



**UNIVERSITI PUTRA MALAYSIA**

***PREPARATION AND CHARACTERIZATION OF POLY(VINYL ALCOHOL)/CHITOSAN BIO-NANOCOMPOSITES REINFORCED WITH CELLULOSE NANOCRYSTALS, CELLULOSE NANOCRYSTALS/ZINC OXIDE AND CELLULOSE NANOCRYSTALS/ZINC OXIDE-SILVER NANOPARTICLES***

**SUSSAN AZIZY**

**FS 2014 21**



**PREPARATION AND CHARACTERIZATION OF POLY(VINYL ALCOHOL)/CHITOSAN BIO-NANOCOMPOSITES REINFORCED WITH CELLULOSE NANOCRYSTALS, CELLULOSE NANOCRYSTALS/ZINC OXIDE AND CELLULOSE NANOCRYSTALS/ZINC OXIDE-SILVER NANOPARTICLES**

**By**

**SUSSAN AZIZY**

**Thesis Submitted to the School of Graduates Studies, Universiti Putra Malaysia, in Fulfilment of Requirements for the Degree of Doctor of Philosophy**

**June 2014**

## COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in  
Fulfilment of the requirement for the degree Doctor of Philosophy

**PREPARATION AND CHARACTERIZATION OF POLY(VINYL  
ALCOHOL)/CHITOSAN BIO-NANOCOMPOSITES REINFORCED WITH  
CELLULOSE NANOCRYSTALS, CELLULOSE NANOCRYSTALS/ZINC  
OXIDE AND CELLULOSE NANOCRYSTALS/ZINC OXIDE-SILVER  
NANOPARTICLES**

By

**SUSSAN AZIZY**

**June 2014**

**Chairman: Professor Mansor Bin Ahmad, PhD**

**Faculty: Science**

Polymer nanocomposites are materials composed of polymer matrices and nano-sized fillers with enhanced physical and chemical properties in comparison with the pure polymers or conventional composites. Strong research attempts have been focused on the use of bio-polymers because of increasing environmental concerns of using petroleum base polymers. The present research aims to study the influence of cellulose nanocrystals (CNCs), cellulose nanocrystals/zinc oxide nanoparticles (CNCs/ZnO-NPs) and cellulose nanocrystals/zinc oxide-silver nanoparticles (CNCs/ZnO-Ag-NPs) on the properties of poly(vinyl alcohol)/chitosan (PVA/CTS), an important bio-polymer blend.

Cellulose nanocrystal was extracted by acid-catalyzed hydrolysis of cotton cellulose. ZnO-NPs and ZnO-Ag-NPs were prepared in suspension of CNCs as a stabilizer via a co-precipitation method. The samples were characterized using Fourier transform infrared (FTIR), energy dispersive x-ray spectroscopy (EDS), ultraviolet-visible (UV-vis), x-ray diffraction (XRD), transmission electron microscope (TEM), field emission scanning electron microscope (FESEM), thermogravimetric analysis (TGA) and antimicrobial tests. According to the XRD and TEM results, polygonal structured ZnO nanocrystallites with a mean size of less than 30 nm were formed. The prepared ZnO-Ag nanocrystallites were spherical with a mean size diameters in a 12–35 nm range.

The bio-nanocomposites were prepared by mixing various percentages of nano-sized fillers and PVA/CTS blends in the ratio of 3:1. The properties of the prepared bio-nanocomposites were studied by XRD, TEM, TGA, UV-vis, barrier, tensile and antimicrobial tests. The PVA/CTS/CNCs bio-nanocomposites with low percentage of the CNCs (1.0 wt%) exhibits maximum mechanical, thermal, and barrier properties.

The PVA/CTS/CNCs/ZnO 5.0 wt% bio-nanocomposite exhibits the maximum tensile, and barrier properties. TGA results show that the maximum thermal decomposition of the PVA/CTS increases by about 34 °C at 1.0 wt%. The UV-visible spectrophotometric study shows that the bio-nanocomposites display an excellent performance of absorbing UV light. The antimicrobial tests reveal a favorable antibacterial effect for those PVA/CTS blend films filled with the high levels of CNCs/ZnO-NPs content.

The PVA/CTS/CNCs/ZnO-Ag 5.0 wt% bio-nanocomposite shows the maximum tensile, thermal and barrier properties. The UV-visible measurements showed that the bio-nanocomposites display perfect absorption in the range of 300-700 nm. The all bio-nanocomposites show antibacterial power, with the maximum effect in the 7.0 wt% of CNCs/ZnO-Ag-NPs loading.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

**PENYEDIAAN DAN PENCIRIAN POLI(VINIL ALKOHOL)/KITOSAN BIO-NANOKOMPOSIT DIPERKUKUH DENGAN SELULOSA NANOKRISTAL, SELULOSA NANOKRISTAL/ZINK OKSIDA DAN SELULOSA NANOKRISTAL/ZINK OKSIDA-PERAK NANOPARTIKEL**

Oleh

**SUSSAN AZIZY**

**Jun 2014**

**Pengerusi: Profesor Mansor Bin Ahmad, PhD**

**Fakulti: Sains**

Nanokomposit polimer adalah bahan terdiri daripada matrik polimer dan pengisi bersaiz nano dengan sifat-sifat fizikal dan kimia yang dipertingkatkan berbanding dengan polimer tulen atau komposit konvensional. dipertingkatkan. Sebaliknya, dalam bidang polimer, sebahagian besar penyelidikan tertumpu kepada polimer terbiodegradasikan disebabkan kebimbangan alam sekitar menggunakan polimer asas petroleum yang semakin meningkat. Penggunaan bio-polimer telah menjadi tumpuan utama usaha penyelidikan kerana peningkatkan kebimbangan alam sekitar terhadap penggunaan polimer berasaskan petroleum. Kajian ini bertujuan untuk mengkaji pengaruh nanokristal selulosa (CNCs), nanokristal selulosa/nanozarah zink oksida (CNCs/ZnO-NPs) dan nanokristal selulosa/nanozarah zink oksida-Perak (CNCs/ZnO-Ag-NPs) terhadap sifat-sifat poli(vinil alkohol)/kitosan (PVA/CTS), satu adunan penting bio-polimer.

Selulosa nanokristal telah diekstrak keluar daripada selulosa kapas melalui hidrolisis bermangkikan asid. ZnO-NPs dan ZnO-Ag-NPs telah disediakan dalam ampaiian CNCs sebagai penstabil melalui kaedah ko-mendakan. Sampel yang telah disediakan disifatkan menggunakan jelman Fourier inframerah (FTIR), spektroskopi tenaga serakan x-ray (EDS), spektroskopi ultraungu/nampak (UV-vis), pembelauan sinar x-ray (XRD), mikroskop elektron transmisi (TEM), bidang pelepasan imbasan elektron mikroskop (FESEM), termogravimetri (TGA) dan antimikrob ujian. Menurut keputusan XRD dan TEM, ZnO nanokristal yang berstruktur polygonal dengan saiz purata kurang daripada

30 nm telah dibentuk. Pembentukan ZnO-Ag nanokristal adalah bulat dengan saiz purata diameter dalam julat 12-35 nm.

Bio-nanokomposit telah disediakan dengan mencampurkan pelbagai peratusan pengisi bersaiz nano dalam campuran PVA/CTS dengan nisbah 3:1. Sifat-sifat bio-nanokomposit kemudian dikaji dengan ujian XRD, TEM, TGA, UV-nampak, halangan, tegangan dan antimikrobial. PVA/CTS/CNCs bio-nanokomposit dengan peratusan CNCs yang rendah (1.0 wt%) mempamerkan sifat mekanikal, terma, dan penghalang yang maksimum.

PVA/CTS/CNCs/ZnO 5.0 wt% bio-nanokomposit mempamerkan sifat tegangan dan halangan yang maksimum. Keputusan TGA menunjukkan bahawa penguraian terma maksimum bagi PVA/CTS meningkat kira-kira 34 °C pada 1.0 wt%. Kajian spektrofotometrik UV-nampak menunjukkan bahawa bio-nanokomposit memaparkan prestasi cemerlang dalam penyerapan cahaya UV. Tambahan pula, ujian antimikrobial menunjukkan bahawa kesan antibakteria yang baik boleh dicapai pada PVA/CTS filem dengan kandungan CNCs/ZnO-NPs yang tinggi.

PVA/CTS/CNCs/ZnO-Ag 5.0 wt% bio-nanokomposit mempamerkan sifat tegangan, terma dan halangan yang maksimum. Pengukuran penyerapan UV menunjukkan bahawa bio-nanokomposit memaparkan penyerapan sempurna dalam julat 300-700 nm. Selain itu, kesemua bio-nanokomposit menunjukkan kuasa antibakteria, dengan kesan maksimum pada 7.0 wt% CNCs/ZnO-Ag-NPs.

## ACKNOWLEDGEMENTS

Praise and gratitude be to ALLAH, almighty, without whose gracious help it would have been impossible to accomplish this work.

I would like to express my sincere gratitude to my supervisor Prof. Dr. Mansor Bin Ahmad, who taught me not only analytical and polymer chemistry, but also about patience and perseverance. Thanks for all your time and advises. As such, I want to express gratitude to members of supervisory committee, Prof. Dr. Mohd Zobir Hussein and Dr. Nor Azowa Ibrahim for their guidance and constant support through the research. I admire their devotion to science.

I also thank to polymer research group, who shared with me not only their knowledge, also the laboratory space.

In addition, I am also very grateful to my family, especially my husband for everything. In gratitude, finally I want to express to all the staff and lecturer of Department of Chemistry, Faculty of Science and Universiti Putra Malaysia that gave me the opportunity to study. I will fondly remember your support, knowledge, assistance, advice, and teaching. I thank the administrators, the Dean and staffs of the chemistry department for the assistance provided throughout the duration of my study at UPM.



This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory committee were as follow:

**Mansor Bin Ahmad, PhD**

Professor  
Faculty of Science  
Universiti Putra Malaysia  
(Member)

**Mohd Zobir Hussein, PhD**

Professor  
Faculty of Science  
Universiti Putra Malaysia  
(Member)

**Nor Azowa Ibrahim, PhD**

Senior Lecturer  
Faculty of Science  
Universiti Putra Malaysia  
(Member)

---

**BUJANG BIN KIM HUAT, PhD**

Professor and Dean  
School of Graduate Studies  
Universiti Putra Malaysia

Date: 23 June 2014

## DECLARATION

### Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (GraduateStudies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia(Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature: \_\_\_\_\_ Date: \_\_\_\_\_

Name and Matric No.: \_\_\_\_\_

## Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: \_\_\_\_\_  
Name of  
Chairman of  
Supervisory  
Committee: \_\_\_\_\_

Signature: \_\_\_\_\_  
Name of  
Member of  
Supervisory  
Committee: \_\_\_\_\_

Signature: \_\_\_\_\_  
Name of  
Member of  
Supervisory  
Committee: \_\_\_\_\_

Signature: \_\_\_\_\_  
Name of  
Member of  
Supervisory  
Committee: \_\_\_\_\_

## TABLE OF CONTENTS

	<b>Page</b>
<b>ABSTRACT</b>	iii
<b>ABSTRAK</b>	v
<b>ACKNOWLEDGEMENTS</b>	vii
<b>APPROVAL</b>	viii
<b>DECLARATION</b>	ix
<b>LIST OF TABLES</b>	xvi
<b>LIST OF FIGURES</b>	xvii
<b>LIST OF ABBREVIATIONS</b>	xx
<b>CHAPTER</b>	<b>1</b>
<b>1 INTRODUCTION</b>	<b>1</b>
1.1 General Background of Research	1
1.2 Problem Statement	3
1.3 Objectives	4
<b>2 LITERATURE REVIEW</b>	<b>5</b>
2.1 Biodegradable Polymer	5
2.2 Chitosan	5
2.3 Poly(vinyl alcohol)	7
2.4 Poly(vinyl alcohol)/Chitosan Blends	7
2.5 Cellulose	7
2.6 Nano sized Cellulose	9
2.6.1 Microfibrillated Cellulose	9
2.6.2 Cellulose Nanocrystals	9
2.7 Polymer Nanocomposites	11
2.8 Polymer Bio-nanocomposites	12
2.9 Polymer Bio-nanocomposite Based on CNCs	12

