



UNIVERSITI PUTRA MALAYSIA

***PREPARATION OF ZINC OXIDE NANOPARTICLES BY LASER
ABLATION IN VARIOUS LIQUID MEDIA***

RAHELEH JORFI

ITMA 2014 8



**PREPARATION OF ZINC OXIDE NANOPARTICLES BY LASER
ABLATION IN VARIOUS LIQUID MEDIA**

By

RAHELEH JORFI

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
In Fulfilment of the Requirements for the Degree of Master of Science**

October 2014

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



DEDICATION

*This thesis dedicated to
My father, my mother, my dear brother, and dear sister
for their love, endless support and encouragement with love*



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Master of Science

PREPARATION OF ZINC OXIDE NANOPARTICLES BY LASER ABLATION IN VARIOUS LIQUID MEDIA

By

RAHELEH JORFI

October 2014

Chairman: Professor Azmi bin Zakaria, PhD

Faculty: Science

A big challenge in the synthesis of nanoparticles (NPs) is particle agglomeration. This tendency can be inhibited by stabilization of NPs and therefore much effort by researchers has been undertaken to use different liquid media as stabilizers. In the present study we have investigated the fabrication of zinc oxide nanoparticles (ZnO-NPs) using Laser Ablation (LA) technique. The first objective is studying the effect of laser parameters on fabrication of ZnO-NPs and secondly investigating the medium effect on optical properties and size of ZnO-NPs. The LA to ZnO plate was carried out in various media such as distilled water, polyvinyl acetate aqueous solution, ethylene glycol, and rice bran oil, and then characterized by UV-vis, photoluminescence (PL), FT-IR spectroscopy, and TEM microscopy. It was revealed the generation of NPs by LA in various liquid media was higher in comparison with in distilled water. The increased NPs generation is attributable to solvent plasma confinement toward the plate. In addition, at longer ablation duration the size decrease of NPs was remarkable. The UV-vis absorption spectra of the ZnO-NPs are monitored to characterize the particle growth because the onset of absorption is associated with the particle size. It is observed that average NPs size decreases with the increase of ablation time. The decrement can be explained by the way that at longer ablation time, there is an increment of produced NPs from the plate since more interactions occurred between target and laser light. NPs with different sizes demonstrate different optical properties; when the particle diameter decreased there is a blue-shifted in absorption spectra. It can be observed that increasing repetition rate lead to larger particle diameter. This size increment is confirmed by the red-shifted absorption spectra. We analyzed PL spectra to see the emission spectra of fabricated NPs. Both the absorption and emission band positions are dependent on the polarity of the medium; in the medium the PL spectrum is much more sensitive than the absorption spectrum. PL spectra of all fabricated ZnO-NPs show two emission peaks; the first one is sharp and narrow UV emission corresponds to excitonic emission and the second one is broad blue-green emission commonly referred to a deep-level or a trap-state emission attributed to the singly ionized

oxygen vacancy of ZnO. TEM images also show the morphology of the ZnO-NPs as they were all spherical. The produced ZnO-NPs in rice bran oil were well dispersed; those fabricated in polyvinyl acetate were more stable than other liquid media. Ethylene glycol has ability to prevent agglomeration of NPs. The average diameter of the obtained ZnO-NPs in distilled water was large; 27.12 nm, on the other hand rice bran oil leads to fabricate fine ones; 14.17 nm.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia Sebagai memenuhi keperluan untuk ijazah Master Sains

PENEYEDIAAN NANOZARAH ZINK OKSIDA MELALUI ABLASI LASER DALAM PELBAGAI MEDIA CECAIR

Oleh

RAHELEH JORFI

Oktober 2014

Pengerusi: Professor Azmi bin Zakaria, PhD

Fakulti : Sains

Cabaran utama dalam sintesis nano-zarah (NPs) adalah aglomerasi zarah. Kecenderongan ini boleh dihalang melalui penstabilan NPs dan oleh itu banyak usaha oleh penyelidik telah dilakukan menggunakan pelbagai media cecair sebagai penstabil. Dalam kajian ini kami telah menyelidiki fabrikasi nano-zarah zink oksida (ZnO-NPs) menggunakan teknik Ablasi Laser (LA). Objektif pertama adalah mengkaji kesan parameter-parameter laser dalam fabrikasi ZnO-NPs dan keduanya mengkaji kesan media cecair ke atas ciri optik dan ke atas saiz ZnO-NPs. LA ke atas plat ZnO telah dilakukan dalam pelbagai media seperti air suling, larutan cair polivinil asitat, etilena glikol, dan minyak dedak beras, dan kemudian dicirikan melalui spektroskopi-spektroskopi UV-vis, kefotopendarcahayaan, FT-IR, dan mikroskopi TEM. Adalah jelas bahawa penghasilan NPs melalui LA dalam pelbagai media cecair adalah lebih tinggi berbanding dengan yang di dalam air suling. Pertambahan penghasilan NPs adalah disabitkan dengan kurungan plasma pelarut kearah plat. Selain dari itu dalam tempoh ablasi lebih lama, pengecilan saiz NPs adalah luar biasa. Spektra penyerapan UV-vis dari ZnO-NPs adalah dimonitor untuk mencirikan pertumbuhan zarah kerana onset penyerapan adalah dikaitkan dengan saiz zarah. Adalah diperhatikan bahawa purata saiz NPs mengurang dengan pertambahan masa ablasi. Pengurangan ini boleh dijelaskan melalui cara bahawa pada tempoh ablasi lebih lama, terdapat pertambahan NPs dihasilkan dari plat kerana lebih saling-tindak berlaku diantara target dan cahaya laser. NP dengan pelbagai saiz menunjukkan ciri-ciri optik berbeza; apabila diameter zarah mengecil akan terdapat anjakan biru dalam spektra penyerapan. Boleh diperhatikan bahawa penambahan kadar ulangan menjurus kepada diameter zarah lebih besar. Penambahan saiz ini disahkan oleh spektra penyerapan anjakan-merah. Kami menganalisa spektra PL untuk melihat spektra pancaran dari NPs terhasil. Kedua-dua kedudukan jalur penyerapan dan pancaran bergantung keatas kekutuban media; dalam media spektra PL adalah jauh lebih sensitif daripada spektra penyerapan. Spektra PL dari semua ZnO-NPs terfabrikasi menunjukkan dua puncak pancaran; yang pertama adalah tajam dan pancaran UV sempit bersabit dengan pancaran

eksotonik dan yang kedua adalah pancaran hijau lebar biasanya dirujukkan kepada paras-dalam atau suatu pancaran keadaan perangkap disabitkan kepada kekosongan oksigen terion tunggal dari ZnO. Imej TEM juga menunjukkan morfologi ZnO-NPs seolah mereka sfera semuanya. ZnO-NPs terhasil dalam minyak dedak beras adalah terserak secara sempurna; manakala yang difabrikasi dalam polivinil asitat adalah lebih stabil berbanding yang didalam medium cecair lain. Etilena glikol mempunyai kebolehan menghalang agglomerasi NPs. Daimeter ZnO-NPs yang diperolehi dalam air suling adalah paling besar, 27.12 nm; manakala yang didalam RBO adalah paling kecil, 14.17 nm.



