

UNIVERSITI PUTRA MALAYSIA

DEVELOPMENT OF GROUND DUNE SAND BLENDED CEMENT

OMER ABDALLA ALAWAD HASSAN

FK 2014 158



DEVELOPMENT OF GROUND DUNE SAND BLENDED CEMENT



By

OMER ABDALLA ALAWAD HASSAN

Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirements for the Degree of Doctor of Philosophy

November 2014

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Compyright© Universiti Putral Malaysia



DEDICATION

To: My Mother, Your love is always with me no matter where I go. My Father, You enlighten me to do all right things and only the right, reminding me that there can be no gain without pain. My Brother, Your unconditional support is treasured for always. My wife, For all that you have been, for all that you are and will always be, I am grateful. Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Doctor of Philosophy

DEVELOPMENT OF GROUND DUNE SAND BLENDED CEMENT

By

OMER ABDALLA ALAWAD HASSAN

November 2014

Chairman: Professor Ir. Mohd Saleh Jaafar, PhD

Faculty: Engineering

Pozzolan materials (e.g. fly ash, slag, silica fume, rice husk ash) have been used successfully as a partial ordinary Portland cement (OPC) replacement material. However, there are some technical and economic drawbacks associated with the use of the existing pozzolan materials. Therefore, there is a growing interest to find an alternative material to be used as a source of siliceous materials for concrete production. This research aims to determine the potential of using ground dune sand (GDS) as partial cement replacement in binary (OPC-GDS) and ternary combinations of OPC-GDSslag and OPC-GDS-lime. The proposed combinations of blended cement system are expected to save large amounts of OPC.

The primary objective of this study is to develop naturally available dune sand as an effective partial cementing material for use in the concrete industry. To achieve this objective, different treatment methods, namely, mechanical, chemical and thermal methods (autoclave curing) have been applied to determine the reactivity of GDS. For the ternary blended combinations, low (15%), moderate (30%) and high (45%) amounts of slag or lime were incorporated into a binder system containing 40% of GDS. Compressive strength, drying shrinkage and durability properties of the cast mixtures were investigated. Moreover, microstructure analyses were carried out using SEM, EDX, XRD, DTA and TGA analyses to characterize the hydrated products. A correlation between CaO/SiO₂ and compressive strength was then carried out. The results revealed that autoclave curing is a promising curing system to utilize the GDS as partial cement replacement. The optimum level of replacement of OPC by GDS was found to be 30%, and up to 40% of GDS can be used without significant loss in the compressive strength. The inclusion of slag or lime as the ternary binder element to the mixture containing 40% of GDS yielded a compressive strength higher or comparable to the control mixture. The drying shrinkage and durability properties of blended autoclave cured mixes were significantly improved. The SEM, EDX, XRD, DTA and TGA analyses explained how GDS contributes to the strength and durability of blended mixtures. The outcome of this research will benefit the Middle East and other countries where supplies of natural dune sands are unlimited.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

PEMBANGUNAN PASIR BUKIT TERKISAR CAMPURAN SIMEN

Oleh

OMER ABDALLA ALAWAD HASSAN

November 2014

Pengerusi: Profesor Ir. Mohd Saleh Jaafar, PhD

Fakulti: Kejuruteraan

Bahan pozzolan (seperti abu terbang, jermang, wasap silika, abu sekam padi) telah digunakan dengan berkesan sebagai sebahagian bahan penggantian simen Portland (OPC). Walau bagaimanapun, terdapat beberapa kekangan teknikal dan ekonomi berkaitan dengan penggunaan bahan pozzolan yang sedia ada. Oleh itu, terdapat usaha untuk mencari pendekatan alternatif bahan yang akan digunakan sebagai sumber bahan bersilica dalam pembuatan konkrit. Penyelidikan ini bertujuan untuk menentukan potensi penggunaan pasir bukit terkisar (*ground dune sand* GDS) sebagai sebahagian pengganti simen dalam penduaan (OPC-GDS) dan pentiga kombinasi OPC-GDS- jermang dan OPC-GDS-kapur. Kombinasi sistem adunan simen yang dicadangkan dijangkakan dapat menjimat penggunaan OPC yang banyak.

Objektif utama dalam penyelidikan ini adalah untuk membangunkan pasir bukit sedia ada sebagai bahan gantian simen yang efektif untuk digunakan dalam industri konkrit. Untuk mencapai objektif ini, pelbagai kaedah rawatan antaranya; kaedah mekanikal, kimia dan terma (pengawetan autoklaf) digunapakai untuk menentukan kereaktifan GDS. Untuk kombinasi pentiga campuran, kuantiti rendah (15%), sederhana (30%) dan tinggi (45%) jermang atau kapur telah digabungkan dalam sistem pengikat yang mengandungi 40% GDS. Kekuatan mampatan, ciri pengecutan keringan, dan ciri ketahanlasakan campuran telah dikaji. Tambahan lagi, analisis mikrostruktur telah dijalankan menggunakan analisis SEM, EDX, XRD, DTA and TGA untuk mencirikan produk terhidrat. Korelasi antara CaO/SiO₂ dan kekuatan mampatan telah dijalankan. Keputusan menunjukkan bahawa pengawetan autoklaf adalah sistem pengawetan yang menjanjikan penggunaan GDS sebagai bahan pengganti simen separa. Tahap optimum penggantian OPC oleh GDS adalah 30%, di mana, sehingga 40% GDS boleh digunakan tanpa kehilangan yang signifikan pada kekuatan mampatan.

Memasukan jermang atau kapur sebagai elemen pengikat ketiga dalam campuran mengandungi 40% GDS menghasilkan kekuatan mampatan lebih tinggi atau setanding dengan campuran kawalan. Pengecutan keringan dan ciri ketahanan adunan campuran terawet secara autoklaf telah meningkat dengan signifikan. Analisis SEM, EDX, XRD, DTA and TGA menerangkan bagaimana GDS menyumbang pada kekuatan dan ketahanan campuran. Hasil daripada penyelidikan ini akan menguntungkan Negara Timur Tengah dan negara lain di mana bekalan pasir bukit adalah hampir tidak terhad.



ACKNOWLEDGEMENTS

In the name of ALLAH, the Beneficent, the Compassionate, and who always gives me strength and patience to complete my duties regardless of the many challenges.

I would like to express my deep thanks to the supervisory committee led by Prof. Dr Saleh, Prof. Dr Jamal, Prof. Dr Alhozaimy, and Dr Farah. Your time, support and advice have been precious to me. I am especially grateful to Prof. Dr Saleh for his fruitful discussions, helpful criticisms, patience and invaluable suggestions to this research. I am deeply indebted to him for guiding me with his extensive knowledge and logical way of thinking. This thesis would never have been possible without him. I am also indebted to Prof. Dr Alhozaimy who was immensely helpful in facilitating the support that I needed for validating the proposed work. Sincere appreciation also goes to the King Saud University for providing financial supports for completion of this project.

Last but by no means least, it has been a privilege for me to study at the Civil Engineering Department, Faculty of Engineering at Universiti Putra Malaysia, where an excellent environment to perform this research was provided. I am also thankful to my fellows at the Faculty for making my stay interesting and enjoyable.

> Omer Alawad, November 2014

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Mohd Saleh Jaafar, PhD

Professor, Faculty of Engineering, Universiti Putra Malaysia. (Chairman)

Jamaloddin Noorzaei, PhD

Professor, Faculty of Engineering, Universiti Putra Malaysia. (Member, 2009-2011)

Farah Nora Aznieta bte Abd. Aziz,PhD

Senior Lecturer, Faculty of Engineering, Universiti Putra Malaysia. (Member)

Abdulrahman Alhozaimy, PhD

Professor, Faculty of Engineering, King Saud University. (Member)

BUJANG BIN KIM HUAT, PhD Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date:

TABLE OF CONTENTS

		Page
ABSTRA	ACT	i
ABSTRA	AK	iii
ACKNO	WLEDGEMENTS	v
APPROV	VAL	vi
DECLAI	RATION	viii
LIST OF	TABLES	xiii
LIST OF	FIGURES	xiv
LIST OF	ABBREVIATIONS	xvii
CHAPTI	ER	
1	INTRODUCTION	1
	1.1 General	1
	1.2 Use of Supplementary Cementitious Materials in Concrete	2
	1.3 Problem Statement	3
	1.4 Objectives of the Study	3
	1.5 Scope of the Study	4
	1.6 Significance of the Study	5
	1.7 Overview of the Thesis	5
2	LITERATURE REVIEW	6
	2.1 Introduction	6
	2.2 Portland cement	6
	2.2.1 Historical Background of Portland Cement	6
	2.2.2 Manufacturing of Portland Cement	6
	2.2.3 Compounds of Portland Cement	8
	2.2.4 Hydration of Portland Cement	8
	2.2.5 Types of Portland Cement	10
	2.3 Supplementary Cementitious Materials	10
	2.3.1 Pozzolanic Materials	11
	2.3.2 Filler additive materials	10
	2.5.5 Pozzolanic Reaction	19
	2.4 Activation Methods	20
	2.4.1 Mechanical Activation Method	20
	2.4.2 Chemical Activation Method	20
	2.4.5 Therman Activation Method	21
	2.6 Physical and Durability Properties of Concrete	25
	2.6.1 Chloride Ion Permeability	25
	2.6.1 Chorace for reinleading	25 26
	2.6.2 Drying Sinnikage 2.6.3 Resistance to Sulfate Attack	20 28
	2 7 Dune Sand	20 30
	2.8 Critical Literature Review	30
	2.8 1 Supplementary Computitious Materials	30
	2.8.2 Activation Methods	31
		~-

