



UNIVERSITI PUTRA MALAYSIA

***REDUCING SOIL ACIDITY IN ULTISOLS AND OXISOLS USING RED
GYPSUM AND BIOCHAR IN MALAYSIA***

TALHA IBRAHIM ZANNAH

FP 2017 19



**REDUCING SOIL ACIDITY IN ULTISOLS AND OXISOLS USING RED
GYPSUM AND BIOCHAR IN MALAYSIA**

By

TALHA IBRAHIM ZANNAH

**Thesis Submitted to the School of Graduate Studies, Universiti Putra
Malaysia, in Fulfillment of the Requirements for the Degree of
Doctor of Philosophy**

January 2017

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photograph and all other artwork, is a copyright material of the Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



DEDICATION

This thesis is dedicated to my parents and family affiliates.



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the Degree of Doctor of Philosophy

REDUCING SOIL ACIDITY IN ULTISOLS AND OXISOLS USING RED GYPSUM AND BIOCHAR IN MALAYSIA

By

TALHA IBRAHIM ZANNAH

January 2017

Chairman : Professor Shamshuddin Jusop, PhD
Faculty : Agriculture

The accruing benefits from RG, rich in Ca and S, produced in over 400,000 tonnes, yearly in Malaysia, can impact positively on the overall growth of crops by providing a readily soluble form of calcium, which is a principal problem in the Malaysian tropical soils. Biochar has 4.49 million ha under oil palm plantation, this connotes huge amount produced yearly. The pursuits to boost food production, the maize grown on these soils which have been below optimal yield are unsatisfactory due to the acid infertility. Therefore, the underlying hypothesis was that red gypsum and/or biochar application can impact positively or negatively on soil chemical properties and productivity of maize crop. The studied soils are representative of Peninsular Malaysia Ultisols and Oxisols comprising four Series, two from each taxonomic class. A series of experiments were conducted in laboratory, greenhouse and in the field to determine i) the effects of sulphate adsorption on pH and the charge properties of Ultisols and Oxisols ii) Al and Mn toxicity in the soils using red gypsum in combination with biochar and iii) the effects of red gypsum application on the growth of maize. Sulphate adsorption capacities differ among soil studied due to variation in physical and chemical properties. Four t ha⁻¹ RG incorporation significantly ($P < 0.05$) increased sulphate adsorption from 22 to 456, 38 to 526, 28 to 474, 70 to 516 mg kg⁻¹ in Bungor, Kuala Brang, Segamat and Kuantan Series respectively, raising soil pH from 4.5 to 5.2 and 4.9 to 5.5 with significant reduction in the activities of Al and Mn in the soil solution from 9.3 to 1.0 and 10.1 to 2.0 μM in Segamat and Kuantan Series, lowering PZC by 0.6 digits in Kuantan Series. Five t ha⁻¹ biochar application significantly increased the sulphate adsorption from 22 to 138, 38 to 170, 28 to 358, 70 to 180 mg kg⁻¹, raising pH from 4.4 to 4.9, 3.9 to 4.5, 4.5 to 5.3 and 4.9 to 5.7 in Bungor, Kuala Brang, Segamat and Kuantan Series respectively,

lowering the PZC by 0.2 to 0.3 digits in Bungor, Kuala Brang and Segamat Series, with significantly decreased Al and Mn activities in all the soil studied. The 3 t ha⁻¹ RG + 2 t ha⁻¹ biochar incorporation significantly ($P < 0.05$) increased sulphate adsorption from 22 to 458, 38 to 494, 28 to 494, 70 to 542 mg kg⁻¹ in Bungor, Kuala Brang, Segamat and Kuantan Series respectively, significantly, raising pH from 3.9 to 4.6, 4.5 to 5.4 to 5.8 in Kuala Brang, Segamat and Kuantan Series due to significant reduction in Al activity, lowering the PZC by 0.6 digits in Kuala Brang and Segamat Series. Calcium levels increased significantly due to respective application of 4, 5 and 3 + 2 t ha⁻¹ RG, biochar and RG + biochar. Both RG and biochar playing a substantial role in reducing toxicities of Al and Mn. The mechanism for the adsorbed sulphates involves both specific and non-specific adsorption processes. Incorporation of red gypsum and/or biochar into the acid soils improved maize growth. The red gypsum treated soils had significantly increased the biomass of the maize crop, which ranged from 17 to 45 g plant⁻¹ in the greenhouse experiment, in Kuala Brang and Kuantan Series. This had contributed 66 and 78 % to the growth of maize crop. The RG and biochar had significant effects on maize growth (761 to 1309 g plot⁻¹) as an amendment under field condition on Bungor Series. In addition to the discrete soil characteristics in terms of nutrient supply, the biomass in the RG amended soils was significantly positively related to available P, exchangeable Ca, Mg and K but negatively related to soil exchangeable Al and Mn ($P < 0.05$). The reduction in acidity by red gypsum was due to increased exchangeable Ca contents.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia
sebagai memenuhi keperluan untuk Ijazah Doktor Falsafah

**MENGURANGKAN KEASIDAN PADA TANAH ULTISOL
DAN OXISOL MENGGUNAKAN GIPSUM MERAH DAN
BIOCHAR DI MALAYSIA**

Oleh

TALHA IBRAHIM ZANNAH

Januari 2017

Pengerusi : Profesor Shamshuddin Jusop, PhD
Fakulti : Pertanian

Gypsum merah merupakan sumber produk sampingan yang kaya dengan Ca dan S. Ia mampu memberikan impak positif terhadap pertumbuhan jagung dengan membekalkan Ca dalam bentuk tersedia diambil oleh tanaman; di mana ia merupakan masalah utama bagi tanah pertanian di kawasan tropika. Terdapat lebih daripada 400, 000 biochar di ladang-ladang kelapa sawit, menunjukkan bahawa jumlah yang besar dihasilkan pada setiap tahun. Usaha meningkatkan pengeluaran hasil makanan; tanaman jagung yang diusahakan di tanah-tanah bermasalah ini memberikan hasil di bawah paras optimum, disebabkan oleh keasidan tinggi dalam tanah. Maka, hipotesis yang dicadangkan adalah aplikasi gypsum merah dan/ atau biochar boleh memberi impak positif atau negatif terhadap sifat kimia tanah dan pengeluaran hasil jagung. Tanah yang dikaji mewakili Ultisols dan Oxisols di Semenanjung Malaysia dan terdiri daripada empat siri, dua daripada setiap kelas taksonomi. Satu siri eksperimen di makmal, rumah hijau dan lapangan telah dijalankan bagi menentukan i) kesan jerapan sulfat terhadap pH dan sifat cas Ultisols dan Oxisols ii) ketoksikan Al dan Mn dalam tanah selepas aplikasi gypsum merah dan/ atau biochar dan iii) kesan aplikasi gypsum ke atas pertumbuhan jagung. Keupayaan jerapan sulfat didapati berbeza pada kesemua tanah yang dikaji akibat daripada perbezaan sifat-sifat fizikal, kimia dan cas. Aplikasi 4 t ha⁻¹ gypsum ke dalam tanah meningkatkan jerapan sulfat secara bererti dengan nilai daripada 22 kepada 256, 38 kepada 536, 28 kepada 474 dan 70 kepada 516 mg kg⁻¹ pada tanah Siri Bungor, Siri Kuala Brang, Siri Segamat dan Siri Kuantan, masing-masing, meningkatkan nilai pH daripada 4.5 kepada 5.2 dan 4.9 kepada 5.5 dengan penurunan signifikan terhadap aktiviti Al dan Mn dalam larutan tanah; daripada 9.3 kepada 1.0 dan 10.1 kepada 2.0 µM dalam tanah Siri Segamat dan Siri Kuantan, selain kesan

penurunan kadar titik cas sifar sebanyak 0.6 digit pada tanah Siri Kuantan. Aplikasi 5 t ha⁻¹ biochar didapati pada jerapan sulfat dengan nilai 22 kepada 138, 38 kepada 170, 28 kepada 358 dan 70 kepada 180 mg kg⁻¹, dengan peningkatan nilai pH 4.4 kepada 4.9, 3.9 kepada 4.5, 4.5 kepada 5.3 dan 4.9 kepada 5.7 pada tanah Siri Bungor, Kuala Brang, Segamat dan Kuantan, masing-masing, dengan penurunan kadar titik cas sifar sebanyak 0.2 hingga 0.3 digit pada tanah Siri Bungor, Kuala Brang dan Segamat, selain penurunan signifikan pada aktiviti Al dan Mn dalam kesemua tanah yang dikaji. Aplikasi 3 t ha⁻¹ gypsum merah bersama 2 t ha⁻¹ biochar meningkatkan ($p < 0.05$) jerapan sulfat daripada 22 kepada 458, 38 kepada 494, 28 kepada 494 dan 70 kepada 542 mg kg⁻¹ dalam kesemua tanah yang dikaji, di samping meningkatkan pH daripada 3.9 kepada 4.6, 4.5 kepada 5.4 dalam tanah Siri Kuala Brang, Siri Segamat dan Siri Kuantan, serta merendahkan titik cas sifar sebanyak 0.6 digit dalam tanah Siri Kuala Brang dan Siri Segamat. Paras Ca meningkat secara signifikan selepas aplikasi 4, 5 dan 3 + 2 t ha⁻¹ RG, biochar dan RG + Biochar, masing-masing. Kedua-dua RG dan biochar memaminkan peranan penting dalam pengurangan kesan ketoksikan Al dan Mn dalam tanah. Mekanisma penjerapan sulfat melibatkan kedua-dua proses penjerapan spesifik dan tidak spesifik. Aplikasi RG dan / atau biochar ke dalam tanah berasid mampu meningkatkan pertumbuhan jagung. Tanah yang dirawat dengan RG memberi kesan signifikan terhadap berat jagung, iaitu antara 17 hingga 45 g pokok⁻¹ dalam eksperimen rumah hijau, pada tanah Siri Kuala Brang dan Kuantan. Ia menyumbang sebanyak 66 dan 78% terhadap pertumbuhan jagung. Aplikasi RG dan biochar memberi kesan signifikan terhadap pertumbuhan jagung (761 hingga 1309 g plot⁻¹) sebagai penambahbaik tanah di lapangan pada tanah Siri Bungor. Selain itu, berat jagung yang ditanam pada tanah yang dirawat dengan RG adalah berkait secara positif dengan kandungan P tersedia, Ca, Mg dan K bolehganti tetapi berkait negatif dengan Al dan Mn bolehganti dalam tanah ($P < 0.05$). Pengurangan keasidan adalah disebabkan peningkatan kandungan Ca bolehganti selepas aplikasi RG.

ACKNOWLEDGEMENTS

First and foremost, Praise is to ALLAH (SWT), the Most Gracious and the Most Merciful for granting me wisdom, sound health and prosperity to sail up to this level. I wish to extend my most sincere message of appreciation to my supervisory committee chairman, Prof Dr Shamshuddin Jusop for his proper supervision, encouragement, constructive criticisms and suggestions. Despite his busy schedule, he always found time to read through the manuscript and made useful comments. His intelligence, valuable guidance, encouragement and meticulous scrutiny of this work ensured the proper articulation of some of the ideas. This is in addition to his words of advice and reading materials made available to me.

