



**STRUCTURAL, MECHANICAL AND BIOACTIVITY PROPERTIES OF  
 $\text{SiO}_2\text{-CaO-B}_2\text{O}_3\text{-Na}_2\text{O-P}_2\text{O}_5\text{-CaF}_2$  GLASS AND GLASS-CERAMICS  
UTILIZED EGGSHELL AS CALCIUM SOURCE**

By

**LOH ZHI WEI**

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,  
in Fulfilment of the Requirements for the Degree of Doctor of Philosophy**

**January 2024**

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment  
of the requirement for the degree of Doctor of Philosophy

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SiO<sub>2</sub>-CaO-B<sub>2</sub>O<sub>3</sub>-Na<sub>2</sub>O-P<sub>2</sub>O<sub>5</sub>-CaF<sub>2</sub> GLASS AND GLASS-CERAMICS  
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**January 2024**

**Chairman : Mohd Hafiz Mohd Zaid, PhD**

**Faculty : Science**

As humans age, the deterioration of teeth in adults and even children are unprotected and easy to fracture due to the loss of mechanical strength over time. Hence, biomaterials such as bioactive glasses have gained significant scientific interest from researchers for application in dentistry. In the present work, the bioactive glass was synthesized through a melt-quenching approach with the configuration of 45SiO<sub>2</sub>-25CaO-10B<sub>2</sub>O<sub>3</sub>-10Na<sub>2</sub>O-(10-x)P<sub>2</sub>O<sub>5</sub>-xCaF<sub>2</sub>, where  $x = 0, 2, 4, 6, 8$  and 10 wt.%. The eggshell waste was chosen as the calcium source. Considering the intrinsic brittleness of the glass, heat treatment is performed to improve the mechanical properties of the bioactive glass. Meanwhile, both samples will be immersed in phosphate buffer saline for 7 and 14 days to evaluate their bioactivity and biodegradability properties. Among the samples, BG3F with a CaF<sub>2</sub>/P<sub>2</sub>O<sub>5</sub> ratio of 6/4 demonstrated the optimal performance based on the structural, mechanical, bioactivity and biodegradability properties. The density of BG3F showed increment with heat treatment temperature,

reaching a maximum value of  $2.626 \text{ g/cm}^3$  at a temperature of  $700^\circ\text{C}$ . X-ray diffraction analysis revealed the phase formation of fluorapatite, wollastonite, and cuspidine in the BG3F samples after heat treatment. In terms of mechanical results, BG3F heat-treated at  $700^\circ\text{C}$  revealed the highest microhardness range of 6.72 GPa to 6.95 GPa and fracture toughness of  $3.55 \text{ MPa}\cdot\text{m}^{1/2}$  to  $3.62 \text{ MPa}\cdot\text{m}^{1/2}$  after 14 days of immersion. Furthermore, the pH of the glass and glass-ceramics samples showed an increasing trend ranging from 7.437 to 9.636 and 7.437 to 8.217, respectively, after being immersed for 14 days, indicating their bioactivity in response to the immersion medium. In conclusion, the BG3F sample with a heat treatment of  $700^\circ\text{C}$  revealed the optimum performance, demonstrating the highest mechanical properties while maintaining bioactivity and biodegradability. The microhardness of the studied samples is comparable to the human enamel (2.00 – 5.00 GPa) and the commercial dental glass-ceramics (4.00 – 6.50 GPa) by more than 39% and 6.92%, respectively. Therefore, this research contributes to a novel approach by using wastes in fabricating bioactive glass while improving the mechanical properties of glass-ceramics for dental applications.

**Keywords:** Waste materials, Bioactive glass, Glass-ceramics, Mechanical, Dental applications

**SDG:** GOAL 4: Quality Education and GOAL 9: Industry, Innovation and Infrastructure

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**SIFAT STRUKTUR, MEKANIKAL DAN BIOAKTIVITI  $\text{SiO}_2\text{-CaO-B}_2\text{O}_3\text{-Na}_2\text{O-P}_2\text{O}_5\text{-CaF}_2$  KACA DAN KACA-SERAMIK MENGGUNAKAN SISA KULIT TELUR SEBAGAI SUMBER KALSIMUM**

Oleh

**LOH ZHI WEI**

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Seiring bertambahnya usia manusia, penurunan keadaan gigi pada orang dewasa dan bahkan kanak-kanak tidak dilindungi dan mudah patah disebabkan kehilangan kekuatan mekanikal sepanjang masa. Oleh itu, bahan bioaktif seperti kaca bioaktif telah menarik minat saintifik yang signifikan daripada penyelidik untuk aplikasi dalam bidang pergigian. Dalam kajian ini, kaca bioaktif disintesis melalui teknik lebur dan pelindap secara konvensional dengan konfigurasi  $45\text{SiO}_2\text{-}25\text{CaO-}10\text{B}_2\text{O}_3\text{-}10\text{Na}_2\text{O-}(10-x)\text{P}_2\text{O}_5\text{-}x\text{CaF}_2$ , di mana  $x = 0, 2, 4, 6, 8$ , dan 10% berat. Sisa kulit telur telah dipilih sebagai sumber kalsium. Mengambil kira kerapuhan intrinsik kaca, rawatan haba dijalankan untuk meningkatkan sifat mekanikal kaca bioaktif. Sementara itu, kedua-dua kaca bioaktif dan kaca-seramik sampel akan direndam dalam larutan garam penampan fosfat selama 7 dan 14 hari untuk menilai sifat bioaktiviti dan biodegradasi mereka. Antara sampel-sampel tersebut, BG3F dengan nisbah  $\text{CaF}_2/\text{P}_2\text{O}_5$  6/4 menunjukkan prestasi optimum berdasarkan sifat-sifat struktur, mekanikal, bioaktiviti, dan biodegradasi. Ketumpatan BG3F meningkat dengan suhu rawatan haba, mencapai

nilai maksimum  $2.626 \text{ g/cm}^3$  pada suhu  $700^\circ\text{C}$ . Analisis pembelauan sinar-X menunjukkan pembentukan fasa fluorapatit, wollastonit, dan cuspidin dalam sampel BG3F selepas rawatan haba. Dari segi keputusan mekanikal, BG3F yang dirawat haba pada  $700^\circ\text{C}$  menunjukkan julat kekerasan mikro Vickers tertinggi dari  $6.72 \text{ GPa}$  hingga  $6.95 \text{ GPa}$  dan kekuatan fraktur dari  $3.55 \text{ MPa}\cdot\text{m}^{1/2}$  hingga  $3.62 \text{ MPa}\cdot\text{m}^{1/2}$  selepas 14 hari rendaman. Selain itu, pH sampel-sampel kaca dan kaca-seramik menunjukkan trend peningkatan dari 7.437 hingga 9.636, dan 7.437 hingga 8.217, masing-masing, selepas direndam selama 14 hari, menunjukkan bioaktiviti mereka sebagai respons kepada medium rendaman. Kesimpulannya, sampel BG3F dengan rawatan haba pada suhu  $700^\circ\text{C}$  menunjukkan prestasi optimum dengan sifat mekanikal tertinggi pada masa yang sama mengekalkan bioaktiviti dan biodegradasi. Kekerasan mikro sampel yang dikaji adalah setanding dengan enamel manusia ( $2.00 - 5.00 \text{ GPa}$ ) dan kaca-seramik pergigian komersial ( $4.00 - 6.50 \text{ GPa}$ ) masing-masing lebih daripada 39% dan 6.92%. Oleh itu, penyelidikan ini menyumbang kepada pendekatan baru dengan menggunakan bahan buangan dalam fabrikasi kaca bioaktif sambil menambah baik sifat mekanikal kaca-seramik untuk aplikasi pergigian.

Kata Kunci: Bahan buangan, Kaca bioaktif, Kaca-seramik, Mekanikal, Aplikasi pergigian

SDG: *GOAL 4: Quality Education and GOAL 9: Industry, Innovation and Infrastructure*

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## LIST OF ABBREVIATIONS

A	Area
a	Half-length of the indented area
Å	Angstrom
atm	Atmospheric pressure
c	Half-length of the cracked area
cm	Centimetre
d	Spacing distance between the adjacent lattice planes
°C	Degree Celsius
F	Maximum load
g	Gram
GPa	Gigapascals
kV	Kilovolts
m	Meter
mA	Milliampere
min(s)	Minute(s)
mL	Millilitre
mm	Millimeter
mol. %	Molar percent
N	Newton
°C	Degree Celsius
s	Second
wt. %	Weight percent
μ	Micro
Al <sub>2</sub> O <sub>3</sub>	Aluminium oxide
B <sub>2</sub> O <sub>3</sub>	Boron Trioxide
Ca(OH) <sub>2</sub>	Calcium Hydroxide
Ca <sub>10</sub> (PO <sub>4</sub> ) <sub>6</sub> F <sub>2</sub>	Fluorapatite
Ca <sub>3</sub> Si <sub>3</sub> O <sub>9</sub>	Pseudo-wollastonite
Ca <sub>4</sub> Si <sub>2</sub> O <sub>7</sub> F <sub>2</sub>	Cuspidine
CaCl <sub>2</sub>	Calcium chloride
CaCO <sub>3</sub>	Calcium carbonate / Calcite

CaF <sub>2</sub>	Calcium fluoride
CaO	Calcium oxide
CaSiO <sub>3</sub>	Wollastonite
CO <sub>2</sub>	Carbon dioxide
CuO	Copper (II) oxide
Fe <sub>2</sub> O <sub>3</sub>	Iron (III) oxide
K <sub>2</sub> O	Potassium oxide
MgF <sub>2</sub>	Magnesium fluoride
MgO	Magnesium oxide
Na <sub>2</sub> CO <sub>3</sub>	Sodium carbonate
Na <sub>2</sub> O	Sodium oxide
P <sub>2</sub> O <sub>5</sub>	Phosphorus pentoxide
SiO <sub>2</sub>	Silicon dioxide
SO <sub>3</sub>	Sulphur Trioxide
SrO	Strontium oxide
AS	Artificial Saliva
BO	Bridging oxygen
CAp	Carbonate Apatite
CS	Compressive strength
EDX	Energy dispersive X-ray
FAp	Fluorapatite
FTIR	Fourier transform infrared spectroscopy
HA	Hydroxyapatite
HCA	Hydroxy carbonate apatite
HV	Vickers microhardness
ICSD	Inorganic Crystal Structure Database
IR	Infrared ray
ISO	International Organization for Standardization
K <sub>IC</sub>	Fracture toughness
M	Molar mass
MPa	Megapascals
NBO	Non-bridging oxygen
PBS	Phosphate buffer saline



pH	Potential of hydrogen
PMMA	polymethyl methacrylate
S	Standard deviation
SBF	Simulated body fluid
SEM	Scanning electron microscopy
TEM	Transmission electron microscopy
$T_f$	Fictive temperature
$T_G$	Maximum crystallization temperature
$T_g$	Glass transition temperature
$T_m$	Glass melting temperature
$T_N$	Nucleation temperature
$T_{NG}$	Crystallization temperature
Tris	Tris-hydroxymethyl-aminomethane buffer
$V_m$	Molar volume
$W$	Weight
$\bar{x}$	Average value
XRD	X-ray diffraction
XRF	X-ray Fluorescence
$\alpha$	Alpha
$\beta$	Beta
$\theta$	Diffraction angle
$\lambda$	Wavelength
$\rho$	Density

# **CHAPTER 1**

## **INTRODUCTION**

### **1.1 Introduction**

This chapter aims to provide an overview of the research background related to biomaterial, particularly bioactive glass and glass-ceramics. The utilization of waste resources, such as eggshell waste, is used in this research to synthesize the bioactive glass and glass-ceramics as potential candidates for dental materials. The chapter extensively addresses the problem statement, outlining the objectives to overcome the challenges. Furthermore, the novelty of research work is discussed, highlighting its unique contributions to the field. The scope of the study is also described, indicating the specific areas and aspects that will be covered. Additionally, the importance of the study is emphasized, highlighting its potential impact and relevance in the relevant field. Lastly, an outline of the thesis is provided, giving a brief overview of the structure and organization of the research work.

### **1.2 Research background**

As the world population ages, the deterioration of bones and teeth in adults and even children are vulnerable to fractures due to the gradual decrease in mechanical strength over a period of time. The challenge is to replace the fractured area with a material that can retain the same properties as the host materials before they become defective. Therefore, the invention of biomaterials has played a crucial role in addressing this challenge. Biomaterials are substances that can interact with the human body and act

as an excellent alternative to conventional materials in many biomedical and dental applications. They are extensively used in bone and tooth replacement, repair and regeneration, and implantation (Filip et al., 2022).

Larry Hench pioneered the creation of the initial bioactive glass, 45S5 Bioglass®, during the 1960s. He developed a material capable of bonding with hard and soft tissues without causing rejection by the body. Consequently, this breakthrough garnered attention from various experts, including orthopedic surgeons, dental researchers, biomechanics specialists, and biologists, who were eager to explore the potential of biomaterials in different applications (Hench, 2006). However, despite its groundbreaking properties, bioglass does have certain limitations. One drawback is its relatively lower mechanical strength compared to other materials. Additionally, the composition of bioglass includes higher amounts of alkali elements, such as sodium, which lead to toxicity and reduced bioactivity, which may pose challenges in specific applications (Damen & Ten Cate, 1992; Ali et al., 2014). These shortcomings highlight the need for further research and development in bioactive materials to address these concerns and enhance their overall performance by modifying their compositions.

In 1977, Professor Ulrich Gross and his colleagues made a significant advancement by discovering bioactive glass-ceramics. They found that these glass-ceramics materials had a stronger bonding strength to the bone than bioactive glass. Furthermore, the progression of glass-ceramics was propelled forward by the innovation of A/W bioactive glass-ceramics (Kokubo et al., 1982). These materials demonstrated enhanced mechanical strength, toughness and stability in the physiological environment. As a result, they have been successfully utilized in vertebral replacement

and spinal repair procedures, showcasing their practical and clinical potential. In addition, the glass-ceramics with better mechanical strength, specifically for dental restoration, were patented by previous researchers in the 1980s (Barrett et al., 1980). With the opened up new possibilities after the discovery of bioactive glass-ceramics, this research highlights the importance of tuning the composition and controlling the heat treatment temperature as crucial steps in the synthesis process. These optimizations ensure that the resulting materials meet the specific requirements of dental applications, further enhancing their suitability.

