

UNIVERSITI PUTRA MALAYSIA

DEVELOPMENT OF RICE HUSK ASH-BASED GEOPOLYMER BINDER FOR FIRE RESISTANT COATING

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DEVELOPMENT OF RICE HUSK ASH-BASED GEOPOLYMER BINDER FOR FIRE RESISTANT COATING

By

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Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirements for the Degree of Doctor of Philosophy

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DEDICATION

To Allah SWT, my beloved parents, sisters, and brothers for supporting and encouraging me with affections, love, and their denoted contribution towards success in my life

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

DEVELOPMENT OF RICE HUSK ASH-BASED GEOPOLYMER BINDER FOR FIRE RESISTANT COATING

Ву

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Despite the growing popularity of geopolymer in various industrial applications, the literature showed gaps on understanding of factors influential on the properties of geopolymer binder (GB). The main objective of this research is to develop an eco-friendly rice husk ash (RHA)-based GB material suitable as fire resistant coating and panel materials for building construction. Four experiments were conducted; namely fire resistant, compression, flexural, and adhesion tests. Fractional factorial design (FrFD) and response surface methodology (RSM) were used to design the experiments incorporating five factors; namely ratio of activated alkaline (AA) solution, RHA to AA (RHA/AA) ratio, curing temperature, curing time, and sodium hydroxide concentration and analyze their interrelationship and effect on GB properties. The optimum range for GB composition was accordingly determined using superimposed contour plotting. Results showed that RHA/AA ratio was the most influential factor on all GB properties (thermal, mechanical, and physical properties). RHA provided effective fire resistant properties since its GB coating exhibited intumescent-like expansion. The newly developed formulation of GB which exhibited semi-brittle behavior was able to produce good fire resistance and compressive strength properties. In addition, good adhesion and flexural properties were important in providing coating with good fire resistant properties. Optimal GB panels were formulated at W/S and Si/Al ratios of 1.05 to 1.13 and 95 to 110, respectively while optimal GB coating formulated at W/S and Si/Al ratios of 110 to 130 and 1.20 to 1.33, respectively. RHA proved to be the best alternative aluminosilicate source and the RHA-based GB, when used as panel or coating materials, can potentially improve fire safety in building construction.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

PEMBANGUNAN PENGIKAT GEOPOLIMER BERASASKAN ABU SEKAM PADI SEBAGAI SALUTAN TAHAN API

Ву

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Walaupun populariti qeopolimer semakin berkembang dalam pelbagai aplikasi industri, tinjauan literatur mengenalpasti jurang dalam memahami faktor-faktor vang mempengaruhi sifat-sifat pengikat geopolimer (GB). Objektif utama penyelidikan ini adalah untuk membangunkan bahan GB berasaskan abu sekam padi yang mesra alam sesuai digunakan sebagai bahan salutan tahan api dan bahan panel untuk pembinaan bangunan. Empat eksperimen telah dijalankan iaitu ujian ketahanan api, mampatan, lekatan dan lenturan. Rekabentuk faktoran pecahan (FFD) dan kaedah tindakbalas permukaan (RSM) telah digunakan untuk menyediakan rekabentuk eksperimen yang melibatkan lima faktor iaitu nisbah larutan aktivasi alkali (AA), nisbah RHA kepada AA (RHA/AA), suhu pengawetan, masa pengawetan, dan kepekatan natrium hidroksida (NaOH) dan mengkaji hubungkait dan kesannya terhadap sifat-sifat GB. Julat optimum bagi komposisi GB dikenalpasti menggunakan plot kontur bertindan. Hasil kajian mendapati bahawa nisbah RHA/AA adalah faktor yang paling mempengaruhi semua sifat-sifat GB (sifat haba, mekanikal, dan fizikal). RHA memberi sifat tahan api yang efektif oleh kerana salutan GB itu mempamerkan kembangan seakan mengembung. Formulasi GB yang baru dibangunkan yang mempamerkan sifat separa rapuh mampu menghasilkan sifat ketahanan api dan kekuatan mampatan yang baik. Tambahan pula, sifat lekatan dan sifat lenturan yang baik adalah penting untuk menyediakan salutan yang mempunyai sifat ketahanan api yang baik. Panel GB yang optimum telah dirumuskan pada nisbah W/S dan Si/Al masing-masing 1.05 ke 1.13 dan 95 ke 110, manakala salutan GB yang optimum dirumuskan pada nisbah W/S dan Si/Al masing-masing 110 ke 130 dan 1.20 ke 1.33. RHA terbukti sebagai sumber bahan alumino-silikat alternatif terbaik dan GB berasaskan RHA berpotensi untuk meningkatkan faktor keselamatan dalam pembinaan bangunan apabila digunakan sebagai bahan panel atau salutan.

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LIST OF ABBREVIATIONS

2-FrFD 2-level Fractional Factorial Design

AA Activated Alkaline

Adj MS Adjusted Mean Square Value Adj SS Adjusted Sum of Errors

ΑI Alumina

ANOVA Analysis of Variant

AS Area of the Inverted Peak

ASTM American Society for Testing and Materials

BFS Blast Furnace Slag

CCD Central Composite Design

Coefficient of Thermal Expansion CTE

DF Degree of Freedom

Differential Scanning Calorimetry DSC

Energy Dispersive X-Ray Spectroscopy **EDX**

EEC Estimated Effect and Coefficient

FΑ Fly Ash Iron Fe

FFD Full Factorial Design **FrFD** Fractional Factorial Design

Fourier Transform Infrared Spectroscopy **FTIR**

GB Geopolymer Binder

GBC Geopolymer Binder Coating

KOH Potassium Hydroxide

Ν Notation Sodium Na

Na₂SiO₃ Sodium Silicate Sodium Hydroxide NaOH

Hydroxide OH

Ordinary Portland Cement OPC **PSD** Particle Size Distribution

RH Rice Husk RHA Rice Husk Ash

RHBA Rice Husk Bark Ash

Response Surface Methodology S/L Solid-to-Liquid SE Standardized Effect

SEC Standard Error of Coefficient SEM Scanning Electron Microscopy

Si

RSM

SS Sequential Sum of Errors Temperature at Equilibrium TAE Thermogravimetry Analysis **TGA**

TT300 Time Taken to Reach 300°C (°C)

Vs Versus

W/S Water-to-Solid Wt. Weight (g) X-Ray Diffraction **XRD** XRF X-Ray Fluorescence

LIST OF SYMBOLS

 $\begin{array}{ccc} \% & & \text{Percent} \\ \overline{x} & & \text{Average} \\ ^{\circ}\text{C} & & \text{Degree Celcius} \\ \mu\text{m} & & \text{Micrometer} \\ \text{cm} & & \text{Centimeter} \end{array}$

E Slope of Proportional Limit (Young's Modulus)

H Inverted Peak Height

M Molar mm Millimeter MPa Mega Pascal

V₁ Ratio of AA Solution

V₂ RHA/AA ratio

V₃ Curing Temperature V₄ NaOH Concentration

V₅ Curing Time

E Compressive Strain Engineering Strain

CHAPTER 1

INTRODUCTION

1.1 Research Background

The collapse of World Trade Center in the United States of America is a turning point in research intensification of fire protection materials. One of the most vital aspects of building regulations is fire protection. Structural integrity, rapid flame spread, smoke and toxic fume emissions are normally the problems caused by fire in buildings. Buildings must be installed with active and passive fire protection systems. Active fire protection systems such as extinguishers, water sprinklers, and fire detectors work to effectively combat fire at the beginning of fire outbreak. At the same time, passive systems act as a sub-defense unit which is extremely important to control rapid fire spread and protect the structural integrity of the buildings.

Coatings and panels are passive fire protection systems. Steel, which is largely used in building construction, normally is unable to withstand high temperatures between 470°C and 500°C causing it slowly to lose its strength. Without coating, the temperature of unprotected steel increases rapidly due to high thermal conductivity of steel. Therefore, unprotected steel structures do not have desirable resistance against fire resulting in rapid creeping, buckling, collapse or other failure. For safety reason, fire resistant coating materials are necessary to keep the temperature of steel structures below the critical temperature during fire incidents. On the other hand, core panel such as in sandwich wall panel or fire door is unduly important to prevent fire from spreading to other side of the wall or door.

New fire resistant building materials are required for higher sustainability and durability. Since overall cost components in building constructions come from building materials and labor, the need for lower-cost and time-efficient technology becomes urgent. 'Geopolymer', a term coined by a French scientist named Joseph Davidovits, is the best alternative which is environmentally friendly and possesses excellent mechanical and thermal properties including good fire resistance and insulation properties (Davidovits, 1989).

For the last 30 years, geopolymers were applied in various field of applications. One of the first objective in geopolymer development is to be an alternative to existing fire and heat resistance organic thermosetting polymer which were insufficient as passive fire protection during a tragic fire in France around 1970 (Davidovits, 2002). Geopolymers or activated-alkaline aluminosilicate are prepared by synthesizing activated alkaline solution with aluminosilicate sources

such as waste sources including fly ash, metakaolin, kaolin, blast furnace slag, palm oil fuel ash and rice husk ash (RHA).

According to Food and Agriculture Organization of the United Nations (2015), global annual rice paddy production between 2014 and 2015 was around 745 million tons. In Malaysia, rice paddy production in the same period was approximately 2.6 million tons. In general, more than 149 million tons of rice husks are produced annually. RH is one of the largest available biomass resources and the low cost of RH may potentially reduce the costs of geopolymer through lowered cost of production compared to other resin systems in the market such as polyimides, cyanide esters, silicone, epoxy, polyurethane and acrylic resins (Han *et al.*, 2008) (Oyetola and Abdullahi, 2006).

