



UNIVERSITI PUTRA MALAYSIA

***EFFECTS OF ACID RAIN ON GEOTECHNICAL PROPERTIES OF
TROPICAL RESIDUAL SOILS***

ZEINAB BAKHSHIPOUR

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**EFFECTS OF ACID RAIN ON GEOTECHNICAL PROPERTIES OF
TROPICAL RESIDUAL SOILS**

By

ZEINAB BAKHSHIPOUR

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfillment of the Requirements for the Degree of Doctor of Philosophy**

January 2017

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DEDICATION

I would like to dedicate my PhD thesis to my parents and my husband



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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the Degree of Doctor of Philosophy

EFFECTS OF ACID RAIN ON GEOTECHNICAL PROPERTIES OF TROPICAL RESIDUAL SOILS

By

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January 2017

Chairman : Professor Bujang Kim Huat, PhD
Faculty : Engineering

Acidic deposit that falls to earth from the atmosphere is named acid rain. Sulfur dioxide (SO_2) and oxide of nitrogen (NO_x) are being constantly released as gases into the atmosphere. These oxides react with the rainwater to produce stronger acids and might lower the pH to 4.5 or even 2.0. This research aimed at determining the effect of acid rain on residual soil and kaolinite in the tropical area. Four types of soils: sedimentary residual soil (SRS), igneous residual soil (IRS), primary kaolinite (PK) and secondary kaolinite (SK) with different properties were chosen. Soaking and infiltration methods were employed to simulate the natural environment and reaction of acid rain on the selected soils based on short-term and long-term reaction. For the soaking method, the specimens were kept in the artificial acid rain (AAR) container for 7, 14, 28, and consequently 90 days, while in the infiltration method, the specimens were kept at flux of AAR for 1, 5, 10, and 20 years at different pH levels for both methods. The AAR was prepared by adding certain volume of 0.005M nitric acid (HNO_3) and sulfuric acid (H_2SO_4) to the deionized distilled water. The pH values of the AAR were adjusted to 2, 3, 4, 5, and 5.6. All soils' properties via physico-chemical and mechanical were assessed before and after treatment of specimens with AAR to identify the effect of the AAR on the soils. For the mechanical properties tests, unconfined compressive strength (UCS), consistency limits, optimum moisture content (OMC), maximum dry density (MDD), hydraulic conductivity, and one-dimensional consolidation tests were carried out. Whereas, to examine the effects of AAR on soil chemical mechanism, the microstructural tests were performed using zeta potential (ζ), Atomic adsorption spectroscopy (AAS), X-ray diffraction (XRD), Fourier transformed infrared spectroscopy (FTIR), scanning electron microscope (SEM), and energy dispersive X-ray spectroscopy (EDX). In soaking method (i.e. short-term), low pH value of AAR changed the characterization of SRS, IRS, PK, and SK, and led to an increase in soil strength and maximum dry density (MDD), and reduction in the hydraulic conductivity, Atterberg limits, and optimum moisture content (OMC). In terms of long-term effect of acid rain, the low pH value and high fluxes of AAR changed the characterization of SRS, IRS, PK, and SK and led to a reduction in soil strength and MDD, and an increase in the hydraulic conductivity,

Atterberg limits, and OMC, initial void ratio, and Cc of SRS, IRS, PK, and SK. Long-term flux of AAR led to a further increase in compressibility of the studied soils. Regarding the influence of flux of acid rain on the soil properties, the acidic pH level of rain has significant effects on the physico-chemical properties. The effect of acid rain on the properties of soils will become more serious if the rainwater is further acidified due to heavier industrialization.

Key words: Acid rain; Residual soil; Kaolinite; Atterberg limits; Compressive strength; Consolidation; Compaction characteristics; Hydraulic conductivity.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia
sebagai memenuhi keperluan untuk Ijazah Doktor Falsafah

KESAN HUJAN ACID PADA SIFAT GEOTEKNIKAL TANAH TANAH TROPIS

Oleh

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Deposit berasid yang jatuh ke bumi dari atmosfera dinamakan hujan asid. Sulfur dioksida (SO_2) dan oksida nitrogen (NO_x) sedang sentiasa dikeluarkan sebagai gas ke atmosfera. Ini oksida bertindak balas dengan air hujan untuk menghasilkan asid kuat dan mungkin menurunkan pH 4.5 atau 2.0. Kajian ini bertujuan untuk menentukan kesan hujan asid pada tanah baki dan kaolinites di kawasan tropika. jenis Fore tanah sedimen sisa tanah (SRS), igneus sisa tanah (IRS), kaolinit utama (PK) dan kaolinit menengah (SK) dengan ciri-ciri yang berbeza telah dipilih. Rendaman dan penyusupan kaedah telah digunakan untuk mensimulasikan persekitaran semula jadi dan tindak balas hujan asid di tanah yang dipilih berdasarkan jangka pendek dan tindak balas jangka panjang. Bagi kaedah rendaman, spesimen disimpan di dalam bekas hujan asid tiruan (AAR) selama 7, 14, 28, dan seterusnya 90 hari, manakala dalam kaedah penyusupan, spesimen telah disimpan di fluks AAR untuk 1, 5, 10, dan 20 tahun pada tahap pH yang berbeza untuk kedua-dua kaedah. The AAR telah disediakan dengan menambah jumlah tertentu 0.005M asid nitrik (HNO_3) dan asid sulfurik (H_2SO_4) untuk air suling yang ternyahion. Nilai pH AAR telah diselaraskan untuk 2, 3, 4, 5, dan 5.6. sifat-sifat semua tanah 'melalui fiziko-kimia dan mekanikal telah dinilai sebelum dan selepas rawatan spesimen dengan AAR untuk mengenal pasti kesan AAR di tanah. Bagi hartanah ujian mekanikal, kekuatan mampatan tak terkurung (UCS), had konsisten, kandungan lembapan optimum (OMC), ketumpatan kering maksimum (MDD), kekonduksian hidraulik, dan ujian pengukuhan dimensi satu- telah dijalankan. Manakala, untuk mengkaji kesan AAR mekanisme kimia tanah, ujian mikrostruktur telah dilakukan dengan menggunakan potensi zeta (ζ), Atomic penyerapan spektroskopi (AAS), X-ray pembelauan (XRD), Fourier berubah spektroskopi inframerah (FTIR), elektron imbasan mikroskop (SEM), dan tenaga serakan X-ray spektroskopi (EDX). Dalam kaedah rendaman (iaitu jangka pendek), nilai pH yang rendah AAR berubah pencirian SRS IRS, PK, dan SK, dan membawa kepada peningkatan dalam kekuatan tanah dan ketumpatan kering maksimum (MDD), dan pengurangan dalam kekonduksian hidraulik, had Atterberg dan kandungan lembapan optimum (OMC). Dari segi kesan jangka panjang hujan asid, nilai pH yang rendah dan fluks tinggi AAR berubah pencirian SRS IRS, PK, dan SK, dan membawa

kepada pengurangan dalam kekuatan tanah dan MDD, dan peningkatan dalam kekonduksian hidraulik, had Atterberg dan OMC, nisbah lompang asal, dan Cc daripada SRS, IRS, PK, dan SK. fluks jangka panjang AAR membawa kepada peningkatan lagi dalam kebolehmampatan tanah dikaji. Mengenai pengaruh fluks hujan asid pada sifat tanah, tahap pH berasid hujan mempunyai kesan yang besar ke atas sifat-sifat fiziko-kimia. Kesan hujan asid kepada sifat-sifat tanah akan menjadi lebih serius jika air hujan itu lebih berasid disebabkan oleh perindustrian yang lebih berat.



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I certify that a Thesis Examination Committee has met on 9 January 2017 to conduct the final examination of Zeinab Bakhshipour on her thesis entitled "Effects of Acid Rain on Geotechnical Properties of Tropical Residual Soils" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xv
LIST OF FIGURES	xvi
LIST OF ABBREVIATIONS	xxiii
LIST OF SYMBOLS	xxvi
CHAPTER	
1 INTRODUCTION	1
1.1 General	1
1.2 Statement of the problem	2
1.3 Objectives	2
1.4 Scope of study	2
1.5 Thesis layout	3
2 LITERATURE REVIEW	4
2.1 Acid rain	4
2.1.1 Acid rain deposition and formation	5
2.1.2 History of acid rain	6
2.1.3 Potential of acid rain threat in south-east Asia	7
2.2 Study of residual soil and clay minerals in Malaysia	9
2.3 Effect of pH and pour fluid on engineering properties of soils	15
2.4 Summary of the literature reviewed	26
2.5 Research gap of the study	28
3 METHODOLOGY	29
3.1 Introduction	29
3.2 Sampling location	30
3.3 Material preparation	33
3.3.1 Soil index properties	33
3.3.1.1 Drying of samples	33
3.3.1.2 Moisture content	33
3.4 Particle size distribution of soil samples	34
3.4.1 Determination of particle size distribution	34
3.4.2 Unified soil classification system (USCS)	34
3.5 Determination of particle density (specific gravity)	34
3.6 Porosity	35
3.7 Organic content	35
3.8 Atterberg limits	36
3.9 Unconfined compression strength	37

