

UNIVERSITI PUTRA MALAYSIA

MORPHOLOGY AND MAGNETIC PROPERTIES OF COBALT AND COBALT-PLATINUM MAGNETIC NANOPARTICLES PREPARED USING REVERSE MICROEMULSION

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By GHAZALEH BAHMANROKH

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Chairman:Associate Professor Mansor Hashim, PhDFaculty:Science

This thesis is focused on the preparation of cobalt and cobalt-platinum type magnetic nanoparticles by the reverse-micelle microemulsion method for the applications in ultra-high density magnetic recording media that could overcome the thermal stability limit of currently materials. Synthesizing nanoparticles (NPs) with a narrow size distribution and high uniformity is one of the prime goals of this research. To achieve that. nanoparticles were synthesized in aqueous cores of cetyltrimethylammonium bromide (CTAB) reverse micelles as a nano-reactor allowing nucleation and growth within a confined space and gaining the advantage of size restriction. A variation was introduced to the molar ratio of water to surfactant (ω) in preparing Co and CoPt₃ nanoparticles for four different ω :5, 10, 15 and 20. The as-prepared Co and CoPt₃ nanoparticles were annealed further under argon atmosphere at 400°C and 800°C respectively. Moreover since Co nanoparticles are rapidly oxidized upon exposure to air, coating of Co and CoPt₃ nanoparticles with a

gold shell was performed in the next step. Physical characteristics of as-prepared and annealed samples were studied as well.

XRD patterns of as-synthesized samples show the evidence that highly crystalline $CoPt_3$ NPs could form during precipitation in reverse-micelle microemulsion. However as-synthesized and annealed Co NPs showed very low crystallinity. Sintering of $CoPt_3$ caused an improvement in the crystallinity and disordered-ordered phase transition at 800°C, which subsequently changed the magnetic properties of $CoPt_3$. Coercivity increased with an increase of the ω ratio or particle size for both Co and $CoPt_3$ NPs. Annealing resulted in a significant increase in the coercivity which showed the highest values of 446.80e and 712.20e respectively for Co and $CoPt_3$ at room temperature which makes them good candidates for recording media. The TEM and FESEM results showed nanoparticles with spherical shape and narrow size distribution. The average size was in good agreement with the crystal size calculated by the Scherer formula and increased with an increased ω ratio. Hysteresis loops at room temperature for the as-synthesized Co and $CoPt_3$ show

Core-shell Co-Au, CoPt₃-Au NPs were synthesised via a two step reduction process in reverse-micelle microemulsion solution. Gold coating increased the crystallinity of Co NPs but decreased that of CoPt₃. TEM results in high magnification show a coreshell structure with shell thickness around 2-3nm. After gold coating of Co NPs the coercivity decreased whereas the saturation magnetization increased due to oxidation protection. A typical high blocking temperature around 50 K was observed for CoPt₃-Au nanoparticles via a simple method. Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

SIFAT MORFOLOGI DAN SIFAT MAGNET KOBALT DAN NANOPARTIKEL MAGNET KOBALT-PLATINUM YANG DISEDIAKAN DENGAN MENGGUNAKAN BERBALIK MICROEMULSION

Oleh

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Tesis ini tertumpu kepada fabrikasi kobalt dan nanopartikel magnet kobalt-platinum dengan menggunakan teknik mikroemulsi berbalik untuk aplikasi media rakaman berketumpatan ultra-tinggi, yang akan mengatasi had kestabilan terma dalam bahanbahan masa kini. Salah satu matlamat utama penyelidikan ini adalah untuk mensintesis nanopartikel-nanopartikel yang mempunyai taburan saiz yang kecil. Untuk mencapai matlamat tersebut, nanopartikel-nanopartikel telah disintesis di dalam teras akues micelles berbalik *cetyltrimethylammonium bromide* (CTAB) sebagai reactor-nano yang membenarkan nukleasi dan pertumbuhan dalam ruang yang terhad dan mendapat kelebihan daripada sekatan saiz. Kami telah mengubah nisbah molar air terhadap surfaktan (ω) kepada empat ω yang berbeza iaitu ω : 5, 10, 15, dan 20 dalam penyediaan nanopartikel-nanopartikel Co and CoPt₃. Nanopartikelnanopartikel Co dan CoPt₃ yang disediakan kemudiannya telah disepuh lindap di dalam suasana argon dengan masing-masingnya pada suhu 400°C dan 800°C. Selain

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itu, memandangkan nanopartikel-nanopartikel Co cepat teroksida apabila terdedah kepada udara, maka penyaduran petala emas pada Co dan CoPt₃ telah dilakukan pada peringkat seterusnya. Pencirian-pencirian fizikal sampel yang telah disediakan dan yang telah disepuh lindap turut dikaji.

Corak-corak XRD untuk sampel yang disediakan menunjukkan nanopartikelnanopartikel CoPt₃ yang berkristaliniti tinggi boleh dibentuk semasa pemendakan micelle berbalik. Walaubagaimanapun, nanopartikel-nanopartikel Co yang telah disediakan dan yang telah disepuh lindap mempunyai kristaliniti yang sangat rendah. Pensinteran pada suhu 800°C CoPt₃ menunjukkan peningkatan dalam kristaliniti dan fasa transisi tidak tertib kepada tertib yang kemudiannya mengubah sifat-sifat magnet CoPt₃. Daya paksa meningkat dengan peningkatan nisbah ω bagi kedua-dua nanopartikel-nanopartikel Co dan CoPt₃. Sepuh lindap telah meningkatkan daya paksa dan menunjukkan nilai tertinggi pada 446.76 Oe untuk Co dan 712.16 Oe untuk CoPt₃ pada suhu bilik yang membuatkan bahan-bahan ini sesuai untuk digunakan sebagai media rakaman. Hasil kajian melalui TEM dan FESEM menunjukkan bentuk sfera, taburan saiz yang kecil, dan saiz purata selari dengan saiz kristal yang dikira melalui formula Scherer dan meningkat dengan peningkatan nisbah ω . Gelung histerisis pada suhu bilik untuk sampel yang disediakan bagi Co dan CoPt₃ menunjukkan sifat superparamagnet kepada ferromagnet di atas suhu pemblokan pada 45K.

