



UNIVERSITI PUTRA MALAYSIA

***DISSOLUTION CHARACTERISTICS OF SELECTED FRUIT
TABLETS WITH EFFERVESCENT AGENTS***

MD. SAIFULLAH

FK 2015 184



**DISSOLUTION CHARACTERISTICS OF SELECTED FRUIT TABLETS
WITH EFFERVESCENT AGENTS**

By

MD. SAIFULLAH

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in
Fulfilment of the Requirements for the degree of Master of Science**

March 2015

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia

DEDICATION



**To My Beloved Parents
And Family Members**

Abstract of thesis presented to Senate of Universiti Putra Malaysia in fulfilment of the requirements for the degree of Master of Science

DISSOLUTION CHARACTERISTICS OF SELECTED FRUIT TABLETS WITH EFFERVESCENT AGENTS

By

MD. SAIFULLAH

March 2015

Chairman: Associate Professor Yus Aniza Yusof, PhD
Faculty: Engineering

Tableting of fruit powder is a unique technique in the area of fruit powder preservation. Dissolution characteristic of tablet is an important parameter for quality control and production. The acceptability and popularity of fruit powder tablet depend on its dissolution rate. Physicochemical properties of tablet ingredients have great influence on dissolution behaviour of tablet. This research was conducted in two major areas namely determination of physicochemical properties of fruit powder and their relationship with the dissolution of tablet with effervescent agents, and dissolution profiling of fruit powder effervescent tablets and dissolution profile comparison. Pitaya, pineapple, mango and guava fruits were used in this research, as these fruits are grown in large quantity in Malaysia. Tablets with 20 mm diameter and 2.5 gm weight were made by using a direct compression method at a constant pressure via a universal testing machine. Dissolution test was carried out in a dissolution tester and dissolved amount of solute was measured by an Ultraviolet spectrophotometric test as a function of time. Distilled water and simulated saliva was used as dissolution medium in the dissolution test at room temperature and 37°C temperature respectively. The investigation on physicochemical analysis showed that fruit powders were different from each other in terms of physical and chemical properties. The results showed that the fat content in powder has inverse relationship with dissolution rate. On the other hand, porosity of the tablet represented proportional relationship with dissolution rate and application of effervescent agents increased the dissolution rate of tablet significantly. In this study three types of method were adopted to compare the dissolution profile which includes model dependent, model independent and statistical method. Five release kinetics mathematical equations were used to compare dissolution profile in model dependent method. According to model dependent method, when distilled water was used as dissolution medium, pineapple, guava and mango powder tablet dissolution profile shows similarity as their dissolution profile was fitted with Higuchi model. However, pitaya powder tablet dissolution profile was different from other types as its dissolution profile was fitted very well with zero order kinetics model. In simulated saliva, same phenomenon was appeared; pineapple, guava and mango powder tablet dissolution profile shows similarity and pitaya powder tablet dissolution profile was different from them. On the other hand, according to model independent method, dissolution profiles of fruit powder tablets were in the similarity range at most

of the points. However, in the statistical method pair-*t* test showed that there was significantly difference among the dissolution profile of fruit powder tablets at the level of $P < 0.01$. Based on this study, a better understanding about physicochemical properties of fruit powder and their relationship with dissolution rate and effect of effervescent agents on dissolution rate are obtained, which are essential for processing and handling of fruit powder and tablet preparation as well as improvement of dissolution rate. In conclusion, dissolution profiling and its comparison will be helpful for further change in formulation, development of new formulation, scale up of the production, and quality control in production line during industrial scale production.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk Ijazah Master Sains

CIRI-CIRI PELARUTAN TABLET BUAH-BUAHAN TERPILIH DENGAN EJEN PEMBUIHAN

Oleh

MD. SAIFULLAH

Mac 2015

Pengerusi: Profesor Madya Yus Aniza Yusof, PhD

Fakulti: Kejuruteraan

Pemadatan serbuk buah-buahan dalam bentuk tablet adalah satu teknik yang unik dalam bidang pemeliharaan serbuk buah-buahan. Ciri pelarutan tablet adalah parameter yang penting untuk kawalan kualiti dan pengeluaran. Penerimaan dan populariti serbuk buah tablet bergantung kepada kadar keterlarutannya. Ciri fiziko-kimia bahan tablet mempunyai pengaruh besar ke atas sifat terlarut bagi tablet. Kajian ini dijalankan dalam dua bidang utama iaitu penentuan sifat fiziko-kimia bagi serbuk buah-buahan dan hubungan mereka dengan sifat terlarut tablet dengan ejen pembuih. Selain itu, corak pelarutan bagi tablet buah yang ditambah agen pembuih dan perbandingan corak pelarutan bagi kesemua tablet buah tersebut. Buah-buahan seperti buah naga, nanas, mangga dan jambu batu telah digunakan dalam kajian ini kerana buah-buahan ini ditanam kuantiti yang banyak di Malaysia. Tablet yang mempunyai ukuran 20 mm bagi diameter dan 2.5 gm bagi berat telah dihasilkan dengan menggunakan kaedah mampatan langsung pada tekanan tetap melalui mesin ujian universal. Tambahan lagi, ujian pelarutan telah dijalankan di dalam alat penguji keterlarutan. Jumlah bahan larut juga telah diukur oleh ujian spektrofotometri ultra-ungu yang berkadar dengan masa. Air suling dan simulasi air liur telah digunakan sebagai medium pelarutan dalam ujian pelarutan pada dua suhu yang berbeza iaitu suhu bilik dan suhu 37°C. Penyelidikan terhadap analisis fizikokimia menunjukkan bahawa serbuk buah adalah berbeza antara satu sama lain dari segi ciri-ciri fizikal dan kimia. Hasil kajian menunjukkan bahawa kandungan lemak dalam serbuk mempunyai hubungan songsang dengan kadar pelarutan. Di samping itu, sifat berongga tablet mempunyai hubungan berkadar terus dengan kadar keterlarutan. Aplikasi ejen pembuih juga meningkatkan kadar keterlarutan bagi tablet buah dengan ketara. Dalam kajian ini tiga jenis kaedah telah digunakan untuk membandingkan profil corak pelarutan tablet buah. Tiga kaedah tersebut ialah model bergantung, model bebas dan kaedah statistik. Lima persamaan matematik bagi kinetik pelepasan telah digunakan untuk membandingkan corak profil pelarutan dalam model kaedah bergantung. Menurut model kaedah bergantung, apabila air suling telah digunakan sebagai medium pelarutan, corak profil pelarutan bagi tablet nanas, jambu batu dan serbuk mangga menunjukkan persamaan kerana corak profil pelarutan bagi ketiga-tiga tablet buah tersebut adalah paling sesuai mengikut model Higuchi. Walau bagaimanapun, corak profil pelarutan bagi serbuk buah naga adalah berbeza daripada serbuk buah yang lain kerana corak profil pelarutan bagi buah naga adalah paling sesuai mengikut model kinetik tertib sifar. Dalam simulasi air liur,

