

UNIVERSITI PUTRA MALAYSIA

DEVELOPMENT OF BIFUNCTIONAL CATALYSTS SYNTHESIZED FROM PYROLYZED AND HYDROTHERMALIZED PALM WASTE FOR BIODIESEL PRODUCTION USING WASTE COOKING OIL

ROSE FADZILAH ABDULLAH

ITMA 2022 11



DEVELOPMENT OF BIFUNCTIONAL CATALYSTS SYNTHESIZED FROM PYROLYZED AND HYDROTHERMALIZED PALM WASTE FOR BIODIESEL PRODUCTION USING WASTE COOKING OIL



By

ROSE FADZILAH ABDULLAH

Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in Fulfilment of the Requirements for the Degree of Doctor of Philosophy

December 2021

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



Dedication

They said, "Exalted are You, we have no knowledge except what You have taught us. Indeed, it is You who is the Knowing, the wise". Quran (Surah Al-Baqarah 2:32)

Mohd Zahid

Nur Aisyah Zulaikha

Abdullah & Zaleha

Shamsudin & Hamimah

Ramlah, Norazlinda & late Normarina

Family & friends

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Doctor of Philosophy

DEVELOPMENT OF BIFUNCTIONAL CATALYSTS SYNTHESIZED FROM PYROLYZED AND HYDROTHERMALIZED PALM WASTE FOR BIODIESEL PRODUCTION USING WASTE COOKING OIL

By

ROSE FADZILAH ABDULLAH

December 2021

Chairman : Umer Rashid, PhD Institute : Nanoscience and Nanotechnology

The abundance of waste biomass from palm oil industries creates a new potential for producing activated carbon as catalyst support. The hydrothermal carbonization (HTC) process provides a brilliant alternative to produce activated carbon from biomass samples with higher water content than conventional pyrolysis techniques. Therefore, this study aims to produce activated carbon from palm kernel shell (PKS), mesocarp fiber (MF) and empty fruit bunch (EFB) using pyrolysis and HTC techniques. The HTC process favours the breakage of large polymeric molecules in the lignocellulosic material into lower molecular weight organic compounds through hydrolysis, dehydration, decarboxylation, aromatization and recondensation process. Subsequently, the performance of the prepared activated carbon was determined through the application as catalyst support of a bifunctional catalyst. Therefore, the activated carbon was impregnated with K_2CO_3 and CuO via wet impregnation provided bifunctional characteristics. The state-of-the-art characterization of the synthesized bifunctional catalysts was conducted including N₂ adsorption-desorption analysis, functional group determination, surface morphology observation, crystallography study, electron dispersive X-ray mapping, elemental distribution analysis, concentration of basicity and acidity measurement and thermal degradation behaviour analysis. The results show that the HTC technique successfully produced a highly mesoporous activated carbon derived from PKS that had better textural properties than pyrolyzed with a BET surface area of 1411.16 m² g⁻¹ and 3368.60 m² g⁻¹, respectively. The produced bifunctional catalyst was applied in the simultaneous esterification and transesterification reaction from the waste cooking oil (WCO) by the conventional reflux system. The high BET surface area provides extra space for active sites impregnation which is very important for producing maximum biodiesel yield from WCO. The catalytic performance results show that the hydrothermalized-activated carbon from PKS (PKSHAC) derived catalyst produced a higher biodiesel yield of 95.3% against 95.0% over the pyrolyzed-activated carbon from PKS (PKSAC) derived catalyst. The activated carbon preparation was continued using MF and EFB, namely hydrothermalized-activated carbon from MF (MFHAC) and EFB (EFBHAC), which exhibited a BET surface area of 3909.33 m² g⁻¹ and 4056.17 m² g⁻¹, and produced a maximum biodiesel yield of 96.4% and 97.1%, respectively. The synthesized catalysts sustained up to 5 reaction cycles with more than 80% of biodiesel yield. The carbon structure collapsed due to multiple calcination during the reactivation procedure, and the catalyst poisoning by glyceroxide on the catalyst surface was identified as a deactivation factor. On the other hand, the transformation of biodiesel from WCO was confirmed via proton nuclei magnetic resonance (¹H NMR), Fourier transform Infrared (FTIR) and thermal gravimetric analysis (TGA). Fuel properties revealed kinematic viscosity of 3.3 mm² s⁻¹, the cetane number of 51, the flashpoint of 160.5 °C, cloud and pour point of 11 °C and -3 °C, respectively. Overall, the finding shows the excellent potential of waste materials, especially PKS, MF, and EFB in producing high quality activated carbon as catalyst support via pyrolysis and HTC techniques. The functionalization with K₂CO₃ and CuO via wet impregnation provides bifunctional that opens the opportunity in utilizing WCO to produce high-quality biodiesel.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

PEMBANGUNAN PEMANGKIN DWI-FUNGSI DARIPADA SISA KELAPA SAWIT SECARA PIROLISIS DAN HIDROTERMAL UNTUK PENGHASILAN BIODIESEL DARIPADA SISA MINYAK MASAK

Oleh

ROSE FADZILAH ABDULLAH

Disember 2021

Pengerusi : Umer Rashid, PhD Institut : Nanosains dan Nan<mark>oteknolo</mark>gi

Lambakan sisa biomass daripada industri kelapa sawit dilihat mempunyai potensi dalam penghasilan karbon teraktif sebagai bahan asas pemangkin. Proses karbonisasi hidrotermal (HTC) adalah alternatif yang terbaik dalam penghasilan karbon teraktif daripada sampel biomas yang mempunyai kandungan air yang tinggi berbanding teknik pirolisis konvensional. Oleh itu, kajian ini bertujuan untuk menghasilkan karbon teraktif daripada tempurung kelapa sawit (PKS), serat mesokap (MF) dan tandan buah yang telah dikosongkan (EFB) dengan menggunakan teknik pirolisis dan HTC. Proses HTC merangsang pemecahan molekul polimerik besar dalam bahan lignoselulosa menjadi sebatian organik yang lebih kecil melalui proses hidrolisis, dehidrasi, dekarboksilasi, aromatisasi dan rekondensasi. Selepas itu, prestasi karbon teraktif diaplikasikan sebagai bahan asas untuk pemangkin dwifungsi. Oleh itu, karbon teraktif diimpregnasikan dengan K₂CO₃ dan CuO melalui proses impregnasi basah yang menghasilkan pemangkin dwifungsi. Pencirian telah dilakukan terhadap pemangkin dwifungsi termasuk analisis penjerapan-penyahjerapan N_2 , penentuan kumpulan fungsi, pemerhatian morfologi, kajian kristalografi, pemetaan sinar-X, analisis taburan unsur, kepekatan bes dan asid dan analisis degradasi terma. Hasil kajian menunjukkan bahawa teknik HTC berjaya menghasilkan karbon aktif daripada PKS yang sangat mesopori dan mempunyai sifat tekstur yang lebih baik daripada pirolisis dengan masing-masing mempunyai luas permukaan BET sebanyak 1411.16 m² g⁻¹ dan 3368.60 m² g⁻¹. Pemangkin dwifungsi telah digunakan dalam reaksi esterifikasi dan transesterifikasi serentak daripada sisa minyak masak (WCO) dengan menggunakan sistem refluks konvensional. Luas permukaan BET yang tinggi memberikan ruang tambahan untuk impregnasi bahan aktif yang sangat penting untuk menghasilkan jumlah biodiesel yang maksimum daripada WCO. Hasil prestasi pemangkin menunjukkan bahawa pemangkin yang menggunakan karbon teraktif hidroterma berasaskan PKS (PKSHAC) menghasilkan biodiesel yang lebih tinggi iaitu sebanyak 95.3% berbanding 95.0% daripada pemangkin yang menggunakan karbon teraktif pirolisis (PKSAC). Penyediaan karbon teraktif kemudiannya dilanjutkan dengan menggunakan MF dan EFB, menghasilkan MFHAC dan EFBHAC, yang menunjukkan luas permukaan BET 3909.33 m² g⁻¹ dan 4056.17 m²

g⁻¹ dan masing-masing menghasilkan jumlah biodiesel yang maksimum iaitu sebanyak 96.4% dan 97.1%. Semua pemangkin yang telah disintesis dapat bertahan sehingga 5 kitaran tindak balas dengan lebih daripada 80% hasil biodiesel. Keruntuhan struktur karbon akibat daripada karbonisasi berulangan untuk pengaktifan semula pemangkin, dan keracunan oleh gliseroksida pada permukaan pemangkin telah dikenal pasti sebagai faktor penyahaktifan. Selain daripada itu, transformasi biodiesel daripada WCO telah disahkan melalui analisis ¹H NMR, FTIR dan TGA. Sifat bahan api menunjukkan kelikatan kinematik pada 3.3 mm² s⁻¹, bilangan cetane 51, titik kilat 160.5 °C, dan titik awan dan titik tuang masing-masing 11 °C dan -3 °C. Secara keseluruhan, penemuan menunjukkan potensi yang sangat baik daripada bahan buangan terutama PKS, MF dan EFB dalam menghasilkan karbon teraktif berkualiti tinggi sebagai bahan asas pemangkin melalui teknik pirolisis dan HTC. Pengungsian dengan K₂CO₃ dan Cu₀ melalui impregnasi basah memberikan sifat dwifungsi yang membuka peluang untuk memanfaatkan WCO dalam menghasilkan biodiesel yang berkualiti tinggi.



ACKNOWLEDGEMENTS

In the name of Allah, The Most Gracious, The Most Merciful.

Alhamdulillah, I would like to express my greatest gratitude to Allah s.w.t. for rewarding me the blessing, strength, spirit, and determination to finish this study.

Appreciation to UPM for providing financial assistance through the Special Graduate Research Assistantship (SGRA) and Graduate Research Fellowship (GRF) fund.

Praise Allah for appointing Ts. Dr. Umer Rashid as my supervisor. His guidance, advice, encouragement and continuous motivation kept me in the right direction until the end of this research. Special thanks are also dedicated to Prof Dr. Taufiq Yap Yun Hin and Dr. Mohd Lokman Ibrahim as my co-supervisor for their exceptional research support throughout the difficulties completing this research.

Thousand thanks to all my colleagues, staff, and technician at the Institute of Advanced Technology (ITMA) for being so accommodating in every way. Not to forget special gratitude to my friends in Catalysis Science and Technology Research Centre (PutraCat) for their invaluable and assistance in carrying out this research.

I also appreciate having my husband and daughter, father, mother and sibling, and family in-laws for their Doa' and uncountable financial and moral support. Not to mention, encouragement from my late best friend and her family has always been my motivation along the journey. This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Umer Rashid, PhD

Senior Fellow Researcher, Ts, Institute of Nanoscience and nanotechnology Universiti Putra Malaysia (Chairman)

Taufiq Yap Yun Hin, PhD

Professor Datuk ChM, Ts. Faculty of Science Universiti Putra Malaysia (Member)

Mohd Lokman Ibrahim, PhD

Senior Lecturer Faculty of Applied Science Universiti Teknologi MARA (Member)

Z<mark>ALILAH M</mark>OHD SHARIFF, PhD

Professor and Dean School of Graduate Studies Universiti Putra Malaysia

Date: 14 April 2022

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any other institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and Innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/ fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software.

Signature:

Date:

Name and Matric No.: Rose Fadzilah Abdullah, GS49854

Declaration by Members of Supervisory Committee

This is to confirm that:

- The research conducted and the writing of this thesis was under our supervision;
- Supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) are adhered to.

Signature: Name of Chairman of Supervisory Committee:	 <u>Umer Rashid</u>
Signature: Name of Member of Supervisory	Taufig Van Vun Hin
Committee:	Taung Yap Yun Hin
Signature: Name of Member of Supervisory	
Committee:	Mohd Lokman Ibrahim

TABLE OF CONTENTS

			Page
ABSTRA	СТ		i
ABSTRA	K		iii
ACKNOV	 WLEDGF	MENTS	V
APPROV	AL		vi
DECLAR	RATION		viii
LIST OF	FIGURE	8	xvi
LIST OF	TABLES	~	xxi
LIST OF	ABBREV	IATIONS	xxiii
СНАРТЕ	ER		
1	INTR	ODUCTION	1
	1.1	Background of Study	1
	1.2	Problem statement	2
	1.3	Objectives	3
	1.4	Scope	3
	1.5	Thesis outline	3
2	LITE	RATURE REVIEW	5
	2.1	Introduction	5
	2.2	Potential of biomass waste as activated carbon	5
	2.3	Availability of palm oil waste in Malaysia	5
	2.4	Technologies in activated carbon production	6
		2.4.1 Pyrolysis	6
		2.4.2 Hydrothermal carbonization (HTC)	7
		2.4.3 Combination of pyrolysis and HTC	8
	2.5	Method of carbon activation	10
	2.6	Application of catalyst in biodiesel	11
		production and techniques	
		2.6.1 Blending	12
		2.6.2 Micro emulsion	12
		2.6.3 Thermal cracking	12
		2.6.4 Esterification and transesterification	12
	2.7	Role of catalyst in esterification and	12
		transesterification reaction	
		2.7.1 Homogeneous catalyst	13
		2.7.2 Heterogeneous catalyst	14
	2.8	Potential of waste cooking oil as feedstock	14
	2.9	Various elements for bifunctional catalyst	16
	2.10	Summary of the literature review	19
3	МАТ	ERIALS AND METHODS	20
č	3.1	Materials and chemicals	20