2.9.1	Mechanical Properties of CNCs/Polymer Bio-nanocomposites	13
2.9.2	Thermal Properties of CNCs/Polymer Bio-nanocomposites	14
2.9.3	Barrier Properties of CNCs/Polymer Bio-nanocomposites	15
2.10	Processing Methods of CNCs/Polymer Bio-nanocomposites	16
2.10.1	Casting-Evaporation Processing	16
2.10.2	Extrusion	18
2.10.3	Sol-gel Processing	18
2.10.4	Electrospinning	18
2.11	Nanoparticles	18
2.12	Polymer-Supported Inorganic Nanoparticles	20
2.13	Zinc oxide Nanoparticle	21
2.13.1	Antibacterial Activities of ZnO Nanoparticles	22
2.13.2	UV Absorbers of ZnO Nanoparticles	22
2.14	Zinc oxide–Silver Hetero-structure	23
2.15	Polymer Nanocomposites Based on Inorganic Nanoparticles	23
2.16	Polymer Nanocomposites Based on Zinc oxide Nanoparticles	25
2.17	Polymer Nanocomposites Based on Zinc oxide-Silver Nanoparticles	25
2.18	Polymer Nanocomposites Based on CNCs /Inorganic Nanoparticles	26
<b>3</b>	<b>MATERIALS AND METHOD</b>	<b>27</b>
3.1	Materials	27
3.2	Extraction of Cellulose Nanocrystals by Acid Hydrolysis	27
3.3	Preparation of CNCs/ZnO Nanocomposites	27
3.4	Preparation of CNCs/ZnO-Ag Nanocomposites	28
3.5	Preparation of PVA/CTS Blend	28
3.6	Preparation of PVA/CTS/CNCs Bio-Nanocomposites	29
3.7	Preparation of PVA/CTS/CNCs/ZnO Bio-Nanocomposites	29
3.8	Characterization of Nanomaterials	29
3.8.1	X-ray Diffraction	29
3.8.2	Fourier Transform Infra-Red Spectroscopy	30
3.8.3	Thermogravimetric Analysis	30

3.8.4	Field Emission Scanning Electron Microscopy	30
3.8.5	Transmission Electron Microscopy	30
3.8.6	Energy Dispersive X-ray Spectroscopy	31
3.8.7	Tensile Properties Measurement	31
3.8.8	UV-Vis Absorption Measurements	31
3.8.9	Antibacterial Measurements	32
3.8.10	Oxygen Transmission Measurements	32
<b>4</b>	<b>RESULTS AND DISCUSSION</b>	<b>33</b>
4.1	Extraction and Characterization of CNCs	33
4.1.1	Extraction of CNCs	33
4.1.2	Characterization of CNCs	34
4.2	Preparation and Characterization of CNCs/ZnO Nanocomposites	41
4.2.1	Preparation of CNCs/ZnO Nanocomposites	41
4.2.2	Characterization of CNCs/ZnO Nanocomposites	41
4.3	Preparation and Characterization of CNCs/ZnO-Ag Nanocomposites	51
4.3.1	Preparation of CNCs/ZnO-Ag Nanocomposites	51
4.3.2	Characterization of CNCs/ZnO-Ag Nanocomposites	51
4.4	Characterization of PVA/CTS/CNCs Bio-nanocomposites	63
4.4.1	X-ray Diffraction	63
4.4.2	Transmission Electron Microscopy	64
4.4.3	Tensile Properties	66
4.4.4	Thermogravimetric Analysis	68
4.4.5	Oxygen Transmission Measurement	70
4.5	Characterization of PVA/CTS/CNCs/ZnO Bio-nanocomposites	72
4.5.1	X-ray Diffraction	72
4.5.2	Transmission Electron Microscopy	73
4.5.3	Tensile Properties	76
4.5.4	Thermogravimetric Analysis	78
4.5.5	Oxygen Transmission Measurement	80
4.5.6	UV-Vis Absorption	81

4.5.7	Antibacterial Activity	83
4.6	Characterization of PVA/CTS/CNCs/ZnO-Ag Bio-nanocomposites	85
4.6.1	X-ray Diffraction	85
4.6.2	Transmission Electron Microscopy	86
4.6.3	Tensile Properties	89
4.6.4	Thermogravimetric Analysis	91
4.6.5	Oxygen Transmission Measurement	93
4.6.6	UV-Vis Absorption	94
4.6.7	Antibacterial Activity	96
<b>5</b>	<b>CONCLUSIONS AND RECOMMENDATIONS FOR FURTHER STUDY</b>	<b>99</b>
5.1	Conclusions	99
5.2	Recommendations for Further Study	101
	<b>REFERENCES</b>	<b>102</b>
	<b>BIODATA OF STUDENT</b>	<b>132</b>
	<b>LIST OF PUBLICATIONS</b>	<b>133</b>

## LIST OF TABLES

Table	Page
4-1. Thermal data of CNCs and original cellulose	39
4-2. UV-absorbance data of CNCs/ZnO nanocomposites	47
4-3. Thermal data of CNCs and CNCs/ZnO nanocomposites	49
4-4. Inhibition zone data of CNCs/ZnO, CNCs and ZnO free-CNCs	50
4-5. UV-vis Absorbance data of C/Z-A samples	59
4-6. Thermal data of and CNCs and CNCs/ZnO-Ag nanocomposites	61
4-7. Inhibition zone data of CNCs/ZnO-Ag-NPs and ZnO-Ag	62
4-8. Results from tensile test of PVA/CTS and its CNCs BNCs	68
4-9. Thermal data of PVA/CTS blend and its CNCs BNCs	70
4-10. OT values of PVA/CTS blend and its CNCs BNCs films	71
4-11. Results from tensile test of PVA/CTS and its CNCs/ZnO BNCs	78
4-12. Thermal data of PVA/CTS blend and its CNCs/ZnO BNCs	80
4-13. OT data of PVA/CTS blend and its CNCs/ZnO BNCs	81
4-14. UV absorbance and transmittance values of PVA/CTS blend and its CNCs/ZnO BNCs	83
4-15. Inhibition zone data of PVA/CTS blend and its CNCs/ZnO BNCs	84
4-16. Results from tensile test of PVA/CTS and PVA/CTS/CNCs/ZnO-Ag BNCs	91
4-17. Thermal data of PVA/CTS blend and its CNCs/ZnO-Ag BNCs	93
4-18. OT data of PVA/CTS blend and its CNCs/ZnO-Ag BNCs	94
4-19. UV absorbance and transmittance values of PVA/CTS blend and its CNCs/ZnO-Ag BNCs	96
4-20. Inhibition zone data of PVA/CTS blend and its CNCs/ZnO-Ag BNCs	97



## LIST OF FIGURES

Figure	Page
2-1. Synthetic Scheme of Chitosan from Chitin	6
2-2. The Basic Chemical Structure of Cellulose	8
2-3. Crystalline and Amorphous Regions in Cellulose Fibers	8
2-4. The Extraction of CNCs Using Sulfuric Acid	11
2-5. Top-down and Bottom-up Strategies	19
2-6. Packing Diagram of ZnO	21
2-7. Illustration of in Situ Preparation Process of PNCs	24
4-1. Acid Hydrolysis Mechanism (a) and Esterification of CNCs (b)	33
4-2. FTIR Spectra of Original Cellulose (a) and CNCs (b)	35
4-3. EDS Spectra of Original Cellulose (a) and CNCs (b)	35
4-4. TEM Images of CNCs	36
4-5. X-ray Spectra of Original Cellulose (a) and CNCs (b)	37
4-6. FESEM Image of CNC	38
4-7. TGA (a) and DTG (b) Thermograms of Original Cellulose (1) and CNCs (2)	39
4-8. Inhibition Zones of CNCs against Gram-positive (a) and Gram-negative 40 (b) Bacteria	40
4-9. FTIR Spectra of CNCs and the CNCs/ZnO Nanocomposites	42
4-10. TEM Images of CNCs/ZnO Nanocomposites (a-d)	43
4-11. XRD Spectra of CNCs and CNCs/ZnO Nanocomposites	44
4-12. FESEM Images of CNCs/ZnO Nanocomposites (a-d)	45
4-13. UV-Absorbance of CNCs and CNCs/ZnO Nanocomposites	46
4-14. TGA Thermograms of CNCs and CNCs/ZnO Nanocomposites	47
4-15. DTG Thermograms of CNCs and CNCs/ZnO Nanocomposites	48
4-16. Inhibition Zone of CNCs/ZnO samples and ZnO against Gram-positive (a) and Gram-negative (b) Bacteria	50
4-17. FTIR Spectra of CNCs and CNCs/ZnO-Ag Nanocomposites	52
4-18. TEM images of C/Z-A <sub>1</sub> (a-a'), C/Z-A <sub>2</sub> (b-b'), C/Z-A <sub>3</sub> (c-c'), C/Z-A <sub>4</sub> (d-d') and 55 C/Z-A <sub>5</sub> (e-e')	55

4-19.	EDS Spectra of C/Z-A <sub>3</sub> Nanocomposites at Circles Marked (a) A, (b) B, (c) C from Figure 4.20 (c')	55
4-20.	XRD Patterns of CNCs and CNCs/ZnO-Ag Nanocomposites	56
4-21.	FESEM Images of CNCs/ZnO-Ag Nanocomposites (a-e)	58
4-22.	UV-vis Absorbance Spectra of CNCs and C/Z-A Nanocomposites	59
4-23.	TGA Thermograms of CNCs and CNCs/ZnO-Ag Nanocomposites	60
4-24.	DTG Thermograms of CNCs/ZnO-Ag Nanocomposites and CNCs	60
4-25.	Inhibition Zones of C/Z-A Samples and ZnO-Ag against Gram-positive (a) and Gram-negative (b) Bacteria	62
4-26.	XRD Pattern of CNCs, PVA/CTS Blend and Its CNCs BNCs	63
4-27.	High Magnification TEM Images of PVA/CTS/CNCs BNCs with 0.5 (a), 1.0 (b), 3.0 (c), and 5.0 (d) wt% CNCs Contents	65
4-28.	Low Magnification TEM Image of PVA/CTS/CNCs BNCs with 0.5wt% CNCs Content	65
4-29.	Tensile Strength and Modulus of PVA/CTS with Different Contents of CNCs	67
4-30.	Elongation at Break of PVA/CTS with Different Contents of CNCs	67
4-31.	TGA Thermograms of PVA/CTS (a) and 0.50(b), 1.0(c), 3.0(d) and 5.0wt% (e) CNCs BNCs	69
4-32.	DTG Thermograms of PVA/CTS (a) and 0.50(b), 1.0(c), 3.0(d) and 5.0wt% (e) CNCs BNCs	69
4-33.	OT plot of PVA/CTS and Its CNCs BNCs Films	71
4-34.	XRD Pattern of CNCs/ZnO, PVA/CTS Blend and Its CNCs/ZnO BNCs	72
4-35.	High Magnification TEM Images of PVA/CTS/CNCs/ZnO BNCs with 1.0(a), 3.0(b), 5.0(c) and 7.0(d) wt% CNCs/ZnO-NPs Contents	74
4-36.	Low Magnification TEM Image of PVA/CTS/CNCs/ZnO BNCs with 7.0wt% CNCs/ZnO-NPs Content	75
4-37.	Tensile Strength and Modulus of PVA/CTS with Different Contents of CNCs/ZnO-NPs	77
4-38.	Elongation at Break of PVA/CTS with Different Contents of CNCs/ZnO-NPs	77
4-39.	TGA Thermograms of the PVA/CTS blend and Its CNCs/ZnO BNCs	79
4-40.	DTG Thermograms of the PVA/CTS blend and Its CNCs/ZnO BNCs	79
4-41.	OT Plot of PVA/CTS Blend and Its CNCs/ZnO BNCs	81
4-42.	UV Absorbance Spectra of PVA/CTS(a), CNCs/ZnO(b) and 1.0(c), 3.0(d), 5.0(e), and 7.0(f) wt% CNCs/ZnO BNCs	82

4-43. Inhibition Zones of PVA/CTS Blend and Its CNCs/ZnO BNCs against Gram-positive (a) and Gram-negative (b) Bacteria	84
4-44. XRD Pattern of CNCs/ZnO-Ag, PVA/CTS Blend and Its CNCs/ZnO-Ag BNCs	85
4-45. High Magnification TEM Images of PVA/CTS/CNCs/ZnO-Ag-NPs BNCs with 1.0(a), 3.0(b), 5.0(c) and 7.0(d) wt% CNCs/ZnO-Ag-NPs Contents	87
4-46. Low Magnification TEM Image of PVA/CTS/CNCs/ZnO-Ag BNCs with 7.0wt% CNCs/ZnO-Ag-NPs Content	88
4-47. Tensile Strength and Modulus of PVA/CTS with Different Contents of CNCs/ZnO-Ag	90
4-48. Elongation at Break of PVA/CTS with Different Contents of CNCs/ZnO-Ag	90
4-49. TGA Thermograms of the PVA/CTS blend and Its CNCs/ZnO-Ag BNCs	92
4-50. DTG Thermograms of the PVA/CTS blend and Its CNCs/ZnO-Ag BNCs	92
4-51. OT Plot of the PVA/CTS Blend and Its CNCs/ZnO-Ag BNCs	94
4-52. UV Absorbance Spectra of the PVA/CTS (a), CNCs/ZnO-Ag NPs(b) and 1.0(c), 3.0(d), 5.0(e) and 7.0(f) wt% CNCs/ZnO-Ag BNCs	95
4-53. Inhibition Zone of CNCs/ZnO-Ag BNCs against Gram-positive (a) and Gram-negative (b) Bacteria	97