fenomena yang sama telah muncul; nanas, jambu batu dan corak profil pelarutan bagi tablet serbuk mangga menunjukkan persamaan dan corak profil pelarutan bagi tablet serbuk pitaya adalah berbeza dari serbuk buah yang lain. Sebaliknya, menurut kaedah model bebas, corak profil pelarutan bagi tablet serbuk buah berada dalam lingkungan persamaan di kebanyakan titik. Walau bagaimanapun, dalam kaedah statistik pasangan-ujian t menunjukkan tidak terdapat perbezaan ketara antara corak profil pelarutan tablet serbuk buah-buahan di peringkat $P < 0.01$. Berdasarkan kajian ini, pemahaman yang lebih baik tentang sifat fizikokimia serbuk buah-buahan dan hubungan mereka dengan kadar pelarutan serta kesan ejen pembuih pada kadar keterlarutan telah dicapai. Hal yang sedemikian, adalah penting untuk pemprosesan dan pengendalian serbuk buah-buahan dan penyediaan tablet serta peningkatan kadar keterlarutan bagi tablet serbuk buah-buahan. Kesimpulannya, corak profil pelarutan dan perbandingannya akan membantu untuk perubahan lagi dalam penggubalan, pembangunan formulasi baru, peningkatan skala pengeluaran dan kawalan kualiti dalam pengeluaran produk dalam skala industri.

ACKNOWLEDGEMENTS

All praise for ALLAH Subhanahu-wa-ta'ala, the Omnipresent, Omnipotent and Omniscient, Who has enabled me to complete this research work and submitting the thesis for the Master of Science degree in Food Engineering.

First and foremost, I would like to express my utmost gratitude and heartfelt appreciation to my supervisor Associate Professor Dr. Yus Aniza Yusof for her continuous support and invaluable guidance for my MS study, for her motivation and enthusiasm. During my MS study, she provided sound advice, good teaching and friendly company, and shared a lot of her expertise, research insight and best ideas. I simply could not imagine having a better advisor and friendlier mentor for MS study.

With a great deal of luck, I got an excellent Supervisory Committee. I would also like to extend appreciation and express gratitude to my co-supervisor, Associate Professor Dr. Ir. Chin Nyuk Ling, Associate Professor Dr. Norashikin Ab. Aziz and Mohd Afandi P. Mohammed for their support, guidance and motivation. I am thankful to Dr. Mohammad Gulzarul Aziz, Former Post-Doctoral fellow Department of Process and Food Engineering, Universiti Putra Malaysia.

I sincerely acknowledge the all technicians of the department of Process and Food Engineering, for their assistance, patience and friendly help during conducting my experimental work. I would like to thank Ministry of Science (MOSTI) for funding to conduct my research.

Lastly, I feel proud to acknowledge the inspirations and incentives of my beloved parents, brothers, sisters and friends for their kind co-operation, blessing and constant source of inspiration that always stand beside to support my higher studies.

I certify that a Thesis Examination Committee has met on 2 March 2015 to conduct the final examination of Md. Saifullah on his thesis entitled "Dissolution Characteristics of Selected Fruit Tablets with Effervescent Agents" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Siti Mazlina bt Mustapa Kamal, PhD

Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Farah Saleena binti Taip, PhD

Professor
Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Rosnita binti A.Talib, PhD

Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Md. Mujibur Rahman, PhD

Associate Professor
Universiti Tenaga Nasional
Malaysia
(External Examiner)



ZULKARNAIN ZAINAL, PhD

Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 15 April 2015

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of supervisory committee were as follows:

Yus Aniza Yusof, PhD
Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Chin Nyuk Ling, PhD
Associate Professor, Ir
Faculty of Engineering
Universiti Putra Malaysia
(Member)

Norashikin Abdul Aziz, PhD
Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Member)

Mohd Afandi P Mohammed, PhD
Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Member)

BUJANG KIM HUAT, PhD
Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- x this thesis is my original work;
- x quotations, illustrations and citations have been duly referenced;
- x this thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- x intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- x written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- x there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature: _____

Date: _____

Name and Matric No.: Md. Saifullah (GS35172)

Declaration by Members of Supervisory Committee

This is to confirm that:

- x the research conducted and the writing of this thesis was under our supervision;
- x supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: _____
Name of
Chairman of
Supervisory
Committee: Yus Aniza Yusof, PhD

Signature: _____
Name of
Member of
Supervisory
Committee: Ir. Chin Nyuk Ling, PhD

Signature: _____
Name of
Member of
Supervisory
Committee: Norashikin Abdul Aziz, PhD

Signature: _____
Name of
Member of
Supervisory
Committee: Mohd Afandi P Mohammed, PhD

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APROVAL	vi
DECLARATION	viii
LIST OF TABLES	xii
LIST OF FIGURES	xiii
CHAPTER	
1 INTRODUCTION	
1.1 Background of study	1
1.2 Problem statement	3
1.3 Objectives	3
1.4 Scope of study	3
1.5 Organization of the thesis	4
2 LITERATURE REVIEW	
2.1 Introduction	5
2.2 Physicochemical properties of fruit powder	5
2.3 Tablet and tablet processing	6
2.3.1 Dry granulation	7
2.3.2 Wet granulation	7
2.3.3 Direct compression	7
2.4 Effervescent tablet	8
2.5 Dissolution of tablet	9
2.6 Mathematical model	11
2.6.1 Zero-order model	11
2.6.2 First order model	12
2.6.3 Hixson-Crowell model	13
2.6.4 Higuchi model	13
2.6.5 Korsmeyer-Peppas model	14
2.6.6 Weibull model	15
2.6.7 Hopfenber model	16
2.7 Model selection approach	17
2.8 Dissolution profiling and profile comparison	17
3 METHODOLOGY	
3.1 Introduction	19
3.2 Materials for fruit powder tablet	19
3.3 Proximate analysis	20
3.4 Tablet preparation	20
3.4.1 Formulation and mixing	20
3.5 Physical properties of mix ingredients	20
3.5.1 Morphology of fruit powder and tablet mixture	20
3.5.2 Bulk density	21