1.2 Problem Statement

Coatings which are widely used for protective purpose contain combustible material that will easily burn during a fire event. Although the coating may have effective fire resistance properties, if it is burnt and emit smoke, especially toxic smoke, the emission of burnt steel smoke may pose a high risk to the community in case of fire events. Current coatings mainly available in the market are epoxy resin, whilst intumescent coating materials are less used due to lack of information

A building structure reinforced with steel but with poor or insufficient coating may fail to protect its function in load-bearing resulting in building collapse. In order to prevent such catastrophic event, the coating must be able to withstand bending moment before the event thus preventing cracks development in the coating. Eco-friendly coating should possess good fire resistance properties in order to function as an effective shield to protect steel from early degradation during fire event. With the use of fire resistant coating, fire rate can be reduced, allowing longer opportunity time for escape and more lives could be saved in fire catastrophes. Initiative must be taken as well in providing not only better durability and longevity of the whole coating system, but also coating with effective good bonding with the steel surface before and during the fire event.

Existing wall panels in the market have poor fire resistance properties in some of the materials which are widely used such as resins, polystyrene, and polyurethane (Zhang et al., 2014). Therefore, environmental-friendly and energy consumption factors have to be crucially considered as alternatives to current wall panels. Although some of the panel structures, used as insulation materials, should possess high fire resistance characteristic, their mechanical requisites including compressive strength for walls and doors must not be unduly compromised. In addition, intensive research studies on the relationship between compressive strength and fire resistance properties of fire resistant materials, using statistical analysis approach, are not as yet well established.

Geopolymer binder has been proven to possess excellent fire resistance properties and various waste materials have been used in producing geopolymer binder including fly ash, metakaolin, kaolin, and blast furnace slag. Although RHA was found to contain the highest silica to alumina ratio among agricultural wastes, the utilization of this resource is still low due probably to several reasons including; (i) low awareness of RH potential among farmers and industrial communities, (ii) information on the material and its utilization was insufficiently known to the public, (iii) socio-economic problems, (iv) low penetration of technology, (v) lack of interest and (vi) lack of awareness about environmental problems (Omatola and Onojah, 2012).

In Malaysia, rice husk is considered biodegradable waste product in the rice mill industry and is burned in open area or dumped in landfills. Rice husk is a good source of renewable energy but if improperly incinerated, this vast quantity of waste may potentially pollute the environment including the air which through open burning, soil and water. Through large-scale industrial utilization of RHA as an aluminosilicate source in fire resistant geopolymer material, the environmental pollution can be reduced through the application of green technology in the manufacture of more sustainable eco-friendly and cost effective geopolymer binder (GB) material.

Despite the growing popularity of geopolymer in various applications, limited research has been devoted to identify the influence of several factors on the properties of RHA-based geopolymer binder. In this study, a new and novel type of geopolymer binder based on RHA was developed for coating and panel applications, particularly as a construction building material. Using statistical approach, this novel formulations of RHA-based geopolymer binder are expected to possess high compressive strength, good adhesion bonding, promising fire resistance properties, and good flexural strength and flexural modulus. The new product, based on RHA, should have efficient and cost-effective properties, with wide potential applications in industry.

1.3 Research Objectives

The main objective of this study is to design experimental work using suitable statistical method and develop a RHA-based geopolymer binder product which is suitable for fire resistant coating and panel materials for various applications, particularly in the construction building industry.

Along with the main objective, there are several specific objectives including:

 To design experimental works using fractional factorial design (FrFD) and response surface methodology (RSM) and conduct experiments on fire resistant, compressive strength, flexural strength, flexural modulus, and adhesion strength properties of RHA-based geopolymer binder.

- ii. To statistically analyze the experimental data to determine the effect of each factor, including ratio of activated alkaline (AA) solution, RHA/AA ratio, curing temperature, sodium hydroxide (NaOH) concentration, and curing time, and its interactions on the properties of the RHA-based geopolymer binder and respectively conduct their optimization tests.
- iii. To investigate and characterize the microstructural and element characteristics of RHA-based geopolymer binder.
- iv. To identify the optimum range for silica to alumina (Si/Al) ratio and water to solid (W/S) ratio in RHA-based geopolymer binder compositions which produced the best mechanical, thermal, and physical properties of RHA-based geopolymer binder.

Although the best binder may be discovered during the research process, it should be noted that the purpose of this study is not to solely develop the best or strongest possible binder material for coating and panel applications, but rather to study and identify the characteristics of RHA-based geopolymer binder that can be the best alternative as fire resistant binder material in the present market.

1.4 Research Scope

In achieving the objectives of this study, experimental works consisting of fire resistant, compressive, three-point bending (flexural), and adhesion tests were conducted based on three main properties; namely, thermal, mechanical, and physical properties. For this purpose, five factors were considered, namely; (i) ratio of activated alkaline (AA) solution, (ii) rice husk ash to activated alkaline (RHA/AA) ratio, (iii) curing temperature, (iv) sodium hydroxide (NaOH) concentration, and (v) curing time. RHA was used as aluminosilicate source whereas sodium silicate and NaOH solution as the AA solution. The geopolymer binder (GB) used in this study was fabricated into two main types of material; coating and panel. Mild steel plate was used as a substrate throughout this study. Statistical methods, including fractional factorial design (FrFD) and response surface methodology (RSM) were employed in experimental design and analysis.

The geopolymer binder materials were characterized into seven main tests including (i) mechanical properties test by using three-point bending (flexural) test, and compressive test, (ii) thermal properties test by using fire resistant test, (iii) physical properties test by using pull-off adhesion test, (iv) thermal analysis by using thermogravimetry analysis (TGA), differential scanning calorimetric (DSC), and coefficient of thermal expansion (CTE), (v) microscopy analysis by using scanning electron microscopy (SEM), (vi) element characterization by using energy dispersive X-ray spectroscopy (EDX), fourier transform infrared spectroscopy (FTIR), x-ray fluorescence (XRF), x-ray diffraction (XRD), differential scanning calorimetric (DSC), and (vii) particle size analysis by using particle size distribution (PSD).

1.5 Thesis Organization

This thesis consists of eight chapters. Chapter one discusses the background of the research and problem statement, followed by an outline of the research objectives, the scope and organization of the dissertation.

Chapter two presents the literature review for related studies that have been carried out and reported previously including rice husk ash, geopolymer, properties of geopolymer, influential factors, and statistical design method including fractional factorial design (FrFD) and response surface methodology (RSM). Chapter three describes the methods in conducting experiments, analyzing data and results on the preliminary findings of rice husk ash.

Results and discussions are presented in chapter four until seven. Chapter four presented the results and discussions of thermal properties (fire resistant) of RHA-based geopolymer binder coating (GBC). Chapter five discussed the properties of the binder under mechanical properties including compressive strength, flexural modulus, flexural strength. Chapter six discusses the results for physical properties (adhesion strength) of RHA-based GBC. Chapter seven discusses the optimum range of GB compositions namely Si/Al ratio and W/S ratio which produced the best properties in terms of thermal, mechanical, and physical properties. Finally, the conclusion, contribution, and recommendations for future research is given in Chapter eight.

REFERENCES

- Abdul Rahim, R. H.; Azizli, K. A.; Man, Z.; Rahmiati, T.; Nuruddin, M. F. (2014). Effect of sodium hydroxide concentration on the mechanical property of non sodium silicate fly ash based geopolymer. *Journal of Applied Sciences*.
- Abdul Rahim, R. H.; Rahmiati, T.; Azizli, M.; Azizi, K.; Man, Z.; Nuruddin, F.; Ismail, L. (2014). Comparison of using NaOH and KOH activated fly ashbased geopolymer on the mechanical properties. *Materials Science Forum*, 803, 179-184.
- Alemán, J. V.; Chadwick, A. V.; He, J.; Mess, M.; Horie, K.; Jones, R. G.; Kratochvil, P.; Meisel, I.; Mita, I.; Moad, G.; Penczek, S.; Stepto, R. F.T. (2009). Definitions of terms relating to the structure and processing of sols, gels, networks, and inorganic-organic hybrid materials (IUPAC Recommendations 2007). In *Pure and Applied Chemistry* (pp. 1081-1829).
- Alonso, S.; Palomo, A. (2001). Alkaline activation of metakaolin and calcium hydroxide mixtures: Influence of temperature, activator concentration and solids ratio. *Materials Letters*, 47(1), 55-62.
- Amat, R. C.; Ibrahim, N. M.; Rahim, N. L.; Tajudin, N. S. B. A.; Ahmad, K. R. (2013). Fire resistance of biomass ash panels used for internal partitions in building. *Procedia Engineering*, *53*, 52-57.
- Anderson, C. E.; Dziuk, J.; Mallow, W. A. (1985). Intumescent reaction mechanisms. *Journal of Fire Sciences*, *3*(3), 161-194.
- Antony, J. (2003). Design of Experiments for Engineers and Scientists. Butterworth-Heinemann: An imprint of Elsevier, Linacre House, Jordon Hill, OxfordOX2 8DP, 200 Wheeler Road, Burlington,MA01803.
- Asif, A.; Man, Z.; Mohd Azizli, K. A.; Nuruddin, M. F.; Ismail, L. (2015). The effect of Si/Al ratio and sodium silicate on the mechanical properties of fly ash based geopolymer for coating. *Material Science Forum*, 803, 355-361.
- ASTM A36. (1999). Standard specification for carbon structural steel.
- ASTM D4541. (n.d.). Standard test method for pull-off strength of coatings using portable adhesion testers.
- ASTM D695. (n.d.). Standard test method for compressive properties of rigid plastics.
- ASTM E176. (2001). Standard terminology of fire standards. *American Society for Testing and Materials*, West Conshohocken, PA.
- ASTM E290-09. (n.d.). Standard test methods for bend testing of material for ductility.
- ASTM E831. (n.d.). Standard test method for linear thermal expansion of solid materials by thermomechanical analysis.
- Aziz, H.; Ahmad, F.; Zia-ul-Mustafa, M. (2015). Nano filler reinforced intumescent fire retardant coating for protection of structural steel. *InCIEC 2014* (pp. 831-844). Springer Singapore.
- Baglin, J.E. E. (1988). Thin film adhesion: New possibilities for interface engineering. *Materials Science and Engineering B, 1*(1), 1-7.
- Bakharev, T. (2004). Effect of curing regime and type of activator on properties of alkali-activated fly ash. *1st International Symposium on Nanotechnology in Construction.* Paisley, Scotland.
- Barbosa, V. F.; MacKenzie, K. J. (2003). Synthesis and thermal behaviour of potassium sialate geopolymers. *Materials Letters*, *57*(9), 1477-1482.