3.10	One-dimensional consolidation	38
3.11	Hydraulic conductivity	39
3.12	Standard proctor compaction	40
3.12.1	Compaction	40
3.12.2	Specimens preparation	41
3.12.3	Acid rain exposure	41
3.13	Soaking and infiltration setup	42
3.13.1	Artificial acid rain preparation	44
3.13.2	Soaking method	44
3.13.3	Infiltration method	45
3.14	Chemical and microstructural method	46
3.14.1	pH test	46
3.14.2	X-ray fluorescence	46
3.14.3	X-ray diffraction	46
3.14.4	Zeta potential	47
3.14.5	Atomic adsorption spectroscopy	47
3.14.6	Scanning electron microscopy and energy dispersive x-ray	48
3.14.7	Fourier transform infrared spectroscopy	49
4	RESULTS AND DISCUSSION	50
4.1	Introduction	50
4.1.1	Engineering index properties of soils	50
4.2	Soil engineering properties of sedimentary residual soil (SRS) and igneous residual soil (IRS)	55
4.3	Short term effect of AAR on SRS and IRS soils by soaking method	55
4.3.1	Variation of unconfined compressive strength of SRS and IRS samples by the effect of AAR in short term effect with soaking method	55
4.3.2	Variation of Atterberg limits of SRS and IRS due to the effect of AAR in short term effect with soaking method	58
4.3.3	Variation of compaction characteristics of SRS and IRS soils brought out by the effect of AAR in short term effect with soaking method	60
4.3.4	Variation of hydraulic conductivity of SRS and IRS soils by effect of AAR in short term effect with soaking method	61
4.3.5	Variation of x-ray diffraction of SRS and IRS soils by the effect of AAR in short term effect with soaking method	63
4.3.6	Variation of zeta potential of SRS and IRS soils by the effect of AAR in short term effect with soaking method	65
4.3.7	Fourier transformed infrared spectroscopy results of SRS and IRS soils by effect of AAR in short term effect with soaking method	66

4.3.8	Scanning electron microscopy and energy dispersive x-ray analysis of SRS and IRS soils in short term effect by soaking method	68
4.4	Long term effect of AAR on SRS and IRS soils by infiltration method	70
4.4.1	Variation of unconfined compressive strength of SRS and IRS due to the effect of AAR in long term effect by infiltration method	70
4.4.2	Variation of Atterberg limits of SRS and IRS samples due to the effect of AAR in long term effect by infiltration method	74
4.4.3	One-dimensional consolidation tests of SRS and IRS in infiltration method	75
4.4.4	Variation of compaction characteristics of SRS and IRS due to effect of AAR in long term effect with infiltration method	78
4.4.5	Variation of hydraulic conductivity of SRS and IRS soils due to the effect of AAR in long term effect with infiltration method	79
4.4.6	Variation of zeta potential of SRS and IRS soils due to the effect of AAR in long term effect with infiltration method	81
4.4.7	Results from atomic adsorption spectroscopy tests of SRS and IRS soils due to the effect of AAR in long term effect with infiltration method	81
4.4.8	Scanning electron microscopy and energy dispersive x-ray analysis of SRS and IRS due to the effect of AAR in long term effect with infiltration method	84
4.5	Engineering properties of primary kaolinite (PK) and secondary kaolinite (SK)	86
4.6	Short term effect of AAR on PK and SK soils by soaking method	87
4.6.1	Variation of unconfined compressive strength of PK and SK soils by effect of AAR in short term effect with soaking method	87
4.6.2	Variation of Atterberg limits for PK and SK soils by effect of AAR in short term effect with soaking method	90
4.6.3	Variation of compaction characteristics of PK and SK soils due to the effect of AAR in short term effect with soaking method	91
4.6.4	Variation of hydraulic conductivity of PK and SK soils due to the effect of AAR in short term effect with soaking method	93
4.6.5	X-ray diffraction of PK and SK soils due to the effect of AAR in short term effect with soaking method	94

4.6.6	Variation of zeta potential of PK and SK soils by effect of AAR in short term effect with soaking method	96
4.6.7	Fourier transformed infrared spectroscopy results of PK and SK soils by the effect of AAR in short term effect with soaking method	98
4.6.8	Scanning electron microscopy and energy dispersive x-ray analysis of PK and SK soils in soaking method	99
4.7	Variation of acid rain on the engineering and other properties of primary kaolinites (PK) and secondary kaolinites (SK) in infiltration method	101
4.7.1	Introduction	101
4.7.2	Unconfined compression tests of PK and SK soils in infiltration method	102
4.7.3	Atterberg limits values of PK and SK in infiltration method	105
4.7.4	One-dimensional consolidation tests of PK and SK soils in infiltration method	106
4.7.5	Compaction of PK and SK soils in infiltration method	108
4.7.6	Variation of hydraulic conductivity of PK and SK soils in infiltration method	110
4.7.7	Zeta potential of PK and SK soils in infiltration method	111
4.7.8	Results from atomic adsorption spectroscopy of PK and SK soils in infiltration method	113
4.7.9	Scanning electron microscopy and energy dispersive x-ray analysis of PK and SK soils in infiltration method	116
4.8	Proposed mechanism of under the effect of acid rain on soils in infiltration and soaking method	118
4.8.1	Proposed mechanism of soaking method due to AAR effect on the soils	118
4.8.2	Proposed mechanism of infiltration method due to AAR effect on the engineering properties of soils	122
4.9	Compare the short-term and long-term and relate them to engineering	123
4.10	Consolidation behaviour and prediction of its characteristics of kaolinite soils with particular reference to geotechnical and chemical reaction	124
5	SUMMARY AND CONCLUSIONS	132
5.1	Summary	132
5.2	Conclusions	132
5.3	Recommendations	133

REFERENCES	134
APPENDICES	148
BIODATA OF STUDENT	160
LIST OF PUBLICATIONS	161



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LIST OF TABLES

Table		Page
2.1	Annual volume-weighted mean rainwater	8
2.2	Nominal annual wet and dry deposition	8
2.3	Weathering Classification Grades	10
2.4	Geotechnical Properties of residual soils of granite origin in Peninsular Malaysia by several researchers	13
2.5	Physical parameters of the clays	18
2.6	Chemical analysis of the clays	18
4.1	Chemical analysis of sedimentary residual soil (SRS) and igneous residual soil (IRS)	51
4.2	Physical properties of sedimentary residual soil (SRS) and igneous residual soil (IRS)	52
4.3	Physical and chemical properties of primary kaolinite (PK) and secondary kaolinite (SK)	53
4.4	The other subsequent equation of Cc	130

LIST OF FIGURES

Figure		Page
2.1	pH Scale, Acidity of Common Substances, and Acidity of Rain and Fog	4
2.2	Origins of acid rain and combination of natural and manmade activities result in the deposition of acidic compounds	5
2.3	Weathering profiles of granitic slope in Kuala Lumpur-Karak Highway, Malaysia	11
2.4	Schematic representation of the mineralogical structure of kaolinite	14
2.5	Schematic representation of the DDL	15
3.1	Overall schematic presentation of the study	30
3.2	The location of the sampling point of the sedimentary residual soil and igneous residual soil at Kuala Lumpur, Malaysia	31
3.3	Location of PK and SK soil samples (a) Primary kaolinite, (b) Secondary kaolinite	32
3.4	Sedimentary residual soil (SRS), igneous residual soil (IRS), primary kaolinite (PK), and secondary kaolinite (SK) before and after grained soils and sieve with 425 μ m	33
3.5	Atterberg limits BS 1377-2 (1990)	36
3.6	Unconfined compression machine	38
3.7	Consolidation test preparation	39
3.8	Mini compaction apparatus	41
3.9	Extruder soil samples from mould	42
3.10	Soaking setup for keeping of specimens in mold and flux for short term effects	43
3.11	Infiltration mold and preparing of specimens for infiltration setup	44
3.12	Artificial acid rain preparations	44
3.13	Experimental setup for soaking test	45

3.14	Experimental setups for infiltration method	46
3.15	Preparation of samples for Atomic adsorption spectroscopy	48
4.1	Particle size distributions of PK and SK	54
4.2	XRD analysis of original PK and SK soils	54
4.3	Stress-strain behaviour of SRS sample in short term effect of AAR in soaking method	56
4.4	Stress-strain behaviour of IRS sample in short term effect of AAR in soaking method	56
4.5	Variation of compressive strength of SRS sample with respect to pH in short term effect of AAR in soaking method	57
4.6	Variation of compressive strength of IRS sample with respect to pH in short term effect of AAR in soaking method	58
4.7	Liquid Limit of SRS and IRS soils with respect to pH in Short Term Effect by soaking method	59
4.8	Plasticity Index (PI) of SRS and IRS soils with respect to pH in Short Term Effect by soaking method	59
4.9	Variation of compaction characteristics of SRS soil with respect to pH in Short Term Effect by soaking method	60
4.10	Evolution of compaction characteristics of IRS soils with respect to pH in Short Term Effect by soaking method	61
4.11	Variation of permeability of SRS soil with respect to pH in Short Term Effect by soaking method	62
4.12	Variation of permeability of IRS soil with respect to pH in Short Term Effect by soaking method	62
4.13	X-ray diffraction of SRS soil in acidic pH in Short Term Effect by soaking method	64
4.14	X-ray diffraction of IRS soil in acidic pH in Short Term Effect by soaking method	65
4.15	Zeta potential of SRS and IRS soils with respect to increase of pH in Short Term Effect by soaking method	66
4.16	FTIR of SRS soil in Short Term Effect of AAR by soaking method	67
4.17	FTIR of IRS soil Short Term Effect of AAR by soaking method	67