3.5.3	Tap density	21
3.5.4	True density	21
3.5.5	Particle size	21
3.5.6	Water activity (Aw)	21
3.5.7	Porosity	21
3.5.8	Hausner ratio and Carr index	22
3.6	Tableting and compressibility test	22
3.7	Dissolution test	23
3.7.1	Dissolution test in water	23
3.7.2	Dissolution test in simulated saliva	23
3.7.3	Measurement of percent solute release/percent dissolve	24
3.8	Dissolution profiling	24
3.9	Dissolution profile comparison	24
4	RESULTS AND DISCUSSION	
4.1	Introduction	25
4.2	Physicochemical properties of fruit powder	25
4.3	Effect of compressive force on tablet volume	30
4.4	Relationship between porosity and dissolution of fruit powder tablets	31
4.5	Dissolution profile of fruit powder tablet	32
4.6	Dissolution profile comparison of fruit powder tablets	34
4.6.1	Model dependent method	35
4.6.2	Model-independent method (similarity and dissimilarity factor)	37
4.6.3	Statistical method	39
5	CONCLUSIONS AND RECOMMENDATIONS	
5.1	Conclusions	43
5.2	Recommendations and future work	44
	REFERENCES	45
	APPENDICES	
I	Calculation of constant a and b from Kawakita and Lüdde model equation	56
II	Data analysis using KinetDS, V:3	57
III	Relationship between fat content versus total dissolution time	58
IV	Relationship between porosity versus total dissolution time	59
V	Relationship between percent protein content versus total dissolution time	60
VI	Relationship between percent moisture content versus total dissolution time	61
VII	Relationship between fibre content versus total dissolution time	62
	BIODATA OF STUDENT	63
	LIST OF PUBLICATIONS	64

LIST OF TABLES

Table		Page
2.1	Apparatus used for novel/special dosage forms	11
2.2 (a)	Diffusional release mechanisms for slab	15
2.2 (b)	Diffusional release mechanisms for cylinder	15
2.3	Relationship between curve characteristics and shape factor	16
3.1	Ranges for Carr index and Hausner ratio	22
4.1	Proximate composition of fruit powder	25
4.2	Physical properties of fruit powder and mix ingredients	27
4.3	SEM image of fruit powder and fruit powder containing effervescent agents	28
4.4	Kawakita and Ludde equation constants for fruit powder tablet.	30
4.5	Model dependent analysis of dissolution profiles of fruit powder tablets in distilled water.	35
4.6	Model dependent analysis of dissolution profiles of fruit powder tablets in simulated saliva	36
4.7	Similarity factor results in water	38
4.8	Difference factor results in water	38
4.9	Similarity factor results in simulated saliva	38
4.10	Difference factor in simulated saliva	39
4.11	Pair <i>t</i> -test result for dissolution of fruit powder tablets in distilled water	40
4.12	Pair <i>t</i> - test result for dissolution of fruit powder tablets in simulated saliva	41

LIST OF FIGURES

Figure		Page
1.1	Plum fruit candy tablet available in market	2
2.1	Direct compression method steps of tableting	8
2.2	The basic steps in the drug dissolution mechanism	9
2.3	Dissolution of tablet in dissolution chamber	10
3.1	Flow chart showing the process involve in this study	19
4.1	Pressure versus density relationship	30
4.2	Porosity of fruit powder tablets	31
4.3	(a) dissolution of fruit powder tablet in water; (b) dissolution of fruit powder tablet in simulated saliva	32
4.4	Dissolution profile in distilled water	33
4.5	Dissolution profile in simulated saliva	34

CHAPTER 1

INTRODUCTION

1.1 Background of study

Like any other agricultural produce, many different fruits are available in the market at peak season. Due to market saturation, the producers do not get their desired price and sometimes postharvest losses occur. People would typically prefer to consume fresh fruits all the year round, but practically it is impossible to provide fresh fruits all the year round. So, people are usually deprived of the natural taste of fruit in the off season, and producers may face financial losses at peak season. Not all fruits are grown in all areas of the world, and most fruits are produced in the tropical and sub-tropical zones. Hence, people of the temperate zones are deprived of most of the fruits grown in the world. Fresh fruits may be exported to temperate zones, but this will be costly and the risk of spoilage is high.

To reduce these problems, fruits are processed mainly into juice and other products. Among the different types of products, fruit juice is equally popular for people of all age groups in every part of world. However, the production of fruit juice requires a large number of unit operations and thermal processing. Due to high moisture and sugar content, it is very susceptible to microbial growth and spoilage risk is high. Fruit juices and drinks containing fruit juices are prone to spoilage by spore-forming bacterium are either fresh (not heat-treated) or pasteurized (but not UHT-treated) and stored unpreserved at ambient temperatures (Pettipher et al., 1997). So, producers need to add preservatives and requires the thermal destruction of microorganisms. Thermal processing may cause the deterioration of heat sensitive nutrients and volatile components and sometimes produce undesirable compounds. Different processing and preservation techniques, such as drying, special packaging, and chemical treatments are used to preserve the fruit juice. Among all preservation methods drying of fruit juice in the powder is a unique technique which increases the shelf life of fruit juice (Chen and Mujumdar, 2009; Kha et al., 2010; Quek et al., 2007). In powder form, moisture content is at microbiologically safe levels, as the water activity becomes very low. There are different types of drying methods available. The most common, convenient and widely practiced method is spray drying to prepare fruit powder (Fernandes et al., 2011; Phisut, 2012). Fruit powder is used to produce cake, candy as well as refreshment drink.

Fruit powder is amorphous dry granular material and is highly hygroscopic in nature. Therefore, careful handling and special packaging is required during marketing, transportation and storage. It is bulky in nature as well. Hence, more space is required for transportation and storage. However, alternative techniques such as compaction of fruit powder in the shape of tablets is more suitable to overcome the post processing problems regarding storage, transportation, and quality degradation. Tableting results in the reduction of surface area and bulk volume of the powder. The advantages of tableting includes good chemical and physical stability, prolonged shelf life, and competitive unit production cost, as well as reduction in transportation and storage cost, elegant appearance and greater acceptance in terms of presentation (Yusof et al., 2012). Normally, people take fruit powder tablets either in juice form after dissolving it in

water or as candy. Hence, fast dissolving fruit powder tablets will be more acceptable to consumers. However, the major drawback of fruit powder tablets is poor dissolution rate. To make the tablets dissolve fast, super dis-integrants or effervescent agents are used with fruit powder during tableting (Ong et al., 2014; Zea et al., 2013).



