



UNIVERSITI PUTRA MALAYSIA

***ENZYMATIC DIRECTED INTERESTERIFICATION OF PALM OIL FOR
PRODUCTION OF HIGHLY UNSATURATED PALM FRACTION***

NOOR LIDA BINTI HABI MAT DIAN

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By

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**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfillment of the Requirements for the Degree of Doctor of Philosophy**

May 2017

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

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May 2017

Chairman : Professor Lai Oi Ming, PhD
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Palm oil (PO) is a balanced oil with equal parts of unsaturated fatty acids (USAFA) mainly oleic acid (monounsaturated fatty acid, MUFA) and saturated fatty acids (SAFA) mainly palmitic acid. USAFA which include polyunsaturated fatty acids and MUFA are classified as good fat that can reduce the risk of coronary heart disease. The ultimate objective of this study was to produce a highly unsaturated palm fraction by enzymatic directed interesterification (EDIE) of PO and fractionation of the interesterified PO (EDIE PO). EDIE is a process whereby randomization of fatty acids on the triacylglycerol (TAG) molecules of oil takes place at a temperature lower than the melting temperature of its highest melting TAGs, normally the trisaturated (S_3) TAGs. The EDIE of oil will eventually result in a structured fat with a high concentration of triunsaturated (U_3) and S_3 TAGs. Since the S_3 TAGs were formed simultaneously with the U_3 TAGs, the second objective of this project was to obtain a highly saturated palm fraction. The first part of the study focused on the optimization of the EDIE processing parameters namely enzyme load, reaction temperature and reaction time using Response Surface Methodology in order to obtain EDIE PO having the highest U_3 TAGs concentration, with a tolerable if not the lowest amount of by-products (free FAs, monoacylglycerol, and diacylglycerol). The optimum reaction parameters were: enzyme load of 10%, reaction temperature of 30 °C and reaction time of 18 hours. EDIE of PO was able to significantly ($p < 0.05$) increase the percentage of U_3 TAG in PO, i.e., from 4.3% before EDIE to 27.6% after EDIE, and S_3 TAGs, i.e., from 5.2% before EDIE to 31.9% after EDIE. Differential Scanning Calorimetry melting profile showed that the EDIE PO had two well-separated high-melting and low-melting endothermic peaks, indicating ease of fractionation. The EDIE PO was subsequently subjected to fractionation at 5, 10, 15, 20 and 25 °C to obtain the highly unsaturated palm fraction (palm olein, EDIE PO_o) and its co-product, a highly saturated palm fraction (palm stearin, EDIE PO_s). The U_3 TAGs and USAFA content of the EDIE PO_o, and S_3 TAGs and SAFA content of the EDIE PO_s fractions varied depending on the

fractionation temperatures. The lower the fractionation temperature, the higher the U₃ TAGs and consequently the higher the USAFA content of the EDIE POo. Reducing the fractionation temperature from 25 °C to 5 °C resulted in the increase of the U₃ TAGs and USAFA content of the EDIE POo from 34.6% to 43.0% and 66.7% to 75.0%, respectively. The S₃ TAGs and SAFA content of the EDIE POs were increased from 50.1% to 92.2% and 82.8% to 97.8%, respectively, when fractionation temperature increase from 5 °C to 25 °C. In conclusion, EDIE of PO followed by fractionation of the EDIE PO were able to produce a highly unsaturated fraction, EDIE POo and yielded a highly saturated fraction, EDIE POs. The highly unsaturated EDIE POo could be used as a healthy cooking oil. The highly saturated EDIE POs could be used as a *trans* fats alternative.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**INTERESTERIFIKASI TERARAH BERENZIM KE ATAS MINYAK
KELAPA SAWIT BAGI PENGHASILAN PECAHAN MINYAK KELAPA
SAWIT KAYA ASID LEMAK TAK TEPU**

Oleh

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Minyak kelapa sawit (PO) adalah minyak yang mempunyai keseimbangan dalam kandungan asid lemak tak tepu (USAFA) terutamanya asid oleik (asid lemak monotaktepu, MUFA) dan asid lemak tepu (SAFA) terutamanya asid palmitik. USAFA yang merangkumi asid lemak politatepu dan MUFA dikelaskan sebagai lemak baik, yang boleh mengurangkan risiko sakit jantung koronari. Objektif utama kajian ini adalah untuk menghasilkan pecahan PO yang kaya dengan USAFA melalui proses interesterifikasi terarah berenzim (EDIE) ke atas PO diikuti dengan pemeringkatan ke atas PO yang telah diesterifikasi. EDIE adalah proses di mana perawakan asid lemak di dalam molekul triasilgliserida (TAG) minyak berlaku pada suhu lebih rendah daripada suhu lebur TAG yang mempunyai suhu takat lebur yang tertinggi, biasanya TAG tritepu (S_3). EDIE yang dilakukan ke atas PO menghasilkan PO terinterester (EDIE PO) yang kaya dengan TAG tritaktepu (U_3) dan S_3 . Oleh kerana berlaku pembentukan TAG S_3 serentak dengan pembentukan TAG U_3 , maka projek ini juga berobjektif untuk mendapatkan pecahan PO pejal yang tinggi SAFA. Bahagian pertama kajian ini tertumpu kepada mengoptimakan parameter pemrosesan EDIE, iaitu jumlah enzim, suhu tindak balas dan masa tindak balas, menggunakan “Kaedah Tindakbalas Permukaan” bagi mendapatkan EDIE PO yang mempunyai kandungan TAG U_3 yang tertinggi dan kandungan produk sampingan (asid lemak bebas, monoasilgliserida dan diasilgliserida) yang rendah. Parameter optimum reaksi ialah jumlah enzim sebanyak 10%; suhu tindak balas pada 30 °C dan masa tindak balas selama 18 jam. EDIE ke atas PO telah berjaya dengan ketara ($p < 0.05$) meningkatkan kandungan TAG U_3 dan TAG S_3 di dalam PO. Kandungan TAG U_3 dan TAG S_3 setiap satunya meningkat daripada 4.3% sebelum EDIE kepada 27.6% selepas EDIE, dan daripada 5.2% sebelum EDIE kepada 31.9% selepas EDIE. Profil peleburan “Differential Scanning Calorimetry” menunjukkan bahawa EDIE PO mempunyai dua puncak peleburan yang ketara berbeza iaitu puncak peleburan bertakat lebur rendah dan puncak peleburan bertakat lebur tinggi,

