



**POWER CONVERSION EFFICIENCY ENHANCEMENT USING SILVER
NANOPARTICLE AND TITANIUM DIOXIDE COMPOSITE FILM FOR
FLEXIBLE TITANIUM FOIL ELECTRODE OF BACK-ILLUMINATED DYE-
SENSITIZED SOLAR CELL**

By

NUR HAZAHSHA BINTI SHAMSUDIN

**Thesis Submitted to the School of Graduate Studies, Universiti
Putra Malaysia, in Fulfilment of the Requirements for the Degree of
Doctor of Philosophy**

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DEDICATION

I dedicate this dissertation to my beloved mom, Zaharah Binti Ibrahim, the woman who sacrifices everything for me unconditionally. You are my source of strength throughout this journey.

Thank you Allah

Thank you ‘



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

POWER CONVERSION EFFICIENCY ENHANCEMENT USING SILVER NANOPARTICLE AND TITANIUM DIOXIDE COMPOSITE FILM FOR FLEXIBLE TITANIUM FOIL ELECTRODE OF BACK-ILLUMINATED DYE-SENSITIZED SOLAR CELL

By

NUR HAZAHSHA BINTI SHAMSUDIN

June 2022

Chair: Prof. Ir. Mohd Zainal Abidin Bin Ab Kadir, PhD
Faculty: Engineering

Power conversion efficiency (PCE) in dye-sensitized solar cells (DSSCs) is a pivotal parameter to gauge the ability to convert light energy into electricity. However, the preference of substrates is among the issues that contribute to a low PCE. While polymer substrates have been exploited as the photoelectrodes for their flexibility and transparency which allows for front illumination, high temperature is required to eliminate the polymer binder from deposited titanium oxide (TiO_2) paste. The polymer substrates are unable to withstand high temperature, and as such affect the PCE with the residue of the polymer binder within TiO_2 film. Herein, a flexible titanium (Ti) substrate, a metal-based material, is used to replace the polymer substrate as the photoelectrode. Inherently, back illumination through counter electrode was opted for instead of front illumination through photoelectrode, owing to the opaque characteristic of metal substrate. However, the longer distance for the light to travel via this route impeded the development of photoelectrons, and thus reducing the power conversion efficiency (PCE). Therefore, the aim of this research is to improve the PCE via back illumination method. Herein, there were two methods proposed so as to increase the PCE. The first method is by coating the counter electrode with a different volume of platinum (Pt) solution at 30 μl , 50 μl , 70 μl and 90 μl . The second method proposed was incorporation of silver nanoparticles (AgNP) prepared via solvothermal into TiO_2 paste. Platinum is integrated in the counter electrode for the optimization of PCE through catalytic activity induced by Pt. It was revealed that 70 μL of Pt solution increased the PCE from 1.248 % to 5.25 %. The incorporation of AgNP into TiO_2 film has been proven to improve the light absorption as compared to solely TiO_2 . Solvothermal, a chemical synthesis method was implemented with a varying amount of capping agent of polyvinylpyrrolidone (PVP) from 0.2 to 2.6 g to synthesize the silver

nanoparticles. TEM images showed that the higher amount of PVP would result in smaller AgNPs size, while the largest AgNPs resulted from the lower amount of PVP. The absorbance of the AgNPs signified that the absorbance decreased with increasing PVP amount. From the results, the AgNP with 28.6 nm sizes was selected to be incorporated in TiO₂ paste for its low agglomeration and high absorbance value. The concentration of AgNPs was thereafter, varied from 1 to 3 wt.%, incorporated into TiO₂ and followed by subsequent annealing process at 450 °C to fabricate AgNP-TiO₂ film. It was observed that addition of 2 wt.% concentration of AgNPs showed the highest improvement of PCE at 4.691 % from that of TiO₂ film with PCE of 2.35 %. Therefore, coating the counter electrode with Pt solution by volume and adding AgNPs into TiO₂ methods have been carried out by which they have been proven to improve the PCE in DSSC. With this quantification, this study provides a way in addressing a durable and wearable flexible structure of portable electrical solar source in a wide range light intensity environment.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia
Sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**PENINGKATAN KECEKAPAN PENUKARAN KUASA MENGGUNAKAN
NANOPARTIKEL PERAK DAN FILEM KOMPOSIT TITANIUM DIOKSIDA
UNTUK ELEKTROD TITANIUM FLEKSIBEL BAGI SEL SURIA SOLAR
PEKA PEWARNA PENCERNA BELAKANG**

Oleh

NUR HAZAHSHA BINTI SHAMSUDIN

Jun 2022

Pengerusi: Prof. Ir. Mohd Zainal Abidin Bin Ab Kadir, PhD
Fakulti: Kejuruteraan

Kecekapan penukaran kuasa (PCE) dalam sel suria pemeka pewarna (DSSC) ialah parameter penting untuk mengukur keupayaan untuk menukar tenaga cahaya kepada elektrik. Walaubagaimanapun, pemilihan substrat menjadi antara isu yang menyumbang kepada permasalahan PCE yang rendah. Walaupun substrat polimer telah dieksploitasi sebagai fotoelektrod untuk fleksibiliti dan ketelusannya yang membolehkan pencahayaan hadapan, suhu tinggi diperlukan untuk menghapuskan pengikat polimer daripada semikonduktor titanium oksida (TiO_2) yang terdapat pada substrat tersebut. Substrat polimer tidak dapat menahan suhu tinggi, dan oleh itu menjejaskan PCE dengan sisa pengikat polimer dalam filem TiO_2 . Di sini, substrat titanium (Ti) fleksibel, bahan berasaskan logam, digunakan untuk menggantikan substrat polimer sebagai fotoelektrod. Secara semulajadi, pencahayaan belakang melalui elektrod kaunter telah dipilih dan bukannya pencahayaan hadapan melalui fotoelektrod kerana ciri substrat logam yang legap. Walau bagaimanapun, jarak yang lebih jauh untuk cahaya bergerak melalui laluan ini menghalang pembangunan fotoelektron, dan dengan itu mengurangkan kecekapan penukaran kuasa (PCE). Oleh itu, tujuan penyelidikan ini adalah untuk menambah baik PCE melalui kaedah pencahayaan belakang. Di sini, terdapat dua kaedah yang dicadangkan untuk meningkatkan PCE. Kaedah pertama ialah dengan menyalut elektrod pembilang dengan isipadu larutan platinum (Pt) yang berbeza daripada 30 μl , 50 μl , 70 μl dan 90 μl . Kaedah kedua yang dicadangkan ialah penggabungan AgNP yang disediakan melalui solvothermal ke dalam TiO_2 . Platinum disepadukan dalam elektrod kaunter untuk pengoptimuman PCE melalui aktiviti pemangkin yang disebabkan oleh Pt. Telah didedahkan bahawa 70 μL larutan Pt meningkatkan PCE daripada 1.248 % kepada 5.25 %. Penggabungan AgNP ke dalam filem TiO_2 telah terbukti

meningkatkan penyerapan cahaya berbanding dengan TiO₂ semata-mata. Solvothermal, kaedah sintesis kimia telah dilaksanakan dengan jumlah agen penutup polivinilpirolidon (PVP) yang berbeza-beza daripada 0.2 hingga 2.6 g untuk mensintesis nanozarah perak. Imej TEM menunjukkan bahawa jumlah PVP yang lebih tinggi akan menghasilkan saiz AgNP yang lebih kecil, manakala AgNP yang terbesar terhasil daripada jumlah PVP yang lebih rendah. Penyerapan AgNP menunjukkan penyerapan menurun dengan peningkatan jumlah PVP. Daripada keputusan, AgNP dengan saiz 28.6 nm telah dipilih untuk dimasukkan ke dalam TiO₂ untuk penggumpalan rendah dan nilai penyerapan yang tinggi. Kepekatan AgNPs selepas itu, diubah daripada 1 hingga 3 wt.%, dimasukkan ke dalam TiO₂ dan diikuti dengan proses pembakaran seterusnya pada 450 °C untuk menghasilkan filem AgNP-TiO₂. Diperhatikan bahawa penambahan 2 wt.% kepekatan AgNPs ke dalam TiO₂ menunjukkan peningkatan tertinggi PCE sebanyak 4.691 % daripada TiO₂ dengan PCE hanya sebanyak 2.35 %. Oleh itu, cara salutan elektrod kaunter dengan larutan Pt mengikut isipadu dan menambah AgNPs ke dalam TiO₂ telah dijalankan yang mana ia telah terbukti dapat meningkatkan PCE dalam DSSC. Dengan kuantifikasi ini, kajian ini dapat menyediakan cara dalam menangani struktur fleksibel yang tahan lama dan boleh pakai bagi sumber suria elektrik mudah alih dalam persekitaran cahaya yang lebih malap.

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This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Mohd Zainal Abidin Ab Kadir, PhD

Professor Ir.
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Suhaidi Bin Shafie, PhD

Professor Ir.
Faculty of Engineering
Universiti Putra Malaysia
(Member)

Yusran Bin Sulaiman, PhD

Associate Professor
Faculty of Science
Universiti Putra Malaysia
(Member)

Fauzan Bin Ahmad, PhD

Senior Lecturer
Malaysia-Japan International Institute of Technology
Universiti Teknologi Malaysia
(Member)

**ZALILAH MOHD SHARIFF,
PhD**

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 13 October 2022

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Signature: _____
Name of Member of
Supervisory
Committee: _____

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LIST OF ABBREVIATIONS

PE	Photoelectrodes
PCE	Power conversion efficiency
Jsc	Current density
FF	Fill factor
Voc	Open circuit voltage
AgNPs	Silver nanoparticles
TW	terawatt
CE	Counter electrode
DSSC	Dye sensitized solar cell
FTO glass	Fluorine-doped Tin Oxide soda lime glass
TCO	Transparent Conductive Oxide
ITO	Indium-doped tin oxide
PET	Polyethylene terephthalate
PEN	Polyethylene naphthalate
LUMO	lowest unoccupied molecular orbital
wt%	Weight percent
TiO ₂	Titanium dioxide
NaBH ₄	sodium borohydrate
Fe ₂ O ₃	Iron (III) oxide
AgNPs	Silver Nanoparticles
AuNPs	Gold Nanoparticles
Ti	Titanium

Al	Aluminium
Cu	Copper
AgNPs-TiO ₂	The incorporation of silver nanoparticles into titanium dioxide
ZnO	Zinc Oxide
Nb ₂ O ₅	Niobium Oxide
Zn ²⁺	Zinc ion
I ³⁻ /I ⁻	triiodide/iodide
QS	quasi solid-state
SS	solid-state
PAN	poly(acrylonitrile)
PVC	poly(vinylchloride)
PEO	poly(ethyleneoxide)
HTM	hole-transporting material
TBP	4-tert-butylpyridine
TBA	tert-Butyl alcohol
PVP	poly(vinylpyrrolidinone)
DI	deionized water
Ni-Se	nickel selenide
In	Inconel
StSt	Stainless steel
SPR	Surface plasmonic resonance
AgNO ₃	Silver Nitrate
EG	Ethylene Glycol

H ₂ O ₂	Hydrogen peroxide
TiCl ₄	Titanium tetrachloride
HNO ₃ HF	Mixture of nitric acid and hydrofluoric acid
Pt	Platinum
AgCl	Silver chloride
EDP	Electrodeposition
CI	chemical ionization
CVD	Chemical Vapor Deposition
DC	Direct Current
RF	Radio Frequency
C	Carbon
N	Nitrogen
Fe	Ferum
S/m	Siemens per meter
l	litre
mg	milligram
eV	Electron volt
TEM	Transmission electron microscopy
FESEM	Field emission scanning electron microscope
UV-Vis	Ultraviolet visible near infrared
XRD	X-ray diffraction
IPCE	Incident photon to current conversion efficiency
TGA	Thermogravimetric analysis

CHAPTER 1

INTRODUCTION

1.1 Research Background

Past research led by the energy researchers' team from Stanford University conducted a roadmap study related to the high performance of renewable energy consisting of sunlight, wind, geothermal, and hydropower in consuming just the 11.808 TW from the global energy demand. To do so, they gathered data covering all energy authorities in 139 countries with the predicted 100 % conversion up to 2050 of the aforementioned renewable energies [1]. Their roadmap provided strong shreds of evidence to support the proposition with optimum electricity consumption while casting new light over the world's worst scenarios such as global warming, air pollution, unpredictable energy expenses market, as well as burden climate and healthcare expenditures

As such, photovoltaic (PV) energy has been thrust into the limelight for years to come. Figure 1.1 gives the upshot of a global inspection commissioned by the National Renewable Energy Laboratory (NREL) to form an up-to-date view of the produced efficiencies by PV research experts across various technologies of PV in experimental scale-based over 45 years. There is a clear divide between the Silicon-based PV, thin films, and the surge of PV evolution through the cells made of organic and inorganic cells, DSSCs, and the perovskite PV-types when it comes to power conversion efficiency (PCE) performances. Among the diverse PV technologies developed, the highest PCE was emphasized by the multijunction cells, aside from crystalline Silicon PV cells. Even though the efficiency of the PV evolution or familiarly called third generation PV is the lowest PCE amongst others, the success rate of the evolved PV generation appears to be between (12 - 28) % compared to (14 - 24) % for the thin-film technologies. In the third generation PV cells, perovskite holds the PCE of 28%, whereby the PCE of DSSCs was around (10 - 13) % [2].