Figure 1.1: Plum fruit candy tablet available in market

Dissolution is a physicochemical process, in which a solute and solvent mix together to produce a uniform mixture. It is very important for the dry and powder food products which are used in liquid or gel/semisolid form. Hence, dissolution rate is a crucial factor for dry foodstuff. Dissolution rate depends upon its physical properties and chemical composition. It also depends on the physical state of material. In the food sector, this test usually used to measure the solubility of powder foodstuff such as food hydrocolloid powder, milk powder, and fruit powder. It is also used for compressed fruit powder such as tablet; for checking total disintegration time, antioxidant release (Adiba et al., 2011). Dissolution testing for tablets is a formal test in pharmaceutical and nutraceutical industries. It is typically used to check the quality of the product, develop new products, and determine the bioavailability of the drug after dissolution (Costa and Lobo, 2001; Dash et al., 2010; Dressman et al., 1998). Dissolution rate of a tablet depends upon the type of ingredients it contains. It may be hydrophilic, moderately hydrophilic, or hydrophobic. The Center for Drug Evaluation and Research at the US Food and Drug Administration (FDA, 1997a) reported three types of dissolution test specification for immediate release products; (1) single point specifications; (2) two-point specifications; and (3) dissolution profile comparison. Among these, the dissolution profile comparison is a more precise technique to specify the product (Sathe et al., 1996; Shah et al., 1998). The methods for comparison of In-vitro dissolution profile may be divided into three groups: (1) Model-dependent method, (Dash et al., 2010; Polli et al., 1997; Sathe et al., 1996; Shah et al., 1992); (2) Model-independent methods (Costa & Lobo, 2001; Dash et al., 2010; Polli et al., 1997; Shah et al., 1992); and (3) Statistical methods (Analysis of variance (ANOVA) based method or *t*-student test) (Costa & Lobo, 2001; Dash et al., 2010; Mauger et al., 1986; Polli et al., 1997). For pharmaceuticals, tablet dissolution performance is measured based on release rate of active pharmaceutical ingredients (API) or drug.

1.2 Problem statement

The demand of ready-to-eat or ready-to-serve foods and drinks is increasing day-by-day. Tableting of fruit powder into ready-to-drink juice tablet/fruit candy is able to meet consumer demands. However, acceptability and popularity of the fruit powder tablets depends on dissolution behavior and dissolution rate. Most fruit powders are sticky in nature due presence of low molecular weight of sugar. To overcome stickiness and increase the glass transition temperature (T_g), a necessary amount of high molecular weight carrier agents must be used during powder production, which able to alter the physicochemical properties and solubility/wettability of powder (Grabowski et al., 2006). The physicochemical properties of powder have a strong influence on compressibility and tableting behavior of fruit powder, which affect the dissolution characteristics of finish products as well.

Dissolution test of fruit powder tablets is vital from the point of large scale production and quality checking of finished products. In-vitro dissolution tests are usually used to: (1) assess the batch-to-batch quality of product; (2) develop new formulations and (3) ensure continued product quality and performance after certain changes, such as changes in the formulation, manufacturing process, the site of manufacture, and the scale-up of the manufacturing process. These are done using dissolution profiling and profile comparison (Yuksel et al., 2000).

From the literature it is clear that although fruit powder effervescent tablets are more beneficial than other products, they are typically not available in local markets. Therefore, based on the demands and benefits of fruit powder tablets, there is a need for large scale production. For large scale production, physicochemical properties of powder are very important in terms of storage, transportation, handling, processing, and quality of finished product as well. Besides, dissolution information of the fruit powder tablets is also very essential for quality control/checking, product development, and scale up in production capacity.

1.3 Objectives

The main objectives of this study are as follows:

1. To investigate the physicochemical properties of fruit powder and their effects on dissolution rate of fruit powder tablets.
2. To investigate dissolution profiling for natural fruit powder fast dissolve tablet and comparison of dissolution profile of natural fruit powder fast-dissolve tablets.

1.4 Scope of study

The physicochemical properties of powder material are of particular importance to food processor and consumer in terms of production and dissolution of fruit powder tablet. Tableting of fruit powder is a special technique of fruit powder preservation and product development. Tablet quality depends on its ingredients' properties, while the suitability of tablet depends upon dissolution behavior and dissolution rate.

Effervescent agents also affect the dissolution characteristics. Literature shows most of the research about fruit powder production, yield, different effects on nutritional quality, drying methods and nutritional properties of fruit powder. However, very limited research may be found in post-processing handling, transportation, storage, preservation and solubility. Therefore, there is room for research about physicochemical properties of fruit powder and their relationship in fruit powder tablet dissolution, dissolution profiling and its comparison for quality control, and large scale production of tablets.

Dissolution rate and behavior of fruit powder tablet differ in different dissolution mediums and environments. In this study, two types of dissolution medium, namely, distilled water and simulated saliva, have been used, as people usually take a fruit powder tablet in juice form after dissolving in water or directly as like candy. The dissolution profile of fruit powder tablets represent the relationship between percent amounts of tablet dissolved and percent solute release from the tablet versus time. Dissolution profiling and profile comparisons of fruit powder tablet are very important in terms of characterization of product, quality control, and scaling up the process. Three types of method are usually used for profile comparison such as model dependent, model independent and statistical method. However, physicochemical properties of fruit powder tablet have a great influence on dissolution rate and behavior of fruit powder tablet and which also affect dissolution profile. Yet, the model based dissolution profile comparison methods does not consider chemical properties of tablet ingredients. The models have been established based on some mathematical metaphor of some aspects of reality and considering physical feature of tablets. Therefore, in this study three types of methods have been used for dissolution profile comparison.

1.5 Organization of the thesis

Chapter 1 contains a general introduction to the thesis. The overview of this research, problem statement, objectives scope of study, and organization of the thesis are also presented.

Chapter 2 discusses and reviews in detail the physicochemical properties of food powder, tablet, tableting method, effervescent tablet, dissolution of tablet, dissolution profiling of tablet, dissolution profile comparison, dissolution kinetics mathematical model, and selection criteria of best fit mathematical model.