Teras-petala nanopartikel-nanopartikel Co-Au dan $CoPt_3$ -Au telah disediakan dengan proses dua peringkat penurunan dalam larutan micelle berbalik. Penyaduran emas meningkatkan kristaliniti nanopartikel-nanopartikel Co tetapi

walaubagaimanapun telah mengurangkan kristaliniti CoPt₃. Keputusan-keputusan TEM pada pembesaran tinggi menunjukkan struktur teras-petala dengan ketebalan petala sekitar 2-3nm. Selepas penyaduran emas pada nanopartikel-nanopartikel Co, daya paksa berkurangan tetapi kemagnetan tepu meningkat disebabkan perlindungan oksidasi. Suhu pemblokan yang tinggi sekitar 50 K telah diperhatikan bagi nanopartikelCoPt₃-Au melalui teknik yang ringkas.



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I certify that a Thesis Examination Committee has met on 15th February 2012 to conduct the final examination of Ghazaleh Bahmanrokh on her thesis entitled "**Morphology and Magnetic Properties of Cobalt and Cobalt-Platinum Magnetic Nanoparticles Prepared Using Reverse-Micelle Microemulsion**" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Uiversiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master Degree.

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DECLARATION

I declare that the thesis is my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.



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LIST OF ABBREVIATIONS

Å	angstroms
AFM	Antiferromagnet
AOT	Sodium bis(2-ethylhexyl) sulfosuccinate
Co	Cobalt
CoPt ₃	Cobalt-platinum
CTAB	Cetyltrimethylammonium bromide
D	Diameter
DDAB	Diodecyldimethylammonium bromide
DL	Double layer
emu/g	Electromagnetic unit per gram
EDX	Energy dispersive x-ray spectroscopy
FC	Field-cooled
fcc	face-centered cubic
fct	face-centered tetragonal
FM	Ferromagnet
hcp	hexagonal closed packed
g	gram
Gb/in ²	Gigabits (10^9 bits) per square inch
н	Magnetic field strength
H _c	Coercivity
$H_{\rm E}$	Exchange bias filed
nm	nanometer
NPs	nanoparticles
$M_{\rm r}$	Remanent magnetization
Ms	Saturation magnetization

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- OeorstedVVolumeχMagnetic susceptibilityXRDX-ray diffraction
- ZFC Zero field-cooled



CHAPTER 1

INTRODUCTION

1.1 Back ground of the study

Through the past several decades, amorphous and more recently nano-crystalline materials have been investigated for application in magnetic devices. An example is the high density magnetic random access memory technology which has grown over the past decades due to its potential to store more data, access that data faster and also to use less power than current memory and technologies. Demands for continues increase in the data storage density bring the challenge to overcome physical limits for currently used magnetic recording media. Benefits were found in the nano-crystalline alloys because of their chemical and structural variations on the nano-scale which are important for developing optimal magnetic devices with high properties.

Magnetic nanoparticles are attracting much attention because they offer the opportunity to study magnetism in between the atomic and bulk limits and because ordered arrays of ferromagnetic properties are of potential interest for applications such as ultra-high-density magnetic recording devices. Recently magnetism recognized as a nanoscle phenomenon (Aharoni, 2000). The atomic exchange interaction that defines ferromagnetism is typically on the length scale of 10nm for most materials (Skomski, 2003).

1.1.1 Magnetic recording media for extremely high density data storage

Magnetic recording media is one of the most important components in any magnetic recording systems. The state of the art and research into hard disk media is approaching nano-scale and sub-nanoscale engineering. Nano-scale means the dimensions smaller than one billionth of a meter. In magnetic media, information is stored in a group of tiny magnets (Yasui *et al.*, 2008). The desire for higher density data storage involves materials with high coercivity. Being able to store data in an ever decreasing amount of space to continue device miniaturization leads one correctly to conclude that the smaller the magnetic bit, the higher the density of data can be stored within a given amount of space. To that end it would seem that magnetic nanoparticles would be an obvious choice for the next generation of magnetic data storage. In a digital recording scheme, if similar poles of two groups of magnets force each other, it is considered as 1, whereas opposite poles force each other, it is considered as 0.

Recently researches achieve small grains with size below 6nm, which will enable a 2.5 inch diameter hard disk to have 500-600 trillion small magnets (Piramanayagam *et al.*, 2007). However when the grain size is reduced down to this scale, a phenomenon called superparamagnetism poses a limit on magnetic properties. This is one of the key challenges in development of recording media.

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1.2 Importance of nanomaterials and magnetic nanomaterials

Nanoscale particles research has recently become an important field in the materials science. Nanoscience and nanothechnology deal with the synthesis, characterization, and exploration of nanostructured materials (McHenry et al., 2000). The nanoparticles are ultrafine particles in the size of nanometer order $(1nm=10^{-9})$. Nanostructure is between atomic and finite bulk systems. The vast interest in nanostructured particles mainly arises from the fact that these structures possess novel physical and magnetic properties that differ from those of bulk materials. This is significant in the case of nanoparticles that have a large surface to volume ratio with a high percentage of surface atoms, resulting in unexpected properties. Individual nanostructures include clusters, quantumdots, nanocrystals, nanowires, and nanotubes. Some nanostructure materials and assemblies are summarized in Table 1.1. Suitable control of the properties of nanomaterials brings new devices and technologies. Nanoscience and nanotechnology have two approaches: one is the bottom-up, that is, the miniaturization of the components, as articulated by Feynman, who stated in the 1959 lecture that "there is plenty of room at the bottom" (Feynman, 1959); and the other ones is the approach of the self-assembly of molecular components, where each nano-structured component becomes part of the suprastructure. Compared with the top-down lithographic techniques, bottom-up chemical synthesis and self-assembly approaches offer much more flexibilities in creating nanostructures and especially magnetic nanostructures with controlled size, shape, composition and so many physical properties.

