

### **UNIVERSITI PUTRA MALAYSIA**

# DEVELOPMENT OF SLOW RELEASE FERTILISERS FOR PINEAPPLE

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# MASTER OF AGRICULTURAL SCIENCE UNIVERSITI PUTRA MALAYSIA

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#### DEVELOPMENT OF SLOW RELEASE FERTILISERS FOR PINEAPPLE

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### Dedicated to

'The estimable nucleus in my life who championed my struggled'

Thank you Mother dearest



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	FERIALS AND METHODS
Prepa	aration of Compound Fertiliser for Pineapple
	Materials
	Methods
	Total Nutrients of Fertiliser
Coat	ing of the Compound Fertiliser
	Materials
	Coating of the Compound Fertilisers with
	Polyacrylamide, Polyvinyl Chloride and
	Natural Rubber
	Coating of the Compound Fertiliser with Polylactic Acid
Phys	ical and Chemical Characteristics of Coated
	lisers
	Infrared Spectral Analysis
	Differential Scanning Calorimetry
	Thermogravimetric Analysis
	Scanning Electron Microscopy
Evalı	uation of the Coated Fertilisers
	Materials
	Methods
An C	Open Field Leaching system
	Treatments
	Materials and Management
	Statistical Analysis
RES	ULTS AND DISCUSSIONS
	ical and Chemical Characteristics of the
-	pound Fertilisers
	nical and Physical Characteristics of
	ing Materials
Cour	The Data from FT-IR Measurements
	Differential Scanning Calorimeter and
	Thermogravimetric Analysis of
	Coated Materials
	Thickness of Coating Materials of
	Coated Fertiliser
Eval	uation of Coated Compound Fertiliser
	hing Experiment in the Laboratory
	ients Content in the Leachate
_ 1001	Nitrogen
	Phosphorus.
	Potassium
	Calcium, Magnesium, and Copper
Distr	ibution of Nutrients in the Soil Column
	Onen Leaching System in the Field



	Distribution of Nutrients in the Soil Column  Nutrients Remained in the Soil Column from the  Laboratory and Field Experiments	88 94
5	SUMMARY CONCLUSION AND FURTHER STUDIES Summary and Conclusion	98 98 102
	SIBLIOGRAPHY	102
A A	2 more and seeming constraints of the commentation	113
	Used in the Study (i) PVC (ii) PA (iii) NR, and(iv) PLA	114
В	Thermogravimetric Analysis of the Coating Materials Used in the Study (i) PVC (ii) PA (iii) NR, and (iv) PLA	119
C	Concentration of (i) Nitrogen 1 and 2, (ii) Phosphorus 1 and 2, (iii) Potassium 1 and 2, (v) Calcium 1 and 2, (iv) Magnesium 1 and 2, (vi) Copper 1 and 2 in the leachate after leaching with distilled water for 30 days of untreated	
<b>X</b> 77	and treated soil column	123



## LIST OF TABLES

Table		Page
1.	Examples of Ionic Functional Groups Which May Used to Impart A negative or Positive Charge in Hydrophilic Polymers	17
2.	Recommended Fertilisation Program For the Singapore Spanish Variety at an Original Population of 27,250 Plants Per Acre.	38
3.	Specification of the Granulation Machine	41
4.	Physical and Chemical Characteristics of Puchong Soil For the Leaching Experiment	47
5.	The Fertiliser Treatments in the Field Experiment	53
6.	Specifications of the fertilisers used in the Field Experiment	54
7.	Total Nutrients in the Compound Fertiliser used in the study	57
8.	The Endothermic and Exothermic Peaks of the Coating Material Using DSC	61
9.	The Thickness of the Coating Materials of the Fertilisers Granule Using SEM	62
10.	The Amount of N Lost and Remained from Uncoated and Coated Fertilisers in the Soil after Leaching for 30 days	68
11.	The Amount of P Lost and Remained from Uncoated and Coated Fertilisers in the Soil after Leaching for 30 days	72
12.	The Amount of K Lost and Remained from Uncoated and Coated Fertilisers in the Soil after Leaching for 30 days	76
13.	The Amount of Ca Lost and Remained from Uncoated and Coated Fertilisers in the Soil after Leaching for 30 days	79
14.	The Amount of Mg Lost and Remained from Uncoated and Coated Fertilisers in the Soil after Leaching for 30 days.	80



