



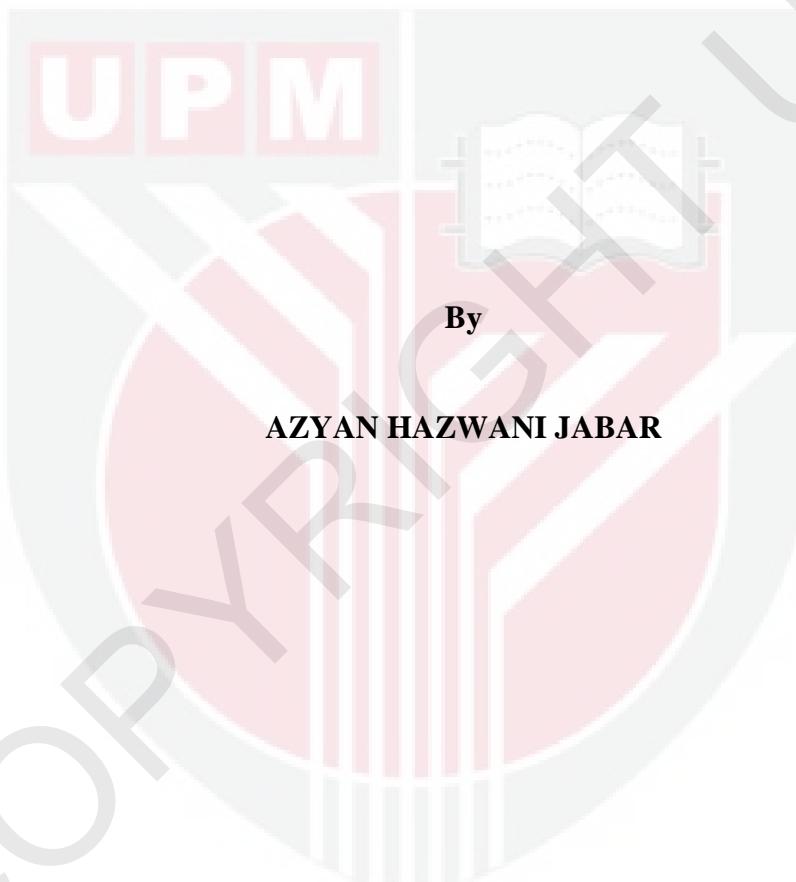
UNIVERSITI PUTRA MALAYSIA

**ULTRASONIC ASSISTED MICROWAVE PROCESSING SYSTEM FOR
OPTIMIZED PRODUCTION OF BIODIESEL FROM COCONUT OIL**

AZYAN HAZWANI JABAR

FS 2011 54

**ULTRASONIC ASSISTED MICROWAVE PROCESSING SYSTEM FOR
OPTIMIZED PRODUCTION OF BIODIESEL FROM COCONUT OIL**



**Thesis Submitted to the School of Graduate Studies, Universiti Putra
Malaysia, in Fulfilment of the Requirements for the Degree of
Master of Science**

November 2011

DEDICATION

To my beloved mother and father whom I owe every success in my life

♥ROHANI BINTI YAAKUB♥
♥JABAR BIN AWANG♥



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of
the requirement for the Degree of Master of Science

**ULTRASONIC ASSISTED MICROWAVE PROCESSING SYSTEM FOR
OPTIMIZED PRODUCTION OF BIODIESEL FROM COCONUT OIL**

By

AZYAN HAZWANI BINTI JABAR

November 2011

Chairman: Professor Kaida Bin Khalid, PhD

Faculty: Science

Biodiesel, a liquid fuel derived from plant oils or animal fats, represents a renewable energy source. The purpose of this research is to improve the coconut biodiesel production process using microwave transesterification and ultrasonic assisted microwave transesterification. This system will identify the yield from the combined ultrasonic and microwave, in terms of its rapidity and efficiency to transesterified. Some of the problems faced when the biodiesel production currently done at 1 to 2 hour in industry, so the technique obtained in this study probably can solved the problem in case of time consuming.