yang menunjukkan kemudahan untuk pemeringkatan. Seterusnya, proses pemeringkatan pada suhu 5, 10, 20, 15 dan 25 °C dilakukan ke atas EDIE PO bagi mendapatkan pecahan lembut (EDIE POo) yang kaya dengan USAFA. Hasil sampingan daripada proses pemeringkatan EDIE PO adalah pecahan pejal (EDIE POs) yang kaya dengan SAFA. Kandungan TAG U₃ dan USAFA di dalam EDIE POo dan kandungan TAG S₃ dan SAFA di dalam EDIE POs berbeza mengikut suhu proses pemeringkatan. Semakin rendah suhu pemeringkatan, semakin tinggi kandungan TAG U₃ dan USAFA di dalam EDIE POo. Penurunan suhu pemeringkatan daripada 25 °C kepada 5 °C mengakibatkan peningkatan kandungan TAG U₃ dan USAFA di dalam EDIE POo masing-masing daripada 34.6% kepada 43.0% dan 66.7% kepada 75.0%. Kandungan TAG S₃ dan SAFA di dalam EDIE POs pula masing-masing meningkat daripada 50.1% kepada 92.2% dan 82.8% kepada 97.8%, apabila suhu pemeringkatan ditingkatkan daripada 5 °C kepada 25 °C. Kesimpulannya, EDIE ke atas PO diikuti dengan pemeringkatan ke atas EDIE PO berjaya menghasilkan pecahan lembut (EDIE POo) yang kaya dengan USAFA (terutamanya asid oleik) dan pecahan pejal (EDIE POs) yang kaya SAFA (terutamanya asid palmitik). EDIE POo boleh digunakan sebagai minyak masak atau minyak salad yang menyihatkan, manakala EDIE POs boleh digunakan sebagai alternatif kepada lemak *trans*.

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I certify that a Thesis Examination Committee has met on 17 May 2017 to conduct the final examination of NOOR LIDA BINTI HABI MAT DIAN on her thesis entitled “ENZYMATIC DIRECTED INTERESTERIFICATION OF PALM OIL FOR PRODUCTION OF HIGHLY UNSATURATED PALM FRACTION” in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiv
LIST OF FIGURES	xvii
LIST OF ABBREVIATIONS	xxi
 CHAPTER	
1 INTRODUCTION	1
2 LITERATURE REVIEW	5
2.1 Overview of oil palm and palm oil	5
2.1.1 Oil palm cultivation in Malaysia	5
2.1.2 Oil palm fruits processing	8
2.1.3 Palm oil and its components	9
2.1.4 Minor components in palm oil	10
2.1.5 Nutritional aspect of palm oil and its fractions	13
2.1.5.1 Unsaturation and essential fatty acids	13
2.1.5.2 Palm oil and cardiovascular disease	13
2.1.5.3 Minor components of palm oil and their health benefits	15
2.1.6 Food applications of palm oil	16
2.1.6.1 Frying/cooking oil	17
2.1.6.2 Solid fat products	17
2.1.6.3 Confectionery fat	18
2.1.6.4 Mayonnaise and salad dressings	18
2.1.6.5 Other food uses	19
2.2 Oils and fats modification	19
2.2.1 Blending	20
2.2.2 Hydrogenation	20
2.2.3 Interesterification	21
2.2.3.1 Alcoholysis and glycerolysis	22
2.2.3.2 Acidolysis	22
2.2.3.3 Transesterification (interesterification)	23
2.2.3.4 Chemical interesterification	24
2.2.3.5 Enzymatic interesterification	30
2.2.3.6 Directed interesterification	35
2.3 Unsaturated oils	37
2.4 <i>Trans</i> fats	38
2.5 Fractionation	40
2.5.1 Principle of fractionation	42

