

UNIVERSITI PUTRA MALAYSIA

AMENDING FORMULATED COMPOUND FERTILIZERS WITH ZEOLITE TO IMPROVE MAIZE (Zea mays L.) GROWTH AND NUTRIENT USE EFFICIENCY

MAGDALINA LIJA WELSON BLUE

FPSM 2013 10

MAGDALINA LIJA W. B. MASTER OF SCIENCE



AMENDING OF FORMULATED COMPOUND FERTILIZERS WITH ZEOLITE TO IMPROVE MAIZE (Zea mays L.) GROWTH AND NUTRIENT USE EFFICIENCY

MAGDALINA LIJA ANAK WELSON BLUE

MASTER OF SCIENCE UNIVERSITI PUTRA MALAYSIA

2013

2013



AMENDING OF FORMULATED COMPOUND FERTILIZERS WITH ZEOLITE TO IMPROVE MAIZE (*Zea mays* L.) GROWTH AND NUTRIENT USE EFFICIENCY

By

MAGDALINA LIJA ANAK WELSON BLUE

Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirement for the Degree of Master of Science

April 2013

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia

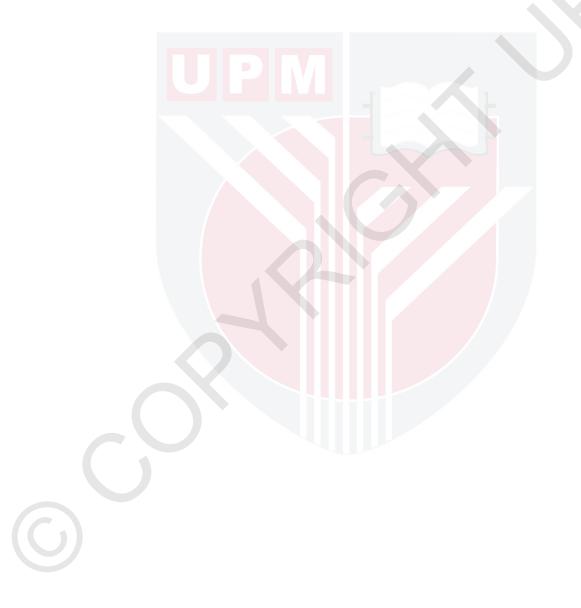


TABLE OF CONTENTS

APPRON DECLAF LIST OF LIST OF	NK WLEDGEMENTS /AL	Page ii v viii x xiii xiii xiii xiv xv
CHAPTE	RUPM	
1	INTRODUCTION	1
2	LITERATURE REVIEW 2.1 Compound Fertilizers 2.2 Zeolites	4
	2.2.1 History	5
	2.2.2 Characteristics 2.2.3 Applications	7 8
	2.2.3 Applications 2.3 Clinoptilolite Zeolite in Improving Nutrien Efficiency	
	 2.4 Effect of Zeolite on Root Development a Community in Rhizosphere 2.5 Straight Fertilizers as Affected by Zeolite 	
	2.5.1 Ammonium Nitrate	13
	2.5.2 Rock Phosphate	16
	2.5.3 Muriate of Potash (Potassium Ch Ammonia Volatilization 2.6.1 Factors Affecting Ammonia Volat	18
	2.6.1.1 Soil pH 2.6.1.2 Cation Exchange Capac 2.6.1.3 Temperature	19
	2.7 Reducing Ammonia Loss Using Ammon with Zeolite	ium Nitrate 20
	2.8 Maize Growth and Development as Affe Compound Fertilizers	cted by 21
3	EFFECT OF ZEOLITE ON AMMONIA VOLATII	
	3.1 Introduction3.2 Materials and Methods	23 24
	3.3 Results and Discussion	24 29
	3.4 Conclusion	33

5 SUMMARY, GENERAL CONCLUSION AND RECOMMENDATION FOR FUTURE RESEARCH 60 REFERENCES/BIBLIOGRAPHY 63 BIODATA OF STUDENT 73 LIST OF PUBLICATIONS 74	4	FORMULATED FERTILIZER WITH CLINOPTILOLITE ZEOLITE ON MAIZE GROWTH AND NUTRIENT USE4.1Introduction4.2Materials and Methods4.3Results and Discussion4.4Conclusion	34 36 38 58
BIODATA OF STUDENT 73	5		60
	BIODATA C	CES/BIBLIOGRAPHY DF STUDENT	73

DEDICATION

I would like to dedicate this thesis to my parent, my siblings and people who are important in my life



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Master of Science

AMENDING OF FORMULATED COMPOUND FERTILIZERS WITH ZEOLITE TO IMPROVE MAIZE (*Zea mays* L.) GROWTH AND NUTRIENT USE EFFICIENCY

By

MAGDALINA LIJA ANAK WELSON BLUE

April 2013

Chairman : Ahmed Osumanu Haruna, PhD

Faculty : Faculty of Agriculture and Food Sciences, Bintulu

Excessive use and poor utilization of fertilizers can cause air, soil and water pollution. These problems could be minimized by amending compound fertilizers with clinoptilolite zeolite. Objectives of this study were to: (i) determine the effect of amending NPK fertilizers with different rates of clinoptilolite zeolite on NH₃ volatilization, soil exchangeable NH₄⁺, and available NO₃⁻, and (ii) determine the effect of NPK fertilizers amending with different rates of clinoptilolite zeolite on maize (*Zea mays* L.) dry matter production, nutrient uptake, and nutrient use efficiency. An ammonia (NH₃) volatilization study was conducted using a closed-dynamic air flow system. Seven treatments evaluated in ammonia volatilization experiment were: 250 g soil only (T1), 250 g soil + 3.28 g ammonium nitrate (AN) without additive (T2), 44.6 g 5:3:2 formulated compound containing 31.77 g zeolite (T3), 44.6 g 5:5:5 formulated compound fertilizer containing 26.36 g zeolite (T4), 37.17 g 6:6:6 formulated compound fertilizer containing 18.92 g zeolite (T5), 22.30

ii

g 10:10:10 formulated compound fertilizer containing 4.06 g zeolite (T6), and 14.87 g 15:15:15 commercial fertilizer (T7). Ammonia loss over 12 days was daily recorded following standard procedure. Soil pH, exchangeable NH4⁺ and available NO₃⁻ at the end of this incubation study were determined using standard procedures. The mixture of the formulated compound fertilizers with clinoptilolite zeolite and T7 were significantly reduced NH₃ volatilization. This was possible because of the temporary retention of NH₄⁺ by clinoptilolite zeolite in the formulated compound fertilizers via sorption reaction. Application of commercial compound fertilizer had superior effect on soil exchangeable NH₄⁺ among treatments, while available NO₃⁻ had similar effect for treatments containing clinoptilolite zeolite and T7. This was probably due to the source of the fertilizers used. Ammonia volatilization experiment was demonstrated that amending N, P, and K compound fertilizers with clinoptilolite zeolite could minimize NH₃ loss as commercial compound fertilizer (T7). A pot experiment by cultivate Masmadu maize variety as test crop was carried out in a greenhouse as further test to examine treatments effect on growth. The objective of this pot experiment was to observe the effect of the compound fertilizers amended with different rates of clinoptilolite zeolite on maize dry matter, nutrient uptake and fertilizer nutrient use efficiency. Treatments for pot experiment evaluated were: 7 kg soil only (T1), 44.6 g 5:3:2 formulated fertilizer containing 31.77 g zeolite (T2), 44.6 g 5:5:5 formulated compound fertilizer containing 26.36 g zeolite (T3), 37.17 g 6:6:6 formulated compound fertilizer containing 18.92 g zeolite (T4), 22.30 g 10:10:10 formulated compound fertilizer containing 4.06 g zeolite (T5), and 14.87 g 15:15:15 commercial fertilizer (T6). Selected soil

