



UNIVERSITI PUTRA MALAYSIA

***THERMAL CONDUCTIVITY OF METAL (Al, Cr, AND Ni) AND
SEMICONDUCTOR (ZnS AND CdS) NANOFUIDS DETERMINED BY A
TRANSIENT HOT WIRE METHOD***

ROSLINA MAT HUSSIN

FS 2014 91



**THERMAL CONDUCTIVITY OF METAL (Al, Cr, AND Ni) AND
SEMICONDUCTOR (ZnS AND CdS) NANOFUIDS DETERMINED BY A
TRANSIENT HOT WIRE METHOD**

By

ROSLINA MAT HUSSIN

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfillment of Requirements for Degree of Master of Science**

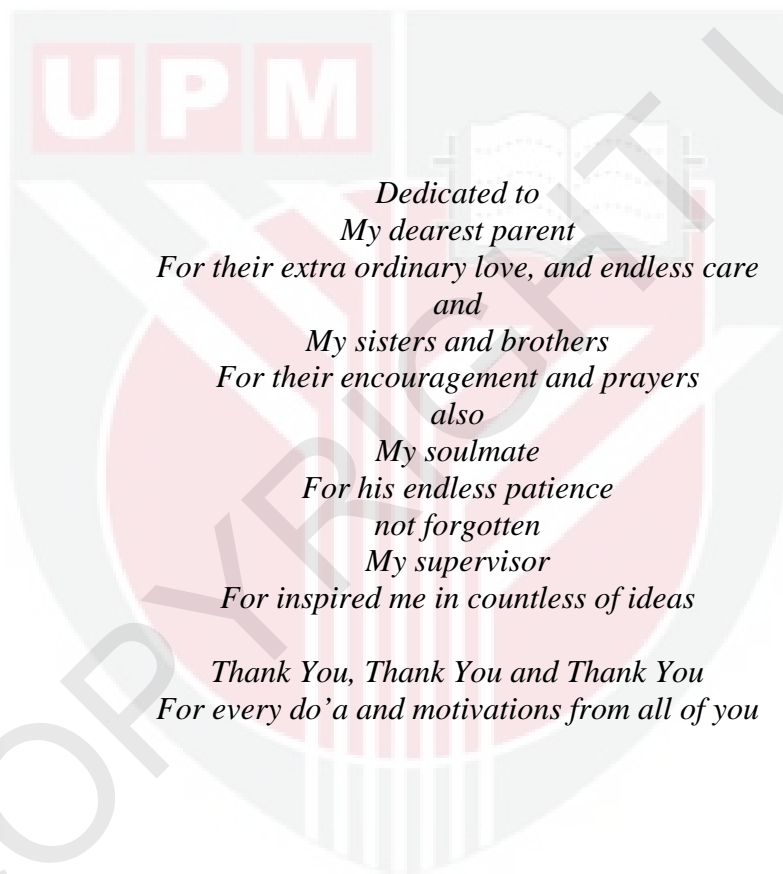
July 2014

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any materialas contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia





*Dedicated to
My dearest parent
For their extra ordinary love, and endless care
and
My sisters and brothers
For their encouragement and prayers
also
My soulmate
For his endless patience
not forgotten
My supervisor
For inspired me in countless of ideas
Thank You, Thank You and Thank You
For every do'a and motivations from all of you*

© COP

UPM

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in Fulfillment of the requirement for degree of Master of Science

THERMAL CONDUCTIVITY OF METAL (Al, Cr, AND Ni) AND SEMICONDUCTOR (ZnS AND CdS) NANOFUIDS DETERMINED BY A TRANSIENT HOT WIRE METHOD

By

ROSLINA MAT HUSSIN

July 2014

Chairman : Professor W. Mahmood Mat Yunus, PhD

Faculty : Science

In heat transfer application, the challenge in development of the nanofluids is in understanding on how the interactions between the particles and fluid affect thermal properties of the fluids. The main objective of this work is to determining the thermal conductivity of the metallic and semiconductor nanofluids. The specific objectives are to study the effect of volume fraction concentration, metal nanoparticles, base fluid, pH value of Al, Ni, and Cr nanofluids and radiation dose of ZnS and CdS nanofluid on the enhancement of the thermal conductivity. The metallic nanofluid samples were prepared at various volume fraction concentrations ranging from 0.01- 0.30 vol.% using a single step technique. For semiconductor nanofluids, the samples were prepared at different radiation dose range from 10-60 KGy. The thermal conductivity of the samples was measured using KD2 Pro Thermal Analyzer which employed transient hot wire (THW) method at the room temperature. The thermal conductivity of pure base fluids measured are 0.254, 0.614, and 0.168 W/m·K, for ethanol, distilled water and ethylene glycol and 0.551W/m·K for pure polyvinyl alcohol (PVA). The result shows that the thermal conductivity of all nanofluids samples was increased proportionally with the increment of volume fraction. The enhancement of thermal conductivity of Al nanoparticles in ethanol, distilled water, and ethylene glycol are increased up to 26.19, 16.48, and 5.62% at 0.02 vol.%. The enhancement of the thermal conductivity of Cr nanoparticles in ethanol, distilled water, and ethylene is 21.42, 17.42, and 15.66% at 0.20 vol.% and for thermal conductivity of Ni nanoparticles in ethanol, distilled water, and ethylene glycol increased up to 21.42, 13.52, and 9.64% at 0.03 vol.%. In measuring effect of pH concentration on the thermal conductivity of Al nanofluids the results show that there exists an optimal pH value for the highest thermal conductivity. The optimal pH value the nanofluids containing a small amount of Al nanoparticles have noticeably higher thermal conductivity than that of the base fluid without nanoparticles. In this work, the optimal pH value were observed at pH 10 for 0.03, 0.05, and 0.10 vol.% with the percentage of enhancement are 8.63, 11.25, and 20.58%. . Interpretation from the results

of the effect of radiation dose shows that the thermal conductivity of CdS and ZnS are decreased by increasing the dose of the radiation exposure on the sample. Thermal conductivity of CdS is 0.558 W/m·K and 0.561 W/m·K for ZnS respectively, after been expose to the radiation at 20 KGy. While the thermal conductivity values is 0.0554 W/m·K for CdS and 0.555 W/m·K for ZnS respectively, after been expose to the radiation at 50 KGy.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

KEKONDUKSIAN TERMA LOGAM (Al, Cr, DAN Ni) DAN SEMI KONDUKTOR (ZnS AND CdS) BENDALIR NANO MENGGUNAKAN KAEDAH WAYAR PANAS FANA SEKETIKA