15.	Fertilisers in the Soil after Leaching for 30 days	82
16.	Distribution of Total N of the Different Coated Fertilisers in Soil column after Leaching for 30 days	83
17.	Distribution of Extractable P of the Different Coated Fertilisers in Soil column after Leaching for 30 days	84
18.	Distribution of Exchangeable K of the Different Coated Fertilisers in Soil column after Leaching for 30 days	85
19.	Distribution of Exchangeable Ca of the Different Coated Fertilisers in Soil column after Leaching for 30 days	86
20.	Distribution of Exchangeable Mg of the Different Coated Fertilisers in Soil column after Leaching for 30 days	86
21.	Distribution of Extractable Cu of the Different Coated Fertilisers in Soil column after Leaching for 30 days	87
22.	Distribution of Total N of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field	89
23.	Distribution of Extractable P of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field	90
24.	Distribution of Exchangeable K of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field.	91
25.	Distribution of Exchangeable Ca of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field.	92
26.	Distribution of Exchangeable Mg of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field.	92
27.	Distribution of Extractable Cu of the Different Coated Fertilisers in Soil column after Leaching for 3 months in the field.	94
28.	Nutrients Remained in the Soil at the Laboratory Experiment and	97



### LIST OF FIGURES

Figure		Pag
1.	Arrangement of Nutrients is Resin Coated controlled Release Fertiliser	1
2.	Poly-S Coated as Controlled-release Fertiliser	1
3.	Schematic Diagram of Granulation Machine Used in preparation of granular Compound Fertiliser	4:
4.	Experimental Setup for Evaluation of Coated Fertiliser in the Laboratory	50
5.	Atypical of the Soil Column Used in An open Leaching Experiment in the Laboratory	5
6.	FT-IR Spectrum of the PVC Coating Material	5
7.	FT-IR Spectrum of the PA Coating Material	59
8.	FT-IR Spectrum of the NR Coating Material	60
9.	FT-IR Spectrum of the PLA Coating Material	60
10.	Distribution of N in Leachate during Leaching of Uncoated and Coated Fertiliser with Distilled Water for 30 Days	6
11.	Cumulative N in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	6′
12.	Distribution of P in Leachate during Leaching of Uncoated and Coated Fertiliser with Distilled Water for 30 Days	7
13.	Cumulative P in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	7
14.	Distribution of K in Leachate during Leaching of Uncoated and Coated Fertiliser with Distilled Water for 30 Days	74
15.	Cumulative K in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	75



## **TABLE OF CONTENTS**

	ABLESIGURES
	LATES
	T
	······································
TER	
I	INTRODUCTION
П	LITERATURE REVIEWS
	Types of Slow Release Fertiliser
	Nitrification Inhibitor-Mixed Fertilisers
	Slowly Soluble Fertilisers
	Coated Fertilisers
	Hydrophilic Polymers Control Nutrient Release
	Sources of Hydrophilic Polymers
	Natural Polymers
	Semi-synthetic Polymers
	Synthetics Polymers
	Release of Nutrients from Hydrophilic Polymers
	Effect of Soluble Salts on Polymer Performance
	Polymer Biodegradation
	Use of Polymers with Fertilisers
	Controlled-Release Nitrogen Fertilisers
	Phosphorus
	Other Nutrients
	Manufacture Process of Mixed Fertilisers
	Granulation
	Chemical Reaction
	Technology Options Available for NPK Production
	Process Description of Granulation
	Steam Granulation
	Moisture Granulation
	Speed of Drum
	Fusion Blend.
	Requirement of Pineapple Plant
	Conclusion



16.	Coated Fertiliser with distilled Water for 30 Days	77
17.	Cumulative Ca in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	77
18.	Distribution of Mg in Leachate during Leaching of Uncoated and Coated Fertiliser with distilled Water for 30 Days	78
19.	Cumulative Mg in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	<b>7</b> 9
20.	Distribution of Cu in Leachate during Leaching of Uncoated and Coated Fertiliser with Distilled Water for 30 Days	81
21.	Cumulative Cu in Leachate from Uncoated and Coated Fertilisers after Leaching with Distilled Water for 30 Days	81



