



UNIVERSITI PUTRA MALAYSIA

***FORMATION AND REDUCTION OF 5-HYDROXYMETHYLFURFURAL
DURING DEEP-FAT FRYING OF BANANAS***

PARVIZ KAVOUSI

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**FORMATION AND REDUCTION OF 5-HYDROXYMETHYLFURFURAL
DURING DEEP-FAT FRYING OF BANANAS**

By

PARVIZ KAVOUSI

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in
Fulfillment of the Requirements for the Degree of Doctor of Philosophy**

November 2014

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DEDICATION

To my beloved parents

To My beloved children

Pouneh and Farshad Kavousi
Thank you for your patience, support and understanding

In memory of my beloved wife

Parvin Aslanbeigui, M.D. 1959-2007

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

FORMATION AND REDUCTION OF 5-HYDROXYMETHYLFURFURAL DURING DEEP-FAT FRYING OF BANANAS

By

PARVIZ KAVOUSI

November 2014

Chairman: Associate Professor Abdul Azis Ariffin, PhD
Faculty: Food Science and Technology

5-hydroxymethylfurfural (HMF) is a heat-induced furanic compound that is formed through Maillard reaction or direct dehydration of hexoses under acidic conditions in carbohydrate-containing food products. Human exposure to HMF mainly takes place through diet. 5-sulfoxymethylfurfural (SMF), as a metabolite of HMF, can react with DNA and other macromolecules, and acts as a toxic and mutagenic compound. This gives rise to concern for reducing the HMF content in thermally processed foods especially fried food products. Up to now the formation and reduction of HMF during deep-fat frying is not clear. This study is conducted to investigate the formation and reduction of HMF during deep-fat frying of banana. The effects of amino acids (15 amino acids) and sugars (fructose, glucose and sucrose) as precursors on the formation and reduction of HMF are evaluated in model systems. The HMF is extracted and separated using an optimized extraction procedure and a validated HPLC method. The results reveal that glutamine, glutamic and aspartic acids enhance the formation of HMF. Conversely, lysine, arginine and histidine reduce the HMF content to non-detectable level. The HMF content in control samples containing glucose, fructose or sucrose alone was 167.99, 232.79 and 18.54 mg/kg, respectively. While, in binary mixtures of these sugars with glutamine, aspartic acid and glutamic acid, the formation of HMF was significantly ($P < 0.05$) increased and reached to 1084.60, 2020.30 and 3816.90 mg/kg, 1115.40, 751.00 and 1232.60 and 1653.30, 1556.60 and 3132.60 mg/kg, respectively ($P < 0.05$). The formation of HMF is significantly ($P < 0.05$) reduced by 73.23% and 90.71% in Abu variety and 76.33% and 82.84% in Tanduk variety, respectively after pre-soaking of bananas in water containing lysine or arginine (0.06 M) for 2 h compared to control samples. The current study reveals that the food compositions have had a more significant effect than frying oil in the formation of HMF ($P < 0.05$). The effect of frying time on HMF formation is significantly ($P < 0.05$) higher than frying temperature. Microwave frying causes the higher HMF formation than conventional deep-fat frying and boiling process. The HMF content in bananas after 6 min cooking was 2168.02,

31.92 and 0.17 mg/kg, respectively. In addition, the effects of the banana variety (7 varieties), ripening stage (ripe and unripe), surface area-to-volume ratio and blanching process on the reduction of HMF are studied. The results show that HMF forms at significantly lower amount in unripe fruits ($P < 0.05$). HMF content in fried Tanduk is lower than Mas, Abu, Berangan, Raja, Ambon and Awak varieties. Moreover, HMF formation is significantly ($P < 0.05$) increased from 26.81 to 124.83 mg/kg by increasing the surface area-to-volume ration from 0.02 mm^{-1} to 0.80 mm^{-1} ($P < 0.05$) The results reveal that the HMF formation is significantly reduced after blanching of bananas at $70 \text{ }^\circ\text{C}$ for 45 min by 80.00% and 85.35% in Abu and Tanduk varieties, respectively ($P < 0.05$).



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PEMBENTUKAN DAN PENGURANGAN 5-HIDROKSIMETHILFURFURAL SEMASA PENGGORENGAN PISANG MINYAK PENUHS

