



UNIVERSITI PUTRA MALAYSIA

***MODIFICATION OF DOLOMITES USING TIN AND ZINC AS EFFICIENT
SOLID CATALYSTS FOR METHYL ESTERS PRODUCTION***

SHAJARATUN NUR BINTI ZDAINAL ABIDIN

FS 2014 88



**MODIFICATION OF DOLOMITES USING TIN AND ZINC AS EFFICIENT
SOLID CATALYSTS FOR METHYL ESTERS PRODUCTION**

By

SHAJARATUN NUR BINTI ZDAINAL ABIDIN

**Thesis Submitted to the School Graduate Studies, Universiti Putra Malaysia,
in Fulfillment of the Requirement for the Degree of Master Science**

July 2014

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



Abstract of the thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the Degree of Master Science

MODIFICATION OF DOLOMITES USING TIN AND ZINC AS EFFICIENT SOLID CATALYSTS FOR METHYL ESTERS PRODUCTION

By

SHAJARATUN NUR BINTI ZDAINAL ABIDIN

July 2014

Chairman : Professor Taufiq Yap Yun Hin, PhD, CChem, FRSC (UK)

Faculty : Science

Rapid energy consumption nowadays has led to the development of renewable resources of energy such as biodiesel. However, high cost of biodiesel production retards further development of this industry. Dolomite is a naturally occurring source of CaO and MgO which promotes a high potential heterogeneous base catalyst for biodiesel production. Transesterification of palm oil via calcined dolomite (AD8) catalyst exhibited 98.3 % conversion of fatty acid methyl ester (FAME). Selective metal oxides, Sn and Zn were as dopants since they have amphoteric properties which can improve the basicity of the parent material (dolomite). In order to obtain modified dolomite catalyst, 1, 3 and 5 % of metal oxides (Sn and Zn) were doped separately on the AD8 via wet impregnation method and exposed for calcination in air at 773 K for 3 h. These catalysts were denoted as 1DSN, 3DSN and 5DSN for Sn-dopant, whereas 1DZN, 3DZN and 5DZN for Zn-dopant. The catalysts were characterized by using X-ray Diffractometer (XRD), Brunauer-Emmet-Teller (BET) surface area, Scanning Electron Microscopy (SEM) and Temperature Programmed Desorption (TPD) of CO₂. The catalysts were then employed for transesterification reaction under different conditions (time of reaction, methanol to oil molar ratio and amount of catalyst) to investigate the catalytic activities of the catalysts. From the result, calcined dolomite (AD8) which has been doped with 3 % of SnO₂ (3DSN) showed optimum conversion of 99.98 % at the least conditions i.e. 1 wt.% of catalyst amount, in 15:1 methanol to oil molar ratio reacted in 4 h compared to 3DZN and AD8. The catalytic activities of these catalysts were found depending on the basicity and the surface area of the catalyst used. Several tests were conducted to study the physicochemical properties such as pour point, flash point, kinematic viscosity, sulphur content and cloud point of biodiesel produced. Based on the results, the synthesized biodiesel is comparable with conventional diesel in the market since it meets the international standards of biodiesel (ASTM, EN) and MS for diesel fuel specifications as well.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk Ijazah Master Sains

PENGUBAHSUAIAN KE ATAS DOLOMIT MENGGUNAKAN TIMAH DAN ZINK SEBAGAI MANGKIN PEPEJAL YANG EFISIEN UNTUK PENGHASILAN ESTER METIL

Oleh

SHAJARATUN NUR BINTI ZDAINAL ABIDIN

Julai 2014

Pengerusi : Profesor Taufiq Yap Yun Hin, PhD, CChem, FRSC (UK)

Fakulti : Sains

Penggunaan tenaga yang pesat pada masa kini telah membawa kepada perlunya pembangunan sumber tenaga alternatif yang boleh diperbaharui seperti biodiesel. Walau bagaimanapun, kos pengeluaran biodiesel yang tinggi melambatkan perkembangan industri ini. Dolomit adalah sumber semula jadi CaO dan MgO yang boleh dijadikan mangkin heterogen alkali yang berpotensi tinggi untuk pengeluaran biodiesel. Trans-pengesteran minyak sawit daripada asid lemak metil ester (FAME) menggunakan mangkin dolomit yang telah dikalsin (AD8) menunjukkan penukaran sebanyak 98.3 %. Logam oksida terpilih iaitu Sn dan Zn telah digunakan sebagai dopan-dopan memandangkan kedua-duanya memiliki sifat-sifat amfoterik yang mana dapat meningkatkan sifat bes katalis asalnya (dolomit). Untuk mengubah suai mangkin dolomit ini, sebanyak 1, 3 dan 5% logam oksida (Sn dan Zn) telah didopkan secara berasingan pada AD8 melalui kaedah pengisitepuan basah kemudian dikalsin di dalam pengaliran udara pada suhu 773 K selama 3 jam. Mangkin-mangkin ini telah dilabelkan sebagai 1DSN, 3DSN dan 5DSN untuk Sn-dopan, manakala 1DZN, 3DZN dan 5DZN untuk Zn-dopan. Pemangkin telah dicirikan dengan menggunakan instrumen pembelauan sinar-X (XRD), pengukuran luas permukaan Brunauer - Emmet - Teller (BET), pengimbas microscopi elektron (SEM) dan penyahjerapan karbon dioksida pada suhu terkawal (CO₂-TPD). Pemangkin kemudiannya digunakan untuk tindak balas trans-pengesteran pada beberapa keadaan (masa tindakbalas, nisbah molar metanol terhadap minyak dan jumlah pemangkin) untuk mengkaji aktiviti kesemua mangkin. Kajian menunjukkan, dolomit yang telah dikalsin (AD8) dan didopkan dengan 3% SnO₂ (3SD) dapat melakukan penukaran optimum sebanyak 99.98 % pada keadaan yang paling rendah berbanding 3ZD dan AD8 iaitu dengan menggunakan 1 wt. % jumlah mangkin, 15:1 nisbah molar metanol kepada minyak dan bertindak balas dalam masa 4 jam. Aktiviti-aktiviti mangkin ini didapati bergantung kepada kealkalian dan keluasan permukaan mangkin tersebut. Mangkin 3DSN telah digunakan semula sehingga 4 kali dan menghasilkan penukaran ester metil yang tinggi (lebih dari 80 %). Beberapa ujian

telah dijalankan untuk menganalisa sifat-sifat fizikokimia biodiesel yang dihasilkan seperti takat tuang, takat kilat, kelikatan, kandungan sulfur dan takat awan. Berdasarkan keputusan tersebut, biodiesel yang dihasilkan didapati setanding dengan diesel konvensional di pasaran memandangkan ianya dapat memenuhi piawaian biodiesel antarabangsa (ASTM, EN) dan juga spesifikasi untuk bahan api diesel di Malaysia (MS).



ACKNOWLEDGEMENT

Alhamdulillah. First and foremost, I would like to express my deepest gratitude to Allah Almighty, whom I might not be able to finish this study without His blessing, strength, patience and supremacy.

I would like to send my greatest appreciation to my project supervisor, Professor Taufiq Yap Yun Hin for his great assistance, invaluable advices and guidance, a lot of encouragements and supports as well as constructive comments throughout this study. I would also like to show my sincere gratitude to my co-supervisor, Professor Mohd Zobir Hussein for his contribution and cooperation in this study.

My warmest thankfulness to my lab officer, Tengku Sharifah Marliza, seniors; Nurul Suziana, Aqilah Noor, Lee Hwei Voon, Ahmad Zaidi, Ahmad Afandi, Sudarno; and my fellow lab-mates; Teo Siow Hwa, Surahim, Aqlili Riana, Noor Asikin, Sivasangar, Nurhaslinda, Siti Noor Syazwani, Nur Syazana, Nur Faizal and Lukman for their assistances, encouragements, supports and contribution to a very good working environment in the lab. Special thanks goes to my buddies; Rabiah Nizah, Siti nursa'udah, Aqilah Fasihah, Norfatiah, Nurul Hidayah, and Alia Farhanis. Thank you for always being next to me and inspire me to hang on until the very last step.