I am deeply indebted to Prof Dr Che Fauziah Ishak, for her intelligence and wealth of academic contributions. In the same stratum, I am deeply obliged to Dr Roslan Ismail for his guidance, understanding and patience in taking his time to give a very good supervision to this work, for his special interest and love right from admission to the completion of this study.

I am obliged to acknowledge contributions of Mr Azali, Puan Rusnah and all the laboratory staff. I equally concede the contributions of all the PhD and MSc students in the Department of Land Management. I am also thankful to members of my family, especially my mother Hajja Ganyeri and all family members for their prayers, encouragement and support throughout the period of this study. I am also beholden to my wives Bintu Liman Kingimi and Fatima Abba Mani and my daughters (Amma and Afnan), for their prayers, love, patience, understanding and endurance they have exhibited throughout my study period. I sincerely appreciate the encouragement and concern of my friends and well-wishers from within and outside the academic community for their support and encouragement at various stages of this study. I am grateful to the brothers we were in the same struggle with in Malaysia, especially Dr Alagie Bah, Late Engr. S D Buba, Dr Hamisu Sadi Ali, Dr Abdullahi Iliyasu, Dr Konto Mohammed, Bobby Abba Nuhu (M. Sc Business Economics), Desmond A. Yakubu (M. Sc Business Economics), Zakaria Umar (M. Sc Mechanical Engineering) for the cooperation, encouragement and support throughout our stay in Malaysia.

I sincerely appreciate the financial support from the Education Trust Fund (ETF). In the same strain, my appreciation also extends to the entire staff and students of the Department of Agricultural Sciences and Technology, Ramat Polytechnic Maiduguri and those of UPM for their support and encouragement. Finally, I wish to appreciate the varying efforts of all those, whose names may not have come into view here, but who have contributed academically, morally, socially, economically and above all with prayers

without which, I may not have come this far. May the Almighty ALLAH bless and reward you all with His best of rewards, Ameen.



I certify that a Thesis Examination Committee has met on 16 January 2017 to conduct the final examination of Talha Ibrahim Zannah on his thesis entitled "Reducing Soil Acidity in Ultisols and Oxisols using Red Gypsum and Biochar in Malaysia" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

Members of the Thesis Examination Committee were as follows:

Mohamed Hanafi bin Musa, PhD

Professor
Faculty of Agriculture
Universiti Putra Malaysia
(Chairman)

Ahmad Husni bin Mohd Haniff, PhD

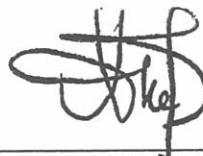
Associate Professor
Faculty of Agriculture
Universiti Putra Malaysia
(Internal Examiner)

Osumanu Haruna Ahmed, PhD

Associate Professor
Faculty of Agriculture and Food Sciences
Universiti Putra Malaysia (Bintulu Campus)
(Internal Examiner)

Zdenko Loncaric, PhD

Professor
University of Osijek
Croatia
(External Examiner)



NOR AINI AB. SHUKOR, PhD

Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

This thesis was submitted to the Senate of the Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Shamshuddin Jusop, PhD

Professor
Faculty of Agriculture
Universiti Putra Malaysia
(Chairman)

Che Fauziah Ishak, PhD

Professor
Faculty of Agriculture
Universiti Putra Malaysia
(Member)

Roslan Bin Ismail, PhD

Senior Lecturer
Faculty of Agriculture
Universiti Putra Malaysia
(Member)

ROBIAH BINTI YUNUS, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software

Signature:  _____ Date: _____

Name and Matric No : Talha Ibrahim Zannah, GS36618

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) were adhered to.

Signature: _____

Name of Chairman
of Supervisory
Committee:

Professor Dr. Shamshuddin Jusop

Signature: _____

Name of Member
of Supervisory
Committee:

Professor Dr. Che Fauziah Ishak

Signature: _____

Name of Member
of Supervisory
Committee:

Dr. Roslan Bin Ismail

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vii
DECLARATION	ix
LIST OF TABLES	xv
LIST OF FIGURES	xvii
LIST OF ABBREVIATIONS	xviii
CHAPTER	
1 INTRODUCTION	1
1.1 Background	1
1.2 Justification	2
1.3 Research Objectives	3
1.4 Organisation of the Thesis	3
2 LITERATURE REVIEW	4
2.1 Soil Acidification Process	4
2.2 Adverse Effects of Soil Acidity	5
2.3 Importance of Soil Acidity Studies	6
2.4 Ultisols and Oxisols	6
2.4.1 Physical, Chemical and Mineralogical Properties	7
2.5 Amendments of Soil Acidity	8
2.5.1 Amendments Effects	8
2.5.2 Red Gypsum	10
2.5.3 Biochar Effects	10
2.6 Sulphate Adsorption	11
2.6.1 Major Mechanisms of Sulphate sorptive in Tropical soils	11
2.6.2 Factors Affecting Sulphate Adsorption in Soil	14
2.7 Surface Charge Characteristics	17
2.7.1 Factors Affecting Surface Charge	18
2.7.2 Effects of Surface Charge on Soil Characteristics	20
2.7.3 Summary	22
3 GENERAL MATERIALS AND METHODS	24
3.1 Description of Location and Soil Samples	24
3.2 Soil Analysis	27
3.3 Plant Analysis	29

3.4	Statistical Analysis	30
4	CHARACTERIZATION OF HIGHLY WEATHERED ULTISOLS AND OXISOLS IN PENINSULAR MALAYSIA	31
4.1	Introduction	31
4.2	Materials and Methods	33
4.2.1	Description of Location and Soil Samples	33
4.2.2	Soil Sample Preparation	33
4.2.3	Soil Analysis	33
4.2.4	Fourier Transmission Infrared Analysis	33
4.2.5	X-ray Diffraction Analysis	34
4.3	Results and Discussion	34
4.3.1	FTIR Spectroscopy	40
4.3.2	XRD of the Clay Fraction	42
4.3.3	Conclusions	43
5	SULPHATE ADSORPTION CHARACTERISTICS OF ULTISOLS AND OXISOLS OF MALAYSIA	44
5.1	Introduction	44
5.2	Materials and Methods	45
5.2.1	Soil Sample Collection	45
5.2.2	Soil Analysis	45
5.2.3	Sulphate Adsorption Experiments	45
5.2.4	Soil Treatments and Analysis after Incubation	47
5.3	Results and Discussion	48
5.3.1	Soils Characteristics	48
5.3.2	Sulphate Adsorption Characteristics of the Soils Studied	50
5.3.3	Affinity of Soil Sulphate and their Colloidal Surfaces	54
5.3.4	Effects of Sulphate Adsorption and Related Soil Properties	56
5.3.5	Effects of Red Gypsum and/or Biochar on Sulphate Adsorption	64
5.3.6	PZC and Soil Solution pH on Charge Generation	67
5.3.7	Effects of PZC on Sulphate Adsorption	69
5.3.8	Conclusions	71
6	EFFECTS OF APPLYING RED GYPSUM AND/OR BIOCHAR ON MAIZE GROWTH IN GREEN HOUSE AND FIELD EXPERIMENT	72
6.1	Introduction	72
6.2	Materials and Methods	73

6.2.1	Soils, Red Gypsum, and Biochar Used in the study	73
6.2.2	Green House Experiment	74
6.2.3	Nitrogen Use Efficiency	75
6.2.4	Nitrogen Balances and Apparent Recovery	76
6.2.5	Effects Red Gypsum and/or Biochar to Biomass	76
6.2.6	Field Experiment	77
6.2.7	Experimental Site, Plot Description and Treatments	77
6.2.8	Soil Sampling and Chemical Analysis	78
6.2.9	Plant tissue analysis	78
6.2.10	Statistical Analyses	79
6.3	Results and Discussion	79
6.3.1	Characteristics of Ultisols and Oxisols Studied	79
6.3.2	Properties of Red Gypsum Used in Study	80
6.3.3	Properties of the Biochar Used in the Study	81
6.3.4	Effects of Red Gypsum and/or Biochar Application on Soil Solution Properties	81
6.3.5	Effects of Red Gypsum and/or Biochar Application on Soil Properties	85
6.3.6	Effects of Red Gypsum and/or Biochar Application on Nitrogen Use Efficiency	89
6.3.7	Effects of Red Gypsum and/or Biochar Application on Maize Biomass	91
6.3.8	Effects of Red Gypsum and/or Biochar Application on Maize Growth	93
6.3.9	Effect of Red Gypsum and/or Biochar Addition on Soil Chemical Characteristics under Field Condition on Bungor Series	95
6.3.10	Effect of Red Gypsum and/or Biochar Addition on Maize Biomass	99
6.3.11	Contributions of Red Gypsum and/or Biochar in the Soils to Maize Growth	100
6.3.12	Effect of Red Gypsum and/or Biochar Addition on N Use Efficiency	101
6.3.13	Conclusions	103
7	GENERAL DISCUSSION, SUMMARY, CONCLUSION AND RECOMMENDATION	104
7.1	General Discussion	104
7.2	Summary and Conclusion	105
7.3	Recommendation for Future Research	106

REFERENCES	107
APPENDICES	131
BIODATA OF STUDENT	132
LIST OF PUBLICATIONS	133



© COPYRIGHT UPM

LIST OF TABLES

Table		Page
3.1	Characteristics Of Sampling Sites	26
4.1	Selected Soil Physical Parameters And Taxonomic Classification	36
4.2	Chemical Characteristics Of Ultisols And Oxisols In Peninsular Malaysia	39
5.1	Selected Physical And Chemical Characteristics Of The Studied Soils	49
5.2	Experimental Data Fitting To The Langmuir And Freundlich Equations Of Sulphate Adsorption	52
5.3	Correlations Between Langmuir Sulphate Adsorption And Soil Characteristics (N=12)	57
5.4	Correlations Between Freundlich Sulphate Adsorption And Soil Characteristics (N=12)	58
5.5	Selected Charge And Chemical Characteristics	65
5.6	Correlations Coefficients Sulphate Adsorbed And Some Soil Chemical Properties	70
6.1	Chemical Properties Of The Red Gypsum Used In The Study	80
6.2	Chemical Properties Of The Biochar Used In The Study	81
6.3	Soil Solutions Chemical Properties Extracted From The Treatments During The Green House Experiment	83
6.4	Effects Of Red Gypsum And/Or Biochar Application On Some Soil Chemical Properties	87
6.5	N Use Efficiency, Apparent N Surplus And Apparent N Recovery In The Soil With Red Gypsum And/Or Biochar Amendment	90

