

# **UNIVERSITI PUTRA MALAYSIA**

STRUCTURAL, ELASTIC AND OPTICAL PROPERTIES OF RICE HUSK SILICATE BOROTELLURITE GLASS SYSTEM DOPED WITH MICRO AND NANOPARTICLES OF ERBIUM OXIDE

ALIYU UMAR SA'AD

FS 2018 88



# STRUCTURAL, ELASTIC AND OPTICAL PROPERTIES OF RICE HUSK SILICATE BOROTELLURITE GLASS SYSTEM DOPED WITH MICRO AND NANOPARTICLES OF ERBIUM OXIDE



Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia in Fulfillment of Requirements for the Degree of Doctor of Philosophy

April 2018

# COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



To them I dedicate this work.

My late father Alh. Sa'adu Aliyu (Rahimahullah) and my mother Khadijatu Sa'adu who have set the guide for my success in life. My Aunt and Uncle, Aishatu Aliyu Sa'ad and Saleh Aliyu Yahaya and my Vice-chancellor, Prof. M. S. Liman who have been there all the time I needed them. To my wife, Fatimah and my son for the love you've shown throughout this journey.



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

# STRUCTURAL, ELASTIC AND OPTICAL PROPERTIES OF RICE HUSK SILICATE BOROTELLURITE GLASS SYSTEM DOPED WITH MICRO AND NANOPARTICLES OF ERBIUM OXIDE

By

#### ALIYU UMAR SA'AD

April 2018

## Chairman : Associated Professor Halimah Mohamed Kamari, PhD Faculty : Science

Successful extraction of high purity SiO<sub>2</sub> (about 99%) from rice husk (waste) was achieved in this work using the cold acid leaching method. Glass series  $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x}$  (SiO<sub>2</sub>)<sub>x</sub> were fabricated using the rice husk silicate (RHS) by meltquenching method. The aim was to extract a very high purity SiO<sub>2</sub> from rice husk with no heat used in leaching to fabricate a system of silicate borotellurite glass system doped with erbium oxide and erbium oxide nanoparticles. The choice of rice husk as silicate source was to add to the commercialization of the rice husk waste management which stand at 30%. Also, to study the structural, elastic and optical properties of the glass systems for Erbium Doped Fiber Amplifier (EDFA) application. The use of rice husk silicate in a borotellurite glass system was expected to improve the glass quality in terms of both elastic and optical properties.

The samples were subjected to X-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FTIR) and UV-Vis characterizations to study the structural nature and optical behavior of the glass system. Density, molar volume and ultrasonic velocities were measured to obtain the elastic constants for the various silicate proportions in the glass using ultrasonic data obtained from non-destructive ultrasonic probing technique. The glasses fabricated were found to be transparent at lower RHS concentrations (0.0-0.2) the glasses became more darkened and opaque from 30% to 40% RHS concentrations. 20% RHS was found to be the best SiO<sub>2</sub> composition in terms of transparency, elastic and optical properties of the glasses for  $Er^{3+}$  ion doping. Using the 0.2 molar fraction of RHS (SiO<sub>2</sub>), a glass system of erbium/erbium NPs doped rice husk silicate borotellurite (Er/Er NPs-doped RHSBT) glass system was fabricated using the chemical composition  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-x}$ 



 $(Er_2O_3)_x$  with x = 0.01, 0.02, 0.03, 0.04 and 0.05 mol was prepared using the meltquenching technique.

The glasses were as well subjected to characterizations as XRD, FTIR, UV-Vis spectroscopy, photoluminescence spectroscopy and non-destructive ultrasonic probing to study the structural, elastic, optical and luminescence properties of the glasses. Theoretical elastic models of Makishima and Mackenzie, Rocherulle, bond compression and ring deformation models were used to study more on the elas tic properties and compare with the experimental data. Density and molar volume measurements were also carried out to further provide structural details while transmission electron microscopy was carried out on the erbium nanoparticles doped series to study the morphological features of the glasses.

In all the three samples, the XRD showed no sharp peak indicating a glassy nature. The density and molar volume in the case of RHSBT glass system decreased with increase in  $SiO_2$  concentration. The ultrasonic velocities, elastic moduli, optical band gaps, Urbach energy, oxygen packing density (OPD), metallization criterion and transmission coefficient increased with RHS concentration increased. The theoretical elastic models confirmed the elastic moduli increase in the silicate borotellurite glass system with increasing  $SiO_2$  content as shown in the experimental study. Only the Makishima and Mackenzie model appeared to be good for the elastic properties of both the erbium oxide doped and erbium oxide NPs doped glass systems.

Using McCumber theory analysis from the UV-Vis data, Absorption and Emission cross-section data was obtained and used to calculate the simple fiber gain of each sample in the two micro and nano erbium doped RHSBT glass systems. The gains were higher in the case of micro erbium doped system than the nano erbium doped system. With high refractive index of between 2.452 to 2.521 and wide effective band with in the S and C bands of between 76 to 106 nm, the micro erbium doped RHSBT glass system provides a good potential for application in the erbium doped fiber amplifiers (EDFA).

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

# SIFAT STRUKTUR, KENYAL DAN OPTIK SISTEM KACA SEKAM PADI SILIKA BOROTELURIT DIDOP DENGAN ERBIUM OKSIDA DAN NANOZARAH ERBIUM OKSIDA

Oleh

#### ALIYU UMAR SA'AD

April 2018

# Pengerusi: Profesor Madya Halimah Mohamed Kamari, PhDFakulti: Sains

Penghasilan SiO<sub>2</sub> yang mempunyai ketulenan yang tinggi (kira-kira 99%) daripada sekam padi berjaya dicapai di dalam kajian ini menggunakan kaedah larut lesap asid sejuk. Siri kaca [(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>1-x</sub> (SiO<sub>2</sub>)<sub>x</sub> telah dihasilkan menggunakan sekam padi silika (RHS) dengan kaedah sepuh lindap. Tujuan kajian ini adalah untuk menghasilkan SiO<sub>2</sub> dengan ketulenan yang tinggi daripada sekam padi tanpa menggunakan haba dalam larut lesap untuk menghasilkan sistem silika borotelurit didop dengan erbium oksida dan nanozarah erbium oksida. Penssunaan sisa sekam padi pada masa kini sebanyak 30% boleh dikumersial apabila sekam padi chipisin sebagai sumber silica. Selain itu, ia juga adalah untuk mempelajari sifat struktur, kenyal dan optik sistem kaca itu bagi aplikasi EDFA. Penggunaan sekam padi silika di dalam sistem kaca borotelurit dijangkakan boleh meningkatkan kualiti kaca itu dalam kedua-dua sifat kekenyalan dan optik.

Sampel-sampel itu diuji dengan pembelauan sinar-X (XRD), spektroskopi jelmaan Fourier Inframerah (FTIR) dan pencirian UV-nampak untuk mengkaji keadaan struktur dan sifat optik kaca tersebut. Ketumpatan, isipadu molar dan halaju ultrabunyi diukur untuk mendapatkan pemalar kenyal bagi kandungan silika yang pelbagai di dalam kaca tersebut menggunakan data ultrabunyi yang diperolehi daripada kaedah pemprobaan ultrabunyi tanpa musnah. Kaca-kaca yang dihasilkan itu ditemui lutsinar pada kepekatan RHS yang rendah (0.0-0.2 mol) dan kaca-kaca itu menjadi lebih gelap dan legap daripada 30% hingga 40% kepekatan. Komposisi yang terbaik untuk transparensi, elastik dan sifat optik untuk pendopan Er<sup>3+</sup> pada gelas adalah 20% RHS. RHS Dengan menggunakan 0.2 mol RHS (SiO<sub>2</sub>), satu sistem kaca erbium/nanozarah erbium didop dengan sekam padi silika borotelurit (Er/Er NPs-doped RHSBT) dihasilkan menggunakan komposisi kimia {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>]<sub>1-x</sub>

 $(Er_2O_3)_x$  dengan x = 0.01, 0.02, 0.03, 0.04 dan 0.05 mol dihasilkan menggunakan teknik sepuh lindap.

Kaca-kaca itu juga dihantar untuk ujikaji XRD, FTIR, spektroskopi UV-Vis, spektroskopi fotopendarcahayaan dan pemprobaan ultrabunyi tanpa musnah untuk mempelajari sifat struktur, kekenyalan, optik dan pendarcahayaan kaca tersebut. Model teori kenyal Makishima dan Mackenzie, Rocherulle, pemampatan ikatan dan model canggaan gegelang digunakan untuk lebih mempelajari mengenai sifat kenyal dan membandingkannya dengan data eksperimen. Pengukuran ketumpatan dan isipadu molar juga dijalankan untuk membekalkan butiran struktur manakala mikroskopi penghantaran electron dilakukan pada siri yang didop nanozarah erbium untuk mempelajari sifat morfologi kaca-kaca tersebut.

Di dalam ketiga-tiga sampel, XRD menunjukkan tiada puncak yang tajam, menandakan keadaan kaca. Ketumpatan dan isipadu molar di dalam kes sistem kaca RHSBT menurun apabila kepekatan SiO<sub>2</sub> bertambah. Halaju ultrabunyi, modulusmodulus kenyal, jurang jalur optik, tenaga Urbach, ketumpatan padatan oksigen (OPD), kriteria penglogaman dan pekali penghantaran menaik dengan kenaikan kepekatan RHS. Model teori kenyal mengesahkan kenaikan modulus-modulus kenyal di dalam sistem silika borotelurit dengan kenaikan kandungan SiO<sub>2</sub> seperti yang ditunjukkan di dalam kajian eksperimen. Hanya model Makishima dan Mackenzie dilihat sesuai untuk sifat kenyal bagi kedua-dua sistem kaca didop dengan erbium oksida dan nanozarah erbium oksida.

Dengan menggunakan teori analisis McCumber daripada data UV-nampak, data keratan rentas penyerapan dan pancaran diperolehi dan digunakan untuk mengira gandaan gentian ringkas untuk setiap sampel di dalam kedua-dua sistem kaca RHSBT didop mikro dan nano erbium. Gandaannya lebih tinggi bagi sistem didop mikro erbium berbanding dengan sistem didop nano erbium. Dengan indeks biasan yang tinggi di antara 2.452 hingga 2.521 dan lebar jalur efektif di dalam jalur S dan C di antara 76 hingga 106 nm, system kaca RHSBT didop mikro erbium memberikan potensi yang baik untuk aplikasi di dalam gentian amplifier didop erbium (EDFA).

## ACKNOWLEDGEMENTS

All praises to Allah (S.W.T) for the completion of this thesis. This would have been impossible without His grace.

To the following people who have contributed immensely to the success of this thesis and the program as whole. First to my supervisor, Professor Halimah Mohamed Kamari who has been so supportive, kind and caring in every respect throughout this journey, I say a big thank you to you.

To my co-supervisors, Dr. Chan Kar Tim and Dr. Amirah Abdul Latif I show my sincere appreciation for your continued contributions and support. The advices, suggestive, constructive criticisms, recommendations and commendations were pillar to success of this work. I say to you a big thank you.

Many thanks to my research group mates and friends, Hamza, Abdulbaset, Abdullahi, Faznny, Zaiti, Dr. Azlan, Dr. Hasnie, Dr. Eevon, Dr. Auwalu, Mahraz, Yati, Nazrin and Salah. Others who have been helpful in this journey are my brothers here, Auwal Yahaya, Dr. Sulaiman Chindo, Dr. Kabiru Maji, Idris Bugaje, Ahmad Bugaje, Mal. Yahaya and others I could not mention.

To my aunts, uncles, siblings, friends and family members I would like to say a big thank you for your endless support. To my wonderful and caring wife, words would not be sufficient enough to explain your support, patience and love that kept me working. You are one in a million. Thank you, may Allah reward you. To you hayatee (soul), I cannot repay your kindness in terms of love, support and patience. You are such a blessing to my life and an irreplaceable soul that I placed high. Jazakillah bi khair.

Finally, to my brothers in struggle, the "Concerned brothers"; Ibrahim Gambo, Barr. Ibrahim Danjuma, Muhammad Inuwa and Auwal Yahaya Tilde. Being part of you gave me comfort, strength and hope to keep pushing throughout the trying times. Thank you for considering and keeping me among you guys. You mean a lot to me guys.

I certify that a Thesis Examination Committee has met on 26 April 2018 to conduct the final examination of Aliyu Umar Sa'ad on his thesis entitled "Structural, Elastic and Optical Properties of Rice Husk Silicate Borotellurite Glass System Doped with Micro and Nanoparticles of Erbium Oxide" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

Members of the Thesis Examination Committee were as follows:

#### Chen Soo Kien, PhD

Associate Professor Faculty of Science Universiti Putra Malaysia (Chairman)

## Abdul Halim bin Shaari, PhD Professor Faculty of Science Universiti Putra Malaysia (Internal Examiner)

Zainal Abidin bin Talib, PhD Professor Faculty of Science Universiti Putra Malaysia (Internal Examiner)

#### Yasser Bakr Saddeek Muhammed, PhD

Professor Al Azhar University Egypt (External Examiner)

NOR AINI AB. SHUKOR, PhD Professor and Deputy Dean School of Graduate Studies Universiti Putra Malaysia

Date: 28 June 2018

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

## Halimah Mohamed Kamari, PhD

Associate Professor Faculty of Science Universiti Putra Malaysia (Chairperson)

### Chan Kar Tim, PhD

Senior Lecturer Faculty of Science Universiti Putra Malaysia (Member)

# Amirah Abdul Latif, PhD Senior Lecturer Faculty of Science Universiti Putra Malaysia

(Member)

**ROBIAH BINTI YUNUS, PhD** Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date

# **Declaration by graduate student**

I hereby confirm that:

- this is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from the supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature:

Date:

Name and Matric No.: Aliyu Umar Sa'ad (GS45200)

# **Declaration by Members of Supervisory Committee**

This is to confirm that:

- the research conducted, and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: Name of Chairman of Supervisory Committee:	Associate Professor Dr Halimah Mohamed Kamari
Signature: Name of Member of Supervisory Committee:	Dr. Chan Kar Tim
Signature: Name of Member of Supervisory Committee:	Dr. Amirah Abdul Latif

# TABLE OF CONTENTS

Al Al Al Al Dl Ll Ll Ll	BSTRACT BSTRAK CKNOWLEDGEMENTS PPROVAL ECLARATION IST OF TABLES IST OF FIGURES IST OF ABBREVIATIONS	i iii v vi vii xiii xvii xvi xxii
СН	(APTER	
1	INTRODUCTION1.1Background of Study1.2Problem Statement1.3Scope of Study1.4Aim and Objectives1.5Hypotheses1.6Significance of the Study1.7Thesis OrganizationLITERATURE REVIEW2.1Rice Husk	1 3 4 5 5 6 6 8 8 8
	<ul> <li>2.1 Rice Husk</li> <li>2.1.1 Rice Husk Ash (RHA)</li> <li>2.2 Glass</li> <li>2.2.1 Tellurite Glasses</li> <li>2.2.2 Silicate Glasses</li> <li>2.2.3 Borate Glasses</li> <li>2.2.4 Erbium Doped Glasses</li> <li>2.3 McCumber Theory/EDFA Gain</li> <li>2.4 Elastic Properties of Glasses</li> </ul>	9 11 12 14 15 16 16 16
3	THEORY	22
	<ul> <li>3.1 Structural Properties</li> <li>3.1.1 The X-ray Diffraction (XRD) Spectroscopy</li> <li>3.1.1.1 Bragg's Law</li> <li>3.1.2 Fourier Transform Infrared (FTIR) Spectroscopy</li> <li>3.1.3 Density and Molar Volume</li> <li>3.2 Optical Properties</li> <li>3.2.1 UV-Vis Spectroscopy</li> <li>3.2.1.1 Energy Band Gap</li> <li>3.2.1.2 Refractive Index, Molar Refractive Index ar Metallization Criterion</li> </ul>	22 22 22 23 24 25 25 26 ad 26