ACKNOWLEDGEMENTS

At the end of this step of my graduate period has allowed for a bit of reflection, and the many people who have contributed to both my work, and my life during of this period of time.

First, I would like to express my full thanks and sincere gratitude to my great dear supervisor, Prof. Dr Azmi Zakaria for all of guidance, discussions, unlimited assistance consultations and support. He also taught me how to look at the life and science. I owe him in whole of my life. I also would like to thank my committee member; Assoc. Prof. Dr Halimah Mohamed Kamari, and also, Prof. Dr. Maarof H.A Moxsin for their invaluable suggestions, beneficial advices and their endless helps.

I wish to acknowledge my gratitude to all lecturers and staffs in ITMA. I would like to express my full thanks and sincere gratitude to my dear family for their encouragements, emotional supports and fortitude efforts in my life time. I also grateful to thank Dr. Arash Toudeshki, Dr. Reza Zamiri , Dr. Hamidreza Bahari, Dr. Amir Reza Sadrolhosseini, and Dr. Maryam Erfani Haghiri for their guidance; and my dear friend Samaneh Babakhani.

Raheleh Jorfi

2014



© COPYRIGHT UPM

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Azmi Zakaria, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Halimah Kamari, PhD

Associate Professor
Faculty of Science
Universiti Putra Malaysia
(Member)



BUJANG KIM HUAT, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Dat

Declaration by Graduate Student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature: _____ Date: _____

Name and Matric No.: _____

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: _____

Name of
Chairman of
Supervisory
Committee: _____

Signature: _____

Name of
Member of
Supervisory
Committee: _____

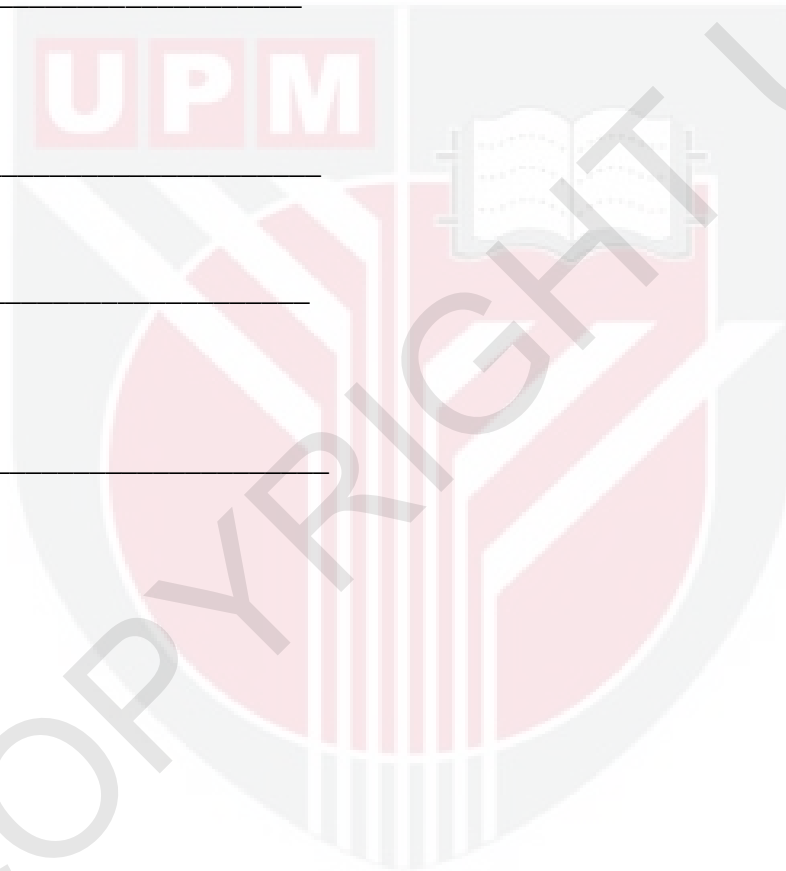


TABLE OF CONTENTS

| | Page |
|--|-------------|
| ABSTRACT | i |
| ABSTRAK | iii |
| ACKNOWLEDGEMENTS | v |
| APPROVAL | vi |
| DECLARATION | viii |
| LIST OF FIGURES | xiii |
| LIST OF TABLES | xvi |
| LIST OF ABBREVIATIONS | xvii |
| | |
| CHAPTER | |
| 1 INTRODUCTION | 1 |
| 1.1 Introduction to Nanoscience | 1 |
| 1.2 Metal Oxide Nanoparticle | 1 |
| 1.3 Zinc Oxide | 2 |
| 1.4 Laser Ablation Technique | 2 |
| 1.5 Scope of Study | 3 |
| 1.6 Statement of Problem | 4 |
| 1.7 Objective of research | 4 |
| | |
| 2 LITERATURE REVIEW | 5 |
| 2.1 Laser Ablation | 5 |
| 2.2 Nanomaterial | 5 |
| 2.3 Metallic NPs' Fabrication | 6 |
| 2.4 Advantage of Formation of ZnO NPs by LAL | 7 |
| 2.5 Zinc Oxide | 8 |
| 2.6 Previous Work in Fabrication of ZnO-NP by Using Laser Ablation | 8 |
| | |
| 3 METHODOLOGY | 11 |
| 3.1 General Setup of Laser Ablation in Liquids | 11 |
| 3.2 Parameters of the Laser | 12 |
| 3.2.1 Pulse Duration | 12 |
| 3.2.2 Laser Wavelength | 12 |
| 3.2.3 Pulse Repetition Rate | 12 |
| 3.3 Sample Preparation | 13 |
| 3.4 Materials | 14 |
| 3.4.1 Zinc Oxide Plate | 14 |
| 3.4.2 Glass Cell | 14 |
| 3.4.3 Liquid | 14 |