2.8.3 Ternary Blended Mixtures	31
2.8.4 Microstructure Analysis	32
3 MATERIALS AND METHODS	33
3.1 Materials	33
3.1.1 Ordinary Portland cement	33
3.1.2 Ground Dune Sand	34
3.1.3 Other supplementary Cementitious Materials	37
3.1.4 Aggregates	39
3.1.5 Water	40
3.1.6 High-Range Water Reduction Agent (Superplasticiser)	40
3.1.7 Chemical Activators	40
3.2 Methods	40
3.2.1 Mix Design	43
3.2.2 Mixing and Casting Procedure	47
3.2.3 Curing	47
3.2.4 Testing the specimens	49
3.3 Microstructure Analysis	54
3.3.1 Morphology Analyses	54
3.3.2 Mineralogical Analysis	58
3.3.3 Thermo-gravimetric Analysis	61
4 RESULTS AND DISCUSSION	65
4.1 Introduction	65
4.2 Result <mark>s of Part One: OPC-GDS Bina</mark> ry Blended Mixtures	65
4.2.1 Effects of Activation Methods on the Reactivity of GDS	65
4.2.2 Effects of Different Autoclave Conditions	67
4.2.3 Consistency, Setting Time and Workability of the	68
4.2.4 Compressive Strength of the OPC-GDS Binary	70
Blended Mixtures	-
4.2.5 Rapid Chloride Ion Permeability of the OPC-GDS	71
A 26 Druing Shrinkage of the OPC CDS Binery Blanded	70
4.2.0 Drying Shinkage of the Of C-GDS binary biended Mixtures	12
4.2.7 Resistance to Sulfate Attack of the GDS Binary Blended Mixtures	74
4 2 8 Microstructure Analyses of the GDS Binary Blended	75
Mixtures	
4.2.9 Mechanism of Reactivity of GDS Under Autoclave	82
Curing	-
4.3 Preliminary Tests of Ternary Ingredient Materials	85
4.4 Results of Part Two: OPC-GDS-Slag- Ternary Blended	88
Mixtures	
4.4.1 Overview	88
4.4.2 Normal Consistency, Setting time and Workability of	88
the OPC-GDS-Slag ternary Blended Mixtures	

Biended Mixtures 90 OPC-GDS-Slag Ternary Blended Mixtures 4.4.5 Drying shrinkage of the OPC-GDS-Slag Ternary 91 Blend Mixtures 4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.5.1 Overview 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 0PC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.7 Incrostructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 <tr< th=""><th>4.4.3 Compressive Strength of the OPC-GDS-Slag Ternary</th><th>89</th></tr<>	4.4.3 Compressive Strength of the OPC-GDS-Slag Ternary	89
4.4.4 Rapid Chloride Ion Permeability of the 90 OPC-GDS-Slag Ternary Blended Mixtures 91 Blend Mixtures 4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag 91 Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.5.7 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 90 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 100 Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carboo Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carboo Dioxide Emissions of Control and 113 Blended Mixtures 111 5.1 Conclusions 118 5.1.	Blended Mixtures	
OPC-GDS-Slag Ternary Blended Mixtures 4.4.5 Drying shrinkage of the OPC-GDS-Slag Ternary 91 Blend Mixtures 4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag 91 Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.5.1 Overview 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 5.1 Conclusions 118	4.4.4 Rapid Chloride Ion Permeability of the	90
4.4.5 Drying shrinkage of the OPC-GDS-Slag Ternary 91 Blend Mixtures 4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 Mixtures 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 91 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 118 5.1 Conclusions 118 5.1.2 Che optimum level of replacement of OPC by GDS and 118 5.1.2 The optimum level of replacement of OPC by GDS and 119	OPC-GDS-Slag Ternary Blended Mixtures	
4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag 91 Blended Mixtures 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Termary Blended Mixtures 4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 Mixtures 99 4.5.2 Normal Consistency, Setting time and Workability 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Termary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 118 5.1.1 Activation Method 118 5.1.2 Che optimum level of replacement of OPC by GDS and 118 5.1.3 Physical and Durability Properties of Binary	4.4.5 Drying shrinkage of the OPC-GDS-Slag Ternary Blend Mixtures	91
Biended Mixtures 12 4.4.7 Microstructure Analyses of the OPC-GDS-Slag 92 Ternary Blended Mixtures 99 4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 100 Ternary Blended Mixtures 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 118 5.1.1 Activation Method 118	4.4.6 Resistance to Sulfate Attack of the OPC-GDS-Slag	91
4.4.7 Microstructure Analyses of the OPC-GDS-Slag Ternary Blended Mixtures 92 4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 4.5.2 Normal Consistency, Setting time and Workability 99 4.5.2 Normal Consistency, Setting time and Workability 99 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and 0ptimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary 121 5.1.4 Microstructure Analysis of Binary and Ternary 122	Blended Mixtures	
Ternary Blended Mixtures 99 Mixtures 99 4.5.1 Overview 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 100 Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Activation Method 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary	4.4.7 Microstructure Analyses of the OPC-GDS-Slag	92
4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended 99 Mixtures 99 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 100 Ternary Blended Mixtures 101 OPC-GDS-Lime Ternary Blended Mixtures 101 OPC-GDS-Lime Ternary Blended Mixtures 103 Blended Mixtures 103 Blended Mixtures 104 Blended Mixtures 104 Blended Mixtures 104 Blended Mixtures 111 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 118 5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and 119 5.	Ternary Blended Mixtures	
Mixtures 4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 4.5.2 Normal Consistency, Setting time and Workability 4.5.3 Compressive Strength of the OPC-GDS-Lime 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 0PC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.5 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Activation Method 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 5.3 Future Studies 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 125 126 127 128 129 129 120 129 120 120 123 124 125 120 125 120 125 120 125 120 125 120 125 120 125 125 120 125 125 120 125 125 120 125 125 120 125 120 125 125 120 125 125 120 125 125 120 125 125 125 125 125 125 125 125	4.5 Result of Part Three: OPC-GDS-Lime Ternary Blended	99
4.5.1 Overview 99 4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 011 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 5.1 Conclusions 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	Mixtures	
4.5.2 Normal Consistency, Setting time and Workability 99 of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.8 Concrete mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 UST OF PUBLICATIONS 143	4.5.1 Overview	99
of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 5.2 Contributions of Study 5.2 Contributions of Study 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 143	4.5.2 Normal Consistency, Setting time and Workability	99
4.5.3 Compressive Strength of the OPC-GDS-Lime 100 Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 101 OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 121 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES	of the OPC-GDS-Lime Ternary Blended Mixtures	
Ternary Blended Mixtures 4.5.4 Rapid Chloride Ion Permeability of the 0PC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Conclusions 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS <	4.5.3 Compressive Strength of the OPC-GDS-Lime	100
4.5.4 Rapid Chloride Ion Permeability of the101OPC-GDS-Lime Ternary Blended Mixtures4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary103Blended Mixtures1044.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime104Blended Mixtures1014.5.7 Microstructure Analyses of the OPC-GDS-Lime104Blended Mixtures1114.7 Energy and Carbon Dioxide Emissions of Control and113Blended Mixtures1155CONCLUSION AND RECOMMENDATIONS FOR1185.1.1 Activation Method1185.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures1115.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures1215.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures1235.2 Contributions of Study1235.3 Future Studies124REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	Ternary Blended Mixtures	200
OPC-GDS-Lime Ternary Blended Mixtures 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary 103 Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS	4.5.4 Rapid Chloride Ion Permeability of the	101
 4.5.5 Drying Shrinkage of the OPC-GDS-Lime Ternary Blended Mixtures 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime Blended Mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.6 Concrete mixtures 4.8 Summary 5 CONCLUSION AND RECOMMENDATIONS FOR 118 5.1 Conclusions 5.1 Conclusions 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 5.3 Future Studies 5.3 Future Studies 	OPC-GDS-Lime Ternary Blended Mixtures	101
Blended Mixtures 105 4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 121 13.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 121 13.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 121 13.1.4 Microstructure Analysis of Binary and Ternary 123 5.3 Future Studies 123 13 124 <th>4.5.5 Drving Shrinkage of the OPC-GDS-Lime Ternary</th> <th>103</th>	4.5.5 Drving Shrinkage of the OPC-GDS-Lime Ternary	103
4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime 104 Blended Mixtures 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 121 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	Blended Mixtures	100
Blended Mixtures 104 4.5.7 Microstructure Analyses of the OPC-GDS-Lime 104 Blended Mixtures 111 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 113 Blended Mixtures 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 121 5.1.3 Future Studies 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	4.5.6 Resistance to Sulfate Attack of the OPC-GDS-Lime	104
4.5.7 Microstructure Analyses of the OPC-GDS-Lime Blended Mixtures 104 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and Blended Mixtures 113 4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 123 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	Blended Mixtures	
Blended Mixtures 111 4.6 Concrete mixtures 111 4.7 Energy and Carbon Dioxide Emissions of Control and 113 Blended Mixtures 113 8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 118 5.1 Conclusions 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 123 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	4.5.7 Microstructure Analyses of the OPC-GDS-Lime	104
4.6 Concrete mixtures1114.7 Energy and Carbon Dioxide Emissions of Control and Blended Mixtures113 Blended Mixtures4.8 Summary1155CONCLUSION AND RECOMMENDATIONS FOR FUTURE STUDIES118 5.1.1 Conclusions5.1 Conclusions118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures5.2 Contributions of Study123 5.3 Future Studies124REFERENCES125 BIODATA OF STUDENT LIST OF PUBLICATIONS143	Blended Mixtures	
4.7 Energy and Carbon Dioxide Emissions of Control and Blended Mixtures113 Blended Mixtures4.8 Summary1155CONCLUSION AND RECOMMENDATIONS FOR FUTURE STUDIES118 5.1 Conclusions5.1 Conclusions118 5.1.1 Activation Method118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures95.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures5.2 Contributions of Study123 5.3 Future Studies124REFERENCES125 BIODATA OF STUDENT LIST OF PUBLICATIONS123 143	4.6 Concrete mixtures	111
Blended Mixtures 4.8 Summary 115 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 5.1 Conclusions 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	4.7 Energy and Carbon Dioxide Emissions of Control and	113
4.8 Summary 115 5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 1.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 121 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	Blended Mixtures	
5CONCLUSION AND RECOMMENDATIONS FOR FUTURE STUDIES1185.1 Conclusions1185.1.1 Activation Method1185.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures1195.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures1215.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures1225.2 Contributions of Study1235.3 Future Studies124REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	4.8 Summary	115
5 CONCLUSION AND RECOMMENDATIONS FOR 118 FUTURE STUDIES 118 5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 122 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143		
FOTORE STUDIES 118 5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 122 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	5 CONCLUSION AND RECOMMENDATIONS FOR	118
5.1 Conclusions 118 5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	FUTURE STUDIES	110
5.1.1 Activation Method 118 5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures 122 Blended Mixtures 123 5.3 Future Studies 124 REFERENCES BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	5.1 Conclusions	118
5.1.2 The optimum level of replacement of OPC by GDS and optimum combinations of ternary blended mixtures 119 5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures 121 5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES 125 BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	5.1.1 Activation Method	118
optimum combinations of ternary blended mixtures1195.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures1215.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures122Blended Mixtures1235.2 Contributions of Study1235.3 Future Studies124REFERENCESBIODATA OF STUDENT142LIST OF PUBLICATIONS143	5.1.2 The optimum level of replacement of OPC by GDS ar	id 110
5.1.3 Physical and Durability Properties of Binary and Ternary Blended Mixtures1215.1.4 Microstructure Analysis of Binary and Ternary Blended Mixtures122Blended Mixtures1235.2 Contributions of Study1235.3 Future Studies124REFERENCESBIODATA OF STUDENT142LIST OF PUBLICATIONS143	optimum combinations of ternary blended mixtures	119
Blended Mixtures1215.1.4 Microstructure Analysis of Binary and Ternary122Blended Mixtures1235.2 Contributions of Study1235.3 Future Studies124REFERENCESBIODATA OF STUDENT142LIST OF PUBLICATIONS143	5.1.3 Physical and Durability Properties of Binary and Terr	ary
5.1.4 Microstructure Analysis of Binary and Ternary 122 Blended Mixtures 123 5.2 Contributions of Study 123 5.3 Future Studies 124 REFERENCES BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	Blended Mixtures	121
Blended Mixtures5.2 Contributions of Study1235.3 Future Studies124REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	5.1.4 Microstructure Analysis of Binary and Ternary	122
5.2 Contributions of Study1235.3 Future Studies124REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	Blended Mixtures	
5.3 Future Studies124REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	5.2 Contributions of Study	123
REFERENCES125BIODATA OF STUDENT142LIST OF PUBLICATIONS143	5.3 Future Studies	124
BIODATA OF STUDENT 142 LIST OF PUBLICATIONS 143	REFERENCES	125
LIST OF PUBLICATIONS 143	BIODATA OF STUDENT	142
	LIST OF PUBLICATIONS	143