## LIST OF ABBREVIATIONS

BCNC	Bacterial cellulose nanocrystal
BNCs	Bio-nanocomposites
CAB	Cellulose acetate butyrate
CCNs	Carboxylate cellulose nanocrystals
CNCs	Cellulose nanocrystals
CMC	Carboxymethyl cellulose
Cs	Chitosan
C/Z	Cellulose nanocrystals/zinc oxide
C/Z-A	Cellulose nanocrystals/zinc oxide-silver
DMF	N,N-dimethylformamide
DMSO	Dimethyl sulfoxide
EDS	Energy dispersive X-ray spectroscopy
e-h	Electron-hole
fcc	Face-center-cubic
FESEM	Field emission scanning electron microscope
FTIR	Fourier transform infrared
GPS	Glycerol plasticized-pea starch
HMDSO	Hexamethyl disiloxane
INPs	Inorganic nanoparticles
LDH	layered double hydroxide
MFC	Microfibrillated cellulose
MMT	Montmorillonite
NCF	Nano-Cellulose fibers
OT	Oxygen transmission
PA6	Polyamide6
PANI	Polyaniline
PBNCs	Polymer Bio-nanocomposites
PCL	Poly(3-caprolactone)
PEO	Polyethylene oxide
PET	Polyethylene terephthalate
PEG	Poly(ethylene glycol)
PHB	Poly(hydroxyl butyrate)
PGA	Poly(glycolic acid)
PLA	Poly(lactic acid)
PLLA	Poly(L-lactide)
PMMA	Poly(methyl methacrylate)
PNC	Polymer Nanocomposites
POE	Polyoxyethylene
Poly(S-co-BuA)	Poly butyl acrylate co-styrene
PS	Polystyrene
PUA	Polyurethane acrylate
PVA	Poly(vinyl alcohol)
PVAc	Poly(vinyl acetate)

SPR	Surface plasmon resonance
T <sub>g</sub>	Glass transition temperature
T <sub>m</sub>	Melting point temperature
T <sub>max</sub>	Maximum degradation temperature
T <sub>onset</sub>	Onset temperature
TEM	Transmission electron microscopy
TGA	Thermogravimetric analysis
TMC	Trimesoylchloride
TPS	Thermoplastic starch
UHMWPE	Ultra-high molecular weight polyethylene
UVA	Ultraviolet-A
WPU	Waterborne polyurethane



## CHAPTER 1

### INTRODUCTION

#### 1.1 General Background of Research

In recent times, using petroleum based plastics has been under attack due to absence of recycling facilities or infrastructure, non-recyclability, non-renewability, non-biodegradability or using toxic additives (Tang and Alavi, 2011). As in 2005, in the USA, 28.9 million tons of plastics packaging was made and only about 5.7% of plastic waste was recycled, 94.3% was dumped in landfill and discarded or combusted into the environment (Marsh and Bugusu, 2007). Recent trends show steady growing interest in the use of biodegradable plastics with increasing accessibility of appropriate materials and owing to social and governmental pressure (Rasato, 2009).

Bio-based polymers often show greater biodegradability and biocompatibility, profiting various applications ranging from packaging to medical devices. However, some factors such as poor mechanical properties, relatively sensitivity to water (Debeaufort *et al.*, 1998), high gas permeability behaviors (Koh *et al.*, 2008) and low heat distortion temperature (Ray and Bousmina, 2005) limit their industrial applications (Zhou and Xanthos, 2009). Therefore, in order to improve and/or control properties of these polymer several methods, such as copolymerization, chain extending or blending these polymers together have been applied by researchers (Kylmä *et al.*, 2001). The use of nano-sized reinforcements to these polymers may open new possibilities for improving not only the chemical and physical properties but also the cost-price-efficiency (Sorrentino *et al.*, 2007).

Polymer nanocomposites are new materials composed of polymers and nano scale inorganic/organic fillers (Gorrasi *et al.*, 2008; Peponi *et al.*, 2009). Nanocomposites can be classified based on the dimensionality of the nano-sized fillers, one can distinguish isodimensional nanoparticles when the all three dimensions are in the nanometer scale, nanotubes or whiskers when two dimensions are on the nanometer range and the third is micrometre, and, lastly, layered minerals, existing in the form of plates of one to a few nanometers thick and hundreds to thousands nanometers in two others dimensions (Alexandre, 2000).

To obtain nanocomposites with acceptable properties, the homogeneous distribution of fillers in the polymer matrices is necessary. The size of fillers and the fillers-matrices interactions are the main parameters affecting the nanocomposite properties (Bianco *et al.*, 2009; Armentano *et al.*, 2009). Advantages of nanocomposites are claimed to be large reinforcement at very small nano-sized fillers content, but functional properties like increased thermal, mechanical, optical and conductivity are often mentioned as well. These materials are useful in various fields such as medical applications, automotive industry, higher performance electronic, magnetic and optical devices manufacturing, packaging industry, etc. (Chae *et al.*, 2005; Raman *et al.*, 2011).

Cellulose nanocrystal (CNC) is one of the emerging renewable materials that has been extensively investigated over the past two decades as a potential nano-sized reinforcement in different polymers (Kvien *et al.*, 2005), particularly within the biopolymer matrices. Cellulose nanocrystals (CNCs) are characteristically rod-formed monocrystals, 2 to 20 nm in diameter and from tens to hundreds of nanometers in length, and extracted after acid hydrolysis of cellulose (Habibi *et al.*, 2010). Some advantages of cellulose nanocrystal are their high aspect ratio, low density, high elastic modulus and strength (Xu *et al.*, 2013; Yu *et al.*, 2013).

Mono-functional nano-sized fillers like CNCs may only enhance a small number of properties of host polymers. To prepare polymer materials with more improved properties, mixtures of various nano-sized fillers are used into polymer matrices.

Inorganic nanoparticles (INPs) are significant kinds of nano-sized fillers which have been successfully utilized in the polymer materials. Inorganic nanoparticles give new properties to the host polymer materials. Nevertheless, the forming of aggregates will significantly decrease inorganic nanoparticles' applicability. How to synthesize inorganic nanoparticles without aggregation during their incorporation into the polymer matrices is a big challenge. The preparation of inorganic nanoparticles is mostly performed through reducing metal salts in the presence of surfactants or polymeric ligands to passivate the cluster surface. Most surfactants and polymeric ligands are prepared from nonrenewable petrochemicals, and finding a renewable biodegradable alternative is essentially important owing to exhausting fossil fuel resources. Cellulose-based materials have been widely used as templates, stabilizers, and carriers in synthesizing metallic nanomaterials (Padalkar *et al.*, 2010; Shinsuke *et al.*, 2009). Considering the functional properties of CNCs and inorganic nanoparticles, the upcoming use of CNCs/INPs nanocomposites as multifunctional nano-sized fillers in polymer matrices is possible.

Among the various types of inorganic nanoparticles, nano-sized zinc oxide (ZnO-NPs) has attained an increased interest and is extensively used in a diversity of applications including functional devices, catalysts, pigments, optical materials, cosmetics, UV-absorbers, and additives in many industrial products ( Kim *et al.*, 2012). Recently, the antimicrobial activity of ZnO nanoparticles with sizes less than 100 nm has been reported (Wang *et al.*, 2012). The use of ZnO-NPs has been considered as a viable solution to stop infectious diseases due to the good antimicrobial properties of these nanoparticles (Stoimenov *et al.*, 2002). Silver has been known to be a bactericide since ancient times. Recently, nanosized silver nanoparticles (Ag-NPs) have been reported to exhibit antimicrobial properties. The outstanding antimicrobial properties of Ag-NPs have led to the development of an extensive diversity of nano-sized silver products, including nano-sized silver-coated wound dressings, contraceptive devices, surgical instruments, and implants (You *et al.*, 2012). For economical and efficient use of ZnO, ZnO nanoparticle composites have been developed and tested for antimicrobial purposes. Additionally, doped silver (Ag) reduced the ionization energy of acceptors in ZnO and thus enhanced the emission (Chen *et al.*, 2011). Therefore, Ag ions can enhance the antimicrobial activity of ZnO. Polymer–inorganic nanoparticle materials with the nanoparticles such as ZnO and Ag-NPs can have the role of antibacterial agents.

## 1.2 Problem Statement

Chitosan is an abundant natural bio-polymer with excellent antimicrobial activity, biocompatibility and non-toxicity. Because of its interesting biological properties, chitosan has long been known and used in pharmaceutical and biomedical applications (Muzzarelli, 2009). Due to its unique bioactivity, the formulation of chitosan with drugs has dual therapeutic outcomes, which make chitosan a new candidate for drug carriers and antimicrobial activity (Muzzarelli *et al.*, 1990). Some synthetic polymers from non-renewable sources are also biodegradable, such as poly(vinyl alcohol). PVA is a water-soluble synthetic polymer, non-toxic with good mechanical properties, film forming, emulsifying, and adhesive properties (No *et al.*, 2007). The blend of PVA/CTS has relatively good physical and chemical properties. Therefore PVA/CTS blend becomes a good candidate to replace petroleum based polymer blends. The PVA/CTS blend has potential applications in the fields of packaging, membrane filtration, biomedical applications, etc. Therefore, scientific studies are needed to improve properties of the polymer blends to produce new products to enhance their performances.

The purpose of this study is to improve the mechanical, thermal, barrier towards oxygen gas, antibacterial, and UV shielding properties of poly(vinyl alcohol)/chitosan (PVA/CTS) blends applications by either the incorporation of mono-functional



reinforcement viz. CNCs or two types of multifunctional fillers viz. CNCs/ZnO-NPs and CNCs/ZnO-Ag-NPs disperse it.

ZnO nanoparticles, alone or coated by other metallic nanoparticles such as silver can be used as filler into polymeric materials with the aim to give UV-shielding, barrier, antibacterial, enhanced mechanical and thermal properties to the obtained nanocomposites, thus making them appropriate for a diversity of target applications such as textile, biomedical and food packaging materials. Although, the formation of particles aggregates significantly decrease their applicability. For preventing from the formation of agglomerated ZnO and ZnO-Ag particles and improve dispersion during their integration into polymer matrix cellulose nanocrystals was used as a stabilizer. Cellulose nanocrystal has plentiful hydroxyl groups on its surface which can absorb metallic ions in the synthesis process. This effect control the size by inhibiting the agglomeration of metallic particles formed in the synthesis procedure. On the other hand, the hydrophilic surface of CNCs allows proper blending with water based host polymer matrices and can provide a homogenous dispersion of inorganic particles into water based polymer. Additionally, cellulose nanocrystal is a potential nano-sized reinforcement with excellent mechanical properties. Considering to these advantages, is expected CNCs, CNCs/ZnO and CNC/ZnO-Ag fillers provide good reinforcing effects into PVA/Cs blends polymer matrices.

To date, there have been no studies based on the preparation of CNCs/ZnO and CNCs/ZnO-Ag nanocomposites, and the use of CNCs, CNCs/ZnO-NPs and CNCs/ZnO-Ag-NPs as nano-sized fillers in the PVA/CTS polymer blends matrices.

### **1.3 Objectives**

The main objectives of this research are:

- 1- To extract CNCs, and prepare CNCs/ZnO and CNCs/ZnO-Ag nano-sized fillers
- 2- To characterize CNCs, CNCs/ZnO and CNCs/ZnO-Ag nano-sized fillers
- 3- To fabricate PVA/CTS/CNCs, PVA/CTS/CNCs/ZnO and PVA/CTS/CNCs/ZnO-Ag bio-nanocomposites
- 4- To characterize and determine properties of PVA/CTS/CNCs, PVA/CTS/CNCs/ZnO and PVA/CTS/CNCs/ZnO-Ag bio-nanocomposites

## REFERENCES

- Abdullah, M., Morimoto, T. and Okuyama, K. Generating blue and red luminescence from ZnO/poly(ethylene glycol) nanocomposites prepared using an in-situ method. *Advanced Functional Materials* 2003; 13: 800–804.
- Ahmad, M.B., Shameli, K., Darroudi, M., Wan Yunus, W.M.Z. and Ibrahim, N.A. Synthesis and characterization of Silver/Clay/Chitosan bio nanocomposites by UV-irradiation method. *American Journal of Applied Sciences* 2009; 6(12): 2030-2035.
- Ahmad, M.B., Shameli, K., Wan Yunus, W.M.Z., Ibrahim, N.A. and Darroudi, M. Synthesis and characterization of Silver/Clay/Starch bio nanocomposites by green method. *Australian Journal of Basic and Applied Sciences* 2010; 4(7): 2158-2165.
- Alexandre, M. and Dubois, P. Polymer-layered silicate nanocomposites: preparation, properties and uses of a new class of materials. *Materials Science and Engineering: R: Reports* 2000; 28: 1–63.
- Alexandrescu, R., Morjan, I., Dumitrache, F., Scarisoreanu, M., Fleaca, C.T. et al. Development of TiO<sub>2</sub> and TiO<sub>2</sub>/Fe-based polymeric nanocomposites by single-step laser pyrolysis. *Applied Surface Science* 2013; 278: 305–312.
- Anz˘lovar, A., Crnjak Orel, Z. and Z˘igon, M. Poly(methyl methacrylate) composites prepared by in situ polymerization using organophilic nano-to-submicrometer zinc oxide particles. *European Polymer Journal* 2010; 46: 1216–1224.
- Araki, J., Wada, M., Kuga, S. and Okano, T. Flow properties of microcrystalline cellulose suspension prepared by acid treatment of native cellulose. *Colloids and Surfaces A* 1998; 42(1): 75–82.
- Aranguren, M.I., Marcovich, N.E., Salgueiro, W. and Somoza, A. Effect of the nano-cellulose content on the properties of reinforced polyurethanes. A study using mechanical tests and positron annihilation spectroscopy. *Polymer Testing* 2013; 32: 115–122.