## LIST OF PLATES

Plate		Page
1.	The Thickness of the PVC Coating Material of the Compound Fertiliser Using SEM	63
2.	The Thickness of the PA Coating Material of the Compound Fertiliser Using SEM	63
3.	The Thickness of the NR Coating Material of the Compound Fertiliser Using SEM	64
4.	The Thickness of the PLA Coating Material of the Compound Fertiliser Using SEM	64



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Science

DEVELOPMENT OF SLOW RELEASE FERTILISERS FOR PINEAPPLE

 $\mathbf{B}\mathbf{y}$ 

ELTAIB SAEED MOHAMED GANAWA

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Chairman: Dr Mohamed Hanafi Musa

Faculty: Agriculture

A compound fertiliser was prepared using different types of straight

fertilisers from urea (46% N), China phosphate rock (CPR) (32% P<sub>2</sub>O<sub>5</sub>), muriate of

potash (60% K<sub>2</sub>O), gypsum (31% CaO), kieserite (27% MgO), and copper sulphate

(32% CuO). Granualator machine was used in the preparation of granular

compound fertilisers. The compound fertiliser showed high percentage of K and N,

16.31 and 15.05%, respectively and between 0.091 and 4.12% for other elements.

The strength of compound fertilisers was measured immediately and after 9 months

of the preparation with the values correspond 3.5 and 0.5 Mpa, respectively.

Four types of coating materials were used in the preparation of the slow release compound fertiliser. With the exception of polylactic acid (PLA), the method used for coating of the compound fertiliser with polyacrylamide (PA), polyvinyl chloride (PVC), and natural rubber (NR) were similar. Differential scanning calorimetry (DSC) and thermogravimetry analysis (TGA) studies showed the thermal behavior of the coating materials, and also TGA indicated the temperature at which the degradation of coating materials may occurr. The thickness of different coating material was determined by scanning electronmicroscope (SEM) and the thickest value was obtained using PVC (3.04 µm).

Evaluation of the coated fertilisers was conducted in the laboratory and field. An open leaching experiment in the laboratory indicated that all coating materials were able to reduce the loss of nutrients compared to uncoat fertiliser. The best coating material (PVC) was able to release lower amount of most nutrients (about 77.16% of N added remained in the soil) followed by NR, PLA, and PA. A similar experiment as in the laboratory was performed to evaluate coated fertilisers under natural conditions in UPM Puchong farm. Open leaching experiment in the field indicated that all coating materials were able to reduce the loss of nutrients. Among the coated materials, PVC was able to reduce effectively the loss of nutrients compared to others. This is in agreement with the physico-chemical characteristic of polymer used. Total N loss in the field was about 24% higher than that in laboratory experiment. However, P, K, Ca, and Mg loss were more



pronounced in the laboratory than that in the field. The difference in the amounts of N lost for both systems is probably due to the other method of N loss in the field conditions, (such as volatilisation and denitrification of N). Copper lost in both systems was similar, this is due to the behavior of Cu reaction with the soil components.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia bagi memenuhi keperluan untuk ijazah Master Sains Pertanian

> PENYDIAAN BAJA-BAJA BEBAS PERLAHAN UNTUK TANAMAN NENAS

> > Oleh

ELTAIB SAEED MOHAMED GANAWA

Mac 1998

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Penyediaan baja sebatian dilakukan dengan menggunakan urea (46% N), batuan fosfat China (CPR) (32% P<sub>2</sub>O<sub>5</sub>), muriate potash (60% K<sub>2</sub>O), gipsum (31% CaO), kieserit (27% MgO) dan kuprum sulfat (32% CuO). Mesin pengranulan digunakan dalam penyediaan baja sebatian bergranul. Baja sebatian yang disediakan mengandungi peratus kandungan N dan K yang tinggi masing-masing 16.31 dan 15.05%, dan di antara 0.09 dan 4.12% bagi unsurunsur lain. Kekuatan fizikal baja sebatian kimia disukat serta-merta setelah

disediakan dan selepas 9 bulan dengan nilai masing-masing 3.5 dan 0.5 Mpa.

