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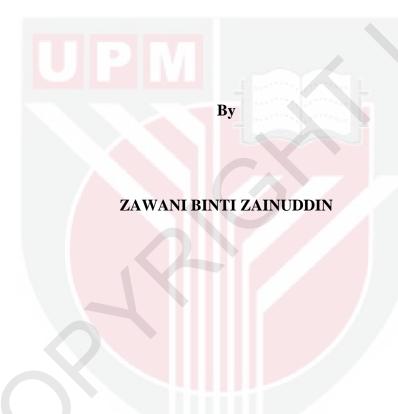
TREATMENT OF WASTEWATER FROM KENAF WATER-RETTING PROCESS

ZAWANI BINTI ZAINUDDIN

FK 2014 170



TREATMENT OF WASTEWATER FROM KENAF WATER-RETTING PROCESS



The submitted to the School of Graduates Studies, Universiti Putra Malaysia, in fulfilment of the requirements for the Degree of Doctor of Philosophy

August 2014

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Abstract of thesis presented to the senate of universiti putra malaysia in fulfilment of the requirement for the degree if doctor of philosophy

TREATMENT OF WASTEWATER FROM KENAF WATER-RETTING PROCESS

By

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August 2014

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Faculty: Engineering

Treatment wastewater from kenaf-retting process was implemented using three types of treatment process, namely coagulation, sequencing batch reactor (SBR) and adsorption process. Each of the tested process has advantages and disadvantages. Characteristic of wastewater was measured to determine presence of chemical composition. Concentration of chemical oxygen demand and colour were recorded in high concentration, which is more than 2000mg/L and 1000mg/L respectively. Most of parameters tested did not comply with the standard and regulation that regulated by Department of Environment (DOE) Malaysia.

In coagulation process, five types of coagulants were selected based on its reputation achieved in previous research. There are ferric chloride, aluminium sulphate, ferric sulphate, *Moringa oleifera* and *Jatropha curcas*. Results obtained from the experiment have given satisfactory results; more than 90% of turbidity was removed from the wastewater by all coagulants tested. All coagulants tested except ferric sulphate required acidic condition to work at optimum condition. Nevertheless, for the removal of COD all coagulants showed an average performance.

Then, the wastewater was tested with biological treatment, which activated sludge from nearby sewerage treatment plant used as source of bacteria. Diversity of microorganisms live in the sludge is a main criterion for the selection. Acclimatization process was implemented before biological treatment executed. Achromobacter, bacillus and acinetobacter were identified as dominant species lived in the wastewater. For biological treatment, sequencing batch reactor was selected to treat the wastewater. Two main parameters were tested; there are food-to-microorganisms (F/M) ratio and hydraulic retention time (HRT). SBR shows the best condition at low F/M ratio, which is 0.25, and the lowest of COD concentration recorded was 163mg/L. whilst, for HRT, SBR showed the best efficiency at HRT 24 per cycle. At higher HRT, the performance of SBR becomes less efficient. The lowest COD value in the final effluent for HRT24 was recorded at 114mg/L.

The last treatment process tested was adsorption process. Activated charcoal was selected as an adsorbent. The highest uptake rate in various initial pH was found at pH8, and the adsorption process efficiency in this experiment was pH dependent. The increasing amount of adsorbent dosage enhanced the COD removal rate, but not the sorption capacity. However, 1.0g of activated charcoal was selected as the best dosage when the cost of adsorbent and treatment take into consideration and the value of COD in the effluent was 339mg/L. In equilibrium studies, COD concentration had shown well agreed with the Redlich-Peterson with high correlation coefficient values compared to other models

Combination of coagulation process with SBR and activated charcoal showed a good performance and high efficiency in removing total suspended solids, chemical oxygen demand and colour. Coagulation process is capable to remove COD and TSS averagely around 80% and 40% respectively. Through SBR-AC process, the value of COD become lower with the reduction recorded around 133mg/L only. Overall the integration process was the best treatment process compared to other three processes.



Abstrak tesis yang dikemukakan kepada Senat Unversiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

RAWATAN AIR SISA DARI PROSES RENDAMAN KENAF

Oleh

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Rawatan air sisa dari proses pemisahan fiber Kenaf dilakukan melalui empat jenis rawatan iaitu process pengumpalan, reaktor penjujukan berkelompok, process penjerapan dan proses pengumpalan bersama reaktor penjujukan berkelompok yang berintegrasi dengan bahan penjerap. Setiap proses yang diuji memberikan kelebihan dan kekurangan yang tersendiri. Karakter bagi air sisa diuji terlebih dahulu bagi memastikan komposisi kimia yang hadir didalamnya. Kehadiran keperluan oksigen kimia, pepejal dan warna yang tinggi melebihi 1000mg/L. Kebanyakan parameter yang diuji menunjukkan bahawa kandungan didalam air sisa tersebut tidak mematuhi piawaian yang ditetapkan oleh Jabatan Alam Sekitar Malaysia.Kandungan bahan organik dan bukan organik juga diuji bagi mengetahui bahan spesifik yang terkandung didalam air sisa yang dikaji. Elemen bukan organik yang diuji seperti mempunyai kepekatan yang rendah. Manakala bahan organik pula terdiri daripada bahan semulajadi yang terdapat pada tumbuhan kenaf.

Melalui proses pengumpalan yang dilakukan, lima jenis pengumpal telah dipilih berdasarkan reputasi yang dicapai melalui kajian yang lepas. Ia adalah, *ferric chloride, aluminium sulphate, ferric sulphate, Moringa Oleifera* dan Jatropha Curcas. Melalui eksperimen yang telah dijalankan, didapati bahawa proses tersebut menunjukkan hasil yang sangat memuaskan dimana, penyingkiran kekeruhan dari sampel air sisa melebihi 90% dari jumlah keseluruhan kepekatan kekeruhan dari sampel. Kesemua pengumpal kecuali *ferric sulphate* memerlukan keadaan berasid bagi membolehkannya berfungsi secara optima. Namun begitu, dari segi penyingkiran keperluan oksigen kimia berada diparas sederhana bagi kesemua jenis pengumpal yang diuji.

