*, 63, 41–50.

- Bridge, B., & Higazy, A. A. (1986). A model of the compositional dependence of the elastic moduli of polycomponent oxide glasses. *Physics and chemistry of glasses*, 27(1), 1-14.
- Bridge, B., Patel, N. D., & Waters, D. N. (1983). On the Elastic Constants and Structure of the Pure Inorganic Oxide Glasses. *Physica Status Solidi (A)*, 77(2), 655–668.
- Byrnes, S. J. F. (2008). Basic theory and phenomenology of polarons. *Department of Physics, University of California at Berkeley, Berkeley, CA 94720*, 2–6.
- Canioni, L., Martin, M.-O., Bousquet, B., & Sarger, L. (1998). Precise measurements and analysis of linear and nonlinear optical properties of glass materials near 1.5 μm . *Optics Communications*, 151, 241-246.
- Çelikbilek, M., Ersundu, A. E., & Aydin, S. (2013). Preparation and characterization of $\text{TeO}_2\text{-WO}_3\text{-Li}_2\text{O}$ glasses. *Journal of Non-Crystalline Solids*, 378, 247–253.
- Çelikbilek, M., Erçin Ersundu, A., & Aydin, S. (2013). Glass formation and characterization studies in the $\text{TeO}_2\text{-WO}_3\text{-Na}_2\text{O}$ system. *Journal of the American Ceramic Society*, 96(5), 1470-1476.
- Chakraborty, S., Satou, H., & Sakata, H. (1997). Direct current conductivity and oxygen gas-sensing properties of ironantimonytellurite glasses. *Journal of Applied Physics Journal of Applied Physics*, 82(11), 5520–5525.
- Chanshetti, U. B., Shelke, V. A., Jadhav, S. M., Shankarwar, S. G., Chondhekar, T. K., Shankarwar, A. G., Sudarsan, V., Jogad, M. S. (2011). Density and molar volume studies of phosphate glasses. *Facta Universitatis - Series: Physics, Chemistry and Technology*, 9(1), 29–36.
- Chen, H., Liu, Y. H., Zhou, Y. F., Zhang, Q. Y., & Jiang, Z. H. (2005). Spectroscopic properties of Er^{3+} doped $\text{TeO}_2\text{-BaO}$ (Li_2O , Na_2O)- La_2O_3 glasses for 1.5- μm optical amplifiers. *Journal of Non-Crystalline Solids*, 351, 3060-3064.
- Chiang, Y.-M., Birnie, D. P., & Kingery, W. D. (1996). Physical ceramics : principles for ceramic science and engineering. *Choice Reviews Online Choice Reviews Online*, 34(03), 34–1566.
- Chimalawong, P., Kaewkhao, J., Kittiauchawal, T., Kedkaew, C., & Limsuwan, P. (2010). Optical Properties of the $\text{SiO}_2\text{-Na}_2\text{O-CaO-Nd}_2\text{O}_3$ Glasses. *American Journal of Applied Sciences*, 7(4), 584–589.
- Christian, G. (1996). Principles of Spectroscopy. In *Analytical Chemistry* (6th ed., p. 104). wiley.

- Da Silva, D. S., Wetter, N. U., de Rossi, W., Kassab, L. R. P., & Samad, R. E. (2018). Production and characterization of femtosecond laser-written double line waveguides in heavy metal oxide glasses. *OPTMAT Optical Materials*, 75, 267–273.
- Damas, P., Coelho, J., Hungerford, G., & Hussain, N. S. (2012). Structural studies of lithium boro tellurite glasses doped with praseodymium and samarium oxides. *Materials Research Bulletin*, 47(11), 3489–3494.
- Devreese, J. T. (1996). Polarons. *Encyclopedia of Applied Physics* (14th ed., pp. 383–409). Wiley-VCH Publishers, Inc.
- Dhara, A., Mishra, R. K., Shukla, R., Valsala, T. P., Sudarsan, V., Tyagi, A. K., & Kaushik, C. P. (2016). A comparative study on the structural aspects of sodium borosilicate glasses and barium borosilicate glasses : Effect of Al₂O₃ addition. *Journal of Non-Crystalline Solids*, 447, 283–289.
- Dharma, J., & Pisal, A. (2012). Simple Method of Measuring the Band Gap Energy Value of TiO₂ in the Powder Form using a UV / Vis / NIR Spectrometer. *PerkinElmer, Inc.* Shelton, CT USA.
- Dias, J. D. M., Melo, G. H. A., Lodi, T. A., Carvalho, J. O., Filho, P. F. F., & Barboza, M. J. (2016). Thermal and structural properties of Nd₂O₃-doped calcium boroaluminate glasses. *Journal of Rare Earths*, 34(5), 521–528.
- Dimitrov, V., & Komatsu, T. (2005). Classification of oxide glasses: A polarizability approach. *Journal of Solid State Chemistry*, 178(3), 831–846.
- Dimitrov, V., & Komatsu, T. (2010). An Interpretation of Optical Properties of Oxides and Oxide Glasses in Terms of the Electronic Ion Polarizability and Average Single Bond Strength. *Journal of the University of Chemical Technology and Metallurgy*, 45(3), 219–250.
- Dimitrov, V., & Komatsu, T. (2013). Electronic Polarizability , Optical Basicity And Single Bond Strength Of Oxide Glasses. *Journal of Chemical Technology and Metallurgy*, 48(6), 549–554.
- Dimitrov, V., & Sakka, S. (1996). Electronic oxide polarizability and optical basicity of simple oxides. I. *Journal of Applied Physics*, 79(3), 1736–40.
- Dimitrov, V., & Sakka, S. (1996). Linear and nonlinear optical properties of simple oxides. II. *Journal of Applied Physics*, 79(3), 1741–1745.
- Djokovic, V., Luyt, A. S., Dramic, M. D., Dramićanin, M. D., Antić, Ž., & Djoković, V. (2009). Morphology, mechanical and thermal properties of composites of polypropylene and nanostructured wollastonite filler. *Polymer Testing*, 28(3), 348–356.

- Dousti, M. R., Amjad, M., Hussain, A., Amjad, R. J., Shaukat, S. F., Carlos, F. D. S., ... Branch, T. (2015). Enhanced Near-Infrared Emission Of Er^{3+} -Doped Tellurite. *Chalcogenide Letters*, 12(3), 123–128.
- Down, C. (1988). Elastic properties of binary, ternary and quaternary rare earth tellurite glasses. *Journal of Materials Science Letters*, 7, 870–874.
- Dresselhaus, M. S. (1966). Solid State Physics Part II: Optical Properties of Solids. *Proceedings of the International School of Physics*, 198.
- Edukondalu, A., Sathe, V., Rahman, S., & Siva Kumar, K. (2014). Thermal, mechanical and Raman studies on mixed alkali borotungstate glasses. *Physica B: Condensed Matter*, 438, 120-126.
- Eevon, C., Halimah, M. K., Zakaria, A., Azurahaman, C. A. C., Azlan, M. N., & Faznny, M. F. (2016). Linear and nonlinear optical properties of Gd^{3+} doped zinc borotellurite glasses for all-optical switching applications. *Results in Physics*, 6, 761-766.
- El-Aal, N. S. A. B. D., & Afifi, H. A. (2009). Structure and Ultrasonic Properties of Vanadium Tellurite Glasses Containing Copper Oxide. *Archives of Acoustics*, 34(4), 641–654.
- El-Deen, L. M. S., Salhi, M. S. A., & Elkholy, M. M. (2008). IR and UV spectral studies for rare earths-doped tellurite glasses. *Journal of Alloys and Compounds*, 465, 333–339.
- El-Diasty, F., Abdel Wahab, F. A., & Abdel-Baki, M. (2006). Optical band gap studies on lithium aluminum silicate glasses doped with Cr^{3+} ions. *Journal of Applied Physics*, 100(9), 93511.
- El-Mallawany, R. (1990). Quantitative analysis of elastic moduli of tellurite glasses. *J. Mater. Res. Journal of Materials Research*, 5(10), 2218–2222.
- El-Mallawany, R. (1993). Longitudinal elastic constants of tellurite glasses. *Journal of Applied Physics*, 73(10), 4878–4880.
- El-Mallawany, R. (1998). Tellurite glasses Part 1. Elastic properties. *Materials Chemistry and Physics*, 53(2), 93–120.
- El-Mallawany, R. (1992). The optical properties of tellurite glasses. *Journal of Applied Physics*, 72(5), 1774–1777.
- El-Mallawany, R. (2000). Specific Heat Capacity of Semiconducting Glasses: Binary Vanadium Tellurite. *PSSA Physica Status Solidi (A)*, 177(2), 439–444.
- El-Mallawany, R. A. H. (2002). *Tellurite Glasses Handbook: Physical Properties and Data*. Florida: CRC Press LLC, 2000 N.W. Corporate Blvd., Boca Raton, Florida 33431.