| | | |
|----------|---|-----------|
| 3.5 | Measurements | 15 |
| 3.5.1 | UV-Visible Spectrometer | 15 |
| 3.5.2 | Transmission Electron Microscopy | 15 |
| 3.5.3 | Fourier Transform Infrared Spectrometer | 15 |
| 3.5.4 | Photoluminescence Spectrometer | 15 |
| 3.6 | ZnO-NPs in Distilled Water | 15 |
| 3.6.1 | Effect of Laser Repetition Rate | 16 |
| 3.6.2 | Effect of LA Time | 16 |
| 3.7 | ZnO-NPs in Polyvinyl Acetate | 16 |
| 3.7.1 | Effect of Laser Repetition Rate | 17 |
| 3.7.2 | Effect of LA Time | 17 |
| 3.8 | ZnO-NPs in Ethylene Glycol | 17 |
| 3.8.1 | Effect of Laser Repetition Rate | 18 |
| 3.8.2 | Effect of LA Time | 18 |
| 3.9 | ZnO-NPs in Rice Bran Oil | 18 |
| 3.9.1 | Effect of Laser Repetition Rate | 19 |
| 3.9.2 | Effect of LA Time | 19 |
| 4 | RESULTS AND DISCUSSION | 20 |
| 4.1 | Introduction | 20 |
| 4.2 | Characterization of the ZnO-NPs in Distilled Water | 20 |
| 4.2.1 | Characterization of the ZnO-NPs in DW Fabricated under Different Laser Repetition Rate | 20 |
| 4.2.2 | Characterization of the ZnO-NPs in DW Fabricated under Different Ablation Times | 23 |
| 4.3 | Characterization of the ZnO-NPs in Polyvinyl acetate | 27 |
| 4.3.1 | Characterization of the ZnO-NPs in PVAc fabricated under Different Laser Repetition Rates | 27 |
| 4.3.2 | Characterization of the ZnO-NPs in PVAc Fabricated under Different Ablation Times | 31 |
| 4.4 | Characterization of the ZnO-NPs in Ethylene glycol | 34 |
| 4.4.1 | Characterization of the ZnO-NPs in EG Fabricated under Different Laser Repetition Rates | 34 |
| 4.4.2 | Characterization of the ZnO-NPs in EG Fabricated under Different Ablation Times | 38 |
| 4.5 | Characterization of the ZnO-NPs in Rice Bran Oil | 42 |
| 4.5.1 | Characterization of the ZnO-NPs in RBO Fabricated under Different Laser Repetition Rates | 42 |
| 4.5.2 | Characterization of the ZnO-NPs in RBO Fabricated under Different Ablation Times | 44 |

| | | |
|----------|--|-----------|
| 5 | CONCLUSIONS AND RECOMMENDATIONS | 50 |
| | REFERENCES | 52 |
| | APPENDIX | 60 |
| | BIODATA OF STUDENT | 63 |
| | LIST OF PUBLICATIONS | 64 |



LIST OF FIGURES

| Figure | | Page |
|--------|---|------|
| 3.1 | The experimental setup for a typical laser ablation of solids into a fluid environment (Shafeev, 2011) | 20 |
| 3.2 | Schematic of fabrication of ZnO-NPs in various media by PLA. | 23 |
| 3.3 | Diagram of the experimental setup for NPs fabrication by PLAL (Zamiri, et al., 2012) | 24 |
| 3.4 | Structure of polyvinyl acetate | 29 |
| 3.5 | Structure of ethylene glycol. | 31 |
| 3.6 | Structure of wheat kernel. | 32 |
| 3.7 | The chemical composition of the rice bran oil. | 32 |
| 4.1 | Photograph of colloidal ZnO-NPs (Ismail, et al., 2011) | 36 |
| 4.2 | UV-visible absorption spectra of fabricated ZnO-NPs in DW with different repetition rates: a=10, b=20, c=30, and d=40 Hz. | 37 |
| 4.3 | UV-visible absorption a) intensity, and b) wavelength versus repetition rate for ZnO-NPs synthesized in DW | 37 |
| 4.4 | TEM image and typical of statistical graph for ZnO-NPs in DW at different repetition rates: a,b) 10 and c,d) 40 Hz. | 38 |
| 4.5 | UV-visible absorption spectra of fabricated ZnO-NPs in DW in different times: a=10, b= 15, c= 30, and d= 45 min. | 40 |
| 4.6 | UV-visible absorption a) intensity, and b) wavelength versus ablation time for ZnO-NPs synthesized in DW. | 41 |
| 4.7 | TEM image and typical of statistical graph for ZnO-NPs in DW under: a,b) 10 and c,d) 45 min ablation times. | 42 |
| 4.8 | Photoluminescence spectra for ZnO-NPs in DW with the excitation wavelength: 350 nm. | 43 |
| 4.9 | UV-visible absorption spectra of fabricated ZnO-NPs in PVAc at different repetition rates: a=10, b=20, c=30 and d=40 Hz. | 45 |
| 4.10 | UV-visible absorption a) intensity, and b) wavelength versus repetition rate for ZnO-NPs synthesized in PVAc. | 46 |