2.5.2	Types of fractionation	43
2.5.2.1	Dry fractionation	43
2.5.2.2	Detergent fractionation	43
2.5.2.3	Solvent fractionation	44
2.6	Summary	45
3	OPTIMIZATION OF ENZYMATIC DIRECTED INTERESTERIFICATION REACTION OF PALM OIL FOR THE PRODUCTION OF HIGH TRIUNSATURATED TRIACYLGLYCEROL USING RESPONSE SURFACE METHODOLOGY (RSM)	46
3.1	Introduction	46
3.2	Materials and Methods	48
3.2.1	Materials	48
3.2.2	Methods	48
3.2.2.1	Response Surface Methodology experimental design	48
3.2.2.2	Lipozyme TLIM lipase conditioning	49
3.2.2.3	Enzymatic directed interesterification of palm oil	49
3.2.2.4	Triacylglycerol composition	50
3.2.2.5	Free fatty acid, monoacylglycerol and diacylglycerol quantification	50
3.3	Results and Discussion	50
3.3.1	Screening of enzymatic directed interesterification reaction parameters range	50
3.3.2	Model fitting	51
3.3.3	The degree of three parameters on enzymatic directed interesterification reaction of palm oil for triunsaturated triacylglycerol production	55
3.3.4	Single factor response	55
3.3.5	Relationship between the processing parameters	58
3.3.6	Optimization of enzymatic directed interesterification of palm oil reaction conditions for the highest triunsaturated triacylglycerol yield	66
3.4	Conclusion	68
4	CHARACTERISTICS OF ENZYMATIC DIRECTED INTERESTERIFIED PALM OIL	69
4.1	Introduction	69
4.2	Materials and Methods	70
4.2.1	Materials	70
4.2.2	Methods	70
4.2.2.1	Enzymatic directed interesterification of palm oil using the optimized reaction parameters	70
4.2.2.2	Lipozyme TLIM lipase conditioning	70
4.2.2.3	Purification of triacylglycerol using short path distillation	70
4.2.2.4	Fatty acid composition	71
4.2.2.5	Triacylglycerol composition	71

4.2.2.6	Slip melting point	71
4.2.2.7	Iodine value	72
4.2.2.8	Thermal analysis by Differential Scanning Calorimetry	72
4.2.2.9	Solid fat content	72
4.2.2.10	Polymorphic behavior	73
4.2.2.11	Statistical analysis	73
4.3	Results and Discussion	73
4.3.1	Purification of enzymatic directed interesterified palm oil	73
4.3.2	Characteristic of enzymatic directed interesterified palm oil	73
4.3.2.1	Fatty acid composition	73
4.3.2.2	Triacylglycerol composition	74
4.3.2.3	Differential Scanning Calorimetry melting profile	77
4.3.2.4	Differential scanning calorimetry crystallization profile	80
4.3.2.5	Solid fat content and slip melting point	84
4.3.2.6	Iodine value	86
4.3.2.7	Polymorphic behavior	87
4.4	Conclusion	88
5	PHYSICOCHEMICAL CHARACTERISTICS OF OLEIN AND STEARIN FRACTIONS OF ENZYMATIC DIRECTED INTERESTERIFIED PALM OIL	94
5.1	Introduction	94
5.2	Materials and Methods	95
5.2.1	Materials	95
5.2.2	Methods	95
5.2.2.1	Enzymatic directed interesterification of palm oil	95
5.2.2.2	Solvent fractionation	96
5.2.2.3	Fatty acid composition	96
5.2.2.4	Triacylglycerol composition	96
5.2.2.5	<i>Sn</i> -2 fatty acid positional distribution	96
5.2.2.6	Thermal analysis by Differential Scanning Calorimetry	97
5.2.2.7	Solid fat content	97
5.2.2.8	Slip melting point	97
5.2.2.9	Iodine value	97
5.2.2.10	Statistical analysis	97
5.3	Results and Discussion	97
5.3.1	Fatty acid composition of olein and stearin fractions of enzymatic directed interesterified palm oil vs. refined, bleached and deodorized palm oil	97
5.3.2	Triacylglycerol composition of olein and stearin fractions of enzymatic directed interesterified palm oil vs. refined, bleached and deodorized palm oil	102

5.3.3	Differential Scanning Calorimetry thermal profile olein and stearin fractions of enzymatic directed interesterified palm oil vs. refined, bleached and deodorized palm oil	106
5.3.3.1	Differential Scanning Calorimetry melting profile	110
5.3.3.2	Differential Scanning Calorimetry crystallization profile	118
5.3.4	Solid fat content and slip melting point of palm olein and palm stearin fraction of enzymatic directed interesterified palm oil vs. refined, bleached and deodorized palm oil	127
5.3.5	Iodine value of palm olein and palm stearin fractions of enzymatic directed interesterified palm oil vs. that of refined, bleached and deodorized palm oil	132
5.4	Conclusion	135
6	SUMMARY, CONCLUSION AND RECOMMENDATION FOR FUTURE WORK	137
6.1	Summary	137
6.2	Conclusion	138
6.3	Recommendation for future work	139
	REFERENCES	140
	APPENDICES	184
	BIODATA OF STUDENT	190
	LIST OF PUBLICATIONS	191

