

UNIVERSITI PUTRA MALAYSIA

In Vitro EVALUATION OF COCKLE SHELL-BASED CALCIUM CARBONATE ARAGONITE POLYMORPH NANOPARTICLE WITH SURFACE FUNCTIONALIZATION FOR DRUG DELIVERY APPLICATIONS

SYAIRAH LIYANA BINTI MOHD ABD GHAFAR

IB 2016 25



In Vitro EVALUATION OF COCKLE SHELL-BASED CALCIUM CARBONATE ARAGONITE POLYMORPH NANOPARTICLE WITH SURFACE FUNCTIONALIZATION FOR DRUG DELIVERY APPLICATIONS



SYAIRAH LIYANA BINTI MOHD ABD GHAFAR

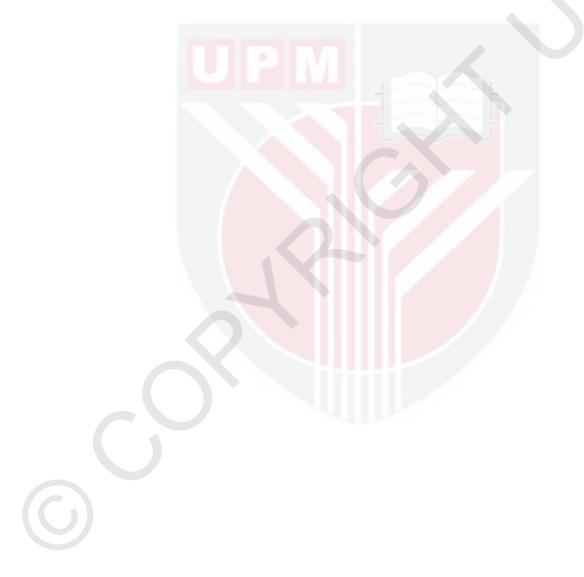
Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirements for the Degree of Master of Science

December 2016

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



Abstract of the thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Master of Science

In Vitro EVALUATION OF COCKLE SHELL-BASED CALCIUM CARBONATE ARAGONITE POLYMORPH NANOPARTICLE WITH SURFACE FUNCTIONALIZATION FOR DRUG DELIVERY APPLICATIONS

By

SYAIRAH LIYANA BINTI MOHD ABD GHAFAR



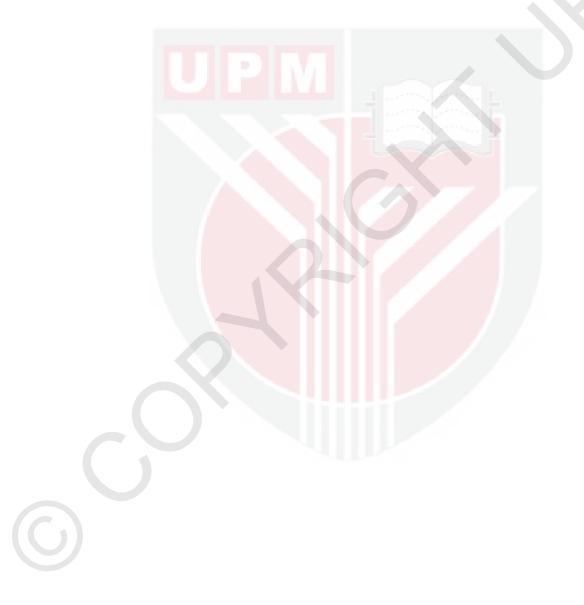
Chairman : Professor Md Zuki Abu Bakar @ Zakaria, PhD Institute : Bioscience

Drug delivery is a current biomedical application employing nano-sized particles. In line with current global interest, employing nature-based materials to construct delivery carriers has been highly preferable due to their environmental friendly, availability, low cost, and good natural mineral purity. Cockle (Anadara granosa) shells was reported to contain comparable mineral compositions to vertebrates bone with high calcium carbon and no evident presence of heavy metal elements with good quality and pure calcium carbonate aragonite crystals. In order to meet stringent qualities of drug carrier for drug delivery applications, an improved synthesis method incorporated with surface functionalization was developed to produce nanoparticles with high homogeneity in size and shape. The study aimed at evaluating the physicochemical characteristics of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticle and its potentials as delivery agent. Cockle shell micron-sized powder was converted into nano-sized particles through a mechanical stirring process in the presence of dodecyl dimethyl betaine (BS-12). The effect of BS-12 surfactant on the surface property of cockle shell-based calcium carbonate aragonite nanoparticles was analyzed through pH evaluation, Fourier Transform Infrared (FTIR) and X-ray diffraction (XRD) analyzes. Transmission Electron Microscopy (TEM) and Field Emission Scanning Electron Microscopy (FESEM) demonstrated agglomeration of nanoparticles after the addition of surfactant. However, with calcium ion adsorption onto the surface of cockle shell-based calcium carbonate aragonite, the dispersion of the nanoparticles has improved as shown by the increase in zeta potential. Purification technique further enhanced the overall distribution of nanoparticles towards more refined size range of less than 100 nanometers, which is favorable for drug delivery applications. The purity of aragonite phase and chemical functionality were verified by FTIR and X-ray diffraction (XRD) analyzes. In vitro biological response on human fetal osteoblast (hFOB 1.19) cell line demonstrated that surface functionalization could decrease cytotoxicity. Surface functionalized cockle shell-based calcium carbonate aragonite nanocarrier showed



better capacity to load drug molecules and was able to sustain incorporation of some drug molecules up to three days *in vitro*. Both the cockle shell-based nanocarrier samples were sensitive to pH changes as they released more drug compounds in the acidic environment of pH 6.4 than pH 7.4. This new delivery agent from cockle shells may provide an alternative source for calcium carbonate aragonite polymorph nanoparticles as an efficient drug carrier for therapeutic applications.

Keywords: cockle shell, calcium carbonate, surface functionalization, drug delivery



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

PENILAIAN *In Vitro* PARTIKEL NANO KALSIUM KARBONAT POLIMORF ARAGONIT DARIPADA KULIT KERANG DENGAN MODIFIKASI PERMUKAAN UNTUK APLIKASI PEMBAWA UBAT

Oleh

SYAIRAH LIYANA BINTI MOHD ABD GHAFAR

Disember 2016

Pengerusi : Profesor Md Zuki Abu Bakar @ Zakaria, PhD Institut : Biosains

Pembawa ubat adalah aplikasi bioperubatan terbaharu yang menggunakan zarah bersaiz nano. Selaras dengan kepentingan global, penggunaan bahan daripada alam semula jadi adalah lebih baik oleh kerana faktor mesra alam, ketersediaan sumber, kos yang rendah, dan memiliki komponen mineral semula jadi yang murni. Kulit kerang (Anadara granosa) memiliki komposisi mineral kalsium karbon yang tinggi yang hampir menyamai tulang vertebra, tidak mengandungi elemen logam berat serta mempunyai kandungan kristal kalsium karbonat aragonit yang berkualiti baik dan tulen. Suatu kaedah penghasilan partikel nano yang dipertambahbaik telah dicipta untuk menghasilkan partikel nano yang mempunyai saiz dan bentuk yang seragam untuk tujuan aplikasi pembawa ubat. Kajian ini dilakukan untuk menilai sifat fizikal, kimia serta potensi partikel nano kalsium karbonat kulit kerang yang dimodifikasi sebagai pembawa ubat. Kulit kering yang telah diproses menjadi serbuk halus bersaiz mikron akan dihasilkan menjadi partikel nano melalui suatu proses mekanikal dengan menggunakan bahan surfaktan iaitu dodecyl dimethyl betaine (BS-12). Kesan penggunaan bahan surfaktan terhadap sifat permukaan partikel nano kulit kerang telah dianalisis melalui pemeriksaan pH, analisis FTIR, dan XRD. Pemeriksaan mikroskop elektron TEM dan FESEM mendapati bahawa penggumpalan partikel nano terjadi setelah penambahan bahan surfaktan. Namun begitu, proses modifikasi permukaan telah menyebabkan berlakunya adsorpsi ion kalsium pada permukaan partikel nano kulit kerang sehingga ia dapat memperbaiki penyebaran partikel yang telah diindikasikan oleh peningkatan nilai potensi zeta. Teknik penulenan yang dianjurkan juga telah berjaya menghasilkan partikel nano yang mempunyai distribusi saiz yang sesuai untuk aplikasi pembawa ubat yaitu kurang daripada 100 nanometer. Pemeriksaan FTIR dan XRD telah mengesahkan ketulenan kristal aragonit dan fungsi kimianya. Hasil penilaian in vitro terhadap sel tulang fetus manusia (hFOB 1.19) mendapati bahawa proses modifikasi yang dilakukan telah berjaya mengurangi ketoksikan bahan terhadap sel. Partikel nano kalsium karbonat kulit kerang yang dimodifikasi juga telah menunjukkan kebolehan membawa ubat yang lebih baik serta ia berjaya mempertahankan kandungan molekul ubat sehingga tiga hari secara in vitro.