More or less, thank you to all lecturers and officers in Chemistry Department, Universiti Putra Malaysia (UPM) for all cooperations and assistances. Financial support from Universiti Putra Malaysia (UPM) through Graduate Research Fellowship (GRF), Ministry of Education through MyMaster programme and chemical (dolomite) support from The Division Group Sdn. Bhd. are gratefully acknowledged.

Last but not least, my primary indebtedness belongs to my parents, Hj. Zdainal Abidin bin Lehan and Hj. Sharipah binti Abdul Salam, my siblings for all continuously motivations, advices and supports all the way in completing my study in UPM.

I certify that a Thesis Examination Committee has met on 4 July 2014 to conduct the final examination of Shajaratun Nur bt Zdainal Abidin on her thesis entitled "Modification of Dolomites using Zinc and Tin as Efficient Solid Catalysts for Methyl Esters Production" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Intan Safinar Ismail, PhD

Associate Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Tan Yen Ping, PhD

Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Internal Examiner)

Zulkarnain bin Zainal, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Internal Examiner)

Mohd Ambar Yarmo, PhD

Professor
Universiti Kebangsaan Malaysia
Malaysia
(External Examiner)



NORITAH OMAR, PhD

Associate Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 19 September 2014

This thesis submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the Degree of Master Science. The members of the Supervisory Committee were as follows:

Taufiq Yap Yun Hin, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Mohd Zobir Hussein, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Member)

BUJANG BIN KIM HUAT, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

DECLARATION

Declaration by graduate student

I hereby confirm that:

- This thesis is my original work;
- Quotations, illustrations and citations have been duly referenced;
- This thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- Intellectual property from the thesis and copyright of the thesis are fully-owned by Universiti Putra Malaysia (Research) Rules 2012;
- Written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- There is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature: _____ Date: _____

Name and Matric No.: Shajaratun Nur binti Zdainal Abidin (GS28381)

Declaration by Members of Supervisory Committee

This is to confirm that:

- The research conducted and the writing of this thesis was under our supervision;
- Supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: _____
Taufiq Yap Yun Hin, PhD

Signature: _____
Mohd Zobir Hussein, PhD



TABLE OF CONTENTS

	Page
ABSTRACT	i
<i>ABSTRAK</i>	ii
ACKNOWLEDGEMENTS	iv
APPROVAL	v
DECLARATION	vii
LIST OF FIGURES	xii
LIST OF TABLES	xiii
LIST OF ABBREVIATIONS	xiv
CHAPTER	
1. INTRODUCTION	
1.1 Energy	1
1.2 Conventional Diesel	2
1.3 Background of Biodiesel	3
1.4 Palm Oil as Malaysian Biodiesel Feedstock	3
1.5 What is Dolomite?	5
1.6 Problem Statement and Scope of Research	6
1.7 Objectives of Research	7
2. LITERATURE REVIEW	
2.1 Transesterification Reaction	8
2.1.1 Transesterification of Palm Oil	9
2.2 Factors Affecting in Transesterification Reaction	9
2.2.1 Methanol to oil molar ratio	9
2.2.2 Reaction Time	10
2.2.3 Reaction Temperature	10
2.2.4 Catalyst Amount	10
2.3 Catalytic Transesterification Reaction	11
2.3.1 Homogeneous Catalyst	11
2.3.2 Heterogeneous Catalyst	12
2.4 Fundamental of CaO in Biodiesel Production	14
2.4.1 Commercialised CaO	14
2.4.2 Readily Occurred/ Waste CaO	15
2.5 Dolomite-Based Catalytic Reactions	16
2.5.1 Tar Cracking and Gasification	17

2.5.2	Transesterification	18
3.	MATERIALS AND METHODOLOGY	
3.1	Materials and Gases	20
3.2	Preparation of catalyst	20
3.3	Characterization of catalyst	21
3.3.1	X-Ray Diffraction (XRD) Analysis	21
3.3.2	X-Ray Fluorescence (XRF) Analysis	22
3.3.3	Surface Area and Pores Size Determination	22
3.3.4	Scanning Electron Microscopy (SEM)	23
3.3.5	Temperature Programmed Desorption of Carbon Dioxide (TPD-CO ₂)	23
3.4	Determination of Saponification Value	23
3.5	Transesterification Reaction	24
3.5.1	Activity of Catalyst	24
3.5.2	Reusability of Catalyst	25
3.6	Characterization of Biodiesel	25
3.6.1	Gas Chromatography-Flame Ionization Detector (GC-FID) Analysis	25
3.6.2	Atomic Absorption Spectroscopy (AAS)	25
3.6.3	Physicochemical Properties of Biodiesel	26
4.	RESULT AND DISCUSSION	
4.1	Unmodified Dolomite Catalyst	27
4.1.1	X-Ray Diffraction (XRD) Analysis	27
4.1.2	Catalytic Activity	28
4.2	Modified Dolomite Catalysts	29
4.2.1	X-Ray Diffraction (XRD) Analysis	29
4.2.2	Catalytic Activity	31
4.3	Characterization of the Catalyst	32
4.3.1	Surface Area and Pores Size Determination	32
4.3.2	Scanning Electron Microscopy (SEM)	33
4.3.3	Temperature Programmed Desorption of Carbon Dioxide (TPD-CO ₂)	34
4.3.4	X-Ray Fluorescence (XRF) Analysis	36
4.4	Catalytic Test	36
4.4.1	Effect of Reaction Parameters	36

4.4.2 Effect of Catalyst Surface Area and Basicity	41
4.5 Reusability of Catalyst	42
4.6 Biodiesel Characterizations	43
4.6.1 Gas Chromatography-Flame Ionization Detector (GC-FID) Analysis	43
4.6.2 Physicochemical Properties of Biodiesel	44
5. CONCLUSION	
5.1 Conclusions	47
5.2 Recommendations for Future Research	48
REFERENCES	49
APPENDIX	59
BIODATA OF STUDENT	63
LIST OF PUBLICATIONS	64

LIST OF FIGURES

Figure		Page
1.1	Non-renewable energy and renewable energy sources	1
1.2	Global total investment in clean energy 2004-2012 (USD Billion)	2
2.1	Transesterification reaction of triglycerides with alcohol to produce biodiesel	8
3.1	Flow chart of catalysts preparation	21
3.2	Flow chart of transesterification reaction	24
4.1	XRD patterns of dolomites	27
4.2	Transesterification reactions of AD4, AD6, AD8 and AD12 catalysts	28
4.3	XRD patterns of AD8 and DSN with different loading percentages	29
4.4	XRD patterns of AD8 and DZN with different loading percentages	30
4.5	Transesterification reactions of AD8, 1DSN, 3DSN and 5DSN catalysts	31
4.6	Transesterification reactions of AD8, 1DZN, 3DZN and 5DZN catalysts	32
4.7	SEM images of (a) FD (b) AD8 (c) 3DZN (d) 3DSN catalysts	34
4.8	TPD-CO ₂ patterns of (a) AD8, (b) 3DSN, (c) 3DZN catalysts	35
4.9	Transesterification reactions of AD8 catalyst affected by: (a) catalyst amount, (b) methanol to oil molar ratio, (c) time of reaction	37
4.10	Transesterification reactions of 3DSN catalyst affected by: (a) catalyst amount, (b) methanol to oil molar ratio, (c) time of reaction	38
4.11	Transesterification reactions of 3DZN catalyst affected by: (a) catalyst amount, (b) methanol to oil molar ratio, (c) time of reaction	39
4.12	The reusability of SD catalyst	42
4.13	GC chromatograms of (a) standard FAME, and (b) synthesized FAME from palm oil.	44