LIST OF TABLES

Table		Page
2.1	Mineral name and typical range of principal cement compounds	9
2.2	Types of portland cement	11
2.3	Types of blended cement	11
2.4	Typical csh phases	25
3.1	Chemical composition of opc, gds, slag, and lime	34
3.2	Physical properties of opc	35
3.3	Mix proportion of blended pastes	46
3.4	Mix proportion of opc-gds binary blended mortars	47
3.5	Mix proportion of opc -slag binary blended mortars	47
3.6	Binder proportion of opc-gds-slag and opc-gds-lime ternary blended mortars and concretes	49
3.7	Rapid chloride permeability test ratings astm c 1202	55
4.1	Normal consistency and setting time of gds binary blended mixtures	73
4.2	Free ch for ctrl and binary blended mixtures	86
4.3	Normal consistency and setting time of opc-gds-slag ternary blended mixtures	92
4.4	Free ch for slag ternary blended mixtures	102
4.5	Normal consistency and setting time of opc-gds-lime ternary blended mixtures	104
4.6	Free ch for lime ternary blended mixtures	116
4.7	Estimated CO ₂ emissions of control and blended mixtures	119

LIST OF FIGURES

Figure		Page
2.1	The flow of Portland cement manufacturing process [3]	8
2.2	Classification of Pozzolan Materials [4]	13
2.3	SEM image of diatomaceous earth particle [42]	14
2.4	SEM image of FA particle [45]	15
2.5	SEM image of slag particle [49]	16
2.6	SEM image of SF particle [52]	17
2.7	SEM image of RHA particle [61]	18
2.8	SEM image of MK particle [69]	19
2.9	Structure of silica (a) crystalline silica (b) amorphous silica [77]	20
2.10	XRD of pattern of kaolin and metakaolin [100]	23
2.11	Delayed ettringite cracking between cement paste and aggregate [3]	24
2.12	Schematic of absorbed water, capillary water and interlayer water [3]	28
2.13	SEM image of AFt [3]	30
3.1	Natural dune sand	35
3.2	Grain size distribution of natural dune sand	36
3.3	Abrasion machine	36
3.4	Figure 3.4 Particle size distribution curves of GDS, GGBS, F-GDS, and lime	37
3.5	Natural dune sand: before and after grinding process	37
3.6	SEM image of GDS	38
3.7	XRD pattern of GDS	39
3.8	SEM image of slag	40
3.9	XRD pattern of slag	40
3.10	XRD pattern of lime	41
3.11	Particle size distribution curve of fine aggregate (mining sand)	42
3.12	Flow chart of experimental program	44
3.13	Autoclave chamber	51
3.14	Autoclave curing cycle	51
3.15	Vicat apparatus used for consistency and setting time tests	52
3.16	Flow table test setup	53
3.17	Compressive strength machines	54
3.18	RCPT processor equipment connected with cells	55
3.19	Shrinkage prism and test setup	56
3.20	Different interactions of an electron beam (PE) with a solid target [154]	58
3.21	SEM of hydrated cement paste (a) early age hydrated (b) more mature paste [161]	59

C

3.22	SEM of hydrated cement in an autoclaved (a) fibrous CSH (b) crystalline platy CSH [163]	59
3 73	SEM of hydrated compart in an autoclayed (a) fibrous CSH	60
5.25	(b) crystalline platy CSH [163]	00
3.24	SEM of paste containing nano-SiO2 particles [52]	60
3.25	SEM of paste containing nano-SiO2 particles [52]	61
3.26	XRD patterns of plain and blended pastes [19]	62
3.27	XRD patterns of different paste specimens at age of 91 days	63
	[166]	
3.28	XRD-6000 machine	64
3.29	DTA/TGA of cement paste [167]	65
3.30	TGA curves of control and blended pastes at 3 and 91 days	66
0.00	[166]	00
3 31	DTA curves for cement and cement-ground quartz pastes	67
0.01	(a) normal curing (b) autoclave cured [79]	07
3 32	TA instrument (model SDT O600)	68
4 1	Effect of various treatment methods on compressive	70
1.1	strength of OPC-GDS mixtures	70
4 2	Effect of different autoclave temperatures and periods on	72
1.2	the strength of OPC-CDS mixtures	12
43	SEM images of OPC and GDS	73
4.5 4 A	Compressive strength of CTRL and OPC-CDS blended	74
1.1	mixtures cured under normal and autoclave conditions	71
4 5	Total charge (coulombs) passed of CTRL and OPC-GDS	75
1.0	blended mixtures cured under NC and AC conditions	70
46	Drying shrinkage of CTRL and OPC-GDS binary blended	77
1.0	mixtures	,,
4.7	Resistance to sulfate attack of CTRL and OPC-GDS blended	78
	mixtures	
4.8	SEM of CTRL and OPC-GDS binary blended mixtures	79
4.9	EDX of CTRL and OPC-GDS binary blended mixtures	81
4.10	XRD patterns of CTRL and OPC-GDS binary blended	83
	mixtures	
4.11	DTA and TGA curves of CTRL and OPC-GDS binary	85
	blended mixtures	
4.12	Schematic phases diagram of autoclave cured OPC-GDS	88
	mixture at 180 °C	
4.13	Compressive strength of blended mixture incorporating	89
	GDS -Slag-OPC	
4.14	Compressive strength of blended mixture incorporating	90
	GDS -Slag-OPC	
4.15	Cracking of cement-less mixture after autoclave curing	91
4.16	Compressive strength OPC-GDS-Slag ternary blended	93
	mixtures cured under NC and AC conditions	
4.17	Total charge (coulombs) passed during the chloride	94
	permeability test of CTRL and OPC-GDS-Slag mixtures	