Oleh

PARVIZ KAVOUSI

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Pengerusi : Professor Madya Abdul Azis Ariffin, PhD
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5- hidroksimetilfurfural (HMF) adalah sebatian furanik teraruh haba yang terbentuk melalui tindak balas Maillard atau dehidrasi langsung heksosa di bawah keadaan berasid dalam produk makanan yang mengandungi karbohidrat. Pendedahan manusia kepada HMF berlaku terutamanya melalui pemakanan. 5- sulfoksimetilfurfural (SMF) merupakan metabolit daripada HMF, boleh bertindak balas dengan DNA dan makromolekul lain, dan bertindak sebagai sebatian toksik dan mutagen . Ini menimbulkan kebimbangan untuk mengurangkan kandungan HMF dalam makanan diproses haba terutamanya produk makanan bergoreng. Sehingga kini pembentukan dan pengurangan HMF sewaktu penggorengan minyak penuh adalah tidak jelas. Kajian ini dijalankan untuk mengkaji pembentukan dan pengurangan HMF sewaktu penggorengan pisang minyak penuh. Kesan asid amino (15 asid amino) dan gula (fruktosa, glukosa dan sukrosa) sebagai prekursor bagi pembentukan dan pengurangan HMF dinilai dalam sistem model. HMF diekstrak dan diasingkan menggunakan proses pengekstrakan optima dan kaedah HPLC yang diperakui. Dapatan kajian menunjukkan bahawa glutamina, asid aspartik dan asid glutamik meningkatkan pembentukan HMF. Lysina, arginina dan histidina sebaliknya mengurangkan kandungan HMF ke paras yang tidak dapat diukur. Kandungan HMF dalam sampel kawalan mengandungi glukosa, fruktosa dan sukrosa masing-masing adalah 167.99, 232.79 dan 18.54 mg/kg. Manakala, dalam campuran dedua gula berkenaan dengan glutamina, asid aspartik dan asid glutamik, pembentukan HMF meningkat dengan ($P < 0.05$) nyata sekali dan masing-masing mencapai 1084.60, 2020.30 dan 3816.90 mg/kg, 1115.40, 751.00 dan 1232.60 mg/kg serta 1653.30, 1556.60 dan 3132.60 mg/kg. Pembentukan HMF menurun ($P < 0.05$) dengan nyata sekali ke 73.23% and 90.71% dalam variasi Abu dan 76.33% dan 82.84% dalam variasi Tanduk, masing-masing selepas pra rendaman pisang di dalam air mengandungi lysina atau arginina (0.06M) selama 2 jam dibandingkan dengan sampel kawalan. Kajian semasa menunjukkan bahawa komposisi makanan mempunyai kesan yang nyata berbanding minyak penggorengan dalam pembentukan HMF ($P < 0.05$). Kesan masa penggorengan terhadap pembentukan HMF adalah nyata sekali ($P < 0.05$) lebih tinggi berbanding suhu penggorengan. Penggorengan gelombang mikro menyebabkan pembentukan HMF lebih tinggi berbanding penggorengan minyak penuh konvensional dan proses pendidihan. Kandungan HMF dalam pisang selepas memasak selama 6 min

adalah masing-masing 2168.02, 31.92 dan 0.17 mg/kg. Di samping itu, kesan variasi pisang (7 variasi), tahap keranuman (ranum atau mengkal), nisbah luas permukaan ke isipadu dan proses celuran ke atas pengurangan HMF turut dikaji. Keputusan menunjukkan bahawa HMF terbentuk pada amaun yang nyata sekali ($P < 0.05$) lebih rendah dalam buah mengkal. Kandungan HMF dalam pisang Tanduk goreng adalah lebih rendah dari variasi Mas, Abu, Berangan, Raja, Ambon dan Awak. Lagi pula, pembentukan HMF meningkat secara nyata sekali ($P < 0.05$) dari 26.81 ke 124.83 mg/kg dengan meningkatkan nisbah luas permukaan ke isipadu dari 0.02 mm^{-1} ke 0.80 mm^{-1} ($P < 0.05$). Hasil dapatan menunjukkan bahawa pembentukan HMF adalah berkurangan dengan nyata ($P < 0.05$) selepas penceluran pisang pada 70°C selama 45 min masing-masing 80.00% dan 85.35% dalam variasi Abu dan Tanduk.



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I certify that a Thesis Examination Committee has met on 6 November 2014 to conduct the final examination of Parviz Kavousi on his thesis entitled "Formation and Reduction of 5-Hydroxymethylfurfural during Deep-Fat Frying of Bananas" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSRTRAK	iii
AKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xv
LIST OF FIGURES	xix
LIST OF ABBREVIATIONS	xxiv
CHAPTER	
1 INTRODUCTION	1
2 LITERATURE REVIEW	4
2.1 Chemistry and Occurrence of 5-Hydroxymethylfurfural (HMF)	4
2.1.1 Chemical Properties	4
2.1.2 Chemical Reactions of HMF	4
2.1.3 Presence of HMF in Foodstuffs	5
2.1.3.1 Presence of HMF in Roasted Products	6
2.1.3.2 Presence of HMF in Cereal products	6
2.1.3.3 Presence of HMF in Fruit and Vegetable Products	7
2.1.3.4 Presence of HMF in Honey	8
2.1.3.5 Presence of HMF in Fried products	8
2.2 Safety Issues of HMF	9
2.2.1 Absorption, Distribution, Metabolism and Excretion	9
2.2.2 Toxicity and Carcinogenicity of HMF	12
2.3 Pathways of HMF Formation	13
2.3.1 Direct Dehydration of Sugars Under Acidic Conditions	13
2.3.2 Caramelization	16
2.3.3 Maillard Reaction	16
2.4 Factors Affecting the Formation of HMF	19
2.4.1 Effect of Temperature and Heating Time on HMF Formation	19
2.4.2 Effect of pH on HMF Formation	20
2.5 Factors Affecting the Reduction of HMF	20
2.5.1 Effect of Row Material	20
2.5.2 Effect of Sugar as Main precursor of HMF	23