LIST OF TABLES

Table		Page
1.1	Production of oil per area from different crops in the world	4
3.1	Determination of physicochemical properties of biodiesel produced	26
4.1	Surface area and pores size determination of dolomites	33
4.2	TPD-CO ₂ analysis of dolomites	34
4.3	XRF analysis of dolomites	36
4.4	Comparison of AD8, 3DSN and 3DZN catalysts	41
4.5	Comparison of 3DSN catalyst with other unmodified and modified calcined dolomite catalysts in previous studies	42
4.6	Retention time of standard and synthesized methyl ester	44
4.7	Comparison of fuel specifications of synthesized biodiesel with ASTM 6751, EN 14214, MS 123:1993 standards and marketed Malaysian Diesel	45

LIST OF ABBREVIATIONS

AAS	Atomic Absorption Spectroscopy
ASTM	American Society for Testing and Materials
BET	Brunauer Emmett Teller
EN	European Standard
FAME	Fatty Acid Methyl Ester
FFA	Free Fatty Acid
GC	Gas Chromatography
JCPDS	Joint Committee on Powder Diffraction Standards
MS	Malaysian Standard
MPOB	Malaysian Palm Oil Board
SEM	Scanning Electron Microscopy
SV	Saponification Value
TPD	Temperature Desorption Programmed
WD	Wave Dispersive
XRD	X-Ray Diffraction
XRF	X-Ray Fluorescence

CHAPTER 1

INTRODUCTION

1.1 Energy

Energy has become global attention today because of high consumption of it, leading towards exhaustion of non-renewable energy sources such as coal, nuclear, natural gas and fossil fuel. Industrial, residential, transportation and commercial buildings sectors entail high energy supplies to retain their activities. Due to severely rely on these finite sources, thus, world is looking for a renewable energy to sustain the energy supplies from several sources of energy for example, solar, hydropower, wind, geothermal, biomass and biofuel energies (Ma and Hanna, 1999; Huber *et al.*, 2006).

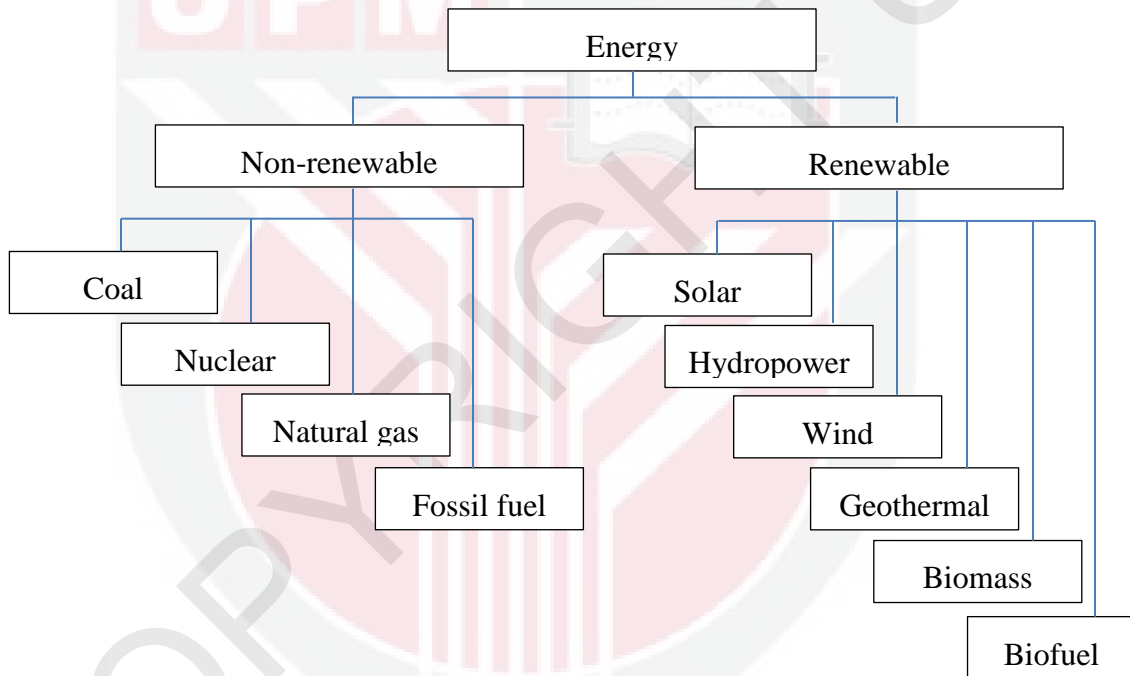


Figure 1.1. Non-renewable energy and renewable energy sources.

As reported in Global Trends in Renewable Energy Investment, (2013), there is a rapid growth of global investment in renewable energy sources until year 2012, mainly supported by corporate and government research and development (R&D). Efforts have been taken since year 2000, realizing that fossil fuel caused the climate change and it is running out sooner or later. Therefore, response from researches to this phenomenon has increased with the same purpose which to preserve the environment and searching for a sustainable energy for the future.

(Investment, %)

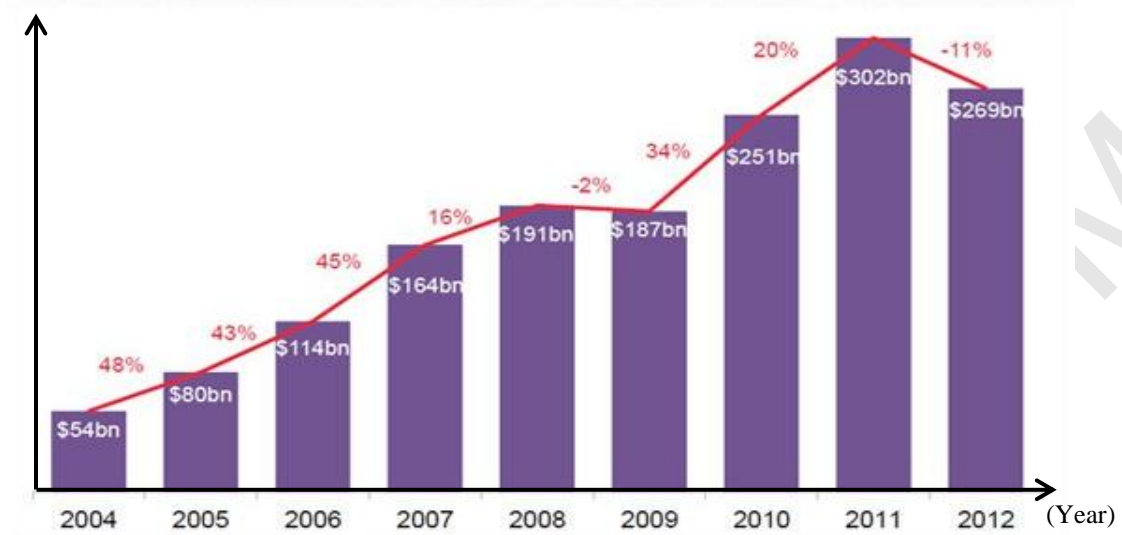


Figure 1.2. Global total investment in clean energy 2004 – 2012 (USD Billion).

(Source: *Global Trends in Renewable Energy Investment 2013*; Bloomberg New Energy Finance, 2013).

1.2 Conventional Diesel

Fossil fuel also known as petroleum has been used worldwide throughout these years as one of our source of energy, especially in transportation sector. Extracted from the core of earth, petroleum cannot be regenerated in a short time since it is naturally made from decaying ancient plants and animals over millions of years. Nowadays, demands on this fuel escalate rapidly for both industrial and living purposes, creating an economic crisis to the world. As predicted, intensive fuel consumption has resulted in rapid depletion of petroleum resources, thus, the price is getting higher, consequently (Lam *et al.*, 2010). Since petroleum is not a renewable energy; amount of petroleum consumed is greater than amount of petroleum being generated; diminishing of this fuel becomes a great global concern on how it will be replaced.

