



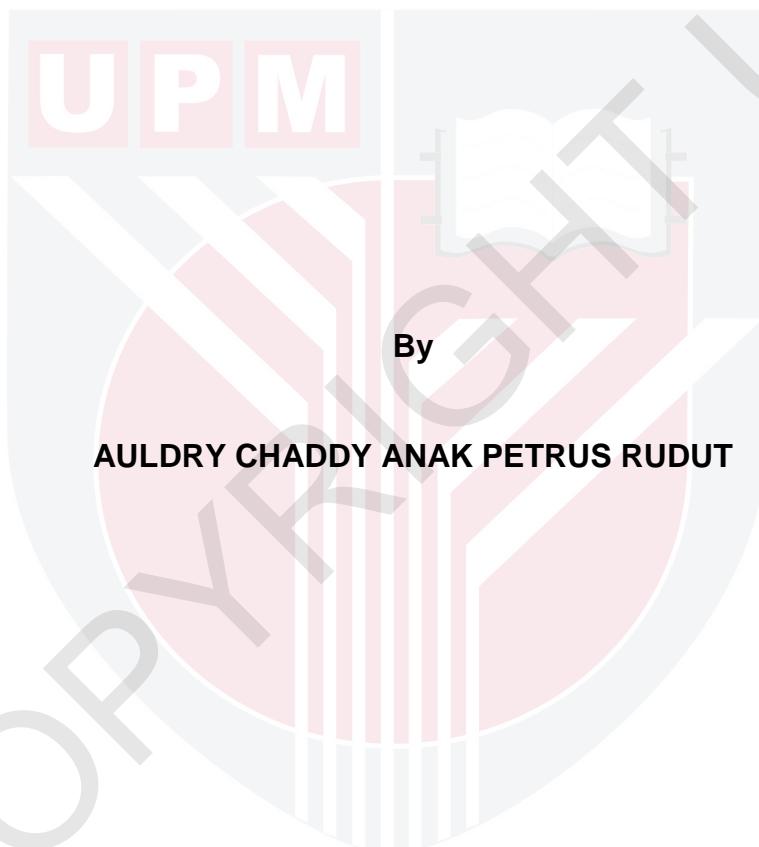
UNIVERSITI PUTRA MALAYSIA

***LABORATORY PRODUCTION OF ORGANIC-BASED FERTILIZER FROM
SAGO (METROXYLON SAGU ROTTB.) WASTE COMPOST***

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Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in
Fulfilment of the Requirements for the Degree of Master of Science

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LABORATORY PRODUCTION OF ORGANIC-BASED FERTILIZER FROM SAGO (*Metroxylon sagu Rottb.*) WASTE COMPOST

By

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May 2010

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Sago waste (SW) has a potential to cause pollution especially when the waste is discarded into rivers and streams. In order to add value to SW, a study was conducted to produce calcium (Ca) and potassium (K) hydroxide, compost and humic acid (HA) from it. The SW was air-dried and some ground. The ground SW was incinerated at 600 °C. Calcium and potassium hydroxide was extracted by dissolving the ash in distilled water at a ratio of 1:500 (ash : water), equilibrated for 24 hours at 150 rpm using a mechanical shaker and filtered. The ungrounded SW was used for compost production. The study had three treatments which were: T1: SW (80%) + chicken feed (5%) + chicken dung slurry (5%) + molasses (5%) + urea (5%), T2: SW (80%) + chicken feed (10%) + chicken dung slurry (5%) + molasses (5%) and T3: SW (80%) + chicken feed (10%) + chicken dung slurry (5%) + urea (5%). Composting was done for 60 days in a white polystyrene box with a size of - 61.5 cm x 49 cm x 33.5 cm. The composts produced were analyzed for pH, total nitrogen, organic carbon, organic matter, ash, cation exchange capacity

(CEC), phosphorus and HA using standard procedures. The hydroxide extracted from ash of SW was used to isolate HA of composted SW. The molarity and pH of the hydroxide were 0.002M and 10 respectively. Calcium (42.88mg kg^{-1}) and potassium (29.51mg kg^{-1}). The hydroxide was able to extract 1.15% of HA from the composted SW. A comparison between the yields of HA extracted from the composted SW using the hydroxide of the SW and that of the analytical grade showed no statistical difference. All three treatments did not reach thermophilic phase. Compost of T2 had high quality (pH, total nitrogen, organic carbon, organic matter, ash, cation exchange capacity (CEC), phosphorus and HA) compared to T1 and T3. The compost characteristics of T1 and T3 were similar. The yield of HA of T2 was also significantly higher compared to those of T1 and T3. The chemical characteristics of HA of the three treatments were within the standard range reported by other researchers. Besides HA, liquid HA and fulvic acid (FA) and humin also been extracted from the compost. The treatments for pot experiment were: control without fertilizer (T1); NPK (4.85 g urea, 4.85 g TSP, 2.5 g KCl) (solid) (T2), 400 mL liquid of FA+HA mixed with 4.85 g of urea and 2.5 g of KCl (T3), liquid HA mixed with 4.85 g of urea and 2.5 g of KCl (T4), 400 mL hydroxide (extracted from ash) mixed with 4.85 g of urea and 2.5 g of KCl + 200 g humin in soil (T5) and 400 mL liquid of FA+HA mixed with 4.85 g of urea and 2.5 g of KCl + 100 g humin in soil (T6). Treatments which had humin (T5 and T6) had the highest total dry weight and nutrient use efficiency. SW can be efficiently utilized by producing valuable products such as compost, Ca-K hydroxide as well as organic based fertilizers for agriculture. Furthermore, future investigation on the effect of soil

physical properties on nutrient use efficiency is encouraged to seek more understanding on soil-fertilizer and plant interaction.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan ijazah Master Sains