2.5.3	Effect of Processing Conditions	23
3	MATERIALS AND METHODS	24
3.1	Materials	24
3.1.1	Chemicals	24
3.1.2	Raw Materials	24
3.2	Experimental Procedures	24
3.2.1	Preparation of Model Systems	25
3.2.1.1	Heating Procedure	25
3.2.1.2	Effect of Moisture on HMF Formation	25
3.2.1.3	Effect of pH on the Formation and Reduction of HMF	25
3.2.1.4	Effect of Heating Time on the Formation and Reduction of HMF	27
3.2.1.5	Effect of Sugars and Amino Acids on HMF Formation	28
3.2.1.6	Effect of Amino Acids on Hydrolysis of Sucrose	29
3.2.1.7	Effect of Amino Acid on Reduction of HMF	29
3.2.1.8	Effect of Concentration of Lysine and Arginine on Reduction of HMF	29
3.2.2	Effect of Pretreatment of Banana with Amino Acids on Reduction of HMF	30
3.2.3	Effect of Food Matrices, Frying Media and Oil Degradation on HMF Formation	31
3.2.4	Effect of Cooking Methods on HMF Formation	31
3.2.5	Kinetic Study of HMF Formation during Deep-fat Frying of Banana	32
3.2.6	Effect of Banana Variety and Ripening Stage on HMF Formation	32
3.2.7	Effect of Initial Moisture Content of Banana on HMF Formation	32
3.2.8	Effect of Surface Area-to-Volume Ratio of Banana on HMF Formation	32
3.2.9	Effect of Pre-soaking of Banana on HMF Formation	33
3.2.10	Effect of Blanching of Banana on HMF Formation	33
3.3	Analytical Procedures	34
3.3.1	Determination of Amino Acid Compositions	34
3.3.2	Quantitative Analysis of HMF	36
3.3.3	Sugar Analysis	37
3.3.4	Determination of Fatty Acid Compositions	38
3.3.5	Measurement of Peroxide Value	39
3.3.6	Determination of p-Anisidine Value	39
3.3.7	Free Fatty Acid (FFA) Analysis	40

3.3.8	Determination of Oil Content	40
3.3.9	Moisture Content Analysis	41
3.3.10	Determination of pH	41
3.3.11	Color Measurement	41
3.3	Experimental Design and Data Analysis	41
4	RESULTS AND DISCUSSION	43
4.1	Optimization and Validation of an HPLC method for Determination of HMF	43
4.1.1	Method Modification	43
4.1.2	Extraction Process	45
4.1.2.1	Preparation of Stripped Oil	45
4.1.2.2	Optimization of Extraction Process	46
4.1.3	Method Validation	50
4.1.3.1	Accuracy and Precision	51
4.1.3.2	Linearity	53
4.1.3.3	Limit of Detection (LOD) and Limit of Quantitation (LOQ)	54
4.1.3.4	Specificity	54
4.1.4	Application of the Validated Method	54
4.2	Formation and Reduction of 5-Hydroxymethylfurfural at Frying Temperature in Model System as a Function of Amino Acid and Sugar Composition	57
4.2.1	Formation of HMF in the Absence of Amino Acids	57
4.2.1.1	Effect of Moisture on the Formation of HMF in the Absence of amino Acids	57
4.2.1.2	Effect of pH on HMF Formation in the Absence of Amino Acids	58
4.2.1.3	Effect of Heating Time on HMF Formation in the Absence of Amino Acids	59
4.2.2	Formation of HMF in the Presence of Amino Acids	61
4.2.2.1	Effect of Amino Acids on HMF Formation from Glucose	62
4.2.2.2	Effect Amino acids on HMF Formation from Fructose	63
4.2.2.3	Effect Amino Acids on HMF Formation from Sucrose	63
4.2.2.4	Effect of pH on the Formation of HMF from Fructose in the Presence of Glutamic Acid	67
4.2.2.5	Effect of Heating Time on the Formation of HMF in the Presence of Glutamic Acid	69
4.2.3	Reduction of HMF at Deep-fat Frying Temperature	72