Seterusnya, air sisa diuji pula dengan rawatan secara biologi dengan menggunakan bakteria yang diambil dari loji rawatan kumbahan yang berdekatan. Melalui proses penyesuaian yang dijalankan, terbukti bakteria yang digunakan mampu merawat air sisa dengan baik. *Achromobacter, bacillius* dan *acinetobacter* merupakan spesis yang mendominasi didalam sampel air sisa . Bagi ujian kaedah penjujukan berkelompok, dua jenis parameter yang diuji iaitu nisbah *food-to-microorganisms*

(F/M) dan juga masa tertahan hidraulik (HRT). Melalui ujian yang telah dijalankan, didapati bahawa, semakin rendah kadar nisbah F/M yang digunakan semakin bagus reaktor merawat air sisa. Nisbah F/M yang paling berkesan dalam rawatan ini adalah 0.25 dan nilai COD didalam sampel akhir adalah sekitar163mg/L. Manakala bagi masa tertahan hidarulik pula yang diperlukan oleh reaktor adalah lebih kurang 24 jam bagi satu pusingan. Semakin tinggi masa tertahan hidraulik yang digunakan semakin rendah kecekapan reaktor merawat air sisa yang diuji. Nilai COD paling rendah yang diperolehi adalah sekitar 114mg/L sahaja.

Seterusnya, proses penjerapan dengan mengunakan bahan karbon teraktif sebagai penjerap. Berdasarkan daripada kepeutusan yang diperolehi menunjukkan bahawa bahan penjerap berupaya menyingkirkan kepekatan COD pada keadaan pH8 dan 283mg/L merupakan nilai COD akhir yang dicatatkan. Bagi ujian perbezaan berat bahan karbon teraktif yang dijalankan menunjukkan bahawa semakin berat bahan karbon teraktif yang digunakan semakin tinggi kadar penyingkiran kepekatan COD diperolehi. Namun begitu, dengan mengambila kira kos bahan dan rawatan, 1.0g bahan penjerap dikira yang paling baik dalam menyingkirkan bahan enapcemar daripada air sisa. Nilai COD akhir yang dicatatkan adalah 339mg/L.

Rawatan melalui proses pengumpalan bersama kaedah penjujukan berkelompok yang berintegrasi dengan bahan penjerap (SBR-AC) merupakan kaedah yang keempat yang diuji. Melalui ujian yang dijalankan, didapati bahawa, kombinasi kaedah rawatan ini mempunyai kecekapan yang tinggi dan efisien dalam menyingkirkan permintaan oksigen kimia, pepejal terampai dan warna yang menjadi punca utama kepada masalah ini. Proses penjerapan berjaya menurunkan kadar kepekatan COD dan pepejal terampai masing-masing sekitar 80% dan 40%. Seterusnya air sisa itu dirawat dengan kaedah SBR-AC, dan nilai COD menurun sehingga 133mg/L sahaja, manakala TSS pula mengalami peningkatan sedikit. Secara keseluruhannya, keputusan ujian yang telah dijalankan menunjukkan bahawa proses integrasi ini berjaya menurunkan kadar kepekatan bahan enapcemar dengan begitu baik sekali. Secara keseluruhannya, proses integrasi menunjukkan prestasi yang terbaik berbanding dengan kaedah rawatan yang lain.

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This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Doctor of Philpsophy. The members of the Supervisory Committee were as follows:

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LIST OF ABBREVIATIONS

Notations / symbols		
°C	Celcius	
Å	Length, $= 10^{-10}$ m	
AA	Activated alumina	
ADMI	American Dye Manufacturer Institute	
AlS	Aluminium sulphate	
APHA	American Public Health Association	
AS	Acitivated sludge	
B.C	Before century	
b_j	Redlich-Peterson constant	
BOD	biological oxygen demand	mg.L ⁻¹
BOD5	biological oxygen demand for 5-days	
Cd	Cadmium	
C_e	Bulk phase concentration	$mg.L^{-1}$
C_o	Initial concentration	$mg.L^{-1}$
CO	Carbon monoxide	
Co	Cobalt	
CO2	Carbon Dioxide	
COD	chemical oxygen demand	$mg.L^{-1}$
Cr	Chromium	
C_t	Concentration of the solution at time t	$mg.L^{-1}$
Cu	Copper	
DOE	Department of Environment	
F/M ratio	food-microorganism ratio	
FC	Ferric chloride	
FS	Ferric sulphate	
g	gram	
GAC	Granular activated carbon	
GCMS	Gas Chromatography Mass Spectrophotometer	
HCl	Hydrochloric acid	
hr HRT	hour hydraulic retention time	hr
ICP	Inter couple Plasma	111
IUPAC	International Union of Pure and Applied Chemistry	
K_F	Freundlich adsorption capacity	mg/g
kg/mol	kilogram per mol	
K_{j}	Adsorption capacity	L.g ⁻¹
K_L	Langmuir constant	$L.g^{-1}$

L	litre	
l/g	litre per gram	
l/mg	litre per miligram	
m	Adsorbent mass	g
mg	miligram	
mg/L	miligram per litre	
min	minute	
min	minute	
mL	mililitre	
MLSS	Mixed liqour suspended solids	mg.L ⁻¹
n	Surface heterogeneity	
N	normality	
NA NaOH	Not available	
NTU	Sodium Hydroxide Nephelometric Turbidity Unit	
O_2	Oxygen	
PAC	powdered activated carbon	
PtCo	Platinum Cobalt	
q_e	Solid phase concentration at equilibrium	mg/g
$q_{e,calc}$	Predicted solid phase concentration at equilibrium	mg/g
$q_{e,calc}$ $q_{e,exp}$	Solid phase concentration at equilibrium obtained	
1 -, <i>T</i>	from experiment	mg/g
q_t	Solid phase concentration at time t	mg/g
R_L	Dimensionless separation factor	
rpm	rotation per minute	
SBR	Sequencing batch reactor	
SS	Suspended solids	
t	Time	minute
TDS	Total dissolved solids	$mg.L^{-1}$
TSS	Total suspended solids	$mg.L^{-1}$
V	Volume of solution	mL
VSS	volatile suspended solids	$mg.L^{-1}$
α_L	Energy of adsorption	dm ³ .mg ⁻¹
β	Heterogeneity factor	

CHAPTER 1

INTRODUCTION

1.1 Background of studies

Recently, a lot of researchers focused on natural fibres in finding new materials to produce environmental friendly and renewable products. There are numbers of researches have been done on various types of natural fibres such as jute, sisal, hemp and kenaf due to capability of these plants to meet the requirement as eco-friendly products. Among natural fibres, kenaf has its own reputation as an alternative material in industry and research field. Kenaf has been recognized as one of the most important fibres in product development like composites, pulp/paper, particle boards etc.