4.18	Drying shrinkage of CTRL and OPC-GDS-Slag ternary blended mixtures	95
4.19	Resistance to sulfate attack of CTRL and OPC-GDS-Slag	96
	mixtures cured under autoclave conditions	
4.20	SEM images of OPC-GDS-Slag blended mixtures	97
4.21	EDX of CTRL and OPC-GDS-Slag mixtures	98
4.22	XRD patterns of OPC-GDS-Slag ternary blended mixtures	100
4.23	DTA and TGA curves of CTRL and OPC-GDS-Slag mixtures	101
4.24	Correlation of compressive strength and CaO/SiO2 ratio of	102
	slag blended mixtures cured under AC conditions	
4.25	Relation between lime replacement and standard	105
	consistency/water-binder ratio of standard flow of mortar	
4.26	Compressive strength of OPC-GDS-Lime ternary blended	106
	mixtures cured under NC and AC conditions	
4.27	Rapid chloride permeability result of CTRL and OPC-GDS-	107
	Lime mixtures	
4.28	Drying shrinkage of CTRL and OPC-GDS-Lime ternary	108
	blended mixtures cured under AC conditions	
4.29	Resistance to sulfate attack of CTRL and OPC-GDS-Lime	109
	mixtures cured under AC conditions	
4.30	SEM of OPC-GDS-Lime ternary blended mixtures	110
4.31	EDX of GDS-LM ternary blended mixtures	111
4.32	XRD patterns of GDS-LM ternary mixtures	113
4.33	Correlation between lime content and peak intensity of SiO2	114
4.34	DTA and TGA curves of CTRL and OPC-GDS-Lime blended mixtures	115
4.35	Correlation between C/S ratio and compressive strength of	116
	lime blended mixtures cured under AC conditions	
4.36	Compressive strength of concrete mixtures incorporating GDS, slag and OPC	117
4.37	Compressive strength of concrete mixture incorporating GDS and lime	118

LIST OF ABBREVIATIONS

SEM	Scanning Electron Microscopy
EDX	Energy DispersiveX-Ray
TGA	Themor-gravimetric Analysis
XRD	X-Ray Diffraction
OPC	Ordinary Portland cement
C3S	3CaO. SiO ₂ Tricalcium silicate
C2S	2CaO. SiO ₂ Dicalcium silicate
C3A	3CaO. Al ₂ O ₃ Tricalcium aluminate
C4AF	4CaO. Al ₂ O ₃ . Fe ₂ O ₃ Ferrite
СН	Ca(OH) ₂ Calcium hydroxide
C-S-H	CaO. Al_2O_3 . SiO ₂ . H_2O Calcium silicate aluminate hydrate
AFt	3CaO. Al ₂ O ₃ . 3CaSO ₄ . 32H ₂ O Ettringite
AFm	$3C_aO_{12}O_{3}$. CaSO ₄ . 12H ₂ O Tricalcium monosulfo aluminate
DTA	Differential Thermal Analysis
PSD	Particle Size Distribution
ASTM	American Society for Testing and Materials
FA	Fly Ash
FT-IR	Fourier Transform Infrared Spectroscopy
SCMs	Supplementary cementing materials
SF	Silica Fume
W/C	Water-to-Cement ratio
XRF	X-ray Fluorescence
МК	Metakaolin
MPa	Megapascals
С	CaO
CO ₂	Carbon dioxide
S	SiO ₂
А	Al ₂ O ₃
F	Fe ₂ O ₃
Н	H ₂ O
Μ	MgO
Ν	Na ₂ O
Κ	K ₂ O
KW	Kilo watt
S ⁻	SO ⁼ ₃
GJ	Giga Joule
Q	Quartz

CHAPTER 1

INTRODUCTION

1.1 General

Concrete is the most extensively used construction material in the world [1]. The consumption rate of concrete is estimated to be one ton for every living human being [2]. Perhaps, the worldwide consumption of concrete is only second to that of water. This could be due to the low cost per cubic meter, availability of raw materials, ease of casting, excellent resistance to water, and ability to be formed in various shapes and sizes [3]. The main ingredients of concrete are ordinary Portland cement (OPC or PC), water and fine and coarse aggregates. OPC is the most important ingredient because it reacts with water to make glue, which bonds the coarse and fine aggregates together.

OPC comprises four main components, which are tri-calcium silicate (C₃S), di-calcium silicate (C₂S), tri-calcium aluminate (C₃A) and tetra-calcium aluminoferrite (C₄FA). When OPC comes into contact with water, several chemical reactions (hydration) occur, resulting in different hydrated products. The hydration of C₃S and C₂S, which is considered to be about 75% of the total weight of OPC, produces calcium silicate hydrate, also known as CSH gel and calcium hydroxide (Ca(OH)₂ or CH). Where, the hydration of C₃A and C₄FA forms ettringite (AFt) and monsolfmonate (AFm) phases [4, 5].

The contributions of CSH gel, CH, AFt, and AFm to concrete properties vary. For instance, the AFt and CSH gel are responsible for initial setting and early strength development, respectively. These phases are mainly formed due to the hydration of C₃A and C₃S, hence their hydration starts after several minutes (i.e. 15 minutes) of mixing. The hydration of C₂S starts after several days of mixing and continues up to hundreds of days. Therefore, C₂S is responsible for the late strength development. On the other hand, CH generated from the hydration of C₃S and C₂S does not induce strengthening properties. CH may weaken the transition zone between the cement paste and aggregate and become a source of concrete deterioration (i.e. carbonation and expansive gypsum formation) [3]. The AFm phase is usually generated from AFt when the amount of C₃A is more than the supplied sulfate ions, or as in the case of elevated curing temperature. AFm has a minor contribution to strength, but could be converted to AFt causing harmful expansion of the hardened concrete properties [6, 7].

The manufacturing of OPC consumes a considerable amount of energy and resources [8]. The production of one ton of OPC requires about four GJ of energy and about two tons of raw materials (limestone, shale, etc.). In addition, the manufacturing of one ton of OPC emits approximately one ton of carbon dioxide (CO_2) into the atmosphere. From the projection made by the cement companies, the consumption rate of cement has risen from two million tons per year in 1880 to about two billion tons in 2006 [9]. Moreover, this proportion is expected to remain steady in the next decade. In particular, the manufacturing of OPC accounts for about seven per cent of the total world CO₂ emissions [10]. However, environmental concern has placed considerable pressure on cement plants to reduce the CO₂ emissions and use alternative eco- friendly materials with lower environmental impact. In fact, the Rio de Junior (1992), Kyoto (1997) and Copenhagen (2009) protocols were essentially established to reduce the total greenhouse gas emissions [11, 12].

1.2 Use of Supplementary Cementitious Materials in Concrete

One option to reduce the CO₂ emissions is to replace large amounts of OPC with supplementary cementitious materials (SCMs) - natural, industrial by product, or agricultural waste materials - which have been used successfully as partial cement replacement materials in concrete production [13-17]. Natural pozzolan materials, such as volcanic tuff, diatomaceous earth and volcanic glass have been used since the days of the ancient Romans. Whereas, the industrial by-product and agricultural waste materials, including ground granulated blast furnace slag (slag or GGBS), fly ash (FA), silica fume (SF), metakaolin (MK), rice husk ash (RHA), and sugar cane, were only introduced to the concrete industry during the last two centuries.

The incorporation of SCMs in concrete production provides technical, economic and ecological benefits [18]. Introducing SCMs in fine form enhances the concrete density due to the filling effect of pores between cement particles and provides a physicochemical effect (nucleating effect), which promotes the hydration of OPC [19]. Most SCMs contain a high amount of siliceous or siliceous and aluminous materials in a non-crystalline (i.e. amorphous or glassy) state. These materials are favoured components in concrete production because they have the ability to react with CH in the presence of moisture to form additional CSH phases [20]. Converting CH to CSH through pozzolanic reaction not only improves the strength but also enhances the physical and durability properties of concrete mixtures. The benefits of using SCMs include: improved workability, increased ultimate strength, low heat of hydration, reduced permeability, and enhanced resistance to chemical attack [16, 21].

1.3 Problem Statement

The incorporation of SCMs as cement replacement material can provide numerous benefits to the concrete industry. In addition, by enhancing the engineering properties of concrete, a significant reduction in the total consumption of OPC can be achieved. This reduction will have a significant positive impact on the environment by way of the reduced total CO₂ emissions. However, there are some technical and economic barriers associated with the use of the existing SCMs. The technical shortfalls include slow rate of strength development; prolonged period of curing; increased water demand; increased chemical admixture dosage; and difficulties in the placing of concrete [22, 23].

Moreover, some SCMs may need a further treatment process, such as grinding and calcined under controlled conditions before being used in concrete production [14, 24]. Furthermore, due to market demand and transportation costs, SCMs can end up being more expensive than the OPC itself when they are imported from other countries [25, 26]. Therefore, there is an urgent need to find alternative materials as good sources for siliceous materials to be used as partial cement replacement material.

In many parts of the world, there is an abundance of natural dune sand. The particle size distribution of dune sand has shown that it does not meet the standard limit of fine aggregate gradations of either BS 882 and ASTM C 33 [27]. This is because the maximum size of the dune sand grains is less than 900 μ m. Therefore, the applications of dune sand have been limited to partial fine aggregate replacement, road construction, backfilling material, and sand concrete [27-30].