- Barbosa, V. F.F.; MacKenzie, K. J.D. (2003). Thermal behaviour of inorganic geopolymers and composites derived from sodium polysialate. *Materials Research Bulletin*, 38(2), 319-331.
- Beck, H.; Siemers, M.; Reuter, M. (2013). Verification of resistance to fire. In *Powder-actuated Fasteners and Fastening Screws in Steel Construction* (p. 4.1.3). John Wiley & Sons.
- Bell, J. L.; Drimeyer, P. E.; Kriven, W. M. (2009). Formation of ceramics from metakaolin-based geopolymers: Part I—Cs-based geopolymer. *Journal of the American Ceramic Society*, *92*(1), 1-8.
- Bezerra, M. A.; Santelli, R. E.; Oliveira, E. P.; Villar, L. S.; Escaleira, L. A. (2008). Response surface methodology (RSM) as a tool for optimization in analytical chemistry. *Talanta*, *76*(5), 965-977.
- Bharadwaj, A.; Wang, Y.; Sridar, S.; Arunachalam, V. S. (2004). Pyrolysis of rice husk. *Current Science*, *87*(7), 981-986.
- Bhowmick, A.; Ghosh, S. (2012). Effect of synthesizing parameters on workability and compressive strength of fly ash based geopolymer mortar. *International Journal of Civil and Structural Engineering, 3*(1), 168-177.
- Box, G. E. P.; Wilson, K. B. (1951). On the experimental attainment of optimum conditions. *Journal of the Royal Statistical Society, 13*, 1-45.
- Box, G. E.; Hunter, J. S. (1957). Multi-factor experimental design for exploring response surfaces. *The Annals of Mathematical Statistics*, 195-241.
- Box, G. E.; Hunter, W. G.; Hunter, J. S. (1978). *Statistics for Experimenters*. New York: John Wiley and Sons.
- Boyle, C. H.; Meguid, S. A. (2015). Mechanical performance of integrally bonded copper coatings for the long term disposal of used nuclear fuel. *Nuclear Engineering and Design*, 293, 403-412.
- Buchler, C.; Rollin, M. (2009). User-friendly composites that take the heat. *Jec Composites Magazine*, 33-35.
- Buchwald, A.; Vicent, M.; Kriegel, R.; Kaps, C.; Monzo, M.; Barba, A. (2009). Geopolymeric binders with different fine fillers—phase transformations at high temperatures. *Applied Clay Science*, *46*(2), 190-195.
- Bulewicz, E. M.; Pelc, A.; Kozlowski, R.; Miciukiewicz, A. (1985). Intumescent silicate-based materials: Mechanism of swelling in contact with fire. *Fire and Materials*, *9*(4), 171-175.
- Caballero, L. A., Eiarque, E. S., Espelleta, C. V., Ferrando, M. F., Garcia, T. J., & Vicent, C. M. (2014). *Zaragoza, Spain Patent No. EP2727894 A1*.
- Camp, C. (n.d.). CIVL 1101 Introduction to Strength of Materials. (Department of Civil Engineering, University of Memphis) Retrieved Feb 03, 2015, from http://www.ce.memphis.edu/1101/notes/concrete/section 1 strength o
 - http://www.ce.memphis.edu/1101/notes/concrete/section_1_strength_of_materials.html
- Cartuyvels, R.; Booth, R.; Kubicek, S.; Dupas, L.; De Meyer, K. M. (1993). A powerful TCAD system including advanced RSM techniques for various engineering optimization problems. *In Simulation of Semiconductor Devices and Processes*, 29-32.
- Chang, W. H.; Scriven, R. L.; Ross, R. B. (1975). Flame-retardant organic coatings. In *Flame-Retardant Polymeric Materials* (pp. 399-453). US: Springer.

- Chaudhary, D. S.; Jollands, M. C.; Cser, F. (2002). Understanding rice hull ash as fillers in polymers: A review. *Silicon Chemistry*, 1(4), 281-289.
- Chen , L.; Song, L.; Lv, P.; Jie, G.; Tai, Q.; Xing, W.; Hu, Y. (2011). A new intumescent flame retardant containing phosphorus and nitrogen: Preparation, thermal properties and application to UV curable coating. *Progress in Organic Coatings*, 70(1), 59-66.
- Chen, C.; Li, Q.; Shen, L.; Zhai, J. (2012). Feasibility of manufacturing geopolymer bricks using circulating fluidized bed combustion bottom ash. *Environmental Technology*, 33(11), 1313-1321.
- Chen, L.; Han, W.; Li, Z.; Wei, T.; Xiao, C. (2011). Preparation and properties of alkali stimulated geopolymer and its application in thermal insulation coating. *Advanced Materials Research*, 233-235, 2443-2446.
- Cheng, S. W.; Wu, C. F. J. (2001). Factor screening and response surface exploration. *Statistica Sinica*, 553-580.
- Cheng, T. W.; Chiu, J. P. (2003). Fire-resistant geopolymer produced by granulated blast furnace slag. *Minerals Engineering*, *16*(3), 205-210.
- Chindaprasirt, P.; Chareerat, T.; Sirivivatnanon, V. (2007). Workability and strength of coarse high calcium fly ash geopolymer. *Cement & Concrete Composites*, 29(3), 224-229.
- Chindaprasirt, P.; Chotetanorm, C.; Rukzon, S. (2010). Use of palm oil fuel ash to improve chloride and corrosion resistance of high-strength and highworkability concrete. *Journal of Materials in Civil Engineering*, 23(4), 499-503.
- Chindaprasirt, P.; Jaturapitakkul, C.; Chalee, W.; Rattanasak, U. (2009). Comparative study on the characteristics of fly ash and bottom ash geopolymers. *Waste Management*, *29*(2), 539-543.
- Ching, F. D.; Winkel, S. R. (2016). Fire-resistive construction. In *Building Codes*Illustrated: A Guide to Understanding the 2015 International Building
 Code (p. Chapter 7). John Wiley & Sons.
- Creagh Precision Joinery. (n.d.). *Fire Resistant Door Range*. (Creagh Precision Joinery) Retrieved January 16, 2016, from http://www.creaghprecisionjoinery.co.uk/fire-resistant-doors.html
- Das, R.; Ghosh, S.; Naskar, M. K. (2014). Synthesis of single crystal zeolite L rods with high aspect ratio using rice husk ash as silica source. *Indian Journal of Chemistry Section A-Inorganic Bio-Inorganic Physical Theoretical & Analytical Chemistry*, *53*(7), 816-819.
- Davidovits, J. (1989). Geopolymers and geopolymeric materials. *Journal of Thermal Analysis*, *35*(2), 429-441.
- Davidovits, J. (1991). Geopolymers: Inorganic polymeric new materials. *Journal of Thermal Analysis and Calorimetry*, 37, 1633-1656.
- Davidovits, J. (1999). Chemistry of geopolymeric systems, terminology. *Geopolymer'99 International Conference.* France.
- Davidovits, J. (2002). 30 years of successes and failures in geopolymer applications. Market trends and potential breakthroughs. *Geopolymer 2002 Conference*. Melbourne, Australia.
- Davidovits, J. (2008). *Geopolymer Chemistry and Applications*. St. Quentin, France: Institut Geopolymere.
- Davidovits, J. (2008). Geopolymer composite sandwiches for heat barrier. In *Geopolymer Chemistry and Application* (pp. 488-490). Institut Geopolymere.