Nanostructure	Size	Materials
Clusters, nanocrystals Quantum dots	Radius, 1-100nm	Insulators, semiconductors, metals, magnetic materials
Other nanoparticles	Radius, 1-100nm	Ceramic oxides
Nanobiomaterials, photosynthetic reaction centre	Radius, 5-10nm	Membrane protein
Nanowires	Diameter, 1-100nm	Metals, semiconductors, oxides, sulphides, nitrides
Nanotubes	Diameter, 1-100nm	Carbon, layered chalcogenides, BN, GaN
Nanobiorods	Diameter, 5nm	DNA
Two-dimensional arrays of nanoparticles	Area, several nm ² -µm ²	Metals, semiconductors, magnetic materials
Surfaces and thin films	Thickness, 1-100nm	Insulators, semiconductors, metals, DNA
Three-dimensional superlattices of nanoparticles	Several nm in three dimensions	Metals, semiconductors, magnetic materials

Table 1.1 Ivanosti uctui es anu then assemble	ructures and their assembli	mblies
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Cobalt and cobalt-platinum alloys are employed primarily for magnetic purposes and numerous applications such as patterned recording media, memory shape devices, magnetoresistive nanosensors and magnetic nanodevices (MEMS). The novelty of these alloys means that it can be advantageously employed in flow meters for handling corrosive liquids and for the magnetic jigs used in electropolishing baths.

1.3 Classification of magnetic materials

The term magnet is typically taken for only certain classes of materials that produce their persistent magnetic field even in the absence of an applied magnetic field. The response of materials to applied magnetic field is a phenomenon known as magnetism. The overall magnetic behavior of a material can vary widely, depending on the structure of the materials, and particularly on its electron configuration. Several forms of magnetic behavior have been observed in different materials, including: ferromagnetic, paramagnetic, diamagnetic, antiferromagnetic, ferrimagnetic and superparamagnetic materials.

1.4 Synthesis and characterization of nanoparticles

In the past few years, nanoparticle production by a size-controlled or shapecontrolled procedure (Lim et al., 2006) has been a new and interesting research focus. Increasing the storage density requires more strict control over the morphology of the magnetic material and strong reduction of its dimensions, down to the sizes of the single domain. Generally, the shape, size and size distribution of nanoparticles can be controlled by employing different synthetic methods. During the last decades, many methods have been employed to prepare magnetic cobalt and cobalt-platinum nanoparticles. The most important key is to avoid agglomeration of nanoparticles during the synthesis because the coalescence of the nanoparticles may loss their characteristic properties. Usually special organic compound such as surfactants, polymers and stabilizing ligands are used to passivate the particles to prevent them from aggregation. Nanocrystalline materials were manufactured passed on physical and chemical processes. It is often difficult to say which technique is better for synthesis nanoparticles but chemical methods are used frequently for their convenience and rapidity. The basic principle of chemical synthesis of nanostructured materials is to initiate chemical reactions and control the nucleation and growth of the reaction products. Among various chemical syntheses, microemulsion based method is creating stable colloids with good size dispersion and are typically carried out at low temperature. Also reverse-micelle microemulsion method is introduced to obtain the un-agglomerated, uniform and size controllable

nanoparticles. They are easy to develop to an industrial level, but they suffer from the disadvantage that the particles are sometimes less crystalline and more polydispersed because of low nucleation rate at low reaction temperature. Restricted reverse-micelle microemulsion (water-in-oil) environment was allowed synthesis of materials in tiny droplets of water encapsulated into oil as the main phase in reversemicelle conditions. Water pools of these reverse-micelles act as nano-reactors for performing simple reaction of synthesis, and the size of the nano-crystals is determined by size of these pools. The size of the water pools are controlled by the molar ratio of water to surfactant in the system. We apply the reverse-micelle microemulsion method for the synthesis of magnetic cobalt and alloy cobaltplatinum nanoparticles. XRD, EDX, TEM, FESEM, VSM and in other studies SQUID was used for analysis and investigate characterization of magnetic nanomaterials.

1.5 Significant of the study

Metal and alloy nanoparticles exhibit very interesting electronic, magnetic, optical, and chemical properties. Their unique features depend on their size, shape, surface composition, and surface atomic arrangement. The advance in storage capacity of magnetic recording media is due to combination of increasing coercivity and magnetization of particles, lowering system noise and improving head design. Providing high coercivity to prevent demagnetization requires that the grains of materials to be as small and uniform in size and shape. Microemulsions have been the subject of extensive research over the last two decades because of their scientific and technological importance. Microemulsions can have characteristic properties such as ultra-low interfacial tension, large interfacial area and capacity to solubilise both aqueous and oil-soluble compounds.

1.6 Problem statements

One of the issues afflicting memory is the fact that it is difficult to store information for as long as 10 years (Schiller, 2009). In order to overcome this problem, scientists and engineers have been looking for a way to increase the density of magnetic grains used for storage, as well as use materials with high magnetic anisotropy energy (MAE). The future of extremely high density recording media beyond 1 Tbits/in² require further reduction of magnetic grain size and thus materials with ultra-high magnetocrystalline anisotropy are needed to overcome the superparamagnetic limit (Wang *et al.*, 2006). The other problem is retaining the magnetization of the medium despite thermal fluctuations that is happened in superparamagnetic edge.

1.7 Objectives of the study

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In this thesis, we are aiming at developing tailored magnetic materials at the nanoscale by chemical method for the application in ultra-high density recording media. Cobalt-platinum alloys are high coercivity materials used as permanent magnets. The objectives of this work are:

1. To prepare mono-dispersed, uniform and sized-controllable Co and CoPt₃ nanoparticles using reverse-micelle microemulsion method.

- To study in detail the influence of molar ratio of water to surfactant on the size and character of Co, CoPt₃ and Co-Au, CoPt₃-Au core-shell nanoparticles, employing the experimental ω ratio (ω=[water]/[surfactant]: 5,10,15 and 20).
- 3. To investigate the morphology, crystal structure and magnetic properties of Co, CoPt₃, Co-Au and CoPt₃-Au nanoparticles for four different ω ratios before and after annealing.
- 4. To attempt measuring blocking temperature with a new simple method.