- El-mallawany, R. a H. (2002). *Tellurite glasses handbook. Solid State Physics*.
- El-Mallawany, R., Abdalla, M. D., & Ahmed, I. A. (2008). New tellurite glass: Optical properties. *Materials Chemistry and Physics*, 109(2–3), 291–296.
- El-Mallawany, R., Abousehly, A., El-Rahamani, A. A., & Yousef, E. (1998). Radiation effect on the ultrasonic attenuation and internal friction of tellurite glasses. *MAC Materials Chemistry and Physics*, 52(2), 161–165.
- El-Mallawany, R., Abousehly, A., & Yousef, E. (2000). Elastic moduli of tricomponent tellurite glasses $\text{TeO}_2\text{-V}_2\text{O}_5\text{-Ag}_2\text{O}$. *Journal of Materials Science Letters*, 19(5), 409–411.
- El-Mallawany, R., & Afifi, H. (2013). Elastic moduli and crosslinking of some tellurite glass systems. *Materials Chemistry and Physics Materials Chemistry and Physics*, 143(1), 11–14.
- El-Mallawany, R., El-Khoshkhany, N., & Afifi, H. (2006). Ultrasonic studies of $(\text{TeO}_2)_{50}\text{-(V}_2\text{O}_5)_{50}\text{-x(TiO}_2)_x$ glasses. *Materials Chemistry and Physics*, 95(2), 321–327.
- El-Mallawany, R., Gaafar, M. S., & Veeraiah, N. (2015). Evaluation of bulk modulus and ring diameter of some tellurite glass systems. *Chalcogenide Letters*, 12(2), 67–74.
- El-Mallawany, R. A., & Saunders, G. A. (1988). Elastic properties of binary, ternary and quaternary rare earth tellurite glasses. *Journal of materials science letters*, 7(8), 870-874.
- El-Mallawany, R., Sayyed, M. I., & Dong, M. G. (2017). Comparative shielding properties of some tellurite glasses: Part 2. *NOC Journal of Non-Crystalline Solids*, 474, 16–23.
- El-Zaidia, M. M., Ammar, A. A., & El-Mallawany, R. A. (1985). Infra-Red Spectra, Electron Spin Resonance Spectra, and Density of $(\text{TeO}_2)_{100-x}\text{-(WO}_3)_x$ and $(\text{TeO}_2)_{100-x}\text{-(ZnCl}_2)_x$ Glasses. *PSSA Physica Status Solidi (A)*, 91(2), 637–642.
- Elazoumi, S. H., Sidek, H. A. A., Rammah, Y. S., El-Mallawany, R., Halimah, M. K., Matori, K. A., & Zaid, M. H. M. (2017). Effect of PbO on optical properties of tellurite glass. *Results in Physics Results in Physics*, 8, 16–25.
- Elkholy, M. M., & El-Mallawany, R. a. (1995). A.c. conductivity of tellurite glasses. *Materials Chemistry and Physics*, 40, 163-167.
- Elkhoshkhany, N. (2014). Optical Properties of $\text{WO}_3\text{-PbO}$ Tellurite Glasses Doped with Rare Earths, *Journal of Chemistry and Chemical Engineering*, 8, 11–20.

- Elkoshkhany, N., Abbas, R., Gaafar, M. S., & El-Mallawany, R. (2015). Elastic properties of quaternary $\text{TeO}_2\text{-ZnO-Nb}_2\text{O}_5\text{-Gd}_2\text{O}_3$ glasses. *Ceramics International*, 41(8), 9862–9866.
- Eraiah, B. (2010). Optical properties of lead-tellurite glasses doped with samarium trioxide. *Bulletin of Materials Science*, 33(4), 391–394.
- Ersundu, M. C., & Ersundu, A. E. (2016). Structure and crystallization kinetics of lithium tellurite glasses. *Journal of Non-Crystalline Solids*, 453, 150–157.
- Escobar-Alarcón, L., Arrieta, A., Camps, E., Muhl, S., Rodil, S., & Viguera-Santiago, E. (2007). An alternative procedure for the determination of the optical band gap and thickness of amorphous carbon nitride thin films. *Applied Surface Science*, 254(1 SPEC. ISS.), 412–415.
- Ferrer, N. (2007). Applications of Fourier transform infrared spectroscopy. Barcelona, Spain.
- Fiocco, L., Ferroni, L., Gardin, C., Zavan, B., Secco, M., Matthews, S., & Bernardo, E. (2016). Wollastonite-diopside glass-ceramic foams from supercritical carbon dioxide-assisted extrusion of a silicone resin and inorganic fillers. *Journal of Non-Crystalline Solids*, 443, 33–38.
- Físicas, F. D. E. C., Martín, D. M., & Mu, D. (2010). TeO_2 -based film glasses for photonic applications: structural and optical properties.
- Gaafar, M. S., Abdeen, M. A. M., & Marzouk, S. Y. (2011). Structural investigation and simulation of acoustic properties of some tellurite glasses using artificial intelligence technique. *Journal of Alloys and Compounds*, 509(8), 3566–3575.
- Gaafar, M. S., & Azzam, Y. A. (2015). Acoustic relaxation of some lead niobium tellurite glasses. *Bulletin of Materials Science*, 38(1), 119–128.
- Gautam, C. R., Kumar, D., Parkash, O., & Singh, P. (2013). Synthesis, IR, crystallization and dielectric study of (Pb, Sr)TiO₃ borosilicate glass-ceramics. *Bulletin of Materials Science*, 36(3), 461–469.
- Gaur, A., & Rastogi, V. (2016). Design and Analysis of Annulus Core Few Mode EDFA for Modal Gain Equalization. *IEEE Photonics Technology Letters*, 28(10), 1057–1060.
- Gebavi, H., Taher, M., Losteau, J., Milanese, D., Taccheo, S., Chimica, I., Schulzgen, A., Ferraris M., Peyghambarian N. (2010). Spectroscopy of Yb: Tm doped tellurite glasses for efficient infrared fiber laser. *Optical Components and Materials*, 7598(0), 1–8.
- Ghobadi, N. (2013). Band gap determination using absorption spectrum fitting procedure. *International Nano Letters*, 3(1), 1-4.