**PENGHASILAN BAJA BERASASKAN ORGANIK DARIPADA KOMPOS
SISA SAGU (*Metroxylon sagu Rottb.*) PADA SKALA MAKMAL**

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Sisa sagu berpotensi untuk menyebabkan pencemaran terutamanya apabila dibuang ke dalam sungai-sungai besar dan kecil. Untuk membuatkan hampas sagu ini bernilai, maka satu kajian telah dilakukan bagi menghasilkan hidroksida kalsium (Ca) dan kalium (K), kompos serta asid humik (AH) dari hampas tersebut. Sisa sagu dikeringkan dengan cara pengeringan udara dan separuh daripadanya dikisar. Sisa sagu yang dikisar dibakar pada suhu 600 °C sehingga menjadi abu. Hidroksida Ca-K diekstrak dari abu sisa sagu dengan melarutkan abu tersebut ke dalam air suling dengan nisbah 1:500 (abu:air) dan kemudiannya digoncang menggunakan penggoncang mekanikal selama 24 jam dengan kelajuan 150 rpm. Sisa sagu yang tidak dikisar digunakan untuk membuat kompos. Terdapat tiga jenis rawatan untuk membuat kompos iaitu: T1: Sisa sagu (80%) + makanan ayam (5%) + cairan tahi ayam (5%) + molases (5%) + urea (5%), T2: Sisa sagu (80%) + makanan ayam (10%) + cairan tahi ayam (5%) +

molases (5%) dan T3: Sisa sagu (80%) + makanan ayam (10%) + cairan tahi ayam (5%) + urea (5%). Pengkomposan dijalankan selama 60 hari di dalam kotak polisterin putih yang bersaiz 61.5 x 49 x 33.5cm. Kompos yang dihasilkan kemudiannya dianalisis untuk pH, jumlah nitrogen, karbon organik, bahan organik, abu, kapasiti pertukaran kation (KPK), fosforus dan AH mengikut prosedur piawai. Hidroksida yang diekstrakkan daripada abu sisa sagu digunakan untuk mengasingkan AH dari kompos sisa sagu. Molariti dan pH hidroksida tersebut adalah 0.002 M dan pH 10. Kandungan Ca (42.88mg kg⁻¹) dan kalium (29.51 mg kg⁻¹). Hidroksida tersebut mampu mengekstrakkan AH sebanyak 1.15% daripada kompos sisa sagu. Perbandingan di antara hasil AH yang diekstrakkan oleh hidroksida sisa sagu adalah sama secara statistiknya dengan gred analitikal. Kesemua rawatan kompos tidak mencapai tahap termofilik. Kompos T2 mempunyai kualiti yang tinggi (pH, jumlah nitrogen, karbon organik, bahan organik, abu, kapasiti pertukaran kation (KPK), fosforus dan AH) berbanding T1 dan T3. Ciri-ciri kompos T1 dan T3 adalah sama. Sifat kimia AH bagi ketiga-tiga rawatan adalah di dalam julat piawaian seperti yang dilaporkan di dalam kajian-kajian lain. Selain asid humik (AH), larutan asid fulvik (AF) serta humin juga diekstrakkan daripada hampas. Rawatan bagi eksperimen pasu adalah: kawalan tanpa baja (T1); NPK (4.85 g urea, 4.85 g TSP, 2.5 g KCl) (pepejal) (T2), 400 mL larutan AF + AH yang dicampurkan dengan 4.85 g urea dan 2.5 g KCl (T3), larutan AH dicampur bersama 4.85 g urea dan 2.5 g KCl (T4), 400 mL hidroksida (diekstrakkan dari abu) dicampurkan bersama 4.85 g urea dan 2.5 g KCl + 200 g humin di dalam tanah (T5) dan 400 mL larutan AF + AH yang dicampurkan dengan 4.85 g urea dan 2.5 g KCl + 100 g humin

dalam tanah (T6). Rawatan yang mempunyai humin (T5 dan T6) mempunyai berat kering dan kecekapan penggunaan nutrient yang tertinggi. Sisa sagu boleh digunakan secara berkesannya dengan menghasilkan produk yang berkualiti seperti kompos, hidroksida Ca-K dan baja berasaskan organik untuk pertanian. Penyelidikan yang mendalam tentang kesan sifat fizik tanah terhadap kecekapan penggunaan nutrien pada masa akan datang adalah sangat digalakkan untuk lebih memahami tentang perkaitan di antara tanah-baja dengan tanaman.



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I certify that a Thesis Examination Committee has met on (7 May 2010) to conduct the final examination of (Auldry Chaddy anak Petrus Rudut) on his (or her) thesis entitled "**Laboratory Production of Organic-based Fertilizer from Sago (*Metroxylon sagu Rottb.*) Waste Compost**" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the (Master of Science).

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DECLARATION

I declare that this thesis is my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.

AULDRY CHADDY ANAK PETRUS RUDUT

Date: 7 May 2010



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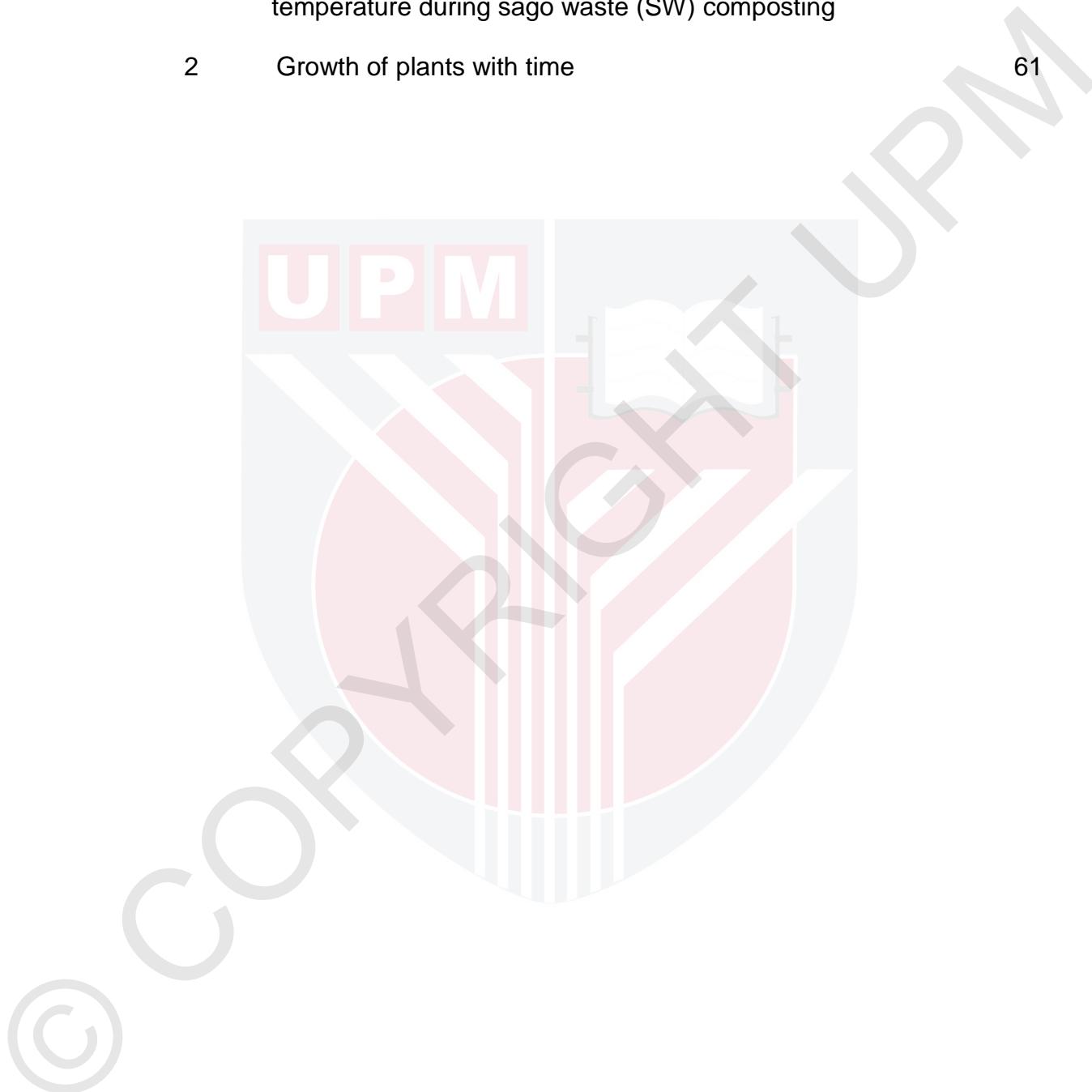
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LIST OF ABBREVIATIONS