| | | |
|------|--|----|
| 4.11 | TEM images of ZnO-NPs prepared by PLA of Zn in PVAc in different repetition rates: a,b) 10 and c,d) 40 Hz | 47 |
| 4.12 | UV-visible absorption spectra of fabricated ZnO-NPs in PVAc in different times: a=10, b= 15, c= 30, and d= 45 min. | 49 |
| 4.13 | UV-visible absorption a) intensity, and b) wavelength versus ablation time for ZnO-NPs synthesized in PVAc. | 49 |
| 4.14 | TEM image and typical of statistical graph for ZnO-NPs in PVAc under: a,b) 10 and c,d) 45 min ablation times. | 50 |
| 4.15 | Photoluminescence spectra for ZnO-NPs in PVAc with the excitation wavelength: 330 nm. | 51 |
| 4.16 | FTIR spectrum of a) PVAc and b) prepared ZnO-NPs in PVAc. | 52 |
| 4.17 | UV-visible absorption spectra of fabricated ZnO-NPs in EG with different repetition rates: a=10, b=20, c=30, and d=40 Hz. | 54 |
| 4.18 | UV-visible absorption a) intensity, and b) wavelength versus repetition rate for ZnO-NPs synthesized in EG. | 54 |
| 4.19 | TEM image and typical of statistical graph for ZnO-NPs in EG at different repetition rates: a,b) 10 and c,d) 40 Hz. | 55 |
| 4.20 | UV-visible absorption spectra of fabricated ZnO-NPs in EG in different times: a=10, b= 15, c= 30, and d= 45 min. | 57 |
| 4.21 | UV-visible absorption a) intensity, and b) wavelength versus ablation time for ZnO-NPs synthesized in EG. | 57 |
| 4.22 | TEM image and typical of statistical graph for ZnO-NPs in EG under: a,b) 10 and c,d) 45 min ablation times. | 58 |
| 4.23 | Photoluminescence spectra for ZnO-NPs in EG with the excitation wavelength: 250 nm. | 59 |
| 4.24 | FTIR spectrum of a) EG. b) prepared ZnO-NPs in EG. | 60 |
| 4.25 | UV-visible absorption spectra of fabricated ZnO-NPs in RBO with different repetition rates: a=10, b=20, c=30, and d=40 Hz. | 62 |
| 4.26 | UV-visible absorption a) intensity, and b) wavelength versus repetition rate for ZnO-NPs synthesized in RBO. | 62 |
| 4.27 | TEM image and typical of statistical graph for ZnO-NPs in RBO at different repetition rates: a,b) 10 and c,d) 40 Hz. | 63 |
| 4.28 | UV-visible absorption spectra of fabricated ZnO-NPs in RBO in different times: a=10, b= 15, c= 30, and d= 45 min. | 65 |

| | | |
|------|--|----|
| 4.29 | UV-visible absorption a) intensity, and b) wavelength versus ablation time for ZnO-NPs synthesized in RBO. | 65 |
| 4.30 | TEM image and typical of statistical graph for ZnO-NPs in RBO under: a,b) 10 and c,d) 45 min ablation times. | 66 |
| 4.31 | Photoluminescence spectra of prepared ZnO-NPs by PLA in RBO. Excitation wavelength: 350 nm | 67 |
| 4.32 | FTIR spectrum of a) RBO. b) prepared ZnO-NPs by PLA in RBO. | 68 |



LIST OF TABLES

| Table | | Page |
|--------------|---|-------------|
| 4.1 | Investigated parameters of ZnO-NPs as S1, S2, S3, and S4 in DW. | 36 |
| 4.2 | The mean diameters of ZnO-NPs and their corresponding standard deviations in DW. | 39 |
| 4.3 | Investigated parameters of ZnO-NPs as S5, S6, S7, and S8 in DW. | 40 |
| 4.4 | The mean diameters of ZnO-NPs and their corresponding standard deviations in DW. | 41 |
| 4.5 | Investigated parameters of ZnO-NPs as S1 ,S2 ,S3, and S4 in PVAc. | 46 |
| 4.6 | The mean diameters of ZnO-NPs and their corresponding standard deviations in PVAc | 48 |
| 4.7 | Investigated parameters of ZnO-NPs as S5, S6 , S7, and S8 in PVAc. | 48 |
| 4.8 | The mean diameters of ZnO-NPs and their corresponding standard deviations in PVAc | 50 |
| 4.9 | Investigated parameters of ZnO-NPs as S1, S2, S3, and S4 in EG. | 53 |
| 4.10 | The mean diameters of ZnO-NPs and their corresponding standard deviations in EG. | 56 |
| 4.11 | Investigated parameters of ZnO-NPs as S5, S6, S7, and S8 in EG. | 56 |
| 4.12 | The mean diameters of ZnO-NPs and their corresponding standard deviations in EG. | 59 |
| 4.13 | Investigated parameters of ZnO-NPs as S1 , S2 , S3, and S4 in RBO. | 61 |
| 4.14 | The mean diameters of ZnO-NPs and their corresponding standard deviations in RBO. | 64 |
| 4.15 | parameters of ZnO-NPs as S5, S6, S7, and S8 in RBO. | 64 |
| 4.16 | The mean diameters of ZnO-NPs and their corresponding standard deviations in RBO. | 66 |
| 4.17 | 4.17: Properties of ZnO-NPs in applied media | 69 |

LIST OF ABBREVIATIONS

| | |
|-----------|----------------------------------|
| LA | Laser ablation |
| NPs | Nanoparticles |
| DW | Distilled water |
| PVAc | Polyvinyl acetate |
| EG | Ethylene glycol |
| RR | Repetition rate |
| PL | Photoluminescence |
| FTIR | Fourier transform infrared |
| TEM | Transmission electron microscopy |
| Hz | Unit of frequency |
| nm | Nanometer |
| t | Time |
| λ | Wavelength |
| min | Minute |





© COPYRIGHT UPM

CHAPTER 1

INTRODUCTION

This chapter presents a brief introduction of the topic and concepts such as metal oxide nanoparticles (NPs), zinc oxide (ZnO) and pulsed laser ablation (PLA) used in this study. The statement of problems and objectives of the research is presented at the end of this chapter.

1.1 Introduction to Nanoscience

The word “nanoparticle” is normally used in the area of materials science to stand for particles in the size of less than 100 nanometres, which lie in the middle of the bulk state and the molecular state. However, similar physical and chemical properties of the bulk material can be seen in NPs larger than a few 10's of nanometres. Therefore, the maximum size limitation of nanoparticle that shows interesting properties different from those of the bulk is at almost 20 nanometers. The quantum size effect has been shown to emerge in NPs lying roughly in the range of 1-10 nanometers. As a result, as the size of a material is reduced down to nanoscale dimensions and more, its electronic properties alter significantly, whereby continuous energy bands of the bulk material changes to distinct energy levels of atoms (Barcikowski and Compagnini, 2013).

1.2 Metal Oxide Nanoparticle

In recent years, the fabrication and application of metal oxide NPs were greatly focused by researchers due to their distinctive physical properties which depend on variety of sizes, shape and the surrounding of NPs. In comparison, NPs are significantly different from those bulk materials as well as atomic or molecular structures. Since the NPs are sized between 100 to 1 nanometers, it exhibits large surface area and high percentage of surface atoms. Hence, when the size of materials approaches the nanoscale, the properties of materials also change and leading to the highest values of the activity to weight ratio (Jeng and Swanson, 2006).