- Golding, B. (1976). Ultrasonic Properties of Glass at Ultra Low Temperatures. *1976 Ultrasonics Symposium*, 692–701.
- Gouraud, F., Chotard, T., & Karray, R. (2015). Structural , mechanical and optical investigations in the TeO₂-rich part of the TeO₂-GeO₂-ZnO ternary glass system. *Solid State Sciences*, 40, 20–30.
- Guo, H., Wang, Y., Gong, Y., Yin, H., Mo, Z., Tang, Y., & Chi, L. (2016). Optical band gap and photoluminescence in heavily Tb³⁺ doped GeO₂-B₂O₃-SiO₂-Ga₂O₃ magneto-optical glasses, 686, 635–640.
- Hager, I. Z., & El-Mallawany, R. (2010). Preparation and structural studies in the (70 - x)TeO₂-20WO₃-10Li₂O-xLn₂O₃ glasses. *Journal of Materials Science*, 45, 897-905.
- Hager, I. Z., El-Mallawany, R., & Bulou, A. (2011). Luminescence spectra and optical properties of TeO₂-WO₃-Li₂O glasses doped with Nd, Sm and Er rare earth ions. *PHYSB Physica B: Physics of Condensed Matter*, 406(4), 972–980.
- Hajer, S. S., Halimah, M. K., Azmi, Z., & Azlan, M. N. (2014). Optical properties of Zinc-Borotellurite doped samarium. *Chalcogenide Letters*, 11(11), 553–566.
- Halimah, M. K., Daud, W. M., & Sidek, H. A. A. (2010). Effect of AgI addition on elastic properties of quaternary tellurite glass systems. *Chalcogenide Letters*, 7(11), 613–620.
- Halimah, M. K., Daud, W. M., & Sidek, H. A. A. (2010). Effect of AgI Addition on Elastic Properties of Quaternary tellurite Glass Systems. *Chalcogenide Letters*, 7(11), 613–620.
- Halimah, M. K., Daud, W. M., & Sidek, H. A. A. (2016). Elastic properties of TeO₂-B₂O₃-Ag₂O glasses, (July), 807–813.
- Halimah, M. K., Faznny, M. F., Azlan, M. N., & Sidek, H. A. A. (2017). Optical basicity and electronic polarizability of zinc borotellurite glass doped La³⁺ ions. *Results in Physics*, 7, 581–589.
- Halimah, M., Sidek, H., Daud, W., Zainul, H., Talib, Z., Zaidan, A., Zainal, A., Mansor, H. (2005). Ultrasonic Study and Physical Properties of Borotellurite Glasses. *American Journal of Applied Sciences*, 2(11), 1541–1546.
- Herzfeld, K. F. (1927). On Atomic Properties which make an Element a Metal. *Phys. Rev. Physical Review*, 29(5), 701–705.
- Honma, T., Sato, R., Benino, Y., Komatsu, T., & Dimitrov, V. (2000). Electronic polarizability, optical basicity and XPS spectra of Sb₂O₃-B₂O₃ glasses. *Journal of Non-Crystalline Solids*, 272(1), 1–13.

- Hooi, M. O., Halimah, M. K., & Wan-Yusoff, W. M. D. (2012). Optical properties of bismuth tellurite based glass. *International Journal of Molecular Sciences*, 13(4), 4623–4631.
- Hussain, N. S., Hungerford, G., El-Mallawany, R., Gomes, M. J. M., Lopes, M. A., Ali, N., Santos, J.D., Buddhudu, S. (2009). Absorption and Emission Analysis of RE³⁺(Sm³⁺ and Dy³⁺): Lithium Boro Tellurite Glasses. *Journal of Nanoscience and Nanotechnology*, 9(6), 3672–3677.
- Inaba, S., Fujino, S., & Morinaga, K. (1999). Young's Modulus and Compositional Parameters of Oxide Glasses. *J. Am. Ceram. Soc.*, 82, 3501–3507.
- Inusa, E. I., Umar, A., Samuel, A. O., & Sanusi, L. M. (2014). Particle-in-cell simulation of electrostatic plasma in one dimension. *Scholars Research Library Archives*, 6(4), 143–156.
- Isakandar, S. M., Halimah, M. K., Daud, W. M., Sidek, H. A. A., & Khairul Zaman, M. D. (2012). Thermal Stability And Physical Properties Of PbO-B₂O₃-TeO₂ Glass System. *Solid State Science and Technology*, 20(1), 48–55.
- Jha, A., Richards, B., Jose, G., Teddy-Fernandez, T., Joshi, P., Jiang, X., & Lousteau, J. (2012). Rare-earth ion doped TeO₂ and GeO₂ glasses as laser materials. *JPMS Progress in Materials Science*, 57(8), 1426–1491.
- Jose, R., Arai, Y., & Ohishi, Y. (2008). Optical properties of MoO₃ containing tellurite glasses. *Applied Physics Letters*, 93(16), 161901.
- Kabalci, I., Ozen, G., Ovecoglu, M. L., & Sennaroglu, A. (2006). Thermal study and linear optical properties of (1-x)TeO₂-(x)PbF₂ (x = 0.10, 0.15 and 0.25 mol) glasses. *Journal of Alloys and Compounds*, 419(1–2), 294–298.
- Kaky, K. M., Lakshminarayana, G., Baki, S. O., Kityk, I. V., Taufiq-Yap, Y. H., & Mahdi, M. A. (2017). Structural, thermal and optical absorption features of heavy metal oxides doped tellurite rich glasses. *Results in Physics*, 7, 166–174.
- Kaur, A., Khanna, A., González-barriuso, M., González, F., Chen, B., & Mas-nmr, A. (2017). Short-range structure and thermal properties of alumino-tellurite glasses. *Journal of Non-Crystalline Solids*, 470, 14-18.
- Kaur, A., Khanna, A., González, F., Pesquera, C., & Chen, B. (2016). Structural , optical , dielectric and thermal properties of molybdenum tellurite and borotellurite glasses. *Journal of Non-Crystalline Solids*, 444, 1–10.
- Kaur, A., Khanna, A., Pesquera, C., González, F., & Sathe, V. (2010). Preparation and characterization of lead and zinc tellurite glasses. *Journal of Non-Crystalline Solids*, 356(18–19), 864–872.

