

UNIVERSITI PUTRA MALAYSIA

GRAPHENE OXIDE/ACRYLONITRILE BUTADIENE RUBBER NANOCOMPOSITES AND THEIR PHYSICOCHEMICAL PROPERTIES

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NASSER ABD-ULLAH MOHAMMED

Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfillment of the Requirements for the Degree of Doctor of Philosophy

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DEDICATION

بسم الله الرحمن الرحيم

"عَلَمَ الإنسانَ ما لَم يَعْلَمُ (العلق) (Taught man that which he knew not (alalak-Quran"

Upon Allah we are rely, every challenging work needs self-efforts as well as guidance of elders especially those who were very close to our heart. Whose affection, love, encouragement and prays of day and night make me able to get such success and honor and the reason of what I become today.

My humble effort I dedicate to my sweet and loving

Father and Mother To my sisters, my brothers, my wife, my children, and to all who taught me in my life. I am really grateful to all of you. You have been my inspiration, and my soul mates. Nasser, 2017

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By

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Studies on the problems pertaining to the rubber industry include looking for ways to address the problem of early deformation mainly due to the weakness in the mechanical and thermal properties of the rubber. Therefore, seeking for alternative fillers (carbonic filler) with desired characteristics such as graphite derivatives is significant for improving the rubber products performance. It is also worth mentioning that in the National Graphene Action Plan 2020, Malaysia plans to invest in graphene for the rubber industry in addition to the other various industrial applications, thus this work is timely.

However, there are some challenges related to the homogeneity of the dispersion of graphene derivatives into dry rubber. Therefore, investigating the compatibility of the solvents to reduce agglomeration of fillers in the rubber matrix is important. Also, gaining substantial improvement in the crosslinking density of the rubber structure and thermal stability are strongly desirable to enhance the rubber performance.

In this study, graphene oxide (GO) was the candidate filler (prepared by the Hummer's method), exfoliated in aqueous solution. The selected matrix was acrylonitrile butadiene rubber (NBR) which is using for producing various parts in the automobile industry. Acetone was the suitable organic solvent for NBR dissolution and compatible with GO suspension. Unvulcanised GO/NBR nanocomposite was prepared as a first objective to verify the level of distribution GO sheets into NBR. Another aim was to investigate the crosslinking formation before vulcanisation treatment. The results showed that tensile strength was increased to 81.2% compared to the unfilled NBR. The improvement is attributed to the ability of

GO networks to restrict the mobility of the NBR molecular chains before vulcanisation treatment. This result could be used effectively in the applications of thermal adhesives.

Vulcanised GO/NBR nanocomposite was prepared by combining effective techniques as a new and facile method. The influence of GO on vulcanised GO/NBR nanocomposite properties with different filler contents of GO (0.2 to 2.4 phr - part per hundred rubbers) was investigated as the second objective. The results showed improvement in the mechanical and thermal properties at 1.2 phr of GO compared to unfilled vulcanised NBR.

For comparison, a commercial filler of carbon black (CB) reinforced vulcanised NBR was fabricated in order to study the possibility of replacing it with GO/NBR nanocomposite. Graphene nanoplatelets (GNP) were also used to prepare vulcanised GNP/NBR nanocomposite for another comparison of using a different type of graphite derivative. The comparisons as the third objective were performed at 1.2 phr of GO (the percolation threshold) without chemical functionalisation. The morphology analysis, identification of functional groups, and cure characteristics of the different nanocomposites were studied.

Based on the filer contents effect, the tensile strength was increased significantly at 1.2 phr in the vulcanised GO/NBR, up to ~149% compared to the unfilled vulcanised NBR. Moreover, based on the filler type, the improvement in GO/NBR nanocomposite was better than those of the vulcanised CB/NBR and GNP/NBR nanocomposites of about 69% and 29.5% respectively. These results were confirmed by the enhancement in the crosslinking density of 38%, 17% and 29% respectively. However, the thermal stability and the glass transition temperatures have the same levels of improvement.

Finally, it is concluded that the desired characteristics of GO and the high level of GO dispersion in a polar elastomer such as NBR matrix, have essential roles in the enhancement of the GO/NBR nanocomposite properties.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

NANOKOMPOSIT GRAFENA OKSIDA / AKRILONITRIL BUTADIENA GETAH DAN SIFAT KIMIK FIZIK MEREKA

Oleh

NASSER ABD-ULLAH MOHAMMED

Disember 2017

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Kajian mengenai masalah berkaitan dengan industri getah termasuklah mencari cara untuk meningkatkan prestasi produk getah. Pencarian pengisi alternatif dengan ciri yang dikehendaki seperti grafit menggalakkan penggunaannya dalam matriks getah pada kandungan yang rendah. Oleh itu, dalam Pelan Tindakan Grafene Nasional 2020, Malaysia merancang untuk melabur di dalam grafit untuk industri getah dan digunakan dalam pelbagai aplikasi industri.

Walan bagaimanapun, terdapat beberapa cabaran yang berkaitan dengan homogenisasi pengagihan graphena derivatif ke dalam getah kering. Oleh itu, pencocokan yang baik dari pelarut memberikan kurang pengumpulan pengisi ke dalam matriks getah adalah penting pada peringkat pertama.

Dalam kajian ini, grafena oksida (GO) telah disediakan menggunakan kaedah Hummer dan dikelupas menggunakan teknik pensonikan ultra dalam larutan akueus. Kemudian, ia disebarkan kedalm getah akrilonitril butadiena (NBR). Pemilihan pelarut yang sesuai telah dipertimbangkan semasa menggunakan kaedah pencampuran larutan. Seterusnya, kajian sifat nanokomposit GO/ NBR sebelum rawatan pemvulkanan telah dilakukan sebagai tujuan pertama. Juga, nanokomposit GO/NBR tervulkan disediakan dengan menggabungkan teknik yang berkesan. Kemudian, sifat mekanikal dan untuk lima kandungan (0.2 hingga 2.4) bgs (bahagian getah seratus) diselidiki sebagai objektif kedua. Sebagai perbandingan, karbon hitam (CB) yang diperkuat NBR tervulkan dibikin yang akan menggantikan GO / NBR pada masa hadapan kerana hasil buangan CB dan ciri-cirinya yang rendah. Tambahan pula, grafena nanoplatelet (GNP) digunakan untuk menyediakan nanokomposit GNP / NBR tervulkan bagi membandingkannya sebagai derivatif grafit yang lain. Perbandingan telah dilakukan pada 1.2 bgs GO (kandungan ambang perkolasi) dengan tiada fungsi kimia. Penilaian ini termasuk analisis morfologi, memperkenalkan kumpulan berfungsi, dan ciri-ciri reologi.

Keputusan GO / NBR tidak tervulkan menunjukkan bahawa tegasan tegangan meningkat kepada 81.2% dan modulus pada kawasan anjal meningkat kepada 238% berbanding NBR yang tak diisi. Peningkatan ini disebabkan oleh keupayaan rangkaian GO untuk menyekat pergerakan rantaian molekul NBR tanpa rawatan pemvulkanan. Hasilnya penting untuk mengesahkan keseragaman penyebaran GO dalam NBR yang boleh digunakan dengan berkesan dalam aplikasi perekat.

Berdasarkan kesan kandungan pengisi, kekuatan tegangan meningkat dengan ketara pada 1.2 bgs paut GO kepada GO / NBR yang lebih kepadatan ~149% daripada NBR tervulkan yang tidak terisi. Selain itu, berdasarkan jenis pengisi peningkatan nanokomposit GO/NBR kenaikan itu lebih baik daripada nanokomposit CB / NBR dan GNP / NBR yang lebih mempunyai 69% dan 29.5%masing-masing. Keputusan ini disahkan oleh nilai pengurangan nisbah bengkak dan peningkatan dalam kepadatan paut silang kepada masing-masing 38%, 17% dan 29%. Kestabilan haba dan suhu peralihan kaca juga mempunyai tahap peningkatan yang sama.

Akhirnya, kekonduksian elektrik GO sangat rendah berbanding pengisi yang lain.

Akhirhya, dapat disimpulkan itu dari hasil-hasil ini bahawa ciri-ciri GO yang baik dan yang diperlukan tinggi serta keseragaman penyebaran GO dalam matriks NBR mempunyai peranan penting dalam meningkatkan sifat nanokomposit GO / NBR Penambahbaikan.

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Nasser, December 2017, Malaysia.