Kedua jenis partikel nano kulit kerang dengan atau tanpa modifikasi bersifat sensitif terhadap perubahan pH di mana, kedua-duanya melepaskan komponen ubat lebih cepat dalam persekitaran berasid yaitu pada pH 6.4 berbanding pH7.4. Partikel nano kalsium karbonat aragonit berasaskan kulit kerang ini dapat menjadi suatu sumber alternatif sebagai pembawa ubat yang efisien untuk pelbagai aplikasi terapi.

Kata kunci: kulit kerang, kalsium karbonat, mofikasi permukaan, pembawa ubat



ACKNOWLEDGEMENTS

First of all, I thank Allah S.W.T for giving me strength and ability to complete the study. I am sincerely grateful to Professor Dr. Md Zuki Abu Bakar @ Zakaria, Deputy Director of Bioscience Institute (IBS) and Professor lecturer of Anatomy Department at Faculty of Veterinary Medicine as the Chairman of supervisory committee, Professor Dr. Mohd Zobir Hussein, Programme Manager for Nanomaterials at Materials Synthesis and Characterization Laboratory (MSCL) of Advanced Technology Institute (ITMA) and Associate Professor Dr. Yaya Rukayadi, lecturer at Faculty of Food Science and Technology and Natural Product Laboratory of Bioscience Institute (IBS) as the members of supervisory committee, for their understandings, support, guidance and invaluable advices throughout the study period and preparation of the thesis. I am greatly indebted and appreciate very much to my father, Mohd Abd Ghafar Mohd Ghazali for his encouragement, support and sacrifices throughout the whole study duration and also to all my dearest brothers and sister, big thanks for their encouragement during the study period. My sincerely appreciation also goes to all members of my research teams for their kind assistance and cooperation while conducting the research and for putting colors in my life and yet, may Allah bless you all. Last but not least, I wish to express my sincere thanks to those who have one way or another helped me accomplishing this study a success.

I certify that a Thesis Examination Committee has met on 8 December 2016 to conduct the final examination of Syairah Liyana binti Mohd Abd Ghafar on her thesis entitled "*In Vitro* Evaluation of Cockle Shell-Based Calcium Carbonate Aragonite Polymorph Nanoparticle with Surface Functionalization for Drug Delivery Applications" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Goh Yong Meng, PhD Associate Professor Faculty of Veterinary Medicine Universiti Putra Malaysia (Chairman)

Rasedee @ Mat bin Abdullah, PhD Professor Faculty of Veterinary Medicine Universiti Putra Malaysia (Internal Examiner)

Ahmad Hafiz bin Zulkifly, PhD Professor International Islamic University Malaysia Malaysia (External Examiner)

NOR AINI AB. SHUKOR, PhD Professor and Deputy Dean School of Graduate Studies Universiti Putra Malaysia

Date: 22 March 2017

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Md. Zuki bin Abu Bakar @ Zakaria, PhD

Deputy Director of Institute of Bioscience Professor of Faculty of Veterinary Medicine Universiti Putra Malaysia (Chairman)

Mohd Zobir Hussein, PhD

Professor Institute of Advanced Technology Universiti Putra Malaysia (Member)

Yaya Rukayadi, PhD

Associate Professor Institute of Bioscience Universiti Putra Malaysia (Member)

ROBIAH BINTI YUNUS, PhD Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any other instituitions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writtings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/ fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection dsoftware.

Signature:

Date:

Name and Matric No.: Syairah Liyana binti Mohd Abd Ghafar, GS 36560

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

| Signature: Name of Chairman of Supervisory Committee: | Professor Dr. Md. Zuki bin Abu Bakar @ Zakaria |
|--|---|
| | |
| | |
| | |
| Signature: | |
| Name of Member | |
| of Supervisory | Professor |
| Committee: | Dr. Mohd Zobir Hussein |
| | |
| Signature: | |
| Name of Member | |
| of Supervisory | Associate Professor |
| Committee: | Dr. Yaya Rukayadi |
| | |

TABLE OF CONTENTS

Page

| ABSTRACT | i |
|-----------------------|------|
| ABSTRAK | iii |
| ACKNOWLEDGEMENTS | V |
| APPROVAL | vi |
| DECLARATION | viii |
| LIST OF TABLES | xiii |
| LIST OF FIGURES | xiv |
| LIST OF ABBREVIATIONS | xvi |
| | |

CHAPTER

| 1 | GEN | ERAL INTRODUCTION | 1 | | |
|---|------|--|----|--|--|
| 2 | LITE | RATURE REVIEW | | | |
| | 2.1 | Nanotechnology | 6 | | |
| | 2.2 | Nanomedicine | 7 | | |
| | 2.3 | Drug Delivery System | 7 | | |
| | 2.4 | Drug Carrier | 8 | | |
| | | 2.4.1 Nanocarrier | 9 | | |
| | 2.5 | Biomaterials | 9 | | |
| | | 2.5.1 Calcium Carbonate Biomaterial | 10 | | |
| | 2.6 | Calcium Carbonate Aragonite Polymorph | 11 | | |
| | | 2.6.1 Calcium Carbonate Aragonite Polymorph | | | |
| | | Derived from Cockle Shell | 12 | | |
| | 2.7 | Calcium Carbonate in Biomedical Applications | 13 | | |
| | 2.8 | Cockle Shell-based Calcium Carbonate Aragonite | | | |
| | | Polymorph as Biomedical Tools | 13 | | |
| | 2.9 | Synthesis of Cockle Shell-based Calcium | | | |
| | | Carbonate Aragonite Polymorph Nanoparticle | 14 | | |
| | 2.10 | Dodecyl Dimethyl Betaine (BS-12) | 16 | | |
| | 2.11 | Nano-sized Drug Carrier | 17 | | |
| | 2.12 | Round and Porous Drug Carrier | 18 | | |
| | 2.13 | Surface Property of Carriers | 19 | | |
| | 2.14 | Surface Functionalization | 21 | | |
| | 2.15 | In vitro Cytotoxicity Studies | 22 | | |
| | 2.16 | Ketoprofen Lysinate as Model Drug | 23 | | |
| | 2.17 | Drug Loading | 23 | | |
| | 2.18 | Drug Release | 25 | | |
| | | | | | |

3 MATERIALS AND METHODS 3.1

| | | |) | | |
|---------|-----------|------------|------------------|-----------|----|
| Synthes | is of | Cockle | Shell-based | Calcium | |
| Carbona | ate Arago | onite Poly | morph Nanopa | rticles | 26 |
| 3.1.1 l | Preparati | on of Mic | ron-sized partie | eles | 26 |
| 3.1.2 l | Preparati | on of Coc | kle Shell-based | d Calcium | |
| (| Carbonat | e Aragoni | te Nanoparticle | es | 27 |
| | | | | | |

| 3.1.3 | Preparation of Surface Functionalized | |
|-----------------|---|----------|
| | Cockle Shell-based Calcium Carbonate | |
| | Aragonite Nanoparticles | 27 |
| 3.1.4 | Purification Procedure | 28 |
| 3.1.5 | Experimental Control | 28 |
| | zation of Surface Functionalized Cockle | |
| Shell-base | | |
| Nanopartic | 8 | |
| 3.2.1 | Size, Morphology, and Distribution of | |
| 5.2.1 | Nanoparticles | 28 |
| | 3.2.1.1 Electron Microscopic Evaluation | |
| | 3.2.1.1.1 Transmission Electron | |
| | Microscopy (TEM) | 29 |
| | 3.2.1.1.2 Field Emission Scanning | |
| | Electron Microscopy | 29 |
| | (FESEM) | |
| | 3.2.1.2 Zetasizer Size Distribution | 29 20 |
| 3.2.2 | Surface Functionalization | 29 |
| 3.2.3 | Surface Properties and Calcium Carbonate | |
| | Polymorphism | 30 |
| | 3.2.3.1 pH Measurement | 30 |
| | 3.2.3.2 Fourier Transform Infrared (FTIR) | |
| | spectrometry | 30 |
| | 3.2.3.3 X-ray Diffraction (XRD) | 30 |
| 3.2.4 | Zeta Potential Measurement | 31 |
| 3.2.5 | In vitro 3-[4,5-dimethylthiazol-2-yl]-2, 5- | |
| | diphenyltetrazolium bromide (MTT) Cell | |
| | Viability Assay | 21 |
| | 3.2.5.1 Cell Line and Culture Conditions | 31 |
| | 3.2.5.2 Cell Seeding and Treatments | 31 |
| | 3.2.5.3 3-[4,5-dimethylthiazol-2-yl]-2 , 5- | |
| | diphenyltetrazolium bromide (MTT) | |
| | Cell-based Assay Procedure | 32 |
| 3.3 Nanopartic | cle-Drug Evaluation | |
| 3.3.1 | Characterization of Nanoparticle-Drug | |
| | Complex | 32 |
| | 3.3.1.1 Transmission Electron Microscopy | |
| | (TEM) | 32 |
| | 3.3.1.2 X-ray Diffraction (XRD) | 32 |
| | 3.3.1.3 Fourier Transform Infrared (FTIR) | |
| | Spectrometry | 33 |
| 3.3.2 | Preparation of Nanoparticle-Drug Loaded | 33 |
| | 3.3.2.1 Procedure to Construct Ketoprofen | |
| | Lysinate Standard Curve | 33 |
| | 3.3.2.2 Determination of Drug Loading | |
| | Contents and Efficiencies | 34 |
| 3.3.3 | Nanocarrier-Drug Complex Release Studies | 34 |
| 3.4 Statistical | | 35 |
| | <i>j</i> | ~~ |