By

ROSLINA MAT HUSSIN

Julai 2014

Pengerusi : Professor W. Mahmood Mat Yunus, PhD

Fakulti : Sains

Dalam bidang aplikasi pemindahan haba, cabaran dalam pembangunan cecair nano adalah untuk memahami bagaimana interaksi antara zarah dan cecair menjejaskan sifat haba daripada cecair tersebut. Objektif utama dalam kajian ini adalah untuk menentukan kekonduksian haba logam dan bendalir nano semikonduktor. Objektif khusus kajian ini adalah untuk mengkaji kesan pecahan isipadu, logam zarah nano, cecair asas, nilai pH bendalir nano Al, Cr, dan Ni dan juga kesan dos radiasi pada peningkatan terhadap kekonduksian terma bendalir nano ZnS dan CdS. Sampel logam bendalir nano telah disediakan pada pelbagai kepekatan yang rendah di antara 0.01-0.30 vol.% menggunakan teknik langkah tunggal. Untuk bendalir nanosemikonduktor, sampel telah disediakan pada julat dos radiasi yang berbeza diantara 10-60 kGy. Kekonduksian terma bagi sampel telah diukur menggunakan KD2 Pro Thermal Analyzer yang menggunakan kaedah wayar panas fana seketika pada suhu bilik. Kekonduksian terma cecair asas tulen diukur adalah 0.254, 0.614, dan 0.168 W/m·K, untuk etanol, air suling dan etilena glikol manakala 0.551 W/m·K untuk polivinil alkohol (PVA) tulen. Hasil kajian menunjukkan bahawa kekonduksian haba semua sampel bendalir nano telah meningkat berkadaran dengan kenaikan pecahan kepekatan jumlah. Peningkatan kekonduksian terma zarah nano Al dalam etanol, air suling, dan etilena glikol yang meningkat sehingga 26.19, 16.48, dan 5.62% pada 0.02 vol.%. Peningkatan kekonduksian haba zarah nano Cr dalam etanol, air suling, dan etilena glikol adalah 21.42, 17.42, dan 15.66% pada 0.20 vol.%. Untuk kekonduksian haba zarah nano Ni dalam etanol, air suling, dan etilena glikol meningkat sehingga 21.42, 13.52, dan 9.64% pada 0.03 vol.%. Dalam mengukur kesan nilai pH ke atas kekonduksian haba bendalir nano Al, keputusan menunjukkan bahawa wujud nilai pH optimum untuk kekonduksian haba yang tinggi. Nilai pH optimum untuk bendalir nano yang mengandungi sejumlah kecil zarah Al mempunyai kekonduksian yang lebih tinggi daripada itu cecair asas tanpa kehadiran zarah nano. Dalam kajian ini pH optimum yang diperolehi adalah pada nilai pH 10 untuk 0.03, 0.05, dan 0.10 vol.% dengan peratusan peningkatan adalah 8.63, 11.25, dan 20.58%. Tafsiran dari hasil kesan

dos radiasi menunjukkan bahwa kekonduksian termal CdS dan ZnS adalah menurun dengan meningkatkan dosis pendedahan radiasi pada sampel. Kekonduksian termal CdS dan ZnS adalah 0.558 dan 0.561 W/m·K, masing-masingnya setelah didedahkan kepada radiasi 20 kGy. Manakala kekonduksiannya adalah 0.0554 dan 0.555 W/m·K, masing-masingnya untuk yang didedahkan pada radiasi 50 kGy.



ACKNOWLEDGEMENTS

In the name of Allah (SWT), the most Beneficent and the most Merciful. Praise to Him who give me strength, faith, courage, and patience to complete this project with the time frame. May blessing and peace be upon our prophet Muhammad (SAW).

First of all, I would like to express sincere appreciation to my supervisor Professor Dr. W. Mahmood Mat Yunus for his keen interest, patience and devotion of time throughout my project. Without his assistance and guidance this report would have never have the ending. I also would like to extend my thanks to my co-supervisor Professor Dr. ZainalAbidinTalib for his suggestion, support and guidance.

Special thanks goes to all the person who participating in this research; members of Department of Physics, Faculty of Science, Institute Bioscience, and Perpustakaan Sultan Abdul Samad of Universiti Putra Malaysia which have provided valuable information and data for my research analysis.

Deepest appreciation is extended to my parents and family members who have been a great source of encouragement to me. Also a bunch of thanks to my entire colleague's at Universiti Putra Malaysia especially my lab mates who had encouraged, supported and help me during the process of completing this work.

I certify that A Thesis Examination Committee has met on 16th July 2014 to conduct the final examination of Roslina Mat Hussin on her thesis entitle “Thermal Conductivity of Metal (Al, Cr, and Ni) and Semiconductor (CdS and Zns)NanofluidsDetermined by aTransient Hot Wire Method” in accordance with the Universities and University College Act 1971 and the constitutions of the Universiti Putra Malaysia [P.U. (A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Dr. Zulkifly Abbas

Ass. Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Dr. Zaidan Abdul Wahab

Ass.Professor
Faculty of Science
Universiti Putra Malaysia
(Internal)

Dr. Azmi Zakaria

Professor
Faculty of Science
Universiti Putra Malaysia
(Internal)

Dr. Muhamad Deraman

Professor
Faculty of Science and Technology
UniversitiKebangsaan Malaysia
(External)

Dr. Noritah Omar

Assoc. Pofessor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

W.Mahmood Mat Yunus, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Zainal Abidin Talib, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Member)

BUJANG BIN KIM HUAT, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

DECLARATION

Declaration by graduate student

I hereby confirm that:

- This thesis is my original work;
- Quotations, illustrations and citations have been duly referenced;
- This thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- Intellectual property from the thesis and copyright of thesis are fully owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- Written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in other form of written, printed or in electronic form) including books, journals, modules, proceeding, popular writing papers, manuscripts, posters, report, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- There is no plagiarism or data falsification / fabrication in the thesis and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature: _____ Date: _____

Name and Matric No.: _____

DECLARATION

Declaration by Members of Supervisory Committee

This is to confirm that:

- The research conducted and the writing of this thesis was under our supervision:
- Supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: _____
Name of
Chairman of
Supervisory
Committee: _____

Signature: _____
Name of
Member of
Supervisory
Committee: _____

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiii
LIST OF FIGURES	xv
CHAPTER	
1 INTRODUCTION	
1.1 Nanotechnology and Transient Hot Wire Method	1
1.1.1 Nanofluids Concept	2
1.1.2 Application of Nanofluids	3
1.2 Problem Statement	5
1.3 Objective	6
1.4 Scope of Research	6
1.5 Thesis Outline	7
2 LITERATURE REVIEW	
2.1 Transient Hot Wire Method	8
2.2 Previous Research Using Transient Hot Wire Method	9
2.3 Effect of Volume Fraction Concentration and Particles Size on Thermal Conductivity Value	22
2.4 Effect of the Base Fluid and Metal Particles on Thermal Conductivity Value	24
2.5 Effect of pH Concentration on Thermal Conductivity Value	29
2.6 Semiconductor Nanofluids	30

3	THEORY	31
	3.1 Introduction to Thermal Conductivity	32
	3.2 Theories for Thermal Conductivity	
4	MATERIALS AND METHODS	
	4.1 Materials	39
	4.2 Preparations of Metallic (Al, Cr, and Ni) Nanofluids at Different Volume Fraction	40
	4.3 Preparations of Al Nanofluids at Different pH Concentrations	41
	4.4 Preparation of CdS and ZnS Nanofluids at Using Different Radiation Dose	42
	4.5 Characterizations of the Material and Nanofluids	43
	4.6 Measuring Thermal Conductivity of Nanofluids	45
5	RESULTS AND DISCUSSION	
	5.1 XRD, TEM and Nanophox Analysis for Metallic Nanofluids	49
	5.2 Thermal Conductivity of Ethanol, Distilled Water and Ethylene Glycol	57
	5.3 Effect of Volume Fraction on Thermal Conductivity of Metallic Nanofluids	58
	5.3.1 Effect of Volume Fraction on Thermal Conductivity of Al Nanofluid	58
	5.3.2 Effect of Volume Fraction on Thermal Conductivity of Cr Nanofluids	63
	5.3.3 Effect of Volume Fraction on Thermal Conductivity of Ni Nanofluids	66
	5.4 Effect of Metallic Nanoparticles on Thermal Conductivity Enhancement of Base Fluids (Ethanol, Distilled water, and Ethylene Glycol)	70