4.18	(a) SEM images, (b) EDX analysis in short term effect of AAR by soaking method in SRS original soils and pH2 in flux of 14 days to 90 days	69
4.19	(a) SEM images, (b) EDX analysis in short term effect of AAR by soaking method in IRS original soils and pH2 in flux of 14 days to 90 days	70
4.20	Stress-strain behaviour of SRS sample due to long term effect by AAR in infiltration method	71
4.21	Stress-strain behaviour of IRS sample in long term effect due to AAR in infiltration method	72
4.22	Variation of compressive strength of SRS sample in long term effect by AAR in infiltration method with respect to pH	73
4.23	Variation of compressive strength of SRS sample in long term effect by AAR in infiltration method with respect to pH	73
4.24	Liquid Limit of SRS and IRS soils due to long term effect by AAR in infiltration method with respect to pH	74
4.25	Plasticity index of SRS and IRS soils due to long term effect by AAR in infiltration method with respect to pH	75
4.26	Influence of pH on the void ratio-effective vertical stress relationship in flux of SAR 1 year to 20 year	76
4.27	Influence of pH on compression index of SRS soil due to long term effect by AAR in infiltration method	77
4.28	Influence of pH on compression index of IRS soil due to long term effect by AAR in infiltration method	77
4.29	Variation of compaction characteristics of SRS soil with respect to pH in long Term Effect by infiltration method	78
4.30	Variation of compaction characteristics of IRS soil with respect to pH in long Term Effect by infiltration method	79
4.31	Variation of hydraulic conductivity of SRS soil with respect to pH in long term effects by infiltration method	80
4.32	Variation of hydraulic conductivity of IRS soil with respect to pH in long term effects by infiltration method	80
4.33	Zeta potential of SRS and IRS soils with respect to pH in long term effect by infiltration method	81

4.34	AAS Analysis of soil in long term effect by infiltration method before and after AAR	82
4.35	AAS Analysis of leachate due to long term effect by infiltration method from	83
4.36	Ion soluble Analysis of SRS and IRS soils before and after infiltration by AAR	84
4.37	(a) SEM images, (b) EDX analysis due to long term effect of AAR by infiltration method in SRS soils in original soil and pH ₂ in flux of 1 year and 20 years	85
4.38	(a) SEM images, (b) EDX analysis due to long term effect of AAR by infiltration method in IRS soils in original soil and pH ₂ in flux of 1 year and 20 years	86
4.39	Stress-strain behaviour of PK sample in short term effect of AAR in soaking method	87
4.40	Stress-strain behaviour of SK sample in short term effect of AAR in soaking method	88
4.41	Variation of compressive strength of PK sample with respect to pH in short term effect of AAR in soaking method	89
4.42	Variation of compressive strength of SK sample with respect to pH in short term effect of AAR in soaking method	89
4.43	Liquid limit of PK and SK soils with respect to pH in Short Term Effect by soaking method	90
4.44	Variation of plasticity index of PK and SK soils with respect to pH in Short Term Effect by soaking method	91
4.45	Variation of compaction characteristics of PK soil with respect to pH in Short Term Effect by soaking method	92
4.46	Variation of compaction characteristics of SK soil with respect to pH in Short Term Effect by soaking method	92
4.47	Variation of hydraulic conductivity of PK soil with respect to pH in Short Term Effect by soaking method	93
4.48	Variation of hydraulic conductivity of SK soil with respect to pH in Short Term Effect by soaking method	94
4.49	X-ray diffraction of PK soil in Short Term Effect by soaking method	95

4.50	X-ray diffraction of SK soil in Short Term Effect by soaking method	96
4.51	Zeta potential of PK soil with respect to pH in Short Term Effect by soaking method	97
4.52	Zeta potential of SK soil with respect to pH in Short Term Effect by soaking method	97
4.53	FTIR of PK soil in Short Term Effect of AAR by soaking method	98
4.54	FTIR of SK soil in Short Term Effect of AAR by soaking method	99
4.55	(a) SEM images, (b) EDX analysis in short term effect of AAR by soaking method in PK soils in original soil and pH2 in flux of 14 days to 90 days	100
4.56	(a) SEM images, (b) EDX analysis in short term effect of AAR by soaking method in SK soils in original soil and pH2 in flux of 14 days to 90 days	101
4.57	Stress-strain behaviour of PK sample in flux of artificial acid rain 1-year to 20-years in infiltration method	102
4.58	Stress-strain behaviour of SK sample in flux of artificial acid rain 1-year to 20-years in infiltration method	103
4.59	Variation of compressive strength of PK sample with respect to pH in long term effect in infiltration method	104
4.60	Variation of compressive strength of SK sample with respect to pH in long term effect in infiltration method	104
4.61	Liquid Limit of PK and SK soils with respect to change in pH in long term effect of AAR in flux of artificial acid rain 1-year to 20-year	105
4.62	Plasticity index of PK and SK soils with respect to change in pH in long term effect of AAR in flux of artificial acid rain, 1-year to 20-year	106
4.63	Influence of pH on the void ratio-effective vertical stress relationships in flux of SAR 1 year to 20 years a) Effect of pH on void ratio-effective vertical stress of PK specimens, b) Effect of pH on void ratio-effective vertical stress of SK specimens	107
4.64	Influence of pH on compression index in long term effect by AAR in PK sample in 1-year to 20-years in infiltration method	108

4.65	Influence of pH on compression index in long term effect by AAR in SK sample in 1-year to 20-years in infiltration method	108
4.66	Compaction characteristics of PK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	109
4.67	Compaction characteristics of SK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	110
4.68	Variation of hydraulic conductivity of PK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	111
4.69	Variation of hydraulic conductivity of SK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	111
4.70	Zeta potential of PK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	112
4.71	Zeta potential of SK soils with respect to pH in long term effect of AAR in 1-year to 20-years with infiltration method	112
4.72	AAS Analysis of soils before and after filtration by AAR in flux of 1-year and 20-year with respect to pH	114
4.73	AAS Analysis of leachate from the soils with respect to pH (a) PK-1 year (b) PK -20 years (c) SK -1 year (d) SK- 20 years	115
4.74	Ion soluble Analysis of soil before and after filtration by AAR with respect to pH (a) PK -1 year (b) PK -20 year (c) SK- 1 year (d) SK-20 year	116
4.75	(a) SEM images, (b) EDX analysis in long term effect of AAR by infiltration method in PK soils in original soil and pH ₂ in flux of 1 year and 20 years	117
4.76	(a) SEM images, (b) EDX analysis in long term effect of AAR by infiltration method in SK soils in original soil and pH ₂ in flux of 1 year and 20 years	118
4.77	Effect of liquid limit (LL) and plastic limit (PL) versus to total compression brought out by changes in pH values in PK and SK samples in infiltration method	125
4.78	Zeta potential (mV) versus pH values in PK samples in infiltration method	126

4.79	Zeta potential (mV) versus pH values in SK samples in infiltration method	126
4.80	Effect of liquid limit (LL) to optimum moisture content (OMC) brought out by changes in pH values in PK samples in infiltration method	127
4.81	Effect of liquid limit (LL) to optimum moisture content (OMC) brought out by changes in pH values in SK samples in infiltration method	128
4.82	Effect of maximum dry density (MDD) to optimum moisture content (OMC) brought out by changes in pH values in PK and SK samples in infiltration method	129
4.83	Effect of maximum dry density (MDD) to optimum moisture content (OMC) brought out by changes in pH values in PK and SK samples in infiltration method	129
4.84	Effect of liquid limit (LL) versus to total compression due to total pressure of 400kPa brought out by changes in pH values in PK samples in infiltration method	131
4.85	Effect of liquid limit (LL) versus to total compression due to total pressure of 400kPa brought out by changes in pH values in SK samples in infiltration method	131
A 1.1	preparation of soaking mould	148
A 1.2	preparation of soaking setup	149
A 1.3	preparation of infiltration setup	150
A 2.1	Soil specimens preparation for UCS test	151
A 3.1	Adjustment of solution by pH meter	152
A 4.1	AAS test	153
A 5.1	Some samples broken after UCS test	154
A 6.1	hydraulic conductivity test	155
A 7.1	Cation exchange capacity (CEC) test	156
A 8.1	SEM and EDX test	157
A 9.1	Zeta potential test	158
A 10.1	Consolidation test	159

