



UNIVERSITI PUTRA MALAYSIA

***EFFECT OF MANGANESE AND CADMIUM ON BIOLOGICAL
ATTRIBUTES OF WILD WATER SPINACH (*Ipomoea aquatica* Forssk.)***

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By

BILLY GUAN TECK HUAT

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in
Fulfillment of the Requirements for the Degree of Doctor of Philosophy**

September 2017

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

EFFECT OF MANGANESE AND CADMIUM ON BIOLOGICAL ATTRIBUTES OF WILD WATER SPINACH (*Ipomoea aquatica* Forssk.)

By

BILLY GUAN TECK HUAT

September 2017

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Heavy metals are inorganic pollutants that are hazardous and toxic to the environment. Agricultural activities have indirectly introduced heavy metals peculiarly manganese (Mn) and cadmium (Cd) to the ecosystem and eventually have polluted aquatic ecosystem which included the ponds located in Universiti Putra Malaysia. Water pollution caused by the heavy metals can greatly affect the life of the wild water spinach (*Ipomoea aquatica* Forssk.), an edible aquatic plant that is living in the ponds. Consequently, human health can be threatened when the metal-contaminated wild water spinach was foraged for consumption. Hence, the metals effects of Mn and Cd on the health status, growth, anatomy, and DNA quality of the wild water spinach were studied. Furthermore, the metal uptake ability by the wild water spinach was determined. The metal bioavailability and health risk were also assessed upon consumption of the metal-contaminated wild water spinach. The mature wild water spinach was hydroponically cultivated under greenhouse conditions and was subjected to Mn and Cd treatments which included low treatment (0.30 mg/L for Mn and 0.10 mg/L for Cd), high treatment (1.50 mg/L for Mn and 0.50 mg/L for Cd), and the control (distilled water) for seven days. ANOVA analysis indicated that significant reduction was observed for roots length and surface area, shoots length, leaves surface area in the metal-contaminated wild water spinach with the increasing Mn and Cd concentrations ($p < 0.05$). Toxicity symptoms such as chlorosis and necrosis also occurred on the wild water spinach from the metal exposure. In the cellular level, the xylem, phloem, epidermis, parenchyma, sclerenchyma, and cell walls of the cross-sectional and longitudinal roots, stems, and leaves have experienced breaking and changes in size, shape, and arrangement that were induced by the metal accumulation. ANOVA results showed that the leaves' DNA concentrations were significantly reduced ranging from 67.73 to 195.54 ng/ μ L and 56.10 to 212.05 ng/ μ L at higher Mn and Cd concentrations; similarly to the changes in DNA purity ($p < 0.05$). The ANOVA statistics showed that the removal efficiency, water-to-shoot bioaccumulation factor (BAF), and root-to-shoot translocation factors (TF) was significantly reduced at higher Mn concentrations ($p < 0.05$). The highest concentration of Mn and Cd was found in the dried (DHS) and raw (RHS) shoots with the highest slope values of 3.75

and 19.50, respectively. Both Mn and Cd had the highest bioaccessibility for absorption in the gastric phase (slope values = 9.68 and 28.28) than intestinal phase (slope values = 0.24 and 17.99). The health risk index showed values > 1 , indicated that the raw (RHS) and cooked (CHS) wild water spinach contaminated with Mn and Cd were not safe to be consumed for the studied population in Selangor, Malaysia. As conclusion, impacts of Mn and Cd were clearly seen when changes occurred in the health status, growth, histological structure, and DNA quality of the metal-contaminated wild water spinach. These metals absorbed in the human gastrointestinal tract could eventually cause health hazards when consuming the metal-contaminated wild water spinach as demonstrated in this work. Nevertheless, wild water spinach can serve as an alternative for phytoremediation on metals-contaminated aqueous medium due to its fairly good metal uptake ability.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**KESAN MANGAN DAN KADMIUM KE ATAS ATRIBUT BIOLOGI
KANGKUNG LIAR (*Ipomoea aquatica* Forssk.)**

