



UNIVERSITI PUTRA MALAYSIA

**THERMOPHYSICAL CHARACTERIZATION OF CERAMIC  
SYNTHESIZED FROM RICE HUSK ASH AND ERBIUM OXIDE**

**HASLINAWATI MOHD MUSTAPHA**

**FS 2010 46**

**THERMOPHYSICAL CHARACTERIZATION OF  
CERAMIC SYNTHESIZED FROM RICE HUSK ASH  
AND ERBIUM OXIDE**

**HASLINAWATI MOHD MUSTAPHA**

**MASTER OF SCIENCE  
UNIVERSITI PUTRA MALAYSIA  
2010**



**THERMOPHYSICAL CHARACTERIZATION OF CERAMIC SYNTHESIZED FROM  
RICE HUSK ASH AND ERBIUM OXIDE**

**By**

**HASLINAWATI MOHD MUSTAPHA**

**Thesis submitted to the School of Graduate Studies, Universiti Putra Malaysia, in  
Fulfilment of the Requirements for the Degree of Master of Science.**

**December 2010**



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of  
the requirement for the degree of Master of Science.

**THERMOPHYSICAL CHARACTERIZATION OF CERAMIC SYNTHESIZED  
FROM RICE HUSK ASH AND ERBIUM OXIDE**

By

**HASLINAWATI MOHD MUSTAPHA**

**December 2010**

**Chair : Khamirul Amin Matori, PhD**

**Faculty : Faculty of Science.**

In this study, the chemical contents of rice husk (RH) were determined using Inductive Couple Plasma (ICP). Various types of acid were used in the leaching process of the RH samples. ICP data shows that, samples that have undergone acid leaching contained a lower percentage of metallic elements compared to untreated sample. Hydrochloric acid (HCl) leaching of husk is superior to Sulfuric acid ( $H_2SO_4$ ) and Nitric acid ( $HNO_3$ ) for removing the metallic elements. It also contained more percentage of silica ( $SiO_2$ ) which is 94.61%, treated with  $HNO_3$  (93.71%), treated with  $H_2SO_4$  (92.56%) and for untreated (85.48%).

For heat treatment, the required combustion temperature is 500 °C and above to produce white ash within a reasonable time (~1 hour in this study). By heat treating rice husk ash (RHA) at 800 °C for 2 hours, all the carbon contained in RHA (black ash) was removed

and an amorphous white rice husk ash (WRHA) is produced. The weight loss of RHA sample treated at 500 °C was found to be about 80.71% and it was increased to 88.64% after treatment at 800 °C.

Ceramics produced from WRHA and Erbium Oxide ( $\text{Er}_2\text{O}_3$ ) was made by mixing, milling, grinding, pressing and sintering procedure. They were well analyzed by XRD, scanning electron microscopy (SEM) and energy dispersive x-ray (EDX), fourier transform infrared (FTIR) and laser flash apparatus (LFA). The XRD revealed the changes in crystal phase due to sintering temperature. Cristobalite and trydimitte phase were observed from all compositions. While for sample with addition of 10% and 20% of  $\text{Er}_2\text{O}_3$ ,  $\text{Er}_2\text{O}_3$  and  $\text{Er}_2\text{O}_7\text{Si}_2$  phase were observed as well as cristobalite and trydimitte phase. SEM is useful tool to analyze structural changes that occur at the surface of ceramics. The morphology analysis of samples showed that the microstructures of samples are related to the phase of the crystal. On the other hand, EDX analysis confirm the composition of elements contain in the samples. FTIR analysis showed the chemical group presents in the samples. The main band is observed as Si-O-Si stretching band in all samples due to its silica composition. Laser flash system was used to measure thermal diffusivity in order to compare the structural of samples with their thermal features. It was shown that thermal diffusivity is increased as sintering temperature increased. Changes in crystal phase as proved by XRD (from cristobalite to tridymite and  $\text{Er}_2\text{O}_3$  to cristobalite and tridymite) are closely followed by thermal diffusivity. Thermal diffusivity is also dominated by their microstructure, density and temperature.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai  
memenuhi keperluan untuk ijazah Master Sains

**TERMOFIZIKAL PENCIRIAN SERAMIK SINTESIS DARIPADA SEKAM  
PADI DAN ERBIUM OXIDE**

Oleh

**HASLINAWATI MOHD MUSTAPHA**

**Disember 2010**

**Pengerusi : Khamirul Amin Matori, PhD**

**Fakulti : Fakulti Sains**

Dalam kajian ini, kandungan kimia di dalam sekam padi (RH) di kira menggunakan ICP.