Initially, the base composition of the bioactive glass includes silicon dioxide ( $\text{SiO}_2$ ), sodium oxide ( $\text{Na}_2\text{O}$ ), calcium oxide ( $\text{CaO}$ ), and phosphorus pentoxide ( $\text{P}_2\text{O}_5$ ). In order to have better performance of the designed glass and glass-ceramics systems, each composition plays a vital role in modifying the glass structure. The bioactive glass consists of three crucial components. For instance, the  $\text{SiO}_2$  content should remain below 60%, with a substantial presence of  $\text{Na}_2\text{O}$  and  $\text{CaO}$ , along with a higher  $\text{CaO}/\text{P}_2\text{O}_5$  ratio (Hench, 1991). In addition to the primary components, additional elements can be integrated into the glass composition to alter its properties. One such element is boron, which, when added in small amounts, can enhance the bioactivity and hardness of the glass. The inclusion of boron has been shown to improve biocompatibility, biodegradability, and anti-bacterial properties. However, it is essential to note that adding excessive amounts of boron beyond a certain threshold can adversely affect the glass (Deilmann et al., 2020; Yusof et al., 2022). Apart from that, incorporating fluoride into the glass system forms fluorapatite, the main component of human enamel and dentin (Chen et al., 2022b). The inclusion of fluoride

helps improve the mechanical properties of the glass and plays a role in preventing dental cavities by inhibiting the demineralization of enamel and dentin (Shah, 2016a). The rapid economic growth in developing countries such as Malaysia has indirectly boosted solid waste generation. Effective ways to manage solid waste should be considered in reducing the percentage of solid waste in Malaysia (Moh & Manaf, 2017). Hence, to address the solid waste issue, applying the concept of recycling, reducing and reusing methods is preferable, cost-effective and environmentally friendly for waste management. These approaches promote the transformation of waste into valuable resources, minimize waste generation at the source, and encourage the reuse of items. Moreover, environmental technology has been promoted in every country, where it is a new technology introduced to protect the Earth against environmental pollution (Doble & Kruthiventi, 2007). Disposing of waste materials in landfills is not a productive approach; however, these materials can be repurposed into valuable products with considerable potential for industrial applications (Waheed et al., 2020).

At the beginning of fabricating the bioactive glasses from previous work, the starting materials used are pure chemical elements, which will lead to the high costs of the final materials (De Caluwe et al., 2016; Pei et al., 2020). The challenge faced by researchers in the twenty-first century is to fabricate a cost-effective bioactive glass system with good performance for the final application (Hench & Thompson, 2010). Therefore, eggshells are selected as the calcium source material for synthesizing the bioactive glass. Eggshells contain 94% of calcium carbonate ( $\text{CaCO}_3$ ). The purity of the CaO obtained from the eggshells after calcination is approximately 96% to 98% (Mohadi et al., 2016; Palakurthy et al., 2020; Jakfar et al., 2023). Utilizing eggshells

in the fabrication of bioactive glass offers an environmentally friendly solution for disposing of waste eggshells, minimizing landfill waste. Furthermore, eggshell-based glass possesses distinct properties such as an anti-bacterial effect and improved hardness (Kamalanathan et al., 2014; Tshizanga et al., 2017; Onwubu et al., 2019) that make it well-suited for dental applications, providing a sustainable substitute for conventional glass materials.

In the present study, a series of bioactive glasses are fabricated using eggshell waste as a source of calcium through a conventional melt-quenching technique. The glass compositions are varied with different fluoride and phosphate ( $\text{CaF}_2/\text{P}_2\text{O}_5$ ) ratios, and a controlled heat treatment is employed to transform the glass into glass-ceramics. The physical, structural and mechanical properties of the bioactive glass and glass-ceramics are characterized, including density, molar volume, phase formation, molecular structure, surface morphology, elemental analysis, compressive strength, microhardness and fracture toughness. Immersion tests are conducted to evaluate the in-vitro bioactivity and biodegradability properties of the synthesized materials. In-vitro bioactivity analysis assesses the phase formation, molecular structure, surface morphology and elemental analysis before and after immersion. In contrast, biodegradability analysis involves the pH changes, weight gain or loss, and mechanical properties before and after immersion. This research aims to develop a novel approach for fabricating bioactive glass from waste materials and improving the mechanical properties of bioactive glass-ceramics for dental applications.

### 1.3 Problem statement

There is a growing demand for dental materials due to various factors like an increasingly ageing population, advancement in dental procedures and rising awareness about oral health. This demand has prompted researchers to focus on developing innovative dental materials, such as bioactive glass, to meet the evolving needs. Initial bioactive glass production often starts with pure chemical elements as starting materials, which can be expensive and lead to higher production costs. Besides, the disposal of waste materials such as eggshells in Malaysia has become a significant concern. Around 13 million tonnes of eggs were produced in 2021, as reported by the Food and Agriculture Organization of the United Nations (2022). Some researchers use waste materials such as clam shells, soda-lime-silica glass waste, and rice husk to synthesize the bioactive glass system for dental fillings and biomedical applications but not for dental prosthetics (Khiri et al., 2020; Nayak et al., 2010b). Hence, in this study, the fabrication of bioactive glass involves the utilization of waste material such as eggshell as a calcium source is developed. Using waste materials as an alternative to the synthetic form reduces the overall production cost and shows several biologically beneficial effects on the bioactive glass system.

In the past few decades, significant research efforts have been focused on the development of biomaterials, particularly calcium phosphate-based ceramics and bioactive glass in bone substitution, while glass ionomer cement for dental restoration materials (Alatawi et al., 2019; Borkowski et al., 2020). However, there has been minimal research on bioactive glass designed explicitly for prosthetics or implants. This is because of the high alkali content and low toughness of the bioactive glass, which restrict their application in the dental field. To address these limitations,



optimizing the composition of the starting materials to achieve properties comparable to those of human enamel and dentin is crucial. Research conducted earlier by Chen et al. (2022), Prasad et al. (2020) and Esmati et al. (2018) have shown that incorporating fluoride into the glass system enhances density and microhardness, promotes apatite development and provides anti-bacterial properties. However, excessive fluoride content can adversely affect the bioactivity and mechanical properties. In addition, the presence of phosphate is essential for the bioactivity of the glass, with optimal apatite growth observed within 2% to 10% of the phosphate content range (Adam et al., 2021). Nevertheless, an excess of more than 7% phosphate inhibits the bioactivity reaction and affects the strength of glass. (Marghussian & Mesgar, 2000; Al-eesa et al., 2017). Therefore, in this study, the fluoride and phosphate ( $\text{CaF}_2/\text{P}_2\text{O}_5$ ) ratios are essential in fabricating the bioactive glass system to achieve optimal properties that improve mechanical properties while maintaining bioactivity.

Considering the intrinsic brittleness of bioactive glass that could not withstand the high mechanical load (Bellucci et al., 2011), a heat treatment process will be performed with the aim of enhancing the mechanical properties of the bioactive glass system. The commercial 45S5 Bioglass® has lower mechanical properties, limiting its properties to be used in various applications (Skallevold et al., 2019). Alternatively, the apatite-wollastonite glass-ceramics developed by Kokubo & Takadaman (1982) exhibit higher mechanical properties but are primarily utilized in bone substitution. Similarly, previous studies reported that apatite-mullite glass-ceramics demonstrate improved mechanical strength for bone substitution and dental restorative materials (Freeman et al., 2003; Fathi et al., 2005). However, the presence of aluminium oxide ( $\text{Al}_2\text{O}_3$ ) content in these materials may inhibit their bioactivity. Hence, developing bioactive



glass-ceramics with properties that closely resemble human enamel, including high microhardness and fracture toughness, is desirable (Dimitriadis et al., 2022). Therefore, selecting the optimal heat treatment temperature for the bioactive glass system is expected to improve the mechanical strength of the final product.

Moreover, in vitro studies are conducted to evaluate the potential for apatite formation and to assess the suitability of the bioactive glass and glass-ceramics for application in living organisms. Previous research was focused on studying the bioactivity and biodegradability properties of bioactive glass (Al-Noaman et al., 2012b; Al-eesa et al., 2017; Sriranganathan et al., 2017). However, there is limited research on the properties of glass-ceramics, as the conversion of bioactive glass to glass-ceramics through heat treatment may potentially affect these properties. The ion exchange phenomena occurring at the interface between the surface of the material and medium are primarily associated with the residual glassy phase. Glass-ceramics, having a denser structure, exhibit a reduced contact area between the materials and the immersed medium (Panah et al., 2022). Conversely, some studies have shown that although apatite formation may slow down when the crystallinity reaches 100%, it does not completely inhibit the bioactivity and biodegradability properties (Kaur et al., 2019). Therefore, a comprehensive analysis of the bioactivity and biodegradability properties of the novel bioactive glass and glass-ceramics, both before and after PBS immersion, will be thoroughly evaluated.

#### **1.4 Research objective**

The main objective of this study is to develop and optimize the mechanical properties while maintaining the bioactivity properties of the glass and glass-ceramics derived from waste materials. This research involved designing glass compositions, implementing the melt-quenching method, precise control of the heat treatment process, and conducting immersion tests to evaluate the performance of the materials.

The specific objectives of this study are as follows:

1. To synthesize the bioactive glass derived from waste eggshell as a source of calcium.
2. To investigate the effect of fluoride and phosphate ratios on the physical, structural and mechanical properties of bioactive glass.
3. To determine the effect of heat treatment temperatures on the physical, structural and mechanical properties of bioactive glass-ceramics.
4. To evaluate the effect of immersion periods on the in-vitro bioactivity and biodegradability properties of bioactive glass and glass-ceramics.

#### **1.5 Novelty of research work**

This study aims to contribute new knowledge regarding the properties and cost-effectiveness of this innovative material. Hence, the research work in this study brings several novel aspects to the field:

1. This innovative approach presents a novel way of utilizing eggshells as a calcium source to produce bioactive glass materials for dental prostheses.

2. Development of sustainable bioactive glass and glass-ceramics with the optimal  $\text{CaF}_2/\text{P}_2\text{O}_5$  ratio and heat treatment temperature to obtain the ideal mechanical properties of the materials for dental materials, an alternative to traditional dental materials.
3. This novel material contributes to new knowledge on physical, structural, mechanical, bioactivity and biodegradability properties.

### 1.6 Scope of study

To achieve the objective of this study, the scope of the study includes the following:

1. A series of bioactive glasses based on the empirical formula of  $45\text{SiO}_2\text{--}25\text{CaO--}10\text{B}_2\text{O}_3\text{--}10\text{Na}_2\text{O--}(10\text{--}x)\text{P}_2\text{O}_5\text{--}x\text{CaF}_2$ , where  $x = 0, 2, 4, 6, 8$  and  $10$  wt.% will be prepared using the conventional melt-quenching technique.
2. The chemical composition of the eggshell waste will be measured using X-ray fluorescence (XRF) spectroscopy to confirm the percentage of the chemical oxide and purity of CaO before synthesizing the bioactive glass.
3. The bioactive glasses will be subjected to heat treatment temperatures varying from  $600\text{ }^\circ\text{C}$  to  $900\text{ }^\circ\text{C}$  to produce bioactive glass-ceramics.
4. The physical, structural and mechanical properties of bioactive glasses and glass-ceramics will be analyzed using the Archimedes method, molar volume formulation, X-ray diffraction (XRD), Fourier transform infrared (FTIR) spectroscopy, scanning electron microscopy with energy dispersive X-ray (SEM-EDX), compressive strength, Vickers microhardness and fracture toughness.
5. The optimum bioactive glass and glass-ceramics samples will be selected and immersed in phosphate buffer saline medium for 7 and 14 days.

6. The bioactivity and biodegradability properties of the optimum samples before and after immersion will be evaluated using XRD, FTIR, SEM-EDX, change in pH, weight gain or loss, compressive strength, Vickers microhardness and fracture toughness.

### **1.7 Importance of study**

The study focuses on developing and optimizing bioactive glass and glass-ceramics derived from waste materials, reducing environmental impact and promoting the circular economy. Repurposing resources like eggshells showcases the transformation of waste into valuable products, lessening reliance on new materials and waste production. This research broadens biomaterial options, offering an eco-friendly alternative to conventional sources.

Besides, the increasing demand for biomaterials, including bioactive glass and glass-ceramics, arises from their appealing aesthetics, functionality, strength and bioactivity. Researchers globally, including from Japan, Malaysia, India and Pakistan, are exploring bioactive glasses for orthopedic and dental applications. Although dental materials are gaining popularity, the susceptibility to fractures remains a significant concern, leading to the need to replace these dental restorations. Indeed, the composition of materials and the parameters employed during processing, such as the heat treatment process, play a crucial role in modifying their properties to ensure suitability for specific applications.

Dental materials demand unique properties, including bioactivity, biodegradability, and strength. Hence, heat treatment is anticipated to enhance the physical, structural

and mechanical properties of the bioactive glass and glass-ceramics. In addition, the immersion tests conducted on the selected bioactive glass and glass-ceramics are expected to provide valuable insights into their bioactivity and biodegradability. Consequently, this research is expected to show the optimum properties of the final products can lead to a novel approach for the low-cost fabrication of material from waste materials with better mechanical properties to apply in dental applications, such as prosthetics and implants. Ultimately, these findings will shed fresh light on the properties and economic viability of this innovative material.

## **1.8 Outline of thesis**

The thesis consists of five chapters that are structured as follows. Chapter 1 provides a general overview of the research field and the need for biomaterials such as bioactive glass and glass-ceramics, the glass composition and eggshell-derived glass. This chapter also presents the problem statements, research objectives, novelty of research work, scope of study and importance of study. Chapter 2 reviews the utilization of eggshell waste, fabrication and types of bioactive glass and glass-ceramics. This chapter also reviews the physical, structural, mechanical, bioactivity and biodegradability properties of the bioactive glass and glass-ceramics from previous works conducted by other researchers in the field, serving as a guideline and reference for the current study. The methods used to prepare the bioactive glass and glass-ceramics samples and the characterization techniques employed throughout this study are explained in Chapter 3. Chapter 4 analyzes and discusses the results obtained from the experiments, focusing on the effects of  $\text{CaF}_2/\text{P}_2\text{O}_5$  ratios, varying heat treatment temperatures, and the outcomes after immersion tests. This chapter covers the physical, structural, mechanical, bioactivity and biodegradability properties of the bioactive

glass and glass-ceramics samples. Finally, Chapter 5 concludes the research findings and their significance and provides suggestions for future studies and areas of further exploration. The references, publications and conferences attended by the author will be placed in the last part of the thesis.