iii

chemical properties after planting were analyzed following standard methods. Dry weight, nutrient concentration, nutrient uptake and nutrient use efficiency at 65 DAP were also measured. Formulated compound fertilizers with zeolite increase soil pH compared to soil alone (T1) and commercial compound fertilizer (T6). Application of the formulated compound fertilizers were significantly similar effect on total N, exchangeable NH₄⁺ and NO₃⁻ availability in soil to commercial compound fertilizer. T6 significantly increased plant height, while total dry weight of treatments with clinoptilolite zeolite was similar to T6. Nitrogen and K concentrations in all parts of maize were statistically similar for treatments with clinoptilolite zeolite and T6, while commercial compound fertilizer had better P concentration. Formulated compound fertilizer and commercial compound fertilizer were had similar Ca concentration in all parts of maize except stems. Mg concentration was greater in maize roots when T6 was applied, while treatments with clinoptilolite zeolite had better Mg concentration in stems and leaves compared to T6. T2, T3, T4 and T5 cause lower uptake of N and P compared to T6. Application of T3, T4, and T5 had statistically similar K uptake to T6 application. Application of T6 was significantly increased N and P use efficiency, while, T2 significantly increased K use efficiency. T2 had high K use efficiency because it had lower rate K applied. A field study for at least three cropping cycles is needed to confirm the findings of this study.

iv

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

MEMINDA FORMULASI BAJA KOMPAUN DENGAN ZEOLIT UNTUK MENINGKATKAN PERTUMBUHAN JAGUNG (*Zea mays* L.) DAN KECEKAPAN PENGGUNAAN NUTRIEN

Oleh

MAGDALINA LIJA ANAK WELSON BLUE

April 2013

Penyelia : Ahmed Osumanu Haruna, PhD

Fakulti : Fakulti Sains Pertanian dan Makanan, Bintulu

Penggunaan baja sebatian berlebihan serta pengurusan yang lemah boleh menyebabkan pencemaran udara, tanah dan air. Masalah ini boleh dikurangkan dengan meminda baja sebatian dengan zeolit klinoptilolit. Objektif kajian ini adalah untuk: (i) menentukan kesan meminda baja NPK pada kadar zeolit klinoptilolit yang berbeza terhadap pemeruapan NH₃, tukarganti NH₄⁺, dan NO₃⁻ tersedia, dan (ii) menentukan kesan baja NPK dipinda dengan kadar zeolit klinoptilolit yang berbeza terhadap berat, pengambialn nutrien dan kecekapan penggunaan nutrien pokok jagung (*Zea mays* L.). Kajian pemeruapan ammonia (NH₃) telah dijalankan menggunakan sistem aliran udara tertutup dinamik. Tujuh rawatan yang telah dinilai dalam eksperimen pemeruapan ammonia adalah: 250 g tanah sahaja (T1), 250 g + 3.28 g AN tanpa aditif (T2), 44.6 g 5:3:2 baja yang diformulasi mengandungi 31.77 g zeolit (T3), 44.6 g 5:5:5 baja yang diformulasi mengandungi 26.36 g zeolit (T4), 37.17 g 6:6:6 baja yang diformulasi mengandungi 18.92 g zeolit

(T5), 22.30 g 10:10:10 baja yang diformulasi mengandungi 4.06 g zeolit (T6), dan 14.87 g 15:15:15 baja komersial (T7). Kehilangan ammonia telah direkod setiap hari menggunakan kaedah piawai. pH tanah, tukarganti NH4⁺ dan NO3⁻ tersedia di akhir kajian inkubasi ini telah ditentukan menggunakan kaedah piawai. Campuran baja sebatian yang diformulasi dengan zeolit klinotilolit dan T7 telah mengurangkan pemeruapan NH3 secara beerti. Ini adalah mungkin disebabkan pemegangan sementara NH4⁺ oleh zeolit klinoptilolit dalam baja sebatian yang diformulasi melalui reaksi penyerapan. Aplikasi baja sebatian komersial mempunyai kesan yang baik terhadap tukarganti NH4⁺ di kalangan rawatan, manakala NO3⁻ tersedia mempunyai kesan yang sama untuk rawatan mengandungi klinoptilolit zeolit dan T7. Ini mungkin disebabkan sumber baja yang digunakan. Eksperimen pemeruapan ammonia telah membuktikan bahawa meminda baja N, P, dan K dengan zeloit klinoptilolit boleh mengurangkan kehilangan ammonia seperti baja sebatian komercial (T7). Eksperimen di dalam pasu dengan mengunakan varieti jagung Masmadu sebagai tumbuhan uji telah dijalankan di rumah hijau sebagai kajian lanjutan untuk menyelidik kesan rawatan terhadap pertumbuhan. Objektif eksperimen pasu ini adalah untuk memerhati kesan pindaan baja sebatian dengan zeolit klinoptilolit pada kadar berbeza terhadap berat kering, pengambilan nutrien dan kecekapan penggunaan nutrien baja oleh pokok jagung. Rawatan yang dinilai untuk ekperimen pasu ini ialah: 7 kg tanah sahaja (T1), 44.6 g 5:3:2 baja yang diformulasi mengandungi 31.77 g zeolit (T2), 44.6 g 5:5:5 baja yang diformulasi mengandungi 26.36 g zeolit (T3), 37.17 g 6:6:6 baja yang diformulasi mengandungi 18.92 g zeolit (T4), 22.30 g 10:10:10 baja yang diformulasi

vi

mengandungi 4.06 g zeolit (T5), dan 14.87 g 15:15:15 baja komersial (T6). Sifat kimia terpilih tanah selepas penanaman dianalisis mengikuti kaedah piawai. Berat kering, kepekatan nutrien, pengambilan nutrien dan kecekapan penggunaan nutrien juga ditentukan. Baja sebatian dengan zeolit telah meningkatkan pH tanah berbanding tanah sahaja (T1) dan baja sebatian komersial (T6). Aplikasi baja sebatian yang diformulasi tidak memberikan perbezaan beerti terhadap jumlah N, tukarganti NH₄⁺ dan ketersediaan NO₃⁻ dalam tanah seperti T6. T6 telah berjaya meningkatkan dengan ketinggian tumbuhan secara beerti, manakala pengeluaran bahan kering bagi rawatan mengandungi zeolit klinoptilolit adalah sama dengan T6. Kepekatan nitrogen, dan K dalam semua bahagian jagung adalah sama secara statistik bagi rawatan mempunyai zeolit klinoptilolit dan T6, manakala baja sebatian komersial mempunyai kepekatan P lebih baik. Baja yang diformulasi dan baja sebatian komersial mempunyai kepekatan Ca yang sama di semua bahagia pokok jagung kecuali stem. Kepekatan Mg di dalam akar jagung adalah tinggi apabila pokok telah diberi T6, manakala rawatan dengan zeolit klinoptilolit mempunyai kepekatan Mg lebih baik di dalam stem dan daun berbanding T6. T2, T3, T4, dan T5 menyebabkan pengambilan N dan P rendah berbanding T6. Aplikasi T3, T4 dan T5 mempunyai pengambilan K yang sama dengan T6 secara statistik. Aplikasi T6 telah meningkatkan kecekapan penggunaan N dan P secara bererti manakala, T2 telah meningkat kecekapan penggunaan K secara bererti. T2 mempunyai kecekapan penggunaan K yang tinggi kerana ia megandungi kadar K yang rendah. Satu kajian lapangan sekurang-kurangnya tiga kitaran tanaman diperlukan untuk mengesahkan penemuan kajian ini.