Amidst a well-performed plateau of PV energy, the excessive costs incurred by solar PV including manufacturing, transportation, assembling, installation, and maintenance are increasing yearly. Even worse, the complex procedures using high amounts of energy to manufacture the final product of cells, higher recombination losses due to the morphological structures of the grain boundaries which hinder the continuous flow of electrons in the cells, and the large losses of material are other baffling issues faced by Silicon cell [3], [4].

Subsequent disclosure of Flexi DSSCs was discovered by a similar group of researchers who developed the first DSSCs in 1993 [5]. Astoundingly, the finding was accidental but unveiled fresh insights into the flexible architectural role of DSSCs, thus garnering momentum on DSSCs application from lab maneuvers to commercial trade. This momentum has been grasped by third-generation-based PV companies of GreatCell Solar, GCell, and 3GSolar. DSSCs have been classified into two categories, and both categories are manufactured from Transparent Conductive Oxide (TCO) material but differ based on the coating layers. The first category known as a solid-state based DSSCs consists of transparent soda-lime glass layered with Fluorine Tin Oxide (FTO) or Indium Tin Oxide (ITO), while the second type is layered with polymer poly(ethylene terephthalate) (PET) or poly(ethylene naphthalate) (PEN) and termed as flexible based DSSCs. In some cases, the metal based substrates are also introduced in the flexible category. The solid-state is designed for extreme and harsh circumstances in wide geographical areas, whilst the criteria for flexibility are meant to be designed in any size or shape relevant to the incessant obligation of performance such as transparent, conductive, good transmittance, and low sheet resistances.

Metals are not as flexible as ITO-PET or ITO-PEN, most likely because they have standard ductility that limits their flexibility. However, metals can withstand intense heat temperatures soaring up to 500°C and are resistive to corrosion that is crucial for stability preservation purposes, in particular during electrolyte instillation. The most pervasive metals used are stainless steel (StSt) and titanium (Ti), other than copper (Cu), zinc (Zn), Inconel (In), and Tungsten (W). Although the highest PCE of flexible DSSCs was exhibited by StSt in contrast to the PCE of Ti as photoelectrodes in a previous study [6], the conclusive findings [7], [8] disclosed that Ti leads ahead StSt when it comes to their stability aptitude.

Surface plasmon resonance (SPR), acting as the optical application in DSSCs, is proven to enhance the scattering effects of light into the metal nanostructure in the cell. This energy arises from the oscillation of electron density which becomes excited when being hit by polarized UV light. The process merely occurs at the surface of the metal nanoparticles. In this case, the absorbed light will be intensified and directed in terms of its intensity. Thus, the dye as the photoactive material would increase its efficiency in absorbing the sunlight. Commonly, the engagement of LSPR is found in boundless applications such as biomedical, sensing, medical, energy, and catalysis [9]–[11]. For more definitive preferences in this research, the plasmonic nano-additive metal particles are used as doping substances in DSSCs. Noble metals including Copper(C), Gold (Au), and Silver (Ag) can affect the SPR performance through their morphological properties such as shapes, sizes, and environment of the utilized metals nanoparticles [11].

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This study is concerned with the development of back-illuminated DSSCs with flexible metal electrode using various volume of Pt solution on counter electrode. By combining the advantageous of metal substrate with AgNPs-TiO₂, the PCE of back-illuminated DSSCs could be enhanced. There is no shadow of doubt that, this research work able to contributes towards many fields such as the photodetector, telecommunication, optoelectronics and chemical based sensor areas.

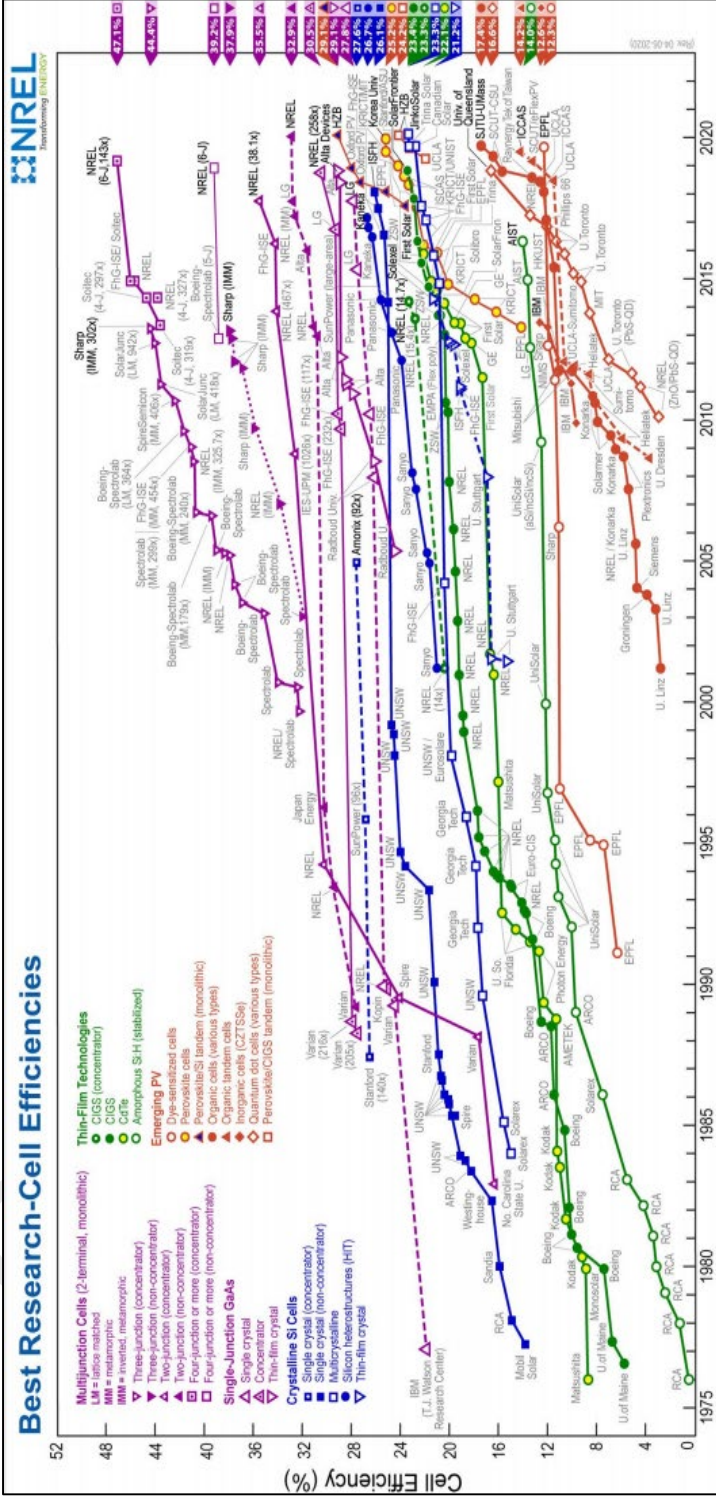


Figure 1.1: The trend of efficiency produced by energy researchers within 1975 -2020 as commissioned by the National Renewable Energy Laboratory (NREL) [2]

1.2 Problem Statement

Back illumination ensues from the solidity and opacity properties of Ti metal substrate as photoelectrodes. It occurs when light is exposed and transferred from the back of the DSSCs cell which is the counter electrode. This leads to the developed photoelectrons being stunted by lower luminous irradiation as a consequence of longer medium-light travel from the counter electrode to the TiO₂ semiconductor. To optimize the loss of back-illuminated performances, previous efforts and research were endeavored in modulating variable circumstances subjected to cathode electrodes playing a pre-eminent role during the light irradiation. This includes investigating the impact of Pt deposition on the counter electrode by manipulating different methods of Pt deposition [12], the deposition parameters [13], thermal annealed temperature [14], and Pt concentration [15]. Unlike the aforementioned past research, the current study employs the variation of Pt volume solution by using the spin coating method. The most common Pt deposition method is sputtering and chemical reduction [16]. Although sputtering provides technologically advanced privileges, especially the ability to deposit into a wide area with multiple controlled parameters, it necessitates a costly and high energy-consuming setup via a vacuum system. As chemical reduction deals with low-temperature heating and the use of sodium borohydride (NaBH₄) as a reagent, spin coating is another alternative of coating Pt, favored for its uniformity, simplicity in the perspectives of preparation and fabrication as well as rein-condition of parameters [17], [18].

A preference of silver (AgNPs) plasmonic impurities doped TiO₂ is therefore implied to improve the deteriorating performance of TiO₂ on the afflicted issues. Instead of other plasmonic metal NPs such as copper(C), aluminium (Al), and gold (Au), AgNPs are known greatly for having the highest conductivity, which allows for more charge transfer pathways into TiO₂ to the external circuit. It also offers a degree of consistency in terms of corrosivity, flammability, and chemical traits. The plasmonic AgNPs excites its light intensity which helps the dye to further elongate the wavelength spectrum potency to the visible range of wavelength [19], [20].

Through years of research, metal plasmonic NPs-TiO₂ doping studies were mostly used for FTO glass substrates. According to a previous study [21], the increase of PCE of from 5.36% to 6.99% of TiO₂ could be observed only when TiO₂ was incorporated with the AgNPs via the chemical reduction technique. The conductivity was attributed to the competent interfacial movement of charge carriers by the AgNPs. Other studies have also reported that the AgNPs enhanced the PCE to 6.06 % [22] and 4.86 % [23] after doping with TiO₂. The solvothermal process on AgNPs on the other hand initiates the electron sink effect inside the photoanode to record the PCE of 6.56% [24]. The exploitation of AgNPs was also disclosed via ion implantation [25]. The implantation of AgNPs-TiO₂ was potent with PCE of 5.59 % contributed from the LSPR light absorption shift to the visible range that facilitates the photoelectron generation.

Nevertheless, there is less disclosed research on the doping of plasmonic nanoparticles-TiO₂NPs on flexible and back-illuminated DSSCs.

This study revolves around the idea of acquiring high temperature resistance and inexpensive flexible and back-illuminated DSSCs with better PCE performance. To do so, the merit plasmonic effect of the solvothermal produced from AgNPs as an additive material is exploited into TiO₂NPs film. The beneficial impacts of AgNPs which were mostly applied in FTO glass and front illuminated, was therefore exploited in this study that used flexible and back-illuminated structures of DSSCs. Before the doping process, different volumes of Pt solution are coated on the counter electrode by spin coating for optimization purposes. Therefore, an improvement must be done to improve the light absorption in photoanode which can be fulfilled by exploiting of LSPR in AgNPs of the photoanode.