4 RESULTS AND DISCUSSION

| • | 4.1 Evalua | ation of Cockle Shell-based Calcium | |
|------------|-------------------|--|------------|
| | Carbo | | |
| | Synthe | | 36 |
| | | cterization of Surface Functionalized Cockle based Calcium Carbonate Aragonite | |
| | | based Calcium Carbonate Aragonite norph Nanoparticles | |
| | 4.2.1 | Size, Morphology, and Distribution of | |
| | 4.2.1 | Nanoparticles | 36 |
| | | 4.2.1.1 Electron Microscopic Evaluation | 36 |
| | | 4.2.1.2 Zetasizer Size Distribution | |
| | | Evaluation | 48 |
| | 4.2.2 | Surface Functionalization Evaluation | 51 |
| | 4.2.3 | Surface Properties and Calcium Carbonate | |
| | | Polymorphism Evaluation | 52 |
| | | 4.2.3.1 pH Evaluation | 53 |
| | | 4.2.3.2 Fourier Transform Infrared (FTIR) | |
| | | Spectrometry | 53 |
| | 121 | 4.2.3.3 X-ray Diffraction (XRD) | 57 |
| | 4.2.4 4.2.5 | Zeta Potential Evaluation In vitro Evaluation of Surface | 59 |
| | 4.2.3 | <i>In vitro</i> Evaluation of Surface Functionalized Cockle Shell-based Calcium | |
| | | Carbonate Aragonite Nanoparticle as a Drug | |
| | | Carrier | 60 |
| | | 4.2.5.1 In vitro MTT Cytotoxicity | |
| | | Evaluation | 60 |
| | | | |
| | - | cle-Drug Evaluation | |
| | 4.3.1 | Characterization of Nanoparticle-Drug | |
| | | Complex | 64 |
| | | 4.3.1.1 Transmission Electron Microscopy | C 1 |
| | | (TEM) | 64 |
| | | 4.3.1.2 X-ray Diffraction (XRD) | 66 |
| | | 4.3.1.3 Fourier Transform Infrared (FTIR) Spectrometry | 67 |
| | 4.3.2 | Drug Loading Evaluation | 07 |
| | 1.3.2 | 4.3.2.1 Standard Calibration Curve of | |
| | | Ketoprofen Lysinate | 69 |
| | | 4.3.2.2 Loading Efficiencies and Contents | 69 |
| | 4.3.3 | Drug Release Evaluation | 71 |
| 5 | GENERAL | DISCUSSION, CONCLUSION AND | |
| | RECOMME | NDATION | 74 |
| REFERENC | CES | | 78 |
| APPENDIC | ES | | 88 |
| | DF STUDENT | | 89 |
| LIST OF PU | BLICATION | S | 90 |

LIST OF TABLES

| Та | Table | | |
|-----|---|----|--|
| 4.1 | Diameter size and polydispersity index of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles. | 48 | |
| 4.2 | Measurement of calcium ion concentrations during five days surface modification. | 51 | |
| 4.3 | pH measurements of samples. | 53 | |
| 4.4 | Zeta potential of (a) cockle shell-based calcium carbonate nanoparticle without surface functionalization, (b) surface functionalized cockle shell-based calcium carbonate nanoparticles, and (c) calcium carbonate precipitate. | 60 | |
| | | | |
| | | | |

LIST OF FIGURES

| | Figur | Page | |
|-----|-------|---|----|
| | | Schematic orientation of carbonate groups in calcium carbonate molecules (a) hydrated calcite and (b) hydrated aragonite. | 10 |
| | 2.2 | Molecular structure of dodecyl dimethyl betaine. | 16 |
| | 3.1 | Cockle shells (a) before and (b) after cleaning with metal brush. | 27 |
| | 4.1 | TEM micrograph of cockle shell-based calcium carbonate aragonite polymorph nanoparticles without filtration. | 37 |
| | | Comparison between cockle shell-based calcium carbonate aragonite nanoparticles after filtration (a) without surface modification and (b) after surface modification. | 38 |
| | | The observation of cockle shell-based calcium carbonate aragonite nanoparticles dispersion after surface functionalization. | 39 |
| | 4.4 | TEM micrograph of single surface functionalized cockle shell- based calcium carbonate aragonite nanoparticle. | 39 |
| | 4.5 | FESEM micrographs of cockle shell-based calcium carbonate aragonite nanoparticles before surface modification. | 41 |
| | | FESEM micrographs of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles. | 42 |
| | 4.7 | Overview of FESEM micrograph (a) cockle shell-based calcium carbonate aragonite nanoparticles before surface modification and (b) surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles. | 45 |
| | | FESEM micrograph of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles after filtration. | 46 |
| (C) | | FESEM micrographs of calcium carbonate nanoparticles produced from precipitation. | 47 |
| | | Plots of size distributions by intensity (%) against diameter size of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles (a) before and (b) after purification procedure. | 48 |
| | 4.11 | Cumulative distribution by intensity (%) against diameter size of surface functionalized cockle shell-based calcium carbonate aragonite nanoparticles. | 49 |

| 4.12 Ad | sorption of calcium ions during surface functionalization. | 52 |
|--------------|---|----|
| add (c) | IR spectra of samples (a) Micron-sized particles before ition of BS-12 (b) Nano-sized particles after BS-12 addition Surface functionalized cockle shell-based calcium carbonate oparticles, and (d) Dodecyl dimethyl betaine (BS-12). | 55 |
| cart surf | ray diffraction patterns of (a) cockle shell-based calcium bonate nanoparticle before surface functionalization (b) face functionalized cockle shell-based calcium carbonate, and calcium carbonate from precipitation. | 58 |
| cyto shel | <i>vitro</i> human osteoblast cell line (hFOB1.19) cell line otoxicity study of blank (a) surface functionalized cockle ll-based calcium carbonate aragonite and (b) without surface dification nanoparticles. | 62 |
| cyto | <i>vitro</i> human osteoblast cell line (hFOB1.19) cell line btoxicity studies on ketoprofen lysinate free drug with (a) CSD (b) CXSD. | 63 |
| 4.17 TEN | M micrographs of (a) CSD and (b) CXSD. | 65 |
| lysi | nparison of X-ray diffraction patterns on (a) ketofropen nate, (b) CSD, and (c) blank surface functionalized cockle ll-based calcium carbonate aragonite nanocarrier. | 66 |
| and | ray diffraction spectra of (a) ketofropen lysinate, (b) CXSD, (c) blank cockle shell-based calcium carbonate aragonite nout surface modification nanocarrier. | 67 |
| 4.20 FTI | R spectra of (a) ketoprofen lysinate, (b) CXSD, and (c) CSD. | 68 |
| | ndard calibration graph of absorbance against ketoprofen nate concentrations. | 69 |
| | ding efficiencies and contents of (a) CSD and (b) CXSD aplexes. | 70 |
| | <i>vitro</i> release profile of ketoprofen lysinate from (a) CSD and CXSD complexes at pH 7.4 in phosphate buffered saline. | 72 |
| | <i>vitro</i> release profile of ketoprofen lysinate from (a) CSD and CXSD complexes at pH 6.4 in phosphate buffered saline. | 72 |

| $\mathbf{X}\mathbf{V}$ | |
|------------------------|--|

LIST OF ABBREVIATIONS

| ANOVA | Analysis of variance |
|--------------------------------------|---|
| ATCC | American type culture collection |
| BS-12 | Dodecyl dimethyl betaine |
| CaCl ₂ .2H ₂ O | Calcium chloride dihydrate |
| CaCO ₃ | Calcium carbonate |
| CSD | surface functionalized cockle shell-based calcium carbonate aragonite nanoparticle loaded drug complex |
| CXSD | cockle shell-based calcium carbonate aragonite without surface modification nanoparticle loaded drug complex |
| DMEM | Dulbecco's modified eagle medium |
| DMSO | Dimethyl sulfoxide |
| FBS | Fetal bovine serum |
| FESEM | Field emission scanning electron microscopy |
| FTIR | Fourier transform infrared spectrophotometer |
| JCPDS | Joint committee of powder diffraction society |
| mL | milliliters |
| MTT | 3-dimethylthiazo-2, 5-diphynyltetrazolium bromide |
| nm | Nanometers |
| PBS | Phosphate buffered solution |
| ppm | parts per million |
| SD | Standard deviation |
| TEM | Transmission electron microscopy |
| μm | Micrometers |
| XRD | X-ray diffractometer |

CHAPTER 1

GENERAL INTRODUCTION

Nanotechnology is no longer a new branch of studies in the science. Generally, nanotechnology does not focus on any specific area of study but it is actually a multidisciplinary field of sciences that covers and greatly influences many areas in our lives. It has successfully contributed a myriad of advancements in the development of various industries including in construction, aerospace defense, engineering, medicine, pharmaceuticals, cosmetics, sports, electronics, automotive, and chemical industries. Some products and applications pertaining to nanotechnology, such as cosmetics and pharmaceutical products, are in fact already available in the market (Mishra *et al.*, 2010; Parveen *et al.*, 2012; Silpa *et al.*, 2012).