- Kesavulu, C. R., Kim, H. J., Lee, S. W., Kaewkhao, J., Wantana, N., Kothan, S., & Kaewjaeng, S. (2016). Influence of Er^{3+} ion concentration on optical and photoluminescence properties of Er^{3+} -doped gadolinium-calcium silica borate glasses. *Journal of Alloys and Compounds*, 683, 590–598.
- Khan, A. A. (1999). *Ultrasonic Testing of Materials at Level 2 Ultrasonic Testing of Materials at Level 2* (10th ed.). Vienna: International Atomic Energy Agency.
- Kim, S.-H., & Yoko, T. (1995). Nonlinear Optical Properties of TeO_2 -Based Glasses: $\text{MO}_x\text{-TeO}_2$ ($M = \text{Sc, Ti, V, Nb, Mo, Ta, and W}$) Binary Glasses. *J American Ceramic Society Journal of the American Ceramic Society*, 78(4), 1061–1065.
- King, P. L., Ramsey, M. S., McMillan, P. F., & Swayze, G. King, P. L., Ramsey, M. S., McMillan, P. F., & Swayze, G. (2004). Laboratory Fourier transform infrared spectroscopy methods for geologic samples. *Infrared Spectroscopy in Geochemistry, Exploration, and Remote Sensing*, 33, 57-91.
- Komatsu, T., Ito, N., Honma, T., & Dimitrov, V. (2012). Temperature dependence of refractive index and electronic polarizability of RO-TeO_2 glasses ($R = \text{Mg, Ba, Zn}$) *Solid State Sciences*, 14(10), 1419–1425.
- Kumar, A. (2010). Preparation of Apatite-Wollastonite and Phlogophite Glass Ceramic and Study of Its Properties. *National institute of technology rourkela*, (Doctoral dissertation), 1-34.
- Kumar, K., Rai, S. B., & Rai, D. K. (2007). Enhancement of luminescence properties in Er^{3+} doped $\text{TeO}_2\text{-Na}_2\text{O-PbX}$ ($X = \text{O and F}$) ternary glasses. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy*, 66(4-5), 1052-1057.
- Kundu, R. S., Dhankhar, S., Punia, R., Nanda, K., & Kishore, N. (2014). Bismuth modified physical, structural and optical properties of mid-IR transparent zinc boro-tellurite glasses. *Journal of Alloys and Compounds*, 587, 66–73.
- Lakshminarayana, G., Kaky, K. M., Baki, S. O., Lira, A., Nayar, P., Kityk, I. V., & Mahdi, M. A. (2017). Physical, structural, thermal, and optical spectroscopy studies of $\text{TeO}_2\text{-B}_2\text{O}_3\text{-MoO}_3\text{-ZnO-R}_2\text{O}$ ($R = \text{Li, Na, and K}$)/ MO ($M = \text{Mg, Ca, and Pb}$) glasses. *Journal of Alloys and Compounds*, 690, 799–816.
- Lambson, E. F., Saunders, G. A., Bridge, B., & El-Mallawany, R. A. (1984). The Elastic Behaviour Of TeO_2 Glass Under Uniaxial And Hydrostatic Pressure. *Journal of Non-Crystalline Solids*, 69, 117–133.
- Landau, L. D., & Pekar, S. I. (1948). Effective Mass Of A Polaron. *Zh. Eksp. Teor. Fiz.*, 18(5), 71–74.
- Lane, K. (1985). 2 . Crystalline phases of the Co-P-O system, 72, 81–108.

- Laopaiboon, R., & Bootjomchai, C. (2015). Physical properties and thermoluminescence of glasses designed for radiation dosimetry measurements. *Materials and Design*, 80, 20–27.
- Laopaiboon, R., Laopaiboon, J., Pencharee, S., Nontachat, S., & Bootjomchai, C. (2016). The effects of gamma irradiation on the elastic properties of soda lime glass doped with cerium oxide. *Journal of Alloys and Compounds*, 666, 292–300.
- Levelut, C., Le Parc, R., Faivre, A., & Champagnon, B. (2006). Influence of thermal history on the structure and properties of silicate glasses. *Journal of Non-Crystalline Solids Journal of Non-Crystalline Solids*, 352(42–49), 4495–4499.
- Levy, R. A. (1968). *Principles of Solid State Physics*. Ohio, U.S.A: Academic Press.
- Lin, G., Tan, D., Luo, F., Chen, D., Zhao, Q., & Qiu, J. (2011). Linear and nonlinear optical properties of glasses doped with Bi nanoparticles. *Journal of Non-Crystalline Solids*, 357(11–13), 2312–2315.
- Linda, D., Duclère, J.-R., Hayakawa, T., Dutreilh-Colas, M., Cardinal, T., Mirgorodsky, A., Kabadou, A., Thomas, P. (2013). Optical properties of tellurite glasses elaborated within the TeO₂–Tl₂O–Ag₂O and TeO₂–ZnO–Ag₂O ternary systems. *Journal of Alloys and Compounds*, 561, 151–160.
- Linganna, K., Narro-garcía, R., Desirena, H., Rosa, E. De, Basavapoornima, C., Venkatramu, V., & Jayasankar, C. K. (2016). Effect of P₂O₅ addition on structural and luminescence properties of Nd³⁺-doped tellurite glasses. *Journal of Alloys and Compounds*, 684, 322–327.
- Lousteau, J., Boetti, N., Chiasera, A., Ferrari, M., Abrate, S., Scarciglia, G., Venturello, A., Milanese, D. (2012). Er³⁺ and Ce³⁺ Co-Doped Tellurite Optical Fiber for Lasers And Amplifiers in the Near Infrared Wavelength Region: Fabrication, Optical Characterization and Prospects. *IEEE Photonics Journal*, 4(1), 194–204.
- Ma, Y., Huang, F., Hu, L., & Zhang, J. (2013). Er³⁺/Ho³⁺-Codoped Fluorotellurite Glasses for 2.7 μm Fiber Laser Materials. *Fibers*, 1(2), 11–20.
- Maheshvaran, K., Linganna, K., & Marimuthu, K. (2011). Composition dependent structural and optical properties of Sm³⁺ doped boro-tellurite glasses. *Journal of Luminescence*, 131(12), 2746–2753.
- Mahraz, Z. A., Sahar, M. R., & Ghoshal, S. K. (2014). Band gap and polarizability of boro-tellurite glass: Influence of erbium ions. *Journal of Molecular Structure Journal*, 1072(1), 238–241.
- Makishima, A., & Mackenzie, J. D. (1973). Direct calculation of Young's modulus of glass. *Journal of Non-Crystalline Solids*, 12(1), 35–45.

- Makishima, A., & Mackenzie, J. D. (1975). Calculation of bulk modulus, shear modulus and Poisson's ratio of glass. *Journal of Non-Crystalline Solids*, 17(2), 147–157.
- Mansour, E. (2012). FTIR spectra of pseudo-binary sodium borate glasses containing TeO₂. *Journal of Molecular Structure*, 1014, 1–6.
- Marzouk, S. Y. (2010). The acoustic properties of borosilicate glass affected by oxide of rare earth gadolinium. *Physica B: Condensed Matter*, 405(16), 3395–3400.
- Matori, K. A., Sayyed, M. I., Sidek, H. A. A., Zaid, M. H. M., & Singh, V. P. (2017). Comprehensive study on physical, elastic and shielding properties of lead zinc phosphate glasses. *Journal of Non-Crystalline Solids*, 457(December), 97–103.
- Meena, S. L., & Bhatia, B. (2016). Polarizability and Optical Basicity of Er³⁺ Ions Doped Zinc Lithium Bismuth Borate Glasses, 6(October), 175–183.
- Merzliakov, M. A., Kouhar, V. V., Malashkevich, G. E., & Pestryakov, E. V. (2018). Spectroscopy of Yb-doped tungstentellurite glass and assessment of its lasing properties. *Optical Materials*, 75(5), 142–149.
- Mirgorodsky, A. A., Colas, M., Smirnov, M., Merle-Méjean, T., El-Mallawany, R., Thomas, P. (2012). Structural peculiarities and Raman spectra of TeO₂/WO₃-based glasses: A fresh look at the problem. *Journal of Solid State Chemistry*, 190, 45–51.
- Mo, Z. X., Guo, H. W., Liu, P., Shen, Y. D., & Gao, D. N. (2016). Luminescence properties of magneto-optical glasses containing Tb³⁺ ions. *Journal of Alloys and Compounds*, 658, 967–972.
- Moawad, H. M. M., Jain, H., El-Mallawany, R., Ramadan, T., El-Sharbiny, M. (2004). Electrical Conductivity of Silver Vanadium Tellurite Glasses. *Journal of the American Ceramic Society*, 85(11), 2655–2659.
- Moelwyn-Hughes, E. A. (1961). *Physical Chemistry*, vols. 1 and 2 Pergamon Press. Oxford.
- Mohamed, E. A., Ahmad, F., & Aly, K. A. (2012). Effect of lithium addition on thermal and optical properties of zinc-tellurite glass. *Journal of Alloys and Compounds*, 538, 230–236.
- Moustafa, E. S., Saddeek, Y. B., & Shaaban, E. R. (2008). Structural and optical properties of lithium borobismuthate glasses. *Journal of Physics and Chemistry of Solids*, 69(9), 2281–2287.
- Mott, N. F., & Davis, E. A. (1971). *Electronic process in non-crystalline materials*. Oxford University Press, London, pp. 293.