Bast-fibres that found in the kenaf plant are playing an important role as a raw material for plant-based or bio-composite products. Nevertheless, bast-fibres cannot be used directly as a raw material because it needs to be separated into individual fibre. This individual fibre can be obtained through retting process.

Retting process can be defined as a slow degradation process that can take up a few days to complete. There are various types of retting, for an example using chemical, dew- or water-retting. Commonly water-retting procedure is selected because it is simple and often produced a good quality of fibre. However, large amount of water in fibre separation (retting) process is required to obtain the fibres. Typically water that has been used in the process will be released to the watercourse without any treatments. The condition of wastewater that turbid and produces odour makes it unacceptable by public. This problem can be resolved with an appropriate treatment process.

For that reason in this research study, wastewater is characterized and the treatment process was designed to treat the wastewater. Therefore, coagulation process, sequencing batch reactor (SBR) and adsorption process were proposed to treat the wastewater. These processes are well-known for their effectiveness in treating wastewater either from municipal or industrial.

1.2 Problem statement

i. Generation of wastewater from kenaf water-retting process

At present, industrial sector are producing fibres in large quantity through waterretting process, hence, the process generating enormous amount of wastewater from the separation process. Generation of wastewater from kenaf retting process in Malaysia is unknown due to lack of information. Nevertheless, according to Mondal & Kaviraj (2008) as reported by Huda et al. (2012) for 10,000kg of jute, approximately 432m³ of water is used in water-retting process. Since kenaf and jute are from one family known as *Malvaceae* so the retting process is similar. In the future, amount of wastewater will be increased tremendously when the amount of fibre production will be greater than before as kenaf become one of the commodities in Malaysia. An effective wastewater treatment process is necessitated to protect the environment, to fulfil the standard required by authority and also to help to recover water resource. However, cost of operation and land area might be the main obstacles for small industry and farmers to treat the wastewater properly.

ii. Presence of foreign matter and compound from the water-retting process to aquatic environment

Solids either suspended or dissolved can be found due to presence of fibres and other component during the retting process. These solids have caused the wastewater become cloudy. Organic substances also can be related to increasing turbidity in the wastewater. A material such as lignin was diffused into the water during the retting process. Therefore, high turbidity which consists of suspended particles, colloids and substances in the wastewater will induce interfere an aquatic ecosystem when the wastewater release into the ecosystem without any treatment due its potential to deplete oxygen (O_2) in the water.

The existence of colour is often related to the presence of certain substance in the wastewater. Generally, colour in the wastewater is unacceptable by the public. Unlike man-made coloured wastewater, it is quite difficult to get rid of the colour from the wastewater without knowing type of compound that responsible for its existence. Furthermore, if the wastewater is released at the point for water intake, it will affect quality of water which will increase the cost of water treatment.

iii. Reclaim and reuse treated wastewater for irrigation

Clean water is an important resource for human and become main attention recently due to declination of water source around the world. The reduction source occurred mainly due to water pollution caused by human activities. Therefore, recycle and reuse program was introduced to optimize an application of treated wastewater for other application such as watering public park, field or farm. However, reuse the wastewater for farming application it is depend on the origin of wastewater, quality of effluent and also whether it is for edible crops or not.

1.3 Objective of studies

The detailed research objectives are listed below.

- i. to evaluate the performance of various coagulants in coagulation/flocculation process in treating kenaf-retting wastewater
- ii. to evaluate the performance of biological treatment, Sequencing Batch Reactor (SBR) in treating kenaf-retting wastewater
- iii. to evaluate the performance of adsorption process in the treatment of kenafretting wastewater.
- iv. to evaluate performance of integrated treatment process by combining coagulation, SBR and adsorption process as one unit treatment to treat kenafretting wastewater

1.4 Scope of studies

There are various types of treatment process consist of physical, chemical and biological were selected and investigated in this study; coagulation, adsorption and sequencing batch reactor (SBR) process and also combination of these three methods in as one treatment unit. These processes are well-known for its good performance in treating municipal or industrial wastewater. The effluent was also investigated for its potential to reuse and reclaim for other activities. The scopes can be summarized as follows;

i. Analyze the characteristic and behaviour of wastewater

There were few or none previous study on characteristic of kenaf-retting wastewater has been reported. For that reason, characteristic and behaviour study of this wastewater is important before any treatment can be done. Therefore, several analyses were implemented to fulfil the intent. American Public Health Association (APHA) standard procedure has been used to serve as guideline to analyze the kenafretting wastewater.

ii. Evaluate performance of selected coagulant and its optimum condition to treat the wastewater

Due to existence of suspended solids (SS) and colloids in the wastewater, thus coagulation process was selected to treat the wastewater. Coagulation is managed to remove any particulate and substances in the wastewater by using several types of selected coagulants that consists of chemical and natural coagulants. Optimum condition for every type of coagulant was determined in terms of coagulant dosage and pH of wastewater sample.

iii. Acclimatization of activated sludge from other source in the wastewater

Activated sludge (AS) from sewerage treatment plant was used as a seeding for SBR since it is difficult to find microbes in the wastewater. Diversity of microorganisms live in the sludge makes it easy to receive different kinds of substrate and pollutants. Acclimatization process was run to introduce the AS with the new wastewater and performance of microorganisms was monitored. Batch method was took place in the acclimatization process, which wastewater sample was introduced to AS for certain period of time and then withdrawn the sample. After that the process was continue introduced new samples until the AS reaches its stabilization. Acclimatized sludge was cultivated and isolated for microbe's species identification. After the AS was acclimatized with the wastewater, the AS is ready to be used as inoculums in biological treatment process.