The characterization of the dune sand shows that it contains a high amount of SiO₂ (91%) in crystalline form. Unlike amorphous silica, crystalline silica does not react or hardly reacts with CH under normal conditions. However, the reactivity of this silica could be improved by a further grinding process or by thermal treatment under special conditions [5, 7, 31] To the best knowledge of the researcher, the potential for using ground dune sand (GDS) as a partial cement replacement material in binary and ternary combinations with slag and hydrated lime has not yet been investigated.

1.4 Objectives of the Study

The main objective of this study is to examine the potential of using GDS as partial cement replacement in binary and ternary blended system of OPC-GDS- slag and OPC-GDS- lime. The ultimate goal of this study is to establish



broad scientific knowledge and the engineering behaviour of concrete containing GDS in binary and ternary combinations of blended cement. The general objectives of this study are shown below:

- 1. To identify an effective method that is suitable to activate the GDS to form a latent reaction in cement-blended mixtures.
- 2. To determine the optimum level of GDS in developing blended cement mixtures and the optimum combinations for the ternary blended system.
- 3. To examine the effectiveness of developing blended cement mixtures on the physical and durability properties.
- 4. To ascertain the microstructural features and underlying mechanism of hydration of mixtures containing GDS blended cement system.

1.5 Scope of the Study

In this study, the examination is limited to the development of blended cement mixtures incorporating GDS, slag and lime in binary and ternary blended mixtures. The scope of this study is designed as follows:

- 1. To examine the reactivity of GDS, a control and blended mixture containing 30% GDS as cement replacement material were fabricated.
- 2. To examine the suitable treatment methods (mechanical, chemical and thermal methods), six different mixtures incorporating treated GDS were cast.
- 3. To examine the optimum replacement level of GDS in binary blended cement, five mixtures with different replacement levels (0%, 10%, 20%, 30% and 40%) were fabricated.
- 4. To examine the suitable combination of slag or lime with blended system containing OPC and GDS as base cementing materials, 12 preliminary mixtures were cast. Then, eight ternary blended cement mixtures containing low (15%), moderate (30%) and high (45%) amounts of slag or lime were introduced into the blended system containing 40% GDS and different amounts of OPC as base cementing materials.
- 5. To examine the fresh properties, compressive strength and physical and durability properties of binary and ternary blended mixtures the normal consistency, setting time, workability, compressive strength, chloride ions permeability, drying shrinkage and resistance to sulfate attack tests were conducted.
- 6. To study the microstructure of control and blended cement mixtures, SEM, EDX, XRD, DTA and TGA analyses were conducted to ascertain and understand the underlying mechanisms of the micro-scale changes that occur in the hydrated blended cement pastes.

1.6 Significance of the Study

The successful use of GDS to reduce OPC consumption has a potential impact on the sustainability and economy of concrete production in the Middle East and other countries, where resources of natural dune sands are unlimited. Moreover, the use of GDS and the proposed combinations of blended cement mixtures would contribute to saving a large amount of OPC and reduce the negative environmental effects of OPC manufacturing.

1.7 Overview of the Thesis

The thesis consists of five chapters. Chapter 1 presents the introduction to the thesis in terms of general background, objectives of the study and the scope of the study. Chapter 2 provides a literature review related to OPC, SCMs, activation methods, and microstructure study on the hydrated cementing mixtures. A critical discussion focusing on the research objectives is presented at the end of this chapter. Chapter 3 discusses the material properties and experimental programme used to carry out this study. Chapter 4 presents the results and discussion of the data obtained from the developed experimental programme. Chapter 5 concludes the problems discussed throughout this thesis and highlights the contributions of this research and recommendations for future studies.

REFERENCES

- [1] Uzal B, Turanli L, Mehta PK. High-volume natural pozzolan concrete for structural applications. ACI Materials Journal 2007; 104(5): 535-538.
- [2] Flower DJ, Sanjayan JG. Green house gas emissions due to concrete manufacture. The International Journal of Life Cycle Assessment 2007; 12(5): 282-288.
- [3] Mehta PK, Monteiro PJ. Concrete: microstructure, properties, and materials. McGraw-Hill; 2006.
- [4] Hewlett P. Lea's chemistry of cement and concrete. Butterworth-Heinemann; 2003.
- [5] Taylor HFW. Cement chemistry. Thomas Telford; 1997.
- [6] Neville AM. Properties of concrete. London: Pitman; 1973.
- [7] Mindess S, Young JF, Darwin D. Concrete. Prentice-Hall: Englewood Cliffs; 2003.
- [8] Worrell E, Price L, Martin N, Hendriks C, Meida LO. Carbon dioxide emissions from the global cement industry 1. Annual Review of Energy and the Environment 2001; 26(1): 303-329.
- [9] Malhotra V. Role of supplementary cementing materials in reducing greenhouse gas emissions. Concrete Technology for a Sustainable Development in the 21st Century, London: E and FN Spon 2000.
- [10] Malhotra V. Making concrete" greener" with fly ash. Concrete International 1999; 21: 61-66.
- [11] Nordhaus WD. Economic aspects of global warming in a post-Copenhagen environment. Proceedings of the National Academy of Sciences 2010; 107(26): 11721-11726.
- [12] Rehan R, Nehdi M. Carbon dioxide emissions and climate change: policy implications for the cement industry. Environmental Science and Policy 2005; 8(2): 105-114.

- [13] Hassan K, Cabrera J, Maliehe R. The effect of mineral admixtures on the properties of high-performance concrete. Cement and Concrete Composites 2000; 22(4): 267-271.
- [14] Sabir B, Wild S, Bai J. Metakaolin and calcined clays as pozzolans for concrete: a review. Cement and Concrete Composites 2001; 23(6): 441-454.
- [15] Malhotra V. Fly ash, slag, silica fume, and rice-husk ash in concrete: a review. Concrete International 1993; 15: 23-23.
- [16] Antcin P. The durability characteristics of high performance concrete: a review. Cement and Concrete Composites 2003; 25(4): 409-420.
- [17] Mehta P, Gjørv O. Properties of Portland cement concrete containing fly ash and condensed silica-fume. Cement and concrete research 1982; 12(5): 587-595.
- [18] Malhotra VM, Mehta PK. Pozzolanic and cementitious materials. Taylor and Francis; 1996.
- [19] Chindaprasirt P, Jaturapitakkul C, Sinsiri T. Effect of fly ash fineness on microstructure of blended cement paste. Construction and Building Materials 2007; 21(7): 1534-1541.
- [20] PCA. Design and Control of Concrete Mixtures. Canadian Portland Cement Association; 2007.
- [21] Long G, Wang X, Xie Y. Very-high-performance concrete with ultrafine powders. Cement and concrete research 2002; 32(4): 601-605.
- [22] Ghrici M, Kenai S, Meziane E. Mechanical and durability properties of cement mortar with Algerian natural pozzolana. Journal of Materials Science 2006; 41(21): 6965-6972.
- [23] Thomas M, Shehata M, Shashiprakash S, Hopkins D, Cail K. Use of ternary cementitious systems containing silica fume and fly ash in concrete. Cement and concrete research 1999; 29(8): 1207-1214.
- [24] Chandrasekhar S, Satyanarayana K, Pramada P, Raghavan P, Gupta T. Review processing, properties and applications of reactive silica from rice husk—an overview. Journal of Materials Science 2003; 38(15): 3159-3168.

- [25] Erdem TK, Kırca Ö. Use of binary and ternary blends in high strength concrete. Construction and Building Materials 2008; 22(7): 1477-1483.
- [26] Cassagnabère F, Escadeillas G, Mouret M. Study of the reactivity of cement/metakaolin binders at early age for specific use in steam cured precast concrete. Construction and Building Materials 2009; 23(2): 775-784.
- [27] Al-Harthy A, Halim MA, Taha R, Al-Jabri K. The properties of concrete made with fine dune sand. Construction and Building Materials 2007; 21(8): 1803-1808.
- [28] Guettala A, Mezghiche B, Chebili R. Strength comparisons between rolled sand concrete and dune sand concrete. Concrete Durability and Repair Technology: Proceedings of the International Conference Held at the University of Dundee, Scotland, 1999.
- [29] Tafraoui A, Lebaili S. Valorization of the Sand of Dune of the Western Erg (Algeria) in the Formulation of the UHPC. Journal of Applied Sciences 2006; 6: 2833-2836.
- [30] Khay SEE, Neji J, Loulizi A. Compacted dune sand concrete for pavement applications. Proceedings of the ICE-Construction Materials 2011; 164(2): 87-93.
- [31] Yang Q, Zhang S, Huang S, He Y. Effect of ground quartz sand on properties of high-strength concrete in the steam-autoclaved curing. Cement and concrete research 2000; 30(12): 1993-1998.
- [32] Spence RJ, Cook DJ. Building materials in developing countries. New York: Wiley Chichester; 1983.
- [33] García Giménez R, Vigil de la Villa R, Goñi S, Frías M. Fly Ash and Paper Sludge on the Evolution of Ternary Blended Cements: Mineralogy and Hydrated Phases. Journal of Materials in Civil Engineering 2014: In Press.
- [34] Moropoulou A, Bakolas A, Anagnostopoulou S. Composite materials in ancient structures. Cement and Concrete Composites 2005; 27(2): 295-300.
- [35] Bogue RH. The chemistry of Portland cement. Reinhold publishing corporation; 1947.