SW	Sago waste
HA	Humic acid
FA	Fulvic acid
CEC	Cation exchange capacity
C/N	Carbon/Nitrogen
KOH	Potassium hydroxide
Ca-K	Calcium-potassium
K-N	Potassium-nitrogen
KCl	Potassium chloride
TSP	Triplesuperphosphate
DAP	Day after planting
MARDI	Malaysian Agriculture Research and Development Institute
BOD	Biochemical oxygen demand
COD	Chemical oxygen demand
AAS	Atomic Absorption Spectrophotometry

CHAPTER 1

INTRODUCTION

Palms are one of the oldest families of plants on earth and many groups have developed cultures based on palm trees. The sago palm (*M. sagu Rottb*) is a monocotyledonous tree local to equatorial swamplands and it is becoming an important source of industrial starch.

The sago (*Metroxylon* sp.) starch industry in Malaysia is based mainly in the state of Sarawak. Sarawak, at present is the principal producer of sago, exporting about 25 000 to 40 000 tonnes of sago starch annually (Apun *et al.*, 2009). The sago industry has become large and as a result more logs are processed thus increasing the quantity of waste products from the logs. The sago palm trunk waste produced by the sago starch industries is type of lignocellulosic waste material available in large quantities but has no commercial value (Akmar and Kennedy, 2001). The residue from starch extraction is a very strong pollutant because of its cellulosic fibrous material (Abd. Aziz, 2002). The amount of waste (fibre and water) from processing sago is about 20 times the total starch production (Haska, 2002) with approximately 7 tonnes of fibre was produced daily from a single sago starch processing mill (Bujang *et al.*, 1996). In addition, about 50-110 tonnes of sago fibre are produced daily especially in Mukah and Sibu Division (Awg-Adeni *et al.*, 2010)

The sago waste is usually washed off into the drain. In some situations, the waste from processing sago is drained into the river or sea and this method of disposal causes water pollution. This is because microbiological degradation which occurs in rivers needs substantial amount of oxygen. This causes reduction in dissolved oxygen in water for fishes which require more than 10 g m^{-3} of dissolved oxygen (Cecil, 2002). Recently, The Borneo Post Online (2010) had reported that four out of six sago flour processing plants in Mukah, Sarawak are found to release pollutants (sago fibre) into the river. Normal COD and BOD should be around 100 mg L^{-1} but Sarawak State Environmental Department had revealed that water samples from affected rivers had showed COD reading of $450 - 700 \text{ mg L}^{-1}$ and BOD level from 150 to 200 mg L^{-1} which contravened the standard limit discharge enacted in the Environmental Quality Act, 1974 (sewage and industrial effluents regulation, 1979) (Awg-Adeni *et al.*, 2010).

It is believed that low nutrients and high moisture content render this waste to be a non-valuable product. At the moment, SW serves as substrate for cultivation of edible mushrooms (Haska, 2002), animal feed (Mohd. Sukri, 1992), production of enzymes (Singhal *et al.*, 2007) and absorbents (Quek *et al.*, 1998; Kadirvelu *et al.*, 2004).

Since there is ever increasing quantity of sago waste, another option of managing this waste efficiently is to add value to it, hence, this study was conducted to: 1) Extract calcium (Ca) and potassium (K) hydroxide from SW so as to use it to isolate humic acid (HA), fulvic acid (FA) and humin from

composted sago waste, 2) Produce organic based K and ammonium fertilizer from the composted sago waste, and 3) Determine the efficiency of the organic fertilizers in maize cultivation.



REFERENCES

1. Abd-Aziz, S. 2002. Sago Starch and Its Utilisation. *Journal of Bioscience and Bioengineering*, 94: 526-529.
2. Adani, F., Genevini, P., Tambone, F. and Montoneri, E. 2006. Compost effect on doil humic acid: A NMR study. *Chemosphere*. 65: 1414-1418.
3. Adhikari, B.K., Barrington, S., Martinez, J. and King, S. 2007. Characterization of food waste and bulking agents for composting. *Waste Management*. 1-10
4. Ahmed, O.H., Husni M.H.A., Anuar A.R., Hanafi, M.M. and Angela E.D.S. 2004. A modify way of producing humic acid from pineapple leaves. *Journal of Sustainable Agriculture*. 25(1):129-139
5. Ahmed, O.H., Husni M.H.A., Anuar A.R., Hanafi, M.M., 2003. Alternative means of recycling pineapple leaf residues. *Fruits*, 58 :53-60
6. Ahmed, O.H., Husni, M.H.A., Hanafi, M.H., Anuar, A.R. and Rastan, S.O.S. 2005. Applied K fertilizer use efficiency in pineapples grown on a tropical peat soil under residues removal. *The Scientific World Journal*, 5: 42-49
7. Akmar, P.F. and Kennedy, J.F. 2001. The potential of oil and sago palm trunk wastes as carbohydrate resources. *Wood Science and Technology*, 35: 467-473
8. Alberts, J.J., Filip, Z. and Hertkorn, N. 1992. Fulvic and humic acids isolated from groundwater: Compositional characteristics and cation binding. *Journal of Contaminant Hydrology*, 11: 317-330
9. Almendros, G. and Gonzalez-Vila, F.J. 1989. Degradative studies on a soil humin fraction-sequential degradation of inherited humin. *Soil Bid. Biokm.* 19: 513-520.
10. Amir, S., Hafidi, M., Merlina, G. and Revel, J-C. 2005. Structural characterization of fulvic acids during composting of sewage sludge. *Process Biochemistry*, 40: 1693–1700
11. Andelkovic, T., Perovic, J., Blagojevic, S., Purenovic, M., Nikolic, R., Bojic, A. and Andelkovic, D. 2006. Acidity of humic acid related to its oxygen-containing functional groups. *Buletin of the Chemists and Technologists of Macedonia*. 25: 131-137. ISSN 0350 – 0136