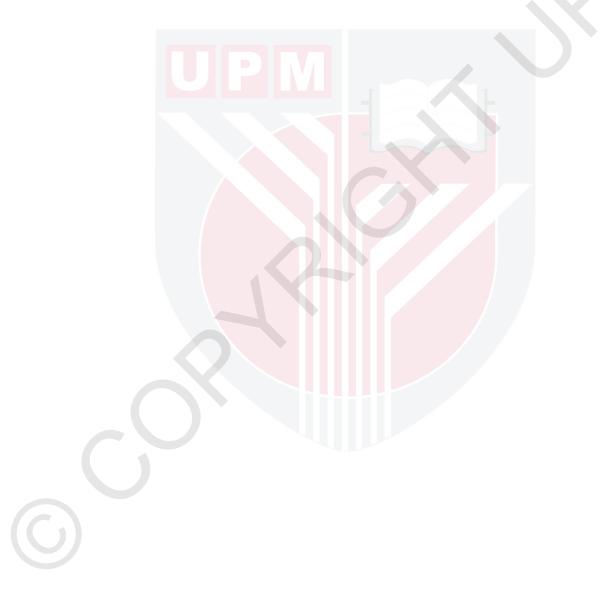
vii

ACKNOWLEDGEMENTS

I am grateful to God for His love and blessings to complete this study. I would like to sincerely thank to my supervisor, Assoc. Prof. Dr. Ahmed Osumanu Haruna for their kindness in providing me guidance, patience and support throughout this study. A special thanks to Dr. Susilawati Kasim for giving me advice and morale support to finish this study. I also extend my thanks to all laboratory staff that helped in completing this study.

My acknowledgement also goes to Universiti Putra Malaysia and School of Graduate Studies for providing me a place and financial support (Graduate Research Fellowship). Also, thanks to Ministry of Higher Education, Malaysia for their kindness in sponsoring my study through MyBrain 15 (MyMaster). Thanks to Chemical Company Malaysia (CCM) for supplying ammonium nitrate fertilizer for my research.

I wish to express my warm and deepest thanks to my beloved parents, Mr. Welson Blue Bana and Madam Monica Belangkat Anthony Buli because without their support I would never have made it on my own through this journey. My most grateful thanks and gratitude goes to Mr. Bantin Tommy, for your morale support that kept me focused on my master research work. Thank you Miss Krystle Amanda Rabai Nasip IIa, Mr. Abenego Pengiran, Miss Piriya Latha Veerasamy, Mr. Palanivel Perumal, Miss Aainaa Hasbullah and my entire friends for their support, encouragement and friendship along this journey. I hope that this study will bring benefits to all of us.



I certify that a Thesis Examination Committee has met on 12 April 2013 to conduct the final examination of Magdalina Lija Anak Welson Blue on her thesis entitled "Amending of Formulated Compound Fertilizers with Zeolite to Improve Maize (*Zea mays* L.) Growth and Nutrient Use Efficiency" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Yiu Pang Hung, PhD

Senior Lecturer Faculty of Agriculture and Food Sciences Universiti Putra Malaysia (Chairman)

Radziah Binti Othman, PhD

Associate Professor Faculty of Agriculture Universiti Putra Malaysia (Internal Examiner)

Samsuri Abd. Wahid, PhD

Senior Lecturer Faculty of Agriculture Universiti Putra Malaysia (Internal Examiner)

Hasnah Md. Jais, PhD

Associate Professor School of Biological Sciences Universiti Sains Malaysia Malaysia (External Examiner)

NORITAH OMAR, PhD

Professor and Deputy Dean School of Graduate Studies Universiti Putra Malaysia

Date: 26 June 2013

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Ahmed Osumanu Haruna, PhD

Associate Professor Faculty of Agriculture and Food Sciences Universiti Putra Malaysia (Chairman)

Susilawati Kasim, PhD

Senior Lecturer Faculty of Agriculture and Food Sciences Universiti Putra Malaysia (Member)

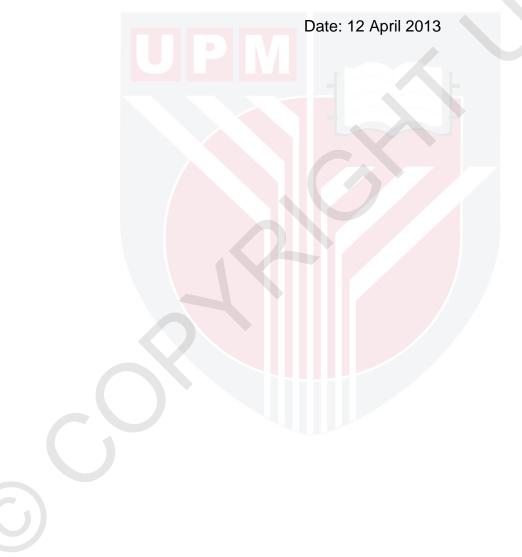
BUJANG BIN KIM HUAT, PhD

Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date:

DECLARATION

I declare that the thesis is my original work except for quotation and citation which have been duly acknowledged. I also declared that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.



MAGDALINA LIJA ANAK WELSON BLUE

xii

LIST OF TABLES

Table		Page
3.1	Selected physical and chemical properties of soil and clinoptilolite zeolite.	29
3.2	Effect of treatments on ammonia loss, pH, exchangeable ammonium and nitrate accumulation with soil over 12 days of incubation.	31
4.1	Selected chemical properties of Bekenu series after planting.	41
4.2	Dry weight (DW) of roots, stems, and leaves of maize (var. Masmadu) at 65 days after planting.	42
4.3	Nitrogen, P, K, Ca, and Mg concentrations in roots, stems, and leaves of maize (var. Masmadu) at 65 days after planting.	43
4.4	Nitrogen, P, and K uptake in roots, stems, and leaves of maize (var. Masmadu) at 65 days after planting.	47
4.5	Nitrogen, P, and K nutrient use efficiency in roots, stems, and leaves of maize (var. Masmadu) at 65 days after planting.	53

LIST OF FIGURES

Figure		Page
2.1	Process of clinoptilolite zeolite formation	6
2.2	Basic of zeolite structure	7
2.3	Structure of clinoptilolite zeolite	8
2.4	Ammonium nitrate production process	15
3.1	Daily ammonia volatilization from incubation.	30
4.1	Effect of treatments on total dry weight of maize (var. Masmadu) at 65 days after planting.	42
4.2	Effect of treatments on N uptake of maize (var. Masmadu) at 65 days after planting.	50
4.3	Effect of treatments on P uptake of maize (var. Masmadu) at 65 days after planting.	51
4.4	Effect of treatments on K uptake of maize (var. Masmadu) at 65 days after planting.	52
4.5	Effect of treatments on N use efficiency of maize (var. Masmadu) at 65 days after planting.	55
4.6	Effect of treatments on P use efficiency of maize (var. Masmadu) at 65 days after planting.	56
4.7	Effect of treatments on K use efficiency of maize (var. Masmadu) at 65 days after planting.	57
4.8	Effect of treatments on height of maize (var. Masmadu) at 65 days after planting.	58