I certify that a Thesis Examination Committee has met on 15 December 2017 to conduct the final examination of Nasser Abd-Ullah Mohammed on his thesis entitled "Graphene Oxide/Acrylonitrile Butadiene Rubber Nanocomposites and their Physicochemical Properties" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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LIST OF ABBREVIATIONS

| | ASTM | American Society for Testing and Material |
|--|---------|--|
| | BR | Butadiene Rubber |
| | CB | Carbon Black |
| | CB/NBR | Carbon black reinforced nitrile butadiene rubber nanocomposite |
| | CNFs | Carbon NanoFillers |
| | CNTs | Carbon NanoTubes |
| | CR | Chloroprene Rubber |
| | DMA | Dynamic Mechanical Analysis |
| | DSC | Differential Scanning Calorimetry |
| | DTG | Differential Thermal Gravimetric |
| | EG | Expanded Graphitic |
| | EPDM | Ethylene Propylene Diene Monomer |
| | FEF | Fast Extrusion Furnace |
| | FTIR | Fourier - Transform Infrared Spectroscopy |
| | GIA | Global Industry Analysts |
| | GNP | Graphite Nanoplatelets |
| | GNP/NBR | Graphene nanoplatelets reinforced acrylonitrile butadiene rubber nanocomposite |
| | GO | Graphene Oxide |
| | GO/NBR | Graphene oxide reinforced acrylonitrile butadiene rubber nanocomposite |
| | GR | Graphite |
| | HAF | High abrasion furnace |
| | HXNBR | Hydrogenated carboxylated nitrile butadiene rubber |
| | ISO | International Standards Organisation |

- MWCNTs Multiwall carbon nanotubes
- NBR Nitrile Butadiene Rubber
- NR Natural Rubber
- PNCs Polymer Nanocomposites
- PU Polyurethane rubber
- RGO Reduced graphene oxide
- RPM Revolution per minutes
- SBR Styrene Butadiene Rubber
- SEM Scanning Electron Microscopy
- SiR, Silicon Rubber
- SWCNTs Single wall carbon nanotubes
- TEGO Thermally exfoliated graphene oxide
- TEM Transmission electron microscopy
- TGA Thermogravimetric Analysis
- VPR Vinyl Pyridine Rubber
- XNBR Carboxylated nitrile rubber
- XRD X-Ray Diffraction

CHAPTER 1

INTRODUCTION

1.1 Background of Study

The technological classification of polymers divides them into three main types; thermoplastic, thermoset, and elastomer (rubber). The first type can be formed by heating (melting) and the second type can be prepared by adding chemicals such as hardeners to the polymer (Carraher Jr, 2010). However, rubbers have a complex preparation process compared to other polymers in which their curing is performed chemically and physically. They have divided into thermoplastic and thermoset elastomers also. Numerous polymers have been developed including ones with high mechanical, thermal and other physical properties when they are prepared in blends or composite forms(Sperling, 2005; Visakh et al., 2013).

Polymeric composites are prepared by the dispersion of the reinforcement materials, such as fibres or particulates of various kinds of fillers, into the polymer matrix. In the 1940s, polymeric composites were used for military applications and in varied fields such as construction and automobiles, so they became alternative materials. This is due to their desired combination characteristics involving the polymer matrix and the filler (Bergmann & Andrade, 2011; Carraher Jr, 2010; McWilliams, 2006). Generally, the type, size, and the aspect ratio of the filler affect the properties of the polymer composites. Therefore, any improvement represents the results of the interaction between the polymer's matrices and the filler's properties at the interfacial region. Consequently, different types of composites have been developed by modifying the structures and the technical preparation (Das et al., 2010; Deepalekshmi et al., 2013; Ponnamma et al., 2013).

A new class of polymeric composite materials, called polymer nanocomposites (PNCs), has been developed; their reinforcement phases have dimensions in the nanometer scale. The start of PNCs was in 1990 when Toyota used nano clay reinforced Nylon-6 in some parts of their automobiles, to enhance the mechanical, thermal and barrier properties (Nguyen & Baird, 2006). Since then, interest has increased in both the academic and industrial circles in developing polymer nanocomposites with different matrices and diverse types of nanofillers (Mensah et al., 2014). Nano-scale fillers with a range of 1-100nm at least in one dimension, such as layered silicates, carbon nanotubes (CNTs), carbon nanofillers (CNFs), and exfoliated graphite derivatives, etc., are dispersed in most polymers types including elastomers (Deepalekshmi et al., 2013). Based on their dimensional morphology, nanofillers can have iso-dimensional elongated or particle shape. When only one dimension is in the nanometer range, then the layered silicates and layered graphite flakes (Kodal & Ozkoc, 2013).



The main advantage of using a nanofiller in conventional composites is the light weight (low contents). Also, the large surface area contact increases the interface of the polymeric matrix with the nanofiller. Thus, it enhances the mechanical strength and the thermal stability, in addition to the other physical characteristics (Kang et al., 2014; Preetha et al., 2013). Reinforcement of nanocomposite elastomers by a nanofiller depends on the interfacial adhesion between the elastomer and the nanoparticles. Hence, homogenous distribution of the nanofiller in the polymer matrix is one of the challenges that determine the optimal properties of nanocomposites. However, the disadvantages caused by using nanoparticles are the increase in viscosity, dispersion difficulties (Agglomerations) and some environmental impacts (Kodal & Ozkoc, 2013; Zeng et al., 2015).

Early research on polymer nanocomposites concerned the development of nano clay filled thermoplastic and thermoset systems. Then, research expanded into elastomers in which the nanofillers used were those such as nano-silica and carbon. Currently, the rubber nanocomposite is being developed in the area of nanomaterials as it has low weight and is an alternative material with high properties compared to the neat elastomer. Therefore, nanocomposite rubber finds application in automobiles, aerospace, coatings, electronics, biotechnology, and healthcare (Deepalekshmi et al., 2013). The last report by Global Industry Analysts (GIA) mentioned that the global market for nanocomposites was projected to reach 1.5 billion pounds by the year 2017. In Malaysia, the National Graphene Action Plan 2020, (Rahim, 2016) reported that Malaysia has taken important steps in the use of graphene in nanomaterials to develop the industry and support employment opportunities via collaboration with local universities and developed countries.

In rubber nanocomposites, fillers such as clays and carbon black are used to improve the rubber properties for extensive applications. The most common application is tyres and parts of the automobile industry in which numerous researchers were and still are attempting to increase the strength, abrasion resistance and thermal resistance with low weight and processing. Carbonic and inorganic fillers with the vulcanization materials, such as zinc oxide, sulfur, and other standards fillers etc., are modifying the material to enhance its mechanical and thermal properties. The literature on rubber nanocomposites has shown that the incorporation of nano clay in rubbers, both natural rubber (NR) and synthetic rubbers such as styrene butadiene rubber (SBR), nitrile rubber (NBR), chloroprene rubber (CR) and ethylene propylene diene monomer (EPDM) rubber etc., enhances the mechanical and other properties (Deepalekshmi et al., 2013; Kodal & Ozkoc, 2013). Still, a few works on rubbers based on using nanofillers such as nano silicate, (CNTs) and graphite have been carried out in recent years (Bergmann & Andrade, 2011; Thomas & Stephen, 2010).

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Also, there are several methods and modified techniques for preparing elastomer nanocomposites such as latex blending, solution mixing, direct mixing and in-situ polymerisation (Kang et al., 2014; Wu et al., 2013). These methodologies have their own advantages and limitations; these will be explained later. Most of the reported literature on elastomer nanocomposites involves using solution mixing techniques,

either using latex rubber or dissolution dry rubbers in suitable solvents(Choi et al., 2005; Kang et al., 2014; Mensah et al., 2015). The process is followed by evaporating the solvent, and mastication with the vulcanisation additives. Finally, rheological curing and standard moulding conditions should be performed to obtain the nanocomposite rubber as in most previous works (Bai et al., 2011; Tarantili, 2013).

An important aspect of elastomeric nanocomposites development is to achieve a good grouping of properties; the performance of the polymeric composites depends on factors including formulation, the dispersion homogeneity of the filler in the matrix, the adhesion between the filler and the polymeric matrix and degree of crosslinking. Moreover, thermal stability and glass transition under a range of temperatures exchange influences with the mechanical strength in rubber nanocomposites. Some studies noted in the literature have reported on the mechanical, electrical, thermal and other physicochemical properties of NBR reinforced by nanoclay (Balachandran & Bhagawan, 2012; Dick, 2014). Although the effects of nano clay content and treatment on the properties of various rubbers have been investigated, studies on other types of nanofiller such as graphite derivatives are still under research in the academic field and research centers. Also, various challenges need to be addressed, such as the dispersion level of the nanoparticles in the elastomers, optimise the characteristics and the availability of nano graphitic fillers in bulk at low cost (Malas & Das, 2015; Mao et al., 2013).