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Logam berat adalah bahan pencemar inorganik yang berbahaya dan bertoksik kepada alam sekitar. Aktiviti pertanian secara tidak langsung menyebabkan logam berat khususnya mangan (Mn) dan kadmium (Cd) memasuki ekosistem dan akhirnya telah mencemarkan ekosistem akuatik termasuklah kolam-kolam yang terletak berhampiran di Universiti Putra Malaysia. Pencemaran air oleh logam berat tersebut boleh memberi kesan kepada kehidupan kangkung liar (*Ipomoea aquatica* Forssk.), iaitu sejenis tumbuhan akuatik yang boleh dimakan yang hidup di dalam kolam. Oleh demikian, kesihatan manusia terancam apabila kangkung liar yang tercemar oleh logam berat dimakan oleh mereka. Jadi, kesan-kesan Mn and Cd terhadap status kesihatan, pertumbuhan, anatomi, dan kualiti DNA bagi kangkung liar dikaji. Tambahan pula, keupayaan pengambilan logam berat oleh kangkung liar perlu ditentukan. Bioavailabiliti logam berat dan risiko kesihatan juga telah dinilai apabila kangkung liar tercemar oleh logam berat dimakan. Kangkung liar yang matang telah ditanam secara hidroponik di dalam rumah hijau dan diberikan rawatan Mn dan Cd pada kepekatan yang rendah (0.30 mg/L untuk Mn dan 0.10 mg/L untuk Cd), kepekatan yang tinggi (1.50 mg/L untuk Mn dan 0.50 mg/L untuk Cd), dan air suling sebagai kawalan selama tujuh hari. Analisis ANOVA menunjukkan pengurangan yang ketara telah diperhatikan bagi panjang dan kawasan permukaan akar, panjang pucuk, dan kawasan permukaan daun kangkung liar tercemar oleh logam berat dengan peningkatan kepekatan Mn dan Cd ($p < 0.05$). Simptom toksik iaitu klorosis dan nekrosis juga berlaku pada kangkung liar selepas diberikan rawatan logam berat. Kajian histologi menunjukkan sel xilem, floem, epidermis, parenkima, sklerenkima, dan dinding sel bagi keratan rentas dan memanjang akar, batang, dan daun telah mengalami pemecahan dan perubahan saiz, bentuk, dan susunan yang disebabkan oleh pengumpulan logam berat. Keputusan ANOVA menunjukkan bahawa pengurangan yang signifikan pada kepekatan DNA daun di antara 67.73 dan 195.54 ng/ μ L dan antara 56.10 dan 212.05 ng/ μ L apabila kepekatan Mn dan Cd semakin meningkat. Pengurangan yang ketara juga berlaku pada ketulen DNA daun ($p < 0.05$). Statistik ANOVA menunjukkan bahawa *removal efficiency*, faktor biokonsentrasi *water-to-shoot* (BAF), dan faktor translokasi *root-to-shoot* (TF) telah dikurangkan dengan ketara pada kepekatan Mn yang tinggi ($p < 0.05$).

Kandungan Mn and Cd yang tertinggi telah dijumpai di CHS and RHS dengan kecerunan tertinggi iaitu 3.75 dan 19.50. Kedua-dua logam berat ini menunjukkan bioasesibiliti tertinggi dalam proses penyerapan dalam fasa gastrik (Nilai kecerunan = 9.68 dan 28.28) berbanding dengan fasa usus (Nilai kecerunan = 0.24 dan 17.99). Indeks risiko bahaya (HRI) menunjukkan nilai > 1 , menunjukkan kangkung liar yang tercemar dengan Mn and Cd adalah tidak selamat untuk dimakan bagi populasi yang telah dikaji di Selangor, Malaysia. Secara kesimpulannya, kesan-kesan toksik Mn dan Cd dapat dilihat dengan jelas apabila perubahan berlaku pada status kesihatan, pertumbuhan, histologi, dan kualiti DNA Logam berat akan diserap dalam saluran pencernaan manusia dan berkemungkinan merbahaya kepada kesihatan. Namun demikian, kangkung liar boleh digunakan sebagai alternatif untuk fitoremediasi bagi medium akueus yang tercemar dengan logam berat kerana tumbuhan ini mempunyai keupayaan pengambilan logam berat yang agak baik.