Metal oxide NPs are receiving much more attention in physics, chemistry, and biotechnology. ZnO-NPs have numerous application, some of the more important applications are biosensors, antibacterial application and catalysis. Basically, the properties of nanostructured material related to the properties of NPs. Among them, the size and shape are the most prominent. Hence, the requirement in fabrication of metal oxide NPs is to develop the efficient method that ensured the stability and size controlled of metal NPs (Garcia and Rodriguez, 2007). They have also various optical properties. The nanoparticle optical spectrum depends on the particle morphology (size and shape), the nature of the metal, and the aggregation state of the particles. The optical properties and electronic structure of metal nanoparticle suspensions is mostly characterize by UV-Visible absorption spectroscopy, because of the relation between absorption bands and diameter of metal NPs.

1.3 Zinc Oxide

Advantages of ZnO are worth mentioning: It is a semiconductor that has a direct large band gap of 3.37 eV. The exciton binding energy of it is 60 meV that exceeds the room-temperature energy of 26 meV. The ZnO is an essential functional oxide that shows a transparent conductivity and exhibits near-ultraviolet emission. Due to its noncentral symmetry, ZnO is also a piezoelectric, that can be used to make transducers and electromechanical coupled sensors. ZnO is biocompatible and bio-safe material. Therefore, it can be utilized in many biomedical applications without need any covering (Garcia *et al.*, 2007).

These three exclusive characteristics of ZnO make it as the most imperative nanomaterials with many applications, in future research. The range of nanostructures which are offered here for ZnO, can open up lots of research fields in nanotechnology. Bulk ZnO is a wurtzite semiconductor with direct band gap which has the valence band with maximum separation by the spin-orbit interaction and crystal field.

ZnO-NPs were also widely examined as an emission material. This is due to the nanocrystallization that can improve the electrical and optical characteristics of wide-gap semiconductors which has a quantum confinement effect. However, impurity levels and surface defects are simply made during the nanoparticle fabrication process. The ZnO's lattice defects, like oxygen vacancies, are identified to make a green emission (Chen *et al.*, 2006). Many surface treatment methods, like annealing in oxygen, surface coating via polymeric materials, implanting in a silica matrix (Chakrabarti *et al.*, 2003), and also annealing face-to-face achieved in smother the green emission strength from ZnO nanostructures (Usui *et al.*, 2005).

1.4 Laser Ablation Technique

This study draws on the Laser ablation (LA) technique to prepare metal oxide NPs in liquids by ablating the metal palette with a high power, Nd:YAG pulsed laser. The resultant suspension of metal oxide NPs by LA is one of the common methods (Zamiri, 2010). Metal atoms, with a smaller amount of metal clusters are ablated from a plate during laser irradiation and are aggregated into sufficiently large sizes of metal oxide clusters. By changing the experimental condition of the ablation, the size distribution and concentration of these NPs can be readily controlled. The advantages of this method are the absence of chemical reagents or ions in the final preparation and the relative simplicity of the procedure (Amendola and Meneghetti, 2009).

This technique is a simple, straightforward and fast way for NPs generation and synthesis, in comparison with other techniques. It does not need a sustained reaction times, multi-step chemical synthetic procedures or high temperatures. It can also manufacture many different types of NPs from metallic to polymeric and semiconducting, as same as NPs of semiconducting alloys or compound multi-element metallic (Semaltianos, 2010).

It does not need the use of hazardous, toxic or pyrophoric chemical precursors to synthesis nanomaterial. Therefore, it is a laboratory safe, green and environmentally

friendly method. In the occasion that production arises in fluid, the out coming NPs, colloidal solutions are ultrapure, and this assists the use of these NPs in biochemical or biological in vivo purposes (Wang and Herron, 1991).

The use of a powerful laser beam with a palette for material removing or ablation must be illustrated by relation between electron-phonon coupling time constant of the material and pulse width of the laser. The energy for the beam of laser is initially transferred to the material carriers. The free electrons are carriers in the case of metals and electrons of valence band which afterward stimulated to the conduction band are carriers in semiconductors. If the laser pulse width is larger than a nanosecond pulse, the energy of the beam remains to pass into the material even though carriers already have finished transferring their energy into the lattice. This causes the energy of laser to waste as heat, from the area where the beam of laser incident to the surface of material. In this case the material ablation even for low fluencies laser is through vaporization and melting, for example, a solid to vapour conversion (Zamiri *et al.*, 2011c).

Various sorts of pulsed lasers have been used for ablation. For example, ruby laser was amongst the foremost applied to ablate solid materials for chemical analysis purposes. Currently, most laser ablation experiments employ excimer lasers or Nd:YAG. Solid state Nd:YAG systems were commonly employed as they are moderately low-cost, need small repairs, and simply built into small commercial ablation systems. Moreover, excimer lasers use chambers filled by halogen gas preferred to solid-state crystals. The laser wavelength effects on ablation process. In most cases, the higher the ablation rate, the shorter the laser wavelength and lower the fractionation. In Nd:YAG lasers, the main wavelength is near infrared at the wavelength of 532 nm (Russo *et al.*, 2002).

1.5 Scope of Study

These days of full challenges to the research trend of synthesis and characterization of metal oxide NPs have led to many advantages and applications to the various field of industry. The demand of new technique in synthesis of metal oxide NPs has been developed to give pure and stable product of NPs.

Based on the objectives, the fundamental works to synthesize metal oxide NPs have been further investigated. The Q-switch Pulsed Nd:YAG laser is utilized to synthesize the ZnO- NPs with the frequency doubled wavelength (532 nm). The prepared ZnO-NPs were ablated in different liquid environment which are distilled water, polyvinyl acetate, ethylene glycol and rice bran oil. In this experiment, 10 ml of each solution is used. The ZnO target was ablated in different liquid at different times and different repetition rates to disperse ZnO-NPs. The different liquid environment are used to observe the stability of ZnO-NPs.

All of the solutions were characterized immediately after preparation for absorbance spectrum and distribution of size respectively. The different concentration of ZnO-NPs produced from laser ablates due to different ablation time. In addition, different particle size and size distribution of ZnO-NPs were obtained by different repetition

rates. The significance is in the properties of NPs which are characterized by four different measurements which are UV-Visible Spectrometer, TEM, FTIR and PL.

1.6 Statement of Problem

Based on the reviewed literature, ZnO-NPs are often synthesized using chemical and physical methods that involves toxic chemical reagents, hazardous procedures or toxic and potentially harmful by-products. So, there is a Strong requirement for “green”, economic, commercially possible as well as environmental friendly procedure for synthesis of ZnO-NPs as laser ablation method. On the other hand, lack of investigation about mpact of important factors (time, repetition rate, and media) on the ablation of ZnO-NPs. Therefore, a research gap for doing more investigation on effect of the medium on optical properties and size of ZnO-NPs is on demand. Before this, knowing the effect of laser parameters on fabrication of the new ZnO-NPs need to be found.