		3.2.1.3 Dielectric Constants, Linear Dielectric Susceptibility, Reflection loss and Transmission	27
		3 2 1 4 Polarizability	27
		3.2.1.5 Oxygen Packing Density Erbium Ion	20
		Concentration (N). Erbium Inter-Ionic Distance.	
		Polaron Radius $(R_p)$ and Optical Basicity	29
		3.2.2 Photoluminescence spectroscopy	30
		3.2.3 The McCumber Theory and Gain	31
	3.3	Non-Destructive Ultrasonic Analysis and Elastic Properties	33
		3.3.1 Theoretical Models	35
		3.3.1.1 Makishima and Mackenzie model	35
		3.3.1.2 Rocherulle Model	36
		3.3.1.3 Bond Compression and Ring Deformation	
		Models	36
4	MAT	TERIALS AND METHODS	39
	4.1	Introduction	39
	4.2	Materials	39
	4.3	Experiments	39
		4.3.1 Silicate Extraction	39
		4.3.2 Glass Fabrication	40
	4.4	Instruments, Measurements and Characterizations	42
		4.4.1 Measurement of Density	42
		4.4.2 Energy Dispersion X-Ray Fluorescence (ED-XRF)	42
		4.4.3 The X-Ray Diffraction (XRD)	42
		4.4.4 Fourier Transform Infrared (FTIR) Analysis	43
		4.4.5 Transmission Electron Microscopy (TEM)	43
		4.4.0 UV-VIS Spectroscopy	43 13
		4.4.8 Non-Destructive Ultrasonic Technique	43 44
		4.4.9 Error Analysis of Measurements	44
		4.4.9 Error Analysis of Weasurements	
5	RES	ULTS AND DISCUSSION	45
	5.1	XRF Oxide Analysis of Rice Husk Silica	45
	5.2	Rice Husk Silicate Borotellurite Glass System	46
		5.2.1 Structural Properties of RHSBT Glass System	46
		5.2.1.1 XRD Studies	46
		5.2.1.2 Fourier Transform Infrared Spectrum (FTIK)	17
		5.2.1.3 Density and Molar Volume	47
		5.2.2. Elastic Properties of the Rice Husk Silicate Borotellurite	<b>ر</b> ۲
		(RHSBT) Glass System	50
		5.2.3 Theoretical Elastic Models	55
		5.2.3.1 Makishima and Mackenzie Model	55
		5.2.3.2 Rocherulle Model	58
		5.2.3.3 Bond Compression Model and Ring Deformation	
		Model	60

		5.2.4	Optical Properties of RHSBT Glass System	63
	5.3	Erbiur	n Doped Rice Husk Silicate Borotellurite glass system	70
		5.3.1	Structural Properties of Er-Doped RHSBT Glasses	71
		5.3.2	Elastic Properties of Er-Doped RHSBT Glass System	74
		5.3.3	The Theoretical Elastic Models	80
			5.3.3.1 Makashima Model	80
			5.3.3.2 Rocherulle's Model	83
			5.3.3.3 Bond Compression Model and Ring Deformation	
			Model	85
		5.3.4	Optical Properties of Er-Doped RHSBT Glass System	88
		5.3.5	McCumber Theory and Simple Gain of the Er-Doped	
			RHSBT Glass System	97
	5.4	Erbiur	n Nanoparticles Doped Rice Husk Silicate Borotellurite	
		(Er-N	Ps Doped RHSBT Glass System	105
		5.4.1	Structural Properties of Er-NPs Doped RHSBT Glasses	105
		5.4.2	Elastic Properties of Er-NPs Doped RHSBT Glass	
			System	110
		5.4.3	Theoretical Elastic Models	114
			5.4.3.1 Makishima Model	114
			5.4.3.2 Rocherulle's Model	116
			5.4.3.3 Bond Compression Model and Ring Deformation	
			Model	117
		5.4.4	Optical Properties of Er-NPs Doped RHSBT Glass	
			System	120
		5.4.5	McCumber Theory and Simple Gain of the Er NPs-	
			Doped RHSBT Glass System	130
	5.5	Comp	arative Assessment of Er-Doped RHSBT and Er NPs-	
		Doped	RHSBT Glass Systems	138
		5.5.1	Physical and Structural Properties	138
		5.5.2	Elastic Properties	139
		5.5.3	Optical Properties	139
		5.5.4	Erbium Doped Fiber Amplifier (EDFA) Applicability of	
			the Er-Doped RHSBT and Er NPs-Doped RHSBT Glass	
			System	140
6	CONC	CLUSI	ON AND RECOMMENDATION	141
	6.1	Conclu	usion	141
	6.2	Recon	nmendations for Future Works	143
DEEE	DENC	FO		144
REFE	KENC.	ES E OTT		144
BIOD		T SIU	DEN I TIONS	100
L191	UF PU	DLICA		100

 $\bigcirc$ 

# LIST OF TABLES

Table		Page
2.1	Summary of some related literatures	20
5.1	XRF analysis result of extracted silica from rice husk	45
5.2	Assignment of IR Absorption Peaks of the FTIR spectra of the $[(TeO_2)_{0.7}$ (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	48
5.3	Relative Areas of the Deconvoluted FTIR Peaks Assigned to TeO <sub>4</sub> , TeO <sub>3</sub> , BO <sub>3</sub> and BO <sub>3</sub> /BO <sub>2</sub> O in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x}$ (SiO <sub>2</sub> ) <sub>x</sub> glass system	49
5.4	Ultrasonic Velocities (V <sub>L</sub> ,V <sub>S</sub> , V <sub>m</sub> ) and Elastic Moduli for $[(TeO_2)_{0.7}$ (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub>	54
5.5	Microhardness (H), Softening Temperature (T <sub>s</sub> ), Poisson's ratio ( $\sigma$ ), Debye Temperature ( $\Theta_D$ ), Molar Weight (M <sub>w</sub> ), and the Fractal Bond Connectivity (d) for [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	54
5.6	Packing Density, Dissociation Energy, Young Modulus, Bulk modulus, Shear Modulus and Poisson Ratio for [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	56
5.7	The Packing Factor, Young Modulus, Bulk Modulus, Shear Modulus and Poisson Ratio for [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system (El-Mallawany, 1998; Saddek, 2005; El-Moneim, 2001)	58
5.8	Oxides' Coordination numbers, packing densities ( $V_i$ and $C_i$ ), Dissociation Energies ( $G_i$ ), Bond Length (r), and Stretching Force constant (F)	59
5.9	Bond per unit volume number (n <sub>b</sub> ), Bulk modulus, Bulk modulus ratio (K <sub>bc</sub> /K <sub>e</sub> ), Atomic ring size ( $\ell$ ) and Stretching Force Constant (F) for for [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-X</sub> (SiO <sub>2</sub> ) <sub>X</sub> glass system	60
5.10	The Elastic Moduli and Poisson for $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-X} (SiO_2)_X$ glass system	62
5.11	Indirect Band Gap ( $E^{1/2}_{opt}$ ), Direct Band Gap ( $E^{2}_{opt}$ ), Urbach Energy ( $\Delta E$ ), Refractive Index (n) and Molar Refractive Index of [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	67
5.12	Metallization Criterion (M), Si Ion Concentration (N), Polaron Radius (R <sub>P</sub> ), Si Ion Inter-Ionic Distance (R <sub>i</sub> ) and Field Strength of Si Ion Yeild (F) for different SiO <sub>2</sub> molar fractions (X) in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x}$ (SiO <sub>2</sub> ) <sub>x</sub> glass system	68

5.13	Optical Basicity ( $\Lambda_{th}$ ), Molar Polarizability ( $\alpha_m$ ), Molar Electronic Polarizability ( $\alpha_{me}$ ), Refractive Index Based Oxide Ion Polarizability ( $\alpha_{O2-}(n)$ ), Energy Band Gap Based Oxide Ion Polarizability ( $\alpha_{O2-}(E)$ ), Dielectric Constant ( $\epsilon$ ), Optical Dielectric	70
5.14	Assignment of IR Absorption Peaks of the FTIR spectra of the $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y \text{ glass system}$	74
5.15	Relative Areas of the Peaks Deconvoluted for the Study of FTIR Spectra of the $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y \text{ glass system}$	74
5.16	Longitudinal Velocity (V <sub>L</sub> ), Shear Velocity (V <sub>S</sub> ), Mean Velocity (V <sub>m</sub> ), Longitudinal (L), Shear (G), Bulk (K) and Young (E) Moduli Values for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$ glass system	75
5.17	Microhardness (H), Softening Temperature (T <sub>S</sub> ), Poisson Ratio ( $\sigma$ ), Debye Temperature ( $\Theta_D$ ) and Fractal Bond Connectivity (d) Values for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	78
5.18	Packing Density (V <sub>t</sub> ), Dissociation Energy (G <sub>t</sub> ), Elastic Moduli (E <sub>m</sub> , K <sub>m</sub> and G <sub>m</sub> ) and Poisson Ratio ( $\sigma_m$ ) for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	83
5.19	: Packing Density (C <sub>t</sub> ), Elastic Moduli (E <sub>R</sub> , K <sub>R</sub> , and G <sub>R</sub> ), Poisson Ratio ( $\sigma_R$ ) for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	85
5.20	: Bond per unit volume number (n <sub>b</sub> ), Bulk modulus, Bulk modulus ratio $(K_{bc}/K_e)$ , Atomic ring size ( $\ell$ ) and Stretching Force Constant (F) for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$ glass system	85
5.21	Elastic Moduli and Poisson Ratio for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	87
5.22	Direct Band Gap ( $E^{2}_{opt}$ ), Indirect Band Gap ( $E^{1/2}_{opt}$ ), Urbach Energy ( $\Delta E$ ), Refractive Index (n) and the Metallization Criterion for varying concentrations of $Er_2O_3$ in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	89
5.23	Molar Volume of Boron Atoms (V <sub>B</sub> ), Average Boron-Boron Separation ( $<$ d <sub>B-B</sub> $>$ ), Molar Concentration of Er <sup>3+</sup> ions (N), Polaron Radius (R <sub>p</sub> ), Inter-Nuclear Distance of Er <sup>3+</sup> (R <sub>i</sub> ), Field Strength of Er <sup>3+</sup> Yields (F)	91
5.24	Optical Basicity ( $\Lambda_{th}$ ), Molar Polarizability ( $\alpha_m$ ), Refractive Index Based Oxide Ion Polarizability ( $\alpha_{O2-}$ (n)), Energy Band Gap Based Oxide Ion Polarizability ( $\alpha_{O2-}$ (E)), Dielectric Constants ( $\epsilon$ , $\epsilon_{opt}$ ) and Dielectric Susceptibility ( $\chi$ ) of {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	93
5.25	Assignment of IR Absorption Peaks of the FTIR spectra of the $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$	108

xiv

G

5.	.26	Relative Areas of the Peaks Deconvoluted for the Study of FTIR Spectra of the $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glasses$	109
5.	.27	Longitudinal Velocity ( $V_L$ , ms <sup>-1</sup> ), Shear Velocity ( $V_s$ , ms <sup>-1</sup> ), Mean Velocity ( $V_m$ , ms <sup>-1</sup> ), Longitudinal Modulus (L, GPa), Shear Modulus (G, GPa), Bulk Modulus (K, GPa) and Young Modulus (E, GPa) of Er <sub>2</sub> O <sub>3</sub> NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	111
5.	.28	Microhardness (H), Poisson Ratio ( $\sigma$ ), Softening Temperature (T <sub>s</sub> ), Fractal Bond Connectivity (d) and the Fugacity ( $f_g$ ) Values for Er <sub>2</sub> O <sub>3</sub> NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	112
5.	.29	Packing Density (V <sub>t</sub> ), Dissociation Energy (G <sub>t</sub> ), Elastic Moduli (E <sub>m</sub> , K <sub>m</sub> and G <sub>m</sub> ) and Poisson Ratio ( $\sigma_m$ ) for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	115
5.	.30	Packing Density (C <sub>t</sub> ), Elastic Moduli (E <sub>R</sub> , K <sub>R</sub> , and G <sub>R</sub> ), Poisson Ratio $(\sigma_R)$ for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	117
5.	.31	Bond per unit volume number $(n_b \ge 10^{28})$ , Bulk modulus, Bulk modulus ratio $(K_{bc}/K_e)$ , Atomic ring size ( $\ell$ ) and Stretching Force Constant (F) for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	118
5.	.32	Elastic Moduli and Poisson Ratio of $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	119
5.	.33	Density, Molar Volume, Molar Mass, Direct Optical Band Gap $(E^{2}_{opt})$ , Indirect Optical Band Gap $(E^{1/2}_{opt})$ and Urbach Energy and the Refractive Index of Er <sub>2</sub> O <sub>3</sub> NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	122
5.	.34	Molar Polarizability ( $\alpha_m$ ), Average Boron-Boron Separation ( $\langle d_{B-B} \rangle$ ), Metallization (M), Molar Concentration of Erbium Ions (N), Polaron Radius (RP), Inter-nuclear distance of $Er^{3+}$ ions (R <sub>i</sub> ) and Field Strength of $Er^{3+}$ ions yields (F) of {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	124
5.	.35	Optical Basicity ( $\Lambda_{th}$ ), Refractive Index Based Oxide Ion Polarizability ( $\alpha_{O2}$ -(n)), Band Gap Based Oxide Ion Polarizability ( $\alpha_{O2}$ -(E)), Electronic Polarizability ( $\alpha_e$ ), Dielectric Constant ( $\epsilon$ ), Optical Dielectric Constant ( $\epsilon_{opt}$ ) and Dielectric Susceptibility ( $\chi$ ) of {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	128

# LIST OF FIGURES

	Figure	e	Page
	4.1	Schematic Flow Chart of Silicate Extraction Process from the Rice Husk	40
	4.2	Schematic Flow Chat of the Glass Fabrication Process for the Preparation of Erbium Oxide and Erbium Oxide NPs Doped Rice Husk Silicate Borotellurite Glass Systems	41
	5.1	The XRD pattern of $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	46
	5.2	The FTIR spectra of the [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	47
	5.3	Density and molar volume variation with molar fraction of SiO <sub>2</sub> in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	50
	5.4	The Elastic Moduli L, G, K and E Variation against SiO <sub>2</sub> Concentration in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	51
5	5.5	The Microhardness and Fractal Bond Connectivity Variation with Silicate Concentration in the $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	52
	5.6	Plot of Poisson's Ratio Variation against the SiO <sub>2</sub> Molar Fraction in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	53
	5.7	Fugacity Variation with SiO <sub>2</sub> Concentration in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x}$ (SiO <sub>2</sub> ) <sub>x</sub> glass system	53
	5.8	Variation of Packing Density and Dissociation Energy with Molar Fraction of SiO <sub>2</sub> for [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	56
	5.9	Elastic Moduli for the $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	57
	5.10	Elastic Moduli for the [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	59
	5.11	Bond per Unit Volume and the Average Stretching Force Constant variation with the Molar Fraction of SiO <sub>2</sub> for $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-X}$ (SiO <sub>2</sub> ) <sub>X</sub> glass system	61
	5.12	Elastic Moduli Variation with the Molar Fraction of $SiO_2$ for the $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-X} (SiO_2)_X$ glass system	62
	5.13	Atomic Ring Size Variation with Molar Fraction of SiO <sub>2</sub> for $[(TeO_2)_{0.7}$ (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-X</sub> (SiO <sub>2</sub> ) <sub>X</sub> glass system	63
	5.14	The UV-Vis Absorption Spectra of $[(TeO_2)_{0.7}(B_2O_3)_{0.3}]_{1-x}(SiO_2)_x$ glass system	64