Berbagai jenis asid digunakan dalam proses penyingkiran sampel sekam padi.

Keputusan ICP menunjukkan sampel yang telah melalui proses penyingkiran mengandungi kandungan bahan logam yang sedikit berbanding sampel yang tidak malalui proses tersebut. Asid hidroklorik (HCl) lebih baik daripada asid sulfuric ( $H_2SO_4$ ) dan asid nitric ( $HNO_3$ ) untuk menyingkirkan bahan logam. Ia juga mengandungi lebih banyak peratus silika ( $SiO_2$ ) iaitu 94.61%, rawatan dengan  $HNO_3$  (93.71%), rawatan dengan  $H_2SO_4$  (92.56%) and tanpa rawatan (85.48%).

Bagi perawatan haba, suhu pembakaran yang diperlukan ialah 500 °C dan keatas untuk menghasilkan abu putih di dalam masa yang sesuai (~1 jam bagi kajian ini). Dengan

merawat RHA pada suhu 800 °C untuk 2 jam, semua kandungan karbon didalam RHA (abu hitam) telah disingkirkan dan abu putih (WRHA) amorfos dihasilkan. Peratus kehilangan berat sampel RHA yang di bakar pada suhu 500 °C adalah sebanyak 80.71% dan nilai tersebut meningkat kepada 88.64% selepas di bakar pada suhu 800 °C.

Seramik yang dihasilkan daripada WRHA dan Erbium Oxide ( $\text{Er}_2\text{O}_3$ ) di buat melalui proses pencampuran, pengisaran, pemampatan, dan pembakaran. Sampel tersebut telah dianalisis menggunakan XRD, FTIR, DTA, SEM and EDX. XRD mendedahkan perubahan didalam fasa hablur berhubung dengan suhu pembakaran. Fasa cristobalite dan trydimite telah di dapati dalam semua komposisi sampel. Manakala penambahan 10 dan 20 peratus komposisi  $\text{Er}_2\text{O}_3$ , fasa yang terlibat ialah  $\text{Er}_2\text{O}_3$  dan  $\text{Er}_2\text{O}_7\text{Si}_2$  serta cristobalite dan trydimite. SEM merupakan alat yang berguna untuk menganalisa perubahan struktur yang berlaku pada permukaan seramik. Analisis morfologi sampel menunjukkan bahawa struktur mikro sampel berkait dengan fasa kristal yang berlaku. Analisis EDX mengesahkan komposisi unsur-unsur di dalam sampel. Analisis FTIR menunjukkan kumpulan kimia yang wujud dalam sampel. Kumpulan utama diperhatikan sebagai kumpulan Si-O-Si dalam semua sampel adalah disebabkan oleh komposisi silikanya. Sistem laser digunakan untuk mengukur penyerakan haba untuk membandingkan struktur sampel dengan ciri-ciri habanya. Ia telah menunjukkan bahawa penyerakan haba meningkat dengan suhu pembakaran. Perubahan fasa kristal sampel sebagaimana dibuktikan melalui XRD (daripada cristobalite kepada trydimite dan  $\text{Er}_2\text{O}_3$  kepada cristobalite kepada trydimite) diikuti rapat oleh penyerakan haba. Penyerakan haba juga didominasi oleh struktur mikro, ketumpatan dan suhu

## **ACKNOWLEDGEMENTS**

Firstly, I would like to thank my lovely parents Mohd Mustapha bin Mat Adam and Wan Ruhani binti Wan Ismail for their support and love. I am also grateful to my brother Hafizullah bin Mohd Mustapha and my sister Nur ‘Izzati binti Mohd Mustapha for their care and understanding. I really appreciate all they have done for me. Not forgotten for my future husband, Mhd Dzul Fadli bin Kharul Anuar for his encouragement and support during my study.