12. Apun, K., Lihan,S., Wong, M.K. and Bilung, L.M. 2009. Microbiological characteristics of trunking and non-trunking sago palm peat soil. Programme and Abstract. 1st ASEAN Sago Symposium 2009. Current trend and development in sago research. October 29-30, 2009. Riverside majestic Hotel, Kuching, Sarawak, Malaysia.
13. Arvanitoyannis, I.S. and Kassaveti, A. 2007. Current and potential uses of composted olive waste. International Journal of Food Science and Technology, 42: 281-295
14. Asik, B.B., Turan, M.A., Celik, H. and Katkat, A.V. 2009. Effects of humic substances on plant growth and mineral nutrients uptake of wheat (*Triticum durum* cv. Salihli) under conditions of salinity. Asian Journal of Crop Science, 1-9
15. Atiyeh, R.M., Lee, S., Edwards, C.A., Arancon, N.Q. and Metzger, J.D. 2002. The influence of humic acids derived from earthworm-processed organic wastes on plant growth. Bioresource Technology 84: 7–14
16. Awg-Adeni, D.S., Abd-Aziz, S., Bujang, K. and Hassan, M.A. 2010. Biconversion of sago residue into value added products. African Journal of Biotechnology, 9 (14): 2016-2021
17. Baglieri, A., Ioppolo, A., Negre, M., and Gennari, M. 2007. A method for isolating soil organic matter after the extraction of humic acids and fulvic acids. Organic Geochemistry. 38: 140-150
18. Barrington, S., Choiniere, D., Trigui, M. and Knight, W. 2002. Effect of carbon source on compost nitrogen and carbon losses. Bioresource Technology 83: 189–194
19. Bar-Tal, A., Yermiyahu, U., Beraud, J., Keinan, M., Rosenber, R., Zohar, D., Rosen, V. and Fine, P. 2004. Nitrogen, phosphorus, potassium uptake by wheat and their distribution in soil following successive, annual compst applications. J. Environ. Qual., 33: 1855-1865.
20. Bernal, M.P., Alburquerque, J.A. and Moral, R. 2008. Composting of animal manures and chemical criteria for compost maturity assessment. A review. Bioresource Technology. 1-10
21. Blondeau, R. 1986. Comparison of soil humic and fulvic acids of similar molecular weight. Org. Geochem, 9:47-50
22. The Borneo Post Online. 2001. Sago processing plants polluting rivers : Dept. 19 February 2010.

23. Brady, N.C. and Weil, R.R. 2002. *The Nature and Properties of Soils*, 13th Edn., ISBN: 0130167630. Pearson Education, Inc, New Jersey.
24. Bremner, J.M. 1965. Total nitrogen. American Society of Agronomy Monograph, 9: 1149-1178
25. Brito, L.M., Coutinho,J. and Smith, S.R. 2008. Methods to improve the composting process of the solid fraction of dairy cattle slurry. *Bioresource Technology*, 99: 8955-8960
26. Bujang, K., Apun, K. and Dieter, B. 1996. A study in the production and bioconversion of sago waste. In Jose C, Rasyad A (eds) *Sago-The future Source of Food and Feed*. Riau University Press, Indonesia, pp. 195-201
27. Bustamante, M.A., Paredes, C., Marhuenda-Egea, F.C., Pérez-Espinosa, A., Bernal, M.P. and Moral, R. 2008. Co-composting of distillery wastes with animal manures: Carbon and nitrogen transformations in the evaluation of compost stability. *Chemosphere*, 72 : 551-557
28. Calace, N., Petronio, B.M., Persia, S., Pietroletti, M. and Pacioni, D. 2007. A new analytical approach for humin determination in sediments and soils. *Talanta*. 71: 1444-1448
29. Cambardella, C.A., Richard, T.L. and Russell A. 2003. Compost mineralization in soil as a function of composting process conditions. *European Journal of Soil Biology*, 39 : 117-127
30. Campitelli P. and Ceppi, S. 2008. Effects of composting technologies on the chemical and physicochemical properties of humic acids. *Geoderma*, 1-9
31. Campitelli, P.A., Velasco M.I. and Ceppi, S.B. 2006. Chemical and physicochemical characteristics of humic acids extracted from compost, soil and amended soil. *Talanta*. 69: 1234-1239
32. Cayuela, M.L., Mondini, C., Insam, H., Sinicco, T. and Franke-Whittle,I. 2009. Plant and animal wastes composting: Effects of the N source on process performance. *Bioresource Technology*. 100: 3097-3106
33. Cayuela, M.L., Sanchez-Monedero, M.A. and Roig A. 2006. Evaluation of two different aeration systems for composting two-phase olive mill wastes. *Process Biochemistry*. 41:616-623. DOI: 10.1016/j.procbio.2005.08.007