- Armentano, I., Del Gaudio, C., Bianco, A., Dottori, M., Nanni, F., Fortunati, E., et al. Processing and properties of poly(e-caprolactone)/carbon nanofibre composite mats and films obtained by electrospinning and solvent casting. *Journal Material Science* 2009; 44: 4789-95.
- Armentano, I., Dottori, M., Fortunati, E., Mattioli, S. and Kenny, J.M. Biodegradable polymer matrix nanocomposites for tissue engineering: A review. *Polymer Degradation and Stability* 2010; 95: 2126-2146.
- Aulin, C., Ahola, S., Josefsson P, et al. Nanoscale cellulose films with different crystallinities and mesostructures their surface properties and interaction with water. *Langmuir* 2009; 25(13): 7675–7685.
- Azeredo, H.M., Mattoso, L.H., Avena-Bustillos, R.J.A., Filho, G.C., Munford, M. L., Wood, D., et al. Nanocellulose reinforced chitosan composite films as affected by nanofiller loading & plasticizer content. *Journal of Food Science* 2010; 75: 19–28.
- Azizi Samir, M.A.S., Alloin, F., Paillet, M. and Dufresne, A. Tangling effect in fibrillated cellulose reinforced nanocomposites. *Macromolecules* 2004a; 37: 4313-4316.
- Azizi Samir, M.A.S., Alloin, F. and Dufresne, A. Review of recent research into cellulosic whiskers, their properties and their application in nanocomposite field. *Biomacromolecules* 2005; 6: 612-626.
- Azizi Samir, M.A.S., Alloin, F., Sanchez, J.Y., El Kissi, N. and Dufresne, A. Preparation of cellulose whiskers reinforced nanocomposites from an organic medium suspension. *Macromolecules* 2004b; 37: 1386-1393.
- Azizi Samir, M.A.S., Alloin, F., Sanchez, J.Y. and Dufresne, A. Cellulose nanocrystals reinforced poly(oxyethylene). *Polymer* 2004c; 45: 4149-4157.
- Beck-Candanedo, S., Roman, M. and Gray, D. Effect of conditions on the properties behavior of wood cellulose nanocrystals suspensions. *Biomacromolecules* 2005; 6: 1048-1054.

- Bendahou, A., Habibi, Y., Kaddami, H. and Dufresne, A. Physico-chemical characterization of palm from Phoenix Dactylifera-L, preparation of cellulose whiskers and natural rubber-based nanocomposites. *Journal of Biobased Materials and Bioenergy* 2009; 3: 81-90.
- Bianco, A., Di Federico, E., Moscatelli, I., Camaioni, A., Armentano, I., Campagnolo, L., et al. Electrospun poly(3-caprolactone)/Ca-deficient hydroxyapatite nanohybrids: microstructure, mechanical properties and cell response by murine embryonic stem cells. *Materials Science and Engineering:C* 2009; 29: 2063-71.
- Bondeson, D., Mathew, A. and Oksman, K. Optimization of the isolation of nanocrystals from microcrystalline cellulose by acid hydrolysis. *Cellulose* 2006; 13: 171-180.
- Bonini, C., Heux, L., Cavaille, J.Y., Lindner, P., Dewhurst, C. and Terech, P. Rodlike cellulose whiskers coated with surfactant: A small-angle neutron scattering characterization. *Langmuir* 2002; 18: 3311-3314.
- Božanic, D.K., Trandafilovi, L.V., Luyt, A.S. and Djokovi, V. Green synthesis and optical properties of silver-chitosan complexes and nanocomposites. *Reactive & Functional Polymers* 2010; 70: 869–873.
- Brännvall, E. 2007. *Aspect on Strength Delivery and Higher Utilisation of Strength Potential of Soft Wood Kraft Pulp Fibres*, PhD Thesis, KTH, Royal Institute of Technology: Stockholm, Sweden.
- Bras, J., Viet, D., Bruzzese, C. and Dufresne, A. Correlation between stiffness of sheets prepared from cellulose whiskers and nanoparticles dimensions. *Carbohydrate Polymers* 2011; 84: 211–215.
- Brayner, R., Ferrari-Iliou, R., Brivois, N., Djediat, S., Benedetti, M.F. and Fievet, F. Toxicological impact studies based on escherichia coli bacteria in ultrafine ZnO nanoparticles colloidal medium. *Nano Letters* 2006; 6: 866–70.
- Cakır, B.A., Budama, L., Topel, Ö. and Hoda, N. Synthesis of ZnO nanoparticles using PS-b-PAA reverse micelle cores for protective, self-cleaning and antibacterial

textile applications. *Colloids and Surfaces A: Physicochemical and Engineering Aspects* 2012; 414: 132– 139.

Cao, X., Chen, Y., Chang, P.R., Muir, A.D. and Falk, G. Starch-based nanocomposites reinforced with flax cellulose nanocrystals. *eXPRESS Polymer Letters* 2008b; 2(7): 502-510.

Cao, X., Chen, Y., Chang, P.R., Stumborg, M. and Huneault, M.A. Green composites reinforced with hemp nanocrystals in plasticized starch. *Journal of Applied Polymer Science* 2008a; 109: 3804-3810.

Cao, X., Dong, H. and Li, C.M. New nanocomposite materials reinforced with flax cellulose nanocrystals in waterborne polyurethane. *Biomacromolecules* 2005; 8(3): 899-904.

Cao, X., Habibi, Y. and Lucia, L.A. One-pot polymerization, surface grafting, and processing of waterborne polyurethane-cellulose nanocrystal nanocomposites. *Journal of Materials Chemistry* 2009; 19: 7137–7145.

Capadona, J.R., Shanmuganathan, K., Tyler, D.J., Rowan, J. and Weder, C. Stimuli-responsive polymer nanocomposites inspired by the sea cucumber dermis. *Science* 2008; 319: 1370-1374.

Chae, H., Young Bum, L., Jin Woo, B., Jae Young, J., Byeong Uk, N. and Tae Won, H. Preparation and mechanical properties of polypropylene/clay nanocomposites for automotive parts application. *Journal of Applied Polymer Science* 2005; 98(1): 427-433.

Chang, B.P., Akil, H.M. and Nasir, R.B.M. Comparative study of micro- and nano-ZnO reinforced UHMWPE composites under dry sliding wear. *Wear* 2013; 297: 1120–1127.

Chen, C.H., Wang, F.Y., Mao, C.F. and Yang, C.H.I. Preparation and characterization of chitosan/poly(vinyl alcohol) blend films. *Journal of Applied Polymer Science* 2007; 105: 1086–1092.

- Chen, C.H., Wang, F.Y., Mao, C.F., Liao, W.T. and Hsieh, C.D.I. Studies of chitosan: II. Preparation and characterization of chitosan/poly(vinyl alcohol)/gelatin ternary blend films. *International Journal of Biological Macromolecules* 2008; 43: 37-42.
- Chen, D. 2012. *Biocomposites Reinforced With Cellulose Nanocrystals Derived From Potato Peelwaste*. Master Thesis, McMaster University.
- Cheng, A. and Samulski, E.T. Hydrothermal synthesis of one-dimensional ZnO nanostructures with different aspect ratios. *Chemical Communications* 2004; 8: 986–987.
- Chen, R.Q., Zou, C.W., Bian, J. M., Sandhu, A. and Gao, W. Microstructure and optical properties of Ag-doped ZnO nanostructures prepared by a wet oxidation doping process. *Nanotechnology* 2011; 22: 105706–105713.
- Chen, S.J., Liu, Y.C., Shao, C.L., Mu, R., Lu, Y.M., Zhang, J.Y., et al. Structural and optical properties of uniform ZnO nanosheets. *Advanced Materials* 2005; 17: 586–90.
- Chen, Z.G., Mo, X.M. and Qing, F.L. Electrospinning of collagen–chitosan complex. *Material Letters* 2007; 61: 3490–3494.
- Costa-Júnior, E.S., Barbosa-Stancioli, E.F., Mansur, A.A.P., Vasconcelos, W.L. and Mansur, H.S. Preparation and characterization of chitosan/poly(vinyl alcohol) chemically crosslinked blends for biomedical applications. *Carbohydrate Polymers* 2009; 76: 472–481.
- Crini, G. Non-conventional low-cost adsorbents for dye removal: a review. *Bioresources Technology* 2006; 97: 1061–1085.
- Cui, F., Liang, S., Zhang, J., Han, Y., Lu, Ch., Cuiab, T. and Yang. B. Formation of nanoparticles in solid-state matrices: a strategy for bulk transparent TiO<sub>2</sub>-polymer nanocomposites. *Polymer Chemistry* 2012; 3: 3296-3300.
- Dahiya, J.B. and Rana, S. Thermal degradation and morphological studies on cotton cellulose modified with various arylphosphorodichloridites. *Polymer International* 2004; 53: 995–1002.

- Damm, C., Munstedt, H. and Rosch, A. The antimicrobial efficacy of polyamide 6/silver-nano- and microcomposites. *Materials Chemistry and Physics* 2008; 108: 61–66.
- Daniel, J.R. 1985. Cellulose structure and properties. In *Encyclopedia of Polymer Science and Engineering*; Kroschwitz, J.I., ed. Wiley-Interscience Publication John Wiley & Sons, Volume 3, pp. 86-123 New York: NY, USA.
- Darder, M., Aranda, P. and Ruiz-Hitzky, E. Bionanocomposites: A new concept of ecological, bioinspired, and functional hybrid materials. *Advanced Materials* 2007; 19: 1309-1319.
- Debeaufort, F., Quezada-Gallo, J.A. and Voilley, A. Edible films and coatings: Tomorrow's packagings. *Critical Reviews in Food Science and Nutrition* 1998; 38: 299-313.
- De, S.L.M.M., Wong, J.T., Paillet, M., Borsali, R. and Pecora, R. Translational and rotational dynamics of rodlike cellulose whiskers. *Langmuir* 2003; 19: 24-29.
- Dinand, E., Vignon, M., Chanzy, H. and Heux, L. Mercerization of primary wall cellulose and its implication for the conversion of cellulose I → cellulose II. *Cellulose* 2002; 9: 7-18.
- Dong, H., Strawhecker, K.E., Snyder, J.F., Orlicki, J.A., Reiner, R.S. and Rudie, A.W. Cellulose nanocrystals as a reinforcing material for electrospun poly(methylmethacrylate) fibers: formation, properties and anomechanical characterization. *Carbohydrate Polymers* 2012; 87: 2488–2495.
- Drogat, N., Granet, R., Sol, V., Memmi, A., Saad, N., Koerkamp, C.K., Bressollier, P. and Krausz, P. Antimicrobial silver nanoparticles generated on cellulose nanocrystals. *Journal of Nanoparticle Research* 2011; 13: 1557–1562.
- Duan, B., Dong, C.H., Yuan, X.Y. and Yao, K.D. Electrospinning of chitosan solutions in acetic acid with poly(ethylene oxide). *Journal of Biomaterials Science, Polymer Edition Brill* 2004; 15: 797–811.