Chapter 3 reports all the materials and experimental procedures involved in performing this research. Proximate analysis of fruit powders, physical properties of fruit powder, dissolution test, data collection, and data analysis are described.

Chapter 4 report and discusses on physicochemical properties of fruit powder, dissolution profile of effervescent fruit powder tablet, best fit mathematical model with the dissolution data of fruit powder effervescent tablet. The comparison of, dissolution profile of fruit powder effervescent tablets are performed based on model dependent, model-independent and statistical method.

A brief summary on all study and findings are presented in **chapter 5**. The recommendations for future study are given in this final chapter.

REFERENCES

- Abdelbary, G., Eouani, C., Prinderre, P., Joachim, J., Reynier, J. P., & Piccerelle, P. H. (2005). Determination of the in vitro disintegration profile of rapidly disintegrating tablets and correlation with oral disintegration. *International Journal of Pharmaceutics*, 292(1), 29-41.
- Abdullah, M. D. A., Bepary, S., & Rouf, A. S. S. (2008). Report: in vitro dissolution studies of different brands of sustained release diclofenac sodium matrix tablet available in Bangladesh. *Pakistan Journal of Pharmaceutical Sciences*, 21(1), 70-77.
- Abdullah, R. (2007). Uniaxial die compaction of Ficus deltoidea. B. Eng. (Hons) (Process and Food Eng.). Thesis, Universiti Putra Malaysia.
- Adiba, B. D., Salem, B., Nabil, S., & Abdelhakim, M. (2011). Preliminary characterization of food tablets from date (*Phoenix dactylifera L.*) and spirulina (*Spirulina sp.*) powders. *Powder Technology*, 208, 225-230.
- Adolfsson, A., & Nyström, C. (1996). Tablet strength, porosity, elasticity and solid state structure of tablets compressed at high loads. *International Journal of Pharmaceutics*, 132, 95-106.
- Anandharamkrishnan, C., Rielly, C. D., Stapley, A. G. F. (2007) Effects of process variable on the denaturation of whey proteins during spray-drying. *Drying Technology*, 25, 799-807.
- AOAC. (1990). Official methods of analysis of the association of official analytical chemists (15th ed.) Arlington, VA.
- Anandharamkrishnan, C., Rielly, C. D. and Stapley A. G. F. (2007). Effects of process variables on the denaturation of whey proteins during spray drying. *Drying Technology*, 25, 799-807, 2007.
- Azarmi, S., Roa, W., & Löbenberg, R. (2007). Current perspectives in dissolution testing of conventional and novel dosage forms. *International Journal of Pharmaceutics*, 328(1), 12-21. doi:10.1016/j.ijpharm.2006.10.001
- Betts, G., Cook, S., Melean, B., Betts, R., Sharpe, T., & Walker, S. (2006). Scientific review of the microbiology risks associated with reduction in fat and added sugar in foods. *Food Standards Agency*, 1-55.
- Brenan. (2003). Food Energy-methods of food analysis and conversion factors. *Food and Nutrition Paper*. 77. 7-12.
- Bultmann, J. M. (2002). Multiple compaction of microcrystalline cellulose in a roller compactor. *European Journal of Pharmaceutics and Biopharmaceutics*, 54(1), 59-64.

- Carr, R. L. 1965. Evaluating flow properties of powders. *Chemical Engineering*, 72, 116-168
- Chaudhary, H., Patel, B., Patel, D., & Patel, C. (2012). Indian Journal of Novel Drug Delivery. *Indian Journal of Novel Drug Delivery*, 4(2), 163-171.
- Chen, X. D., & Mujumdar, A. S. (2009). *Drying technologies in food processing*. John Wiley & Sons. Retrieved from <http://books.google.com.my/books?hl>
- Curley, T., Forsyth, R., Sun, S., Fliszar, K., Colletto, M., Martin, G. P. (2004). Measurement of Dissolved Oxygen as a determination of media equilibrium during dissolution testing. *Dissolution technologies*, 5, 6-11.
- Costa, P., & Sousa Lobo, J. M. (2001). Modeling and comparison of dissolution profiles. *European Journal of Pharmaceutical Sciences*, 13(2), 123-133.
- Dash, S., Murthy, P. N., Nath, L., & Chowdhury, P. (2010). Kinetic modeling on drug release from controlled drug delivery systems. *Acta Poloniae Pharmaceutica*, 67(3), 217-223.
- Desai, S. J., Singh, P., Simonelli, A. P., & Higuchi, W. I. (1966a). Investigation of factors influencing release of solid drug dispersed in inert matrices III. Quantitative studies involving the polyethylene plastic matrix. *Journal of Pharmaceutical Sciences*, 55(11), 1230-1234.
- Desai, S. J., Singh, P., Simonelli, A. P., & Higuchi, W. I. (1966b). Investigation of factors influencing release of solid drug dispersed in inert matrices IV. Some studies involving the polyvinyl chloride matrix. *Journal of Pharmaceutical Sciences*, 55(11), 1235-1239.
- Dokoumetzidis, A., & Macheras, P. (2006). A century of dissolution research: From Noyes and Whitney to the Biopharmaceutics Classification System. *International Journal of Pharmaceutics*, 321(1-2), 1-11. doi:10.1016/j.ijpharm.2006.07.011
- 'RUR*\ VNL 3 .XOLQRZVNL 3 0HQG\N \$ -DFKRZLF] 5
drug delivery systems with l-dopa based on carrageenans and hydroxypropylmethylcellulose. *International Journal of Pharmaceutics*, 404(1-2), 169-175.
- Dressman, J. B., Amidon, G. L., Reppas, C., & Shah, V. P. (1998). Dissolution Testing as a Prognostic Tool for Oral Drug Absorption: Immediate Release Dosage Forms. *Pharmaceutical Research*, 15(1), 11-22. doi:10.1023/A:1011984216775
- El-Arini, S. K., & Leuenberger, H. (1998). Dissolution properties of praziquantel-PVP systems. *Pharmaceutica Acta Helveticae*, 73(2), 89-94.
- FDA, (1995). Guidance for Industry: Immediate Release Solid Oral Dosage Forms ± Scale-up and Postapproval Changes: Chemistry, Manufacturing, and Controls; In Vitro Dissolution Testing, and In Vivo Bioequivalence

DocumentationCenter for Drug Evaluation and Research, Rockville, MD
November.