There are many definitions of nanotechnology that was made to elaborate the term precisely. For example, the National Nanotechnology Initiative (NNI) by the National Science and Technology Council of the United States government defines nanotechnology as "an encompassing term for nanoscale science, engineering, and technology that involves imaging, measuring, modeling, and manipulating matter via the understanding and control of matter at dimensions between approximately 1 and 100 nanometers, where unique phenomena enable novel applications" (NSTC, 2004). The Malaysian Ministry of Science, Technology, and Innovation, via its National Nanotechnology Directorate Division (NND) in July 2010, has come out with a simpler definition of nanotechnology that is "the creation and utilization of material, devices and systems through the manipulation of matter at scales of less than 100 nanometers" (MOSTI, 2010). Although those definitions come from different sources, their essential objective is directed towards exploitation of nano-sized matter for advanced innovation in various applications.

Implementation of nanotechnology that basically focuses on the medical science field is known as nanomedicine. The term was defined as "monitoring, repair, construction and control of human biological systems at the molecular level using engineered nanodevices and nanostructures" (Robert, 1999). Some applications of nanotechnology in medicine are promising and have indeed offered many advantages in various medical areas such as drug delivery to specific sites, gene delivery in gene targeted therapy, molecular imaging for diagnostic and therapeutic intervention, cardiac therapy, dental care as well as orthopedic applications (Sahoo *et al.*, 2007; Robert, 2009; Mishra *et al.*, 2010). In fact, numerous applicable nanotechnology-based biomedical devices are already available in the current market and there are also some products undergoing clinical trials (Sahoo *et al.*, 2007; Parveen *et al.*, 2012).

Drug delivery is one of the current biomedical advances employing nano-sized particles for either diagnostic or therapeutic purposes based on nanomedicine concept. Ideally, drug delivery system is for the transport drug molecules to particular sites without affecting normal tissues or other internal organs (Zhi-Ping *et al.*, 2006; Amir

and Peter, 2009; Mishra *et al.*, 2010; Parveen *et al.*, 2012). On the other hand, there are some limitations encountered by the conventional drug delivery system. For instance, direct delivery by conventional system might experience problem relating to enzymatic degradation of drugs compounds before they could safely arrive at the specified targets (Zhi-Ping *et al.*, 2006; Rajesh and James, 2009; Amir and Peter, 2009). Besides, some therapeutically active molecules could also experience difficulty in reaching arduous sites of action, attaining optimal therapeutic level while targeting desired location, and controlling drug release (Zhi-Ping *et al.*, 2006; Sahoo *et al.*, 2007; Amir and Peter, 2009; Rajesh and James, 2009; Parveen *et al.*, 2012).

To date, extensive efforts was devoted in the search for an appropriate key to overcome some of the problems of drug delivery via research and development programs (Ueno *et al.*, 2005; Rajesh and James, 2009; Jae-Hyung *et al.*, 2010; Parveen *et al.*, 2012; Anil *et al.*, 2012; Sang-Kyoon *et al.*, 2012; Yulia *et al.*, 2013; Shafiu *et al.*, 2013). The demanding exploration for delivery agents remains ongoing covering many aspects of research including on types, physical, and chemical properties of materials as well as surface characterization of delivery carriers. In fact, many materials were intensively studied in the construct of efficient and effective carriers including inorganic materials, carbon tube, gold, silver, and polymer-based materials (Ueno *et al.*, 2005; Mahendra *et al.*, 2009; Chen *et al.*, 2011; Satya *et al.*, 2011; Forrest *et al.*, 2011; Anil *et al.*, 2012; Folusho and Richard, 2012; Shafiu *et al.*, 2013). Although there are variety choices of materials, calcium carbonate represents one of the inorganic materials of choice used to devise the delivery vehicles in drug delivery system (Ueno *et al.*, 2005; Dong *et al.*, 2012; Sang-kyoon *et al.*, 2012; Shafiu *et al.*, 2013; Yulia *et al.*, 2013; Ping *et al.*, 2013).

A large number of studies was also conducted to investigate some properties of the calcium carbonate material (Jiaguo *et al.*, 2004; Zeshan and Yulin, 2004; Ueno *et al.*, 2005; Guowei *et al.*, 2009; Meng-Chun and Clifford, 2010; Hoang-Vinh *et al.*, 2010; Kwang-Min and Kazuyuki, 2011; Sargheini *et al.*, 2012; Dong *et al.*, 2012; Nobuyoshi *et al.*, 2013). Calcium carbonate consists of three kind polymorphs which are calcite, aragonite, and vaterite. Each of these polymorphisms possesses different properties that define their special characteristics. Currently, the physical and chemical properties of calcium carbonate polymorphisms have been established (Kamiya *et al.*, 1977; Sohnel and Mullin, 1982; Laifeng *et al.*, 1999; Mitsutaka, 2002; Xiang *et al.*, 2004; Chengyu *et al.*, 2006; Rizzuti and Lionelli, 2008; Zeshan *et al.*, 2009; Hongxia *et al.*, 2011; Sargheini *et al.*, 2012; Nobuyoshi *et al.*, 2013).

Aragonite polymorph of calcium carbonate has high potential as a good biomedical material that could be integrated, resolved, and replaced by bones owing to its some unique characteristics (Stupp and Braun, 1997; Chengyu *et al.*, 2006; Islam *et al.*, 2011). Although aragonite is thermodynamically less stable than calcite at ambient temperature and pressure, but it is denser than calcite, sensitive towards temperature changes, possesses beneficial properties such as high mechanical strength, biocompatibility, and biodegradability (Stupp and Braun, 1997; Zuki *et al.*, 2004; Chengyu *et al.*, 2006; Awang-Hazmi *et al.*, 2007; Islam *et al.*, 2011; Shafiu *et al.*, 2013). For those reasons, this material has become a current focus in many fields of

biomedical research including drug delivery and bone tissue engineering (Zuki *et al.*, 2011; Shafiu *et al.*, 2013).

Surface functionalization can improve the interaction between carriers and specific cell membranes through manipulation on the surface charge of particles and thus help the carrier binding to targeted locations (Zhi-Ping *et al.*, 2006; Amir and Peter, 2009; Rajesh and James, 2009). In fact, the interaction among particles as well as the interaction between nanoparticles and drug compounds in suspension can be enhanced through modifications of surface property of the particles to affect their surface energy and electrical potential hence improving the dispersion as well as distribution of nanoparticles in the body (Huang *et al.*, 1990; Xu *et al.*, 2005; Chengyu *et al.*, 2007; Hoang-vinh *et al.*, 2010; Ping *et al.*, 2013).

From the fabrication of micron-sized calcium carbonate particles in the early days up to the synthesis of desirable nano-sized particles today, many studies have used the bottom-up approach via the precipitation process, either through carbonation or solution route for the production of inorganic calcium carbonate raw material over the past few years (Chengyu *et al.*, 2006; Meng-Chun and Clifford, 2010; Hoang-Vinh *et al.*, 2010; Hongxia *et al.*, 2011; Sargheini *et al.*, 2012; Dong *et al.*, 2012; Yulia *et al.*, 2013). Many researchers recently started diverting their interest from synthetically methods to produce calcium carbonate to the utilization of nature-based biogenic materials using the top-down process of calcium carbonate nanoparticle production. In fact, a number of studies have derived calcium carbonate compound from naturally occurring byproduct of cockle shells for many purposes especially in biomedical applications (Zuki *et al.*, 2004; Awang-Hazmi *et al.*, 2007; Islam *et al.*, 2011; Islam *et al.*, 2012; Shafiu *et al.*, 2013).

The cockle belongs to the *Anadara* granosa species, a type of sea molluscan that could be found in the intertidal sea and mudflat areas in the coastal region of Southeast Asian countries such as Malaysia, Indonesia, and Thailand (Awang-Hazmi *et al.*, 2007). In fact, several studies (Zuki *et al.*, 2004; Awang-Hazmi *et al.*, 2007; Islam *et al.*, 2011; Shafiu *et al.*, 2013; Hemabarathy *et al.*, 2014) have successfully shown that calcium carbonate derived from cockle shells is suitable and safe for biomedical applications. Besides the vast availability and abundance, cockle shells are convenient, economic and environmental-friendly resource of calcium carbonate. Furthermore, calcium carbonate biomaterial derived from naturally occurring cockle shells contains good quality and pure aragonite polymorph of calcium carbonate and therefore, it is a very suitable biomaterial source for application in biomedical and tissue engineering studies (Zuki *et al.*, 2004; Awang-Hazmi *et al.*, 2007; Islam *et al.*, 2011; Zuki *et al.*, 2011; Shafiu *et al.*, 2013).