1.3 Motivation

The low-temperature sintering for plastic and polymer photoanode substrate appears to be troublesome, in flexible DSSCs. Therefore, this has allured the current study to focus on Ti metal, instead. Unfortunately, the downside of Ti metal is the opacity feature that is unable to absorb light from the front which leads to back-illuminated circumstances. Therefore, the amount of Pt coated on CE provides an impact in optimizing the light penetration and catalytic activity into the CE during the process of light irradiation for back-illuminated performances. Hence, the various volume of Pt solution was investigated towards the back-illuminated DSSCs. For front illumination DSSCs, the optimum Pt volume of Pt solution is 50 μ l. Therefore, for back illumination, the range selection is varied between 10 μ l to 100 μ l. Meanwhile, the synthesis of AgNPs via physical route commonly involves high energy and cost, despite the assurance of high purity in material production. In order to provide a simple and size-controlled product without a high degree of thermal requisites procedures, the preparation of customized AgNPs through the solvothermal method was executed.

1.4 Aim and Objectives

The research aims to improve the power conversion efficiency (PCE) performances of back-illuminated DSSC. This can be achieved by few objectives as highlighted below:

- 1) To investigate the effect of Platinum catalyst deposition by spin coating under 30 μ l to 90 μ l volumes of Pt solution towards the back-illuminated DSSC to improve light penetration of the counter electrode.

2) To synthesize the Silver nanoparticles (AgNPs) between 20 nm to 40 nm using the solvothermal method to be utilized in DSSC photoanode for improvement of light absorption.

3) To investigate on current-voltage (I-V) characteristics of back-illuminated DSSCs by embedding the synthesized AgNPs into TiO₂NPs based on a variation of concentration.

1.5 Scope of Research

This research work focuses on the development of flexible DSSCs, which was limited to the following circumstances:

- A Titanium foil with a thickness of 0.2 mm was tested as photoelectrode with the optimized film thickness at two layers Scotch tape thicknesses.
- The counter electrode (CE) is FTO coated glass substrate with a catalytic layer of Pt. This is due to the limitation of equipment for Pt deposition on ITO-PET substrates. The ranges of Pt volume solution at 30 μ l, 50 μ l, 70 μ l and 90 μ l were measured by micropipette under spin coating method.
- Method used to obtain Silver Nanoparticles (AgNPs) was the Solvothermal method, which was then embedded with TiO₂ nanoparticles under different concentrations. This research only conducted on the preliminary implementation of a composite AgNPs-TiO₂ film on the Ti foil photoelectrodes based DSSCs in performing the PCE performances. It also does not scrutinize the analysis and effect in regards to oxidation as well as corrosion of AgNPs resulting from the injected non-aqueous iodide/triiodide electrolyte.
- The experimental works are conducted Functional Nanotechnology Devices Laboratory (FNDL) and Nanomaterials Processing and Technology Laboratory (NPTL), Institute of Nanoscience and Nanotechnology (ION2), UPM Serdang.

1.6 Thesis Outline

The thesis is structured into five (5) chapters, which outlines as follows as:

Chapter 1 introduces the background of this research, the declaration of the research problem, and its objectives. The limitation of research activities is also asserted.

Chapter 2 presents the inclusive fundamental of Dye-Sensitized Solar Cells (DSSCs) and defines the Power Conversion Efficiency (PCE) Performances which includes Current Density (J_{sc}), Open Circuit Voltage (V_{oc}), Fill Factor

(FF). The flexible DSSCs in particular of polymer, plastic and metal-based anodes are also reviewed. For metal-based anodes, Titanium is highlighted as the main preference. The review also comprises the surface modification of TiO_2 via TiO_2 nanostructured and Localized Surface Plasmon Resonance (LSPR) that exerted metal nanoparticles, specifically AgNPs as a doping substance.

Chapter 3 provides the chemicals and methods utilized to synthesize the AgNPs and the methodology of cell construction. The characterizations in verifying the optimized studies are also described.

Chapter 4 discusses the PCE and analysis of characterization from the experimental results.

Chapter 5 concludes the final finding of the research as well as suggestions for further extending of the research.

REFERENCES

- [1] C. Q. Choi, "A Road Map to 100 Percent Renewable Energy in 139 Countries by 2050," *IEEE Spectrum*, 2017. <https://spectrum.ieee.org/100-percent-renewable-energy-for-139-countries-by-2050>.
- [2] National Renewable Energy Laboratory, "Best Research-Cell Efficiency Chart," 2020. <https://www.nrel.gov/pv/cell-efficiency.html>.
- [3] A. L. Fahrenbruch, R. H. Bube, and R. V. D'Aiello, "Fundamentals of Solar Cells (Photovoltaic Solar Energy Conversion)," *J. Sol. Energy Eng.*, vol. 106, no. 4, p. 497, 2010, doi: 10.1115/1.3267632.
- [4] T. Soga, "Chapter 1 - Fundamentals of Solar Cell," in *Nanostructured Materials for Solar Energy Conversion*, T. Soga, Ed. Amsterdam: Elsevier, 2006, pp. 3–43.
- [5] N. Vlachopoulos, P. Liska, J. Augustynski, and M. Gratzel, "Very Efficient Visible Light Energy Harvesting and Conversion by Spectral Sensitization of High Surface Area Polycrystalline Titanium Dioxide Films," pp. 1216–1220, 1988.
- [6] J. H. Park, Y. Jun, H.-G. Yun, S.-Y. Lee, and M. G. Kang, "Fabrication of an Efficient Dye-Sensitized Solar Cell with Stainless Steel Substrate," *J. Electrochem. Soc.*, vol. 155, no. 7, p. F145, 2008, doi: 10.1149/1.2909548.
- [7] K. Onoda, S. Ngamsinlapasathian, T. Fujieda, and S. Y. Å, "The superiority of Ti plate as the substrate of dye-sensitized solar cells," vol. 91, pp. 1176–1181, 2007, doi: 10.1016/j.solmat.2006.12.017.
- [8] K. Miettunen *et al.*, "Stability of Dye Solar Cells with Photoelectrode on Metal Substrates," *J. Electrochem. Soc.*, vol. 157, no. 6, p. B814, 2010, doi: 10.1149/1.3374645.
- [9] R. Alharbi, M. Irannejad, and M. Yavuz, "A short review on the role of the metal-graphene hybrid nanostructure in promoting the localized surface plasmon resonance sensor performance," *Sensors (Switzerland)*, vol. 19, no. 4, 2019, doi: 10.3390/s19040862.
- [10] M. Sui, S. Kunwar, P. Pandey, and J. Lee, "Strongly confined localized surface plasmon resonance (LSPR) bands of Pt, AgPt, AgAuPt nanoparticles," *Sci. Rep.*, vol. 9, no. 1, p. 16582, 2019, doi: 10.1038/s41598-019-53292-1.

- [11] K. K. Paul and P. K. Giri, *Plasmonic metal and semiconductor nanoparticle decorated TiO₂-based photocatalysts for solar light driven photocatalysis*. Elsevier, 2018.
- [12] S. Cells, "Investigation of Electrochemically Deposited and Chemically Reduced Platinum Nanostructured Thin Films as Counter Electrodes in Dye-Sensitized," 2018, doi: 10.3390/coatings8020056.
- [13] T. Ponken, K. Tagsin, C. Suwannakhun, J. Luecha, and W. Choawunklang, "Preparation of Platinum (Pt) Counter Electrode Coated by Electrochemical Technique at High Temperature for Dye-sensitized Solar Cell (DSSC) Application," *J. Phys. Conf. Ser.*, vol. 901, p. 12084, Sep. 2017, doi: 10.1088/1742-6596/901/1/012084.
- [14] N. T. Q. Hoa, V. D. Dao, and H. S. Choi, "Fabrication of platinum nanoparticle counter electrode for highly efficient dye-sensitized solar cells by controlled thermal reduction time," *J. Mater. Sci.*, vol. 49, no. 14, pp. 4973–4978, 2014, doi: 10.1007/s10853-014-8199-y.
- [15] L. Chen, W. Tan, J. Zhang, X. Zhou, X. Zhang, and Y. Lin, "Fabrication of high performance Pt counter electrodes on conductive plastic substrate for flexible dye-sensitized solar cells," *Electrochim. Acta*, vol. 55, no. 11, pp. 3721–3726, 2010, doi: 10.1016/j.electacta.2010.01.108.
- [16] M. Wu and T. Ma, "Recent Progress of Counter Electrode Catalysts in Dye-Sensitized Solar Cells," *Phys. Chem.*, vol. 118, no. 30, pp. 16727–16742, 2014, doi: 10.1021/jp412713h.
- [17] F. Mandoj, S. Nardis, C. Di Natale, and R. Paolesse, "Porphyrinoid Thin Films for Chemical Sensing," in *Porphyrinoid thin films for chemical sensing*, 2018, pp. 422–43, doi: <https://doi.org/10.1016/B978-0-12-409547-2.11677-4>.
- [18] S. Ahmadi *et al.*, "The Role of Physical Techniques on the Preparation of Photoanodes for Dye Sensitized Solar Cells," vol. 2014, 2014.
- [19] A. K. Gupta, P. Srivastava, and L. Bahadur, "Improved performance of Ag-doped TiO₂ synthesized by modified sol–gel method as photoanode of dye-sensitized solar cell," *Appl. Phys. A Mater. Sci. Process.*, vol. 122, no. 8, 2016, doi: 10.1007/s00339-016-0241-2.
- [20] A. Lalis, G. Tessier, J. Plain, and G. Baffou, "Quantifying the Efficiency of Plasmonic Materials for Near-Field Enhancement and Photothermal Conversion," *J. Phys. Chem. C*, vol. 119, no. 45, pp. 25518–25528, 2015, doi: 10.1021/acs.jpcc.5b09294.

- [21] D. N. Joshi, P. Ilaiyaraja, C. Sudakar, and R. A. Prasath, "Facile one-pot synthesis of multi-shaped silver nanoparticles with tunable ultra-broadband absorption for efficient light harvesting in dye-sensitized solar cells," *Sol. Energy Mater. Sol. Cells*, vol. 185, no. March, pp. 104–110, 2018, doi: 10.1016/j.solmat.2018.05.018.
- [22] H. Chang, C. H. Chen, M. J. Kao, and H. H. Hsiao, "Effect of core-shell Ag@TiO₂ volume ratio on characteristics of TiO₂-based DSSCs," *J. Nanomater.*, vol. 2014, 2014, doi: 10.1155/2014/264108.
- [23] D. N. Joshi, S. Mandal, R. Kothandraman, and R. A. Prasath, "Efficient light harvesting in dye sensitized solar cells using broadband surface plasmon resonance of silver nanoparticles with varied shapes and sizes," *Mater. Lett.*, vol. 193, pp. 288–291, 2017, doi: 10.1016/j.matlet.2017.02.008.
- [24] S. Buda, S. Shafie, S. A. Rashid, H. Jaafar, and N. F. M. Sharif, "Enhanced visible light absorption and reduced charge recombination in AgNP plasmonic photoelectrochemical cell," *Results Phys.*, vol. 7, pp. 2311–2316, 2017, doi: 10.1016/j.rinp.2017.07.009.
- [25] N. Kaur *et al.*, "Ag ion implanted TiO₂ photoanodes for fabrication of highly efficient and economical plasmonic dye sensitized solar cells," *Chem. Phys. Lett.*, vol. 740, p. 137070, 2020, doi: <https://doi.org/10.1016/j.cplett.2019.137070>.
- [26] L. Escoubas, J. J. Simon, J. Le Rouzo, and V. Bermudez, "16 - Innovative approaches in thin film photovoltaic cells," in *Optical Thin Films and Coatings*, A. Piegari and F. Flory, Eds. Woodhead Publishing, 2013, pp. 596–630.
- [27] S. Bose, V. Soni, and K. R. Genwa, "Recent Advances and Future Prospects for Dye Sensitized Solar Cells : A Review," vol. 5, no. 4, pp. 1–9, 2015.
- [28] G. M. O'Regan B, "A low-cost, high-efficiency solar cell based on dye-sensitized colloidal TiO₂ films," *Nature*, vol. 353, no. 6346, pp. 737–740, 1991, doi: <https://doi.org/10.1038/353737a0>.
- [29] D. Devadiga, M. Selvakumar, P. Shetty, and M. S. Santosh, "Dye-Sensitized Solar Cell for Indoor Applications: A Mini-Review," *J. Electron. Mater.*, vol. 50, no. 6, pp. 3187–3206, 2021, doi: 10.1007/s11664-021-08854-3.
- [30] M. Ye *et al.*, "Recent advances in dye-sensitized solar cells: From photoanodes, sensitizers and electrolytes to counter electrodes," *Mater. Today*, vol. 18, no. 3, pp. 155–162, 2015, doi:

10.1016/j.mattod.2014.09.001.