1.8 Outline of the thesis

Given the importance of detailed understanding of the structural, morphology and magnetic properties of nanostructures materials for the applications described above, this study will focus on the general introduction about the research background, scope, problem statement and objectives of the study. Chapter 2 concerns with the background and synthesis of magnetic nanoparticles and core-shell nanoparticles. Preparation of nanomaterials with microemulsion method and related literatures in view of preparation techniques together with characterization of magnetic nanoparticles were discussed as well. Chapter 3 focused on theoretical background which includes brief introduction to magnetism and the chemistry of metals and alloys. Chapter 4 highlights the methodology of the study; including materials, sample preparation and characterization of as-prepared and annealed nanoparticles before and after gold coating. Chapter 6 summarizes the results and gives some suggestions for future work.

REFERENCES

- Aharoni A: Introduction to the Theory of Ferromagnetism: Oxford University Press, USA, 2000.
- Ahmed J, Ahmad T, Ramanujachary KV, Lofland SE, Ganguli AK: Development of a microemulsion-based process for synthesis of cobalt (Co) and cobalt oxide (Co3O4) nanoparticles from submicrometer rods of cobalt oxalate. Journal of colloid and interface science 321:434-441, 2008.
- Ahmed J, Sharma S, Ramanujachary KV, Lofland SE, Ganguli AK: Microemulsionmediated synthesis of cobalt (pure fcc and hexagonal phases) and cobaltnickel alloy nanoparticles. Journal of colloid and interface science 336:814-819, 2009.
- Akulov N: Über das magnetische Quadrupolmoment des Eisenatoms. Zeitschrift für Physik A Hadrons and Nuclei 57:249-256, 1929.
- Anders S, Toney M, Thomson T, Thiele JU, Terris B, Sun S, Murray C: X-ray studies of magnetic nanoparticle assemblies. Journal of Applied Physics 93:7343, 2003.
- Armendariz V, Parsons JG, Lopez ML, Peralta-Videa JR, Jose-Yacaman M, Gardea-Torresdey JL: The extraction of gold nanoparticles from oat and wheat biomasses using sodium citrate and cetyltrimethylammonium bromide, studied by x-ray absorption spectroscopy, high-resolution transmission electron microscopy, and UV-visible spectroscopy. Nanotechnology 20:105607, 2009.
- Atwater JE, Akse JR, Holtsnider JT: Cobalt-poly (amido amine) superparamagnetic nanocomposites. Materials Letters 62:3131-3134, 2008.
- Ban Z, O'Connor C: A Novel Synthesis of Co@ Au Nanoparticles and Characterization, in, 2004, 2004, pp. 320–322.
- Bansal P, Hall M, Realff MJ, Lee JH, Bommarius AS: Multivariate statistical analysis of X-ray data from cellulose: A new method to determine degree of crystallinity and predict hydrolysis rates. Bioresource technology 101:4461-4471, 2010.
- Bao Y, Krishnan KM: Preparation of functionalized and gold-coated cobalt nanocrystals for biomedical applications. Journal of Magnetism and Magnetic Materials 293:15-19, 2005.
- Bao Y, Calderon H, Krishnan KM: Synthesis and characterization of magneticoptical Co-Au core-shell nanoparticles. The Journal of Physical Chemistry C 111:1941-1944, 2007.

- Barker P, Mason F, Barham P: Density and crystallinity of poly (3hydroxybutyrate/3-hydroxyvalerate) copolymers. Journal of Materials Science 25:1952-1956, 1990.
- Berkowitz A, Takano K: Exchange anisotropy--a review. Journal of Magnetism and Magnetic Materials 200:552-570, 1999.
- Boutonnet M, Kizling J, Stenius P, Maire G: The preparation of monodisperse colloidal metal particles from microemulsions. Colloids and Surfaces 5:209-225, 1982.
- Braidy N, Behal S, Adronov A, Botton G: Investigation of the oxide shell forming on-Co nanocrystals. Micron 39:717-722, 2008.
- Brooks H: Ferromagnetic anisotropy and the itinerant electron model. Physical Review 58:909, 1940.
- Capek I: Preparation of metal nanoparticles in water-in-oil (w/o) microemulsions. Advances in colloid and interface science 110:49-74, 2004.
- Carpenter EE: Synthesis of magnetic nanoparticles using reverse micelles/--by Everett E. Carpenter. University of New Orleans, 1999.
- Chatterjee J, Haik Y, Chen CJ: Polyethylene magnetic nanoparticle: a new magnetic material for biomedical applications. Journal of Magnetism and Magnetic Materials 246:382-391, 2002.
- Chen D, Liu S, Li J, Zhao N, Shi C, Du X, Sheng J: Nanometre Ni and core/shell Ni/Au nanoparticles with controllable dimensions synthesized in reverse microemulsion. Journal of Alloys and Compounds 475:494-500, 2009.
- Chen M: Synthesis and self-assembly of monodispersed cobalt-platinum, iron-cobaltplatinum, iron-paladium, and iron-platinum nanoparticles. 2002.
- Cheng G, Hight Walker AR: Synthesis and characterization of cobalt/gold bimetallic nanoparticles. Journal of Magnetism and Magnetic Materials 311:31-35, 2007.
- Chiang CL, Hsu MB, Lai LB: Control of nucleation and growth of gold nanoparticles in AOT/Span80/isooctane mixed reverse micelles. Journal of solid state chemistry 177:3891-3895, 2004.
- Cho SJ, Idrobo JC, Olamit J, Liu K, Browning ND, Kauzlarich SM: Growth mechanisms and oxidation resistance of gold-coated iron nanoparticles. Chemistry of materials 17:3181-3186, 2005.
- Cho SJ, Shahin AM, Gary J, Davies JE, Liu K, Grandjean F, Kauzlarich SM: Magnetic and Mössbauer spectral study of core/shell structured Fe/Au nanoparticles. Chemistry of materials 18:960-967, 2006.