1.2 Problem Statement

Rubber is an important material in various industrial applications but it doesn't have the ability to withstand high mechanical and thermal effects without reinforcement. Therefore, it has been strengthened by many types of available and necessary fillers to improve its properties. The technological reinforcement of rubbers involves overcoming obstacles of the rubber composites preparation by using effective and compatible fillers. However, various problems such as mechanical deformation and thermal decomposition tend to emerged after a period of time.

The characteristics of rubber also play an important role in selecting it for use in various applications with acceptable performance. For example, synthetic rubber such as acrylonitrile Butadiene Rubber (NBR) is considered as a backbone in rubber used in the automobile industry. However, there are some challenges related to improving its resistance to the swelling due to the solvents and oils. Also, although the vulcanisation treatment is used with some fillers for enhancing NBR structure, the mechanical properties are still affected by the types of filler (Deepalekshmi et al., 2013).

Fillers such as carbon black (CB), silica and other clays are still the main fillers in rubber composites and used with relatively high weight percentages. Currently, CB faces several challenges since it is derived from crude oil and generates excessive

wastes. Also, the functional groups on its surfaces, such as phenol, carboxylic acids, and lactones showed incomplete interaction with the rubber matrix. Moreover, the tendency of CB particles to aggregate is mostly expected, which reduces its solidity in terms of low crosslinking density in the elastomer chains. Therefore, finding new oleo chemical independent fillers, such as graphite derivatives, have received increasing attention for the replacement of CB (Mao et al., 2013; Sperling, 2005). This is due to their high characteristics and performance as reported (Li et al., 2012). Furthermore, their thicknesses, which are less than 10 nm, make them an extremely efficient structural reinforcement in the elastomers.

Other carbon-based nanofillers such as carbon nanotubes CNTs and expanded graphitic (EG) derivatives in nanocomposite rubber exhibit outstanding properties at low loading of the filler compared with CB and unfilled rubber compounds (Bai et al., 2011; Xie et al., 2005). Some research studied the characteristics of graphite derivatives and their influences on the polymers generally and elastomers properties in specific.

Graphene Oxide (GO) is a promising nanomaterial that is commonly prepared by the chemical oxidation of graphite. However, there are some problems facing the preparation process, such as the agglomeration of the GO sheets in the rubber matrix. The agglomerates can be one of the causes of the micro-defects in the rubber composites, or they reduce the cross-linking between the NBR chains. This is attributed to the incomplete exfoliation of the GO sheets, also due to the heterogeneous dispersion of GO in the rubber nanocomposite.

It is worth mentioning that the direct mixing technique does not provide a homogeneous dispersion of nanofiller into the rubber matrix as reported by several researches (Fryczkowski et al., 2013; Gudarzi & Sharif, 2012; She et al., 2014; Wang; et al., 2013). Therefore, dissolution of dry NBR in a suitable solvent with compatible viscosity makes the solution mixing method more effective method. The further critical issue should be investigated, is the compatibility between the GO suspension and the rubber solution.

Even though Graphite Nanoplatelets (GNP) is distinct graphitic nanofiller with high surface area, its characteristics is dissimilar compared to GO and CB. In this case, the comparison among the properties of GO/NBR, GNP/NBR and CB/NBR nanocomposites are required. This investigation could provide necessary information about the differences in the properties between two different groups of carbonic fillers.

Regarding the thermal properties, early thermal decomposition is a common problem facing rubber in high-temperature environments in hoses and seals applications. Therefore, dispersion of GO in the GO/NBR nanocomposites could improve the thermal stability. Accordingly, enhancement of the cross-linking density can increase the glass transition temperature T_g at low specific contents having an essential

influence on the storage modulus and the weight residue for the rubber nanocomposite, compared with the unfilled NBR (Malas & Das, 2015; Wang et al., 2012).

This research is an important contribution for studying the possibility of using graphite derivatives in the rubber industry. Thus, it has not yet been used commercially, because it is still under academic and research work. Based on the recent reports, Malaysia has an interest in developing the technology of using graphite derivatives in the industry and in the rubber manufacturing in specific (Ali et al., 2017). This is one of the main justifications to set this proposal.

1.3 Hypothesis of the Study

Graphene possesses high physical and chemical characteristics. This encourages the study of improving the mechanical and thermal properties of rubber composites reinforced by one of the graphite derivatives. Thus, to search the possibility of replacing carbon black was hypothesised as in the following;

A few amount of graphene oxide (between 0.2 to 1.2 phr) can reinforced effectively a synthetic rubber such as NBR. The improvement in the mechanical properties is attributed to the presence of the functional group of oxygen on the GO surfaces. Also, the increase of the cross-linking of the rubber chains should increase the tensile strength of the GO/NBR nanocomposite.

The large surface area of GO sheets contributes to increase the interfacial interaction between NBR and GO. This characteristic reduces the swelling ratio when the GO/NBR nanocomposite is placed in the oil or solvents medium, thus its enhancing the crosslinking density of the composite.

High thermal capacity of GO and its ability to restrict the rubber chains mobility can increase the thermal stability of the GO/NBR nanocomposite. Furthermore, the increase in the glass transition temperature is attributed to the increment in the storage modulus. This result led to increase the resistance of the rubber nanocomposite to the thermal decomposition.

1.4 Objectives of the Study

The present study was undertaken to explore the potential of GO-reinforced NBR in the preparation of unvulcanised and vulcanised GO/NBR rubber nanocomposites. Also, it aimed to compare the characteristics of the vulcanised GO/NBR nanocomposite with the other carbonic/NBR nanocomposites. The salient objectives of the current research were:

- **1.4.1** To investigate the effectiveness of the dispersion of GO into NBR (dry) via the preparation of an unvulcanised GO/NBR rubber nanocomposite. Also, to characterise unvulcanised GO/NBR nanocomposites and their mechanical properties for different contents of GO.
- **1.4.2** To investigate the influence of GO on the mechanical strength and thermal stability of the vulcanized GO/NBR rubber nanocomposite at different low contents of GO compared to unfilled NBR.
- **1.4.3** To determine the variances in the characteristics and properties between the vulcanised GO/NBR nanocomposite and other rubber nanocomposites of NBR reinforced by CB and GNP at the optimal contents of GO.

1.5 Scope and Limitation of the Study

The scope of this study focuses on investigating the influence of the desired characteristics of GO on the physicochemical properties of GO/NBR nanocomposite. Therefore, unvulcanised and vulcanised GO/NBR rubber nanocomposites were prepared and characterized. This research covers experimental work based on the objectives, so the scope is summarized as follows;

- 1. Unvulcanised GO/NBR nanocomposites were prepared with five different contents of GO. Suitable NBR solution was prepared via selection the compatible solvent based on the ability of dissolution NBR and its viscosity. The parameters of the sonication process were adjusted in order to exfoliate the GO in a compatible aqueous solution. The mixing process and the qualitative of the coagulant mixture were monitored. Finally, the verification of the homogeneity was observed by characterising the GO/NBR nanocomposites using XRD, FTIR and SEM analysis. Also, by testing the tensile strength and the electrical conductivity at different contents of GO. The purpose was to verify the evidence of the functional groups role in the interfacial interaction between the GO sheets and the NBR chains. The expected findings could approve the effectiveness of using the product for coating and adhesives applications as a novel work.
- 2. Vulcanised GO/NBR nanocomposites preparation were performed for five different contents of GO as the second objective. This is for studying the influence of GO on the cure characteristics of the composites compared to unfilled NBR at an optimal temperature and time. The enhancement in the maximum torque and the curing time at 90% of torque were determine at optimal content of GO too. This test was essential to support the crosslinking forming interpretation and the curing time .

Furthermore, analysing the vulcanised GO/NBR nanocomposites crosslinking density by swelling test, the tensile strength properties by tensile stress-strain test, and the storage modulus and loss energy using the DMA tests in the range

of temperatures between -50 to +50 $^{\circ}$ C were performed. The results were supported by the morphology observation analysis of the SEM micrograph for the fractured surfaces of the tensile test samples .

In the same context, the thermal decomposition behaviour of the unfilled NBR and the GO/NBR nanocomposites, were studied based on TGA and DTG analysis. The temperature equivalent to the initial thermal degradation temperature (T onset) at percolation threshold of GO contents was significant for analysing the thermal/mechanical properties. Also, the heat flow capacity provided needed information about T_g using DSC tests at the same five contents of GO in the vulcanised GO/NBR nanocomposites. Finally, from the electrical properties test of this rubber nanocomposite, high electrical resistance of GO in the nanocomposites was approved.