LIST OF ABBREVIATIONS

%	Percentage
°C	Degree Celcius
AAS	Atomic absorption spectrometer
AN	Ammonium Nitrate
ANOVA	Analysis of variance
Са	Calcium
CaCO ₃	Calcium carbonate
CEC	Cation exchange capacity
CsCl	Caesium chloride
DAP	Day after planting
DNMRT	Duncan new multiple range test
DW	Dry weight
ERP	Egypt rock phosphate
g	Gram
h	Hour
H ₂ SO ₄	Sulphuric acid
ha	Hectare
HCI	Hydrochloric acid
HNO ₃	Nitric acid
К	Potassium
K ₂ SO ₄	Potassium sulphate
KCI	Potassium chloride

 \bigcirc

Kg	Kilogram
L	Liter
Mg	Magnesium
MgO	Magnesium oxide
min	Minute
mL	Mililiter
MOP	Muriate of potash
Ν	Nitrogen
NH3	NH3
NH4 ⁺	Ammonium
NH₄OAc	Ammonium acetate
NO3 ⁻	Nitrate
NUE	Nutrient use efficiency
ОМ	Organic matter
Р	Phosphorus
SAS	Statistical analysis system
UV	Ultra violet

C

CHAPTER 1

INTRODUCTION

Malaysia has 4.06 million hectares of agricultural land distributed in the country (Mattassan, 2008; Murad *et al.*, 2009). The agricultural sector is important as it contributes to the Malaysian economy (Murad *et al.*, 2009). According to Matassan (2008), some of the Malaysian soils are highly leached, infertile, and acidic. Crops in general require sufficient nutrients in soil to sustain performance and yield. The high demand for fertilizers to sustain high crop yield annually contribute to the high cost of their use in agriculture.USDA (2010) reported that fertilizer prices in 2010 were \$517 per ton anhydrous ammonia, \$421 per ton urea, \$245 per ton liquid N (28%), \$447 per ton DAP, and \$495 per ton potash. Malaysia spent about RM9.17 billion for 4.16 million ton of imported mineral fertilizers (Sabri, 2009).

Since fertilizers are usually the highest variable costs item in crop production budget, it is essential to improve nutrient use efficiency of fertilizers. In 2006, Malaysia produced about 1.1 million ton of granular and compacted fertilizers (Shaddick, 2007). Blending of straight fertilizers can reduce fertilizer application costs and time for nutrients distribution (McCauley *et al.*, 2009). Compound fertilizers can be defined as a fertilizer that has a declarable content of at least two of plant nutrients obtained chemically or by blending (UNIDO and IFDC, 1998). Maize like other crops requires sufficient amount of nutrients for optimum growth and development. Nitrogen, P, and K fertilizers amended with clinoptilolite zeolite may improve soil nutrients retention to sustain maize growth because zeolites contain essential nutrients such as Ca, K, and Mg that could improve growth and development of plants. Zeolite based fertilizers have several advantages namely, they are prepared based on a non-toxic natural material and easily applied at the beginning of the vegetation period yet supplies an even fertilizing effect throughout the whole period of plant growth and development (Rehakova *et al.*, 2004).

Zeolites have high CEC, high water holding capacity, and high adsorption capacity (Mumpton, 1999). Clinoptilolite zeolite decreases NH₃ volatilization by retaining NH₃ in its pore structure. In general, N loss and NO₃⁻ leached from agricultural soils occur through NH₃ volatilization and denitrification, respectively (He *et al.*, 2002). Zeolites improve nutrient use efficiency by increasing Ca²⁺ exchange from phosphate rock during dissolution of P. Moreover, the utilization of NH₄⁺ and NO₃⁻ as well as reducing losses *via* cations exchange reaction, for instance Ca²⁺ and K⁺ (Ghorbani and Babaei, 2008). Bernardi *et al.* (2012) reported that concentrated zeolite enriched with N, P and K was an adequate slow-release source of nutrients for plants. Zeolites also improves soil water holding capacity and its availability to plants (Xiubin and Zhanbin, 2001; Bernardi *et al.*, 2008).

2

In recent times, studies are focused on NH₃ volatilization from urea and direct application of zeolites (Ahmed *et al.*, 2006a, 2008a; Shamsuddin *et al.*, 2009; Latifah *et al.*, 2010). Currently, there is dearth of information on nutrient use efficiency *via* mixing ammonium nitrate (AN), Egypt rock phosphate (ERP), and muriate of potash (MOP) with clinoptilolite zeolite. These kinds of compound fertilizers with clinoptilolite zeolite may not only improve retention of nutrients but they will also aid in slow-release of nutrients for optimum plant uptake and utilization. Therefore, the objectives of this study were to: (1) determine the effect of amending NPK fertilizers with different rates of clinoptilolite zeolite on NH₃ volatilization, soil exchangeable NH₄⁺, and available NO_{3⁻}, and (2) determine the effect of NPK fertilizers amending with different rates of clinoptilolite zeolite on maize (*Zea mays* L.) dry matter production, nutrient uptake, and nutrient use efficiency.

REFERENCES

- Abdi, G., Khosh-Khui, M., & Eshghi, S. (2006). Effect of natural zeolite on growth and flowering of strawberry (*Fragaria x xananassa* Duch.). *International Journal of Agricultural Research*, 1: 384-389.
- Adeyemi, A.A. (2011). Yield and yield components of maize as influenced by cultivars and fertilizer rates. *International Journal of Applied Agriculture and Apiculture Research*, 7: 72-78.
- Ahmed, O., Sumalatha, G., & Nik Muhamad, A.M. (2010b). Use of zeolite in maize (Zea mays) cultivation on nitrogen, potassium and phosphorus uptake and use efficiency. *International Journal of the Physical Sciences*, 5: 2393-2401.
- Ahmed, O.H., Aminuddin, H., & Husni, M.H A. (2006b). Reducing ammonia loss from urea and improving soil-exchangeable ammonium retention through mixing triple superphosphate, humic acid and zeolite. Soil Use and Management, 22: 315-319.
- Ahmed, O.H., Aminuddin, H., & Husni, M.H.A. (2006a). Effects of urea, humic acid and phosphate interactions in fertilizer microsites on ammonia volatilization and soil ammonium and nitrate contents. *International Journal Agricultural Research*, 1: 25-31.
- Ahmed, O.H., Braine, C.H., & Muhamad, A.M. (2010a). Minimizing ammonia loss from urea through mixing with zeolite and acid sulphate soil. *International Journal Physical Science*, 5: 2198-2202.
- Ahmed, O.H., Husin, A., & Husni, M.H.A. (2008a). Ammonia volatilization and ammonium accumulation from urea mixed with zeolite and triple superphosphate. *Acta Agriculturae Scandinavica Section B-Soil and Plant Science*. 58: 182-186.
- Ahmed, O.H., Hussin, A., Ahmad, H.M., Rahim, A.A., & Majid, N.M.A. (2008b). Enhancing the urea-N use efficiency in maize (*Zea mays*) cultivation on acid soils amended with zeolite and TSP. *The Scientific World Journal*, 8: 394-399.
- Al-Busaidi, A., Yamamoto, T., Tanigawa, T., & Rahman, H.A. (2011). Use of zeolite to alleviate water stress on subsurface drip irrigated barley under hot environments. *Irrigation and Drainage*, 60: 473-480.
- Aleksandrov, V.G., Blagodyr, R.N. & liiev, I.P. (1967). Liberation of phosphoric acid from apatite by silicate bacteria. *Mikrobiyol Zh. (Kiev)*, 29: 111-114.
- Asman, W.A., Sutton, M.A., & Schjørring, J.K. (1998). Ammonia: emission, atmospheric transport and deposition. *New Phytologist*, 139: 27-48.
- Association of American Plant Food Control Officials (AAPFCO). (1995). Official Publication No. 48. Association of American Plant Food Control Officials, Inc. West Lafayette, Indiana, USA.
- Bachand, P.A.M. (2003). Potential application of adsorptive media to enhance phosphorus uptake in stormwater basins and wetlands at Lake Tahoe. University of California Davis, Tahoe Research Group.
- Bennett, P.C., Choi, W.J. & Rogera, J.R. (1998). Microbial destruction of feldspars. *Mineral Management*, 8: 149–150.