Petroleum was claimed as one of the most prevailing causes of environmental devastations as regards toxicity, oil spillage, air pollution, acid rain and waste oil, resulting by burning of the fossil fuel. The emission of traceable amount of poisonous chemicals and gases such as plumbum, benzene, carbon monoxides (CO), nitrogen oxides (NO_x) and sulfur oxides (SO_x) into the surroundings are hazardous for humans and the ecosystems. Excessive amount of CO₂ emits from vehicles, generators and power plants cause the climate change and global warming. Meanwhile, the petroleum also contributes negative charge to water and land since it is not biodegradable and hydrophobic to water (Lim and Teong, 2010).

Petroleum diesel is called petrodiesel to distinguish from other alternatives diesel (synthetic biodiesel). It has been used in most and heavy duty engines (buses, trucks,

trains, ships and barges) because of their greater energy and more power compared to gasoline (Kim *et al.*, 2004). In 2012, 19 % of petroleum products stated was petrodiesel production, second largest after gasoline as the main product (Global Trends in Renewable Energy Investment, 2013). However, due to the complications regarding petroleum products as mentioned before, the invention of biofuel such as biodiesel promotes a feasible solution to these matters (Silitonga *et al.*, 2013),

1.3 Background of biodiesel

Biodiesel is one of renewable energy resources. It has been considered as an alternative oil or replacement for the conventional diesel. In 1893, Rudolf Diesel tried by using biodiesel for his invented diesel engine; after he failed to use pure vegetable oil for the same purpose (Lin *et al.*, 2011). The vegetable oil is not suitable as fuel oil due to its high viscosity that blocked the engine, and then reduced the engine's performance. Similar characteristics of biodiesel with petrodiesel in some ways make it suitable for the replacement. Since then, production of biodiesel is boosted proportional to the world's demand. Furthermore, since the petroleum industries were mostly controlled by Middle-East regions, European and American countries were now struggled on the research and production of biofuels to cut demands on them later on.

Biodiesel comprised of fatty acid methyl ester (FAMES) through the reaction involves triglycerides that hail from plant oils or animal fats, instead of petroleum based fuel (Meher *et al.*, 2006). However, the animal fats are not commonly used due to high viscosity and readily solidify in room temperature. Therefore, biodiesel derived from plant oil offers solution to fossil fuel depletion and environmental degradation issues due to its ability to be renewed, sustained and biodegraded easily. Unlike petrodiesel, biodiesel does not contribute to excessive amount of carbon dioxide (CO₂) to the atmosphere since the CO₂ emission will be recycled for photosynthesis. Since there is no sulphur, plumbum and benzene contained in the biodiesel and it can lower the emission of carbon monoxide (CO) and nitrogen oxides (NO_x) to the atmosphere as well, therefore, biodiesel is not harmful to the environment and living things (Balat, 2011).

Despite the fact that biodiesel promotes the best answer to the issues related to petroleum based diesel, it was still not comparable with the conventional diesel since it is still expensive. This is due to higher cost of production and feedstocks hence increased the price of biodiesel. The cost of production is the cost of processes which includes the preparation, reaction, separation, and purification stages while the cost of feedstocks include the cost of vegetable oil, catalyst (for catalyzed reaction) and methanol (Atadashi *et al.*, 2010).

1.4 Palm Oil as Malaysian Biodiesel Feedstock

There are various types of edible oil that have been studied as feedstock for biodiesel production worldwide for example rapeseed oil, soybean oil, peanut oil, coconut oil, sunflower oil and corn oil (Yusuf, 2011). However, the consumption of rapeseed and soy bean oil in US and EU countries to produce biodiesel resulting in high price of biodiesel, proposing that it derives from high quality vegetable oils. In Malaysia, high

production of palm oil (*Elaeis guineensis*) annually gave some ideas on developing biodiesel production by using palm oil as the feedstock due to its availability and economic price rather than using other vegetable oils, suggesting that the production of the biodiesel using palm oil will be sustained (Abdullah, 2009).

Palm oil industry has been debated intensely regarding palm oil plantation that triggered deforestation and created global warming issues for over the past few years. However, palm oil has distinct advantages in biodiesel production industry including larger percentage of forest (60 %) in Malaysia was able to be reserved compared to many developing countries since less land was required for palm oil plantation than the other crops as tabulated in Table 1 (Pahl, 2005). Palm oil crop was also proclaimed to absorb a lot of CO₂ gas for production of biomass and release more oxygen during photosynthesis compared to the other crops such as rapeseed and soybean oil (Basiron, 2007). Still, the usage of oil palm to produce biodiesel turns out to be a massive concern worldwide because they have to compete with the food resources, thus non-edible oils such as jathropa and algae oils were introduced to replace them (Gui *et al.*, 2008)

Table 1.1. Production of oil per area from different crops in the world.

Source	Yield (liters/hectare)	Comparison of yields
Palm	5,950	1.00
Coconut	2,689	0.45
Jatropha	1,818	0.31
Rapeseed	1,190	0.20
Soybean	446	0.07

(Source: Pahl, 2005).

Malaysia is the second world's largest palm oil producer after Indonesia (Palm Oil World, 2011). Despite the argument that oil palm is a food resource, it offers the most reasonable price among market vegetable oils and non-edible oils due to its efficient production (Table 1). In fact, land area devoted for oil palm plantation is the lowest compared to the other crops. High yields of palm oil per hectare only utilizes around 0.26 hectares of land to have a tonne of oil in return whereas low-yielding crops such as sunflower, soybean and rapeseed consume even more than a hectare to produce the same amount of oil (Malaysian Palm Oil Council, 2012). With the rapid development of technologies nowadays, the production of palm oil is believed to increase in future. Therefore, it is better to produce palm oil intensively which can guarantee a great return in the future.

Biodiesel implementation in Malaysia is still in its early age. In 2006, National Biofuel Policy announcement has been made by the government, thus many researches were done and numbers of biodiesel plants were established ever since. In order to reduce of dependence on petrodiesel, several petrol pumps in Selangor have started using B5 and B10 in 2010 followed by the other central region including Putrajaya, Kuala Lumpur, Melaka and Negeri Sembilan in 2011 (Wahab, 2012), where:

B5 comprised of 5 % of biodiesel blended with 95 % of petrodiesel, B10 comprised of 10 % of biodiesel blended with 90 % of petrodiesel, and B100 comprised of 100 % of biodiesel.

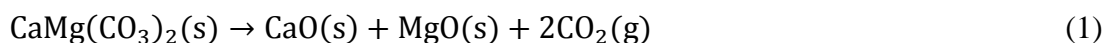
Unfortunately, due to uncertainty price of biodiesel feedstock and low in demand, some of the plants were forced to close down and there are only a few biodiesel plants that are under operation thus far. Malaysian government's decision to postpone the application of biodiesel nationally from 2011 to 2014 also retards the development of biodiesel industry in this country to date (Malaysia Debt Ventures Berhad, 2011).

Though the implementation of biodiesel is proposed to increase from B5 to B10 in 2014 by the government, the programme seems doubtful to be fulfilled and has been criticized by industry observers since the old programme (B5) still struggling in the market currently. Moreover, B5 programme is not nationwide implemented yet, signifying that the new programme (B10) is totally irrational. Significantly, decreasing in global biodiesel exports and investment from year 2011 to year 2012 are the reflections of this problem. Therefore, since B10 programme will be launched in this few months away, the implementation of biodiesel must be increasing nationwide from this point to make this programme possible.