1.7 Objective of research

The present study aims to investigate the characteristics of ZnO-NPs produced from PLA from ZnO plate in the aqueous solutions and oil. Furthermore, it attempts to reveal the effects of surrounding molecules on size distribution and defect structure. The change of the liquid environment used in LA provides an easy and flexible method to modify NPs properties. More specifically the study pursues the following objectives:

1. To examine the effect of laser parameters as ablation duration, repetition rate on ZnO-NPs yield.
2. To investigate the medium effect on optical properties and size of ZnO-NPs.

REFERENCES

- Abid, J., Wark, A., Brevet, P., and Girault, H. (2002). Preparation of Silver Nanoparticles in Solution from a Silver Salt by Laser Irradiation. *Chemical Communications*(7), 792-793.
- Amendola, V., and Meneghetti, M. (2009). Laser Ablation Synthesis in Solution and Size Manipulation of Noble Metal Nanoparticles. *Physical chemistry chemical physics*, 11(20), 3805-3821.
- Amini, M. M., Fardad, S., Massudi, R., and Manteghi, A. (2007). *Synthesis, Size and Colloidal Stability of Zno Nanoparticles in Ionic Solutions*. Paper presented at the Nanotechnology, 2007. 7th IEEE Conference on.
- Ashe, B. (2011). *A Detail Investigation to Observe the Effect of Zinc Oxide and Silver Nanoparticles in Biological System*. Master of Technology, National Institute Of Technology Rourkela, Orissa, India.
- Baladi, A., and Sarraf, R. (2010). Investigation of Different Liquid Media and Ablation Times on Pulsed Laser Ablation Synthesis of Aluminum Nanoparticles. *Applied Surface Science*, 256(24), 7559-7564.
- Barcikowski, S., and Compagnini, G. (2013). Advanced Nanoparticle Generation and Excitation by Lasers in Liquids. *Physical Chemistry Chemical Physics*, 15(9), 3022-3026.
- Bäuerle, D. (2011). Nanosecond-Laser Ablation *Laser Processing and Chemistry* (Vol. 4, pp. 237-278): Springer.
- Becker, M. F., Brock, J. R., Cai, H., and Henneke, D. E. (1998). Metal Nanoparticles Generated by Laser Ablation. *Nanostructured Materials*, 10.
- Chakrabarti, S., Das, D., Ganguli, D., and Chaudhuri, S. (2003). Tailoring of Room Temperature Excitonic Luminescence in Sol-Gel Zinc Oxide-Silica Nanocomposite Films. *Thin Solid Films*, 441(1), 228-237.
- Chen, Li, L.-H., Chen, X.-T., Xue, Z., Hong, J.-M., and You, X.-Z. (2003a). Preparation and Characterization of Nanocrystalline Zinc Oxide by a Novel Solvothermal Oxidation Route. *Journal of crystal growth*, 252(1), 184-189.

- Chen, S., Li, L.-H., Chen, X.-T., Xue, Z., Hong, J.-M., and You, X.-Z. (2003b). Preparation and Characterization of Nanocrystalline Zinc Oxide by a Novel Solvothermal Oxidation Route. *Journal of crystal growth*, 252(1), 184-189.
- Chen, S., Liu, Y. C., Lu, Y. M., Zhang, J. Y., Shen, D. Z., and Fan, X. W. (2006). Photoluminescence and Raman Behaviors of ZnO Nanostructures with Different Morphologies. *Journal of crystal growth*, 289(1), 55-58.
- Cho, J., Kyu Song, J., and Min Park, S. (2009). Characterization of ZnO Nanoparticles Grown by Laser Ablation of a Zn Target in Neat Water. *Bulletin of The Korean Chemical Society*, 30(7), 1616-1618.
- Darroudi, M., Ahmad, M. B., Zak, A. K., Zamiri, R., and Hakimi, M. (2011a). Fabrication and Characterization of Gelatin Stabilized Silver Nanoparticles under Uv-Light. *Int J Mol Sci*, 12(12), 6346-6356.
- Darroudi, M., Ahmad, M. B., Zamiri, R., Zak, A. K., Abdullah, A. H., and Ibrahim, N. A. (2011b). Time-Dependent Effect in Green Synthesis of Silver Nanoparticles. *Int J Nanomedicine*, 6, 677-681.
- Faraji, M., Yamini, Y., and Rezaee, M. (2010). Magnetic Nanoparticles: Synthesis, Stabilization, Functionalization, Characterization, and Applications. *Journal of the Iranian Chemical Society*, 7(1), 1-37.
- Garcia, M., and Rodriguez, J. A. (2007). Metal Oxide Nanoparticles *Encyclopedia of Inorganic and Bioinorganic Chemistry*: John Wiley & Sons.
- Garcia, M. A., Merino, J. M., Pinel, E. F. n., Quesada, A., Venta, J. d. l., Gonza'lez, M. L. R. z., . . . Hernando†, A. (2007). Magnetic Properties of ZnO Nanoparticles. *Nano Letters*, 7, 1489-1494.
- Gill, S. K. (2010). *Synthesis and Characterization of Metal-Semiconductor and Metal-Oxide Porous Materials*. Texas Tech University.
- Haase, M., Weller, H., and Henglein, A. (1988). Photochemistry and Radiation Chemistry of Colloidal Semiconductors. 23. Electron Storage on Zinc Oxide Particles and Size Quantization. *The Journal of Physical Chemistry*, 92(2), 482-487.