LIST OF ABBREVIATIONS

SO_4^{2-}	Sulfate
NO_3^-	Nitrate
Al_2O_3	Aluminum dioxide
Fe_2O_3	Iron dioxide
NaO	Sodium oxide
CaO	Calcium oxide
XRD	X-Ray Diffraction
SEM	Scanning electron microscope
EM	Electrophoretic mobility at actual temperature
V_t	Velocity of the suspending liquid
Dt	Dielectric constant
AAS	Atomic adsorption spectroscopy
SEM	scanning electron microscope
Cc	Compression Index
PK	Primary kaolinite
SK	Secondary kaolinite
SRS	Sedimentary residual soil
IRS	Igneous residual soil
CEC	Cation exchange capacity
Fe	Iron
Al	Aluminum
Si	Silicate
Ca	Calcium

K	Potassium
Na	Sodium
Mg	Magnesium
Fe ₂ O ₃	Iron oxide
Al ₂ O ₃	Aluminum oxide
SiO ₂	Sulfur oxide
SO ₂	Sulfur dioxide
NO _x	Nitrogen oxides
OH	Hydroxide
H	Hydrogen
H ₂ SO ₄	Sulfuric acid
HNO ₃	Nitric acid
BSI	British Standard Classification
a.u.	Arbitrary unit
ASTM	American Society for Testing and Material
BS	British Standard
USCS	Unified Soil Classification System
CL	low medium plasticity
MH	Elastic Silt
Rave	Stands for annual average precipitation
A	Stands for the cross sectional area of the specimen
AAR	Artificial acid rain
kg/m ³	Kilograms per Cubic Meter
kN/m ³	Kilonewtons per Cubic Meter
m	Meter

cm	Centimeter
mm	Millimeter
μm	Micrometer
I	Illite
K	Kaolinite
G	Goethite
Q	Quartz
H	Hematite
K-Feldspar	Kaolinite Feldspar
S	Smectite
Mg/m ³	Milligram per Cubic Meter
kPa	Kilo Pascal



LIST OF SYMBOLS

LL	Liquid limit	%
PL	Plastic limit	%
PI	Plasticity Index	%
SL	Shrinkage limit	
W	Natural water content	%
G _s	Specific gravity	
MDD	Maximum dry density	mg/m ³
OMC	Optimum moisture content	%
ζ	Zeta potential	(mV)
π	Constant	
UCS	Unconfined compressive strength	kPa
strain	Compressive strain	%
e	void ratio	
σ' _v	Effective vertical stress	
m ₁	mass of density bottle including stopper	g
m ₂	mass of density bottle, stopper and oven dried soil	g
m ₃	mass of density	g
m ₄	mass of density bottle, stopper and distilled water	g
<i>p_d</i>	Dry density	g/cm ³
<i>p_w</i>	Wet density	g/cm ³
m _w	Wet sample	g
<i>D</i>	The mean diameter of the specimen	cm
<i>L</i>	The mean length of the soil specimen	cm

k	Hydraulic conductivity	m/s
LS	Linear shrinkage	%
A	cross section area	m ²
L	length of sample	m
Φ	Porosity	%



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CHAPTER 1

INTRODUCTION

1.1 General

Research conducted over the past few decades indicates that changes in pore water chemistry may significantly alter the geotechnical properties of soils (Bolt, 1956, Mitchell, 1960, Rosenqvist, 1959, Santamarina et al., 2002). Geotechnical practices indicate that engineering structures built on such soils can experience significant deformation due to changes in the soil's structure (Sridharan et al., 1981). Therefore, the engineering properties of soil have received special consideration by researchers. One of the most important and obvious sources of soil pollution is acid rain. The major sources of acid rain are sulphur dioxide (SO_2) and nitrogen oxides (NO_x) (Streets and Waldhoff, 2000). These oxides react with rain water to produce stronger acids with a low pH of 4.5, or even 3.0. The engineering properties of soils are significantly influenced by the chemical properties of the soil and is termed "chemical-mechanical coupling" (Santamarina et al., 2002). The acid rain and its annual rate in the various areas cause to increase weathering rate in the soils (Denman et al., 2007). It is true that the weathering factor is the main fact which owing to the formation of residual soils in the tropical areas (Kato and Akimoto, 1992). The major types of soil in Malaysia as a tropical area are residual soils (Galloway, 1989).

In this study two common types of residual soil, including sedimentary residual soil and igneous residual soil were chosen. It is noticeable that in almost all of the tropical residual soils in Malaysia, kaolinite is one of the dominating clay minerals (Tan, 2009). Kaolinite is classified into two broad categories which is called primary and secondary (Bignell et al., 2002). The primary kaolinite or residual kaolin originates from in situ alteration of rocks by hot fluids, which permeate mainly in pegmatite. Generally primary kaolinite has not been altered chemically since its deposition and crystallization as a result of large-scale geological processes (Bolland et al., 1976). In addition, the secondary minerals from the decomposition and restructuring of primary minerals or from precipitation reactions concerning chemical constituents of primary kaolinite released during dissolution (Bloodworth et al., 1993, Bolland et al., 1976). Secondary kaolinite is generally characterized by smaller particle size, because the particle size of primary kaolinite is decreased during weathering and release of soluble materials (Bloodworth et al., 1993, Scorzelli et al., 2008b). Based on the existing literature, it can be concluded that the following chemical and physical processes might describe the effect of acid rain on soil (Benna et al., 2002, Jozefaciuk and Bowanko, 2002): (I) Dissolution and leaching of cations or anions from soil, based on their solubility at different pH values; (II) Changes in surface electrical properties in the colloidal fraction of soil; (III) Absorption of anions in acid rain such as SO_4^{2-} , NO_3^- , and CO_3^{2-} in the soil environment; and (IV) Changes in diffuse double layer of the clay fraction due to the ion exchange and consequent changes in the van der Waals forces acting between the clay particles, which may result in a change in engineering properties of soil affected by acid rain.

1.2 Statement of the Problem

One of the most important and sources of soil pollution is acid rain. Geotechnical properties can be changed under effect of acid rain. Moreover, increasing urbanisation and industrialisation in developing countries, such as Malaysia, have extended the geographical occurrence of acid rain. Malaysia is one of the areas that have undergone to experience the effects of acid rain. The major types of soil in Malaysia as a tropical area are residual soils. It is noticeable that among all of the tropical residual soils in Malaysia, kaolinite is one of the dominating clay minerals. The reaction of variety of soils and minerals affected by acid rain is different. The literature review show that the effects of acid rain on physico-chemical and engineering properties of soil could vary due to the complexity of mineral composition of the parent soil and chemical properties of the pore fluid. However, there is still a lack of comprehensive knowledge on the physical and chemical mechanisms of influence of acid rain on the minerals in tropical area. It should be noted that although the effect of each minerals on the properties of clay in an acidic medium seems to be well understood, whereas the behaviour of different kinds of kaolinite in tropical area still remains unclear. The primary aim of this study is to determine the effects of acid rain in tropical area on two common types of residual soils and different kaolinites.

1.3 Objectives

The objectives of this study are:

- I. To investigate short term effects of acid rain on geotechnical properties of sedimentary residual soil, igneous residual soil, primary kaolinite, and secondary kaolinite.
- II. To investigate long term effects of acid rain on geotechnical properties of sedimentary residual soil, igneous residual soil, primary kaolinite, and secondary kaolinite.
- III. To determine the underlying mechanisms controlling the effect of acid rain on the sedimentary residual soil, igneous residual soil, primary kaolinite, and secondary kaolinite.

1.4 Scope of Study

The purpose of this research is to observe the changing in tropical residual natural soils and clay minerals. The scopes and limitations of the research is including of the sedimentary residual soil (SRS) and igneous residual soil (IRS) were selected from Batu Cave and Taman Ukay Perdana respectively in Kuala Lumpur, Malaysia and primary kaolinite (PK) and secondary kaolinite (SK). Moreover, in this research the short and long term effect of acid rain with choosing two methods of soaking and infiltration method used to investigate the reality event. The strength and compressibility evaluation tests for the samples are UCS test and consolidation test respectively. In addition, the maximum flux of artificial acid rain used for the strength evaluations of samples are 90 days and 20 years in soaking and infiltration method

respectively. Furthermore, the chemical properties of acid rain on soils and kaolinites are evaluated by X-ray diffraction (XRD), scanning electron microscopy (SEM), Scanning Electron Microscopy (EDX), zeta potential, and atomic adsorption spectroscopy (AAS).

1.5 Thesis Layout

The chapters of the thesis are organized as follows:

Chapter 2 presents the literature review on acid rain and deposition, history of acid rain and the potential of acid rain in different country. This chapter also includes the principle of residual soil and clay minerals that include of primary and secondary kaolinite and the most important clay minerals of residual soils. Furthermore, this chapter presents the effect of acid rain on physical and engineering properties of different soils.