5.5. Effect of pH concentration on Thermal Conductivity Al Nanofluids	75
5.6 Effect of Radiation Dose on Thermal Conductivity of ZnS and CdS Nanofluids	78
6 CONCLUSION	
6.1 General Conclusion	81
6.2 Future Study	83
REFERENCES	86
APENDICES	96
BIODATA OF STUDENT	104
LIST OF PUBLICATION	105

LIST OF TABLES

Table		Page
2.1	The summary of previous researcher using Transient Hot Wire method to study nanofluids	26
5.1	Particles Size changing of Ni nanoparticles by 2 minutes interval	55
5.2	The thermal conductivity values of pure base fluids at room temperature	57
5.3	Thermal conductivity value for pure Cr in ethanol, distilled water and ethylene glycol	63
5.4	Thermal conductivity value for Ni in ethanol, distilled water and ethylene glycol	67
A	Thermal Conductivity of pure metal at room temperature	103

LIST OF FIGURES

Figure		Page
1.2	Illustrations of nanofluids	5
2.1	Schematic diagram for THW method proposed by Alvarado et al. (2012)	16
2.2	The entire assembly for the measurement cell (Alvarado et al. 2012)	16
2.3	Experimental setup used by Hong et al., 2011	17
2.4	Experimental set up used by Minta et al. 2009.	20
4.1	Illustration of schematic diagram in measuring Ph	44
4.2	Illustration of experimental set up	46
5.1	X-ray diffractogram of Al powder	50
5.2	TEM image of Al/distilled water at 0.11 vol.% (a) and (b) is before been sonicated (c) and (d) is after 9 hours sonicated	51
5.3	X-ray diffractogram of Cr powder	52
5.4	TEM image of Cr in (a) ethanol, (b) distilled water and (c) in ethylene glycol at 0.11 vol.% is after 9 hours sonication.	53
5.5	X-ray diffractogram of Ni powder	54

5.6	TEM image of Ni sample on copper grid (a) Ni nanoparticles in ethylene glycol (b) Ni nanoparticles in distilled water (c) Ni nanoparticles in ethanol	56
5.7	The thermal conductivity of Al nanoparticles dispersed in ethanol as function of volume fraction	60
5.8	The thermal conductivity of Al nanoparticles dispersed in distilled water as function of volume fraction	60
5.9	The thermal conductivity of Al nanoparticles dispersed in ethylene glycol as function of volume fraction	61
5.10	The comparison of the enhancement thermal conductivity of Al nanoparticles in three base fluid (ethanol, distilled water, and ethylene glycol)	61
5.11	The thermal conductivity of Cr nanoparticles dispersed in ethanol as function of volume fraction	64
5.12	The thermal conductivity of Cr nanoparticles dispersed in distilled water as function of volume fraction	64
5.13	The thermal conductivity of Cr nanoparticles dispersed in ethanol as function of volume fraction	65
5.14	The comparison of the enhancement thermal conductivity of Cr nanoparticles in ethanol, distilled water, and ethylene glycol	65
5.15	The thermal conductivity of Ni nanoparticles dispersed in ethanol as function of volume fraction	68

5.16	The thermal conductivity of Ni nanoparticles dispersed in distilled water as function of volume fraction	68
5.17	The thermal conductivity of Ni nanoparticles dispersed in ethylene glycol as function of volume fraction	69
5.18	The comparison of the enhancement thermal conductivity of Ni nanoparticles in three base fluid (ethanol, distilled water, and ethylene glycol)	69
5.19	Thermal Conductivity of metal (Al, Cr and Ni) in ethanol	72
5.20	Enhancement of Thermal Conductivity of metal (Al, Cr and Ni) in ethanol	72
5.21	Thermal Conductivity of metal (Al, Cr and Ni) in distilled water	73
5.22	Enhancement of Thermal Conductivity of metallic (Al, Cr and Ni) in distilled water	73
5.23	Thermal Conductivity of metallic (Al, Cr and Ni) in ethylene glycol	74
5.24	Enhancement of Thermal Conductivity of metallic (Al, Cr and Ni) in ethylene glycol	74
5.25	Thermal Conductivity value of 0.03 vol.% Al/distilled water at different pH concentration	76
5.26	Thermal Conductivity value of 0.05 vol.% Al/distilled water at different pH concentration	77

5.27	Thermal Conductivity value of 0.10 vol.% Al/distilled water at different pH	77
5.28	Percentage different of Al/distilled water from the base fluid at different pH	78
5.29	Thermal Conductivity of ZnSnanofluids in PVA at different radiation dose	79
5.30	Thermal Conductivity of CdSnanofluids in PVA at different radiation dose	79
5.31	The de-enhancement of ZnS and CdSnanofluids in PVA at different radiation dose	80
A	XRD results of Al powder (Nanostructured and Amourhous Material, Inc)	98
B	XRD results of Ni powder (Nanostructured and Amourhous Material, Inc)	99

CHAPTER 1 INTRODUCTION

This chapter presents general overview and the concept of nanotechnology and definition of the nanofluids. The rationales of nanofluids with their application and advantages are also discussed elaborately in this chapter. This chapter also presents the problems statement, objective, scopes of research and the outline for this thesis.

1.1 Nanotechnology and Transient Hot Wire Method

Nanotechnology is a derivative of functional materials, devices and system by controlling the specific matter at nanosize level. The concepts that seeded nanotechnology were first discussed in 1959 by Richard Feynman, by his statement on the possibility of synthesis via direct manipulation of atoms (Rohrer, 1996). Rohrer also reported that the term nanotechnology was first used by Norio Taniguchi in 1974. In most industrial technologies, an ultrahigh-performance cooling is one of the crucial requirement. Unfortunately, the conventional fluids like distilled water and ethylene glycol has low thermal conductivity. Due to that it is became a primary goals in developing energy efficient heat transfers fluids which are required for ultrahigh-performance cooling. As a result, many research and development activities are being carried out to improve the heat transport properties of fluid.

Maxwell is the first person who develops an innovative idea in enhancing the thermal conductivity by adding solid particles into heat transfer fluids (Maxwell, 1873). At the same time, modern fabrications technology had also provide a great opportunity for an active material fabrication at nanometer size. Nanomaterials have a unique mechanical, optical, electrical, magnetic and thermal property. In the early twenty-first century, nanotechnology has given a big impact on our economy and society. Bharat (2004) stated that, the nanotechnology will be the next industrial revolution. Early review of research programs on nanotechnology in the United State, China, Europe, and Japan shows that nanotechnology will be an emerging and exciting technology in the twenty-first century and that is why many universities, national laboratories, small businesses and large multinational corporations have established their own nanotechnology research groups or interdisciplinary centers that focus on nanotechnology.

For determining the thermal conductivity of fluids with a high degree of accuracy Transient Hot Wire method has been widely used by many researchers since decades (Nagasaka and Nagashima, 1981; Castro et al., 1976; Mani, 1971; Haarman, 1969; Pittman 1968). Thermal conductivity of nanofluids can be measured by many optical technique as suggested by the previous researcher (Putman et al. 2006) such as hot wire laser beam (W.Mahmood et al., 2013; Faris et al. 2010) beam deflection technique and also transient hot wire method (Walvekar et al. 2012; Pang and Kang, 2012; Timofeeva et al., 2009; Garg et al., 2008; Singh, 2008).