6.6	Correlations Coefficients Of Nue, Ans And Anr With Some Soil Chemical Properties	90
6.7	Correlation Coefficients Between Maize Biomass And Soil Solution Properties	93
6.8	Effect Of Red Gypsum And/Or Biochar To Crop Growth In The Green House Experiment	94
6.9	Changes In Soil Chemical Rank As Amended With Red Gypsum And/Or Biochar	96
6.10	Correlation Coefficients Between Maize Biomass And Soil Chemical Properties	100
6.11	Contributions Of Red Gypsum And/Or Biochar In The Soils Each To Crop Growth	100
6.12	Effects of Red Gypsum and/or Biocharon N Use Efficiency, Apparent N Surplus and Apparent N Recovery	102
6.13	Correlations Coefficients of N Use Efficiency, Apparent Nutrient Surplus and Apparent Nutrient Recovery with Some Soil Properties	102

LIST OF FIGURES

Figure		Page
3.1	Location of Selected Sampling Sites	25
4.1	Means of Chemical Characteristics of Ultisols and Oxisols in Peninsular Malaysia	38
4.2	FTIR Spectra of Topsoil (0 to 15cm) of Bungor (a), Kuala Brang (b), Kuantan (c), and Segamat (d) Series	41
4.3	XRD Diffractogram From the Top Soil of Bungor (a), Kuala Brang (b), Kuantan (c), and Segamat (d) Series	43
5.1	Sulphate Adsorbed and S Concentrations in Soil Solution	50
5.2	Sulphate Adsorption Maximum and Free Fe_2O_3	59
5.3	Sulphate Adsorption Maximum and Free Al_2O_3	60
5.4	Point Zero Charge of the Selected Acid Soils	61
5.5	Effects of Red Gypsum and/or Biochar on pH and PZC	69
6.1	Experimental Design/Layout	77
6.2	XRD Diffractogram of the Red Gypsum Used in the Study	80
6.3	Relationships between Soil pH and Exchangeable Al (a) and Extractable Mn (b) Concentrations	86
6.4	Effects of Treatments on Maize Biomass	92
6.5	Effects of Treatments on Maize Biomass	99

LIST OF ABBREVIATIONS

MBC	Maximum Buffering Capacity
K	Bonding energy or affinity constant
B	Sulphate sorption maxima
PZC or pH ₀	Point zero charge
CEC	Cation Exchange Capacity
AEC	Anion Exchange Capacity
CEC _B	Basic Cation Exchange Capacity
CEC _T	Total Cation Exchange Capacity
RG	Red Gypsum
EFB	Empty Fruit Bunch
ANC	Acid Neutralizing Capacity
DDL	Diffuse Double Layer
FTIR	Fourier Transmission Infrared
XRD	X-Ray Diffraction
ICP-OES -	Inductively coupled plasma emission spectrometry
AAS	Atomic Absorption Spectrophotometer
AA	Auto Analyser
CNS	Carbon, Nitrogen and Sulphur
C:N	Carbon Nitrogen
OH	hydroxyl groups
C-C, C=O, -COO	Carboxylate groups, carbon and oxygen atoms
Si-O	Silicon and Oxygen.

CHAPTER 1

INTRODUCTION

1.1 Background

Acid Ultisols and Oxisols which occur during geological evolution, especially in areas of high rainfall due to relatively easy leaching from soils, leaving them acidic (Rengel, 2003), is one of the major causes of soil degradation. It is essential decline in soil quality or reduction in its productivity and environmental regulatory capacity (Lal, 1997), that limit crop production. It is a natural process, which can be accelerated by the activity of the living organisms or can be impeded by sound management practices. Developments of acid soils continue to increase in the tropics and worldwide, primarily because of continuous leaching by heavy rains and agricultural practices. This has become a constraint and one of the most yield-limiting factors for crop production. This trend poses a high risk of soil and environmental degradation because of harsh climate and resource-poor farmers (Lal and Stewart, 2011).

The tropical regions which cover significant portion of the world's land acreage are characterized by a large portion of the world's rapidly increasing population (Lal, 2002), projected to have a huge increase in food demand for cereals. World population is expected to increase to 8 billion by 2025 and over 9 billion by 2050 (Fageria and Baligar, 2008). It is estimated that global cereal demand (rice, wheat, and maize) will increase from 1657 million Mg in 1995 to 2436 million Mg in 2025 at the mean rate of 1.29% per year (Cassman et al., 2003). Despite the gains in global mean average crop yields, grain yields of cereals in developing countries are still low (Lal, 2007). The acid tropical soils cover a significant part of at least 48 developing countries (Narro et al., 2001), being more frequent in Ultisols and Oxisols in Malaysia.

In view of the foregoing, the accruing benefits from waste generated in the majority of industrial production processes recycling with special relations to soil properties cannot be neglected, as these waste materials become a subject of research in different field of endeavours in recent years. A growing number of soil scientists and environmentalists are becoming interestingly alarmed over the importance of waste material recycling in resolving issues related to soil and environmental degeneration. These are of great concern for both economy as well as environmental reasons/or benefits.

Red gypsum (RG) and the empty fruit bunch (EFB) of palm oil which are typically by-products of industry and farming respectively, commonly called wastes because they are not the primary product. Such wastes are a resource that can be utilized and not just discarded. The possible dividend of such wastes include use as soil amendment/or fertilizer, thus improving soil physical, chemical and biological characteristics. Therefore, the underlying hypothesis is that, red gypsum and/or biochar application can impact on positively or negatively on soil chemical properties and hence improved productivity. Energy production from varying sources such as crop residues, and/or other organic wastes has been utilized in agriculture to varying degrees in different parts of the world (Westerman and Bicudo, 2005). The utilization of a specific waste in agriculture depends among several factors, including the characteristics of the waste such as nutrient content, availability and transportation costs, benefits to agriculture and regulatory considerations.

1.2 Justification

Red gypsum is rich in Ca, S and Fe contents (Fauziah et al., 1996; 2011; Rodriguez-Jorda et al., 2010) which can be harnessed for enhancing soil fertility. Over one million tonnes of red gypsum is produced each year (Azdarpour et al., 2014). Its total accumulation in Malaysia is about 400,000 tonnes per year (Kamarudin and Zakaria, 2007). Red gypsum is 75% gypsum (Azdarpour et al., 2014) which is a calcium rich source. Therefore, proposed as a new feedstock for mineral CO₂ sequestration (Rahmani et al., 2014).

This can impact positively on overall growth of maize plants by providing a readily soluble form of calcium which is a principal problem to the Malaysian tropical soils. And biochar being carbonaceous recalcitrant product can be used as a tool to modify or improve soil productivity. In addition to its carbon sequestration potential, biochar has been shown to improve soil quality and crop yields (Brunn, 2011; Abdulrazzaq et al., 2014). The great carbon sequestration potential of biochar, together with biochar's apparent ability to improve soil quality, has become a highly interesting concept in recent time. The pursuits to boost food production, the maize grown on these soils which have been below optimal yield are unsatisfactory due to the acid infertility. The maize crop has been used as test crop because it is acid sensitive affecting its growth, principally due to toxicities of Al and Mn which led to nutrient deficiencies of Ca, Mg and K and thus requires great attention. This necessitates exploration and efficient management of waste from surrounding environment with a view to have effective soil, plant and water relations for sustainable agriculture.

Numerous strategies have been pursued to manage these acid soils. Adequately limed soils have been proven positive as it enhances sustainability of the cropping systems. However, there has been increased interest on alternative multiple liming agents due to multiple benefits. But it appears that, results around the world toward addressing problems of this nature or related to soil acidity heavily rely on locally sourced materials. As this acknowledges or addresses the problems of cost and practicability to the target beneficiaries and hence the dearth of information on the effects of red gypsum in combination with biochar application in reducing soil acidity in Ultisols and Oxisols necessitate the present investigation. In addition, adsorption and exchange effects of the anions such as sulphate in soils are significantly less understood as compared to cations adsorption and exchange.

1.3 Research Objectives

The general objective of this study was to understand the effects of red gypsum and/or biochar application in reducing soil acidity, and the consequential soil properties on chemical and charge characteristics. The specific objectives are as follows:

- 1) To determine the effects of sulphate adsorption on pH and the charge properties of Ultisols and Oxisols;
- 2) To alleviate Al and Mn toxicity in the soils using red gypsum in combination with biochar; and
- 3) To determine the effects of gypsum application on the growth of maize.

1.4 Organisation of the Thesis

The thesis has been structured into seven chapters. Chapter one finds background and justification, research objectives and organisation of the thesis. Chapter two reviewed related literature. Chapter three describes general materials and methods used. Chapter four presents characterization of soils under study. Chapter five identifies sulphate adsorption characteristics of the selected soils and its effects on soil chemical properties after red gypsum and/or biochar application. Chapter six determines effects of applying red gypsum and/or biochar on maize growth after greenhouse and field studies. Chapter seven provides a general discussion, summary of the major findings. It also concludes the study and articulates recommendations and likely future research.

REFERENCES

- Abdulrazzaq, H., Jol, H., Husni, A., & Abu-Bakr, R. 2014. Biochar from Empty Fruit Bunches, Wood, and Rice Husks: Effects on Soil Physical Properties and Growth of Sweet Corn on Acidic Soil. *Journal of Agricultural Science*, 7(1), p192.
- Adams, F. 1984. Crop response to lime in the southern United States. *Soil acidity and liming*, (soilacidityandl), 211-265.
- Adams, M. B., DeWalle, D. R., Peterjohn, W. T., Gilliam, F. S., Sharpe, W. E., & Williard, K. W. 2006. Soil chemical response to experimental acidification treatments. In *The Fernow Watershed Acidification Study* (pp. 41-69). Springer Netherlands.
- Adcock, K. G., Gartrell, J. W., & Brennan, R. F. 2001. Calcium deficiency of wheat grown in acidic sandy soil from Southwestern Australia. *Journal of plant nutrition*, 24(8), 1217-1227.
- Adnan, A., Mavinic, D. S., & Koch, F. A. 2003. Pilot-scale study of phosphorus recovery through struvite crystallization examining the process feasibility. *Journal of Environmental Engineering and Science*, 2(5), 315-324.
- Adriano, D. C. 2001. Bioavailability of trace metals. In *Trace Elements in Terrestrial Environments* (pp. 61-89). Springer New York.
- Adriano, D. C., Wenzel, W. W., Vangronsveld, J., & Bolan, N. S. 2004. Role of assisted natural remediation in environmental cleanup. *Geoderma*, 122(2), 121-142.
- Aguiar, N. O., Novotny, E. H., Oliveira, A. L., Rumjanek, V. M., Olivares, F. L., & Canellas, L. P. 2013. Prediction of humic acids bioactivity using spectroscopy and multivariate analysis. *Journal of Geochemical Exploration*, 129, 95-102.
- Ahmad, M., Rajapaksha, A. U., Lim, J. E., Zhang, M., Bolan, N., Mohan, D., & Ok, Y. S. 2014. Biochar as a sorbent for contaminant management in soil and water: a review. *Chemosphere*, 99, 19-33.
- Alamgir, M. 2016. The Effects of Soil Properties to the Extent of Soil Contamination with Metals. In *Environmental Remediation Technologies for Metal-Contaminated Soils* (pp. 1-19). Springer Japan.