iv. to examine the effectiveness of sequencing batch reactor (SBR)

Wastewater produced from water-retting process contains high organic compound and unpleasant odour. Biological treatment like sequencing batch reactor (SBR) has capability to treat the wastewater with a lot of advantage such as capability to receive high shock of loading rate and eliminate odour. Acclimatized AS was used as inoculums in the SBR process. Main parameters were exploited in the study such as hydraulic retention time (HRT) and food to microorganism (F/M) ratio.

v. to explore the potential and the efficiency of adsorption process

Adsorption process was considered in this study due to its efficiencies in treating innumerable types of wastewater. Granular charcoal activated carbon was picked as an adsorbent in this study. Batch experiment was implemented in the process such as equilibrium study, effect of adsorbent dosage and pH of sample.

vi. to measure performance and effectiveness of integrated system to treat the wastewater

Coagulation process as a pre-treatment process, combining with SBR that integrated with activated carbon was implemented in this study. Optimum condition employed in the experiment was obtained from previous experiment – coagulation, SBR and adsorption.

REFERENCES

- Abdessemed, D., Nezzal, G., & Aim, R. Ben. (2000). Coagulation—adsorption ultrafiltration for wastewater treatment and reuse. *Desalination*, 131(October), 307–314.
- Abdulmoniem, M. S. & Zaid, I. A. (2011). An In-vitro antimicrobial activity of Moringa Oleifera L. seed extracts against different groups of microorganisms. *Australian Journal of Basic and Applied Sciences*. 129-134.
- Abidin, Z. et al. (2011). A preliminary study on Jatropha Curcas as coagulant in wastewater treatment. *Environ. Technol.* 32(9): 971-977.
- Abidin, Z. Z., Mohd Shamsudin, N. S., Madehi, N., & Sobri, S. (2013). Optimisation of a method to extract the active coagulant agent from Jatropha curcas seeds for use in turbidity removal. *Industrial Crops and Products*. 41: 319–323.
- Achten, W. et al. (2007). Jatropha biodiesel fueling sustainability. *Biofuels, Bioproducts and Biorefining*. 1(4): 283-291.
- Adamson, A. & Gast, A. (1997). *Physical Chemistry of Surfaces*. Sixth ed. ed. New York: Wiley-Interscience.
- AECOM, M. &. E. I. a. et al. (2007). *Water reuse: Issuses, Technologies, and application.* New York: McGraw Hill Professional.
- Ahmad, A, Wong, S., Teng, T., & Zuhairi, a. (2008). Improvement of alum and PACl coagulation by polyacrylamides (PAMs) for the treatment of pulp and paper mill wastewater. *Chemical Engineering Journal*. 137(3): 510–517.
- Ahmad, A. L., & Puasa, S. W. (2007). Reactive dyes decolourization from an aqueous solution by combined coagulation/micellar-enhanced ultrafiltration process. *Chemical Engineering Journal*. 132(1-3): 257–265.
- Ahmed, A. M. (1993). *Modelling of biological wastewater treatment in SBR*. King Fadh University of Petroleum and Minerals, Saudi Arabia.
- Aksu, Z., & Tezer, S. (2000). Equilibrium and kinetic modelling of biosorption of Remazol Black B by Rhizopus arrhizus in a batch system: effect of temperature. *Process Biochemistry*. 36(5): 431–439.
- Aldegs, Y., Elbarghouthi, M., Elsheikh, a, & Walker, G. (2008). Effect of solution pH, ionic strength, and temperature on adsorption behavior of reactive dyes on activated carbon. *Dyes and Pigments*: 77(1), 16–23.
- Alexander, B., Mishra, S., & Lampke, T. (2005). Plant Fibers as Reinforcement Green Composites. In L. K. Mohanty, Amar; Misra, Manjusri; T. Drzal (Ed.), Natural Fibers, Biopolymers, and Biocomposites (p. 896). CRC Press.
- Alexopoulou, E., Christou, M., Mardikis, M., & Chatziathanassiou, A. (2000). Growth and yields of kenaf varieties in central Greece. *Industrial Crops and*

Products. 11(2-3): 163–172.

- Allen, S., McKay, G. & Porter, J. (2004). Adsorption isotherm model for basic dyeadsorption by peat in single and binary component systems. *J. Colloid Interface Sci.*280: 322-333.
- AlMubaddal, F., AlRumaihi, K., & Ajbar, A. (2009). Performance optimization of coagulation/flocculation in the treatment of wastewater from a polyvinyl chloride plant. *Journal of Hazardous Materials*. 161(1): 431–438.
- Amuda, O. S., & Alade, a. (2006). Coagulation/flocculation process in the treatment of abattoir wastewater. *Desalination*. 196(1-3): 22–31.
- APHA, A. P. H. A. (2005). Standard methods for the examinations for water and wastewater. Washington DC: APHA.
- Aritkhodzhaev, K. (1995). Pectin from green kenaf bast. *Chemistry of Natural* Compounds. 31(2): 159–162.
- Ayoub, G. M., Hamzeh, A., & Semerjian, L. (2011). Post treatment of tannery wastewater using lime/bittern coagulation and activated carbon adsorption. *Desalination*. 273(2-3): 359–365.
- Bansal, R. & Goya, M. (2005). Activated carbon adsorption. Florida, USA: CRC Press.
- Bennet, T. A. (2007). Evaluation of bench-scale sequencing batch reactor swine waste treatment under continuous and cyclic operation. North Carolina State University, North Carolina, USA.
- Beychok, M. R. (1967). Aqueous wastes from petroleum and petrochemical plants. New York: John Wiley & Sons, Inc.
- Bhatia, S., Othman, Z., & Ahmad, A. L. (2007). Pretreatment of palm oil mill effluent (POME) using Moringa oleifera seeds as natural coagulant. *Journal of Hazardous Materials*. 145(1-2): 120–6.
- Bhatia, S., Zalina, O. & Abdul Latif, A. (2007). Coagulation-flocculation process for POME treatment using Moringa oleifera seeds extract. *Chemical Engineering Journal*. 133: 205-212.
- Boulinguiez, B., Cloirec, P. L. & Wolber, D. (2008). Revisiting the determination of Langmuir parameters application to tetrahydrothiophene adsorption onto activated carbon. *Langmuir* .24: 6420-6424.
- Bramer, H. C. & Hoak, R. D. (1967). *Water reclamation. In: Water reuse*. (pp. 92-95) New York: American Institute of Chemical Engineers.
- Button, D. & Robertson, B. (1985). Effect of toulene exposure time and concentration on induction of high affinity values for toulene oxidation by bacteria of estuarine seawater samples. *Mar. Ecol. Prog. Ser.* 26: 53-65.