1.5 What is dolomite?

Dolomite is a type of rock that occurs naturally all around the world. It appears in different colours (white, brown, yellow, pink, grey, etc.) and shapes like saddles or flakes. Chemically, it consists of calcium carbonate (CaCO_3) and magnesium carbonate (MgCO_3) as the main compounds with the presence of very small percentages of other compounds and has molecular weight around 184.4 g/mol. Dolomite comprises of alternating layers of Ca, Mg and CO_3 which CO_3 comes between Ca and Mg elements in a sequence (Wilson, 2008).

In Malaysia, the price of dolomite is very cheap; US \$20 - 40 per metric ton, since it is abundant and easy to get. Numbers of quarries for dolomite manufacturing can be found at Hulu Langat, Selangor and Chuping, Perlis. Due to these reasons, dolomite is mainly used in constructions as well as agricultural fields (Ngamcharussrivicahi *et al.*, 2007). It is used in agricultural field as fertilizers which supplies Mg mineral to the plants that is necessary for their growth. In catalytic applications, dolomite is a promising heterogeneous catalyst since it promotes a non-toxic, basic catalyst which important for the reaction processes. Similar to other natural sources of CaCO_3 such as limestone and seashell, the active phases of the dolomite were obtained by thermal decomposition of dolomite into CaO and MgO as shown in Equation 1 below (Ilgen, 2011):



In renewable energy's field, dolomites were commonly used in biomass and tar reforming areas. There were a few studies focusing on using activated dolomite as catalyst in biodiesel production. By using dolomite as catalyst for biodiesel production, the cost of chemical used in the reaction will be reduced.

1.6 Problem Statements

Global three-facet problems concerning energy are rising recently, they are:

1. Depletion of non-renewable energy resources
2. High cost of petroleum based fuel
3. Negative environmental impacts

To date, energy demands were increasing significantly since 1900s, resulting in the depletion of petroleum resources. The rate of this non-renewable energy has been consumed is greater than it has been produced or discovered. Consequently, the price of petroleum is getting higher. The consumption of fossil fuel also triggered environmental degradation which can be harmful to the ecosystems. These problems urge the findings of new resources especially renewable energy which can be sustained in the future.

Biodiesel is one of the renewable energy which able to solve this problem. However, biodiesel is not comparable with the petroleum based diesel due to its high production cost and feedstock price. Palm oil as feedstock can reduce the cost of feedstock to be compared to other edible and non-edible oils such as soybean and Jatropha oils.

It was claimed that catalytic transesterification reaction can promote an efficient way to reduce the cost of biodiesel production, yet the usage of homogeneous catalyst entails high cost of separation of the catalyst from the reaction medium. Invention of heterogeneous catalyst will reduce the cost of purification; however, it is usually higher in price and also requires higher reaction conditions. In order to overcome these problems, a cheap, heterogeneous catalyst which has good properties for transesterification reaction has been actively investigated.

Dolomite is a cheap, heterogeneous, basic catalyst which provides good properties for transesterification reaction. By incorporating Sn and Zn (amphoteric metal oxides) onto the dolomite, the properties of the catalyst was enhanced, hence the reaction conditions were reduced.

1.7 Scope of Research

Biodiesel production by using base-catalyzed transesterification method has been investigated in this study. The advantages of using palm oil as feedstock has been revised comprehensively and the influence of dolomites based catalysts with and without dopants in the biodiesel production were also discussed. Dolomite catalyst was modified by doping metal oxides with amphoteric properties (SnO_2 and ZnO), in order to increase the properties of the catalyst, thus improve the production of biodiesel. Furthermore, the effects of reaction conditions such as methanol to oil molar ratio, reaction time and catalyst amount were also investigated.

Chapter 1 introduces the energies which comprised of non-renewable and renewable energy sources. The biodiesel backgrounds, advantages and implementations in Malaysia were also reached. In Chapter 2, the previous studies related on studies and researched are reviewed. The experimental part is in Chapter 3, where the particular

information of all materials, methods and instruments used in this study were presented. The preparation and the characterization of catalyst followed by transesterification reaction were described as well. The results and discussions part in Chapter 4 will discuss over all the results obtained. Finally, the conclusions and recommendations of this study will be made as whole in Chapter 5.

1.7 Objectives of Research

The objectives of this research are:

1. To modify dolomite catalyst by doping selective metal oxides which were SnO₂ and ZnO.
2. To characterize catalysts using various techniques:
 - a. X-Ray diffraction (XRD) analysis
 - b. Scanning electron microscopy (SEM) analysis
 - c. Temperature programmed desorption of carbon dioxide (TPD-CO₂) analysis
 - d. Brunauer-Emmett-Teller (BET) surface area analysis
 - e. X-Ray fluorescence (XRF) analysis
3. To determine the reaction parameters in optimizing the production of biodiesel:
 - a. reaction time
 - b. catalyst amount
 - c. methanol to oil molar ratio
 - d. reusability of the catalyst

REFERENCES

- Abdullah, A.Z., Salamatinia, H., Mootabadi, H. and Bhatia, S. (2009). Current status and biodiesel industry in Malaysia as the world's leading producer of palm oil. *Energy Policy*. 37: 5440-5448.
- Alamu, O.J., Waheed, M.A., Jekayinfa, S.O. and Akintola, T.A. (2007). Optimal transesterification duration for biodiesel production from Nigeria palm kernel oil. *International Journal of Agricultural Engineering: the CIGR E-journal*. 9: Manuscript EE 07 018.
- Alamu, O.J., Waheed, M.A. and Jekayinfa, S.O. (2008). Effect of ethanol-palm kernel oil ratio on alkali-catalysed biodiesel yields. *Fuel*. 87: 1529-1533.
- Aminul, I., Taufiq-Yap, Y.H., Chi-Ming, C., Pogaku, R. and Eng-Seng, C. (2013). Studies on design of heterogeneous catalysts for biodiesel production. *Process Safety and Environmental Protection*. 91: 131-144.
- Atadashi, I.M., Aroua, M.K. and Abdul Aziz, A. (2010). High quality biodiesel and its diesel engine application: a review. *Renewable and Sustainable Energy Reviews*. 14: 1999-2008.
- Balat, M. Potential alternatives to edible oils for biodiesel production – a review of current work. (2011). *Energy Conversion and Management*. 52: 1479=1492.
- Basiron, Y. (2012). Deforestation and agricultural development: why the double standards? *Palm Oil – Getting the Facts Right*. <http://www.ceopalmoil.com/2012/11/deforestation-and-agricultural-development-why-the-double-standards/>. (Dated: November 7th, 2012)
- Basiron, Y. The palm-oil advantage in biofuel. *New Straits Times*, February 24th, 2007, pp. 22.
- Bhatti, H.N., Hanif, M.A., Qasim, M. and Rehman, A. (2008). Biodiesel production from waste tallow. *Fuel*. 87: 2961-2966.
- Birla A., Singh, B., Upadhyay, S.N. and Sharma, Y.C. (2012). Kinetics studies of synthesis of biodiesel from waste frying oil using a heterogeneous catalyst derived from snail shell. *Bioresource Technology*. 106: 95-100.
- Bo, X., Guomin, X., Lingfeng, C., Ruiping, W. and Lijing, G. (2007). Transesterification of palm oil with methanol to biodiesel over a KF/Al₂O₃ heterogeneous base catalyst. *Energy and Fuels*. 21(6): 3109-3112.
- Boey P.L., Maniam., G.P. and Hamid, S.A. (2009). Biodiesel production via transesterification of plam olein using waste mud crab (*Scylla serrata*) shell as a heterogeneous catalyst. *Bioresource Technology*. 100: 6362-6368.