- Davidovits, J. (2011). Geopolymers in toxic waste management. In *Geopolymer, Chemistry and Applications 3rd edition* (p. 585). Saint-Quentin, France: Institut Géopolymère.
- Delair, S.; Prud'homme, E.; Peyratout, C.; Smith, A.; Michaud, P.; Eloy, L.; Joussein, E.; Rossignol, S. (2012). Durability of inorganic foam in solution: The role of alkali elements in the geopolymer network. *Corrosion Science*, *59*, 213-221.
- Della, V. P.; Kuhn, I.; Hotza, D. (2002). Rice husk ash as an alternative source for active silica production. *Materials Letters*, *57*(4), 818-821.
- Deventer, V.; Jannie, SJ.; Provis, J. L.; Duxson, P. (2012). Technical and commercial progress in the adoption of geopolymer cement. *Minerals Engineering*, 29, 89-104.
- Dhakal, M.; Kupwade-Patil, K.; Allouche, E. N.; Johnson, C. C. L. B.; Ham, K. (2013). Optimization and characterization of geopolymer mortars using response surface methodology. *Developments in Strategic Materials and Computational Design IV*, 135-149.
- Dimanshteyn, F. A., & Barone, R. J. (2005, September 27). *United States Patent No. US 11/664,465.*
- Domke, P. V. (2012). Improvement in the strength of concrete by using industrial and agri-cultural waste. *IOSR Journal of Engineering*, *24*(2), 755-759.
- Draft Malaysian Standard. (2014). Fire Safety in the Design, Management and Use of Buildings Code of Practice (First Revision). Department of Standards Malaysia.
- Duquesne, S.; Magnet, S.; Jama, C.; Delobel, R. (2004). Intumescent paints: Fire protective coatings for metallic substrates. *Surface and Coatings Technology*, 180, 302-307.
- Duquesne, S.; Magnet, S.; Jama, C.; Delobel, R. (2005). Thermoplastic resins for thin film intumescent coatings—towards a better understanding of their effect on intumescence efficiency. *Polymer Degradation and Stability*, 88(1), 63-69.
- Duxson, P.; Lukey, G. C.; van Deventer, J. S.J. (2007). Physical evolution of Nageopolymer derived from metakaolin up to 1000 °C. *Journal of Materials Science*, 42(9), 3044-3054.
- Fernández-Jiménez, A.; Palomo, A. (2005). Composition and microstructure of alkali activated fly ash binder: Effect of the activator. *Cement and Concrete Research*, *35*(10), 1984-1992.
- Ferone, C.; Roviello, G.; Colangelo, F.; Cioffi, R.; Tarallo, O. (2013). Novel hybrid organic-geopolymer materials. *Applied Clay Science*, *73*, 42-50.
- Ferraris, C. F.; Obla, K. H.; Hill, R. (2001). The influence of mineral admixtures on the rheology of cement paste and concrete. *Cement and Concrete Research*, 31(2), 245-255.
- Ferreira, S. L. C.; Bruns, R. E.; da Silva, E. G. P.; dos Santos, W. N. L.; Quintella, C. M.; David, J. M.; de Andrade, J. B.; Breitkreitz, M. C.; Jardim, I. C. S. F.; Neto, B. B. (2007). Statistical designs and response surface techniques for the optimization of chromatographic systems. *Journal of Chromatography A, 1158*(1), 2-14.
- Fletcher, R. A.; MacKenzie, K. J.; Nicholson, C. L.; Shimada, S. (2005). The composition range of aluminosilicate geopolymers. *Journal of the European Ceramic Society*, *25*(9), 1471-1477.
- FLIR Systems. (2006). ThermaCam Researcher User's Manual. FLIR Systems.

- Foo, K. Y.; Hameed, B. H. (2009). Utilization of rice husk ash as novel adsorbent: A judicious recycling of the colloidal agricultural waste. *Advances in Colloid and Interface Science*, *152*(1), 39-47.
- Food and Agriculture Organization of the United Nations. (2015). *Rice Market Monitor.* The FAO Rice Market Monitor (RMM).
- Funk, J. E.; Dinger, D. (1994). *Predictive process control of crowded particulate suspensions: Applied to ceramic manufacturing.* Springer Science & Business Media.
- Ganesan, N.; Indira, P. V.; Santhakumar, A. (2013). Prediction of ultimate strength of reinforced geopolymer concrete wall panels in one-way action. *Construction and Building Materials*, 48, 91-97.
- Gardelle, B.; Duquesne, S.; Vandereecken, P.; Bourbigot, S. (2014). Resistance to fire of silicone based coatings: Fire protection of steel against cellulosic fire. *Journal of Fire Sciences*, *32*(4), 374-387.
- Ghosh, R.; Bhattacherjee, S. (2013). A review study on precipitated silica and activated carbon from rice husk. *Journal of Chemical Engineering & Process Technology*, 4(4).
- Giancaspro, J.; Balaguru, P. N.; Lyon, R. E. (2006). Use of inorganic polymer to improve the fire response of balsa sandwich structures. *Journal of Materials in Civil Engineering*, 18(3), 390-397.
- Giannopoulou, I.; Panias, D. (2007). Structure, design and applications of geopolymeric materials. *In Third International Conference on Deformation Processing and Structure of Materials*. Belgrade, Serbia.
- Goga, V. (2011). New phenomenological model for solid foams. Springer Netherlands.
- Grand View Research. (2015). Intumescent Coatings Market Analysis By Application (Cellulosic, Hydrocarbons), By End-Use (Oil & Gas, Construction, Automotive) And Segment Forecasts To 2022. United States: Grand View Research, Inc.
- Gredel, T. E. (1997). Industrial ecology: Definition and implementation. In *Industrial Ecology and Global Change* (pp. 23-40). R. Socolow: Ed. Cambridge: University Press.
- Grillet, A. M.; Gloe, L. M.; Wyatt, N. B.; Vicente, J. D. (Ed.). (2012). Polymer gel rheology and adhesion. In *Rheology* (p. 61). INTECH Open Access Publisher.
- Gu, J. W.; Zhang, G. C.; Dong, S. L.; Zhang, Q. Y.; Kong, J. (2007). Study on preparation and fire-retardant mechanism analysis of intumescent flame-retardant coatings. *Surface and Coatings Technology*, *201*(18), 7835-7841.
- Guo, X.; Shi, H.; Dick, W. A. (2010). Compressive strength and microstructural characteristics of class C fly ash geopolymer. *Cement and Concrete Composites*, *32*(2), 142-147.
- Gurit. (2011). Guide to Composites Delivering the Future of Composite Solutions. Gurit.com.
- Habeeb, G. A.; Mahmud, H. B. (2010). Study on properties of rice husk ash and its use as cement replacement material. *Materials Research*, *13*(2), 185-190.
- Han, R.; Ding, D.; Xu, Y.; Zou, W.; Wang, Y.; Li, Y.; Zou, L. (2008). Use of rice husk for the adsorption of congo red from aqueous solution in column mode. *Bioresources Technology*, *99*(8), 2938-2946.

- Hanrahan, G.; Lu, K. (2006). Application of factorial and response surface methodology in modern experimental design and optimization. *Critical Reviews in Analytical Chemistry*, *36*(3-4), 141-151.
- Hardjito, D.; Wallah, S. E.; Sumajouw, D. M.; Rangan, B. V. (2004). On the development of fly ash-based geopolymer concrete. *ACI Materials Journal*, 101(6).
- He, J.; Jie, Y.; Zhang, J.; Yu, Y.; Zhang, G. (2013). Synthesis and characterization of red mud and rice husk ash-based geopolymer composites. *Cement & Concrete Composites*, *37*, 108-118.
- Heah, C. Y.; Kamarudin, H.; Al Bakri, A. M.; Bnhussain, M.; Luqman, M.; Nizar, I. K.; Ruzaidi, C. M.; Liew, Y. M. (2012). Study on solids-to-liquid and alkaline activator ratios on kaolin-based geopolymers. *Construction and Building Materials*, *35*, 912-922.
- Heah, C. Y.; Kamarudin, H.; Al Bakri, A. M.; Binhussain, M.; Luqman, M.; Nizar, I. K.; Ruzaidi, C. M.; Liew, Y. M. (2011). Effect of curing profile on kaolin-based geopolymers. *Physics Procedia*, *22*, 305-311.
- Heo, U. H.; Sankar, K.; Kriven, W. M.; Musil, S. S. (2015). Rice husk ash as a silica source in a geopolymer formulation. *Developments in Strategic Materials and Computational Design V: A Collection of Papers Presented at the 38th International Conference on Advanced Ceramics and Composites January 27-31, 2014 Daytona Beach, Florida, 87-101.*
- Hirschler, M. M. (2000). Chemical aspects of thermal decomposition of polymeric materials. In *Fire Retardancy of Polymeric Materials* (p. 31). New York: CRC Press.
- Hollingbery , L. A.; Hull, T. R. (2010). The fire retardant behaviour of huntite and hydromagnesite A review. *Polymer Degradation and Stability*, 95(12), 2213-2225.
- Hshieh, F. Y. (1998). Shielding effects of silica-ash layer on the combustion of silicones and their possible applications on the fire retardancy of organic polymers. *Fire and Materials*, 22(2), 69-76.
- Hull, T. R.; Colligon, J. S.; Hill, A. E. (1987). Measurement of thin film adhesion. *Vacuum*, *37*, 327-330.
- Hunt, G. M.; Truax, T. R. (1933). Fire Proofing Wood with Chemicals. Madison, Wisconsin: University of Wisconsin.
- Hussein, M. Z.; Al Ali, S. H.; Zainal, Z.; Hakim, M. N. (2011). Development of antiproliferative nanohybrid compound with controlled release property using ellagic acid as the active agent. *Int J Nanomedicine*, *6*, 1373-1383.
- Hwang, C. L.; Huynh, T. P. (2015). Evaluation of the performance and microstructure of ecofriendly construction bricks made with fly ash and residual rice husk ash. *Advances in Materials Science and Engineering*.
- Iji, M.; Serizawa, S. (1998). Silicone derivatives as new flame retardants for aromatic thermoplastics used in electronic devices. *Polymers for Advanced Technologies, 9*(10-11), 593-600.
- ISO 834. (n.d.). Fire resistance tests -- Elements of building construction.
- Jarvis, P.; Jefferson, B.; Parsons, S. A. (2005). Breakage, regrowth, and fractal nature of natural organic matter flocs. *Environmental Science & Technology*, *39*(7), 2307-2314.
- Jaynes, J.; Ding, X.; Xu, H.; Wong, W. K.; Ho, C. M. (2013). Application of fractional factorial designs to study drug combinations. *Statistics in Medicine*, *32*(2), 307-318.