3. Influence of dispersion different types of carbonic filler in NBR matrix was determined in this study. Therefore, two different groups of filler of carbon black (CB)-N550 and graphene nanoplatelets (GNP)-C750 were used. The variances in the characteristics and properties between the vulcanised GO/NBR nanocomposite (which was mentioned above at 2) and other rubber nanocomposites of CB/NBR and GNP/NBR at the optimal contents of GO were studied.

In this part of the scope, as a novel work, the study involved analysing and comparing the mechanical and thermal properties in order to determine the variances in the enhancement due to using different fillers. Morphology, nonfunctionalisation, cure characteristics, and swelling ratio reduction of the graphitic rubber nanocomposites provide essential indications. This is concerning the expected improvement in the tensile strength, elasticity modulus, storage modulus, and the hardness property better than the carbon black base rubber composite.

Correspondently, their thermal properties were influenced also by optimal filler contents. The thermal decomposition, stability, and Glass transition temperatures provide further evidence of the level of the filler dispersion (agglomerations and aggregations) in the NBR matrix. Also, analysis can interpret the variances built on the structure, size, shape and the surface interaction of the fillers as concluded .

The electrical conductivity of the rubber nanocomposites was tested for GNP, CB, and GR reinforced NBR. As these fillers were not functionalised, the low impedance against the electrical charges in the rubber composites is expected.

The limitations of the study are to improve the mechanical and thermal properties of vulcanised NBR and reduce the weight. In the conclusion, the study mainly aimed to obtain high performance of rubber parts of automobiles, aerospace, and miscellaneous applications in the industry.

1.6 Outline of the Thesis

The thesis consists of five chapters, and the contents of each chapter are as outlined below:

Chapter 1 includes an introduction which presents a brief background of the field of elastomeric composites materials focusing on filler types and the nanosize effect. Also, the chapter highlights the research problem statements, the hypothesis, the research objectives and the scopes of the thesis.

Chapter 2 reviews the available literature concerning the fundamentals of polymeric composites. Also, information is given about NBR, the types of fillers available in general and graphene more specifically, and the preparation methods. In the characterisation section, the rheological curing and the crosslinking density, in addition to the morphology, are demonstrated in this chapter along with the main physicochemical properties.

Chapter 3 covers the materials which were used and the methodology. Also, the chapter includes GO exfoliation and dispersion into NBR. The preparation steps of unvulcanised (GO/NBR) nanocomposites with different contents of GO are showed. The vulcanization treatment of GO/NBR nanocomposite followed by the cure characteristics is displayed. For comparison, there is the preparation of another two rubber nanocomposites which were CB/NBR and GNP/NBR. The characterisation and properties of the rubber nanocomposites are demonstrated using the techniques of the rheometer, FTIR, XRD, SEM, the swelling test, tensile tests, DMA, hardness tests, TGA-DTG, DSC, and the electrical conductivity using two techniques.

Chapter 4 represents the results and discussions of all the tests including the data, the graphs, and the results. In the first section, the unvulcanised GO/NBR nanocomposite preparation includes the dissolution of NBR and the exfoliation of GO. The dispersion of the GO into the NBR is demonstrated and discussed. Then, there is a verification of the dispersion process by testing the characteristics and properties of the unvulcanised GO/NBR nanocomposite at different contents of GO.

The second section in this chapter is related to a discussion of the effect of the filler content on the vulcanised GO/NBR nanocomposites. The following characteristics are assessed: the cure characteristics, the morphology using (SEM, XRD, FTIR), the crosslinking density, the mechanical properties (tensile strength, storage modulus, loss modulus and loss energy (tan δ)), and the hardness property. Also, the thermal properties are determined, including the thermal stability (TGA-DTG) and the glass transition temperature T_g using DSC tests. In addition, the electrical conductivity property was studied.

The third section is related to the comparison of the previous section in this chapter involving vulcanised GO/NBR nanocomposites, with the other rubber nanocomposites which are filled with GNP and CB.

Chapter 5 includes the conclusions which were achieved and the suggested opportunities for future studies. These involve the essential conclusions of preparation and characterisation of unvulcanised GO/NBR nanocomposites. Also, the vulcanised GO/NBR nanocomposite characteristics and properties are displayed. Then, the main variances between the properties of the vulcanised GO/NBR, GNP/NBR, and CB/NBR nanocomposites are listed. Finally, future works and recommendations are suggested.



REFERENCES

- Abdullah, Shivan Ismael, & Ansari, MNM. (2015). Mechanical properties of graphene oxide (GO)/epoxy composites. *HBRC Journal*, 11(2), 151-156.
- Ali, Murni, Burtt, Chester, Neto, Antonio H Castro, Cho, Nam Sung, Correia, Antonio, Ferrari, Andrea, . . . Huang, Kuan-Tsae. (2017). CONTRIBUTIONS-INDEX. *INDEX*, 32.
- Araby, Sherif, Meng, Qingshi, Zhang, Liqun, Kang, Hailan, Majewski, Peter, Tang, Youhong, & Ma, Jun. (2014). Electrically and thermally conductive elastomer/graphene nanocomposites by solution mixing. *Polymer*, 55(1), 201-210.
- ASTM, D. (1765). 06 "Standard Classification System for Carbon Blacks Used in Rubber Products.
- ASTM, D. (1997). 1646-96a-" Standard Test Methods for Rubber-Viscosity, Stress Relaxation, and Pre-vulcanization Characteristics (Mooney Viscometer),". *American Society for Testing & Materials, Annual Book of ASTM Standards,* 9, 313-322.
- ASTM D1646-15, "Standard Test Methods for Rubber-Viscosity, Stress Relaxation, and Pre-vulcanization Characteristics (Mooney Viscometer)", American Society for Testing & Materials, United States, 2015.
- ASTM D5668-14, "Standard Test Methods for Rubber From Synthetic Sources-Volatile Matter", American Society for Testing & Materials, United States, 2014.
- ASTM D5667-15, "Standard Test Methods for Rubber From Synthetic Sources-Total and Water Soluble Ash", American Society for Testing & Materials, United States, 2015.
- ASTM D1765-06, "Standard Classification System for Carbon Blacks Used in Rubber Products", American Society for Testing & Materials, United States, 2015.
- ASTM D1513-05, "Standard Classification System for Carbon Blacks Pelleted-Pour Density", American Society for Testing & Materials, United States, 2017.
- ASTM D3187-06, "Standard Test Methods for Rubber-Evalution of NBR (Acrylonitrile-Butadiene Rubber)", American Society for Testing & Materials, United States, 2016.
- ASTM D3182-16, "Standard Practice for Rubber-Materials, Equipment, and Procedures for Mixing Standard Compound and Preparing Standard Vulcanized Sheets", American Society for Testing & Materials, United States, 2016.

- ASTM D2084-17, "Standard Test Method for Rubber Property-Vulcanization Using Oscillating Disk Cure Meter", American Society for Testing & Materials, United States, 2017
- ASTM D471-06, "Standard Test Method for Rubber Property-Effect of Liquids", American Society for Testing & Materials, United States, 2006
- ASTM D7028-07, "Standard Test Method for Method for Glass Transition Temperature (DMA-Tg) of Polymer Composites by DMA", American Society for Testing & Materials, United States, 2015
- ASTM D2240-15e1, "Standard Test Method for Rubber Property—Durometer Hardness", ASTM International, West Conshohocken, PA, 2015.
- ASTM D6370-99(2014), Standard Test Method for Rubber—Compositional Analysis by Thermogravimetry (TGA), ASTM International, West Conshohocken, PA, 2014.
- ASTM D257-14, Standard Test Methods for DC Resistance or Conductance of Insulating Materials, ASTM International, West Conshohocken, PA, 2014.
- Bai, Xin, Wan, Chaoying, Zhang, Yong, & Zhai, Yinghao. (2011). Reinforcement of hydrogenated carboxylated nitrile–butadiene rubber with exfoliated graphene oxide. *Carbon*, 49(5), 1608-1613.
- Balachandran, Meera, & Bhagawan, SS. (2011). Studies on acrylonitrile-butadiene copolymer (NBR) layered silicate composites: mechanical and viscoelastic properties. *Journal of Composite Materials*, 0021998311399484.
- Balachandran, Meera, & Bhagawan, SS. (2012). Mechanical, thermal and transport properties of nitrile rubber (NBR)—nanoclay composites. *Journal of Polymer Research, 19*(2), 9809.
- Behabtu, Natnael, Dan, Budhadipta, Martinez, Angel, Evans, Julian S, Kosynkin, Dmitry V, Tour, James M, . . . Smalyukh, Ivan I. (2011). Liquid crystals of aqueous, giant graphene oxide flakes. *Soft Matter*, 7(23), 11154-11159.
- Bergmann, Carlos Pérez, & Andrade, Mônica Jung. (2011). Nanostructured materials for engineering applications: Springer.
- Beyler, Craig L, & Hirschler, Marcelo M. (2002). Thermal decomposition of polymers. *SFPE handbook of fire protection engineering*, *2*, 111-131.
- Bhowmick, AK, Bhattacharya, M, Mitra, S, Kumar, K Dinesh, Maji, PK, Choudhury, A, . . . Basak, GC. (2010). Morphology–property relationship in rubber-based nanocomposites: some recent developments *Advanced Rubber Composites* (pp. 1-83): Springer.
- Bishop, Chris. (2001). The relationship between loss, conductivity, and dielectric constant. *Advanced Engineering Electromagnetics*.