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I certify that a Thesis Examination Committee has met on 8 September 2017 to conduct the final examination of Billy Guan Teck Huat on his thesis entitled “Effect of Manganese and Cadmium on Biological Attributes of Wild Water Spinach (*Ipomoea aquatica* Forssk.)” in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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TABLE OF CONTENTS

ABSTRACT	Page
ABSTRAK	i
ACKNOWLEDGEMENTS	iii
APPROVAL	v
DECLARATION	vi
LIST OF TABLES	viii
LIST OF FIGURES	xiii
LIST OF SYMBOL	xv
LIST OF ABBREVIATIONS	xvii
	xix

CHAPTER

1 INTRODUCTION

2 LITERATURE REVIEW

2.1	Heavy Metals Pollution in General	4
2.1.1	Agricultural Pollution for Heavy Metals in Soils, Water and Air	4
2.1.2	Indirect Heavy Metals Pollution in Soils, Water, and Food Chain from Surface Runoff	5
2.1.3	The Threats from Less Popular Heavy Metals	7
2.2	Status of Manganese and Cadmium Pollution in the Surface Water and Other Water Sources in Malaysia	8
2.3	Previous Studies and Their Limitations in Malaysia	9
2.3.1	Effects of Manganese and Cadmium on the Biological Attributes in Plants	9
2.3.2	Phytoremediation on Manganese and Cadmium Pollution	9
2.3.3	Heavy Metals Bioavailability through <i>In Vitro</i> Human Gastrointestinal Digestion	10
2.3.4	Health Risk Assessment on the Consumption of Heavy Metals Contaminated Food	10
2.4	The Threats from Manganese and Cadmium to the Environment and Biological System	11
2.4.1	Source of Manganese Pollution and the Risk of Manganese to the Biological System	11
2.4.2	Source of Cadmium Pollution and the Risk of Cadmium to the Biological System	12
2.5	Surface Water Quality and Maximum Permissible Limit for Manganese and Cadmium	13
2.6	Past and Present Heavy Metals Mitigation Approach	16
2.7	Bioremediation and Phytoremediation for Heavy Metals	19
2.8	Various Concepts of Phytoremediation	21
2.8.1	Techniques and Application of Phytoremediation	21

2.8.2	Phytofiltration	23
2.8.3	Rhizofiltration	26
2.8.4	Handling and Disposal of Phyto-remediated Residue	27
2.9	Choice of Phytoremediator	28
2.9.1	Hyperaccumulator for Heavy Metals	28
2.9.2	Aquatic Plants	30
2.9.3	Edible Aquatic Plants	31
2.9.4	Water Spinach	32
2.9.5	Wild Water Spinach	33
2.10	Heavy Metals Uptake Mechanism in Plants	35
2.11	Bioavailability and Bioaccessibility of Heavy Metals	39
2.12	Assessment on the Impacts of Heavy Metals on Plants	41
2.12.1	Growth and Morphology	41
2.12.2	Histological Structure	42
2.12.3	Genetic Assessment	46
3	MATERIALS AND METHODS	
3.1	Screening of Heavy Metals Pollution in the Selected Ponds Water	51
3.2	Collection and Cultivation of Wild Water Spinach	55
3.3	Setting-up Hydroponic System and Running the Heavy Metal Uptake Experiments	57
3.4	Harvesting of the Control and Metal-contaminated Wild Water Spinach	60
3.4.1	Health Status and Growth Study	60
3.4.2	Histological Study on the Control and Metal-contaminated Wild Water Spinach	61
3.4.3	DNA Quality Study on the Control and Metal-contaminated Wild Water Spinach	63
3.4.4	Acid Digestion on the Control and Metal-contaminated Wild Water Spinach	65
3.4.5	<i>In Vitro</i> Gastrointestinal Digestion on the Dried, Raw, and Cooked of the Control and Metal-contaminated Wild Water Spinach	66
3.5	Data Collection and Analysis	69
3.5.1	Heavy Metals Uptake Assessment	69
3.5.2	Heavy Metals Bioaccessibility Assessment	71
3.5.3	Health Risk Assessment	71
3.5.4	Statistical Analysis	72
4	RESULTS AND DISCUSSION	
4.1	Results	74
4.1.1	Characteristics of the Health Status for the Control and Metal-contaminated Wild Water Spinach	74
4.1.2	Characteristics of the Growth for the Control and Metal-contaminated Wild Water Spinach	79
4.1.3	Characteristics of the Histological Structure for the Control and Metal-contaminated Wild Water Spinach	82