- Jimenez, M.; Duquesne, S.; Bourbigot, S. (2006). High-throughput fire testing for intumescent coatings. *Industrial & Engineering Chemistry Research*, 45(22), 7475-7481.
- Jimenez, M.; Duquesne, S.; Bourbigot, S. (2006). Intumescent fire protective coating: Toward a better understanding of their mechanism of action. *Thermochimica Acta*, *449*(1), 16-26.
- Kamarudin, H.; Al Bakri, A. M.; Binhussain, M.; Ruzaidi, C. M.; Luqman, M.; Heah, C. Y.; Liew, Y. M. (2011). Preliminary study on effect of NaOH concentration on early age compressive strength of kaolin-based green cement. *Proceedings of International Conference on Chemistry and Chemical Process (ICCCP 2011).*
- Karim, M.R.; Zain, M.F.M.; Jamil, M.; Lai, F.C. (2013). Fabrication of a non-cement binder using slag, palm oil fuel ash and rice husk ash with sodium hydroxide. *Construction and Building Materials*, 49, 894-902.
- Karthikeyan, K.; Nanthakumar, K.; Shanthi, K.; Lakshmanaperumalsamy, P. (2010). Response surface methodology for optimization of culture conditions for dye decolorization by a fungus, Aspergillus Niger HM11 isolated from dye affected soil. *Iranian Journal of Microbiology*, 2(4), 213-222.
- Khale, D.; Chaudhary, R. (2007). Mechanism of geopolymerization and factors influencing its development: a review. *Journal of Materials Science*, 42(3), 729-746.
- Khan, M. I.; Azizli, K.; Sufian, S.; Man, Z. (2014). Effect of Na/Al and Si/Al ratios on adhesion strength of geopolymers as coating material. *Applied Mechanics and Materials*, 625, 85-89.
- Khan, M. I.; Azizli, K.; Sufian, S.; Man, Z. (2014). Sodium silicate-free geopolymers as coating materials: Effects of Na/Al and water/solid ratios on adhesion strength. *Ceramics International*, *4*(1), 2794-2805.
- Kilinc, F. S. (2013). Intumescent coatings. In *Handbook of Fire Resistant Textiles* (p. 332). Elsevier.
- Kim, D.; Lai, H. T.; Chilingar, G. V.; Yen, T. F. (2006). Geopolymer formation and its unique properties. *Environmental Geology*, *51*(1), 103-111.
- Kim, Y. Y.; Lee, B.J.; Saraswathy, V.; Kwon, S.J. (2014). Strength and durability performance of alkali-activated rice husk ash geopolymer mortar. *The Science World Journal*.
- Komnitsas, K.; Zaharaki, D. (2007). Geopolymerisation: A review and prospects for the minerals industry. *Minerals Engineering*, *20*(14), 1261-1277.
- Kong, Daniel L.Y.; Sanjayan, J. G.; Safoe-Crentsil, K. (2007). Comparative performance of geopolymers made with metakaolin and fly ash after exposure to elevated temperatures. *Cement and Concrete Research*, 37(12), 1583-1589.
- Konsta-Gdoutos, M. S. (2003). Calculation of strain energy release rate from load displacement crack area equation. *Problems of Fracture Mechanics and Fatigue*, 117-120.
- Krishnarao, R. V.; Godkhindi, M. M. (1992). Distribution of silica in rice husks and its effect on the formation of silicon carbide. *Ceramics International*, 18, 243-249.
- Kriven, W. M.; Bell, J. L.; Gordon, M. (2003). Microstructure and microchemistry of fully-reacted geopolymers and geopolymer matrix composites. *Advances in Ceramic Matrix Composites IX, 153,* 227-250.

- Kuenzel, C.; Vandeperre, L. J.; Donatello, S.; Boccaccini, A. R.; Cheeseman, C. (2012). Ambient temperature drying shrinkage and cracking in metakaolin-based geopolymers. *Journal of the American Ceramic Society*, 95(10), 3270-3277.
- Laosiritaworn, W. (2015). Improving multi-panel lamination process optimization using response surface methodology and neural network. *Progress in Systems Engineering*, 221-226.
- Laoutid, F.; Bonnaud, L.; Alexandre, M.; Lopez-Cuesta, J. M.; Dubois, P. (2009). New prospects in flame retardant polymer materials: From fundamentals to nanocomposites. *Materials Science and Engineering: R: Reports*, 63(3), 100-125.
- Laza, J. M.; Julian, C. A.; Larrauri, E.; Rodriguez, M.; Leon, L. M. (1999). Thermal scanning rheometer analysis of curing kinetic of an epoxy resin: 2. An amine as curing agent. *Polymer*, *40*(1), 35-45.
- Leiva, C.; Vilches, L. F.; Querol, X.; Vale, J.; Fernandez, C. P. (2007). Use of zeolitised fly ashes in fire resistant plates. *Proceedings of the World of Coal Ash: Science, Applications and Sustainability.* Covington, KY, United States.
- Leiva, C.; Vilches, L. F.; Vale, J.; Fernández-Pereira, C. (2005). Influence of the type of ash on the fire resistance characteristics of ash-enriched mortars. *Fuel*, *84*(11), 1433-1439.
- Leiva, C.; Vilches, L. F.; Vale, J.; Fernández-Pereira, C. (2009). Fire resistance of biomass ash panels used for internal partitions in building. *Fire Safety Journal*, *44*(4), 622-628.
- Li, W. Z.; Yao, Y.; Li, Y. Q.; Li, J. B.; Gong, J.; Sun, C.; Jiang, X. (2008). Damage behavior of the NiCrAlY coating systems with or without barrier layer during three-point bending. *Materials Science and Engineering A, 512*, 117-125.
- Li, X.; Cui, X.; Liu, S.; Mo, B.; Cui, L. (2013). Preparation and characterization of inorganic zinc-rich coatings based on geopolymers. *Key Engineering Materials*, 537, 261-264.
- Li, Z.; Ding, Z.; Zhang, Y. (2004). Development of sustainable cementitious materials. *Proceedings of International Workshop on Sustainable Development and Concrete Technology.* Beijing, China.
- Liou, T. H. (2004). Preparation and characterization of nano-structured silica from rice husk. *Materials Science and Engineering: A, 364*(1), 313-323.
- Lipiäinen, H.; Chen, Q.; Larismaa, J.; Hannula, S. P. (2016). The effect of fire retardants on the fire resistance of unsaturated polyester resin coating. In *Key Engineering Materials* (pp. 277-283). Trans Tech Publications.
- Liu, Y.; Fischer, T. E.; Dent, A. (2003). Comparison of HVOF and plasmasprayed alumina/titania coatings—microstructure, mechanical properties and abrasion behavior. *Surface and Coatings Technology*, 167(1), 68-76.
- Liu, Z.; Sheng, S.; Cui, X.; Liu, H.; Chao, Q. (2012). The preparation of geopolymer-based external thermal insulating materials [J]. *New Building Materials*, 1(032).
- Lopez, F. J.; Sugita, S.; Tagaya, M.; Kobayashi, T. (2014). Geopolymers using rice husk silica and metakaolin derivatives; preparation and their characteristics. *Journal of Materials Science and Chemical Engineering*, 35-43.

- Lotfy , W. A.; Ghanem, K. M.; El-Helow, E. R. (2007). Citric acid production by a novel Aspergillus niger isolate: II. Optimization of process parameters through statistical experimental designs. *Bioresource Technology*, 98(18), 3470-3477.
- Luna-Galiano, Y.; Cornejo, A.; Leiva, C.; Vilches, L. F.; Fernandez-Pereira, C. (2015). Properties of fly ash and metakaolín based geopolymer panels under fire resistance tests. *Materiales de Construcción*, *65*(319), e059.
- Lundstedt, T.; Seifert, E.; Abramo, L.; Thelin, B.; Nyström, Å.; Pettersen, J.; Bergman, R. (1998). Experimental design and optimization. *Chemometrics and Intelligent Laboratory Systems, 42*(1), 3-40.
- Lyon, R. E.; Balaguru, P. N.; Foden, A.; Sorathia, U.; Davidovits, J.; Davidovics, M. (1997). Fire-resistant aluminosilicate composites. *Fire and Materials*, 21(2), 67-73.
- Lyon, R. E.; Foden, A. J.; Balaguru, P. N.; Davidovits, J.; Davidovics, M. (1997). Fire resistant aluminosilicate composite. *Fire and Materials*, *21*, 67-73.
- Madejska, L.; Jarosiński, A.; Żelazny, S.; Kusnierova, M.; Chachlowska, M. (2011). Properties of geopolymer binder obtained from fly ash. *Czasopismo Techniczne. Chemia, 108*(1-Ch), 113-118.
- Mathur, A.; Singh, U.; Vijay, Y. K.; Hemlata, M.; Sharma, M. (2013). Analyzing performance for generating power with renewable energy source using rice husk as an alternate fuel. *Control Theory and Informatics*, *3*(2), 64-71.
- Mattox, D. M. (1978). Thin film adhesion and adhesive failure A perspective. In *Adhesion Measurement of Thin Films, Thick Films and Bulk Coatings* (pp. 54-62). K.L. Mittal, Editor, American Society for Testing and Materials.
- Meyers, R. H.; Montgomery, D. C.; Anderson-Cook, C. M. (2009). Response Surface Methodology: Process and Product Optimization Using Designed Experiments. New York: John Wiley & Sons.
- Minitab 17: Getting started with Minitab 17. (2013). Minitab Inc.
- Mohd Yusof, A.; Nizam, N. A.; Abd Rashid, N. A. (2010). Hydrothermal conversion of rice husk ash to faujasite-types and NaA-type of zeolites. *Journal of Porous Material*, 17(1), 39-47.
- Mohd Yusof, Alias; Nizam, Nik Ahmad; Abd Rashid, Noor Aini. (2010). Hydrothermal conversion of rice husk ash to faujasite-types and NaAtype of zeolites. *Journal of Porous Material*, 17(1), 39-47.
- Mohsen, Q.; Mostafa, N. Y. (2010). Investigating the possibility of utilising low kaolinitic clays in production of geopolymer bricks. *Ceramics-Silikaty*, *54*(2), 160-168.
- Montgomery, D. C. (2008). Design and Analysis of Experiments. John Wiley & Sons
- Moser, R. D.; Allison, P. G.; Williams, B. A.; Weiss Jr., C. A.; Diaz, A. J.; Gore, E. R.; Malone, P. G. (2013). Improvement in the geopolymer-to-steel bond using a reactive vitreous enamel coating. *Construction and Building Materials*, 49, 62-69.
- Mouritz, A. P.; Gibson, A. G. (2007). Chapter 8 Flame retardant composites. In Fire Properties of Polymer Composite Materials (pp. 237-286). Springer Science & Business Media.
- Mouritz, A. P.; Mathys, Z. (2001). Post-fire mechanical properties of glass-reinforced polyester composites. *Composites Science and Technology*, *61*(4), 475-490.