- Bo, Zheng, Shuai, Xiaorui, Mao, Shun, Yang, Huachao, Qian, Jiajing, Chen, Junhong, . . . Cen, Kefa. (2014). Green preparation of reduced graphene oxide for sensing and energy storage applications. *Scientific reports*, *4*, 4684.
- Bokobza, Liliane. (2007). Multiwall carbon nanotube elastomeric composites: a review. *Polymer*, 48(17), 4907-4920.
- Bourlinos, Athanasios B, Georgakilas, Vasilios, Zboril, Radek, Steriotis, Theodore A, Stubos, Athanasios K, & Trapalis, Christos. (2009). Aqueous-phase exfoliation of graphite in the presence of polyvinylpyrrolidone for the production of water-soluble graphenes. *Solid State Communications*, 149(47), 2172-2176.
- Brandrup, J, & Immergut, EH. (1974). Handbook of Polymer Chemistry: Wiley: New York.
- Carraher Jr, Charles E. (2010). *Carraher's polymer chemistry* (J. J. Lagowski Ed. 6th ed.). NY, USA: Marcel Dekker.
- Chandrasekaran, Swetha, Seidel, Christian, & Schulte, Karl. (2013). Preparation and characterization of graphite nano-platelet (GNP)/epoxy nano-composite: mechanical, electrical and thermal properties. *European Polymer Journal*, 49(12), 3878-3888.
- Chen, Ma, Nan, Bai, Xin, Zhang, Hongmei, & Zhang, Yong. (2012). Effects of graphene oxide on surface energy, mechanical, damping and thermal properties of ethylene-propylene-diene rubber/petroleum resin blends. *Rsc Advances*, 2(11), 4683-4689.
- Chen, Yao, Bowen, Li, Chun, & Shi, Gaoquan. (2013). An improved Hummers method for eco-friendly synthesis of graphene oxide. *Carbon, 64*, 225-229.
- Child, Michael. (2014). Industrial-scale hydrothermal carbonization of waste sludge materials for fuel production.
- Choi, Dongcheol, Kader, M Abdul, Cho, Baik- Hwan, Huh, Yang- il, & Nah, Changwoon. (2005). Vulcanization kinetics of nitrile rubber/layered clay nanocomposites. *Journal of applied polymer science*, *98*(4), 1688-1696.
- Chung, Deborah DL. (2010). *Composite materials: science and applications:* Springer Science & Business Media.
- Corcione, Carola Esposito, & Frigione, Mariaenrica. (2012). Characterization of nanocomposites by thermal analysis. *Materials*, *5*(12), 2960-2980.
- Das, Amit, Kasaliwal, Gaurav R, Jurk, René, Boldt, Regine, Fischer, Dieter, Stöckelhuber, Klaus Werner, & Heinrich, Gert. (2012). Rubber composites based on graphene nanoplatelets, expanded graphite, carbon nanotubes and their combination: a comparative study. *Composites Science and Technology*, 72(16), 1961-1967.

- Das, Amit, Wang, De-Yi, Stöckelhuber, Klaus Werner, Jurk, René, Fritzsche, Juliane, Klüppel, Manfred, & Heinrich, Gert. (2010). Rubber–clay nanocomposites: some recent results *Advanced Rubber Composites* (pp. 85-166): Springer.
- Daud, Noraniza Ahmad, Chieng, Buong Woei, Ibrahim, Nor Azowa, & Talib, Zainal Abidin. (2017). Synthesis and Characterisation of Functionalised-Graphene Oxide by Gamma-Ray Irradiation. *Journal of Engineering Science*, 13, 1-17.
- Deepalekshmi, P, Visakh, PM, Mathew, Aji P, Chandra, Arup K, & Thomas, Sabu. (2013). Advances in Elastomers: Their Composites and Nanocomposites: State of Art, New Challenges and Opportunities *Advances in Elastomers II* (pp. 1-9): Springer.
- Deshmukh, Kalim, Khatake, SM, & Joshi, Girish M. (2013). Surface properties of graphene oxide reinforced polyvinyl chloride nanocomposites. *Journal of Polymer Research*, 20(11), 286.
- Dick, John S. (2014). *Rubber technology: compounding and testing for performance:* Carl Hanser Verlag GmbH Co KG.
- Dogadkin, BA, Dontsov, AA, & Shershnev, VA. (1981). Chemistry of elastomers. *Moscow, Chimia*.
- Doty, Paul, & Zable, Helen S. (1946). Determination of polymer–liquid interaction by swelling measurements. *Journal of Polymer Science*, 1(2), 90-101.
- El-Sabbagh, SH, & Yehia, AA. (2007). Detection of crosslink density by different methods for natural rubber blended with SBR and NBR. *Egyptian Journal of Solids*, 30(2), 157-173.
- Etmimi, Hussein M, & Sanderson, Ronald D. (2011). New approach to the synthesis of exfoliated polymer/graphite nanocomposites by miniemulsion polymerization using functionalized graphene. *Macromolecules*, 44(21), 8504-8515.
- Facio, Adali Castañeda, Galindo, Aide Saenz, Cepeda, Lorena Farias, López, Lluvia López, & de León-Gómez, Ramón Díaz. (2015). Thermal Degradation of Synthetic Rubber Nanocomposites *Thermal Degradation of Polymer Blends*, *Composites and Nanocomposites* (pp. 157-191): Springer.
- Flory, Paul J. (1950). Statistical mechanics of swelling of network structures. *The Journal of Chemical Physics*, 18(1), 108-111.
- Fryczkowski, Ryszard, Gorczowska, Marta, Ślusarczyk, Czesław, Fryczkowska, Beata, & Janicki, Jarosław. (2013). The possibility of obtaining graphene/polymer composites from graphene oxide by a one step process. *Composites Science and Technology*, 80, 87-92.

- Fu, DH, Zhan, YH, Yan, N, & Xia, HS. (2015). A comparative investigation on strain induced crystallization for graphene and carbon nanotubes filled natural rubber composites. *Express Polymer Letters*, 9(7).
- Galpaya, Dilini, Wang, Mingchao, Liu, Meinan, Motta, Nunzio, Waclawik, Eric, & Yan, Cheng. (2012). Recent advances in fabrication and characterization of graphene-polymer nanocomposites.
- Geng, Yan, Wang, Shu Jun, & Kim, Jang-Kyo. (2009). Preparation of graphite nanoplatelets and graphene sheets. *Journal of colloid and interface science*, 336(2), 592-598.
- Gibson, Ronald F. (1994). Principles of composite material mechanicsMcGraw-Hill. *New York*.
- Govindjee, Sanjay, & Simo, Juan. (1992). Transition from micro-mechanics to computationally efficient phenomenology: carbon black filled rubbers incorporating Mullins' effect. *Journal of the Mechanics and Physics of Solids*, 40(1), 213-233.
- Green, Maya, Marom, Gad, Li, Jing, & Kim, Jang- Kyo. (2008). The electrical conductivity of graphite nanoplatelet filled conjugated polyacrylonitrile. *Macromolecular rapid communications*, 29(14), 1254-1258.
- Gudarzi, M Moazzami, & Sharif, F. (2012). Enhancement of dispersion and bonding of graphene-polymer through wet transfer of functionalized graphene oxide. *Express Polymer Letters*, 6(12), 1017-1031.
- Guo, Hui-Lin, Wang, Xian-Fei, Qian, Qing-Yun, Wang, Feng-Bin, & Xia, Xing-Hua. (2009). A green approach to the synthesis of graphene nanosheets. *ACS nano*, *3*(9), 2653-2659.
- Heinrich, Gert. (2011). Advanced rubber composites (Vol. 239): Springer Science & Business Media.
- Hernández, Marianella, del Mar Bernal, María, Verdejo, Raquel, Ezquerra, Tiberio
 A, & López-Manchado, Miguel A. (2012). Overall performance of natural rubber/graphene nanocomposites. *Composites Science and Technology*, 73, 40-46.
- Huang, Lou, Ching-Wen, Liu, Chi-Fan, Huang, Chen-Hung, Song, Xiao-Min, & Lin, Jia-Horng. (2015). Polypropylene/graphene and polypropylene/carbon fiber conductive composites: Mechanical, crystallization and electromagnetic properties. *Applied Sciences*, 5(4), 1196-1210.
- Huang, Yin, Zongyou, Wu, Shixin, Qi, Xiaoying, He, Qiyuan, Zhang, Qichun, . . . Zhang, Hua. (2011). Graphene- based materials: synthesis, characterization, properties, and applications. *small*, *7*(14), 1876-1902.