Boey P.L., Maniam G.P, Hamid, S.A. and Ali, D.M.H. (2011). Utilization of waste cockle shell (*Anadara granosa*) in biodiesel production from palm olein: Optimization using response surface methodology. *Fuel*. 90: 2353-2358.

Buasri, A., Chaiyut, N., Loryuenyong, V., Wongweang, C. and Khamsrisuk, S. (2013). Application of eggshell wastes as a heterogeneous catalyst for biodiesel production. *Sustainable Energy*. 1(2): 7-13.

Cantrell, D.G., Gillie, L.J., Lee, A.F and Wilson, K. (2005). Structure-reactivity correlations in MgAL hydrotalcite catalysts for biodiesel synthesis. *Applied Catalysis A*. 287: 183-190.

Chakraborty, R., Bepari, S. and Banerjee, A. (2011). Application of calcined waste fish (*Labeo rohita*) scale as low-cost heterogeneous catalyst for biodiesel synthesis. *Bioresource Technology*. 100: 6362-6368.

Chitra, P., Venkatachalam, P. and Sampathrajam, A. (2005). Optimization of experimental procedure for biodiesel production from alkaline-catalysed transesterification in supercritical methanol. *International Journal of Energy Conversion and Management*. 43: 2349-2356.

Corella, J., Toledo, J.M. and Padilla, R. (2004). Olivine or dolomite as in-bed additive in biomass gasification with air in a fluidized bed: which is better?. *Energy Fuel*. 18(3):713-720.

Corma, A., Bee Abd Hamid, S., Iborra, S. and Velty, A. (2005). Lewis and Bronsted basic active sites on solid catalysts and their role in the synthesis of monoglycerides. *Journal of Catalysis*. 234: 340-347.

Darnoko, D. and Cheryan, M. (2000). Kinetics of palm oil transesterification in a batch reactor. *Journal of the American Oil Chemists' Society*. 77: 1263-1267.

Demirbas, A. (2007). *Biodiesel: a realistic fuel alternative for diesel engines*. Springer.

Demirbas, A. (2009). Progress and recent trends in biodiesel fuels. *Energy Conversion and Management*. 50: 14-34.

Dorado, M.P., Ballesteros, E. and Mittelbach, M. (2004). Kinetic parameters affecting the alkali-catalysed transesterification process of used olive oil. *Energy Fuels*. 18: 1457-1462.

Eevera, T., Rajendran, K. and Saradha, S. (2009). Biodiesel production process optimization and characterization to access the suitability of the product for varied environmental conditions. *Renewable Energy*. 34: 762-765.

Encinar, J.M., Gonzalez, J.F. and Rodriguez-Reinares, A. (2005). Biodiesel from used frying oil. Variables affecting the yields and characteristics of the biodiesel. *Industrial and Engineering Chemistry Research*. 44(15): 5491-5499.

Encinar, J.M., Juan, F., Gonzalez, J.F. and Rodriguez-Reinares, A. (2007). Ethanolysis of used frying oils: biodiesel preparation and characterization. *Fuel Processing Technology*. 88(5): 513-522.

Enweremadu, C.C. and Mbarawa, M.M. (2009). Technical aspects of production and analysis of biodiesel from used cooking oil – a review. *Renewable and Sustainable Energy Reviews*. 13:2205-2224.

Gao, L., Xu, B., Xiao, G. and Lv, J. (2008). Transesterification of palm oil with methanol to biodiesel over a KF/hydrotalcite solid catalyst. *Energy Fuels*. 22: 3531-3535.

Gerpen, J.V. (2005). Biodiesel processing and production. *Fuel Processing Technology*. 86: 1097-1107

Global Trends in Renewable Energy Investment 2013; Bloomberg New Energy Finance, London, 2013

Granados, M.L., Zafra, Poves, M.D., Alonso, D.M., Mariscal, R., Galisteo, F.C., Moreno-Tost, R., Santamaría, J. and Fierro, J.L. (2007). Biodiesel from sunflower oil by using activated calcium oxide. *Applied Catalysis B*. 73: 317-326.

Gryglewicz, S. (1999). Rapeseed oil methyl esters preparation using heterogeneous catalysts. *Bioresource Technology*. 79: 249-253.

Gui, M.M., Lee, K.T. and Bhatia, S. (2008). Feasibility of edible oil vs. non-edible oil vs. waste edible oil as biodiesel feedstock. *Energy*. 33: 1646-1653.

Helwani, Z., Othman, M.R., Aziz, N., Kim, J. and Fernando, W.J.N. (2009). Solid heterogeneous catalysts for transesterification of triglycerides with methanol: a review. *Applied Catalyst A*. 363: 1-10.

Hingu, M.H., Gorate, P.R. and Rathod, V.K. (2010). Synthesis of biodiesel from waste cooking oil using sonochemical reactors. *Ultrasonics Sonochemistry*. 17(5): 827-832.

Huber, G., Iborra, S. and Corma, A.V. (2006). Synthesis of transportation fuels from biomass: chemistry, catalysts, and engineering. *Chemical Reviews*. 106: 4044-4098.

Ilgen, O. (2011). Dolomite as a heterogeneous catalyst for transesterification of canola oil. *Fuel Processing Technology*. 92: 452-455.

Isa, F.M. *Malaysian fuel quality and bio-fuel initiative*. 5th Asian Petroleum Technology Symposium, Jakarta. January 2007.

Kafuku, G., Lee, K.T. and Mbarawa, M. (2010). The use of sulfated tin oxide as solid superacid catalyst for heterogeneous transesterification of *Jathropa curcas* oil. *Chemical Papers*. 64(6): 734-740.

Karatas, H., Olgun, H. and Akgun, F. (2013). Coal and coal and calcined dolomite gasification experiments in a bubbling fluidised bed gasifier under air atmosphere. *Fuel Processing Technology*. 105: 555-572.

Kim, H.J., Kang, B.S., Kim, M.J., Park, Y.M., Kim, D.K., Lee, J.S. and Lee, K.Y. (2004). Transesterification of vegetable oil to biodiesel using heterogeneous base catalyst. *Catalysis Today*. 106: 190-192.

Koh, T.S. and Chung, K.H. (2008). Production of biodiesel from waste frying oil by transesterification on zeolite catalysts with different acidity. *Journal of Industrial and Engineering Chemistry*. 19(2): 214-221.

Komintarachat, C. and Chuepeng, S. (2009). Solid acid catalyst for biodiesel production from waste used cooking oils. *Industrial and Engineering Chemistry Research*. 48(20): 9350-9353.

Komintarachat, C. and Chuepeng, S. (2010). Methanol-based transesterification optimization of waste used cooking oil over potassium hydroxide catalyst. *American Journal of Applied Sciences*. 7(8): 1073-1078.

Kouzu, M., Kasuno, T., Tajika, M., Sugimoto, Y., Yamanaka, S. and Hidaka, J. (2008). Calcium oxide as a solid base catalyst for transesterification of soybean oil and its application to biodiesel production. *Fuel*. 87: 2798-2806.

Kouzu, M. and Hidaka, J.-s. (2012). Transesterification of vegetable oil into biodiesel catalysed by CaO: a review. *Fuel*. 93: 1-12.

Kulkarni, M.G. and Dalai, A.K. (2006). Waste cooking oil - an economical source for biodiesel: a review. *Industrial and Engineering Chemistry Research*. 45: 2901-2913.

Lam, M.K., Teong, L.K. and Rahmanmohamed, A. (2010). Life cycle assessment for the production of biodiesel: a case study in Malaysia for palm oil versus jathropa oil. *Biofuels, Bioproducts and Biorefining*. 3(6): 610-612.

Lee, D.W., Park, Y.M. and Lee, K.Y. (2009). Heterogeneous base catalysts for transesterification in biodiesel synthesis. *Catalysis Surveys from Asia*. 13: 63-77.