- [36] Shetty M. Concrete technology: theory and practice. S. Chand; 2005.
- [37] Gartner E. Industrially interesting approaches to "low-CO2" cements. Cement and concrete research 2004; 34(9): 1489-1498.
- [38] Oner A, Akyuz S. An experimental study on optimum usage of GGBS for the compressive strength of concrete. Cement and Concrete Composites 2007; 29(6): 505-514.
- [39] Rao GA. Influence of silica fume on long-term strength of mortars containing different aggregate fractions. Cement and concrete research 2001; 31(1): 7-12.
- [40] Zhang M, Lastra R, Malhotra V. Rice-husk ash paste and concrete: some aspects of hydration and the microstructure of the interfacial zone between the aggregate and paste. Cement and concrete research 1996; 26(6): 963-977.
- [41] Frias M, Villar-Cociña E, Valencia-Morales E. Characterisation of sugar cane straw waste as pozzolanic material for construction: Calcining temperature and kinetic parameters. Waste Management 2007; 27(4): 533-538.
- [42] Tagnit-Hamou A, Petrov N, Luke K. Properties of concrete containing diatomaceous earth. ACI Materials Journal 2003; 100(1): 73-78.
- [43] Mehta P. Studies on blended Portland cements containing Santorin earth. Cement and concrete research 1981; 11(4): 507-518.
- [44] Shannag MJ, Yeginobali A. Properties of pastes, mortars and concretes containing natural pozzolan. Cement and concrete research 1995; 25(3): 647-657.
- [45] Temuujin J, Williams R, Van Riessen A. Effect of mechanical activation of fly ash on the properties of geopolymer cured at ambient temperature. Journal of Materials Processing Technology 2009; 209(12): 5276-5280.
- [46] Yamei Z, Wei S, Lianfei S. Mechanical properties of high performance concrete made with high calcium high sulfate fly ash. Cement and concrete research 1997; 27(7): 1093-1098.
- [47] Siddique R. Performance characteristics of high-volume Class F fly ash concrete. Cement and concrete research 2004; 34(3): 487-493.

- [48] Berry E, Malhotra VM. Fly ash for use in concrete-a critical review. ACI Materials Journal 1980; 77(2): 59-73.
- [49] Gao J, Qian C, Liu H, Wang B, Li L. ITZ microstructure of concrete containing GGBS. Cement and concrete research 2005; 35(7): 1299-1304.
- [50] Yeau KY, Kim EK. An experimental study on corrosion resistance of concrete with ground granulate blast-furnace slag. Cement and concrete research 2005; 35(7): 1391-1399.
- [51] Osborne G. Durability of Portland blast-furnace slag cement concrete. Cement and Concrete Composites 1999; 21(1): 11-21.
- [52] Jo BW, Kim CH, Tae G, Park JB. Characteristics of cement mortar with nano-SiO2 particles. Construction and Building Materials 2007; 21(6): 1351-1355.
- [53] Langan BW, Weng K, Ward MA. Effect of silica fume and fly ash on heat of hydration of Portland cement. Cement and concrete research 2002; 32(7): 1045-1051.
- [54] Hooton RD. Permeability and pore structure of cement pastes containing fly ash, slag, and silica fume. Blended Cements, ASTM STP 1986; 897: 128-143.
- [55] Mazloom M, Ramezanianpour A, Brooks J. Effect of silica fume on mechanical properties of high-strength concrete. Cement and Concrete Composites 2004; 26(4): 347-357.
- [56] Bhanja S, Sengupta B. Influence of silica fume on the tensile strength of concrete. Cement and concrete research 2005; 35(4): 743-747.
- [57] Rao GA. Development of strength with age of mortars containing silica fume. Cement and concrete research 2001; 31(8): 1141-1146.
- [58] Al-Khalaf MN, Yousif HA. Use of rice husk ash in concrete. International Journal of Cement Composites and Lightweight Concrete 1984; 6(4): 241-248.
- [59] Givi AN, Rashid SA, Aziz FNA, Salleh MAM. Contribution of rice husk ash to the properties of mortar and concrete: a review. Journal of American Science 2010; 6(3): 157-165.

- [60] El-Dakroury A, Gasser M. Rice husk ash (RHA) as cement admixture for immobilization of liquid radioactive waste at different temperatures. Journal of Nuclear Materials 2008; 381(3): 271-277.
- [61] Habeeb G, Fayyadh M. Rice husk ash concrete: The effect of RHA average particle size on mechanical properties and drying shrinkage. Australian Journal of Basic and Applied Sciences 2009; 3(3): 1616-1622.
- [62] Zhang MH, Malhotra VM. High-performance concrete incorporating rice husk ash as a supplementary cementing material. ACI Materials Journal 1996; 93(6): 629-636.
- [63] Givi AN, Rashid SA, Aziz FNA, Salleh MAM. Assessment of the effects of rice husk ash particle size on strength, water permeability and workability of binary blended concrete. Construction and Building Materials 2010; 24(11): 2145-2150.
- [64] Ramlochan T, Thomas M, Gruber KA. The effect of metakaolin on alkali-silica reaction in concrete. Cement and concrete research 2000; 30(3): 339-344.
- [65] Poon C, Kou S, Lam L. Compressive strength, chloride diffusivity and pore structure of high performance metakaolin and silica fume concrete. Construction and Building Materials 2006; 20(10): 858-865.
- [66] Al-Akhras NM. Durability of metakaolin concrete to sulfate attack. Cement and concrete research 2006; 36(9): 1727-1734.
- [67] Poon C-S, Azhar S, Anson M, Wong Y-L. Performance of metakaolin concrete at elevated temperatures. Cement and Concrete Composites 2003; 25(1): 83-89.
- [68] Khatib J, Hibbert J. Selected engineering properties of concrete incorporating slag and metakaolin. Construction and Building Materials 2005; 19(6): 460-472.
- [69] Shvarzman A, Kovler K, Grader G, Shter G. The effect of dehydroxylation/amorphization degree on pozzolanic activity of kaolinite. Cement and concrete research 2003; 33(3): 405-416.
- [70] Heikal M, El-Didamony H, Morsy M. Limestone-filled pozzolanic cement. Cement and concrete research 2000; 30(11): 1827-1834.

- [71] Poppe A-M, De Schutter G. Cement hydration in the presence of high filler contents. Cement and concrete research 2005; 35(12): 2290-2299.
- [72] Matschei T, Lothenbach B, Glasser FP. The role of calcium carbonate in cement hydration. Cement and concrete research 2007; 37(4): 551-558.
- [73] Tennis P, Thomas M, Weiss W. State-of-the-Art Report on Use of Limestone in Cements at Levels of up to 15%. Portland Cement Association, 2011.
- [74] Lothenbach B, Le Saout G, Gallucci E, Scrivener K. Influence of limestone on the hydration of Portland cements. Cement and concrete research 2008; 38(6): 848-860.
- [75] Celik K, Jackson M, Mancio M, Meral C, Emwas A-H, Mehta P, Monteiro P. High-volume natural volcanic pozzolan and limestone powder as partial replacements for portland cement in selfcompacting and sustainable concrete. Cement and Concrete Composites 2014; 45: 136-147.
- [76] Velea A. New insight into phase change memories. Journal of Non-Crystalline Solids 2011; 357(14): 2626-2631.
- [77] Crucq P. Development of a method to measure the mechanical behavior of ASR gels. Part I: Literature Study; Alkali-silica reaction, causes, effects and prevention. Internal publication TU Delft: 2005.
- [78] Lawrence P, Cyr M, Ringot E. Mineral admixtures in mortars: effect of inert materials on short-term hydration. Cement and concrete research 2003; 33(12): 1939-1947.
- [79] Klimesch DS, Ray A, Sloane B. Autoclaved cement-quartz pastes: The effects on chemical and physical properties when using ground quartz with different surface areas Part I: Quartz of wide particle size distribution. Cement and concrete research 1996; 26(9): 1399-1408.
- [80] Chan C, Sakiyama M, Mitsuda T. Kinetics of the CaO---quartz---H2O reaction at 120° to 180° C in suspensions. Cement and Concrete Research 1978; 8(1): 1-5.
- [81] Uzal B, Turanlı L, Yücel H, Göncüoğlu M, Çulfaz A. Pozzolanic activity of clinoptilolite: a comparative study with silica fume, fly ash

and a non-zeolitic natural pozzolan. Cement and concrete research 2010; 40(3): 398-404.