Sebanyak empat jenis bahan penyalut digunakan bagi menyediakan baja sebatian kimia bebas perlahan. Melainkan asid polilaktik, kaedah yang sama digunakan untuk menyalut baja sebatian kimia dengan poliakrilamid (PA), polivinil klorida (PVC) dan getah asli (NR). Sifat-sifat therma bahan-bahan dilakukan dengan menggunakan "kalorimetri pengimbasan penvalut pembezaan" (DSC) dan "analysis termogravimetri" (TGA), di samping itu TGA

menunjukkan suhu di mana bahan-bahan penyalut boleh terurai. Ketebalan beberapa bahan penyalut ditentukan dengan menggunakan "scanning electron microscope" (SEM) dan nilai yang paling tebal sekali diperolehi dari PVC (3.04 μM).

Penilaian baja-baja sebatian bersalut dijalankan di makmal dan ladang. Satu sistem larutlesap terbuka di makmal menunjukkan kesemua bahan penyalut boleh mengurangkan kehilangan beberapa unsur pemakanan dalam baja sebatian dibandingkan dengan tanpa bersalut. Bahan penyalut terbaik dari PVC boleh mengurangkan pembebasan unsur-unsur pemakanan paling rendah di dalam larutan (contohnya sebanyak 77.2% N masih berada di dalam tanah), kemudian diikuti oleh NR, PLA dan PA. Kajian serupa dengan di makmal dilakukan untuk menilai baja bersalut dalam keadaan semulajadi di ladang UPM, Puchong. Sistem larutlesap terbuka di ladang menunjukkan kesemua bahan penyalut dapat mengurangkan kehilangan unsur-unsur pemakanan. Di kalangan bahan-bahan penyalut, bahan PVC masih juga boleh mengurangkan kehilangan unsur-unsur pemakanan dibandingkan dengan yang lain. Ini adalah selari dengan sifat-sifat kimia fizik polimer yang digunakan. Kehilangan jumlah N di ladang adalah sekitar 24% lebih tinggi dibandingkan dengan di makmal. Walau bagaimanapun, kehilangan unsur-unsur P, K, Ca dan Mg adalah lebih ketara di makmal dibandingkan di ladang. Perbezaan dalam jumlah kehilangan N bagi kedua-dua sistem berkemungkinan disebabkan oleh cara lain bagaiman N boleh hilang di ladang, termasuk melalui pemeruapan dan denitrifikasi N. Kehilangan kuprum sulfat bagi kedua-dua sistem adalah sama dan ini mungkin disebabkan oleh tindakbalas unsur Cu dengan komponen-komponen tanah.



#### **CHAPTER I**

#### INTRODUCTION

Fertilisation is important key to successful pineapple production. The proportion of the total cost incurred by fertilisers varies depending on the intensity of cultivation and on labours. Nevertheless, it always represents a large proportion of total running costs. In Malaysia, pineapples are mainly grown on peat. Smallholders had grown pineapples on deep peat and mineral soils for many years without using fertilisers causing depletion of natural fertility of peat and soil, and consequently the yield declining. To overcome this problem various fertiliser programmes have been put forward by several researchers (Anon., 1956; Grist, 1961). Since the demand of fertiliser by pineapples is high and labour intensive, a new strategy such as slow release fertiliser technology is becoming increasingly important.

Slow-release fertilizers (SRF) are fertilizers made to release their nutrient content gradually, and if possible, to coincide with the nutrient requirements of a plant. Evaluation of the release rate and its mechanism is



essential for the selection of the proper fertilizers for a given set of condition or toward the development of proper SRF formulation. The rate of the nutrient release was described by a number of investigators as being controlled by the slow diffusion of the nutrient ions through the membrane to the soil (Ahmed *et al.*, 1963; Lunt and Orteli, 1962; Orteli and Lunt, 1962). As such, the process is deterministic and equal to all the SRF granule population. Kochba *et al.* (1990) proposed the mechanism leading to the nutrient release. This is caused by the diffusion of water vapor into the granule through the hydrophilic membrane and the subsequent bursting or expansion of the membrane, both leading to an accelerate outward flow of the saturated solution from the coated granule. Timing of the nutrient release of the individual granules was considered to be a random phenomenon described by first-order kinetics i.e., similar to radioactive decay.

A number of SRF formulations have been developed during the last decades (Hauke, 1995). This trend is motivated by the effort to produce and apply fertilisers so the crop nutrient needs are satisfied while a minimum of fertiliser salts is leached or wasted. This reasoning is valid for economic consideration (i.e., maximising utilisation of added fertiliser) and, even more, to minimising fertiliser leaching.