## REFERENCES

- Abd El Rahim, S. H., Melegy, A. A., & Hamzawy, E. M. A. (2017). Wollastonite-pseudowollastonite from silica fume, limestone and glass cullet composite. *InterCeram: International Ceramic Review*, 66(6), 232–236.
- Abdullah, A., & Mohammed, A. (2019). Scanning electron microscopy (SEM): A review. *Proceedings of 2018 International Conference on Hydraulics and Pneumatics – HERVEX, 2018*, 77–85.
- Abdulrahman, I., Tijani, H. I., Mohammed, B. A., Saidu, H., Yusuf, H., Ndejiko Jibrin, M., & Mohammed, S. (2014). From garbage to biomaterials: An overview on egg shell based hydroxyapatite. *Journal of Materials*, 2014, 1–6.
- Ab Llah, N., Jamaludin, S. B., Daud, Z. C., Zaludin, M. A. F., Jamal, Z. A. Z., Idris, M. S., & Osman, R. A. M. (2016). Corrosion behavior of Mg-3Zn/bioglass (45S5) composite in simulated body fluid (SBF) and phosphate buffered saline (PBS) solution. *AIP Conference Proceedings*, 1756(1), 030001.
- Abo-Naf, S. M., Khalil, E. S. M., El-Sayed, E. S. M., Zayed, H. A., & Youness, R. A. (2015). In vitro bioactivity evaluation, mechanical properties and microstructural characterization of Na<sub>2</sub>O–CaO–B<sub>2</sub>O<sub>3</sub>–P<sub>2</sub>O<sub>5</sub> glasses. *Spectrochimica Acta – Part A: Molecular and Biomolecular Spectroscopy*, 144, 88–98.
- Adam, S. N. F. S., Zainuddin, F., & Osman, A. F. (2021). Effect of varying phosphate content on the structure and properties of sol-gel derived SiO<sub>2</sub>–CaO–P<sub>2</sub>O<sub>5</sub> bio-glass. *Journal of Physics: Conference Series*, 2080(1), 012018.
- Agalit, H., Zari, N., Grosu, Y., Faik, A., & Maaroufi, M. (2020). Synthesis of high temperature TES materials from silicates wastes for application in solar tower power plants. *Solar Energy Materials and Solar Cells*, 218, 110763.
- Ahmad, R., Rohim, R., & Ibrahim, N. (2015). Properties of waste eggshell as calcium oxide catalyst. *Applied Mechanics and Materials*, 754, 171–175.
- Al-eesa, N. A., Wong, F. S. L., Johal, A., & Hill, R. G. (2017). Fluoride containing bioactive glass composite for orthodontic adhesives – ion release properties. *Dental Materials*, 33(8), 1324–1329.
- Al-Harbi, O. A., & Hamzawy, E. M. A. (2014). Stable wollastonite-cuspidine glass-ceramic using inexpensive raw materials. *Silicon*, 6(4), 257–264.
- Al-Harbi, N., Mohammed, H., Al-Hadeethi, Y., Bakry, A. S., Umar, A., Hussein, M. A., Abbassy, M. A., Vaidya, K. G., Al Berakdar, G., Mkawi, E. M., & Nune, M. (2021). Silica-based bioactive glasses and their applications in hard tissue regeneration: A review. *Pharmaceuticals*, 14(2), 1–20.
- Al-Khafaji, T. J., Wong, F., Fleming, P. S., Karpukhina, N., & Hill, R. (2019). Novel fluoride and strontium-containing bioactive glasses for dental varnishes-design



- and bioactivity in tris buffer solution. *Journal of Non-Crystalline Solids*, 503, 120–130.
- Al-Noaman, A., Rawlinson, S. C. F., & Hill, R. G. (2012a). The influence of  $\text{CaF}_2$  content on the physical properties and apatite formation of bioactive glass coatings for dental implants. *Journal of Non-Crystalline Solids*, 358(15), 1850–1858.
- Al-Noaman, A., Rawlinson, S. C. F., & Hill, R. G. (2012b). The role of  $\text{MgO}$  on thermal properties, structure and bioactivity of bioactive glass coating for dental implants. *Journal of Non-Crystalline Solids*, 358(22), 3019–3027.
- Alatawi, R. A. S., Elsayed, N. H., & Mohamed, W. S. (2019). Influence of hydroxyapatite nanoparticles on the properties of glass ionomer cement. *Journal of Materials Research and Technology*, 8(1), 344–349.
- Ali, S., Farooq, I., & Iqbal, K. (2014). A review of the effect of various ions on the properties and the clinical applications of novel bioactive glasses in medicine and dentistry. *Saudi Dental Journal*, 26(1), 1–5.
- Alsohaimi, I. H., Nassar, A. M., Seaf Elnasr, T. A., & Cheba, B. A. (2020). A novel composite silver nanoparticles loaded calcium oxide stemming from eggshell recycling: A potent photocatalytic and antibacterial activities. *Journal of Cleaner Production*, 248, 119274.
- Alzahrani, A. S. (2022). A review of glass and crystallizations of glass-ceramics. *Advances in Materials Physics and Chemistry*, 12(11), 261–288.
- Amaral, M. C., Siqueira, F. B., Destefani, A. Z., & Holanda, J. N. F. (2013). Soil-cement bricks incorporated with eggshell waste. *Proceedings of Institution of Civil Engineers: Waste and Resource Management*, 166(3), 137–141.
- Amudha, S., Ramya, J. R., Arul, K. T., Deepika, A., Sathiamurthi, P., Mohana, B., Asokan, K., Dong, C. L., & Kalkura, S. N. (2020). Enhanced mechanical and biocompatible properties of strontium ions doped mesoporous bioactive glass. *Composites Part B: Engineering*, 196, 108099.
- Angell, C. A., & Torell, L. M. (1982). Short time structural relaxation processes in liquids: Comparison of experimental and computer simulation glass transitions on picosecond time scales. *The Journal of Chemical Physics*, 78(2), 937–945.
- Angioni, D., Orrù, R., Cao, G., Garroni, S., Iacomini, A., Bellucci, D., & Cannillo, V. (2022). Spark plasma sintering, mechanical and in-vitro behavior of a novel Sr and Mg-containing bioactive glass for biomedical applications. *Journal of the European Ceramic Society*, 42(4), 1776–1783.
- Arango-Ospina, M., Hupa, L., & Boccaccini, A. R. (2019). Bioactivity and dissolution behavior of boron-containing bioactive glasses under static and dynamic conditions in different media. *Biomedical Glasses*, 5(1), 124–139.



- Araujo, M., Miola, M., Baldi, G., Perez, J., & Verné, E. (2016). Bioactive glasses with low Ca/P ratio and enhanced bioactivity. *Materials*, 9(4), 1–14.
- Araujo, M. S., Bartolomé, J. F., & Mello-Castanho, S. (2020). Tribological and mechanical behaviour of 45S5 Bioglass®-based compositions containing alumina and strontium. *Ceramics International*, 46(15), 24347–24354.
- Awogbemi, O., Inambao, F., & Onuh, E. I. (2020). Modification and characterization of chicken eggshell for possible catalytic applications. *Heliyon*, 6(10), e05283.
- Ayodeji, A. A., Ojewumi, M. E., Rasheed, B., & Ayodele, J. M. (2018). Data on CaO and eggshell catalysts used for biodiesel production. *Data in Brief*, 19, 1466–1473.
- Ayukawa, Y., Suzuki, Y., Tsuru, K., Koyano, K., & Ishikawa, K. (2015). Histological comparison in rats between carbonate apatite fabricated from gypsum and sintered hydroxyapatite on bone remodeling. *BioMed Research International*, 2015(1), 1–8.
- Azarian, M. H., & Sutapun, W. (2022). Tuning polymorphs of precipitated calcium carbonate from discarded eggshells: Effects of polyelectrolyte and salt concentration. *RSC Advances*, 12(23), 14729–14739.
- Babu, M. M., Rao, P. V., Singh, R. K., Kim, H. W., Veeraiah, N., Özcan, M., & Prasad, P. S. (2021). ZnO incorporated high phosphate bioactive glasses for guided bone regeneration implants: Enhancement of in vitro bioactivity and antibacterial activity. *Journal of Materials Research and Technology*, 15, 633–646.
- Bachar, A., Mercier, C., Tricoteaux, A., Hampshire, S., Leriche, A., & Follet, C. (2013). Effect of nitrogen and fluorine on mechanical properties and bioactivity in two series of bioactive glasses. *Journal of the Mechanical Behavior of Biomedical Materials*, 23, 133–148.
- Bajraktarova-Valjakova, E., Korunoska-Stevkovska, V., Kapusevska, B., Gigovski, N., Bajraktarova-Misevska, C., & Grozdanov, A. (2018). Contemporary dental ceramic materials, a review: Chemical composition, physical and mechanical properties, indications for use. *Open Access Macedonian Journal of Medical Sciences*, 6(9), 1742–1755.
- Bakry, A. S., Abbassy, M. A., Alharkan, H. F., Basuhail, S., Al-Ghamdi, K., & Hill, R. (2018). A novel fluoride containing bioactive glass paste is capable of remineralizing early caries lesions. *Materials*, 11(9), 1–10.
- Balamurugan, A., Balossier, G., Kannan, S., Michel, J., Rebelo, A. H. S., & Ferreira, J. M. F. (2007). Development and in vitro characterization of sol-gel derived CaO–P<sub>2</sub>O<sub>5</sub>–SiO<sub>2</sub>–ZnO bioglass. *Acta Biomaterialia*, 3(2), 255–262.
- Baliga, S., Muglikar, S., & Kale, R. (2013). Salivary pH: A diagnostic biomarker. *Journal of Indian Society of Periodontology*, 17(4), 461–465.

- Barrett, J. M., Clark, D. E., & Hench, L. L. (1980). *Glass-ceramic dental restorations* (U.S. Patent No. 4,189,325). Washington, DC: U.S. Patent and Trademark Office.
- Bellucci, D., Cannillo, V., & Sola, A. (2011). Calcium and potassium addition to facilitate the sintering of bioactive glasses. *Materials Letters*, 65(12), 1825–1827.
- Bellucci, D., Sola, A., Anesi, A., Salvatori, R., Chiarini, L., & Cannillo, V. (2015). Bioactive glass/hydroxyapatite composites: Mechanical properties and biological evaluation. *Materials Science and Engineering C*, 51, 196–205.
- Bellucci, D., Veronesi, E., Dominici, M., & Cannillo, V. (2020). A new bioactive glass with extremely high crystallization temperature and outstanding biological performance. *Materials Science and Engineering C*, 110, 110699.
- Birkholz, M. (2006). Principles of X-ray diffraction. *Thin film analysis by X-Ray scattering* (pp. 1-11). Wiley-Vch Verlag GmbH & Co. KGaA.
- Blaß, C., Müller, R., Poologasundarampillai, G., & Brauer, D. S. (2019). Sintering and concomitant crystallization of bioactive glasses. *International Journal of Applied Glass Science*, 10(4), 449-462.
- Borges, A. L., Soares, S. M., Freitas, T. O. G., Oliveira Júnior, A. D., Ferreira, E. B., & Ferreira, F. G. D. S. (2021). Evaluation of the pozzolanic activity of glass powder in three maximum grain sizes. *Materials Research*, 24, e20200496.
- Borkowski, L., Przekora, A., Belcarz, A., Palka, K., Jozefaciuk, G., Lübek, T., Jojczuk, M., Nogalski, A., & Ginalska, G. (2020). Fluorapatite ceramics for bone tissue regeneration: Synthesis, characterization and assessment of biomedical potential. *Materials Science and Engineering C*, 116, 111211.
- Brauer, Delia S., Karpukhina, N., O'Donnell, M. D., Law, R. V., & Hill, R. G. (2010). Fluoride-containing bioactive glasses: Effect of glass design and structure on degradation, pH and apatite formation in simulated body fluid. *Acta Biomaterialia*, 6(8), 3275–3282.
- Brauer, D. S., Al-Noaman, A., Hill, R. G., & Doweidar, H. (2011a). Density-structure correlations in fluoride-containing bioactive glasses. *Materials Chemistry and Physics*, 130(1–2), 121–125.
- Brauer, D. S., Mneimne, M., & Hill, R. G. (2011b). Fluoride-containing bioactive glasses: Fluoride loss during melting and ion release in tris buffer solution. *Journal of Non-Crystalline Solids*, 357(18), 3328–3333.
- Brauer, D. S., Anjum, M. N., Mneimne, M., Wilson, R. M., Doweidar, H., & Hill, R. G. (2012). Fluoride-containing bioactive glass-ceramics. *Journal of Non-Crystalline Solids*, 358(12–13), 1438–1442.
- Brett, D. J. L., Atkinson, A., Brandon, N. P., & Skinner, S. J. (2008). Intermediate temperature solid oxide fuel cells. *Chemical Society Reviews*, 37(8), 1568–1578.

- Brisi, C. (1956). Role of Cuspidine ( $3\text{CaO}-2\text{SiO}_2-\text{CaF}_2$ ) in the System  $\text{CaO}-\text{SiO}_2-\text{CaF}_2$ . *Journal of the American Ceramic Society*, 40(5), 174–178.
- Brouwer, P. (2010). *Theory of XRF*. PANalytical BV.
- Bunaciu, A. A., Udriștioiu, E. G., & Aboul-Enein, H. Y. (2015). X-Ray diffraction: Instrumentation and applications. *Critical Reviews in Analytical Chemistry*, 45(4), 289–299.
- Buxbaum, E. (2011). Bond properties. *Biophysical Chemistry of Proteins*, 3(1), 453–453.
- Callister Jr, W. D. (2007). *Materials science and engineering: An introduction*. John Wiley & Sons.
- Cerruti, M., & Sahai, N. (2006). Silicate biomaterials for orthopaedic and dental implants. *Reviews in Mineralogy and Geochemistry*, 64(1), 283–313.
- Chaikina, M. V., Bulina, N. V., Prosanov, I. Y., & Ishchenko, A. V. (2023). Formation of Fluorapatite in the Equilibrium System  $\text{CaO}-\text{P}_2\text{O}_5-\text{HF}-\text{H}_2\text{O}$  at 298 K in a Nitrogen Atmosphere. *Crystals*, 13(8), 1264.
- Chanshetti, U. B., Shelke, V. A., Jadhav, S. M., Shankarwar, S. G., Chondhekar, T. K., Shankarwar, A. G., Sudarsan, V., & Jogad, M. S. (2011). Density and molar volume studies of phosphate glasses. *Facta Universitatis – Series: Physics, Chemistry and Technology*, 9(1), 29–36.
- Charmforoushan, A., Roknabadi, M. R., Shahtahmassebi, N., Malaekheh-Nikouei, B., & Bagherabadi, M. (2022). Synthesis and controlled drug release behavior of micro-mesoporous wollastonite nanoparticles. Effect of calcination temperature on the structural and biodegradability properties. *Materials Chemistry and Physics*, 280, 125825.
- Chatterjee, A. K. (2000). X-ray diffraction. *Handbook of analytical techniques in concrete science and technology* (pp.275–332). William Andrew Publishing/Noyes Publications
- Chatzistavrou, X., Fenno, J. C., Faulk, D., Badylak, S., Kasuga, T., Boccaccini, A. R., & Papagerakis, P. (2014). Fabrication and characterization of bioactive and antibacterial composites for dental applications. *Acta Biomaterialia*, 10(8), 3723–3732.
- Chen, X., Chen, X., Brauer, D. S., Wilson, R. M., Hill, R. G., & Karpukhina, N. (2014). Novel alkali free bioactive fluorapatite glass ceramics. *Journal of Non-Crystalline Solids*, 402, 172–177.
- Chen, X., Chen, X., Pedone, A., Apperley, D., Hill, R. G., & Karpukhina, N. (2018). New insight into mixing fluoride and chloride in bioactive silicate glasses. *Scientific Reports*, 8(1), 1316.