- Dufresne, A. Cellulose-based composites and nanocomposites. In monomers, polymers and composites from renewable resources. *1st ed.*; Gandini, A., Belgacem, M.N., Eds.; Elsevier: Oxford, UK, 2008a; 401-418.
- Dufresne, A. Polysaccharide nanocrystal reinforced nanocomposites. *Canadian Journal of Chemistry* 2008b; 86: 484–494.
- Dufresne, A. Processing of polymer nanocomposites reinforced with polysaccharide nanocrystals. *Molecules* 2010; 15: 4111-4128.
- Ebeling, T., Paillet, M., Borsali, R., Diat, O., Dufresne, A. et al., Shear-induced orientation phenomena in suspensions of cellulose microcrystals, revealed by small angle X-ray scattering. *Langmuir* 1999; 15(19): 6123.
- Eichhorn, S.J., Defense, A., Aranguren, M., Marcovich, N.E., Capadona, J.R., Rowan, S.J. et al. Review: current international research into cellulose nanofibres and nanocomposites. *Journal of Materials Science Letters* 2010; 45: 1–33.
- Elen, K., Murariu, M., Peeters, R., Dubois, Ph., Mullens, J., et al. Towards high-performance biopackaging: barrier and mechanical properties of dual-action polycaprolactone/zinc oxide nanocomposites. *Polymers for Advanced Technologies* 2012; 23: 1422–1428.
- Fang, M., Chen, J.H., Xu, X.L., Yang, P.H., and Hildebrand, H.F. Antibacterial activities of inorganic agents on six bacteria associated with oral infections by two susceptibility tests. *International Journal of Antimicrobial Agents* 2006; 27: 513-517.
- Favier, V., Canova, G.R., Cavallé, J.Y., Chanzy, H., Dufresne, A. and Gauthier, C. Nanocomposite materials from latex and cellulose whiskers. *Polymer Advance Technology* 1995a; 6: 351-355.
- Favier, V., Chanzy, H. and Cavaille, J.Y. Polymer nanocomposites reinforced by cellulose whiskers. *Macromolecules* 1995b; 28: 6365–6367.



- Feris, K., Caitlin, O., Tinker, J., Wingett, D., Punnoose, A., Thurber, A., et al. Electrostatic interactions affect nanoparticle-mediated toxicity to the Gram-negative bacterium *Pseudo monasaeruginosa* PAO1. *Langmuir* 2010; 26: 4429-4436.
- Fortunati, E., Armentano, I., Zhou, Q., Puglia, D., Terenzi, A., Berglund, L.A., et al. Microstructure and non-isothermal cold crystallization of PLA composites based on silver nanoparticles and nanocrystalline cellulose. *Polymer Degradation and Stability* 2012b; 97(10): 2027-2036.
- Fortunati, E., Peltzer, M., Armentano, I., Torre, L., Jiménez, A. and Kenny, J.M. Effects of modified cellulose nanocrystals on the barrier and migration properties of PLA nano-biocomposites. *Carbohydrate Polymers* 2012a; 90: 948-956.
- Garcia, D.R.N.L., Thielemans, W. and Dufresne, A. Sisal cellulose whiskers reinforced polyvinyl acetate nanocomposites. *Cellulose* 2006; 13: 261-270.
- George, J., Ramana, K.V., Bawa, A.S. and Siddaramaiah. Bacterial cellulose nanocrystals exhibiting high thermal stability and their polymer nanocomposites. *International Journal of Biological Macromolecules* 2011; 48: 50-57.
- Ghosh chaudhuri, R. and Paria, S. Core/shell nanoparticles: classes, properties, synthesis mechanisms, characterization, and applications. *Chemical Reviews* 2012; 112: 2373-2433.
- Gorrasi, G., Vittoria, V., Murariu, M., Ferreira, A.D.S., Alexandre, M. and Dubois, P. Effect of filler content and size on transport properties of water vapor in PLA/calcium sulfate composites. *Composite Science Technology* 2008; 9: 627-32.
- Gunatillake, P., Mayadunne, R., Adhikari, R. and El-Gewely, M.R. Recent developments in biodegradable synthetic polymers. *Biotechnology Annual Review* 2006; 12: 301-347.
- Grunert, M. and Winter, W.T. Nanocomposites of cellulose acetate butyrate reinforced with cellulose nanocrystals. *Journal of Polymers and the Environment* 2002; 10: 27-30.

- Habibi, Y. and Dufresne, A. Highly filled bionanocomposites from functionalized polysaccharides nanocrystals. *Biomacromolecules* 2008; 9: 1974–1980.
- Habibi, Y., Goffin, A.L., Schiltz, N., Duquesne, E., Dubois, P. and Dufresne, A. Bionanocomposites based on Poly(Epsilon-Caprolactone)-grafted cellulose nanocrystals by ring-opening polymerization. *Journal of Materials Chemistry* 2008; 18: 5002-5010.
- Habibi, Y., Lucia, L.A. and Rojas, O.J. Cellulose nanocrystals: chemistry, self-assembly, and applications. *Chemical Reviews* 2010; 110: 3479–3500.
- Heinlaan, M., Ivask, A., Blinova, I., Dubourguier, H.C. and Kahru, A. Toxicity of nanosized and bulk ZnO, CuO and TiO<sub>2</sub> to bacteria vibrio fischeri and crustaceans daphnia magna and thamocephalus platyurus. *Chemosphere* 2008; 71: 1308–16.
- He, J.H., Kunitake, T. and Nakao, A. Facile in situ synthesis of noble metal nanoparticles in porous cellulose fibers. *Journal of Materials Chemistry* 2003; 15: 4401-4406.
- He, J.H., Lao, C.S., Chen, L.J., Davidovic, D. and Wang, Z.L. Large-scale Ni-doped ZnO nanowire arrays and electrical and optical properties. *Journal of the American Chemical Society* 2005; 127: 16376–16377.
- He, J., Shao, W., Zhang, L., Deng, C. and Li, Ch. Crystallization behavior and UV-protection property of PET-ZnO nanocomposites prepared by in situ polymerization. *Wiley Inter Science* 2009; 114: 1303–1311.
- Hem, R.P., Bishweshwar, P., Han, J.K., Altangerel, A., Chan, H.P. et al. A green and facile one-pot synthesis of Ag-ZnO/RGO nanocomposite with effective photocatalytic activity for removal of organic pollutants. *Ceramics International* 2013; 39: 5083–5091.
- Herrick, F.W., Casebier, R.L., Hamilton, J.K. and Sandberg, K.R. Microfibrillated cellulose: morphology and accessibility. *Journal of Applied Polymer Science* 1983; 37: 797-813.

- Hirai, A., Inui, O., Horii, F. and Tsuji, M. Phase separation behavior in aqueous suspensions of bacterial cellulose nanocrystals prepared by sulfuric acid treatment. *Langmuir* 2009; 25: 497-502.
- Hidayat, D., Ogi, T., Iskandar, F. and Okuyama, K. Single crystal ZnO:Al nanoparticles directly synthesized using low-pressure spray pyrolysis. *Materials Science and Engineering:B* 2008; 151(3): 231-237.
- Hon, D.N.S. and Shiraishi, N. Wood and cellulosic chemistry. *New York: M.Dekker* 1991.
- Hong, R.Y., Li, J.H., Chen, L.L., Liu, D.Q., Li, H.Z., Ding, J. and Zheng, Y. Synthesis, surface modification and photocatalytic property of ZnO nanoparticles. *Powder Technology* 2009; 189: 426-432.
- Hosokawa, M., Nogi, K., Naito, M. and Yokoyama, T. 2007. *Nanoparticle Technology Handbook*, First Ed, Elsevier B.V., AE Amsterdam.
- Hubbe, M.A., Rojas, O.J., Lucia, L.A. and Sain, M. Cellulosic nanocomposites: a review. *Bioresource* 2008; 3: 929-980.
- Hunne, M.A., Rojas, O.J., Lucia, L.A. and Sain, M. Cellulosic nanocomposites: a review. *Bioresources* 2008; 3: 929-980.
- Hyder, M.N. and Chen, P. Pervaporation dehydration of ethylene glycol with chitosan-poly (vinyl alcohol) blend membranes: effect of CS-PVA blending ratios. *Journal of Membrane Science* 2009; 340 (1): 171-180.
- Iwamoto, S., Nakagaito, A.N. and Yano, H. Nano-fibrillation of pulp fibers for the processing of transparent nanocomposites. *Applied Physics A: Materials Science & Processing* 2007; 89: 461-466.
- Jagannatha, A., Reddy, M.K., Kokila, H., Nagabhushana, J.L., Rao, C., et al. Combustion synthesis, characterization and raman studies of ZnO nanopowders. *Spectrochimica Acta Part A* 2011; 81: 53-58.

- Jegal, J. and Lee, K. Nanofiltration membranes based on poly(vinyl alcohol) and ionic polymers. *Journal of Applied Polymer Science* 1999; 72: 1755-1762.
- Jin, T., Sun, D., Su, J.Y., Zhang, H. and Sue, H.J. Antimicrobial efficacy of zinc oxide quantum dots against *Listeria monocytogenes*, *Salmonella enteritidis*, and *Escherichia coli* O157:H7. *Journal of Food Science* 2009; 74(1): 46-52.
- Jones, N., Ray, B., Ranjit, T.K. and Manna, C.A. Antibacterial activity of ZnO nanoparticle suspensions on a broad spectrum of microorganisms. *FEMS Microbiology Letters* 2008; 279(1): 71-76.
- Joni, I.M., Purwanto, A., Iskandar, F., Hazata, M. and Okuyam, K. Intense UV-light absorption of ZnO nanoparticles prepared using a pulse combustion-spray pyrolysis method. *Chemical Engineering Journal* 2009; 155: 433-441.
- Kanagaraj, S., Varanda, F.R., Zhil'tsova, T.V., Oliveira, M.S. and Simões, J.A.O. Mechanical properties of high density polyethylene/carbon nanotube composites. *Composites Science and Technology* 2007; 67(15-16): 3071-3077.
- Kannusamy, P. and Sivalingam, Th. Chitosan ZnO/polyaniline hybrid composites: polymerization of aniline with chitosan ZnO for better thermal and electrical property. *Polymer Degradation Stability* 2013; 98: 988-996.
- Karunakaran, C., Rajeswari, V. and Gomathisankar, P. Optical, electrical, photocatalytic, and bactericidal properties of microwave synthesized nanocrystalline Ag-ZnO and ZnO. *Solid State Sciences* 2011; 13: 923-928.
- Kelnar, I., Rotrekl, J., Kotek, J., Kaprálková, L. and Hromádková, J. Effect of montmorillonite on structure and properties of nanocomposite with PA6/PS/elastomer matrix. *European Polymer Journal* 2009; 45: 2760-2766.
- Khorsand zak, A., Ebrahimzadeh abrishami, M., Majid, W.H., Yousefi, R. and Hosseini, S.M. Effects of annealing temperature on some structural and optical properties of ZnO nanoparticles prepared by a modified sol-gel combustion method. *Ceramics International* 2011; 37: 393-398.

- Kikuchi, Y., Sunada, K., Iyoda, T., Hashimoto, K. and Fujishima, A. Photocatalytic bactericidal effect of TiO<sub>2</sub> thin films: Dynamic view of the active oxygen species responsible for the effect. *Journal of Photochemistry and Photobiology A: Chemistry* 1997; 106: 51–56.
- Kim, D., Jeon, K., Lee, Y., Seo, J., Seo, K., Han, H. and Khan, Sh.B. Preparation and characterization of UV-cured polyurethane acrylate/ZnO nanocomposite films based on surface modified ZnO. *Progress in Organic Coatings* 2012; 74: 435– 442.
- Kim, H.J., Pant, R.H., Amarjargal, A. and Kim, Ch.S. Incorporation of silver-loaded ZnO rods into electrospun nylon-6spider-web-like nanofibrous mat using hydrothermal process. *Colloids and Surfaces A: Physicochemical and Engineering Aspects* 2013; 434: 49–55.
- Klemm, D., Schumann, D., Kramer, F., Hessler, N., Hornung, M., Schmauder, H.P. and Marsch, S. Nanocelluloses as innovative polymers in research and application. *Polysaccharides* 2006; 205:49–96.
- Koh, H.C., Park, J.S., Jeong, M.A., Hwang, H.Y., Hong, Y.T., et al. Preparation and gas permeation properties of biodegradable polymer/layered silicate nanocomposite membranes. *Desalination* 2008; 233: 201-209.
- Koehler, W.S. and Griffith, L.G. Osteoblast response to PLGA tissue engineering scaffolds with PEO modified surface chemistries and demonstration of patterned cell response. *Biomaterials* 2004; 25: 2819-30.
- Kojima, Y., Usuki, A., Kawasumi, M., Okada, A., Kurauchi, T. and Kamigaito, O. Synthesis of nylon-6-clay hybrid by montmorillonite intercalated with ε-caprolactam. *Journal of Polymer Science Part A: Polymer Chemistry* 1993; 31: 983-986.
- Kuila, T., Srivastava, S.K., Bhowmick, A.K. and Saxena, A.K. Thermoplastic polyolefin based polymer–blend-layered double hydroxide nanocomposites. *Composites Science and Technology* 2008; 68: 3234–3239.