х

 \bigcirc

3.2	Preparation of activated carbon from PKS	20
2.2	using pyrolysis techniques	21
5.5	Preparation of hydro-activated carbon from	21
	PKS, MF and EFB using HTC technique	21
	3.5.1 Preparation of PKS hydro-activated	21
	carbon	21
	3.3.2 Preparation of MF and EFB hydro-	21
2.4	activated carbon	22
3.4	Preparation of bifunctional catalyst	22
	3.4.1 Screening for the best combination	22
	of basic and acidic agent	22
	3.4.2 Optimization of basic and actuic	23
	agent for PKSAC, PKSHAC,	
25	Characterization of optalust	25
3.5	2.5.1 Ultimate analysis of CHNS	25
	2.5.2 Numbusison	23
	3.5.2 Norphological and elemental	20
	3.5.5 Morphological and elemental	20
	2.5.4 Crystallinity study	27
	3.5.5 Eunctional group determination	27
	2.5.6 Amount of basicity and acidity	27
	3.5.0 Amount of basicity and actuity	21
3.6	Waste cooking oil analysis	20
5.0	3.6.1 Acid value and FFA content	28
	3.6.2 Saponification value	20
	3.6.3 Profiling by Gas-chromatography-	29
	mass spectrometer (GC-MS)	2)
37	Experiment set up for the catalytic activity of	29
5.7	the catalyst	2)
3.8	Reusability study and characterization of	30
5.0	spent catalyst	50
3.9	Determination of biodiesel yield using gas-	30
	chromatography with flame ionization	•••
	detector (GC-FID)	
3.10	FAME profiling by Gas-chromatography-	31
	mass spectrometer (GC-MS)	
3.11	Confirmation of biodiesel production	31
	3.11.1 Confirmation 1: Functional group	31
	determination	
	3.11.2 Confirmation 2: ¹ H nuclear	31
	magnetic resonance (¹ H -NMR)	
	analysis	
	3.11.3 Confirmation 3: Thermogravimetric	31
	analysis (TGA)	
3.12	Biodiesel properties	32
	3.12.1 Density and kinematic viscosity	32
	3.12.2 Moisture content	32
	3.12.3 Flash point	32
	3.12.4 Cloud and pour point	32
	3.12.5 Cetane number	33

SYN1 FROM	THESIS OF BIFUNCTIONAL CATALYST M WASTE PALM KERNEL SHELL USING	
PVR	OI VSIS METHOD AND ITS	
	ICATION FOD DIODIESEI	
	DICTION FOR BIODIESEL	
PKU 4 1	DUCTION Introduction	
4.1	Introduction Dependence of WCO	
4.2	Catalant allowed institution and actuality	
4.3	Catalyst characterization and catalytic	
4 4	activity for screening	
4.4	Catalysi characterization of PKSAC-	
	$K_2 CO_3(x^{(k)}) CuO_{(y^{(k)})}$ catalyst	
	4.4.1 Surface characterization by N ₂	
	4.4.2 Crystallography study	
	4.4.2 Crystanography study	
	4.4.5 Allouit of basicity and actuity by	
	4.4.4 Eurotional group evaluation by	
	4.4.4 Functional group evaluation by	
	1 4 5 Morphological analysis	
	4.4.5 Morphological analysis	
	4.4.0 Elemental composition analysis	
15	Application of the catalyst in biodiesel	
ч.у	production	
16	Effect of reaction parameters	
 0	4.6.1 Effect of different catalyst loading	
	on biodiesel vield	
	4.6.2 Effect of reaction temperature on	
	hiodiesel vield	
	4.6.3 Effect of methanol to oil molar ratio	
	on biodiesel vield	
	464 Effect of reaction duration on	
4	biodiesel vield	
4.7	Reusability study and characterization of the	
	spent catalyst	
4.8	Comparison with previous study	
4.9	Summary	
BIFU	NCTIONAL CATALYST PRODUCED	
FRO	M PALM KERNEL SHELL VIA	
HYD	ROTHERMAL - ASSISTED	
CAR	BONIZATION FOR BIODIESEL	
PRO	DUCTION FROM WASTE COOKING OIL	
5.1	Introduction	
5.2	Evaluation of catalyst characterization	
	5.2.1 Surface characterization by N_2	
	adsorption-desorption	
	5.2.2 Morphological analysis	
	5.2.3 Elemental composition analysis	
	5.2.4 Crystallinity study by X-ray	
	diffraction	

	5.2.5 Functional group evaluation by FTIR	59
	5.2.6 Active sites determination by temperature programme desorption	60
	5.2.7 Thermal degradation behaviour	61
5.3	Application of the catalyst in biodiesel	62
5 1	Effect of reaction normators	62
5.4	Effect of reaction parameters	03
	on biodiesel yield	03
	5.4.2 Effect of methanol to oil molar ratio	64
	on biodiesel yield	
	5.4.3 Effect of reaction temperature on	64
	biodiesel yield	
	5.4.4 Effect of reaction duration on	64
	biodiesel yield	
5.5	Reusability study and characterization of	65
	spent catalyst	
5.6	Summary	67
BIFUN	ICTIONAL BIOMASS-BASED	68
САТА	LYST FOR BIODIESEL PRODUCTION	
VIA	HYDROTHERMAL CARBONIZATION	
(HTC)	PRETREATMENT – SYNTHESIS,	
CHAR	ACTERIZATION AND OPTIMIZATION	
6.1	Introduction	68
6.2	Evaluation of catalyst characterization	68
	6.2.1 Morphological analysis	68
	6.2.2 Textural properties	69
	6.2.3 Crystallinity profile	71
	6.2.4 Concentration of basicity and	72
	acidity	12
	6.2.5 Functional groups profile	73
	6.2.6 Degradation behavior	74
63	Application of the catalyst in biodiesel	76
0.5	production	70
6.4	Optimization of reaction parameters	77
0.4	6.4.1 Influence of experimental duration	77
	on vield	//
	6.4.2 Influence of establish amount on	77
	viald	//
	(4.2) Influence of methodel to oil molor	70
	0.4.5 Influence of methanol to off molar	/0
	ratio on yield	70
<i>.</i> -	6.4.4 Influence of temperature on yield	/8
6.5	Reusability study and characterization of the	/9
	spent catalyst	
6.6	Reaction mechanism of esterification and	81
	transesterification reactions over MFHAC-	
	4K2CO3/CuO catalyst	
6.7	Summary	82

POT	ENTIAL HETEROGENEOUS CATALYST	83
VIA	INTEGRATING HYDROTHERMAL	
CAR	BONIZATION FOR BIODIESEL	
PRO	DUCTION USING WASTE COOKING OIL	
7.1	Introduction	83
7.2	Evaluation of catalyst characterization	83
	7.2.1 Surface characterization by N_2	83
	adsorption-desorption and particle	
	size distribution	
	7.2.2 Morphological analysis	86
	7.2.3 Elemental composition analysis	87
	7.2.4 Active sites determination by	88
	temperature programmed	
	desorption	
	7.2.5 Crystallinity study by X-ray	89
	diffraction	
	7.2.6 Functional group evaluation by	90
	FTIR	
7.3	Application of the catalyst in biodiesel	91
	production	
7.4	Effect of reaction parameters	92
	7.4.1 Effect of different catalyst loading	92
	on biodiesel yield	
	7.4.2 Effect of methanol to oil molar ratio	93
	on biodiesel yield	
	7.4.3 Effect of reaction temperature on	93
	biodiesel yield	
	7.4.4 Effect of reaction duration on	93
	biodiesel yield	
7.5	Reusability study and characterization of the	94
	spent catalyst	
7.6	Summary	95
BIOL	DIESEL ASSESSMENT AND PROPERTIES	96
8.1	Introduction	96
8.2	Determination of biodiesel yield using Gas	96
	chromatography-flame ionization detector	
	(GC-FID) analysis	
8.3	FAME profiling by Gas chromatography –	98
	mass spectrometer (GC-MS) analysis	
8.4	Confirmation 1: ¹ H nuclear magnetic	98
	resonance (¹ H-NMR) analysis	
8.5	Confirmation 2: Fourier transform infrared	99
	spectroscopy (FTIR) analysis	
8.6	Confirmation 3: Thermogravimetric analysis	100
	(TGA)	
8.7	Biodiesel properties	101
8.8	Summary	104

9	SUM REC	MARY, GENERAL CONCLUSION AND OMMENDATIONS	105
	9.1	Summary and general conclusion	105
	9.2	Recommendations for Future Work	106
REFE	RENCES		107
APPEN	NDICES		128
BIODA	ATA OF ST	UDENT	132
LIST C	OF PUBLIC	CATIONS	133



 \bigcirc

LIST OF FIGURES

Figure		Page
2.1	General mechanism of (a) esterification reaction by acid catalyst and (b) transesterification reaction by basic catalyst. Adapted from Aransiola et al., (2014)	14
2.2	General mechanism of simultaneous esterification and transesterification by the bifunctional catalyst. Adapted from Faruque et al., (2020).	16
3.1	Schematic diagram of the autoclave reactor.	22
3.2	Schematic flow diagram of bifunctional catalyst preparation using pyrolyzed activated carbon.	23
3.3	Schematic flow diagram of bifunctional catalyst preparation using hydrothermalized activated carbon.	25
3.4	Schematic diagram of the conventional reflux system for esterification/transesterification reaction.	30
4.1	TGA profile for PKSAC-KOH _(30%) CuO _(5%) , PKSAC- $K_{2CO3(30%)}$ CuO _(5%) , PKSAC-K ₃ PO _{4(30%)} CuO _(5%) and PKSAC-NaOH _(30%) CuO _(5%) .	36
4.2	Nitrogen adsorption-desorption isotherms of the raw PKS, PKSAC and PKSAC derived catalysts.	38
4.3	Pore diameter distribution of the raw PKS, PKSAC and PKSAC derived catalysts.	38
4.4	XRD diffractogram for PKSAC and PKSAC derived catalysts.	39
4.5	FTIR spectra for PKSAC and PKSAC derived catalysts.	41
4.6	FESEM images for (a) raw PKS, (b) PKSAC, (c) PKSAC-K ₂ CO _{3(20%)} CuO(5%), (d) PKSAC-K ₂ CO _{3(30%)} CuO(5%), (e) PKSAC-K ₂ CO _{3(40%)} CuO(5%) and (f) PKSAC-K ₂ CO _{3(50%)} CuO(5%) analysed at 20,000x magnification.	42
4.7	TGA profile of PKSAC and PKSAC- K2CO3(30%)CuO(5%) catalyst.	44

4.8	(a) Effect of K ₂ CO ₃ wt.% with Cu(NO ₃) ₂ fixed at 5 wt.% for transesterification-esterification of WCO and (b) Effect of Cu(NO ₃) ₂ wt.% with K ₂ CO ₃ fixed at 30 wt.% for transesterification-esterification of WCO. Reaction condition: 5 wt.% catalyst loading, 15:1 methanol-to-oil ratio at 70 °C for 4 h.	45
4.9	The effect of (a) catalyst loading (methanol to oil molar ratio of 15:1 at 70 °C for 4 h) (b) temperature (catalyst loading of 5 wt.%, methanol to oil molar ratio of 15:1 for 4 h) (c) methanol to oil molar ratio (catalyst loading of 5 wt.% at 80 °C for 4 h) (d) duration on the FAME yield percentage (catalyst loading of 5 wt.%, methanol to oil molar ratio of 12:1 at 80 °C) on FAME yield.	47
4.10	(a) The effect of reusability study on the FAME yield percentage. Reaction condition; catalyst loading of 5 wt.%, methanol to oil molar ratio of 12:1, 80 °C for 4 h (b) XRD for fresh and spent PKSAC-K2CO3(30%)CuO(5%) catalyst (c) FTIR for fresh and spent PKSAC- K2CO3(30%)CuO(5%) catalyst.	49
5.1	N2 adsorption-desorption isotherm curves of (a) raw PKS (b) PKSAC (c) PKSHC (d) PKSHAC (e) PKSHAC-K2CO3(10%)CuO(5%) (f) PKSHAC- K2CO3(20%)CuO(5%) (g) PKSHAC-K2CO3(30%)CuO(5%) (h) PKSHAC-K2CO3(40%)CuO(5%).	55
5.2	Pore volume distribution of raw PKS, PKSAC, PKSHC, PKSHAC and PKSHAC derived catalysts.	56
5.3	FESEM images of (a) PKSHC (b) PKSHAC (c) PKSHAC-K2CO3(10%)CuO(5%), (d) PKSHAC- K2CO3(20%)CuO(5%), (e) PKSHAC-K2CO3(30%)CuO(5%), (f) PKSHAC-K2CO3(40%)CuO(5%) captures at 50,000x magnification.	57
5.4	EDX-mapping of PKSHAC-K2CO3(20%)CuO(5%).	58
5.5	XRD diffractogram profiles of PKSHAC and PKSHAC derived catalyst.	59
5.6	FTIR spectrum of PKSHAC and PKSHAC derived catalysts.	60
5.7	TGA/DTG thermogram of (a) PKSHAC and (b) PKSHAC-K2CO3(20%)CuO(5%) catalyst.	62
5.8	Effect of (a) K ₂ CO ₃ loading (wt.%) on the catalyst surface and (b) CuO from Cu(NO ₃) ₂ loading (wt.%) on the catalyst surface on FAME yield. Reaction condition:	63

	5 wt.% catalyst loading, 15:1 methanol to oil molar ratio at 70 °C for 4 h.	
5.9	The effect of (a) catalyst loading (methanol to oil molar ratio of 15:1 at 70 °C for 4 h) (b) reaction temperature (catalyst loading of 4 wt.%, methanol to oil molar ratio of 15:1 for 4 h) (c) methanol to oil molar ratio (catalyst loading of 4 wt.% at 70 °C for 4 h) and (d) duration (catalyst loading of 4 wt.%, methanol to oil molar ratio of 12:1 at 70 °C) on FAME yield over PKSHAC-K ₂ CO _{3(20%)} CuO _(5%) catalyst.	65
5.10	(a) Reusability study on biodiesel production by using PKSHAC-K ₂ CO _{3(20%)} CuO(5%) (Reaction condition: catalyst loading of 4 wt.%, methanol to oil molar ratio of 12:1 at 70 °C for 2 h), (b) FTIR of fresh PKSHAC-K ₂ CO _{3(20%)} CuO(5%) and spent PKSHAC-K ₂ CO _{3(20%)} CuO(5%) catalyst.	66
6.1	FESEM images of (a) raw MF, (b) MFHAC, (c) MFHAC-2K ₂ CO ₃ /CuO, (d) MFHAC-4K ₂ CO ₃ /CuO, (e) MFHAC-6K ₂ CO ₃ /CuO and (f) MFHAC-8K ₂ CO ₃ /CuO catalysts captured at 20,000x magnification.	69
6.2	Isotherm profile of (a) raw MF, (b) MFHAC (c) MFHAC-2K ₂ CO ₃ /CuO, (d) MFHAC-4K ₂ CO ₃ /CuO, (e) MFHAC-6K ₂ CO ₃ /CuO and (f) MFHAC-8K ₂ CO ₃ /CuO catalysts.	71
6.3	Distribution of pore volume of raw MF, MFHAC and impregnated MFHAC catalysts.	71
6.4	XRD diffractogram of MFHAC and impregnated MFHAC catalysts.	72
6.5	TGA/DTG profiles of (a) raw MF, (b) MFHAC and (c) MFHAC-4K ₂ CO ₃ /CuO catalysts.	76
6.6	Effect of different K ₂ CO ₃ to Cu(NO ₃) ₂ wt.% ratio on biodiesel yield. Reaction condition: 3 wt.% of catalyst loading, 9:1 of methanol to oil molar ratio at 60 °C for 2 h.	77
6.7	Influence of (a) experimental duration (3 wt.% catalyst, 9:1 molar ratio of methanol to WCO at 60 °C), (b) catalyst loading (9:1 of methanol to WCO molar ratio for 2 h at 60 °C), (d) methanol to WCO molar ratio (5 wt.% catalyst for 2 h at 60 °C), and (d) temperature (5 wt.% catalyst, 12:1 molar ratio of methanol to oil for 2 h) over MFHAC-4K ₂ CO ₃ /CuO catalyst on biodiesel yield.	79