- Chen, H., Sun, Q., Zhang, J., & Sheng, J. (2022a). Effect of  $\text{MgF}_2$  addition on sinterability and mechanical properties of fluorapatite ceramic composites fabricated by wollastonite and phosphate glass. *Ceramics International*, 48(14), 20400–20408.
- Chen, X., Wang, M., Kenny, C., Chen, X., Karpukhina, N., & Hill, R. G. (2022b). Novel fluoride- and chloride-containing bioactive glasses for use in air abrasion. *Journal of Dentistry*, 125, 104252.
- Chhetri, K. B. (2012). Computation of errors and their analysis on physics experiments. *The Himalayan Physics*, 3, 78–86.
- Chitra, S., Bargavi, P., Durgalakshmi, D., Rajashree, P., & Balakumar, S. (2019). Role of sintering temperature dependent crystallization of bioactive glasses on erythrocyte and cytocompatibility. *Processing and Application of Ceramics*, 13(1), 12–23.
- Chong, B. W., Rokiah, O., Ramadhansyah, P. J., Doh, S. I., & Li, X. (2020). Properties of concrete with eggshell powder: A review. *Physics and Chemistry of the Earth*, 120, 102951.
- Cocchi, M., & Durante, C. (2012). Evaluation of the behaviour of fluorine-containing bioactive glasses: Reactivity in a simulated body fluid solution assisted by multivariate data analysis. *Journal of Materials Science: Materials in Medicine*, 23, 639–648.
- Crovace, M. C., Souza, M. T., Chinaglia, C. R., Peitl, O., & Zanotto, E. D. (2016). Biosilicate® – A multipurpose, highly bioactive glass-ceramic. in vitro, in vivo and clinical trials. *Journal of Non-Crystalline Solids*, 432, 90–110.
- Dahiya, M. S., Tomer, V. K., & Duhan, S. (2018). Bioactive glass/glass ceramics for dental applications. *Applications of Nanocomposite Materials in Dentistry*, 1–25.
- Damen, J. J. M., & Ten Cate, J. M. (1992). Silica-induced Precipitation of calcium phosphate in the presence of inhibitors of hydroxyapatite formation. *Journal of Dental Research*, 71(3), 453–457.
- De Caluwe, T., Vercruysse, C. W. J., Declercq, H. A., Schaubroeck, D., Verbeeck, R. M. H., & Martens, L. C. (2016). Bioactivity and biocompatibility of two fluoride containing bioactive glasses for dental applications. *Dental Materials*, 32(11), 1414–1428.
- De Caluwé, T., Vercruysse, C. W. J., Ladik, I., Convents, R., Declercq, H., Martens, L. C., & Verbeeck, R. M. H. (2017). Addition of bioactive glass to glass ionomer cements: Effect on the physico-chemical properties and biocompatibility. *Dental Materials*, 33(4), e186–e203.
- de Siqueira, L., Campos, T. M. B., Camargo, S. E. A., Thim, G. P., & Trichês, E. S. (2021). Structural, crystallization and cytocompatibility evaluation of the 45S5 bioglass-derived glass-ceramic containing niobium. *Journal of Non-Crystalline*



*Solids*, 555, 120629.

- Deilmann, L., Winter, O., Cerrutti, B., Bradtmüller, H., Herzig, C., Limbeck, A., Lahayne, O., Hellmich, C., Eckert, H., & Eder, D. (2020). Effect of boron incorporation on the bioactivity, structure, and mechanical properties of ordered mesoporous bioactive glasses. *Journal of Materials Chemistry B*, 8(7), 1456–1465.
- Deliormanli, A. M., Ensoylu, M., Issa, S. A. M., Elshami, W., Al-Baradi, A. M., Al-Buriah, M. S., & Tekin, H. O. (2021). WS<sub>2</sub>/bioactive glass composites: Fabrication, structural, mechanical and radiation attenuation properties. *Ceramics International*, 47(21), 29739–29747.
- Deubener, J., Allix, M., Davis, M. J., Duran, A., Höche, T., Honma, T., Komatsu, T., Krüger, S., Mitra, I., Müller, R., Nakane, S., Pascual, M. J., Schmelzer, J. W. P., Zanutto, E. D., & Zhou, S. (2018). Updated definition of glass-ceramics. *Journal of Non-Crystalline Solids*, 501, 3–10.
- Dietschi, D., Marret, N., & Krejci, I. (2003). Comparative efficiency of plasma and halogen light sources on composite micro-hardness in different curing conditions. *Dental Materials*, 19(6), 493–500.
- Dimitriadis, K., Vasilopoulos, K. C., Vaimakis, T. C., Karakassides, M. A., Tulyaganov, D. U., & Agathopoulos, S. (2020). Synthesis of glass-ceramics in the Na<sub>2</sub>O/K<sub>2</sub>O–CaO–MgO–SiO<sub>2</sub>–P<sub>2</sub>O<sub>5</sub>–CaF<sub>2</sub> system as candidate materials for dental applications. *International Journal of Applied Ceramic Technology*, 17(4), 2025–2035.
- Dimitriadis, K., Tulyaganov, D. U., & Agathopoulos, S. (2021). Development of novel alumina-containing bioactive glass-ceramics in the CaO–MgO–SiO<sub>2</sub> system as candidates for dental implant applications. *Journal of the European Ceramic Society*, 41(1), 929–940.
- Dimitriadis, K., Sfikas, A. K., Kamnis, S., Tsolka, P., & Agathopoulos, S. (2022). Influence of heat treatment on the microstructure and the physical and mechanical properties of dental highly translucent zirconia. *Journal of Advanced Prosthodontics*, 14(2), 96–107.
- Doble, M., & Kruthiventi, A. K. (2007). Conclusions and future trends. *Green Chemistry and Engineering* (pp. 297–312). Academic Press.
- Dorozhkin, S. V. (2016). *Calcium orthophosphate-based bioceramics and biocomposites*. John Wiley & Sons.
- Ebrahimi, M., Manafi, S., & Sharifianjazi, F. (2023). The effect of Ag<sub>2</sub>O and MgO dopants on the bioactivity, biocompatibility, and antibacterial properties of 58S bioactive glass synthesized by the sol-gel method. *Journal of Non-Crystalline Solids*, 606, 122189.

- Effendy, N., Ab Aziz, S. H., Kamari, H. M., Zaid, M. H. M., & Wahab, S. A. A. (2020). Ultrasonic and artificial intelligence approach: Elastic behavior on the influences of ZnO in tellurite glass systems. *Journal of Alloys and Compounds*, 835, 1–11.
- Eilaghi, M., Montazerian, M., & Yekta, B. E. (2016). Effect of partial substitution of  $K_2O$  for  $Na_2O$  on sintering, crystallization and mechanical properties of  $SiO_2$ – $CaO$ – $K_2O$ – $Na_2O$ – $CaF_2$  glass-ceramics. *Transactions of the Indian Ceramic Society*, 75(1), 1–6.
- El-Meliegy, E., & Noort, R. V. (2012). Models of bioactive glass ceramics. *Glasses and Glass Ceramics for Medical Applications* (pp. 229–238). Springer.
- Elbatal, F. H., Ouis, M. A., & Elbatal, H. A. (2016). Comparative studies on the bioactivity of some borate glasses and glass-ceramics from the two systems:  $Na_2O$ – $CaO$ – $B_2O_3$  and  $NaF$ – $CaF_2$ – $B_2O_3$ . *Ceramics International*, 42(7), 8247–8256.
- Eletta, O. A. A., Ajayi, O. A., Ogunleye, O. O., & Akpan, I. C. (2016). Adsorption of cyanide from aqueous solution using calcined eggshells: Equilibrium and optimisation studies. *Journal of Environmental Chemical Engineering*, 4(1), 1367–1375.
- Eliaz, N., & Metoki, N. (2017). Calcium phosphate bioceramics: A review of their history, structure, properties, coating technologies and biomedical applications. *Materials*, 10(4).
- Eriksson, A. (2019). Bioactivity testing of dental materials. *Uppsala Universitet*, 1–35.
- Esmati, N., Khodaei, T., Salahinejad, E., & Sharifi, E. (2018). Fluoride doping into  $SiO_2$ – $MgO$ – $CaO$  bioactive glass nanoparticles: Bioactivity, biodegradation and biocompatibility assessments. *Ceramics International*, 44(14), 17506–17513.
- Fan, W. Di, Liu, B., Luo, X., Yang, J., Guo, B., & Zhang, S. G. (2019). Production of glass–ceramics using municipal solid waste incineration fly ash. *Rare Metals*, 38(3), 245–251.
- Fathi, H., Johnson, A., Noort, R. V., & Ward, J. M. (2005). The influence of calcium fluoride ( $CaF_2$ ) on biaxial flexural strength of apatite-mullite glass-ceramic materials. *Dental Materials*, 21(9), 846–851.
- Fathi, M. H., & Mohammadi, A. D. (2008). Preparation and characterization of sol-gel bioactive glass coating for improvement of biocompatibility of human body implant. *Materials Science and Engineering A*, 474(1–2), 128–133.
- Faupel, F., Frank, W., Macht, M. P., Mehrer, H., Naundorf, V., Rätzke, K., Schober, H. R., Sharma, S. K., & Teichler, H. (2003). Diffusion in metallic glasses and supercooled melts. *Reviews of Modern Physics*, 75(1), 237–280.

- Fernandes, H. R., Gaddam, A., Tulyaganov, D. U., & Ferreira, J. M. F. (2014). Structure, properties and crystallization of non-stoichiometric lithium disilicate glasses containing  $\text{CaF}_2$ . *Journal of Non-Crystalline Solids*, 406, 54–61.
- Fernandes, H. R., Gaddam, A., Rebelo, A., Brazete, D., Stan, G. E., & Ferreira, J. M. F. (2018). Bioactive glasses and glass-ceramics for healthcare applications in bone regeneration and tissue engineering. *Materials*, 11(12), 1–54.
- Fernandes, F. A. da S., Arcaro, S., Junior, E. F. T., Serra, J. C. V., & Bergmann, C. P. (2019). Glass foams produced from soda-lime glass waste and rice husk ash applied as partial substitutes for concrete aggregates. *Process Safety and Environmental Protection*, 128, 77–84.
- Ferraris, S., Yamaguchi, S., Barbani, N., Cazzola, M., Cristallini, C., Miola, M., Vernè, E., & Spriano, S. (2020). Bioactive materials: In vitro investigation of different mechanisms of hydroxyapatite precipitation. *Acta Biomaterialia*, 102, 468–480.
- Filho, O. P., Latorre, G. P., & Hench, L. L. (1996). Effect of crystallization on apatite-layer formation of bioactive glass 45S5. *Journal of Biomedical Materials Research*, 30(4), 509–514.
- Filip, D. G., Surdu, V., Paduraru, A. V., & Andronesu, E. (2022). Current development in biomaterials – Hydroxyapatite and bioglass for applications in biomedical Field : A Review. *Journal of Functional Biomaterials*, 13(248), 1021.
- Fiume, E., Cao, S. P. O., Na, M., Migneco, C., & Vern, E. (2020). Comparison between bioactive sol-gel and melt-derived glasses/glass-ceramics based on the multicomponent  $\text{SiO}_2\text{--P}_2\text{O}_5\text{--CaO--MgO--Na}_2\text{O--K}_2\text{O}$ . *Materials*, 13(540), 1–13.
- Fiume, E., Magnaterra, G., Rahdar, A., Verné, E., & Baino, F. (2021). Hydroxyapatite for biomedical applications: A short overview. *Ceramics*, 4(4), 542–563.
- Food and Agriculture Organization of the United Nations. (2022). Crops and livestock products. Faostat.
- Florian, P., Fayon, F., & Massiot, D. (2009).  $^2\text{J}$  Si–O–Si scalar spin-spin coupling in the solid state: crystalline and glassy wollastonite  $\text{CaSiO}_3$ . *Journal of Physical Chemistry C*, 113(6), 2562–2572.
- Forensics, G. (2017). Scanning electron microscopy–energy-dispersive X-ray (SEM/EDX): a rapid diagnostic tool to aid the identification of burnt bone and contested remains. *Journal of Forensic Sciences* 63, 63(2), 504–510.
- Freeman, C. O., Brook, I. M., Johnson, A., Hatton, P. V., Hill, R. G., & Stanton, K. T. (2003). Crystallization modifies osteoconductivity in an apatite-mullite glass-ceramic. *Journal of Materials Science: Materials in Medicine*, 14(11), 985–990.
- Freund, F., & Knobel, M. (1977). Distribution of fluorine in hydroxyapatite studied by infrared spectroscopy. *Journal of the Chemical Society*, 11, 1136–1140.

- Fu, L., Engqvist, H., & Xia, W. (2020). Glass-ceramics in dentistry: A review. *Materials*, 13(5), 1–23.
- Fuji, E., Kawabata, K., Yoshimatsu, H., Hayakawa, S., Tsuru, K., & Osaka, A. (2003). Structure and biomineralization of calcium silicate glasses containing fluoride ions. *Journal of the Ceramic Society of Japan*, 111(1298), 762–766.
- Ganzoury, M. A., Allam, N. K., Nicolet, T., & All, C. (2015). Introduction to Fourier transform infrared spectrometry. *Renewable and Sustainable Energy Reviews*, 50, 1–8.
- Garcia-Alvarez, G., Escobedo-Bocardo, J. C., Cortés-Hernández, D. A., Almanza-Robles, J. M., & Sánchez-Escobedo, B. A. (2018). Effect of wollastonite and a bioactive glass-ceramic on the in vitro bioactivity and compressive strength of a calcium aluminate cement. *Ceramics International*, 44(16), 19077–19083.
- Gautam, C., Joyner, J., Gautam, A., Rao, J., & Vajtai, R. (2016). Zirconia based dental ceramics: Structure, mechanical properties, biocompatibility and applications. *Dalton Transactions*, 45(48), 19194–19215.
- Ghosh, S., Dandapat, N., & Balla, V. K. (2015). Preparation and in vitro characterization of fluoroapatite based bioactive glass-ceramics for biomedical applications. *Materials Today: Proceedings*, 2(4–5), 1326–1331.
- Goswami, P., & O’Haire, T. (2016). Developments in the use of green (biodegradable), recycled and biopolymer materials in technical nonwovens. *Advances in Technical Nonwovens*, 97–114.
- Goudouri, O. M., Kontonasaki, E., Papadopoulou, L., Kantiranis, N., Lazaridis, N. K., Chrissafis, K., Chatzistavrou, X., Koidis, P., & Paraskevopoulos, K. M. (2014). Towards the synthesis of an experimental bioactive dental ceramic. Part I: Crystallinity characterization and bioactive behavior evaluation. *Materials Chemistry and Physics*, 145(1–2), 125–134.
- Grodzińska-Jurczak, M. S. (2001). Management of industrial and municipal solid wastes in Poland. *Resources, Conservation and Recycling*, 32(2), 85–103.
- Groh, D., Döhler, F., & Brauer, D. S. (2014). Bioactive glasses with improved processing. Part 1. Thermal properties, ion release and apatite formation. *Acta Biomaterialia*, 10(10), 4465–4473.
- Gumus, H. S., Polat, N. T., & Yildirim, G. (2018). Evaluation of fracture resistance of inlay-retained fixed partial dentures fabricated with different monolithic zirconia materials. *Journal of Prosthetic Dentistry*, 119(6), 959–964.
- Gunawidjaja, P. N., Mathew, R., Lo, A. Y. H., Izquierdo-Barba, I., García, A., Arcos, D., Vallet-Regí, M., & Edén, M. (2012). Local structures of mesoporous bioactive glasses and their surface alterations in vitro: Inferences from solid-state nuclear magnetic resonance. *Philosophical Transactions of the Royal Society A: Mathematical, Physical and Engineering Sciences*, 370(1963), 1376–1399.