- [31] H. Chen, N. Li, Y. H. Wu, J. Bin Shi, B. X. Lei, and Z. F. Sun, "A novel cheap, one-step and facile synthesis of hierarchical TiO₂ nanotubes as fast electron transport channels for highly efficient dye-sensitized solar cells," *Adv. Powder Technol.*, vol. 31, no. 4, pp. 1556–1563, 2020, doi: 10.1016/j.apt.2020.01.020.
- [32] F. Babar *et al.*, "Nanostructured photoanode materials and their deposition methods for efficient and economical third generation dye-sensitized solar cells: A comprehensive review," *Renew. Sustain. Energy Rev.*, vol. 129, no. May, p. 109919, 2020, doi: 10.1016/j.rser.2020.109919.
- [33] C. Sima, C. Grigoriu, and S. Antohe, "Comparison of the dye-sensitized solar cells performances based on transparent conductive ITO and FTO," *Thin Solid Films*, vol. 519, no. 2, pp. 595–597, 2010, doi: 10.1016/j.tsf.2010.07.002.
- [34] S. Geiger, O. Kasian, A. M. Mingers, K. J. J. Mayrhofer, and S. Cherevko, "Stability limits of tin-based electrocatalyst supports," *Sci. Rep.*, vol. 7, no. 1, pp. 3–9, 2017, doi: 10.1038/s41598-017-04079-9.
- [35] M. A. Aouaj, R. Diaz, A. Belayachi, F. Rueda, and M. Abd-lefdil, "Comparative study of ITO and FTO thin films grown by spray pyrolysis," vol. 44, pp. 1458–1461, 2009, doi: 10.1016/j.materresbull.2009.02.019.
- [36] D.-P. Tran, H.-I. Lu, and C.-K. Lin, "Conductive Characteristics of Indium Tin Oxide Thin Film on Polymeric Substrate under Long-Term Static Deformation," *Coatings*, vol. 8, no. 6, 2018, doi: 10.3390/coatings8060212.
- [37] T. Yamaguchi, N. Tobe, D. Matsumoto, and H. Arakawa, "Highly efficient plastic substrate dye-sensitized solar cells using a compression method for preparation of TiO₂ photoelectrodes," *Chem. Commun.*, no. 45, p. 4767, 2007, doi: 10.1039/b709911h.
- [38] M. Dürr, A. Schmid, M. Obermaier, S. Rosselli, A. Yasuda, and G. Nelles, "Low-temperature fabrication of dye-sensitized solar cells by transfer of composite porous layers," *Nat. Mater.*, vol. 4, no. 8, pp. 607–611, 2005, doi: 10.1038/nmat1433.
- [39] M. Shamimul, H. Choudhury, N. Kishi, and T. Soga, "Compression of ZnO nanoparticle films at elevated temperature for flexible dye-sensitized solar cells," *J. Alloys Compd.*, vol. 656, pp. 476–480, 2016, doi: 10.1016/j.jallcom.2015.09.138.

- [40] S. Zen, Y. Inoue, and R. Ono, "Low temperature (150 C) fabrication of high-performance TiO₂ films for dye-sensitized solar cells using ultraviolet light and plasma treatments of TiO₂ paste containing organic binder," *J. Appl. Phys.*, vol. 117, no. 10, 2015, doi: 10.1063/1.4914873.
- [41] J. Liu, Y. Li, S. Arumugam, J. Tudor, and S. Beeby, "Investigation of Low Temperature Processed Titanium Dioxide (TiO₂) Films for Printed Dye Sensitized Solar Cells (DSSCs) for Large Area Flexible Applications," *Mater. Today Proc.*, vol. 5, no. 5, Part 3, pp. 13846–13854, 2018, doi: <https://doi.org/10.1016/j.matpr.2018.02.026>.
- [42] C. Wu, B. Chen, X. Zheng, and S. Priya, "Scaling of the flexible dye sensitized solar cell module," *Sol. Energy Mater. Sol. Cells*, vol. 157, pp. 438–446, 2016, doi: 10.1016/j.solmat.2016.07.021.
- [43] Y. B. C. H.C.Weerasinghe, P.M. Sirimanne, G.P. Simon, "Fabrication of efficient solar cells on plastic substrates using binder-free ball milled titania slurries," *J. Photochem. Photobiol. A Chem.*, vol. 206, pp. 64–70, 2009, doi: 10.1016/j.jphotochem.2009.05.013.
- [44] H. Horiuchi *et al.*, "Electron injection efficiency from excited N₃ into nanocrystalline ZnO films: Effect of (N₃-Zn²⁺) aggregate formation," *J. Phys. Chem. B*, vol. 107, no. 11, pp. 2570–2574, 2003, doi: 10.1021/jp0220027.
- [45] K. Keis, J. Lindgren, S. E. Lindquist, and A. Hagfeldt, "Studies of the adsorption process of Ru complexes in nanoporous ZnO electrodes," *Langmuir*, vol. 16, no. 10, pp. 4688–4694, 2000, doi: 10.1021/la9912702.
- [46] K. Sayama, H. Sugihara, and H. Arakawa, "Photoelectrochemical properties of a porous Nb₂O₅ electrode sensitized by a ruthenium dye," *Chem. Mater.*, vol. 10, no. 12, pp. 3825–3832, 1998, doi: 10.1021/cm980111l.
- [47] X. Liu, R. Yuan, Y. Liu, S. Zhu, J. Lin, and X. Chen, "Niobium pentoxide nanotube powder for efficient dye-sensitized solar cells," *New J. Chem.*, vol. 40, no. 7, pp. 6276–6280, 2016, doi: 10.1039/c6nj00159a.
- [48] S. N. F. Zainudin, H. Abdullah, and M. Markom, "Electrochemical studies of tin oxide based-dye-sensitized solar cells (DSSC): a review," *J. Mater. Sci. Mater. Electron.*, vol. 30, no. 6, pp. 5342–5356, 2019, doi: 10.1007/s10854-019-00929-6.
- [49] K. Kakiage, Y. Aoyama, T. Yano, K. Oya, J. Fujisawa, and M. Hanaya, "Highly-efficient dye-sensitized solar cells with collaborative sensitization by silyl-anchor and carboxy-anchor dyes," *Chem. Commun.*, vol. 51, no. 88, pp. 15894–15897, 2015, doi: 10.1039/C5CC06759F.

- [50] S. Mathew *et al.*, "Dye-sensitized solar cells with 13% efficiency achieved through the molecular engineering of porphyrin sensitizers," *Nat. Chem.*, vol. 6, no. 3, 2014, doi: 10.1038/nchem.1861.
- [51] F. Zhu *et al.*, "Amorphous silicon carbide photoelectrode for hydrogen production directly from water using sunlight," *Philos. Mag.*, vol. 89, no. 28–30, pp. 2723–2739, 2009, doi: 10.1080/14786430902740729.
- [52] S. Datta, A. Dey, N. R. Singha, and S. Roy, "Enhanced performance of dye-sensitized solar cell with thermally stable natural dye-assisted TiO₂/MnO₂ bilayer-assembled photoanode," *Mater. Renew. Sustain. Energy*, vol. 9, no. 4, 2020, doi: 10.1007/s40243-020-00185-3.
- [53] G. Richhariya, A. Kumar, P. Tekasakul, and B. Gupta, "Natural dyes for dye sensitized solar cell: A review," *Renew. Sustain. Energy Rev.*, vol. 69, no. April 2015, pp. 705–718, 2017, doi: 10.1016/j.rser.2016.11.198.
- [54] E. K. Kalyanasundaram *et al.*, *DYE-SENSITIZED SOLAR CELLS*, First. Lausanne, Switzerland: EPFL Press, 2010.
- [55] B. O'Regan and M. Grätzel, "A low-cost, high-efficiency solar cell based on dye-sensitized colloidal TiO₂ films," *Nature*, vol. 353, no. 6346, pp. 737–740, 1991, doi: 10.1038/353737a0.
- [56] M. K. Nazeeruddin *et al.*, "Conversion of light to electricity by cis-X₂bis(2,2'-bipyridyl-4,4'-dicarboxylate)ruthenium(II) charge-transfer sensitizers (X = Cl-, Br-, I-, CN-, and SCN-) on nanocrystalline titanium dioxide electrodes," *J. Am. Chem. Soc.*, vol. 115, no. 14, pp. 6382–6390, 1993, doi: 10.1021/ja00067a063.
- [57] M. K. Nazeeruddin, P. Péchy, and M. Grätzel, "Efficient panchromatic sensitization of nanocrystalline TiO₂ films by a black dye based on a trithiocyanato-ruthenium complex," *Chem. Commun.*, vol. 1, no. 18, pp. 1705–1706, 1997, doi: 10.1039/a703277c.
- [58] N. C. D. Nath, J. C. Kim, K. P. Kim, S. Yim, and J. J. Lee, "Deprotonation of N₃ adsorbed on TiO₂ for high-performance dye-sensitized solar cells (DSSCs)," *J. Mater. Chem. A*, vol. 1, no. 43, pp. 13439–13442, 2013, doi: 10.1039/c3ta12298k.
- [59] *et al.* Nazeeruddin, M. K., De Angelis, F., "S1. Experimental section: SE 1. Materials: All the solvents and the chemicals (puriss grade) were obtained from Fluka," *J. Am. Chem. Soc.*, vol. 2, no. 127, pp. 1–11, 2005.
- [60] K. Sharma, V. Sharma, and S. S. Sharma, "Dye-Sensitized Solar Cells: Fundamentals and Current Status," *Nanoscale Res. Lett.*, vol. 13, no. 1, p. 381, 2018, doi: 10.1186/s11671-018-2760-6.

- [61] L.-C. Tseng, M. Kuo, and R.-H. Lee, "An imidazolium iodide-containing hyperbranched polymer ionic liquid that improves the performance of dye-sensitized solar cells," *J. Polym. Res.*, vol. 23, 2016, doi: 10.1007/s10965-016-1054-x.
- [62] P. T. Nguyen, P. E. Hansen, and T. Lund, "The effect of 4-tert-butylpyridine and Li⁺ on the thermal degradation of TiO₂-bound ruthenium dye N719," *Sol. Energy*, vol. 88, pp. 23–30, 2013, doi: 10.1016/j.solener.2012.11.012.
- [63] F.-T. Kong, S.-Y. Dai, and K.-J. Wang, "Review of Recent Progress in Dye-Sensitized Solar Cells," *Adv. Optoelectron.*, vol. 2007, pp. 1–13, 2007, doi: 10.1155/2007/75384.
- [64] S. Chandra, "Recent trends in high efficiency photo-electrochemical solar cell using dye-sensitized photo-electrodes and ionic liquid based redox electrolytes," *Proc. Natl. Acad. Sci. India Sect. A - Phys. Sci.*, vol. 82, no. 1, pp. 5–19, 2012, doi: 10.1007/s40010-012-0001-4.
- [65] H. Iftikhar, G. G. Sonai, S. G. Hashmi, A. F. Nogueira, and P. D. Lund, *Progress on electrolytes development in dye-sensitized solar cells*, vol. 12, no. 12. 2019.
- [66] J. J. Nelson, T. J. Amick, and C. M. Elliott, "Mass Transport of Polypyridyl Cobalt Complexes in Dye-Sensitized Solar Cells with Mesoporous TiO₂ Photoanodes," *J. Phys. Chem. C*, vol. 112, no. 46, pp. 18255–18263, 2008, doi: 10.1021/jp806479k.
- [67] J. Wu *et al.*, "Electrolytes in Dye-Sensitized Solar Cells," *Chem. Rev.*, vol. 115, no. 5, pp. 2136–2173, Mar. 2015, doi: 10.1021/cr400675m.
- [68] M. Ye *et al.*, "Recent advances in dye-sensitized solar cells: From photoanodes, sensitizers and electrolytes to counter electrodes," *Mater. Today*, vol. 18, no. 3, pp. 155–162, 2015, doi: 10.1016/j.mattod.2014.09.001.
- [69] M. Suzuka *et al.*, "A Quasi-Solid State DSSC with 10.1% Efficiency through Molecular Design of the Charge-Separation and -Transport," *Sci. Rep.*, vol. 6, no. March, pp. 1–7, 2016, doi: 10.1038/srep28022.
- [70] M. Gerosa *et al.*, "Toward Totally Flexible Dye-Sensitized Solar Cells Based on Titanium Grids and Polymeric Electrolyte," *IEEE J. Photovoltaics*, vol. 6, no. 2, pp. 498–505, Mar. 2016, doi: 10.1109/JPHOTOV.2016.2514702.
- [71] M. K. Nazeeruddin *et al.*, "Combined Experimental and DFT-TDDFT Computational Study of Photoelectrochemical Cell Ruthenium

Sensitizers," *J. Am. Chem. Soc.*, vol. 127, no. 48, pp. 16835–16847, Dec. 2005, doi: 10.1021/ja052467l.