34. Cecil, J. 2002. The development of technology for the extraction of sago. New Frontiers of Sago Palm Studies: Proceedings of the International Symposium on SAGO (SAGO 2001). Kainuma, K., Okazaki, M., Toyoda, Y and Cecil, J.E (ed). Tsubuka International Congress Center, Japan. October 15-17 2001. Universal Academy Press, Inc., Tokyo, Japan. ISBN: 4-946443-71-1. pp: 83-91
35. Ceppi, S.B., Velasco, M.I. and De Pauli, C.P. 1999. Differential scanning potentiometry: surface charge development and apparent dissociation constants of natural humic acids. *Talanta*, 50: 1057-1063
36. Chang, J.I., Tsai, J.J. and Wu, K.H. 2006. Thermophilic composting of food waste. *Bioresource Technology*. 116:122
37. Chefetz, B., P.H. Hatcher, Y. Hadar and Y. Chen, 1996. Chemical and biological characterization of organic matter during composting of municipal solid waste. *J. Environ. Qual.*, 25: 776-785
38. Chilom, G., Bruns, A.S. and Rice, J.A. 2009. Aggregation of humic acid in solution: Contributions of different fractions. *Organic Geochemistry*, 40: 455-460
39. Coles, C.A. and Yong, R.N. 2006. Humic acid preparation, properties and interactions with metals lead and cadmium. *Engineering Geology*, 85:26-32
40. Cottenie, A. 1980. Soil testing and plant testing as a basis of fertilizer recommendation. *FAO Soils Bull.* 38: 70-73
41. CRAUN Research Sdn. Bhd. 2010.
42. Department of Environment, (DOE), 1974. Environmental Quality Act, 1974. Department of Environment, Malaysia
43. Department of Statistics. 2007. Department of Agriculture Sarawak, 2007. <http://www.doa.sarawak.gov.my>
44. Drori, Y., Aizenshtat, Z. and Chefetz, B. 2008. Sorption of organic compounds to humin from soils irrigated with reclaimed wastewater. *Geoderma* 145: 98-106
45. Eango, D., Thinakaran, N., Panneerselvam, P. and Sivanesan, S. 2009. Thermophilic composting of municipal solid waste. *Applied Energy*. 86: 663-668
46. Eyheraguibel, B., Silvestre, J. and Morard, P. 2008. Effects of humic substances derived from organic waste enhancement

- on the growth and mineral nutrition of maize. *Bioresource Technology* 99: 4206–4212
47. Flach, M. 1997. Sago palm. *Metroxylon sagu Rottb.* International Plant Genetic Resources Institute, Rome, Italy. Pp. 8-11. ISBN: 92-9043-314-X
 48. Fong, S.S. and Mohamed, M. 2007. Chemical characterization of humic substances occurring in the peats of Sarawak, Malaysia. *Organic Geochemistry*, 38: 967-976
 49. Frederickson , J., Howell, G. and Hobson, A.M. 2007. Effect of pre-composting and vermicomposting on compost characteristics. *European Journal of Soil Biology*, 43: S320-S326
 50. Freemantle, M. 1987. *Chemistry in action*. Macmillan Education Ltd, Hounds Mills, Basingstoke, Hampshire, London. ISBN: 0-333-44497-3
 51. Fukushima, M., Yamamoto, K., Ootsuka, K., Komai, T., Aramaki, T., Ueda, S. and Horiya, S. 2008. Effects of the maturity of wood waste compost on the structural features of humic acids. *Bioresource Technology*. 1-7
 52. Garcia, C., Hernandez, T. and Costa, F. 1992. Characterization of humic acids from uncomposted and composted sewage sludge by degradative and non-degradative techniques. 41:53-57
 53. Garcia, C., Hernandez, T., Costa, F. and del Rio, J.C. 1989. Study of the lipidic and humic fractions from organic wastes before and after the composting process. *The Science of the Total Environment*, 89/82: 551-560
 54. Garcia-Gil, J.C., Ceppi, S.B., Velasco, M.I., Polo, A. and Senesi, N. 2004. Long-term effects of amendment with municipal solid waste compost on the elemental and acidic functional group composition and buffer capacity of soil humic acids. *Geoderma*, 121: 135-142
 55. Garcia-Gil, J.C., Plaza, C., Fernandez, J.M., Senesi, N. and Polo, A. 2008. Soil fulvic acid characteristics and proton binding behaviour as affected by long-term municipal waste compost amendment under semi-arid environment. *Geoderma*. 146: 363-369
 56. Gil, M.V., Carballo, M.T. and Calvo, L.F. 2007. Fertilization of maize with compost from cattle manure supplemented with additional mineral nutrients. *Waste Management*, 1-9

57. Gil, M.V., Carballo, M.T. and Calvo, L.F. 2007. Fertilization of maize with compost from cattle manure supplemented with additional mineral nutrients. *Waste Management*, 1-9
58. Gomez-Brandon, M., Lazcano, C. and Dominguez, J. 2008. The evaluation of stability and maturity during the composting of cattle manure. *Chemosphere*. 70: 436-444
59. Gong Chun-ming, 2007. Microbial safety control of compost material with cow dung by heat treatment. *Journal of Environmental Sciences* 19: 1014–1019
60. Govindasamy, R. and Chandrasekaran, S. 1992. Effect of humic acids on the growth, yield and nutrient content of sugarcane. *The Science of the Total Environment*, 117/118: 575-581
61. Guardia, A., Petiot, C., Rogeau, D. and Druilhe, C. 2008. Influence of aeration rate on nitrogen dynamics during composting. *Waste Management*. 28: 575-587
62. Guedes de Carvalho, R.A., Gonzfilez Beqa, C.G., Neves, O.R. and Sol Pereira, M.C. 1991. Composting of pine and eucalyptus barks. *Bioresource Technology* 38: 51-63
63. Haska, N. 2002. The utilization of the fibrous residue of sago palm as a substrate for the cultivation of edible mushrooms. *New Frontiers of Sago Palm Studies: Proceedings of the International Symposium on SAGO (SAGO 2001)*. Kainuma, K., Okazaki, M., Toyoda, Y and Cecil, J.E (ed). Tsubuka International Congress Center, Japan. October 15-17 2001. Universal Academy Press, Inc., Tokyo, Japan. ISBN: 4-946443-71-1. pp: 133-140
64. Helal, A.A., Imam, D.M., Khalifa, S.M. and Aly, H.F. 1998. Effect of some environmental ligands and fertilizers on humic acid complexation with strontium. *Journal of Radioanalytical and Nuclear Chemistry*, 232: 159-161
65. Horiuchi, J-I., Ebie, K., Tada, K., Kobayashi, M. and Kanno, T. 2003. Simplified method for estimation of microbial activity in compost by ATP analysis. *Bioresource Technology*. 86: 95-98
66. Imbeah, M. 1998. Composting piggery waste: A review. *Bioresource Technology*, 63:197-203
67. Inbar, Y., Chen, Y. and Hadar, Y. 1990. Humic Substances Formed during the Composting of Organic Matter. *Soil Sci. Soc. Am. J.*, 54:1316-1323