- Christodoulides J, Huang Y, Zhang Y, Hadjipanayis G, Panagiotopoulos I, Niarchos D: CoPt and FePt thin films for high density recording media. Journal of Applied Physics 87:6938, 2000.
- Coffey KR, Howard JK, Parker MA: Thin film magnetic recording medium having high coercivity, in. Edited by, US Patent 5,989,728, 1999.
- Coffey WT, Crothers D, Dormann J, Kalmykov YP, Kennedy E, Wernsdorfer W: Thermally activated relaxation time of a single domain ferromagnetic particle subjected to a uniform field at an oblique angle to the easy axis: Comparison with experimental observations. Physical review letters 80:5655-5658, 1998.
- Cullity BD, Graham CD: Introduction to magnetic materials: Wiley-IEEE Press, 2009.
- Darbandi M, Thomann R, Nann T: Single quantum dots in silica spheres by microemulsion synthesis. Chemistry of materials 17:5720-5725, 2005.
- Darling A: Cobalt-Platinum Alloys. Platinum Metals Review 7:96-104, 1963.
- Deraz N: Size and crystallinity-dependent magnetic properties of copper ferrite nanoparticles. Journal of Alloys and Compounds 501:317-325, 2010.
- Du X, Inokuchi M, Toshima N: Preparation and characterization of Co-Pt bimetallic magnetic nanoparticles. Journal of Magnetism and Magnetic Materials 299:21-28, 2006.
- Duguet E, Vasseur S, Mornet S, Devoisselle JM: Magnetic nanoparticles and their applications in medicine. Nanomedicine 1:157-168, 2006.
- Duke CB, Plummer EW: Frontiers in surface and interface science: North-Holland, 2002.
- Ennas G, Casula M, Falqui A, Gatteschi D, Marongiu G, Piccaluga G, Sangregorio C, Pinna G: Nanocrystalline iron-cobalt alloys supported on a silica matrix prepared by the sol-gel method. Journal of non-crystalline solids 293:1-9, 2001.
- Ermakov A, Maikov V: Temperature dependence of magnetic crystallographic anisotropy and spontaneous magnetization of single crystals of FePd and CoPt alloys. Physics of Metals and Metallography(USSR) 69:198-201, 1990.
- Fang J, Stokes KL, Wiemann J, Zhou W: Nanocrystalline bismuth synthesized via an in situ polymerization-microemulsion process. Materials Letters 42:113-120, 2000.

Feynman RP: Plenty of Room at the Bottom, in, 1959, 1959.

- Fruchart O, Klaua M, Barthel J, Kirschner J: Self-organized growth of nanosized vertical magnetic Co pillars on Au (111). Physical review letters 83:2769-2772, 1999.
- Fu W, Yang H: Preparation and characteristics of core-shell structure cobalt/silica nanoparticles. Materials chemistry and physics 100:246-250, 2006.
- Giannakas A, Vaimakis T, Ladavos A, Trikalitis P, Pomonis P: Variation of surface properties and textural features of spinel ZnAl2O4 and perovskite LaMnO3 nanoparticles prepared via CTAB-butanol-octane-nitrate salt microemulsions in the reverse and bicontinuous states. Journal of colloid and interface science 259:244-253, 2003.
- Gong JL, Jiang JH, Liang Y, Shen GL, Yu RQ: Synthesis and characterization of surface-enhanced Raman scattering tags with Ag/SiO2 core-shell nanostructures using reverse micelle technology. Journal of colloid and interface science 298:752-756, 2006.
- Gubin SP, Koksharov YA, Khomutov G, Yurkov GY: Magnetic nanoparticles: preparation, structure and properties. Russian chemical reviews 74:489-520, 2005.
- Han D, Yang H, Zhu C, Wang F: Controlled synthesis of CuO nanoparticles using TritonX-100-based water-in-oil reverse micelles. Powder Technology 185:286-290, 2008.
- Harada M, Saijo K, Sakamoto N, Einaga H: Small-angle X-ray scattering study of metal nanoparticles prepared by photoreduction in aqueous solutions of sodium dodecyl sulfate. Colloids and Surfaces A: Physicochemical and Engineering Aspects 345:41-50, 2009.
- Heisenberg W: Zur theorie des ferromagnetismus. Zeitschrift für Physik A Hadrons and Nuclei 49:619-636, 1928.
- Hermans P, Weidinger A: Quantitative X Ray Investigations on the Crystallinity of Cellulose Fibers. A Background Analysis. Journal of Applied Physics 19:491-506, 1948.
- Ho KM, Li P: Design and synthesis of novel magnetic core-shell polymeric particles. Langmuir 24:1801-1807, 2008.

Hosokawa M: Nanoparticle technology handbook: Elsevier Science, 2007.

- Hu X, Li G, Yu JC: Design, fabrication, and modification of nanostructured semiconductor materials for environmental and energy applications. Langmuir 26:3031-3039, 2009.
- Israelachvili JN: Intermolecular and surface forces: with applications to colloidal and biological systems: London, 1985.