A fundamental study by Islam *et al.* (2011) has established a basic preparation method to process the cockle shells into micron-sized powder and also developed a novel preparation method using simple mechanical stirring in the presence of Dodecyl dimethyl betaine or commonly known as BS-12. Shafiu *et al.* (2013) has also introduced another preparation method of cockle shell-based calcium carbonate

nanoparticles production using a high pressure homogenizer (HPH) technique via microemulsion route in the presence of polysorbate 80 (Tween 80) surfactant. Even though the technique has shown promising results for drug delivery application, but given considerable limitations such as requirement of complex and expensive equipments with high energy input to operate, the simple stirring method by Islam *et al.* (2011; 2012) is a more appealing and convenient option for facile preparation of calcium carbonate aragonite nanoparticles even for a large-scale production of calcium carbonate aragonite polymorph nanoparticles.

Adjuring to the recent trend in drug delivery system, the present research employed the method introduced by Islam *et al.* (2012) with some improvisation, in the synthesis of calcium carbonate aragonite polymorph nanoparticle. The nanoparticle production method was also designed to incorporate surface functionalization process in order to achieve the required bioactivity and efficiency as a delivery carrier in terms of size, morphology, surface property, drug loading and release while attaining low cytotoxic towards biological systems.

Problem Statements

In line with current global interest, employing nature-based materials to construct delivery carriers is highly preferable due to their environmental friendly, availability, low cost, and good natural mineral purity. Cockle (*Anadara granosa*) shells was reported to contain comparable mineral compositions to vertebrates bone with high calcium carbon and no evident presence of heavy metal elements with good quality and pure calcium carbonate aragonite crystals.

The synthesis of calcium carbonate nanoparticles from naturally occurring cockle shells in the presence of dodecyl dimethyl betaine was previously developed, but further research is required to improve the synthesized method for high homogeneity production of nanoparticles in size and shape for drug delivery application.

To the best of our knowledge, there is no published study investigating on the controlled surface of cockle shells-based calcium carbonate aragonite polymorph nanoparticles as present drug carrier. Therefore, the effect of surface functionalization on the cockle shells-based calcium carbonate aragonite polymorph nanoparticles was evaluated.

The investigation of physicochemical characteristics of surface functionalized cockle shell-based calcium carbonate aragonite polymorph nanoparticles is important to assess its potential qualities as a promising delivery vehicle in drug delivery systems. An efficient drug carrier should have low cytotoxicity with good abilities to load and release drugs. In this study, the cytocompatibility of surface functionalized cockle shells-based calcium carbonate aragonite polymorph nanoparticle and its efficacy to load and release drug were thus investigated.

Study Hypothesis

Surface functionalized cockle shell-based calcium carbonate aragonite polymorph nanoparticle has characteristics of an efficient drug nanocarrier in terms of zeta potentials, dispersion, and stability in solution, cytocompatibility, low cytotoxicity, as well as excellent drug loading and release.



This research aims to assess the potential of surface functionalized cockle shell-based calcium carbonate aragonite polymorph nanoparticles for a drug delivery system.

Specific objectives

- i. To prepare and evaluate the physical and chemical properties of surface functionalized cockle shell-based calcium carbonate aragonite polymorph nanoparticles for drug delivery.
- ii. To determine the *in vitro* efficacy of surface functionalized cockle shell-based calcium carbonate aragonite polymorph nanoparticles in terms of cytotoxicity, drug loading and release as a drug nanocarrier system.

REFERENCES

- American Type Culture Collection. 2011. MTT cell proliferation assay instruction guide. *American Type Culture Collection*, Manassas, VA, USA: 1-6.
- American Type Culture Collection. 2013. Product sheet guideline of hFOB 1.19 (ATCC® CRL11372 TM) cell line. *American Type Culture Collection*, Manassas, VA, USA: 1-3.
- Anacardio, R., Perilli, O., Bartolini, S., Gentile, M.M., Mazzeo, P. and Carlucci, G. 2003. Physicochemical compatibility between ketoprofen lysine salt injections (artrosilene[†]) and pharmaceutical products frequently used for combined therapy by intravenous administration. *Journal of Pharmaceutical and Biomedical Analysis*, 32: 1235-1241.
- Awang-Hazmi, A.J., Zuki, A.B.Z., Nordin, M.M., Jalila, A. and Norimah, Y. 2007. Mineral composition of the cockle (anadara granosa) shells of west coast of peninsular Malaysia and its potential as biomaterial for use in bone repair. *Journal* of Animal and Veterinary Advances, 6(5): 591–594.
- Bachhav, R.M. and Deore, S.N. 2015. A Review on —Nanomaterials. *International Journal of Science and Research*, 4(9): 1451-1457.
- Berlin, T.S. and Kabakov, A.V. 1961. Different in the electrokinetics potentials of carbonate sedimentary rocks of different origin and composistion. *Geochemistry*, 3: 217-230.
- Bharatham, H., Zakaria, A.B.Z., Perimal, E.K., Yusof, L.M. and Hamid, M. 2014. Mineral and Physiochemical Evaluation of Cockle Shell (*Anadara granosa*) and Other Selected Molluscan Shell as Potential Biomaterials. Sains Malaysiana, 43(7): 1023–1029.
- Bihari, P., Vippola, M., Schultes, S., Praetner, M., Khandoga, A.G., Reichell, C.S., Coester, C., Tuomi, T., Rehberg1, M. and Krombach, F. 2008. Optimized dispersion of nanoparticles for biological *in vitro* and *in vivo* studies. *Particle and Fibre Toxicology*, 5; doi:10.1186/1743-8977-5-14:14. Retrieved December 2015 from http://www.particleandfibretoxicology.com/content/5/1/14.
- Casanova, H. and Higuita, L.P. 2011. Synthesis of calcium carbonate nanoparticles by reactive precipitation using a high pressure jet homogenizer. *Chemical Engineering Journal*, 175: 569–578.
- Champion, J.A., Katare, Y.K. and Mitragotri, S. 2007. Particle shape: A new design parameter for micro- and nanoscale drug delivery carriers. *Journal of Controlled Release*, 121: 3–9.
- Chang, M.C and Tai, C.Y. 2010. Effect of the magnetic field on the growth rate of aragonite and the precipitation of caco3. *Chemical Engineering Journal*, 164: 1-9.

- Chen, S., Li, F., Zhuo, R.X. and Cheng, S.X. 2011. Efficient non-viral gene delivery mediated by nanostructured calcium carbonate in solution-based transfection and solid-phase transfection. *Molecular BioSystems*, 7: 2841–2847.
- Choi, K-M. and Kuroda, K. 2012. Polymorph Control of Calcium Carbonate on the Surface of Mesoporous Silica. *Crystal Growth and Design*, 12: 887–893.
- Chorny, M., Fishbein, I., Danenberg, H.D. and Golomb, G. 2002. Lipophilic drug loaded nanospheres prepared by nanoprecipitation: effect of formulation variables on size, drug recovery and release kinetics. *Journal of Controlled Release*, 83: 389–400.
- Dilanthi, H. M. D. R., Cabot, P. J., Shaw, P. N. and Hewavitharana, A. K. 2012. Study of beta endorphin metabolism in inflamed tissue, serum and trypsin solution by liquid chromatography-tandem mass spectrometric analysis. *Analytical and Bioanalytical Chemistry*, 402(6): 2089-2100.
- Ding, M., Zhou, L., Fu, X., Tan, H., Li, J. and Fu, Q. 2010. Biodegradable Gemini multiblock poly (E- caprolactone urethane)s towards controllable micellization. Supplementary material (ESI) for soft matter. The Royal Society of Chemistry; *Soft Matter*, 6: 2087-2092.
- Dobrovolskaia, M.A., Patri, A.K., Zheng, J., Clogston, J.D., Ayub, N., Aggarwal, P., Neun, B.W., Hall, J.B., and McNeil, S.E. 2009. Interaction of colloidal gold nanoparticles with human blood: effects on particle size and analysis of plasma protein binding profiles. *Nanomedicine: Nanotechnology, Biology, and Medicine*, 5: 106–117.
- Falini, G., Albeck, S., Weiner, S. and Addadi, L. 1996. Control of aragonite or calcite polymorphism by mollusk shell macromolecules. *Science*, 271: 67-69.
- Faraji, A.H. and Wipf, P. 2009. Nanoparticles in cellular drug delivery. *Bioorganic* and Medicinal Chemistry, 17: 2950–2962.
- Fellner, P., Jurišová, J., Kozánková, J. and Pach, L. 2012. Preparation of needle—like aragonite particles from calcium nitrate solution in batch and flow reactors. *Acta Chimica Slovaca*, 5(1): 5-11.
- Forrest M. Kievit, F.M., Wang, F.Y., Fang, C., Mok, H., Wang, K., Silber, J.R., Ellenbogen, R.G. and Zhang, M. 2011. Doxorubicin loaded iron oxide nanoparticles overcome multidrug resistance in cancer in vitro. *Journal of Controlled Release*, 152: 76–83.
- Gaudio, P.D., Russo, P., Lauro, M.R., Colombo, P. and Aquino, R.P. 2009. Encapsulation of ketoprofen and ketoprofen lysinate by prilling for controlled drug release. *American Association of Pharmaceutical Scientists*, 10(4): 1178-1185.