- Humers, WS, & Offeman, RE. (1958). Preparation of graphitic oxide [J]. J Am Chem Soc, 80(6), 1339.
- Hussain, Farzana, Hojjati, Mehdi, Okamoto, Masami, & Gorga, Russell E. (2006). Review article: polymer-matrix nanocomposites, processing, manufacturing, and application: an overview. *Journal of composite materials, 40*(17), 1511-1575.
- Ignatz-Hoover, F, To, BH, & Rodgers, B. (2004). Rubber compounding: chemistry and applications. *Rubber Compounding: Chemistry and Applications*.
- Ismail, MN, & Khalaf, AI. (2011). Styrene–butadiene rubber/graphite powder composites: rheometrical, physicomechanical, and morphological properties. *Journal of Applied Polymer Science, 120*(1), 298-304.
- Jang, Suk-Hee, Kim, Wook-Soo, Kang, Yong-Gu, Han, Min-Hyun, & Chang, Sang-Mok. (2013). Study on Mixing Condition of the Rubber Composite Containing Functionalized S-SBR, Silica and Silane: I. Effect of Mixing Temperature. *Elastomers and Composites*, 48(2), 94-102.
- Janowska, G, Rybiński, P, & Jantas, R. (2006). Effect of the modification of silica on thermal properties and flammability of cross-linked butadiene-acrylonitrile rubbers. *Journal of thermal analysis and calorimetry*, 87(2), 511-517.
- Jeffrey, R Potts. (2012). Preparation, Properties, and Structure-Property Relationships of Graphene-Polymer Nanocomposites. (PhD), The University of Texas.
- Jovanović, Vojislav, Budinski-Simendić, Jaroslava, Samardžija-Jovanović, Suzana, Marković, Gordana, & Marinović-Cincović, Milena. (2009). The influence of carbon black on curing kinetics and thermal aging of acrylonitrile-butadiene rubber. *Chemical Industry and Chemical Engineering Quarterly/CICEQ*, 15(4), 283-289.
- Kang, Hailan, Zuo, Kanghua, Wang, Zhao, Zhang, Liqun, Liu, Li, & Guo, Baochun.
 (2014). Using a green method to develop graphene oxide/elastomers nanocomposites with combination of high barrier and mechanical performance. *Composites Science and Technology*, 92, 1-8.
- Khademeh, Fateme, Soltani, Sedigheh, Naderi, Ghasem, & Bagheri, Rohollah. (2016). Effect of multi-walled carbon nanotube on mechanical and rheological properties of silane modified EPDM rubber. *Polyolefins Journal*, 3(2), 69-77.
- Khimi, S Raa, & Pickering, Kim L. (2014). A new method to predict optimum cure time of rubber compound using dynamic mechanical analysis. *Journal of Applied Polymer Science*, 131(6).
- Kim, Abdala, Ahmed A, & Macosko, Christopher W. (2010). Graphene/polymer nanocomposites. *Macromolecules*, 43(16), 6515-6530.

- Kim, Chang, Hwan, Pal, Kaushik, & Sridhar, V. (2011). Role of Different Nanoparticles in Elastomeric Nanocomposites *Recent Advances in Elastomeric Nanocomposites* (pp. 3-55): Springer.
- Knite, Maris, & Linarts, Artis. (2015). Polymer/Nanographite Composites for Mechanical Impact Sensing Graphene-Based Polymer Nanocomposites in Electronics (pp. 223-252): Springer.
- Kodal, Mehmet, & Ozkoc, Guralp. (2013). Micro and nanofillers in rubbers Advances in Elastomers I (pp. 303-356): Springer.
- Konios, Dimitrios, Stylianakis, Minas M, Stratakis, Emmanuel, & Kymakis, Emmanuel. (2014). Dispersion behaviour of graphene oxide and reduced graphene oxide. *Journal of colloid and interface science, 430*, 108-112.
- Kuila, Tapas, Bose, Saswata, Mishra, Ananta Kumar, Khanra, Partha, Kim, Nam Hoon, & Lee, Joong Hee. (2012). Chemical functionalization of graphene and its applications. *Progress in Materials Science*, 57(7), 1061-1105.
- Kuilla, Tapas, Bhadra, Sambhu, Yao, Dahu, Kim, Nam Hoon, Bose, Saswata, & Lee, Joong Hee. (2010). Recent advances in graphene based polymer composites. *Progress in polymer science*, *35*(11), 1350-1375.
- Kumar, Annamalai Pratheep, Depan, Dilip, Tomer, Namrata Singh, & Singh, Raj Pal. (2009). Nanoscale particles for polymer degradation and stabilization trends and future perspectives. *Progress in Polymer Science*, 34(6), 479-515.
- Lawandy, SN, Halim, SF, & Darwish, NA. (2009). Structure aggregation of carbon black in ethylene-propylene diene polymer. *Express Polymer Letters*, 3(3), 152-158.
- Le, HH, Hoang, XT, Das, A, Gohs, U, Stoeckelhuber, K-W, Boldt, R, . . . Radusch, H-J. (2012). Kinetics of filler wetting and dispersion in carbon nanotube/rubber composites. *Carbon*, *50*(12), 4543-4556.
- Li, Umer, Rehan, Samad, Yarjan Abdul, Zheng, Lianxi, & Liao, Kin. (2013). The effect of the ultrasonication pre-treatment of graphene oxide (GO) on the mechanical properties of GO/polyvinyl alcohol composites. *Carbon, 55*, 321-327.
- Li, Wang, Qihua, Wang, Tingmei, & Pan, Guangqin. (2012). Preparation and tribological properties of graphene oxide/nitrile rubber nanocomposites. *Journal of Materials Science*, 47(2), 730-738.
- Li, Zhang, J, & Chen, SJ. (2008). Effects of carbon blacks with various structures on vulcanization and reinforcement of filled ethylene-propylene-diene rubber. *Express Polym. Lett, 2*(10), 695-704.
- Li, & Zhong, Wei-Hong. (2011). Review on polymer/graphite nanoplatelet nanocomposites. *Journal of materials science*, 46(17), 5595-5614.

- Liu, Becerril, Héctor A, Mao, Jie, Stoltenberg, Randall M, Bao, Zhenan, & Chen, Yongsheng. (2008). Evaluation of solution-processed reduced graphene oxide films as transparent conductors. ACS nano, 2(3), 463-470.
- Liu, Kuang, Wenyi, & Guo, Baochun. (2015). Preparation of rubber/graphene oxide composites with in-situ interfacial design. *Polymer*, *56*, 553-562.
- Mahajan, GV, & Aher, VS. (2012). Composite material: A review over current development and automotive application. *International Journal of Scientific and Research Publications*, 2(11), 1-5.
- Malas, Asish, & Das, Chapal Kumar. (2015). Effect of graphene oxide on the physical, mechanical and thermo-mechanical properties of neoprene and chlorosulfonated polyethylene vulcanizates. *Composites Part B: Engineering*, 79, 639-648.
- Manna, Rakesh, & Srivastava, Suneel Kumar. (2017). Fabrication of functionalized graphene filled carboxylated nitrile rubber nanocomposites as flexible dielectric materials. *Materials Chemistry Frontiers*.
- Mao, Yingyan, Wen, Shipeng, Chen, Yulong, Zhang, Fazhong, Panine, Pierre, Chan, Tung W, . . . Liu, Li. (2013). High performance graphene oxide based rubber composites. *Scientific reports*, *3*, 2508.
- Mazrouaa, Azza M. (2012). Polypropylene nanocomposites: INTECH Open Access Publisher.
- McWilliams, Andrew. (2006). Nanocomposites, nanoparticles, nanoclays, and nanotubes. *NANO21C*, *BCC Research*, *Norwalk*, *CT*.
- Menard, Kevin P. (2008). *Dynamic mechanical analysis: a practical introduction:* CRC press.
- Mensah, Bismark, Kim, Sungjin, Arepalli, Sivaram, & Nah, Changwoon. (2014). A study of graphene oxide- reinforced rubber nanocomposite. *Journal of Applied Polymer Science*, 131(16).
- Mensah, Bismark, Kumar, Dinesh, Lim, Dong- Kwon, Kim, Seung Gyeom, Jeong, Byeong- Heon, & Nah, Changwoon. (2015). Preparation and properties of acrylonitrile–butadiene rubber–graphene nanocomposites. *Journal of Applied Polymer Science, 132*(36).
- Molavi, Fatemeh Khademeh, Bagheri, Rohollah, Naderi, Ghasem, & Soltani, Sedigheh. (2013). Compatibilization of Multi-Wall Carbon Nanotube/EPDM: Studies on the Properties of Nanocomposite. *Science and Technology*, *26*(2), 115-123.