- Jamet M, Négrier M, Dupuis V, Tuaillon-Combes J, Mélinon P, Pérez A, Wernsdorfer W, Barbara B, Baguenard B: Interface magnetic anisotropy in cobalt clusters embedded in a platinum matrix. Journal of Magnetism and Magnetic Materials 237:293-301, 2001.
- Jewett JW, Serway RA: Physics for scientists and engineers with modern physics: Cengage Learning EMEA, 2008.
- Kakeshita T, Kim JH, Fukuda T: Microstructure and transformation temperature in alloys with a large magnetocrystalline anisotropy under external fields. Materials Science and Engineering: A 481:40-48, 2008.
- Kim D, Zhang Y, Voit W, Rao K, Muhammed M: Synthesis and characterization of surfactant-coated superparamagnetic monodispersed iron oxide nanoparticles. Journal of Magnetism and Magnetic Materials 225:30-36, 2001.
- Kittel C, McEuen P: Introduction to solid state physics, Vol. 4: Wiley New York, 1986.
- Klug HP, Alexander LE: X-ray diffraction procedures: for polycrystalline and amorphous materials. X-Ray Diffraction Procedures: For Polycrystalline and Amorphous Materials, 2nd Edition, by Harold P. Klug, Leroy E. Alexander, pp. 992. ISBN 0-471-49369-4. Wiley-VCH, May 1974. 1, 1974.
- Langford J, Wilson A: Scherrer after sixty years: a survey and some new results in the determination of crystallite size. Journal of Applied Crystallography 11:102-113, 1978.
- Leslie-Pelecky DL, Rieke RD: Magnetic properties of nanostructured materials. Chemistry of materials 8:1770-1783, 1996.
- Li F, Vipulanandan C: Production and characterization of YBCO nanoparticles. Applied Superconductivity, IEEE Transactions on 13:3196-3198, 2003.
- Lin CR, Wang CC, Chen I: Magnetic behavior of core-shell particles. Journal of Magnetism and Magnetic Materials 304:e34-e36, 2006.
- Lin J, Zhou W, Kumbhar A, Wiemann J, Fang J, Carpenter E, O'Connor C: Goldcoated iron (Fe@ Au) nanoparticles: synthesis, characterization, and magnetic field-induced self-assembly. Journal of solid state chemistry 159:26-31, 2001a.
- Lin J, Zhou W, O'Connor CJ: Formation of ordered arrays of gold nanoparticles from CTAB reverse micelles. Materials Letters 49:282-286, 2001b.
- Lin X, Sorensen C, Klabunde K, Hadjipanayis G: Temperature dependence of morphology and magnetic properties of cobalt nanoparticles prepared by an inverse micelle technique. Langmuir 14:7140-7146, 1998.

- Lin XM, Samina A: Synthesis, assebly and physical properties of magnetic nanoparticles. Journal pf Magnetism and Magnetic Materials 305:100-109, 2006.
- Liu J, Liu Y, Luo C, Shan Z, Sellmyer D: Magnetic hardening in FePt nanostructured films. Journal of Applied Physics 81:5644, 1997.
- Liu JP: Nanoscale magnetic materials and applications: Springer Verlag, 2009.
- Lopez-Quintela MA: Synthesis of nanomaterials in microemulsions: formation mechanisms and growth control* 1. Current opinion in colloid & interface science 8:137-144, 2003.
- Lu AH, Salabas EL, Schuth F: Magnetic nanoparticles: synthesis, protection, functionalization, and application. Angewandte Chemie-International Edition 46:1222-1245, 2007a.
- Lu AH, Salabas EL, Schüth F: Magnetic nanoparticles: synthesis, protection, functionalization, and application. Angewandte Chemie International Edition 46:1222-1244, 2007b.
- Lu X, Liang G, Sun Q, Yang C: The influence of silica matrix on the crystal structure and high frequency performance of cobalt nanoparticles. Journal of solid state chemistry 183:1555-1560, 2010.
- Mandal M, Das B, Mandal K: Synthesis of CoxPt1-x alloy nanoparticles of different phase by micellar technique and their properties study. Journal of colloid and interface science 335:40-43, 2009a.
- Mandal M, Pal D, Mandal K: Negatively charged micelles directed synthesis of snow-ball flower like superparamagnetic Ni nanoparticles and investigation of their properties. Colloids and Surfaces A: Physicochemical and Engineering Aspects 348:35-38, 2009b.
- Maria M, Agostiano A, Manna L, Monica M, Catalano M, Chiavarone L: Spagolo. V.; Lugara, M. J. Phys. Chem. B 104:8391-8397, 2000.
- Martin D: Crystallographic and magnetic properties of pseudobinary alloys based on cobalt-platinum. Journal of Physics F: Metal Physics 5:1031, 1975.
- Masala O, Seshadri R: Spinel ferrite/MnO core/shell nanoparticles: Chemical synthesis of all-oxide exchange biased architectures. Journal of the American Chemical Society 127:9354-9355, 2005.
- Mathew DS, Juang RS: Role of alcohols in the formation of inverse microemulsions and back extraction of proteins/enzymes in a reverse micellar system. Separation and purification technology 53:199-215, 2007a.

- Mathew DS, Juang RS: An overview of the structure and magnetism of spinel ferrite nanoparticles and their synthesis in microemulsions. Chemical Engineering Journal 129:51-65, 2007b.
- May S, Ben-Shaul A: Molecular theory of the sphere-to-rod transition and the second CMC in aqueous micellar solutions. The Journal of Physical Chemistry B 105:630-640, 2001.
- McHenry M, Laughlin D: Nano-scale materials development for future magnetic applications* 1. Acta materialia 48:223-238, 2000.
- Narita I, Oku T, Tokoro H, Suganuma K: Synthesis and structures of iron nanoparticles coated with boron nitride nanomaterials. Journal of electron microscopy 55:123, 2006.
- Newman R, Hemmingson J: Determination of the degree of cellulose crystallinity in wood by carbon-13 nuclear magnetic resonance spectroscopy. Holzforschung-International Journal of the Biology, Chemistry, Physics and Technology of Wood 44:351-356, 1990.
- Ninjbadgar T, Yamamoto S, Fukuda T: Synthesis and magnetic properties of the [gamma]-Fe2O3/poly-(methyl methacrylate)-core/shell nanoparticles. Solid state sciences 6:879-885, 2004.
- O'connor CJ, Carpenter EE, Sims JA: Sequential synthesis of core-shell nanoparticles using reverse micelles, in. Edited by, Google Patents, 2001a.
- O'Connor CJ, Kolesnichenko V, Carpenter E, Sangregorio C, Zhou W, Kumbhar A, Sims J, Agnoli F: Fabrication and properties of magnetic particles with nanometer dimensions. Synthetic metals 122:547-557, 2001b.
- Paland D, Santosh K: Shape controlled synthesis of iron-cobalt alloy magnetic nanoparticles using soft template method. Materials Letters 64:1127-1129, 2010.
- Pellegrino T, Fiore A, Carlino E, Giannini C, Cozzoli PD, Ciccarella G, Respaud M, Palmirotta L, Cingolani R, Manna L: Heterodimers based on CoPt3-Au nanocrystals with tunable domain size. Journal of the American Chemical Society 128:6690-6698, 2006.
- Peng S, Xie J, Sun S: Synthesis of Co/MFe2O4 (M= Fe, Mn) core/shell nanocomposite particles. Journal of solid state chemistry 181:1560-1564, 2008.
- Penuelas J, Andreazza-Vignolle C, Andreazza P, Ouerghi A, Bouet N: Temperature effect on the ordering and morphology of CoPt nanoparticles. Surface Science 602:545-551, 2008.