A process for the production of the methyl ester of coconut oil for use as a biodiesel fuel has been studied. The essential part of the process is the transesterification of the coconut oil with methanol, in the presence of a catalyst, to yield the methyl ester as a product and glycerine as a by-product. Experiments have been performed to determine the optimum conditions for the preparation of the coconut biodiesel. The optimum conditions were:
(1) 1.0% sodium hydroxide catalyst (dissolved in methanol) based on weight of coconut

oil and 100% excess of the stoichiometric amount of required anhydrous alcohol; (2) extremely vigorous agitation with 24 kHz ultrasonic frequency until the reaction mixture (oil and methoxide) becomes thoroughly mixed; (3) microwave irradiation of the mixture. Maximum yields of 95%, were obtained for coconut oil by microwave transesterification in 3 mins. Whereas the washing process only takes 7 mins for every cycle by using microwave method compared to 45 mins in conventional method. When the reaction was assisted with ultrasonic, the optimum yield is 97.20%, where the time taken for ultrasonic was about 5mins and microwave about 3min, which gives total time 8mins. This results extremely favorably to the hour of processing required by conventional methods.

A factor that is highly important during microwave heating is dielectric properties of material. Dielectric properties of coconut oil, methanol, sodium hydroxide, mixture of methanol and catalyst and first mixture between coconut oil and methoxide were determined using an open-ended, connected to a network analyzer. The important part of this dielectric measurement is the microwave dielectric detection of transesterification process at frequency 2.45 GHz. Result of the measurement over the frequency range shows drastic changes on the dielectric properties in the first 3 minutes of the reaction and after 3 minutes, the dielectric properties slowly decreases and approaches to the dielectric properties of biodiesel and glycerin. This result gives valuable information on the optimum mixing time of transesterification reaction. These properties could also be used to estimate absorbed power by the mixture for the application in microwave transesterification process as dielectric loss of the mixture change from 20 to about 0.5 as the transesterification reaction complete.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk Ijazah Master Sains

**SISTEM PEMPROSESAN ULTRASONIK DIBANTU GELOMBANGMIKRO
UNTUK PENGHASILAN OPTIMUM BIODIESEL DARIPADA MINYAK
KELAPA**

Oleh

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Biodiesel, bahan bakar cair yang berasal dari minyak tumbuhan atau lemak haiwan, merupakan sumber tenaga boleh diperbaharui. Tujuan daripada projek ini adalah untuk membantu dan meningkatkan proses pengeluaran biodiesel kelapa dengan menggunakan penyinaran gelombang mikro dan ultrasonik. Sistem ini akan mengenalpasti hasil dari gabungan gelombang mikro dan ultrasonik, bagi kelajuan dan kecekapan untuk ditransesterifikasi. Di dalam industry sebahagian daripada masalah yang dihadapi apabila pengeluaran biodiesel mengambil masa selama1 hingga 2 jam, maka teknik yang diperolehi dalam kajian ini mungkin boleh menyelesaikan masalah dalam penggunaan masa.

Satu proses untuk penghasilan metil ester minyak kelapa untuk digunakan sebagai bahan bakar biodiesel telah dikaji. Bahagian penting dari proses ini adalah pengtransesteran

minyak kelapa dengan methanol bersama pemangkin, untuk menghasilkan metil ester sebagai produk dan gliserin sebagai produk kedua. Eksperimen telah dilakukan untuk menentukan keadaan optimum untuk penyediaan biodiesel kelapa. Keadaan optimum adalah: (1) natrium hidroksida 1.0% mangkin (dilarutkan dalam metanol) berdasarkan berat minyak kelapa dan kelebihan 100% dari nilai stoikiometri alcohol anhidrat diperlukan; (2) tindakan yang berkesan dengan frekuensi ultrasonik 24 kHz sehingga reaksi campuran (minyak dan metoksida) menjadi campuran sebenar; (3) iradiasi gelombang mikro pada campuran. Hasil maksima kira-kira 95% telah berjaya diperoleh dengan pengtransesteran gelombang mikro dalam masa 3 minit. Sedangkan proses pencucian hanya memerlukan masa 7 minit untuk setiap kitaran dengan menggunakan kaedah gelombang mikro berbanding dengan 45 minit dalam kaedah konvensional. Ketika reaksi itu dibantu dengan penyinaran ultrasonik, hasil optimum adalah 97.20%, di mana masa yang diperlukan untuk ultrasonik adalah sekitar 5 minit dan gelombang mikro sekitar 3 minit, iaitu 8 minit jumlah masa keseluruhan. Keputusan ini sangat baik jika dibandingkan dengan masa pemprosesan yang diperlukan dengan kaedah konvensional.