- Gbureck, U., Vorndran, E. and Barralet, J.E. 2008. Modeling vancomycin release kinetics from microporous calcium phosphate ceramics comparing static and dynamic immersion conditions. *Acta Biomaterialia*, 4: 1480–1486.
- Ge Li, G., Li, Z. and Ma, H. 2013. Synthesis of aragonite by carbonization from dolomite without any additives. *International Journal of Mineral Processing*, 123: 25–31.
- Hamdani, D.A., Javeed, A., Ashraf, M., Nazir, J., Ghafoor, A., Altaf, I. and Yousaf, M.S. 2014. *In vitro* cytotoxic and genotoxic evaluation to ascertain toxicological potential of ketoprofen. African Journal of Pharmacy and Pharmacology, *Academic Journals*, 8(14): 386-391.
- Hashim, U., Nadia, E. and Salleh, S. 2009. Nanotechnology development status in Malaysia: industrialization strategy and practices. *Int. Journal of Nanoelectronics and Materials*, 2(1): 119-134.
- Henriksen, I., Sande, S.A, Smistad, G., Agren, T. and Karlsen, J. 1995. In vitro evaluation of drug release kinetics from liposomes by fractional dialysis. *International Journal of Pharmaceutics*, 119: 231-238.
- Hongxia, G., Zhenping, Q., Peng, Q., Peng, Y., Suping, C., and Wei, W. 2011. Crystallization of aragonite CaCO3 with complex structures. Advanced Powder Technology, 22: 777–783.
- Hosoda, N., Sugawara, A. and Kato, T. 2003. Template effect of crystalline poly(vinyl alcohol) for selective formation of aragonite and vaterite caco3 thin films. *Macromolecules*, 36: 6449-6452.
- Hu, Z. and Deng, Y. 2003. Supersaturation control in aragonite synthesis using sparingly soluble calcium sulfate as reactants. *Journal of Colloid and Interface Science*, 266: 359–365.
- Hu, Z. and Deng, Y. 2004. Synthesis of needle-like aragonite from calcium chloride and sparingly soluble magnesium carbonate. *Powder Technology*, 140: 10–16.
- Hu, Z., Shao, M., Cai, Q., Ding, S., Zhong, C., Wei, X., and Deng, Y. 2009. Synthesis of needle-like aragonite from limestone in the presence of magnesium chloride. *Journal of materials processing technology*, 209: 1607–1611.
- Huang, Y.C., Fowkes, F.M., Lloyd, T.B. and Sanders, N.D. 1991. Adsorption of calcium ions from calcium chloride solutions onto calcium carbonate particles. *Langmuir*, 7: 1742-1748.
- International Standard ISO13321. 1996. Methods for Determination of Particle Size Distribution Part 8: Photon Correlation Spectroscopy. *International Organisation for Standardisation* (ISO). Retrieved December 2015 from https://www.iso.org/obp/ui/#iso:std:iso:13321:ed-1:v1:en.

- International Standard ISO22412. 2008. Particle Size Analysis Dynamic Light Scattering. *International Organisation for Standardisation* (ISO). Retrieved December 2015 from http://www.iso.org/iso/catalogue_detail.htm?csnumber=40942.
- Islam, K.N., Zakaria, M.Z.A.B., Ali, M.E., Hussein, M.Z. Noordin, M.M., Loqman, M.Y., Miah, G., Wahid, H. and Hashim, U. 2013. A novel method for the synthesis of calcium carbonate (aragonite) nanoparticles from cockle shells. *Powder Technology*, 235: 70–75.
- Islam, K.N., Zakaria, M.Z.A.B., Wahid, H., Ali, M.E., Hussein, M.Z.B., Noordin, M.M., Loqman, M.Y., Wahid, H., Hakim, M.A. and Hamid, S.B.A. 2012. Facile synthesis of calcium carbonate nanoparticles from cockle shells. *Journal of Nanomaterials*, 2012: 2.
- Jaji, A.Z., Bakar, M.Z.B.A., Mahmud, R., Loqman, M.Y., Hezmee, M.N.M., Isa, T., Wenliang, F. and Hammadi, N.I. 2017. Synthesis, characterization, and cytocompatibility of potential cockle shell aragonite nanocrystals for osteoporosis therapy and hormonal delivery. *Nanotechnology, Science and Applications*, 10: 23-33.
- Kalita, S., Devi, B., Kandimalla, R., Sharma, K.K., Sharma, A., Kalita, K., Kataki, A.C. and Kotoky J. 2015. Chloramphenicol encapsulated in poly-εcaprolactone-pluronic composite: nanoparticles for treatment of MRSA-infected burn wounds. *International Journal Nanomedicine*, 10: 2971-2984.
- Kamba, A.S., Ismail, M., Al-Ali, S.H.H., Ibrahim, T.A.T. and Zakaria, Z.A.B. 2013. In vitro delivery and controlled release of doxorubicin for targeting osteosarcoma bone cancer. *Molecules*, 2013 (18): 10580-10598.
- Kamba, A.S., Ismail, M., Ibrahim, T.A.T. and Zakaria, Z.A.B. 2013. A pH-sensitive, biobased calcium carbonate aragonite nanocrystal as a novel anticancer delivery system. *BioMed Research International*, 2013: 1–10.
- Kamba, A.S., Ismail, M., Ibrahim, T.A.T. and Zakaria, Z.A.B. 2013. Synthesis and characterisation of calcium carbonate aragonite nanocrystals from cockle shell powder (*Anadara granosa*). *Journal of Nanomaterials*, 2013: 1–9.
- Kamba, A.S. and Zakaria, Z.A.B. 2014. Osteoblasts growth behaviour on bio-based calcium carbonate aragonite nanocrystal. *BioMed Research International*, 2014: 1-9.
- Kim, S.K., Foote, M.B.and Huang, L. 2013. Targeted delivery of ev peptide to tumor cell cytoplasm using lipid coated calcium carbonate nanoparticles. *Cancer Letters*, 334(2): 311-8
- Kitamura, M. 2002. Controlling factor of polymorphism in crystallization process. *Journal of Crystal Growth*, 237-239: 2205–2214.

- Koga, N., Kasahara, D and Kimura, T. 2013. Aragonite crystal growth and solid-state aragonite–calcite transformation: a physico–geometrical relationship via thermal dehydration of included water. *American Chemical Society: Crystal Growth and Design*, 13(5): 2238-2246.
- Koo, O.M., Rubinstein, I. and Onyuksel, H. 2005. Role of nanotechnology in targeted drug delivery and imaging: a concise review. *Nanomedicine: Nanotechnology*, *Biology, and Medicine*, 1: 193–212.
- Kumar, A., Zhang, X. and Liang, X-J. 2013. Gold nanoparticles: Emerging paradigm for targeted drug delivery system. *Biotechnology Advances*, 31(5): 593-606.
- Lardner, A. 2001. The effects of extracellular pH on immune function. *Journal of Leukocyte Biology*, 69: 522–530.
- Lemos, A.F., Rocha, J.H.G., Quaresma, S.S.F., Kannan, S., Oktar, F.N., Agathopoulos, S. and Ferreira, J.M.F. 2006. Hydroxyapatite nano-powders produced hydrothermally from nacreous material. *Journal of the European Ceramic Society*, 26 (16): 3639-3646.
- Liang, P., Zhao, D., Wang, C.Q., Zongg, J.Y., Zhuo, R.X. and Cheng, S.X. 2013. Facile preparation of heparin/CaCO₃/CaP hybrid nano-carriers with controllable size for anticancer drug delivery. *Colloids and Surfaces B: Biointerfaces* 102: 783–788.
- Ma, M-G. and Sun, R-C. 2011. Biomineralization and biomimetic synthesis of biomineral and nanomaterials. *Advances in Biomimetics*, Cavrak, M., (Ed.), ISBN:978-953-307-191-6, InTech: 13-50. Retrieved 20 August 2014 from http://www.intechopen.com/books/advances-in-biomimetics/biomineralizationand-biomimeticsynthesis-of-biomineral-and-nanomaterials.
- Maestrelli, F., Zerrouk, N., Cirri, M., Mennini, N. nd Mura, P. 2008. Microspheres for colonic delivery of ketoprofen-hydroxypropyl-ß-cyclodextrin complex. *European Journal of Pharmaceutical Sciences*, 34: 1-11.
- Magenheim, B., Levy, M.Y. and Benita, S. 1993. A new in vitro technique for the evaluation of drug release profile from colloidal carriers- ultrafiltration technique at low pressure. *International Journal of Pharmaceutics*, 94: 115-123.
- Mahendra Rai, M., Yadav, A. and Gade, A. 2009. Silver nanoparticles as a new generation of antimicrobials. *Biotechnology Advances*, 27: 76–83.
- Mishra, B., Patel, B.B. and Tiwari, S. 2010. Colloidal nanocarriers: a review on formulation technology, types and applications toward targeted drug delivery. *Nanomedicine: Nanotechnology, Biology, and Medicine*, 6: 9–24.
- Moghimi, S.M., Hunter, A.C. and Murray, J.C. 2005. Nanomedicine: current status and future prospects. *The FASEB Journal*, 19: 311-330.