6.8	(a) The results on reusability study on biodiesel production by using MFHAC-4K ₂ CO ₃ /CuO catalyst (5 wt.% of catalyst loading, 12:1 of methanol to WCO molar ratio at 70 °C for 2 h), (b) FESEM images of fresh MFHAC-4K ₂ CO ₃ /CuO and (c) FESEM images of spent MFHAC-4K ₂ CO ₃ /CuO catalyst captured at 20,000x magnifications.	80
6.9	Proposed mechanism of (a) esterification and (b) transesterification reaction of WCO over MFHAC-4K2CO3/CuO catalyst. Modified from Mansir et al., (2017b)	82
7.1	N ₂ adsorption-desorption isotherms of the (a) raw EFB, EFBH, EFBP and EFBHAC and (b) EFBHAC and EFBHAC derived catalysts.	85
7.2	Morphology structure photographs with magnification of 20,000x magnificent of (a) raw EFB (b) close-up image of raw EFB (c) untreated EFB char (d) treated EFB char with H ₃ PO4 (e) EFBHAC-K ₂ CO ₃ (10%)CuO(5%) (f) EFBHAC-K ₂ CO ₃ (20%)CuO(5%) (g) EFBHAC- K ₂ CO ₃ (30%)CuO(5%) (h) EFBHAC-K ₂ CO ₃ (40%)CuO(5%) catalysts.	87
7.3	Relationship between BET surface area with the amount of basicity and acidity for each catalyst	89
7.4	XRD pattern of EFBHAC and EFBHAC derived catalysts	90
7.5	FTIR spectrum of EFBHAC and EFBHAC derived catalysts	90
7.6	Effect of (a) K_2CO_3 and (b) CuO percentage on the catalyst surface on the FAME yield. The experiment was carried out at reaction condition of 5 wt.% of catalyst loading, 15:1 of methanol to oil molar ratio at 60 °C for 4 h.	92
7.7	The effect of (a) catalyst loading (15:1 of methanol to oil molar ratio at 60 °C for 4 h) (b) methanol to oil molar ratio (5 wt.% of catalyst loading at 60 °C for 4 h) (c) reaction temperature (5 wt.% of catalyst loading, 12:1 of methanol to oil molar ratio for 4 h) and (d) duration (5 wt.% of catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C) on the FAME yield.	94
7.8	Reusability cycle of EFBHAC-K2CO3(20%)CuO(5%) catalyst under optimized condition of 5 wt.% of catalyst	95

	loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h, and its relationship with BET surface area and amount of basicity and acidity.	
8.1	GC-FID chromatogram of (a) internal and reference standard and (b) produced biodiesel over EFBHAC- K ₂ CO _{3(20%)} CuO _(5%) catalyst with the reaction condition of 5 wt.% of the catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h.	97
8.2	GC-MS chromatogram of the produced biodiesel over EFBHAC-K ₂ CO _{3(20%)} CuO(5%) catalyst with the reaction condition of 5 wt.% of the catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h.	98
8.3	¹ H NMR spectrum of the produced biodiesel over EFBHAC-K ₂ CO ₃ (20%)CuO(5%) catalyst with the reaction condition of 5 wt.% of the catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h.	99
8.4	FTIR of (a) WCO and (b) produced biodiesel over EFBHAC-K ₂ CO ₃ (20%)CuO(5%) catalyst with the reaction condition of 5 wt.% of the catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h.	100
8.5	Thermal composition of WCO and produced biodiesel over EFBHAC-K ₂ CO _{3(20%)} CuO(5%) catalyst with the reaction condition of 5 wt.% of the catalyst loading, 12:1 of methanol to oil molar ratio at 70 °C for 2 h.	101

LIST OF TABLES

	Table		Page
	2.1	Comparison study for process technique and condition to produce activated carbon.	9
	2.2	Production of biodiesel from WCO using various catalysts.	17
	2.3	Catalysts for esterification-transesterification of high FFA feedstock.	18
	4.1	Properties and fatty acid profile of WCO.	34
	4.2	Physicochemical properties of the PKSAC and PKSAC derived catalysts.	36
	4.3	Physicochemical properties of raw PKS, PKSAC and PKSAC derived catalysts.	37
	4.4	Physicochemical properties of raw PKS, PKSAC and PKSAC derived catalysts.	40
	4.5	List of functional groups available on the raw PKS, PKSAC and PKSAC derived catalysts.	41
	4.6	Elemental composition of raw PKS, PKSAC and PKSAC derived catalysts.	43
	4.7	TPD results for the fresh PKSAC-K ₂ CO ₃ (30%)CuO(5%) and spent PKSAC-K ₂ CO ₃ (30%)CuO(5%) catalyst after the sixth reaction cycle.	50
	4.8	Comparison of the catalytic activities of the previously synthesized catalysts.	50
	5.1	Ultimate analysis of raw PKS, PKSHC and PKSHAC.	53
	5.2	Physicochemical properties of raw PKS, PKSAC, PKSHC, PKSHAC and PKSHAC derived catalysts.	54
	5.3	List of functional groups available on the raw PKS, PKSHAC and PKSHAC derived catalysts.	60
	5.4	Amount of CO ₂ and NH ₃ desorbed of PKSHAC and PKSHAC derived catalysts.	61
	5.5	Amount of CO ₂ and NH ₃ desorbed of fresh and spent PKSHAC-K ₂ CO _{3(20%)} CuO _(5%) catalyst.	67

6.1	Physicochemical properties of raw MF, MFHAC and impregnated MFHAC catalysts.				
6.2	Concentration of basicity and acidity of MFHAC and 73 impregnated MFHAC catalysts.				
6.3	List of functional groups available on the raw MF, 74 MFHAC and impregnated MFHAC catalysts.				
6.4	Comparison of BET surface area, BJH pore volume, BJH pore diameter, concentration of basicity and acidity of fresh and spent MFHAC-4K ₂ CO ₃ /CuO catalyst.				
7.1	Physicochemical properties of the raw EFB, EFBH, EFBP, EFBHAC and EFBHAC derived catalysts.	85			
7.2	Elemental composition of EFBHAC and EFBHAC derived catalysts.	87			
7.3	Amount of CO ₂ and NH ₃ desorbed of EFBHAC and 88 EFBHAC derived catalysts.				
7.4	List of functional groups available on the raw EFB, EFBHAC and EFBHAC derived catalysts.	91			
8.1	Methyl ester components of biodiesel derived from the WCO using GC-FID.	97			
8.2	Comparison of fuel properties of the produced biodiesel from WCO over EFBHAC-K ₂ CO _{3(20%)} CuO _(5%) catalyst with previous studies according to ASTM D6751.	103			

 $\left[\mathbf{G} \right]$

LIST OF ABBREVIATIONS

¹ H-NMR	Hydrogen-1 nuclei magnetic resonance
BET	Brunauer-Emmett-Teller
ВЈН	Barret-Joyner-Halenda
EDX	Energy dispersive X-ray
EFB	Empty fruit bunch
EFBH	EFB-hydrochar
EFBHAC	Empty fruit bunch hydro-activated carbon
FAME	Fatty acid methyl ester
FESEM	Field emission scanning electron microscopy
FFA	Free fatty acid
FTIR	Fourier transform infrared spectroscopy
GC-FID	Gas chromatography - flame ionization detector
HTC	Hydrothermal carbonization
MF	Mesocarp fiber
MFHAC	Mesocarp fiber hydro activated carbon
PKS	Palm kernel shell
PKSHAC	Palm kernel shell hydro activated carbon
PKSHC	Palm kernel shell hydro carbon
TGA/DTG	Thermogravimetric analysis/ differential thermogravimetric
TPD-CO ₂	Temperature programmed desorption carbon dioxide
TPD-NH ₃	Temperature programmed desorption ammonia
WCO	Waste cooking oil
XRD	X-ray diffraction
	IH-NMRBETBJHEDXEFBEFBHACFAMEFESEMFFAFTIRGC-FIDHTCMFMFHACPKSHACPKSHACTGA/DTGTD-CO2TPD-NH3WCOXRD

CHAPTER 1

INTRODUCTION

1.1 Background of study

Biodiesel is one of the best alternatives for conventional fossil fuels and its use is growing all over the world. In conjunction with global economic evaluation and the expansion of the automobile industry sectors, researchers consistently provide the most remarkable discoveries and inventions to produce alternatives to fossil fuel (F. Martins et al., 2019). Biodiesel has gained popularity worldwide because it is renewable and energy efficient as it is primarily made from crop waste and recycled resources. Biodiesel was found a promising new source of energy as it is highly biodegradable and could reduce global warming gas emissions as it emits lesser hydrocarbon (HC), carbon monoxide (CO), 10 diesel engines with no or only minor modifications. The potential of waste cooking oil (WCO) was discovered as the best candidate for biodiesel production in this study. The non-value by-product from daily cooking and frying activities reported that 15 Mt had been disposed of annually worldwide (A. F. Lee et al., 2014a). Most of the time, the WCO is drained off into the drainage system which will cause water pollution. This situation has caused difficulties for authorities in waste management, especially when dealing with WCO. Thus, utilizing WCO for biodiesel production in extent will help to reduce the cost of waste management.

Theoretically, biodiesel was produced via esterification or transesterification reaction. Each reaction requires a catalyst to accelerate the reaction. The acidic catalyst was involved in the esterification reaction of feedstock with a high content of free fatty acid (FFA), while the basic catalyst facilitates the transesterification reaction of feedstock with high content percentages of triglycerides. The development of a catalyst for biodiesel production shows an excellent opportunity for activated carbon as catalyst support. The use of biomass-derived activated carbon for heterogeneous catalysts seems to be a promising option as it eliminates the tediousness and problems faced by homogeneous reactions. In addition, biomass is ubiquitous in nature, wide distribution, CO_2 neutrality and economic feasibility (Ma et al., 2012). Therefore, the synthesizing activated carbon originating from palm oil biomass such as palm kernel shell (PKS), empty fruit bunch (EFB) and mesocarp fiber (MF) provides a promising opportunity in the utilization of waste material into valuable activated carbon as catalyst support for biodiesel production.

The introduction of the hydrothermal carbonization (HTC) process provides a brilliant alternative in producing activated carbon from high moisture biomass. Instead of drying the materials, the HTC operated in the presence of subcritical water conditions with autogenous pressure in the range of 2-10 Mpa in the sealed reactor, thus promoting dissolution of lignocellulosic components into the hot water (He et al., 2013). The HTC technique involves hydrolysis, dehydration, decarboxylation, aromatization, recondensation, and other complicated unknown processes to produce biochar, namely

hydrochar, bio-oil, and gases at the end of the process (Fang et al., 2018; Kalderis et al., 2014). The subcritical water accelerates the large polymeric molecules contained in the lignocellulosic materials into lower molecular weight organic compounds (Khan et al., 2019). Other than that, the HTC process has an advantage in producing 70-80% of yield and the produced hydrochar exhibited high aromaticity structure with a higher content of oxygen-containing functional groups (Yihunu et al., 2019).

1.2 Problem Statement

Nowadays, the biodiesel production process suffers from technological challenges in producing highly effective catalysts to accommodate feasibility and sustainable development. Homogeneous catalyst is widely used by industry nowadays cause several technical problems such as reaction corrosion due to high acidity/basicity, generate large amounts of waste water via product washing and acid neutralization step, separation problems caused by unfavourable aqueous emulsion; which thereby increase the overall cost of biodiesel production (Farooq, Ramli, Naeem, & Saleem khan, 2016; Mardhiah et al., 2017). In contrast, a heterogeneous catalyst has the ability to be recovered and reused provides excellent economic benefits to the product liability (Meher et al., 2013). Other than that, different heterogeneous catalysts such as zirconia, ferric alginate, titanium dioxide, silica nanoparticles, and others require expensive materials, have complicated synthesis routes, and record low catalytic activity. To overcome this problem, an activated carbon was prepared using waste biomass as a catalyst support.

Conventionally, the activated carbon was produced by the pyrolysis technique. The pyrolysis techniques require high temperature up to 700 °C in an inert atmosphere to produce activated carbon. As a result, the produced activated carbon, namely biochar, will have high BET surface area and porosity. However, the pyrolysis technique requires an energy-intensive drying process and is inconvenient for biomass-based material as it contains high moisture content. The reaction condition is also critical as the oxygen should be avoided throughout the process by providing inert gas such as nitrogen; otherwise, it will produce ash instead of activated carbon. Besides that, pyrolysis emits toxic gasses during carbonization, requiring specific safety precautions to handle the emission. In this work, the environmentally friendly of HTC process was introduced to skip the drying process of the biomass material by using water as solvent to produce hydrochar that has been used as catalyst support to produce biodiesel.