LIST OF TABLES

Table		Page
2.1	World production of 17 oils and fats ('000 Tonnes) in 2014-2016	6
2.2	Fatty acid composition of palm oil	10
2.3	Acylglycerol composition of palm oil	11
2.4	Physical characteristics of palm oil	11
2.5	Chemical and physical characteristics of fractionated palm oil	12
2.6	Minor components of crude palm oil	13
2.7	Specificity of lipases and their sources	25
3.1	Experiment variables in coded and actual unit	49
3.2	Central Composite Design for twenty experimental runs and experimental data for two responses namely the triunsaturated triacylglycerols and by-products yield	53
3.3	ANOVA of quadratic model representing triunsaturated triacylglycerol yield from enzymatic directed interesterification of palm oil	54
3.4	ANOVA of linear model representing by-products yield from enzymatic directed interesterification of palm oil	54
3.5	The yield of triunsaturated triacylglycerol (U ₃ TAGs) and by-products from three validation runs using the optimized reaction condition as compared to the predicted value of RSM optimization	66
4.1	Fatty acid composition of enzymatic directed interesterified (EDIE PO) and refined bleached and deodorized (RBD PO) palm oil	74
4.2	Triacylglycerol composition (% area) of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	76
4.3	Differential Scanning Calorimetry total (ΔH_f) and partial ($\Delta H_{f^{\circ}C}$) melting enthalpy of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	80
4.4	Differential Scanning Calorimetry partial liquid (%) at various temperature of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	81
4.5	Differential Scanning Calorimetry total ($-\Delta H_f$) and partial ($-\Delta H_{f^{\circ}C}$) crystallization enthalpy of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	83
4.6	Differential Scanning Calorimetry partial solid (%) at various temperature of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	84

4.7	Slip melting point of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	86
4.8	Iodine value of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil	86
4.9	Crystal polymorphic form of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil at storage and determination temperature of 5, 10, 15, 20 and 25 °C	87
5.1	Fatty acid composition (%) of olein fraction of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	98
5.2	Fatty acid composition (%) of stearin fraction of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at 5, 10, 15, 20 and 25 °C	100
5.3	Triacylglycerol composition (% area) of olein fraction of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	103
5.4	Triacylglycerol composition (% area) of stearin fraction of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at 5, 10, 15, 20 and 25 °C	107
5.5	Fatty acid composition (%) at the sn-2 position of triacylglycerol molecules of enzymatic directed interesterified (EDIE PO) and refined, bleached and deodorized (RBD PO) palm oil, and the olein fractions of EDIE PO (EDIE POo) and RBD PO (RBD POo) fractionated at 5 °C and 15 °C	112
5.6	Differential Scanning Calorimetry total melting enthalpy, onset of melting and complete melting temperature of olein fractions of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	114
5.7	Differential Scanning Calorimetry partial liquid and partial melting enthalpy at -5 °C and 0 °C of olein fractions of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	115
5.8	Differential Scanning Calorimetry total melting enthalpy, onset of melting and complete melting temperature of stearin fractions of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at 5, 10, 15, 20 and 25 °C	119

5.9	Differential Scanning Calorimetry total crystallization enthalpy, onset of crystallization and complete crystallization temperature of olein fractions of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	122
5.10	Differential Scanning Calorimetry total crystallization enthalpy, onset of crystallization and complete crystallization temperature of stearin fractions of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at 5, 10, 15, 20 and 25 °C	125
5.11	Slip melting point of olein fraction of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	127
5.12	Slip melting point of stearin fractions of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at 5, 10, 15, 20 and 25 °C	130
5.13	Iodine value of olein fraction of enzymatic directed interesterified (EDIE POo) and refined, bleached and deodorized (RBD POo) palm oil fractionated at 5, 10, 15, 20 and 25 °C	132
5.14	Iodine value of stearin fraction of enzymatic directed interesterified (EDIE POs) and refined, bleached and deodorized (RBD POs) palm oil fractionated at different temperatures	134

LIST OF FIGURES

Figure	Page
2.1	6
Percentage of the Area (Total: 258.9 Million Hectares) Used for Ten Major Oilseeds Cultivation in 2012	
2.2	7
Efficiency of Oil Production from Four Major Oilseed Crops	
2.3	8
Structure of Oil Palm Fruit	
2.4	9
Flow Chart of Palm and Palm Kernel Oil Processing	
2.5	19
Molecular Structure of A Triacylglycerol Molecule	
2.6	22
Alcoholysis Reaction between Triacylglycerol and Alcohol	
2.7	22
Acidolysis Reaction between Triacylglycerol and Fatty Acid	
2.8	23
Transesterification Reaction between Triacylglycerol Molecules	
2.9	24
Transesterification between 1,3-distearoyl-2-oleoylglycerol and 1,3-dipalmitoyl-2-oleoylglycerol catalyzed by Nonspecific Lipases or Chemical Catalyst	
2.10	27
Enolate Mechanism: (A) Formation of Enolate Anion, (B) Formation of Glycelorate Anion	
2.11	31
The Catalytic Mechanism for Enzymatic Interesterification	
2.12	41
Dry Multiple Fractionation of Palm Oil	
3.1	52
The triunsaturated (U_3), Diunsaturated-monosaturated (U_2S), Monounsaturated-disaturated (S_2U), Trisaturated (S_3) Triacylglycerols and By-product Yield Throughout the 72 Hours of Enzymatic Directed Interesterification of Palm Oil Produced in the Presence of 10% Lipozyme TLIM Lipase at 30 °C	
3.2	56
Perturbation Graph Showing the Single Factor Effect on Triunsaturated (U_3) Triacylglycerols (TAGs) Yield	
3.3	58
Perturbation Graph Showing the Single Factor Effect on By-Products Yield	
3.4	59
Contour and Response Surface Plot on the Effect of Interaction between Enzyme Load (w/w %) and Reaction Time (Hour) at Reaction Temperature of 30 °C on Triunsaturated Triglycerides (U_3 TAGs) Yield	
3.5	61
The Contour and Response Surface Plot of the Effect of Interaction between Reaction Temperature (°C) and Reaction Time (Hour) at Enzyme Load of 10% on Triunsaturated Triglycerides (U_3 TAGs) Yield	
3.6	62
The Contour and Response Surface Plot on the Effect of Interaction between Enzyme Load (w/w %) and Reaction Temperature (°C) at Reaction Time of 18 Hours on Triunsaturated Triglycerides (U_3 TAGs) Yield	