- [72] U. Ahmed, M. Alizadeh, N. A. Rahim, S. Shahabuddin, M. S. Ahmed, and A. K. Pandey, "A comprehensive review on counter electrodes for dye sensitized solar cells: A special focus on Pt-TCO free counter electrodes," *Sol. Energy*, vol. 174, no. October, pp. 1097–1125, 2018, doi: 10.1016/j.solener.2018.10.010.
- [73] N. A. Karim, U. Mehmood, H. F. Zahid, and T. Asif, "Nanostructured photoanode and counter electrode materials for efficient Dye-Sensitized Solar Cells (DSSCs)," *Sol. Energy*, vol. 185, no. January, pp. 165–188, 2019, doi: 10.1016/j.solener.2019.04.057.
- [74] A. Kay and M. Grätzel, "Low cost photovoltaic modules based on dye sensitized nanocrystalline titanium dioxide and carbon powder," *Sol. Energy Mater. Sol. Cells*, vol. 44, no. 1, pp. 99–117, 1996, doi: 10.1016/0927-0248(96)00063-3.
- [75] M. R. Samantaray *et al.*, *Synergetic effects of hybrid carbon nanostructured counter electrodes for dye-sensitized solar cells: A review*, vol. 13, no. 12. 2020.
- [76] Y. Shibata *et al.*, "Quasi-solid dye sensitised solar cells filled with ionic liquid - Increase in efficiencies by specific interaction between conductive polymers and gelators," *Chem. Commun.*, no. 21, pp. 2730–2731, 2003, doi: 10.1039/B305368G.
- [77] A. Kancierzewska, E. Dobruchowska, A. Baranzahi, E. Carlegrim, M. Fahlman, and M. A. Gîrțu, "Study on poly(3,4-ethylene dioxythiophene)-poly(styrenesulfonate) as a plastic counter electrode in dye sensitized solar cells," *J. Optoelectron. Adv. Mater.*, vol. 9, no. 4, pp. 1052–1059, 2007.
- [78] H. Lindström, A. Holmberg, E. Magnusson, L. Malmqvist, and A. Hagfeldt, "A new method to make dye-sensitized nanocrystalline solar cells at room temperature," *J. Photochem. Photobiol. A Chem.*, vol. 145, no. 1–2, pp. 107–112, 2001, doi: 10.1016/S1010-6030(01)00564-0.
- [79] A. Holmberg, E. Magnusson, S. Lindquist, L. Malmqvist, and A. Hagfeldt, "A New Method for Manufacturing.pdf," 2001.
- [80] K. Ezaka, T. Yamamura, T. Yasufuku, N. Kishi, and T. Soga, "Low-temperature fabrication of dye-sensitized solar cells on plastic films by hot-pressing method," *Chem. Lett.*, vol. 42, no. 10, pp. 1263–1264, 2013, doi: 10.1246/cl.130582.

- [81] J. Shao *et al.*, "Low temperature preparation of TiO₂ films by cold isostatic pressing for flexible dye-sensitized solar cells," *Mater. Lett.*, vol. 68, pp. 493–496, 2012, doi: 10.1016/j.matlet.2011.11.040.
- [82] M. S. Haque Choudhury, N. Kishi, and T. Soga, "Hot-compress: A new postdeposition treatment for ZnO-based flexible dye-sensitized solar cells," *Mater. Res. Bull.*, vol. 80, pp. 135–138, 2016, doi: 10.1016/j.materresbull.2016.03.037.
- [83] B. C. Kocaoglu, K. C. Icli, and M. Ozenbas, "Optimization of Selective Electrophoretic Deposition and Isostatic Compression of Titania Nanoparticles for Flexible Dye-Sensitized Solar Cells," *Electrochim. Acta*, vol. 196, pp. 535–546, 2016, doi: 10.1016/j.electacta.2016.02.198.
- [84] B. S. Han, S. Caliskan, W. Sohn, M. Kim, J. K. Lee, and H. W. Jang, "Room Temperature Deposition of Crystalline Nanoporous ZnO Nanostructures for Direct Use as Flexible DSSC Photoanode," *Nanoscale Res. Lett.*, vol. 11, no. 1, 2016, doi: 10.1186/s11671-016-1437-2.
- [85] Y. Yang *et al.*, "Enhanced Photoelectrochemical Performances in Flexible Mesoscopic Solar Cells: An Effective Light-Scattering Material," *ChemPhotoChem*, vol. 2, no. 11, pp. 986–993, 2018, doi: 10.1002/cptc.201800089.
- [86] H. W. Chen *et al.*, "Plastic dye-sensitized photo-supercapacitor using electrophoretic deposition and compression methods," *J. Power Sources*, vol. 195, no. 18, pp. 6225–6231, 2010, doi: 10.1016/j.jpowsour.2010.01.009.
- [87] G. Boschloo, H. Lindström, E. Magnusson, A. Holmberg, and A. Hagfeldt, "Optimization of dye-sensitized solar cells prepared by compression method," vol. 148, pp. 11–15, 2002.
- [88] K. G. Baiju, B. Murali, R. Subba Rao, K. Jayanarayanan, and D. Kumaresan, "Heat sink assisted elevated temperature sintering process of TiO₂ on polymer substrates for producing high performance flexible dye-sensitized solar cells," *Chem. Eng. Process. - Process Intensif.*, vol. 149, p. 107817, 2020, doi: <https://doi.org/10.1016/j.cep.2020.107817>.
- [89] P. J. Holliman, A. Connell, M. Davies, M. Carnie, D. Bryant, and E. W. Jones, "Low temperature sintering of aqueous TiO₂ colloids for flexible, co-sensitized dye-sensitized solar cells," *Mater. Lett.*, vol. 236, pp. 289–291, 2019, doi: 10.1016/j.matlet.2018.10.118.
- [90] J. Liu, Y. Li, S. Arumugam, J. Tudor, and S. Beeby, "Investigation of Low Temperature Processed Titanium Dioxide (TiO₂) Films for Printed Dye

Sensitized Solar Cells (DSSCs) for Large Area Flexible Applications,” vol. 5, pp. 13846–13854, 2018, doi: 10.1016/j.matpr.2018.02.026.

- [91] A. Kunzmann *et al.*, “Hybrid Dye-Titania Nanoparticles for Superior Low-Temperature Dye-Sensitized Solar Cells,” vol. 1702583, pp. 1–12, 2018, doi: 10.1002/aenm.201702583.
- [92] M. M. Maitani *et al.*, “Low-temperature annealing of mesoscopic TiO₂ films by interfacial microwave heating applied to efficiency improvement of dye-sensitized solar cells,” *Sol. Energy Mater. Sol. Cells*, vol. 147, 2016, doi: 10.1016/j.solmat.2015.12.020.
- [93] T. O. S. Zen, D. Saito, R. Ono, “Low-temperature sintered dye-sensitized solar cell using surface treatment of TiO₂ photoelectrode with ultraviolet light,” *Chem. Lett.*, vol. 42, no. 6, pp. 624–626, 2013, doi: 10.1246/cl.130147.
- [94] S. Zen, Y. Ishibashi, and R. Ono, “Low-temperature sintering for plastic dye-sensitized solar cells using conventional TiO₂ paste containing organic binders Low-temperature sintering for plastic dye-sensitized solar cells using conventional TiO₂ paste containing organic binders,” vol. 213904, pp. 1–5, 2014, doi: 10.1063/1.4880117.
- [95] S. Ito *et al.*, “High-efficiency (7.2%) flexible dye-sensitized solar cells with Ti-metal substrate for nanocrystalline-TiO₂ photoanode,” *Chem. Commun.*, no. 38, pp. 4004–4006, 2006, doi: 10.1039/b608279c.
- [96] K. Onoda, S. Ngamsinlapasathian, Y. Susumu, and A. Fujieda, “Application of Metallic Titanium as the Substrate for Photoelectrode of Dye-sensitized Solar Cells,” *2nd Int. Conf. Sustainable Energy Environ. (SEE 2006)* in 21-23 Novemb. 2006, Bangkok, Thai., vol. 025, no. November, pp. 21–26, 2006.
- [97] K. Onoda, S. Ngamsinlapasathian, T. Fujieda, and S. Yoshikawa, “The superiority of Ti plate as the substrate of dye-sensitized solar cells,” *Sol. Energy Mater. Sol. Cells*, vol. 91, no. 13, pp. 1176–1181, 2007, doi: 10.1016/j.solmat.2006.12.017.
- [98] K. Fan, T. Peng, B. Chai, J. Chen, and K. Dai, “Fabrication and photoelectrochemical properties of TiO₂ films on Ti substrate for flexible dye-sensitized solar cells,” *Electrochim. Acta*, vol. 55, no. 18, pp. 5239–5244, 2010, doi: 10.1016/j.electacta.2010.04.051.
- [99] N. Fu, X. Xiao, X. Zhou, J. Zhang, and Y. Lin, “Electrodeposition of Platinum on Plastic Substrates as Counter Electrodes for Flexible Dye-Sensitized Solar Cells,” 2012.

- [100] K. Lee, L. Lin, C. Chen, V. Suryanarayanan, and C. Wu, "Preparation of High Transmittance Platinum Counter Electrode at an Ambient Temperature for Flexible Dye-Sensitized Solar Cells," *Electrochim. Acta*, vol. 135, pp. 578–584, 2014.
- [101] A. Hauch and A. Georg, "Diffusion in the electrolyte and charge-transfer reaction at the platinum electrode in dye-sensitized solar cells," vol. 46, pp. 3457–3466, 2001.
- [102] X. Fang, T. Ma, G. Guan, M. Akiyama, and E. Abe, "Performances characteristics of dye-sensitized solar cells based on counter electrodes with Pt films of different thickness," vol. 164, pp. 179–182, 2004, doi: 10.1016/j.jphotochem.2003.12.024.
- [103] X. Fang, T. Ma, G. Guan, M. Akiyama, T. Kida, and E. Abe, "Effect of the thickness of the Pt film coated on a counter electrode on the performance of a dye-sensitized solar cell," *J. Electroanal. Chem.*, vol. 570, no. 2, pp. 257–263, 2004, doi: 10.1016/j.jelechem.2004.04.004.
- [104] C. P. Cho, H. Y. Wu, and C. C. Lin, "Impacts of sputter-deposited platinum thickness on the performance of dye-sensitized solar cells," *Electrochim. Acta*, vol. 107, pp. 488–493, 2013, doi: 10.1016/j.electacta.2013.06.023.
- [105] L. Y. Lin, C. P. Lee, R. vittal, and K. C. Ho, "Improving the durability of dye-sensitized solar cells through back illumination," *J. Power Sources*, vol. 196, no. 3, pp. 1671–1676, 2011, doi: 10.1016/j.jpowsour.2010.08.032.
- [106] Y. L. Lee, C. L. Chen, L. W. Chong, C. H. Chen, Y. F. Liu, and C. F. Chi, "A platinum counter electrode with high electrochemical activity and high transparency for dye-sensitized solar cells," *Electrochem. commun.*, vol. 12, no. 11, pp. 1662–1665, 2010, doi: 10.1016/j.elecom.2010.09.022.
- [107] X. Yin, Z. Xue, and B. Liu, "Electrophoretic deposition of Pt nanoparticles on plastic substrates as counter electrode for flexible dye-sensitized solar cells," *J. Power Sources*, vol. 196, no. 4, pp. 2422–2426, 2011, doi: 10.1016/j.jpowsour.2010.09.047.
- [108] K. Lee, L. Lin, V. Suryanarayanan, and C. Wu, "Titanium dioxide coated on titanium / stainless steel foil as photoanode for high efficiency flexible dye-sensitized solar cells," *J. Power Sources*, vol. 269, pp. 789–794, 2014, doi: 10.1016/j.jpowsour.2014.07.013.
- [109] C. H. Lee, P. T. Hsiao, M. De Lu, and J. M. Wu, "Light harvesting enhancement for Ti-based dye-sensitized solar cells by introducing a grooved texture underlayer," *RSC Adv.*, vol. 3, no. 7, pp. 2216–2218, 2013, doi: 10.1039/c2ra22463a.