	5.15	Variation of Energy Band Gaps (Direct and Indirect) with SiO <sub>2</sub> Molar Fraction in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	65
	5.16	Variation of Refractive Index with SiO <sub>2</sub> Concentration in $[(TeO_2)_{0.7}$ (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	66
	5.17	Variation of Molar Refractive Index and OPD with SiO <sub>2</sub> Molar Fraction in $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{1-x} (SiO_2)_x$ glass system	68
	5.18	Transmission Coefficient (T) and Reflection Loss ( $R_L$ ) with SiO <sub>2</sub> Molar Fraction in [(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>1-x</sub> (SiO <sub>2</sub> ) <sub>x</sub> glass system	69
	5.19	The XRD Pattern of {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	71
	5.20	Density and Molar Volume Variation with $Er_2O_3$ Molar Fraction in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$ glass system	72
	5.21	The FTIR Spectra of $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y \text{ glass system}$	73
	5.22	Longitudinal, Shear, Bulk and Young Moduli Variation with $Er_2O_3$ Concentration in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	76
	5.23	Poisson Ratio Variation with $Er_2O_3$ Concentration in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	77
	5.24	Microhardness and Fractal Bond Connectivity Variation with $Er_2O_3$ Concentration in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	79
	5.25	Fugacity Value Variation with $Er_2O_3$ Concentration in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	80
	5.26	Packing Density and Dissociation Energy Variation with Er <sub>2</sub> O <sub>3</sub> Molar Fraction in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	81
	5.27	The Elastic Moduli Variation with Molar Fraction of $Er_2O_3$ in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y \text{ glass system} \}$	82
	5.28	Elastic Moduli Variation with Molar Fraction of $Er_2O_3$ for {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> ) <sub>y</sub> glass system	84
	5.29	Bond per Unit Volume and Stretching Force Constant Variation with Molar Fraction of $Er_2O_3$ for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ $(Er_2O_3)_y$ glass system	86
	5.30	Elastic Moduli Variation with Molar Fraction of $Er_2O_3$ for $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$ glass system	86

5.31 Atomic Ring Size Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> for {[(TeO<sub>2</sub>)<sub>0.7</sub> 87  $(B_2O_3)_{0.3}]_{0.8}$   $(SiO_2)_{0.2}\}_{1-y}$   $(Er_2O_3)_y$  glass system 5.32 UV-Vis Spectra of { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> glass system 88 5.33 Direct and Indirect Optical Band Gap Variation with Er<sub>2</sub>O<sub>3</sub> Concentration in  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$  glass system 90 5.34 Refractive Index and Oxygen Packing Density Variation with Er<sub>2</sub>O<sub>3</sub> Concentration in { $[(TeO_2)_{0,7} (B_2O_3)_{0,3}]_{0,8} (SiO_2)_{0,2}$ }<sub>1-v</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>v</sub> glass 91 system 5.35 Transmission Coefficient and Reflection Loss Variation with Er2O3 Concentration in { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> glass 92 system Optical Basicity Variation against the Metallization Criterion in 5.36 94  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y \text{ glass system}$ 5.37 Linear Electric Susceptibility Variation against the Electronic Polarizability in { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> glass 95 system 5.38 Variation of Molar Polarizabilities and Molar Electronic Polarizability with Molar Refractive Index in { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }\_1-y 95  $(Er_2O_3)_y$  glass system 5.39 Emission Spectra { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> glass 96 system ( $\lambda_{ext} = 350 \text{ nm}$ ) Emission transitions of  $Er^{3+}$  ions for the {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> 5.40  $(SiO_2)_{0.2}$   $_{1-y}$   $(Er_2O_3)_y$  glass system 97 5.41 Absorption Cross-Section of  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> Glass System 98 5.42Emission Cross-Section of  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3)_y$ **Glass System** 99 5.43 Gain Coefficient for 0.01  $\text{Er}_2\text{O}_3$  in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>v</sub> Glass System 100 5.44 Gain Coefficient for 0.02  $\text{Er}_2\text{O}_3$  in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> 101 (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> Glass System Gain Coefficient for 0.03 Er<sub>2</sub>O<sub>3</sub> in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> 5.45 101 (Er<sub>2</sub>O<sub>3</sub>)<sub>v</sub> Glass System 5.46 Gain Coefficient for 0.04  $\text{Er}_2\text{O}_3$  in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>v</sub> Glass System 102

5.47 Gain Coefficient for 0.05 Er<sub>2</sub>O<sub>3</sub> in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub>)<sub>v</sub> Glass System 102 5.48 The Overall Simple Gain (dB) against the Population Inversion Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1500 nm signal for Er 0.01 103 5.49 The Overall Simple Gain (dB) against the Population Inversion Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1530 nm signal for Er 0.01 104 The Overall Simple Gain (dB) against the Population Inversion 5.50 Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1550 nm signal for Er 0.01 104 5.51 XRD pattern for { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>y</sub> glass system 106 Density and Molar Volume Variation with Molar Fraction of Nano 5.52  $Er_2O_3$  in {[(TeO\_2)\_{0.7} (B\_2O\_3)\_{0.3}]\_{0.8} (SiO\_2)\_{0.2}}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>y</sub> Glass System 107 5.53 FTIR Spectra of { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>y</sub> Glass 108 System 5.54 TEM micrograph of { $[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}$ }<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>y</sub> Glass with 3% Er<sub>2</sub>O<sub>3</sub> NPs 109 5.55 Elastic Moduli Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> NPs in  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$ 110 5.56 Microhardness and Fractal Bond Connectivity Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> NPs in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>v</sub> Glass System 111 Poisson Ration Value Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> NPs in 5.57  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$ 112 Fugacity Value Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> NPs in 5.58 $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$ 113 Elastic Moduli Variation with Molar Fraction of Er2O3 NPs in 5.59  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$ 114 5.60 Packing Density and Dissociation Energy Variation with Molar Fraction of Er<sub>2</sub>O<sub>3</sub> NPs in {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>0.8</sub> (SiO<sub>2</sub>)<sub>0.2</sub>}<sub>1-y</sub> (Er<sub>2</sub>O<sub>3</sub> NPs)<sub>y</sub> Glass System 116 5.61 Elastic Moduli Variation with Molar Fraction of Er2O3 NPs in  $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$ 117

	5.62	Bond per Unit Volume and Average Stretching Force Constant Variation with Molar Fraction of $Er_2O_3$ NPs in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y}$ ( $Er_2O_3$ NPs) <sub>y</sub> Glass System	118
	5.63	Elastic Moduli Variation with the Molar Fraction of $Er_2O_3$ NPs in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$	119
	5.64	Atomic Ring Size Variation with Molar Fraction of $Er_2O_3$ NPs in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$	120
	5.65	UV-Vis Spectra of $Er_2O_3$ NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	121
	5.66	Direct and Indirect Optical Band Gaps Variation with Molar Fraction of $Er_2O_3$ NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	122
	5.67	Molar Refractive Index and Oxygen Parking Density (OPD) Variation with Molar Fraction of $Er_2O_3$ NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	123
5.0 5.0 5.2	5.68	Variation of Transmission Coefficient (T) and Reflection Loss with Molar Fraction of $Er_2O_3$ NPs in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	125
	5.69	Molar Polarizability and Molar Electronic Polarizability Variation with Molar Refractive Index in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	126
	5.70	Linear Dielectric Susceptibility Variation with Electronic Polarizability in {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System	127
	5.71	Optical Basicity Variation with Metallization Criterion in $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glass System$	129
	5.72	Luminescence Spectra of {[(TeO <sub>2</sub> ) <sub>0.7</sub> (B <sub>2</sub> O <sub>3</sub> ) <sub>0.3</sub> ] <sub>0.8</sub> (SiO <sub>2</sub> ) <sub>0.2</sub> } <sub>1-y</sub> (Er <sub>2</sub> O <sub>3</sub> NPs) <sub>y</sub> Glass System at $\lambda_{Ext.} = 350$ nm	129
	5.73	Absorption Cross-Section of $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glasses$	131
	5.74	Emission Cross-Section of $\{[(TeO_2)_{0.7} (B_2O_3)_{0.3}]_{0.8} (SiO_2)_{0.2}\}_{1-y} (Er_2O_3 NPs)_y Glasses$	131
	5.75	Simple Gain of Er <sub>2</sub> O <sub>3</sub> NPs 0.01 Sample	133
	5.76	Simple Gain of Er <sub>2</sub> O <sub>3</sub> NPs 0.02 Sample	133
	5.77	Simple Gain of Er <sub>2</sub> O <sub>3</sub> NPs 0.03 Sample	134
	5.78	Simple Gain of Er <sub>2</sub> O <sub>3</sub> NPs 0.04 Sample	134

- 5.79 Simple Gain spectrum of Er<sub>2</sub>O<sub>3</sub> NPs 0.05 sample
- 5.80 The Overall Simple Gain against the Population Inversion Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1500 nm signal for Er<sub>2</sub>O<sub>3</sub> NPs 0.03
- 5.81 The Overall Simple Gain against the Population Inversion Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1530 nm signal for Er<sub>2</sub>O<sub>3</sub> NPs 0.03
- 5.82 The Overall Simple Gain against the Population Inversion Coefficient for Fiber Lengths L=3, 5, 10 and 15 m for 1550 nm signal for Er<sub>2</sub>O<sub>3</sub> NPs 0.03



135

137

137

# LIST OF ABBREVIATIONS

	Symbols	Description	Units
	RHA	Rice Husk Ash	-
	RHS	Rice Husk Silicate	-
	XRD	X-ray diffraction	-
	FTIR	Fourier transform infrared	
	UV-Vis	Ultraviolet-Visible	-
	TEM	Transmission Electron Microscopy	-
	BO	Bridging Oxygen	-
	NBO	Non-Bridging Oxygen	-
	ρ	Density	g/cm <sup>3</sup>
	Vm	Molar Volume	cm <sup>3</sup>
	OPD	Oxygen Packing Density	cm <sup>-3</sup>
	$V_L$	Longitudinal Velocity	m/s
	Vs	Shear Velocity	m/s
	L	Longitudinal Modulus	GPa
	G	Shear Modulus	GPa
	Е	Young Modulus	GPa
	K	Bulk Modulus	GPa
	σ	Poisson Ratio	-
	$T_s$	Softening Temperature	Κ
	$ heta_D$	Debye Temperature	К
	$C_t/V_t$	Packing Density	-
	$G_t$	Dissociation Energy	kJ/cm <sup>3</sup>
	$N_c$	Bond Number per Unit Formula	-
	F	Average Stretching Force Constant	(Nm <sup>-1</sup> )
	R	Bond Length	(nm)

n <sub>b</sub>	Bond Number per Unit Volume	$(cm^{-3})$
α	Coefficient of Absorption	$(cm^{-1})$
$\lambda_c$	Cut-off Wavelength	(nm)
$E_{opt}$	Optical Energy Band Gap	(eV)
$\Delta E$	Urbach Energy	(eV)
п	Refractive Index	-
$R_m$	Molar Refractive Index	
$\alpha_m$	Molar Polarizability	(Å <sup>3</sup> )
$\alpha_e$	Electronic Polarizability	(Å <sup>3</sup> )
$\alpha_{0^{2-}}^{(n)}$	Refractive Index Based Oxide Ion Polarizabilit	ty (Å <sup>3</sup> )
$\alpha_{o^{2-}}^{(E)}$	Band Gap Based Oxide Ion Polarizability	(Á <sup>3</sup> )
X	Linear Dielectric Susceptibility	-
$\Lambda_{th}$	Optical Basicity	-
Ε	Dielectric Constant	-
Eopt	Optical Dielectric Constant	-
R <sub>i</sub>	Ion-Ion Separation	(Å)
$R_p$	Polaron Radius	(Å)
$\sigma_{ab}$	Absorption Cross-section	$(cm^2)$
σ <sub>em</sub>	Emission Cross-Section	$(cm^2)$
g	Simple Gain	-
G	Fiber Gain	dB
$\Delta\lambda_{e\!f\!f}$	Effective Band Width	nm

## **CHAPTER 1**

#### **INTRODUCTION**

## **1.1 Background of Study**

Over the last decades, tellurite glasses have been under comprehensive investigation due to their high refractive index, better transparency in mid-infrared (0.35-6 mm), thermal and chemical stability, rare earth (RE) ion solubility, relatively low-phonon energy (~700-800 cm<sup>-1</sup>), better glass stability, high dielectric constant and good corrosion resistance. Because of their large refractive index, good semiconducting properties, and low melting point, tellurite glasses as semi-conducting glasses are used for a wide range of applications when containing oxides of ferroelectric materials, transition metal oxides and rare earth oxides (Saddeek, 2005).

Among the oxide glasses, tellurite glasses have advantage over other oxide glasses due also to their good mechanical strength, chemical durability, malleability and low processing temperatures among others (Hesham et al., 2003). High hyper polarity of the lone electron pair in the 5s orbit of the tellurium atom makes the tellurite-based glasses excellent nonlinear optical glasses. Weak Te-O bond in the tellurite based glasses which are easily broken during glass formation gives the ability to accommodate rare earth metal ions and heavy metal oxides (Maheshvaran et al., 2011).

Tellurite glasses have found many applications in non-linear optical devices such as high speed optical- switches, up-conversion frequency systems and others like production of optical fibers and planar waveguides, high performance optics, laser technology and in optical data transmission, detection and sensing (Joshi et al., 2012). Because of their high third order nonlinear susceptibility, tellurium oxides glasses are considered excellent for optical amplifiers (Pavani et al., 2011). Their applications in fields of science, technology and even industrial are informed by their high linear and non-linear refractive indices as well as their physical, chemical and mechanical qualities (Linda et al., 2013).

Because tellurium oxide (TeO<sub>2</sub>) alone does not form glass, and is considered a conditional glass former, another oxide is required to form a tellurite based glasses (Gaafar et al., 2014). Previous researches have proven boron oxide ( $B_2O_3$ ), another glass former to be the best candidate for combination with tellurium oxide. Normally, borate glasses are considered in glass science and technology because of their high transparency, relatively high thermal stability, good rare earth (RE) ions solubility, low melting point and variable coordination numbers (Kesavulu et al., 2016).

Borate glasses in the face of science and technology has found special attention due to their application in electrochemical and optics. These applications include fabrication of solid-state lasers, optical waveguide, Bio-materials and optical luminescent materials (Janek et al., 2016). In borate glasses, borate matrix BO<sub>3</sub> triangles and BO<sub>4</sub> tetrahedra forms, diborate, triborate, tetraborate and so on, which are stable borate groups (Maheshvaran et al., 2013).

To improve the glass quality in terms of thermal and chemical stability, hardness, glass solidification and also the forming ability, boron oxide  $(B_2O_3)$  is used in combination with tellurium oxide  $(TeO_2)$  to form what is called borotellurite glass (Hajer et al., 2014). The incorporation of tellurium oxide with boron oxide in the glass network also reduces the hygroscopic nature of the glass (Azlan et al., 2017). Due to their advantage in terms of optical and electrical properties, borotellurite glasses have been excellent in microelectronics and opto-acoustics applications (Gaafar et al., 2014).

Because of their low stability in terms of chemical durability pure borotellurite absorbs atmospheric water vapors to form crystalline precipitates of  $TeO_2$  in the glass matrix. This informs the need for glass modification to correct or reduce such effect (Azlan et al., 2017). Although they are limited by their low solubility of rare earth ions, low infrared transparency and high phonon energy, silicon oxide glasses possess strong mechanical strength, high laser threshold and good thermal stability. Hence, silicon oxide is a good choice for borotellurite combination for use in optical applications and high gain photonic crystal fibers (PCFs) (Tingting et al., 2016).