- Nagral, M. R.; Ostwal, T.; Chitawadagi, M. V. (2014). Effect of curing temperature and curing hours on the properties of geopolymer concrete. *International Journal of Computational Engineering Research*, 4(9).
- Narayanan, K.; Subramanyam, V. M.; Venkata Rao, J. (2014). A fractional factorial design to study the effect of process variables on the preparation of hyaluronidase loaded plga nanoparticles. *Enzyme Research*.
- Nath, P.; Sarker, P. K.; Rangan, V. B. (2015). Early age properties of low-calcium fly ash geopolymer concrete suitable for ambient curing. *Procedia Engineering*, 125, 601-607.
- Nazari, A. (2011). Properties of geopolymer with seeded fly ash and rice husk bark ash. *Materials Science and Engineering: A*, *528*(24), 7395-7401.
- Nguyen, H. T.; Gallardo, S. M.; Bacani, F. T.; Hinode, H.; Do, Q. M.; Do, M. H.; Promentilla, M. A. B. (2014). Evaluating thermal properties of geopolymer produced from red mud, rice husk ash and diatomaceous earth.
- Noor Syuhadah, S.; Rohasliney, H. (2012). Rice husk as biosorbent: A review. Health Environ. J., 3, 89-95.
- NZ Builders Ltd. (n.d.). Concrete Insulated Panel (CIP) Construction. (NZ Builders Ltd.) Retrieved July 26, 2016, from http://nzbuilders.com/construction-method/concrete-insulated-panel-construction/
- Omatola, K. M.; Onojah, A. D. (2012). Rice husk as a potential source of high technology raw materials: A review. *Journal of Physical Sciences and Innovation*, *4*, 1-6.
- Oyane, M.; Sato, T.; Okimoto, K.; Shima, S. (1980). Criteria for ductile fracture and their applications. *Journal of Mechanical Working Technology, 4*(1), 65-81.
- Oyetola, E. B.; Abdullahi, M. (2006). The use of rice husk ash in low-cost sandcrete block production. *Leonardo Electronic Journal of Practices and Technologies*, *8*, 58-70.
- Padture, N. P.; Gell, M.; Jordan, E. H. (2002). Thermal barrier coatings for gasturbine engine applications. *Science*, 280-284.
- Palanikumar, K. (2007). Modeling and analysis for surface roughness in machining glass fibre reinforced plastics using response surface methodology. *Materials & Design*, 28(10), 2611-2618.
- Palomo, A.; Blanco-Varela, M. T.; Granizo, M. L.; Puertas, F.; Vazquez, T.; Grutzeck, M. W. (1999). Chemical stability of cementitious materials based on metakaolin. *Cement and Concrete Research*, *29*(7), 997-1004.
- Palomo, A.; Grutzeck, M. W.; Blanco, M. T. (1999). Alkali-activated fly ashes: A cement for the future. *Cement and Concrete Research*, *29*(8), 1323-1329.
- Pan, Z.; Thompson, J. F. (2009). Study of Rice for Improved Quality and Processing Efficiency Annual Report Comprehesive Research on Rice. California, USA: Department of Biological and Agricultural Engineering, University of California.
- Panias, D.; Gainnopoulou, I.; Boufounos, D. (2014). Valorization of alumina red mud for production of geopolymeric bricks and tiles. *Light Metals 2014*, 155-159.
- Panias, D.; Giannopoulou, I. P.; Perraki, T. (2007). Effect of synthesis parameters on the mechanical properties of fly ash-based geopolymers.

- Colloids and Surfaces A: Physicochemical and Engineering Aspects, 301(1), 246-254.
- Poppovics, S. (1998). Fundamental of the fracture mechanism of concrete. In Strength and Related Properties of Concrete: A Quantitative Approach (p. 151). John wiley & sons.
- Prasad, R.; Pandey, M. (2012). Rice husk ash as a renewable source for the production of value added silica gel and its application: An overview. *Bulletin of Chemical Reaction Engineering & Catalysis*, 7(1), 1-25.
- Provis, J. L.; Harrex, R. M.; Bernal, S. A.; Duxson, P.; van Deventer, J. S.J. (2012). Dilatometry of geopolymers as a means of selecting desirable fly ash sources. *Journal of Non-Crystalline Solids*, *358*(16), 1930-1937.
- Provis, J. L.; van Deventer, J. S. J. (2009). Producing fire- and heat-resistant geopolymers. In *Geopolymers: Structures, Processing, Properties and Industrial Applications* (p. 262). Woodhead Publishing.
- Rahier, H.; Van Mele, B.; Biesemans, M.; Wastiels J.; Wu, X. (1996). Low-temperature synthesized aluminosilicate glasses. Part I Low-temperature reaction stoichiometry and structure of a model compound. *Journal of Materials Science*, *31*(1), 71-79.
- Rahier, H.; Van Mele, B.; Biesemans, M.; Wastiels, J.; Wu, X. (1996). Low-temperature synthesized aluminosilicate glasses part II. *Journal of Materials Science*, *31*(1), 71-79.
- Rahmiati, T.; Azizli, K. A.; Man, Z.; Ismail, L.; Nuruddin, M. F. (2015). Effect of solid/liquid ratio during curing time fly ash based geopolymer on mechanical property. *Materials Science Forum*, 803, 120-124.
- Ramezanianpour, A. A.; Mahdikhani, M.; Ahmadibeni, G. (2009). The effect of rice husk ash on mechanical properties and durability of sustainable concretes. *International Journal of Civil Engineering*, 7(2), 83-91.
- Ramujee, K.; Potharaju, M. (2013). Development of low calcium fly ash based geopolymer concrete. *Indian Journal of Applied Research*, *3*(6), 180-182.
- Rao, K. K.; Rao, K. J.; Sarwade, A. G.; Chandra, M. S. (2012). Strength analysis on honeycomb sandwich panels of different materials. *International Journal of Engineering Research and Applications*, 2(3), 365-374.
- Rattanasak , U.; Chindaprasirt, P. (2009). Influence of NaOH solution on the synthesis of fly ash geopolymer. *Minerals Engineering*, 22(12), 1073-1078.
- Rees, C. A.; Provis, J. L.; Lukey, G. C.; van Deventer, J. S. (2007). In situ ATR-FTIR study of the early stages of fly ash geopolymer gel formation. *Langmuir*, 23(17), 9076-9082.
- Rickard, W. D.; Kealley, C. S.; Riessen, A. (2015). Thermally induced microstructural changes in fly ash geopolymers: Experimental results and proposed model. *Journal of the American Ceramic Society*, *98*(3), 929-939.
- Rickard, W. D.A.; Temuujin, J.; van Riessen, A. (2012). Thermal analysis of geopolymer pastes synthesised from five fly ashes of variable composition. *Journal of Non-Crystalline Solids*, *358*(15), 1830-1839.
- Rickard, W. D.A.; van Riessen, A.; Walls, P. (2010). Thermal character of geopolymers synthesized from class F fly ash containing high concentrations of iron and α-quartz. *International Journal of Applied Ceramic Technology*, 7(1), 81-88.

- Roche, A. A.; Dole, P.; Bouzziri, M. (1994). Measurement of the practical adhesion of paint coatings to metallic sheets by the pull-off and three-point flexure tests. *Journal of Adhesion Science and Technology, 8*(6), 587-609.
- Rodríguez, E. D.; Bernal, S. A.; Provis, J. L.; Gehman, J. D.; Monzó, J. M.; Payá, J.; Borrachero, M. V. (2013). Geopolymers based on spent catalyst residue from a fluid catalytic cracking (FCC) process. *Fuel, 109*, 493-502.
- Rowell, R. M. (2012). Chemicals used to form a coating on the wood surface. In Handbook of Wood Chemistry and Wood Composites (p. 145). CRC Press.
- Ryu, G. S.; Lee, Y. B.; Koh, K. T.; Chung, Y. S. (2013). The mechanical properties of fly ash-based geopolymer concrete with alkaline activators. *Construction and Building Materials*, *47*, 409-418.
- Sabitha, D.; Dattatreya, J. K.; Sakthivel, N.; Bhuvaneshwari, M.; Sathik, S. J. (2012). Reactivity, workability and strength of potassium versus sodium-activated high volume fly ash-based geopolymers. *Current Science* (*Bangalore*), 103(11), 1320-1327.
- Sakkas, K.; Kapelari, S.; Panias, D.; Nomikos, P.; Sofianos, A. (2014). Fire resistant K-based metakaolin geopolymer for passive fire protection of concrete tunnel linings. *Open Access Library Journal*, 1(6).
- Sakkas, K.; Nomikos, P.; Sofianos, A.; Panias, D. (2015). Sodium-based fire resistant geopolymer for passive fire protection. *Fire and Materials*, 39(3), 259-270.
- Schulte, R. (2014, January 30). *Fire Protection History*. Retrieved from Building Code Resource Library: http://buildingcoderesourcelibrary.com/Fire-Protection-History-Part-257.00.pdf
- Seeman, M.; Ganesan, G.; Karthikeyan, R.; Velayudham, A. (2010). Study on tool wear and surface roughness in machining of particulate aluminum matel matrix composite-response surface methodology approach. *Int. J. Adv. Manuf. Technol.*, 48, 613-624.
- Shi, C.; Jiménez, A. F.; Palomo, A. (2011). New cements for the 21st century: The pursuit of an alternative to portland cement. *Cement and Concrete Research*, *41*(7), 750-763.
- Shinohara, Y.; Kohyama, N. (2004). Quantitative analysis of tridymite and cristobalite crystallized in rice husk ash by heating. *Industrial Health*, 42(2), 277-285.
- Silva, P. D.; Sagoe-Crenstil, K.; Sirivivatnanon, V. (2007). Kinetics of geopolymerization: Role of Al2O3 and SiO2. *Cement and Concrete Research*, *37*(4), 512-518.
- Sindhunata; van Deventer, J. S. J.; Lukey, G. C.; Xu, H. (2006). Effect of curing temperature and silicate concentration on fly-ash-based geopolymerization. *Industrial & Engineering Chemistry Research*, 45(10), 3559-3568.
- Singh, B. J. (2014). Response surface methodology. In *RSM: A Key to Optimize Machining: Multi-Response Optimization of CNC Turning with Al-7020 Alloy* (pp. 51-59). Anchor Academic Publishing.
- Sisman, C. B.; Gezer, E. (2011). Effects of rice husk ash on characteristics of the briquette produced for masonry units. *Scientific Research and Essays*, *6*(4), 984-992.