- [110] W. Tan, X. Yin, X. Zhou, J. Zhang, X. Xiao, and Y. Lin, "Electrophoretic deposition of nanocrystalline TiO₂ films on Ti substrates for use in flexible dye-sensitized solar cells," *Electrochim. Acta*, vol. 54, no. 19, pp. 4467–4472, 2009, doi: 10.1016/j.electacta.2009.03.037.
- [111] Y. Jun and M. G. Kang, "The Characterization of Nanocrystalline Dye-Sensitized Solar Cells with Flexible Metal Substrates by Electrochemical Impedance Spectroscopy," *J. Electrochem. Soc.*, vol. 154, no. 1, p. B68, 2006, doi: 10.1149/1.2374943.
- [112] J. Linnemann, J. Giorgio, K. Wagner, G. Mathieson, G. G. Wallace, and D. L. Officer, "A simple one step process for enhancement of titanium foil dye sensitised solar cell anodes," *J. Mater. Chem. A*, vol. 3, no. 7, pp. 3266–3270, 2015, doi: 10.1039/c4ta05407e.
- [113] Y. Rui *et al.*, "In-situ construction of three-dimensional titania network on Ti foil toward enhanced performance of flexible dye-sensitized solar cells," *Appl. Surf. Sci.*, vol. 380, pp. 210–217, 2016, doi: 10.1016/j.apsusc.2016.01.156.
- [114] J. An, W. Guo, and T. Ma, "Enhanced photoconversion efficiency of all-flexible dye-sensitized solar cells based on a Ti substrate with TiO₂ nanoforest underlayer," *Small*, vol. 8, no. 22, pp. 3427–3431, 2012, doi: 10.1002/smll.201200802.
- [115] S. S. Taleghani, M. Reza, Z. Meymian, and M. Ameri, "Interfacial modification to optimize stainless steel photoanode design for flexible dye sensitized solar cells: an experimental and numerical modeling approach," *J. Phys. D: Appl. Phys.*, p. 405601, doi: 10.1088/0022-3727/49/40/405601.
- [116] Y. Jun, J. Kim, and M. G. Kang, "A study of stainless steel-based dye-sensitized solar cells and modules," *Sol. Energy Mater. Sol. Cells*, vol. 91, no. 9, pp. 779–784, 2007, doi: 10.1016/j.solmat.2007.01.007.
- [117] C. H. Huang, Y. W. Chen, and C. M. Chen, "Chromatic Titanium Photoanode for Dye-Sensitized Solar Cells under Rear Illumination," *ACS Appl. Mater. Interfaces*, vol. 10, no. 3, pp. 2658–2666, 2018, doi: 10.1021/acsami.7b18351.
- [118] E. Gaul, "Coloring titanium and related metals by electrochemical oxidation," *J. Chem. Educ.*, vol. 70, no. 3, pp. 176–178, 1993, doi: 10.1021/ed070p176.
- [119] S.-Y. Wu, Y.-H. Chen, K.-C. Chen, and J.-L. He, "Using Micro-Arc Oxidation and Alkali Etching to Produce a Nanoporous TiO₂ Layer on Titanium Foil for Flexible Dye-Sensitized Solar Cell Application," *Jpn.*

- J. Appl. Phys.*, vol. 49, no. 9R, p. 92301, Sep. 2010, doi: 10.1143/JJAP.49.092301.
- [120] Y.-Y. Tsai, P.-L. Song, and J. Fang, "Colorful Oxide Film Formation on Titanium by using EDM Process," *Int. J. Precis. Eng. Manuf.*, vol. 14, pp. 1933–1937, 2013, doi: 10.1007/s12541-013-0262-0.
- [121] Y. W. Chen, M. Chandra Sil, and C. M. Chen, "Increasing solar light efficiency by engineering cell structures with modified Ti foil and specific concentrations of electrolyte in liquid dye-sensitized solar cells," *Electrochim. Acta*, vol. 334, p. 135631, 2020, doi: 10.1016/j.electacta.2020.135631.
- [122] J. Wang, X. Miao, Q. Fengzhao, C. Ren, Z. Yang, and L. Wang, "7.35%–Efficiency rear–irradiated flexible dye–sensitized solar cell by sealing liquid electrolyte in a groove," *Chem. Commun.*, vol. 51, pp. 491–494, 2015, doi: 10.1039/C4CC07549H.
- [123] M. Jeyaraj, S. Gurunathan, M. Qasim, M. H. Kang, and J. H. Kim, "A comprehensive review on the synthesis, characterization, and biomedical application of platinum nanoparticles," *Nanomaterials*, vol. 9, no. 12, 2019, doi: 10.3390/nano9121719.
- [124] J. Ma, S. Qingfeng, Z. Fengbao, and W. Mingxing, "Improvement on the catalytic activity of the flexible PEDOT counter electrode in dye-sensitized solar cells," *Mater. Res. Bull.*, vol. 100, no. December 2017, pp. 213–219, 2018, doi: 10.1016/j.materresbull.2017.12.031.
- [125] A. Khataee and G. A. Mansoori, *Nanostructured Titanium Dioxide Materials Properties, Preparation and Applications*. World Scientific Publishing Co. Pte. Ltd., 2012.
- [126] D. Dambournet, I. Belharouak, and K. Amine, "Tailored preparation methods of TiO₂ anatase, rutile, brookite: Mechanism of formation and electrochemical properties," *Chem. Mater.*, vol. 22, no. 3, pp. 1173–1179, 2010, doi: 10.1021/cm902613h.
- [127] S. Shen *et al.*, "Titanium dioxide nanostructures for photoelectrochemical applications," *Prog. Mater. Sci.*, vol. 98, no. May, pp. 299–385, 2018, doi: 10.1016/j.pmatsci.2018.07.006.
- [128] C. Cavallo, F. Di Pascasio, A. Latini, M. Bonomo, and D. Dini, "Nanostructured Semiconductor Materials for Dye-Sensitized Solar Cells," *J. Nanomater.*, vol. 2017, 2017, doi: 10.1155/2017/5323164.
- [129] L. Wu *et al.*, "Surface chemistry and photochemistry of small molecules on rutile TiO₂(001) and TiO₂(011)-(2 × 1) surfaces: The crucial roles of

- defects," *J. Chem. Phys.*, vol. 152, no. 4, 2020, doi: 10.1063/1.5135945.
- [130] N. Park, J. Van De Lagemaat, and A. J. Frank, "Comparison of Dye-Sensitized Rutile- and Anatase-Based TiO₂ Solar Cells," pp. 8989–8994, 2000.
- [131] I. M. Low, H. Albetran, V. M. Prida, V. Vega, P. Manurung, and M. Ionescu, "A comparative study on crystallization behavior, phase stability, and binding energy in pure and Cr-doped TiO₂ nanotubes," *J. Mater. Res.*, vol. 28, no. 3, pp. 304–312, 2013, doi: 10.1557/jmr.2012.275.
- [132] S. V. Umale, S. N. Tambat, V. Sudhakar, S. M. Sontakke, and K. Krishnamoorthy, "Fabrication, characterization and comparison of DSSC using anatase TiO₂ synthesized by various methods," *Adv. Powder Technol.*, vol. 28, no. 11, pp. 2859–2864, 2017, doi: 10.1016/j.apt.2017.08.012.
- [133] K. A. and P. D. L. Minna Toivola, Janne Halme, Kati Miettunen, "Nanostructured dye solar cells on flexible substrates—Review," *Int. J. ENERGY Res.*, vol. 33, pp. 1145–1160, 2009, doi: 10.1002/er.1605.
- [134] M. Toivola, F. Ahlskog, and P. Lund, "Industrial sheet metals for nanocrystalline dye-sensitized solar cell structures," *Sol. Energy Mater. Sol. Cells*, vol. 90, no. 17, pp. 2881–2893, 2006, doi: 10.1016/j.solmat.2006.05.002.
- [135] K. Miettunen *et al.*, "Stability of Dye Solar Cells with Photoelectrode on Metal Substrates," *J. Electrochem. Soc.*, vol. 157, no. 6, p. B814, 2010, doi: 10.1149/1.3374645.
- [136] Y. J. and M. G. K. Ho-Gyeong Yun, Byeong-Soo Bae, "Effective Methods for the High Efficiency Dye-Sensitized Solar Cells Based on the Metal Substrates, Solar Cells - Dye-Sensitized Devices," 2nd ed., L. A. Kosyachenko, Ed. Chernivtsi National University, Ukraine: Intech, 2011, pp. 267–278.
- [137] K. M. Lee, L. C. Lin, V. Suryanarayanan, and C. G. Wu, "Titanium dioxide coated on titanium/stainless steel foil as photoanode for high efficiency flexible dye-sensitized solar cells," *J. Power Sources*, vol. 269, pp. 789–794, 2014, doi: 10.1016/j.jpowsour.2014.07.013.
- [138] W. Q. Wu, Y. F. Xu, H. S. Rao, C. Y. Su, and D. Bin Kuang, "Trilayered photoanode of TiO₂ nanoparticles on a 1D-3D nanostructured TiO₂-grown flexible Ti substrate for high-efficiency (9.1%) dye-sensitized solar cells with unprecedentedly high photocurrent density," *J. Phys. Chem. C*, vol. 118, no. 30, pp. 16426–16432, 2014, doi: 10.1021/jp4116782.

- [139] L.-Y. Lin *et al.*, "Low-temperature flexible Ti/TiO₂ photoanode for dye-sensitized solar cells with binder-free TiO₂ paste," *Prog. Photovoltaics Res. Appl.*, vol. 20, 2012, doi: 10.1002/pip.1116.
- [140] T. Watson, G. Reynolds, D. Wragg, G. Williams, and D. Worsley, "Corrosion Monitoring of Flexible Metallic Substrates for Dye-Sensitized Solar Cells," *Int. J. Photoenergy*, vol. 2013, p. 791438, 2013, doi: 10.1155/2013/791438.
- [141] M. G. Kang, A. N. Park, K. S. Ryu, S. H. Chang, and K. K. Åy, "Flexible Metallic Substrates for TiO₂ Film of Dye-sensitized Solar Cells," vol. 34, no. 6, pp. 1–2, 2005, doi: 10.1246/cl.2005.804.
- [142] C.-H. Lee, W.-H. Chiu, K.-M. Lee, W.-F. Hsieh, and J.-M. Wu, "Improved performance of flexible dye-sensitized solar cells by introducing an interfacial layer on Ti substrates," *J. Mater. Chem.*, vol. 21, no. 13, p. 5114, 2011, doi: 10.1039/c0jm04099a.
- [143] A. K. Baranwal, "Transparent conductive oxide-less back contact dye-sensitized solar cells using flat titanium sheet with microholes for photoanode fabrication," vol. 7, no. 1, 2017, doi: 10.1117/1.JPE.7.015501.
- [144] H. G. Yun, B. S. Bae, and M. G. Kang, "A simple and highly efficient method for surface treatment of Ti substrates for use in dye-sensitized solar cells," *Adv. Energy Mater.*, vol. 1, no. 3, pp. 337–342, 2011, doi: 10.1002/aenm.201000044.
- [145] J. Wu, S. Hayakawa, K. Tsuru, and A. Osaka, "Porous titania films prepared from interactions of titanium with hydrogen peroxide solution," *Scr. Mater.*, vol. 46, pp. 101–106, 2002.
- [146] T. Tsai, C. Chen, S. Cherng, and S. Suen, "An efficient titanium-based photoanode for dye-sensitized solar cell under back-side illumination," no. September 2011, pp. 226–231, 2013, doi: 10.1002/pip.
- [147] Y. T. Huang, H. Lee, W. Di Li, and S. P. Feng, "Engineered platinum nanoparticles via pulse electrochemical deposition for bifacially transparent and efficient full-plastic dye-sensitized solar cells," *J. Power Sources*, vol. 435, no. June, pp. 1–6, 2019, doi: 10.1016/j.jpowsour.2019.226801.
- [148] R. S. Moraes, E. Saito, D. M. G. Leite, M. Massi, and A. S. Da Silva Sobrinho, "Optical, electrical and electrochemical evaluation of sputtered platinum counter electrodes for dye sensitized solar cells," *Appl. Surf. Sci.*, vol. 364, pp. 229–234, 2016, doi: 10.1016/j.apsusc.2015.12.114.