Chapter 3 discusses the sampling location of soils and minerals, moreover explained the geotechnical test, the preparation of acid rain exposure, short and long term method which include of soaking and infiltration methods. This chapter also describes the microstructure analysis of soils as affected by acid rain at different time and pH.

Chapter 4 describes the effect of acid rain on residual soils and kaolinites at short and long term effect at different pH value on some engineering properties of soil. The study evaluated the microstructure analysis of soils and minerals by effect of acid rain at long at short term effect. Consequently the proposed mechanism under effect of acid rain is explained.

Chapter 5 presents the conclusions arrived from this study and recommendations for future research.

REFERENCES

- ACT, E. P. 2005. 109th Congress. *Public Law*, 109, 54.
- ALIAS, N. E. & TAKARA, K. 2013. Estimating the Probable Maximum Precipitation of Kuala Lumpur, Malaysia and Yodo River Basin, Japan Using Statistical Methods. *Journal of Disaster Research Vol*, 8, 197.
- ASSAAD, J. J. & HARB, J. 2012. Use of the falling-head method to assess permeability of freshly mixed cementitious-based materials. *Journal of Materials in Civil Engineering*, 25, 580-588.
- ASTM. Standard Specification for Portland Cement, Annual Book of ASTM Standards. 2004. ASTM Philadelphia, PA.
- AYERS, G., PENG, L. C., FOOK, L. S., KONG, C. W., GILLET, R. & MANINS, P. 2000. Atmospheric concentrations and deposition of oxidised sulfur and nitrogen species at Petaling Jaya, Malaysia, 1993–1998. *Tellus B*, 52.
- AZZOUZ, A. S., KRIZEK, R. J. & COROTIS, R. B. 1976. Regression analysis of soil compressibility. 16, 19-29.
- BELL, F. 1994. A survey of the engineering properties of some anhydrite and gypsum from the north and midlands of England. *Engineering Geology*, 38, 1-23.
- BELL, F. G. 2013. *Fundamentals of engineering geology*, Elsevier.
- BENNA, M., KBIR-ARIGUIB, N., CLINARD, C. & BERGAYA, F. 2002. Card-house microstructure of purified sodium montmorillonite gels evidenced by filtration properties at different pH. *Adsorption and Nanostructure*. Springer.
- BENNETT, R. & HULBERT, M. 2012. *Clay microstructure*, Springer Science & Business Media.
- BERGNA, H. E. & ROBERTS, W. O. 2005. *Colloidal silica: fundamentals and applications*, CRC Press.
- BHATTI, N., STREETS, D. G. & FOELL, W. K. 1992. Acid rain in Asia. *Environmental Management*, 16, 541-562.
- BIDOGLIO, G. & STUMM, W. 2013. *Chemistry of aquatic systems: local and global perspectives*, Springer Science & Business Media.
- BIGNELL, D. E., HOLT, J. A. & TERMITES, R. L. 2002. *Encyclopedia of Soil Science*. Marcel Dekker, New York.
- BLOODWORTH, A., HIGHLEY, D. E. & MITCHELL, C. 1993. *Industrial minerals laboratory manual: kaolin*.

- BOLLAND, M. D. A., POSNER, A. & QUIRK, J. 1976. Surface charge on kaolinites in aqueous suspension. *Soil Research*, 14, 197-216.
- BOLT, G. 1956. Physico-chemical analysis of the compressibility of pure clays. *Geotechnique*, 6, 86-93.
- BOWDERS JR, J. J. & DANIEL, D. E. 1987. Hydraulic conductivity of compacted clay to dilute organic chemicals. *Journal of Geotechnical Engineering*, 113, 1432-1448.
- BRADY, N. C. & WEIL, R. R. 2010. *Elements of the nature and properties of soils*, Pearson Prentice Hall.
- BRANDENBURG, U. & LAGALY, G. 1988. Rheological properties of sodium montmorillonite dispersions. *Applied Clay Science*, 3, 263-279.
- BRIMBLECOMBE, P. 1996. *Air composition and chemistry*, Cambridge University Press.
- BRODERICK, G. P. & DANIEL, D. E. 1990. Stabilizing compacted clay against chemical attack. *Journal of Geotechnical Engineering*, 116, 1549-1567.
- BROWN & BRINDLEY, G. 1980. Crystal structures of clay minerals and their X-ray identification. *Mineralogical Society, London*, 361-410.
- BROWN, I. 1981. The bond-valence method: an empirical approach to chemical structure and bonding. *Structure and bonding in crystals*, 2, 1-30.
- BS1377-2, B. S.-. 2003. Code of practice for soil engineering purposes: classification tests. *British Standards Institution, Milton, UK*.
- BS1377-4, B. S.-. 2003. Code of practice for soil engineering purposes: Compaction-related *British Standards Institution, Milton, UK*.
- BS1377-7, B. S.-. 2003. Code of practice for soil engineering purposes: Shear strength tests (total stress). *British Standards Institution, Milton, UK*.
- CALACE, N., FIORENTINI, F., PETRONIO, B. & PIETROLETTI, M. 2001. Effects of acid rain on soil humic compounds. *Talanta*, 54, 837-846.
- CASAGRANDE, A. 1958. Notes on the design of the liquid limit device. *Geotechnique*, 8, 84-91.
- CHANG, S., RYAN, M. & GUPTA, R. 1993. The effect of pH, ionic strength, and temperature on the rheology and stability of aqueous clay suspensions. *Rheologica acta*, 32, 263-269.
- CHEN, J., ANANDARAJAH, A. & INYANG, H. 2000. Pore fluid properties and compressibility of kaolinite. *Journal of Geotechnical and Geoenvironmental Engineering*, 126, 798-807.

- CHEW, S., KAMRUZZAMAN, A. & LEE, F. 2004. Physicochemical and engineering behavior of cement treated clays. *Journal of geotechnical and geoenvironmental engineering*, 130, 696-706.
- CHOLLET, M. & HORGNIES, M. 2011. Analyses of the surfaces of concrete by Raman and FT-IR spectroscopies: comparative study of hardened samples after demoulding and after organic post-treatment. *Surface and Interface Analysis*, 43, 714-725.
- CHOROM, M. & RENGASAMY, P. 1995. Dispersion and zeta potential of pure clays as related to net particle charge under varying pH, electrolyte concentration and cation type. *European Journal of Soil Science*, 46, 657-665.
- COZZOLINO, V. Statistical forecasting of compression index. Proceedings of the 5th international conference on soil mechanics and foundation engineering Paris, 1961. 51-53.
- DE ANIL, K. 2003. *Environmental chemistry*, New Age International.
- DEERE, D. & PATTON, F. Slope stability in residual soils. Proceedings of the 4th Panamerican Conference on Soil Mechanics and Foundation Engineering, San Juan, Puerto Rico, 1971. 87-110.
- DENMAN, K. L., BRASSEUR, G., CHIDTHAISONG, A., CIAIS, P., COX, P. M., DICKINSON, R. E., HAUGLUSTAINE, D., HEINZE, C., HOLLAND, E. & JACOB, D. 2007. Couplings between changes in the climate system and biogeochemistry. *Climate change*, 2007, 541-584.
- DI MAIO, C. 1996. Exposure of bentonite to salt solution: osmotic and mechanical effects. *Geotechnique*, 46, 695-707.
- DIXON, J. B. & SCHULZE, D. G. 2002. *Soil mineralogy with environmental applications*, Soil Science Society of America Inc.
- DOLINAR, B. & TRAUNER, L. 2007. The impact of structure on the undrained shear strength of cohesive soils. *Engineering geology*, 92, 88-96.
- DUDEK, D. J. & FUND, E. D. 1989. Climate change impacts upon agriculture and resources: A case study of California. *The Potential Effects of Global Climate Change on the United States*, 1, 3.
- ERSOY, B., KAVAS, T., EVCIN, A., BAŞPINAR, S., SARUŞIK, A. & DIKMEN, S. 2009. Production Of Fired Ceramic Materials From Fly Ash With Witherite Additive. *Afyon Kocatepe Üniversitesi Fen Ve Mühendislik Bilimleri Dergisi*, 9, 45-52.
- EVANS, G. O. 1992. *Principles of acarology*, CAB international Wallingford.
- FANG, H.-Y. & DANIELS, J. L. 2006. *Introductory geotechnical engineering: an environmental perspective*, CRC Press.