Among of them, the transient wire method is the most common and popular technique which is widely employed among the researchers (Mursheed et al., 2006; Cyril et al.,

2010) due to its advantage in eliminating the errors due to natural convection (Paul et al., 2010; Garg et al. 2010). THW method also requires a simple and fast apparatus set up compared to other techniques (Hong et al. 2011; Kostic and Kalyan, 2009; Singh, 2008). It is also low cost and easy implementation (Clement and Yu 2011). THW method is based on the measurement of the time dependent temperature rise in a heat source such as hot wire which has already been immersed in the fluid. The hot wire serves simultaneously as an electrical resistance heater and resistance thermometer (Gross and Song, 1996; Nagasaka and Nagasima, 1981; Anderson and Backstrom, 1976).

1.1.1 Nanofluids Concept

Nanofluids are a novel heat transfer fluid that has attracted great interest in recent years due to the report that nanofluids can greatly enhance thermal properties. These nanoparticles have gained an interest of researchers from around the world due to their application which can be widely used in many industries such as cooling system, nuclear reactor, transportation, electronics and as well as biomedicine (Faris, 2011). Meanwhile, the conventional fluids such as water, ethylene glycol and ethanol are currently being extensively employed as heat transfer fluids which have many applications in industry. These fluids have relatively poor thermal conductivity. Thus several methods and techniques have been proposed to enhance the abilities of their heat transfer performance such as adding particles, applying pressure and heat, changing the pH value and many others. Nanofluid is a fluid consisting of solid nanoparticles with sizes typically in the order of 1-100 nm, dispersed in a liquid (Keblikinski, 2010).

Since solid materials have higher thermal conductivity compared with those of fluids (Touloukian et al., 1970), researchers have tried to improve the thermal conductivity of conventional fluids by suspending solid particles in them (Choi, 1995). The objective of the nanofluid is to enhance the thermal properties at the lowest concentrations by uniform dispersion and stable suspension of nanoparticles in the base fluids. Figure 1.1 shows the illustration of nanofluid in the sample cell. It consists of base fluid as a solvent and nanoparticles which have an additive around the particles as a stabilizer to make the nanoparticles well suspended.

For more than centuries, many scientists and engineers have made great efforts to enhance the inherently poor thermal conductivity of liquids by dispersing solid particles in the liquids. The situation changed when Choi and Eastman in Argonne National Laboratory revisited this field with their metallic particles and carbon nanotube suspension in 1992 (Choi and Eastman 1995; Eastman et al. 1996). Both of them have tried to disperse various metal and metal oxides nanoparticles in several different fluids. Their efforts have produced interesting results in the enhancement of thermal conductivity value which is 30% (Choi et al. 2001; Chon et al. 2005).

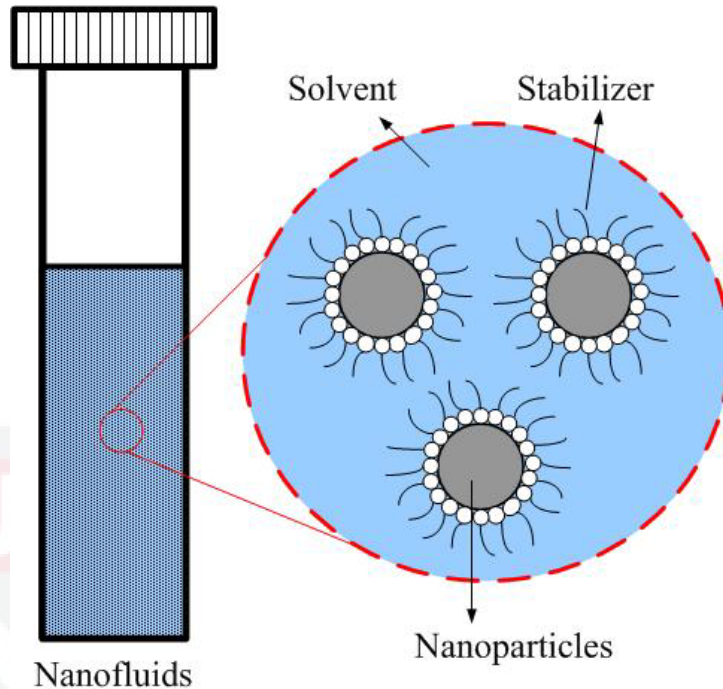


Figure 1.1. Illustrations of nanofluids

Stable suspensions of nanoparticles in conventional heat transfer are produced by two methods either single step technique or two step technique (Das et al., 2007). Single step technique is a process which involves a simultaneous process. A method in preparing nanofluids by using single step method is by dispersing the nanoparticles directly into the based fluids. However in the two step technique it involved two phase. First phase is the nanoparticle is prepared by using chemical or physical process and then followed by second phase by dispersing the nanoparticles into the based fluids. To stabilize the nanofluids after being prepared in both method, some surfactant will be added such as cetyltrimetylammonium bromide (CTAB), changing the pH concentration and also by sonicate the sample for hours in a sonic clean machine.

1.1.2 Application of Nanofluids

Nowadays many industries around the world are facing the thermal challenge and there is pressing needs for ultrahigh performance substance. Therefore, nanofluids can be used to improve the heat transfers and energy efficiency in most thermal controls system such as high cooling rates, smaller and lighter cooling systems, improved wear resistance, decreased pumping power needs, and reduced inventory of heat transfers fluids. Even though it was still at the early stage, many works in the field of nanofluids is being done in national laboratories around the world.

Great industrial interest shows that the nanofluids can be used and applied in a wide range of industries such as in the industries of transportation, electronics, process heating and cooling, and also in medical application (Das et al., 2007). The most attractive characteristic of nanofluids is that even by a small addition of nanoparticles,

they show anomalous enhancement in thermal conductivity which is over 10 times higher than the theoretically predicted using theoretical calculation. Eastman et al., (2001) reported a 40% thermal conductivity increase in ethylene glycol by adding only 0.3 vol% of Cu nanoparticles with a diameter smaller than 10 nm. However, more research work is needed for a more in depth understanding of the thermal conductivity and enhancement in the convective heat transfer of nanofluids.

As for the transportation field, a mixture of ethylene glycol and water is widely used vehicle coolant due to its lowered freezing point for anti freezing as well as the elevated boiling point. However, thermal conductivity of ethylene glycol is relatively low compared to water. At the same time, engine oils are even much worse heat transfer fluids than ethylene glycol in thermal transport performance. The addition of nanoparticles and nanotubes to these coolants to form nanofluids can increase the thermal conductivity, and has the potential to improve the heat exchange rates and fuel efficiency. The improvements can be used to reduce the size of cooling systems or remove more heat from the vehicle engine exhaust while maintaining the same cooling system.

Tseng and co-worker have studied the effects of nanofluids in the cooling of automatic transmission (Tseng et al., 2005). In his study, CuO and Al₂O₃ at 4.4 wt % of nanoparticles and antifoam agents were dispersed into transmission fluid and then the transmission fluid was used in a real-time four wheel automatic transmission. The results showed that by using nanofluids, more heat were removed from the transmission system and the automatic transmission can be kept at lower temperature distribution at both high and low rotating speeds. Gosselin and Silva (2004) carried out an investigation on the optimization of particle loading in attempt to maximize the thermal transport performance of nanofluids with appropriate constraint conditions. They found that when the volume fraction is low, the improvement of heat transfer rate is small, however when more nanoparticles were added, the increase in viscosity led to large shear stresses and then larger pumping power is necessary.