- Sivaraja, M.; Kandasamy, S. (2011). Potential reuse of waste rice husk as fibre composites in concrete. *Asian Journal of Civil Engineering (Building and Housing)*, 12(2), 205-217.
- Sivaraos; Milkey, K. R.; Samsudin, A. R.; Dubey, A. K.; Kidd, P. (2014). Comparison between taguchi method and response surface methodology (RSM) in modelling CO2 laser machining. *Jordan Journal of Mechanical and Industrial Engineering*, 8(1), 35-42.
- Škvára, F.; Kopecký, L.; Namecek, J.; Bittnar, Z.D.E.N.Ì.K. (2006). Microstructure of geopolymer materials based on fly ash. *Ceramics-Silikaty*, *50*(4), 208-215.
- Songpiriyakij, S.; Kubprasit, T.; Jaturapitakkul, C.; Chindaprasirt, P. (2010). Compressive strength and degree of reaction of biomass- and fly ashbased geopolymer. *Construction and Building Materials*, *24*, 236-240.
- Sperazza, M.; Moore, J. N.; Hendrix, M. S. (2004). High-resolution particle size analysis of naturally occurring very fine-grained sediment through laser diffractometry: research methods papers. *Journal of Sedimentary Research*, 74(5), 736-743.
- Stuart, B. (2005). Organic molecules. In *Infrared Spectroscopy: Fundamentals and Applications* (p. 83). John Wiley & Sons, Inc.
- Sun, L.; Gong, K. (2001). Silicon-based materials from rice husks and their applications. *Industrial & Engineering Chemistry Research*, 40(25), 5861-5877.
- Tarley, C. R. T.; Silveira, G.; dos Santos, W. N. L.; Matos, G. D.; da Silva, E. G. P.; Bezerra, M. A.; Miro, M.; Ferreira, S. L. C. (2009). Chemometric tools in electroanalytical chemistry: Methods for optimization based on factorial design and response surface methodology. *Microchemical Journal*, *92*(1), 58-67.
- Temuujin, J.; Minjigmaa, A.; Rickard, A.; Lee, M.; Williams, I.; van Riessen, A. (2010). Fly ash based geopolymer thin coatings on metal substrates and its thermal evaluation. *Journal of Hazardous Materials*, 180(1), 748-752.
- Temuujin, J.; Minjigmaa, A.; Rickard, W.; Lee, M.; Williams, I.; van Riessen, A. (2009). Preparation of metakaolin based geopolymer coatings on metal substrates as thermal barriers. *Applied Clay Science*, *46*(3), 265-270.
- Temuujin, J.; Minjigmaa, A.; Rickard, W.; van Riessen, A. (2012). Thermal properties of spray-coated geopolymer-type compositions. *J Therm Anal Calorim*, 107, 287-292.
- Temuujin, J.; Rickard, W.; Lee, M.; van Riessen, A. (2011). Preparation and thermal properties of fire resistant metakaolin-based geopolymer-type coatings. *Journal of Non-Crystalline Solids*, 357(5), 1399-1404.
- Temuujin, J.; Rickard, W.; Lee, M.; van Riessen, A. (2011). Preparation and thermal properties of fire resistant metakaolin-based geopolymer-type coatings. *Journal of Non-Crystalline Solids*, *357*(5), 1399-1404.
- Temuujin, J.; van Riessen, A. (2009). Effect of fly ash preliminary calcination on the properties of geopolymer. *Journal of Hazardous Materials*, *164*(2), 634-639.
- Temuujin, J.; Williams, R. P.; Van Riessen, A. (2009). Effect of mechanical activation of fly ash on the properties of geopolymer cured at ambient temperature. *Journal of Materials Processing Technology*, 209(12), 5276-5280.

- Terbendalir, S. L.; Taib, M. R. B. (2007). *Production of Amorphous Silica From Rice Husk in Fluidised Bed System.* Universiti Teknologi Malaysia: Faculty of Chemical Engineering and Natural Resources Engineering.
- Tewarson, A.; Macaione, D. P. (1993). Polymers and composites-an examination of fire spread and generation of heat and fire products. *Journal of Fire Sciences*, *11*(5), 421-441.
- Thuc, C. N. H.; Thuc, H. H. (2013). Synthesis of silica nanoparticles from Vietnamese rice husk by sol–gel method. *Nanoscale Research Letters*, 8(1), 1-10.
- Tsai, C. W.; Tong, L. I.; Wang, C. H. (2010). Optimization of multiple responses using data envelopment analysis and response surface methodology. *Tamkang Journal of Science and Engineering*, 13(2), 197-203.
- Tsai, P. W.; Gilmour, S. G.; Mead, R. (2006). Statistical isomorphism of three-level fractional factorial designs. *Utilitas Mathematica*, *70*, 3-10.
- UL 1709. (2011). Standard for rapid rise fire tests of protection materials for structural steel.
- Ullah, S.; Ahmad, F.; Shariff, A. M.; Bustam, M. A. (2014). Synergistic effects of kaolin clay on intumescent fire retardant coating composition for fire protection of structural steel substrate. *Polymer Degradation and Stability*, 110, 91-103.
- Ullah, S.; Ahmad, F.; Yusoff, P. S. M. (2013). Effect of boric acid and melamine on the intumescent fire-retardant coating composition for the fire protection of structural steel substrates. *Journal of Applied Polymer Science*, 128(5), 2983-2993.
- Umesh, G.; Mallikarjun, B.; Ramesh, C. S. (2015). Analysis of bending stresses on coating materials by experimental and FE method. *International Journal of Mechanical and Industrial Technology*, 2(2), 41-50.
- Van Jaarsveld, J. G. S.; Van Deventer, J. S. J.; Schwartzman, A. (1999). The potential use of geopolymeric materials to immobilise toxic metals: Part II. Material and leaching characteristics. *Minerals Engineering, 12*(1), 75-91.
- Van Jaarsveld, J. G. S.; Van Deventer, J. S. J.; Lorenzen, L. (1997). The potential use of geopolymeric materials to immobilise toxic metals: Part I. Theory and applications. *Minerals Engineering*, *10*(7), 659-669.
- Van Jaarsveld, J. G. S.; Van Deventer, J. S. J.; Lukey, G. C. (2002). The effect of composition and temperature on the properties of fly ash-and kaolinite-based geopolymers. *Chemical Engineering Journal*, 89(1), 63-73.
- van Jaarsveld, J.G.S.; va Deventer, J.S.J; Lukey, G.C. (2003). The characterisation of source materials in fly ash-based geopolymers. *Materials Letters*, *57*(7), 1272-1280.
- van Riessen, A. (2007). Thermo-mechanical and microstructural characterisation of sodium-poly (sialate-siloxo)(Na-PSS) geopolymers. *Journal of Materials Science*, 42(9), 3117-3123.
- van Riessen, A.; Rickard, W. (2009). Thermal properties of geopolymers. In *Geopolymers: Structures, Processing, Properties and Industrial Applications* (pp. 315-342). Australia: Woodhead Publishing Limited.
- Vaou, V.; Panias, D. (2010). Thermal insulating foamy geopolymers from perlite. *Minerals Engineering*, 23(14), 1146-1151.