The use of hydrophilic polymers with fertilisers offers concept for controlling nutrient release in soil. Further work remains to be done on the exact nature of the polymer and nutrient interaction. Cooperation with polymer chemical, modification of the polymer structure should be made to optimise the nutrient release pattern to fit specific crop needs. By modifying the polymers, properties such as enhance ion retention, biodegradation rate, and adverse salts effect can be manipulated to correspond with desire nutrients release rates. The side benefits of hydrophilic polymers, such as increased moisture retention and improves physical conditions of soil need to be systematically quantified. Low cost polymers should be further explored to identify promising carriers that can be used at an affordable price. Further research is also needed on field application equipment that would allow precise placement of the nutrient containing polymers into the soil. Carefully designed field experiments are needed to determine the value of the polymer technology in improving nutrient recovery in a variety of agricultural conditions. Using polymers to supply plant nutrients may provide an additional method of providing nutrients in controlled manner to more accurately meet the crop needs while protecting the quality of the environment and conserving natural resources.

One of the main causes for nutrient losses is through leaching and this is particularly serious in sandy soil and in areas of high rainfall or intensive irrigation. Chemical instabilities, such as volatilisation of



ammonia from urea and ammonium containing fertilizers or conversion of relatively immobile  $\mathrm{NH_4}^+$  to mobile  $\mathrm{NO_3}^-$  and its subsequently denitrification to  $\mathrm{N_2}$  and  $\mathrm{N_2O}$  are other causes for nutrient loss. Therefore, losses of fertilizers whether through leaching, chemical instabilities or by any other means make their use quite costly. Repeated fertilizers are therefore necessary to replenish the losses. A single heavy application of higher than normal quantity to avoid repeated applications is not advisable as it may cause salt injury to the plant. Apart from losses, the hygroscopic nature of much conventional fertiliser creates both handling and storage problems.

Considerable efforts are being directed to minimise the losses of nutrients from water-soluble fertilizers. One approach is to develop fertilizers which will dissolve slowly enabling only a small quantity of nutrients to be released at any one time to the growing plant. This may be achieved by synthesising fertilisers with slow rate of dissolution as in the cause of condensation reaction of urea with various aldehydes (Hays and Hadden, 1969). Quastel (1965) also stated that fusing micronutrients with large amount of silicate, loss of nitrogen through denitrification could be reduced by chemically inhibiting the nitrifying organism. A third approach is to encapsulate or coat granulated fertilisers with a film of insoluble material such plastic, resin, synthetic fibre, glass, wax, gum, asphalt etc. Coated fertilisers can delay the rate of release of nutrients (Soong et al.,



1977), and have advantages over other type of SRF also can be obtained without undergoing any chemical reaction. However, because of the high cost of the coating materials most of coated fertilisers have limited commercials production and are relegated to specific uses for cultivation of turf and ornamentals. The production of SRF by the encapsulation process, although technically feasible, depends very much on the availability of effective coating materials.

### **Objectives**

The present project is focused on the development of slow release fertiliser for the pineapple and the aims of the current project are:

- To prepare compound fertilisers which is suitable for pineapple production.
- 2. To coat the compound fertiliser with different types of polymers.
- To study the performance of the coated fertiliser with polymers, using leaching technique.



#### **CHAPTER II**

#### LITERATURE REVIEW

Fertiliser burn, fast leaching, and surface runoff are several shortcoming of normal chemical fertilisers. Extensive efforts have been undertaken to search for a new type of fertiliser that can overcome these disadvantages. Since the 1960s the United States, Japan and some European countries have taken the lead in manufacturing such new fertiliser. They are chemically compound fertilisers or controlled-release fertiliser. Thus far, many of the Slow Release Fertiliser (SRF) have been manufactured and tested on various crops.

#### **Types of Slow-Release Fertiliser**

The term 'controlled-release fertiliser' or 'slow-release fertiliser' refers to a fertiliser that for any reason release its nutrient contents over an extended period or release plant nutrients slowly throughout a growing season or even several continuous growing season (Hinget, 1974). In other words, slow-release fertiliser are the ones that release their plant nutrients at a rate which possibly permits maximum plant uptake and reduces minimally nutrient loss due to leaching and surface runoff (Tajudin, 1979).