Rare earth oxides are used sometimes in high concentrations (depending on the need and application) in oxide glasses for wide range of applications from optical data transmission, detection, sensing to laser technologies and many others (El-Mallawany, 2000). The choice of the RE ions depends on the relationship between the radiative or non-radiative RE ions and the host glass composition (Ashur et al., 2014). Hence, the phonon energy must be low to attend high efficiency in terms of luminescence and the tunability between the host and the dopant ions is necessary. Of the rare earth metal oxide ions, erbium ions were found to be excellent for doping in borotellurite glasses due to their high solubility in the network and unique properties they possess (Azlan et al., 2017).

 $\bigcirc$ 

Through their transition from  ${}^{4}S_{3/2}$  to  ${}^{4}I_{15/2}$  at around 18300 cm<sup>-1</sup> and subsequent emission due to the transition, Er<sup>3+</sup> ions are preferable for use in green up-conversion lasers. Er<sup>3+</sup> ions, when excited with 980 nm signal exhibits in the NIR region of 1400-1600 nm an optically transparent window. This window possesses high signal to noise ratio, high detector sensitivity, low auto-fluorescence background, as well as high penetration depth in biological tissues (Annapoorani et al., 2016). Due to their potentials for applications in the areas of microchip lasers, infrared lasers, medical eye-safe lasers and erbium doped fiber amplifiers (EDFA) in the communication technology. EDFA used in the communication technology (Swapna et al., 2015). Nippon Telephone and Telegraph (NTT) produced a tellurite-based  $Er^{3+}$  doped fiber amplifiers (EDFA) that showed a gain-flattened amplification of 20 dB across 1530 to 1610 nm wavelength range. Because of the quality in terms of efficiency, tellurite glasses doped with  $Er^{3+}$  ions are used in many radiative and non-radiative applications as optical fiber amplifiers, optical devices, up-conversion lasers, nonlinear and many others (Sayed et al., 2016).

This work presents the utilization of rice husk (waste from rice milling) to obtain high purity silicon oxide for use in fabrication of silicate borotellurite glasses. The optical, structural and elastic properties of the glasses where studied to obtain the best amount of  $SiO_2$  for borotellurite glasses. The best proportion of  $SiO_2$  was used to fabricate erbium and erbium nano-particles doped silicate borotellurite glasses. The optical, structural and elastic properties of the glasses were studied to ascertain the applicability of the glasses from the studied properties.

## 1.2 Problem Statement

Scientists and technologist have been engaged over the years in the study of utilization of tellurite glasses in various areas of science, technology and industry. Since  $TeO_2$  alone does not form glass, it must be combined with one or more other oxide to form glass (El-Mallawany, 2002). Structural, optical, electrical and elastic properties of tellurium oxide-based glasses have been and are still under study.

The work of Halimah *et al.* (2005) studied the ultrasonic and physical properties of borotellurite glass. From the work, study was made of the proportion of tellurium oxide and boron oxide combination in terms physical and mechanical properties for the various scientific, technological and industrial applications of the glasses (Halimah et al., 2005).

For their properties that made them promising materials in various applications for photonics and opto-electronics, tellurite based glasses has been given special research attention (Bahari Poor et al., 2013). Applications particularly in fabrication of both active and passive optical fibers as well as high-gain optical amplifiers used in the communication technology has made  $TeO_2$  based glasses of high research interest (Chillcce et al., 2011).

In the report of Damas *et al.*, (2013), tellurium oxide as a host material in glasses have the lowest phonon energy of around 780 cm<sup>-1</sup> with high transparency in the visible, near and mid infrared spectral region. Having refractive indices that are larger than those of quartz, phosphate glasses or borosilicate glasses, they appeared to be better in terms of losses due scattering and absorption and possesses higher chemical durability (Damas et al., 2012).

The combination of  $B_2O_3$  with  $TeO_2$  in the borotellurite glass represents wonderful compromise between the requirements of the two oxides with each complimenting the other. Thus making it easy to achieve a better quality of low phonon energy and a relatively high thermal stability, high chemical durability and ease of fabrication that none of the two has combined (Maheshvaran et al., 2013). Because of their near white luminescence characteristics,  $Dy^{3+}$  ion doping on tellurite or borotellurite glass medium has been under study nowadays. They are considered very promising as laser active material that emits radiation at around ~3µm and in radiation dosimetry, they are used in identifying new materials on glass (Ab et al., 2014).

Different researchers studied different effects of  $Er^{3+}$  ions doping on many different borotellurite glass mediums. Most recent works include the work that study the doping effect of erbium oxide NPs on the green emission, Linear and non-linear optical properties, polarizability and optical basicity and spectral intensity of zinc borotellurite glass (Azlan et al., 2013; Azlan et al., 2014; Azlan et al., 2015; Azlan et al., 2017). The erbium doping effect on optical, elastic and DTA of borotellurite glasses containing lead and zinc were also studied (Sayed et al., 2016).

There was no report found in the literature reporting a study of the optical, structural and elastic properties of erbium or erbium NPs doped rice husk silicate borotellurite glasses. Hence in this work, silicate is extracted from rice husk and utilized to prepare erbium and erbium NPs doped silicate borotellurite glasses for the study of the optical, structural and elastic properties.

## 1.3 Scope of Study

The study includes the use of rice milling waste called "rice husk" to extract high purity silica (SiO<sub>2</sub>). The percentage purity of the extracted silica was determined using the XRF characterization. The silica extracted was used to fabricate silicate borotellurite glasses using the chemical formula [  $(TeO_2)_{0.7}$  (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>1-X</sub> (SiO<sub>2</sub>)<sub>X</sub> with x= 0.0-0.4. The silicate boro tellurite glasses were subjected XRD, FTIR, UV-Vis and Ultrasonic characterization for optical, structural and ultrasonic study to obtain the best SiO<sub>2</sub> proportion for optical glass applications. The best amount of SiO<sub>2</sub> was used to fabricate two series of erbium doped silicate borotellurite and erbium NPs doped silicate borotellurite glasses using the chemical formula {[(TeO<sub>2</sub>)<sub>0.7</sub> (B<sub>2</sub>O<sub>3</sub>)<sub>0.3</sub>]<sub>1-X</sub> (SiO<sub>2</sub>)<sub>X</sub>  $_{1-y}$  (Er<sub>2</sub>O<sub>3</sub>)<sub>y</sub> with y=0.01-0.05 for both the series. The two series were also subjected to XRD, FTIR, UV-Vis characterizations and Photoluminescence spectroscopy to study the erbium or erbium NPs doping effects on the optical, structural and ultrasonic properties of the silicate borotellurite glasses.

## **1.4** Aim and Objectives

The aim of the research is to study the optical, structural and elastic properties of micro and nanoparticle erbium oxide doped rice husk silicate borotellurite glasses. The objectives through which the aim will be achieved are as follows;

- 1. To extract high purity silicate from rice husk and fabricate silicate borotellurite glass.
- 2. To study the optical, structural and elastic properties of silicate (RHA) borotellurite glasses.
- 3. To fabricate micro/nanoparticle erbium oxide doped glasses using the best Silicate boro-tellurite proportion obtained and study the structural, optical and elastic properties of the two glass systems.
- 4. To determine the applicability of the two erbium oxide doped series in the EDFA technology using the McCumber theory.

## 1.5 Hypotheses

High purity SiO<sub>2</sub> can be extracted from the rice husk using the simple cold acid leaching method (Abu, Yahya, & Neon, 2016; Sharifnasab & Alamooti, 2017).

The pattern of glass XRD exhibits no sharp peaks but a broad hump indicating an amorphous nature of the material (Tagiara et al., 2017)

The FTIR spectra for silicate borotellurite glass system shows structural units such as SiO<sub>4</sub>, TeO<sub>4</sub>, TeO<sub>3</sub>, BO<sub>4</sub>, BO<sub>3</sub> units.

The use of silicate in a borotellurite glass system improves the glass mechanical quality and thermal resistance (Oana et al., 2017).

 $Er_2O_3$  and  $Er_2O_3$  NPs doping on silicate borotellurite glass system might increase the elastic moduli of the glass by increasing the rigidity of the glasses through increase in network connectivity (Yousef, 2013).

The Optical band gap of the  $Er_2O_3$  and  $Er_2O_3$  NPs doped silicate borotellurite glasses might increase with increase in the  $Er_2O_3$  content. This is due to the possible changes in the glass structure leading to the formation of bridging oxygens (BOs)(Yousef, 2013). The glasses are expected to have high refractive index as the presence of  $TeO_2$ enhances refractive index.

The  $Er_2O_3$  and  $Er_2O_3$  NPs doped glasses are expected to have good signal power amplification ability for fiber optics application. This is expected as the use of rice



husk ash and borotellurite is expected to improve both optical and mechanical quality of the glasses.

## **1.6** Significance of the Study

The study was carried out to study the optical, structural and the elastic properties of erbium and erbium doped silicate borotellurite glasses. With over 120 tonnes of rice husk produced globally every year without proper management, the world is risk of environmental pollution and land damages (Mohanta et al., 2012). Various scientific and technological approaches are being employed by scientist to manage and recycle the agricultural waste and control the possible environmental hazard that may be caused. These includes the use of the rice husk in cement production (Khan et al., 2014), as an alternative fuel in power generation plants, as a raw material for production of silica (SiO<sub>2</sub>) and silicon compounds in chemical industries (Kumar et al., 2013) and in the production of activated carbon (Abu et al., 2016). The utilization will there for bring about ease in disposal and save cost (Nagrale et al., 2012). Erbium doped glasses has been under study due their variety of applications in optical data transmission and optical fiber amplifies. The work attempts to determine the possibility of utilization of rice husk in the fabrication of glasses for optical glass technology. SiO<sub>2</sub> based glasses are known to have chemical durability, thermal stability, high viscosity and optical transparency at excitation and lasing wavelength (Berwal et al., 2017). Hence, the rice husk silicate (SiO<sub>2</sub>) was employed to help in managing the rice milling waste and to improve the physical, mechanical, optical and life span of the erbium doped borotellurite glasses used in the fabrication optical fiber and optical fiber amplifiers (Sayed et al., 2016).

To improve the glass quality in terms of thermal and chemical stability, hardness, glass solidification and also the forming ability, boron oxide  $(B_2O_3)$  is used in combination with tellurium oxide  $(TeO_2)$  to form what is called borotellurite glass (Hajer et al., 2014). The incorporation of tellurium oxide with boron oxide in the glass network also reduces the hygroscopic nature of the glass (Azlan et al., 2017). Due to their advantage in terms of optical and electrical properties, borotellurite glasses have been excellent in microelectronics and opto-acoustics applications (Gaafar et al., 2014).

## 1.7 Thesis Organization

The thesis consists of six chapters, with each chapter describing the sequence of study as described as follows;

**Chapter 1** shows the introduction to tellurite, borotellurite, silicate and erbium doped glasses with their applications in the background. It contains the problem statement, scope of the study, aim and objective, significance of study and thesis organization. **Chapter 2** shows an over view of related literatures on rice husk, glass, tellurite

glasses, borate glasses, silicate glasses, borotellurite glasses, doping and erbium doped glasses and applications. Chapter 3 describes the principles and theories employed in the study. The theories applied in some of the characterizations as XRF, XRD, FTIR, UV-Vis, Photoluminescence and the Ultrasonic non-destructive characterizations. The chapter also shows the theory used in density measurement. Also discussed here is the concept of glass free volumes, molar crystalline volume, molar crystalline volume based free volume, molar ionic volume, molar ionic volume based free space/ glass cubic interstitial space and the glass degree of compactness. Chapter 4 describes the materials, methods and the experimental techniques employed in the study. This includes the rice husk ash (silica) extraction and XRF characterization, Silicate borotellurite glass fabrication, density measurements and characterizations and Erbium/Erbium NPs dope silicate borotellurite glass fabrication, density measurements and characterizations. Also mentioned are the equipment and chemicals used in the process of the study. Chapter 5 presents and discusses the results from the XRF analysis result, through the optical, structural and elastic results of the various characterizations, measurements and calculation carried out on the three series of glass under study. Chapter 6 presents the overall conclusion based on the obtained results and gives recommendations on some future study in the field.

#### REFERENCES

- Ab, A., Wagiran, H., Hashim, S., Hussin, R., & Ibrahim, Z. (2014). Optical Properties of Undoped and Dy 3 + -doped Boro-Tellurite Glass. *Advanced Materials Research*, 895(4028), 194–199. https://doi.org/10.4028/www.scientific.net/AMR.895.194
- Abd, G., Saddeek, Y. B., Mohamed, G. Y., Mostafa, A. M. A., & Hassan, H. S. (2017). Nuclear Instruments and Methods in Physics Research B Effect of cement kiln dust and gamma irradiation on the ultrasonic parameters of HMO borate glasses. *Nuclear Inst. and Methods in Physics Research, B*, 394, 44–49. https://doi.org/10.1016/j.nimb.2016.12.041
- Abu, R., Yahya, R., & Neon, S. (2016). Production of High Purity Amorphous Silica from Rice Husk. *Procedia Chemistry*, 19, 189–195. https://doi.org/10.1016/j.proche.2016.03.092
- Agazzi, L. (2012). Spectroscopic excitation and quenching processes in rare-earthion-doped Al203 and their impact on amplifier and laser performance. https://doi.org/10.3990/1.9789036534239
- Alazoumi, S. H., Sidek, H. A. A., Halimah, M. K., Matori, K. A., Zaid, M. H. M., & Abdulbaset, A. A. (2017). Synthesis and elastic properties of ternary ZnO-PbO-TeO2 glasses. *Chalcogenide Letters*, 14(8), 303–320.
- Annapoorani, K., Basavapoornima, C., Murthy, N. S., Marimuthu, K., Suriya Murthy, N., & Marimuthu, K. (2016). Investigations on structural and luminescence behavior of Er3 + doped Lithium Zinc borate glasses for lasers and optical amplifier applications. *Journal of Non-Crystalline Solids*, 447, 273–282. https://doi.org/10.1016/j.jnoncrysol.2016.06.021
- Annapoorani, K., Maheshvaran, K., Arunkumar, S., Murthy, N. S., & Marimuthu, K. (2015). Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy Structural and luminescence behavior of Er 3 + ions doped Barium tellurofluoroborate glasses. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 135, 1090–1098. https://doi.org/10.1016/j.saa.2014.08.003
- Annapoorani, K., Murthy, N. S., Ravindran, T. R., & Marimuthu, K. (2016). Influence of Er 3+ ion concentration on spectroscopic properties and luminescence behavior in Er 3+ doped Strontium telluroborate glasses. *Journal of Luminescence*, 171, 19–26. https://doi.org/10.1016/j.jlumin.2015.10.071
- Ashiedu, F. I. and Akpan, E. E. (2015). Effects of Mixed Alkaline Earth Oxides in Potash Silicate Glass. *Nigerian Journal of Technology*, *34*(4), 731–736.
- Ashur, Z., Mahraz, S., Sahar, M. R., & Ghoshal, S. K. (2014). Band gap and polarizability of boro-tellurite glass: Influence of erbium ions. *Journal of Molecular Structure Journal*, 1072, 238–241.