I am also extending my deepest gratitude and appreciation to my supervisor Dr. Khamirul Amin Matori. Without his assistance and guidance, valuable advice, constant encouragement, critical comments and suggestion, it is pretty hard for me to complete this study. Also, I would like to express my thanks to Assoc. Prof. Dr. Zaidan Abd Wahab for being such a helpful and dedicated advisor towards the end of my study.

My thanks also go to my fellow friends for giving me emotional stability and helped me with my work.

Finally I would like to acknowledge University Putra Malaysia for funding this research through Graduate Research Fellowship scheme (GRF) and giving me the opportunity to undertake and complete this field of study.

I certify that an Examination Committee has met on 30 December 2010 to conduct the final examination of Haslinawati binti Mohd Mustapha on his Master of Science thesis entitled “Thermophysical Characterization of Ceramic Synthesized from Rice Husk Ash and Erbium Oxide” in accordance with University Putra Malaysia (Higher Degree) Act 1980 and University Putra Malaysia (Higher Degree) Regulation 1981. The Committee recommends that the student be awarded the degree of Master of Science.

Members of Examination Committee were as follows:

Prof. Dr. Azmi Zakaria  
Lecturer  
Faculty of Science  
University Putra Malaysia.  
(Chairman)

Dr. Zulkifly Abbas  
Lecturer  
Faculty of Science  
University Putra Malaysia  
(Internal Examiner)

Assoc. Prof. Dr. Jumiah Hassan  
Lecturer  
Faculty of Science  
University Putra Malaysia  
(Internal Examiner)

Assoc. Prof. Dr. Umi Sarah Jais  
Lecturer  
Faculty of Applied Sciences  
University Technology MARA (UiTM)  
(External Examiner)

---

**SHAMSUDDIN SULAIMAN, PhD.**  
Professor and Deputy Dean  
School of Graduate Studies  
University Putra Malaysia  
Date: 30 December 2010

This thesis submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for degree of Master of Science. The members of the Supervisory Committee are as follows:

**Khamirul Amin Matori, PhD**

Lecturer

Faculty of Science

Universiti Putra Malaysia

(Chairman).

**Zaidan Abdul Wahab, PhD**

Associate Professor

Faculty of Science

Universiti Putra Malaysia

(Member).

---

**HASANAH MOHD GHAZALI, PhD**

Professor and Dean

School of Graduate Studies

Universiti Putra Malaysia

Date:



## **DECLARATION**

I declare that the thesis is my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institutions.

---

**HASLINAWATI MOHD MUSTAPHA**

Date: 30 December 2010

## TABLE OF CONTENTS

	Page
<b>ABSTRACT</b>	ii
<b>ABSTRAK</b>	iv
<b>ACKNOWLEDGEMENTS</b>	vi
<b>APPROVAL</b>	vii
<b>DECLARATION</b>	ix
<b>LIST OF TABLES</b>	xiii
<b>LIST OF FIGURES</b>	xiv
<b>LIST OF ABBREVIATIONS</b>	xvii
 <b>CHAPTER</b>	
<b>1 INTRODUCTION</b>	
1.1 Problem Statement	1
1.2 Introduction	1
1.3 Objectives	4
<b>2 LITERATURE REVIEW</b>	
2.1 Introduction	5
2.2 Ceramic	
2.2.1 Basic Theory of Ceramic	5
2.2.2 Properties of Ceramic	6
2.2.3 Ceramic Processing	7
2.2.3.1 Calculating and Processing of the Batch	8
2.2.3.2 Forming Process	9
2.2.3.3 Sintering	9
2.3 Glass and Ceramic From Silica	11
2.4 Silica From Rice Husk	
2.4.1 Rice Husk Production	14
2.4.2 Chemical Composition in Rice Husk Ash	15
2.4.3 Phase Formation of Rice Husk Ash	17
2.4.4 Microstructure of Rice Husk Ash	18
2.4.5 Infrared Spectroscopy for Basic Structure of Rice Husk	20
2.4.6 Thermal Analysis	21
2.5 Applications of Rice Husk	
2.5.1 Rice Husk in Cement	24