- Morgan, T.T., Muddana, H.S., Altinoglu, E.I., Rouse, S.M., Tabakovic´, A., Tabouillot, T., Russin, T.J., Shanmugavelandy, S.S., Butler, P.J., Eklund, P.C., Yun, J.K., Kester, M., and Adair, J.H. 2008. Encapsulation of organic molecules in calcium phosphate nanocomposite particles for intracellular imaging and drug delivery. *Nano Letters*, 8(12): 4108-4115.
- Mohanraj, V.J. and Chen, Y. June, 2006. Nanoparticles a review. *Tropical Journal* of Pharmaceutical Research, 5(1): 561-573.
- Moulin, P. and Roques, H. 2003. Zeta potential measurement of calcium carbonate. *Journal of Colloid and Interface Science*, 261: 115–126.
- Malaysian Ministry of Science, Technology and Innovation. 2010. National Nanotechnology Directorate Division (NND). Retrieved 20 December 2015 from http://www.mosti.gov.my/en/corporate-profile/divisionsdepartments/nationalnanotechnology-directorate-division-nnd/.
- Mello, V.A.D. and Ricci-Junior, E. 2011. Encapsulation of naproxen in nanostructured system: structural characterization and *in vitro*release studies. *Quím. Nova*, 34(6): ISSN 0100-4042. Retrieved June, 2015 from http://dx.doi.org/10.1590/S0100-40422011000600004.
- Newcomb, R.W. 2009. Nanotechnology for biomedicine: past, present and future. Petra '09 proceedings of the 2nd international conference on pervasive technologies related to assistive environments, article no. 69, ACM ISBN 978-1-60558-409-6: 1-4.
- Nora, H.D.L. and Stephen C.P. 1998. Surface Structure and Morphology of Calcium Carbonate Polymorphs Calcite, Aragonite, and Vaterite: An Atomistic Approach. *Journal of Physical Chemistry B*, 102: 2914-2922.
- Oyerokun, F.T. and Vaia R.A. 2012. Distribution in the grafting density of endfunctionalized polymer chains adsorbed onto nanoparticle surfaces. *Macromolecules*, 45: 7649–7659.
- Pai, R.K., Jansson, K. and Hedin, N. 2009. Transport-Mediated Control of Particles of Calcium Carbonate. American Chemical Society: Crystal Growth and Design, 9: 4581–4583.
- Panyam, J. and Patil, Y. 2008. Distribution: Movement of Drugs through the Body. Preclinical development handbook: ADME and biopharmaceutical properties. *Wiley-Interscience Publication*, Cox, S: 332.
- Park, J.H., Saravanakumar, G., Kim, K. and Kwon, I.C. 2010. Targeted delivery of low molecular drugs using chitosan and its derivatives. *Advanced Drug Delivery Reviews*, 62: 28–41.
- Park, W.K., Ko, S.J., Lee, S.W., Cho, K.H, Ahn, J.W. and Han, C. 2008. Effects of magnesium chloride and organic additives on the synthesis of aragonite precipitated calcium carbonate. *Journal of Crystal Growth*, 310: 2593–2601.

- Parveen, S., Misra, R. and Sahoo, S.K. 2012. Nanoparticles: a boon to drug delivery, therapeutics, diagnostics and imaging. *Nanomedicine: Nanotechnology*, *Biology, and Medicine*, 8: 147–166.
- Peng, C., Zhao, Q. and Gao, C. 2010. Sustained delivery of doxorubicin by porous caco3 and chitosan/alginate multilayers-coated caco3 microparticles. *Colloids* and Surfaces A: Physicochemical and Engineering Aspects, 353: 132–139.
- Peng, J. Feng, L-N., Zhang, K., Li, X-H, Jiang, L-P. and Zhu, J-J. 2012. Calcium Carbonate-Gold Nanocluster Hybrid Spheres: Synthesis and Versatile Application in Immunoassays, *Chemistry European Journal*, 18: 5261-5268.
- Pharmacopoeia commission of the people Republic of China. 2000. *Pharmacopoeia* of the Peoples Republic of China; Chemistry Industry Press, Beijing, China.
- Powers, K.W., Brown,S.C., Krishna,V.B., Wasdo, S.C., Moudgil,B.M. and Roberts, S.M. 2006. Research strategies for safety evaluation of nanomaterials part vi characterization of nanoscale particles for toxicological evaluation. *Toxicological Sciences* 90(2): 296-303.
- Prakash, S., Malhotra, M., Shao, W., Tomaro-Duchesneau, C. and Abbasi, S. 2011. Polymeric nanohybrids and functionalized carbon nanotubes as drug delivery carriers for cancer therapy. *Advanced Drug Delivery Reviews*, 63: 1340–1351.
- Qiu, N., Yin, H., Ji, B., Klauke, N., Glidle, A., Zhang, Y., Song, H., Cai, L., Ma, L., Wang, G., Chen, L. And Wang, W. 2012. Calcium carbonate microspheres as carriers for the anticancer drug camptothecin. *Materials Science and Engineering C*, 32: 2634-2640.
- Raj, S., Jose, S., Sumod, U.S. and Sabitha, M. 2012. Nanotechnology in cosmetics: opportunities and challenges. *Journal of Pharmacy and Bioallied Sciences*, 4(3): 186-193.
- Reddy, M.M and Nancollas, G.H. 1976. The crystallization of calcium carbonate, iv the effect of magnesium, strontium and sulfate ions. *Journal of Crystal Growth*, 35: 33-38.
- Riss, T.L., Moravec, R.A., Niles, A.L., Benink, H.A., Worzella, T.J. and Minor, L. 2013. Cell viability assays. *Assay guidance manual*: 1-23. Retrieved 5 December 2015 from http://www.ncbi.nlm.nih.gov/books/NBK144065/.
- Rizzuti, A. and Leonelli, C. 2008. Crystallization of aragonite particles fromsolution under microwave irradiation. *Powder Technology*, 186: 255–262.
- Roco, M.C. 2005. International perspective on government nanotechnology funding in 2005. *Journal of Nanoparticle Research*, Report 7(6): 1-9.
- Sargheini, J., Ataie, A., Salili, S.M. and Hoseinion, A.A. 2012. One-step facile synthesis of CaCO3 nanoparticles via mechano-chemical route. *Powder Technology*, 219: 72–77.

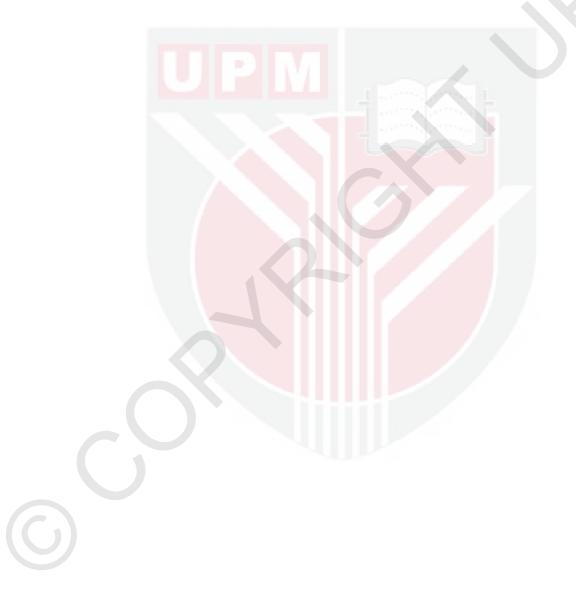
- Sahoo, S.K., Parveen, S. and Panda, J.J. 2007. The present and future of nanotechnology in human health care. *Nanomedicine: Nanotechnology, Biology, and Medicine*, 3: 20-31.
- Santos, R.M., Bodor, M., Dragomir, P.N., Vraciu, A.G., Vlad, M. and Gerven, T.V. 2004. Magnesium chloride as a leaching and aragonite-promoting selfregenerative additive for the mineral carbonation of calcium-rich materials. *Minerals Engineering*, 59: 71-81.
- Sheva Naahidi, S., Jafari, M., Edalat, F., Raymond, K., Khademhosseini, A. and Chen, P. 2013. Biocompatibility of engineered nanoparticles for drug delivery. *Journal* of Controlled Release, 166: 182–194.
- Singh, R. and Lillard Jr, J.W. 2009. Nanoparticle-based targeted drug delivery. *Experimental and Molecular Pathology*, 86(3): 215–223.
- Sohnel, O. and Mullin, J.W. 1982. Precipitation of calcium carbonate. *Journal of Crystal Growth*, 60: 239-250.
- Stigliani, M., Aquino, R.P., Gaudio, P.D, Mencherini, T., Sansone, F. and Russo, P. 2013. Non-steroidal anti-inflammatory drug for pulmonary administration: design and investigation of ketoprofen lysinate fine dry powders. *International Journal of Pharmaceutics*, 448: 198–204.
- Stockert, J.C., Blázquez-Castro, A., M., Horobin, R.W. and Villanueva, A. 2012. MTT assay for cell viability: Intracellular localization of the formazan product is in lipid droplets. *Acta Histochemica*, 114 (8): 785-796.
- Stupp, S.I. and Braun, P.V. 1997. Molecular manipulations of materials: biomaterials, ceramics and semiconductors. *Science*, 277: 1242–1248.
- Svenskaya, Y., Parakhonskiy, B., Haase, A., Atkin, V., Lukyanets, E., Gorin, D. and Antolini, R. 2013. Anticancer drug delivery system based on calcium carbonate particles loaded with a photosensitizer. *Biophysical Chemistry*, 182: 11–15.
- Takiyama, H., Minamisono, T., Osada, Y. and Matsuoka, M. 2010. Operation design for controlling polymorphism in the anti-solvent crystallization by using ternary phase diagram. *Chemical Engineering Research and Design*, 88: 1242–1247.
- Tran, H.V., Tran, L.D., Vu, H.D. and Thai, H. 2010. Facile surface modification of nanoprecipitated calcium carbonate by adsorption of sodium stearate in aqueous solution. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 366: 95–103.
- Ueno, T., Tsuchiya, H., Mizogami, M., Takakura, K. 2008. Local anesthetic failure associated with inflammation: verification of the acidosis mechanism and the hypothetic participation of inflammatory peroxynitrite. *Journal of Inflammation Research*, 1: 41–48.