4.2.3.1	Hydrolysis of HMF to Levulinic Acid	72
4.2.3.2	Effect of Amino Acids on the Reduction of HMF in Model System	74
4.2.3.3	Effect of pH on the Reduction of HMF in the Presence of Lysine and Arginine	76
4.2.3.4	Effect of Heating Time on of reduction of HMF in the Presence of Lysine and Arginine	77
4.2.3.5	Effect of Arginine and Iysine Concentrations on the Reduction of HMF	78
4.2.3.6	Application of Amino Acids to Reduce the Formation of HMF during Deep-Fat Frying of Bananas	78
4.3	Factors Affecting the Formation of HMF during Deep-fat Frying	82
4.3.1	Effect of Different Food Matrices, Frying Media, and Oil Degradation on Formation of HMF during Deep-Fat Frying	82
4.3.1.1	Effect of Food Type on HMF Formation	82
4.3.1.2	Effect of Oil Type on HMF Formation	84
4.3.1.3	Effect of Oil Degradation on HMF Formation	86
4.3.1.4	Accumulation of HMF in Frying Oil	92
4.3.2	Effect of Cooking Methods on HMF and Its Precursors in Banana	95
4.3.2.1	Effect of Cooking Methods on Banana Compositions	95
4.3.2.1	Effect of Cooking Methods on HMF Formation	103
4.3.3	Kinetics Study of HMF Formation during Deep-fat Frying of Banana	107
4.3.3.1	Effect of Frying Time and Temperature on HMF Formation	107
4.3.3.2	Relationship between Moisture Loss and HMF Formation	112
4.3.3.3	Relationship between sugar Loss and HMF Formation	113
4.4	Factors Affecting the Reduction of HMF during Deep-Fat Frying of Banana	117
4.4.1	Effect of Banana Varieties and Ripening Stages on HMF Formation	117
4.4.2	Effect of Moisture Content on HMF Formation	126
4.4.3	Effect of Surface Area-to-Volume Ratio on HMF Formation	131
4.4.4	Effect of Pre-Soaking of Banana Slices on HMF Formation	132
4.4.5	Effect of Blanching on HMF Formation	141

5	SUMMARY, CONCLUSION AND RECOMMENDATION	149
	REFERENCES	152
	APPENDICES	174
	BIODATA OF STUDENT	180
	PUBLICATIONS	181



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LIST OF TABLES

Table		Page
2.1	Common banana cultivars in Malaysia	22
3.1	Surface area-to-volume ration of banana discs (Tanduk variety)	33
3.2	Gradient elution of mobile phase A and B for separation of amino acids	35
3.3	Relative response factor, molar mass and concentration of individual amino acid	36
4.1	Design of experiment (general full factorial design) and corresponding experimental data	47
4.2	General linear model analysis (p-value and F-ratio) for the independent variables and their interactions	48
4.3	Accuracy and precision of the method for determination of 5-hydroxymethylfurfural (HMF) in oil	52
4.4	Accuracy and precision of the method for determination of 5-hydroxymethylfurfural (HMF) in fried banana	53
4.5	Results of the linearity study for 5-Hydroxymethylfurfural (HMF)	53
4.6	Limit of detection (LOD) and limit of quantification (LOQ) for 5-Hydroxymethylfurfural (HMF)	54
4.7	Levels of 5-hydroxymethylfurfural (HMF) in crude palm oil (CPO) and fried food samples	55
4.8	Formation of HMF in model systems at frying temperature (175 ± 5 °C) in the absence and presence of water	57
4.9	Effect of pH on the formation of HMF from fructose at frying temperature (175 ± 5 °C) in the absence of amino acids	59
4.10	The influence of heating time on HMF yield, conversion of fructose and selectivity of HMF at frying temperature (175 ± 5 °C) in the absence of amino acids	61
4.11	Effect of amino acids on the formation of 5-hydroxymethylfurfural from glucose in model systems at frying temperature (175 ± 5 °C)	62

4.12	Effect of amino acids on the formation of 5-hydroxymethylfurfural (HMF) from fructose in model systems at frying temperature (175±5 °C)	63
4.13	Effect of amino acids on the formation of 5-hydroxymethylfurfural (HMF) from sucrose in model systems at frying temperature (175±5 °C)	64
4.14	Effect of amino acids on the hydrolysis of sucrose in aqueous solution after heating at 80 °C for 30 min	67
4.15	Effect of pH on the formation of 5-hydroxymethylfurfural (HMF) from fructose in the presence of glutamic acid at frying temperature (175±5 °C)	68
4.16	The influence of heating time on the formation of 5-hydroxymethylfurfural (HMF) and reduction of fructose at frying temperature (175±5 °C) in the presence of glutamic acid	69
4.17	Effect of amino acids on the reduction of 5-hydroxymethylfurfural (HMF) after heating at 80 °C in model systems	74
4.18	Effect of heating time on the reduction of 5-hydroxymethylfurfural (HMF) at frying temperature (175±5 °C) in model systems	77
4.19	General linear model analysis (p-value and F-ratio) for the effect independent variables (soaking time, amino acid concentration and banana variety) and their interactions	79
4.20	Effect of lysine and arginine on 5-hydroxymethylfurfural (HMF) content in fried banana slices (Abu and Tanduk varieties) after soaking for 0.5, 1 and 2 hours	80
4.21	Amino acids, sugar and moisture contents of fresh and fried banana, potato and fish and their reduction after frying represented as mean (standard deviation) from two individual replications	83
4.22	Fatty acid compositions and quality characteristics of fresh oils represented as mean from two individual replications	85
4.23	Effect of frying oils on 5-hydroxymethylfurfural (HMF) formation during deep-fat frying of banana, potato, and fish	86
4.24	Pearson correlation coefficient (r) between 5-hydroxymethylfurfural (HMF) accumulation in fried foods oil quality characteristics during 20 h frying of banana and potato	92