- Perez N, Guardia P, Roca AG, Morales MP, Serna CJ, Iglesias O, Bartolome F, Garcia LM, Batlle X, Labarta A: Surface anisotropy broadening of the energy barrier distribution in magnetic nanoparticles. Nanotechnology 19:475704, 2008.
- Petit C, Pileni M: Physical properties of self-assembled nanosized cobalt particles. Applied surface science 162:519-528, 2000.
- Pileni M: Reverse micelles used as templates: a new understanding in nanocrystal growth. Journal of Experimental Nanoscience 1:13-27, 2006.
- Piramanayagam S, Shi J, Zhao H, Pock C, Mah C, Ong C, Zhao J, Zhang J, Kay Y, Lu L: Magnetic and microstructural properties of CoCrPt: Oxide perpendicular recording media with novel intermediate layers. Magnetics, IEEE Transactions on 43:633-638, 2007.
- Pita M, Abad JM, Vaz-Dominguez C, Briones C, Mateo-Martí E, Martín-Gago JA, del Puerto Morales M, Fernández VM: Synthesis of cobalt ferrite core/metallic shell nanoparticles for the development of a specific PNA/DNA biosensor. Journal of colloid and interface science 321:484-492, 2008.
- Puntes VF, Krishnan KM, Alivisatos AP: Colloidal nanocrystal shape and size control: the case of cobalt. Science 291:2115, 2001.
- Qin W, Yang C, Ma X, Lai S: Selective synthesis and characterization of metallic cobalt; cobalt/platinum; and platinum microspheres. Journal of Alloys and Compounds, 2010.
- Rauscher F, Veit P, Sundmacher K: Detailed analysis of a technical grade w/omicroemulsion and its application for the precipitation of calcium carbonate particles. Colloid Colloids and Surfaces A: Physicochemical Engineering Aspects 254:183–191, 2005.
- Rivas J, Garcia-Bastida A, Lopez-Quintela M, Ramos C: Magnetic properties of Co/Ag core/shell nanoparticles prepared by successive reactions in microemulsions. Journal of Magnetism and Magnetic Materials 300:185-191, 2006.
- Rodriguez-Hernandez J, Chécot F, Gnanou Y, Lecommandoux S: Toward [] smart'nano-objects by self-assembly of block copolymers in solution. Progress in polymer science 30:691-724, 2005.
- Rodriguez-Sanchez L, Blanco M, Lopez-Quintela M: Electrochemical synthesis of silver nanoparticles. The Journal of Physical Chemistry B 104:9683-9688, 2000.
- Rohart S, Raufast C, Favre L, Bernstein E, Bonet E, Dupuis V: Magnetic anisotropy of Co_ {x} Pt_ {1- x} clusters embedded in a matrix: Influences of the cluster chemical composition and the matrix nature. Physical Review B 74:104408, 2006.

- Salavati-Niasari M, Fereshteh Z, Davar F: Synthesis of cobalt nanoparticles from [bis (2-hydroxyacetophenato) cobalt (II)] by thermal decomposition. Polyhedron 28:1065-1068, 2009.
- Sawicki M, Bowden G, De Groot P, Rainford B, Beaujour JML, Ward R, Wells M: Exchange springs in antiferromagnetically coupled DyFe_ {2}-YFe_ {2} superlattices. Physical Review B 62:5817, 2000.
- Schenzel K, Fischer S, Brendler E: New method for determining the degree of cellulose I crystallinity by means of FT Raman spectroscopy. Cellulose 12:223-231, 2005.
- Scher EC, Manna L, Alivisatos AP: Shape control and applications of nanocrystals. Philosophical Transactions of the Royal Society of London. Series A: Mathematical, Physical and Engineering Sciences 361:241, 2003.
- Schiller J: Quantum Computers: PublishAmerica, 2009.
- Schmid G: Nanoparticles: from theory to application: Wiley VCH, 2006.
- Schulman JH, Stoeckenius W, Prince LM: Mechanism of formation and structure of micro emulsions by electron microscopy. The Journal of Physical Chemistry 63:1677-1680, 1959.
- Segal L, Creely J, Martin A, Conrad C: An empirical method for estimating the degree of crystallinity of native cellulose using the X-ray diffractometer. Textile Research Journal 29:786, 1959.
- Shobana M, Rajendran V, Jeyasubramanian K, Suresh Kumar N: Preparation and characterisation of NiCo ferrite nanoparticles. Materials Letters 61:2616-2619, 2007.
- Shokrollahi H: The magnetic and structural properties of the most important alloys of iron produced by mechanical alloying. Materials & Design 30:3374-3387, 2009.
- Skomski R: Nanomagnetics. Journal of Physics: Condensed Matter 15:R841, 2003.
- Smirnov S, Komogortsev S: Magnetization curves of randomly oriented ferromagnetic single-domain nanoparticles with combined symmetry of magnetic anisotropy. Journal of Magnetism and Magnetic Materials 320:1123-1127, 2008.
- Sottys J, Lisowski Z, Knapczyk J: X-ray diffraction study of the crystallinity index and the structure of the microcrystalline cellulose. Acta pharmaceutica technologica 30:174-181, 1984.
- Spaldin NA: Magnetic materials: fundamentals and device applications: Cambridge Univ Pr, 2003.