- FDA, (1997a). Guidance for Industry: Dissolution Testing of Immediate Release Solid Oral Dosage Forms. Center for Drug Evaluation and Research, Rockville, MD August.
- FDA, (1997b). Guidance for Industry: SUPAC-MR: Modified Release Solid Oral Dosage Forms \pm Scale-up and Postapproval Changes: Chemistry, Manufacturing, and Controls; In Vitro Dissolution Testing, and In Vivo Bioequivalence DocumentationCenter for Drug Evaluation and Research, Rockville, MD September.
- Fleury, N., & Lahaye, M. (1991). Chemical and physicochemical characterization of fibers from *Laminaria digitata* (Kombu Breton): a physiological approach. *Journal of the Science of Food and Agriculture*, 55, 389 \pm 400.
- Faure, A., York, P., & Rowe, R. C. (2001). Process control and scale-up of pharmaceutical wet granulation processes: a review. *European Journal of Pharmaceutics and Biopharmaceutics*, 52(3), 269 \pm 277. doi:10.1016/S0939-6411(01)00184-9
- Fernandes, F. A., Rodrigues, S., Law, C. L., & Mujumdar, A. S. (2011). Drying of exotic tropical fruits: A comprehensive review. *Food and Bioprocess Technology*, 4(2), 163 \pm 85.
- Fitzpatrick, J. J., (2007). Particle Properties and the Design of Solid Food Particle Processing Operations. *Food and Bioproducts Processing*, 85(4), 308 \pm 14. doi:10.1205/fbp07056
- FMC Biopolymer 2009. Ac-Di-Sol® And Avicel Ph101 Are Trademark Of FMC , 2009 FMC Corporation. All Rights Reserved.
- Freitag, F., & Kleinebudde, P. (2003). How do roll compaction/dry granulation affect the tableting behaviour of inorganic materials? Comparison of four magnesium carbonates. *European Journal of Pharmaceutical Sciences*, 19(4), 281 \pm 289.
- Freitas, M. N., & Marchetti, J. M. (2005). Nimesulide PLA microspheres as a potential sustained release system for the treatment of inflammatory diseases. *International Journal of Pharmaceutics*, 295(1 \pm), 201 \pm 211. doi:10.1016/j.ijpharm.2005.03.003
- Fusayama T, Katayori T, Nomoto S. (1963). Corrosion of gold and amalgam placed in contact with each other. *Journal Dental Research*, 42, 1183 \pm 7.

- Ganesan, V., Rosentrater, K. A., & Muthukumarappan, K. (2008). Flowability and handling characteristics of bulk solids and powders ± a review with implications for DDGS5. *Bio Systems Engineering*, 101, 425-435.
- Gibaldi, M., & Feldman, S. (1967). Establishment of sink conditions in dissolution rate determinations. Theoretical considerations and application to nondisintegrating dosage forms. *Journal of Pharmaceutical Sciences*, 56(10), 1238-1242. doi:10.1002/jps.
- Goula, A. M., Adamopoulos, K. G., & Kazakis, N. A. (2004). Influence of spray drying conditions on tomato powder properties. *Drying Technology*, 22(5), 1129-1151.
- Grabowski, J. A., Truong, V. D., & C. R. Daubert, C. R. (2006). Spray-drying of amylase hydrolyzed sweet potato puree and physicochemical properties of powder. *Journal of Food Science*, 71(5), E209-217.
- Hausner, H.H. 1967 Fraction condition in a mass of metal powder. *International Journal of Powder Metallurgy*. 3 (4), 7-13.
- Hopfenberg HB (1976) In Controlled Release Polymeric Formulations. Paul, D. R., Harris, F. W. (Eds.), ACS Symposium Series 33. American Chemical Society, Washington, DC, pp. 26-31.
- Higuchi W I (1962). Analysis of data on the medicament release from ointments. *Journal of Pharmaceutical Sciences* 5: 802-804.
- Higuchi T (1963). Mechanism of sustained action medication. Theoretical analysis of rate of release of solid drugs dispersed in solid matrices. *Journal of Pharmaceutical Sciences* 52, 1145-1149.
- Hixson, A. W., & Crowell, J. H. (1931). Dependence of reaction velocity upon surface and agitation. *Industrial & Engineering Chemistry*, 23(8), 923-931.
- Jain, C. P., & Naruka, P. S. (2009). Formulation and evaluation of fast dissolving tablets of valsartan. *International Journal of Pharmacy and Pharmaceutical Sciences*, vol. 1, issue 1.
- Jamzad S., Tutunji L., & Fassihi R. (2005). Analysis of macromolecular changes and drug release from hydrophilic matrix systems. *International Journal of Pharmaceutics*, 292, 75-85.
- Jaya S., & Das, H. (2009). Glass transition and sticky point temperatures and stability/mobility diagram of fruit powders. *Food Bioprocess Technology*, 2, 89-95.
- Johanson, J. R. (1978). Know your material-how to predict and use the properties of bulk solids. *Chemical Engineering*, 9-17.

REFERENCES

- Extended release lipophilic indomethacin microspheres: formulation factors and mathematical equations fitted drug release rates. *European Journal of Pharmaceutical Sciences*, 19(2), 99–104.
- Katzhendler, I., Hoffman, A., Goldberger, A., & Friedman, M. (1997). Modeling of drug release from erodible tablets. *Journal of Pharmaceutical Sciences*, 86(1), 110–115.
- Kaunisto, E., Nilsson, B., & Axelsson, A. (2009). Drug dissolution rate measurements ± evaluation of the rotating disc method. *Pharmaceutical Development and Technology*, 14(4), 400–408. doi:10.1080/10837450802712641
- Kaur, H. (2012). Pharmaceutical tablets and tablet compression machines: a review. *Novel Science International Journal of Pharmaceutical Sciences*, 1(8). Retrieved from <http://pharmacy.novelscience.info/index.php>
- Kawakita, K., & Lüdde, K.-H. (1971). Some considerations on powder compression equations. *Powder Technology*, 4(2), 61–68. doi:10.1016/0032-5910(71)80001-3
- Kha, T. C., Nguyen, M. H., & Roach, P. D. (2010). Effects of spray drying conditions on the physicochemical and antioxidant properties of the Gac (*Momordica cochinchinensis*) fruit aril powder. *Journal of Food Engineering*, 98(3), 385–392.
- Kim, E. H.-J., Chen, X. D., & Pearce, D. (2005). Effect of surface composition on the flowability of industrial spray-dried dairy powders. *Colloids and Surfaces B: Biointerfaces*, 46(3), 182–187. doi:10.1016/j.colsurfb.2005.11.005
- Knowlton, T. M., Carson, J. W., Klinzing, G. E., & Yang, W. C. (1994). The importance of storage, transfer and collection. *Chemical Engineering Progress*, 90, 44–54.
- Korsmeyer, R. W., Gurny, R., Doelker, E., Buri, P., & Peppas, N. A. (1983). Mechanisms of solute release from porous hydrophilic polymers. *International Journal of Pharmaceutics*, 15(1), 25–35. doi:10.1016/0378-5173(83)90064-9
- Krögel, I., & Bodmeier, R. (1999). Floating or pulsatile drug delivery systems based on coated effervescent cores. *International Journal of Pharmaceutics*, 187(2), 175–184.
- Langenbucher, F. (1972). Letters to the Editor: Linearization of dissolution rate curves by the Weibull distribution. *Journal of Pharmacy and Pharmacology*, 24(12), 979–981.
- Lebrun, P., Krier, F., Mantanus, J., Grohganz, H., Yang, M., Rozet, E., Hubert, P. (2012). Design space approach in the optimization of the spray-drying

