ADSORPTION OF ACID FUCHSIN AND FAST GREEN ON ACTIVATED CARBONS

By

HO SOON MIN

Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia in Fulfilment of the Requirement for the Degree of Master of Science

December 2003

DEDICATION

I would like to dedicate this work to my beloved family members, especially my parents for their full support to carry out my Master Degree study in Universiti Putra Malaysia. Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirements for the degree of Master of Science

ADSORPTION OF ACID FUCHSIN AND FAST GREEN ON ACTIVATED CARBONS

By

HO SOON MIN

December 2003

Chairman: Professor Mohd Zobir Bin Hussein, Ph.D.

Faculty: Science and Environmental Studies

Activated carbons produced from palm oil shell with H_3PO_4 activation (AC 34) and physical activation (AC 00) were used to adsorb dye solutions, namely Acid Fuchsin and Fast Green. The precursors were carbonized under an inert nitrogen atmosphere for three hours followed by carbon dioxide gas activation for another hour in an electric furnace at 500 °C. Analysis of the adsorption isotherms of N₂ at 77 K shows that the surface area of AC 34 (1017 m²/g) is higher than AC 00 (426 m²/g). The average pore diameter for AC 34 and AC 00 is 25.2 and 15.1 Å, respectively. Characterization of activated carbons were studied by using various analytical techniques such as Fourier Transform Infrared (FTIR), Scanning Electron Microscopy (SEM), determination of ash and moisture content. The contact time, adsorbent dosage, adsorbate concentration, temperature and pH of the adsorbate that affecting the adsorption process were also studied. The result shows that the AC 34 is very effective in adsorbing Acid Fuchsin and Fast Green compared to AC 00. The adsorption process agreed to the Langmuir and Freundlich isotherms with endothermic nature. In addition to , the two samples prepared in this study, another two commercial activated carbons (AC 6070 and AC 3040) were also used for comparison. A favorable adsorption of both dye solutions on AC 6070 compared to the other activated carbons is due to its higher BET surface area and micropore volume. The amount of dye adsorbed on adsorbent is in the order of AC 34>AC 00>AC 6070>AC 3040.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

PENJERAPAN ACID FUCHSIN DAN FAST GREEN OLEH KARBON TERAKTIF

Oleh

HO SOON MIN

Disember 2003

Pengerusi: Profesor Mohd Zobir Bin Hussein, Ph.D.

Fakulti: Sains dan Pengajian Alam Sekitar

Karbon teraktif yang dihasilkan daripada tempurung kelapa sawit secara pengaktifkan kimia dengan H₃PO₄ (AC 34) dan pengaktifkan fizikal (AC 00) telah diguna untuk menjerap pencelup, iaitu Acid Fuchsin dan Fast Green. Bahan permulaan bagi karbon teraktif telah dikarbonkan pada 500 °C selama 3 jam di bawah aliran gas N₂ dan diikuti dengan sejam lagi untuk gas CO₂. Analisis isoterma penjerapan N₂ pada 77 K menunjukkan bahawa luas permukaan bagi AC 34 (1017 m²/g) adalah lebih tinggi daripada AC 00 (426 m²/g). Garis pusat keliangan purata bagi AC 34 dan AC 00 ialah masing-masing 25.2 dan 15.1 Å. Pencirian karbon teraktif telah dikaji menggunakan teknik-teknik spektra infra merah transformasi Fourier (FTIR), mikroskop pengimbasan elektron (SEM), penentuan kandungan lembapan dan bahan tak organik. Masa sentuhan, jisim penjerap, kepekatan bahan terjerap, suhu dan pH telah didapati mempengaruhi proses penjerapan. Keputusan juga menunjukkan bahawa AC 34 sangat berkesan sebagai

bahan menjerap pencelup berbanding dengan AC 00. Di samping itu, didapati proses penjerapan juga mematuhi model Langmuir dan Freundlich, dan proses tersebut adalah bersifat endotermik. Selain daripada itu, dua karbon teraktif komersil, iaitu AC 6070 dan AC 3040 telah juga digunakan dalam penyelidikan ini bagi tujuan perbandingan. Peratusan penjerapan bagi kedua-dua jenis pencelup adalah tertinggi untuk AC 6070 berbanding yang lain disebabkan AC 6070 mempunyai luas permukaan BET dan isipadu keliangan mikro yang lebih tinggi. Jumlah pewarna yang dapat dijerap oleh karbon teraktif adalah seperti siri berikut, iaitu AC 34>AC 00>AC 6070>AC 3040.

ACKNOWLEDGEMENTS

First of all, I wish to express my deepest thanks and appreciation to my honourable project supervisor Prof. Dr. Mohd Zobir Bin Hussein for his continuous supervision, suggestions, guidance and contribution throughout the duration of this project.