3.7	The contour and Response Surface Plot on the Effect of Interaction between Reaction Temperature (°C) and Reaction Time (Hour) at 10% Enzyme Load on the By-Product Yield	63
3.8	The Contour and Response Surface Plot on the Effect of Interaction between Enzyme Load (w/w %) and Reaction Time (Hour) at Reaction Temperature of 30 °C on the By-Product Yield	64
3.9	The Contour and Response Surface Plot on the Effect of Interaction between Reaction Temperature (°C) and Enzyme Load (w/w %) at Reaction Time of 18 Hours on the By-Product Yield	65
3.10	Contour and Response Surface Plot of the Effect of Reaction Time and Reaction Temperature at 10% Enzyme Load on the Triunsaturated Triacylglycerol (U ₃ TAGs) and By-Product Yield	67
4.1	Triacylglycerol Chromatogram of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil	76
4.2	Differential Scanning Calorimetry Melting Thermogram of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil	77
4.3	Differential Scanning Calorimetry Crystallization Thermogram of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil	81
4.4	Solid Fat Content Profile of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil	85
4.5	X-ray Diffraction Crystallography of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil at Storage and Determination Temperature of 5 °C	89
4.6	X-ray Diffraction Crystallography of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil at Storage and Determination Temperature of 10 °C	90
4.7	X-ray Diffraction Crystallography of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil at Storage and Determination Temperature of 15 °C	91
4.8	X-ray Diffraction Crystallography of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil at Storage and Determination Temperature of 20 °C	92

4.9	X-ray Diffraction Crystallography of Enzymatic Directed Interesterified (EDIE PO) and Refined, Bleached and Deodorized (RBD PO) Palm Oil at Storage and Determination Temperature of 25 °C	93
5.1	Effect of Fractionation Temperature on Fatty Acid Composition of Olein Fractions of Enzymatic Directed Interesterified (EDIE POo) and Refined, Bleached and Deodorized (RBD POo) Palm Oil	99
5.2	Effect of Fractionation Temperature on Fatty Acid Composition of Palm Stearin Fractions of Enzymatic Directed Interesterified (EDIE POs) and Refined, Bleached and Deodorized (RBD POs) Palm Oil	101
5.3	Effect of Fractionation Temperature on Triacylglycerol (TAG) Composition of Olein Fractions of Enzymatic Directed Interesterified (EDIE POo) and Refined, Bleached and Deodorized (RBD POo) Palm Oil	104
5.4	Triacylglycerol Chromatogram of Olein Fraction of Enzymatic Directed Interesterified (EDIE POo) and Refined, Bleached and Deodorized (RBD POo) Palm Oil Fractionated at 5 °C	105
5.5	Triacylglycerol Chromatogram of Palm Stearin Fraction of Enzymatic Directed Interesterified (EDIE POs) and Refined, Bleached and Deodorized (RBD POs) Palm Oil Fractionated at 5 and 25 °C	108
5.6	Effect of Fractionation Temperature on Triacylglycerol Composition of Palm Stearin Fractions of Enzymatic Directed Interesterified (EDIE POs) and Refined, Bleached and Deodorized (RBD POs) Palm Oil	109
5.7	Differential Scanning Calorimetry Melting Thermogram of Olein Fractions of Enzymatic Directed Interesterified (EDIE POo) and Refined Bleached Deodorized (RBD POo) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	110
5.8	Differential Scanning Calorimetry Melting Thermogram of Stearin Fractions of Enzymatic Directed Interesterified (EDIE POs) and Refined Bleached Deodorized (RBD POs) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	117
5.9	Differential Scanning Calorimetry Crystallization Thermogram of Olein Fractions of Enzymatic Directed Interesterified (EDIE POo) and Refined Bleached Deodorized (RBD POo) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	120
5.10	Differential Scanning Calorimetry Crystallization Thermogram of Stearin Fractions of Enzymatic Directed Interesterified (EDIE POs) and Refined Bleached Deodorized (RBD POs) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	124