- Alcordero, I. S., & Rechcigl, J. E. 1993. Phosphogypsum in agriculture: a review. *Advances in Agronomy*, 49, 55-118.
- Alewel, C. 2001. Predicting reversibility of acidification: the European sulfur story. *Water, Air, and Soil Pollution*, 130(1-4), 1271-1276.
- Alloway, B. J. 2013. Bioavailability of elements in soil. In *Essentials of medical geology* (pp. 351-373). Springer Netherlands.
- Almaroai, Y. A., Usman, A. R., Ahmad, M., Moon, D. H., Cho, J. S., Joo, Y. K., & Ok, Y. S. 2014. Effects of biochar, cow bone, and eggshell on Pb availability to maize in contaminated soil irrigated with saline water. *Environmental earth sciences*, 71(3), 1289-1296.
- Alva, A. K., Sumner, M. E., & Miller, W. P. 1990. Reactions of gypsum or phosphogypsum in highly weathered acid subsoils. *Soil Science Society of America Journal*, 54(4), 993-998.
- Alves, M. E., & Lavoretti, A. 2004. Sulphate adsorption and its relationships with properties of representative soils of the Sao Paulo State, Brazil. *Geoderma*, 118(1), 89-99.
- Amonett, J. E., Dixon, J. B., & Schulze, D. G. 2002. Soil Mineralogy with Environmental Applications. *Soil Science Society of America, Madison, WI*.
- Anda, M., Shamshuddin, J., & Fauziah, C. I. 2013. Increasing negative charge and nutrient contents of a highly weathered soil using basalt and rice husk to promote cocoa growth under field conditions. *Soil and Tillage Research*, 132, 1-11.
- Appel, C., & Ma, L. 2002. Concentration, pH, and surface charge effects on cadmium and lead sorption in three tropical soils. *Journal of Environmental Quality*, 31(2), 581-589.
- Arbestain, M. C., Barreal, M. E., & Macías, F. 2001. Sulphate sorption in nonvolcanic Andisols and andic soils from Galicia, NW Spain. *Geoderma*, 104(1), 75-93.
- Ashman, M., & Puri, G. 2013. *Essential soil science: a clear and concise introduction to soil science*. John Wiley & Sons.
- Atwell, B. J., Kriedemann, P. E., & Turnbull, C. G. 1999. *Plants in action: adaptation in nature, performance in cultivation*. Macmillan Education AU.

- Azdarpour, A., Asadullah, M., Junin, R., Manan, M., Hamidi, H., & Mohammadian, E. 2014. Direct carbonation of red gypsum to produce solid carbonates. *Fuel Processing Technology*, 126, 429-434.
- Barber, S. A. 1995. *Soil nutrient bioavailability: a mechanistic approach*. John Wiley & Sons.
- Bhatti, J. S., Foster, N. W., & Evans, L. J. 1997. Sulphate sorption in relation to properties of podzolic and brunisolic soils in northeastern Ontario. *Canadian journal of soil science*, 77(3), 397-404.
- Bhogal, N. S., Choudhary, K. C., & Sakal, R. 1996. Sulphur adsorption by calciorthents. *Journal of the Indian Society of Soil Science*, 44(2), 326-328.
- Biswas, T. D., & Mukherjee, S. K. 2001. *Textbook of Soil Sciences*. Tata McGraw-Hill Education.
- Blackwell, P., Riethmuller, G., & Collins, M. 2009. Biochar application to soil. *Biochar for environmental management: science and technology*, 1, 207-226.
- Bohn, H. L., McNeal, B. L., & O'Connor, G. A. 1985. *Soil Chemistry*. A Wiley-Interscience Publication.
- Bohn, H. L., Myer, R. A., & O'Connor, G. A. 2002. *Soil chemistry*. John Wiley & Sons.
- Bolan, N. S., Adriano, D. C., & Curtin, D. 2003. Soil acidification and liming interactions with nutrient and heavy metal transformation and bioavailability. *Advances in Agronomy*, 78, 215-272.
- Bolan, N. S., Naidu, R., Syers, J. K., & Tillman, R. W. 1999. Surface charge and solute interactions in soils. *Advances in agronomy*, 67, 87-140.
- Bolan, N. S., Scotter, D. R., Syers, J. K., & Tillman, R. W. 1986. The effect of adsorption on sulphate leaching. *Soil Science Society of America Journal*, 50(6), 1419-1424.
- Bolan, N. S., Syers, J. K., & Sumner, M. E. 1993. Calcium-induced sulphate adsorption by soils. *Soil Science Society of America Journal*, 57(3), 691-696.
- Bolan, N. S., Syers, J. K., & Sumner, M. E. 1993. Calcium-induced sulphate adsorption by soils. *Soil Science Society of America Journal*, 57(3), 691-696.

- Bolan, N. S., Syers, J. K., Tillman, R. W., & Scotter, D. R. 1988. Effect of liming and phosphate additions on sulphate leaching in soils. *Journal of soil science*, 39(4), 493-504.
- Bolan, N., Kunhikrishnan, A., Thangarajan, R., Kumpiene, J., Park, J., Makino, T., & Scheckel, K. 2014. Remediation of heavy metal (loid) s contaminated soils–to mobilize or to immobilize?. *Journal of hazardous materials*, 266, 141-166.
- Bradl, H. B. 2004. Adsorption of heavy metal ions on soils and soils constituents. *Journal of Colloid and Interface Science*, 277(1), 1-18.
- Bray, R. H., & Kurtz, L. T. 1945. Determination of total, organic, and available forms of phosphorus in soils. *Soil science*, 59(1), 39-46.
- Bremner, J. M. 1960. Determination of nitrogen in soil by the Kjeldahl method. *The Journal of Agricultural Science*, 55(01), 11-33.
- Brown, G., & Brindley, G. W. (Eds.). 1980. *Crystal structures of clay minerals and their X-ray identification* (Vol. 5, pp. 305-360). London: Mineralogical Society.
- Bruun, E. W., Hauggaard-Nielsen, H., Ibrahim, N., Egsgaard, H., Ambus, P., Jensen, P. A., & Dam-Johansen, K. 2011. Influence of fast pyrolysis temperature on biochar labile fraction and short-term carbon loss in a loamy soil. *Biomass and Bioenergy*, 35(3), 1182-1189.
- Bujdák, J. (2006). Effect of the layer charge of clay minerals on optical properties of organic dyes. A review. *Applied Clay Science*, 34(1), 58-73.
- Butnan, S., Deenik, J. L., Toomsan, B., Antal, M. J., & Vityakon, P. 2015. Biochar characteristics and application rates affecting corn growth and properties of soils contrasting in texture and mineralogy. *Geoderma*, 237, 105-116.
- Caires, E. F., Churka, S., Garbuio, F. J., Ferrari, R. A., & Morgano, M. A. 2006. Soybean yield and quality a function of lime and gypsum applications. *Scientia Agricola*, 63(4), 370-379.
- Campbell, C. G., Garrido, F., Illera, V., & García-González, M. T. 2006. Transport of Cd, Cu and Pb in an acid soil amended with phosphogypsum, sugar foam and phosphoric rock. *Applied geochemistry*, 21(6), 1030-1043.

- Campillo, R., Urquiaga, S., Undurraga, P., Pino, I., & Boddey, R. M. 2005. Strategies to optimise biological nitrogen fixation in legume/grass pastures in the southern region of Chile. *Plant and soil*, 273(1-2), 57-67.
- Carter, S., Shackley, S., Sohi, S., Suy, T. B., & Haefele, S. 2013. The impact of biochar application on soil properties and plant growth of pot grown lettuce (*Lactuca sativa*) and cabbage (*Brassica chinensis*). *Agronomy*, 3(2), 404-418.
- Cassman, K. G., Dobermann, A., Walters, D. T., & Yang, H. 2003. Meeting cereal demand while protecting natural resources and improving environmental quality. *Annual Review of Environment and Resources*, 28(1), 315-358.
- Chan, K. Y., & Xu, Z. 2009. Biochar: nutrient properties and their enhancement. *Biochar for environmental management: science and technology*, 67-84.
- Chan, K. Y., Van Zwieten, L., Meszaros, I., Downie, A., & Joseph, S. 2008. Agronomic values of greenwaste biochar as a soil amendment. *Soil Research*, 45(8), 629-634.
- Chen, L., Ramsier, C., Bigham, J., Slater, B., Kost, D., Lee, Y. B., & Dick, W. A. 2009. Oxidation of FGD-CaSO₃ and effect on soil chemical properties when applied to the soil surface. *Fuel*, 88(7), 1167-1172.
- Chepkwony, C. K., Haynes, R. J., Swift, R. S., & Harrison, R. 2001. Mineralization of soil organic P induced by drying and rewetting as a source of plant-available P in limed and unlimed samples of an acid soil. *Plant and Soil*, 234(1), 83-90.
- Chitrakar, R., Tezuka, S., Sonoda, A., Sakane, K., Ooi, K., & Hirotsu, T. 2006. Phosphate adsorption on synthetic goethite and akaganeite. *Journal of Colloid and Interface Science*, 298(2), 602-608.
- Chwil, S. 2002. Changes of basic indices of acidification in soil profile as affected by fertilization. *Zeszyty problemowe postepów nauk rolniczych*, (482), 79-85.
- Cichota, R., Vogeler, I., Bolan, N. S., & Clothier, B. E. 2007. Cation influence on sulfate leaching in allophanic soils. *Soil Research*, 45(1), 49-54.
- Clark, R. B., & Baligar, V. C. 1995. Acidic and alkaline soil constrains on plant mineral. *Plant Environment Interactions (2nd ed.)*, Marcel Dekker, Inc., New York, 133-177.