- Cecen, F. & Aktas, O. (2012). Activated Carbon for Water and Wastewater Treatment. Weiham: Wiley-VCH Verlag & Co.
- Chen, W., & Liu, J. (2011). The possibility and applicability of coagulation-MBR hybrid system in reclamation of dairy wastewater. *Desalination*. 285: 226–231.
- Chern, J. M., & Wu, C. Y. (2001). Desorption of dye from activated carbon beds: effects of temperature, pH, and alcohol. *Water Research*. 35(17): 4159–65.
- Clark, R. M. & Jr., B. W. L. (1989). *Granular Activated Carbon*. Michigan: Lewis Publisher, Inc.
- Cooper, J. C. & Hanger, D. G. (1967). *Water reclamation with Granular Activated Carbon. In: Water Reuse.* (pp. 185-192). New York: American Institute of Chemical Engineers.
- Coulson, J. & Richarson, J. (1991). *Chemical Engineering. In: Particle Technology and Separation Processes.* Fourth Edition ed. Headington Hill Hall, Oxford: Pergamon Press.

Department of Statistics Malaysia, 2013. Department of Statistics Malaysia. [Online]

- Dialynas, E., & Diamadopoulos, E. (2008). Integration of immersed membrane ultrafiltration with coagulation and activated carbon adsorption for advanced treatment of municipal wastewater. *Desalination*. 230(1-3): 113–127.
- Dosta, J., Rovira, J., Galí, a, Macé, S., & Mata-Alvarez, J. (2008). Integration of a Coagulation/Flocculation step in a biological sequencing batch reactor for COD and nitrogen removal of supernatant of anaerobically digested piggery wastewater. *Bioresource Technology*. 99(13):5722–30.
- Douglas, M., Shamsuzzaman, F. & Knaebel, K. (1994). *Pressure Swing Adsorption*. New York: VCH Publishers Inc.
- Eichner, C., Erb, R., Timmis, K. & Wagner-Dobler, I. (1999). Thermal gradient gel electrophoresis analysis of bio-protection from pollutant shocks in the activated sludge microbial community. *Appl Environ Microbiol*. 65:102-109.
- El Bestawy, E., Helmy, S., Hussein, H., & Fahmy, M. (2013). Optimization and/or acclimatization of activated sludge process under heavy metals stress. World *Journal of Microbiology & Biotechnology*. 29(4):693–705.
- El-Gohary, F., & Tawfik, A. (2009). Decolorization and COD reduction of disperse and reactive dyes wastewater using chemical-coagulation followed by sequential batch reactor (SBR) process. *Desalination*. 249(3): 1159–1164.
- Fink, W. (1984). Identifizierung, Reindarstellung und strukturaufklarung flockungsaktiver wirkstoffe aus hoheren planzen zur wasserreinigung. s.l.:Univ. Heidelberg.
- Foo, K. & Hameed, B. (2010). Insights into the modeling of adsorption isotherms

systems. Chemical Engineering Journal. 156: 2-10.

- Fournier, J., Codaccioni, P. & Soulas, G., 1981. Soil adaptation to 2,4-D degradation in relation to the application rates and the metabolic behaviour of the degrading microflora. *Chemosphere*. 10: 977-984.
- Gassen, H. G., Gassenschmidt, U., D., J. K. & Tauscher, B. (1991). Chemical Properties of flocculating active proteins from Moringa Oleifera. *Biol. Chem. Hoppe-Seyler.* 372: 659.
- Gerardi, M. H. (2006). *Wastewater Bacteria* (p. 208). New Jersey: John Wiley & Sons.
- Gosset, S. & Dentel, J. (1988). Mechanisms of coagulation with aluminum salts. J. Am. Water Work Assoc. 80 187-198.
- Guida, M., Mattei, M., Rocca, C. Della, Melluso, G., & Meriç, S. (2007). Optimization of alum-coagulation/flocculation for COD and TSS removal from five municipal wastewater. *Desalination*. 211(1-3): 113–127.
- Gunay, A., Arslankaya, E. & Tosun, I. (2007). Lead removal from aqueous solution by natural and pretreated clinoptilolite: adsorption equilibrium and kinetics. *J. Hazard. Mater.* 146: 362-371.
- Guo, W., Vigneswaran, S., Ngo, H. & Xing, W. (2007). Experimental investigation on acclimated wastewater for membrane bioreactors. *Desalination*. 207: 383-391.
- Haigler, C. (1985). The functions and biogenesis of native cellulose, in cellulose chemistry and its application. West Sussex, England: Ellis Horwood, Chichester.
- Hajiabadi, H. (2009). Effect Of Sludge Retention Time On Treating High Load Synthetic Wastewater Using Aerobic Sequencing Batch Reactors. *Iranian Journal of Environ. Health Sci. Eng.* 6(4): 217–222.
- Hall, K., Eagleton, L., Acrivos, A. & Vermeulen, T. (1966). Pore and solid diffusion kinetics in fixed bed adsorption under constant pattern conditions. *Ind. Eng. Chem. Fundam.* 5: 212-217.
- He, S., Xue, G., Kong, H., & Li, X. (2007). Improving the performance of sequencing batch reactor (SBR) by the addition of zeolites powder. *Journal of Hazardous Materials*. 142(1-2): 493–9.
- Higuchi, T. (1980). Lignin biodegradation: microbiology, chemistry, and potential applications. (T. K. Kirk, T. Higuchi, & H. Chang, Eds.) (Vol. 1, p. 256). Florida: CRC Press.
- Huda, A. H. N., Tajuddin, R. M., Ahmad, Z., & Nazrin, O. M. (2012). Determination of optimum debark time in retting process of kenaf fibre production. *International Sustainabilty and Civil Engineering Journal*. 1(2): 14–18.