- Spasova M, Wiedwald U, Farle M, Radetic T, Dahmen U, Hilgendorff M, Giersig M: Temperature dependence of exchange anisotropy in monodisperse cobalt nanoparticles with a cobalt oxide shell. Journal of Magnetism and Magnetic Materials 272:1508-1509, 2004.
- Spasova M, Salgueiriño-Maceira V, Schlachter A, Hilgendorff M, Giersig M, Liz-Marzán LM, Farle M: Magnetic and optical tunable microspheres with a magnetite/gold nanoparticle shell. Journal of Materials Chemistry 15:2095-2098, 2005.
- Sui Y, Skomski R, Sorge KD, Sellmyer DJ: Nanotube magnetism. Applied physics letters 84:1525, 2004.
- Summers M, Eastoe J, Davis S: Formation of BaSO4 nanoparticles in microemulsions with polymerized surfactant shells. Langmuir 18:5023-5026, 2002.
- Sun L, Zhang Y, Zhang J, Yan C, Liao C, Lu Y: Fabrication of size controllable YVO4 nanoparticles via microemulsion-mediated synthetic process. Solid state communications 124:35-38, 2002.
- Tai CY, Chen C: Particle morphology, habit, and size control of CaCO3 using reverse microemulsion technique. Chemical Engineering Science 63:3632-3642, 2008.
- Tracy JB, Bawendi MG: Defects in CoO in oxidized cobalt nanoparticles dominate exchange biasing and exhibit anomalous magnetic properties. Physical Review B 74:184434, 2006.
- Uskokovic V, Drofenik M: Synthesis of materials within reverse micelles. Surface Review and Letters 12:239-277, 2005.
- Van Vleck J: On the anisotropy of cubic ferromagnetic crystals. Physical Review 52:1178, 1937.
- Varadan VK, Chen L, Xie J: Nanomedicine: design and applications of magnetic nanomaterials, nanosensors and nanosystems: John Wiley & Sons Inc, 2008.
- Vidal-Vidal J, Rivas J, López-Quintela M: Synthesis of monodisperse maghemite nanoparticles by the microemulsion method. Colloids and Surfaces A: Physicochemical and Engineering Aspects 288:44-51, 2006.
- Wang H: A study of the structural, microstructural and magnetic properties of ironplatinum and cobalt-platinum type nanoparticles. University of Delaware, 2007.
- Wang JP, Qiu JM, Taton T, Kim BS: Direct Preparation of Highly Ordered \$ rm L1_0 \$ Phase FePt Nanoparticles and Their Shape-Assisted Assembly. Magnetics, IEEE Transactions on 42:3042-3047, 2006.

- Wen M, Liu QY, Wang YF, Zhu YZ, Wu QS: Positive microemulsion synthesis and magnetic property of amorphous multicomponent Co-, Ni-and Cu-based alloy nanoparticles. Colloids and Surfaces A: Physicochemical and Engineering Aspects 318:238-244, 2008.
- Witthayapanyanon A, Phan TT, Heitmann TC, Harwell JH, Sabatini DA: Interfacial Properties of Extended-Surfactant-Based Microemulsions and Related Macroemulsions. Journal of surfactants and detergents 13:127-134, 2010.
- Xiao Q, Bruck E, Zhang Z, De Boer F, Buschow K: Remanence enhancement in nanocrystalline CoPt bulk magnets. Journal of Alloys and Compounds 336:41-45, 2002.
- Xie Y, Ye R, Liu H: Synthesis of silver nanoparticles in reverse micelles stabilized by natural biosurfactant. Colloids and Surfaces A: Physicochemical and Engineering Aspects 279:175-178, 2006.
- Xu N, Chen J, Deng S: Effect of heat treatment on the properties of nano-diamond under oxygen and argon ambient. Diamond and related materials 11:249-256, 2002.
- Yadav O, Palmqvist A, Cruise N, Holmberg K: Synthesis of platinum nanoparticles in microemulsions and their catalytic activity for the oxidation of carbon monoxide. Colloids and Surfaces A: Physicochemical and Engineering Aspects 221:131-134, 2003.
- Yang M, Yang Y, Liu Y, Shen G, Yu R: Platinum nanoparticles-doped solgel/carbon nanotubes composite electrochemical sensors and biosensors. Biosensors and Bioelectronics 21:1125-1131, 2006.
- Yang Y, Jing L, Yu X, Yan D, Gao M: Coating aqueous quantum dots with silica via reverse microemulsion method: Toward size-controllable and robust fluorescent nanoparticles. Chemistry of materials 19:4123-4128, 2007.
- Yao L, Xu G, Yang X, Luan Y: CdS@ SiO2 nanoparticles synthesized from polyoxyethylene (10) tertoctylphenyl ether based reverse microemulsion. Colloids and Surfaces A: Physicochemical and Engineering Aspects 333:1-6, 2009.
- Yasui N, Ichihara S, Nakamura T, Imada A, Saito T, Ohashi Y, Den T, Miura K, Muraoka H: Characterization of high-density patterned media fabricated by a new anodizing process. Journal of Applied Physics 103:07C515, 2008.
- Yu M, Liu Y, Sellmyer D: Nanostructure and magnetic properties of composite CoPt: C films for extremely high-density recording. Journal of Applied Physics 87:6959, 2000.
- Zeng H, Li J, Wang Z, Liu J, Sun S: Bimagnetic core/shell FePt/Fe3O4 nanoparticles. Nano Letters 4:187-190, 2004.

- Zhang J, Sun L, Liao C, Yan C: Size control and photoluminescence enhancement of CdS nanoparticles prepared via reverse micelle method. Solid state communications 124:45-48, 2002.
- Zhang W, Qiao X, Chen J: Synthesis of silver nanoparticles--Effects of concerned parameters in water/oil microemulsion. Materials Science and Engineering: B 142:1-15, 2007.
- Zhang WW, Cao QQ, Xie JL, Ren XM, Lu CS, Zhou Y, Yao YG, Meng QJ: Structural, morphological, and magnetic study of nanocrystalline cobaltnickel-copper particles. Journal of colloid and interface science 257:237-243, 2003.
- Zhang X, Wen G, Xiao G, Sun S: Magnetic relaxation of diluted and self-assembled cobalt nanocrystals. Journal of Magnetism and Magnetic Materials 261:21-28, 2003.
- Zhong CJ, Luo J, Njoki PN, Mott D, Wanjala B, Loukrakpam R, Lim S, Wang L, Fang B, Xu Z: Fuel cell technology: nano-engineered multimetallic catalysts. Energy Environ. Sci. 1:454-466, 2008.