- Clough, T. J., Condrón, L. M., Kammann, C., & Müller, C. (2013). A review of biochar and soil nitrogen dynamics. *Agronomy*, 3(2), 275-293.
- Connor, D. J., Loomis, R. S., & Cassman, K. G. 2011. *Crop ecology: productivity and management in agricultural systems*. Cambridge University Press.
- Cottenie, A., G. Velghe, M. Verloo, and L. Kiekens. "Analytical methods for plants and soils." (1979).
- Courchesne, F., & Hendershot, W. H. 1990. The role of basic aluminium sulphate minerals in controlling sulphate retention in the mineral horizons of two spodosols. *Soil science*, 150(3), 571-578.
- Coventry, D. R., Slattery, W. J., Burnett, V. F., & Ganning, G. W. 1997. Longevity of wheat yield response to lime in south-eastern Australia. *Animal Production Science*, 37(5), 571-575.
- Curtin, D., & Syers, J. K. 1990. Extractability and adsorption of sulphate in soils. *Journal of soil science*, 41(2), 305-312.
- Dam-ampai, S., Nilnond, C., & Onthong, J. (2007). Effect of some soil amendments on soil properties and plant growth in southern Thailand acid upland soil. *Songklanakarin Journal of Science and Technology (Thailand)*.
- De Castro, O. M., De Camargo, O. A., Vieira, S. R., & Filho, J. V. 1999. Effect of two types of lime on some soil physical attributes of an oxisol from Brazil. *Communications in Soil Science & Plant Analysis*, 30(15-16), 2183-2195.
- Delhaize, E., & Ryan, P. R. 1995. Aluminium toxicity and tolerance in plants. *Plant physiology*, 107(2), 315.
- Dijkstra, J. J., Meeussen, J. C., & Comans, R. N. 2004. Leaching of heavy metals from contaminated soils: an experimental and modeling study. *Environmental science & technology*, 38(16), 4390-4395.
- Dixon, J. B., & Schulze, D. G. 2002. *Soil mineralogy with environmental applications*. Soil Science Society of America Inc.
- Du, C., Linker, R., & Shaviv, A. 2008. Identification of agricultural Mediterranean soils using mid-infrared photoacoustic spectroscopy. *Geoderma*, 143(1), 85-90.

- Du, C., Zhou, J., Wang, H., Chen, X., Zhu, A., & Zhang, J. 2009. Determination of soil properties using Fourier transform mid-infrared photoacoustic spectroscopy. *Vibrational Spectroscopy*, 49(1), 32-37.
- Edwards, P. J. 1998. Sulphur cycling, retention, and mobility in soils: a review.
- Eidukeviciene, M. J., Ozheraitiene, D. J., Tripolskaja, L. N., & Marcinkonis, S. I. 2001. The effect of long-term liming on the chemical properties of Lithuanian soils. *Eurasian soil science*, 34(9), 999-1005.
- Essington, M. E. 2015. *Soil and water chemistry: An integrative approach*. CRC press.
- Evangelou, V. P. 1995. *Pyrite oxidation and its control*. CRC press.
- Evans Jr, A. 1991. The interaction of aliphatic acids with basic aluminium sulphates in a forested Ultisols. *Soil Science*, 152(1), 53-60.
- Evans Jr, A., & Anderson, T. J. 1990. Influence of aliphatic acids on sulphate mobility in a forest soil. *Soil Sci. Soc. Am. J*, 54(4), 1136-1139.
- Fageria, N. K., & Baligar, V. C. 2008. Ameliorating soil acidity of tropical Oxisols by liming for sustainable crop production. *Advances in agronomy*, 99, 345-399.
- Fageria, N. K., Baligar, V. C., & Jones, C. A. 2010. *Growth and mineral nutrition of field crops*. CRC Press.
- Farina, M. P. W., & Channon, P. 1988. Acid-subsoil amelioration: II. Gypsum effects on growth and subsoil chemical properties. *Soil Science Society of America Journal*, 52(1), 175-180.
- Fauziah, C. I., Hanani, M. N., Zauyah, S., Samsuri, A. W., & Rosazlin, A. 2011. Co-application of Red Gypsum and Sewage Sludge on Acidic Tropical Soils. *Communications in soil science and plant analysis*, 42(21), 2561-2571.
- Fauziah, I., Zauyah, S., & Jamal, T. (1996). Characterization and land application of red gypsum: a waste product from the titanium dioxide industry. *Science of the total environment*, 188(2), 243-251.
- Fernández-Getino, A. P., Hernandez, Z., Piedra Buena, A., & Almendros, G. 2013. Exploratory analysis of the structural variability of forest soil humic acids based on multivariate processing of infrared spectral data. *European Journal of Soil Science*, 64(1), 66-79.

- Fox, T. R., Comerford, N. B., & McFee, W. W. 1990. Phosphorus and aluminium release from a spodic horizon mediated by organic acids. *Soil Science Society of America Journal*, 54(6), 1763-1767.
- Galli, U., Schüepp, H., & Brunold, C. 1994. Heavy metal binding by mycorrhizal fungi. *Physiologia Plantarum*, 92(2), 364-368.
- Ganor, J., Reznik, I. J., & Rosenberg, Y. O. 2009. Organics in water-rock interactions. *Reviews in Mineralogy and Geochemistry*, 70(1), 259-369.
- García-Montero, L. G., Quintana, A., Valverde-Asenjo, I., & Díaz, P. 2009. Calcareous amendments in truffle culture: a soil nutrition hypothesis. *Soil Biology and Biochemistry*, 41(6), 1227-1232.
- García-Rodeja, E., Nóvoa, J. C., Pontevedra, X., Martínez-Cortizas, A., & Buurman, P. 2007. Aluminium and iron fractionation of European volcanic soils by selective dissolution techniques. In *Soils of Volcanic Regions in Europe* (pp. 325-351). Springer Berlin Heidelberg.
- Garrido, F., Illera, V., Vizcayno, C., & García-González, M. T. 2003. Evaluation of industrial by-products as soil acidity amendments: chemical and mineralogical implications. *European Journal of Soil Science*, 54(2), 411-422.
- Gazey, C., Andrew, J., & Griffin, T. 2013. Soil acidity. *Report card on sustainable natural resource use in agriculture – status and trend in the agricultural areas of the southwest of Western Australia*. (Department of Agriculture and Food, Western Australia: South Perth, W. Aust.).
- Gazey, C., Andrew, J., & Griffin, T. 2013. Soil acidity. *Report card on sustainable natural resource use in agriculture – status and trend in the agricultural areas of the south-west of Western Australia*. (Department of Agriculture and Food, Western Australia: South Perth, W. Aust.).
- Gazquez M.J., J.P. Bolivar, F. Vaca, R. García-Tenorio, & A. Caparros 2013. Evaluation of the use of TiO₂ industry red gypsum waste in cement production. *Cem. Concr. Compos.* 37: 76-81.
- Gencoglan, C. 1996. *Determining the water production function, root distribution and crop water stress index of maize and validation of CERES maize crop growth model under Cukurova* (Doctoral dissertation, Ph. D. Thesis, Cukurova University, Adana, Turkey, 220 p.
- Ghosal, P., Chakraborty, T., & Banik, P. 2011. Phosphorus fixing capacity of the Oxic Rhodustalf – alfisol soil in the Chotanagpur plateau region of Eastern India. *Agricultural Sciences*, 2(04), 487.

- Ghosha, G. K., & Dashb, N. R. 2012. Sulphate Sorption–Desorption Characteristics of Lateritic Soils of West Bengal, India. *International Journal of Plant, Animal and Environmental Sciences*, 2(1), 167-170.
- Gillman, G. P. 1984. Using variable charge characteristics to understand the exchangeable cation status of oxic soils. *Soil Research*, 22(1), 71-80.
- Gillman, G. P. 2007. An analytical tool for understanding the properties and behaviour of variable charge soils. *Soil Research*, 45(2), 83-90.
- Glaser, B., Lehmann, J., & Zech, W. 2002. Ameliorating physical and chemical properties of highly weathered soils in the tropics with charcoal—a review. *Biology and fertility of soils*, 35(4), 219-230.
- Gobran, G. R., Selim, H. M., Hultberg, H., & Andersson, I. 1998. Sulphate adsorption-desorption in a Swedish forest soil. *Water, Air, and Soil Pollution*, 108(3-4), 411-424.
- Gundersen, P., & Beier, C. 1988. Aluminium sulphate solubility in acid forest soils in Denmark. *Water, Air, and Soil Pollution*, 39(3-4), 247-261.
- Harris, R. G., Johnson, B. B., & Wells, J. D. 2006. Studies on the adsorption of dyes to kaolinite. *Clays and clay minerals*, 54(4), 435-448.
- Harris, W. I. L. L. I. E., & White, G. N. 2008. X-ray diffraction techniques for soil mineral identification. *Methods of soil analysis: Part, 5*, 81-115.
- Harrison, R. B., Johnson, D. W., & Todd, D. E. 1989. Sulphate adsorption and desorption reversibility in a variety of forest soils. *Journal of Environmental Quality*, 18(4), 419-426.
- Harter, R. D. 2007. Acid soils of the tropics. *Echo technical note*.
- Havlin, J. L., Beaton, J. D., Nelson, W. L., & Tisdale, S. L. (2005). *Soil fertility and fertilizers: An introduction to nutrient management* (Vol. 515). Upper Saddle River, NJ: Pearson Prentice Hall.
- Hayes, M. H. B. 2013. Some Properties of Clays and of. *Soil Colloids and Their Associations in Aggregates*, 214, 39.
- Haynes, R. J., & Mokolobate, M. S. 2001. Amelioration of Al toxicity and P deficiency in acid soils by additions of organic residues: a critical review of the phenomenon and the mechanisms involved. *Nutrient cycling in agroecosystems*, 59(1), 47-63.

- He, L. M., Zelazny, L. W., Martens, D. C., Baligar, V. C., & Ritchey, K. D. 1997. Ionic strength effects on sulfate and phosphate adsorption on γ -alumina and kaolinite: Triple-layer model. *Soil Science Society of America Journal*, 61(3), 784-793.
- He, Z. L., Yang, X. E., Baligar, V. C., & Calvert, D. V. 2003. Microbiological and biochemical indexing systems for assessing quality of acid soils. *Advances in Agronomy*, 78, 89-138.
- Helyar, K. R. 1991. The management of acid soils. In *Plant-soil interactions at low pH* (pp. 365-382). Springer Netherlands.
- Ho, Y. S., Huang, C. T., & Huang, H. W. (2002). Equilibrium sorption isotherm for metal ions on tree fern. *Process Biochemistry*, 37(12), 1421-1430.
- Horst, W. J., Puschel, A. K., & Schmohl, N. 1997. Induction of callose formation is a sensitive marker for genotypic aluminium sensitivity in maize. *Plant and Soil*, 192(1), 23-30.
- Howell, T. A., Tolk, J. A., Schneider, A. D., & Evett, S. R. 1998. Evapotranspiration, yield, and water use efficiency of corn hybrids differing in maturity. *Agronomy Journal*, 90(1), 3-9.
- Huang, P. M., & Violante, A. 1986. Influence of organic acids on crystallization and surface properties of precipitation products of aluminium. *Interactions of soil minerals with natural organics and microbes, (interactionsofs)*, 159-221.
- Hue, N. V. 2005. Responses of coffee seedlings to calcium and zinc amendments to two Hawaiian acid soils. *Journal of plant nutrition*, 27(2), 261-274.
- Hue, N. V., Adams, F., & Evans, C. E. 1985. Sulphate retention by an acid BE horizon of an Ultisol. *Soil Science Society of America Journal*, 49(5), 1196-1200.
- Huete, A. R., & Mc Coll, J. G. 1984. Soil cation leaching by "acid rain" with varying nitrate-to-sulfate ratios. *Journal of Environmental Quality*, 13(3), 366-371.
- Hutchison, C. S., & Tan, D. N. K. (Eds.). 2009. *Geology of Peninsular Malaysia*. University of Malaya.