https://doi.org/10.1016/j.molstruc.2014.05.017

- Assadi, A. A., Herrmann, A., Lachheb, R., Damak, K., Rüssel, C., & Maâlej, R. (2016). Experimental and theoretical spectroscopic study of erbium doped aluminosilicate glasses. *Journal of Luminescence*, 176, 212–219. https://doi.org/10.1016/j.jlumin.2016.02.022
- Awshah, A. A. A., Kamari, H. M., Tim, C. K., Shah, N. M., Alazoumi, S. H., Umar, S. A., & Noorazlan, M. A. (2017). Effect of Neodymium Nanoparticles on Elastic Properties of Zinc-Tellurite Glass System. *Advances in Materials Science and Engineering*, 2017, 7.
- Azlan, M. N., Halimah, M. K., Shafinas, S. Z., & Daud, W. M. (2013). Effect of Erbium Nanoparticles on Optical Properties of Zinc Borotellurite Glass System. *Journal of Nanomaterials* ·, 940917(November), 8.
- Azlan, M. N., Halimah, M. K., Shafinas, S. Z., & Daud, W. M. (2014). POLARIZABILITY AND OPTICAL BASICITY OF Er 3 + IONS DOPED TELLURITE BASED GLASSES. *Chalcogenide Letters*, 11(7), 319–335.
- Azlan, M. N., Halimah, M. K., & Sidek, H. A. A. (2017). Linear and nonlinear optical properties of erbium doped zinc borotellurite glass system. *Journal of Luminescence*, 181, 400–406. https://doi.org/10.1016/j.jlumin.2016.09.047
- Azlan, M. N., Halimah, M. K., Zulkefly, S. S., & Mohamad, D. W. (2013). Effect of Erbium Nanoparticles on Optical Properties of Zinc Borotellurite Glass System. *Journal of Nanomaterials*, 8(2), 49–59. https://doi.org/10.4028/www.scientific.net/MSF.846.63
- Azuraida, A., Halimah, M. K., Sidek, A. A., Azurahanim, C. A. C., Iskandar, S. M., Ishak, M., & Nurazlin, A. (2015). Comparative Studies Of Bismuth And Barium Boro-Tellurite Glass System: Structural And Optical Properties. *Chalcogenide Letters*, 12(10), 497–503.
- Bahari Poor, H.-R., Sidek, H. A. A., & Zamiri, R. (2013). Ultrasonic and optical properties and emission of Er3+/Yb3+ doped lead bismuth-germanate glass affected by Bi+/Bi2+ ions. *Journal of Luminescence*, *143*, 526–533. https://doi.org/10.1016/j.jlumin.2013.05.053
- Balaji, S., Allu, A. R., Biswas, K., Gupta, G., Ghosh, D., & Annapurna, K. (2017). Bandwidth enhancement of MIR emission in Yb3+/Er3+/Dy3+ triply doped fluoro-tellurite glass. *Laser Physics Letters*, 14(3). https://doi.org/10.1088/1612-202X/aa5c1b
- Balda, R., Miguel, A., Morea, R., & Gonzalo, J. (2016). Luminescence Properties of Er 3 + Ions in Nanocrystalline. *IEEE Journal Of Quantum Electronics*, 2–5.
- Basavapoornima, C., Linganna, K., Kesavulu, C. R., Ju, S., Kim, B. H., Han, W.-T., & Jayasankar, C. K. (2017). Spectroscopic and pump power dependent upconversion studies of Er 3 b -doped lead phosphate glasses for photonic applications. *Journal of Alloys and Compounds*, 699, 959–968.

https://doi.org/10.1016/j.jallcom.2016.12.199

- Battisha, I., & Nahrawy, A. El. (2012). Physical Properties of Nano-Composite Silica-Phosphate Thin Film Prepared by Sol Gel Technique. *New Journal of Glass and Ceramics*, 2, 17–22. https://doi.org/10.4236/njgc.2012.21004
- Benmadani, Y., Kermaoui, A., Chalal, M., Khemici, W., Kellou, A., & Pellé, F. (2013). Erbium doped tellurite glasses with improved thermal properties as promising candidates for laser action and amplification. *Optical Materials*, 35(12), 2234–2240. https://doi.org/10.1016/j.optmat.2013.06.014
- Bernard, V. (2001). Absorption of UV visible light. In *Molecular Fluorescence: Principles and Applications* (Vol. 8, pp. 20–33). Wiley-VCH Verlag GmbH.
- Berwal, N., Dhankhar, S., Sharma, P., Kundu, R. S., Punia, R., & Kishore, N. (2017). Physical, structural and optical characterization of silicate modi fi ed bismuthborate-tellurite glasses. *Journal of Molecular Structure*, *1127*, 636–644. https://doi.org/10.1016/j.molstruc.2016.08.033
- Bhatia, B., Meena, S. L., Parihar, V., & Poonia, M. (2015). Optical Basicity and Polarizability of Nd 3 + -Doped Bismuth Borate Glasses. *New Journal of Glass and Ceramics*, (July), 44–52.
- Bilir, G., Kaya, A., Cinkaya, H., & Eryürek, G. (2016). Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy Spectroscopic investigation of zinc tellurite glasses doped with Yb 3 + and Er 3 + ions. Spectrochimica Acta Part A:Molecular and Biomolecular Spectroscopy, 165, 183–190. https://doi.org/10.1016/j.saa.2016.04.042
- Bordes-Richard, E., & Courtine, P. (2006). Optical basicity: a scale of acidity/basicity of solids and its application to oxidation catalysis. Optical basicity: a scale of acidity/basicity of solids and its application to oxidation catalysis. *Chemical Industries*, *108*, 319–352.
- Bounakhla, M., & Tahri, M. (2014). X-Ray Fluorescence Analytical Techniques. Rabat, Morocco.
- Bourhis, K., Massera, J., Petit, L., Koponen, J., Fargues, A., Cardinal, T., ... Ferraris, M. (2015). Erbium-doped borosilicate glasses containing various amounts of P<inf>2</inf>O<inf>5</inf> and Al<inf>2</inf>O<inf>3</inf>: Influence of the silica content on the structure and thermal, physical, optical and luminescence properties. *Materials Research Bulletin*, 70, 47–54. https://doi.org/10.1016/j.materresbull.2015.04.017

Byrnes, S. J. F. (2008). Basic theory and phenomenology of polarons. Berkeley.

Cabello, F., Sanchez-cortes, S., Jiménez de Castro, M., & Castro, M. J. De. (2016). Influence of the preparation conditions of erbium-doped bismuth germanate glasses on its optical response. *Journal of Non-Crystalline Solids*, 445–446, 110–115. https://doi.org/10.1016/j.jnoncrysol.2016.05.029

- Cai, M., Wei, T., Zhou, B., Tian, Y., Zhou, J., Xu, S., & Zhang, J. (2015). Analysis of energy transfer process based emission spectra of erbium doped germanate glasses for mid-infrared laser materials. *Journal of Alloys and Compounds*, 626, 165–172. https://doi.org/10.1016/j.jallcom.2014.11.077
- Çelikbilek, M., Ersundu, A. E., & Ayd, S. (2012). Crystallization Kinetics of Amorphous Materials. In *Advances in Crystallization Processes* (p. 648). Rijeka, Croatia: InTech. https://doi.org/10.5772/35347
- Chillcce, E. F., Mazali, I. O., Alves, O. L., & Barbosa, L. C. (2011). Optical and physical properties of Er 3+ -doped oxy-fluoride tellurite glasses. *Optical Materials*, *33*, 389–396. https://doi.org/10.1016/j.optmat.2010.09.027
- Chimalawong, P., Kaewkhao, J., Kittiauchawal, T., Kedkaew, C., & Limsuwan, P. (2010). Optical Properties of the SiO 2 -Na 2 O-CaO-Nd 2 O 3 Glasses. *American Journal of Applied Sciences* 7, 7(4), 584–589.
- Chinnamat, W., Laopaiboon, R., Laopaiboon, J., Pencharee, S., & Bootjomchai, C. (2017). Influence of ionic radius modifying oxides on the elastic properties of glasses using ultrasonic techniques and FTIR spectroscopy. *Physics and Chemistry of Glasses: European Journal of Glass Science and Technology Part B*, 58(5), 207–216. https://doi.org/10.13036/17533562.58.5.101
- Christian, G. (1996). Principles of Spectroscopy. In *Analytical Chemistry* (6th ed., p. 104). wiley.
- Damas, P., Coelho, J., Hungerford, G., & Hussain, N. S. (2012). Structural studies of lithium boro tellurite glasses doped with praseodymium and samarium oxides. *Materials Research Bulletin*, 47, 3489–3494. https://doi.org/10.1016/j.materresbull.2012.06.071
- Deepa, A. V, Priya, M., & Suresh, S. (2016). Influence of Samarium Oxide ions on structural and optical properties of borate glasses. *Scientific Research and Essays*, 11(5), 57–63. https://doi.org/10.5897/SRE2015.6359
- Della, V. P., Kuthn, I., & Hotza, D. (2002). Rice husk ash as an alternate source for active silica production. *Materials Letters*, 57(December 2002), 818–821. https://doi.org/10.1016/S0167-577X(02)00879-0
- Devreese, J. T. (2000). Polarons. *Encyclopedia of Applied Physics*, 383–409. Retrieved from http://arxiv.org/abs/cond-mat/0004497
- Dhara, A., Mishra, R. K., Shukla, R., Valsala, T. P., Sudarsan, V., Tyagi, A. K., & Kaushik, C. P. (2016). A comparative study on the structural aspects of sodium borosilicate glasses and barium borosilicate glasses : Effect of Al 2 O 3 addition. *Journal of Non-Crystalline Solids*, 447, 283–289. https://doi.org/10.1016/j.jnoncrysol.2016.04.040
- Dharma, J., & Pisal, A. (2012). Simple Method of Measuring the Band Gap Energy Value of TiO 2 in the Powder Form using a UV / Vis / NIR Spectrometer. PerkinElmer, Inc. Shelton, CT USA.

- Dias, J. D. M., Melo, G. H. A., Lodi, T. A., Carvalho, J. O., Filho, P. F. F., & Barboza, M. J. (2016). Thermal and structural properties of Nd 2 O 3 -doped calcium boroaluminate glasses. *Journal of Rare Earths*, 34(5), 521–528. https://doi.org/10.1016/S1002-0721(16)60057-1
- Dimitrov, V., & Komatsu, T. (2010). AN INTERPRETATION OF OPTICAL PROPERTIES OF OXIDES AND OXIDE GLASSES IN TERMS OF THE ELECTRONIC ION POLARIZABILITY AND AVERAGE SINGLE BOND STRENGTH (REVIEW). Journal of the University of Chemical Technology and Metallurgy, 45(3), 219–250.
- Dimitrov, V., Komatsu, T., & Sato, R. (1999). Optical Basicity and O1s Binding Energy of Simple Oxides. *Journal of the Ceramic Society of Japan*, 107(1), 21–26.
- Dimitrov, V., & Takayuki, K. (2005). Classification of oxide glasses : A polarizability approach. *Journal of Solid State Chemistry*, 178, 831–846. https://doi.org/10.1016/j.jssc.2004.12.013
- Doweidar, H., El-Damrawi, G. M., Moustafa, Y. M., & Ramadan, R. M. (2005). Density of mixed alkali borate glasses: A structural analysis. *Physica B*, 362, 123–132. https://doi.org/10.1016/j.physb.2005.02.001
- Doweidar, H., & Saddeek, Y. B. (2009). FTIR and ultrasonic investigations on modified bismuth borate glasses. *Journal of Non-Crystalline Solids*, 355(2009), 348–354. https://doi.org/10.1016/j.jnoncrysol.2008.12.008
- Dresselhaus, M. S. (1966). Solid State Physics Part II: Optical Properties of Solids. *Proceedins of the International School of Physics*, 198. Retrieved from http://web.mit.edu/course/6/6.732/www/6.732-pt2.pdf
- Eevon, C., Halimah, M. K., Zakaria, A., Azurahanim, C. A. C., Azlan, M. N., & Faznny, M. F. (2016). Linear and nonlinear optical properties of Gd 3 + doped zinc borotellurite glasses for all-optical switching applications. *Results in Physics*, 6, 761–766. https://doi.org/10.1016/j.rinp.2016.10.010
- Ehrt, D. (2009). Photoluminescence in Glass and Ceramics. In *IOP Conference Series: Material Science and Engineering* (Vol. 2).
- El-Mallawany, R. (1998). Tellurite glasses Part 1. Elastic properties. *Materials Chemistry and Physics*, 53, 93–120.
- El-Mallawany, R. (2000). Structural Interpretations on tellurite glasses. *Materials Chemistry and Physics*, 63, 109–115.
- El-Mallawany, R. (2005). An Introduction to Tellurite Glasses Module 5 Optical Properties.
- El-Mallawany, R. A. H. (2002). *Tellurite Glasses Handbook: Physical Properties and Data*. Florida: CRC Press LLC, 2000 N.W. Corporate Blvd., Boca Raton, Florida 33431.

- El-Mallawany, R. A. H. (2002). Tellurite Glasses Handbook: Physical Properties and Data. Florida: CRC Press LLC, 2000 N.W. Corporate Blvd., Boca Raton, Florida 33431.
- El-Mallawany, R., Afifi, H. A., El-Gazery, M., & Ali, A. A. (2018). Effect of Bi2O3addition on the ultrasonic properties of pentaternary borate glasses. *Measurement: Journal of the International Measurement Confederation*, 116(November 2017), 314–317. https://doi.org/10.1016/j.measurement.2017.11.028
- El-mallawany, R., Gaafar, M. S., Abdeen, M. A. M., & Marzouk, S. Y. (2014).
   Simulation of acoustic properties of some tellurite glasses. *Ceramics International*, 40(5), 7389–7394.
   https://doi.org/10.1016/j.ceramint.2013.12.084
- El-moneim, A. A. (2016). Correlation between acoustical and structural properties of glasses: Extension of Abd El-Moneim model for bioactive silica based glasses. *Materials Chemistry and Physics*, 173, 372–378. https://doi.org/10.1016/j.matchemphys.2016.02.025
- El-Moneim, A. A. (2001). Bond compression bulk modulus and Poisson's ratio of the polycomponent silicate glasses. *Materials Chemistry and Physics*, 70(3), 340–343. https://doi.org/10.1016/S0254-0584(00)00519-8
- El-Moneim, A. A. (2012). Correlation between acoustical and compositional parameters of borate and tellurite glasses. *Materials Chemistry and Physics*, 135(2–3), 653–657. https://doi.org/10.1016/j.matchemphys.2012.05.040
- El-Moneim, A. A. (2016). Correlation between acoustical and structural properties of glasses: Extension of Abd El-Moneim model for bioactive silica based glasses. *Materials Chemistry and Physics*, 173, 372–378. https://doi.org/10.1016/j.matchemphys.2016.02.025
- El-Moneim, A. A. (2017). Theoretical analysis for ultrasonic properties of vanadatephosphate glasses over an extended range of composition: Part II. *Journal of Non-Crystalline* Solids, 465, 49–54. https://doi.org/10.1016/j.jnoncrysol.2017.02.008
- El-Moneim, A. A., & Hassan, Y. A. (2018). Approach to dissociation energy and elastic properties of vanadate and V 2 O 5 -contained glasses from single bond strength: Part I. *Materials Chemistry and Physics*, 207, 271–281. https://doi.org/10.1016/j.matchemphys.2017.12.057
- El-Moneim, A., Youssof, I. M., & Abd El-Latif, L. (2006). Structural role of RO and Al2O3in borate glasses using an ultrasonic technique. *Acta Materialia*, *54*(14), 3811–3819. https://doi.org/10.1016/j.actamat.2006.04.012
- Elbatal, F. H., Marzouk, M. A., & Elbatal, H. A. (2016). Optical and crystallization studies of titanium dioxide doped sodium and potassium silicate glasses. *Journal of Molecular Structure*, *1121*, 54–59.