Biodiesel was formerly produced from edible oils creates unhealthy competition between food and biodiesel crops. The used of inedible oil of WCO become a challenge as the it contains of high triglycerides (85%), FFA (15%), moisture and solid residue. Previously, the biodiesel production from WCO was catalyzed with a homogeneous catalyst by two-step esterification-transesterification, where esterification is to reduce the FFA before transesterification reaction. Thus, the two-step reaction has been double the production cost and time and it needs a neutralization step to the treatment system before it is safe for the environment (Baroutian et al., 2010). Therefore, the acid-base bifunctional catalyst has been synthesized to catalyze both transesterification and esterification reactions simultaneously to fully utilize the oil under mild conditions exhibits great potential for industrial application.

1.3 Objectives

This study aims to produce biomass based bifunctional catalyst for biodiesel production from WCO. The aim will be supported with the following objectives:

- i. to discover potential of palm waste in producing activated carbon for catalyst support using pyrolysis technique,
- ii. to access capability of HTC technique in producing activated carbon for catalyst support,
- iii. to determine the physical and chemical properties of the bifunctional catalyst for simultaneous esterification-transesterification reaction, and
- iv. to identify the deactivation factor that caused the deterioration of biodiesel yield that effect the reusability of the bifunctional catalyst.

1.4 Scope

This study was focused on the utilization of PKS, EFB and MF in producing activated carbon via pyrolysis and HTC techniques. Initially, the activated carbon was produced from PKS using pyrolysis and HTC techniques. A comparison of its physical and chemical properties was done to choose the best technique to produce high-quality activated carbon. Later, both of the produced activated carbon were doped with potassium carbonate (K_2CO_3) and copper nitrate ($Cu(NO_3)_2$) via wet impregnation followed by calcination to produced bifunctional catalyst and tested its catalytic performance. Subsequently, after considering best results, the catalysts derived from EFB, and MF were prepared using the HTC technique. All catalysts were subjected to detailed characterizations to assess their surface characteristics, morphological and elemental analysis, amount of basicity and acidity, crystallinity, surface functional group and thermal degradation behaviour. The optimum loading of K₂CO₃ and Cu(NO₃)₂ were determined by various percentages loading onto the activated carbon to find the best combination as a bifunctional catalyst for biodiesel production using a conventional reflux system for simultaneous esterification and transesterification reaction of WCO. The reaction condition of catalyst loading, methanol to oil molar ratio, temperature and reaction duration were optimized through one-factor-at-a-time techniques. Afterwards, the detailed reusability study was carried out using the best catalyst and the properties of the spent catalyst were characterized to discover the deactivation factor. This study also includes a quality assessment of the produced biodiesel to confirm the conversion of WCO to biodiesel and was further compared with ASTM D-6751.

1.5 Thesis Outline

This thesis contains nine chapters that include four chapters based on published work. Chapter 1 provides the background of this study and the deliberation of problem statements that initiate the study. This chapter also highlights the aim and objectives to solve the problems and scope of the study. Chapter 2 discusses palm oil-based waste's potential as a source of activated carbon with its availability in Malaysia and technologies in producing the activated carbon. Afterward, the application of the homogeneous and heterogeneous catalysts in biodiesel production was reviewed and quality standards associated with biodiesel were listed. Chapter 3 started with a list of chemicals and materials used in this study. It also delivers a methodology covering the preparation of activated carbon, synthesizing bifunctional catalyst, and illustrates the experiment set up for the esterification-transesterification reaction. Other than that, the details of catalysts characterization were included. Additionally, analyses relating to biodiesel assessment and properties were inserted. Chapter 4, 5, 6 and 7 present the potential of PKS, MF and EFB as catalyst support of bifunctional catalysts in biodiesel production. These chapters include the characterizations of the synthesized catalysts, the experiment set up for application in producing biodiesel from WCO, the optimization of the reaction condition, the reusability study, and its characterization. Chapter 8 focuses on the biodiesel evaluation and deliberation of its properties. Chapter 9 summarizes by highlighting the main contribution and significant findings of this study and recommendations for future investigation.

REFERENCES

- Abdelhady, H. H., Elazab, H. A., Ewais, E. M., Saber, M., & El-Deab, M. S. (2020). Efficient catalytic production of biodiesel using nano-sized sugar beet agroindustrial waste. *Fuel*, 261(August 2019), 116481.
- Abdollahi Asl, M., Tahvildari, K., & Bigdeli, T. (2020). Eco-friendly synthesis of biodiesel from WCO by using electrolysis technique with graphite electrodes. *Fuel*, 270(August 2019), 117582.
- Abduh, M. Y., Syaripudin, Putri, L. W., & Manurung, R. (2019). Effect of storage time on moisture content of Reutealis trisperma seed and its effect on acid value of the isolated oil and produced biodiesel. *Energy Reports*, 5, 1375–1380.
- Abdulkareem-Alsultan, G., Asikin-Mijan, N., Lee, H. V., & Taufiq-Yap, Y. H. (2016). A new route for the synthesis of La-Ca oxide supported on nano activated carbon via vacuum impregnation method for one pot esterification-transesterification reaction. *Chemical Engineering Journal*, 304, 61–71.
- Abdullah, N. H., Hasan, S. H., & Yusoff, N. R. M. (2013). Biodiesel production based on waste cooking oil (WCO). *International Journal of Materials Science and Engineering*, 1(2), 94–99.
- Abu-Jrai, A. M., Jamil, F., Al-Muhtaseb, A. H., Baawain, M., Al-Haj, L., Al-Hinai, M., Al-Abri, M., & Rafiq, S. (2017). Valorization of waste Date pits biomass for biodiesel production in presence of green carbon catalyst. *Energy Conversion and Management*, 135, 236–243.
- AbuKhadra, M. R., Basyouny, M. G., El-Sherbeeny, A. M., El-Meligy, M. A., & Abd Elgawad, A. E. E. (2020). Transesterification of commercial waste cooking oil into biodiesel over innovative alkali trapped zeolite nanocomposite as green and environmental catalysts. *Sustainable Chemistry and Pharmacy*, 17(April).
- Abukhadra, M. R., & Sayed, M. A. (2018). K+ trapped kaolinite (Kaol/K+) as low cost and eco-friendly basic heterogeneous catalyst in the transesterification of commercial waste cooking oil into biodiesel. *Energy Conversion and Management*, 177(July), 468–476.
- Adinata, D., Wan Daud, W. M. A., & Aroua, M. K. (2007). Preparation and characterization of activated carbon from palm shell by chemical activation with K2CO3. *Bioresource Technology*, 98(1), 145–149.
- Afolalu, S. A., Adejuyigbe, S. B., Adetunji, O. R., & Olusola, O. I. (2015). Production of cutting tools from recycled steel with palm kernel shell as carbon additives. *International Journal of Innovation and Applied Studies*, *12*(1), 110–122.
- Ahmad Farid, M. A., Hassan, M. A., Taufiq-Yap, Y. H., Ibrahim, M. L., Othman, M. R., Ali, A. A. M., Shirai, Y., Abdillah, M., Farid, A., Ali, M., Hin, Y., Lokman, M., Ridzuan, M., Amiruddin, A., & Ali, M. (2017). Production of methyl esters from waste cooking oil using a heterogeneous biomass-based catalyst. *Renewable Energy*, 114, 638–643.

Ahmad, J., Rashid, U., Patuzzi, F., Alamoodi, N., Choong, T. S. Y., Soltani, S.,

Ngamcharussrivichai, C., Nehdi, I. A., & Baratieri, M. (2020). Mesoporous acidic catalysts synthesis from dual-stage and rising co-current gasification char: Application for fame production from waste cooking oil. *Materials*, *13*(4), 871.

- Ahmadpour, A., & Do, D. D. (1997). The preparation of activated carbon from macadamia nutshell by chemical activation. *Carbon*, 35(12), 1723–1732.
- Akbarzadeh, O., Mohd Zabidi, N. A., Abdul Wahab, Y., Hamizi, N. A., Chowdhury, Z. Z., Merican Aljunid Merican, Z., Rahman, M. A., Akhter, S., Shalauddin, M., & Johan, M. R. (2019). Effects of cobalt loading, particle size, and calcination condition on Co/CNT catalyst performance in Fischer-Tropsch reactions. *Symmetry*, 11(1), 1–18.
- Akinfalabi, S.-I. I., Rashid, U., Yunus, R., Hin Taufiq-Yap, Y., & Taufiq-Yap, Y. H. (2017). Synthesis of biodiesel from palm fatty acid distillate using sulfonated palm seed cake catalyst. *Renewable Energy*, 111, 611–619.
- Al-Muhtaseb, A. H., Jamil, F., Al-Haj, L., Zar Myint, M. T., Mahmoud, E., Ahmad, M. N. M., Hasan, A. O., & Rafiq, S. (2018). Biodiesel production over a catalyst prepared from biomass-derived waste date pits. *Biotechnology Reports*, 20.
- Al-Muhtaseb, A. H., Osman, A. I., Jamil, F., Al-Riyami, M., Al-Haj, L., Alothman, A. A., Htet Kyaw, H., Tay Zar Myint, M., Abu-Jrai, A., & Kumar Ponnusamy, V. (2020). Facile technique towards clean fuel production by upgrading waste cooking oil in the presence of a heterogeneous catalyst. *Journal of King Saud University Science, xxxx*, 1–7.
- Al-Saadi, A., Mathan, B., & He, Y. (2020). Esterification and transesterification over SrO–ZnO/Al2O3 as a novel bifunctional catalyst for biodiesel production. *Renewable Energy*, 158, 388–399.
- Alexandre-franco, M., & Ferna, C. (2016). Activated carbon surface chemistry: Changes upon impregnation with Al (III), Fe (III) and Zn (II) -metal oxide catalyst precursors from NO À 3 aqueous solutions. 3.
- Alhassan, F. H., Rashid, U., & Taufiq-Yap, Y. H. (2015). Synthesis of waste cooking oil-based biodiesel via effectual recyclable bi-functional Fe2O3MnOSO42-/ZrO2 nanoparticle solid catalyst. *Fuel*, 142, 38–45.
- Ali, B., Yusup, S., Quitain, A. T., Alnarabiji, M. S., Kamil, R. N. M., & Kida, T. (2018). Synthesis of novel graphene oxide/bentonite bi-functional heterogeneous catalyst for one-pot esterification and transesterification reactions. *Energy Conversion and Management*, 171(April), 1801–1812.
- Alias, N. I., Kumar, J., Jayakumar, A. / L., & Zain, S. M. (2018). Characterization of waste cooking oil for biodiesel production (Pencirian sisa minyak masak untuk penghasilan biodisel). Jurnal Kejuruteraan SI, 1(2), 79–83.
- Aliasa, N., Ahmad Zaini, M. A., & Kamaruddin, M. J. (2017). Roles of impregnation ratio of K2CO3 and NaOH in chemical activation of palm kernel shell. *Journal of Applied Science & Process Engineering*, 4(2), 195–204.
- Alonso, D. M., Mariscal, R., Moreno-Tost, R., Poves, M. D. Z., & Granados, M. L. (2007). Potassium leaching during triglyceride transesterification using K/γ-

Al2O3 catalysts. Catalysis Communications, 8(12), 2074–2080.

- Alsultan, A., Mijan, A., & Yap, T. (2016). Preparation of activated carbon from walnut shell doped La and Ca catalyst for biodiesel production from waste cooking oil. *Materials Science Forum*, 840(3), 348–352.
- Alsultan, G. A., Asikin-Mijan, N., Lee, H. V., Albazzaz, A. S., & Taufiq-Yap, Y. H. (2017). Deoxygenation of waste cooking to renewable diesel over walnut shellderived nanorode activated carbon supported CaO-La2O3 catalyst. *Energy Conversion and Management*, 151, 311–323.
- Amani, H., Ahmad, Z., Asif, M., & Hameed, B. H. (2014). Transesterification of waste cooking palm oil by MnZr with supported alumina as a potential heterogeneous catalyst. *Journal of Industrial and Engineering Chemistry*, 20(6), 4437–4442.
- Ambat, I., Srivastava, V., Haapaniemi, E., & Sillanpää, M. (2019). Nano-magnetic potassium impregnated ceria as catalyst for the biodiesel production. *Renewable Energy*, 139, 1428–1436.
- Amelkovich, Y. A., Nazarenko, O. B., Sechin, A. I., & Visakh, P. M. (2015). Characterization of copper nanopowders after natural aging. *IOP Conference Series: Materials Science and Engineering*, 81(1).
- Aransiola, E. F., Ojumu, T. V., Oyekola, O. O., Madzimbamuto, T. F., & Ikhu-Omoregbe, D. I. O. (2014). A review of current technology for biodiesel production: State of the art. *Biomass and Bioenergy*, 61(January), 276–297.
- Ashok, A., Kumar, A., Bhosale, R. R., Saleh, M. A. H., & Van Den Broeke, L. J. P. (2015). Cellulose assisted combustion synthesis of porous Cu-Ni nanopowders. *RSC Advances*, *5*(36), 28703–28712.
- Atadashi, I. M., Aroua, M. K., Abdul Aziz, A. R., & Sulaiman, N. M. N. (2013). The effects of catalysts in biodiesel production: A review. *Journal of Industrial and Engineering Chemistry*, 19(1), 14–26.
- Ayoob, A. K., & Fadhil, A. B. (2019). Biodiesel production through transesterification of a mixture of non-edible oils over lithium supported on activated carbon derived from scrap tires. *Energy Conversion and Management*, 201(October), 112149.
- Aziz, M., Kurniawan, T., Oda, T., & Kashiwagi, T. (2017). Advanced power generation using biomass wastes from palm oil mills. *Applied Thermal Engineering*, 114, 1378–1386.
- Aziz, N. A. M., Yunus, R., Hamid, H. A., Ghassan, A. A. K., Omar, R., Rashid, U., & Abbas, Z. (2020). An acceleration of microwave-assisted transesterification of palm oil-based methyl ester into trimethylolpropane ester. *Scientific Reports*, 10(1).
- Azlina Wan Ab Karim Ghani, W., Salleh, A., Zalikha Rebitanim, N., Khalid Mahmoud, D., Akmal Rebitanim, N., & Amran Mohd Salleh, M. (2012). Adsorption capacity of raw empty fruit bunch biomass onto methylene blue dye in aqueous solution. *Journal of Purity, Utility Reaction and Environment, 1*(August 2015), 45–60. https://www.researchgate.net/publication/267844981