4.1.4	Characteristics of the DNA Quality for the Control and Metal-contaminated Wild Water Spinach	92
4.1.5	Characteristics of the Nutrient Quality Before and After Heavy Metal Treatment	95
4.1.6	Characteristics of Manganese And Cadmium Uptake by the Wild Water Spinach	97
4.1.7	Characteristics of the Metal Bioavailability for the Control and Metal-contaminated Wild Water Spinach	100
4.1.8	Health Risk Assessment	105
4.2	Discussion	106
4.2.1	Plant Health Status	106
4.2.2	Plant Growth	108
4.2.3	Plant Histological Structure	115
4.2.4	Plant DNA Quality	121
4.2.5	Plant Heavy Metal Uptake	123
4.2.6	Plant Heavy Metal Bioavailability	131
4.2.7	Human Health Risk Assessment from the Consumption of the Metal-contaminated Wild Water Spinach	134
5	CONCLUSIONS AND RECOMMENDATIONS	
5.1	Conclusions	135
5.2	Recommendations for Future Studies	136
	REFERENCES	137
	APPENDICES	193
	BIODATA OF STUDENT	247
	LIST OF PUBLICATIONS	248

LIST OF TABLES

Table	Page
2.1 Cadmium contamination in various types of plants	6
2.2 Cadmium contamination in various types of aquatic animals	7
2.3 Highest Mn reported in surface waters in certain countries	15
2.4 Highest Cd reported in surface waters in certain countries	15
2.5 Permissible limits of Mn and Cd regulated by the authorities from different countries	15
2.6 Conventional technologies for heavy metals treatment	17
2.7 The differences for the selected water treatment methods	18
2.8 Concepts and applications of bioremediation and phytoremediation	20
2.9 Phytoremediation techniques and their mechanisms and applications	22
2.10 The overall advantages and disadvantages of phytoremediation technique	23
2.11 The overall strengths and limitations of rhizofiltration technique	26
2.12 Phyto-remediated residue treatment methods and their potential resource utilization	27
2.13 Hyperaccumulators for different type of heavy metals	29
2.14 Metals accumulation found in edible aquatic plants	32
2.15 The characteristics between water spinach and wild water spinach	34
2.16 Differences between bioavailability and bioaccessibility of heavy metals at various aspects	40
2.17 Affected tissues in plant organs from the heavy metals toxicity	43
3.1 The locations and coordinates of the selected sites for the water sampling	51
3.2 Baseline data on the elements and <i>in situ</i> water quality parameters at sites A, B, and C	54
3.3 Baseline data on the Mn and Cd concentration detected in the wild water spinach roots and shoots from the sites A, B and C (mean \pm SE, n = 3)	56
3.4 Initial weights of mature cultivated wild water spinach before treatment (mean \pm SE, n = 3)	58
3.5 Chlorosis rating scale for plant	61
3.6 Analysis tools used in this work	73
4.1 The mean number of plants with different conditions after exposure to Mn and Cd (n = 3)	76
4.2 The number of plants that associated with chlorosis at different conditions scales after exposure to Mn and Cd (n = 3)	77
4.3 The range of reduction for the growth parameters from the metal treatment	79
4.4 Parameters of plant growth for the uncontaminated and Mn-contaminated wild water spinach (mean \pm SE, n = 3 ^a)	80
4.5 Parameters of plant growth for the uncontaminated and Cd-contaminated wild water spinach (mean \pm SE, n = 3)	81
4.6 DNA concentration in the different organs of wild water spinach for each metal (mean \pm SE, n = 3)	94
4.7 Means of DNA purity detected in different organs of wild water spinach for each metal (mean \pm SE, n = 3)	94

