

Novel monoclinic zirconolite in Bi₂O₃-CuO-Ta₂O₅ ternary system: phase equilibria, structural and electrical properties

ABSTRACT

Synthesis of novel monoclinic zirconolite, Bi_{1.92}Cu_{0.08}(Cu_{0.3}Ta_{0.7})₂O_{7.06} (-BCT) using solid state reaction had been finalised at the firing temperature of 900 °C over 24 h. The X-ray diffraction pattern of -BCT was fully indexed on a monoclinic symmetry, space group, C2/c with lattice constants, a = 13.1052 (8), b = 7.6749 (5), c = 12.162 (6), $\beta = 90^\circ$ and $\alpha = 101.32^\circ$ (1), respectively. The reaction mechanism study indicated phase formation was greatly influenced by the reaction between intermediate bismuth tantalate binary phases and CuO at elevated temperatures. -BCT was thermally stable up to a temperature of 900 °C and contained spherulite grains with sizes ranging from 1 to 14 μ m. Electrical properties of this material were characterised over a broad temperature range covering temperatures from 10 K to 874 K. At the temperature of 304 K, two semicircles were discernible in complex Coleó Cole plot showing an insulating grain boundary with $C_{gb} = 6.63 \times 10^{-9}$ F cm⁻¹ and a bulk response capacitance, $C_b = 6.74 \times 10^{-12}$ F cm⁻¹. The Power law frequency-dependent ac conductivity of -BCT was apparent in three frequency regimes; a low-frequency plateau regime, a high-frequency plateau regime and a dispersive regime taking place in the temperature range of 220-6576 K. The frequency-dependent ac conductivity of -BCT with increasing temperature was attributed to the thermal activated electrical conduction mechanism within the structure.

Keyword: Ceramics; Solid state reaction; Dielectric response; Ac impedance spectroscopy