- Bernardi, A.C.D.C., Oliviera, P.P.A., de Melo Monte, M.B., & Souza-Barros, F. (2012). Brazilian sedimentary zeolite use in agriculture. *Microporous* and Mesoporous Materials. 167: 16-21.
- Bernardi, A.C.D.C., Werneck, C.G., Haim, P.G., Rezende, N.D.G.D.A., Paiva, P.R.P., & Monte, M.B.D.M. (2008). Growth and mineral nutrition of rangpur lime rootstock cultivated in substrate with zeolite enriched with NPK. *Revista Brasileira de Fruticultura*, *30*: 794-800.
- Beychok, M. (2009). Drawing of the chemical structure of Zeolites. United States. <u>http://en.citizendium.org/wiki/File:ZeoliteChemicalStructure.png</u>
- Boettinger, J.L. & Ming, D.W. (2002). Chapter 19: Zeolites. In J.B. Dixon and D.G. Schulze (Ed), *Soil mineralogy with environmental applications.* Soil Science Society of America Book Series. No. 7, Madison, WI.
- Bogdanov, B., Georgiev, D., Angelova, K., Yaneva, K. Natural zeolites: clinoptilolite review. In *Economics and society development on the base of knowledge*, Proceedings of the International Science Conference, Stara Zagora, Bulgaria, June 4-5, 2009. Natural & Mathematical science Volume IV: Bulgaria, 2009.
- Bremner, J.M. (1965). Total nitrogen. In Black *et al. Method of Soil Analysis, Part 2*, (pp. 1149-1178). American Society of Agronomy, Madison, Wisconcin.
- Bushby, H.V.A. & Marshall, K.C. (1977). Some factors affecting the survival of root-nodule bacteria on desiccation. *Soil Biology and Biochemistry*, 9: 143-147.
- Cai, G.X., Chen, D.L., Ding, H., Pacholski, A., Fan, X.H., & Zhu, Z.L. (2002). Nitrogen losses from fertilizers applied to maize, wheat and rice in the North China plan. *Nutrient Cycle Agroecosystem*, 63: 187-195.
- Camberato, J.J. (2001). Cation exchange capacity: everything you want to know and much more. South Carolina Turfgrasss Foundation News. Clemson University.
- Carlino, J.L., Williams, K.A. & Allen, E.R. (1998). Evaluation of zeolite based soilless root media for potted chrysanthemum production. *Hotricultural Technology*, 8: 373-378.
- Chen, N.Y., Degnan, T.F., & Smith, C.M. (1994). Molecular transport and reaction in zeolites: Design and application of shape selective catalysis. Wiley, VCH, Hoboken, NJ.
- Colella, C. (2007). Natural zeolites and environment. In J. Cejka, Van Bekkum, H., Corma, A. & Schiith, F. (Ed.), *Introduction to Zeolite Science and Practice.* 3rd Revised Edition.
- Cottenie, A. (1980). Soil Testing and Plant Testing as a Basis of Fertilizer Recommendation. *FAO Soils Bulletin*, 38: 70-73.
- Davis, T. (2009). Fertilizer and pH. Cymbidium Club of Australia.
- Deer, A., Howie, R., Wise, W.S., and Zussman, J. (2004). Rock Forming Minerals. vol. 4B.Framework Silicates: Silica Minerals, Feldspathoids and the Zeolites. The Geological Society, London. In General clinoptilolite-Na information. <u>http://webmineral.com/data/Clinoptilolite-Na.shtml</u>. Weinrich Minerals, Inc.
- Department for Enviroment Food and Rural Affair (DEFRA). (2010). Fertilizer and Types and Quality. In *Fertilizer Manual (RB209)* 8th Edition. TSO (The Stationery Office), Ireland.

64

- Desutter, T.M. & Pierzynski, G.M. (2005). Evaluation of soils for use as liner materials: a soil chemistry approach. *Journal Environment Quality*, 34: 951-962.
- Di, H.J. & Cameron, K.C. (2002). Nitrate leaching in temperate agroecosystems: Sources, factors and mitigating strategies. *Nutrient Cycling In Agroecosystems.* 46: 237-256.
- Dondur, V., Raki, V., Damjanovi, L. & Auroux, A. (2005). Comparative study of the active sites in zeolites by different probe molecules. *Journal Serbian Chemistry Society*, 70: 457-474.
- Eberl, D.D. (1993). Controlled-release fertilizers using zeolites. U.S. Department of the Interior and U.S. Geological Survey
- Environmental Protection Agency (EPA). Zeolites. http://www.epa.gov/radiation/docs/cleanup/nanotechnology/chapter-2zeolites.pdf
- Fan, M.X. & Mackenzie, A.F. (1993). Urea and phosphate interactions in fertilizer microsites: ammonia volatilization and pH changes. *Soil Science Society American Journal*, 57: 839-845.
- Fenn, L.B. & Kissel, D.E. (1974). Ammonia volatilization from surface applications of ammonium compounds on calcareous soils: II. Effects of temperature and rate of ammonium nitrogen application. Soil Science Society American Journal, 38: 606–610. (Abstract)
- Fenn, L.B. & Richard, J. (1989). Ammonia loss from surface applied urea acid products. *Fertilizer Research*, 9: 265-275.
- Fenn, L.B., Tatum, G., Horst, G. (1990). Ammonia loss from surface-applied mixtures of urea-calcium-potassium salts in the presence of phosphorus. *Fertilizer Research*, 21: 125-131.
- Ferguson, G.A. & Pepper, G. (1987). Ammonium retention in sand amended with clinoptillolite. *Soil Science Society American Journal*, 51: 231-234.
- Flanigen, M., Broach, R.W. & Wilson, S.T. (2007). Introduction. In S. Kulprathipanja (Ed.), *Zeolites in Industrial Separation and Catalysis*.
- Főldesová, M., Hudec, P., Dillinger, P. (2007). Chemically modified zeolites: Surfaces and interaction with Cs and Co. *Petroleum and Coal.* 49: 61-64.
- Follett, R.F., Keeney, D.R. & Crusel, R.M. (1991). Managing nitrogen for groundwater quality and farm profitability, SSSA, Madison, WI.
- Food and Agriculture Organization (FAO). (2000). Fertilizers and their use a pocket guide or extension officers (4th Ed.). Food and Agriculture Organization of the United Nations and International Fertilizer Industry Association.
- Food and Agriculture Organization (FAO). (2008). Current world fertilizer trends and outlook to 2011/12. Food and Agriculture Organization of the United Nations.
- Freibauer, A. (2001). Biogenic emissions of greenhouse gases caused by arable and animal agriculture. *Nutrient Cycling Agroecosystem Special Issue.* 60: 1-326.
- Ghorbani, H. & Babaei, A.A. (2008). The effects of natural zeolite on ions adsorption and reducing solution electrical conductivity Na and K solutions. *International Meeting on Soil Fertility Land Management and Agroclimatology.* Turkey. pp. 947-955.