5.11	Solid Fat Content Profile of Olein Fractions of Enzymatic Directed Interesterified Palm Oil (EDIE POo) and Refined Bleached Deodorised (RBD POo) Palm Oil Fractionated at (A) 5 °C, (B) 10 °C, (C) 15 °C, (D) 20 °C and (E) 25 °C	128
5.12	Effect of Fractionation Temperature on Solid Fat Content at 5 °C (A), 10 °C (B), 15 °C (C), 20 °C (D) and 25 °C (E) of Olein Fractions of Enzymatic Directed Interesterified (EDIE POo) and Refined Bleached Deodorized (RBD POo) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	129
5.13	Solid Fat Content Profile of Stearin Fractions of Enzymatic Directed Interesterified Palm Oil (EDIE POs) and Refined Bleached and Deodorized (RBD POs) Palm Oil Fractionated at (A) 5 °C, (B) 10 °C, (C) 15 °C, (D) 20 °C and (E) 25 °C	131
5.14	Effect of Fractionation Temperature on Solid Fat Content at 5 °C (A), 10 °C (B), 15 °C (C), 20 °C (D), 25 °C (E), 30 °C (F), 35 °C (G), 40 °C (H), 45 °C (I), 50 °C (J), 55 °C (K) and 60 °C (L) of Stearin Fractions of Enzymatic Directed Interesterified (EDIE POs) and Refined Bleached Deodorized (RBD POs) Palm Oil Fractionated at 5, 10, 15, 20 and 25 °C	133

LIST OF ABBREVIATIONS

$-\Delta H_f$	Total crystallization enthalpy
$-\Delta H_{i^\circ C}$	Partial crystallization enthalpy
$^\circ C$	Degree celcius
%	Percent
ΔH_f	Total melting enthalpy
$\Delta H_{i^\circ C}$	Partial melting enthalpy
1,2-DAGs	1,2- diacylglycerol
2-MAGs	2-monoacylglycerol
2,3-DAGs	2,3-diacylglycerol
ANOVA	Analysis of variance
AOCS	American Oil and Chemist Society
CCD	Central Composite Design
CIE	Chemical interesterification
CPKO	Crude palm kernel oil
CPO	Crude palm oil
CVD	Cardiovascular disease
DHA	Docosahexaenoic acid
DIE	Directed interesterification
DSC	Differential Scanning Calorimetry
EDIE	Enzymatic directed interesterification
EDIE PO	Enzymatic directed interesterified palm oil
EIE	Enzymatic interesterification
ELCFA	Essential long-chain fatty acids
ELSD	Evaporative Light Scattering Detector
EPA	Eicosapentaenoic acid
FA	Fatty acid
FAC	FA composition
FAO	Food and Agriculture Organization
FDA	Food and Drug Administration
FFA	Free fatty acid
g	Gram
HDL	High density lipoprotein

HMFS	Human milk fat substitutes
HPLC	High Performance Liquid Chromatography
IV	Iodine value
LDL	Low density lipoprotein
MAG	Monoacylglycerol
MFA	Medium chain fatty acid
mg	Miligram
mL	Mililitre
mol	Mole
MUFA	Monounsaturated fatty acid
Na ₂ S ₂ O ₃	Sodium thiosulphate
OLL	1-octadecenoate, 2,3- octadecadienoate sn-glycerol
OLO	1,3- octadecenoate, 2- octadecadienoate sn-glycerol
OOO	Triolein
p-NMR	Pulsed Nuclear Magnetic Resonance
PLL	1-hexadecenoate, 2,3- octadecadienoate sn-glycerol
PLO	1- hexadecenoate, 2- octadecadienoate, 3- octadecenoate sn-glycerol
PLP	1- hexadecenoate, 2- octadecadienoate,3- hexadecenoate sn-glycerol
PMF	Palm mid fraction
PMP	1- hexadecenoate, 2-tetradecanoate, 3-hexadecenoate sn-glycerol
PO	Palm oil
POO	1- hexadecenoate,2,3- octadecenoate sn-glycerol
POo	Palm olein
POs	Palm stearin
POP	1,3- hexadecenoate, 3- octadecenoate sn-glycerol
POS	1- hexadecenoate, 2- octadecenoate, 3-octadecanoic sn-glycerol
PPP	Tripalmitin
PPS	1,2- hexadecenoate, 3- octadecanoic sn-glycerol
PUFA	Polyunsaturated fatty acid
RBD	Refined, bleached and deodorized
RSM	Response surface methodology

S ₂ U	Disaturated-monounsaturated triacylglycerol
S ₃	Trisaturated triacylglycerol
SAFA	Saturated fatty acid
SFC	Solid fat content
SMP	Slip melting point
SOO	1- octadecanoic, 2,3 octadecenoate sn-glycerol
TAG	Triacylglycerol
U ₂ S	Diunsaturated-monosaturated triacylglycerol
U ₃	Triunsaturated triacylglycerol
USAFA	Unsaturated fatty acid
v/v	Volume/volume
vs.	Versus
w/v	Weight/volume
α	Alpha
β	Beta
β'	Beta prime

CHAPTER 1

INTRODUCTION

Dietary fat intake is required for optimal health. Dietary fat supplies the energy for most of our life functions. It provides the essential fatty acid (FA) such as linoleic and linolenic acids that cannot be synthesized in the body and the transport vehicle for fat-soluble vitamins A, D, E and K through the bloodstream. Dietary fats are also important for the functioning of nerves and brain, forming steroid and hormone needed to regulate many bodily processes. Choosing the right types of dietary fats to consume is one of the most important factors in maintaining a healthy life (Hulbert, 2005).