Faktor yang sangat penting semasa pemanasan gelombang mikro adalah sifat dielektrik bahan. Sifat dielektrik minyak kelapa, metanol, sodium hidroksida, campuran metanol dan pemangkin dan campuran pertama antara minyak kelapa dan metoksida ditentukan dengan menggunakan sensor sepaksi hujung terbuka yang disambung ke penganalisa rangkaian. Bahagian penting dari pengukuran ini adalah pengesanan dielektrik gelombang mikro proses pengtransesteran pada frekuensi 2.45 GHz. Hasil pengukuran menunjukkan perubahan yang cepat pada sifat dielektrik dalam 3 minit pertama

tindakbalas dan selepas 3 minit, sifat dielektrik perlahan-lahan menurun dan menghampiri sifat dielektrik biodiesel dan gliserin. Keputusan ini memberikan maklumat yang penting bagi menentukan masa pencampuran optimum tindakbalas pengtransesteran. Sifat ini juga boleh digunakan untuk menganggarkan kuasa yang diserap oleh campuran untuk aplikasi dalam proses pengtransesteran gelombang mikro dimana faktor kehilangan dielektrik dari campuran berubah dari 20 menjadi sekitar 0.5 pada pengtransesteran lengkap.

ACKNOWLEDGEMENTS

I am thankful to Allah S.W.T. for the opportunity to pursue my studies at University Putra Malaysia; I have been blessed to meet many bright and special individuals here. The encouragement and support of many people have contributed to the completion of this thesis. There are, however, several people that deserve special mention. Words cannot express my deepest appreciation and respect that I have for my major professor, Prof Kaida Khalid. He has served as my mentor, teacher, role-model, father and friend. The guidance, constant enthusiasm, kindness, approachability and inspiration that he provided were indispensable throughout my graduate studies and completion of the thesis. Appreciation is extended to my committee members, Dr. Jumiah Hassan and Dr. Irmawati Ramli. To all the people from our research group En. Roslim, all student workers and for their help and assistance during this project, I express a very special thank you.

A special acknowledgment is made to my parents, Jabar Awang and Rohani Yaakub;- without whom I would not be here-, and all my family members, Rodziah Yaakub and Azmi Jabar. There is absolutely no way I could have done this without the love and support that they provided. I feel your love and support despite the distance of a half-globe.

In addition, I want to thanks my labmates, Nor Aina, Izzatul Hidayah, Bibi Sabrina, Nora Salina, Hasnidar, and Khairul Najmie; the friendship that they have freely given me is one of the greatest sources of comfort in my life. Thank you all!

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of **Master of Science**. The members of the Supervisory Committee were as follows:

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DECLARATION

I declare that the thesis is my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.