4.8	<i>In situ</i> measurements for the uncontaminated and metal-contaminated nutrient solution	95
4.9	Manganese and cadmium concentration in the uncontaminated and metal-contaminated nutrient solution (mean \pm SE, n = 3)	96
4.10	Manganese and cadmium concentration in the uncontaminated and metal-contaminated wild water spinach (mean \pm SE, n = 3)	97
4.11	Removal efficiency for Mn and Cd by the wild water spinach at different treatment concentrations (mean \pm SE, n = 3)	98
4.12	Bioaccumulation factor of Mn and Cd for the wild water spinach at different treatment concentrations (mean \pm SE, n = 3)	99
4.13	Translocation factor of Mn and Cd for the wild water spinach at different treatment concentrations (mean \pm SE, n = 3)	99
4.14	Manganese and cadmium concentrations detected in the wild water spinach samples at different treatment concentrations and phases (mean \pm SE, n = 3)	101
4.15	Comparison of bioaccessibilities of Mn between the DHS, RHS, and CHS at different digestion phases and treatment concentrations (mean \pm SE, n = 3)	103
4.16	Comparison of bioaccessibilities of Cd between the DHS, RHS, and CHS at different digestion phases and treatment concentrations (mean \pm SE, n = 3)	104
4.17	Daily intake of metals from the consumption of metal-contaminated wild water spinach (mean \pm SE, n = 3)	105
4.18	Health risk index for Mn and Cd in raw and cooked wild water spinach (mean \pm SE, n = 3)	106
4.19	Summary of the statistical results of all the studied components in the plant growth of wild water spinach	108
4.20	Comparisons of the changes observed in plants' tissues caused by metal toxicity	117
4.21	DNA degradation found on plant species resulted from heavy metal toxicity	122

LIST OF FIGURES

Figure		Page
2.1	General Phytofiltration Process in a Hydroponic System (Own Drawing)	25
2.2	Four Main Mechanisms in Heavy Metals Uptake by Plants (Own Drawing)	38
2.3	DNA Gel Electrophoresis of the Southern Cutgrass Leaves under Chromium (Cr) Treatment (Cai <i>et al.</i> , 2014)	47
2.4	Agarose Gel Electrophoresis Showing DNA Degradation in Chickpea Leaves Contaminated with Vanadium (Imtiaz <i>et al.</i> , 2016)	47
3.1	Flow Chart of Research Design of the Study	50
3.2	Pond Water Sampling at Sites A, B, and C around Universiti Putra Malaysia, Selangor, Malaysia	52
4.1	Physical Appearance for the Wild Water Spinach (a) Healthy Plant; (b) Unhealthy Plant with Chlorosis	78
4.2	Cross Section of Wild Water Spinach Roots (Magnification 400×) (a) Mn Experiment; (b) Cd Experiment (n = 3). Abbreviation: Epidermis (ep), Parenchyma (p), Sclerenchyma (scl), Xylem (xyl), and Phloem (phl). Scale: 100 µm. Arrow Indicates the Breaking of Cortex Cells and Changes in Size, Shape, and Arrangement of Vascular Bundle	84
4.3	Cross Section of Wild Water Spinach Stems (Magnification 400×) (a) Mn Experiment; (b) Cd Experiment (n = 3). Abbreviation: Epidermis (ep), Collenchyma (c), Parenchyma (p), Sclerenchyma (scl), Xylem (xyl), and Phloem (phl). Scale: 100 µm. Arrow Indicates the Breaking of Cortex Cells and Changes in Size, Shape, and Arrangement of Vascular Bundles	85
4.4	Cross Section of Wild Water Spinach Leaves (Magnification 100×) (a) Mn Experiment; (b) Cd Experiment (n = 3). Abbreviation: Epidermis (ep), Collenchyma (c), Parenchyma (p), Sclerenchyma (scl), Xylem (xyl), and Phloem (phl). Scale: 100 µm. Arrow Indicates the Breaking of Cortex Cells, Vascular Bundles, Etc.	86
4.5	Longitudinal Sections of Wild Water Spinach (a) Root (Magnification 100×; Scale: 100 µm); (b) Stem (Magnification 400×; Scale: 150 µm); (c) Leaf (Magnification 100×; Scale: 100 µm). Abbreviation: Xylem (xyl), Phloem (phl), Cortex (ct), Guard Cell (gc), Stoma (st), Epidermis (ep), and Mesophyll (mp)	88
4.6	Longitudinal Sections (Magnification 400×) of Wild Water Spinach Roots' Cortex (Vacuole Region) for the Selected Sample (a) Mn-C1; (b) Mn-T1a; (c) Mn-T2a; (d) Cd-C1; (e) Cd-T1a; (f) Cd-T2a. Scale: 150 µm. Arrow Indicates the Localization of Metal in the Cortex Regions	89
4.7	Longitudinal Sections (Magnification 400×) of Wild Water Spinach Stems' Cortex (Vacuole Region) for the Selected Sample (a) Mn-C1; (b) Mn-T1a; (c) Mn-T2a; (d) Cd-C1; (e) Cd-T1a; (f) Cd-T2a. Scale: 150 µm. Arrow Indicates the Localization of Metal in the Cortex Regions and Thickening of Cell Walls	90