- Hahn, A., Barcikowski, S., and Chichkov, B. N. (2008). Influences on Nanoparticle Production During Pulsed Laser Ablation. *Pulse*, 40(45), 50.
- Hajiesmaeilbaigi, F., Mohammadalipour, A., Sabbaghzadeh, J., Hoseinkhani, S., and Fallah, H. (2006). Preparation of Silver Nanoparticles by Laser Ablation and Fragmentation in Pure Water. *Laser Physics Letters*, 3(5), 252.
- He, C., Sasaki, T., Usui, H., Shimizu, Y., and Koshizaki, N. (2007). Fabrication of ZnO Nanoparticles by Pulsed Laser Ablation in Aqueous Media and Ph-Dependent Particle Size: An Approach to Study the Mechanism of Enhanced Green Photoluminescence. *Journal of Photochemistry and Photobiology A: Chemistry*, 191(1), 66-73.
- Hirota, K., Sugimoto, M., Kato, M., Tsukagoshi, K., Tanigawa, T., and Sugimoto, H. (2010). Preparation of Zinc Oxide Ceramics with a Sustainable Antibacterial Activity under Dark Conditions. *Ceramics International*, 36(2), 497-506.
- Hong, R., Pan, T., Qian, J., and Li, H. (2006). Synthesis and Surface Modification of ZnO Nanoparticles. *Chemical Engineering Journal*, 119(2-3), 71-81.
- Hsieh, C. (2007). Spherical Zinc Oxide Nano Particles from Zinc Acetate in the Precipitation Method. *Journal of the Chinese Chemical Society*, 54(1), 31-34.
- Ishikawa, Y., Shimizu, Y., Sasaki, T., and Koshizaki, N. (2006). Preparation of Zinc Oxide Nanorods Using Pulsed Laser Ablation in Water Media at High Temperature. *J Colloid Interface Sci*, 300(2), 612-615.
- Ismail, R. A., Ali, A. K., Ismail, M. M., and Hassoon, K. I. (2011). Preparation and Characterization of Colloidal ZnO Nanoparticles Using Nanosecond Laser Ablation in Water. *Applied Nanoscience*, 1(1), 45-49.
- Jalal, R., Goharshadi, E. K., Abareshi, M., Moosavi, M., Yousefi, A., and Nancarrow, P. (2010). ZnO Nanofluids: Green Synthesis, Characterization, and Antibacterial Activity. *Materials Chemistry and Physics*, 121(1-2), 198-201.
- Jamali, F., Joag, D. S., More, M. A., Singh, J., and Srivasatva, O. (2010). Low Temperature Growth of Aligned ZnO Nanowires and Their Application as Field Emission Cathodes. *Materials Chemistry and Physics*, 120(2), 691-696.

- Jeng, H. A., and Swanson, J. (2006). Toxicity of Metal Oxide Nanoparticles in Mammalian Cells. *J Environ Sci Health A Tox Hazard Subst Environ Eng*, 41(12), 2699-2711.
- Kadhim, R. G., Ali, A. K., and Noori, M. F. (2012). Nanofabrication and Characterization of Gold Metal by Pulsed Laser Ablation in Sds Solution. *Advances in Physics Theories and Applications*, 11, 23-27.
- Khorsand, A., Abd Majid, W. H., Darroudi, M., and Yousefi, R. (2011). Synthesis and Characterization of ZnO Nanoparticles Prepared in Gelatin Media. *Materials Letters*, 65(1), 70-73.
- Kumar, P., Kumar, P., Deep, A., and Bharadwaj, L. M. (2013). Synthesis and Conjugation of ZnO Nanoparticles with Bovine Serum Albumin for Biological Applications. *Applied Nanoscience*, 3(2), 141-144.
- Li, W.-J., Shi, E.-W., Zheng, Y.-Q., and Yin, Z.-W. (2001). Hydrothermal Preparation of Nanometer ZnO Powders. *Journal of materials science letters*, 20(15), 1381-1383.
- Ma, H., Williams, P. L., and Diamond, S. A. (2013). Ecotoxicity of Manufactured ZnO Nanoparticles--a Review. *Environmental Pollution*, 172, 76-85.
- Ma, X., Yan, Z., Yuan, B., and Li, B. (2005). The Light-Emitting Properties of Ge Nanocrystals Grown by Pulsed Laser Deposition. *Nanotechnology*, 16(6), 832.
- Mafuné, F., Kohno, J.-y., Takeda, Y., Kondow, T., and Sawabe, H. (2000). Structure and Stability of Silver Nanoparticles in Aqueous Solution Produced by Laser Ablation. *The Journal of Physical Chemistry B*, 104(35), 8333-8337.
- Moezzi, A., McDonagh, A. M., and Cortie, M. B. (2012). Zinc Oxide Particles: Synthesis, Properties and Applications. *Chemical Engineering Journal*, 185-186, 1-22.
- Moosavi, M., Goharshadi, E. K., and Youssefi, A. (2010). Fabrication, Characterization, and Measurement of Some Physicochemical Properties of ZnO Nanofluids. *International Journal of Heat and Fluid Flow*, 31(4), 599-605.

- Oliveira, A. P. A., Hochepped, J.-F., Grillon, F., and Berger, M.-H. (2003). Controlled Precipitation of Zinc Oxide Particles at Room Temperature. *Chemistry of Materials*, 15(16), 3202-3207.
- Pal, U., Serrano, J. G., Santiago, P., Xiong, G., Ucer, K. B., and Williams, R. T. (2006). Synthesis and Optical Properties of ZnO Nanostructures with Different Morphologies. *Optical Materials*, 29(1), 65-69.
- Premanathan, M., Karthikeyan, K., Jeyasubramanian, K., and Manivannan, G. (2011). Selective Toxicity of ZnO Nanoparticles toward Gram-Positive Bacteria and Cancer Cells by Apoptosis through Lipid Peroxidation. *Nanomedicine*, 7(2), 184-192.
- Rakshit, R., and Budhani, R. (2006). Magnetic Relaxation and Superparamagnetism of Non-Interacting Disordered CoPt Nanoparticles. *Journal of Physics D: Applied Physics*, 39(9), 1743.
- Raoufi, D. (2013). Synthesis and Microstructural Properties of ZnO Nanoparticles Prepared by Precipitation Method. *Renewable Energy*, 50, 932-937.
- Rostami, M. R., and Semsarzadeh, M. A. (2008). Study of Macroinitiator Efficiency and Microstructure–Thermal Properties in the Atom Transfer Radical Polymerization of Methyl Methacrylate. *Journal of Polymer Research*, 10.
- Russo, R. E., Mao, X., Liu, H., Gonzalez, J., Samuel, and Mao, S. (2002). Laser Ablation in Analytical Chemistry—a Review. *Talanta*, 57 425–451.
- Sasaki, T., Shimizu, Y., and Koshizaki, N. (2006). Preparation of Metal Oxide-Based Nanomaterials Using Nanosecond Pulsed Laser Ablation in Liquids. *Journal of Photochemistry and Photobiology A: Chemistry*, 182(3), 335-341.
- Semaltianos, N. G. (2010). Nanoparticles by Laser Ablation. *Critical Reviews in Solid State and Materials Sciences*, 35(2), 105-124.
- Shafeev, G. A. (2011). Formation of Nanoparticles under Laser Ablation of Solids in Liquids. In S. E. BLACK (Ed.), *Laser Ablation: Effects and Applications*. New York: Nova Science Publishers, Inc.