68. Janos, P., Krizenecka, S. and Madronova, L. 2008. Acid-base titration curves of solid humic acids. *Reactive and Functional Polymers*, 68: 242-247
69. Jimenez, E.I. and Garcia, V.P. 1992. Determination of maturity indices for city refuse composts. *Agriculture, Ecosystems and Environment*, 38: 331-343
70. Jouraiphy, A., Amir, S., El Gharous, M., Revel, J.C. and Hafidi, M. 2005. Chemical and spectroscopic analysis of organic matter transformation during composting of sewage sludge and green plant waste. *International Biodeterioration and Biodegradation*, 56: 101-108
71. Kadirvelu, K., Kavipriya, M., Karthika, C., Vennilamani, N and Pattabhi, S. 2004. Mercury (II) adsorption by activated carbon made from sago waste. *Carbon* 42: 745–752
72. Kalamdhad, A.S., Pasha, M. and Kazmi, A.A. 2008. Stability evaluation of compost by respiration techniques in a rotary drum composter. *Resources, Conservation and Recycling*. 52: 829-834
73. Kanwar, R.S. 1988. The nitrogen loss in drained soils. *Math1 Comput. Modelling*, 11 : 519-522
74. Keeney, D.R. and D.W. Nelson, 1982. Nitrogen- Inorganic Forms. In: *Methods of Soil Analysis*, Part 2, Page, A.L., D.R. Keeney, D.E. Baker, R.H. Miller, R. Jr. Ellis and J.D. Rhoades (Eds.). 2nd Edn., Agron. Monogr. 9. ASA and SSSA, Madison, WI. ISBN: 0891180729 (pt. 2)
75. Khandelwal, K.C. 1977. Effect of a formulation of urea, thiourea and humate on nitrogen availability and yield of wheat. *Plant and Soil*, 47: 717-719.
76. Kohl, S.D. and Rice, J.A. 1998. The Binding of Contaminants to Humin: A Mass Balance. *Chemosphere*, 36: 251-261
77. Koivula, N., Hanninen, K. and Tolvanen, O. 2000. Windrow composting of source separated kitchen biowastes in Finland. *Waste Manage. Res*, 18: 160-173
78. Koivula, N., Raikkonen, T., Urpilainen, S., Ranta, J. and Hanninen, K. 2004. Ash in composting of source-separated catering waste. *Bioresource Technology*. 93: 291-299

79. Kostov, O., Petkova, G. and Van Cleemput, O. 1994. Microbial indicators for sawdust and bark compost stability and humification processes. *Bioresource Technology*, 50: 193-200
80. Kulcu, R. and Yaldiz, O. 2004. Determination of aeration rate and kinetics of composting some agricultural wastes. *Bioresouce Technology*, 93: 49-57
81. Kulcu, R. and Yaldiz,O. 2007. Composting of goat manure and wheat straw using pine cones as a bulking agent. *Bioresouce Technology*. 98: 2700-2704
82. Kulcu, R. and Yaldiz,O. 2007. Effects of air flow directions on composting process temperature profile. *Waste Management*. 1-7
83. Lakhdar, A., Rabhia, M., Ghnayaa,T. and Montemurroc,F. 2009. Effectiveness of compost use in salt-affected soil. *Journal of Hazardous Materials*, 1-9
84. Lazcano, C., Gómez-Brandón, M. and Domínguez, J. 2008. Comparison of the effectiveness of composting and vermicomposting for the biological stabilization of cattle manure. *Chemosphere* 72:1013–1019
85. Lima, J.S., de Queiroz, J.E.G and Freitas, H.B. Effect of selected and non-selected urban waste compost on the initial growth of corn. *Resources, Conservation and Recycling* 42: 309–315
86. Linder, P.W. and Murray, K. 1987. Statistical determination of the molecular structure and the metal binding sites of fulvic acids. *The Science of the Total Environment*. 64: 149-161
87. Linehan, D.J. 1976. Some effect of a fulvic acid component of soil organic-matter on the growth of cultured excised tomato roots. *Sod Bml Bmchrm.* 8: 511 - 517.
88. Looper, M. 2002. Whole animal composting of dairy cattle. College of Agriculture and Home Economics. Guide D-108
89. Lulakis, M.D. and Petsas, S.I. 1995. Effect of humic substances from vine-canapes mature compost on tomato seedling growth. *Bioresouce Technology* 54 :179-182
90. Malik, K.A. and Azam, F. 1985. Effect of humic acid on wheat (*Triticum aestivul* L.) seedling growth. *Environmentaland Experimental Botany*, 25: 245-252
91. Institut Penyelidikan dan Kemajuan Pertanian Malaysia (MARDI), 1990. Jagung Manis Baru: Masmadu.

92. Masciandaro, G., Ceccanti, B. and Garcia, C. 2000. "In situ" vermicomposting of biological sludges and impacts on soil quality. *Soil Biology and Biochemistry*, 32: 1015-1024
93. Masini, J.C., Abate, G., Lima, E.C., Hahn, L.C., Nakamura, M.S., Lichtig, J. and Nagatomo, R. 1998. Comparison of methodologies for determination of carboxylic and phenolic groups in humic acids. *Analytics Chimica Acta*, 364: 223-233
94. Mengel, K. and Kirby, E.A. 1996. Principles of plant nutrition (4th edition). Panina Publishing Corporation, New Delhi, pp:147-149
95. Mikki, V., Senesi, N. and Hanninen, K. 1997. Characterization of humic material formed by composting of domestic and industrial biowastes. Part 2: Spectroscopic evaluation of humic acid structures. *Chemosphere*. 8: 1639-1651
96. Minar, J. and Laztuvka, Z. 1969. The Dynamics of the Accumulation of Nitrogen, Phosphorus and Potassium in Maize and Peas in the First Growth Phases at Constant Mineral Nutrition. *Biologia Plantarum (PRAHA)*, 11 (2) : 149—157
97. Mohamad Ishak Jaafar. 1987. Siri pemakanan ternakan: Sumber makanan dan perumusannya. Dewan Bahasa dan Pustaka, Kementerian Pendidikan Malaysia, Kuala Lumpur
98. Mohd. Sukri, H.I. 1992. Batang sagu sebagai sumber makanan ternakan. Institut Penyelidikan dan Kemajuan Pertanian Malaysia (MARDI), Kuala Lumpur, Malaysia. ISSN 0127-4007. ISBN 967-936-134-9
99. Mondini, C., Sanchez-Monedero, M.A., Sinicco, T. and Leita, L. 2006. Evaluation of extracted organic carbon and microbial biomass as stability parameters in ligno-cellulosic waste composts. *J. Environ. Qual.*, 35: 2313-2320
100. Murphy, J. and Riley, J.I. 1962. A modified single solution method for the determination of phosphate in natural waters. *Anal. Chm. Acta*, 27: 31-36
101. Mylavaram, R.S and Zinati, G.M. 2009. Improvement of soil properties using compost for optimum parshley production in sandy soils. *Scientia Horticulturae*, 120: 426-430
102. Nelson, M.I., Marchant, T.R., Wake, G.C., Balakrishnan, E. and Chen, X.D. 2007. Self-heating in compost piles due to biological effects. *Chemical Engineering*, 62: 4612-4619