Oils high in unsaturated FA (USAFA) especially oils high in monounsaturated FA (MUFA) such as marine oils containing long-chain MUFA and oils high in oleic acid are considered to be healthy dietary oils. Marine oils containing long-chain MUFA are able to decrease cardiovascular disease risk (Yang *et al.*, 2017; Yang *et al.*, 2016). High oleic oils such as canola and olive with oleic acid content around 60% and 70%, respectively, are able to protect against cardiovascular disease by lowering total blood cholesterol levels, increasing the good high density lipoprotein (HDL) cholesterol and reducing the bad low-density lipoprotein (LDL) cholesterol levels (Crouse *et al.*, 2016; Huang and Sumpio, 2008; Teres *et al.*, 2007; Farinelli *et al.*, 2005). High oleic oils have a protective effect against LDL cholesterol oxidation, which plays an important role in atherogenesis (Teres *et al.*, 2007; Bonanome *et al.*, 1992; Reaven *et al.*, 1991). Artificial *trans* fats (or *trans* FAs) created in partial hydrogenation process to convert liquid vegetable oils to more solid oil is the bad fat. *Trans* FAs raise bad (LDL) cholesterol levels, lower good (HDL) cholesterol levels, and associated with a higher risk of developing heart disease, stroke, cancer, and type II diabetes (Iqbal, 2014).

Palm oil (PO) and its liquid fraction palm olein (POo) are considerably high in oleic acid. PO contains about 37.4-44.1% of oleic acid whereas the regular POo which has iodine value (IV) of 56-59 and super POo with IV of 60-67 have approximately 39.8-43.9% and 43.2-49.2% oleic acid, respectively (MPOB, 2009). Despite having significantly lower oleic acid than the commonly known high oleic oils like olive and canola oils, PO and POo have been reported to have a comparable effect on the blood cholesterol levels. Numerous reputable nutrition studies have evidenced that PO and POo diet significantly increases the level of HDL cholesterol, and decreases the total blood cholesterol and LDL cholesterol levels in human subject which have normal blood cholesterol level (Zhang *et al.*, 1997a; Choudhury *et al.*, 1995; Ghafoorunissa *et al.*, 1995; Sundram *et al.*, 1995; Khosla and Hayes 1994; Khosla and Hayes 1992; Sundram *et al.*, 1992; Truswell *et al.*, 1992; Hayes *et al.*, 1991; Marzuki *et al.*, 1991). The non-raising cholesterol effect of PO might be due to the influence of the amount of saturated FAs (SAFA) at the *sn*-2 position of the triacylglycerol (TAG) molecules backbone, as reported by Decker (1996). Despite the high palmitic acid content of around 39.2-45.8%, only 13-14% of the *sn*-2

position in PO TAG molecules is occupied by palmitic acid. More than 85% of *sn*-2 position of PO's TAG molecules are occupied by USAFA, mainly oleic acid (Ng *et al.*, 1992; Padley *et al.*, 1986; Berger, 1983). According to Kubow (1996) and Small (1991), the absorption of FAs dependent not only on its chain length and degree of saturation but also on its positional distribution in the glycerol backbone. The gastric and pancreatic lipases hydrolyze FAs at the *sn*-1 and *sn*-3 positions of dietary TAGs to produce free FAs (FFA) and 2-monoacylglycerol (MAG) (Carriere *et al.*, 1993; Small 1991). The FAs at the *sn*-2 position of dietary TAGs are preferentially absorbed through the intestinal wall while the FAs esterified at the *sn*-1 and *sn*-3 position, especially long-chain SAFAs such as palmitic acid are not well absorbed (Innis *et al.*, 1997; Pronczuk *et al.*, 1994). The FA at *sn*-1 and *sn*-3 position, especially the long chain SAFA are not absorbed because of their melting points that is substantially above body temperatures, causing them to have a strong tendency to form insoluble soaps with divalent cations, such as calcium and magnesium which in turn is excreted in the feces, reducing their absorption by animals (Innis *et al.*, 1997). Since the majority of FA attached at the *sn*-2 position of PO and POo is of oleic acid, most of this oleic acid will be absorbed in the gut and as such PO and POo can behave like a high oleic oil despite their high SAFA content.

The quest for an ideal oil has always been a longstanding concern for the food industry. An ideal oil should have excellent functional properties and yet should not pose adverse health effect to human upon consumption. Over the years, deliberations over healthy oil and beneficial FAs have resulted in recommendations for a diet high in USAFA, i.e., high in MUFA and polyunsaturated FAs (PUFA), and low in SAFA. This declaration by some means had tarnished the image of PO and POo as a good and versatile food oil as despite their high content of MUFA, they also have high content of SAFA. Therefore, it is desirable to be able to convert PO and POo into products that are high in USAFA especially oleic acid and low in SAFA to improve the nutritive and functional values as well to counter the negative perceptions of PO and POo as unhealthy oil having limited functional properties due to their SAFA content.