Leclercq, E., Finiels, A. and Moreau, C.J. (2001). Transesterification of rapeseed oil in the presence of basic zeolites and related solid catalysts. *Journal of American Oil Chemists' Society*. 78: 1161-1165.

Leung, D.Y.C. and GuO, Y. (2006). Transesterification of neat and used frying oil: optimization for biodiesel production. *Fuel Processing Technology*. 87: 883-890.

Lim, S. and Teong, L.K. (2010). Recent trends, opportunities and challenges of biodiesel in Malaysia: an overview. *Renewable and Sustainable Energy Reviews*. 14: 938-954.

Li, H. and Xie, W. (2008). Fatty acid methyl ester synthesis over Fe^{3+} - vanadyl phosphate catalysts. *Journal of the American Oil Chemists' Society*. 71: 1179-1187.

Li, E., Xu, Z.P. and Rudolph, V. (2009) MgCoAl LDH derived heterogeneous catalysts for the ethanol transesterification of canola oil to biodiesel. *Applied Catalysis*. 88: 42-49.

Li, L., Kunii, H., Yamauchi, M., Kim, H.-J. and Shimizu, T. (2013). In Steam Gasification for Biomass Tar with Natural Ores of Limonite and Dolomite. *Advanced Materials Research*. 608-609: 201-205.

Li, Y., Tian, G., Xu, H. (2012). *Application of Biodiesel in Automotive Diesel Engines, Biodiesel-feedstocks, Production and Applications*. Prof. Zhen Fang (Ed.). InTech Publishing.

Lin, L., Cunshan, Z., Vittayapadung, S., Xiangqian, S. and Mingdong, D. (2011). Opportunities and challenges for biodiesel fuel. *Applied Energy*. 88: 1020-1031.

Liu, X., He, H., Wang, Y. and Zhu, S. (2007). Transesterification of soybean oil to biodiesel using SrO as a solid base catalyst. *Catalysis Communications*. 8: 1107-1111.

Lotero, E., Goodwin, J.G., Bruce, D.A., Suwannakarn, K., Liu, Y. and Lopez, D.E. (2006). The catalysis of biodiesel synthesis. *Catalysis*. 19:41-83.

Lowell, S., Shields, J.E., Thomas, M.A. and Thommes. (2004). *Characterization of Porous Solids and Powders: Surface Area, Pore Size and Density*. Kluwer Academic Publisher. Dordrecht.

Ma, F. and Hanna, M.A. (1999). Biodiesel production: a review. *Bioresource Technology*. 70(1): 1-15

Macleod, C.S., Harvey, A.P., Lee, A.F. and Wilson, K. (2008). Evaluation of the activity and stability of alkali-doped metal oxide catalysts for application to an intensified method of biodiesel production. *Chemical Engineering Journal*. 135: 63-70.

Malaysia Debt Ventures Berhad (MDVB). (2011). MPOB: Only 10 biodiesel plants operating. www.mdv.com.my/v2/archives/news_post/ (Dated: March 22nd, 2011)

Malaysian Palm Oil Council (MPOC). (2012). The oil palm tree, http://www.mpoc.org.my/The_Oil_Palm_Tree.aspx. (Date viewed: April 10th, 2013)

Margaretha Y.Y., Prastyo, H.S., Ayucitra, A. and Ismadji, S. (2012). Calcium oxide from *Pomacea sp.* shell as a catalyst for biodiesel production. *International Journal of Energy and Environmental Engineering*. 3: 33-41.

Math, M.C., Kumar, S.P. and Chetty, S.V. (2010). Technologies for biodiesel production from used cooking oil - a review. *Energy for Sustainable Development*. 14: 339-345.

Meher, L.C., Sagar, D.V. and Naik, S.N. (2006). Technical aspects of biodiesel production by transesterification – a review. *Renewable and Sustainable Energy Review*. 10: 248-268.

Meng, X., Chen, G. and Wang, Y. (2008). Biodiesel production from waste cooking oil via alkali catalyst and its engine test. *Fuel Processing Technology*. 89(9): 851-857.

Mohamed Shamsuddin, S.Z. and Nagaraju, N. (2006). Transesterification: salol synthesis over solid acids. *Catalysis Communications*. 7(3). 593-599.

Moriyasu, H., Koshi, K. and Kouzu, M. (2012). CaO catalysts prepared from a variety of limestone-deriving industrial materials for transesterification of soybean oil with methanol. *Journal of the Japan Institute of Energy*. 91: 34-40 (2012).

Myrén, C., Hörnell, C., Björnbom, E. and Sjöström, K. (2002). Catalytic tar decomposition of biomass pyrolysis gas with a combination of dolomite and silica. *Biomass and Bioenergy*. 23: 217-227.

Nair, P., Singh, B., Upadhyay, S.N. and Sharma, Y.C. (2012). Synthesis of biodiesel from low FFA waste frying oil using calcium oxide derived from *Meretrix meretrix* as a heterogeneous catalyst. *Journal of Cleaner Production*. 29-30: 82-90.

Nakatani, N., Takamori, H., Tadeka, K. and Sakugawa, H. (2009). Transesterification of soybean oil using combusted oyster shell waste as a catalyst. *Bioresource Technology*. 100: 1510-1513.

Ngamcharussrivichai, C., Wiwatnimit, W. and Wangnoy, S. (2007). Modified dolomites as catalysts for palm kernel oil transesterification. *Journal of molecular catalysis A*. 276: 24-33.

Ngamcharussrivichai, C., Nunthasanti, P., Tanachai, S. and Bunya, K. (2010). Biodiesel production through transesterification over natural calciums. *Fuel Processing Technology*. 91: 1409-1415.

O'Neill, R.E., Vanoye, L., Bellefon, C.D. and Aiouache, F. (2014). Aldol-condensation of furfural by activated dolomite catalyst. *Applied Catalysis B. Environmental*. 144: 46-56.

Obadiah, A., Swaroopa, G.A., Kumar, S.V., Jeganathan, K.R. and Ramasubbu, A. (2012). Biodiesel production from palm oil using calcined waste animal bone as catalyst. *Bioresource Technology*. 116: 512-516.

Om Tapanes, N.C., Gomes, Aranda, D.A., de Mesquita Carneiro, J.W. and Ceva Antunes, O.A. (2008). Transesterification of Jathropa curcas oil glycerides: theoretical and experimental studies of biodiesel reaction. *Fuel*. 87(10-11): 2286-2295.

Pahl, G. (2005). *Biodiesel: Growing a New Energy Economy*. Vermont. Chelsea Green Publishing.

Palm Oil World. (2011). Malaysian Palm Oil Industry: Oil Palm in Malaysia. http://www.palmoilworld.org/about_malaysian-industry.html. (Date viewed: December 25th, 2012)

Pinto, A.C., Guarieiro, L.L.N., Rezende, J.C., Ribeiro, N.M., Torres, E.A., Lopes, W.A., Pereira, P.A.D.P. and Andrade, J.B.D. (2005). Biodiesel: an overview. *Journal of the Brazilian Chemical Society*. 16: 1313-1330.

Pinto, F., Lopes, H., Andre, R.N., Gulyurtlu, I. and Cabriata, I. (2007). Effect of catalysts in the quality of syngas and by-products obtained by co-gasification of coal and wastes. 1. Tars and nitrogen compounds abatement. *Fuel*. 86: 2052-2063.

Ramos, M.J., Casas, A., Rodríguez, L., Romero, R. and Pérez, Á. (2008). Transesterification of sunflower oil over zeolites using different metal loading: a case of leaching and agglomeration studies. *Applied Catalysis A*. 346: 79-85.

Refaat, A.A. (2009). Different techniques for the production of biodiesel from waste vegetable oil. *International Journal of Environmental Science and Technology*. 7(1): 185-213.