4.25	Amino acids, sugar and moisture contents of fresh banana (Abu variety)	95
4.26	Effect of cooking methods and time on sugar contents of banana (Abu variety)	97
4.27	Effect of cooking methods and time on amino acids of banana (Abu variety)	98
4.28	Effect of cooking method and time on the formation of 5-hydroxymethylfurfural (HMF) in banana (Abu variety)	103
4.29	Pearson correlation (r) coefficients between HMF, moisture and sugar during microwave frying and conventional deep-fat frying of banana (Abu variety)	104
4.30	Peel color of seven banana varieties at two unripe and ripe stages of maturity	118
4.31	Amino acid compositions of seven fresh banana varieties at two ripening stages (unripe and ripe)	119
4.32	Amino acid compositions of banana varieties at two ripening stages (unripe and ripe) after frying	120
4.33	Change in pH values of banana varieties as a function of ripening stage	122
4.34	Change in moisture content of fresh banana varieties at two ripening stages during deep-fat frying	123
4.35	Effect of ripening stage and deep-fat frying on sugar content in banana varieties	124
4.36	Reduction in sugar contents in unripe and ripe banana varieties after deep-fat frying	125
4.37	Changes in moisture content of banana (Abu variety) as a function of drying time at 60 °C	126
4.38	Changes in the formation of 5-hydroxymethylfurfural (HMF) during frying of fresh and pre-dried banana (Abu variety)	128
4.39	Oil uptake in fresh and pre-dried banana slices during deep-fat frying	131
4.40	Effect of pre-soaking of bananas (Abu and Tanduk varieties) in water on amino acid content during deep-fat frying	134

4.41	Reduction in sugar contents in fried banana (Abu and Tandul varieties) slices as a function of soaking time compared to control samples	136
4.42	Changes in the pH of banana slices (Abu and Tanduk varieties) as a function of pre-soaking time in citric acid solutions (0.5 and 1%)	138
4.43	Effect of pre-soaking in citric acid solutions (0.5 and 1%) on the amino acid content of fried banana (Abu and Tanduk varieties) slices compared to fresh samples	140
4.44	Reduction in sugar contents in fried banana (Abu and Tanduk varieties) slices pre-soaked in citric acid solutions (0.5 and 1%) as a function of soaking time compared to control samples	141
4.45	Reduction of sugar contents in fried bananas (Abu and Tanduk varieties) compared to the control samples as a function of blanching treatments	142
4.46	Effect of blanching on the amino acid content of fried bananas (Tanduk variety)	144
4.47	Effect of blanching on the amino acid content of fried bananas (Abu variety)	145
4.48	Effect of blanching on the formation of 5-hydroxymethylfurfural (HMF) in bananas (Abu and Tanduk varieties) during deep-fat frying	146
4.49	Change in the pH values of fresh bananas (Abu and Tanduk varieties) as a function of blanching treatments	147

LIST OF FIGURES

Figure		Page
2.1	Chemical structure of 5-hydroxymethyl-2-furaldehyde (HMF)	4
2.2	Chemical structure of levulinic acid	5
2.3	5-Hdroxymethylfurfural (HMF) biotransformation pathway	10
2.4	Biotransformation of 5-hydroxymethylfurfural (HMF) into 5-sulfoxymethylfurfural (SMF) in the presence of sulfotransferases (SULT)	11
2.5	Acyclic pathway in the dehydration of hexoses to 5-hydroxymethylfurfural (HMF)	14
2.6	Cyclic pathway in the dehydration of fructose to 5-hydroxymethylfurfural (HMF)	15
2.7	Condensation of carbonyl group of sugar with amino group in the first stage of Maillard reaction	17
2.8	Amadori rearrangements leading to the Amadori compound, the N-substituted 1-amino-2-deoxy-2-ketose	17
2.9	Maillard reaction	18
3.1	Flow diagram of experimental work	25
4.1	RP-HPLC chromatogram of 5-Hydroxymethylfurfural (HMF) and furfural standards	44
4.2	RP-HPLC chromatogram of stripped oil	46
4.3	Main effects plot for the peak area of 5-hydroxymethylfurfural (HMF)	48
4.4	Surface plot of 5-hydroxymethylfurfural (HMF) versus solvent composition and mixing time	49
4.5	Surface plot of 5-hydroxymethylfurfural (HMF) versus solvent composition and extraction replications	49
4.6	Surface plot of 5-hydroxymethylfurfural (HMF) versus extraction replications and mixing time	50