- Igbinosa, O., Igbinosa, E. & Aiyegoro, O. (2009). Antimicrobial activity and phytochemical screening of stem bark extracts from Jatropha curcas (Linn). *African Journal of Pharmacy and Pharmacology*. 3(2): 58-62.
- Inglezakis, V. & Poulopoulos, S. (2006). Adsorption, ion exchange and catalysis: Design of Operations and Environmental Applications. Elsevier Science & Technology.
- Irvine, R. & Ketchum Jr., L. (1989). Sequencing batch reactors for biological wastewater treatment. *CRC Critical Reviews in Environmental Control.* 18(4): 225-294.
- Jahn, S. (1988). Using Moringa seeds as coagulants in developing countries. J. Am. Wat. Wks. Assoc.80: 43-50.
- Jahn, S. (1991). The traditional domestication of a multipurpose tree Moringa Stenopetala in Ethiopian Rift Valley. *Ambio.* 20: 244-247.
- Jongschaap, R. et al. *The water footprint of bio energy from Jatropha curcas*. Proceedings of the National Academy of Sciences of the United States of America, 1 September, 106(35), 2009.
- Junkins, R., Deeny, K. & Eckhoff, T. (1983). *The activated sludge process: Fundamentals of Operation.* Pennsylvania: Ann ArborScience Publisher.
- Kaldor, A., Karlgren, C. & Verwest, H. (1990). Kenaf a fast growing fiber source for papermaking. *TAPPI*. 73: 205-208.
- Kapoor, A. & Yang, R. (1989). Correlation of equilibrium adsorption data of condensable vapours on porous adsorbents. *Gas Sep. Purif.* 3: 187-192.
- Kulikowska, D., Klimiuk, E., & Drzewicki, A. (2007). BOD₅ and COD removal and sludge production in SBR working with or without anoxic phase. *Bioresource Technology*. 98(7): 1426–32.
- Lee, B., Choo, K., Chang, D., & Choi, S. (2009). Optimizing the coagulant dose to control membrane fouling in combined coagulation/ultrafiltration systems for textile wastewater reclamation. *Chemical Engineering Journal*. 155(1-2): 101– 107.
- Lee, J.-W., Choi, S.-P., Thiruvenkatachari, R., Shim, W.-G., & Moon, H. (2006). Submerged microfiltration membrane coupled with alum coagulation/powdered activated carbon adsorption for complete decolorization of reactive dyes. *Water Research*. 40(3): 435–44.
- Lembaga Kenaf dan Tembakau Negara. (2009). Pengeluaran Kenaf 2009. Retrieved April 16, 2013, from http://www.lktn.gov.my/page.php?102
- Lewis, D., Kollig, H. & Hodson, R. (1986). Nutrient limitation and adaptation of microbial populations to chemical transformations. *Appl. Environ. Microbiol.* 51: 598-603.

- Li, A., Li, X., & Yu, H. (2011). Effect of the food-to-microorganism (F/M) ratio on the formation and size of aerobic sludge granules. *Process Biochemistry*. 46(12): 2269–2276.
- Li, Y., Liu, Y., & Xu, H. (2008). Is sludge retention time a decisive factor for aerobic granulation in SBR? *Bioresource Technology*. 99(16): 7672–7.
- Lim, P.-E., Lim, S.-P., Seng, C.-E., & Noor, A. M. (2010). Treatment of landfill leachate in sequencing batch reactor supplemented with activated rice husk as adsorbent. *Chemical Engineering Journal*. 159(1-3): 123–128.
- Limousin, G. et al., 2007. Sorption Isotherms: a review on physical bases, modeling and measurement. *Appl. Geochem.* 249-275.
- Limousin, G., Gaudet, J.-P., Charlet, L., Szenknect, S., Barthès, V., & Krimissa, M. (2007). Sorption isotherms: A review on physical bases, modeling and measurement. *Applied Geochemistry*. 22(2): 249–275.
- Lin, S. H., & Cheng, K. W. (2001). A new sequencing batch reactor for treatment of municipal sewage wastewater for agricultural reuse. *Desalination*. 133(1): 41– 51.
- Lin, S. H., & Jiang, C. D. (2003). Fenton oxidation and sequencing batch reactor (SBR) treatments of high-strength semiconductor wastewater. *Desalination*. 154(2): 107–116.
- LTKN, 2014. Lembaga Kenaf dan Tembakau Negara. [Online]
- Ma, J., & Liu, W. (2002). Effectiveness and mechanism of potassium ferrate (VI) preoxidation for algae removal by coagulation. *Water Research*. 36(4): 871–878.
- Makkar, H., Becker, K., Sporer, F. & Wink, M. (1997). Studies on nutritive potential and toxic constituents of different provenances of Jatropha curcas. J. Agric. Food. Chem. 45: 3152-3157.
- Mane, V., Mall, I. & Srivastava, V. (2007). Kinetic and equilibrium isotherm studies for the adsorptive removal of Brilliant Green dye from aqueous solution by rice husk ash. J. Environ. Manage. 84: 390-400.
- Marañón, E. (2010). Colour, turbidity and COD removal from old landfill leachate by coagulation-flocculation treatment. *Waste Management & Research.* 28(8): 731–737.
- McCabe, W., Smith, J. & Harriot, P. (1993). *Unit operation of chemical engineering*. 5th ed. New York: McGraw-Hill.
- Metcalf & Eddy, Inc. (1991). *Wastewater Engineering: treatment, disposal & reuse*. 3 ed. Singapore: McGraw-Hill Inc.
- Mignoni, G. (1994). Il kenaf, non solo una nuova materia prima cellulosica. ISMEA,

Roma.