- Moly, KA, Bhagawan, SS, Groeninckx, Gabriël, & Thomas, S. (2006). Correlation between the morphology and dynamic mechanical properties of ethylene vinyl acetate/linear low- density polyethylene blends: Effects of the blend ratio and compatibilization. *Journal of applied polymer science*, 100(6), 4526-4538.
- Morsy, RM, Ismaiel, MN, & Yehia, AA. (2013). Conductivity studies on acrylonitrile butadiene rubber loaded with different types of carbon blacks. *Int J Mater Methods Technol, 1*(4), 22-35.
- Mulla, SM, Phale, PS, & Saraf, MR. (2012). Use of X-Ray Diffraction Technique for Polymer Characterization and Studying the Effect of Optical Accessories. *ARAI, AdMet paper NOM, 6.*
- Nah, Changwoon, Kim, Seung Gyeom, Shibulal, Gopi Sathi, Yoo, Yong Hwan, Mensah, Bismark, Jeong, Byeong-Heon, . . . Ahn, Jou-Hyeon. (2015). Effects of curing systems on the mechanical and chemical ageing resistance properties of gasket compounds based on ethylene-propylene-dienetermonomer rubber in a simulated fuel cell environment. *International Journal of Hydrogen Energy*, 40(33), 10627-10635.
- Nah, Changwoon, Lim, Jee Young, Cho, Baik Hwan, Hong, Chang Kook, & Gent, Alan N. (2010). Reinforcing rubber with carbon nanotubes. *Journal of applied polymer science*, 118(3), 1574-1581.
- Nguyen, Quang T, & Baird, Donald G. (2006). Preparation of polymer-clay nanocomposites and their properties. *Advances in Polymer Technology*, 25(4), 270-285.
- Novoselov, KS, Ferrari, Andrea C, Meyer, JC, Scardaci, V, Casiraghi, C, Lazzeri, Michele, . . . Roth, S. (2006). Raman spectrum of graphene and graphene layers. *Physical review letters*, 97(18), 187401.
- Pang, Huan, Xu, Ling, Yan, Ding-Xiang, & Li, Zhong-Ming. (2014). Conductive polymer composites with segregated structures. *Progress in Polymer Science*, *39*(11), 1908-1933.
- Papageorgiou, Dimitrios G, Kinloch, Ian A, & Young, Robert J. (2015). Graphene/elastomer nanocomposites. *Carbon*, 95, 460-484.
- Peters, Stanley T. (2013). *Handbook of composites*: Springer Science & Business Media.
- Pingot, Tomasz, Pingot, Martyna, & Zaborski, Marian. (2011). The Effect of Chemical Modification on Mechanical Properties of Carbon Black Filled Elastomer *Trends in Colloid and Interface Science XXIV* (pp. 143-146): Springer.

- Ponnamma, Deepalekshmi, Maria, Hanna J, Chandra, Arup K, & Thomas, Sabu. (2013). Rubber nanocomposites: latest trends and concepts Advances in elastomers II (pp. 69-107): Springer.
- Potts, Jeffrey R, Dreyer, Daniel R, Bielawski, Christopher W, & Ruoff, Rodney S. (2011). Graphene-based polymer nanocomposites. *Polymer*, 52(1), 5-25.
- Preetha, Rani, Joseph, & Mathiazhagan, A. (2013). *Studies on the use of Nanokaolin, MWCNT and Graphene in NBR and SBR*. (phD), Cochin University of Science and Technology.
- Prud'Homme, Robert, Ozbas, Bulent, Aksay, Ilhan, Register, Richard, & Adamson, Douglas. (2010). Functional graphene-rubber nanocomposites: Google Patents.
- Rahim, Ahmad Kh., Abdul. (2016). Speaker profile. Organisers 21 sponsors 24 Exhibitors 26, 6.
- Ramanathan, T, Abdala, AA, Stankovich, S, Dikin, DA, Herrera-Alonso, M, Piner, RD, . . . Ruoff, RS. (2008). Functionalized graphene sheets for polymer nanocomposites. *Nature nanotechnology*, 3(6), 327-331.
- Rodgers, Brendan, & Waddell, Walter. (2005). The Science of Rubber Compounding-9.
- Roldughin, VI, & Vysotskii, VV. (2000). Percolation properties of metal-filled polymer films, structure and mechanisms of conductivity. *Progress in Organic Coatings*, 39(2), 81-100.
- Sadasivuni, Kishor Kumar, Ponnamma, Deepalekshmi, Thomas, Sabu, & Grohens, Yves. (2014). Evolution from graphite to graphene elastomer composites. *Progress in Polymer Science*, 39(4), 749-780.
- Saeb, Mohammad Reza, Sarami, Raha, Iranpak, Bijan, & Gaffari, Ramtin. (2013). Tensile Properties of Natural Rubber Nanocomposites Affected by Crosslink Density. *Advances in Materials Physics and Chemistry*, 2(04), 260.
- Sangwichien, Chayanoot, Sumanatrakool, Panita, & Patarapaiboolchai, Orasa. (2008). Effect of filler loading on curing characteristics and mechanical properties of thermoplastic vulcanizate. *Chiang Mai J. Sci*, *35*(1), 141-149.
- Shanmugharaj, AM, Bae, JH, Lee, Kwang Yong, Noh, Woo Hyun, Lee, Se Hyoung, & Ryu, Sung Hun. (2007). Physical and chemical characteristics of multiwalled carbon nanotubes functionalized with aminosilane and its influence on the properties of natural rubber composites. *Composites Science and technology*, 67(9), 1813-1822.

- She, Xiaodong, He, Canzhong, Peng, Zheng, & Kong, Lingxue. (2014). Molecularlevel dispersion of graphene into epoxidized natural rubber: Morphology, interfacial interaction and mechanical reinforcement. *Polymer*, 55(26), 6803-6810.
- Shen, Jing- Wei, Chen, Xiao- Mei, & Huang, Wen- Yi. (2003). Structure and electrical properties of grafted polypropylene/graphite nanocomposites prepared by solution intercalation. *Journal of applied polymer science*, 88(7), 1864-1869.
- Singh, Virendra, Joung, Daeha, Zhai, Lei, Das, Soumen, Khondaker, Saiful I, & Seal, Sudipta. (2011). Graphene based materials: past, present and future. *Progress in materials science*, 56(8), 1178-1271.
- Sobhy, Maged S, El-Nashar, DE, & Maziad, Nabila A. (2003). Cure characteristics and physicomechanical properties of calcium carbonate reinforcement rubber composites. *Egypt. J. Sol, 26*(02), 241-257.
- Song, Sung Ho, Jeong, Ho Kyun, Kang, Yong Gu, & Cho, Chun Taek. (2010). Physical and thermal properties of acid-graphite/styrene-butadiene-rubber nanocomposites. *Korean Journal of Chemical Engineering*, 27(4), 1296-1300.
- Sperling, Leslie H. (2005). *Introduction to physical polymer science*: John Wiley & Sons.
- Stankovich, Sasha, Dikin, Dmitriy A, Piner, Richard D, Kohlhaas, Kevin A, Kleinhammes, Alfred, Jia, Yuanyuan, . . . Ruoff, Rodney S. (2007). Synthesis of graphene-based nanosheets via chemical reduction of exfoliated graphite oxide. *carbon*, 45(7), 1558-1565.
- Stephen, Ranimol, Alex, Rosamma, Cherian, Treesa, Varghese, Siby, Joseph, Kuruvilla, & Thomas, Sabu. (2006). Rheological behavior of nanocomposites of natural rubber and carboxylated styrene butadiene rubber latices and their blends. *Journal of applied polymer science*, 101(4), 2355-2362.
- Strankowski, Michał, Włodarczyk, Damian, Piszczyk, Łukasz, & Strankowska, Justyna. (2016). Polyurethane nanocomposites containing reduced graphene oxide, FTIR, raman, and XRD studies. *Journal of Spectroscopy, 2016*.
- Tang, Gongqing, Jiang, Zhi-Guo, Li, Xiaofeng, Zhang, Hao-Bin, Hong, Song, & Yu, Zhong-Zhen. (2014). Electrically conductive rubbery epoxy/diaminefunctionalized graphene nanocomposites with improved mechanical properties. *Composites Part B: Engineering*, 67, 564-570.
- Tarantili, Petroula A. (2013). Reinforced elastomers: Interphase modification and compatibilization in rubber-based nanocomposites *Advances in Elastomers II* (pp. 109-154): Springer.