Nanofluids have been considered as the working fluid in heat pipes for electronic cooling applications. In a recent study, Ma and co-worker investigated the effect of nanofluids on the heat transport capability of an oscillating heat pipe (Ma et al. 2006). Nanofluids at a nanoparticles loading of 1 vol.% have shows to be capable to reduce the temperature difference between the evaporator and the condenser from 40.9 to 24.3 °C. Nanofluids also have the capability to provide the required cooling capacity in such applications as well as in other military systems, including military vehicles, submarines and high-power lasers. Nanofluids for military applications sometimes involve multifunctional nanofluids by adding thermal storage materials or energy harvesting materials through chemical reactions.

For the applications in cooling and thermal management, nanofluids technology can be used to improve oils and lubricants. Recent nanofluids research demonstrates the potential of adding nanoparticles into lubricants to improve the tribological properties, such as load-carrying capacity, wear resistance and friction reduction between the

mechanical components in motion. It has been verified by experimental results that the addition of surface-modified nanoparticles can be easily dispersed into lubricants to form stable nanofluids which are effective in reducing friction and improving load-carrying capacitance.

1.2 Problems Statement

Based on the previous study and development effort on the heat transfer enhancement, major improvement in cooling systems capabilities have been limited because of low thermal conductivity of the conventional heat transfers fluids such as water, ethanol, and ethylene glycol. Since solid materials have much higher thermal conductivities than fluids, it was a great idea to add particles in the fluid to enhance the thermal conductivities of the fluids. However, if solid particles of micrometer, even millimeter magnitudes are added into the base fluids it will cause many troublesome problems. Large particles and the agglomeration in fluid flows carry too much momentum and kinetic energy, which may cause damage to the surface. Thus it was very useful to replace micro particles with nanoparticles in this research.

Noticeable thermal conductivity enhancement is based on particle concentration of the liquid. Thus with the presence of nanoparticles will increase the thermal conductivity proportional to the volume fraction of the nanofluids. Even a numbers of articles published on thermal conductivity of nanofluids each year, there are very limited data reported on low volume concentration of nanofluids especially at below dilute limits of nanofluids. Dilute is a process of making a substance less concentrated by adding a solvent. For the nanofluids, the dilute limits is with a 0.20 vol.% of the nanofluid (Kwak and Kim, 2005). If the particles concentration is less than the dilute limit, the particles can move and rotate freely. However, if the concentration of the nanofluids is more than 0.20 vol.% the particles will be in the semi dilute form which will then affected the particles to become unable to move and rotate freely hence influence the heat transfer and the enhancement will be limited.

In addition, the thermal conductivity for metallic nanoparticles is higher than of the metal oxide particles. This gives an advantage to the metallic nanoparticles for material research because they can give a high rate in the enhancement of the thermal conductivity. Even though it was clear that the thermal conductivity of metallic particles is higher compare to the oxide particle, the data reported on the thermal conductivity of metallic nanofluids is still limited due to less publication by previous researcher.

There are several aspects that can influence the thermal conductivity of nanofluids such as, particle size, temperature, base fluid, pressure and viscosity. The viscosity of the liquid can also be controlled by the pH. Previous research also predicted that nanofluids synthesis by adjusting the pH of the solution will show the tendency of a larger enhancement. This implies that, by controlling the pH of the nanofluids will lead to enhancement of thermal conductivity. However, the effect of pH on thermal conductivity of nanofluids is still in the infant stage which is the same case with low

volume fraction and metallic nanofluids. These affect the data analysis on thermal conductivity enhancement by these factors to be limited and less report were published.

Due to the limitation from the previous research and advantages of the nanoparticles, hence the main objective of this research is to study the effect of volume fraction on the enhancement on thermal conductivity of Al, Cr, and Ni nanofluids. At the same time, we also study the influence of metal nanoparticles and the base fluid on the enhancement of the thermal conductivity. To observe the behavior of thermal conductivity of nanofluids, we also investigate the behavior of thermal conductivity enhancement by manipulating the pH concentration on Al nanofluids in distilled water.

The publications and reports on semiconductor nanofluids on thermal conductivity also very limited. This reason had brings this study to one step further which is expended this research by measuring the thermal conductivity of semiconductor nanofluids prepared at different radiation dose. In this study, the semiconductor nanofluids that had being prepared are CdS and ZnS nanofluid using PVA water base fluid prepared at different radiation dose from 10 - 60 kGy.

1.3 Objective

The main objective of this work is to study the enhancement of the thermal conductivity on metallic nanofluids that varies with the volume fraction concentration. To achieve the main objective, several objective needs to be complete as followed.

- 1) To study the effect of the volume fraction concentration of Al, Cr, and Ni nanoparticles in distilled water, ethanol, and ethylene glycol on the enhancement of thermal conductivity nanofluids.
- 2) To examine the optimum of the thermal conductivity Al nanoparticles in distilled water affected by the pH value.
- 3) To study the thermal conductivity of CdS and ZnS nanofluid using PVA water base fluid prepared at different radiation dose.

1.4 Scope of Research

This work has a scope of limitation in instrumentation and sample preparation. The measurement of the thermal conductivity metallic nanofluids were measured using transient hot wire (THW) method due to it advantages especially in sample preparation and experimental set up. In this experimental, we used KD2 Pro Thermal Analyzer supplied by Decagon Device was done at room temperature in the laboratory (27 °C).

For metallic nanofluids, the sample were only prepared using Al (18 nm), Ni (20 nm) and Cr (30 nm) of nanoparticles and used three type conventional fluid. The base fluids used in this study are ethanol, distilled water and ethylene glycol. For the experiment in

observing the pH effect on thermal conductivity of nanofluids, we only measured the thermal conductivity of Al/distilled water at five different pH value ranges 1-14. In addition, this research only measured the thermal conductivity of nanofluids at the range 0.01 - 0.30 vol.% which mostly cover for dilute limits of metallic nanofluids. For semiconductor nanofluids, the samples were only subjected on CdS and ZnS nanofluids in PVA which had prepared using radiation dose range from 10-60 kGy.

1.5 Thesis Outline

Chapter 1 presents the general view and concept of nanofluids and the technique. The description on the concept of nanotechnology and definition of the nanofluids has been briefly introduced in this chapter. This chapter also includes the problems statement, objective, scopes of research and an outline for the thesis.

In Chapter 2, we will discuss on literature review of the method that we used in measuring the thermal conductivity of nanofluids from the previous researchers who had a success with the technique. The summary of published articles respecting to the effect of volume fractions, base fluids, particles size and pH of nanofluids on thermal conductivity enhancement were also present in this chapter.

Chapter 3 devoted to present the theories on the method that used in this work and explain the deviation of the thermal conductivity from the Fourier's Law. We also present the theories from Maxwell and Hamilton Crosser in 1891 on thermal conductivity.

Chapter 4 presents the preparations of all sample and experimental set up were presented. The method was used in preparing the nanofluids present with details in this chapter. Besides that, this section also describes in detail in the procedure and instrumentation that we use in measuring the thermal conductivity of nanofluids.

Chapter 5 discussed the data collections and analysis in the experiment on thermal conductivity of metallic nanofluids and also the characteristic of the materials. The effect on volume fraction, base fluid, pH and metal particles were also discussed in this chapter.

Chapter 6 is devoted to present the main conclusion for this work. In addition, this chapter also presents some suggestion for future work and research to create more interest in nanofluids and expands the nanotechnology fields.