- Sibbald, M. S., Chumanov, G., and Cotton, T. M. (1996). Reduction of Cytochrome C by Halide-Modified, Laser-Ablated Silver Colloids. *The Journal of Physical Chemistry*, 100(11), 4672-4678.
- Simakin, A. V., Voronov, V. V., Kirichenko, N. A., and Shafeev, G. A. (2004). Nanoparticles Produced by Laser Ablation of Solids in Liquid Environment. *Applied Physics A*, 79(4-6).
- Singh, A. K. (2010). Synthesis, Characterization, Electrical and Sensing Properties of ZnO Nanoparticles. *Advanced Powder Technology*, 21(6), 609-613.
- Singh, S., and Gopal, R. (2007). Zinc Nanoparticles in Solution by Laser Ablation Technique. *Bulletin of Materials Science*, 30(3), 291-293.
- Singh, S. C., and Gopal, R. (2008). Synthesis of Colloidal Zinc Oxide Nanoparticles by Pulsed Laser Ablation in Aqueous Media. *Physica E: Low-dimensional Systems and Nanostructures*, 40(4), 724-730.
- Sun, D., Wong, M., Sun, L., Li, Y., Miyatake, N., and Sue, H.-J. (2007). Purification and Stabilization of Colloidal ZnO Nanoparticles in Methanol. *Journal of Sol-Gel Science and Technology*, 43(2), 237-243.
- Takeshi, S., Shimizu, Y., and Koshizaki, N. (2006). Preparation of Metal Oxide-Based Nanomaterials Using Nanosecond Pulsed Laser Ablation in Liquids. *Journal of Photochemistry and Photobiology A: Chemistry*, 182(3), 335-341.
- Talam, S., Karumuri, S. R., and Gunnam, N. (2012). Synthesis, Characterization, and Spectroscopic Properties of ZnO Nanoparticles. *ISRN Nanotechnology*, 2012, 1-6.
- Tanaka, H., Fujioka, A., Futoyu, A., Kandori, K., and Ishikawa, T. (2007). Synthesis and Characterization of Layered Zinc Hydroxychlorides. *Journal of Solid State Chemistry*, 180(7), 2061-2066.
- Thareja, R. K., and Shukla, S. (2007). Synthesis and Characterization of Zinc Oxide Nanoparticles by Laser Ablation of Zinc in Liquid. *Applied Surface Science*, 253(22), 8889-8895.

- Tso, C. P., Zhung, C. M., Shih, Y. H., Tseng, Y. M., Wu, S. C., and Doong, R. A. (2010). Stability of Metal Oxide Nanoparticles in Aqueous Solutions. *Water Science and Technology*, 61(1), 127-133.
- Usui, H., Shimizu, Y., Sasaki, T., and Koshizaki, N. (2005). Photoluminescence of ZnO Nanoparticles Prepared by Laser Ablation in Different Surfactant Solutions. *J. Phys. Chem. B* 109, 120-124.
- Villanueva, Y. Y., Liu, D.-R., and Cheng, P. T. (2006). Pulsed Laser Deposition of Zinc Oxide. *Thin Solid Films*, 501(1-2), 366-369.
- Wang, Y., and Herron, N. (1991). Nanometer-Sized Semiconductor Clusters: Materials Synthesis, Quantum Size Effects, and Photophysical Properties. *The Journal of Physical Chemistry*, 95(2), 525-532.
- Wang, Z. G., Zu, X. T., Zhu, S., and Wang, L. M. (2006). Green Luminescence Originates from Surface Defects in ZnO Nanoparticles. *Physica E: Low-dimensional Systems and Nanostructures*, 35(1), 199-202.
- Wang, Z. L. (2004). Zinc Oxide Nanostructures: Growth, Properties and Applications. *Journal of Physics: Condensed Matter*, 16(25), R829-R858.
- Wu, Y. L., Tok, A. I. Y., Boey, F. Y. C., Zeng, X. T., and Zhang, X. H. (2007). Surface Modification of ZnO Nanocrystals. *Applied Surface Science*, 253(12), 5473-5479.
- Yang, G. (2012). *Laser Ablation in Liquids*: Pan Stanford.
- Zamiri, R. (2010). *Fabrication and Optical Characterization of Nanoparticles in Liquid Media*. Doctor of Philosophy, Universiti Putra Malaysia, Universiti Putra Malaysia.
- Zamiri, R., Azmi, B. Z., Sadrolhosseini, A. R., Ahangar, H. A., Zaidan, A. W., and Mahdi, M. A. (2011a). Preparation of Silver Nanoparticles in Virgin Coconut Oil Using Laser Ablation. *International Journal of Nanomedicine*, 6, 71-75.
- Zamiri, R., Zakaria, A., Abbastabar, H., Darroudi, M., Husin, M. S., and Mahdi, M. A. (2011b). Laser-Fabricated Castor Oil-Capped Silver Nanoparticles. *International Journal of Nanomedicine*, 6, 565-568.

- Zamiri, R., Zakaria, A., Ahangar, H. A., Darroudi, M., Zak, A. K., and Drummen, G. P. C. (2012). Aqueous Starch as a Stabilizer in Zinc Oxide Nanoparticle Synthesis Via Laser Ablation. *Journal of Alloys and Compounds*, 516, 41-48.
- Zamiri, R., Zakaria, A., Husin, M. S., Abd Wahab, Z., and Forough Kalaei, N. (2011c). Formation of Silver Microbelt Structures by Laser Irradiation of Silver Nanoparticles in Ethanol. *International Journal of Nanomedicine*, 2221.
- Zeng, H., Du, X. W., Singh, S. C., Kulinich, S. A., Yang, S., He, J., and Cai, W. (2012). Nanomaterials Via Laser Ablation/Irradiation in Liquid: A Review. *Advanced Functional Materials*, 22(7), 1333-1353.
- Zhiyong, F., and Jia, G. (2005). *Zinc Oxide Nanostructures: Synthesis and Properties*. University of California, Irvine.
- Zhou, J., Wang, Y., Zhao, F., Wang, Y., Zhang, Y., and Yang, L. (2006). Photoluminescence of ZnO Nanoparticles Prepared by a Novel Gel-Template Combustion Process. *Journal of Luminescence*, 119-120, 248-252.
- Zhu, Y., Mei, T., Wang, Y., and Qian, Y. (2011). Formation and Morphology Control of Nanoparticles Via Solution Routes in an Autoclave. *Journal of Materials Chemistry*, 21(31), 11457.