process. *European Journal of Pharmaceutics and Biopharmaceutics*, 80(1), 226–234.

Lee, R.E. (2008). Amerilab technologies: Growing with global market. Retrieved 29/24, 2014, from www.nutraceuticalworld.com

Lin, S. Y., & Yang, J. C. (1989). In-vitro dissolution behavior of some sustained-release theophylline dosage forms. *Pharmaceutica Acta Helvetica*, 64(8), 236–240.

Lokhandwala, H., Deshpande, A., & Deshpande, S. (n.d.). Kinetic modeling and dissolution profiles comparison: An overview. *International Journal of Pharma and Bio Sciences*, 4(1), 728–737.

Liu, L. X., Marziano, I., Bentham, A. C., Litster, J. D., E.T.White, & Howes, T., (2008). Effect of particle properties on the flowability of ibuprofen powders. *International Journal of Pharmaceutics*, 362(1–2), 109–117. doi:10.1016/j.ijpharm.2008.06.023

Mareci, D., Chelariu, R., Dan, I., Gordin, D.-M., & Gloriant, T. (2010). Corrosion of Ti-6Al-4V alloy in artificial saliva. *Journal of Materials Science: Materials in Medicine*, 21(11), 2907–2913.

Mauger, J. W., Chilko, D., & Howard, S. (1986). On the analysis of dissolution data. *Drug Development and Industrial Pharmacy*, 12(7), 969–992.

Wang, Y., & Wang, Y. (2010). Dissolution profiles comparison methods. Retrieved from <http://www.pharmtech.com> /pharmtech.

Meher P. S., Saroj Jain and Neeraj. (2012). Dissolution specifications, dissolution profiling and dissolution profiles comparison methods. *International journal of drug research and technology*, 2 (4S), 297-305

Sharma, S. (2012). KinetDS: an open source software for dissolution test data analysis. *Disso-Lut Technol*, 19(1), 6–11.

Miao, Y., Liu, J., He, X., Huang, Z., & Wang, X. (2010). Study on Development of Capsaicin Effervescent Tablets. *Food Research and Development*, 5, 028.

Miller, R. W. (1997). Roller compaction technology. *Drugs and the Pharmaceutical Sciences*, 81, 99–150.

Millqvist-Fureby, A., Ulla Elofsson, U., & Bergenstahl, B. (2001). Surface composition of spray-dried milk protein-stabilized emulsions in relation to pre-heat treatment of proteins. *Colloids and Surfaces B: Biointerfaces*, 21, 47–58.

- Moore J. W., Flanner H. H. (1996) Mathematical comparison of curves with an emphasis on dissolution profiles. *Pharma Tech* 20, 64-74.
- Mudbidri, A. (2010). Tablet compression principles. *Pharma Times*, 42(11), 44–47.
- Narasimhan, B. (2001). Mathematical models describing polymer dissolution: consequences for drug delivery. *Advanced Drug Delivery Reviews*, 48(2), 195–210.
- Nayak, A. K., & Pal, D. (2013). Formulation optimization and evaluation of jackfruit seed starch–alginate mucoadhesive beads of metformin HCl. *International Journal of Biological Macromolecules*, 59, 264–272. doi:10.1016/j.ijbiomac.2013.04.062
- Ngwuluka, N. C., Idiakhwa, B. A., Nep, E. I., Ogaji, I., & Okafor, I. S. (2010). Formulation and evaluation of paracetamol tablets manufactured using the dried fruit of Phoenix dactylifera Linn as an excipient. *Research in Pharmaceutical Biotechnology*, 2(3), 025–032.
- Ning, W., Xuefeng, C., & Ruiping, W. (2007). The Rrocessing Technology of Effervescent Bubbles Tea Drinks Tables. *Food and Fermentation Industries*, 33(3), 151.
- O'hara, T., Dunne, A., Butler, J., & Devane, J. (1998). A review of methods used to compare dissolution profile data. *Pharmaceutical Science & Technology Today*, 1(5), 214–223.
- Ong, M. Y., Yusof, Y. A., Aziz, M. G., Chin, N. L., & Amin, N. A. (2014). Characterisation of fast dispersible fruit tablets made from green and ripe mango fruit powders. *Journal of Food Engineering*, 125, 17–23.
- Ozdikicierler, O., Dirim, S. N., & Pazir, F. (2014). The effects of spray drying process parameters on the characteristic process indices and rheological powder properties of microencapsulated plant (< i> Gypsophila</i>) extract powder. *Powder Technology*, 253, 474–480.
- Peppas, N. A. (1984). Analysis of Fickian and non-Fickian drug release from polymers. *Pharmaceutica Acta Helvetiae*, 60(4), 110–111.
- Peppas, N. A. (1985). Analysis of Fickian and non-Fickian drug release from polymers. *Pharmaceutica Acta Helvetiae*, 60, 110–111.
- Pettipher, G. L., Osmundson, M. E., & Murphy, J. M. (1997). Methods for the detection and enumeration of Alicyclobacillus acidoterrestris and investigation of growth and production of taint in fruit juice and fruit juice-containing drinks. *Letters in Applied Microbiology*, 24(3), 185–189. doi:10.1046/j.1472-765X.1997.00373.x
- Phisut, N. (2012). Spray drying technique of fruit juice powder: some factors influencing the properties of product. *International Food Research Journal*, 19(4). Retrieved from <http://search.ebscohost.com/login.aspx?direct=t>