- FELDMAN, R. & RAMACHANDRAN, V. S. 1966. The influence of $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ upon the hydration character of $3\text{CaO} \cdot \text{Al}_2\text{O}_3$. *Magazine of Concrete Research*, 18, 185-196.
- FELT, E. J. 1955. Factors influencing physical properties of soil-cement mixtures. *Highway Research Board Bulletin*.
- FENN, L. & KISSEL, D. 1973. Ammonia volatilization from surface applications of ammonium compounds on calcareous soils: I. General theory. *Soil Science Society of America Journal*, 37, 855-859.
- FERKEL, H. & HELLMIG, R. 1999. Effect of nanopowder deagglomeration on the densities of nanocrystalline ceramic green bodies and their sintering behaviour. *Nanostructured materials*, 11, 617-622.
- FONTES, M. P. F. & ALLEONI, L. R. F. 2006. Electrochemical attributes and availability of nutrients, toxic elements, and heavy metals in tropical soils. *Scientia Agricola*, 63, 589-608.
- FOOKES, P. G. Tropical residual soils: A Geological Society Engineering Group working party revised report. 1997. Geological Society of London.
- FROST, R. L., WILLIAMS, P. A., MARTENS, W., LEVERETT, P. & KLOPROGGE, J. T. 2004. Raman spectroscopy of basic copper (II) and some complex copper (II) sulfate minerals: Implications for hydrogen bonding. *American Mineralogist*, 89, 1130-1137.
- FURUKAWA, Y., FUJITA, T. & HIRAYAMA, K. 1994. Influence of acid rain on stabilized decomposed granite soil. *Proc JSSMFE*, 34, 123-135.
- GAJO, A. & MAINES, M. 2007. Mechanical effects of aqueous solutions of inorganic acids and bases on a natural active clay. *Géotechnique*, 57, 687-699.
- GALLOWAY, J. N. 1989. Atmospheric acidification: projections for the future. *Ambio*, 161-166.
- GARCÍA LODEIRO, I., MACPHEE, D., PALOMO, A. & FERNÁNDEZ-JIMÉNEZ, A. 2009. Effect of alkalis on fresh C-S-H gels. FTIR analysis. *Cement and Concrete Research*, 39, 147-153.
- GARG, N., WANG, K. & MARTIN, S. W. 2013. A Raman spectroscopic study of the evolution of sulfates and hydroxides in cement-fly ash pastes. *Cement and Concrete Research*, 53, 91-103.
- GIVI, A. N., RASHID, S. A., AZIZ, F. N. A. & SALLEH, M. A. M. 2011a. Investigations on the development of the permeability properties of binary blended concrete with nano-SiO₂ particles. *Journal of Composite Materials*, 45, 1931-1938.

- GIVI, A. N., RASHID, S. A., AZIZ, F. N. A. & SALLEH, M. A. M. 2011b. Particle size effect on the permeability properties of nano-SiO₂ blended Portland cement concrete. *Journal of Composite Materials*, 45, 1173-1180.
- GIVI, A. N., RASHID, S. A., AZIZ, F. N. A. & SALLEH, M. A. M. 2012. Influence of 15 and 80 nano-SiO₂ particles addition on mechanical and physical properties of ternary blended concrete incorporating rice husk ash. *Journal of Experimental Nanoscience*, 1-18.
- GOLDBERG, S. & GLAUBIG, R. A. 1987. Effect of saturating cation, pH, and aluminum and iron oxide on the flocculation of kaolinite and montmorillonite. *Clays and Clay Minerals*, 35, 220-7.
- GORI, U. 1994. The pH influence on the index properties of clays. *Bulletin of the International Association of Engineering Geology-Bulletin de l'Association Internationale de Géologie de l'Ingénieur*, 50, 37-42.
- GRATCHEV, I. & TOWHATA, I. 2011. Compressibility of natural soils subjected to long-term acidic contamination. *Environmental Earth Sciences*, 64, 193-200.
- GRATCHEV, I. & TOWHATA, I. 2015. Compressibility of soils containing kaolinite in acidic environments. *KSCE Journal of Civil Engineering*, 1-8.
- GRATCHEV, I. B. & SASSA, K. 2009. Cyclic behavior of fine-grained soils at different pH values. *Journal of geotechnical and geoenvironmental engineering*, 135, 271-279.
- GRIBBLE, C. & MCLEAN, A. 2003. *Geology for civil engineers*, CRC Press.
- GUTHRIE, W., SHEA, M. & EGGETT, D. Hydraulic Conductivity of Cement-Treated Soils and Aggregates after Freezing. Cold Regions Engineering 2012@sustainable Infrastructure Development in a Changing Cold Environment, 2012. ASCE, 93-103.
- HAVAS, M., HUTCHINSON, T. C. & LIKENS, G. E. 1984. Effect of low pH on sodium regulation in two species of *Daphnia*. *Canadian journal of zoology*, 62, 1965-1970.
- HE, F. & ZHAO, D. 2005. Preparation and characterization of a new class of starch-stabilized bimetallic nanoparticles for degradation of chlorinated hydrocarbons in water. *Environmental Science & Technology*, 39, 3314-3320.
- HEIL, D. & SPOSITO, G. 1993. Organic matter role in illitic soil colloids flocculation: II. Surface charge. *Soil Science Society of America Journal*, 57, 1246-1253.
- HEINE, J. J., PATELKE, R. L. & ORESKOVICH, J. A. 1998. Characterization of the Kaolin Occurrences in Northern and Central Minnesota.

- HENRIKSEN, A., LIEN, L., TRAAEN, T. S., SEVALDRUD, I. S. & BRAKKE, D. F. 1988. Lake acidification in Norway-present and predicted chemical status. *Ambio (Sweden)*.
- HILEMAN, B. 1983. Acid fog. *Environmental Science & Technology*, 17, 117A-120A.
- HORGNIES, M. & CHOLLET, M. 2011. Composition of Concrete Surfaces after Demoulding and Coating: Comparative Study by XPS, FTIR and Raman Spectroscopies. *Key Engineering Materials*, 466, 215-223.
- HORGNIES, M., WILLIEME, P. & GABET, O. 2011. Influence of the surface properties of concrete on the adhesion of coating: Characterization of the interface by peel test and FT-IR spectroscopy. *Progress in Organic Coatings*, 72, 360-379.
- HUAT, B. B., ALI, F. H. & LOW, T. 2006. Water infiltration characteristics of unsaturated soil slope and its effect on suction and stability. *Geotechnical & Geological Engineering*, 24, 1293-1306.
- HUNTER, R. J. 2013. *Zeta potential in colloid science: principles and applications*, Academic press.
- IMAI, G., KOMATSU, Y. & FUKUE, M. 2006. Consolidation yield stress of Osaka-Bay pleistocene clay with reference to calcium carbonate contents. *J ASTM Int*, 3, 1-9.
- JANCSEK, S. 1991. Definition geotechnischer Parameter für den Einsatz von Schildvortriebsmaschinen mit suspensionsgestützter Ortsbrust. *Forschung+Praxis*, 34, 34-60.
- JAYARAMAN, K., CHOUDHURY, A. & RAO, C. 2002. Sulfates of organic diamines: hydrogen-bonded structures and properties. *Solid state sciences*, 4, 413-422.
- JO, B.-W., KIM, C.-H., TAE, G.-H. & PARK, J.-B. 2007. Characteristics of cement mortar with nano-SiO₂ particles. *Construction and Building materials*, 21, 1351-1355.
- JONES JR, J. B. 2001. *Laboratory guide for conducting soil tests and plant analysis*, CRC press.
- JOZEFACIUK, G. & BOWANKO, G. 2002. Effect of acid and alkali treatments on surface areas and adsorption energies of selected minerals. *Clays and Clay Minerals*, 50, 771-783.
- KAI-LI, J., QUN-QING, L. & SHOU-SHAN, F. 2003. Continuous carbon nanotube yarns and their applications. *Physics*, 8, 506-510.

- KAMON, M., YING, C. & KATSUMI, T. 1997. Effect of acid rain on physico-chemical and engineering properties of soils. *Soils and foundations*, 37.
- KASHIR, M. & YANFUL, E. K. 2001. Hydraulic conductivity of bentonite permeated with acid mine drainage. *Canadian Geotechnical Journal*, 38, 1034-1048.
- KATO, N. & AKIMOTO, H. 1992. Anthropogenic emissions of SO₂ and NO_x in Asia: emission inventories. *Atmospheric Environment. Part A. General Topics*, 26, 2997-3017.
- KAWASHIMA, S., HOU, P., CORR, D. J. & SHAH, S. P. 2012. Modification of Cement-based Materials with Nanoparticles. *Cement and Concrete Composites*.
- KELLER 1964a. The origin of high-alumina clay minerals-a review. *Clays and Clay Minerals*.
- KELLER, J. B. 1964b. A theorem on the conductivity of a composite medium. *Journal of Mathematical Physics*, 5, 548-549.
- KENNEDY, A. & PAPENDICK, R. 1995. Microbial characteristics of soil quality. *Journal of soil and water conservation*, 50, 243-248.
- KIM HUAT, B. 2010. Problematic soils: in search for solution.
- KIRKPATRICK, R. J., YARGER, J., MCMILLAN, P. F., PING, Y. & CONG, X. 1997. Raman spectroscopy of CSH, tobermorite, and jennite. *Advanced Cement Based Materials*, 5, 93-99.
- KÖLLE, P. & DRONSKOWSKI, R. 2004. Hydrogen bonding in the crystal structures of the ionic liquid compounds butyldimethylimidazolium hydrogen sulfate, chloride, and chloroferrate (II, III). *Inorganic chemistry*, 43, 2803-2809.
- KONTOLEONTOS, F., TSAKIRIDIS, P., MARINOS, A., KALOUIDAS, V. & KATSIOTI, M. 2012. Influence of colloidal nanosilica on ultrafine cement hydration: Physicochemical and microstructural characterization. *Construction and Building Materials*, 35, 347-360.
- LAMBE, T. W. 1958. The structure of compacted clay. *Journal of the Soil Mechanics and Foundations Division, ASCE*, 84, 1-34.
- LANE, C. N. 2003. *Acid rain: overview and abstracts*, Nova Publishers.
- LEVY, M. A. 1995. International cooperation to combat acid rain. *Green Globe Yearbook*, 1995, 59-68.
- LIEBAU, F. 2012. *Structural chemistry of silicates: structure, bonding, and classification*, Springer Science & Business Media.