- Kumar, A., Vemula, P.K., Ajayan, P.M. and John, G. Silver-nanoparticle-embedded antimicrobial paints based on vegetable oil. *Nature Materials* 2008; 7: 236–241.
- Kumar, A.P., Depan, D., Tomer, N. S. and Singh, R. P. Nanoscale particles for polymer degradation and stabilization-Trends and future perspectives. *Progress in Polymer Science*, 2009a; 34(6): 479–515.
- Kumar, K., Tripathi, B.P. and Shahi, V.K. Crosslinked chitosan/polyvinyl alcohol blend beads for removal and recovery of Cd(II) from wastewater. *Journal of Hazardous Materials* 2009b; 172: 1041–1048.
- Kuo, S.T., Tuan, W.H., Shieh, J. and Wang, S.F. Effect of Ag on the micro structure and electrical properties of ZnO. *Journal of the European Ceramic Society* 2007; 27(16): 4521-4527.
- Kvien, I., Tanem, B.S. and Oksman, K. Characterization of cellulose whiskers and their nanocomposites by atomic force and electron microscopy. *Biomacromolecules* 2005; 6: 3160–3165.
- Kylmä, J., Tuominen, J., Helminen, A. and Seppälä, J. Chain extending of lactic acid oligomers. Effect of 2, 2'-bis(2-oxazoline) on 1,6-hexamethylene diisocyanate linking reaction. *Polymer* 2001; 42(8): 3333-3343.
- Lee, Ch.J., Karim, M.R., Vasudevan, T., Kim, H.J., Raushan, K., et al. A comparison method of silver nanoparticles prepared by the gamma irradiation and in situ reduction methods. *The Bulletin of the Korean Chemical Society* 2010; 31: 1993–1996.
- Lee, J., Bhattacharyya, D., Easteal, A.J. and Metson, J.B. Properties of nano-ZnO/poly(vinyl alcohol)/poly(ethylene oxide) composite thin films. *Current Applied Physics* 2008; 8: 42–47.
- Lewandowska, K. Miscibility and thermal stability of poly(vinyl alcohol)/chitosan mixtures. *Thermochimica Acta* 2009; 493: 42-48.

- Liang, S., Liu, L., Huang, Q. and Yam, K.L. Preparation of single or double-network chitosan/poly(vinyl alcohol) gel films through selectively cross-linking method. *Carbohydrate Polymers* 2009; 77: 718–724.
- Liaw, W. and Chen, K. Preparation and characterization of poly(imide siloxane) (PIS)/titania (TiO<sub>2</sub>) hybrid nanocomposites by sol–gel processes. *European Polymer Journal* 2007; 43(6): 2265–2278.
- Liepins, R., and Pearce, E.M. Chemistry and toxicity of flame retardants for plastics. *Environmental Health Perspectives*, 1976; 17: 55- 63.
- Li, J., Lu, X.L. and Zheng, Y.F. Effect of surface modified hydroxyapatite on the tensile property improvement of HA/PLA composite. *Applied Surface Science* 2008; 255: 494-497.
- Lin, X., Zhou, R., Zhang, J. and Fei, S. A novel one-step electron beam irradiation method for synthesis of Ag/Cu<sub>2</sub>O nanocomposites. *Applied Surface Science* 2009; 256: 889–893.
- Liu, H., Liu, D., Yao, F. and Wu, Q. Fabrication and properties of transparent polymethylmethacrylate/cellulose nanocrystals composites. *Bioresource Technology* 2010; 101: 5685–5692.
- Liu, H., Wang, D., Song, Z. and Shang, S. Preparation of silver nanoparticles on cellulose nanocrystals and the application in electrochemical detection of DNA hybridization. *Cellulose* 2011; 18: 67-74.
- Liu, H., Song, J., Shang, Sh., Song, Zh. and Wang, D. Cellulose nanocrystal/silver nanoparticle composites as bifunctional nanofillers within waterborne polyurethane. *Applied Materials & Interfaces* 2012; 4: 2413–2419.
- Liu, H., Cui, Sh., Shang, Sh., Wang, D. and Song, J. Properties of rosin-based waterborne polyurethanes/cellulose nanocrystals composites. *Carbohydrate Polymers* 2013; 96: 510– 515.

- Liu, J.P., Huang, X.T., Li, Y.Y., Duan, J.X. and Ai, H.H. Large-scale synthesis of flower-like ZnO structures by a surfactant-free and low-temperature process. *Materials Chemistry and Physics* 2006; 98: 523–527.
- Liu, Sh., Zhang, L., Zhou, J. and Wu, R. Structure and properties of cellulose/Fe<sub>2</sub>O<sub>3</sub> nanocomposite fibers spun via an effective pathway. *The Journal of Physical Chemistry C* 2008; 112: 4538-4544.
- Liu, Y. and Kim, H.II. Characterization and antibacterial properties of genipin-crosslinked chitosan/poly(ethylene glycol)/ZnO/Ag nanocomposites. *Carbohydrate Polymers* 2012; 89: 111–116.
- Liu, Y., He, L., Mustapha, A., Li, H., Hu, Z.Q. and Lin, M. Antibacterial activities of zinc oxide nanoparticles against Escherichia coli O157:H7. *Journal of Applied Microbiology* 2009; 107(4): 1193-1201.
- Li, X., Wang, D., Cheng, G., Luo, Q., An, J. and Wang, Y. Preparation of polyaniline modified TiO<sub>2</sub> nanoparticles and their photocatalytic activity under visible light illumination. *Applied Catalysis B: Environmental* 2008; 81(3–4): 267–273.
- Li, Y.Q., Fu, S.Y. and Mai, Y.W. Preparation and characterization of transparent ZnO epoxy nanocomposites with high-UV shielding efficiency. *Polymer* 2006; 47: 2127–2132.
- Ljungberg, N., Bonini, C., Bortolussi, F., Boisson, C., Heux, L. and Cavaille, J.Y. New nanocomposite material reinforced with cellulose whiskers in atactic polypropylene: Effect of surface and dispersion characteristics. *Biomacromolecules* 2005; 6: 2732-2739.
- Ljungberg, N., Cavaille, J.Y. and Heux, L. Nanocomposites of isotactic polypropylene reinforced with rod-like cellulose whiskers. *Polymer* 2006; 47: 6285–6292.
- Lu, H., Hu, Y., Li, M., Chen, Z. and Fan, W. Structure characteristics and thermal properties of silane-grafted-polyethylene/clay nanocomposite prepared by reactive extrusion. *Composite Science and Technology* 2006; 66: 3035–3039.
- Lu, J., Askeland, P. and Drzal, L.T. Surface modification of microfibrillated cellulose for epoxy composite applications. *Polymer* 2008a; 49: 1285-1298.



- Lu, P. and Hsieh, Y.L. Preparation and properties of cellulose nanocrystals: Rods, spheres, and network. *Carbohydrate Polymers* 2010; 82: 329–336.
- Lu, W., Liu, G., Gao, S., Xing, S. and Wang, J. Tyrosine-assisted preparation of Ag/ZnO nanocomposites with enhanced photocatalytic performance and synergistic antibacterial activities. *Nanotechnology* 2008b; 19: 1-10.
- Lu, Y. and Chen, S.C. Micro and nano-fabrication of biodegradable polymers for drug delivery. *Advanced Drug Delivery Reviews* 2004; 56(11): 1621-1633.
- Lu, Y.S., Weng, L.H. and Cao, X.D. Biocomposites of plasticized starch reinforced with cellulose crystallites from cottonseed linter. *Macromolecular Bioscience* 2005; 5: 1101-1107.
- Ly, B., Thielemans, W., Dufresne, A., Chaussy, D. and Belgacem, M.N. Surface functionalization of cellulose fibres and their incorporation in renewable polymeric matrices. *Composite Science Technology* 2008; 68: 3193-3201.
- Ma, H., Williams, P.L. and Diamond, S.A. Ecotoxicity of manufactured ZnO nanoparticles - A review. *Environmental Pollution* 2013; 172: 76-85.
- Ma, X., Chang, P.R., Yang, J., Yu, J. Preparation and properties of glycerol plasticized-peastarch/zinc oxide-starch bionanocomposites. *Carbohydrate Polymers* 2009; 75: 472–478.
- Ma, X.Y. and Zhang, W.D. Effects of flower-like ZnO nanowhiskers on the mechanical, thermal and antibacterial properties of waterborne polyurethane. *Polymer Degradation and Stability* 2009; 94: 1103–1109.
- Mahmoud, Kh.A., Male, K.B., Hrapovic, S. and Luong, J.H.T. Cellulose nanocrystal/gold nanoparticle composite as a matrix for enzyme immobilization. *ACS Applied Materials & Interfaces* 2009; 1(7):1383-6.

- Mangalam, A.P., Simonsen, J. and Benight, A. Cellulose/DNA hybrid nanomaterials. *Biomacromolecules*, 2009; 10: 497-504.
- Man, Z., Muhammad, N., Sarwono, A., Bustam, M. A., Kumar, M.V. and Rafiq, S. Preparation of cellulose nanocrystals using an ionic liquid. *Journal of Polymers and the Environment* 2011; 19: 726–731.
- Mao, Zh., Shi, Q., Zhang, L. and Cao, H. The formation and UV-blocking property of needle-shaped ZnO nanorod on cotton fabric. *Thin Solid Films* 2009; 517: 2681–2686.
- Mansur, H.S. and Costa, H.S. Nanostructured poly(vinyl alcohol)bioactive glass and poly(vinyl alcohol)/Chitosan/bioactive glass hybrid scaffold for biomedical application. *Chemical Engineering Journal* 2008; 137: 72–83.
- Marchessault, R.H., Morehead, F.F. and Koch, M.J. Hydrodynamic properties of neutral suspensions of cellulose crystallites as related to size and shape. *Journal of Colloid Science* 1961; 16: 327–344.
- Martino, V. P., Ruseckaite, R. A., Jiménez, A. and Averous, L. Correlation between composition structure and properties of poly(lactic acid)/polyadipate-based nanobiocomposites. *Macromolecular Materials and Engineering* 2010; 295:551–558.
- Marsh, K. and Bugusu, B. Food packaging: roles, materials, and environmental issues. *Journal of Food Science* 2007; 72: 39–55.
- Mathew, J. and Kodama, M. Study of blood compatible polymers I. modification of poly (vinyl alcohol). *Polymer Journal* 1992; 24: 31-41.
- Mathew, A.P., Chakraborty, A., Oksman, K. and Sain, M. In Cellulose Nanocomposites: processing, characterization, and properties; Oksman, K., Sain, M., Eds.; ACS Symposium Series 938; American Chemical Society: Washington, DC 2006.
- Moller, T. and Sylvester, P. Effect of silica and pH on arsenic uptake by resin/iron oxide hybrid media. *Water resistance* 2008; 42: 1760–1766.

Morin, A. and Dufresne, A. Nanocomposites of chitin whiskers from riftia tubes and poly(caprolactone). *Macromolecules* 2002; 35: 2190-2199.

Mo, Z.L., Zhao, Z.L., Chen, H., Niu, G.P. and Shi, H.F. Heterogeneous preparation of cellulose-polyaniline conductive composites with cellulose activated by acids and its electrical properties. *Carbohydrate Polymer* 2009; 75(4): 660–664.

Mulvaney, P. Surface plasmon spectroscopy of nanosized metal particles. *Langmuir*. 1996; 12: 788–800.

Muzzarelli, R.A.A. Genipin crosslinked chitosan hydrogels as biomedical and pharmaceutical aids. *Carbohydrate Polymers* 2009; 77: 1–9.

Muzzarelli, R.A.A., Tarsi, R., Filippini, O., Giovanetti, E., Biagini, G. and Varaldo, P.E. Antimicrobial properties of N-carboxybutyl chitosan. *Antimicrobial Agents and Chemotherapy* 1990; 34: 2019–2023.

Nadanathangam, V. and Satyamurthy, P. Preparation of spherical nanocellulose by anaerobic microbial consortium. *International Proceedings of Chemical, Biological & Environmental Engineering* 2011; 7: 181-183.

Nakano, Y., Bin, Y., Bando, M., Nakashima, T., Okuno, T., Kurosu, H. and Matsuo, M. Structure and mechanical properties of chitosan/poly(vinyl alcohol) blend films. *Macromolecular Symposia* 2007; 258: 63–81.

Nam, Y.J. and Lead, J.R. Manufactured nanoparticles: An overview of their chemistry, interactions and potential environmental implications. *Science of the Total Environment*, 2008; 400: 396-414.

Ng, L.Y., Mohammad, A.W., Leo, Ch.P. and Hilal, N. Polymeric membranes incorporated with metal/metal oxide nanoparticles: A comprehensive review. *Desalination* 2013; 308: 15–33.