- Mohammed, M. (1989). *Characterization and treatment of a petrochemical wastewater for reuse*. King Fadh University of Petroleum and Minerals, Saudi Arabia.
- Mondal, D. K. & Kaviraj, A. (2008). Ecotoxicological effects of jute retting on the survival two freshwater fish and two invertebrates. *Ecotoxicology*. 17: 207-211.
- Mor, S., Ravindra, K., & Bishnoi, N. R. (2007). Adsorption of chromium from aqueous solution by activated alumina and activated charcoal. *Bioresource Technology*. 98(4): 954–7.
- Moreno, G. & Buitron, G. (2004). Influence of the origin of the inoculum and the acclimation strategy on the degradation of 4-chlorophenol. *Bioresource technology*. 94: 215-218.
- Morrison III, W. H., Akin, D. E., Archibald, D. D., Dodd, R. B., & Raymer, P. L. (1999). Chemical and instrumental characterization of maturing kenaf core and bast. *Industrial Crops and Products*. 10(1): 21–34.
- Morrison, W. H., Akin, D. E., Ramaswamy, G., & Baldwin, B. (1996). Evaluating Chemically Retted Kenaf Using Chemical, Histochemical, and Microspectrophotometric Analyses. *Textile Research Journal*. 66(10): 651–656.
- Moussavi, G., & Khosravi, R. (2010). Removal of cyanide from wastewater by adsorption onto pistachio hull wastes: parametric experiments, kinetics and equilibrium analysis. *Journal of Hazardous Materials.* 183(1-3): 724–30.
- Muyibi, S. & Evison, L. (1996). Coagulation of turbid water and softening of hard water with Moringa Oleifera seeds. *Int. J. Environ. Stud.* 49: 247-259.
- Muyibi, S. & Okuofu, C. (1995). Coagulation of low turbidity surface water with Moringa oleifera seeds. *Int. J. Environ. Stud.* 48: 263-273.
- Ndabigengesere, A. & Narasiah, K. (1998). Quality of water treated by coagulation using Moringa Oleifera seeds. *Water Res.* 781-791.
- Ndabigengesere, A., Narasiah, K. S. & Talbot, B. G., 1995. Active agents and mechanism of coagulation of turbid waters using Moringa oleifera. *Wat. Res.* 29(2): 703-710.
- Ng, H. (2002). *Performance of a membrane reactor and completely mixed activated sludge system at short solid retention times*, Berkeley: University of California, California, USA.
- Ng, J., Cheung, W. & McKay, G. (2002). Equilibrium studies of the sorption of Cu(II) ions onto chitosan. J. Colloid Interface Sci.255: 64-74.
- Nicholas, P. & Paul, N. (1993). *Carbon adsorption for Pollution Control*. Englewood Cliffs, New Jersey: PTR Prentice-Hall Inc.

- Nieschlag, H., Nelson, G., Wolff, I. & R.E. Perdue, J. (1960). A search for new fiber crops. *TAPPI*. 43: 193-201.
- Nishimura, N., Izumi, A., & Kuroda, K. (2002). Structural characterization of kenaf lignin: differences among kenaf varieties. *Industrial Crops and Products*. 15(2): 115–122.
- Norulaini, N., & Zuhair, A. (2001). Chemical Coagulation Of Settleable Solid-Free Palm Oil Mill Effluent (POME) For Organic Load Reduction. *Journal of Industrial*. 10(1): 55–72.
- Ogawa, T., Idaka, E., & Yatome, C. (1981). Acclimation of activated sludge to dye. *Bulletin of Environmental Contamination and Toxicology*. 26(1), pp 31–37.
- Özbelge, T. a., Özbelge, H. Ö., & Altınten, P. (2007). Effect of acclimatization of microorganisms to heavy metals on the performance of activated sludge process. *Journal of Hazardous Materials*. 142(1-2): 332–339.
- Paliwal, R., Sharma, V. & Pracheta (2011). A Review on Horse Radish Tree (Moringa Oleifera): a multipurpose tree with high Economic and Commercial Importance. *Asian Journal of Biotechnology*. 3(4): 317-328.
- Perez-Marin, A. et al., 2007. Removal of cadmium from aqueous solutions by adsorption onto orange waste. J. Hazard. Mater. B139: 122-131.
- Popinigis, F. (1985). Fisiologia da semente. Brasilia: Agiplan.
- Prasad, R. & Srivastava, S., 2009. Sorption of distillery spent washonto fly ash: kinetic and mass transfer studies. *Chem. Eng. J.* 146(1): 90-97.
- Ramaswamy, G. N., Ruff, C. G., & Boyd, C. R. (1994). Effect of Bacterial and Chemical Retting on Kenaf Fiber Quality. *Textile Research Journal*. 64(5): 305–308.
- Reza, M. & Seyedeh, M. B. (2011). Removal of orange 7 dye from wastewater used by natural absorbent of Moringa oleifera seeds. *American Journal of Environmental Engineering*. 1(1): 1-9.
- Richmond, M. (1968). Enzymic adaptation in bacteria: its biochemical and genetic basis. *Essays Biochem*. 4: 105-154.
- Rittmann, B. (1987). A critical evaluation of soluble microbial formation in biological processes. *Water Sci. Technol.* 19: 517-528.
- Samudro, G. & Mangkoedihardjo, S. (2010). Review on BOD, COD and BOD/COD ratio: a triangle zone for toxic, biodegradable and stable level. *International Journal of Academic Research*. 2(4): 235-239.
- Scmidt, E., Hellwig, M. & Knackmuss, H. (1983). Degradation of chlorophenols by a defined mixed microbial community. *Appl. Environ. Microbiol.* 46: 1038-1044.