103. Oates, C. and Hicks, A. 2002. Sago starch production in Asia and the Pacific-Problems and Prospects. New Frontiers of Sago Palm Studies: Proceedings of the International Symposium on SAGO (SAGO 2001). Kainuma, K., Okazaki, M., Toyoda, Y and Cecil, J.E (ed). Tsubuka International Congress Center, Japan. October 15-17 2001. Universal Academy Press, Inc., Tokyo, Japan. ISBN: 4-946443-71-1. pp: 27-36
104. Ozawa, T., Ueno, T., Negishi, O., Masaki S. and Amari, M. 2002. Fundamental research on the production of feed from sago residue after the extraction of starch. New Frontiers of Sago Palm Studies: Proceedings of the International Symposium on SAGO (SAGO 2001). Kainuma, K., Okazaki, M., Toyoda, Y and Cecil, J.E (ed). Tsubuka International Congress Center, Japan. October 15-17 2001. Universal Academy Press, Inc., Tokyo, Japan. ISBN: 4-946443-71-1. pp. 321-324
105. Pandeya, S.B., Singh, A.K. and Dhar,P. 1998. Influence of fulvic acid on transport of iron in soils and uptake by paddy seedlings. Plant and Soil 198: 117–125.
106. Paramanathan, S., 2000. Soils of Malaysia: Their Characteristics and Identification. Vol. 1. Academy of Sciences Malaysia, Kuala Lumpur. ISBN: 9839445065
107. Parrado J., Bautista, J., Romero, E.J., Garcí'a-Martí'nez, A.M., Friaza, V. and Tejada, M. 2007. Production of a carob enzymatic extract: Potential use as a biofertilizer. Bioresource Technology, pp. 1-7
108. Pena-Mendez E.M., J. Havel and J. Patocka, 2005. Humic substances-compounds of still unknown structure: applications in agriculture, industry, environment, and biomedicine. Journal of Applied Biomedicine. 3 :13-24
109. Piccolo, A. 1996.Humis and soil conservation. Humic Substances in Terrestrial Ecosystems, 1996, Pages 225-264
110. Poapst, P.A. and Schnitzer, M. 1971. Fulvic acid and adventitious root formation. Soil Biol.Biochem. 3: 215-219.
111. Pomares-Garcia F. and P.F. Pratt, 1987. Recovery of ¹⁵N-labelled fertilizer from manured and sludged-amended soils. *Soil Science Society of American Journal*, 42: 717-720
112. Quek, S.Y., Wase, D.A.J. and Forster, C.F. 1998. The use of sago waste for the sorption of lead and copper. Water S.A. 24: 251-256.

113. Rice, J.A. and MacCarthy, P. 1988. Comments on the Literature of the Humin Fraction of Humus. *Geoderma*, 43: 65-73
114. Rice, J.A. and MacCarthy, P. 1992. Disaggregation and characterization of humin. *The Science of the Total Environment*. 117/118: 83-88
115. Rivero, C., Chirenje, T., Ma, L.Q. and Martinez, G. 2004. Influence of compost on soil organic matter quality under tropical conditions. *Geoderma*, 123: 355-361
116. Ros, M., Klammer, S., Knapp B., Aichberger, K. and Insam H. 2006. Long-term effects of compost amendment of soil on functional and structural diversity and microbial activity. *Soil Use and Management*, 22: 209- 218
117. Ryckeboer, J., Mergaert, J., Coosemans, J., Deprins, K. and Swings, J. 2003. Microbiological aspects of biowaste during composting in a monitored compost bin. *Journal of Applied Microbiology*. 94:127-137.
118. Saludes, R.B., Iwabuchi, K., Kayanuma, A. and Shiga, T. 2007. Composting of dairy cattle manure using a thermophilic-mesophilic sequence. *Biosystems Engineering*, 98: 198-205
119. Sanchez-Monedero, M.A., Cegarra, J., Garcia, D. and Roig, A. 2002. Chemical and structural evolution of humic acids during organic waste composting. *Biodegradation*. 13: 361-371
120. Sasaki, S., Yamaguchi, C., Tanaka, H. , Ohmi, M., Tominaga, H. and Fukuda, K. 2002. Characteristics of esterified sago residue. *New Frontiers of Sago Palm Studies: Proceedings of the International Symposium on SAGO (SAGO 2001)*. Kainuma, K., Okazaki, M., Toyoda, Y and Cecil, J.E (ed). Tsubuka International Congress Center, Japan. October 15-17 2001. Universal Academy Press, Inc., Tokyo, Japan. ISBN: 4-946443-71-1. pp: 337-341
121. Schnitzer, M. and Kerndorff, H. 1980. Reactions of fulvic acid with metal ions. *Water, Air, and Soil Pollution* 15: 97-108
122. Sellami, F., Hachicha, S., Chtourou, M., Medhioub, K. and Ammar, E. 2008. Maturity assessment of composted olive mill wastes using UV spectra and humification parameters. *Bioresource Technology*, 1- 8
123. Senesi, N., Miano, T.M., Provenzano, M.R. and Brunetti, G. 1989. Spectroscopic and compositional comparative characterization of I.H.S.S reference and standard fulvic and humic acids of various