AZYAN HAZWANI BINTI JABAR

Date: 11 November 2011

TABLE OF CONTENTS

| | Page |
|--|-------------|
| DEDICATION | ii |
| ABSTRACT | iii |
| ABSTRAK | v |
| ACKNOWLEDGEMENTS | viii |
| APPROVAL | ix |
| DECLARATION | xi |
| LIST OF TABLES | xvi |
| LIST OF FIGURES | xviii |
| LIST OF ABBREVIATIONS | xxii |
| LIST OF SYMBOLS | xxiii |
| CHAPTER | |
| 1 INTRODUCTION | |
| 1.1 Introduction | 1 |
| 1.2 Motivations of the study | 1 |
| 1.3 Ultrasonic-reactor method | 2 |
| 1.4 Microwave method | 3 |
| 1.5 Problem statement | 3 |
| 1.6 Scope and limitations of the study | 4 |
| 1.7 Research Objectives | 4 |
| 1.8 Index of Thesis | 5 |
| 2 LITERATURE REVIEW | |
| 2.1 Introduction | 6 |
| 2.2 Microwave-assisted Transesterification | 7 |
| 2.3 Ultrasonic-assisted Transesterification | 8 |
| 2.4 Combined Ultrasonic and Microwave-assisted Transesterification Method | 9 |
| 2.5 Conventional Method versus Microwave Transesterification | 9 |
| 2.6 Biodiesel Standard | 11 |
| 2.7 Vegetable oil as fuel | 11 |
| 2.8 Transesterification Reaction | 13 |
| 2.9 Summary | 16 |
| 3 THEORY | |
| 3.1 Introduction | 17 |
| 3.2 The Electromagnetic Spectrum and the Electromagnetic Wave Propagation | 18 |
| 3.3 Microwave Energy | 19 |
| 3.4 Dielectric Properties | 20 |
| 3.5 Factors Influencing Dielectric Properties | 22 |
| 3.5.1 Frequency Effect | 23 |
| 3.5.2 Dielectric Properties of Reactants Contribute in Transesterification | 26 |

| | | |
|----------|--|----|
| 3.5.3 | Temperature Effects | 29 |
| 3.6 | Power absorption | 30 |
| 3.6.1 | Calculate power output of heating source $PO = PA_w = (PA/Vol)_w$ | 32 |
| 3.7 | Permittivity Measurement Technique | 32 |
| 3.7.1 | Open-Ended Coaxial Probe (OECP) | 33 |
| 3.8 | Ultrasonic Energy | 35 |
| 3.8.1 | Sonochemistry | 35 |
| 3.8.2 | Acoustic Cavitation | 35 |
| 3.8.3 | Applications in Chemistry | 37 |
| 3.8.4 | Laboratory Ultrasonic Instrument | 37 |
| 3.9 | Summary | 37 |
| 4 | MATERIALS AND METHODOLOGY | |
| 4.1 | Introduction | 39 |
| 4.2 | Part I - Production and Characterization of Coconut Oil | 42 |
| 4.2.1 | Feedstock Preparation | 42 |
| 4.2.2 | Analytical methods to measure oil characteristics - Oil Analysis | 43 |
| 4.2.2.1 | Analytical Method - Iodine Value Determination | 43 |
| 4.2.2.2 | Analytical Method - Saponification Value Determination | 44 |
| 4.2.2.3 | Analytical method - Free Fatty Acid and Acid Value Determination | 44 |
| 4.2.2.4 | Physical test – Kinematic Viscosity | 45 |
| 4.3 | Part II - Monitoring Microwave Dielectric Properties (ϵ' , ϵ'') during Transesterification Reaction and Prediction of Power Absorption by sample(PA/Vol_s) | 45 |
| 4.3.1 | Calibration Method | 46 |
| 4.3.2 | Monitoring Microwave Dielectric Properties during Transesterification Reactions | 48 |
| 4.3.2.1 | Materials | 48 |
| 4.3.2.2 | Experimental Set-up for Monitoring Dielectric Properties of Transesterification Reaction | 48 |
| 4.3.3 | Power output of heating source, $PO = PA_w = (PA/Vol)_w$ by water heating process measurement | 50 |
| 4.3.3.1 | Water heating process procedure | 51 |
| 4.4 | Part III – Optimization of Yield Percentage of Coconut Biodiesel through Combination of Ultrasonic and Microwave Processing | 51 |
| 4.4.1 | Transesterification Assisted by Microwave Heating | 51 |
| 4.4.2 | Ultrasonic Mixing and Microwave irradiation-Assisted Transesterification | 55 |
| 4.4.3 | Preparation of coconut biodiesel from Transesterification through Mechanical Stirring | 57 |