https://doi.org/10.1016/j.molstruc.2016.05.052

- Elkhoshkhany, N., El-Mallawany, R., & Syala, E. (2016). Mechanical and thermal properties of TeO2–Bi2O3–V2O5–Na2O–TiO2glass system. *Ceramics International*, 42(16), 19218–19224. https://doi.org/10.1016/j.ceramint.2016.09.086
- Elkhoshkhany, N., Gaafar, M. S., & El-mallawany, R. (2015). Elastic properties of quaternaryTeO2–ZnO–Nb2O5–Gd2O3 glasses. *Ceramics International*, *41*, 9862–9866. https://doi.org/10.1016/j.ceramint.2015.04.060
- Ersundu, M. C., & Ersundu, A. E. (2016). Structure and crystallization kinetics of lithium tellurite glasses. *Journal of Non-Crystalline Solids*, 453, 150–157. https://doi.org/10.1016/j.jnoncrysol.2016.10.007
- Ferrer, N. (2007). Applications of Fourier transform infrared spectroscopy. Barcelona, Spain.
- Fiocco, L., Ferroni, L., Gardin, C., Zavan, B., Secco, M., Matthews, S., & Bernardo, E. (2016). Wollastonite-diopside glass-ceramic foams from supercritical carbon dioxide-assisted extrusion of a silicone resin and inorganic fi llers. *Journal of Non-Crystalline Solids*, 443, 33–38. https://doi.org/10.1016/j.jnoncrysol.2016.04.012
- Gaafar, M. S., Abdeen, M. A. M., & Marzouk, S. Y. (2011). Structural investigation and simulation of acoustic properties of some tellurite glasses using artificial intelligence technique. *Journal of Alloys and Compounds*, 509(8), 3566–3575. https://doi.org/10.1016/j.jallcom.2010.12.064
- Gaafar, M. S., & Marzouk, S. Y. (2007). Mechanical and structural studies on sodium borosilicate glasses doped with Er 2 O 3 using ultrasonic velocity and FTIR spectroscopy. *Physica B*, 388, 294–302. https://doi.org/10.1016/j.physb.2006.06.132
- Gaafar, M. S., Marzouk, S. Y., Zayed, H. A., Soliman, L. I., & El-deen, A. H. S. (2013). Structural studies and mechanical properties of some borate glasses doped with different alkali and cobalt oxides. *Current Applied Physics*, 13(1), 152–158. https://doi.org/10.1016/j.cap.2012.07.007
- Gaafar, M. S., Shaarany, I., & Alharbi, T. (2014). Structural investigations on some cadmium-borotellurate glasses using ultrasonic, FT-IR and X-ray techniques. *Journal of Alloys and Compounds*, 616, 625–632. https://doi.org/10.1016/j.jallcom.2014.07.145
- Gaur, A., & Rastogi, V. (2016). Design and Analysis of Annulus Core Few Mode EDFA for Modal Gain Equalization. *IEEE Photonics Technology Letters*, 28(10), 1057–1060. https://doi.org/10.1109/LPT.2016.2528502
- Gayathri Pavani, P., Sadhana, K., & Chandra Mouli, V. (2011). Optical, physical and structural studies of boro-zinc tellurite glasses. *Physica B: Condensed Matter*, 406(6–7), 1242–1247. https://doi.org/10.1016/j.physb.2011.01.006

- Gebavi, H., Taher, M., Losteau, J., Milanese, D., Taccheo, S., Chimica, I., ... Duca, C. (2010). Spectroscopy of Yb: Tm doped tellurite glasses for efficient infrared fiber laser. *Optical Components and Materials*, 7598(0), 1–8. https://doi.org/10.1117/12.843017
- Ghorbani, F., Sanati, A. M., & Maleki, M. (2015). Production of Silica Nanoparticles from Rice Husk as Agricultural Waste by Environmental Friendly Technique. *Environmental Studies of Persian Gulf*, 2(1), 56–65.
- Ghosh, R., & Bhattacherjee, S. (2013). Chemical Engineering & Process Technology A Review Study on Precipitated Silica and Activated Carbon from Rice Husk. J Chem Eng Process Technol, 4(4). https://doi.org/10.4172/2157-7048.1000156
- Gouraud, F., Chotard, T., & Karray, R. (2015). Structural, mechanical and optical investigations in the TeO 2 -rich part of the TeO 2 e GeO 2 e ZnO ternary glass system. *Solid State Sciences*, 40, 20–30. https://doi.org/10.1016/j.solidstatesciences.2014.12.009
- Grelowska, I., Reben, M., Burtan, B., Sitarz, M., Cisowski, J., Sayed, E., ... Dudek, M. (2016). Structural and optical study of tellurite - barium glasses. *Journal of Molecular* Structure, 1126, 219–225. https://doi.org/10.1016/j.molstruc.2016.01.034
- Hafiz, M., Zaid, M., Matori, K. A., Aziz, S. H. A., Zakaria, A., Sabri, M., & Ghazali, M. (2012). Effect of ZnO on the Physical Properties and Optical Band Gap of Soda Lime Silicate Glass. *Int. J. Mol. Sci. Int. J. Mol. Sci*, 13, 7550–7558. https://doi.org/10.3390/ijms13067550
- Hajer, S. S., Halimah, M. K., Azmi, Z., & Azlan, M. N. (2014). Optical properties of Zinc-Borotellurite doped samarium. *Chalcogenide Letters*, 11(11), 553–566.
- Hajer, S. S., Halimah, M. K., Azmi, Z., & Azlan, M. N. (2016). Effect of Samarium Nanoparticles on Optical Properties of Zinc Borotellurite Glass System. *Materials Science Forum*, 846. https://doi.org/10.4028/www.scientific.net/MSF.846.63
- Halimah, M. K., Daud, W. M., & Sidek, H. A. A. (2010). Effect of AgI addition on elastic properties of quaternary tellurite glass systems. *Chalcogenide Letters*, 7(11), 613–620.
- Halimah, M. K., Daud, W. M., Sidek, H. A. A., Zaidan, A. W., & Zainal, A. S. (2010). Optical properties of ternary tellurite glasses. *Material Science-Poland*, 28(1).
- Halimah, M. K., Hasnimulyati, L., Zakaria, A., Halim, S. A., Ishak, M., Azuraida, A., & Al-Hada, N. M. (2017). Influence of gamma radiation on the structural and optical properties of thulium-doped glass. *Materials Science and Engineering B: Solid-State Materials for Advanced Technology*, 226(September), 158–163. https://doi.org/10.1016/j.mseb.2017.09.010

- Halimah, M. K., Sidek, H. A. A., Daud, W. M., Zainul, H., Talib, Z. A., Zaidan, A., ... Mansor, H. (2005). Ultrasonic Study and Physical Properties of Borotellurite Glasses. *American Journal of Applied Sciences*, 2(11), 1541– 1546.
- Hamezan, M., Sidek, H., Zaidan, A., Kaida, K., & Zainal, A. (n.d.). Elastic Constants and Thermal Properties of Lead-bismuth Borate Glasses.
- Hasnimulyati, L., Halimah, M. K., Zakaria, A., Halim, S. A., & Ishak, M. (2017). A comparative study of the experimental and the theoretical elastic data of Tm3+doped zinc borotellurite glass. *Materials Chemistry and Physics*, 192, 228–234. https://doi.org/10.1016/j.matchemphys.2017.01.086
- Hesham, A., & Samier, M. (2003). Ultrasonic velocity and elastic moduli of heavy metal tellurite glasses. *Material Chemistry and Physics*, 80, 517–523. https://doi.org/10.1016/S0254-0584(03)00099-3
- Hesham, A., & Samier, M. (2003). Ultrasonic velocity and elastic moduli of heavy metal tellurite glasses. *Material Chemistry and Physics*, 80, 517–523.
- Hraiech, S., Bouzidi, C., & Férid, M. (2017). Luminescence properties of Er3+-doped phosphate glasses. *Physica B: Condensed Matter*, 522(July), 15–21. https://doi.org/10.1016/j.physb.2017.07.047
- Hwa, L.-G., Lu, C. L., & Liu, L.-C. (2000). Elastic moduli of calcium alumino-silicate glasses studied by Brillouin scattering. *Materials Research Bulletin*, 35(8), 1285–1292. https://doi.org/10.1016/S0025-5408(00)00317-2
- IAEA. (1988). Ultrasonic Testing of Materials At Level 2. In A Technical Document Issued By The Internal Atomic Energy Agency, Vienna (p. 278). Vienna.
- Jágerská, J., Salavcová, L., Míka, M., & Spirková, J. (2008). Er Yb Waveguide Amplifiers in Novel Silicate Glasses. *IEEE Journal of Quantum Electronics*, 44(6), 536–541.
- Jamalaiah, B. C., Srinivasa, R. C., Vijaya Kumar, M. V, Rama Gopal, K., & Reddy, R. R. (2015). Enhanced 1.53 µm luminescence in Er3+ -doped sodium boro silicate glasses by Yb3+ co-doping. *Applied Science Letters*, 1(3), 82–85. https://doi.org/10.17571/appslett.2015.01019
- Janek, J., So, M., Zur, L., Pietrasik, E., Pisarska, J., & Pisarski, W. A. (2016). Luminescence investigations of rare earth doped lead-free borate glasses modi fi ed by MO ( M <sup>1</sup>/<sub>4</sub> Ca , Sr , Ba ), *180*, 237–243. https://doi.org/10.1016/j.matchemphys.2016.06.001
- Jarosław, B., JAKUBCZAK, P., MAJERSKI, K., OSTAPIUK, M., & SUROWSKA, B. (2013). METHODS OF ULTRASONIC TESTING, AS AN EFFECTIVE WAY OF ESTIMATING DURABILITY AND DIAGNOSING OPERATIONAL CAPABILITY OF COMPOSITE LAMINATES USED IN AEROSPACE INDUSTRY. *NIEZAWODNOSC – MAINTENANCE AND RELIABILITY*, 15(3), 284–289.

- Joshi, C., Rai, R. N., & Rai, S. B. (2012). Structural, thermal, and optical properties of Er 3 b /Yb 3 b co-doped oxyhalide tellurite glasses, glass-ceramics and ceramics. *Journal of Quantitative Spectroscopy and Radiative Transfer*, *113*, 397–404. https://doi.org/10.1016/j.jqsrt.2012.01.004
- Kaewkhao, J., & Limsuwan, P. (2012). Utilization of rice husk fly ash in the color glass production. *Procedia Engineering*, *32*(32), 670–675. https://doi.org/10.1016/j.proeng.2012.01.1325
- Kaky, K. M., Lakshminarayana, G., Baki, S. O., Kityk, I. V, Taufiq-yap, Y. H., & Mahdi, M. A. (2017). Structural, thermal and optical absorption features of heavy metal oxides doped tellurite rich glasses. *Results in Physics*, 7, 166–174. https://doi.org/10.1016/j.rinp.2016.12.013
- Kallel, T., Koubaa, T., Dammak, M., Pandya, S. G., Kordesch, M. E., Wang, J., ...
  Wang, Y. (2016). Spectra , energy levels and crystal fi eld calculation of Er 3
  b doped in AlN nanoparticles. *Journal of Luminescence*, 171, 42–50. https://doi.org/10.1016/j.jlumin.2015.11.002
- Kassab, L. R. P., Courrol, L. C., Seragioli, R., Wetter, N. U., Tatumi, S. H., & Gomes, L. (2004). Er3+ laser transition in PbO-PbF2-B2O 3 glasses. *Journal of Non-Crystalline Solids*, *348*, 94–97. https://doi.org/10.1016/j.jnoncrysol.2004.08.132
- Kaur, A., Khanna, A., González, F., Pesquera, C., & Chen, B. (2016). Structural, optical, dielectric and thermal properties of molybdenum tellurite and borotellurite glasses. *Journal of Non-Crystalline Solids*, 444, 1–10. https://doi.org/10.1016/j.jnoncrysol.2016.04.033
- Kek, B., & Grum, T. (2016). Pulse-Echo Ultrasonic Testing of Adhesively Bonded Joints in Glass Façades, 62, 147–153. https://doi.org/10.5545/svjme.2015.2988
- Kesavulu, C. R., Kim, H. J., Lee, S. W., Kaewkhao, J., Wantana, N., Kothan, S., & Kaewjaeng, S. (2016). Influence of Er 3 b ion concentration on optical and photoluminescence properties of Er 3 b -doped gadolinium-calcium silica borate glasses. *Journal of Alloys and Compounds*, 683, 590–598. https://doi.org/10.1016/j.jallcom.2016.04.314
- Kesavulu, C. R., Kiran Kumar, K., & Jayasankar, C. K. (2014). Upconversion properties of Er 3+ -doped oxyfluoride glass-ceramics containing SrF 2 nanocrystals. SPIE Photonics West 2014-OPTO: Optoelectronic Devices and Materials, 8987, 89871J. https://doi.org/10.1117/12.2039308
- Khan, A. A. (1999). Ultrasonic Testing of Materials at Level 2 Ultrasonic Testing of Materials at Level 2 (10th ed.). Vienna: Internatinal Atomic Energy Agency.
- Khan, M. N. N., Jamil, M. A., Kaish, A. B. M. A., & Zain, M. F. M. (2014). An Overview on Manufacturing of Rice Husk Ash as Supplementary Cementitious Material. *Australian Journal of Basic and Applied Sciences*,

8(19), 176–181.

- Kindrat, I. I., Padlyak, B. V., & Drzewiecki, A. (2015). Luminescence properties of the Sm-doped borate glasses. *Journal of Luminescence*, 166, 264–275. https://doi.org/10.1016/j.jlumin.2015.05.051
- King, P. L., Ramsey, M. S., McMillan, P. F., & Swayze, G. Laboratory Fourier Transform Infrared Spectroscopy Methods For Geologic Samples (2004).
- Koskenvaara, H. (2008). Photoluminescence Spectroscopy And Carrier Dynamics Modeling Of Quantum Dot Structures Doctoral Dissertation Photoluminescence Spectroscopy And Carrier Dynamics Modeling Of Quantum Dot Structures Doctoral Dissertation. Espoo; Findland.
- Kumar, A. (2010). Preparation of Apatite-Wollastonite and Phlogophite Glass Ceramic and Study of Its Properties. National Institute Of Technology Rourkela.
- Kumar, S., Sangwan, P., V, D. R. M., & Bidra, S. (2013). Utilization of Rice Husk and Their Ash : A Review. *Journal of Chemical and Environmental Sciences*, *1*(5), 126–129.
- Kundu, R. S., Dhankhar, S., Punia, R., Nanda, K., & Kishore, N. (2014). Bismuth modified physical, structural and optical properties of mid-IR transparent zinc boro-tellurite glasses. *Journal Of Alloys And Compounds*, 587, 66–73. https://doi.org/10.1016/j.jallcom.2013.10.141
- Lakshminarayana, G., Kaky, K. M., Baki, S. O., Lira, A., Nayar, P., Kityk, I. V, & Mahdi, M. A. (2017). Physical, structural, thermal, and optical spectroscopy studies of TeO2 B 2 O 3 e MoO 3 e ZnO e R 2 O (R<sup>1</sup>/<sub>4</sub> Li, Na, and K)/ MO (M<sup>1</sup>/<sub>4</sub> Mg, Ca, and Pb) glasses. *Journal of Alloys and Compounds Journal*, 690. https://doi.org/10.1016/j.jallcom.2016.08.180
- Landau, L. D., & Pekar, S. I. (1948). Effective Mass of Polarons. *Journal of Physics*, 18(5), 71–74.
- R., Bootjomchai, C. Laopaiboon, & (2015).Physical properties and thermoluminescence of glasses designed for radiation dosimetry measurements. **Materials** Design, 80, 20-27. and https://doi.org/10.1016/j.matdes.2015.05.002
- Laopaiboon, R., Laopaiboon, J., Pencharee, S., Nontachat, S., & Bootjomchai, C. (2016). The effects of gamma irradiation on the elastic properties of soda lime glass doped with cerium oxide. *Journal of Alloys and Compounds*, 666, 292– 300. https://doi.org/10.1016/j.jallcom.2016.01.098
- Lima, S. M., Souza, A. K. R., Langaro, A. P., Silva, J. R., Costa, F. B., Moraes, J. C. S., ... Andrade, L. H. C. (2017). Fluorescence quantum yield of Yb 3 b -doped tellurite glasses determined by thermal lens spectroscopy. *Optical Materials*, 63, 19–25. https://doi.org/10.1016/j.optmat.2016.08.042