- origin. *The Science of the Total Environment*, 81/82 : 143-156
124. Senn, M., Durst, B., Kaufmann, A. and Langhans, W. Feeding Patterns of Lactating Cows of Three Different Breeds Fed Hay, Corn Silage and Grass Silage. *Physiology and Behavior*, 58: 229-236
 125. Shen, Y-H. 1999. Sorption of humic acid to soil: The role of soil mineral composition. *Chemosphere*. 38: 2489-2499
 126. Singer, M.J. and Munns, D.N. 2006. Soils: An introduction. Pearson, Prentice Hall, Upper Saddle River, New Jersey, Columbus, Ohio. ISBN: 0-13-119019-9
 127. Singhal, R.S., Kennedy, J.F., Sajilata, M.G., Kaczmarek, A., Knill, C.J. and Akmar, P.F. 2007. Industrial production, processing, and utilization of sago palm-derived products. *Carbohydrate Polymers*, 1-20
 128. Smidt, E., Meissl, K., Schmutzler, M. and Hinterstoisser, B. 2007. Co-composting of lignin to build of humic substances- Strategies in waste management to improve compost quality. *Industrial Crops and Products*, 1-6
 129. Soane, B.D. 1990. The Role of Organic Matter in Soil Compactibility: A Review of Some Practical Aspects. *Soil and Tillage Research*, 16: 179-201
 130. Stevenson, F.J., 1994. Humus chemistry: genesis, composition, reactions, 2nd Edn., John Wiley and Sons, New York, ISBN: 0471594741. pp. 36
 131. Sundberg, C and Jo"nsson, H. 2008. Higher pH and faster decomposition in biowaste composting by increased aeration. *Waste Management*, 28: 518–526
 132. Sundberg, C. and Jonsson, H. 2007. Higher pH and faster decomposition in biowaste composting by increased aeration. *Waste Management*, 1-9
 133. Susilawati, K. and Ahmed, O.H., Nik Muhammad, A.M. and Khanif,M.Y. 2008. Simple method of purifying humic acids isolated from tropical hemists (peat soil). *American Journal of Applied Sciences*, 5 (12) :1812-1815
 134. Susilawati, K. Ahmed, O.H., Nik Muhammad, A.B., Khanif, M.Y. and Jalloh, M.B. 2009. Effect of organic based N fertilizer on dry matter (*Zea mays L.*), ammonium and nitrate recovery in an acid soil of Sarawak, Malaysia. *American Journal of Applied Sciences*, 6(7): 1282-1287

135. Tan, K.H. 2004. Soil Sampling, Preparation, and Analysis. Second Edition. CRC Press Taylor and Francis Group, New York. ISBN: 0-8493-3499-3
136. Tan, K.H. and Binger, A. 1986. Effect of humic acid on aluminium toxicity in corn plants. *Soil Science*, 20-25
137. Tan, K.H. and Nopamornbodi, V. 1979. Effect of different levels of humic acids on nutrient content and growth of corn (*Zea mays L.*). *Plant and Soil*, 51: 283-287
138. Tan, K.H., 2003. Humic Matter in Soil and the Environment: Principles and Controversies. Marcel Dekker, Inc., New York, ISBN: 0-8247-4272-9, pp: 39
139. Tan, K.H. 1978. Effects of humic and fulvic acids on release of fixed potassium. *Geoderma*, 21:67-74
140. Tatzber, M., Stemmel, M., Splegel, H., Katzlberger, C., Haberhauer, G., Mentler, A. and Gerzabek, M.H. 2007. FTIR-spectroscopic characterization of humic acids and humin fractions obtained by advanced NaOH, Na₄P₂O₇ and Na₂CO₃ extraction procedures. *J. Plant. Nutr. Soil Sci.* 170: 522-529
141. Timofeevna Shirshova, L., Ghabbour , E.A. and Davies, G. 2006. Spectroscopic characterization of humic acid fractions isolated from soil using different extraction procedures. *Geoderma*. 206-216
142. Tiquia, S.M. 2005. Microbiological parameters as indicators of compost maturity. *Journal of Applied Microbiology*. 99: 816-828
143. Tiquia, S.M., Tam, N.F.Y. and Hodgkiss, I.J. 1997. Effects of turning frequency on composting of spent pig-manure sawdust litter. *Bioresource Technology*, 62: 37-42
144. Tognetti, C., Mazzarino, M.J. and Laos, F. 2007. Improving the quality of municipal organic waste compost. *Bioresource Technology*, 98: 1067–1076
145. Tomati, U., Galli, E., Fiorella, F. and Pasetti, L. 1996. Fertilizers from composting of olive-mill wastewaters. *International Biodeterioration and Biodegradation*. 1196: 155-162
146. Tuomela, M., Vikman, M., Hatakka, A. and Itavaara, M. 2000. Biodegradation of lignin in a compost environment: a review. *Bioresource Technology* 72: 169-183

147. Umi Kalsom, M.S., Nur, H., Norlea, A.A., and Ngaspan, S. 2006. Characterization of humic acid from humification of oil palm empty fruit brunch using *Trichoderma viride*. *J. Trop. Agric. And Fd. Sc.*, 34(1): 165-172
148. Unsal, T and Sozudogru Ok, S. 2001. Description of characteristics of humic substances from different waste materials substances from different waste materials. *Bioresource Technology*, 78: 239-242
149. Veeken, A., Nierop, K., de Wilde, V. and Hamelers, B. 2000. Characterisation of NaOH-extracted humic acids during composting of a biowaste. *Bioresource Technology*, 72: 33-41
150. Vikineswary, S., Shim, Y.L, Thambirajah and J.J. and Blakebrough N. 1994. Possible microbial utilization of sago processing wastes. *Resources, Conservation and Recycling*, 11: 289-296
151. Wang, P., Chang, C.M., Watson M.E., Dick, W.A., Chen, Y. and Hoitink, H.A.J. 2004. Maturity indices for composted dairy and pig manures. *Soil Biology and Biochemistry*. 36 :767-776
152. Wang, W.J., Powell, A.D. and Oates, C.G. 1996. Sago starch as a biomass source: Raw sago starch hydrolysis by commercial enzymes. *Bioresource Technology* 55: 55-61
153. Wei, Z., Xi, B., Zhao, Y., Wang, S., Liu, H., and Jiang, Y. 2007. Effect of inoculating microbes in municipal solid waste composting on characteristics of humic acid. *Chemosphere*, 68 : 368-374
154. You, S-J., Yin, Y. and Allen, H.E. 1999. Partitioning of organic matter in soils: effects of pH and water/soil ratio. *The Science of The Total Environment*, 227: 155-160
155. Zaccone, C., Cocozza, C., D' Orazio, V., Plaza, C., Cheburkin, A. and Miano, T.M. 2007. Influence of extractant on quality and trace elements content of peat humic acids. *Talanta*. 73: 820-830
156. Zbytniewski, R. and Buszewski, B. 2005. Characterization of natural organic matter (NOM) derived from sewage sludge compost. Part 1: chemical and spectroscopic properties. *Bioresource Technology* 96: 471–478
157. Zmora-Nahum, S., Markovitch, O., Tarchitzky, J. and Chen, Y. 2005. Dissolved organic carbon (DOC) as a parameter of compost maturity. *Soil Biology and Biochemistry*. 37: 2109-2116