- Nishino, T., Matsuda, I. and Hirao, K. All-cellulose composite. *Macromolecules* 2004; 37(20): 7683-7687.
- No, H. K., Meyers, S. P., Prinyawiwatkul, W. and Xu, Z. Applications of chitosan for improvement of quality and shelf-life of foods: a review. *Journal of Food Science* 2007; 72: 87-100.
- Oh, S.Y., Yoo, D.I., Shin, Y. and Seo, G. FTIR analysis of cellulose treated with sodium hydroxide and carbon dioxide. *Carbohydrate Research* 2005; 340: 417–428.
- Oksman, K., Mathew, A.P., Bondeson, D. and Kvien, I. Manufacturing process of cellulose whisker/polylactic acid nanocomposites. *Composites Science and Technology* 2006; 66: 2776-2784.
- Padalkar, S., Capadona, J.R., Rowan, S.J., Weder, C., Won, Y.H., Stanciu, L.A. and Moon, R.J. Natural bio-polymers: novel templates for the synthesis of nanostructures. *Langmuir* 2010; 26: 8497-8502.
- Pandey, J.K., Chu, W.S., Kim, C.S., Lee, C.S. and Ahn, S.H. Bio-nano reinforcement of environmentally degradable polymer matrix by cellulose whiskers from grass. *Composites: Part B* 2009; 40: 676–680.
- Paralikal, S.A., Simonsen, J. and Lombardi, J. Poly(vinyl alcohol)/cellulose nanocrystals barrier membranes. *Journal of Membrane Science* 2008; 320: 248-258.
- Filson, P.B. and Dawson-Andoh, B.E. Sono-chemical preparation of cellulose nanocrystals from lignocellulose derived materials. *Bioresource Technology* 2009; 100: 2259–2264.
- Pei, A., Zhou, Q. and Berglund, L.A. Functionalized cellulose nanocrystals as bio based nucleation agents in poly(L-lactide) (PLLA)–Crystallization and mechanical property effects. *Composites Science and Technology* 2010; 70: 815–821.

- Peponi, L., Tercjak, A., Torre, L., Mondragon, I. and Kenny, J.M. Nanostructured physical gel of SBS block copolymer and Ag/DT/SBS nanocomposites. *Journal Material Science* 2009; 44: 1287-93.
- Peresin, M.S., Habibi, Y., Zope, J.O., Pawlak, J.J. and Rojas, O.J. Nanofiber composites of polyvinyl alcohol and cellulose nanocrystals: manufacture and characterization. *Biomacromolecules* 2010; 11(3): 674-681.
- Petit, C., Lixon, P. and Pileni, M. In situ synthesis of silver nanocluster in AOT reverse micelles. *Journal of Physics and Chemistry* 1993; 97: 12974–12983.
- Piskorz, J., Radlein, D., Scott, D.S. and Czernik, S. Pretreatment of wood and cellulose for production of sugars by fast pyrolysis. *Journal of Analytical and Applied Pyrolysis* 1989; 16: 127-142.
- Prashanth, R. and Tharanathan, R. Chitin/chitosan: modifications and their unlimited application potentialan overview. *Trends in Food Science & Technology* 2007; 18: 117-131.
- Premanathan, M., Karthikeyan, K., Jeyasubramanian, K. and Manivannan, G. Selective toxicity of ZnO nanoparticles toward Gram-positive bacteria and cancer cells by apoptosis through lipid peroxidation. *Nanomedicine: Nanotechnology, Biology, and Medicine* 2011; 7(2): 184-192.
- Qin, Y.M., Zhu, C.J., Chen, J., Chen, Y.Z. and Zhang, C. The absorption and release of silver and zinc ions by chitosan fibers. *Journal Applied Polymer Science* 2006; 101: 766–771.
- Raman, N., Sudharsan, S. and Pothiraj, K. Synthesis and structural reactivity of inorganic-organic hybrid nanocomposites-A review. *Journal of Saudi Chemical Society* 2011; 16: 339–352.
- Ramires, E.C. and Dufresne, A. A review of cellulose nanocrystals and nanocomposites. *Tappi Journal* 2011; 158: 9-16.

Raghupathi, K.R., Koodali, R.T. and Manna, A.C. Size-dependent bacterial growth inhibition and mechanism of antibacterial activity of zinc oxide nanoparticles. *Langmuir* 2011; 27(7): 4020-4028.

Rånby, B.G. and Ribí, E. Über den Feinbau des Zellulose. *Experientia* 1950; 6: 12–14.

Rao, C.N.R., Muller, A. and Cheetham, A.K. The chemistry of nanomaterials: Synthesis, properties and applications. Weinheim: Wiley-VCH Verlag GmbH & Co. KGaA 2004.

Rasato, D. 2009. The emerging world of bioplastics: An industry ‘father’ looks forward <http://www.omnexus.com/resources/print.aspx?id=22050>

Ray, S.S. and Bousmina, M. Biodegradable polymers and their layered silicate nanocomposites: In greening the 21st century materials world. *Progress in Materials Science* 2005; 50(8): 962-1079.

Reneker, D.H. and Yarin, A.L. Electrospinning jets and polymer nanofibers. *Polymer* 2008; 49: 2387-2425.

Rhim, J.W. Preparation and characterization of vacuum sputter silver coated PLA film. *LWT-Food Science and Technology* 2013; 54: 477-484.

Roohani, M., Habibi, Y., Belgacem, N.M., Ebrahim, G., Karimi, A.N. and Dufresne, A. *European Polymer Journal* 2008; 44: 2489-2498.

Roman, M. and Winter, W.T. Effect of sulfate groups from sulfuric acid hydrolysis on the thermal degradation behavior of bacterial cellulose. *Biomacromolecules* 2004; 5(5): 1671-1677.

Rueda, L., Saralegui, A., Fernández d’Arlas, B., Zhou, Q., Berglund, L.A., et al. Cellulose nanocrystals/polyurethane nanocomposites. study from the viewpoint of microphase separated structure. *Carbohydrate Polymers* 2013; 92: 751–757.

- Sadaf, J.R., Israr, M.Q., Kishwar, S., Nur, O. and Willander, M. White electroluminescence using ZnO nanotubes/GaN heterostructure light-emitting diode. *Nanoscale Research Letters* 2010; 5: 957–960.
- Sakurada, I., Nukushina, Y., et al. Experimental determination of the elastic modulus of crystalline regions in oriented polymers. *Journal of Polymer Science* 1962; 57(165): 651–660.
- Salavati, M.N., Davar, F. and Mazaheri, M. Preparation of ZnO nanoparticles from [bis(acetylacetonato)zinc(II)]-oleylamine complex by thermal decomposition. *Materials Letters* 2008; 62(12-13): 1890-1892.
- Sanchez-Garcia, M.D., Gimenez, E. and Lagaron, J.M. Morphology and barrier properties of solvent cast composites of thermoplastic bio-polymers and purified cellulose fibers. *Carbohydrate Polymers* 2008; 71: 235–244.
- Sanchez-Garcia, M.D., Lugarno, J. M. and Hoa, S.V. Effect of addition of carbon nanofibers and carbon nanotubes on properties of thermoplastic biopolymers *Composites Science and Technology* 2010; 70: 1095.
- Savadekar, N.R. and Mhaske. S.T. Synthesis of nano cellulose fibers and effect on thermoplastics starch based films. *Carbohydrate Polymers* 2012; 89: 146– 151.
- Saxena, I.M. and Brown, R.M.J. Cellulose biosynthesis: current views and evolving concepts. *Ann Bot* 2005; 96: 9-21.
- Shafei, A.E. and Abou-Okeil, A. ZnO/carboxymethyl chitosan bionano-composite to impart antibacterial and UV protection for cotton fabric. *Carbohydrate Polymers* 2011; 83: 920–925.
- Shen, L.M., Bao, N.Z. and Yanagisawa, K. Direct synthesis of ZnO nanoparticles by a solution free mechanochemical reaction. *Nanotechnology* 2006; 17: 5117–5123.
- Shi, Q., Zhou, Ch., Yue, Y., Guo, W., Wu, Y. and Wu, Q. Mechanical properties and in vitro degradation of electrospun bio-nanocomposite mats from PLA and cellulose nanocrystals. *Carbohydrate Polymers* 2012; 90: 301– 308.

- Shin, Y. and Exarhos, G.J. Template synthesis of porous titania using cellulose nanocrystals. *Materials Letters* 2007; 61: 2594–2597.
- Shin, Y., Bae, I., Arey, B.W., and Exarhos, G.J.J. Facile stabilization of gold-silver alloy nanoparticles on cellulose nanocrystals. *The Journal of Physical Chemistry C* 2008; 112: 4844-4848.
- Shinsuke, I., Manami, T., Minoru, M., Hiroyuki, S. and Hiroyuki, Y. Synthesis of silver nanoparticles templated by TEMPO-mediated oxidized bacterial cellulose nanofibers. *Biomacromolecules* 2009; 10: 2714–2717.
- Siddaramaiah, J.G. High performance edible nanocomposite films containing bacterial cellulose nanocrystals. *Carbohydrate Polymers* 2012; 87: 2031– 2037.
- Siqueira, G., Bras, J., Follain, N., Belbekhouche, S., Marais, S. and Dufresne, A. Thermal and mechanical properties of bio-nanocomposites reinforced by *Luffacylindrica* cellulose nanocrystals. *Carbohydrate Polymers* 2013; 91: 711–717.
- Siró, I. and Plackett, D. Microfibrillated cellulose and new nanocomposite materials: a review. *Cellulose* 2010; 17(3): 459–494.
- Sorrentino, A., Gorrasi, G. and Vittoria, V. Potential perspectives of bio nanocomposites for food packaging applications. *Trends in Food Science & Technology* 2007; 18(2): 84–95.
- Staggs, J.E.J. Discrete bond-weighted random scission of linear polymers. *Polymer* 2006; 47(3): 897-906.
- Stoimenov, P. K., Klinger, R. L., Marchin, G. L. and Klabunde, K. J. Metal oxide nanoparticles as bactericidal agents. *Langmuir* 2002; 18: 6679–6686.
- Sue, K., Kimura, K., Yamamoto, M. and Arai, K. Rapid hydrothermal synthesis of ZnO nanorods without organics. *Materials Letters* 2004; 58(26): 3350-3352.



Sugiyama, J., Chanzi, H. and Revol, J.F. On the polarity of cellulose in the cell wall of *Valonia*. *Planta* 1994; 193: 260-265.

Sun, J.H., Dong, S.Y., Wang, Y.K. and Sun, S.P. Preparation and photocatalytic property of a novel dumbbell-shaped ZnO microcrystal photocatalyst. *Journal of Hazardous Materials* 2009; 172: 1520–1526.

Sun, Y.G. and Xia, Y.N. Shape-controlled synthesis of gold and silver nanoparticles. *Science* 2002; 298: 2176–2179.

Tang, X. and Alavi, S. Recent advances in starch, polyvinyl alcohol based polymer blends, nanocomposites and their biodegradability. *Carbohydrate Polymers* 2011; 85: 7–16.

Tao, J.C., Chen, X., Sun, Y., Shen, Y. and Dai, N. Controllable preparation of ZnO hollow microspheres by self-assembled block copolymer. *Colloids and Surfaces A* 2008; 330: 67–71.

Terakado, O. and Hirasawa, M. Effect of metal oxides on the pyrolysis residues of poly(ethyleneterephthalate): formation of carbonaceous submicron, nano-scale filaments and mesoporous compounds. *Journal of Analytical and Applied Pyrolysis* 2005; 73: 248–256.

Thielemans, W., Warbey, C.R. and Walsh, D.A. Permselective nanostructured membranes based on cellulose nanowhiskers. *Green Chemistry* 2009; 11: 531-537.

Tian, Ch., Li, W., Pan, K., Zhang, Q., Tian, G., Zhou, W. and Fu, H. One pot synthesis of Ag nanoparticle modified ZnO microspheres in ethylene glycolmedium and their enhanced photocatalytic performance. *Journal of Solid State Chemistry* 2010; 183: 2720-2725.

Tong, M., Yuan, S., Long, H. and Zheng, M. Reduction of nitrobenzene in groundwater by iron nanoparticles immobilized in PEG/nylon membrane. *Journal of Contaminant Hydrology* 2010; 122(1-4): 16-25.