- Gomez, K.A. & Gomez, A.A. (1984). Statistical procedures for agricultural research (2nd Ed). John Wiley and Sons, Inc.
- Gul, A., Erogul, D. & Ongun, A.R. (2005). Comparison of the use of zeolite and perlite as substrate for crisp-head lettuce. *Science Horticulturae*, 106: 464-471.
- Hattenschwiller, S. & Vitousek, P.M. (2000). The role of polyphenols in terrestrial ecosystem nutrient cycling. *Tree*, 15: 238-243.
- Havlin, J., Beaton, J.D., Tisdale, S.L., & Nelson, W.L. (2005). Soil fertility and fertilizers: An introduction to nutrient management (Vol. 515). Upper Saddle River^ eNJ NJ: Pearson Prentice Hall.
- He, Z.L., Calvert, D.V., Alva, A.K., Li, Y.C., & Banks, D.J. (2002). Clinoptilolite zeolite and cellulose amendments to reduce ammonia volatilization in a calcareous sandy soil. *Plant and Soil*. 247: 253-260.
- He, Z. L., Baligar, V. C., Martens, D. C., Ritchey, K. D., & Elrashidi, M. (1999). Effect of byproduct, nitrogen fertilizer, and zeolite on phosphate rock dissolution and extractable phosphorus in acid soil. *Plant and soil*, 208: 199-207.
- Heffer, P. & Prud'homme, M. Medium-Term outlook for global fertilizer demand, supply and trade: 2008-2012. Paper presented at the meeting of *76th IFA Annual Conference*, Vienna. May 2008.
- Hemingway, B.S. & Robie, R.A. (1984). Thermodynamic properties of zeolites: low-temperature heat capacities and thermodynamic functions for phillipsite and clinoptilolite. Estimates of the thermochemical properties of zeolitic water at low temperature. *American Mineralogist*. 69: 692-700.
- Hernandez, M.A. & Torero, M. (2011). Fertilizer market situation: market structure, consumption and trade patterns, and pricing behavior. In *IFPRI Discussion Paper 01058*. International Food Policy Research Institute.
- Hershey, D.R., Paul, J.L. & Carlson, R.M. (1980). Evaluation of potassiumenriched clinoptilolite as a potassium source of potting media. *Horticultural Sciences*, 15: 87-89.
- Hussaini, M.A., Ogunlela, V.B., Ramalan, A.A. & Falaki, A.M. (2008). Mineral composition of dry season maize (*Zea mays* L.) in response to varying levels of nitrogen, phosphorus and irrigation at Kadawa, Nigeria. *World Journal of Agricultural Sciences*. 4: 775-780.
- Ibrahim, S.A. (2007). *Synthesis and characterization of zeolites from sodium aluminosilicate solution.* Master Thesis, School Of Material and Mineral Resources Engineering, USM.
- Incitec Pivot Limited. Ammonium nitrate fertilizers Cal-Am® & N-Sure®. Ammonium Nitrate Agritopic, 2006.
- Inglezakis, V.J. (2005). The concept of "capacity" in zeolite ion-exchange systems. *Journal of Colloid and Interface Science*. 281: 68-79.
- Inglezakis, V.J., & Poulopoulos, S.G. (2006). Adsorption and ion exchange. In Adsorption, Ion Exchange and Catalysis. pp. 243-353.
- Ippolito, J.A, Tarkalson, D.D., & Lehrsch, G.A. (2011). Zeolite soil application method affects inorganic nitrogen, moisture, and corn growth. *Soil Science*. 176: 136-142.
- Jr, D.D.D. & Johnson, N.C. (2007). Contributions of arbuscular mycorrhizas to soil biological fertility. In Abbott, L.K. & Murphy, D.V. (Ed), Soil

Biological Fertility: A Key to Sustainable Land Use in Agriculture. pp. 129-162. Springer.

- Junejo, N., Khanif, M.Y., Dharejo, K.A., Abdul-Hamid, H. & Abdul, A. (2012). Evaluation of coated urea for ammonia volatilization loss, nitrogen mineralization and microsite pH in selected soil series. *African Journal* of *Biotechnology*. 11: 366-378
- Junrungreang, S., Limtong, P., Wattanaprapat, K., & Patsarayeangyong, T. Effect of zeolite and chemical fertilizer on the change of physical and chemical properties on Lat Ya soil series for sugar cane. Paper presented at the meeting of the 17th World Congress of Soil Science, Thailand. 2002.
- Kawamura, S. (2000). Integrated design and operation of water treatment facilities. Wiley, Hoboken, NJ.
- Keeney, D.R. & Nelson, D.W. (1982). Nitrogen-Inorganic forms. In Page *et al.* (Ed), *Methods of Soil Analysis, Part 2, (2nd Ed)*. Agron. Monograph, 9. ASA and SSSA, Madison, Wisconsin.
- Kithome, M., Paul, J.W., Lavkulich, L.M. & Bomke, A.A. (1998). Kinetics of ammonium adsorption and desorption by the natural zeolite clinoptilolite. *Soil Science Society American Journal*, 62: 622-629.
- Kowalchuk, G.A. & Stephen, J.R. (2001). Ammonia-Oxidizing bacteria: A model for molecular microbial ecology. *Annual Review of Microbiology*, 55: 485-529. (Abstract)
- Krason, J., & Knud-Hansen, C.F. (2004). Natural zeolites-remedy for concentrated animal feeding operations and sustainable agriculture. In *Organic Waste Treatments: Safety Implications*, Proceeding Sustainable Organic Waste Management for Environmental Protection and Food Safety.
- Lai, T.M. & Eberl, D.D. (1986). Controlled and renewable release of phosphorous in soils from mixtures of phosphate rock and NH4-exchanged clinoptilolite. Zeolites, 6: 129-132.
- Latifah, O., Ahmed, O.H. & Nik Muhamad, A.M. (2010). Minimizing ammonia volatilization in waterlogged soils through mixing of urea with zeolite and sago waste water. *International Journal Physical Science*, 5: 2193-2197.
- Lear, H. (2012). Ammonium nitrate outlook 2012. http://www.noble.matthey.com/affinity/issue3/ano.htm. The Noble Metals N₂O Abatement and Related Products Newsletter-Spring (2012).
- Lewis, M.D., Moore, F.D.I. & Goldsberry, K.L. (1984). ammonium-exchanged clinoptilolite and granulated clinoptilolite with urea as nitrogen fertilizers. In Pond W.G., Mumpton, F.A. (Ed.), *Zeo-Agriculture: Use Natural Zeolites in Agriculture and Aquaculture* (pp. 105-111) Westview Press, Boulder, CO.
- Liu, G. Li, Y. & Alva, A.K. (2007). High water regime can reduce ammonia volatilization from soils under potato production. USDA-ARS / UNL Faculty. Paper 512.
- Lo'pez-Bucio, J., Cruz-Rami'rez, A. & Herrera-Estrella, L. (2003). The role of nutrient availability in regulating root architecture. *Plant Biology*, 6: 280-287.