- Linda, D., Duclère, J.-R., Hayakawa, T., Dutreilh-Colas, M., Cardinal, T., Mirgorodsky, A., ... Thomas, P. (2013). Optical properties of tellurite glasses elaborated within the TeO 2 –Tl 2 O–Ag 2 O and TeO 2 –ZnO–Ag 2 O ternary systems. *Journal of Alloys and Compounds*, 561, 151–160. https://doi.org/10.1016/j.jallcom.2013.01.172
- Linganna, K., Narro-garcía, R., Desirena, H., Rosa, E. De, Basavapoornima, C., Venkatramu, V., & Jayasankar, C. K. (2016). Effect of P 2 O 5 addition on structural and luminescence properties of Nd 3 b -doped tellurite glasses. *Journal of Alloys and Compounds*, 684, 322–327. https://doi.org/10.1016/j.jallcom.2016.05.082
- Liou, T.-H. (2004). Preparation and characterization of nano-structured silica from rice husk. *Materials Science and Engineering: A*, *364*, 313–323. https://doi.org/10.1016/j.msea.2003.08.045
- M. Çelikbilek, A.E. Ersundu, E.O. Zayimb, S. A. (2015). Thermochromic behavior of tellurite glasses. *Journal of Alloys and Compounds*, 637, 162–170. https://doi.org/10.1016/j.jallcom.2015.03.013
- Ma, Y., Huang, F., Hu, L., & Zhang, J. (2013). Er3+/Ho3+-Codoped Fluorotellurite Glasses for 2.7 μm Fiber Laser Materials. *Fibers*, 1(2), 11–20. https://doi.org/10.3390/fib1020011
- Madrid, R., Nogueira, C. A., & Margarido, F. (2012). Production And Characterisation Of Amorphous Silica. In 4th International Conference on Engineering for Waste and Biomass Valorisation (pp. 10–12).
- Maheshvaran, K., Linganna, K., & Marimuthu, K. (2011). Composition dependent structural and optical properties of Sm 3 b doped boro-tellurite glasses. *Journal* of *Luminescence*, *131*(12), 2746–2753. https://doi.org/10.1016/j.jlumin.2011.06.047
- Maheshvaran, K., Veeran, P. K., & Marimuthu, K. (2013). Structural and optical studies on Eu 3 b doped boro-tellurite glasses. *Solid State Sciences*, *17*, 54–62. https://doi.org/10.1016/j.solidstatesciences.2012.11.013
- Mahraz, Z. A. S., Sahar, M. R., & Ghoshal, S. K. (2014). Band gap and polarizability of boro-tellurite glass: Influence of erbium ions. *Journal of Molecular Structure*, 1072(1), 238–241. https://doi.org/10.1016/j.molstruc.2014.05.017
- Makishima, A., & Mackenzie, J. D. (1973). Direct calculation of Young's modulus of glass. *Journal of Non-Crystalline Solids*, 12(1), 35–45. https://doi.org/10.1016/0022-3093(73)90053-7
- Makishima, A., & Mackenzie, J. D. (1975). Calculation of bulk modulus, shear modulus and Poisson's ratio of glass. *Journal of Non-Crystalline Solids*, *17*(2), 147–157. https://doi.org/10.1016/0022-3093(75)90047-2
- Mansour, E. (2012). FTIR spectra of pseudo-binary sodium borate glasses containing TeO 2. Journal of Molecular Structure, 1014, 1–6.

https://doi.org/10.1016/j.molstruc.2012.01.034

- Markom, A. M., Paul, M. C., Dhar, A., Das, S., Pal, M., & Bhadra, S. K. (2017). Performance comparison of enhanced Erbium – Zirconia – Yttria – Aluminum co-doped conventional erbium-doped fiber amplifiers. *Optik*, 132, 75–79. https://doi.org/10.1016/j.ijleo.2016.12.041
- Marzouk, S. Y., & Gaafar, M. S. (2007). Ultrasonic study on some borosilicate glasses doped with different transition metal oxides. *Solid State Communications*, *144*, 478–483. https://doi.org/10.1016/j.ssc.2007.09.017
- Matori, K. A., Hafiz, M., Zaid, M., Quah, H. J., Hj, S., Aziz, A., ... Ghazali, M. (2015). Studying the Effect of ZnO on Physical and Elastic Properties of (ZnO) x (P 2 O 5) 1 - x Glasses Using Nondestructive Ultrasonic Method. *Advances in Materials Science and Engineering*, 2015, 6.
- Matori, K. A., Sayyed, M. I., Sidek, H. A. A., Zaid, M. H. M., & Singh, V. P. (2017). Comprehensive study on physical, elastic and shielding properties of lead zinc phosphate glasses. *Journal of Non-Crystalline Solids*, 457, 97–103. https://doi.org/10.1016/j.jnoncrysol.2016.11.029
- Mauro, J. C. (2011). Topological constraint theory of glass. *American Ceramic Society Bulletin*, 90(4), 31–37.
- Meena, S. L., & Bhatia, B. (2016). Polarizability and Optical Basicity of Er 3 + Ions Doped Zinc Lithium Bismuth Borate Glasses. *Journal of Pure Applied and Industrial Physics*, 6(10), 175–183.
- Mhareb, M. H. A., Hashim, S., Ghoshal, S. K., Alajerami, Y. S. M., Saleh, M. A., Maqableh, M. M. A., & Tamchek, N. (2015). Optical and erbium ion concentration correlation in lithium magnesium borate glass. *Optik*, 126(23), 3638–3643. https://doi.org/10.1016/j.ijleo.2015.08.222
- Michalina, W., Tomasz, L., Anna, S., Marcin, Ł., Wojciech, S., & Barbara, K. (2017).
  Eu 3 b doped tellurite glass ceramics containing SrF 2 nanocrystals:
  Preparation, structure and luminescence properties. *Journal of Alloys and Compounds*, 696, 619–626. https://doi.org/10.1016/j.jallcom.2016.11.301
- Miniscalco, W. J., & Quimby, R. S. (1991). General procedure for the analysis of Er^3+ cross sections. *Optics Letters*, 16(4), 258. https://doi.org/10.1364/OL.16.000258
- Mo, Z. X., Guo, H. W., Liu, P., Shen, Y. D., & Gao, D. N. (2016). Luminescence properties of magneto-optical glasses containing Tb 3 p ions. *Journal of Alloys* and Compounds, 658, 967–972. https://doi.org/10.1016/j.jallcom.2015.10.236
- Mohanta, K., Kumar, D., & Parkash, O. (2012). Properties and Industrial Applications of Rice husk : A review. *International Journal of Emerging Technology and Advanced Engineering*, 2(10), 86–90.

- Munoz-Martín, D., Villegas, M. A., Gonzalo, J., & Fernández-Navarro, J. M. (2009). Characterisation of glasses in the TeO2-WO3-PbO system. *Journal of the European Ceramic Society*, 29(14), 2903–2913. https://doi.org/10.1016/j.jeurceramsoc.2009.04.018
- Mustafa, I. S., Ain, N., Razali, N., Ibrahim, A. R., Yahaya, Z., & Kamari, H. M. (2015). From Rice Husk to Transparent Radiation Protection Material. *Jurnal Intelek*, 9(2), 1–6.
- Nagrale, S. D., Hajare, H., & Modak, P. R. (2012). Utilization Of Rice Husk Ash. International Journal of Engineering Research and Applications (IJERA), 2(4), 1–5.
- Narayanan, M. K., & Shashikala, H. D. (2015). Thermal and optical properties of BaO–CaF 2 –P 2 O 5 glasses. *Journal of Non-Crystalline Solids*, 422, 6–11. https://doi.org/10.1016/j.jnoncrysol.2015.04.038
- Nobarzad, A. E. K., Masoumi, K. M., & Heirdari, K. (2014). *Phase Identification by X-ray diffraction*.
- Noorazlan, A. M., Kamari, H. M., Baki, S. O., & Mohamad, D. W. (2015). Green Emission of Tellurite Based Glass Containing Erbium Oxide Nanoparticles. *Journal of Nanomaterials*, 952308(1155). https://doi.org/10.1155/2015/952308
- Oana, C., Mocioiu, A., Marin, A., & Zaharescu, M. (2017). Study of historical lead silicate glasses and their preservation by silica coating. *Ceramics International Journal*, 43, 77–83. https://doi.org/10.1016/j.ceramint.2016.09.055
- Ono, S., & Tanabe, S. (2004). Evidence of Enhanced Hypersensitive Transition inErbium-Doped FibersWith Different Al2O3 Content. *IEEE JOURNAL OF QUANTUM ELECTRONICS*, 40(12), 1704–1708.
- Owen, T. (1996). *Principles and applications of UV-visible spectroscopy*. Copyright Hewlett-Packard Company.
- Oyawale, F. A. (2012). Characterization of Rice Husk via Atomic Absorption Spectrophotometer for Optimal Silica Production. *International Journal of Science and Technology*, 2(4), 210–213.
- Pal, I., Agarwal, A., & Sanghi, S. (2012). Spectral analysis and structure of Cu 2+ doped cadmium bismuth borate glasses. *Indian Journal of Pure & Applied Physics*, 50, 237–244.
- Pandarinath, M. A., Upender, G., Rao, K. N., & Babu, D. S. (2016). Thermal, optical and spectroscopic studies of boro-tellurite glass system containing ZnO. *Journal of Non-Crystalline Solids*, 433, 60–67. https://doi.org/10.1016/j.jnoncrysol.2015.11.028

- Pavani, P. G., Sadhana, K., & Mouli, V. C. (2011). Optical, physical and structural studies of boro-zinc tellurite glasses. *Physica B: Physics of Condensed Matter*, 406, 1242–1247. https://doi.org/10.1016/j.physb.2011.01.006
- Pawar, P. P., Munishwar, S. R., Gautam, S., & Gedam, R. S. (2017). Physical, thermal , structural and optical properties of Dy 3 b doped lithium alumino-borate glasses for bright W-LED. *Journal of Luminescence*, 183, 79–88. https://doi.org/10.1016/j.jlumin.2016.11.027
- PerkinElme, L. (2000). An Introduction to Fluorescence Spectroscopy. Buckinghamshire, United Kingdom.
- Pönitzsch, A., Nofz, M., Wondraczek, L., & Deubener, J. (2016). Bulk elastic properties, hardness and fatigue of calcium aluminosilicate glasses in the intermediate-silica range. *Journal of Non-Crystalline Solids*, 434, 1–12. https://doi.org/10.1016/j.jnoncrysol.2015.12.002
- Pontuschka, W. M., Giehl, J. M., Miranda, A. R., Costa, Z. M. Da, & Alencar, A. M. (2016). Effect of the Al2O3 addition on the formation of silver nanoparticles in heat treated soda-lime silicate glasses. *Journal of Non-Crystalline Solids*, 453, 74–83. https://doi.org/10.1016/j.jnoncrysol.2016.09.028
- Poor, H. B., Aziz, H. A., & Zamiri, R. (2013). Ultrasonic and optical properties and emission of Er 3 + / Yb 3 + doped lead bismuth-germanate glass affected by Bi + / Bi 2 + ions. Journal of Luminescence, 143, 526–533. https://doi.org/10.1016/j.jlumin.2013.05.053

Pörner Ingenieurgesellschaft mbH. (2006). Rice Husk Technology.

- Prasad, R., & Pandey, M. (2012). Rice Husk Ash as a Renewable Source for the Production of Value Added Silica Gel and its Application: An Overview. *Bulletin of Chemical Reaction Engineering & Catalysis*, 7(1), 1–25. https://doi.org/10.9767/bcrec.7.1.1216.1-25
- Putra Hashim Syed Hashim, S., Abdul Aziz Sidek, H., Kamari Halimah, M., Amin Matori, K., Mohamad Daud Wan Yusof, W., & Hafiz Mohd Zaid, M. (2013). The Effect of Remelting on the Physical Properties of Borotellurite Glass Doped with Manganese. *Int. J. Mol. Sci, 14*, 1022–1030. https://doi.org/10.3390/ijms14011022
- Qu, W., Liu, S., Xiao, Z., Ma, J., & Wei, M. (2016). Rheological and thermodynamic properties of alkaline earth oxide bearing silicate glasses: Implication of structural role of Mg 2 b in peralkaline glasses. *Journal of Alloys and Compounds*, 661, 1–5. https://doi.org/10.1016/j.jallcom.2015.11.165
- Química, D. D. E., Politécnica, E., Paulo, U. D. S., Prof, A., & Prestes, L. (2011). Production of Silica Gel from Residual Rice Husk Ash. *Quim Noya*, *34*(1), 71–75.

- Rajesh, S., Saravanan, S., & Palani, R. (2015). Structural and Elastic Properties of Li + and W 6 + Metal Ions Doped With Sodium Borate Glass Using Pulser – Receiver Technique. *International Journal of Science and Research (IJRS)*, (January), 22–24.
- Ramadevudu, G., & Chary, M. N. (2017). Physical and spectroscopic studies of Cr 3 b doped mixed alkaline earth oxide borate glasses. *Materials Chemistry and Physics*, *186*, 382–389. https://doi.org/10.1016/j.matchemphys.2016.11.009
- Rao, V. H., Prasad, P. S., Rao, P. V., Santos, L. F., & Veeraiah, N. (2016). Influence of Sb 2 O 3 on tellurite based glasses for photonic applications. *Journal of Alloys and Compounds*, 687, 898–905. https://doi.org/10.1016/j.jallcom.2016.06.256
- Rao, Y. R., Goud, K. K., Kumar, E. R., Chandra, M., Reddy, S., & Rao, B. A. (2014). Upconversion Luminescence in Er 3+ / Yb 3+ Codoped Lead Bismuth Indium Borate Glasses. *International Journal of Recent Development in Engineering* and Technology, 3(1), 2347–6435.
- Rao, Y. R., Goud, K. K., Kumar, E. R., Reddy, M. C. S., Rao, B. A., Chandra, M., ... Rao, B. A. (2014). Upconversion Luminescence in Er 3+ / Yb 3+ Codoped Lead Bismuth Indium Borate Glasses. *International Journal of Recent Development in Engineering and Technology*, 3(1), 2347–6435.
- Rasool, S. N., Jamalaiah, B. C., Suresh, K., Moorthy, L. R., & Jayasankar, C. K. (2017). Spectroscopic properties of Er 3 b -doped phosphate based glasses for. *Journal of Molecular Structure*, *1130*, 837–843. https://doi.org/10.1016/j.molstruc.2016.10.090
- Rawan, E. H., Frédérique, F., Pierre, F., Alexander, P., & Daniel, R. N. (2017). Structure and properties of lime alumino-borate glasses. *Chemical Geology*, 461(2017), 75–81. https://doi.org/10.1016/j.chemgeo.2016.11.025
- Rocherulle, J., Ecolivet, C., Poulain, M., Verdier, P., & Laurent, Y. (1989). Elastic moduli of oxynitride glasses. Extension of Makishima and Mackenzie's theory. *Journal of Non-Crystalline Solids*, 108(2), 187–193. https://doi.org/10.1016/0022-3093(89)90582-6
- Rodriguez, O., Curran, D. J., Papini, M., Placek, L. M., Wren, A. W., Schemitsch, E. H., ... Towler, M. R. (2016). Characterization of silica-based and boratebased, titanium-containing bioactive glasses for coating metallic implants. *Journal of Non-Crystalline Solids*, 433, 95–102. https://doi.org/10.1016/j.jnoncrysol.2015.09.026
- Rojek, M., Stabik, J., & Sokół, S. (2007). Fatigue and ultrasonic testing of epoxy-glass composites. *Journal of Achievements in Material and Manufacturing Engineering*, 20(1–2), 183–186.