2.5.2	Zeolite from Rice Husk	25
2.5.3	Silicon Carbide from Rice Husk	26
2.5.4	Silicon Nitride from Rice Husk	27
2.5.5	Rice Husk in Polymers	28
<b>3 EXPERIMENTAL TECHNIQUE</b>		
3.1	Introduction	31
3.2	Rice Husk Preparation	31
3.3	Chemical Analysis	32
3.4	Composition of Samples	36
3.5	Sample Preparation	
3.5.1	Mixing and Milling	36
3.5.2	Binder	37
3.5.3	Pressing	37
3.5.4	Sintering	38
3.6	Physical Appearance	38
3.7	Phase Formation	
3.7.1	X-ray Diffraction	39
3.7.2	Preparation of Samples for XRD	40
3.8	Microscopy	
3.8.1	Scanning Electron Microscopy	41
3.8.2	Energy Dispersive X-ray	42
3.8.3	Preparation of Samples for SEM/EDX	43
3.9	Fourier Transform Infrared Spectroscopy	
3.9.1	FTIR	44
3.9.1	Preparation of Samples for FTIR	45
3.10	Laser Flash Apparatus	
3.10.1	Laser Flash Apparatus (NETZSCH-LFA 457 MicroFlash)	46
3.10.2	Preparation of Samples for Laser Flash Apparatus	48
<b>4 RESULTS AND DISCUSSION</b>		
4.1	Introduction	50
4.2	The Weight Loss of Rice Husk	50
4.3	Chemical Composition	52
4.4	Physical Properties	
4.4.1	Density	56
4.4.2	Linear Shrinkage	60
4.5	Phase Formation	

4.5.1	Phase Formation of 100 wt. % WRHA	62
4.5.2	Phase Formation of 90 wt. % WRHA and 10 wt. % Er <sub>2</sub> O <sub>3</sub>	66
4.5.3	Phase Formation of 80 wt. % WRHA and 20 wt. % Er <sub>2</sub> O <sub>3</sub>	69
4.6	<b>Microstructural Characterization</b>	
4.6.1	Microstructural Characterization of 100 wt. % WRHA	72
4.6.2	Microstructural Characterization of 90 wt. % WRHA and 10 wt. % Er <sub>2</sub> O <sub>3</sub>	76
4.6.3	Microstructural Characterization of 80 wt. % WRHA and 20 wt. % Er <sub>2</sub> O <sub>3</sub>	81
4.7	<b>FTIR</b>	
4.7.1	FTIR Analysis of 100 wt. % WRHA	86
4.7.2	FTIR Analysis of 90 wt. % WRHA and 10 wt. % Er <sub>2</sub> O <sub>3</sub>	90
4.7.3	FTIR Analysis of 80 wt. % WRHA and 10 wt. % Er <sub>2</sub> O <sub>3</sub>	96
4.8	<b>Thermal Diffusivity</b>	
4.8.1	Thermal Diffusivity of 100 wt. % WRHA	101
4.8.2	Thermal Diffusivity of 90 wt. % WRHA and 10 wt. % Er <sub>2</sub> O <sub>3</sub>	102
4.8.3	Thermal Diffusivity of 80 wt. % WRHA and 20 wt. % Er <sub>2</sub> O <sub>3</sub>	103
<b>5 CONCLUSION AND FUTURE WORK</b>		
5.1	Conclusion	107
5.1	Future Work	109
<b>REFERENCES/BIBLIOGRAPHY</b>		110
<b>APPENDICES</b>		117
<b>BIODATA OF STUDENT</b>		118
<b>LIST OF PUBLICATIONS</b>		119