4.8	Longitudinal Sections (Magnification 400×) of Wild Water Spinach Leaf Tissues for the Selected Sample (a) Mn-C1; (b) Mn-T1a; (c) Mn-T2a; (d) Cd-C1; (e) Cd-T1a; (f) Cd-T2a. Scale: 150 μm. Arrow Indicates the Thickening of Mesophyll and Spiral	91
4.9	Agarose Gel Electrophoresis of the DNA extracted from the Wild Water Spinach (a) Roots; (b) Stems; (c) Leaves with Identical Sample Arrangement. Lanes 1 and 20 = The Lambda HindIII DNA Marker (fragments from 564 to 2027, 2322, 4361, 6557, 9416, and 23130 bp); Lanes 2 to 7 = The Mn-Control Specimen of 1 to 6; Lanes 8 to 13 = The Mn-T1-Treated Specimen of 1 to 6; Lanes 14 to 19 = The Mn-T2-Treated Specimen of 1 to 6; Lanes 21 to 26 = The Cd-T2-Treated Specimen of 6 to 1; Lanes 27 to 32 = The Cd-T1-Treated Specimen of 6 to 1; Lanes 33 to 38 = The Cd-Control Specimen of 6 to 1	93
4.10	Comparisons between the Mean Cd Concentrations and Maximum Permissible Limits (mean ± SE, n = 3)	102

LIST OF SYMBOL

%	Percentages
°C	Celsius
μmol	Micromols
rpm	Revolutions per minute
mBar	Millibars
H	Hours
min	Minutes
ms	Millisiemens
μS/cm	Microsiemens per centimeter
L	Liters
mL	Milliliters
μL	Microliters
cm ²	Square centimeters
cm	Centimeters
mm	Millimeters
μm	Micrometers
nm	Nanometers
kg	Kilograms
g	Grams
G	Gravity forces
mg	Milligrams
μg	Micrograms
mg/kg	Milligrams per kilogram
mg/g	Milligrams per gram

$\mu\text{g/g}$	Micrograms per gram
mg/L	Milligrams per liter
$\mu\text{g/L}$	Micrograms per liter
mg/mL	Milligrams per milliliter
$\text{mg}/\mu\text{L}$	Milligrams per microliter
$\text{ng}/\mu\text{L}$	Nanograms per microliter
g/d	Grams per day
mg/d	Milligrams per day
mg/kg/d	Milligrams per kilogram per day
$\mu\text{g/d}$	Micrograms per day
kg/d	Kilograms per day
g/cm^3	Grams per cubic centimeter
ppm	Parts per million
$\text{mg/m}^2\text{year}$	Milligrams per square meter per year
ng/m^3	Nanograms per cubic meter
$\mu\text{g/m}^3$	Micrograms per cubic meter
gm/Nm^3	Grams per normal cubic meter
mA	Microamperes
g/mL	Grams per milliliter
M	Molars
mM	MilliMolars
μM	MicroMolars
$\mu\text{M/L}$	MicroMolars per liter
mg/dm^3	Milligrams per cubic decimeter