- Mott, N. F., & Davis, E. A. (1979). *Electronic Process in Non-Crystalline Materials*. second edition (Clarendon., Clarendon Press), Oxford, UK.
- Munoz-Martín, D., Villegas, M. A., Gonzalo, J., & Fernández-Navarro, J. M. (2009). Characterisation of glasses in the TeO₂-WO₃-PbO system. *Journal of the European Ceramic Society*, 29(14), 2903–2913.
- Murawski, L. (1982). Electrical conductivity in iron-containing oxide glasses. *Journal of Materials Science*, 17(8), 2155-2163.
- Mustafa, I. S., Kamari, H. M., Wan Yusoff, W. M. D., Aziz, S. A., & Rahman, A. A. (2013). Structural and optical properties of lead-boro-tellurite glasses induced by Gamma-ray. *International Journal of Molecular Sciences*, 14(2), 3201–3214.
- Narayanan, M. K., & Shashikala, H. D. (2015). Thermal and optical properties of BaO–CaF₂–P₂O₅ glasses. *Journal of Non-Crystalline Solids*, 422, 6–11.
- Narayanan, R. A., & Zwanziger, J. W. (2003). The glass forming ability of tellurites: A rigid polytope approach. *Journal of Non-Crystalline Solids*, 316(2-3), 273-280.
- Narazaki, A., Tanaka, K., Hirao, K., & Soga, N. (1999). Induction and relaxation of optical second-order nonlinearity in tellurite glasses. *Journal of Applied Physics Journal of Applied Physics*, 85(4), 2046–2051.
- Nobarzad, A. E. K., Masoumi, K. M., & Heirdari, K. (2014). Phase Identification by X-ray diffraction.
- Noranizah, A., Azman, K., Azhan, H., Nurbaisyatul, E. S., & Mardhiah, A. (2014). Spectroscopic Properties of Rare Earth Ion Doped TeO₂-B₂O₃-PbO Glass. *Jurnal Teknologi (Sciences and Engineering)*, 69(6), 49–52.
- Owen, T. (1996). *Principles and applications of UV-visible spectroscopy*. Copyright Hewlett-Packard Company.
- Ozdanova, J., Ticha, H., & Tichy, L. (2007). Remark on the optical gap in ZnO-Bi₂O₃-TeO₂ glasses. *Journal of Non-Crystalline Solids*. 353(29), 2799-2802.
- Pandarinath, M. A., Upender, G., Rao, K. N., Babu, D. S. (2016). Thermal, optical and spectroscopic studies of boro-tellurite glass system containing ZnO. *Journal of Non-Crystalline Solids*, 433, 60–67.
- Paul, A., Roychoudhury, P., Mukherjee, S., & Basu, C. (2000). Ultrasonic study of (CuO)_x(TeO₂)_{1-x} glass system. *Journal of Non-Crystalline Solids*, 275, 83-92.
- Pavani, P. G., Sadhana, K., & Mouli, V. C. (2011). Optical, physical and structural studies of boro-zinc tellurite glasses. *Physica B: Condensed Matter*, 406(6–7), 1242–1247.

- Pavani, P. G., Suresh, S., & Mouli, V. C. (2011). Studies on boro cadmium tellurite glasses. *Optical Materials*, 34(1), 215–220.
- Pawar, P. P., Munishwar, S. R., Gautam, S., & Gedam, R. S. (2017). Physical, thermal, structural and optical properties of Dy³⁺ doped lithium aluminoborate glasses for bright W-LED. *Journal of Luminescence*, 183, 79–88.
- Pawar, P. P., Munishwar, S. R., & Gedam, R. S. (2017). Intense white light luminescent Dy³⁺ doped lithium borate glasses for W-LED : A correlation between physical , thermal , structural and optical properties. *Solid State Sciences*, 64, 41–50.
- Pereira, C., Barbosa, J., Cassanjes, F. C., Gonçalves, R. R., Ribeiro, S. J. L., & Poirier, G. (2016). Thermal , structural and optical properties of new TeO₂-Sb₂O₃-GeO₂ ternary glasses. *Optical Materials*, 62, 95–103.
- Pereira, C., Cassanjes, F. C., Barbosa, J. S., Gonçalves, R. R., Ribeiro, S. J. L., & Poirier, G. (2017). Structural and optical study of glasses in the TeO₂-GeO₂-PbF₂ ternary system. *Journal of Non-Crystalline Solids*, 463, 158–162.
- Poor, H. B., Aziz, H. A., & Zamiri, R. (2013). Ultrasonic and optical properties and emission of Er³⁺/Yb³⁺ doped lead bismuth-germanate glass affected by Bi⁺/Bi²⁺ ions. *Journal of Luminescence*, 143, 526–533.
- Qin, G., Jose, R., & Ohishi, Y. (2007). Stimulated Raman scattering in tellurite glasses as a potential system for slow light generation. *Journal of Applied Physics*, 101(9), 93109.
- Rachkovskaya, G. E., & Zakharevich, G. B. (2002). Vitrification, Properties, and Structure of Lead-Tellurite Borate Glasses. *Glass and Ceramics Glass and Ceramics*, 59(3–4), 123–126.
- Rada, S., Dehelean, A., & Culea, E. (2011). FTIR and UV-VIS spectroscopy investigations on the structure of the europium-lead-tellurate glasses. *Journal of Non-Crystalline Solids*, 357(16–17), 3070–3073.
- Rajendran, V., Palanivelu, N., Chaudhuri, B. K., & Goswami, K. (2003). Characterisation of semiconducting V₂O₅-Bi₂O₃-TeO₂ glasses through ultrasonic measurements. *Journal of Non-Crystalline Solids*, 320, 195-209.
- Rakov, N., Maciel, G. S., Sundheimer, M. L., Menezes, L. de S., Gomes, A. S. L., Messaddeq, Y., Cassanjes, F.C., Poirier, G., Ribeiro, S. J. L. (2002). Blue upconversion enhancement by a factor of 200 in Tm³⁺-doped tellurite glass by codoping with Nd³⁺ ions. *Journal of Applied Physics*, 92(10), 6337-6339.
- Ramadevudu, G., & Chary, M. N. (2017). Physical and spectroscopic studies of Cr³⁺ doped mixed alkaline earth oxide borate glasses. *Materials Chemistry and Physics*, 186, 382–389.