- Han, R., Buchanan, F., Ford, L., Julius, M., & Walsh, P. J. (2021). A comparison of the degradation behaviour of 3D printed PDLGA scaffolds incorporating bioglass or biosilica. *Materials Science and Engineering C*, 120, 111755.
- Hanao, M., Kawamoto, M., & Watanabe, T. (2004). Influence of Na<sub>2</sub>O on phase relation between mold flux composition and cuspidine. *ISIJ International*, 44(5), 827–835.
- Hanifi, A. R., Genson, A., Redington, W., Pomeroy, M. J., & Hampshire, S. (2012). Effects of nitrogen and fluorine on crystallisation of Ca–Si–Al–O–N–F glasses. *Journal of the European Ceramic Society*, 32(4), 849–857.
- Harada, K., Shinya, A., Gomi, H., Hatano, Y., Shinya, A., & Raigrodski, A. J. (2016). Effect of accelerated aging on the fracture toughness of zirconias. *Journal of Prosthetic Dentistry*, 115(2), 215–223.
- Harripersadth, C., Musonge, P., Makarfi Isa, Y., Morales, M. G., & Sayago, A. (2020). The application of eggshells and sugarcane bagasse as potential biomaterials in the removal of heavy metals from aqueous solutions. *South African Journal of Chemical Engineering*, 34, 142–150.
- Hellström, K., Diószegi, A., & Diaconu, L. (2017). A broad literature review of density measurements of liquid cast iron. *Metals*, 7(5), 165.
- Hench, L. L., Spilman, D. B., & Hench, J. W. (1988). *Fluoride-containing bioglass compositions* (U.S. Patent number 4,775,646). Washington, DC: U.S. Patent and Trademark Office.
- Hench, L. L. (1991). Bioceramics: From concept to clinic. *Journal of the American Ceramic Society*, 74(7), 1487–1510.
- Hench, L. L. (1999). Bioactive glasses and glass-ceramics. *Materials Science Forum*, 293, 37–64.
- Hench, L. L. (2006). The story of Bioglass®. *Journal of Materials Science: Materials in Medicine*, 17(11), 967–978.
- Hench, L. L. (2009). Genetic design of bioactive glass. *Journal of the European Ceramic Society*, 29(7), 1257–1265.
- Hench, L. L., & Thompson, I. (2010). Twenty-first century challenges for biomaterials. *Journal of the Royal Society Interface*, 7(4), 379–391.
- Higazy, M. G. (2019). Analysis and accuracy of experimental methods. *International Journal of Petrochemistry and Research*, 3(1), 262–268.
- Hisham, N. A. N., Zaid, M. H. M., Matori, K. A., & Shabdin, M. K. (2022). Effect of ark clam shell on crystal growth and mechanical evaluation of foam glass-ceramic derived from cullet glass waste. *Materials Science & Engineering B*, 281, 115730.

- Höland, W., Rheinberger, V., Apel, E., Hoen, C. V., Höland, M., Dommann, A., Obrecht, M., Mauth, C., & Graf-Hausner, U. (2006). Clinical applications of glass-ceramics in dentistry. *Journal of Materials Science: Materials in Medicine*, 17(11), 1037–1042.
- Hong, Z., Reis, R. L., & Mano, J. F. (2009). Preparation and in vitro characterization of novel bioactive glass ceramic nanoparticles. *Journal of Biomedical Materials Research - Part A*, 88(2), 304–313.
- Hoornweg, D., & Bhada-Tata, P. (2012). What a waste: A global review of solid waste management. *Urban Development Series* (Vol. 15). World Bank.
- Hoppe, A., Güldal, N. S., & Boccaccini, A. R. (2011). A review of the biological response to ionic dissolution products from bioactive glasses and glass-ceramics. *Biomaterials*, 32(11), 2757–2774.
- Hossain, S. S., Yadav, S., Majumdar, S., Krishnamurthy, S., Pyare, R., & Roy, P. K. (2020). A comparative study of physico-mechanical, bioactivity and hemolysis properties of pseudo-wollastonite and wollastonite glass-ceramic synthesized from solid wastes. *Ceramics International*, 46(1), 833–843.
- Hung, G. Y., Chen, P. Y., Wang, C. Y., Tu, C. S., Chen, C. S., Lai, P. L., & Feng, K. C. (2022). Tailoring bioactive and mechanical properties in polycrystalline CaO–SiO<sub>2</sub>–P<sub>2</sub>O<sub>5</sub> glass-ceramics. *Ceramics International*, 49(5), 7289–7298.
- Hupa, L. (2011). Melt-derived bioactive glasses. *Bioactive Glasses: Materials, Properties and Applications* (pp. 3–28). Woodhead Publishing Limited.
- International Standard of Organization. (2015). *Dentistry – Ceramic materials*.
- Ismail, N. Q. A., Sa'at, N. K., Zaid, M. H. M., Zainuddin, N., & Mayzan, M. Z. H. (2024). Effect of Na<sub>2</sub>CO<sub>3</sub>/Al<sub>2</sub>O<sub>3</sub> ratio on the calcium fluoroaluminosilicate-based bioactive glass-ceramics derived from waste materials. *Materials Chemistry and Physics*, 312, 128556.
- Jaafar, S. H., Zaid, M. H. M., Matori, K. A., Aziz, S. H. A., Kamari, H. M., Honda, S., & Iwamoto, Y. (2021). Influence of calcination temperature on crystal growth and optical characteristics of Eu<sup>3+</sup> doped ZnO/Zn<sub>2</sub>SiO<sub>4</sub> composites fabricated via simple thermal treatment method. *Crystals*, 11(2), 115–131.
- Jafari, N., Habashi, M. S., Hashemi, A., Shirazi, R., Tanideh, N., & Tamadon, A. (2022). Application of bioactive glasses in various dental fields. *Biomaterials Research*, 26(1), 31.
- Jakfar, N. H., Khor, S. F., Johar, B., Adzali, N. M. S., Yunus, S. N. H. M., & Cheng, E. M. (2023). Synthesis of sustainable binary calcium monosilicate ceramics from bio-waste: Effect of sintering temperature on microstructure and electrical properties. *International Journal of Nanoelectronics and Materials*, 16(1), 63–79.

- Jalil, R. A., Matori, K. A., Zaid, M. H. M., Zainuddin, N., Khiri, M. Z. A., Rahman, N. A. A., Jusoh, W. N. W., & Kul, E. (2020). A study of fluoride-containing bioglass system for dental materials derived from clam shell and soda lime silica glass. *Journal of Spectroscopy*, 2020, 1–9.
- Jereme, I. A., MahmudulAla, M., & Siwar, C. (2015). Waste recycling in malaysia: Transition from developing to developed country. *Journal of Education and Information Management*, 4(1), 1–14.
- Jiang, Z. H., & Zhang, Q. Y. (2014). The structure of glass: A phase equilibrium diagram approach. *Progress in Materials Science*, 61, 144–215.
- Jusoh, W. N. W., Matori, K. A., Zaid, M. H. M., Zainuddin, N., Khiri, M. Z. A., Rahman, N. A. A., Jalil, R. A., & Kul, E. (2019). Effect of sintering temperature on physical and structural properties of alumino-silicate-fluoride glass ceramics fabricated from clam shell and soda lime silicate glass. *Results in Physics*, 12, 1909–1914.
- Kah, P., Väst, H., Layus, P., Kah, P., Martikainen, J., & Layus, P. (2011). Methods of evaluating weld quality in modern production (Part 2). *Proceedings of 16th International Conference*, 7–8.
- Kamalanathan, P., Ramesh, S., Bang, L. T., Niakan, A., Tan, C. Y., Purbolaksono, J., Chandran, H., & Teng, W. D. (2014). Synthesis and sintering of hydroxyapatite derived from eggshells as a calcium precursor. *Ceramics International*, 40, 16349–16359.
- Kannan, M. (2018). Scanning electron microscopy: Principle, components and applications. *A Textbook on Fundamentals And Applications of Nanotechnology* (pp. 81–90). Astral.
- Kansal, I., Tulyaganov, D. U., Goel, A., Pascual, M. J., & Ferreira, J. M. F. (2010). Structural analysis and thermal behavior of diopside-fluorapatite-wollastonite-based glasses and glass-ceramics. *Acta Biomaterialia*, 6(11), 4380–4388.
- Kanwal, N., Brauer, D. S., Earl, J., Wilson, R. M., Karpukhina, N., & Hill, R. G. (2018). In-vitro apatite formation capacity of a bioactive glass-containing toothpaste. *Journal of Dentistry*, 68, 51–58.
- Karakuzu-Ikizler, B., Terzioğlu, P., Basaran-Elalmis, Y., Tekerek, B. S., & Yücel, S. (2020a). Role of magnesium and aluminum substitution on the structural properties and bioactivity of bioglasses synthesized from biogenic silica. *Bioactive Materials*, 5(1), 66–73.
- Karakuzu-Ikizler, B., Terzioğlu, P., Oduncu-Tekerek, B. S., & Yücel, S. (2020b). Effect of selenium incorporation on the structure and in vitro bioactivity of 45S5 bioglass. *Journal of the Australian Ceramic Society*, 56(2), 697–709.
- Karamanov, A., Hamzawy, E. M. A., Karamanova, E., Jordanov, N. B., & Darwish, H. (2020). Sintered glass-ceramics and foams by metallurgical slag with addition

- of  $\text{CaF}_2$ . *Ceramics International*, 46(5), 6507–6516.
- Karasu, B., Yanar, A. O., Kocak, A., & Kisacik, Ö. (2019). Bioactive glasses. *Journal of Science and Engineering*, 4(3), 436–471.
- Karimi, A. Z., Rezabeigi, E., & Drew, R. A. L. (2019). Glass ionomer cements with enhanced mechanical and remineralizing properties containing 45S5 bioglass-ceramic particles. *Journal of the Mechanical Behavior of Biomedical Materials*, 97, 396–405.
- Karmakar, B. (2016). Fundamentals of glass and glass nanocomposites. *Glass nanocomposites: Synthesis, properties and applications* (pp. 3–53). Elsevier Inc.
- Karmakar, B. (2017). Glasses and glass-ceramics for biomedical applications. *Functional glasses and glass-ceramics* (pp. 253–280). Elsevier Inc.
- Kaur, G., Pandey, O. P., Singh, K., Homa, D., Scott, B., & Pickrell, G. (2014). A review of bioactive glasses: Their structure, properties, fabrication and apatite formation. *Journal of Biomedical Materials Research – Part A*, 102(1), 254–274.
- Kaur, G., Pickrell, G., Sriranganathan, N., Kumar, V., & Homa, D. (2016). Review and the state of the art: Sol-gel and melt quenched bioactive glasses for tissue engineering. *Journal of Biomedical Materials Research – Part B Applied Biomaterials*, 104(6), 1248–1275.
- Kaur, G., Kumar, V., Baino, F., Mauro, J. C., Pickrell, G., Evans, I., & Bretcanu, O. (2019). Mechanical properties of bioactive glasses, ceramics, glass-ceramics and composites: State-of-the-art review and future challenges. *Materials Science and Engineering C*, 104(6), 1–14.
- Kaur, D., Reddy, M. S., & Pandey, O. P. (2020). Synthesis, characterization, drug loading and in-vitro bioactivity studies of rice husk derived  $\text{SiO}_2\text{--P}_2\text{O}_5\text{--MgO--CaO--SrO}$  bio-active glasses. *Journal of Drug Delivery Science and Technology*, 61, 102154.
- Kaza, S., Yao, L., Bhada-Tata, P., & Woerden, F. Van. (2018). *What a waste 2.0: A global snapshot of solid waste management to 2050* (Issue 1). The World Bank.
- Khalil, E. M. A., Youness, R. A., Amer, M. S., & Taha, M. A. (2018). Mechanical properties, in vitro and in vivo bioactivity assessment of  $\text{Na}_2\text{O--CaO--P}_2\text{O}_5\text{--B}_2\text{O}_3\text{--SiO}_2$  glass-ceramics. *Ceramics International*, 44(7), 7867–7876.
- Khiri, M. Z. A., Matori, K. A., Zaid, M. H. M., Abdullah, A. C., Zainuddin, N., Jusoh, W. N. W., Jalil, R. A., Rahman, N. A. A., Kul, E., Wahab, S. A. A., & Effendy, N. (2020). Soda lime silicate glass and clam Shell act as precursor in synthesize calcium fluoroaluminosilicate glass to fabricate glass ionomer cement with different ageing time. *Journal of Materials Research and Technology*, 9(3), 6125–6134.