- Mackown, C.T. & Tucker, T.C. (1985). Ammonium nitrogen movement in a course-textured soil amended with zeolite. *Soil Science Society American Journal*, 49: 235-238.
- Macnack, N.E., Chim, B.K. & Raun, W.R. (2012). Applied model for estimating potential ammonia loss from surface applied urea. *Oklahoma State University*. Stillwater, OK.
- Malaysia Agriculture Research and Development (MARDI). (1990). jagung manis baru (New Sweet Corn): Masmadu. MARDI, Kuala Lumpur.
- Manolov, I., Antonov, D., Stoilov, G., Tsareva, I. & Baev, M. (2005). Jordanian zeolitic tuff as a raw material for the preparation of substrates used for plant growth. *Journal Central European Agriculture*. 6: 485-494.
- Marcille-Kerslake, V. (1989). Sedimentary Phosphates in the Fernie Basin: Development of New Technology for Direct Application to Soils (82G and 82J). *Geological Fieldwork*. British Columbia Geological Survey Branch. University of Guelph. pp. 489-492.
- Maria-Ramirez, A., Osuna-Ceja, E.S. & Limón-Ortega, A. (2011). Two sources of zeolite as substitutes of nitrogen fertilizer for wheat (*Triticum Aestivum*) production in Tlaxcala, *Mexico. Tropical and Subtropical Agroecosystems.* 13: 533-536.
- Mattassan, Z.A. The malaysian fertilizer market. Paper presented at the meeting of the *IFA Crossroads Asia-Pacific in Melbourne*. Fertilizer Industry Association of Malaysia (FIAM), Malaysia, 2008.
- Mbah, E.U., Muoneke, C.O., & Okpara, D.A. (2007). Effect of compound fertilizer on the yield and productivity of soybean/maize intercrop in Southteastern Nigeria. *Tropical and Subtropical Agroecosystem*. 7: 87-95.
- Mccauley, A., Jones, C., & Jacobsen, J. (2009). Module 10: Commercial fertilizers and soil amendments. *Nutrient Management*. Montana State University, Bozeman.
- Meisinger, J. & Jokela, W.E. (2000). *Ammonia Volatilization from Dairy and Poultry Manure*, Proceedings from Managing Nutrients and Pathogens from Animal Agriculture. pp. 334-354.
- Milosevic, T. & Milosevic, N. (2009). The effect of zeolite, organic and inorganic fertilizers on soil chemical properties, growth and biomass yield of apple trees. *Plant Soil Environ.* 55: 528–535.
- Ming, D.W. & Allen, E.R. (2001). Use of natural zeolites in agronomy, horticulture and environmental soil remediation. *Mineralogy and Geochemistry*. 45: 619-654.
- Ming, D.W. & Dixon, J.B. (1986). Clinoptilolite in South Texas soils. Soil Science Society American Journal, 50: 1618-1622.
- Ming, D.W. & Dixon, J.B. (1987). Quantitative determination of clinoptilolite in soils by a cation-exchange capacity method. *Clays and Clay Minerals*, 35: 463-468.
- Ming, D.W., & Boettinger, J.L. (2001). Zeolites in soil environments. In Bish, D.L. and Ming, D.W. (Ed.), *Natural Zeolites: Occurrence, Properties, Applications. Mineralogy and Geochemistry*. 45: 323-345. Mineralogical Society of America and Geochemical Society, Washington, DC.

- Mumpton, F.A. (1999). La Roca Magica: Uses of natural zeolites in agriculture and industry. *Proceeding National Academic Science*, 96: 3463-3470.
- Murad, M.W., Siwarl, C., Mustapha, N.H.N., Kamil, N.F.N.M., Muhamad, S. & Aziz, A.A. (2009). Emergence to develop an appraisal system for agricultural practices in Malaysia. *Journal Human Ecology*, 28: 191-198
- Murphy, J. & Riley, J.P. (1962). A modified single solution method for the determination of phosphate in natural waters. *Analytica Chimia Acta*. 27: 31-36.
- Noori, M., Zendehdel, M., & Ahmadi, A. (2006). Using natural zeolite for the improvement of soil salinity and crop yield. *Toxicological Environment Chemistry*, 88: 77-84.
- Nottidge, D.O., Ojeniyi, S.O., & Asawalam, D.O. (2005). Comparative effect of plant residue and NPK fertilizer on nutrient status and yield of maize (*Zea mays* I.) in a humid ultisol. *Nigerian Journal of Soil Science 15.*
- Oladiran, A.O. (2010). Effect of sources of fertilizer on the growth and yield of maize. University of Agriculture, Abeokuta, Nigeria.
- Olesen, J.E. & Sommmer, S.G. (1993). Modelling effects of wind speed and surface cover on ammonia volatilization from stored pig slurry. *Atmospheric Environment*, 27: 2567-2574.
- Onasanya, R.O., Aiyelari, O. P., Onasanya, A., Oikeh, S., Nwilene, F.E., & Oyelakin, O.O. (2009). Growth and yield response of maize (*Zea Mays* L.) to different rates of nitrogen and phosphorus fertilizers in Southern Nigeria. *World Journal of Agricultural Sciences*. 5: 400-407.
- Paramananthan, S. (2000). Soils of Malaysia: Their characteristics and identification (Volume 1). Academy of Sciences Malaysia. Kuala Lumpur, Malaysia. pp. 616.
- Parmar, P. & Sindhu, S.S. (2013). Potassium solubilization by rhizosphere bacteria: Influence of nutritional and environmental conditions. *Journal of Microbiology Research*, 3: 25-31. Scientific & Academic Publishing.
- Pavelic, K. & Hadz'lja, M. (2003). Chapter 24: Medical applications of zeolites. In Auerbach, S.M., Carrado, K.A., Dutta, P.K. (Ed.), Handbook of Zeolite Science and Technology. Marcel Dekker, Inc.
- Payra, P. & Dutta, P.K. (2003). Zeolites: A primer. In Auerbach, S. M., Carrado, K. A., Dutta, P. K. (ed.), *Handbook of Zeolite Science and Technology*, Marcel Dekker, Inc.
- Peech, H.M. (1965). Hydrogen ion activity. In Black et al. (ed.), Methods of Soil Analysis Part 2. American Society of Agronomy. Madison, Wisconsin. pp. 914-926.
- Perez-Caballero, R., Gil, J., Benitez, C. & Gonzalez, J.L. (2008). The effect of adding zeolite to soils in order to improve the N-K nutrition of olive trees, preliminary results. *American Journal Agriculture Biological Sciences*, 2: 321-324.
- Philippot, L., Hallin, S. Börjesson, G. & Baggs, E.M. (2009). Biochemical cycling in the rhizosphere having an impact on global change. *Plant Soil*, 321: 61-81
- Piccolo, A. (1996). Humus and soil conservation. *Humic Substances in Terrestrial Ecosystem.* Elseiver, Amsterdam. pp. 225-264.
- Plaster, E.J. (2009). Chapter 14: Fertilizers. Soil Science and Management 5th Edition. Delmar, Cengage Learning. United State. pp. 236-237.