REFERENCES

- Akerman, M.E., Chan, W.C., Laakkonen, P., Bhatia, S.N., and Ruoslahti, E. (2002). Nanocrystal targetting in vivo. *Proc Natl Acad Sci USA*. 99: 12617-12621.
- Alvarado, S., Marin E., Juarez , A.G., Calderon, A., and Ivanov, R. (2012). A hot-wire method based thermal conductivity measurement apparatus for teaching purposes. *European Journal of Physics*. 33(4): 897-906.
- Anderson, P. and Backstrom, G., (1976). Thermal conductivity of solids under pressure by transient hot wire method. *Review of Scientific Instrument*. 47(2): 205-209.
- Assael, M.J., Metaxs, I.N., Kakosimos, K. and Constantinou, D. (2006). Thermal Conductivity of Nanofluids : Experimental and Theoretical. *Int. J. Thermophys*. 27 (4): 999-1017.
- Backlund, N.G. (1969). *Thermal Conductivity 8, edited by Ho CY and Tayloy RE*. New York: Plenum Press.
- Beck M.P., Yuan Y., Warriar, P., and Teja, A.S. (2010). The thermal conductivity of alumina nanofluids in water, ethylene glycol, and a water and ethylene glycol mixtures. *Journal Nanoparticles Res*. 12:1469.
- Beck, M.P., Warriar, P., and Yuan Y. (2009). The effect of particles size on the thermal conductivity of alumina nanofluids. *Journal Nanoparticles Res*. 11: 1129-1136.
- Bertolini D. and Tani A. (1997). Thermal Conductivity of water. Molecular Dynamic and Generalized Hydrodynamics results. *Physics Review E*., 56: 212 - 215.
- Bharat, B. (2004). Nanotechnology. *Introduction to Nanotechnology* (pp. 1-13). New York: Springer.
- Cao, A., Liu, Z., Chu, S., Wu, M., Ye, Z., Cai, Z., Chang, Y., Wang, S., Gong, Q., and Liu, Y. (2010). A facile one step method to produce grapheme CdS quantum dot nanocomposites as promising optoelectronic materials. *Adv Mater*. 22: 103-106.
- Carslaw, H.S., and Jaeger, J.C., (1959). *Conduction of Heat in Solids, 2nd edition*. Oxford: Clarendon Press.
- Castro, C.A.N., Calado, J.C.G., Wakeham, W.A., and Dix, M., (1976). An apparatus to measure the thermal conductivity of liquids. *Journal Physics E: Science Instrument* 9: 1073-1080.

Chanal, R.P., Mahendia, S. Tomar, A.K., and Kumar, S. (2012). γ -Irradiated PVA/Ag nanocomposite films: Materials for optical applications. *Journals Alloys Comd.* 538: 212-219.

Chiese M. and Simonsen A.J. (2010). The importance of suspension stability for hot wire measurement of thermal conductivity of colloidal suspensions, 16th Australasian Fluid Mechanics Conference, Gold Coast Australia.

Choi, S.U.S. (1995). *Enhancing thermal conductivity of fluid with nanoparticles: Developments and application of non-newtonian flows*, Conference: 1995 International mechanical engineering congress and exhibition, San Francisco, CA (United States), Nov.12-17, 1995.

Chon, C.H., Kihm, K.D., Lee, S.P., and Choi, S.U.S., (2005). Empirical correlation finding the role of temperature and particle size for nanofluids (Al_2O_3) thermal conductivity enhancement. *Applied Physics Letter* 87(15):153107-1531.

Clement, K. and Yu F. (2011) Experimental and theoretical studies of nanofluids thermal conductivity enhancement: a review. *Nanoscale Research Letters* 6: 229 (1-13).

Cook, J.G., Moore, J.P., Matsumura, T., and Vandermeer, M.P., (1976). *Thermal Conductivity 14*, edited by Klements PG and Chu TK. New York: Plenum Press.

Dae H.Y., Hong, K.S., and Ho-Song, Y. (2007). Study of thermal conductivity of nanofluids for the application of heat transfer fluid. *Thermochimica Acta* 45: 66-69.

Das, S.K., Choi, S.U.S., Yu, W., and Pradeep, T., (2007). *Nanofluids: Science and technology*. New Jersey: John Wiley Press.

David R.L. (2009). CRC handbook of chemistry and physics, Wiley-VCH, Germany

Davis, R. H. (1996). The effective thermal conductivity of a composite material with spherical inclusions. *Int. J. Thermophys.* 7: 609 -614.

Dean J.A., Lange's Handbook of Chemistry, McGraw Hill Professional (1998)

Eastman J A, Choi U S, Li S, Thompson L J and Lee S (1996). *Fall Meeting of the Materials Research Society*. MRS, Pittsburgh Boston: 2-6.

- Eastman J.A., Choi S.U.S., Li S., Yu W., and Thompson L.J. (2001). Anomalously increased effective thermal conductivities of ethylene glycol-based nanofluids containing copper nanoparticles, *Appl. Phys. Lett.* 78: 718-710.
- Eastman, J.A., Choi S.U.S., Li S., Thompson L.J. (1996) Enhanced thermal conductivity through the development of nanofluids. *Fall Meeting of the Materials Research Society (MRS)*, Boston, USA.
- Eastman, J.A., Choi, S.U.S., Zhang, Z.G., Yu, W., Lockwood, F., and Grulke, E.A., (2001). Anomalous thermal conductivity enhancement in nanotube suspension. *International Applied Physics Letter* 79(14): 2252-2254.
- Faris (2011). *Development of hot wire laser beam displacement technique for determining thermal conductivity and thermal diffusivity of nanofluids*, Universiti Putra Malaysia, Serdang, Malaysia.
- Faris, M.A., W.Mahmood, M.Y., Maarof, M., and Zainal, T. (2010). The effect of volume fraction concentration on the thermal conductivity and thermal diffusivity of nanofluids: Numerical and experimental. *Review of Scientific Instrument* 81(7): 074901(1-9).
- Faris, M.A., W.Mahmood, M.Y., Maarof, M., and Zainal T. (2013). Study of the effect particles size and volume fraction concentration on the thermal conductivity and thermal diffusivity of Al₂O₃ nanofluids. *International Journal of Physical Science* 8 (28): 1442-1457.
- Garg J., Poudel B., Chiesa M., Gordon J.B., Ma J.J., Wang J.B., Ren Z.F., Kang Y.T., Ohtani H., Nanda J., McKinley G.H., and Chen G. (2008). Enhanced thermal conductivity and viscosity of copper nanoparticles in ethylene glycol nanofluids. *Journal App. Phys.* 103: 13251-13255.
- Garg, J., Poudel, B., Chiesa, M., Gordon, J.B., Ma, J.J., Wang, J.B., Ren, Z.F., Kang, Y.T., Ohtani, H., Nanda, J., McKinley, G.H., and Chen, G. (2008). Enhanced thermal conductivity and viscosity of copper nanoparticles in ethylene glycol nanofluids. *Journal of Applied Physics* 103: 074301(1-6).
- Gosselin, L, and Silva, A.K. (2004). Combined heat transfer and power dissipation optimization of nanofluid flows. *Applied Physics Letters* 85(18): 4160-4162.
- Gross, U. and Song, Y.W. (1996). Thermal conductivity of new refrigerants R125 and R32 measured by transient hot wire method. *International Journal of Thermo Physics*, 17 (3): 607-619.

Haarman, J.W., (1969). *An accurate method for the determinations of the thermal conductivity coefficient of gases*. PhD Thesis, Technische Hageschool, Delft, Netherlands.

Hong, S.W., Kang, T., Kleinstreuer, C., and Koo, J. (2011). Impact analysis of natural convection on thermal conductivity measurement of nanofluids using transient hot wire method. *International Journal of Heat and Mass Transfers* 54: 3448-3456.