- Pitt, K., & Sinka, C. (2007). Chapter 16 Tableting. In M. J. H. and J. P. K. S. A.D. Salman (Ed.), *Handbook of Powder Technology* (Vol. Volume 11, pp. 735 ± 778). Elsevier Science B.V. Retrieved from <http://www.sciencedirect.com/>
- Podczec, F., & Sharma, M. (1996). The influence of particle size and shape of components of binary powder mixtures on the maximum volume reduction due to packing. *International Journal of Pharmaceutics*, 137(1), 41 ± 47. doi:10.1016/0378-5173(95)04420-5
- Polli, J. E., Rekhi, G. S., Augsburger, L. L., & Shah, V. P. (1997). Methods to compare dissolution profiles and a rationale for wide dissolution specifications for metoprolol tartrate tablets. *Journal of Pharmaceutical Sciences*, 86(6), 690 ± 700.
- Quek, S. Y., Chok, N. K., & Swedlund, P. (2007). The physicochemical properties of spray-dried watermelon powders. *Chemical Engineering and Processing: Process Intensification*, 46(5), 386 ± 392.
- Ranganna, S. (1997). Manual of analysis of fruit and vegetables products. New Delhi: MacGrw Hill Company Ltd.
- Renard, C. M. G. C., Crepeau, M. J., & Thibault, J. F. (1994). Influence of ionic strength, pH and dielectric constant on hydration properties of native and modified fiber from sugar-beet and wheat bran. *Industrial Crops and Products*, 3, 75 ± 84.
- Riepma, K. A., Vromans, H., Zuurman, K., & Lerk, C. F. (1993). The effect of dry granulation on the consolidation and compaction of crystalline lactose. *International Journal of Pharmaceutics*, 97(1 ± 3), 29 ± 38. doi:10.1016/0378-5173(93)90123-W
- Riippi, M., Yliruusi, J., Niskanen, T., & Kiesvaara, J. (1998). Dependence between dissolution rate and porosity of compressed erythromycin acistrate tablets. *European Journal of Pharmaceutics and Biopharmaceutics*, 46, 169 ± 175.
- Robinson, J. R., McGinity, J. W., & Delmas, P. (2003, November 18). Effervescent granules and methods for their preparation. Retrieved from <http://www.americanpharmaceuticalreview.com/Featured-Articles/>
- Sahoo, P. K. (2007). Asstt. Professor Delhi Institute of Pharmaceutical Sciences and Research Pusph Vihar-III, MB Road, New Delhi-110017. Retrieved from <http://www.share-pdf.com/1d662dae049146adaec26c7c848ae205/Tablets.pdf>
- Sathe, P. M., Tsong, Y., & Shah, V. P. (1996). In-vitro dissolution profile comparison: statistics and analysis, model dependent approach. *Pharmaceutical Research*, 13(12), 1799 ± 803.
- Schwartz, J. B., Simonelli, A. P., & Higuchi, W. I. (1968a). Drug release from wax matrices I. Analysis of data with first-order kinetics and with the diffusion-

- controlled model. *Journal of Pharmaceutical Sciences*, 57(2), 274–277. doi:10.1002/jps.2600570206
- Schwartz, J. B., Simonelli, A. P., & Higuchi, W. I. (1968b). Drug release from wax matrices II. Application of a mixture theory to the sulfanilamide² wax system. *Journal of Pharmaceutical Sciences*, 57(2), 278–282. doi:10.1002/jps.2600570207
- Schmitta, C., Sanchez, C., Desobry-Banona, S. and Hardya, J., (2010). Structure and Technofunctional Properties of Protein-Polysaccharide Complexes: A Review
- Shah, V. P., Midha, K. K., Dighe, S., McGilveray, I. J., Skelly, J. P., Yacobi, A., Spector, S. (1992). Analytical methods validation: Bioavailability, bioequivalence, and pharmacokinetic studies. *Journal of Pharmaceutical Sciences*, 81(3), 309–312. doi:10.1002/jps.2600810324
- Shah, V. P., Tsong, Y., Sathe, P., & Liu, J.-P. (1998). In Vitro Dissolution Profile Comparison²Statistics and Analysis of the Similarity Factor, f₂. *Pharmaceutical Research*, 15(6), 889–896. doi:10.1023/A:1011976615750
- Shangraw, R., Mitrejev, A., & Shah, M. (1980). A new era of tablet disintegrants. *Pharmaceutical Technology*, 4, 49–57.
- Sharma, S., Sher, P., Badve, S., Pawar, A.P., 2005. Adsorption of meloxicam on porous calcium silicate: Characterization and tablet formulation. *AAPS PharmSciTech* 6, 618–625. doi:10.1208/pt060476
- Shoib, M. H., Tazeen, J., Merchant, H. A., & Yousuf, R. I. (2006). Evaluation of drug release kinetics from ibuprofen matrix tablets using HPMC. *Pakistan Journal of Pharmaceutical Sciences*, 19(2), 119–124.
- Shittu, T. A., & Lawal, M. O. (2007). Factors affecting instant properties of powdered cocoa beverages. *Food Chemistry*, 100, 91–98.
- Siepmann, J., Kranz, H., Peppas, N. A., & Bodmeier, R. (2000). Calculation of the required size and shape of hydroxypropyl methylcellulose matrices to achieve desired drug release profiles. *International Journal of Pharmaceutics*, 201(2), 151–164.
- Siewert, M., Dressman, J., Brown, C. K., Shah, V. P., Aiache, J.-M., Aoyagi, N., Williams, R. (2003). FIP/AAPS guidelines to dissolution/in vitro release testing of novel/special dosage forms. *AAPS PharmSciTech*, 4(1), 43–52. doi:10.1208/pt040107
- Singhvi, G., & Singh, M. (2011). Review: in vitro drug release characterization models. *International Journal of Pharmaceutical Studies and Research*, 2, 77–84.
- Skibsted L H, Risbo J, Andersen M L (2010). Chemical deterioration and physical instability of food and beverage. Elsevier ±Technology & Engineering.
- Srinath, K. R., Chowdary, C. P., Palanisamy, P., Krishna, A. V., Aparna, S., & others. (2011). Formulation and evaluation of effervescent tablets of paracetamol.