- Thammathadanukul, Valeeporn, O'Haver, John H, Harwell, Jeffrey H, Osuwan, Somchai, Na- Ranong, Nuchanat, & Waddell, Walter H. (1996). Comparison of rubber reinforcement using various surface- modified precipitated silicas. *Journal of applied polymer science*, *59*(11), 1741-1750.
- Thomas, Sabu, & Stephen, Ranimol. (2010). *Rubber nanocomposites: preparation, properties and applications:* John Wiley & Sons.
- Tian, Ming, Zhang, Jing, Zhang, Liqun, Liu, Suting, Zan, Xiaoqing, Nishi, Toshio, & Ning, Nanying. (2014). Graphene encapsulated rubber latex composites with high dielectric constant, low dielectric loss and low percolation threshold. *Journal of colloid and interface science*, 430, 249-256.
- Turku, Irina, & Kärki, Timo. (2014). Research progress in wood-plastic nanocomposites: A review. Journal of Thermoplastic Composite Materials, 27(2), 180-204.
- Valentini, L, Bon, S Bittolo, Lopez-Manchado, MA, Verdejo, R, Pappalardo, L, Bolognini, A, . . . Pugno, NM. (2016). Synergistic effect of graphene nanoplatelets and carbon black in multifunctional EPDM nanocomposites. *Composites Science and Technology*, 128, 123-130.
- Varghese, Tony V, Kumar, H Ajith, Anitha, S, Ratheesh, S, Rajeev, RS, & Rao, V Lakshmana. (2013). Reinforcement of acrylonitrile butadiene rubber using pristine few layer graphene and its hybrid fillers. *Carbon, 61*, 476-486.
- Verge, Pierre, Peeterbroeck, Sophie, Bonnaud, Leila, & Dubois, Philippe. (2010). Investigation on the dispersion of carbon nanotubes in nitrile butadiene rubber: Role of polymer-to-filler grafting reaction. *Composites Science and Technology*, 70(10), 1453-1459.
- Visakh, PM, Thomas, Sabu, Chandra, Arup K, & Mathew, Aji P. (2013). Advances in Elastomers: Springer.
- Wang, Ji, Dandan , Sun, Yi, Gong, Xuedong, & Ding, Lifeng. (2013). Enhancements of the mechanical properties and thermal conductivity of carboxylated acrylonitrile butadiene rubber with the addition of graphene oxide. *Journal of Materials Science*, 48(4), 1571-1577.
- Wang, Zhang, Li Qun, & Tian, Ming. (2012). Mechanical and tribological properties of acrylonitrile–butadiene rubber filled with graphite and carbon black. *Materials & Design*, *39*, 450-457.
- Wang;, Nanying, Zhang, Liqun, Lu, Yonglai, Tian, Ming, & Chan, Tung. (2013). Filler dispersion evolution of acrylonitrile–butadiene rubber/graphite nanocomposites during processing. *Composites Part A: Applied Science and Manufacturing*, 47, 135-142.

- Winget, Paul, Dolney, Derek M, Giesen, David J, Cramer, Christopher J, & Truhlar, Donald G. (1999). Minnesota solvent descriptor database. *Dept. of Chemistry* and Supercomputer Inst., University of Minnesota, Minneapolis, MN, 55455.
- Wu, Jinrong, Huang, Guangsu, Li, Hui, Wu, Siduo, Liu, Yufeng, & Zheng, Jing. (2013). Enhanced mechanical and gas barrier properties of rubber nanocomposites with surface functionalized graphene oxide at low content. *Polymer, 54*(7), 1930-1937.
- Xie, Xiao-Lin, Mai, Yiu-Wing, & Zhou, Xing-Ping. (2005). Dispersion and alignment of carbon nanotubes in polymer matrix: a review. *Materials Science and Engineering: R: Reports, 49*(4), 89-112.
- Xiong, Xiaogang, Wang, Jingyi, Jia, Hongbing, Fang, Eryuan, & Ding, Lifeng. (2013). Structure, thermal conductivity, and thermal stability of bromobutyl rubber nanocomposites with ionic liquid modified graphene oxide. *Polymer degradation and stability*, 98(11), 2208-2214.
- Yakovlev, AV, Finaenov, AI, Zabud'kov, SL, & Yakovleva, EV. (2006). Thermally expanded graphite: synthesis, properties, and prospects for use. *Russian journal of applied chemistry*, 79(11), 1741-1751.
- Yang, Tian, Ming, Jia, Qing-Xiu, Shi, Jun-Hong, Zhang, Li-Qun, Lim, Szu-Hui, ... Mai, Yiu-Wing. (2007). Improved mechanical and functional properties of elastomer/graphite nanocomposites prepared by latex compounding. Acta Materialia, 55(18), 6372-6382.
- Yang, Tian, Ming, Jia, Qing- Xiu, Zhang, Li- Qun, & Li, Xiao- Lin. (2006). Influence of graphite particle size and shape on the properties of NBR. *Journal of Applied Polymer Science*, 102(4), 4007-4015.
- Yang, Zhu, Junwu, Qiu, Ling, & Li, Dan. (2011). Bioinspired effective prevention of restacking in multilayered graphene films: towards the next generation of high- performance supercapacitors. *Advanced Materials*, 23(25), 2833-2838.
- Yasmin, Asma, Luo, Jyi-Jiin, & Daniel, Isaac M. (2006). Processing of expanded graphite reinforced polymer nanocomposites. *Composites Science and Technology*, 66(9), 1182-1189.
- Yi, YB, & Tawerghi, E. (2009). Geometric percolation thresholds of interpenetrating plates in three-dimensional space. *Physical Review E*, 79(4), 041134.
- Yin, Biao, Wang, Jingyi, Jia, Hongbing, He, Junkuan, Zhang, Xumin, & Xu, Zhaodong. (2016). Enhanced mechanical properties and thermal conductivity of styrene–butadiene rubber reinforced with polyvinylpyrrolidone-modified graphene oxide. *Journal of Materials Science*, 51(12), 5724-5737.
- Young, Robert J, Kinloch, Ian A, Gong, Lei, & Novoselov, Kostya S. (2012). The mechanics of graphene nanocomposites: a review. *Composites Science and Technology*, 72(12), 1459-1476.

- Yuan, & Misra, RDK. (2006). Polymer nanocomposites: current understanding and issues. *Materials science and technology*, 22(7), 742-755.
- Yuan, Zou, LL, Liao, CC, & Dai, JW. (2012). Improved properties of chemically modified graphene/poly (methyl methacrylate) nanocomposites via a facile in-situ bulk polymerization. *Express Polymer Letters*, 6(10), 847-858.
- Zaeimoedin, TZ, & Yong, KC. (2012). Capillary Rheometer: A Practical Rheological Test Instrument for Both Natural and Synthetic Rubbers. *JOURNAL OF RUBBER RESEARCH, 15*(1), 46-63.
- Zeng, Lu, Shaorong, Xiao, Xiane, Gao, Jian, Pan, Lulu, He, Zihai, & Yu, Jinhong. (2015). Enhanced thermal and mechanical properties of epoxy composites by mixing noncovalently functionalized graphene sheets. *Polymer Bulletin*, 72(3), 453-472.
- Zeng, Yu, AB, Lu, GQ, & Paul, DR. (2005). Clay-based polymer nanocomposites: research and commercial development. *Journal of nanoscience and nanotechnology*, 5(10), 1574-1592.
- Zhu, Murali, Shanthi, Stoller, Meryl D, Velamakanni, Aruna, Piner, Richard D, & Ruoff, Rodney S. (2010). Microwave assisted exfoliation and reduction of graphite oxide for ultracapacitors. *Carbon*, 48(7), 2118-2122.
- Zhu, Uhl, Fawn M, Morgan, Alexander B, & Wilkie, Charles A. (2001). Studies on the mechanism by which the formation of nanocomposites enhances thermal stability. *Chemistry of Materials*, 13(12), 4649-4654.