Hong, Y. and Yang, H. (2005). Nanoparticles dispersion dependent thermal conductivity in nanofluids. *Journals of Korean Physical Society*. 47: 321-324.

Hrishikesh, E.P., Sundararajan, T., and Das, S.K. (2009). An experimental investigations into the thermal conductivity enhancement in oxide and metallic nanofluids. *Journal of Nanoparticles Research* 12 (3): 1015-1031.

Huaqing, X., Wonjin, H.L., and Monsoo, Y.C. (2003). Nanofluids containing multiwalled carbon nanotubes and their enhanced thermal conductivity. *Journal of Applied Physics*. 94 (8): 4967-4971.

Huynh, W.U., Dittmer, J.J., and Alivisatos, A.P. (2002). Hybrid nanorod - polymer solar cell. *Science* 295: 2425-2427.

Hwang, Y., Ahn, Y.C., Lee, J.K., Lee, C.G., Shin, H.S., Kim G.T., and Park, H.S. (2006). Investigation characteristic of thermal conductivity enhancement of nanofluids. *Current Applied Physics* 6: 1068-1071.

Hwang, Y., Lee, J.K., Lee, C.H., Jung, Y.M., Cheong, S., Lee, C.G., Ku, B.C., and Jang, S.P. (2007). Stability and thermal conductivity characteristic of nanofluids. *Thermochimica Acta* 455: 70-74.

Jana, S., Khojin, A.S., and Zhong, W.H. (2007). Enhancement of fluid thermal conductivity by the addition of single and hybrid nano-additives. *Thermochimica Acta* 462 (1-2): 45-55.

Jang S.P. and Choi S.U.S. (2004). Role of Brownian motion in the enhanced thermal conductivity of nanofluids. *App. Phys. Rev. Lett.* 94: 513-517.

Jeffrey, D. J. (1973). Conduction through a random suspension of spheres. *Proc. R. Soc. London Ser. A*. 335: 355-367.

John J.M. Encyclopedia of chemical processing and design, Marcel Dekker, Inc, New York, USA (1984).

Kebllinski P., Thermal nanosystem and nanomaterial in topic in applied physics.

S.Volz, Springer (2010).

- Kityk, I., Kasperczyk, J., Sahraoui, B., Yasinskii, M and Holan, B. (1997). Low temperature anomalies in polyvinyl alcohol photopolymers. *Polymer*. 38: 4803-4806.
- Kostic, M. and Kallian, C.S. (2009). Computerized transient hot wire thermal conductivity (HWTC) apparatus for nanofluids, Proceedings of the 6th WSEAS International Conference on Heat and Mass Transfer (HMT'09), USA.
- Kumar D.H., Patel H.E., Rajeev V.R., Sundararajan T., Pradeep T., and Das S.K. (2004). Model for Heat Conduction in Nanofluids. *Phys. Rev. Lett.* 93: 4423-4426.
- Kwak, K. and Kim, C. (2005). Viscosity and thermal conductivity of copper oxide nanofluid dispersed in ethylene glycol. *Korea-Australia Rheology Journal* 17 (2): 35-40.
- Lee S. and Choi S.U.S (1999). Measuring thermal conductivity of fluid containing oxide nanoparticles, *Journal of Heat Transfer* 212: 521-524.
- Lee, S. and Choi, S.U.S. (1999). Measuring thermal conductivity of fluid containing oxide nanoparticles, *Journal of Heat Transfer* 121 (2): 280-289.
- Li, C.H. and Peterson, G.P. (2006). Experimental investigation of temperature and volume fraction variations on the effective thermal conductivity of nanoparticle suspensions (nanofluids). *Journal of Applied Physics*. 99 (8): 084314.
- Li, J.M., Li, Z.L., and Wang, B.X.(2002). Experimental viscosity measurements for copper oxide nanoparticle suspensions. *Tsinghua Sci. Tech.* 7(2): 198 – 204.
- Li, X., Zhu, D., and Wang, X. (2007). Evaluation on dispersion behavior of the aqueous copper nano-suspensions. *Journal of Colloid and Interface Science*. 31 (2): 456–463.
- Lu, S. and Lin, H. (1996). Effective conductivity of composites containing aligned spheroidal inclusions of finite conductivity. *J. Appl. Phys.* 79 (9): 6761–6769.
- Ma, H.B., Wilson, C., Borgmeyer, B., Park, K., Yu, Q., Choi, S. U. S., and Tirumala, M. (2006). Effect of nanofluid on the heat transport capability in an oscillating heat pipe. *Applied Physics Letters* 88(14): 143116(1-3).
- Mani, N. (1971). Precise determination of the thermal conductivity of fluids using absolute transient hot wire technique *PhD Thesis*, University of Calgary, Canada.

Masuda, H., Ebata, A., Teramae, K., and Hishinuma, N. (1993). Alternation of thermal conductivity and viscosity of liquid dispersing ultra fine particles (dispersion of γ -Al₂O₃, SiO₂ and TiO₂ ultra fine particles). *Netsu Bussei*, Japan, 4 (4): 227-233

Maxwell, J.C. (1873). *A treatise on electricity and magnetism*, Dower Publications.

Mintsa, H.A., Roy, G., Nguyen, C.T., and Doucet, D. (2009). New temperature dependent thermal conductivity data for water based nanofluids, *International Journal of Thermal Science* 48: 2926-2932.

Moghadassi, A.R., Hosseini, S.M., and Henneka, D.E. (2010). Effect of CuO nanoparticles in enhancing the thermal conductivity of monoethylene glycol and paraffin fluids. *India Engineering Chemical Res.*49: 1900-1904.

Moore, J.P., Williams, R.K., and McElroy, D.L., (1968). *Thermal Conductivity 7*, edited by Flynn DR and Peavy BA. Washington NBS Spec. Publication.

Muccini, M. (2006). A bright future for organic field effect transistor. *Nat Mater.* 5: 605-613.

Mursheed S.M.S., Leong K.C., and Yang C. (2006). Determination of the effective thermal diffusivity of nanofluids by double hot wire technique, *Journal Appl. Phys* 39: 560-568.

Musheed, S.M.S., Leong, K.C., and Yang, C. (2009) A combined model for effective thermal conductivity of nanofluids. *Applied Thermal Engineering* 29: 571-580.

Nagasaka Y. and Nagashima Y. (1981). Absolute measurement of the thermal conductivity of electrically conducting liquids by the transient hot-wires method. *Journal Phys. E: Sci. Instrum.*14: 2231-2234.

Nagasaka, Y. and Nagashima, Y. (1981). Absolute measurement of the thermal conductivity of electrically conducting liquids by the transient hot-wires method. *Journal of Physics. E: Sci. Instrum.*14: 1435-1440.

Nan, C.W., Birringer, R., and Clarke, D.R. (1997). Effective thermal conductivity of particulate composites with interfacial thermal resistance. *Journal Applied Physics* 81: 6692-6699.