- Polat, E., Karaca, M., Demir, H., & Onus, A.N. (2004). Use of natural zeolite (clinoptilolite) in agriculture. *Journal of Fruit and Ornamental Plant Research Special Edition*, 12: 183-189.
- Pomares-Gracia, F. & Pratt, P.F. (1987). Recovery of 15N-labelled fertilizer from manured and sludged-amended soils. *Soil Science Society American Journal*, 42: 717-720.
- Prasertsak, P., Freney, J.R., Saffiga, P.G., Denmead, O.T. & Prove, B.G. (2001). Fate of urea nitrogen applied to a banana crop in the wet tropics of Queensland. *Nutrient Cycling Agroecosystem*, 59: 65-73.
- Ramesh, K. & Reddy, D.D. (2011). Zeolites and their potential uses in agriculture. Advances in Agronomy. 113 (4).
- Reddy, S.R. (1999). Principle of agronomy. In *Chapter 6: Mineral Nutrition, Manures, and Fertilizers.* Kalyani Publishers. New Delhi. pp. 197-248
- Rehakova, M., Cuvanova, S., Dzivak, M., Rimar, J., & Gavalova, Z. (2004). Agricultural and agrochemical uses of natural zeolite of the clinoptilolite type. *Current Opinion in Solid State and Materials Science* 8: 397–404.
- Rengel, Z. & Damon, P.M. (2008). Crops and genotypes differ in efficiency of potassium uptake and use. *Physiologia Plantarum*. 133: 624-636.
- Robinson, D. (2005). Integrated root responses to variations in nutrient supply. In BassiriRad, H. (Ed.), Nutrient Acquisition by Plants An Ecological Perspective. *Ecological Studies*, 181.
- Roy, R.N., Finck, A., Blair, G.J., & Tandon, H.L.S. (2006). Plant nutrition for food security: A guide for integrated nutrient management. Food and Agriculture Organization of the United Nations, Rome.
- Sabri, M.A. (2009). Evolution of fertilizer use by crops in malaysia: recent trends and prospects. Fertilizer Industry Association of Malaysia.
- Sakrabani, R. Evaluation of zeolite as a binding agent to mitigate ammonia loss from pig slurry and manure. Paper presented at the conference of the Treatment and Use of Organic Residues in Agriculture: Challenges and Opportunities towards Sustainable Management in 14th Ramiran International Conference, 2010.
- SAS. (2008). SAS/STAT user's guide. Version 9.2. SAS Institute Inc., Cary, NC. United State.
- Shaddick, P. (2007). Compound fertilizers in South East Asia. *IFA Asia-Pacific Crossroads in Bali*. International Fertilizer Industry Association.
- Shamsuddin, R., Ahmed, O.H., Nik Muhamad, A.M. & Jalloh, M.B. (2009). Reduction of ammonia volatilization through mixing urea with humic and fulvic acids isolated from palm oil mill effluent sludge. *American Journal* of Environmental Sciences. 5: 382-386.
- Siva, K.B, Aminuddin, H., Husni, M.H.A., & Manas, A.R. (1999). Ammonia volatilization from urea as affected by tropical-based palm oil effluent (pome) and peat. *Comm. Soil Sci. Plant Anal.* 30: 785-804.
- Soh, K.G. (2001). A review of the global fertilizer use by product. 7th AFA Annual Conference.
- Soil Survey Staff. (1999). Soil taxonomy: A basic system of soil classification for making and interpreting soil surveys. Agricultural Handbook No. 436, 2nd ed. USDA, Natural Resource Conservation Service, Washington, D.C.

- Sommer, S.G., Schjoerring, J.K. & Denmead, O.T. (2004). Ammonia emission from mineral fertilizers and fertilized crops. *Advances in Agronomy*, 82.
- Stumpe, J.M., Vlek, P.L.G. & Linday, W.L. (1984). Ammonia volatilization from urea and urea phosphates in calcareous soils. *Soil Science Society American Journal*, 48: 921-926.
- Tan, K.H. (2005). Soil Sampling, Preparation, and Analysis, Second Edition. CRC Press. Taylor & Francis Group. Boca Raton, Florida, USA. pp. 154-174.
- Tariq, M., Saeed, A., Nisar, M., Mian, I.A. & Afzal, M. (2011). Effect of potassium rates and sources on the growth performance and on chloride accumulation of maize in two different textured soils of Haripur, Hazara Division. Sarhad Journal Agricultural, 27: 415-422.
- Tarkalson, D.D. & Ippolito, J.A. (2010). Clinoptilolite zeolite influence on inorganic nitrogen in silt loam and sandy agricultural soils. *Soil Science*, 175: 357-362.
- Torkashvand, A.M., Karami, A., Khomami, A.M. & Shadparvar, V. (2012). Zeolite: An appropriate alternative to peat in the growth medium of ornamental plants. *Indian Journal of Fundamental and Applied Life Sciences*, 2: 127-132.
- Ullman, W.J., Kirchman, D.L. & Welch, S.A. (1996). Laboratory evidence by microbially mediated silicate mineral dissolution in nature. *Chemistry and Geology*, 132: 11-17.
- United Nations Industrial Development Organization (UNIDO) and International Fertilizer Development Center (IFDC). (1998). *Fertilizer manual 3rd edition*. Kluwer Academic Publishers, The Netherlands.
- United States Department of Agriculture (USDA). (2010). Fertilizer prices in 2008, 2009, and 2010. In *Farm Economics: Facts And Opinions*. University Of Illinois Extension.
- United States International Trade Commission (USITC). (1998). Ammonium nitrate: A comparative analysis of factors affecting global trade. In *Investigation No. 332-393*. Washington.
- United States International Trade Commission (USITC). (2011). Ammonium nitrate from Russia. In *Investigation No. 731-TA-856* (2nd Review). Washington.
- Virta, R.L. (2002). Zeolites. U.S. Geological Survey Minerals Yearbook. pp. 84.
- Wu, L. & Liu, M. (2008). Preparation and properties of chitosan-coated NPK compound fertilizer with controlled-release and water-retention. *Carbohydrate Polymers*, 72: 240-247.
- Xie, W.J., Wang, H.Y., Xia, J.B., & Yao, Z.G. (2011). Influence of N, P, and K application on *Zea mays* L. growth and Cu and Pb accumulation. *Plant Soil Environment*, 57: 28-34.
- Xiubin, H., Zhanbin, H. (2001). Zeolite application for enhancing water infiltration and retention in loess soil. *Resources, Conservation and Recycling*, 34: 45-52.
- Yagoub, S.O., Ahmed, W.M.A. & Mariod, A.A. (2012). Effect of urea, npk and compost on growth and yield of soybean (*Glycinemax* L.), in semi-arid region of Sudan. *International Scholarly Research Network.*

ZEO Inc. (2006). Zeolite formation. <u>http://www.zeoinc.com/zeolite_formation.html</u>. In *The Leader in Natural Zeolite Products and Technology*. ZEO, Inc.



BIODATA OF STUDENT

Magdalina Lija Anak Welson Blue was born on 29th March 1987 at Bintulu, Sarawak. She attended a pre-school at St Anthony Kindergarten. Magdalina received her primary education at SK Kidurong No. 2. Later, she attended SMK Kidurong secondary school for her secondary education and entered a pre-university at Labuan Matriculation College. She had her first degree (Bachelor of Science Bioindustry) at Universiti Putra Malaysia in 2010. She is currently a fulltime student pursuing Master of Science (Agronomy) at Universiti Putra Malaysia Bintulu Sarawak Campus (UPMKB). In 2011, she attended National Horticulture Conference organized by Malaysia Agricultural Research and Development Institute (MARDI). She is currently working as a research assistant at Ta Ann Plantation Sdn. Bhd.

LIST OF PUBLICATIONS

- 1. Lija, W.B.M., Ahmed, O.H., & Kasim, S. (2012). Reducing ammonia volatilization from compound fertilizers amended with zeolite. *African Journal of Biotechnology*, 11(74): 13903-13906.
- 2. Lija, W.B.M., Ahmed, O.H., & Kasim, S. (2014). Maize (*Zea mays* L.) nutrient use efficiency as affected by formulated fertilizer with Clinoptilolite Zeolite. *Emirates Journal of Food and Agriculture*, 26(2).