- Kim, H. W., Noh, Y. J., Koh, Y. H., Kim, H. E., & Kim, H. M. (2002). Effect of  $\text{CaF}_2$  on densification and properties of hydroxyapatite-zirconia composites for biomedical applications. *Biomaterials*, 23(20), 4113–4121.
- Kim, S. H., Kim, Y. S., Bang, H. G., & Park, S. Y. (2007). Synthesis of bio-glass ceramics in  $\text{Na}_2\text{O}$ – $\text{CaO}$ – $\text{SiO}_2$ – $\text{P}_2\text{O}_5$  system with fluoride additives. *Solid State Phenomena*, 124–126(7), 759–762.
- Kokubo, T., Shigematsu, Masazumi Nagashima, Y., Megumi, T., Nakamura, T., Yamamuro, T., & Higashi, S. (1982). Apatite- and wollastonite-containing glass-ceramics for prosthetic application. *Bulletin of the Institute for Chemical Research, Kyoto University*, 60(3–4), 260–268.
- Kothiyal, G. P., Ananthanarayanan, A., & Dey, G. K. (2012). Glass and glass-ceramics. *Functional materials* (pp. 323–386). Elsevier Inc.
- Krishnan, V., & Lakshmi, T. (2013). Bioglass: A novel biocompatible innovation. *Journal of Advanced Pharmaceutical Technology and Research*, 4(2), 78–83.
- Kumari, C. V., Gandhi, Y., Sobhanachalam, P., Reddy, A. S. S., Venkatramaiah, N., Rao, P. V., & Kumar, V. R. (2019). Bioactive behaviour of NiO substituted  $\text{CaF}_2$ – $\text{CaO}$ – $\text{B}_2\text{O}_3$ – $\text{BaO}$ – $\text{P}_2\text{O}_5$  glasses by means of spectroscopic studies. *Optical Materials*, 97, 109394.
- Lee, Y. K., Kim, A. Y., Min, S. G., & Kwak, H. S. (2016). Characteristics of milk tablets supplemented with nanopowdered eggshell or oyster shell. *International Journal of Dairy Technology*, 69(3), 337–345.
- Lewis, M. H. (1989). *Glasses and glass-ceramics*. Chapman Hall Ltd.
- Li, X., Cai, S., Zhang, W., Xu, G., & Zhou, W. (2009). Effect of pH values on surface modification and solubility of phosphate bioglass-ceramics in the  $\text{CaO}$ – $\text{P}_2\text{O}_5$ – $\text{Na}_2\text{O}$ – $\text{SrO}$ – $\text{ZnO}$  system. *Applied Surface Science*, 255(22), 9241–9243.
- Li, H. C., Wang, D. G., Hu, J. H., & Chen, C. Z. (2013). Crystallization, mechanical properties and in vitro bioactivity of sol-gel derived  $\text{Na}_2\text{O}$ – $\text{CaO}$ – $\text{SiO}_2$ – $\text{P}_2\text{O}_5$  glass-ceramics by partial substitution of  $\text{CaF}_2$  for  $\text{CaO}$ . *Journal of Sol-Gel Science and Technology*, 67(1), 56–65.
- Li, H. C., Wang, D. G., Hu, J. H., & Chen, C. Z. (2014). Influence of fluoride additions on biological and mechanical properties of  $\text{Na}_2\text{O}$ – $\text{CaO}$ – $\text{SiO}_2$ – $\text{P}_2\text{O}_5$  glass-ceramics. *Materials Science and Engineering C*, 35(1), 171–178.
- Li, Z., Huang, B., Mai, S., Wu, X., Zhang, H., Qiao, W., Luo, X., & Chen, Z. (2015). Effects of fluoridation of porcine hydroxyapatite on osteoblastic activity of human MG63 cells. *Science and Technology of Advanced Materials*, 16(3), 35006.
- Li, F., Jiang, Y., Chen, M., Yu, B., Wang, J., & Wang, F. (2022). Effect of  $\text{ZrO}_2$  addition on in-vitro bioactivity and mechanical properties of  $\text{SiO}_2$ – $\text{Na}_2\text{O}$ – $\text{CaO}$ –

- P<sub>2</sub>O<sub>5</sub> bioactive glass-ceramic. *Ceramics International*, 48(13), 18541–18550.
- Lin, L., Zhang, L., Wang, J., Xie, K., Yang, X., Chen, X., Yang, G., Gao, C., & Gou, Z. (2014). Low-temperature sintering of 45S5 Bioglass®-based glass ceramics: Effect of biphasic mixing approach on the mechanical and biological properties. *Materials Letters*, 126, 154–158.
- Liu, H., Luo, G., Wei, H., & Yu, H. (2018). Strength, permeability, and freeze-thaw durability of pervious concrete with different aggregate sizes, porosities, and water-binder ratios. *Applied Sciences (Switzerland)*, 8(8).
- Llah, N. A., Jamaludin, S. B., Daud, Z. C., Zaludin, M. A. F., Jamal, Z. A. Z., Idris, M. S., & Osman, R. A. M. (2016). Corrosion behavior of Mg–3Zn/bioglass (45S5) composite in simulated body fluid (SBF) and phosphate buffered saline (PBS) solution. *AIP Conference Proceedings*, 1756, 030001.
- Lopes, G., Rodrigues, I. F., Sth, S., Pereira, S., Martins, G., Fontoura, G., Reis, A. S., Pedrochi, F., & Steimacher, A. (2022). Bioactive antibacterial borate glass and glass-ceramics. *Journal of Non-Crystalline Solids*, 595, 121829.
- Lu, X., Kolzow, J., Chen, R. R., & Du, J. (2019). Effect of solution condition on hydroxyapatite formation in evaluating bioactivity of B<sub>2</sub>O<sub>3</sub> containing 45S5 bioactive glasses. *Bioactive Materials*, 4, 207–214.
- Lu, X., & Du, J. (2022). Effects of boron oxide on the structure, properties and bioactivities of bioactive glasses: A review. *Journal of Non-Crystalline Solids: X*, 16, 100118.
- Lubis, M., Ginting, M. H. S., Dalimunthe, N. F., Hasibuan, D. M. T., & Sastrodihardjo, S. (2017). The influence of chicken egg shell as fillers on biocomposite acrylic resin for denture based. *IOP Conference Series: Materials Science and Engineering*, 180, 012006.
- Luborsky, F. E. (1983). *Amorphous metallic alloys*. Butterworth & Co Ltd.
- Lucacel, R. C., Ponta, O., Licarete, E., Radu, T., & Simon, V. (2016). Synthesis, structure, bioactivity and biocompatibility of melt-derived P<sub>2</sub>O<sub>5</sub>–CaO–B<sub>2</sub>O<sub>3</sub>–K<sub>2</sub>O–MoO<sub>3</sub> glasses. *Journal of Non-Crystalline Solids*, 439, 67–73.
- Lukito, D., Xue, J. M., & Wang, J. (2005). In vitro bioactivity assessment of 70(wt.%)SiO<sub>2</sub>–30(wt.%)CaO bioactive glasses in simulated body fluid. *Materials Letters*, 59(26), 3267–3271.
- Lunge, S., Thakre, D., Kamble, S., Labhsetwar, N., & Rayalu, S. (2012). Alumina supported carbon composite material with exceptionally high defluoridation property from eggshell waste. *Journal of Hazardous Materials*, 237–238, 161–169.
- Luo, Z., He, F., Zhang, W., Xiao, Y., Xie, J., Sun, R., & Xie, M. (2020). Effects of fluoride content on structure and properties of steel slag glass-ceramics. *Materials*

- Lusvardi, G., Malavasi, G., Menabue, L., Aina, V., & Morterra, C. (2009). Fluoride-containing bioactive glasses: Surface reactivity in simulated body fluids solutions. *Acta Biomaterialia*, 5(9), 3548–3562.
- Lutpi, H. A., Mohamad, H., Abdullah, T. K., & Ismail, H. (2021). Effect of sintering treatment time on the sintering behaviour and thermal shock resistance of  $\text{Li}_2\text{O}-\text{Al}_2\text{O}_3-\text{SiO}_2$  glass-ceramics. *Journal of Asian Ceramic Societies*, 9(2), 507–518.
- Lynch, E., Brauer, D. S., Karpukhina, N., Gillam, D. G., & Hill, R. G. (2012). Multi-component bioactive glasses of varying fluoride content for treating dentin hypersensitivity. *Dental Materials*, 28(2), 168–178.
- Ma, J., Chen, C. Z., Wang, D. G., Meng, X. G., & Shi, J. Z. (2010). Influence of the sintering temperature on the structural feature and bioactivity of sol-gel derived  $\text{SiO}_2-\text{CaO}-\text{P}_2\text{O}_5$  bioglass. *Ceramics International*, 36(6), 1911–1916.
- Mahdy, M. A., El Zawawi, I. K., Kenawy, S. H., Hamzawy, E. M. A., & El-Bassyouni, G. T. (2022). Effect of zinc oxide on wollastonite: Structural, optical, and mechanical properties. *Ceramics International*, 48(5), 7218–7231.
- Majhi, M. R., Pyare, R., & Singh, S. P. (2011). Preparation and characterization of  $\text{CaF}_2$  doped bioglass ceramics. *Journal of Biomimetics, Biomaterials, and Tissue Engineering*, 1489(11), 45–66.
- Maleki, S., Barzegar-Jalali, M., Zarrintan, M. H., Adibkia, K., & Lotfipour, F. (2015). Calcium carbonate nanoparticles; potential applications in bone and tooth disorders. *Pharmaceutical Sciences*, 20(4), 175–182.
- Manafi, S., Mirjalili, F., & Reshadi, R. (2019). Synthesis and evaluation of the bioactivity of fluorapatite–45S5 bioactive glass nanocomposite. *Progress in Biomaterials*, 8(2), 77–89.
- Marghussian, V. K., & Mesgar, A. S.-M. (2000). Effects of composition on crystallization behaviour and mechanical properties of bioactive glass-ceramics in the  $\text{MgO}-\text{CaO}-\text{SiO}_2-\text{P}_2\text{O}_5$  system. *Ceramics International*, 26(4), 415–420.
- Marguí, E., Queralt, I., & de Almeida, E. (2022). X-ray fluorescence spectrometry for environmental analysis: Basic principles, instrumentation, applications and recent trends. *Chemosphere*, 303, 135006.
- Martin-Luengo, M. A., Yates, M., Ramos, M., Salgado, J. L., Aranda, R. M., Plou, F., Sanz, J. L., Pirrongelli, R. L., Rojo, E. S., Gil, L. G., Serrano, A. M., & Ruiz-Hitzky, E. (2011). Renewable raw materials for advanced applications. *2011 World Congress on Sustainable Technologies, WCST 2011*, 19–22.
- Martin-Sedeno, M. C., Losilla, E. R., Leon-Reina, L., Bruque, S., Marrero-Lopez, D., Nunez, P., & Aranda, M. A. G. (2004). Enhancement of oxide ion conductivity in cuspidine-type materials. *Chemistry of Materials*, 16(24), 4960–4968.

- Mirhadi, B., & Mehdikhani, B. (2012). Effect of calcium fluoride on sintering behavior of  $\text{SiO}_2\text{--CaO--Na}_2\text{O--MgO}$  glass-ceramic system. *Processing and Application of Ceramics*, 6(3), 159–164.
- Mirza, A., Riaz, M., Zia, R., Hussain, T., Bashir, F., Riaz, M., Zia, R., Hussain, T., & Bashir, F. (2017). Effect of temperature on mechanical and bioactive properties of glass-ceramics. *Journal of Alloys and Compounds*, 726, 348–351.
- Moghanian, A., Tajer, M. H. M., Zohourfazeli, M., Miri, Z., & Yazdi, M. S. (2021). Sol-gel derived silicate-based bioactive glass: Studies of synergetic effect of zirconium and magnesium on structural and biological characteristics. *Journal of Non-Crystalline Solids*, 554, 120613.
- Moh, Y. C., & Manaf, L. A. (2017). Solid waste management transformation and future challenges of source separation and recycling practice in Malaysia. *Resources, Conservation and Recycling*, 116(2017), 1–14.
- Mohadi, R., Anggraini, K., Riyanti, F., & Lesbani, A. (2016). Preparation calcium oxide from chicken eggshells. *Sriwijaya Journal of Environment*, 1(2), 32–35.
- Mollazadeh, S., Yekta, B. E., Javadpour, J., Yusefi, A., & Jafarzadeh, T. S. (2013). The role of  $\text{TiO}_2$ ,  $\text{ZrO}_2$ ,  $\text{BaO}$  and  $\text{SiO}_2$  on the mechanical properties and crystallization behavior of fluorapatite-mullite glass-ceramics. *Journal of Non-Crystalline Solids*, 361(1), 70–77.
- Monmaturapoj, N., Lawita, P., & Thepsuwan, W. (2013). Characterisation and properties of lithium disilicate glass ceramics in the  $\text{SiO}_2\text{--Li}_2\text{O--K}_2\text{O--Al}_2\text{O}_3$  system for dental applications. *Advances in Materials Science and Engineering*, 763838.
- Montazeri, N., Jahandideh, R., & Biazar, E. (2011). Synthesis of fluorapatite-hydroxyapatite nanoparticles and toxicity investigations. *International Journal of Nanomedicine*, 6, 197–201.
- Montazerian, M., & Zanolto, E. D. (2017). Bioactive and inert dental glass-ceramics. *Journal of Biomedical Materials Research – Part A*, 105(2), 619–639.
- Montazerian, M., Bairo, F., Fiume, E., Migneco, C., Alaghmandfard, A., Sedighi, O., DeCeanne, A. V., Wilkinson, C. J., & Mauro, J. C. (2023). Glass-ceramics in dentistry: Fundamentals, technologies, experimental techniques, applications, and open issues. *Progress in Materials Science*, 132, 101023.
- Morena, R., Lockwood, P. E., & Fairhurst, C. W. (1986). Fracture toughness of commercial dental porcelains. *Dental Materials*, 2(2), 58–62.
- Mozafari, M., Banijamali, S., Bairo, F., Kargozar, S., & Hill, R. G. (2019). Calcium carbonate: Adored and ignored in bioactivity assessment. *Acta Biomaterialia*, 91, 35–47.



- Muniz, R. F., Soares, V. O., Montagnini, G. H., Medina, A. N., & Baesso, M. L. (2021). Thermal, optical and structural properties of relatively depolymerized sodium calcium silicate glass and glass-ceramic containing  $\text{CaF}_2$ . *Ceramics International*, 47(17), 24966–24972.
- Naga, S. M., El-Maghraby, H. F., Sayed, M., & Saad, E. A. (2015). Highly porous scaffolds made of nanosized hydroxyapatite powder synthesized from eggshells. *Journal of Ceramic Science and Technology*, 6(3), 237–243.
- Najah, M. I., Journal, I., Najah, M. I., Razak, A., Aida, N., Shukur, C., Adzila, S., Othman, R., & Nordin, N. (2020). Characterization of calcium carbonate extracted from eggshell waste at various calcination temperature. *International Journal of Emerging Trends in Engineering Research*, 8(10), 6725–6731.
- Nandi, S. K., Mahato, A., Kundu, B., & Mukherjee, P. (2016). Doped bioactive glass materials in bone regeneration. *Advanced Techniques in Bone Regeneration* (pp. 276). InTech.
- Naresh, P., Narsimlu, N., Srinivas, C., Shareefuddin, M., & Kumar, K. S. (2020).  $\text{Ag}_2\text{O}$  doped bioactive glasses: An investigation on the antibacterial, optical, structural and impedance studies. *Journal of Non-Crystalline Solids*, 549, 120361.
- Nayak, J. P., & Bera, J. (2010a). Effect of sintering temperature on mechanical behaviour and bioactivity of sol-gel synthesized bioglass-ceramics using rice husk ash as a silica source. *Applied Surface Science*, 257(2), 458–462.
- Nayak, J. P., Kumar, S., & Bera, J. (2010b). Sol-gel synthesis of bioglass-ceramics using rice husk ash as a source for silica and its characterization. *Journal of Non-Crystalline Solids*, 356(28–30), 1447–1451.
- Neel, E. A. A., & Knowles, J. C. (2008). Biocompatibility and other properties of phosphate-based glasses for medical applications. *Cellular response to biomaterials* (pp. 156–290). Woodhead Publishing Limited.
- Neel, E. A. A., Aljabo, A., Strange, A., Ibrahim, S., Coathup, M., Young, A. M., Bozec, L., & Mudera, V. (2016). Demineralization–remineralization dynamics in teeth and bone. *International journal of nanomedicine*, 11, 4743–4763.
- Ni, M., & Ratner, B. D. (2008). Differentiating calcium carbonate polymorphs by surface analysis techniques—an XPS and TOF-SIMS study. *Surface and Interface Analysis: An International Journal devoted to the development and application of techniques for the analysis of surfaces, interfaces and thin films*, 40(10), 1356–1361.
- Nicholson, J. W. (2020a). Synthetic Materials in Medicine. *The chemistry of medical and dental materials* (Vol. 7). Royal Society of Chemistry.
- Nicholson, J. W., Sidhu, S. K., & Czarnecka, B. (2020b). Enhancing the mechanical properties of glass-ionomer dental cements: A review. *Materials*, 13(11), 1–14.