- Babel, S., Arayawate, S., Faedsura, E., & Sudrajat, H. (2018). Microwave assisted transesterification of waste cooking oil for biodiesel production. *Utilization and Management of Bioresources*, 165–174.
- Baby, R., Saifullah, B., & Hussein, M. Z. (2019). Palm kernel shell as an effective adsorbent for the treatment of heavy metal contaminated water. *Scientific Reports*, 9(1), 1–11.
- Baroutian, S., Aroua, M. K., Raman, A. A. A., & Sulaiman, N. M. N. (2010). Potassium hydroxide catalyst supported on palm shell activated carbon for transesterification of palm oil. *Fuel Processing Technology*, 91(11), 1378–1385.
- Baskar, G., Aberna Ebenezer Selvakumari, I., & Aiswarya, R. (2018). Biodiesel production from castor oil using heterogeneous Ni doped ZnO nanocatalyst. *Bioresource Technology*, 250, 793–798.
- Basso, D., Weiss-Hortala, E., Patuzzi, F., Castello, D., Baratieri, M., & Fiori, L. (2015). Hydrothermal carbonization of off-specification compost: A byproduct of the organic municipal solid waste treatment. *Bioresource Technology*, 182, 217–224.
- Becker, R., Dorgerloh, U., Paulke, E., Mumme, J., & Nehls, I. (2014). Hydrothermal carbonization of biomass: Major organic components of the aqueous phase. *Chemical Engineering and Technology*, 37(3), 511–518.
- Berge, N. D., Ro, K. S., Mao, J., Flora, J. R. V., Chappell, M. A., & Bae, S. (2011). Hydrothermal carbonization of municipal waste streams. *Environmental Science* and Technology, 45(13), 5696–5703.
- Bhatia, S. K., Gurav, R., Choi, T. R., Kim, H. J., Yang, S. Y., Song, H. S., Park, J. Y., Park, Y. L., Han, Y. H., Choi, Y. K., Kim, S. H., Yoon, J. J., & Yang, Y. H. (2020). Conversion of waste cooking oil into biodiesel using heterogenous catalyst derived from cork biochar. *Bioresource Technology*, 302(January), 122872.
- Bilgic, E., Yaman, S., Haykiri-Acma, H., & Kucukbayrak, S. (2016). Is torrefaction of polysaccharides-rich biomass equivalent to carbonization of lignin-rich biomass? *Bioresource Technology*, 200, 201–207.
- Bilgin, A., & Gulum, M. (2018). Effects of various transesterification parameters on the some fuel properties of hazelnut oil methyl ester. *Energy Procedia*, 147, 54–62.
- Borah, M. J., Das, A., Das, V., Bhuyan, N., & Deka, D. (2019). Transesterification of waste cooking oil for biodiesel production catalyzed by Zn substituted waste egg shell derived CaO nanocatalyst. *Fuel*, 242(January), 345–354.
- Borhan, A., Abdullah, N. A., Rashidi, N. A., & Taha, M. F. (2016). Removal of Cu2+ and Zn2+ from single metal aqueous solution using rubber-seed shell based activated carbon. *Procedia Engineering*, *148*, 694–701.
- Boz, N., Degirmenbasi, N., & Kalyon, D. M. (2013). Transesterification of canola oil to biodiesel using calcium bentonite functionalized with K compounds. *Applied Catalysis B: Environmental*, 138–139, 236–242.
- Cazetta, A. L., Vargas, A. M. M., Nogami, E. M., Kunita, M. H., Guilherme, M. R., Martins, A. C., Silva, T. L., Moraes, J. C. G., & Almeida, V. C. (2011). NaOH-

activated carbon of high surface area produced from coconut shell: Kinetics and equilibrium studies from the methylene blue adsorption. *Chemical Engineering Journal*, 174(1), 117–125.

- Changmai, B., Rano, R., Vanlalveni, C., & Rokhum, L. (2021). A novel Citrus sinensis peel ash coated magnetic nanoparticles as an easily recoverable solid catalyst for biodiesel production. *Fuel*, 286(P2), 119447.
- Chen, C., Qu, S., Guo, M., Lu, J., Yi, W., Liu, R., & Ding, J. (2021). Waste limescale derived recyclable catalyst and soybean dregs oil for biodiesel production: Analysis and optimization. *Process Safety and Environmental Protection*, 149, 465–475.
- Chen, W., Chen, Y., Yang, H., Li, K., Chen, X., & Chen, H. (2018). Investigation on biomass nitrogen-enriched pyrolysis: Influence of temperature. *Bioresource Technology*, 249(October 2017), 247–253.
- Chen, X., Li, Z., Chun, Y., Yang, F., Xu, H., & Wu, X. (2020). Effect of the formation of diglycerides/monoglycerides on the kinetic curve in oil transesterification with methanol catalyzed by calcium oxide. ACS Omega, 5(9), 4646–4656.
- Cheng, F., & Li, X. (2018). Preparation and Application of Biochar-Based Catalysts for Biofuel Production. *Catalysts*, 8(9), 346.
- Chowdhury, Z. Z., Hamid, S. B. A., Das, R., Hasan, M. R., Zain, S. M., Khalid, K., & Uddin, M. N. (2013). Preparation of carbonaceous adsorbents from lignocellulosic biomass and their use in removal of contaminants from aqueous solution. *BioResources*, 8(4), 6523–6555.
- Conte, P., Bertani, R., Sgarbossa, P., Bambina, P., Schmidt, H. P., Raga, R., Lo Papa, G., Chillura Martino, D. F., & Lo Meo, P. (2021). Recent developments in understanding biochar's physical-chemistry. *Agronomy*, 11(4).
- Daligaux, V., Richard, R., & Manero, M.-H. (2021). Deactivation and regeneration of zeolite catalysts used in pyrolysis of plastic wastes—A process and analytical review. *Catalysts*, 11(7), 770.
- Dantas, J., Leal, E., Cornejo, D. R., Kiminami, R. H. G. A., & Costa, A. C. F. M. (2018). Biodiesel production evaluating the use and reuse of magnetic nanocatalysts Ni0.5Zn0.5Fe2O4 synthesized in pilot-scale. *Arabian Journal of Chemistry*.
- Degfie, T. A., Mamo, T. T., & Mekonnen, Y. S. (2019). Optimized biodiesel production from waste cooking oil (WCO) using calcium oxide (CaO) nano-catalyst. *Scientific Reports*, 9(1), 1–8.
- Dehghani, S., & Haghighi, M. (2020). Sono-enhanced dispersion of CaO over Zr-Doped MCM-41 bifunctional nanocatalyst with various Si/Zr ratios for conversion of waste cooking oil to biodiesel. *Renewable Energy*, 153, 801–812.
- Deris, N. H., Rashid, U., Soltani, S., Choong, T. S. Y., & Nehdi, I. A. (2020). Study the effect of various sulfonation methods on catalytic activity of carbohydrate-derived catalysts for ester production. *Catalysts*, *10*(6), 1–13.
- di Bitonto, L., Reynel-Ávila, H. E., Mendoza-Castillo, D. I., Bonilla-Petriciolet, A.,

Durán-Valle, C. J., & Pastore, C. (2020). Synthesis and characterization of nanostructured calcium oxides supported onto biochar and their application as catalysts for biodiesel production. *Renewable Energy*, *160*, 52–66.

- Diaz, E., Manzano, F. J., Villamil, J., Rodriguez, J. J., & Mohedano, A. F. (2019). Lowcost activated grape seed-derived hydrochar through hydrothermal carbonization and chemical activation for sulfamethoxazole adsorption. *Applied Sciences* (*Switzerland*), 9(23).
- dos Santos, L. K., Hatanaka, R. R., de Oliveira, J. E., & Flumignan, D. L. (2019). Production of biodiesel from crude palm oil by a sequential hydrolysis/esterification process using subcritical water. *Renewable Energy*, 130, 633–640.
- Du, L., Ding, S., Li, Z., Lv, E., Lu, J., & Ding, J. (2018). Transesterification of castor oil to biodiesel using NaY zeolite-supported La2O3 catalysts. *Energy Conversion and Management*, 173(August), 728–734.
- El-Hendawy, A. N. A., Samra, S. E., & Girgis, B. S. (2001). Adsorption characteristics of activated carbons obtained from corncobs. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 180(3), 209–221.
- Elias, S., Rabiu, A. M., Okeleye, B. I., Okudoh, V., & Oyekola, O. (2020). Bifunctional heterogeneous catalyst for biodiesel production from waste vegetable oil. *Applied Sciences (Switzerland)*, 10(9).
- Elkelawy, M., Alm-Eldin Bastawissi, H., Esmaeil, K. K., Radwan, A. M., Panchal, H., Sadasivuni, K. K., Ponnamma, D., & Walvekar, R. (2019). Experimental studies on the biodiesel production parameters optimization of sunflower and soybean oil mixture and DI engine combustion, performance, and emission analysis fueled with diesel/biodiesel blends. *Fuel*, 255(June), 115791.
- Erchamo, Y. S., Mamo, T. T., Workneh, G. A., & Mekonnen, Y. S. (2021). Improved biodiesel production from waste cooking oil with mixed methanol-ethanol using enhanced eggshell-derived CaO nano-catalyst. *Scientific Reports*, 11(1), 6708.
- Escobar, J. C., Lora, E. S., Venturini, O. J., Yáñez, E. E., Castillo, E. F., & Almazan, O. (2009). Biofuels: Environment, technology and food security. *Renewable and Sustainable Energy Reviews*, 13(6–7), 1275–1287.
- Falowo, O. A., Ojumu, T. V., Pereao, O., & Betiku, E. (2020). Sustainable biodiesel synthesis from honne-rubber-neem oil blend with a novel mesoporous base catalyst synthesized from a mixture of three agrowastes. *Catalysts*, *10*(2).
- Falowo, O. A., Oloko-Oba, M. I., & Betiku, E. (2019). Biodiesel production intensification via microwave irradiation-assisted transesterification of oil blend using nanoparticles from elephant-ear tree pod husk as a base heterogeneous catalyst. *Chemical Engineering and Processing - Process Intensification*, 140(January), 157–170.
- Fang, J., Zhan, L., Ok, Y. S., & Gao, B. (2018). Minireview of potential applications of hydrochar derived from hydrothermal carbonization of biomass. *Journal of Industrial and Engineering Chemistry*, 57, 15–21.

- Farabi, M. S. A., Ibrahim, M. L., Rashid, U., & Taufiq-Yap, Y. H. (2019). Esterification of palm fatty acid distillate using sulfonated carbon-based catalyst derived from palm kernel shell and bamboo. *Energy Conversion and Management*, 181(September 2018), 562–570.
- Farooq, M., Ramli, A., Naeem, A., & Saleem khan, M. (2016). Effect of different metal oxides on the catalytic activity of γ-Al 2 O 3 –MgO supported bifunctional heterogeneous catalyst in biodiesel production from WCO. *RSC Adv.*, 6(2), 872– 881.
- Farooq, M., Ramli, A., Naeem, A., & Saleem Khan, M. (2016a). Effect of different metal oxides on the catalytic activity of γ-Al2O3-MgO supported bifunctional heterogeneous catalyst in biodiesel production from WCO. *RSC Advances*, *6*(2), 505–512.
- Farooq, M., Ramli, A., Naeem, A., & Saleem Khan, M. (2016b). Effect of different metal oxides on the catalytic activity of γ -Al2O3-MgO supported bifunctional heterogeneous catalyst in biodiesel production from WCO. *RSC Advances*, 6(2), 505–512.
- Farooq, M., Ramli, A., & Subbarao, D. (2013). Biodiesel production from waste cooking oil using bifunctional heterogeneous solid catalysts. *Journal of Cleaner Production*, 59, 131–140.
- Faruque, M. O., Razzak, S. A., & Hossain, M. M. (2020). Application of heterogeneous catalysts for biodiesel production from microalgal oil—A review. *Catalysts*, 10(9), 1–25.
- Furusawa, T. (2017). Biodiesel fuel production using CaO-loaded alginate capsules. Journal of the Japan Petroleum Institute, 60(4), 170–185.
- Gamal, M. S., Asikin-Mijan, N., Khalit, W. N. A. W., Arumugam, M., Izham, S. M., & Taufiq-Yap, Y. H. (2020). Effective catalytic deoxygenation of palm fatty acid distillate for green diesel production under hydrogen-free atmosphere over bimetallic catalyst CoMo supported on activated carbon. *Fuel Processing Technology*, 208(July).
- García-Martín, J. F., Alés-Álvarez, F. J., López-Barrera, M. del C., Martín-Domínguez, I., & Álvarez-Mateos, P. (2019). Cetane number prediction of waste cooking oilderived biodiesel prior to transesterification reaction using near infrared spectroscopy. *Fuel*, 240(November 2018), 10–15.
- Garcia, E., Junior, S., Rodriguez, O., Haber, V., Melo, S., Reyero, I., Serrano-lotina, A., & Mompean, F. J. (2020). Biodiesel synthesis using a novel monolithic catalyst with magnetic properties (K 2 CO 3 / γ -Al 2 O 3 / Sepiolite / γ -Fe 2 O 3) by ethanolic route. *Fuel*, 271(November 2019), 117650.
- Gardy, J., Hassanpour, A., Lai, X., Ahmed, M. H., & Rehan, M. (2017). Biodiesel production from used cooking oil using a novel surface functionalised TiO2nano-catalyst. *Applied Catalysis B: Environmental*, 207, 297–310.
- Gebremariam, S. N., & Marchetti, J. M. (2017). Biodiesel production technologies: Review. In *AIMS Energy* (Vol. 5, Issue 3).