- Illera, V., Garrido, F., Vizcayno, C., & García-González, M. T. 2004. Field application of industrial by-products as Al toxicity amendments: chemical and mineralogical implications. *European journal of soil science*, 55(4), 681-692.
- Illés, E., & Tombácz, E. (2006). The effect of humic acid adsorption on pH-dependent surface charging and aggregation of magnetite nanoparticles. *Journal of Colloid and Interface Science*, 295(1), 115-123.
- Inskip, W. P. 1989. Adsorption of sulphate by kaolinite and amorphous iron oxide in the presence of organic ligands. *Journal of environmental quality*, 18(3), 379-385.
- Iqbal, M. T. 2012. Acid tolerance mechanisms in soil grown plants. *Malaysian Journal of Soil Science*, 16, 1-21.
- Jin-Hua, Y. U. A. N., Ren-Kou, X. U., Ning, W. A. N. G., & Jiu-Yu, L. I. 2011. Amendment of acid soils with crop residues and biochars. *Pedosphere*, 21(3), 302-308.
- Johnson, D. W., & Todd, D. E. 1983. Relationships among iron, aluminum, carbon, and sulfate in a variety of forest soils. *Soil Science Society of America Journal*, 47(4), 792-800.
- Jones Jr, J. B. 2002. *Agronomic handbook: management of crops, soils and their fertility*. CRC press.
- Jones, C., & Jacobsen, J. 2005. Plant nutrition and soil fertility. *Nutrient management module*, (2), 11.
- Joussein, E., Petit, S., Churchman, J., Theng, B., Righi, D., & Delvaux, B. 2005. Halloysite clay minerals – a review. *Clay Minerals*, 40(4), 383-426.
- Jung, K., Ok, Y. S., & Chang, S. X. 2011. Sulphate adsorption properties of acid-sensitive soils in the Athabasca oil sands region in Alberta, Canada. *Chemosphere*, 84(4), 457-463.
- Jungwirth, P., & Tobias, D. J. 2006. Specific ion effects at the air/water interface. *Chemical reviews*, 106(4), 1259-1281.
- Kabata-Pendias, A. 2010. *Trace elements in soils and plants*. CRC press.
- Kacimi, L., Simon-Masseron, A., Ghomari, A., & Derriche, Z. 2006. Reduction of clinkerization temperature by using phosphogypsum. *Journal of hazardous materials*, 137(1), 129-137.

- Kaiser, K., & Kaupenjohann, M. 1998. Influence of the soil solution composition on retention and release of sulfate in acid forest soils. *Water, Air, and Soil Pollution*, 101(1-4), 363-376.
- Kamarudin, R. A., & Zakaria, M. S. 2007. The utilization of red gypsum waste for glazes. *Malaysian Journal of Analytical Sciences*, 11(1), 57-64.
- Kenzo, T., Yoneda, R., Matsumoto, Y., Azani, M. A., & Majid, N. M. 2008. Leaf photosynthetic and growth responses on four tropical tree species to different light conditions in degraded tropical secondary forest, Peninsular Malaysia. *Japan Agricultural Research Quarterly: JARQ*, 42(4), 299-306.
- Kinraide, T. B. 2003. Toxicity factors in acidic forest soils: attempts to evaluate separately the toxic effects of excessive Al^{3+} and H^+ and insufficient Ca^{2+} and Mg^{2+} upon root elongation. *European Journal of Soil Science*, 54(2), 323-333.
- Kleber, M., Eusterhues, K., Keiluweit, M., Mikutta, C., Mikutta, R., & Nico, P. S. 2014. Mineral-organic associations: Formation, properties, and relevance in soil environments. *Adv. Agron*, 130.
- Kochian, L. V., Hoekenga, O. A., & Pineros, M. A. 2004. How do crop plants tolerate acid soils? Mechanisms of aluminum tolerance and phosphorous efficiency. *Annu. Rev. Plant Biol.*, 55, 459-493.
- Kogelmann, W. J., & Sharpe, W. E. 2006. Soil acidity and manganese in declining and nondeclining sugar maple stands in Pennsylvania. *Journal of environmental quality*, 35(2), 433-441.
- Korcak, R. F. 1988. Fluidized bed material applied at disposal levels: Effects on an apple orchard. *Journal of environmental quality*, 17(3), 469-473.
- Kovacevic, V., & Rastija, M. 2010. Impacts of liming by dolomite on the maize and barley grain yields. *Poljoprivreda*, 16(2), 3-8.
- Kretschmar, R., Sticher, H., & Hesterberg, D. 1997. Effects of adsorbed humic acid on surface charge and flocculation of kaolinite. *Soil Science Society of America Journal*, 61(1), 101-108.
- Krstic, D., Bjelic, D., Nikezic, D., & Djalovic, I. 2012. *Aluminium in Acid Soils: Chemistry, Toxicity and Impact on Maize Plants*. INTECH Open Access Publisher.
- Lal, R. (1997). Degradation and resilience of soils. *Philosophical Transactions of the Royal Society of London B: Biological Sciences*, 352(1356), 997-1010.

- Lal, R. 2002. The potential of soils of the tropics to sequester carbon and mitigate the greenhouse effect. *Advances in Agronomy*, 76, 1-30.
- Lal, R. 2007. Anthropogenic influences on world soils and implications to global food security. *Advances in Agronomy*, 93, 69-93.
- Lal, R., & Stewart, B. A. 2011. Sustainable management of soil resources and food security. *Advances in Soil Science World Soil Resources and Food Security*, 1-10.
- Lalande, R., Gagnon, B., & Royer, I. 2009. Impact of natural or industrial liming materials on soil properties and microbial activity. *Canadian journal of soil science*, 89(2), 209-222.
- Lavelle, P., & Spain, A. V. 2001. *Soil ecology*. Springer Science & Business Media.
- Lebron, I. 2011. Soil Colloidal Behaviour. *Handbook of Soil Sciences: Properties and Processes*.
- Lee, J. W., Kidder, M., Evans, B. R., Paik, S., Buchanan Iii, A. C., Garten, C. T., & Brown, R. C. 2010. Characterization of biochar's produced from corn stovers for soil amendment. *Environmental Science & Technology*, 44(20), 7970-7974.
- Lehmann, J., & Rondon, M. 2006. Bio-char soil management on highly weathered soils in the humid tropics. *Biological approaches to sustainable soil systems*. CRC Press, Boca Raton, FL, 517-530.
- Lehmann, J., da Silva Jr, J. P., Rondon, M., Cravo, M. D. S., Greenwood, J., Nehls, T., & Glaser, B. 2002. Slash-and-char-a feasible alternative for soil fertility management in the central Amazon. In *Proceedings of the 17th World Congress of Soil Science* (pp. 1-12).
- Lehmann, J., Rillig, M. C., Thies, J., Masiello, C. A., Hockaday, W. C., & Crowley, D. 2011. Biochar effects on soil biota—a review. *Soil Biology and Biochemistry*, 43(9), 1812-1836.
- Liang, B., Lehmann, J., Solomon, D., Kinyangi, J., Grossman, J., O'neill, B., & Neves, E. G. 2006. Black carbon increases cation exchange capacity in soils. *Soil Science Society of America Journal*, 70(5), 1719-1730.
- Linker, R., Weiner, M., Shmulevich, I., & Shaviv, A. 2006. Nitrate determination in soil pastes using attenuated total reflectance mid-infrared spectroscopy: Improved accuracy via soil identification. *Biosystems Engineering*, 94(1), 111-118.

- Liu, F., He, J., Colombo, C., & Violante, A. 1999. Competitive adsorption of sulphate and oxalate on goethite in the absence or presence of phosphate. *Soil science*, 164(3), 180-189.
- MacDonald, N. W., & Hart, J. B. 1990. Relating sulphate adsorption to soil properties in Michigan forest soils. *Soil Science Society of America Journal*, 54(1), 238-245.
- Madari, B. E., Reeves, J. B., Machado, P. L., Guimarães, C. M., Torres, E., & McCarty, G. W. 2006. Mid-and near-infrared spectroscopic assessment of soil compositional parameters and structural indices in two Ferralsols. *Geoderma*, 136(1), 245-259.
- Marcano-Martinez, E., & McBride, M. B. 1989. Calcium and sulphate retention by two Oxisols of the Brazilian cerrado. *Soil Science Society of America Journal*, 53(1), 63-69.
- Marschner, H. 1995. Functions of mineral nutrients: macronutrients. *Mineral nutrition of higher plants*, 2, 379-396.
- Marschner, H. 2011. *Marschner's mineral nutrition of higher plants*. Academic press.
- Marsh, K. B., Tillman, R. W., & Syers, J. K. 1987. Charge relationships of sulphate sorption by soils. *Soil Science Society of America Journal*, 51(2), 318-323.
- Martinez, C. E., Kleinschmidt, A. W., & Tabatabai, M. A. 1998. Sulphate adsorption by variable charge soils: Effect of low-molecular-weight organic acids. *Biology and fertility of soils*, 26(3), 157-163.
- McCarty, G. W., & Reeves, J. B. 2006. Comparison of near infrared and mid infrared diffuse reflectance spectroscopy for field-scale measurement of soil fertility parameters. *Soil Science*, 171(2), 94-102.
- McKeague, J. A., & Day, J. H. 1966. Dithionite-and oxalate-extractable Fe and Al as aids in differentiating various classes of soils. *Canadian Journal of Soil Science*, 46(1), 13-22.
- Mengel, K., Kosegarten, H., Kirkby, E. A., & Appel, T. (Eds.). 2001. *Principles of plant nutrition*. Springer Science & Business Media.
- Merino-Gergichevich, C., Alberdi, M., Ivanov, A. G., & Reyes-Díaz, M. 2010. Al³⁺-Ca²⁺ Interaction in plants growing in acid soils: al-phytotoxicity response to calcareous amendments. *Journal of soil science and plant nutrition*, 10(3), 217-243.