- LORENZ, P. B. 1969. Surface conductance and electrokinetic properties of kaolinite beds. *Clays and Clay minerals*, 17, 223-231.
- MAAIL, S., HUAT, B. B., JAMALUDIN, S., SEW, G. & ALI, F. 2004. Index, engineering properties and classification of tropical residual soils. *Tropical residual soils engineering*, 37-55.
- MALLELA, J., HAROLD VON QUINTUS, P., SMITH, K. L. & CONSULTANTS, E. 2004. Consideration of lime-stabilized layers in mechanistic-empirical pavement design. *The National Lime Association*.
- MINDESS, S. 2002. Applications of Fracture Mechanics to Concrete: Where Do We Go From Here? *ACI Special Publication*, 206.
- MITCHELL & SITAR, N. 1982. Engineering properties of tropical residual soils. *Engineering and Construction in Tropical and Residual Soils*, 30-57.
- MITCHELL & SOGA, K. 1976. *Fundamentals of soil behavior*, Wiley New York.
- MITCHELL, J. 1960. The application of colloidal theory to the compressibility of clays. *Commonwealth Science and Industry Research Organization*, 2, 92-97.
- MITCHELL, J. K. 1976. *Fundamentals of soil behavior*, Wiley New York.
- MITCHELL, J. K. & SOGA, K. 2005. *Fundamentals of soil behavior*.
- MOH, Z.-C. & MAZHAR, M. F. 1900. Effects of method of preparation on index properties of lateritic soils. *Soil Mech & Fdn Eng Conf Proc/Mexico/*.
- MOHAMEDZEIN, Y. E.-A. & ABOUD, M. H. 2006. Compressibility and shear strength of a residual soil. *Geotechnical & Geological Engineering*, 24, 1385-1401.
- MOORE, R. & BRUNSDEN, D. 1996. Physico-chemical effects on the behaviour of a coastal mudslide. *Geotechnique*, 46, 259-278.
- MÜLLER-VONMOOS, M. & LØKEN, T. 1988. Das scherverhalten der tone. *Mitteilungen des Institutes für Grundbau und Bodenmechanik, Eidgenössische Technische Hochschule Zürich*, 23-37.
- NAGARAJ, H. & SRIDHARAN, A. 2010. An improved k o-stress method of determining liquid limit of soils. *International Journal of Geotechnical Engineering*, 4, 557-563.
- NAJI GIVI, A., ABDUL RASHID, S., AZIZ, F. N. A. & SALLEH, M. A. M. 2010. Experimental investigation of the size effects of SiO₂ nano-particles on the mechanical properties of binary blended concrete. *Composites Part B: Engineering*, 41, 673-677.

- NAJI GIVI, A., ABDUL RASHID, S., AZIZ, F. N. A. & SALLEH, M. A. M. 2011. The effects of lime solution on the properties of SiO₂ nanoparticles binary blended concrete. *Composites Part B: Engineering*, 42, 562-569.
- NAZARI, A. & RIAHI, S. 2010. The effect of TiO₂ nanoparticles on water permeability and thermal and mechanical properties of high strength self-compacting concrete. *Materials Science and Engineering: A*, 528, 756-763.
- NORTON, R. 1985. Aspects of the biology and systematics of soil arachnids, particularly saprophagous and mycophagous mites. *Quaestiones Entomologicae*, 21, 523-541.
- OLA, S. 1978. Geotechnical properties and behaviour of some stabilized Nigerian lateritic soils. *Quarterly Journal of Engineering Geology and Hydrogeology*, 11, 145-160.
- OLSEN, R. & MESRI, G. 1970. Mechanisms controlling compressibility of clays. *J. Soil Mechanics and Foundations Division, Proceedings of the American Society of Civil Engineers*, 96, 1863-1978.
- PALOMINO, A. M. & SANTAMARINA, J. C. 2005. Fabric map for kaolinite: effects of pH and ionic concentration on behavior. *Clays and Clay minerals*, 53, 211-223.
- PARIA, S. 2008. Surfactant-enhanced remediation of organic contaminated soil and water. *Advances in Colloid and Interface Science*, 138, 24-58.
- PENNER, D. & LAGALY, G. 2001. Influence of anions on the rheological properties of clay mineral dispersions. *Applied Clay Science*, 19, 131-142.
- PETERSON, S. R. & GEE, G. W. 1985. Interaction between acidic solutions and clay liners: permeability and neutralisation. *Hydraulic Barriers in Soil and Rock, ASTM STP*, 874, 229-245.
- RAJ, J. 1985. Characterisation of the weathering profile developed over a porphyritic biotite granite in Peninsular Malaysia. *Bulletin of the International Association of Engineering Geology-Bulletin de l'Association Internationale de Géologie de l'Ingénieur*, 32, 121-129.
- RAJ, J. K. 1993. Clay minerals in the weathering profile of a quartz-muscovite schist in the Seremban Area, Negeri Sembilan. *Pertanika Journal of Tropical Agricultural Science*, 16, 129-136.
- RAKHECHA, P. 2007. Probable maximum precipitation for 24-h duration over an equatorial region: Part 2-Johor, Malaysia. *Atmospheric research*, 84, 84-90.
- ROSENQVIST, I. T. 1955. *Investigations in the clay-electrolyte-water system*, Norwegian Geotechnical Institute.

- ROSENQVIST, I. T. 1959. Physico-chemical properties of soils: soil-water systems. *Journal of the Soil Mechanics and Foundations Division*, 85, 31-54.
- RUHL, J. L. & DANIEL, D. E. 1997. Geosynthetic clay liners permeated with chemical solutions and leachates. *Journal of Geotechnical and Geoenvironmental Engineering*, 123, 369-381.
- SANTAMARINA, J., KLEIN, K., WANG, Y. & PRENCKE, E. 2002. Specific surface: determination and relevance. *Canadian Geotechnical Journal*, 39, 233-241.
- SARIOSSEIRI & MUHUNTHAN, B. 2008. Geotechnical properties of Palouse loess modified with cement kiln dust and Portland cement. *Proceedings of Geocongress*.
- SARIOSSEIRI, RAZAVI, M., CARLSON, K. & GHAZVINIAN, B. Stabilization of Soils with Portland Cement and CKD, and Application of CKD on Slope Erosion Control. *Geo-Frontiers 2011@ sAdvances in Geotechnical Engineering*, 2011. ASCE, 778-787.
- SARKAR, A. 2009. *Global Climate Change & Sustainable Energy Development*, Pentagon Press.
- SCHLICK, G. 1989. *Adhäsion im Boden-Werkzeug-System*, na.
- SCHOFIELD, R. & SAMSON, H. 1953. *The deflocculation of kaolinite suspensions and the accompanying change-over from positive to negative chloride adsorption*, Blackwell scientific publ.
- SCORZELLI, G., PAOLUZZI, A. & PASCUCCHI, V. 2008a. Parallel solid modeling using BSP dataflow. *International Journal of Computational Geometry & Applications*, 18, 441-467.
- SCORZELLI, R., BERTOLINO, L., LUZ, A., DUTTINE, M., SILVA, F. & MUNAYCO, P. 2008b. Spectroscopic studies of kaolin from different Brazilian regions. *Clay Minerals*, 43, 129-135.
- SHAMSHUDDIN, J. & ANDA, M. 2008a. Charge properties of soils in Malaysia dominated by kaolinite, gibbsite, goethite and hematite. *Bull Geol Soc Malays*, 54, 27-31.
- SHAMSHUDDIN, J. & ANDA, M. 2008b. Charge properties of soils in Malaysia dominated by kaolinite, gibbsite, goethite and hematite. *Bulletin of the Geological Society of Malaysia*, 54, 27-31.
- SHARMA, P., VYAS, S., MAHURE, N., KACHHAL, P., SIVAKUMAR, N. & RATNAM, M. 2011. Effect of acid rain on geotechnical properties of soils of high and low plasticity.