- Tong, Y., Li, Y., Xie, F. and Ding, M. Preparation and characteristics of polyimide-TiO<sub>2</sub> nanocomposite film. *Polymer International* 2000; 49: 1543–1547.
- Tripathi, S., Mehrotra, G.K. and Dutta, P.K. Physicochemical and bioactivity of cross-linked chitosane PVA film for food packaging applications. *International Journal of Biological Macromolecules* 2009; 45: 372-376.
- Turbak, A.F., Snyder, F.W. and Sandberg, K.R. Microfibrillated cellulose, a new cellulose product: Properties, uses, and commercial potential. *Journal of Applied Polymer Science* 1983; 37: 815-827.
- Umare, S.S., Chandure, A.S. and Pandey, R.A. Synthesis, characterization and biodegradable studies of 1, 3-propanediol based polyesters. *Polymer Degradation and Stability* 2007; 92(3): 464-479.
- Ureña, B.E.E. 2011. *Cellulose Nanocrystals Properties and Applications in Renewable Nanocomposites*, PhD Thesis, Clemson University.
- Vanden B.O., Capadona, J.R. and Weder, C. Preparation of homogeneous dispersions of tunicate cellulose whiskers in organic solvents. *Biomacromolecules* 2007; 8: 1353-1357.
- Vankrevelen, D.W. Some basic aspects of flame resistance of polymeric materials. *Polymer* 1975; 16(8): 615–620.
- Vatutsina, O.M., Soldatov, V.S., Sokolova, V.I., Johann, J., Bissen, M. and Weissenbacher, A. A new hybrid (polymer/inorganic) fibrous sorbent for arsenic removal from drinking water. *Reactive and Functional Polymers* 2007; 67: 184–201.
- Vicentini, S.D., Jr, A.S., Laranjeira and M.C.M. Chitosan/poly (vinyl alcohol) films containing ZnO nanoparticles and plasticizers. *Materials Science and Engineering C* 2010; 30: 503–508.
- Viet, D., Beck-Candanedo, S. and Gray, D.G. Dispersion of cellulose nanocrystals in polar organic solvents. *Cellulose* 2007; 14: 109-113.

- Wada, M., Kondo, T. and Okano, T. Thermally induced crystal transformation from cellulose I. alpha. to I. beta. *Polymer Journal (Tokyo Japan)* 2003; 35(2): 155–159.
- Wahab, R., Mishra, A., Yun, S.I., Kim, Y.S. and Shin, H.S. Antibacterial activity of ZnO nanoparticles prepared via non-hydrolytic solution route. *Applied Microbiology and Biotechnology* 2010; 87 (5): 1917-1925.
- Wahab, R., Mishra, A., Yun, S., Hwang, I.H., Mussarat, J., et al. Fabrication, growth mechanism and antibacterial activity of ZnO micro-spheres prepared via solution process. *Biomass and Biotechnology* 2012; 39: 227-236.
- Wan, T., Wang, Y.C. and Feng, F. Preparation of titanium dioxide/polyacrylate nanocomposites by sol-gel process in reverse micelles and in situ photopolymerization. *Journal of Applied Polymer Science* 2006; 102: 5105–5112.
- Wang, F.L., Sun, Y., Wang, J., Yu, A., Zhang, H. and Song, D. Water-soluble ZnO–Au nanocomposite-based probe for enhanced protein detection in a SPR biosensor system. *Journal of Colloid and Interface Science* 2010a; 351: 392–397.
- Wang, H. and Xie, C. Effect of annealing temperature on the microstructures and photocatalytic property of colloidal ZnO nanoparticles. *Journal of Physics and Chemistry of Solids* 2008; 69: 2440-2444.
- Wang, L. and Wang, A. Adsorption properties of Congo Red from aqueous solution onto N,O-carboxymethyl-chitosan. *Bioresource Technology* 2008; 99, 1403–1408.
- Wang, N., Ding, E. and Cheng, R. Thermal degradation behaviors of spherical cellulose nanocrystals with sulfate groups. *Polymer* 2007; 48: 3486–3493.
- Wang, Q., Du, Y.M. and Fan, L.H. Properties of chitosan/poly(vinyl alcohol) films for drug-controlled release. *Journal of Applied Polymer Science* 2005; 96: 808–813.
- Wang, Y., Zhang, C., Bi, S. and Luo, G. Preparation of ZnO nanoparticles using the direct precipitation method in a membrane dispersion micro-structured reactor. *Powder Technology, In Press, Corrected Proof* 2010b.

- Wang, Y., Zhang, Q., Zhang, Ch.L. and Li, P. Characterisation and cooperative antimicrobial properties of chitosan/nano-ZnO composite nanofibrous membranes. *Food Chemistry* 2012; 132: 419–427.
- Wang, Z.H., Lu, Y.L., Liu, J., Dang, Z.M., Zhang, L.Q. and Wang, W. Preparation of nanoalumina/EPDM composites with good performance in thermal conductivity and mechanical properties. *Polymers for Advanced Technologies* 2010c.
- Wang, Z., Lu, Y., Liu, J., Dang, Z., Zhang, L. and Wang, W. Preparation of nano-zinc oxide/EPDM composites with both good thermal conductivity and mechanical properties. *Journal of Applied Polymer Science* 2011; 119 (2): 1144–1155.
- Wang, Z., Qian, X.F., Yin, J. and Zhu, Z.K. Large-scale fabrication of tower-like, flowerlike, and tube-like ZnO arrays by a simple chemical solution route. *Langmuir* 2004; 20: 3441–3448.
- Wen, X. and Tresco, P.A. Fabrication and characterization of permeable degradable poly(DL-lactide-co-glycolide) (PLGA) hollow fiber phase inversion membranes for use as nerve tract guidance channels. *Biomaterials* 2006; 27: 3800-9.
- Whang, T.J. Hsieh, M.T. and Chen, H.H. Visible-light photocatalytic degradation of methylene blue with laser-induced Ag/ZnO nanoparticles. *Applied Surface Science* 2012; 258: 2796–2801.
- Wu, D.M., Meng, Q.Y., Liu, Y., Ding, Y.M., Chen, W.H., Xu, H. and Ren, D.Y. In situ bubble-stretch dispersion mechanism for additives in polymers. *Journal of Polymer Science Part B: Polymer Physics* 2003; 41(10): 1051–1058.
- Xiang, Q., Meng, G., Zhanga, Y., Xua, J. and Xu, P. Ag nanoparticle embedded-ZnO nanorods synthesized via a photochemical method and its gas-sensing properties. *Sensors and Actuators B* 2010; 143: 635–640.
- Xie, Y., He, Y., Irwin, P.L., Jin, T. and Shi, X. Antibacterial activity and mechanism of action of zinc oxide nanoparticles against *Campylobacter jejuni*. *Applied Environmental Microbiology* 2011; 77(7): 2325-2331.

- Xu, J.C., Liu, W.M. and Li, H.L. Titanium dioxide doped polyaniline. *Materials Science and Engineering: C* 2005; 25(4):444-7.
- Xu, W., Qin, Z.Y., Yu, H.Y., Liu, Y., Liu, N., Zhou, Z. and Chen, L. Cellulose nanocrystals as organic nanofillers for transparent polycarbonate films. *Journal of Nanoparticle Research* 2013; 15: 1562–1570.
- Xu, X., Wang, Q. and Choi, H.C. Encapsulation of iron nanoparticles with PVP nanofibrous membranes to maintain their catalytic activity. *Journal of Membrane Science* 2010; 1348: 231–237.
- Yew, S.P. Tang, H.Y. and Sudesh, K. Photocatalytic activity and biodegradation of polyhydroxybutyrate films containing titanium dioxide. *Polymer Degradation and Stability* 2006; 91(8): 1800–1807.
- Yi, X.X., Ying, Q.Y., Jiu, Y.X., Shi, F.Y. and Wang, F. Properties of novel polyvinyl alcohol/cellulose nanocrystals/silver nanoparticles blend membranes. *Carbohydrate Polymer* 2013; 98: 1573–1577.
- Yin, X., Que, W., Fei, D., Shen, F. and Guo, Q. Ag nanoparticle/ZnO nanorods nanocomposites derived by a seed-mediated method and their photocatalytic properties. *Journal of Alloys and Compounds* 2012; 524: 13–21.
- Yoon, P.J., Hunter, D.L. and Paul, D.R. Polycarbonate nanocomposites, effect of organoclay structure on morphology and properties. *Polymer* 2003; 44: 5323–5339.
- You, C., Han, C., Wang, X., Zheng Y., Li, Q., Hu, X. and Sun, H. The progress of silver nanoparticles in the antibacterial mechanism, clinical application and cytotoxicity. *Molecular Biology Reports* 2012; 39: 9193–9201.
- Yuan, J., Choo, E.S., Tang, X., Sheng, Y., Ding, J. and Xue, J. Synthesis of ZnO-Pt nanoflowers and their photocatalytic applications. *Nanotechnology* 2010; 21: 185–189.

- Yu, H.Y., Qin, Z.Y., Liang, B.L., Liu, N., Zhou, Z. and Chen, L. Facile preparation of thermally stable cellulose nanocrystals with high yield of 93% through hydrochloric acid hydrolysis under hydrothermal condition. *Journal of Materials Chemistry A* 2013; 1: 3938–3944.
- Yu, J., Yang, J., Liu, B. and Ma, X. Preparation and characterization of glycerol plasticized-pea starch/ZnO–carboxymethylcellulose sodium nanocomposites. *Bioresource Technology* 2009a; 100: 2832–2841.
- Yun, K. and Veerapandian, M. Integrated strategy for multifunctional hybrid nanocomposites. *Plastics research online* 2010, DOI: 10.2417/spetro.002876.
- Zhang, J., Roll, D., Geddes, C.D. and Lakowicz, J.R. Aggregation of silver nanoparticle-dextran adducts with concanavalin A and competitive complexation with glucose. *The Journal of Physical Chemistry B* 2004; 108: 12210–12214.
- Zhang, J., Sun, L., Yin, J., Su, H., Ch, Liao. and Yan, Ch. Control of ZnO morphology via a simple solution route. *Chemistry of Materials* 2002; 14: 4172–7.
- Zhang, L.L., Jiang, Y.H., Ding, Y.L., Povey, M. and York, D. Investigation into the antibacterial behaviour of suspensions of ZnO nanoparticles (ZnO nanofluids). *Journal of Nanoparticle Research* 2007; 9: 479–89.
- Zhang, T.J., Wang, W., Zhang, D.Y., Zhang, X.X., Ma, Y.Y., Zhou, Y.L., et al. Biotemplated synthesis of gold nanoparticles-bacteria cellulose nanofibers nanocomposites and their application in biosensing. *Advanced Functional Materials* 2010; 20(7): 1152-60.
- Zhao, X., Lv, L., Pan, B., Zhang, W., Zhang, Sh. and Zhang, Q. Polymer-supported nanocomposites for environmental application: A review. *Chemical Engineering Journal* 2011; 170: 381–394.
- Zhao, C.X. and Zhang, W.D. Preparation of waterborne polyurethane nanocomposites: polymerization from functionalized hydroxyapatite. *European Polymer Journal* 2008; 44: 1988–95.

Zheng, Y.F., Yu, X.J., Xu, X.Q., Jin, D.L. and Yue, L.H. Preparation of ZnO particle with novel nut-like morphology by ultrasonic pretreatment and its luminescence property. *Ultrasonics Sonochemistry* 2010; 17: 7–10.

Zhong, J., Cheng, K., Hu, B., Gong, H., Zhou, Sh. and Du, Z. Temperature-controlled growth and optical properties of ZnO nanorods with quadrangular and hexagonal cross sections. *Materials Chemistry and Physics* 2009; 115: 799–803.

Zhou, G. and Deng, J.C. Preparation and photocatalytic performance of Ag/ZnO nanocomposites. *Materials Science in Semiconductor Processing* 2007; 10: 90–96.

Zhou, Q. and Xanthos, M. Nanosize and microsize clay effects on the kinetics of the thermal degradation of polylactides. *Polymer Degradation and Stability* 2009; 94: 327-338.

Zimmermann, T., Pohler, E. and Geiger, T. Cellulose fibrils for polymer reinforcement. *Advanced Engineering Materials* 2004; 6: 754-761.

Zou, C.W., Rao, Y.F., Alyamani, A., Chu, W., Chen, M.J. et al. Heterogeneous lollipop-like V<sub>2</sub>O<sub>5</sub>/ZnO array: a promising composite nanostructure for visible light photocatalysis. *Langmuir* 2010; 26: 11615–11620.

[https://www.google.com.my/webhp?gws\\_rd=cr&ei=GG6IU\\_roA4\\_k8AWmr4CwBg#q=deacetylation+of+chitin+image](https://www.google.com.my/webhp?gws_rd=cr&ei=GG6IU_roA4_k8AWmr4CwBg#q=deacetylation+of+chitin+image)

[https://www.google.com/search?q=Zno+nanoparticles+image&tbm=isch&tbo=u&source=univ&sa=X&ei=3EmoU6uWDsKE8gXH0YHoAQ&ved=0CCMQsAQ&biw=1366&bih=667#facrc=&imgdii=&imgrc=kfUZTOCCyAyhNM%253A%3QB5pzJCZ\\_H1ENM%3Bhttp%253A%252F%252Fnanos](https://www.google.com/search?q=Zno+nanoparticles+image&tbm=isch&tbo=u&source=univ&sa=X&ei=3EmoU6uWDsKE8gXH0YHoAQ&ved=0CCMQsAQ&biw=1366&bih=667#facrc=&imgdii=&imgrc=kfUZTOCCyAyhNM%253A%3QB5pzJCZ_H1ENM%3Bhttp%253A%252F%252Fnanos)