- [149] G. Calogero, P. Calandra, A. Irrera, A. Sinopoli, I. Citro, and G. Di Marco, "A new type of transparent and low cost counter-electrode based on platinum nanoparticles for dye-sensitized solar cells," *Energy Environ. Sci.*, vol. 4, no. 5, pp. 1838–1844, 2011, doi: 10.1039/c0ee00463d.
- [150] H. Wender, P. Migowski, A. F. Feil, S. R. Teixeira, and J. Dupont, "Sputtering deposition of nanoparticles onto liquid substrates: Recent advances and future trends," *Coord. Chem. Rev.*, vol. 257, no. 17–18, pp. 2468–2483, 2013, doi: 10.1016/j.ccr.2013.01.013.
- [151] K.-H. Dahmen, "Chemical Vapor Deposition," in *Encyclopedia of Physical Science and Technology (Third Edition)*, Third Edit., R. A. Meyers, Ed. New York: Academic Press, 2003, pp. 787–808.
- [152] A. S. H. Makhlof, *Current and advanced coating technologies for industrial applications*. Woodhead Publishing Limited, 2011.
- [153] B. S. Yilbas, A. Al-Sharafi, and H. Ali, "Chapter 3 - Surfaces for Self-Cleaning," in *Self-Cleaning of Surfaces and Water Droplet Mobility*, B. S. Yilbas, A. Al-Sharafi, and H. Ali, Eds. Elsevier, 2019, pp. 45–98.
- [154] C. Longo and M. A. De Paoli, "Dye-Sensitized Solar Cells: A Successful Combination of Materials," *J. Braz. Chem. Soc.*, vol. 14, no. 6, pp. 889–901, 2003, doi: 10.1590/S0103-50532003000600005.
- [155] J. Wu *et al.*, "Counter electrodes in dye-sensitized solar cells," *Chem. Soc. Rev.*, vol. 46, no. 19, pp. 5975–6023, 2017, doi: 10.1039/c6cs00752j.
- [156] J. Orava, T. Kohoutek, and T. Wagner, *Deposition techniques for chalcogenide thin films*. 2013.
- [157] S. Iwata, H. Furuta, M. Kanayama, N. Imawaka, and K. Yoshino, "Fabrication of a patterned Pt counter electrode for dye-sensitized solar cells using neutralized H₂PtCl₆·6H₂O paste," *Mater. Today Commun.*, vol. 18, no. December 2018, pp. 163–166, 2019, doi: 10.1016/j.mtcomm.2018.12.004.
- [158] M. Ali *et al.*, "Synthesis and utilization of platinum (II) dialkyldithiocarbamate precursors in aerosol assisted chemical vapor deposition of platinum thin films as counter electrodes for dye-sensitized solar cells," vol. 166, pp. 186–195, 2019, doi: 10.1016/j.poly.2019.03.058.
- [159] M. Zalas and K. Jelak, "Optimization of platinum precursor concentration for new, fast and simple fabrication method of counter electrode for DSSC application," *Optik (Stuttg.)*, vol. 206, no. January, p. 164314, 2020, doi:

10.1016/j.ijleo.2020.164314.

- [160] A. Ghifari, D. X. Long, S. Kim, B. Ma, and J. Hong, "Transparent platinum counter electrode prepared by polyol reduction for bifacial, dye-sensitized solar cells," *Nanomaterials*, vol. 10, no. 3, 2020, doi: 10.3390/nano10030502.
- [161] K. Shimada, T. Toyoda, and T. Taima, "Platinum counter electrodes for dye-sensitized solar cells prepared by one-step dipping process," *Jpn. J. Appl. Phys.*, vol. 58, no. 12, p. 124001, 2019, doi: 10.7567/1347-4065/ab50ea.
- [162] S. Mukherji, S. Bharti, G. Shukla, and S. Mukherji, "Synthesis and characterization of size- And shape-controlled silver nanoparticles," *Phys. Sci. Rev.*, vol. 4, no. 1, pp. 1–73, 2019, doi: 10.1515/psr-2017-0082.
- [163] Y. Kamikoriyama, H. Imamura, A. Muramatsu, and K. Kanie, "Ambient Aqueous-Phase Synthesis of Copper Nanoparticles and Nanopastes with Low-Temperature Sintering and Ultra-High Bonding Abilities," *Sci. Rep.*, vol. 9, no. 1, pp. 1–10, 2019, doi: 10.1038/s41598-018-38422-5.
- [164] R. G. Nair, S. Ojah, P. M. Kumar, S. K. Nikhil, and S. K. Samdarshi, "Role of copper and silver modified titania photoanode on performance engineering of dye sensitized solar cells," *Mater. Lett.*, vol. 221, pp. 313–317, 2018, doi: 10.1016/j.matlet.2018.03.150.
- [165] J. Y. Park *et al.*, "Copper and nitrogen doping on TiO₂ photoelectrodes and their functions in dye-sensitized solar cells," *J. Power Sources*, vol. 306, pp. 764–771, 2016, doi: 10.1016/j.jpowsour.2015.12.087.
- [166] A. Gupta, K. Sahu, M. Dhonde, and V. V. S. Murty, "Novel synergistic combination of Cu/S co-doped TiO₂ nanoparticles incorporated as photoanode in dye sensitized solar cell," *Sol. Energy*, vol. 203, no. April, pp. 296–303, 2020, doi: 10.1016/j.solener.2020.04.043.
- [167] C. Thambiliyagodage and S. Mirihana, "Photocatalytic activity of Fe and Cu co-doped TiO₂ nanoparticles under visible light," *J. Sol-Gel Sci. Technol.*, vol. 99, no. 1, pp. 109–121, 2021, doi: 10.1007/s10971-021-05556-4.
- [168] H. R. Ghorbani, "A review of methods for synthesis of Al nanoparticles," *Orient. J. Chem.*, vol. 30, no. 4, pp. 1941–1949, 2014, doi: 10.13005/ojc/300456.
- [169] C. R. Jacobson, D. Solti, D. Renard, L. Yuan, M. Lou, and N. J. Halas, "Shining light on aluminum nanoparticle synthesis," *Acc. Chem. Res.*, vol.

53, no. 9, pp. 2020–2030, 2020, doi: 10.1021/acs.accounts.0c00419.

- [170] A. Ziashahabi and R. Poursalehi, “The Effects of Surface Oxidation and Interparticle Coupling on Surface Plasmon Resonance Properties of Aluminum Nanoparticles as a UV Plasmonic Material,” *Procedia Mater. Sci.*, vol. 11, pp. 434–437, 2015, doi: 10.1016/j.mspro.2015.11.030.
- [171] N. C. Bigall and A. Eychmüller, “Synthesis of noble metal nanoparticles and their non-ordered superstructures,” *Philos. Trans. R. Soc. A Math. Phys. Eng. Sci.*, vol. 368, no. 1915, pp. 1385–1404, 2010, doi: 10.1098/rsta.2009.0274.
- [172] Y. Gutiérrez, A. S. Brown, F. Moreno, and M. Losurdo, “Plasmonics beyond noble metals: Exploiting phase and compositional changes for manipulating plasmonic performance,” *J. Appl. Phys.*, vol. 128, no. 8, 2020, doi: 10.1063/5.0020752.
- [173] S. Li *et al.*, “Recent Advances in Plasmonic Nanostructures for Enhanced Photocatalysis and Electrocatalysis,” *Adv. Mater.*, vol. 33, no. 6, pp. 1–19, 2021, doi: 10.1002/adma.202000086.
- [174] S. Akter and M. A. Huq, “Biologically rapid synthesis of silver nanoparticles by *Sphingobium* sp. MAH-11T and their antibacterial activity and mechanisms investigation against drug-resistant pathogenic microbes,” *Artif. Cells, Nanomedicine Biotechnol.*, vol. 48, no. 1, pp. 672–682, 2020, doi: 10.1080/21691401.2020.1730390.
- [175] M. Mavaei, A. Chahardoli, Y. Shokoohinia, A. Khoshroo, and A. Fattahi, “One-step Synthesized Silver Nanoparticles Using Isoimperatorin: Evaluation of Photocatalytic, and Electrochemical Activities,” *Sci. Rep.*, vol. 10, no. 1, pp. 1–12, 2020, doi: 10.1038/s41598-020-58697-x.
- [176] Y. Gutiérrez, R. A. de la Osa, D. Ortiz, J. M. Saiz, F. González, and F. Moreno, “Plasmonics in the ultraviolet with aluminum, gallium, magnesium and rhodium,” *Appl. Sci.*, vol. 8, no. 1, 2018, doi: 10.3390/app8010064.
- [177] P. Slepíčka, N. S. Kasálková, J. Siegel, Z. Kolská, and V. Švorčík, “Methods of gold and silver nanoparticles preparation,” *Materials (Basel)*, vol. 13, no. 1, p. 1, 2020, doi: 10.3390/ma13010001.
- [178] L. Wei, X. Xia, Y. Yang, P. Wang, Y. Dong, and T. Luan, “Variable temperature spectroelectrochemistry study of silver-doped TiO₂ and its influence on the performance of dye sensitized solar cells,” *RSC Adv.*, vol. 6, no. 72, pp. 68341–68350, 2016, doi: 10.1039/c6ra10747h.
- [179] K. B. Bhojanaa, M. Ramesh, and A. Pandikumar, “Complementary

properties of silver nanoparticles on the photovoltaic performance of titania nanospheres based photoanode in dye-sensitized solar cells," *Mater. Res. Bull.*, vol. 122, p. 110672, 2020, doi: <https://doi.org/10.1016/j.materresbull.2019.110672>.