- Rolli, R., Montagna, M., Chaussedent, S., Monteil, A., Tikhomirov, V. K., & Ferrari, M. (2003). Erbium-doped tellurite glasses with high quantum efficiency and broadband stimulated emission cross section at 1.5 lm. *Optical Materials*, 21, 743–748.
- Rouxel, T. (2007). Elastic Properties and Short-to Medium-Range Order in Glasses. *American Ceramic Society Journal*, 90(10), 3019–3039. https://doi.org/10.1111/j.1551-2916.2007.01945.x
- Rudiger, P. (2017). Fiber Amplifiers: Self-consistent Solutions for the Steady State. Retrieved from https://www.rp?photonics.com/tutorial\_fiber\_amplifiers3.html
- Rudiger, P. (2017). Fiber Amplifiers for Ultrashort Pulses. Retrieved from https://www.rp?photonics.com/tutorial fiber amplifiers8.html
- Saddeek, Y. B. (2004a). Structural analysis of alkali borate glasses. *Physica B*, 344, 163–175. https://doi.org/10.1016/j.physb.2003.09.254
- Saddeek, Y. B. (2004b). Ultrasonic study and physical properties of some borate glasses. *Materials Chemistry and Physics*, 83, 222–228. https://doi.org/10.1016/j.matchemphys.2003.09.051
- Saddeek, Y. B. (2005). Elastic properties of Gd 3+ -doped tellurovanadate glasses using pulse-echo technique. *Materials Chemistry and Physics*, 91, 146–153. https://doi.org/10.1016/j.matchemphys.2004.11.005
- Saddeek, Y. B. (2009). Structural and acoustical studies of lead sodium borate glasses. *Journal of Alloys and Compounds*, 467, 14–21. https://doi.org/10.1016/j.jallcom.2007.11.126
- Saddeek, Y. B. (2011). Study of elastic moduli of lithium borobismuthate glasses using ultrasonic technique. *Journal of Non-Crystalline Solids*, 357(15), 2920– 2925. https://doi.org/10.1016/j.jnoncrysol.2011.03.034
- Saddeek, Y. B., Abd, L., & Latif, E. (2004). Effect of TeO 2 on the elastic moduli of sodium borate glasses. *Physica B*, 348, 475–484. https://doi.org/10.1016/j.physb.2004.02.001
- Saddeek, Y. B., & Elsayed, N. Z. (2015). Structural and mechanical features of some lanthanum tellurite glasses. *NRC Research Press*, *465*(September 2014), 460–465.
- Saddeek, Y. B., & Latif, L. A. El. (2004). Effect of TeO2 on the elastic moduli of sodium borate glasses. *Physica B: Condensed Matter*, 348(1–4), 475–484. https://doi.org/10.1016/j.physb.2004.02.001
- Said Mahraz, Z. A., Sahar, M. R., Ghoshal, S. K., Ashur, Z., Mahraz, S., Sahar, M. R., ... Ghoshal, S. K. (2014). Band gap and polarizability of boro-tellurite glass: Influence of erbium ions. *Journal of Molecular Structure Journal*, 1072(1), 238–241. https://doi.org/10.1016/j.molstruc.2014.05.017

- Sandu, V., Cimpoiasu, E., Greculeasa, S., Kuncser, A., Nicolescu, M. S., & Kuncser, V. (2017). Magnetite-based glass-ceramics prepared by controlled crystallization of borosilicate glasses : Effect of nucleating agents on magnetic properties and relaxation. *Ceramics International*, 43(3), 3405–3413. https://doi.org/10.1016/j.ceramint.2016.11.188
- Sayed, E., Elokr, M. M., & Aboudeif, Y. M. (2016). Optical, elastic properties and DTA of TNZP host tellurite glasses doped with Er 3 b ions. *Journal of Molecular Structure*, 1108, 257–262. https://doi.org/10.1016/j.molstruc.2015.11.066
- Schmitt, J., & Flemming, H. (1998). FTIR-spectroscopy in microbial and material analysis. *International Biodeterioration & Biodegradation*, 41, 1–11.
- Selvi, S., Marimuthu, K., Murthy, N. S., & Muralidharan, G. (2016). Red light generation through the lead boro À telluro À phosphate glasses activated by Eu 3 þ ions. *Journal of Molecular Structure*, 1119, 276–285. https://doi.org/10.1016/j.molstruc.2016.04.073
- Sharifnasab, H., & Alamooti, M. Y. (2017). Preparation of silica powder from rice husk. *Agricultural Engineering International: CIGR Journal*, 19(1), 158–161.
- Sharma, B. I., Goswami, M., Sengupta, P., Shrikhande, V. K., Kale, G. B., & Kothiyal, G. P. (2004). Study on some thermo-physical properties in Li 2 O–ZnO–SiO 2 glass-ceramics. *Materials Letters*, 58, 2423–2428. https://doi.org/10.1016/j.matlet.2004.02.038
- Shelby, J. E. (2005). *Introduction to Glass Science and Technology* (Second Edi). New York State College of Ceramics at Alfred University School of Engineering, Alfred, NY, USA: Royal Society of Chemitry.
- Shelke, V. R., Bhagade, S. S., & Mandavgane, S. A. (2010). Mesoporous Silica from Rice Husk Ash. *Bulletin of Chemical Reaction Engineering & Catalysis*, 5(2), 63–67. https://doi.org/10.9767/bcrec.5.2.793.63-67
- Shinde, K. N., Dhoble, S. J., Swart, H. C., & Park, K. (2012). Basic Mechanisms of Photoluminescence. In *Phosphate Phosphors for Solid-State Lighting* (pp. 41– 60). https://doi.org/10.1007/978-3-642-34312-4
- Sidek, H. A. A., El-Mallawany, R. A., Hariharan, K., & Rosmawati, S. (2014). Effect of concurrent ZnO addition and AlF3 reduction on the elastic properties of tellurite based glass system. *Advances in Condensed Matter Physics*, 2014. https://doi.org/10.1155/2014/174362
- Sidek, H. A. A., El-Mallawany, R., Siti, S. B., Halimah, M. K., & Khamirul, A. M. (2015). Optical Properties of Erbium Zinc Tellurite Glass System. Advances in Material Science and Engineering, 628954(10.1155/2015/628954). https://doi.org/10.1155/2015/628954
- Sidek, H. A. A., Rosmawati, S., Talib, Z. A., Halimah, M. K., & Halim, S. A. (2013). Effect of Zinc on the Elastic Behaviour of (TeO 2) 90 (AlF 3) 10-x (ZnO)

x Glass System. International Journal of Basic & Applied Sciences IJBAS-IJENS, 90(October 2009), 41–44.

- Solids, N., Ramakrishna, R., Reddy, R. R., Ahammed, Y. N., Azeem, P. A., Gopal, K. R., ... Ramakrishna, R. (2001). Electronic Polarizability and Optical Basicity Properties of Oxide Glasses Through Average Electronegativity. *Journal of Non-Crystalline Solids*, 286(February 2015), 169–180. https://doi.org/10.1016/S0022-3093(01)00481-1
- Soundararajan, G. (2009). *Optical Characterization of Rare Earth Doped Glasses*. University of Saskatchewan, Saskatoon. https://doi.org/Department of Electric and Electronic Engineering, University of Saskatchewan, Canada.
- Souri, D. (2017). Ultrasonic velocities, elastic modulus and hardness of ternary Sb-V2O5-TeO2 glasses. *Journal of Non-Crystalline Solids*, 470(May), 112–121. https://doi.org/10.1016/j.jnoncrysol.2017.05.006
- Stuart, B. (2005). *Infrared Spectroscopy: Fundamentals and Applications*. Chichester, UK.: Spinger.
- Subramanian, A. Z., Murugan, G. S., Zervas, M. N., & Wilkinson, J. S. (2012). Spectroscopy, Modeling, and Performance of. *Journal of Lightwave Technology*, 30(10), 1455–1462.
- Suresh, B., Reddy, M. S., Ashok, J., Sesha, A. S., Rao, P. V., Kumar, V. R., & Veeraiah, N. (2016). Enhancement of orange emission of Co 2 b ions with Bi 3 b ions in lead silicate glasses. *Journal of Luminescence*, 172, 47–52. https://doi.org/10.1016/j.jlumin.2015.11.018
- Suresh, B., Zhydachevskii, Y., Brik, M. G., Suchocki, A., & Reddy, M. S. (2016). Amplification of green emission of Ho 3 b ions in lead silicate glasses by sensitizing with Bi 3 b ions. *Journal of Alloys and Compounds*, 683, 114–122. https://doi.org/10.1016/j.jallcom.2016.05.056
- Swapna, K., Mahamuda, S., Venkateswarlu, M., Srinivasa Rao, A., Jayasimhadri, M.,
  Shakya, S., & Prakash, G. V. (2015). Visible, Up-conversion and NIR (???1.5 ??m) luminescence studies of Er3+ doped Zinc Alumino Bismuth
  Borate glasses. *Journal of Luminescence*, 163, 55–63. https://doi.org/10.1016/j.jlumin.2015.02.036
- Tagiara, N. S., Palles, D., Simandiras, E. D., Psycharis, V., Kyritsis, A., & Kamitsos, E. I. (2017). Synthesis, thermal and structural properties of pure TeO2 glass and zinc-tellurite glasses. *Journal of Non-Crystalline Solids*, 457, 116–125. https://doi.org/10.1016/j.jnoncrysol.2016.11.033
- Tanner, D. B. (2013). *Optical effects in solids*. Department of Physics, University of Florida.
- Ter-Mikirtychev, V. (2014). Fundamentals of fiber lasers and fiber amplifiers. *Springer Series in Optical Sciences*, *181*, 1–253. https://doi.org/10.1007/978-3-319-02338-0\_9

- Thakur, V., Singh, A., Punia, R., Dahiya, S., & Singh, L. (2017). Structural properties and electrical transport characteristics of modi fi ed lithium borate glass ceramics. *Journal of Alloys and Compounds*, 696, 529–537. https://doi.org/10.1016/j.jallcom.2016.11.230
- Tian, Y., Li, B., Chen, R., Xia, J., Jing, X., & Zhang, J. (2016). Thermal stability and 2 . 7 m m spectroscopic properties in Er 3 b doped tellurite glasses. *Solid State Sciences*, 60, 17–22. https://doi.org/10.1016/j.solidstatesciences.2016.07.012
- Tingting, Z. H. U., Guowu, T., Xiaodong, C., & Min, S. U. N. (2016). Enhanced 1 . 8 μ m emission in Er 3 + / Tm 3 + co-doped lead silicate glasses under different excitations for near infrared laser. *Journal of Rare Earths*, 34(10), 978–985. https://doi.org/10.1016/S1002-0721(16)60124-2
- Todkar, B. S., Deorukhkar, O. A., & Deshmukh, S. M. (2016). Extraction of Silica from Rice Husk. International Journal of Engineering Research and Development, 12(3), 69–74.
- Trinel, J., Cocq, G. Le, Andresen, E. R., Quiquempois, Y., & Bigot, L. (2016). Optical Fiber Technology Latest results and future perspectives on Few-Mode Erbium Doped Fiber Amplifiers. *Optical Fiber Technology*, (2016), 1–8. https://doi.org/10.1016/j.yofte.2016.09.004
- Ugheoke, I. B., & Mamat, O. (2012). A critical assessment and new research directions of rice husk silica processing methods and properties. *Maejo Int. J. Sci. Technol*, 6(03), 430–448.
- Umar, S. A., Halimah, M. K., Chan, K. T., & Latif, A. A. (2017a). Physical, structural and optical properties of erbium doped rice husk silicate borotellurite (Erdoped RHSBT) glasses. *Journal of Non-Crystalline Solids*, 472(July), 31–38. https://doi.org/10.1016/j.jnoncrysol.2017.07.013
- Umar, S. A., Halimah, M. K., Chan, K. T., & Latif, A. A. (2017b). Polarizability, optical basicity and electric susceptibility of Er3 + doped silicate borotellurite glasses. *Journal of Non-Crystalline Solids*, 471(March), 101–109. https://doi.org/10.1016/j.jnoncrysol.2017.05.018
- Umar, S. A., Ibrahim, G. G., Ibrahim, I. E., Najib, M. U., & Liman, M. S. (2015). Simulation of heating in space plasma. *Pelagia Research Library*, *6*(6), 103–113.
- Veit, U., & Rüssel, C. (2016). Density and Young 's Modulus of ternary glasses close to the eutectic composition in the CaO Al 2 O 3 SiO 2 -system. *Ceramics International*, 42(5), 5810–5822. https://doi.org/10.1016/j.ceramint.2015.12.123
- Wright, M. W., Yao, H., & Marciante, J. R. (2012). Resonant Pumping of Er-Doped Fiber Amplifiers for Improved Laser Efficiency in Free-Space Optical Communications. *The Interplanetary Network Progress Report*, 42(189), 1– 20.

- Xie, J., Tang, H., Wang, J., Wu, M., Han, J., & Liu, C. (2018). Network connectivity and properties of non-alkali aluminoborosilicate glasses. *Journal of Non-Crystalline* Solids, 481(November 2017), 403–408. https://doi.org/10.1016/j.jnoncrysol.2017.11.023
- Yamane, M., & Ashara, Y. (2000). Glasses for Photonics. Cambridge University Press. Cambridge, New York, Port Melbourne, Madrid, Cape Town Yinnon,: Cambridge University Press. https://doi.org/10.1017/CBO9780511541308
- Yan, Y. (1997). Optical waveguide amplifiers based on Er-doped phosphate glasses.pdf. https://doi.org/10.6100/IR493128
- Ye, R., & Barron, A. R. (2011). *Photoluminescence Spectroscopy and its Applications* (Vol. 38357).
- Yildirim, S., Yurddaskal, M., Dikici, T., & Aritman, I. (2016). Structural and luminescence properties of undoped, Nd 3 b and Er 3 b doped TiO 2 nanoparticles synthesized by fl ame spray pyrolysis method. *Ceramics International*, 49(9), 10579–10586. https://doi.org/10.1016/j.ceramint.2016.03.131
- Yousef, E. S. (2013). Er 3+ ions doped tellurite glasses with high thermal stability, elasticity, absorption intensity, emission cross section and their optical application. *Journal of Alloys and Compounds Journal*, 561, 234–240. https://doi.org/10.1016/j.jallcom.2013.01.199
- Yousef, E. S., El-Adawy, A., & El-KheshKhany, N. (2006). Effect of rare earth (Pr2O3, Nd2O3, Sm2O3, Eu2O3, Gd2O3and Er2O3) on the acoustic properties of glass belonging to bismuth-borate system. *Solid State Communications*, 139(3), 108–113. https://doi.org/10.1016/j.ssc.2006.05.022
- Yu, C., Yang, Z., Huang, A., Chai, Z., Qiu, J., Song, Z., & Zhou, D. (2017). Photoluminescence properties of tellurite glasses doped Dy 3 + and Eu 3 + for the UV and blue converted WLEDs. *Journal of Non-Crystalline Solids*, 457, 1–8. https://doi.org/10.1016/j.jnoncrysol.2016.11.025
- Zhang, D.-L., Qi, L., Hua, P.-R., Yu, D.-Y., Vallés, J.-A., & Yue-Bun Pun, E. (2011). Emission and absorption cross section spectra of Er3+ in LiNbO3 crystals codoped with indium. *Journal of Materials Research*, 26(10), 1316–1325. https://doi.org/10.1557/jmr.2011.53
- Zhao, Q., Luo, Y., Wang, W., Canning, J., & Peng, G. D. (2017). Enhanced broadband near-IR luminescence and gain spectra of bismuth/erbium co-doped fiber by 830 and 980 nm dual pumping. *AIP Advances*, 7(4). https://doi.org/10.1063/1.4981903
- Zheng, T., Qin, J.-M., Jiang, D.-Y., Lü, J.-W., & Xiao, S.-C. (2012). Spectroscopic properties in Er 3+ /Yb 3+ Co-doped fluorophosphate glass. *Chinese Physics B*, 21(4), 043302. https://doi.org/10.1088/1674-1056/21/4/043302