- Nunez-Rodriguez, L. A., Encinas-Romero, M. A., Gomez-Alvarez, A., Valenzuela-Garcia, J. L., & Tiburcio-Munive, G. C. (2018). Evaluation of bioactive properties of  $\alpha$  and  $\beta$  wollastonite bioceramics soaked in a simulated body fluid. *Journal of Biomaterials and Nanobiotechnology*, 09(3), 263–276.
- Odermatt, R., Par, M., Mohn, D., Wiedemeier, D. B., Attin, T., & Tauböck, T. T. (2020). Bioactivity and physico-chemical properties of dental composites functionalized with nano- vs. micro-sized bioactive glass. *Journal of Clinical Medicine*, 9(3), 1–13.
- Ohtsuki, C., Kokubo, T., & Yamamuro, T. (1992). Mechanism of apatite formation on  $\text{CaO-SiO}_2\text{-P}_2\text{O}_5$  glasses in a simulated body fluid. *Journal of Non-Crystalline Solids*, 143, 84–92.
- Ojovan, M. I. (2021). Glass formation kinetic theory of vitrification. *Encyclopedia of Glass Science, Technology, History, and Culture: Vol. I* (pp. 249–259). John Wiley & Sons, Inc.
- Oliveira, D. A., Benelli, P., & Amante, E. R. (2013). A literature review on adding value to solid residues: Egg shells. *Journal of Cleaner Production*, 46, 42–47.
- Onwubu, S. C., Mdluli, P. S., Singh, S., Madikizela, L., & Ngombane, Y. (2019). Characterization and in vitro evaluation of an acid resistant nanosized dental eggshell-titanium dioxide material. *Advanced Powder Technology*, 30(4), 766–773.
- Owoeye, S. S., Abegunde, S. M., Folorunso, D. O., Adigun, B. O., & Kingsley, U. (2021). Microstructure, phase and physical evaluation of non-bioactive wollastonite glass – Ceramic prepared from waste glass by sintering method. *Open Ceramics*, 5(10), 1–9.
- Oyedotun, T. D. T. (2018). X-ray fluorescence (XRF) in the investigation of the composition of earth materials: A review and an overview. *Geology, Ecology, and Landscapes*, 2(2), 148–154.
- Padilla, S., Román, J., Sánchez-Salcedo, S., & Vallet-Regí, M. (2006). Hydroxyapatite/ $\text{SiO}_2\text{-CaO-P}_2\text{O}_5$  glass materials: In vitro bioactivity and biocompatibility. *Acta Biomaterialia*, 2(3), 331–342.
- Pajor, K., Pajchel, L., & Kolmas, J. (2019). Hydroxyapatite and fluorapatite in conservative. *Materials*, 12(2683), 1–16.
- Pal, M., Pal, U., Jiménez, J. M. G. Y., & Pérez-Rodríguez, F. (2012). Effects of crystallization and dopant concentration on the emission behavior of  $\text{TiO}_2$ : Eu nanophosphors. *Nanoscale Research Letters*, 7, 1–12.
- Palakurthy, S., Reddy, K. V., Patel, S., & Azeem, P. A. (2020). A cost effective  $\text{SiO}_2\text{-CaO-Na}_2\text{O}$  bio-glass derived from bio-waste resources for biomedical applications. *Progress in Biomaterials*, 9, 239–248.

- Pallan, N. F. B., Matori, K. A., Hashim, M., Azis, R. S., Zainuddin, N., Pallan, N. F. B., Idris, F. M., Ibrahim, I. R., Wah, L. C., Rusly, S. N. A., Adnin, N., Khiri, M. Z. A., Alassan, Z. N., Mohamed, N., & Zaid, M. H. M. (2019). Effects of different sintering temperatures on thermal, physical, and morphological of  $\text{SiO}_2\text{--Na}_2\text{O--CaO--P}_2\text{O}_5$  based glass-ceramic system from vitreous and ceramic wastes. *Science of Sintering*, 51(4), 377–387.
- Panah, N. G., Atkin, R., & Sercombe, T. B. (2022). Bioactivity and biodegradability of high temperature sintered 58S ceramics. *Journal of the European Ceramic Society*, 42(8), 3614–3623.
- Pankaew, P., Kaewwiset, W., Naemchanthara, K., Hoonnivathana, E., & Limsuwan, P. (2014). Characterization of  $\text{CaCO}_3$  polymorphs prepared from waste eggshell. *Journal of Applied Science Research*, 9(12), 6085–6090.
- Paramita, P., Ramachandran, M., Narashiman, S., Nagarajan, S., Sukumar, D. K., Chung, T. W., & Ambigapathi, M. (2021). Sol–gel based synthesis and biological properties of zinc integrated nano bioglass ceramics for bone tissue regeneration. *Journal of Materials Science: Materials in Medicine*, 32(1), 1–11.
- Paul, A. (1982). Phase transformations in glass. *Chemistry of Glasses* (pp. 16–50). Springer Dordrecht.
- Peccati, F., Bernocco, C., Ugliengo, P., & Corno, M. (2018). Properties and reactivity towards water of A-type carbonated apatite and hydroxyapatite surfaces. *The Journal of Physical Chemistry C*, 122(7), 3934–3944.
- Pei, F., Zhu, G., Li, P., Guo, H., & Yang, P. (2020). Effects of  $\text{CaF}_2$  on the sintering and crystallisation of  $\text{CaO--MgO--Al}_2\text{O}_3\text{--SiO}_2$  glass-ceramics. *Ceramics International*, 46(11), 17825–17835.
- Peitl, O., Zanutto, E. D., & Hench, L. L. (2001). Highly bioactive  $\text{P}_2\text{O}_5\text{--Na}_2\text{O--CaO--SiO}_2$  glasses-ceramics. *Journal of Non-Crystalline Solids*, 292, 115–126.
- Persson, C., Unosson, E., Ajaxon, I., Engstrand, J., Engqvist, H., & Xia, W. (2012). Nano grain sized zirconia-silica glass ceramics for dental applications. *Journal of the European Ceramic Society*, 32(16), 4105–4110.
- Piatti, E., Verné, E., & Miola, M. (2022). Synthesis and characterization of sol-gel bioactive glass nanoparticles doped with boron and copper. *Ceramics International*, 48(10), 13706–13718.
- Pornchai, T., Imkum, A., & Apipong, P. (2016). Effect of calcination time on physical and chemical properties of  $\text{CaO}$ -catalyst derived from industrial-eggshell wastes. *Journal of Science and Technology Mahasarakham University*, 35(6), 693–697.
- Poskus, L. T., Placido, E., & Cardoso, P. E. C. (2004). Influence of placement techniques on Vickers and Knoop hardness of class II composite resin restorations. *Dental Materials*, 20(8), 726–732.

- Possolli, N. M., da Silva, D. F., Vieira, J., Maurmann, N., Pranke, P., Demétrio, K. B., Angioletto, E., Montedo, O. R. K., & Arcaro, S. (2022). Dissolution, bioactivity behavior, and cytotoxicity of  $19.58\text{Li}_2\text{O} \cdot 11.10\text{ZrO}_2 \cdot 69.32\text{SiO}_2$  glass-ceramic. *Journal of Biomedical Materials Research - Part B Applied Biomaterials*, 110(1), 67–78.
- Prasad, S., Vyas, V. K., Ershad, M., & Pyare, R. (2017a). Crystallization and mechanical properties of (45S5-HA) biocomposite for biomedical implantation. *Ceramics - Silikaty*, 61(4), 378–384.
- Prasad, S., Vyas, V. K., Mani, K. D., Ershad, M., & Pyare, R. (2017b). Preparation, in-vitro bioactivity and mechanical properties of reinforced 45S5 bioglass composite with HA-ZrO<sub>2</sub> powders. *Oriental Journal of Chemistry*, 33(3), 1286–1296.
- Prasad, S., Datta, S., Adarsh, T., Diwan, P., Annapurna, K., Kundu, B., & Biswas, K. (2018). Effect of boron oxide addition on structural, thermal, in vitro bioactivity and antibacterial properties of bioactive glasses in the base S53P4 composition. *Journal of Non-Crystalline Solids*, 498, 204–215.
- Prasad, S., Ganiseti, S., Jana, A., Kant, S., Sinha, P. K., Tripathy, S., Illath, K., Ajithkumar, T. G., Annapurna, K., Allu, A. R., & Biswas, K. (2020). Elucidating the effect of CaF<sub>2</sub> on structure, biocompatibility and antibacterial properties of S53P4 glass. *Journal of Alloys and Compounds*, 831, 154704.
- Pushie, M. J., Pickering, I. J., Korbas, M., Hackett, M. J., & George, G. N. (2014). Elemental and chemically specific X-ray fluorescence imaging of biological systems. *Chemical Reviews*, 114(17), 8499–8541.
- Putra, W. P., Kamari, A., Yusoff, S. N. M., Ishak, C. F., Mohamed, A., Hashim, N., & Isa, I. M. (2014). Biosorption of Cu(II), Pb(II) and Zn(II) ions from aqueous solutions using selected waste materials: adsorption and characterisation studies. *Journal of Encapsulation and Adsorption Sciences*, 4(1), 25–35.
- Rad, R. M., Alshemary, A. Z., Evis, Z., Keskin, D., Altunbaş, K., & Tezcaner, A. (2018). Structural and biological assessment of boron doped bioactive glass nanoparticles for dental tissue application. *Ceramics International*, 44(11), 13453.
- Rahman, N. A. A., Matori, K. A., Zaid, M. H. M., Zainuddin, N., Aziz, S. A., Khiri, M. Z. A., Jalil, R. A., & Jusoh, W. N. W. (2019). Fabrication of alumino-silicate-fluoride based bioglass derived from waste clam shell and soda lime silica glasses. *Results in Physics*, 12, 743–747.
- Rahyussalim, A. J., Supriadi, S., Marsetio, A. F., Pribadi, P. M., & Suharno, B. (2019). The potential of carbonate apatite as an alternative bone substitute material. *Medical Journal of Indonesia*, 28(1), 92–97.
- Rajaramakrishna, R., & Kaewkhao, J. (2019). Glass material and their advanced applications. *KnE Social Sciences*, 796–807.



- Rasteiro, M. G., Gassman, T., Santos, R., & Antunes, E. (2007). Crystalline phase characterization of glass-ceramic glazes. *Ceramics International*, 33(3), 345–354.
- Raszewski, Z., Kulbacka, J., & Nowakowska-Toporowska, A. (2022). Mechanical properties, cytotoxicity, and fluoride ion release capacity of bioactive glass-modified methacrylate resin used in three-dimensional printing technology. *Materials*, 15(3), 1133.
- Rawlings, R. D., Wu, J. P., & Boccaccini, A. R. (2006). Glass-ceramics: Their production from wastes – A Review. *Journal of Materials Science*, 41(3), 733–761.
- Razali, N., Jumadi, N., Jalani, A. Y., Kamarulzaman, N. Z., & Pa'ee, K. F. (2022). Thermal decomposition of calcium carbonate in chicken eggshells: Study on temperature and contact time. *Malaysian Journal of Analytical Sciences*, 26(2), 347–359.
- Reza, H., Hassan, R., Rizi, B., Mahdi, M., Khamseh, R., & Öchsner, A. (2020). *A review on dental materials*. Springer International Publishing.
- Rezaee, T., Bouxsein, M. L., & Karim, L. (2020). Increasing fluoride content deteriorates rat bone mechanical properties. *Bone*, 136, 115369.
- Riaz, S., Mills, K. C., Nagata, K., Ludlow, V., & Normanton, A. S. (2004). Determination of mould powder crystallinity using X-ray diffractometry. *High Temperature Materials and Processes*, 22(5–6), 379–386.
- Richet, P., Robie, R. A., & Hemingway, B. S. (1991). Thermodynamic properties of wollastonite, pseudowollastonite and  $\text{CaSiO}_3$  glass and liquid. *European Journal of Mineralogy*, 3(3), 475–484.
- Rintoul, L., Wentrup-Byrne, E., Suzuki, S., & Grøndahl, L. (2007). FT-IR spectroscopy of fluoro-substituted hydroxyapatite: Strengths and limitations. *Journal of Materials Science: Materials in Medicine*, 18(9), 1701–1709.
- Rocton, N., Oudadesse, H., Lefeuvre, B., Peisker, H., & Rbii, K. (2020). Fine analysis of interaction mechanism of bioactive glass surface after soaking in SBF solution: AFM and ICP-OES investigations. *Applied Surface Science*, 505, 144076.
- Rodriguez, O., Alhalawani, A., Arshad, S., & Towler, M. (2018). Rapidly-dissolving silver-containing bioactive glasses for cariostatic applications. *Journal of Functional Biomaterials*, 9(2), 28.
- Rohim, R., Ahmad, R., Ibrahim, N., Hamidin, N., & Abidin, C. Z. A. (2014). Characterization of calcium oxide catalyst from eggshell waste. *Advances in Environmental Biology*, 8(22), 35–38.
- Saadaldin, S. A., & Rizkalla, A. S. (2014). Synthesis and characterization of wollastonite glass-ceramics for dental implant applications. *Dental Materials*,

30(3), 364–371.

Saint-Jean, S. J. (2014). Dental glasses and glass-ceramics. *Advanced Ceramics for Dentistry* (pp. 255–277). Elsevier Inc.

Sakamoto, A., & Yamamoto, S. (2010). Glass-ceramics: Engineering principles and applications. *International Journal of Applied Glass Science*, 1(3), 237–247.

Saparuddin, D. I., Hisham, N. A. N., Aziz, S. A., Matori, K. A., Honda, S., Iwamoto, Y., & Zaid, M. H. M. (2020). Effect of sintering temperature on the crystal growth, microstructure and mechanical strength of foam glass-ceramic from waste materials. *Journal of Materials Research and Technology*, 9(3), 5640–5647.

Saranti, A., Koutselas, I., & Karakassides, M. A. (2006). Bioactive glasses in the system  $\text{CaO-B}_2\text{O}_3\text{-P}_2\text{O}_5$ : Preparation, structural study and in vitro evaluation. *Journal of Non-Crystalline Solids*, 352(5), 390–398.

Scherer, G. W. (2001). Viscous sintering. *Encyclopedia of Materials: Technical Ceramics and Glasses* (pp. 9536–9540). Elsevier Science Ltd.

Schindler, H. J. (2018). Fracture toughness for engineering application: There is a need for more suitable testing standards. *Procedia Structural Integrity*, 13, 398–403.

Schubert, U. S., & Husing, N. (2019). *Synthesis of Inorganic Materials*. John Wiley & Sons.