Romero, R., Martínez, S.L. and Natividad, R. (2011). Alternative Fuel. Dr. Maximino Manzanera. *Biodiesel Production by Using Heterogeneous Catalysts*. (pp. 3-20). InTech Publishers.

Seki, T., Kabashima, H., Akutsu, K., Tachikawa, H. and Hattori, H. (2001). MixedTishchenko reaction over solid base catalysts. *Journal of Catalysis*. 204: 393-401.

Serio, M.D., Ledda, M., Cozzolino, M., Minutillo, G., Tesser, R. and Santacesaria, E. (2006). Transesterification of Soybean Oil to Biodiesel by Using Heterogeneous Basic Catalyst. *Industrial and Engineering Chemistry Research*. 45: 3009-3014.

Shah, S. and Gupta, M.N. (2007). Lipase catalysed preparation of biodiesel from jathropa oil in a solvent free system. *Process Biochem*. 42: 409-414.

Sharma, Y.C., Singh B. and Korstad, J. (2010). Application of an efficient nonconventional heterogeneous catalyst for biodiesel synthesis from pongamia pinnata oil. *Energy Fuels*. 24: 3223-3231.

Shumaker, J.L., Crofcheck, C., Tackett, S.A., Santillan-Jimenez, E. and Crocker, M. (2007). Biodiesel production from soybean oil using calcined Li-AL layered double hydroxide catalysts. *Catalysis Letters*. 115: 56-61.

Siano, D., Nastasi, M., Santacesaria, E., Serio, M.D., Tesser, R., Minutillo, G., Ledda, M. and Tenore, T. (2006). *Process for producing esters from vegetable oils or animal fats using heterogeneous catalysts*. PCT Application No. WO2006/050925.

Silitonga, A.S., Masjuki, H.H., Mahlia, T.M.I., Ong, H.C. and Chong, W.T. (2013). Experimental study on performance and exhaust emissions of a diesel engine fuelled with *Ceiba pentandra* biodiesel blends. *Energy Conversion and Management*. 76: 828-836.

Sun, Y., Jiang, J., Kantarelis, E., Xu, J., Li, L., Zhao, S. and Yang, W. (2012). Development of a bimetallic dolomite based tar cracking catalyst. *Catalysis Communications*. 20: 36-40.

Suppes, G.J., Dasari, M.A., Duskocil, E.J., Mankidy, P.J. and Goff, M.J. (2004). Transesterification of sunflower oil over zeolites using different metal loading: a case of leachinh and agglomeration studies. *Applied Catalysis A*. 346: 79-85.

Tateno, T. and Sasaki, T. (2004). *Process for producing fatty acid fuels comprising fatty acids esters*. U.S Patent 6, 818, 026.

Taufiq-Yap, Y.H., Lee, H.V., Hussein, M.Z. Yunus, R. (2011). Calcium-based mixed oxide catalysts for methanolysis of *Jathropa curcas* oil to biodiesel. *Biomass and Bioenergy*. 35(2): 827-834.

Taufiq-Yap, Y.H., Lee, H.V., Yunus, R. and Juan, J.C. (2011). Transesterification of non-edible *Jathropa curcas* oil to biodiesel using binary Ca-Mg mixed oxide catalyst: effect of stoichiometric composition. *Chemical Engineering Journal*. 178: 342-347.

Taufiq-Yap, Y.H., Nur-Faizal, A.R., Sivasangar, S., Hussein, M.Z. and Aishah, A. (2013). Modification of Malaysian dolomite using mechanochemical treatment via different media for oil palm fronds gasification. *International Journal of Energy Research*. DOI: 10.1002/er.

Tsuji, H., Yagi, F., Hattori, H. and Kita, H. (1994). Self-condensation of n-butyraldehyde over solid base catalysts. *Journal of Catalysis*. 148: 759-770.

Wahab, A.G. (2012). Malaysia Biofuel Annual Report. *Global Agricultural Information Network*. GAIN Report No: MY2006. Kuala Lumpur.

Wang, T.J., Chang, J., Wu, C.Z., Fu, Y. and Chen, Y. (2005). The steam reforming of naphthalene over a nickel-dolomite cracking catalyst. *Biomass and Bioenergy*. 28: 508-514.

Wang, Y., Ou, S., Liu, P., Xue, F. and Tang, S. (2006). Comparison of two different processes to synthesize biodiesel by waste cooking oil. *Journal of Molecular Catalysis A: Chemical*. 252: 107-112.

Wang, R., Li, H., Chang, F., Luo, J., Hanna, M.A., Tan, D., Hu, D., Zhang, Y., Song, B. and Yang, S. (2013). A facile, low-cost route for the preparation of calcined porous calcite and dolomite and their application as heterogeneous catalysts in biodiesel production. *Catalysis Science and Technology*. 3: 2244-2251.

Watkins, R.S., Lee, A.F. and Wilson, K. (2004). Li-CaO catalysed triglyceride transesterification for biodiesel applications. *Green Chemistry*. 6: 335-340.

Wei, Z., Xu, C. and Li, B. (2009). Application of waste eggshell as low-cost solid catalyst for biodiesel production. *Bioresource Technology*. 100: 2883-2885.

Wilson, K., Hardacre, C., Lee, A.F., Montero, J.M. and Shellard, L. (2008). The application of natural dolomitic rock as a solid base catalyst in triglyceride transesterification for biodiesel synthesis. *Green Chemistry*. 10: 654-659.

Wongmanerod, C., Zangoie, S. and Arwin, H. (2001). Determination of pore size distribution and surface area of thin porous silicon layers by spectroscopic ellipsometry. *Applied Surface Science*. 172: 117-125.

Xie, W., Peng, H. and Chen, I. (2006). Transesterification of soybean oil catalysed by potassium loaded on alumina as a solid-base catalyst. *Applied Catalysis A*. 300: 67-74.

Xie, W., Huang, X. and Li, H. (2007). Soybean oil methyl esters preparation using NaX zeolite loaded with KOH as a heterogeneous catalyst. *Bioresource Technology*. 98: 936-939.

Yan, S., Lu, H. and Liang, B. (2008). Supported CaO catalysts used in the transesterification of rapeseed oil for the purpose of biodiesel production. *Energy and Fuels*. 22: 646-651.

Yan, S., DiMaggio, C., Mohan, S., Kim, M., Salley, S.O. and Ng, KYS. (2010). Advancements in heterogeneous catalysis for biodiesel synthesis. *Topics in Catalysis*. 53: 721-736.

Yang, L., Zhang, A. and Zheng, X. (2009). Shrimp shell catalyst for biodiesel production. *Energy Fuels*. 23: 3859-3865.

Yu, Q., -Z., Brage, C., Nordgreen, T. and Sjoström, K. (2009). Effects of Chinese dolomites on tar cracking in gasification of birch. *Fuel*. 88: 1922-1926.

Yuan, X., Liu, J., Zeng, G., Shi, J., Tong, J. and Huang, G. (2008). Optimization of conversion of waste rapeseed oil with high FFA to biodiesel using response surface methodology. *Renewable Energy*. 33: 1678-1684.

Yusuf, N.N.A.N, Kamarudin, S.K. and Yaakub, Z. (2011). Overview on the current trends in biodiesel production. *Energy Conversion and Management*. 52: 2741-2751.

Zagonel, G.F., Peralta-Zamora, P.G. and Ramos, L.P. (2002). Production of ethyl esters from crude soybean oil: optimization of reaction yields using a 23 experimental design and development of a new analytical strategy for reaction control. Preprints Symposium: *American Chemical Society, Division of Fuel Chemistry*. 47: 363-364.

Zheng, S., Kates, M., Dubé, M.A. and McLean, D.D. (2006). Acid-catalysed production of biodiesel from waste frying oil. *Biomass and Bioenergy*. 30: 267-272.