- Mora, M. L., Cartes, P., Demanet, R., & Cornforth, I. S. 2002. Effects of lime and gypsum on pasture growth and composition on an acid Andisol in Chile, South America. *Communications in soil science and plant analysis*, 33(13-14), 2069-2081.
- Naidu, R., Bolan, N. S., Kookana, R. S., & Tiller, K. G. 1994. Ionic-strength and pH effects on the sorption of cadmium and the surface charge of soils. *European Journal of Soil Science*, 45(4), 419-429.
- Naidu, R., Sumner, M. E., & Harter, R. D. 1998. Sorption of heavy metals in strongly weathered soils: an overview. *Environmental Geochemistry and Health*, 20(1), 5-9.
- Naidu, R., Syers, J. K., Tillman, R. W., & Kirkman, J. H. 1990. Effect of liming and added phosphate on charge characteristics of acid soils. *Journal of Soil Science*, 41(1), 157-164.
- Nandi, B. K., Goswami, A., & Purkait, M. K. (2009). Adsorption characteristics of brilliant green dye on kaolin. *Journal of Hazardous Materials*, 161(1), 387-395.
- Narro, L., Pandey, S., De León, C., Salazar, F., & Arias, M. P. 2001. Implications of soil-acidity tolerant maize cultivars to increase production in developing countries. In *Plant Nutrient Acquisition* (pp. 447-463). Springer Japan.
- Nelson, D. W., & Sommers, L. E. (1996). Total carbon, organic carbon, and organic matter. *Methods of soil analysis part 3 – chemical methods*, (methodsofsoilan3), 961-1010.
- Nierves, M. C. P., & Salas, F. M. 2015. Assessment of Soil Phosphorus and Phosphorus Fixing Capacity of Three Vegetable Farms at Cabintan, Ormoc City, Leyte. *World Journal of Agricultural Research*, 3(2), 70-73.
- Noble, A. D., Gillman, G. P., Nath, S., & Srivastava, R. J. 2001. Changes in the surface charge characteristics of degraded soils in the wet tropics through the addition of beneficiated bentonite. *Soil Research*, 39(5), 991-1001.
- Nodvin, S. C., Driscoll, C. T., & Likens, G. E. 1986. The effect of pH on sulphate adsorption by a forest soil. *Soil Science*, 142(2), 69-75.
- Novak, J. M., Busscher, W. J., Laird, D. L., Ahmedna, M., Watts, D. W., & Niandou, M. A. 2009. Impact of biochar amendment on fertility of a southeastern coastal plain soil. *Soil science*, 174(2), 105-112.

- Nurlaeny, N., Marschner, H., & George, E. 1996. Effects of liming and mycorrhizal colonization on soil phosphate depletion and phosphate uptake by maize (*Zea mays* L.) and soybean (*Glycine max* L.) grown in two tropical acid soils. *Plant and soil*, 181(2), 275-285.
- Osman, K. T. 2012. *Soils: principles, properties and management*. Springer Science & Business Media.
- Palm, C., Sanchez, P., Ahamed, S., & Awiti, A. (2007). Soils: A contemporary perspective. *Annu. Rev. Environ. Resour.* 32, 99-129.
- Paramanathan, S. 2000. *Soils of Malaysia: Their characteristics and Identification*. 1: 121-384, Akademi Sains Malaysia.
- Parker, D. R., & Bertsch, P. M. 1992. Formation of the " Al13" tridecameric aluminium polycation under diverse synthesis conditions. *Environmental science & technology*, 26(5), 914-921.
- Pärn, J., Pinay, G., & Mander, Ü. 2012. Indicators of nutrients transport from agricultural catchments under temperate climate: A review. *Ecological indicators*, 22, 4-15.
- Pavan, M. A., Bingham, F. T., & Pratt, P. F. 1984. Redistribution of exchangeable calcium, magnesium, and aluminium following lime or gypsum applications to a Brazilian Oxisol. *Soil Science Society of America Journal*, 48(1), 33-38.
- Pavia, D., Lampman, G., Kriz, G., & Vyvyan, J. 2008. *Introduction to spectroscopy*. Cengage Learning.
- Peel, M. C., Finlayson, B. L., & McMahon, T. A. 2007. Updated world map of the Köppen-Geiger climate classification. *Hydrology and Earth System Sciences Discussions Discussions*, 4(2), 439-473.
- Peng, X., Ye, L. L., Wang, C. H., Zhou, H., & Sun, B. 2011. Temperature-and duration-dependent rice straw-derived biochar: Characteristics and its effects on soil properties of an Ultisol in southern China. *Soil and Tillage Research*, 112(2), 159-166.
- Pigna, M., & Violante, A. 2003. Adsorption of sulphate and phosphate on Andisols. *Communications in soil science and plant analysis*, 34(15-16), 2099-2113.
- Pinkerton, A., & Simpson, J. R. 1986. Responses of some crop plants to correction of subsoil acidity. *Animal Production Science*, 26(1), 107-113.

- Porter, G. S., Bajita-Locke, J. B., Hue, N. V., & Strand, D. 2004. Manganese solubility and phytotoxicity affected by soil moisture, oxygen levels, and green manure additions. *Communications in soil science and plant analysis*, 35(1-2), 99-116.
- Qafoku, N. P., Van Ranst, E., Noble, A., & Baert, G. 2004. Variable charge soils: their mineralogy, chemistry and management. *Advances in Agronomy*, 84, 159-215.
- Rabileh, M. A., Shamshuddin, J., Panhwar, Q. A., Rosenani, A. B., & Anuar, A. R. 2015. Effects of biochar and/or dolomitic limestone application on the properties of Ultisol cropped to maize under glasshouse conditions. *Canadian Journal of Soil Science*, 95(1), 37-47.
- Rahmani, O., Junin, R., Tyrer, M., & Mohsin, R. 2014. Mineral Carbonation of Red Gypsum for CO₂ Sequestration. *Energy & Fuels*, 28(9), 5953-5958.
- Railsback, L. B. 2006. Some fundamentals of mineralogy and geochemistry. *On-line book, quoted from: www.gly.uga.edu/railsback*.
- Rajan, S. S. S. 1978. Sulphate adsorbed on hydrous alumina, ligands displaced, and changes in surface charge. *Soil Science Society of America Journal*, 42(1), 39-44.
- Rao, S. M., & Sridharan, A. 1984. Mechanism of sulphate adsorption by kaolinite. *Clays Clay Miner*, 32(5), 414-418.
- Raphael, L. 2011. *Application of FTIR spectroscopy to agricultural soils analysis*. INTECH Open Access Publisher.
- Rashid, A. 1996. Secondary and micronutrients. *Soil science*, 374.
- Rengel, Z. (Ed.). 2003. *Handbook of soil acidity* (Vol. 94). CRC Press.
- Reuss, J. O., & Johnson, D. W. (2012). *Acid deposition and the acidification of soils and waters* (Vol. 59). Springer Science & Business Media.
- Rieuwerts, J. S. 2007. The mobility and bioavailability of trace metals in tropical soils: a review. *Chemical Speciation & Bioavailability*, 19(2), 75-85.
- Ritchey, K. D., & Snuffer, J. D. 2002. Limestone, gypsum, and magnesium oxide influence restoration of an abandoned Appalachian pasture. *Agronomy journal*, 94(4), 830-839.
- Ritchie, G. S. P. 1989. The chemical behaviour of aluminium, hydrogen and manganese in acid soils. *Soil acidity and plant growth*, 1-60.

- Rodriguez-Cruz, M. S., Sanchez-Martin, M. J., & Sanchez-Camazano, M. 2005. A comparative study of adsorption of an anionic and a non-ionic surfactant by soils based on physicochemical and mineralogical properties of soils. *Chemosphere*, 61(1), 56-64.
- Rodríguez-Jordá, M. P., Garrido, F., & García-González, M. T. 2010. Assessment of the use of industrial by-products for induced reduction of As, and Se potential leachability in an acid soil. *Journal of hazardous materials*, 175(1), 328-335.
- Rose, S. 1998. Anion adsorption and desorption characteristics of a Piedmont ultisol: Some implications for the fate of sulfate deposition. *Water, Air, and Soil Pollution*, 101(1-4), 333-347.
- Rowell, D. L. (2014). *Soil science: Methods & applications*. Routledge.
- Rushton, G. T., Karns, C. L., & Shimizu, K. D. 2005. A critical examination of the use of the Freundlich isotherm in characterizing molecularly imprinted polymers (MIPs). *Analytica chimica acta*, 528(1), 107-113.
- Sahin, U., Oztas, T., & Anapali, O. 2003. Effects of consecutive applications of gypsum in equal, increasing, and decreasing quantities on soil hydraulic conductivity of a saline-sodic soil. *Journal of Plant Nutrition and Soil Science*, 166(5), 621-624.
- Sakurai, K., Ohdate, Y., & Kyuma, K. 1989. Factors affecting zero point of charge (ZPC) of variable charge soils. *Soil Science and Plant Nutrition*, 35(1), 21-31.
- SAS, S., & Guide, S. U. S. 2008. Version 9.4. Cary, NC, USA: SAS Institute Inc.
- Seta, A. K., & Karathanasis, A. D. 1996. Water dispersible colloids and factors influencing their dispersibility from soil aggregates. *Geoderma*, 74(3), 255-266.
- Shaff, J. E., Schultz, B. A., Craft, E. J., Clark, R. T., & Kochian, L. V. 2010. GEOCHEM-EZ: a chemical speciation program with greater power and flexibility. *Plant and Soil*, 330(1-2), 207-214.
- Shamshuddin, J., & Anda, M. 2008. Charge properties of soils in Malaysia dominated by kaolinite, gibbsite, goethite and hematite. *Bulletin of the Geological Society of Malaysia*, 54, 27-31.
- Shamshuddin, J., & Daud, N. W. 2011. *Classification and management of highly weathered soils in Malaysia for production of plantation crops*. INTECH Open Access Publisher.

- Shamshuddin, J., & Fauziah, C. I. 2010. Weathered tropical soils: the ultisols and oxisols. *UPM Press, Serdang*.
- Shamshuddin, J., & Ismail, H. 1995. Reactions of ground magnesium limestone and gypsum in soils with variable-charge minerals. *Soil Science Society of America Journal*, 59(1), 106-112.
- Shamshuddin, J., Fauziah, C. I., Anda, M., Kapok, J., & Shazana, M. A. R. S. 2011. Using ground basalt and/or organic fertilizer to enhance productivity of acid soils in Malaysia for crop production. *Malaysian Journal Soil Science*, 15, 127-146.
- Shamshuddin, J., Fauziah, I. C., & Sharifuddin, H. A. H. 1991. Effects of limestone and gypsum application to a Malaysian Ultisol on soil solution composition and yields of maize and groundnut. In *Plant-Soil Interactions at Low pH* (pp. 501-508). Springer Netherlands.
- Shoji, s., & Fujiwara, y. 1984. Active aluminium and iron in the humus horizons of andosols from north-eastern japan: their forms, properties, and significance in clay weathering. *Soil Science*, 137(4), 216-226.
- Singh, B. R. 1984. Sulphate sorption by acid forest soils: 2. Sulphate adsorption isotherms with and without organic matter and oxides of aluminium and iron1. *Soil Science*, 138(4), 294-297.
- Singh, B. R., Abrahamsen, G., & Stuanes, A. 1980. Effect of simulated acid rain on sulfate movement in acid forest soils. *Soil Science Society of America Journal*, 44(1), 75-80.
- Singh, B., Singh, B. P., & Cowie, A. L. 2010. Characterisation and evaluation of biochars for their application as a soil amendment. *Soil Research*, 48(7), 516-525.
- Slattery, W. J., Ridley, A. M., & Windsor, S. M. 1991. Ash alkalinity of animal and plant products. *Animal Production Science*, 31(3), 321-324.
- Soil Survey Staff, 2014. Soil survey laboratory methods manual. *Soil Survey Laboratory Investigations Report*, (42), Washington, DC.
- Sokolova, T. A., & Alekseeva, S. A. 2008. Adsorption of sulphate ions by soils (a review). *Eurasian Soil Science*, 41(2), 140-148.
- Sokolova, T. A., Dronova, T. Y., & Tolpeshta, I. I. 2005. Glinistye mineraly v pochvakh. *Clay Minerals in Soils*, Moscow.