| | | |
|----------|---|----|
| | (Conventional Transesterification) | |
| 4.4.4 | Washing Process for conventional method | 57 |
| 4.4.5 | Sample Preparation | 58 |
| 4.5 | Part IV – Analysis of Physical and Chemical Characteristics of Coconut Biodiesel | 60 |
| 4.5.1 | Biodiesel fuel properties | |
| 4.6 | Part V – Analysis of Diesel Engine Performance using coconut biodiesel | 61 |
| 4.7 | Summary | 63 |
| 5 | RESULTS AND DISCUSSIONS | |
| 5.1 | Introduction | 64 |
| 5.2 | Result I - Production of Feedstock and Characterization | 64 |
| 5.2.1 | Feedstock Production | 64 |
| 5.2.2 | Physical and Chemical Properties of Coconut oil | 65 |
| 5.3 | Result II - Dielectric Properties and Determination of Power Absorption | 66 |
| 5.3.1 | Dielectric Constant and Dielectric Loss Factor of Coconut Oil, Methanol and Sodium Hydroxide | 66 |
| 5.3.2 | Dielectric Properties of Transesterification Process | 71 |
| 5.3.2.1 | Dielectric Properties of Reaction Process | 71 |
| 5.3.3 | Monitoring During Transesterification Process with Time of Reaction | 74 |
| 5.3.4 | Power absorbed | 76 |
| 5.4 | Result III - Optimum Conditions for Coconut Methyl Ester (CME) Production by using Ultrasonic and Microwave Combined Method (Quantity) and Washing Technique using Microwave Method (Quality) | 77 |
| 5.4.1 | Microwave Method | 77 |
| 5.4.1.1 | Effect of Molar Ratio on CME | 78 |
| 5.4.1.2 | Effect of Power and Temperature of Microwave on CME | 81 |
| 5.4.1.3 | Effect of Time of Microwave Irradiation on CME | 84 |
| 5.4.2 | Ultrasonic Assisted Microwave Method | 86 |
| 5.4.2.1 | Effect of Amplitude of Ultrasonic | 86 |
| 5.4.3 | Washing biodiesel | 92 |
| 5.5 | Result IV - Physical Properties of Coconut Biodiesel | 93 |
| 5.5.1 | Pour Point and Flash Point | 94 |
| 5.5.2 | Cloud Point | 95 |
| 5.5.3 | Specific Gravity | 95 |
| 5.5.4 | Gross Calorific Value (Energy Content) | 95 |
| 5.5.5 | Acid Number and Sulfur | 96 |
| 5.5.6 | Total and Free Glycerol | 96 |
| 5.5.7 | Sulfated Ash | 96 |
| 5.5.8 | Kinematic Viscosity | 96 |
| 5.5.9 | Derived Cetane Number (DCN) | 97 |
| 5.6 | Result V - Engine performance characteristics of coconut biodiesel | 97 |

| | | |
|-----------------------------|--|-----|
| 5.6.1 | Power and Torque | 98 |
| 5.6.2 | Brake Specific Fuel Consumption (BSFC) | 100 |
| 5.6.3 | Engine exhausts properties: temperature and emissions | 102 |
| 5.6.3.1 | Exhaust Temperature | 102 |
| 5.6.3.2 | Pollutant Gases: Nitrogen Oxide (NO _x) and Carbon Dioxide (CO ₂) | 105 |
| 6 | CONCLUSION AND RECOMMENDATION | 109 |
| 6.1 | Conclusion | 109 |
| 6.2 | Recommendations | 111 |
| REFERENCES | | |
| APPENDIX A | | 112 |
| APPENDIX B | | 117 |
| APPENDIX C | | 119 |
| APPENDIX D | | 123 |
| APPENDIX E | | 129 |
| APPENDIX F | | 132 |
| BIODATA OF STUDENT | | 134 |
| LIST OF PUBLICATIONS | | 140 |
| | | 141 |