- Vilches, L. F.; Fernández-Pereira, C.; Olivares del Valle, J.; Rodríguez-Piñero, M.; Vale, J. (2002). Development of new fire-proof products made from coal fly ash: The CEFYR project. *Journal of Chemical Technology and Biotechnology*, 77(3), 361-366.
- Vilches, L. F.; Leiva, C.; Vale, J.; Fernández-Pereira, C. (2005). Insulating capacity of fly ash pastes used for passive protection against fire. *Cement and Concrete Composites*, *27*(7), 776-781.
- Vilches, L. F.; Leiva, C.; Vale, J.; Olivares, J.; Fernández-Pereira, C. (2007). Fire resistance characteristics of plates containing a high biomass-ash proportion. *Industrial & Engineering Chemistry Research, 46*(14), 4842-4829.
- Waijarean, N.; Asavapisit, S.; Sombatsompop, K.; MacKenzie, K. J. D. (2014). The effect of SiO2/Al2O3 ratios on the properties of geopolymers prepared from water treatment residue (WTR) in the presence of heavy metals. *Gmsarn International Journal*, 1(1.78), 97.
- Wang, G.; Yang, J. (2011). Influences of glass flakes on fire protection and water resistance of waterborne intumescent fire resistive coating for steel structure. *Progress in Organic Coatings, 70*(2), 150-156.
- Wang, H.; Li, H.; Yan, F. (2005). Synthesis and mechanical properties of metakaolinite-based geopolymer. *Colloids and Surface A: Physicochem. Eng. Apects*, *268*(1), 1-6.
- Wang, K.; Yan, P. S.; Cao, L. X. (2014). Chitinase from a novel strain of Serratia marcescens JPP1 for biocontrol of aflatoxin: Molecular characterization and production optimization using response surface methodology. *BioMed Research International*, 2014, 1-8.
- Wang, Z. Y.; Han, E. H.; Ke, W. (2007). Fire-resistant effect of nanoclay on intumescent nanocomposite coatings. *Journal of Applied Polymer Science*, 103(3), 1681-1689.
- Wang, Z.; Han, E.; Ke, W. (2005). Influence of nano-LDHs on char formation and fire-resistant properties of flame-retardant coating. *Progress in Organic Coatings*, 29-37.
- Warson, H.; Finch, C. A. (2001). Intumescent coatings. In Applications of Synthetic Resin Lattices Volume 2: Lattices in Surface Coatings; Emulsion Paints (p. 996). Chichester: John Wiley & Sons, Ltd.
- Wass, J. A. (2010). First steps in experimental design The screening experiment. *Journal of Validation Technology*, *16*(2), 49.
- Wastiels, J.; Wu, X.; Faignet, S.; Patfoort, G. (1993). Mineral polymer based on fly ash. *Proceedings of the Ninth International Conference on Solid Waste Management*. Philadelphia, USA.
- Wegman, R. F.; Tullos, T. R. (1992). Sandwich construction. In *Handbook of Adhesive Bonded Structural Repair* (p. 34). William Andrew.
- Wei, S.; Sun, Z. Y.; Zongjin, L. (2009). Preparation and microstructure of na-psds geopolymeric matrix. *Ceramics-Silikáty*, *53*(2), 88.
- Wennberg, D. (2011). Light-weighting methodology in rail vehicle design through introduction of load carrying sandwich panels. Design Department of Aeronautical and Vehicle Engineering.
- Wladyka-Przybylak, M.; Kozlowski, R. (1999). The thermal characteristics of different intumescent coatings. *Fire and Materials*, *23*(1), 33-43.
- Wu, C. J.; Hamada, M. S. (2009). Fundamental principles for factorial effects: Effect hierarchy, effect sparsity, and effect heredity. In *Experiments:*

- Planning, Analysis, and Optimization (pp. Chapter 4, Sec. 4.6). New Jersey: John Wiley & Sons.
- Wu, C. J.; Hamada, M. S. (2011). *Experiments: Planning, Analysis, and Optimization*. New Jersey: John Wiley & Sons.
- Xia, Y.; Yang, Z.; Mokaya, R. (2005). Synthesis of hollow spherical mesoporous N-doped carbon materials with graphitic framework. *Studies in Surface Science and Catalysis*, *156*, 565-572.
- Xu, H.; Van Deventer, J. S. (2003). Effect of source materials on geopolymerization. *Industrial & Engineering Chemistry Research*, *42*(8), 1698-1706.
- Xu, H.; Van Deventer, J. S. J. (2000). The geopolymerisation of alumino-silicate minerals. *International Journal of Mineral Processing*, *59*(3), 247-266.
- Yang, Gui-Rong; Song, Wen-ming; Lu, Jin-Jun; Hao, Yuan; Ma, Ying. (2008). Three-point bending behavior of surface composite Al2O3/Ni on bronze substrate produced by vacuum infiltration casting. *Journal of Material Processing Technology*, 202, 195-200.
- Yao, X.; Zhang, Z.; Zhu, H.; Chen, Y. (2009). Geopolymerization process of alkali-metakaolinite characterized by isothermal calorimetry. *Thermochimica Acta*, 493(1), 49-54.
- Yew, M. C.; Ramli Sulong, N. H.; Yew, M. K.; Amalina, M. A.; Johan, M. R. (2014). Fire propagation performance of intumescent fire protective coatings using eggshells as a novel biofiller. *The Scientific World Journal*, 2014.
- Yew, M. C.; Sulong, N. R. (2012). Fire-resistive performance of intumescent flame-retardant coatings for steel. *Materials & Design*, *34*, 719-724.
- Yong, S.L.; Feng, D.W.; Lukey, G.C.; van Deventer, J.S.J. (2007). Chemical characterisation of the steel–geopolymeric gel interface. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 302(1-3), 411-423.
- Zain , M. F. M.; Islam , M. N.; Mahmud, F.; Jamil, M. (2011). Production of rice husk ash for use in concrete as a supplementary cementitious material. *Construction and Building Materials*, 25(2), 798-805.
- Zhang, G.; Wang, B.; Ma, L.; Wu, L.; Pan, S.; Yang, J. (2014). Energy absorbtion and low velocity impact response polyurethane foam filled pyramidal lattice core sandwich panels. *Composite Structures*, *108*, 304-310.
- Zhang, H. Y.; Kodur, V.; Qi, S. L.; Cao, L.; Wu, B. (2014). Development of metakaolin–fly ash based geopolymers for fire resistance applications. *Construction and Building Materials*, *55*, 38-45.
- Zhang, L. P.; Yu, X. J.; Ge, Z. W.; Dong, Y. H.; Li, D. G.; Zhang, Y. L. (2011). Research on properties of SiC coating inert anode for aluminum electrolysis. *In Materials Science Forum, 686*, 623-629.
- Zhang, M.; El-Korchi, T.; Zhang, G.; Liang, J.; Tao, M. (2014). Zhang, M., El-Korchi, T., Zhang, G., Liang, J., & Tao, M. (2014). Synthesis factors affecting mechanical properties, microstructure, and chemical composition of red mud–fly ash based geopolymers. *Fuel, 134*, 315-325.
- Zhang, R.; Li, P.; Zhao, S.; Ai, M. (2008). A general minimum lower-order confounding criterion for two-level regular designs. *Statistica Sinica*, 1689-1705.
- Zhang, Z.; Yao, X.; Wang, H. (2012). Potential application of geopolymers as protection coatings for marine concrete III. Field experiment. *Applied Clay Science*, 67-68, 57-60.

- Zhang, Z.; Yao, X.; Zhu, H. (2010). Potential application of geopolymers as protection coatings for marine concrete: I. Basic properties. *Applied Clay Science*, *49*(1), 1-6.
- Zhao, Q.; Zhang, B.; Quan, H.; Yam, R. C.; Yuen, R. K.; Li, R. K. (2009). Flame retardancy of rice husk-filled high-density polyethylene ecocomposites. *Composites Science and Technology, 69*(15), 2675-2681.
- Zhou, W.; Yang, H. (2007). Flame retarding mechanism of polycarbonate containing methylphenyl-silicone. *Thermochimica Acta, 452*(1), 43-48.
- Zia-ul-Mustafa, M.; Ahmad, F.; Ullah, S.; Amir, N.; Gillani, Q. F. (2017). Thermal and pyrolysis analysis of minerals reinforced intumescentfire retardant coating. *Progress in Organic Coatings*, 201-216.
- Zinno, A.; Prota, A.; Di Maio, E.; Bakis, C. E. (2011). Experimental characterization of phenolic-impregnated honeycomb sandwich structures for transportation vehicles. *Composite Structures*, *93*(11), 2910-2924.
- Zolgharnein, J.; Shahmoradi, A.; Ghasemi, J. B. (2013). Comparative study of Box–Behnken, central composite, and Doehlert matrix for multivariate optimization of Pb (II) adsorption onto Robinia tree leaves. *Journal of Chemometrics*, 27(1-2), 12-20.
- Zuhua, Z.; Xiao, Y.; Huajun, Z.; Yue, C. (2009). Role of water in the synthesis of calcined kaolin-based geopolymer. *Applied Clay Science*, *43*(2), 218-223.

LIST OF PUBLICATIONS

Patent

Mohd Salahuddin, M. B.; Mustapha, F.; Norkhairunnisa, M.; Ishak, M. R. Fire Retardant Coating Composition. App. no. PI 2016701334, filing date. 12 April 2016.

Journals

- **Mohd Salahuddin, M. B.**; Norkhairunnisa, M.; Mustapha, F. (2015). A review on thermophysical evaluation of alkali-activated geopolymers. *Ceramics International*, *41*, 4273-4281. (Thomson Reuters, IF: 2.64)
- **Mohd Salahuddin, M. B.**; Mustapha, F.; Norkhairunnisa, M.; Ishak, M. R. (2017). Novel formulation of rice husk ash-based geopolymer binder coating and its flexural behavior. *Construction & Building Materials*. (submitted) (Thomson Reuters, IF: 3.24)
- Mohd Salahuddin Mohd Basri; Faizal Mustapha; Norkhairunnisa Mazlan; Mohamad Ridzwan Ishak. (2015). Fire retardant performance of rice husk ash-based geopolymer coated mild steel A factorial design and microstructure analysis. In Materials Science Forum, Trans Tech Publications, 841, 48-54. (Scopus)
- Basri, M. S. M.; Mazlan, N.; Mustapha, F. (2015). Effects of stirring speed and time on water absorption performance of blank epoxy and silica aerogel reinforced epoxy nanocomposite. *ARPN Journal of Engineering and Applied Sciences*, *10*, 9982-9991. (Scopus)

Proceeding

Mohd Salahuddin, M. B.; Faizal, M.; Norkhairunnisa, M., (2014). A review on the insulation technology of liquefied natural gas (LNG) pipe insulation system. The Postgraduate Symposium on Composites Science and Technology 2014 & 4th Postgraduate Seminar on Natural Fiber Composites 2014, IOI Palm Garden Hotel, Putrajaya, 80-85.



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