- Gebremariam, S. N., & Marchetti, J. M. (2021). Biodiesel production process using solid acid catalyst: influence of market variables on the process's economic feasibility. *Biofuels, Bioproducts and Biorefining*, 15(3), 815–824.
- Giakoumis, E. G., & Sarakatsanis, C. K. (2018). Estimation of biodiesel cetane number, density, kinematic viscosity and heating values from its fatty acid weight composition. *Fuel*, 222(January), 574–585.
- Glorius, M., Markovits, M. A. C., & Breitkopf, C. (2018). Design of specific acid-baseproperties in CeO2-ZrO2-mixed oxides via templating and Au modification. *Catalysts*, 8(9), 1–25.
- Gollakota, A. R. K., Volli, V., & Shu, C. M. (2019). Transesterification of waste cooking oil using pyrolysis residue supported eggshell catalyst. *Science of the Total Environment*, 661, 316–325.
- Gónzalez, J. P. C., Gutiérrez, P. E. Á., Medina, M. A., Zapata, B. Y. L., Guerrero, G. V. R., & Valdés, L. G. V. (2020). Effects on biodiesel production caused by feed oil changes in a continuous stirred-tank reactor. *Applied Sciences (Switzerland)*, 10(3).
- Guo, S., Dong, X., Liu, K., Yu, H., & Zhu, C. (2015). Chemical, energetic, and structural characteristics of hydrothermal carbonization solid products for lawn grass. *BioResources*, 10(3), 4613–4625.
- Hanisah, Kumar S, & Ay, T. (2013). The Management of Waste Cooking Oil: A Preliminary Survey. In *Health and the Environment Journal* (Vol. 4, Issue 1).
- Hasnain, S. M. M., & Sharma, R. P. (2019). Evaluation of the performance and emission and spectroscopic analysis of an improved soy methyl ester. *RSC Advances*, 9(46), 26880–26893.
- Hayashi, H., & Hakuta, Y. (2010). Hydrothermal Synthesis of Metal Oxide Nanoparticles in Supercritical Water. 3794–3817.
- He, C., Giannis, A., & Wang, J. Y. (2013). Conversion of sewage sludge to clean solid fuel using hydrothermal carbonization: Hydrochar fuel characteristics and combustion behavior. *Applied Energy*, 111, 257–266.
- Heidari, A., Younesi, H., Rashidi, A., & Ghoreyshi, A. A. (2014). Evaluation of CO2 adsorption with eucalyptus wood based activated carbon modified by ammonia solution through heat treatment. *Chemical Engineering Journal*, *254*, 503–513.
- Heidarinejad, Z., Dehghani, M. H., Heidari, M., Javedan, G., Ali, I., & Sillanpää, M. (2020). Methods for preparation and activation of activated carbon: a review. *Environmental Chemistry Letters*, 18(2), 393–415.
- Hindryawati, N., Maniam, G. P., Karim, M. R., & Chong, K. F. (2014). Transesterification of used cooking oil over alkali metal (Li, Na, K) supported rice husk silica as potential solid base catalyst. *Engineering Science and Technology, an International Journal*, 17(2), 95–103.
- Huang, H. J., Yuan, X. Z., Li, B. T., Xiao, Y. D., & Zeng, G. M. (2014). Thermochemical liquefaction characteristics of sewage sludge in different organic solvents. *Journal*

of Analytical and Applied Pyrolysis, 109, 176-184.

- Hurst, K. E., Heben, M. J., Blackburn, J. L., Gennett, T., Dillon, A. C., & Parilla, P. A. (2013). A dynamic calibration technique for temperature programmed desorption spectroscopy. *Review of Scientific Instruments*, 84(2), 0–9.
- Ibrahim, M. L., Nik Abdul Khalil, N. N. A., Islam, A., Rashid, U., Ibrahim, S. F., Sinar Mashuri, S. I., & Taufiq-Yap, Y. H. (2020). Preparation of Na2O supported CNTs nanocatalyst for efficient biodiesel production from waste-oil. In *Energy Conversion and Management* (Vol. 205).
- Ibrahim, N. A., Rashid, U., Taufiq-Yap, Y. H., Yaw, T. C. S., & Ismail, I. (2019). Synthesis of carbonaceous solid acid magnetic catalyst from empty fruit bunch for esterification of palm fatty acid distillate (PFAD). *Energy Conversion and Management*, 195, 480–491.
- Ibrahim, S. F., Asikin-Mijan, N., Ibrahim, M. L., Abdulkareem-Alsultan, G., Izham, S. M., & Taufiq-Yap, Y. H. (2020). Sulfonated functionalization of carbon derived corncob residue via hydrothermal synthesis route for esterification of palm fatty acid distillate. *Energy Conversion and Management*, 210(March).
- Illán-Gómez, M. J., García-García, A., Salinas-Martinez De Lecea, C., & Linares-Solano, A. (1996). Activated carbons from Spanish coals. 2. Chemical activation. *Energy and Fuels*, 10(5), 1108–1114.
- Intarapong, P., Jindavat, C., Luengnaruemitchai, A., & Jai-In, S. (2014). The transesterification of palm oil using KOH supported on bentonite in a continuous reactor. *International Journal of Green Energy*, *11*(9), 987–1001.
- Islam, A., Taufiq-Yap, Y. H., Chan, E. S., Moniruzzaman, M., Islam, S., & Nabi, M. N. (2014). Advances in solid-catalytic and non-catalytic technologies for biodiesel production. *Energy Conversion and Management*, 88, 1200–1218.
- Islam, M. A., Ahmed, M. J., Khanday, W. A., Asif, M., & Hameed, B. H. (2017). Mesoporous activated carbon prepared from NaOH activation of rattan (Lacosperma secundiflorum) hydrochar for methylene blue removal. *Ecotoxicology and Environmental Safety*, 138(August 2016), 279–285.
- Jain, A., Balasubramanian, R., & Srinivasan, M. P. (2016). Hydrothermal conversion of biomass waste to activated carbon with high porosity: A review. *Chemical Engineering Journal*, 283(December 2017), 789–805.
- Jain, A., Jayaraman, S., Balasubramanian, R., & Srinivasan, M. P. (2014). Hydrothermal pre-treatment for mesoporous carbon synthesis: Enhancement of chemical activation. *Journal of Materials Chemistry A*, 2(2), 520–528.
- Jain, S., & Sharma, M. P. (2010). Kinetics of acid base catalyzed transesterification of Jatropha curcas oil. *Bioresource Technology*, *101*(20), 7701–7706.
- Jamil, F., Murphin Kumar, P. S., Al-Haj, L., Tay Zar Myint, M., & Al-Muhtaseb, A. H. (2020). Heterogeneous carbon-based catalyst modified by alkaline earth metal oxides for biodiesel production: Parametric and kinetic study. *Energy Conversion* and Management: X, March, 100047.

- Januszewicz, K., Kazimierski, P., Klein, M., Kardaś, D., & Łuczak, J. (2020). Activated carbon produced by pyrolysis of waste wood and straw for potential wastewater adsorption. *Materials*, 13(9), 19–288.
- Jiang, X., Xu, W., Liu, W., Yue, M., Zhu, Y., & Yang, M. (2018). Facile preparation of cuprous oxide decorated mesoporous carbon by one-step reductive decomposition for deep desulfurization.
- Joshi, G., Rawat, D. S., Lamba, B. Y., Bisht, K. K., Kumar, P., Kumar, N., & Kumar, S. (2015). Transesterification of Jatropha and Karanja oils by using waste egg shell derived calcium based mixed metal oxides. *Energy Conversion and Management*, 96, 258–267.
- Joshi, S., Gogate, P. R., Moreira, P. F., & Giudici, R. (2017). Intensification of biodiesel production from soybean oil and waste cooking oil in the presence of heterogeneous catalyst using high speed homogenizer. *Ultrasonics Sonochemistry*, 39, 645–653.
- Kalderis, D., Kotti, M. S., Méndez, A., & Gascó, G. (2014). Characterization of hydrochars produced by hydrothermal carbonization of rice husk. *Solid Earth*, 5(1), 477–483.
- Kambo, H. S., & Dutta, A. (2015). A comparative review of biochar and hydrochar in terms of production, physico-chemical properties and applications. In *Renewable and Sustainable Energy Reviews*.
- Kang, S., Li, X., Fan, J., & Chang, J. (2012). Characterization of hydrochars produced by hydrothermal carbonization of lignin, cellulose, d-xylose, and wood meal. *Industrial and Engineering Chemistry Research*, 51(26), 9023–9031.
- Kang, S., Li, X., Fan, J., & Chang, J. (2016). A direct synthesis of adsorbable hydrochar by hydrothermal conversion of lignin. *Energy Sources, Part A: Recovery, Utilization and Environmental Effects*, 38(9), 1255–1261.
- Kapadia, H., Brahmbhatt, H., Dabhi, Y., & Chourasia, S. (2019). Investigation of emulsion and effect on emission in CI engine by using diesel and bio-diesel fuel: A review. *Egyptian Journal of Petroleum*, 28(4), 323–337.
- Karmakar, A., Karmakar, S., & Mukherjee, S. (2010). Properties of various plants and animals feedstocks for biodiesel production. *Bioresource Technology*, 101(19), 7201–7210.
- Karmakar, R., Kundu, K., & Rajor, A. (2018). Fuel properties and emission characteristics of biodiesel produced from unused algae grown in India. *Petroleum Science*, 15(2), 385–395.
- Kazmierczak-Razna, J., Nowicki, P., Wiśniewska, M., Nosal-Wiercińska, A., & Pietrzak, R. (2017). Thermal and physicochemical properties of phosphoruscontaining activated carbons obtained from biomass. *Journal of the Taiwan Institute of Chemical Engineers*, 80, 1006–1013.
- Kefas, H. M., Yunus, R., Rashid, U., & Taufiq-Yap, Y. H. (2018). Modified sulfonation method for converting carbonized glucose into solid acid catalyst for the esterification of palm fatty acid distillate. *Fuel*, 229(May), 68–78.

- Khan, T. A., Saud, A. S., Jamari, S. S., Rahim, M. H. A., Park, J. W., & Kim, H. J. (2019). Hydrothermal carbonization of lignocellulosic biomass for carbon rich material preparation: A review. *Biomass and Bioenergy*, 130(September), 105384.
- Kheang, L. S., May, C. Y., Foon, C. S., & Ngan, M. A. (2006). Recovery and conversion of palm olein-derived used frying oil to methyl esters for biodiesel. *Journal of Oil Palm Research*, 18, 247–252.
- Kim, Y., Thomas, A. E., Robichaud, D. J., Iisa, K., St. John, P. C., Etz, B. D., Fioroni, G. M., Dutta, A., McCormick, R. L., Mukarakate, C., & Kim, S. (2020). A perspective on biomass-derived biofuels: From catalyst design principles to fuel properties. *Journal of Hazardous Materials*, 400(April), 123198.
- Koberg, M., & Gedanken, A. (2013). Using microwave radiation and SrO as a catalyst for the complete conversion of oils, cooked oils, and microalgae to biodiesel. In *New and Future Developments in Catalysis: Catalytic Biomass Conversion*. Elsevier B.V.
- Kolhe, N. S., Gupta, A. R., & Rathod, V. K. (2017). Production and purification of biodiesel produced from used frying oil using hydrodynamic cavitation. *Resource-Efficient Technologies*, 3, 198–203.
- Krisnangkura, K., & Simamaharnnop, R. (1992). Continuous transmethylation of palm oil in an organic solvent. *Journal of the American Oil Chemists' Society*, 69(2), 166–169.
- Kumar, A., Osembo, S., Namango, S., & Kiriamiti, K. (2012). Heterogeneous basic catalysts for tansesterification of vegetable oils: A review. *Mechanical Engineering Conference on Sustainable Research and Innovation*, 4(May), 59–68.
- Kumar Sangar, S., Syazwani, O. N., Farabi, M. S. A., Razali, S. M., Shobhana, G., Teo, H., & Hin Taufiq-Yap, Y. (2019). Effective biodiesel synthesis from palm fatty acid distillate (PFAD) using carbon-based solid acid catalyst derived glycerol.
- Lee, A. F., Bennett, J. A., Manayil, J. C., & Wilson, K. (2014a). Heterogeneous catalysis for sustainable biodiesel production via esterification and transesterification. *Chemical Society Reviews*, 43(22), 7887–7916.
- Lee, A. F., Bennett, J. A., Manayil, J. C., & Wilson, K. (2014b). Heterogeneous catalysis for sustainable biodiesel production *via* esterification and transesterification. *Chem. Soc. Rev.*, 43(22), 7887–7916.
- Lee, C. L., H'ng, P. S., Paridah, M. T., Chin, K. L., Rashid, U., Maminski, M., Go, W. Z., Nazrin, R. A. R., Rosli, S. N. A., & Khoo, P. S. (2018). Production of bioadsorbent from phosphoric acid pretreated palm kernel shell and coconut shell by two-stage continuous physical activation via N₂ and air. *Royal Society Open Science*, *5*(12), 180775.
- Lee, J., Kim, K. H., & Kwon, E. E. (2017). Biochar as a Catalyst. *Renewable and Sustainable Energy Reviews*, 77(February), 70–79.
- Leggieri, P. A., Senra, M., & Soh, L. (2018). Cloud point and crystallization in fatty acid ethyl ester biodiesel mixtures with and without additives. *Fuel*, 222(September 2017), 243–249.