- Sinsabaugh, R., Hoehn, R., Knocke, W. & Linlins, A. (1986). Removal of dissolved organic carbon by coagulation with iron sulfate. J. Am. Wat. Wks. Assoc. 78(5): 74-82.
- Sips, R. (1948). Combined form of Langmuir and Freundlich equations. J. Chem. Phys. 16: 490-498.
- Sirianuntapiboon, S., & Sansak, J. (2008). Treatability studies with granular activated carbon (GAC) and sequencing batch reactor (SBR) system for textile wastewater containing direct dyes. *Journal of Hazardous Materials*. 159(2-3): 404–11.
- Skoulikides, T. (1989). Physical chemistry I 1.2. Athens, Greece: Symetria Editions.
- Šostar-Turk, S., PetriniÄ[‡], I., & SimoniÄ[•], M. (2005). Laundry wastewater treatment using coagulation and membrane filtration. *Resources, Conservation* and Recycling. 44(2): 185–196.
- Spain, J., Pritchard, P. H. & Bourquin, A. (1980). Effects of adaptation on biodegradation rates in sediment/water cores from estuarine and freshwater environments. *Appl. Environ. Microbiol.* 40: 726-734.
- Stephenson, J., R. & Duff, S. J. (1996). Coagulation and precipitation of a mechanical pulping effluent-I: Removal of carbon, colour, and turbidity. *Wat. Res.* 30(4): 781-792.
- Stephenson, T., Lester, J. & Perry, R., 1984. Acclimatisation to nitrilotriacetic acid in the activated sludge process. *Chemosphere*. 13: 1033-1040.
- Suzuki, M. (1990). Adsorption Engineering. Tokyo, Japan: Elsevier.
- Tebbutt, T. (1997). *Principles of water quality control.* 5 ed. s.l.:Butterworth-Heinemann.
- Torstensson, N., Stark, J. & Goransson, B. (1975). The effect of repeated applications of 2,4-D and MCPA on their breakdown in soil. *Weed Res*.15: 159-154.
- Toth, J. (1971). State equations of the solid gas interface layer. Acta. Chem. Acad. Hung. 69: 311-317.
- Valverde, K. C., Moraes, L. C. K., Bongiovani, M. C., Camacho, F. P., & Bergamasco, R. (2013). Coagulation diagram using the Moringa oleifera Lam and the aluminium sulphate, aiming the removal of color and turbidity of water. *Acta Scientiarum. Technology*. 35(3): 485–489.
- Vashon, R., Jones, W. & Payne, A. (1982). The effect of water hardness on nitrilotriaceate removal and microbial acclimation in activated sludge. *Water Res.*16: 1429-1432.

Ventullo, R. & Larson, R. (1986). Adaptation of aquatic communities to quaternary

ammonium compounds. Appl. Environ. Microbiol. 51: 356-361.

- Verma, S., Prasad, B. & Mishra, I. M. (2010). Pretreatment of petrochemical wastewater by coagulation and flocculation and sludge characteristic. *Journal of Hazardous Material.* 178: 1055-1064.
- Vijayaraghavan, K., Padmesh, T., Palanivelu, K. & Velan, M. (2006). Biosorption of Nickel (II) into Sargassum Wightii: application of two-parameter and three parameter isotherm models. J. Hazard. Mater. B133: 304-308.
- Waber, W. & Morris, J. (1963). Kinetic of adsorption on carbon solution. J. Sanit. Eng. Div. Am. Soc. Civ. Eng. 89: 31-59.
- Walker, R. & Newman, A. (1956). Microbial decomposition of 2,4dichlorophenoxyacetic acid. *Appl. Microbiol.* 4: 201-206.
- Wang, Z. et al. (2002). Landfill leachate treatment by a coagulation-photooxidation process. J. Hazardous Mater. 95: 153-159.
- Wanner, J. (1992). Comparison of biocenoses from continuous and sequencing batch reactors. *Water Science and Technology*. 25(6): 239-249.
- Warren, L., Julian, C. & Harriot, P. (1982). *Unit Operation of Chemical Engineering*. s.l.:McGraw-Hill International Edition.
- Watson, H. (1993). A comparison of the effects of two methods of acclimation on aerobic biodegradability. *Environmental toxicology and chemistry*. 12: 2023-2030.
- Webber III, C., & Bledsoe, V. (2002). Kenaf yield components and plant composition. *Trends in New Crops and New Uses*, pp. 348–357.
- Weber, W. (1972). *Physicochemical Processes for Water Quality Control*. New York, USA: Wiley Interscience.
- White, G. et al. (1970). *Cultural and harvesting methods for Kenaf*. Washington D.C.: USDA Prod. Res. Rpt.
- Wiggins, B. A., Jones, S. H. & Alexander, M. (1987). Explanations for the Acclimation Period Preceding the Mineralization of Organic Chemicals in the Aquatic Environment. *Applied and Environmental Microbiology*. 791-796.
- Wilderer, P., Irvine, R. & Goronszy, M. (2001). Sequencing Batch Reactor *Technology*. London: IWA Publishing.
- Yonge, D. & Keinath, T. (1986). The effects of non-ideal competition on multicomponent adsorption equilibria. J. Water Pollut. Control Fed. 58: 77-81.
- Yonge, D., Keinath, T., Poznanska, K. & Jiang, Z. (1985). Single-solute irreversible adsorption on granular activated carbon. *Environ. Sci. Technol.* 19: 690-694.
- Yu, H., & Yu, C. (2007). Study on microbe retting of kenaf fiber. Enzyme and

Microbial Technology. 40(7): 1806–1809.

- Yu, H., & Yu, C. (2010). Influence of various retting methods on properties of kenaf fiber. *Journal of the Textile Institute*. 101(5): 452–456.
- Zhang, J. et al., 2005. The active component in the flax-retting system of the zygomycete rhizopus oryzae sb is a family 28 polygalacturonase. J. Ind. Microbial. Biotechnol. 32: 431-438.
- Zhou, Y., Liang, Z., & Wang, Y. (2008). Decolorization and COD removal of secondary yeast wastewater effluents by coagulation using aluminum sulfate. *Desalination*. 225(1-3): 301–311.
- Zhou, Y., Xing, X.-H., Liu, Z., Cui, L., Yu, A., Feng, Q., & Yang, H. (2008). Enhanced coagulation of ferric chloride aided by tannic acid for phosphorus removal from wastewater. *Chemosphere*. 72(2): 290–8.