- Ramamoorthy, R. K., & Bhatnagar, A. K. (2015). Effect of ZnO and PbO/ZnO on structural and thermal properties of tellurite glasses. *Journal of Alloys and Compounds*, 623, 49–54.
- Rao, V. H., Prasad, P. S., Rao, P. V., Santos, L. F., & Veeraiyah, N. (2016). Influence of Sb₂O₃ on tellurite based glasses for photonic applications. *Journal of Alloys and Compounds*, 687, 898–905.
- Rao, V., Prasad, P.S., Babu, M. M., Rao, P. V., Satyanarayana, T., Santos, L. F & Veeraiyah, N. (2018). Spectroscopic studies of Dy³⁺ ion doped tellurite glasses for solid state lasers and white LEDs. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 188, 516–524.
- Rao, Y. R., Goud, K. K., Kumar, E. R., Reddy, C. S., Rao, B. A. (2014). Upconversion Luminescence in Er³⁺/Yb³⁺ Codoped Lead Bismuth Indium Borate Glasses. *International Journal of Recent Development in Engineering and Technology*, 3(1), 2347–6435.
- Rasool, S. N., Jamalayah, B. C., Suresh, K., Moorthy, L. R., & Jayasankar, C. K. (2017). Spectroscopic properties of Er³⁺-doped phosphate based glasses for broadband 1.54 μm emission. *Journal of Molecular Structure*, 1130, 837–843.
- Rawson, H. (1980). *Properties and applications of glass*. Amsterdam: North-Holland.
- Rocherulle, J., Ecolivet, C., Poulain, M., Verdier, P., & Laurent, Y. (1989). Elastic moduli of oxynitride glasses. Extension of Makishima and Mackenzie's theory. *Journal of Non-Crystalline Solids*, 108(2), 187–193.
- Rojek, M., Stabik, J., & Sokół, S. (2007). Fatigue and ultrasonic testing of epoxy-glass composites. *Journal of Achievements in Material and Manufacturing Engineering*, 20(1–2), 183–186.
- Reddy, R. R., Ahammed, Y. N., Azeem, P. A., Gopal, K. R., Ramakrishna, R. (2001). Electronic Polarizability and Optical Basicity Properties of Oxide Glasses Through Average Electronegativity. *Journal of Non-Crystalline Solids*, 286, 169–180.
- Saddeek, Y. B. (2004). Ultrasonic study and physical properties of some borate glasses. *Materials Chemistry and Physics*, 83(2–3), 222–228.
- Saddeek, Y. B., Afifi, H. A., & Abd El-Aal, N. S. (2007). Interpretation of mechanical properties and structure of TeO₂-Li₂O-B₂O₃ glasses. *Physica B: Condensed Matter*, 398(1), 1–7.
- Saddeek, Y. B., & Elsayed, N. Z. (2015). Structural and mechanical features of some lanthanum tellurite glasses, 465(September 2014), 460–465.

- Sahar, M. R., & Noordin, N. (1995). Oxychloride glasses based on the $\text{TeO}_2\text{-ZnO-ZnCl}_2$ system. *Journal of Non-Crystalline Solids*, *184*, 137–140.
- Samanta, B., Dutta, D., & Ghosh, S. (2017). Synthesis and different optical properties of Gd_2O_3 doped sodium zinc tellurite glasses. *Physica B: Condensed Matter*, *515*, 82–88.
- Sayyed, M. I., & El-Mallawany, R. (2017). Shielding properties of $(100-x)\text{TeO}_2-(x)\text{MoO}_3$ glasses. *Materials Chemistry and Physics*, *201*, 50–56.
- Scannell, G., Huang, L., & Rouxel, T. (2015). Elastic properties and indentation cracking behavior of $\text{Na}_2\text{O-TiO}_2\text{-SiO}_2$ glasses. *Journal of Non-Crystalline Solids*, *429*, 129–142.
- Selvi, S., Marimuthu, K., Murthy, N. S., & Muralidharan, G. (2016). Red light generation through the lead boro telluro phosphate glasses activated by Eu^{3+} ions. *Journal of Molecular Structure*, *1119*, 276–285.
- Shelby, J. E. (2005). Introduction to glass science and technology. Cambridge: Royal Society of Chemistry.
- Shen, S., Richards, B., & Jha, A. (2006). Enhancement in pump inversion efficiency at 980 nm in Er^{3+} , $\text{Er}^{3+}/\text{Eu}^{3+}$ and $\text{Er}^{3+}/\text{Ce}^{3+}$ doped tellurite glass fibers. *Optics Express*, *14*, 5050-5054.
- Sidebottom, D. L., Hruschka, M. a., Potter, B. G., & Brow, R. K. (1997). Structure and optical properties of rare earth-doped zinc oxyhalide tellurite glasses. *Journal of Non-Crystalline Solids*, *222*, 282-289.
- Sidek, H. A. A., El-Mallawany, R., Hariharan, K., & Rosmawati, S. (2014). Effect of concurrent ZnO addition and AlF_3 reduction on the elastic properties of tellurite based glass system. *Advances in Condensed Matter Physics*, *2014*.
- Sidek, H. A. A., El-Mallawany, R., Matori, K. A., & Halimah, M. K. (2016). Effect of PbO on the elastic behavior of $\text{ZnO-P}_2\text{O}_5$ glass systems H.A.A. *Results in Physics*, *6*, 449–455.
- Sidek, H. A. A., Halimah, M. K., Daud, W. M., Khamirul, A. M., Talib, Z. A., Zaidan, A. W., Zainal, A. S., Z Azmi, Z. A., Zulkifly, A., Halim, S. A. (2007). Quantitative Analysis on the Elastic Behaviour of Borotellurite Glass. *Journal of Wuhan University of Technology–Mater. Sci. Ed*, *12*, 863–867.
- Sidek, H. H. A., Kamari, H. M., & Hui, A. J. (2012). Quantitative Analysis on the Elastic Moduli of Borate and Phosphate Glasses Containing Lead Oxide. *Solid St Sci Technol Express*, *8*(1), 16–20.

- Sidek, H.H.A., Rosmawati, S., Azmi, B. Z. & Shaari, A. H. (2013). Effect of ZnO on the thermal properties of tellurite glass. *Advances in Condensed Matter Physics*, 2013, 1-6.
- Sidek, H.H.A., Rosmawati, S., Azmi, B. Z. & Shaari, A. H. (2013). Effect of Zinc on the Physical Properties of Tellurite Glass. *Journal of Applied Sciences*, 8(10), 1956–1961.
- Sidek, H. A. A., Rosmawati, S., Talib, Z. A., Halimah, M. K., & Halim, S. A. (2013). Effect of Zinc on the Elastic Behaviour of $(\text{TeO}_2)_{90}(\text{AlF}_3)_{10-x}(\text{ZnO})_x$ Glass System. *International Journal of Basic & Applied Sciences*, 90, 41–44.
- Sidek, H. A. A., Rosmawati, S., Yahya, A. K., Al-, S., Campus, T., Enstek, B., ... Alam, S. (2016). Characteristic Temperatures And Microhardness Of $(\text{ZnO})_x(\text{AlF}_3)_y(\text{TeO}_2)_z$ Tellurite Glass Systems. *Chalcogenide Letters*, 13(4), 169–176.
- Sidkey, M. A., El Aal, N. A., El Latif, L. A., & El Moneim, A. A. (2001). Correlation between some physical properties for tri-component tellurite glasses and low temperature ultrasonic relaxation. *Archives of Acoustics*, 26(4), , 361-369.
- Sidkey, M. A., Abd El-Moneim, A., & Abd El-Latif, L. (1999). Ultrasonic studies on ternary $\text{TeO}_2\text{-V}_2\text{O}_5\text{-Sm}_2\text{O}_3$ glasses. *Materials Chemistry and Physics*, 61(2), 103-109.
- Sidkey, M. A., El-Mallawany, R., Nakhla, R. I., & Abd El-Moneim, A. (1997). Ultrasonic Attenuation at Low Temperature of $\text{TeO}_2\text{-V}_2\text{O}_5$ Glasses. *PSSA Physica Status Solidi (A)*, 159(2), 397–404.
- Sidkey, M. A., El-moneim, A. A., & El-latif, L. A. (1999). Ultrasonic studies on ternary $\text{TeO}_2\text{-V}_2\text{O}_5\text{-Sm}_2\text{O}_3$ glasses. *Materials Chemistry and Physics*, 61(February), 103–109.
- Sidkey, M. A., El Mallawany, R. A., Abousehly, A. A., & Saddeek, Y. B. (2002). Relaxation of longitudinal ultrasonic waves in some tellurite glasses. *Materials Chemistry and Physics*, 74, 222-229.
- Sidkey, M. A., & Gaafar, M. S. (2004). Ultrasonic studies on network structure of ternary $\text{TeO}_2\text{-WO}_3\text{-K}_2\text{O}$ glass system. *Physica B: Condensed Matter*, 348(1–4), 46–55.
- Sidkey, M., & Afifib, H. (1994). Materials chemistry and attenuation. *Materials Chemistry and Physics*, 37 (1994), 197-200.
- Silva, M. A. P., Messaddeq, Y., Ribeiro, S. J. L., Poulain, M., Villain, F., & Briois, V. (2001). Structural studies on $\text{TeO}_2\text{-PbO}$ glasses. *Journal of Physics and Chemistry of Solids*, 62(6), 1055–1060.