- [82] Al-Jabri K, Shoukry H. Use of nano-structured waste materials for improving mechanical, physical and structural properties of cement mortar. Construction and Building Materials 2014; 73(1): 636-644.
- [83] Mofa N, Chervyakova O, Ketegenov T, Mansurov Z. Magnetic sorbets obtained by mechanochemical treatment of quartz containing mixtures. Chemistry in the Interests of Stable Development 2003; 11: 755-761.
- [84] Kumar S, Kumar R. Mechanical activation of fly ash: Effect on reaction, structure and properties of resulting geopolymer. Ceramics International 2011; 37(2): 533-541.
- [85] Frost RL, Mako E, Kristof J, Kloprogge JT. Modification of kaolinite surfaces through mechanochemical treatment – a mid-IR and near-IR spectroscopic study. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy 2002; 58(13): 2849-2859.
- [86] Vizcayno C, De Gutierrez R, Castello R, Rodriguez E, Guerrero C. Pozzolan obtained by mechanochemical and thermal treatments of kaolin. Applied Clay Science 2010; 49(4): 405-413.
- [87] Boldyrev VV. Mechanochemistry and mechanical activation of solids. Russian Chemical Reviews 2006; 75(3): 177-189.
- [88] Baláž P. Mechanochemistry in Minerals Engineering. Springer; 2008.
- [89] Bakharev T, Sanjayan JG, Cheng YB. Alkali activation of Australian slag cements. Cement and concrete research 1999; 29(1): 113-120.
- [90] Collins F, Sanjayan J. Workability and mechanical properties of alkali activated slag concrete. Cement and concrete research 1999; 29(3): 455-458.
- [91] Shi C, Day RL. Chemical activation of blended cements made with lime and natural pozzolans. Cement and concrete research 1993; 23(6): 1389-1396.
- [92] Jeon D, Jun Y, Jeong Y, Oh JE. Microstructural and strength improvements through the use of Na2CO3 in a cementless Ca(OH)2-

activated Class F fly ash system. Cement and concrete research 2015; 67: 215-225.

- [93] Lier JV, Bruyn PD, Overbeek JTG. The solubility of quartz. The Journal of Physical Chemistry 1960; 64(11): 1675-1682.
- [94] Aimin X, Sarkar SL. Microstructural study of gypsum activated fly ash hydration in cement paste. Cement and concrete research 1991; 21(6): 1137-1147.
- [95] Wu X, Jiang W, Roy DM. Early activation and properties of slag cement. Cement and concrete research 1990; 20(6): 961-974.
- [96] Allahverdi A, Ghorbani J. Chemical activation and set acceleration of lime-natural pozzolan cement. Ceramics Silikaty 2006; 50(4): 193-199.
- [97] Ambroise J, Murat M, Pera J. Investigations on synthetic binders obtained by middle-temperature thermal dissociation of clay minerals. Silicates Industriels 1986; 51(7-8): 99-107.
- [98] Zhang M, Malhotra V. Characteristics of a thermally activated alumino-silicate pozzolanic material and its use in concrete. Cement and concrete research 1995; 25(8): 1713-1725.
- [99] Said-Mansour M, Kadri E-H, Kenai S, Ghrici M, Bennaceur R. Influence of calcined kaolin on mortar properties. Construction and Building Materials 2011; 25(5): 2275-2282.
- [100] Morsy MS, Alsayed SH, Salloum YA. Development of eco-friendly binder using metakaolin-fly ash-lime-anhydrous gypsum. Construction and Building Materials 2012; 35(0): 772-777.
- [101] Siddique R, Klaus J. Influence of metakaolin on the properties of mortar and concrete: A review. Applied Clay Science 2009; 43(3): 392-400.
- [102] Aldea C-M, Young F, Wang K, Shah SP. Effects of curing conditions on properties of concrete using slag replacement. Cement and concrete research 2000; 30(3): 465-472.
- [103] Demir I, Serhat Baspinar M. Effect of silica fume and expanded perlite addition on the technical properties of the fly ash-lime-gypsum mixture. Construction and Building Materials 2008; 22(6): 1299-1304.

- [104] Jupe AC, Wilkinson AP, Luke K, Funkhouser GP. Class H cement hydration at 180 °C and high pressure in the presence of added silica. Cement and concrete research 2008; 38(5): 660-666.
- [105] Eilers LH, Nelson EB, Moran LK. High-Temperature Cement Compositions-Pectolite, Scawtite, Truscottite, or Xonotlite: Which Do You Want? Journal of Petroleum Technology 1983; 35(7): 1373-1377.
- [106] Meller N, Hall C, Kyritsis K, Giriat G. Synthesis of cement based CaO-Al₂O₃-SiO₂-H₂O (CASH) hydroceramics at 200 and 250 °C: Ex-situ and in-situ diffraction. Cement and concrete research 2007; 37(6): 823-833.
- [107] Luke K. Phase studies of pozzolanic stabilized calcium silicate hydrates at 180 °C. Cement and concrete research 2004; 34(9): 1725-1732.
- [108] Bágel L. Strength and pore structure of ternary blended cement mortars containing blast furnace slag and silica fume. Cement and concrete research 1998; 28(7): 1011-1022.
- [109] Menéndez G, Bonavetti V, Irassar EF. Strength development of ternary blended cement with limestone filler and blast-furnace slag. Cement and Concrete Composites 2003; 25(1): 61-67.
- [110] Wongkeo W, Thongsanitgarn P, Ngamjarurojana A, Chaipanich A. Compressive strength and chloride resistance of self-compacting concrete containing high level fly ash and silica fume. Materials and Design 2014; 64: 261-269.
- [111] Shannag MJ. High strength concrete containing natural pozzolan and silica fume. Cement and Concrete Composites 2000; 22(6): 399-406.
- [112] Li G, Zhao X. Properties of concrete incorporating fly ash and ground granulated blast-furnace slag. Cement and Concrete Composites 2003; 25(3): 293-299.
- [113] Berry EE, Malhotra VM. Fly ash for use in concrete-a critical review. ACI Materials Journal 1980; 77(2): 59-73.
- [114] Snyder KA, Feng X, Keen BD, Mason TO. Estimating the electrical conductivity of cement paste pore solutions from OH–, K+ and Na+ concentrations. Cement and concrete research 2003; 33(6): 793-798.

- [115] Shi C. Effect of mixing proportions of concrete on its electrical conductivity and the rapid chloride permeability test (ASTM C1202 or ASSHTO T277) results. Cement and concrete research 2004; 34(3): 537-545.
- [116] Halamickova P, Detwiler RJ, Bentz DP, Garboczi EJ. Water permeability and chloride ion diffusion in portland cement mortars: Relationship to sand content and critical pore diameter. Cement and concrete research 1995; 25(4): 790-802.
- [117] Chindaprasirt P, Rukzon S, Sirivivatnanon V. Resistance to chloride penetration of blended Portland cement mortar containing palm oil fuel ash, rice husk ash and fly ash. Construction and Building Materials 2008; 22(5): 932-938.
- [118] Ganesan K, Rajagopal K, Thangavel K. Rice husk ash blended cement: Assessment of optimal level of replacement for strength and permeability properties of concrete. Construction and Building Materials 2008; 22(8): 1675-1683.
- [119] Dogan UA, Kurt EB, Saran AG, Ozkul MH. Benchmarking concretes with pozzolanic materials in terms of rapid chloride penetration test. ACI Materials Journal 2009; 106(3): 251-257.
- [120] Ramezanianpour AA, Mousavi R, Kalhori M. Influence of zeolite additive on chloride durability and carbonation of concretes. Applied mathematics in Engineering, Management and Technology 2014; The special issue in Management and Technology: 1081-1093.
- [121] Papadakis VG. Effect of supplementary cementing materials on concrete resistance against carbonation and chloride ingress. Cement and concrete research 2000; 30(2): 291-299.
- [122] Naik TR, Singh SS, Ramme BW. Mechanical properties and durability of concrete made with blended fly ash. ACI Materials Journal 1998; 95(4): 454-462.
- [123] Wee TH, Suryavanshi AK, Tin SS. Influence of aggregate fraction in the mix on the reliability of the rapid chloride permeability test. Cement and Concrete Composites 1999; 21(1): 59-72.
- [124] Wee T, Suryavanshi AK, Tin S. Evaluation of rapid chloride permeability test (RCPT) results for concrete containing mineral admixtures. ACI Materials Journal 2000; 97(2): 221-232.

- [125] Stanish K, Hooton R, Thomas M. A rapid migration test for evaluation of the chloride penetration resistance of high performance concrete. PCI/FHWA/FIB International Symposium on High Performance Concrete, 2000.
- [126] Zheng L, Beaudoin J. The permeability of cement systems to chloride ingress and related test methods. Cement, Concrete and Aggregates 2000; 22(1): 16-23.
- [127] Rao GA. Long-term drying shrinkage of mortar influence of silica fume and size of fine aggregate. Cement and concrete research 2001; 31(2): 171-175.
- [128] Wongkeo W, Thongsanitgarn P, Chaipanich A. Compressive strength and drying shrinkage of fly ash-bottom ash-silica fume multi-blended cement mortars. Materials and Design 2012; 36: 655-662.
- [129] El Hindy E, Miao B, Chaallal O, Aitcin PC. Drying shrinkage of readymixed high-performance concrete. ACI Materials Journal 1994; 91(3): 300-305.
- [130] Šelih J, Bremner TW. Drying of saturated lightweight concrete: an experimental investigation. Materials and Structures 1996; 29(7): 401-405.
- [131] Rao GA. Influence of silica fume replacement of cement on expansion and drying shrinkage. Cement and concrete research 1998; 28(10): 1505-1509.
- [132] Chindaprasirt P, Homwuttiwong S, Sirivivatnanon V. Influence of fly ash fineness on strength, drying shrinkage and sulfate resistance of blended cement mortar. Cement and concrete research 2004; 34(7): 1087-1092.
- [133] Yunfeng L. Early Age Shrinkage of Cement Mortar with Steel Slag Admixture. Power and Energy Engineering Conference (APPEEC), Chengdu: China 2010.
- [134] Wee T, Suryavanshi AK, Wong S, Rahman AA. Sulfate resistance of concrete containing mineral admixtures. ACI Materials Journal 2000; 97(5): 536-549.
- [135] Mehta PK. Mechanism of sulfate attack on portland cement concrete Another look. Cement and concrete research 1983; 13(3): 401-406.