- Sparks, D. L. 1999. Kinetics and mechanisms of chemical reactions at the soil mineral/water interface. *Soil physical chemistry*, 2, 135-191.
- Sparks, D. L. 2003. *Environmental soil chemistry*. Academic press.
- Spokas, K. A., Cantrell, K. B., Novak, J. M., Archer, D. W., Ippolito, J. A., Collins, H. P., ... & Lentz, R. D. 2012. Biochar: a synthesis of its agronomic impact beyond carbon sequestration. *Journal of environmental quality*, 41(4), 973-989.
- Sposito, G. 2008. *The chemistry of soils*. Oxford university press.
- Stenberg, B., Rossel, R. A. V., Mouazen, A. M., & Wetterlind, J. 2010. Chapter five-visible and near infrared spectroscopy in soil science. *Advances in agronomy*, 107, 163-215.
- Stern, L. A., & Milliken, K. L. 2004. Silicate surface chemistry and dissolution kinetics in dilute aqueous systems.
- Stevenson, F. J. 1994. *Humus chemistry: genesis, composition, reactions*. John Wiley & Sons.
- Stumm, W. 1992. Chemistry of the solid-water interface: processes at the mineral-water and particle-water interface in natural systems.
- Stumpe, B., Weihermüller, L., & Marschner, B. 2011. Sample preparation and selection for qualitative and quantitative analyses of soil organic carbon with mid-infrared reflectance spectroscopy. *European Journal of Soil Science*, 62(6), 849-862.
- Sumner, M. E., & Noble, A. D. 2003. Soil acidification: the world story. *Handbook of soil acidity*. Marcel Dekker, New York, 1-28.
- Sumner, M. E., Fey, M. V., & Noble, A. D. 1991. Nutrient status and toxicity problems in acid soils. In *Soil acidity* (pp. 149-182). Springer Berlin Heidelberg.
- Sumner, M. E., Miller, W. P., Sparks, D. L., Page, A. L., Helmke, P. A., Loeppert, R. H., ... & Johnston, C. T. 1996. Cation exchange capacity and exchange coefficients. *Methods of soil analysis. Part 3-chemical methods*, 1201-1229.
- Takahashi, T., Fukuoka, T., & Dahlgren, R. A. 1995. Aluminium solubility and release rates from soil horizons dominated by aluminium-humic complexes. *Soil Science and plant nutrition*, 41(1), 119-131.

- Takahashi, T., Ikeda, Y., Fujita, K., & Nanzyo, M. 2006. Effect of liming on organically complexed aluminium of nonallophanic Andosols from north-eastern Japan. *Geoderma*, 130(1), 26-34.
- Talha I. Z, J. Shamsuddin, C.I. Fauziah and I. Roslan. 2016. FTIR and XRD analyses of highly weathered Ultisols and Oxisols in Peninsular Malaysia, *Asian Journal of Agriculture and Food Sciences* 4 (4), 191-201.
- Tang, J., Zhu, W., Kookana, R., & Katayama, A. 2013. Characteristics of biochar and its application in remediation of contaminated soil. *Journal of bioscience and bioengineering*, 116(6), 653-659.
- Teh, C. B. S., & Talib, J. 2006. *Soil physics analyses: volume 1*. Universiti Putra Malaysia Press.
- Thakre, Y. G., Choudhary, M. D., & Raut, R. D. 2014. Mineralogical studies of red and black soils in Wardha region. *Der Pharma Chemica*, 6(1), 407-410.
- Theng, B. K., & Yuan, G. 2008. Nanoparticles in the soil environment. *Elements*, 4(6), 395-399.
- Thermo, N. 2001. Introduction to Fourier transform infrared spectrometry. *Thermo Nicolet Corporation: Madison-USA*.
- Thomas, G. W. 1982. Exchangeable cations. *Methods of soil analysis. Part 2. Chemical and microbiological properties, (methodsofsoil2)*, 159-165.
- Toma, M., & Saigusa, M. 1997. Effects of phosphogypsum on amelioration of strongly acid nonallophanic andosols. *Plant and soil*, 192(1), 49-55.
- Toma, M., Saigusa, M., Qafoku, N., & Sumner, M. E. 2005. Effects of Gypsum on Amelioration of Subsoil Acidity of Andisols. *Journal of Integrated Field Science*, 2, 69-71.
- Tombacz, E., & Szekeres, M. 2006. Surface charge heterogeneity of kaolinite in aqueous suspension in comparison with montmorillonite. *Applied Clay Science*, 34(1), 105-124.
- Tombacz, E., Libor, Z., Illes, E., Majzik, A., & Klumpp, E. (2004). The role of reactive surface sites and complexation by humic acids in the interaction of clay mineral and iron oxide particles. *Organic Geochemistry*, 35(3), 257-267.
- Tuna, A. L., Kaya, C., Ashraf, M., Altunlu, H., Yokas, I., & Yagmur, B. 2007. The effects of calcium sulphate on growth, membrane stability and nutrient uptake of tomato plants grown under salt stress. *Environmental and Experimental Botany*, 59(2), 173-178.

- Turner, L. J., & Kramer, J. R. 1991. Sulphate ion binding on goethite and hematite. *Soil Science*, 152(3), 226-230.
- Ulrich, B., & Sumner, M. E. (Eds.). 2012. *Soil acidity*. Springer Science & Business Media.
- Uzoho, B. U., & Ekeh, C. (2014). Potassium status of soils in relation to land use types in Ngor-Okpala, Southeastern Nigeria. *J Nat Sci Res*, 4(6), 105-114.
- van Zwieten, L., Kimber, S., Morris, S., Chan, K. Y., Downie, A., Rust, J., & Cowie, A. 2010. Effects of biochar from slow pyrolysis of papermill waste on agronomic performance and soil fertility. *Plant and soil*, 327(1-2), 235-246.
- Vaneckhaute, C., Ghekiere, G., Michels, E., Vanrolleghem, P. A., Tack, F., & Meers, E. 2014. Assessing nutrient use efficiency and environmental pressure of macronutrients in biobased mineral fertilizers: a review of recent advances and best practices at field scale. *Advances in Agronomy*, 128, 137-180.
- Velde, B. (Ed.). 2013. *Origin and mineralogy of clays: clays and the environment*. Springer Science & Business Media.
- Verheijen, F., Jeffery, S., Bastos, A. C., Van Der Velde, M., & Diafas, I. 2010. Biochar Application to Soils-A Critical Scientific Review of Effects on Soil Properties. *Processes and Functions, European Commission Joint Research Centre for scientific and Technical reports*, 51-68.
- Violante, A., Cozzolino, V., Perelomov, L., Caporale, A. G., & Pigna, M. 2010. Mobility and bioavailability of heavy metals and metalloids in soil environments. *Journal of soil science and plant nutrition*, 10(3), 268-292.
- Violante, A., Krishnamurti, G. S. R., & Pigna, M. 2007. Factors affecting the sorption-desorption of trace elements in soil environments. *Biophysico Chemical Processes of Heavy Metals and Metalloids in Soil Environments*, 169-214.
- Violante, A., Zhu, J., Pigna, M., Jara, A., Cozzolino, V., & Mora, M. L. 2013. Role of Biomolecules in Influencing Transformation Mechanisms of Metals and Metalloids in Soil Environments. In *Molecular Environmental Soil Science* (pp. 167-191). Springer Netherlands.
- von Uexküll, H. R., & Mutert, E. 1995. Global extent, development and economic impact of acid soils. *Plant and soil*, 171(1), 1-15.

- von Willert, F. J., & Stehouwer, R. C. 2003. Compost and calcium surface treatment effects on subsoil chemistry in acidic mine spoil columns. *Journal of environmental quality*, 32(3), 781-788.
- Wang, D., Tang, H., & Gregory, J. 2002. Relative importance of charge neutralization and precipitation on coagulation of kaolin with PACl: effect of sulphate ion. *Environmental science & technology*, 36(8), 1815-1820.
- Warnock, D. D., Lehmann, J., Kuyper, T. W., & Rillig, M. C. 2007. Mycorrhizal responses to biochar in soil—concepts and mechanisms. *Plant and Soil*, 300(1-2), 9-20.
- Watmough, S. A., Eimers, M. C., & Dillon, P. J. 2007. Manganese cycling in central Ontario forests: response to soil acidification. *Applied geochemistry*, 22(6), 1241-1247.
- Westerman, P. W., & Bicudo, J. R. 2005. Management considerations for organic waste use in agriculture. *Bioresource Technology*, 96(2), 215-221.
- White, R. E. 2013. *Principles and practice of soil science: the soil as a natural resource*. John Wiley & Sons.
- Wijnja, H., & Schulthess, C. P. 2000. Interaction of carbonate and organic anions with sulfate and selenate adsorption on an aluminium oxide. *Soil Science Society of America Journal*, 64(3), 898-908.
- Wild, A. 2003. *Soils, land and food: managing the land during the twenty-first century*. Cambridge University Press.
- Yao, Y., Gao, B., Zhang, M., Inyang, M., & Zimmerman, A. R. 2012. Effect of biochar amendment on sorption and leaching of nitrate, ammonium, and phosphate in a sandy soil. *Chemosphere*, 89(11), 1467-1471.
- Yavitt, J. B., & Wright, S. J. 2002. Charge characteristics of soil in a lowland tropical moist forest in Panama in response to dry-season irrigation. *Soil Research*, 40(2), 269-281.
- Yu, T. R. 1997. *Chemistry of variable charge soils*. Oxford University Press.
- Yuan, J. H. and Xu, R. K. 2011. The amelioration effects of low-temperature biochar generated from nine crop residues on an acidic Ultisol. *Soil Use Manage.* 27: 110 - 115.
- Yuan, J. H., Xu, R. K., Wang, N. & Li, J. Y. 2011. Amendment of acid soils with crop residues and biochars. *Pedosphere*, 21(3), 302-308.

- Zeigler, R. S., Pandey, S., Miles, J., Gourley, L. M., & Sarkarung, S. 1995. Advances in the selection and breeding of acid-tolerant plants: Rice, maize, sorghum and tropical forages. In *Plant-Soil Interactions at Low pH: Principles and Management* (pp. 391-406). Springer Netherlands.
- Zhang, G. Y., Zhang, X. N., & Yu, T. R. (1987). Adsorption of sulphate and fluoride by variable charge soils. *Journal of soil science*, 38(1), 29-38.
- Zhu, Q. H., Peng, X. H., Huang, T. Q., Xie, Z. B. & Holden, N. M. 2014. Effect of biochar addition on maize growth and nitrogen use efficiency in acidic red soils. *Pedosphere*, 24(6), 699-708.