4.7	RP-HPLC chromatograms for 5-hydroxymethylfurfural (HMF) in crude palm oil and fried banana	55
4.8	Effect of heating time on the formation of 5-hydroxymethylfurfural (HMF) and reduction of fructose in model systems at frying temperature (175 ± 5 °C) in the absence of amino acids	60
4.9	Effect of pH on the formation of 5-hydroxymethylfurfural (HMF) from fructose in the presence and absence of glutamic acid at frying temperature (175 ± 5 °C) in model systems	68
4.10	The reduction of fructose as a function of heating time at frying temperature (175 ± 5 °C) in the presence and absence of glutamic acid in model systems	71
4.11	The formation of 5-hydroxymethylfurfural (HMF) from fructose as a function of heating time at frying temperature (175 ± 5 °C) in the presence and absence of glutamic acid in model systems	71
4.12	RP-HPLC chromatogram of 5-hydroxymethylfurfural (HMF) and levulinic acid	72
4.13	Hydrolysis of 5-hydroxymethylfurfural (HMF) into levulinic acid at frying temperature (175 ± 5 °C) in binary mixtures of amino acid and sugar	73
4.14	Effect of lysine and arginine on the reduction of 5-hydroxymethylfurfural (HMF) at frying temperature (175 ± 5 °C)	75
4.15	HPLC overlay chromatograms for evaluating the effect of pH (4 to 10) on the reduction of 5-hydroxymethylfurfural (HMF) at frying temperature (175 ± 5 °C) in model system containing fructose with lysine (a) and arginine (b)	76
4.16	Effect of lysine and arginine concentrations on the reduction of 5-hydroxymethylfurfural (HMF) at frying temperature (175 ± 5 °C) in model systems	78
4.17	Effect of soaking of banana slices (Abu and Tanduk varieties) in water on the formation of 5-hydroxymethylfurfural (HMF) compared to unsoaked bananas	79
4.18	Effect of food type on the formation of 5-hydroxymethylfurfural (HMF) during deep-fat frying	82

4.19	Changes in peroxide and p-anisidine values of frying oils during frying banana, potato and fish for 20 h	87
4.20	Changes in Totox value and free fatty acid content of frying oils during frying banana, potato and fish for 20 h	89
4.21	Accumulation of 5-hydroxymethylfurfural (HMF) in fried potato and banana during 20 h frying in palm olein, sunflower and olive oils	90
4.22	Changes in 5-hydroxymethylfurfural (HMF) to levulinic acid peak area ratios as a function of frying time	91
4.23	Accumulation of 5-hydroxymethylfurfural (HMF) in oils during frying of the banana	93
4.24	Accumulation and elimination of 5-hydroxymethylfurfural (HMF) in oil at frying temperature (175 ± 5 °C)	93
4.25	Changes in moisture content of banana (Abu variety) as a function of cooking time during microwave frying, deep-fat frying, boiling and steaming	102
4.26	Relationship between 5-hydroxymethylfurfural (HMF) formation and moisture loss in banana (Abu variety) as a function of frying time in the microwave and conventional deep-fat frying	105
4.27	Relationship between 5-hydroxymethylfurfural (HMF) formation and sugar loss in banana (Abu variety) as a function of frying time in the microwave frying	106
4.28	Formation of levulinic acid (a), and hydrolysis of 5-hydroxymethylfurfural (HMF) to levulinic acid (b) as a function of time during microwave frying of banana (Abu variety)	106
4.29	Effect of frying time and temperature on the formation of 5-hydroxymethylfurfural (HMF) in banana (Abu variety)	107
4.30	Formation of 5-hydroxymethylfurfural (HMF) as a function of frying temperature at different frying time during deep-fat frying of banana (Abu variety)	109
4.31	Interaction of frying time and temperature on the formation of 5-hydroxymethylfurfural (HMF) during deep-fat frying of banana (Abu variety)	110