- Nie, S., Xing, Y., Kim, G.J., and Simon, J.W. (2007). Nanotechnology applications cancer. *Annu Rev Biomed Eng.* 9: 257-288.
- Okhio, C., Hudes, D., and Black, J. (2010). Review of literature on nanofluids flow and heat transfer properties. *Journals of Selected Areas in Nanotechnology. December Edition:* 1-8.
- Ong, P.L. and Levitsky I.A. (2010). Organic/IV, III-V Semiconductor hybrid solar cells. *Energies.* 3: 313-334.
- Ozisik, M.N. (1993) .Heat Conduction, John Wiley & Sons, INC.
- Pang, C. and Kang, Y.T. (2012). *Stability and thermal conductivity characteristic of nanofluids (H₂O/CH₃OH + H₂Cl + Al₂O₃ nanoparticles) for CO₂ absorptions applications.* International refrigeration and air conditioning conference: Purdue.
- Patel E., Sundararajan T., Pradeep T., Dasgupta A., Dasgupta N., and Das S.K., (2005). A micro convention model for thermal conductivity of nanofluids. *Pramana Journal Phys.* 65:718-722
- Patel, H.E., Das, S.K., and Surdarajan, T. (2003). Thermal conductivity of naked and monolayer protected metal nanoparticles base nanofluids: Manifestation of anomalous enhancement and chemical effects. *Applied Physics Letter.* 83 (14): 2931-2933.
- Paul G., Chopkar M., Manna I., Das P.K. (2010). Technique for measuring the thermal conductivity : A review, *Renewable and Sustainable Energy Review* 14: 3168-3172.
- Peng, X., Schlamp, M.C., Kadavanich, A.V., and Alivisatos A. (1997). Epitaxial growth of highly luminescent CdSe/CdS core/shell nanocrystals with photostability and electronic accesibility. *Journal of Applied Physics.* 119: 7019-7029.
- Pittman, J.F.T. (1968). *Fluid thermal conductivity determination by transient hot-wires method* PhD Thesis, University of London, United Kingdom.
- Powell R.W., Tye R.P., and Hickman M.J. (1964). The thermal conductivity of nickel, *Int.J.Heat Mass Trans.* 8: 979-688
- Putman, S.A., Cahill, D.G., Braun, P.V., Ge, Z., and Shimmin, R.G, (2006). Thermal conductivity of nanoparticle suspensions *Journal.Appy. Phys.* 99: 084308-084324.

- Roehr, T. (2001) *The biotechnology of ethanol classical and future applications*, Wiley-Vch, Germany.
- Rogach, A.L., Gaponik, N., Lupton, J.M., Bertoni, C., Gallardo, D.E., Dunn, S., Li, P.N., Paderi, M., Reppeto, P., and Romanov S.G. (2008). Light emitting diodes with semiconductor nanocrystals. *Angew Chem Int. Ed.* 47: 6538-6549.
- Rohrer H., (1996). *The nanoworld: changes and challenges*, Microelectron. England.
- Singh A.K. (2008). Thermal Conductivity of Nanofluids. *Defence Science Journal.* 58: 600-607
- Singh A.K. and Reddy N.S. (2003). Field instrumentation for thermal conductivity measurement *Ind. Journal Pure and Appl. Phys.*, 41: 1-5
- Taniguchi, N. (1974). *On the Basic Concept of 'Nano-Technology'*, Proc. Intl. Conf. Prod. Eng. Tokyo, Part II, Japan Society of Precision Engineering.
- Timofeeva E.V., Roubort J.L., and Singh D. (2009). Particles shape effects on thermal physical properties of alumina nanofluids, *Journal. App. Phys.* 106: 1821 (1-7).
- Timofeeva E.V., Yu W., France D.M., Roubort J.L., and Singh D., (2011) Nanofluids for heat transfer: An engineering approach, *Springler Open Journal*: 1-5
- Timofeeva, E.V., Yu,W., Franc, D.M., Singh, D., and Roubort, J.L. (2011). Based fluid temperature effect on the heat transfer characteristics of Sic in EG/H₂O and H₂O nanofluids, *Journal Applied Physics.* 109; 014914 (1-5).
- Timofeeva, E.V., Smith, D.S., Yu, W., France , D.M., Singh, D., and Roubort, J.L. (2010). Particles size and interfacial effects on thermo physical properties and heat transfer characteristic of water based α -Sic nanofluids, *Nanotechnology.* 21: 215703 (1-10).
- Touloukian, Y. (1970). *Thermophysical Properties of Matter, vol. 2.* New York: Plenum Press.
- Tseng, S.C., Lin, C.W., and Huang, K. (2005). Heat transfer enhancement of nanofluids in rotary blade coupling of four-wheel-drive vehicles. *Acta Mechanica.* 179: 11–23.

- Twardowaki, T.T. .(2007). *Introduction to nanocomposite material: Properties, processing, characterization*. Lancaster, Pa: Destech Publication.
- Venkatram, N., Kumar, R., and Narayana, R.D. (2006). Nonlinear absorption and scattering properties of cadmium sulphine nanocrystalines with its application as apotential optical limiter. *Journal of Applied Physics*. 100: 074308(1-8).
- Vilimpoc V., Chen T., Cole R., and Sukanek P.C. (1989). A method for determining the thermal diffusivity of fluid. *Int. Journal of Heat and Mass transfer*, 32: 121-214
- W. Mahmood, M.Y., Faris M.A., and Zainal A.T. (2013). Study on the effect of particles size and volume fraction concentration on the thermal conductivity and thermal diffusivity of Al₂O₃ nanofluid. *International Journal of Physical Sciences* 8(28): 1442-1457
- Wakeham, W.A., Nagashima, Y., and Sengers, J.V. (1991). *Measurement of The Transport Properties of Fluids*. Amsterdam: Blackwell Oxford.
- Walvekar ,R., Faris, .A., and Khalid, M. (2012). Thermal conductivity of carbon nanotube nanofluids: Experimental and theoretical study. *Heat transfer, Asian Res.*41: 145-463.
- Wang S., *Thermal Conductivity of Nanocrystalline Nickel*, University of Toronto 2011.
- Wang, X.J. and Li, X.F. (2009) Influence of pH on nanofluids viscosity and thermal conductivity. *Chinese Physics Letter*. 26 (5): 056601(1-4).
- Wang, X.Q. and Mujumdar, A.S. (2007). Heat transfer characteristics of nanofluids: a review. *International Journal of Thermal Sciences*, 46: 1-19.
- Wen, D. and Ding Y. (2004). Effective thermal conductivity of aqueous suspensions of carbon nanotube (carbon nanotube nanofluids). *Journal of Thermophysics and Heat Transfer*. 18(4): 482-485.
- Xie H., Lee H., Youn W., and Choi M. (2003). Nanofluids containing multiwalled carbon nanotubes and their enhanced thermal conductivities *Journal Appl. Phys. Lett* 94: 412-415.
- Xie, H.Q., Wang, J.C., Xi, T.G., Liu, Y., Ai, F., and Wu, Q.R. (2002). Thermal conductivity enhancement of suspensions containing nanosized alumina particles. *Journal of Applied Physics Letter*. 91 (7): 4568-4572.

- Xuan Y. and Li Q. (2000). Heat transfer enhancement of nanofluids, *Journal of Heat and Fluid Flow* 21: 58-64.
- Xuan Y. and Li Q. (2003). Investigation on convective heat transfer and flow features of nanofluids, *Journal of Heat Transfer* 125: 4967-4971
- Xuan, Y. and Li, Q. (2000). Conceptions for heat transfer correlation of nanofluids. *International Journal of Heat and Fluid Flow*. 21: 58-64.
- Xue Q.Z. (2003). Model for effective thermal conductivity of nanofluids, *Phys. Lett A* 307: 313-317
- Yaws C.L. Handbook of transport property data, Houston Gulf Publishing Company (1995)
- Yeganeh, M., Shahtahmaseb, N., Kompany, A., Goharshadi E.K, Yourssefi, A., and Siller, L. (2010) Volume fraction and temperature variations of the effective thermal conductivity of nanodiamond fluids in deionized water. *International Journal of Heat and Mass Transfer* 53 (15): 3186-3192.