- Li, H., Niu, S., Lu, C., & Wang, Y. (2015). Comprehensive investigation of the thermal degradation characteristics of biodiesel and its feedstock oil through TGA-FTIR. *Energy and Fuels*, 29(8), 5145–5153.
- Li, Z., Ding, S., Chen, C., Qu, S., Du, L., Lu, J., & Ding, J. (2019). Recyclable Li/NaY zeolite as a heterogeneous alkaline catalyst for biodiesel production: Process optimization and kinetics study. *Energy Conversion and Management*, *192*(April), 335–345.
- Libra, J. A., Ro, K. S., Kammann, C., Funke, A., Berge, N. D., Neubauer, Y., Titirici, M.-M., Fuhner, C., Bens, O., Kern, J., & Emmerich, K.-H. (2017). *Hydrothermal* carbonization of biomass residuals: A comparative review of the chemistry, processes and apllication of wet and dry pyrolysis. 2(January 2011), 71–106.
- Libra, J. A., Ro, K. S., Kammann, C., Funke, A., Berge, N. D., Neubauer, Y., Titirici, M. M., Fühner, C., Bens, O., Kern, J., & Emmerich, K. H. (2011). Hydrothermal carbonization of biomass residuals: A comparative review of the chemistry, processes and applications of wet and dry pyrolysis. *Biofuels*, 2(1), 71–106.
- Lillo-Ródenas, M. A., Cazorla-Amorós, D., & Linares-Solano, A. (2003). Understanding chemical reactions between carbons and NaOH and KOH: An insight into the chemical activation mechanism. *Carbon*, 41(2), 267–275.
- Liu, D., Xu, B., Zhu, J., Tang, S., Xu, F., Li, S., Jia, B., & Chen, G. (2020). Preparation of highly porous graphitic activated carbon as electrode materials for supercapacitors by hydrothermal pretreatment-assisted chemical activation. ACS Omega, 5(19), 11058–11067.
- Liu, W. J., Jiang, H., & Yu, H. Q. (2015). Development of Biochar-Based Functional Materials: Toward a Sustainable Platform Carbon Material. In *Chemical Reviews*.
- Liu, W., Yin, P., Liu, X., & Qu, R. (2014). Design of an effective bifunctional catalyst organotriphosphonic acid-functionalized ferric alginate (ATMP-FA) and optimization by Box-Behnken model for biodiesel esterification synthesis of oleic acid over ATMP-FA. *Bioresource Technology*, *173*, 266–271.
- Liu, Z., Zhang, F. S., & Wu, J. (2010). Characterization and application of chars produced from pinewood pyrolysis and hydrothermal treatment. *Fuel*, 89(2), 510– 514.
- Lozano-Castelló, D., Lillo-Ródenas, M. A., Cazorla-Amorós, D., & Linares-Solano, A.
 (2001a). Preparation of activated carbons from Spanish anthracite. *Carbon*, 39(5), 741–749.
- Lozano-Castelló, D., Lillo-Ródenas, M. A., Cazorla-Amorós, D., & Linares-Solano, A. (2001b). Preparation of activated carbons from Spanish anthracite - I. Activation by KOH. *Carbon*, 39(5), 741–749.
- Lu, X., Flora, J. R. V., & Berge, N. D. (2014). Influence of process water quality on hydrothermal carbonization of cellulose. *Bioresource Technology*, 154, 229–239.
- Luo, J., Kamasamudram, K., Currier, N., & Yezerets, A. (2018). NH3-TPD methodology for quantifying hydrothermal aging of Cu/SSZ-13 SCR catalysts. *Chemical Engineering Science*, 190, 60–67.

- M. Ali, R. (2015). Preparation and characterization of CaSO4–SiO2–CaO/SO42composite for biodiesel production. *American Journal of Applied Chemistry*, 3(3), 38.
- Ma, H., Li, S., Wang, B., Wang, R., & Tian, S. (2008). Transesterification of rapeseed oil for synthesizing biodiesel by K/KOH/γ-Al2O3 as heterogeneous base catalyst. *JAOCS, Journal of the American Oil Chemists' Society*, 85(3), 263–270.
- Ma, Z., Chen, D., Gu, J., Bao, B., & Zhang, Q. (2015). Determination of pyrolysis characteristics and kinetics of palm kernel shell using TGA-FTIR and model-free integral methods. *Energy Conversion and Management*, 89, 251–259.
- Madhu, D., & Sharma, Y. C. (2017). Synthesis of a reusable novel catalyst (β-tricalcium phosphate) for biodiesel production from a common Indian tribal feedstock. *Resource-Efficient Technologies*, *3*(2), 144–157.
- Madzaki, H., Ghani, W. A. W. A. K., Yaw, T. C. S., Rashid, U., & Muda, N. (2018). Carbon dioxide adsorption on activated carbon hydrothermally treated and impregnated with metal oxides. *Jurnal Kejuruteraan*, 30(1), 31–38.
- Małecka, B., Łącz, A., Drozdz, E., & Małecki, A. (2015). Thermal decomposition of dmetal nitrates supported on alumina. *Journal of Thermal Analysis and Calorimetry*, 119(2), 1053–1061.
- Malins, K. (2018). The potential of K3PO4, K2CO3, Na3PO4 and Na2CO3 as reusable alkaline catalysts for practical application in biodiesel production. *Fuel Processing Technology*, *179*(July), 302–312.
- Mansir, N., Hwa Teo, S., Lokman Ibrahim, M., & Yun Hin, T. Y. (2017). Synthesis and application of waste egg shell derived CaO supported W-Mo mixed oxide catalysts for FAME production from waste cooking oil: Effect of stoichiometry. *Energy Conversion and Management*, 151(August), 216–226.
- Mansir, N., Taufiq-Yap, Y. H., Rashid, U., & Lokman, I. M. (2017). Investigation of heterogeneous solid acid catalyst performance on low grade feedstocks for biodiesel production: A review. *Energy Conversion and Management*, 141(December), 171–182.
- Mansir, N., Teo, S. H., Rabiu, I., & Taufiq-Yap, Y. H. (2018). Effective biodiesel synthesis from waste cooking oil and biomass residue solid green catalyst. *Chemical Engineering Journal*, 347(April), 137–144.
- Mansir, N., Teo, S. H., Rashid, U., & Taufiq-Yap, Y. H. (2018). Efficient waste Gallus domesticus shell derived calcium-based catalyst for biodiesel production. *Fuel*, 211(November 2017), 67–75.
- Marchetti, J. M., Miguel, V. U., & Errazu, A. F. (2008). Techno-economic study of different alternatives for biodiesel production. *Fuel Processing Technology*, 89(8), 740–748.
- Mardhiah, H. H., Ong, H. C., Masjuki, H. H., Lim, S., & Lee, H. V. (2017). A review on latest developments and future prospects of heterogeneous catalyst in biodiesel production from non-edible oils. *Renewable and Sustainable Energy Reviews*, 67, 1225–1236.

- Martins, A. C., Pezoti, O., Cazetta, A. L., Bedin, K. C., Yamazaki, D. A. S., Bandoch, G. F. G., Asefa, T., Visentainer, J. V., & Almeida, V. C. (2015). Removal of tetracycline by NaOH-activated carbon produced from macadamia nut shells: Kinetic and equilibrium studies. *Chemical Engineering Journal*, 260, 291–299.
- Martins, F., Felgueiras, C., Smitkova, M., & Caetano, N. (2019). Analysis of fossil fuel energy consumption and environmental impacts in european countries. *Energies*, 12(6), 1–11.
- Mbarki, F., Selmi, T., Kesraoui, A., Seffen, M., Gadonneix, P., Celzard, A., & Fierro, V. (2019). Hydrothermal pre-treatment, an efficient tool to improve activated carbon performances. *Industrial Crops and Products*, 140(August).
- Meher, L. C., Churamani, C. P., Arif, M., Ahmed, Z., & Naik, S. N. (2013). Jatropha curcas as a renewable source for bio-fuels - A review. *Renewable and Sustainable Energy Reviews*, 26, 397–407.
- Mello, V. M., Oliveira, F. C. C., Fraga, W. G., Do Nascimento, C. J., & Suareza, P. A. Z. (2008). Determination of the content of fatty acid methyl esters (FAME) in biodiesel samples obtained by esterification using 1H-NMR spectroscopy. *Magnetic Resonance in Chemistry*, 46(11), 1051–1054.
- Mendonça, I. M., Paes, O. A. R. L., Maia, P. J. S., Souza, M. P., Almeida, R. A., Silva, C. C., Duvoisin, S., & de Freitas, F. A. (2019). New heterogeneous catalyst for biodiesel production from waste tucumã peels (Astrocaryum aculeatum Meyer): Parameters optimization study. *Renewable Energy*, 130, 103–110.
- Meri, N. H., Alias, A. B., Talib, N., Rashid, Z. A., & Ghani, W. A. W. A. K. (2017). Effect of washing pre-Treatment of empty fruit bunch hydrogel biochar composite properties as potential adsorbent. *Chemical Engineering Transactions*, 56, 1255– 1260.
- Mochida, I., Nakamura, E. I., Maeda, K., & Takeshita, K. (1976). Carbonization of aromatic hydrocarbons-IV. Reaction path of carbonization catalyzed by alkali metals. *Carbon*, 14(2), 123–129.
- Montalbo, K. D., De Leon, R. L., Sophiphun, O., Manadee, S., Prayoonpokarach, S., & Wittayakun, J. (2013). Characterization and catalytic performance of potassium loaded on rice husk silica and zeolite nay for transesterification of jatropha seed oil. *Quimica Nova*, 36(8), 1116–1120.
- Mora, E., Blanco, C., Pajares, J. A., Santamaría, R., & Menéndez, R. (2006). Chemical activation of carbon mesophase pitches. In *Journal of Colloid and Interface Science* (Vol. 298, Issue 1, pp. 341–347).
- Mukarakate, C., Mittal, A., Ciesielski, P. N., Budhi, S., Thompson, L., Iisa, K., Nimlos, M. R., & Donohoe, B. S. (2016). Influence of crystal allomorph and crystallinity on the products and behavior of cellulose during fast pyrolysis. *ACS Sustainable Chemistry and Engineering*, 4(9), 4662–4674.
- Musa, I. A. (2016). The effects of alcohol to oil molar ratios and the type of alcohol on biodiesel production using transesterification process. In *Egyptian Journal of Petroleum*.

- Nam, H., Choi, W., Genuino, D. A., & Capareda, S. C. (2018). Development of rice straw activated carbon and its utilizations. *Journal of Environmental Chemical Engineering*, 6(4), 5221–5229.
- Narowska, B., Kułażyński, M., Łukaszewicz, M., & Burchacka, E. (2019). Use of activated carbons as catalyst supports for biodiesel production. *Renewable Energy*, 135, 176–185.
- Ngamcharussrivichai, C., Totarat, P., & Bunyakiat, K. (2008). Ca and Zn mixed oxide as a heterogeneous base catalyst for transesterification of palm kernel oil. *Applied Catalysis A: General*, 341(1–2), 77–85.
- Nicholas, A. F., Hussein, M. Z., Zainal, Z., & Khadiran, T. (2018). Palm kernel shell activated carbon as an inorganic framework for shape-stabilized phase change material. *Nanomaterials*, 8(9), 689.
- Nisar, J., Razaq, R., Farooq, M., Iqbal, M., Khan, R. A., Sayed, M., Shah, A., & Rahman, I. ur. (2017). Enhanced biodiesel production from Jatropha oil using calcined waste animal bones as catalyst. *Renewable Energy*, 101, 111–119.
- Nizamuddin, S., Jayakumar, N. S., Sahu, J. N., Ganesan, P., Bhutto, A. W., & Mubarak, N. M. (2015). Hydrothermal carbonization of oil palm shell. *Korean Journal of Chemical Engineering*, 32(9), 1789–1797.
- Okoroigwe, E. C., & Saffron, C. M. (2012). Determination of bio-energy potential of palm kernel shell by physicochemical characterization. *Nigerian Journal of Technology (NIJOTECH)*, 31(3), 329–335.
- Oschatz, M., Van Deelen, T. W., Weber, J. L., Lamme, W. S., Wang, G., Goderis, B., Verkinderen, O., Dugulan, A. I., & De Jong, K. P. (2016). Effects of calcination and activation conditions on ordered mesoporous carbon supported iron catalysts for production of lower olefins from synthesis gas. *Catalysis Science and Technology*, 6(24), 8464–8473.
- Pari, G., Darmawan, S., & Prihandoko, B. (2014). Porous carbon spheres from hydrothermal carbonization and KOH activation on cassava and tapioca flour raw material. *Procedia Environmental Sciences*, 20, 342–351.
- Pariatamby, A. (2014). MSW management in Malaysia-changes for sustainability. In Environmental Science and Engineering (Subseries: Environmental Science) (Issue 9789814451727, pp. 195–232).
- Park, S. H., Khan, N., Lee, S., Zimmermann, K., Derosa, M., Hamilton, L., Hudson, W., Hyder, S., Serratos, M., Sheffield, E., Veludhandi, A., & Pursell, D. P. (2019). Biodiesel production from locally sourced restaurant waste cooking oil and grease: Synthesis, characterization, and performance evaluation. ACS Omega, 4(4), 7775– 7784.
- Piloto-Rodríguez, R., Sánchez-Borroto, Y., Lapuerta, M., Goyos-Pérez, L., & Verhelst, S. (2013). Prediction of the cetane number of biodiesel using artificial neural networks and multiple linear regression. *Energy Conversion and Management*, 65, 255–261.

Putra, M. D., Irawan, C., Udiantoro, Ristianingsih, Y., & Nata, I. F. (2018). A cleaner

process for biodiesel production from waste cooking oil using waste materials as a heterogeneous catalyst and its kinetic study. *Journal of Cleaner Production*, 195, 1249–1258.