- Souri, D. (2010). DSC and FTIR Spectra of Tellurite-Vanadate Glasses Containing Molybdenum. *Middle-East Journal of Scientific Research*, 5, 44-48.
- Souza, R. F., Alencar, M. A. R. C., Hickmann, J. M., Kobayashi, R., & Kassab, L. R. P. (2006). Femtosecond nonlinear optical properties of tellurite glasses. *Applied Physics Letters*, 89(17), 171917.
- Stambouli, W., Elhouichet, H., & Ferid, M. (2012). Study of thermal, structural and optical properties of tellurite glass with different TiO₂ composition. *Journal of Molecular Structure*, 1028, 39-43.
- Stuart, B. (2005). *Infrared Spectroscopy: Fundamentals and Applications*. Chichester, UK.: Springer.
- Suehara, S., Yamamoto, K., Hishita, S., Aizawa, T., & Inoue, S. (1995). Bonding nature in tellurite glass. *Physical Review B*, 51(21), 919-922.
- Sun, Y., Yang, Q., Wang, H., & Shao, Y. (2018). Sensitization of Ho³⁺ on the 2.7 μm emission of Er³⁺ in (Y_{0.9}La_{0.1})₂O₃ transparent ceramics. *Journal of Luminescence*, 194, 50-55.
- Sushama, D., & Predeep, P. (2014). Thermal and Optical Studies of Rare Earth Doped Tungsten-Tellurite Glasses. *International Journal of Applied Physics and Mathematics*, 4(2), 139-143.
- Swapna, K., Mahamuda, S., Venkateswarlu, M., Srinivasa Rao, A., Jayasimhadri, M., Shakya, S., & Prakash, G. V. (2015). Visible, Up-conversion and NIR (~1.5 μm) luminescence studies of Er³⁺ doped Zinc Alumino Bismuth Borate glasses. *Journal of Luminescence*, 163, 55-63.
- Tagiara, N. S., Palles, D., Simandiras, E. D., Psycharis, V., Kyritsis, A., & Kamitsos, E. I. (2017). Synthesis, thermal and structural properties of pure TeO₂ glass and zinc-tellurite glasses. *Journal of Non-Crystalline Solids*, 457, 116-125.
- Tanaka, K., Kashima, K., Hirao, K., & Soga, N. (1995). Second harmonic generation in electrically poled Li₂O-Nb₂O₅-TeO₂ glasses. *Journal of non-crystalline solids*, 185(1-2), 123-126.
- Tanner, D. B. (2013). Optical effects in solids. Lecture notes for PHY7097, Department of Physics, University of Florida USA.
- Thakur, V., Singh, A., Punia, R., Dahiya, S., & Singh, L. (2017). Structural properties and electrical transport characteristics of modified lithium borate glass ceramics. *Journal of Alloys and Compounds*, 696, 529-537.
- Thirumaran, S., & Karthikeyan, N. (2013). Structural Elucidation of Some Borate Glass Specimen by Employing Ultrasonic and Spectroscopic Studies. *Journal of Ceramics*, 2013, 1-10.

- Thirumaran, S., & Sathish, K. (2014). Spectroscopic investigations on structural characterization of borate glass specimen doped with transition metal ions. *Research Journal of Chemistry and Environment*, 18(10), 77–82.
- Thirumaran, S., & Sathish, K. (2015). Spectroscopic and ultrasonic investigations on structural characterization of borate glass specimen doped with transition metal ions. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy*, 147(10), 163–172.
- Thombre, D. B. (2014). The Estimation of Oxide Polarizability using the Electronegativity for $\text{Li}_2\text{O} : \text{B}_2\text{O}_3 : \text{SiO}_2$ Glass System. *International Journal of Science Engineering and Technology*, 1050(3), 1047–1050.
- Ticha, H., Schwarz, J., & Tichy, L. (2017). On the structural arrangement and optical band gap $(\text{PbO})_x(\text{ZnO})_{10}(\text{TeO}_2)_{90-x}$ glasses. *Journal of Non-Crystalline Solids*, 459, 63–67.
- Tauc, J. 1974. Optical properties of amorphous semiconductor. In *Amorphous and liquid semiconductors*, ed. J. Tauc, pp 159-220 Plenum Press, London and New York.
- Tauc, J., Grigorovici, R., & Vancu, A. (1966). Optical properties and electronic structure of amorphous germanium. *physica status solidi (b)*, 15(2), 627-637.
- Uma, V., Vijayakumar, M., Marimuthu, K., & Muralidharan, G. (2018). Luminescence and energy transfer studies on $\text{Sm}^{3+}/\text{Tb}^{3+}$ codoped telluroborate glasses for WLED applications. *MOLSTR Journal of Molecular Structure*, 1151, 266–276.
- Umair, M. M., Yahya, A. K., Halimah, M. K., & Sidek, H. A. A. (2015). Effects of Increasing Tungsten on Structural, Elastic and Optical Properties of $x\text{WO}_3-(40-x)\text{Ag}_2\text{O}-60\text{Te}_2\text{O}$ Glass System. *Journal of Materials Science & Technology*, 31(1), 83-90.
- Umar, S. A., Halimah, M. K., Chan, K. T., & Latif, A. A. (2017). Physical, structural and optical properties of erbium doped rice husk silicate borotellurite (Er-doped RHSBT) glasses. *Journal of Non-Crystalline Solids*, 472, 31-38.
- Umar, S. A., Halimah, M. K., Chan, K. T., & Latif, A. A. (2017). Polarizability, optical basicity and electric susceptibility of Er^{3+} doped silicate borotellurite glasses. *Journal of Non-Crystalline Solids*, 471, 101–109.
- Umar, S. A., Ibrahim, G. G., Ibrahim, I. E., Najib, M. U., & Liman, M. S. (2015). Simulation of heating in space plasma. *Pelagia Research Library*, 6(6), 103–113.
- Varshneya, A. K. (1994). *Fundamentals of inorganic glasses*. Boston: Academic Press.