- SINGER, A. 1990. Quartz synthesis by Soxhlet leaching of basaltic rock. *Soil Science*, 150, 630-636.
- SIVAPULLAIAH, P. Effects of soil pollution on geotechnical behaviour of soils. Indian Geotechnical Conference, Guntur, 2009. 933-940.
- SKEMPTON, A. W. & JONES, O. 1944. Notes on the compressibility of clays. *Quarterly Journal of the Geological Society*, 100, 119-135.
- SO, H. B. & LAL, R. 2002. Encyclopedia of Soil Science. Taylor & Francis.
- SOGA, K. 2005. Fundamentals of soil behavior. Hoboken: John Wiley & Sons, Inc.
- SPAGNOLI, G., RUBINOS, D., STANJEK, H., FERNÁNDEZ-STEEGER, T., FEINENDEGEN, M. & AZZAM, R. 2012. Undrained shear strength of clays as modified by pH variations. *Bulletin of Engineering Geology and the Environment*, 71, 135-148.
- SPOSITO, G., SKIPPER, N. T., SUTTON, R., PARK, S.-H., SOPER, A. K. & GREATHOUSE, J. A. 1999. Surface geochemistry of the clay minerals. *Proceedings of the National Academy of Sciences*, 96, 3358-3364.
- SREE, D. & EVANGELINE, Y. S. Study on the shrinkage, swelling and strength characteristics of clay soils under different environmental conditions. Proceedings of IGC, 2011.
- SRIDHARAN, A. 2002. Engineering behaviour of clays: influence of mineralogy. *Chemo-mechanical coupling in clays: from nano-scale to engineering applications*. Swets and Zeitlinger, Lisse, 3-28.
- SRIDHARAN, A., EL-SHAFEI, A. & MIURA, N. 2002. Mechanisms controlling the undrained strength behavior of remolded Ariake marine clays. *Marine Georesources and Geotechnology*, 20, 21-50.
- SRIDHARAN, A., NAGARAJ, T. & SIVAPULLAIAH, P. 1981. Heaving of soil due to acid contamination. *Proc. of International Conference on Soil Mechanics Foundation Engineering*.
- SRIDHARAN, A. & PRAKASH, K. 1999. Mechanisms controlling the undrained shear strength behaviour of clays. *Canadian Geotechnical Journal*, 36, 1030-1038.
- SRIDHARAN, A. & RAO, G. V. 1973. Mechanisms controlling volume change of saturated clays and the role of the effective stress concept. *Geotechnique*, 23, 359-382.
- SRIDHARAN, A., RAO, S. & MURTHY, N. 1986. Liquid limit of montmorillonite soils.

- SRIDHARAN, A., RAO, S. & MURTHY, N. 1988. Liquid limit of kaolinitic soils. *Geotechnique*, 38, 191-198.
- SRIDHARAN, A. & SIVAPULLAIAH, P. V. 2005. Mini compaction test apparatus for fine grained soils. *Geotechnical Testing Journal*, 28, 1-7.
- SRIDHARAN, A. & VENKATAPPA RAO, G. 1979. Shear strength behaviour of saturated clays and the role of the effective stress concept. *Geotechnique*, 29, 177-193.
- STEVENSON, F. J. 1994. *Humus chemistry: genesis, composition, reactions*, John Wiley & Sons.
- STREETS, D. & WALDHOFF, S. 2000. Present and future emissions of air pollutants in China:: SO₂, NO_x, and CO. *Atmospheric Environment*, 34, 363-374.
- SUNIL, B., NAYAK, S. & SHRIHARI, S. 2006. Effect of pH on the geotechnical properties of laterite. *Engineering geology*, 85, 197-203.
- SZYNKARCZUK, J., KAN, J., HASSAN, T. & DONINI, J. 1994. Electrochemical coagulation of clay suspensions. *Clays and clay minerals*, 42, 667-673.
- TAHA, M. & DEBNATH, D. 1999. Interaction of cyanide with residual soil and kaolinite in batch adsorption tests. *J Inst Eng Malaysia*, 60, 49-57.
- TAHA, M. R. & TAHA, O. M. E. 2012. Influence of nano-material on the expansive and shrinkage soil behavior. *Journal of Nanoparticle Research*, 14, 1-13.
- TAHA, M. R. & YING, T. 2008. effects of carbon nanotube on kaolinite: basic geotechnical behavior.
- TAN, B., HUAT, B., SEW, G. & ALI, F. 2004. Country case study: engineering geology of tropical residual soils in Malaysia. *Tropical residual soils engineering*, 237-244.
- TAN, K. & BARTLETT, R. J. 1994. Principles of soil Chemistry. *Soil Science*, 157, 330.
- TAN, K. H. 2009. *Environmental soil science*, CRC Press.
- TAN, K. H. 2010. *Principles of soil chemistry*, CRC Press.
- TAYLOR 1993. Nanostructure of C-S-H: Current status. *Advanced cement based materials*, 1, 38-46.
- TAYLOR, R. K. 1985. Cation exchange in clays and mudrocks by methylene blue. *Journal of Chemical Technology and Biotechnology. Chemical Technology*, 35, 195-207.

- TERZAGHI, K., PECK, R. B. & MESRI, G. 1996. *Soil mechanics in engineering practice*, John Wiley & Sons.
- THEWES, M. 2003. Adhäsion von Tonböden beim Tunnelvortrieb mit Flüssigkeitsschilden. *geotechnik*, 26.
- TODD, D. K. & MAYS, L. W. 1980. Groundwater Hydrology Edition.
- TOMBÁ CZ, E. & SZEKERES, M. 2004. Colloidal behavior of aqueous montmorillonite suspensions: the specific role of pH in the presence of indifferent electrolytes. *Applied Clay Science*, 27, 75-94.
- TOMBÁ CZ, E. & SZEKERES, M. 2006. Surface charge heterogeneity of kaolinite in aqueous suspension in comparison with montmorillonite. *Applied Clay Science*, 34, 105-124.
- TREFIL, J. S. 2001. *Encyclopedia of science and technology*, Taylor & Francis.
- TSUYUKI, N., WATANABE, R., KOIZUMI, K., UMEMURA, Y. & MACHINAGA, O. 2000. Effects of barium salt on the fixation of chloride ions in hardened mortars. *Cement and concrete research*, 30, 1435-1442.
- VAN OLPHEN, H. & HSU, P. H. 1978. An Introduction to Clay Colloid Chemistry. *Soil Science*, 126, 59.
- VANCE, G. F. & DAVID, M. B. 1991. FOREST SOIL RESPONSE TO ACID AND SALT ADDITIONS-OF SULFATE: III. SOLUBILIZATION AND COMPOSITION OF DISSOLVED ORGANIC CARBON. *Soil Science*, 151, 297-305.
- VICKERS, B. 1983. *Laboratory work in soil mechanics*.
- WANG, Y.-H. & SIU, W.-K. 2006. Structure characteristics and mechanical properties of kaolinite soils. I. Surface charges and structural characterizations. *Canadian Geotechnical Journal*, 43, 587-600.
- WARKENTIN, B. 1972. Use of the liquid limit in characterizing clay soils. *Canadian Journal of Soil Science*, 52, 457-464.
- WATKINS, J. A. 2015. Tectonic and Aqueous Processes in the Formation of Mass-wasting Features on Mars and Earth.
- WEH, M., ZIEGLER, M. & ZWICK, O. 2009. Verklebungen bei EPB-Vortrieben in wechselndem Baugrund: eintrittsbedingungen und Gegenmassnahmen. *Forschung+ Praxis*, 43, 185-189.
- WONG, C., VENNEKER, R., UHLENBROOK, S., JAMIL, A. & ZHOU, Y. 2009. Variability of rainfall in Peninsular Malaysia. *Hydrology and Earth System Sciences Discussions*, 6, 5471-5503.

- YAMAGUCHI, H., UI, M., FUKUDA, M. & KUROSHIMA, I. 1993. Considerations on physico-chemical properties of soils taking acid rain into consideration. *Tsuchi-to-Kiso, JGS*, 41, 47-52.
- YULE, C. M. & GOMEZ, L. N. 2009. Leaf litter decomposition in a tropical peat swamp forest in Peninsular Malaysia. *Wetlands Ecology and Management*, 17, 231-241.
- ZHAO, H., BHATTACHARJEE, S., CHOW, R., WALLACE, D., MASLIYAH, J. H. & XU, Z. 2008. Probing surface charge potentials of clay basal planes and edges by direct force measurements. *Langmuir*, 24, 12899-12910.
- ZHAO, Y., HU, L. B., CUI, P. & HUECKEL, T. Evolution of shear strength of clayey soils in a landslide due to acid rain: a case study in the area of Three Gorges, China. Proc. 1st North Am. Landslide Conference. Vail, Colorado USA, 2007. 1656-1666.
- ZUO, Y. & HOIGNE, J. 1992. Formation of hydrogen peroxide and depletion of oxalic acid in atmospheric water by photolysis of iron (III)-oxalato complexes. *Environmental Science & Technology*, 26, 1014-1022.
- ZYGANITIDIS, I., STEFANIDOU, M., KALFAGIANNIS, N. & LOGOTHETIDIS, S. 2011. Nanomechanical characterization of cement-based pastes enriched with SiO₂ nanoparticles. *Materials Science and Engineering: B*, 176, 1580-1584.