4.32	Formation of levulinic acid as a function of frying time at different frying temperatures during deep-fat frying of banana (Abu variety)	111
4.33	Interaction of frying time and temperature on the formation of levulinic acid during deep-fat frying of banana (Abu variety)	111
4.34	Changes in moisture content as a function of frying temperature and time during deep-fat frying of banana (Abu variety)	112
4.35	Formation of 5-hydroxymethylfurfural (HMF) as a function of moisture loss at different frying temperatures during deep-fat frying of banana (Abu variety)	113
4.36	Changes in sugar content as a function of frying temperature and time during deep-fat frying of banana (Abu variety)	114
4.37	Relationship between of 5-hydroxymethylfurfural (HMF) formation) and reduction in sugar content at different frying time and temperatures during deep-fat frying of banana (Abu variety)	115
4.38	Reduction in total amino acid contents after frying of banana slices (Tanduk, Abu, Awak, Mas, Berangan, Ambon and Raja varieties) at two ripening stages	121
4.39	Formation of 5-hydroxymethylfurfural (HMF) during deep-fat frying of banana (Tanduk, Abu, Awak, Mas, Berangan, Ambon and Raja) varieties at two ripening stages	126
4.40	Changes in moisture content during deep-fat frying of banana (Abu variety) as a function of drying time	127
4.41	Effect of moisture content of banana (Abu variety) prior to frying on the formation of 5-hydroxymethylfurfural (HMF) during deep-fat frying	128
4.42	Relationship between reduction of moisture in banana before frying and formation of 5-hydroxymethylfurfural (HMF) during deep-fat frying	129
4.43	Relationship between reduction of moisture in banana before frying and hydrolysis of 5-hydroxymethylfurfural (HMF) to levulinic acid during deep-fat frying	130
4.44	Formation of 5-hydroxymethylfurfural (HMF) in fried bananas (Tanduk variety) as a function of surface area-to-volume ratio	131

4.45	Changes in moisture content in fried bananas (Tanduk variety) as a function of surface area-to-volume ratio	132
4.46	Changes in sugar contents in fried bananas (Tanduk variety) as a function of surface area-to-volume ratio	133
4.47	Effect of pre-soaking on the sugar content of fried banana (Abu and Tanduk varieties) slices compared to fresh and control samples	135
4.48	Effect of pre-soaking of bananas (Abu and Tanduk varieties) in water for 1 and 2 hours on the formation of 5-hydroxymethylfurfural (HMF)	137
4.49	Changes in moisture content of the pre-soaked banana (Abu and Tanduk varieties) slices during frying	137
4.50	Effect of pre-soaking of banana slices (Abu and Tanduk varieties) in citric acid solutions (0.5 and 1%) on 5-hydroxymethylfurfural (HMF) formation during frying	139
4.51	Changes in sugar contents of fried bananas (Abu and Tanduk varieties) as a function of blanching time and temperature	143
4.52	Change in moisture content of fried bananas (Abu and Tanduk varieties) as a function of blanching treatments	147

LIST OF ABBREVIATIONS

%	Percentage
/	Per
<	Less than
>	Higher than
ANOVA	Analysis of Variance
AOAC	Association of Official Analytical Chemist
AOCS	American Oil Chemists' Society
dw	Dried weight
et al.	And others
g	Gram
h	Hour
H ⁺	Hydrogen ion
i.d.	Internal diameter
IOOC	International Olive Oil Council
L	Liter
M	Molar
mg	Milligram
min	Minute
mL	Milliliter
°C	Degrees Celsius
p	Probability
pH	Hydrogen ion concentration
ppm	Parts per million
rpm	Rotation per minute
s	Second
SD	Standard Deviation
<i>SPP.</i>	species
UV	Ultra Violet ray
v/v	Volume per volume
α	Alpha
β	Beta
μ L	Micro liter

CHAPTER 1

INTRODUCTION

Deep-fat frying, as a popular method for food preparation, is widely used in homes, restaurants and food industries. It is well-known for producing the desirable flavour, colour and texture in fried food products. In deep-fat frying, food is immersed in frying oil at high temperature (150 to 190°C) for a specified time, and the oil acts as a heat transfer medium (Choe & Min, 2007). Although, using high temperature leads to the formation of desirable surface colour and textural characteristics of fried foods, heat-induced toxicants such as HMF are also produced in carbohydrate-containing foods during deep-fat frying (Capuano & Fogliano, 2011). Deep-fat fried snack foods such as French fries, fried seafood, egg rolls, doughnuts, fried chicken and chips (potato, corn and tortilla) are very popular worldwide.

5-Hydroxymethylfurfural (HMF) as a heat-induced furanic compound is formed by dehydration of hexoses through Maillard reaction, caramelization and direct dehydration of hexoses under acidic conditions (Capuano & Fogliano, 2011). Although, hexoses are the forerunner of HMF, sugars such as sucrose that release hexoses after hydrolysis can convert into HMF. The hydrolysis of sucrose during processing of food products leads to formation of fructofuranosyl cation and glucose. This reactive cation can directly convert into HMF in dry conditions (Perez Locas & Yaylayan, 2008) or convert into fructose in the presence of water or contribute in formation of HMF through the reaction with a free amino group during Maillard reaction. Fructose and glucose are the main precursors of HMF. They can react with amino acids in the early stage of Maillard reaction and form corresponding N-substituted amine (Hodge, 1953). After the formation of Amadori rearrangement products, the degradation pathway is pH-dependent. HMF forms from dehydration of sugar through 1, 2-enolization pathway at pH values lower than 7 (Anese & Suman, 2013). Moreover, HMF can be formed by dehydration of sugars through 1, 2-enolization pathway under acidic condition.