Sincere thanks are extended to Assoc. Prof. Dr. Zulkarnain Zainal and Prof. Dr. Anuuar Kassim as co-supervisors, for their assistance in making this project possible. I am also grateful to all the staff in Chemistry Department, especially to Encik Zainal Abidin, Encik Nazari, Encik Abbas, Mr. Ho and Kak Ita from Institute of Bio Science, without their help and encouragement, the project would never have been started and carried through to completion.

Finally, I would like to take this opportunity to thank my labmates and all those who had contributed to the success of this project in one way or another. To all these truly outstanding people and also to many others, my warm personal regards. I certify that an Examination Committee met on 15th December 2003 to conduct the final examination of Ho Soon Min on his Master of Science thesis entitled "Adsorption of Acid fuchsin and Fast Green on Activated Carbons" in accordance with Universiti Pertanian Malaysia (Higher Degree) Act 1980 and Universiti Pertanian Malasyia (Higher Degree) Regulations 1981. The Committee recommends that the candidate be awarded the relevant degree. Members of the Examination Committee are as follows:

TAN WEE TEE, Ph.D.

Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Chairman)

MOHD ZOBIR BIN HUSSEIN, Ph.D.

Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Member)

ZULKARNAIN ZAINAL, Ph.D.

Associate Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Member)

ANUAR KASSIM, Ph.D.

Associate Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Member)

GULAM RUSUL RAHMAT ALI, Ph.D.

Professor/Deputy Dean School of Graduate studies Universiti Putra Malaysia

Date:

This thesis submitted to the Senate of Universiti Putra Malaysia and has been accepted as partial fulfillment of the requirement for the degree of Master of Science. The members of the supervisory Committee are as follows:

MOHD ZOBIR Bin HUSSEIN, Ph.D.

Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Chairman)

ZULKARNAIN ZAINAL, Ph.D.

Associate Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Member)

ANUAR KASSIM, Ph.D.

Associate Professor Faculty of Science and Environmental Studies Universiti Putra Malaysia (Member)

AINI IDERIS, Ph.D.

Professor/Dean School of Graduate studies Universiti Putra Malaysia

Date:

DECLARATION

I hereby declare that the thesis is based on my original work except for the quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted or any other degree at UPM or other institutions.

HO SOON MIN

Date:

TABLE OF CONTENTS

DEDICATION	ii
ABSTRACT	iii
ABSTRAK	V
ACKNOWLEDGEMENTS	vii
APPROVAL	viii
DECLARATION	Х
LIST OF TABLES	xiii
LIST OF FIGURES	XV
LIST OF ABBREVIATIONS	XX

CHAPTER

1

INTRODUCTION Textile Industry Waste water

-	Textile Industry Waste water	1
	Activated Carbon	3
	Historical Discoveries	4
	Structure of Activated Carbon	5
	The Properties of Activated Carbon Functional Group	7
	Pore Structure	10
	Raw Materials	12
	Manufacturing of Activated Carbon	17
	Chemical Activation	18
	Physical Activation	19
	Application of Activated Carbon	19
	Theory of Adsorption	24
	Factors which Influenced Adsorption	25
	Adsorption Kinetics	30
	Adsorption Isotherms	31
	Classification of Adsorption Isotherm	36
	Textile Industry	39
	Classification of Dyes	40
	Palm Oil Industry	43
	Acid Fuchsin	44
	Fast Green	45
	Objectives	47
CHAPTER		
II	MATERIALS AND METHODS	
	Materials	48
	Impregnation of Activated Carbon	49
	Preparation of Activated Carbon	49
	Characterization of Activated Carbon	52
	BET Surface Area Measurement	52
	Scanning Electron Microscope	52
	Fourier Transform Infrared Spectrometer	53
	Determination of Moisture Content	53

	Determination of pH	53
	Determination of Ash content	54
	Density of Samples	54
	Adsorption Studies	55
	The Effect of initial Concentration	55
	The Effect of pH	56
	The Effect of Dosage of Carbon	57
	The Effect of Temperature	57
	Kinetics of Adsorption	58
	The Adsorption Isotherms	58
CHAPTER		
III	RESULTS AND DISCUSSIONS	
	Characterization of Activated Carbon	59
	pH	59
	Ash Content	60
	Moisture Content	62
	Morphology of Activated Carbon	63
	Surface Area	65
	Shape of Isotherms and Hysteresis	66
	Porosity	69
	Pore size distribution	69
	Density of Adsorbent	72
	Fourier Transform Infrared spectroscopy	73
	EDX – Mapping Analysis	77
	EDX Analysis	84
	Adsorption Study	88
	The Effect of pH	88
	The Effect of Initial Concentration	94
	Adsorption of Dye on Activated Carbons	103
	The Effect of Dosage of Carbon	104
	Adsorption Isotherms	108
	The Effect of Temperature	137
	Adsorption Kinetics Rate Constant	143
CHAPTER IV	CONCLUSION	150
CHAPTER V	SUGGESTION	154
BIBLIOGRAPHY		155
APPENDICES		161
BIODATA OF THE AUTHOR		185