- [136] Odler I, Colan-Subauste J. Investigations on cement expansion associated with ettringite formation. Cement and concrete research 1999; 29(5): 731-735.
- [137] Stark J, Bollmann K. Delayed ettringite formation in concrete. Nordic Concrete Research Publication 2000; 23: 4-28.
- [138] Saito T, Khamhou S, Yumoto T, Otsuki N. Permeability of Sulfate Ions in Cementitious Materials Containing γ-Ca2SiO₄ after Autoclave Curing and Accelerated Carbonation. Journal of Advanced Concrete Technology 2011; 9(3): 223-230.
- [139] Luo FJ, He L, Pan Z, Duan WH, Zhao XL, Collins F. Effect of very fine particles on workability and strength of concrete made with dune sand. Construction and Building Materials 2013; 47: 131-137.
- [140] Cisse I, Laquerbe M. Mechanical characterisation of filler sandcretes with rice husk ash additions: Study applied to Senegal. Cement and concrete research 2000; 30(1): 13-18.
- [141] Laquerbe M, Cisse I, Ahouansou G. Pour une utilisation rationnelle des graveleux latéritiques et des sables de dunes comme granulats à béton Application au cas du Sénégal. Materials and Structures 1995; 28(10): 604-610.
- [142] Damene Z, Goual M, Saiti I, Ahmida F. Study of the Effect of Aluminum Content and C/S Ratio on the Physico-Mechanical and Thermal Properties of a Lightweight Concrete Made From Sand Dune. Journal of Fundamental and Applied Sciences 2014; 6(2): 222-230.
- [143] Benabed B, Azzouz L, Kadri E-H, Kenai S, Belaidi ASE. Effect of fine aggregate replacement with desert dune sand on fresh properties and strength of self-compacting mortars. Journal of Adhesion Science and Technology 2014; 28(21): 2182-2195.
- [144] Duval R, Kadri E. Influence of silica fume on the workability and the compressive strength of high-performance concretes. Cement and concrete research 1998; 28(4): 533-547.
- [145] Blanco F, Garcia M, Ayala J, Mayoral G, Garcia M. The effect of mechanically and chemically activated fly ashes on mortar properties. Fuel 2006; 85(14): 2018-2026.

- [146] Guettala S, Mezghiche B. Compressive strength and hydration with age of cement pastes containing dune sand powder. Construction and Building Materials 2011; 25(3): 1263-1269.
- [147] Shi C, Day RL. Comparison of different methods for enhancing reactivity of pozzolans. Cement and concrete research 2001; 31(5): 813-818.
- [148] Bougara A, Lynsdale C, Ezziane K. Activation of Algerian slag in mortars. Construction and Building Materials 2009; 23(1): 542-547.
- [149] Lange F, Mörtel H, Rudert V. Dense packing of cement pastes and resulting consequences on mortar properties. Cement and concrete research 1997; 27(10): 1481-1488.
- [150] Ezziane K, Bougara A, Kadri A, Khelafi H, Kadri E. Compressive strength of mortar containing natural pozzolan under various curing temperature. Cement and Concrete Composites 2007; 29(8): 587-593.
- [151] Shi C, Day RL. Pozzolanic reaction in the presence of chemical activators: Part I. Reaction kinetics. Cement and concrete research 2000; 30(1): 51-58.
- [152] Klieger P, Lamond JF. Significance of tests and properties of concrete and concrete-making materials. ASTM International; 1994.
- [153] Diamond S. The microstructure of cement paste and concrete-a visual primer. Cement and Concrete Composites 2004; 26(8): 919-933.
- [154] Ramachandran VS, Beaudoin JJ. Handbook of analytical techniques in concrete science and technology: principles, techniques and applications. New Jersey: 2000.
- [155] Omotoso O, Ivey D, Mikula R. Characterization of chromium doped tricalcium silicate using SEM/EDS, XRD and FTIR. Journal of Hazardous Materials 1995; 42(1): 87-102.
- [156] Scrivener KL. Backscattered electron imaging of cementitious microstructures: understanding and quantification. Cement and Concrete Composites 2004; 26(8): 935-945.
- [157] Yılmaz B, Olgun A. Studies on cement and mortar containing lowcalcium fly ash, limestone, and dolomitic limestone. Cement and Concrete Composites 2008; 30(3): 194-201.

- [158] Nochaiya T, Sekine Y, Choopun S, Chaipanich A. Microstructure, characterizations, functionality and compressive strength of cementbased materials using zinc oxide nanoparticles as an additive. Journal of Alloys and Compounds 2014: In Press.
- [159] Yajun J, Cahyadi JH. Effects of densified silica fume on microstructure and compressive strength of blended cement pastes. Cement and concrete research 2003; 33(10): 1543-1548.
- [160] Grabowski E, Gillott J. Effect of replacement of silica flour with silica fume on engineering properties of oilwell cements at normal and elevated temperatures and pressures. Cement and concrete research 1989; 19(3): 333-344.
- [161] Stutzman PE. Scanning electron microscopy in concrete petrography. Materials Science of Concrete Special Volume: Calcium Hydroxide in Concrete, Proceedings-Anna Maria Island-FL 2000: 59-72.
- [162] Kjellsen KO, Detwiler RJ, Gjørv OE. Development of microstructures in plain cement pastes hydrated at different temperatures. Cement and concrete research 1991; 21(1): 179-189.
- [163] Klimesch DS, Ray A. Autoclaved cement-quartz pastes with metakaolin additions. Advanced Cement Based Materials 1998; 7(3): 109-118.
- [164] Hamish C, Abhi R, Paul T. Autoclaved lime-colloidal silica slurries and formation of tobermorite. Journal of the Australasian Ceramic Society 2007; 43(2): 150-153.
- [165] Li H, Xiao H, Yuan J, Ou J. Microstructure of cement mortar with nano-particles. Composites Part B: Engineering 2004; 35(2): 185-189.
- [166] Ashraf M, Naeem Khan A, Ali Q, Mirza J, Goyal A, Anwar A. Physicochemical, morphological and thermal analysis for the combined pozzolanic activities of minerals additives. Construction and Building Materials 2009; 23(6): 2207-2213.
- [167] Pane I, Hansen W. Investigation of blended cement hydration by isothermal calorimetry and thermal analysis. Cement and concrete research 2005; 35(6): 1155-1164.
- [168] Vessalas K, Thomas P, Ray A, Guerbois J-P, Joyce P, Haggman J. Pozzolanic reactivity of the supplementary cementitious material

pitchstone fines by thermogravimetric analysis. Journal of Thermal Analysis and Calorimetry 2009; 97(1): 71-76.

- [169] Alarcon-Ruiz L, Platret G, Massieu E, Ehrlacher A. The use of thermal analysis in assessing the effect of temperature on a cement paste. Cement and concrete research 2005; 35(3): 609-613.
- [170] Midgley H. The determination of calcium hydroxide in set Portland cements. Cement and concrete research 1979; 9(1): 77-82.
- [171] Peschard A, Govin A, Grosseau P, Guilhot B, Guyonnet R. Effect of polysaccharides on the hydration of cement paste at early ages. Cement and concrete research 2004; 34(11): 2153-2158.
- [172] Alhozaimy A, Fares G, Al-Negheimish A, Jaafar MS. The autoclaved concrete industry: An easy-to-follow method for optimization and testing. Construction and Building Materials 2013; 49: 184-193.
- [173] Noorvand H, Abang Ali AA, Demirboga R, Farzadnia N, Noorvand H. Incorporation of nano TiO2 in black rice husk ash mortars. Construction and Building Materials; 2013; 47(0): 1350-1361.
- [174] Garcés P, Pérez Carrión M, García-Alcocel E, Payá J, Monzó J, Borrachero M. Mechanical and physical properties of cement blended with sewage sludge ash. Waste Management 2008; 28(12): 2495-2502.
- [175] Saikia N, Kato S, Kojima T. Thermogravimetric investigation on the chloride binding behaviour of MK–lime paste. Thermochimica Acta 2006; 444(1): 16-25.
- [176] Kikuma J, Tsunashima M, Ishikawa T, Matsuno S, Ogawa A, Matsui K, Sato M. Effects of quartz particle size and water-to-solid ratio on hydrothermal synthesis of tobermorite studied by in-situ time-resolved X-ray diffraction. Journal of Solid State Chemistry 2011; 184(8): 2066-2074.
- [177] Shi C, Hu S. Cementitious properties of ladle slag fines under autoclave curing conditions. Cement and concrete research 2003; 33(11): 1851-1856.
- [178] Hong SY, Glasser FP. Phase relations in the CaO–SiO₂–H₂O system to 200 °C at saturated steam pressure. Cement and concrete research 2004; 34(9): 1529-1534.

- [179] Ghrici M, Kenai S, Said-Mansour M. Mechanical properties and durability of mortar and concrete containing natural pozzolana and limestone blended cements. Cement and Concrete Composites 2007; 29(7): 542-549.
- [180] Andrade C. Calculation of chloride diffusion coefficients in concrete from ionic migration measurements. Cement and concrete research 1993; 23(3): 724-742.
- [181] Shi C. Strength, pore structure and permeability of high performance alkali-activated slag mortars. Cement and concrete research 1996; 26 (12): 1789-1800.