The formation of HMF is affected by food composition and processing conditions. Generally, besides the sugars as the main precursor of HMF, amino acid composition, temperature, pH and heating time are influential factors (Anese & Suman, 2013). Thus, the concentration of HMF in food products varies as a function of the dominant factors. To reduce the formation of HMF during processing of foods such as deep-fat frying, two preventive and removal strategies can be applied (Anese & Suman, 2013). The changes in processing conditions such as temperature and heating time, and reduction of HMF precursors prior to processing by pre-treatment of the foods are regarded as preventive strategy. However, the removal strategy deals with reduction of HMF after formation.

HMF has been used as an indicator to determine the extent of heat load applied to foods during their processing or as a quality marker for many products such as honey (Khalil, Sulaiman, & Gan, 2010). The presence of HMF has been reported in fruit juices (Lee, Sakai, Manaf, Rodhi, & Saad, 2014), dried fruits (Murkovic & Pichler, 2006), coffee

(Bignardi, Cavazza, & Corradini, 2014), bakery products (Petisca, Henriques, Pérez-Palacios, Pinho, & Ferreira, 2014), fried foods (Göncüoğlu & Gökmen, 2013) and infant formulae (Chávez-Servín, de la Torre Carbot, García-Gasca, Castellote, & López-Sabater, 2015). Human exposure to HMF mainly takes place through the diet, and based on the data reported in literature, its dietary intake is often higher than other food toxicants such as acrylamide (Capuano & Fogliano, 2011).

The risk associated with HMF can be induced by HMF itself or its metabolites. Despite controversial reports that are derived from *in vitro* studies on the mutagenicity and genotoxicity of HMF (Severin, Dumont, Jondeau-Cabaton, Graillot, & Chagnon, 2012), the influence of HMF on human health is not clear (Capuano & Fogliano, 2011). The major concern for HMF is related to its bioconversion *in vitro* into 5-sulfoxymethylfurfural (SMF) through sulfonation of its allylic hydroxyl functional group, catalyzed by sulfotransferases (SLUTs) (Abraham et al., 2011). The latter compound can react with DNA and other macromolecules, thereby resulting in toxic and mutagenic effects (Surh & Tannenbaum, 1994). The European Food Safety Authority (EFSA) has confirmed the mutagenicity of SMF (EFSA, 2005). A recent study showed that SMF could be detected in the blood of FVB/N mice after HMF intravenous administration (Monien, Frank, Seidel, & Glatt, 2009). Owing to the higher activity of human SLUTs compared to those expressed by rodents (Bauer-Marinovic, Taugner, Florian, & Glatt, 2012), the risk associated with HMF in human may be higher than in rodents. Thus, due to a potentially high HMF intake from the food, even a limited conversion can negatively affect the human health.

As mentioned earlier, the formation of HMF mainly depends on sugar composition. The foods containing hexose (such as banana) are more susceptible to have HMF after deep fat frying. Deep-fat fried banana is also a very popular snack food in many countries especially in Malaysia. Banana is a general term embracing a number of species or hybrids in the genus *Musa* of the family Musaceae (Happi Emaga, Andrianaivo, Wathelet, Tchango, & Paquot, 2007). Generally, there are two broad groups of banana namely dessert bananas, and cooking banana or plantain. The dessert bananas are sweeter than plantain and can be eaten raw, while, plantain is more starchy than sweet and is used for cooking purposes. There is a limited literature data on the formation and reduction of HMF in fried products. To reduce HMF content in fried products, the mechanism that regulates the formation of HMF, and the effects of factors on HMF formation during deep-fat frying should be elucidated.

The first objective of this study was:

- To validate HPLC method to extract and quantify HMF from frying oils and fried foods.

In this study, three hypotheses were tested in relation to investigate the formation and reduction of HMF during deep-fat frying of banana. These hypotheses and objectives were as follows.

Hypothesis 1: The formation of HMF from sugars at frying temperature can be affected by type of amino acids.

Objective:

- To determine the effect of different sugars and amino acids on formation and reduction of HMF in different model systems

Hypothesis 2: Formation of HMF during deep-fat frying is dependent on food composition, type of frying media and frying conditions.

Objective:

- To investigate the effect of different oils, food matrices and frying condition on the formation and reduction of HMF during deep-fat frying

Hypothesis 3: Banana variety, ripening stage and pre-treatment of banana can affect the formation of HMF during deep-fat frying.

Objective:

- To investigate the effects of variety, ripening stage and pre-treatments of banana on the formation and reduction of HMF during deep-fat frying

Studies were designed, conducted and results were analyzed to test these hypotheses.

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