Oleic acid enhancement of PO and POo could be made possible by various means. Through plant breeding, a new variety of oil palm known as PS12 could produce oil palm fruits that could yield PO having oleic acid content of more than 48% (Isa *et al.*, 2006). Fractionation of POo IV56 produced super POo which has about 50% oleic acid, with IV of more than 62 (MPOB, 2009). Additionally, enzymatic acidolysis of POo IV56 with oleic acid produced POo with the oleic acid content of about 56% (Saw and Siew, 2009). Chemical interesterification (CIE) of POo IV62 with FA methyl ester high in oleic acid (methyl oleate) followed by fractionation produced POo with 57% oleic acid (Ramli *et al.*, 2009). Enzymatic interesterification (EIE) of POo IV62 followed by fractionation yielded POo having oleic acid content of up to 67.0% (Siew and Saw, 2009).

Directed interesterification (DIE) is a catalytic process which redistributes FAs on the glycerol backbone of TAG molecules of oil and fats. DIE is carried out at a temperature lower than the melting temperature of the highest melting TAGs of the

oils and fats. The reaction is initialized by melting the oils and fats prior to addition of a catalyst to expedite the reaction. Subsequently, the DIE temperature is reduced to a level that is low enough for the higher melting TAGs, normally trisaturated (S_3) TAGs, to crystallize out as they are formed. The high melting TAGs are then withdrawn from the reaction system (liquid phase) as once the crystallized TAGs are in the solid phase they do not take part in the DIE reaction. Theoretically, the reaction will continue to redistribute the residual FAs and thereby continuously form higher melting TAGs to re-establish the equilibrium. As such, continuous crystallization caused the reaction mixture to become less saturated. The DIE eventually results in a reaction mixture containing two main types of TAGs, which are S_3 and triunsaturated (U_3) TAGs (Rousseau and Marangoni, 2008). Through DIE, it is possible to convert a liquid oil into a plastic product with the consistency of a margarine or shortening, as reported in United States Patent 4,482,576 (Boot *et al.*, 1984) and United States Patent 4,419,291 (Lathauwer *et al.*, 1983). DIE of lard and tallow significantly increased the level of their S_3 and U_3 TAGs, in simultaneous with the reduction in diunsaturated-monosaturated (U_2S) and disaturated-monounsaturated (S_2U) TAGs (MacKenzie and Stevenson, 2000; Chobanov and Topalova, 1978). In addition, DIE of PO in the presence of sodium-potassium alloy as a catalyst at 30 °C in a nitrogen atmosphere resulted in POo having an IV of more than 95 and softening point below -8 °C (Lago and Hartman, 1986). Separation or fractionation of the DIE lipids resulted in a soft or liquid fraction high in U_3 TAGs and a solid fraction high in S_3 TAGs.

Fractional crystallization or fractionation is one of the most important techniques for oils and fats modification. It is a fully reversible modification process. Fractionation is a thermo-mechanical separation process in which oils and fats especially those containing TAGs with a range of melting points, are physically separated into two or more fractions of distinct physical and chemical properties (Kellens *et al.*, 2007; Harris, 2005; Sreenivasan, 1978). Fractionation principally involves the partial crystallization of TAG components followed by physical separation of the solid crystallized higher-melting TAGs from the liquid phase containing the low-melting TAGs (Deffense, 1985). It works based on differences in solubility of the solid TAGs in the liquid phase, which further depends on their molecular weight and degree of unsaturation of the TAGs (Gibon, 2006). Three types of fractionation techniques are available, namely the dry fractionation, solvent fractionation and detergent fractionation. Dry fractionation is the most widely used as it is the simplest, cheapest and safest process (Kellen *et al.*, 2007; Harris, 2005).

The main objective of the current study was to produce a high USAFA especially MUFA or oleic acid palm fraction having a USAFA level of more than 70%, MUFA or oleic acid level of more than 60% and SAFA level of preferably less than 30%. The high USAFA or high MUFA palm fraction was produced *via* a combination of enzymatic directed interesterification (EDIE) of PO followed by fractionation of the EDIE PO. Both the steps above will result in the increase of USAFA and MUFA or oleic acid content in PO. As a solid palm fraction concentrated with S_3 TAGs and highly saturated was the by-product of the fractionation process, it was also the objective of this project to obtain the highly saturated palm fraction from the

fractionation of the EDIE PO. Thus, the specific objectives of the present study were as follows:

1. To determine the optimum processing parameters of EDIE of PO to produce EDIE PO containing a high concentration of U₃ TAG, preferably the OOO TAG.
2. To investigate the changes in the physicochemical characteristics of the PO after EDIE process.
3. To fractionate the EDIE PO and subsequently study the physicochemical characteristics of the PO_o and PO_s fractions of the EDIE PO.

This thesis is presented in the following lay out:-

Chapter 1 presents the introduction to subjects associated with highly unsaturated oil and potency to produce high oleic palm fraction using modification process namely EDIE and fractionation, problem statement and the objectives of research along with this outline.

Chapter 2 covers the literature review on the topics related to this study.

Chapter 3 discusses on determination of the optimum processing parameters of EDIE of PO using Response Surface Methodology (RSM) to produce EDIE PO containing a high concentration of U₃ TAG, preferably the OOO TAG.

Chapter 4 discusses the changes in the physicochemical characteristics of PO following EDIE.

Chapter 5 discusses the physicochemical characteristics of PO_o and PO_s fractions of the EDIE PO. This chapter also discusses the effect of fractionation temperature characteristics of the PO_o and PO_s fractions of the EDIE PO.

Chapter 6 presents summary of the discussion, conclusion and recommendation for future work.

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