Seo, Y., Goto, T., Nishida, H., Cho, S. H., Zarkov, A., Yamamoto, T., & Sekino, T. (2020). Low-temperature mineralization sintering process for fabrication of fluoridated hydroxyapatite-containing bioactive glass. *Journal of the Ceramic Society of Japan*, 128(10), 783–789.

Sha, S., Omid, M., Nasehi, F., Golzar, H., & Mohammadrezaei, D. (2019). Egg shell-derived calcium phosphate/carbon dot nano fibrous scaffolds for bone tissue engineering: Fabrication and characterization. *Materials Science and Engineering C*, 100, 564–575.

Shah, F. A., Brauer, D. S., Hill, R. G., & Hing, K. A. (2015). Apatite formation of bioactive glasses is enhanced by low additions of fluoride but delayed in the presence of serum proteins. *Materials Letters*, 153, 143–147.

Shah, F. A. (2016a). Fluoride-containing bioactive glasses: Glass design, structure, bioactivity, cellular interactions, and recent developments. *Materials Science and Engineering C*, 58, 1279–1289.

Shah, A. T., Batool, M., Chaudhry, A. A., Iqbal, F., Javaid, A., Zahid, S., Ilyas, K., Qasim, S. B., Khan, A. F., Khan, A. S., & Rehman, I. U. (2016b). Effect of calcium hydroxide on mechanical strength and biological properties of bioactive glass. *Journal of the Mechanical Behavior of Biomedical Materials*, 61, 617–626.

- Shahrouzifar, M. R., Salahinejad, E., & Sharifi, E. (2019). Co-incorporation of strontium and fluorine into diopside scaffolds: Bioactivity, biodegradation and cytocompatibility evaluations. *Materials Science and Engineering C*, 103, 109752.
- Shari, F., Parvin, N., & Tahriri, M. (2017). Sythesis and characteristics of sol gel bioactive  $\text{SiO}_2\text{-P}_2\text{O}_5\text{-CaO-Ag}_2\text{O}$  glasses. *Journal of Non-Crystalline Solids*, 476, 108–113.
- Sharifianjazi, F., Moradi, M., Abouchenari, A., Pakseresht, A. H., Esmailkhanian, A., Shokouhimehr, M., & Asl, M. S. (2020). Effects of Sr and Mg dopants on biological and mechanical properties of  $\text{SiO}_2\text{-CaO-P}_2\text{O}_5$  bioactive glass. *Ceramics International*, 46(14), 22674–22682.
- Sharma, S. K., Verma, D. S., Khan, L. U., Kumar, S., & Khan, S. B. (2018a). Fourier transform infrared spectroscopy: Fundamentals and application in functional groups and nanomaterials characterization. *Handbook of Materials Characterization* (pp. 317–344). Springer International Publishing.
- Sharma, S. K., Verma, D. S., Khan, L. U., Kumar, S., & Khan, S. B. (2018b). Scanning electron microscopy: Principle and applications in nanomaterials characterization. *Handbook of Materials Characterization* (pp. 113–145). Springer International Publishing.
- Shearer, A., Montazerian, M., & Mauro, J. C. (2023). Modern definition of bioactive glasses and glass-ceramics. *Journal of Non-Crystalline Solids*, 608, 122228.
- Shekhawat, M. S. (2015). A review paper on glass-ceramics. *International Journal of Materials Physics*, 6(1), 1–6.
- Shelby, J. E. (2020). *Introduction to Glass Science and Technology*. Royal society of chemistry.
- Siemiradzka, W., Dolinska, B., & Ryszka, F. (2018). New Sources of calcium (chicken eggshells, chelates) – Preparation of raw material and tablets. *Current Pharmaceutical Biotechnology*, 19(7), 566–572.
- Singh, K., & Walia, T. (2021). Review on silicate and borosilicate-based glass sealants and their interaction with components of solid oxide fuel cell. *International Journal of Energy Research*, 45(15), 20559–20582.
- Singh, S., Patil, A., Mali, S., & Jaiswal, H. (2022). Bioglass: A new era in modern dentistry. *European Journal of General Dentistry*, 11(1), 001–006.
- Skallevold, H. E., Rokaya, D., Khurshid, Z., & Zafar, M. S. (2019). Bioactive glass applications in dentistry. *International Journal of Molecular Sciences*, 20(23), 1–24.
- Srinath, P., Azeem, P. A., Reddy, K. V., & Kumar, S. R. (2019). Synthesis and in vitro bioactivity of  $\text{SiO}_2\text{-CaO-Na}_2\text{O}$  glass using bio-waste resources. *AIP Conference*

- Sriranganathan, D., Chen, X., Hing, K. A., Kanwal, N., & Hill, R. G. (2017). The effect of the incorporation of fluoride into strontium containing bioactive glasses. *Journal of Non-Crystalline Solids*, 457, 25–30.
- Srivatsan, T. S. (2009). *Processing and Fabrication of Advanced Materials, XVII: Part 8: Polymer-based Composites and Nano Composites: Volume Two*. IK International Pvt Ltd.
- Stábile, F. M., & Volzone, C. (2019). Crystallization and sintering of glasses formulated from different theoretical contents of leucite (L) and bioglass 45S5 (Bg) (LxBg(100-x) x:25, 30, 40, 50): A comparative study on raw materials influence. *Materials Chemistry and Physics*, 238, 121892.
- Stavros, V. G. (2014). A bright future for glass-ceramic. *Nature Chemistry*, 6(11), 955–956.
- Sujirote, K., Rawlings, R. D., & Rogers, P. S. (1998). Effect of fluoride on sinterability of a silicate glass powder. *Journal of the European Ceramic Society*, 18(9), 1325–1330.
- Sund-Levander, M., Forsberg, C., & Wahren, L. K. (2002). Normal oral, rectal, tympanic and axillary body temperature in adult men and women: A systematic literature review. *Scandinavian Journal of Caring Sciences*, 16(2), 122–128.
- Tangboriboon, N., Kunanurksapong, R., & Sirivat, A. (2012). Preparation and properties of calcium oxide from eggshells via calcination. *Materials Science-Poland*, 30(4), 313–322.
- Tilocca, A. (2010). Models of structure, dynamics and reactivity of bioglasses: A review. *Journal of Materials Chemistry*, 20(33), 6848–6858.
- Tsai, W. T., Hsien, K. J., Hsu, H. C., Lin, C. M., Lin, K. Y., & Chiu, C. H. (2008). Utilization of ground eggshell waste as an adsorbent for the removal of dyes from aqueous solution. *Bioresource Technology*, 99(6), 1623–1629.
- Tshizanga, N., Aransiola, E. F., & Oyekola, O. (2017). Optimisation of biodiesel production from waste vegetable oil and eggshell ash. *South African Journal of Chemical Engineering*, 23, 145–156.
- Unyi, T., Juhász, A., Tasnádi, P., & Lendvai, J. (2000). Changes of the mechanical properties during the crystallization of bio-active glass-ceramics. *Journal of Materials Science*, 35(12), 3059–3068.
- Uskoković, V., Abuna, G., Ferreira, P., Wu, V. M., Gower, L., Pires-de-Souza, F. C. P., Murata, R. M., Sinhoreti, M. A. C., & Geraldini, S. (2021). Synthesis and characterization of nanoparticulate niobium- and zinc-doped bioglass-ceramic/chitosan hybrids for dental applications. *Journal of Sol-Gel Science and Technology*, 97(2), 245–258.



- Vaidya, A., & Pathak, K. (2019a). Mechanical stability of dental materials. *Applications of Nanocomposite Materials in Dentistry* (pp. 285–305). Elsevier Inc.
- Vaidya, N., & Bastwadkar, M. P. (2019b). Experimental study of partial replacement of cement with eggshell powder in concrete. *International Journal of Engineering Development and Research*, 7(4), 212–216.
- Van de Voort, F. R. (1992). Fourier transform infrared spectroscopy applied to food analysis. *Food Research International*, 25(5), 397–403.
- Vercamer, V. (2016). *Spectroscopic and Structural Properties of Iron in Silicate Glasses*. Université Pierre et Marie Curie - Paris VI.
- Vyas, V. K., Kumar, A. S., Ali, A., Prasad, S., Srivastava, P., Mallick, S. P., Ershad, M., Singh, S. P., & Pyare, R. (2016). Assessment of nickel oxide substituted bioactive glass-ceramic on in vitro bioactivity and mechanical properties. *Boletín de La Sociedad Espanola de Ceramica y Vidrio*, 55(6), 228–238.
- Waheed, M., Yousaf, M., Shehzad, A., Inam-Ur-Raheem, M., Khan, M. K. I., Khan, M. R., Ahmad, N., Abdullah, & Aadil, R. M. (2020). Channelling eggshell waste to valuable and utilizable products: A comprehensive review. *Trends in Food Science and Technology*, 106, 78–90.
- Wallace, K. E., Hill, R. G., Pembroke, J. T., Brown, C. J., & Hatton, P. V. (1999). Influence of sodium oxide content on bioactive glass properties. *Journal of Materials Science: Materials in Medicine*, 10(12), 697–701.
- Wang, L., D'Alpino, P. H. P., Lopes, L. G., & Pereira, J. C. (2003). Mechanical properties of dental restorative materials: Relative contribution of laboratory tests. *Journal of Applied Oral Science*, 11(3), 162–167.
- Wang, W. H. (2012). The elastic properties, elastic models and elastic perspectives of metallic glasses. *Progress in Materials Science*, 57(3), 487–656.
- Wei, M., Evans, J. H., Bostrom, T., & Grondahl, L. (2003). Synthesis and characterization of hydroxyapatite, fluoride-substituted hydroxyapatite and fluorapatite. *Journal of Materials Science: Materials in Medicine*, 14(1), 311–320.
- Wei, Z., Xu, C., & Li, B. (2009). Application of waste eggshell as low-cost solid catalyst for biodiesel production. *Bioresource Technology*, 100(11), 2883–2885.
- Wen, C., Bai, N., Luo, L., Ye, J., Zhan, X., Zhang, Y., & Sa, B. (2021). Structural behavior and in vitro bioactivity evaluation of hydroxyapatite-like bioactive glass based on the SiO<sub>2</sub>–CaO–P<sub>2</sub>O<sub>5</sub> system. *Ceramics International*, 47(13), 18094–19104.
- Winter, A. (1957). Glass formation. *Journal of the American Ceramic Society*, 40(2), 54–58.

- Wojcik, N. A., Wolff, S., Karczewski, J., Ryl, J., & Ali, S. (2022). Effect of crystallinity on structural, thermal, and in vitro dissolution properties of  $\text{Na}_2\text{O}-\text{CaO}-\text{Nb}_2\text{O}_5/\text{MgO}-\text{P}_2\text{O}_5$  glass-ceramics. *Journal of the European Ceramic Society*, 43(5), 2234–2244.
- Workie, A. B., & Shih, S.-J. (2022). A study of bioactive glass–ceramic’s mechanical properties, apatite formation, and medical applications. *RSC Advances*, 12(36), 23143–23152.
- Xia, L., Ma, W., Zhou, Y., Gui, Z., Yao, A., Wang, D., Takemura, A., Uemura, M., Lin, K., & Xu, Y. (2019). Stimulatory Effects of boron containing bioactive glass on osteogenesis and angiogenesis of polycaprolactone: In vitro study. *BioMed Research International*, 2019, 8961409.
- Xie, K., Zhang, L., Yang, X., Wang, X., Yang, G., Zhang, L., Shao, H., He, Y., Fu, J., & Gou, Z. (2015). Preparation and characterization of low temperature heat-treated 45S5 bioactive glass-ceramic analogues. *Biomedical Glasses*, 1(1), 80–92.
- Xu, S., Yang, X., Chen, X., Shao, H., He, Y., Zhang, L., Yang, G., & Gou, Z. (2014). Effect of borosilicate glass on the mechanical and biodegradation properties of 45S5-derived bioactive glass-ceramics. *Journal of Non-Crystalline Solids*, 405, 91–99.
- Xu, H., Zou, J., Wang, W., Wang, H., Ji, W., & Fu, Z. (2021). Densification mechanism and microstructure characteristics of nano- and micro- crystalline alumina by high-pressure and low temperature sintering. *Journal of the European Ceramic Society*, 41(1), 635–645.
- Xuan, N., Tram, T., Ishikawa, K., Minh, T. H., Benson, D., & Tsuru, K. (2021). Characterization of carbonate apatite derived from chicken bone and its in-vitro evaluation using MC3T3-E1 cells. *Materials Research Express*, 8, 025401.
- Yahyazadehfard, M., Ivancik, J., Majd, H., An, B., Zhang, D., & Arola, D. (2014). On the mechanics of fatigue and fracture in teeth. *Applied Mechanics Reviews*, 66(3), 1–19.
- Yanikoglu, N. D., & Sakarya, R. E. (2020). Test methods used in the evaluation of the structure features of the restorative materials: A literature review. *Journal of Materials Research and Technology*, 9(5), 9720–9734.
- Yeo, T. M., Jeon, J. M., Hyun, S. H., Ha, H. M., & Cho, J. W. (2022). Effects of  $\text{Li}_2\text{O}$  on structure of  $\text{CaO}-\text{SiO}_2-\text{CaF}_2-\text{Na}_2\text{O}$  glasses and origin of crystallization delay. *Journal of Molecular Liquids*, 347, 117997.
- Yi, J., Dai, Q., Weir, M. D., Melo, M. A., Lynch, C. D., Oates, T. W., Zhang, K., Zhao, Z., & Xu, H. H. (2019). A nano- $\text{CaF}_2$ -containing orthodontic cement with antibacterial and remineralization capabilities to combat enamel white spot lesions. *Journal of dentistry*, 89, 103172.

- Yusof, N. N., Aziz, S. M., Noor, F. M., Yaacob, S. N. S., & Hashim, S. (2022). A novel borate-based 45S5 Bioglass®: In vitro assessment in phosphate-buffered saline solution. *Journal of Non-Crystalline Solids*, 596, 121843.
- Zachariasen, W. H. (1932). The atomic arrangement in glass. *Journal of the American Chemical Society*, 54(10), 3841–3851.
- Zhao, S., Liu, B., Ding, Y., Zhang, J., Wen, Q., Ekberg, C., & Zhang, S. (2020). Study on glass-ceramics made from MSWI fly ash, pickling sludge and waste glass by one-step process. *Journal of Cleaner Production*, 271, 122674.
- Zhao, S. Z., Zhang, X. Y., Liu, B., Zhang, J. J., Shen, H. L., & Zhang, S. G. (2021). Preparation of glass-ceramics from high-chlorine MSWI fly ash by one-step process. *Rare Metals*, 40(11), 3316–3328.
- Zhou, Q., Xu, L., Liu, L., Wang, W., Zhu, C., & Gan, F. (2004). Study on the laser-induced darkening in Nd-doped laser glasses. *Optical Materials*, 25(3), 313–319.
- Zulkipli, F., Nopiah, Z. M., Jamian, N. H., Basri, N. E. A., & Kie, C. J. (2022). Mean score analysis on awareness of solid waste management in Malaysia. *International Journal of Academic Research in Business and Social Sciences*, 12(6), 649–658.