LIST OF ABBREVIATIONS

AAS	Atomic absorption spectroscopy
ANOVA	Analysis of variance
ANVISA	National Agency for Sanitary Vigilance
BAF/BCF	Bioaccumulation factor/bioconcentration factor
C, T1, and T2	Control, low treatment, and high treatment
CAC	Codex Alimentarius Commission
Cd	Cadmium
CTAB	Cetyltrimethylammonium bromide
DHS, RHS, and CHS	Dry-harvest shoots, raw-harvest shoots, and cook-harvest shoots
DNA	Deoxyribonucleic acid
DO	Dissolved oxygen
DOE	Department of Environment of Malaysia
DSM	Department of Statistics Malaysia
EC	Electrical conductivity/ European Commission
EQA	Malaysia Environmental Quality Act
EU	European Union
FAA	Formalin, acetic acid, and alcohol
FAMA	Federal Agricultural Marketing Authority
FAO/WHO	Joint Food and Agriculture Organization and World Health Organization
G1, G2, and G3	Greenhouse 1, greenhouse 2, and greenhouse 3
GT	Gastrointestinal tract
HKFEHD CFS	Hong Kong Food and Environmental Hygiene Department, Centre for Food Safety
HMs	Heavy metals

HRI	Health risk index
ICP-OES	Inductively coupled plasma optical emission spectrometry
INWQS	Interim National Water Quality Standards Malaysian
MHPRC	Ministry of Health of the People's Republic of China
MFR	Malaysian Food Regulations
Mn	Manganese
MWA	Malaysian Water Association
PFA	Prevention of Food Adulteration Act
ROS	Reactive oxygen species
SRM	Standard reference material
TF	Translocation factor
UK	United Kingdom
USA	United States of America
USDA	United States Department of Agriculture
USDHHS	United States Department of Health and Human Services
USEPA	United States Environmental Protection Agency
WEPs	Wild edible plants
WHO	World Health Organization
WHO/EU	World Health Organization Regional Office for Europe

CHAPTER 1

INTRODUCTION

Surface water serves as the breeding habitat for aquatic life. However, the quality of surface water is deteriorating due to the increasing of anthropogenic activities. Huang *et al.* (2015) have reported that the number of clean rivers in Malaysia was reduced from 338 to 278 when compared to year 2005 with 2012. Surface water pollution occurs when there is excessive of organic or inorganic pollutant present in the water. Heavy metals (HMs) such as chromium (Cr), copper (Cu), zinc (Zn), manganese (Mn), iron (Fe), magnesium (Mg), nickel (Ni), and cobalt (Co), mercury (Hg), arsenic (As), cadmium (Cd), and lead (Pb) are examples of inorganic pollutants. Agriculture activity is one of the anthropogenic sources for heavy metals particularly Mn and Cd. Many of the agrochemicals used in the agriculture contain Mn and Cd (Zhao *et al.*, 2015). Thus the uncontrollable usage of fertilizers and pesticides can indirectly pollute the surface waters like lakes, ponds, and streams that are located near to the agricultural land through surface runoff (Parris, 2011; Wang *et al.*, 2016).