- [180] Y. X. Dong, X. L. Wang, E. M. Jin, S. M. Jeong, B. Jin, and S. H. Lee, "One-step hydrothermal synthesis of Ag decorated TiO₂ nanoparticles for dye-sensitized solar cell application," *Renew. Energy*, vol. 135, pp. 1207–1212, 2019, doi: 10.1016/j.renene.2018.12.062.
- [181] P. Nbelayim, G. Kawamura, W. Kian Tan, H. Muto, and A. Matsuda, "Systematic characterization of the effect of Ag@TiO₂ nanoparticles on the performance of plasmonic dye-sensitized solar cells," *Sci. Rep.*, vol. 7, no. 1, pp. 1–12, 2017, doi: 10.1038/s41598-017-15541-z.
- [182] M. Fallah and I. Maleki, "Enhancing the efficiency of dye-sensitized solar cell by increasing the light trapping and decreasing the electron-hole recombination rate due to Ag @ TiO₂ core-shell photoanode structure Enhancing the efficiency of dye-sensitized solar cell by increas," *Mater. Res. Express*, vol. 7, no. 1, p. 16409, 2020, doi: 10.1088/2053-1591/ab5c8a.
- [183] G. Castillo Dalí and D. Torres Lagares, "Chapter 1 - Nanobiomaterials in hard tissue engineering," in *Nanobiomaterials in Hard Tissue Engineering*, A. M. Grumezescu, Ed. William Andrew Publishing, 2016, pp. 1–31.
- [184] Y. Battie *et al.*, "Influence of the laser light absorption by the colloid on the properties of silver nanoparticles produced by laser ablation in stirred and stationary liquid Influence of the laser light absorption by the colloid on the properties of silver nanoparticles p," vol. 113103, 2015, doi: 10.1063/1.4915277.
- [185] J. A. Gan and C. C. Berndt, "4 - Plasma surface modification of metallic biomaterials," in *Surface Coating and Modification of Metallic Biomaterials*, C. Wen, Ed. Woodhead Publishing, 2015, pp. 103–157.
- [186] V. Sittler, A. J. S. Ahammad, A. Subhan, and J. Uddin, "Enhancing the Performance of Dye Sensitized Solar Cells Using Silver Nanoparticles Modified Photoanode," 2020.
- [187] N. Kaur, V. Bhullar, D. P. Singh, and A. Mahajan, "Bimetallic Implanted Plasmonic Photoanodes for TiO₂ Sensitized Third Generation Solar Cells," pp. 1–16, 2020, doi: 10.1038/s41598-020-64653-6.
- [188] N. Kaur, A. Mahajan, V. Bhullar, and D. P. Singh, "Ag ion implanted TiO₂ photoanodes for fabrication of highly efficient and Abstract :," 2019.

- [189] R. Selvapriya, "Screen printed multifunctional TiO₂ photoanode with plasmonic Ag nanoparticles for performance enhancement of dye sensitized solar cell," 2020.
- [190] D. N. Joshi, P. Ilaiyaraja, C. Sudakar, and R. A. Prasath, "Facile one-pot synthesis of multi-shaped silver nanoparticles with tunable ultra-broadband absorption for efficient light harvesting in dye-sensitized solar cells," *Sol. Energy Mater. Sol. Cells*, vol. 185, no. November 2017, pp. 104–110, 2018, doi: 10.1016/j.solmat.2018.05.018.
- [191] M. A. Al-azawi, N. Bidin, M. Bououdina, and S. M. Mohammad, "Preparation of gold and gold – silver alloy nanoparticles for enhancement of plasmonic dye-sensitized solar cells performance," vol. 126, pp. 93–104, 2016, doi: 10.1016/j.solener.2015.12.043.
- [192] S. Ranjitha, V. Aroulmoji, T. Selvankumar, C. Sudhakar, and V. Hariharan, "Synthesis and development of novel sensitizer from spirulina pigment with silver doped TiO₂ nano particles for bio-sensitized solar cells," *Biomass and Bioenergy*, vol. 141, p. 105733, 2020, doi: <https://doi.org/10.1016/j.biombioe.2020.105733>.
- [193] W.-Y. Wu, C.-F. Hsu, M.-J. Wu, C.-N. Chen, and J.-J. Huang, "Ag–TiO₂ composite photoelectrode for dye-sensitized solar cell," *Appl. Phys. A*, vol. 123, no. 5, p. 357, 2017, doi: 10.1007/s00339-017-0963-9.
- [194] S. Buda, "Enhancement of Dye-Sensitized Solar Cell Power Conversion Efficiency by Improving Carrier Generation and Charge Collection," 2017.
- [195] M. M. and H. S. A. Berni, "Doctor blades," in *Sol-gel technologies for glass producers and users*, M. Aegerter, M.A. & Mennig, Ed. USA, MA, Boston:Springer, 2004, pp. 89–92.
- [196] S. Phomma, T. Wutikhun, P. Kasamechonchung, T. Eksangsri, and C. Sapcharoenkun, "Effect of calcination temperature on photocatalytic activity of synthesized TiO₂ nanoparticles via wet ball milling sol-gel method," *Appl. Sci.*, vol. 10, no. 3, 2020, doi: 10.3390/app10030993.
- [197] P. Nyamukamba, O. Okoh, H. Mungondori, R. Taziwa, and S. Zinya, "Synthetic Methods for Titanium Dioxide Nanoparticles: A Review," *Titan. Dioxide - Mater. a Sustain. Environ.*, 2018, doi: 10.5772/intechopen.75425.
- [198] D. Nunes *et al.*, *Synthesis, design, and morphology of metal oxide nanostructures*. 2019.
- [199] D. Z. Yiwei Tan, Yongfang Li, "Noble Metal Nanoparticles," *Encycl. Nanosci. Nanotechnol.*, vol. 8, pp. 9–40, 2004.

- [200] D. H. Sliney, M. Bitran, and W. Murray, "Infrared, Visible, and Ultraviolet Radiation," *Patty's Toxicol.*, vol. 6, pp. 169–208, 2012, doi: 10.1002/0471435139.tox102.pub2.
- [201] S. Shaban, "Optimized Titanium Dioxide Photoanode on Flexible Substrates For Dye-Sensitized Solar Cell," Universiti Putra Malaysia, 2020.
- [202] K. Kumari, N. Sanjeevadarshini, L. Dissanayake, G. K. R. Senadeera, and C. Thotawatthage, "The effect of TiO₂ photo anode film thickness on photovoltaic properties of dye-sensitized solar cells," *Ceylon J. Sci.*, vol. 45, p. 33, 2016, doi: 10.4038/cjs.v45i1.7362.
- [203] M. G. Kang, K. S. Ryu, S. H. Chang, N. G. Park, J. S. Hong, and K. Kim, "Dependence of TiO₂ Film Thickness on Photocurrent-Voltage Characteristics of Dye-Sensitized Solar Cells," *Bull. Korean Chem. Soc.*, vol. 25, pp. 742–744, 2004.
- [204] S. Ito, S. M. Zakeeruddin, P. Comte, P. Liska, D. Kuang, and M. Grätzel, "Bifacial dye-sensitized solar cells based on an ionic liquid electrolyte," *Nat. Photonics*, vol. 2, no. 11, pp. 693–698, 2008, doi: 10.1038/nphoton.2008.224.
- [205] H. Wang *et al.*, "Low resistance dye-sensitized solar cells based on all-titanium substrates using wires and sheets," *Appl. Surf. Sci.*, vol. 255, pp. 9020–9025, 2009, doi: 10.1016/j.apsusc.2009.06.085.
- [206] E. S. Teixeira *et al.*, "Building and testing a spin coater for the deposition of thin films on DSSCS," *Mater. Res.*, vol. 23, no. 6, 2020, doi: 10.1590/1980-5373-MR-2020-0214.
- [207] M. A. Ehsan *et al.*, "Synthesis and utilization of platinum(II) dialkyldithiocarbamate precursors in aerosol assisted chemical vapor deposition of platinum thin films as counter electrodes for dye-sensitized solar cells," *Polyhedron*, vol. 166, pp. 186–195, 2019, doi: 10.1016/j.poly.2019.03.058.
- [208] U. S. Maret, M. Faculty, and U. S. Maret, "Preparation of colloidal silver nanoparticles," no. October 2020, pp. 754–760, 2021.
- [209] A. Mirzaei, K. Janghorban, B. Hashemi, M. Bonyani, S. G. Leonardi, and G. Neri, "Characterization and optical studies of PVP-capped silver nanoparticles," *J. Nanostructure Chem.*, vol. 7, no. 1, pp. 37–46, 2017, doi: 10.1007/s40097-016-0212-3.
- [210] D. R. H. Craig F. Bohren, *Absorption and scattering of light by small particles*, Reprint, R. John Wiley & Sons, 2008, 2008.

- [211] J. An, W. Guo, and T. Ma, "Enhanced photoconversion efficiency of all-flexible dye-sensitized solar cells based on a Ti substrate with TiO₂ nanoforest underlayer," *Small*, vol. 8, no. 22, pp. 3427–3431, 2012, doi: 10.1002/sml.201200802.
- [212] J. Villanueva-Cab, J. L. Montaño-Priede, and U. Pal, "Effects of Plasmonic Nanoparticle Incorporation on Electrostatics and Photovoltaic Performance of Dye Sensitized Solar Cells," *J. Phys. Chem. C*, vol. 120, no. 19, pp. 10129–10136, 2016, doi: 10.1021/acs.jpcc.6b01053.
- [213] E. V. Salomatina, D. G. Fukina, A. V. Koryagin, D. N. Titaev, E. V. Suleimanov, and L. A. Smirnova, "Preparation and photocatalytic properties of titanium dioxide modified with gold or silver nanoparticles," *J. Environ. Chem. Eng.*, vol. 9, no. 5, p. 106078, 2021, doi: 10.1016/j.jece.2021.106078.
- [214] S. J. Lin, K. C. Lee, J. L. Wu, and J. Y. Wu, "Plasmon-enhanced photocurrent in dye-sensitized solar cells," *Sol. Energy*, vol. 86, no. 9, pp. 2600–2605, 2012, doi: 10.1016/j.solener.2012.05.027.
- [215] D. T. Gangadharan, Z. Xu, Y. Liu, R. Izquierdo, and D. Ma, "Recent advancements in plasmon-enhanced promising third-generation solar cells," *Nanophotonics*, vol. 6, no. 1, pp. 153–175, 2017, doi: 10.1515/nanoph-2016-0111.
- [216] W. R. Erwin, H. F. Zarick, E. M. Talbert, and R. Bardhan, "Light trapping in mesoporous solar cells with plasmonic nanostructures," *Energy Environ. Sci.*, vol. 9, no. 5, pp. 1577–1601, 2016, doi: 10.1039/c5ee03847b.
- [217] J. Disdier, J.-M. Herrmann, and P. Pichat, "Platinum/titanium dioxide catalysts. A photoconductivity study of electron transfer from the ultraviolet-illuminated support to the metal and of the influence of hydrogen," *J. Chem. Soc., Faraday Trans. 1*, vol. 79, no. 3, pp. 651–660, 1983, doi: 10.1039/F19837900651.
- [218] J. Li *et al.*, "Silver nanoparticle doped TiO₂ nanofiber dye sensitized solar cells," *Chem. Phys. Lett.*, vol. 514, pp. 141–145, 2011.
- [219] S. Shah *et al.*, "Plasmonic effects of quantum size metal nanoparticles on dye-sensitized solar cell," *Opt. Mater. Express*, vol. 7, no. 6, p. 2069, 2017, doi: 10.1364/ome.7.002069.
- [220] F. Meng and Z. Sun, "A mechanism for enhanced hydrophilicity of silver nanoparticles modified TiO₂ thin films deposited by RF magnetron sputtering," *Appl. Surf. Sci.*, vol. 255, pp. 6715–6720, 2009.

- [221] S. P. Lim, A. Pandikumar, N. M. Huang, H. N. Lim, G. Gu, and T. L. Ma, "Promotional effect of silver nanoparticles on the performance of N-doped TiO₂ photoanode-based dye-sensitized solar cells," *RSC Adv.*, vol. 4, no. 89, pp. 48236–48244, 2014, doi: 10.1039/c4ra09775k.
- [222] F. Saadmim *et al.*, "Enhancing the Performance of Dye Sensitized Solar Cells Using Silver Nanoparticles Modified Photoanode.," *Molecules*, vol. 25, no. 17, Sep. 2020, doi: 10.3390/molecules25174021.
- [223] G. A. Alamu, O. Adedokun, I. T. Bello, and Y. K. Sanusi, "Plasmonic enhancement of visible light absorption in Ag-TiO₂ based dye-sensitized solar cells," *Chem. Phys. Impact*, vol. 3, p. 100037, 2021, doi: 10.1016/j.chphi.2021.100037.
- [224] S. Shaban, S. Shafie, Y. Sulaiman, F. Ahmad, M. Q. Lokman, and N. F. M. Sharif, "Flexible Photoanode on Titanium Foil for Back-Illuminated Dye Sensitized Solar Cells," in *2018 IEEE International Conference on Semiconductor Electronics (ICSE)*, 2018, pp. 197–200, doi: 10.1109/SMELEC.2018.8481325.