- Veeraiah, N. (1997). Elastic properties of ZnF / -PbO-TeO₂ glasses doped with certain rare earth ions, *20*(5), 667–675.
- Viezbicke, B. D., Patel, S., Davis, B. E., & Birnie, D. P. (2015). Evaluation of the Tauc method for optical absorption edge determination: ZnO thin films as a model system. *Physica Status Solidi (B)*, *11*(8), 1700–1710.
- Villegas, M. A., & Navarro, J. M. F. (2007). Physical and structural properties of glasses in the TeO₂-TiO₂-Nb₂O₅ system. *Journal of the European Ceramic Society*, *27*(7), 2715–2723.
- Wakkad, M. M., Shokr, E. K., & Mohamed, S. H. (2000). Optical and calorimetric studies of Ge-Sb-Se glasses. *Journal of Non-Crystalline Solids*, *265*(1), 157–166.
- Wang, G., Zhang, J., Dai, S., Yang, J., & Jiang, Z. (2005). Thermal analyses, spectral characterization and structural interpretation of Yb³⁺ doped TeO₂-ZnO-ZnCl₂ glasses. *Physics Letters A*, *341*(1-4), 285-290.
- Wang, W. H. (2012). The elastic properties, elastic models and elastic perspectives of metallic glasses. *Progress in Materials Science*, *57*(3), 487–656.
- Wang, Y., Dai, S., Chen, F., Xu, T., & Nie, Q. (2009). Physical properties and optical band gap of new tellurite glasses within the TeO₂-Nb₂O₅-Bi₂O₃ system. *Materials Chemistry and Physics*, *113*(1), 407–411.
- Weber, M. J., Myers, J. D., & Blackburn, D. H. (1981). Optical properties of Nd³⁺ in tellurite and phosphotellurite glasses. *Journal of Applied Physics Journal of Applied Physics*, *52*(4), 2944–2949.
- Wunderlich, B. (2005). Thermal analysis of polymeric materials. *Springer Science & Business Media*.
- Yamane, M., & Asahara, Y. (2000). Glasses for photonics. Cambridge [England]; New York: Cambridge University Press.
- Yanmin, Y., Yanzhou, L. I. U., Peiqing, C. A. I., Maalej, R., & Seo, H. J. (2015). Thermal stability and spectroscopic properties of Ho³⁺ doped tellurite-borate glasses. *Journal of Rare Earths*, *33*(9), 939–945.
- Yonesaki, Y., Tanaka, K., Narazaki, A., Si, J., & Hirao, K. (2007). Relaxation phenomena in second-order nonlinearity of thermally and optically poled Nb₂O₅-TeO₂ glasses. *J. Phys. D: Appl. Phys. Journal of Physics D: Applied Physics*, *35*(16), 2026–2031.
- Yousef, E. S. (2013). Er³⁺ ions doped tellurite glasses with high thermal stability, elasticity, absorption intensity, emission cross section and their optical application. *Journal of Alloys and Compounds Journal*, *561*, 234–240.

- Yousef, E. S., El-Adawy, A., El Koshkhany, N., & Shaaban, E. R. (2006). Optical and acoustic properties of TeO₂/WO₃ glasses with small amount of additive ZrO₂. *Journal of Physics and Chemistry of Solids*, 67, 1649–1655.
- Yousef, E. S., Elokr, M. M., & Aboudeif, Y. M. (2016). Optical, elastic properties and DTA of TNZP host tellurite glasses doped with Er³⁺ ions. *Journal of Molecular Structure*, 1108, 257–262.
- Yousef, E. S. S. (2005). Characterization of oxyfluoride tellurite glasses through thermal, optical and ultrasonic measurements. *Journal of Physics D: Applied Physics*, 38(21), 3970–3975.
- Zaid, M. H. M., Matori, K. a, Wah, L. C., Sidek, H. a a, Halimah, M. K., Wahab, Z. a, & Azmi, B. Z. (2011). Elastic moduli prediction and correlation in soda lime silicate glasses containing ZnO. *International Journal of the Physical Sciences*, 6(6), 1404–1410.
- Zhao, X., Wang, X., Lin, H., & Wang, Z. (2008). A new approach to estimate refractive index, electronic polarizability, and optical basicity of binary oxide glasses. *Physica B: Condensed Matter*, 403, 2450-2460.
- Zhou, Y., Yang, Y., Huang, F., Ren, J., Yuan, S., & Chen, G. (2014). Characterization of new tellurite glasses and crystalline phases in the TeO₂-PbO-Bi₂O₃-B₂O₃ system. *Journal of Non-Crystalline Solids*, 386, 90–94.

BIODATA OF STUDENT

Salah Hassan Almohtar Alazoumi was born in Al Maamoura, Libya on 17th November 1977. He received early education at Al Maamoura primary school. Then he continued his secondary education at Wadi Alhai secondary school, Gharyan, Libya. He then continued his higher education at University of Gharyan, Libya and obtained his Bachelor of Science (1999), after which he worked as a demonstrator for three years and then proceeded to get his Master of Science degree (2008) at Tripoli University, Libya. In 2015 he secured a scholarship for a PhD study from the Libyan Government, which enabled him to start his PhD in Material Science at Universiti Putra Malaysia in the same year.



LIST OF PUBLICATIONS

Publications

- S. H. Alazoumi, H. A. A. Sidek, M. K. Halimah, K. A. Matori, M. H. M. Zaid, and A. A. Abdulbaset, "Synthesis and elastic properties of ternary ZnO-PbO-TeO₂ glasses," *Chalcogenide Letters*, vol. 14, no. 8, pp. 303–320, 2017.
- S. H. Elazoumi, H. A. A. Sidek, Y. S. Rammah, R. El-Mallawany, M. K. Halimah, K. A. Matori, and M. H. M. Zaid, "Effect of PbO on optical properties of tellurite glass," *Results in Physics*, vol. 8, pp. 16–25, 2017.
- Salah Hassan Alazoumi, Sidek Abdul Aziz, R. El-Mallawany, Umar Sa'ad Aliyu, Halimah Mohamed Kamari, Mohd Hafiz Mohd Mohd Zaid, Khamirul Amin Matori, "Optical properties of zinc lead tellurite glasses," *Results in Physics*, vol. 9, no. March, pp. 1371–1376, 2018.

Conferences

- S.H.Alazoumi, H. A. A. Sidek, M.K. Halimah, K.A. Matori "Synthesis and Elastic Moduli of PbO-TeO₂ System." Fundamental Science Congress, FSC2016, August 9-10, 2016 Universiti Putra Malaysia, 2016.



UNIVERSITI PUTRA MALAYSIA

STATUS CONFIRMATION FOR THESIS / PROJECT REPORT AND COPYRIGHT

ACADEMIC SESSION : _____

TITLE OF THESIS / PROJECT REPORT :

ELASTIC, OPTICAL AND THERMAL PROPERTIES OF ZINC-LEAD TELLURITE GLASS SYSTEMS

NAME OF STUDENT: SALAH HASSAN ALMOKHTAR ALAZOUMI

I acknowledge that the copyright and other intellectual property in the thesis/project report belonged to Universiti Putra Malaysia and I agree to allow this thesis/project report to be placed at the library under the following terms:

1. This thesis/project report is the property of Universiti Putra Malaysia.
2. The library of Universiti Putra Malaysia has the right to make copies for educational purposes only.
3. The library of Universiti Putra Malaysia is allowed to make copies of this thesis for academic exchange.

I declare that this thesis is classified as :

*Please tick (v)

CONFIDENTIAL

(Contain confidential information under Official Secret Act 1972).

RESTRICTED

(Contains restricted information as specified by the organization/institution where research was done).

OPEN ACCESS

I agree that my thesis/project report to be published as hard copy or online open access.

This thesis is submitted for :

PATENT

Embargo from _____ until _____
(date) (date)

Approved by:

(Signature of Student)
New IC No/ Passport No.:

(Signature of Chairman of Supervisory Committee)
Name:

Date :

Date :

[Note : If the thesis is CONFIDENTIAL or RESTRICTED, please attach with the letter from the organization/institution with period and reasons for confidentiality or restricted.]