Heavy metal contamination in surface water can endanger the aquatic life that is living in the water. Aquatic plants absorb nutrients from the water through roots that are essential for photosynthesis. Meanwhile, heavy metals that are existed in the water are being absorbed by the aquatic plants as well. Consequently, the continuous accumulation of heavy metals can disrupt the plant growth and trigger photo-oxidative stress (Lambert and Davy, 2011). Heavy metals contaminated aquatic plants in the water become a human health concern because some species of aquatic plants are edible. Examples of edible aquatic plants are wild water spinach, wild taro, cattails, wild rice, etc. The edible aquatic plants mentioned previously are actually being harvested or foraged for consumption by the locals in some countries including Malaysia. The heavy metals that were bioaccumulated in the edible aquatic plants can be absorbed, transferred, and stored in the human bodies from ingestion; in the long-term, the central nervous system, liver, kidneys, heart, lungs, skin, reproduction can be damaged due to the carcinogenicity of heavy metals (Panagos *et al.*, 2013). One of the most serious cases of heavy metal poisoning was happened in Toyama, Japan in the early 1950s where the locals suffered a disease called as *itai-itai* disease that was caused by acute cadmium toxicity (Bhattacharya, 2009; Yang *et al.*, 2012). The outbreak of the disease was due to the consumption of cadmium contaminated rice.

Different countries have different mitigation approaches to overcome the water pollution issues. In Malaysia, legislations such as Environmental Quality Act (EQA) 1974, National Water Quality Standards (NWQS), Malaysian Water Association's (MWA) raw water quality criteria, and water quality index (WQI) are adopted to control the water pollution; besides that, swale, infiltration facility, bioretention, gross pollutant traps (GPTs), sediment ponds, wet ponds, wetlands, and wastewater treatment plant were implemented which were proposed in the Urban Stormwater Management Manual for Malaysia (MSMA) to improve the water quality (Mamum and Zainudin, 2013). On the other hand, a hands-on approach is applied in China to deal with the water pollution which includes water diversions, dredging, and wetland construction

(Yang *et al.*, 2010). In addition, physical, chemical, and biological methods, for example membrane filtration, ion exchange, electrodialysis, and biosorption can be carried out to solve the water pollution problems (Gunatilake, 2015). These techniques are effective but also expensive, labor and energy intensive, hazardous, and complicated (Barakat, 2011).

Phytoremediation is a promising method that is relatively low cost, safe, and easy to remove unwanted heavy metals from the contaminated water. Phytoremediation is the use of plants to remediate contamination. In order to effectively remove heavy metals from the water, it is crucial to select suitable plant species that able to adapt well in the aqueous environment. Aquatic plants are ideal choices because of their free-floating and submerge capability in water. Water hyacinth, water lettuce, and duckweed are examples of heavy metal hyperaccumulating aquatic plants. Generally, heavy metals is taken, accumulated, translocated, and stored in plant organs. The metal uptake mechanisms by a plant can be through adsorption, accumulation, and absorption. Phytoremediation is becoming increasingly popular, trendy, and fast growing especially in the United States and Europe (Lelie *et al.*, 2001). Nevertheless, phytoremediation is still not well-known in the Asian countries and thus it is deserved to be further explored.

This research has proposed an edible aquatic plant that is commonly found in the ponds or lakes to be added into the existing list of potential plants for phytoremediation. Wild water spinach or *Kangkung* is one of the native plants in Malaysia and it is merely considered as a type of vegetable; despite that, this underrated plant can be exploited for the application of phytoremediation to clean the heavy metals contaminated surface water. It will be beneficial to promote the establishment of many research and development (R & D) companies to focus in phytoremediation technology in the future. Since wild water spinach is easily available and abundant but most importantly it is effective in eliminating heavy metals, therefore it will certainly be an attractive addition to other aquatic plants species such as water hyacinth and duckweed that were hugely studied for remediating heavy metal polluted water. Furthermore, this research will help to promote public awareness in regards to food safety. Wild water spinach is able to uptake heavy metals from its surrounding and it will be a public health concern when eating the metal-contaminated wild water spinach. So far it is yet to discover any casualty involved due to the consumption of metal-contaminated wild water spinach.

The objectives of this research are listed as follows:

1. To examine the health status and growth of the metal-contaminated wild water spinach.
2. To identify and investigate the changes on the microscopic cell structure and DNA quality of the metal-contaminated wild water spinach.
3. To determine the effectiveness of Mn and Cd uptake by wild water spinach.
4. To assess the bioavailability of metals for absorption from the *in vitro* gastrointestinal digestion of wild water spinach.

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