- Ueno, Y., Futagawa, H., Takagi, Y., Ueno, A. and Mizushima, Y. 2005. Drugincorporating calcium carbonate nanoparticles for a new delivery system. *Journal of Controlled Release*, 103: 93–98.
- Vallet-Reg1, M., Balas, F., Colilla, M. and Manzano, M. 2008. Bone-regenerative bioceramic implants with drug and protein controlled delivery capability, *Progress in Solid State Chemistry*, 36: 163-191.
- Wang, C., Liu, H., Gao, Q., Liu, X. and Tong, Z. 2008. Alginate–calcium carbonate porous microparticle hybrid hydrogels with versatile drug loading capabilities and variable mechanical strengths. *Carbohydrate Polymers*, 71: 476–480.
- Wang, C., Liu, Y., Bala, H., Pan, Y., Zhao, J., Zhao, X. and Wang, Z. 2007. Facile preparation of caco3 nanoparticles with self-dispersing properties in the presence of dodecyl dimethyl betaine. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 297: 179-182.
- Wang, C., He, C., Tong, Z., Liu, X., Ren, B. and Zeng, F. 2006. Combination of adsorption by porous caco3 microparticles and encapsulation by polyelectrolyte multilayer films for sustained drug delivery. *International Journal of Pharmaceutics*, 308: 160–167.
- Wang, C., Zhao, J., Zhao, X., Bala, H. and Wang, Z. 2006. Synthesis of nanosized calcium carbonate (aragonite) via a polyacrylamide inducing process. *Powder Technology*, 163: 134–138.
- Wang, L., Sondi, I. and Matijevic', E. 1999. Preparation of uniform needle-like aragonite particles by homogeneous precipitation. *Journal of Colloid and Interface Science*, 218: 545–553.
- Wang, Z., Qian, L., Wang, X., Yang, F. and Yanga, X. 2008. Construction of hollow DNA/PLL microcapsule as a dual carrier for controlled delivery of DNA and drug. Colloids and Surfaces A: Physicochemal and Engineering Aspects, 326: 29–36.
- Westesen, K., Bunjes, H. and Koch, M.H.J. 1997. Physicochemical characterization of lipid nanoparticles and evaluation of their drug loading capacity and sustained release potential. *Journal of Controlled Release*, 48: 223–236.
- Xiang, L., Xiang, Y., Wen, Y. and Wei, F. 2004. Formation of CaCO₃ nanoparticles in the presence of terpineol. *Materials Letters*, 58: 959-965.
- Xu, Z.P., Zeng, Q.H., Lu,G.Q. and Yu, A.B. 2006. Inorganic nanoparticles as carriers for efficient cellular delivery. *Chemical Engineering Science*, 61: 1027-1040.
- Yan, G., Wang, L. and Huang, J. 2009. The crystallization behavior of calcium carbonate in ethanol/water solution containing mixed nonionic/anionic surfactants. *Powder Technology*, 192: 58–64.

- Ye, Z., Guo, G., Chen, H. and Shu⁷Z. 2014. Interaction between aqueous solutions of hydrophobically associating polyacrylamide and dodecyl dimethyl betaine. *Journal of Chemistry*, 2014: 1-8.
- Yu, J., Lei, M. and Cheng, B. 2004. Facile preparation of monodispersed calcium carbonate spherical particles via a simple precipitation reaction. *Materials Chemistry and Physics*, 88: 1-4.
- Yu, J., Lei, M., Cheng, B. and Zhao, X. 2004. Facile preparation of calcium carbonate particles with unusual morphologies by precipitation reaction. *Journal of Crystal Growth*, 261: 566–570.
- Zakaria, A.B.Z., Hussein, B.F. and Noordin, M.M. 2011. Cockle shell-based biocomposite scaffold for bone tissue engineering. *Regenerative Medicine and Tissue Engineering Cells and Biomaterials*, Eberli, D., ISBN: 978-953-307-663-8: 365-390. Retrieved 30 June 2014 from http://www.intechopen.com/books/regenerative-medicine-and-tissue-engineering-cells-andbiomaterials/cockle-shell-based-biocomposite-scaffold-for-bone-tissue-engineering.
- Zakaria, A.B.Z., Zakaria, N. and Kasim, Z. 2004. Mineral composition of the cockle (anadara granosa) shells, hard clamps (meretrix meretrix) shells and corals (porites. spp): a comparative study. *Journals of Animals and Veterinary Advances*, 3(7): 445-447.
- Zetasizer Nano Series User Manual. 2004. MAN 0317, Issue 1.1: Malvern Instruments Ltd. United Kingdom Malvern.
- Vallet-Regí, M., Balas, F., Colilla, M. and Manzano, M. Drug Confinement and Delivery in Ceramic Implants. 2007. *Drug Metabolism Letters*, 1: 37-40.
- Zhao, D., Liu, C-J., Zhuo, R-X., and Cheng, S-X. 2012. Alginate/caco3 hybrid nanoparticles for efficient co-delivery of antitumor gene and drug. *Molecular Pharmaceutics*, 9: 2887–2893.
- Zhao, Q. and Li, B. 2008. pH-controlled drug loading and release from biodegradable microcapsules. *Nanomedicine: Nanotechnology, Biology, and Medicine,* 4: 302–310.

LIST OF PUBLICATIONS

- Syairah L.M.A.G., Mohd, Z.H. and Zuki, A.B.Z. 2017. Synthesis and characterization of cockle shell-based calcium carbonate aragonite polymorph nanoparticles with surface functionalization. *Journal of Nanoparticles*, 2017: 12.
- Syairah L.M.A.G., Mohd, Z.H., Yaya, R. and Zuki, A.B.Z. Surface-functionalized cockle shell-based calcium carbonate aragonite polymorph for nanocarrier of drug. *Nanotechnology, Science and Applications*, (Status: In-press).





UNIVERSITI PUTRA MALAYSIA

STATUS CONFIRMATION FOR THESIS / PROJECT REPORT AND COPYRIGHT

ACADEMIC SESSION : Second Semester 2016/2017

TITLE OF THESIS / PROJECT REPORT :

In Vitro EVALUATION OF COCKLE SHELL-BASED CALCIUM CARBONATE ARAGONITE POLYMORPH NANOPARTICLE WITH SURFACE FUNCTIONALIZATION FOR DRUG DELIVERY APPLICATIONS

NAME OF STUDENT : SYAIRAH LIYANA BINTI MOHD ABD GHAFAR

I acknowledge that the copyright and other intellectual property in the thesis/project report belonged to Universiti Putra Malaysia and I agree to allow this thesis/project report to be placed at the library under the following terms:

- 1. This thesis/project report is the property of Universiti Putra Malaysia.
- 2. The library of Universiti Putra Malaysia has the right to make copies for educational purposes only.
- 3. The library of Universiti Putra Malaysia is allowed to make copies of this thesis for academic exchange.

I declare that this thesis is classified as :

*Please tick (V)



CONFIDENTIAL

RESTRICTED

OPEN ACCESS

(Contain confidential information under Official Secret Act 1972).

(Contains restricted information as specified by the organization/institution where research was done).

I agree that my thesis/project report to be published as hard copy or online open access.

This thesis is submitted for :

| Embargo from | | until | | |
|--------------|--------|-------|--------|--|
| | (date) | | (date) | |

Approved by:

(Signature of Student) New IC No/ Passport No.:

(Signature of Chairman of Supervisory Committee) Name:

Date :

Date :

[Note : If the thesis is CONFIDENTIAL or RESTRICTED, please attach with the letter from the organization/institution with period and reasons for confidentially or restricted.]