- Puziy, A. M., Poddubnaya, O. I., Martínez-Alonso, A., Suárez-García, F., & Tascón, J. M. D. (2002). Synthetic carbons activated with phosphoric - Acid I. Surface chemistry and ion binding properties. *Carbon*, 40(9), 1493–1505.
- Qi, L., Tang, X., Wang, Z., & Peng, X. (2017). Pore characterization of different types of coal from coal and gas outburst disaster sites using low temperature nitrogen adsorption approach. *International Journal of Mining Science and Technology*, 27(2), 371–377.
- Rahman, N. J. A., Ramli, A., Jumbri, K., & Uemura, Y. (2019a). Tailoring the surface area and the acid-base properties of ZrO2 for biodiesel production from Nannochloropsis sp. *Scientific Reports*, 9(1).
- Rahman, N. J. A., Ramli, A., Jumbri, K., & Uemura, Y. (2019b). Tailoring the surface area and the acid-base properties of ZrO2 for biodiesel production from Nannochloropsis sp. *Scientific Reports*, 9(1), 1–12.
- Rahmani Vahid, B., Haghighi, M., Alaei, S., & Toghiani, J. (2017). Reusability enhancement of combustion synthesized MgO/MgAl2O4 nanocatalyst in biodiesel production by glow discharge plasma treatment. *Energy Conversion and Management*, 143, 23–32.
- Rajgopal, S., Karthikeyan, T., Prakash Kumar, B. G., & Miranda, L. R. (2006). Utilization of fluidized bed reactor for the production of adsorbents in removal of malachite green. *Chemical Engineering Journal*, 116(3), 211–217.
- Rana, S., Haque, M., Poddar, S., Sujan, S., Hossain, M., & Jamal, M. (2015). Biodiesel production from non-edible Mahogany seed oil by dual step process and study of its oxidation stability. *Bangladesh Journal of Scientific and Industrial Research*, 50(2), 77–86.
- Rashid, U., Anwar, F., & Knothe, G. (2011). Biodiesel from Milo (Thespesia populnea L.) seed oil. *Biomass and Bioenergy*, 35(9), 4034–4039.
- Rashidi, N. A., & Yusup, S. (2017a). A review on recent technological advancement in the activated carbon production from oil palm wastes. *Chemical Engineering Journal*, 314, 277–290.
- Rashidi, N. A., & Yusup, S. (2017b). Potential of palm kernel shell as activated carbon precursors through single stage activation technique for carbon dioxide adsorption. *Journal of Cleaner Production*, 168, 474–486.
- Rashtizadeh, E., Farzaneh, F., & Talebpour, Z. (2014). Synthesis and characterization of Sr3Al2O6 nanocomposite as catalyst for biodiesel production. *Bioresource Technology*, *154*, 32–37.
- Rechnia-gor, P., Malaika, A., Rechnia-Gorący, P., Malaika, A., & Kozłowski, M. (2018). Acidic activated carbons as catalysts of biodiesel formation. *Diamond and Related Materials*, 87(May), 124–133.

- Reneta Nafu, Y., Foba-Tendo, J., Njeugna, E., Oliver, G., & Omar Cooke, K. (2015). Extraction and characterization of fibres from the stalk and spikelets of empty fruit bunch. *Journal of Applied Chemistry*, 2015, 1–10.
- Rezayan, A., & Taghizadeh, M. (2018). Synthesis of magnetic mesoporous nanocrystalline KOH/ZSM-5-Fe3O4 for biodiesel production: Process optimization and kinetics study. *Process Safety and Environmental Protection*, 117, 711–721.
- Román, S., Libra, J., Berge, N., Sabio, E., Ro, K., Li, L., Ledesma, B., Alvarez, A., & Bae, S. (2018). Hydrothermal carbonization: Modeling, final properties design and applications: A review. *Energies*, 11(1), 1–28.
- Sabzevari, M., Sajjadi, S. A., & Moloodi, A. (2016). Physical and mechanical properties of porous copper nanocomposite produced by powder metallurgy. *Advanced Powder Technology*, 27(1), 105–111.
- Saeedi Dehaghani, A. H., & Rahimi, R. (2019). An experimental study of diesel fuel cloud and pour point reduction using different additives. *Petroleum*, 5(4), 413–416.
- Saif, S., Tahir, A., Asim, T., & Chen, Y. (2016). Plant mediated green synthesis of CuO nanoparticles: Comparison of toxicity of engineered and plant mediated CuO nanoparticles towards Daphnia magna. *Nanomaterials*, 6(11), 1–15.
- Sajjadi, B., Chen, W. Y., & Egiebor, N. O. (2019). A comprehensive review on physical activation of biochar for energy and environmental applications. *Reviews in Chemical Engineering*, 35(6), 735–776.
- Saleem, J., Shahid, U. Bin, Hijab, M., Mackey, H., & McKay, G. (2019). Production and applications of activated carbons as adsorbents from olive stones. *Biomass Conversion and Biorefinery*, 9(4), 775–802.
- Salmasi, M. Z., Kazemeini, M., & Sadjadi, S. (2020). Transesterification of sunflower oil to biodiesel fuel utilizing a novel K2CO3/Tale catalyst: Process optimizations and kinetics investigations. *Industrial Crops and Products*, 156(May), 112846.
- Sano, N., Yamada, K., Tsunauchi, S., & Tamon, H. (2017). A novel solid base catalyst for transesterification of triglycerides toward biodiesel production: Carbon nanohorn dispersed with calcium ferrite. *Chemical Engineering Journal*, 307, 135–142.
- Santana Costa, J. A., & Paranhos, C. M. (2018). Systematic evaluation of amorphous silica production from rice husk ashes. *Journal of Cleaner Production*, *192*, 688–697.
- Sarwono, R., Tursiloadi, S., & Sembiring, K. C. (2016). Carbonization of palm oil empty fruit bunch (EFB) in hydrothermal processes to produce biochar. *Jurnal Kimia Terapan Indonesia*, 18(02), 116–123.
- Schonvogel, D., Nowotny, M., Woriescheck, T., Multhaupt, H., Wagner, P., Dyck, A., Agert, C., & Wark, M. (2019). Hydrothermal carbonization-derived carbon from waste biomass as renewable Pt support for fuel cell applications: Role of carbon activation. *Energy Technology*, 7(11), 1–13.

- Seffati, K., Honarvar, B., Esmaeili, H., & Esfandiari, N. (2019). Enhanced biodiesel production from chicken fat using CaO/CuFe2O4 nanocatalyst and its combination with diesel to improve fuel properties. *Fuel*, 235(August 2018), 1238–1244.
- Seguel, J., García, R., Chimentão, R. J., García-Fierro, J. L., Ghampson, I. T., Escalona, N., & Sepúlveda, C. (2020). Thermal modification effect on supported cu-based activated carbon catalyst in hydrogenolysis of glycerol. *Materials*, 13(3).
- Shafie, S. M., Mahlia, T. M. I., Masjuki, H. H., & Ahmad-Yazid, A. (2012). A review on electricity generation based on biomass residue in Malaysia. *Renewable and Sustainable Energy Reviews*, 16(8), 5879–5889.
- Shahbaz, M., Yusup, S., Pratama, A., Inayat, A., Patrick, D. O., & Ammar, M. (2016). Parametric study and optimization of methane production in biomass gasification in the presence of coal bottom ash. *Procedia Engineering*, 148, 409–416.
- Shan, R., Shi, J., Yan, B., Chen, G., Yao, J., & Liu, C. (2016). Transesterification of palm oil to fatty acids methyl ester using K2CO3/palygorskite catalyst. *Energy Conversion and Management*, 116, 142–149.
- Sharma, A., Kodgire, P., & Kachhwaha, S. S. (2020). Investigation of ultrasoundassisted KOH and CaO catalyzed transesterification for biodiesel production from waste cotton-seed cooking oil: Process optimization and conversion rate evaluation. *Journal of Cleaner Production*, 259, 120982.
- Shuit, S. H., Yee, K. F., Lee, K. T., Subhash, B., & Tan, S. H. (2013). Evolution towards the utilisation of functionalised carbon nanotubes as a new generation catalyst support in biodiesel production: An overview. RSC Advances, 3(24), 9070–9094.
- Skupien, E., Berger, R. J., Santos, V. P., Gascon, J., Makkee, M., Kreutzer, M. T., Kooyman, P. J., Moulijn, J. A., & Kapteijn, F. (2014). Inhibition of a gold-based catalyst in benzyl alcohol oxidation: Understanding and remediation. *Catalysts*, 4(2), 89–115.
- Smyrnioti, M., Tampaxis, C., Steriotis, T., & Ioannides, T. (2020). Study of CO2 adsorption on a commercial CuO/ZnO/Al2O3 catalyst. *Catalysis Today*, 357(May), 495–502.
- Soltani, S., Rashid, U., Al-Resayes, S. I., & Nehdi, I. A. (2017). Recent progress in synthesis and surface functionalization of mesoporous acidic heterogeneous catalysts for esterification of free fatty acid feedstocks: A review. In *Energy Conversion and Management* (Vol. 141, pp. 183–205). Elsevier Ltd.
- Soltani, S., Rashid, U., Nehdi, I. A., Al-Resayes, S. I., & Al-Muhtaseb, A. H. (2017). Sulfonated mesoporous zinc aluminate catalyst for biodiesel production from high free fatty acid feedstock using microwave heating system. *Journal of the Taiwan Institute of Chemical Engineers*, 70, 219–228.
- Soltani, S., Rashid, U., Yunus, R., & Taufiq-Yap, Y. H. (2016). Biodiesel production in the presence of sulfonated mesoporous ZnAl2O4catalyst via esterification of palm fatty acid distillate (PFAD). *Fuel*, *178*, 253–262.
- Sudaryanto, Y., Hartono, S. B., Irawaty, W., Hindarso, H., & Ismadji, S. (2006). High surface area activated carbon prepared from cassava peel by chemical activation.

Bioresource Technology, 97(5), 734–739.

- Sulaiman, N. F., Hashim, A. N. N., Toemen, S., Rosid, S. J. M., Mokhtar, W. N. A. W., Nadarajan, R., & Bakar, W. A. W. A. (2020). Biodiesel production from refined used cooking oil using co-metal oxide catalyzed transesterification. *Renewable Energy*, 153, 1–11.
- Sun, H., Sun, K., Wang, F., Liu, Y., Ding, L., Xu, W., Sun, Y., & Jiang, J. (2021). Catalytic self-activation of Ca-doped coconut shell for in-situ synthesis of hierarchical porous carbon supported CaO transesterification catalyst. *Fuel*, 285(September 2020), 119192.
- Syazwani, O. N., Teo, S. H., Islam, A., & Taufiq-Yap, Y. H. (2017). Transesterification activity and characterization of natural CaO derived from waste venus clam (Tapes belcheri S.) material for enhancement of biodiesel production. *Process Safety and Environmental Protection*, 105(February 2018), 303–315.
- Tang, Z. E., Lim, S., Pang, Y. L., Ong, H. C., & Lee, K. T. (2018). Synthesis of biomass as heterogeneous catalyst for application in biodiesel production: State of the art and fundamental review. *Renewable and Sustainable Energy Reviews*, 92(April), 235–253.
- Tang, Z. E., Lim, S., Pang, Y. L., Shuit, S. H., & Ong, H. C. (2020). Utilisation of biomass wastes based activated carbon supported heterogeneous acid catalyst for biodiesel production. *Renewable Energy*, 158, 91–102.
- Taufiq-Yap, Y. H., Teo, S. H., Rashid, U., Islam, A., Hussien, M. Z., & Lee, K. T. (2014). Transesterification of Jatropha curcas crude oil to biodiesel on calcium lanthanum mixed oxide catalyst: Effect of stoichiometric composition. *Energy Conversion* and Management, 88, 1290–1296.
- Teng, H., Lin, H. C., & Teng, H. (1998). Activated carbon production from low ash subbituminous coal with CO2 activation. AIChE Journal, 44(5), 1170–1177.
- Thangarasu, V., & Anand, R. (2019). Comparative evaluation of corrosion behavior of aegle marmelos correa diesel, biodiesel, and their blends on aluminum and mild steel metals. In Advanced Biofuels: Applications, Technologies and Environmental Sustainability. Elsevier Ltd.
- Thirumarimurugan, M., Sivakumar, V. M., Xavier, A. M., Prabhakaran, D., & Kannadasan, T. (2013). Preparation of biodiesel from sunflower oil by transesterification. *International Journal of Bioscience, Biochemistry and Bioinformatics*, 2(6), 441–444.
- Titiladunayo, I. F., McDonald, A. G., & Fapetu, O. P. (2012). Effect of temperature on biochar product yield from selected lignocellulosic biomass in a pyrolysis process. *Waste and Biomass Valorization*, *3*(3), 311–318.
- Toufiq Reza, M., Freitas, A., Yang, X., Hiibel, S., Lin, H., & Coronella, C. J. (2016). Hydrothermal carbonization (HTC) of cow manure: Carbon and nitrogen distributions in HTC products. *Environmental Progress and Sustainable Energy*, 35(4), 1002–1011.

Tran, T. T. V., Kaiprommarat, S., Kongparakul, S., Reubroycharoen, P., Guan, G.,

Nguyen, M. H., & Samart, C. (2016). Green biodiesel production from waste cooking oil using an environmentally benign acid catalyst. *Waste Management*, *52*, 367–374.

- Tseng, R. L. (2006). Mesopore control of high surface area NaOH-activated carbon. Journal of Colloid and Interface Science, 303(2), 494–502.
- Ukanwa, K. S., Patchigolla, K., Sakrabani, R., Anthony, E., & Mandavgane, S. (2019). A review of chemicals to produce activated carbon from agricultural waste biomass. Sustainability (Switzerland), 11(22), 1–35.
- Vakros, J. (2018). Biochars and Their Use as Transesterification Catalysts for Biodiesel Production: A Short Review. *Catalysts*, 8(11), 562.
- Valášek, P., Ruggiero, A., & Müller, M. (2017). Experimental description of strength and tribological characteristic of EFB oil palm fibres/epoxy composites with technologically undemanding preparation. *Composites Part B: Engineering*, 122, 79–88.
- Veljković, V. B., Biberdžić, M. O., Banković-Ilić, I. B., Djalović, I. G., Tasić, M. B., Nježić, Z. B., & Stamenković, O. S. (2018). Biodiesel production from corn oil: A review. *Renewable and Sustainable Energy Reviews*, 91(April), 531–548.
- Wadumesthrige, K., Smith, J. C., Wilson, J. R., Salley, S. O., & Ng, K. Y. S. (2008). Investigation of the parameters affecting the cetane number of biodiesel. *JAOCS*, *Journal of the American Oil Chemists' Society*, 85(11), 1073–1081.
- Wahyudi, A., Kurniawan, W., & Hinode, H. (2017). Study on deactivation and regeneration of modified red mud catalyst used in biodiesel production. *Green and Sustainable Chemistry*, 07(04), 247–258.
- Wassilkowska, A., Czaplicka-Kotas, A., Zielina, M., & Bielski, A. (2014). An analysis of the elemental composition of micro-samples using EDS technique. *Technical Transactions*, 1, 133–148.
- Weber, B., Stadlbauer, E. A., Eichenauer, S., Koch, C., Albert, K., Kramer, M., & Steffens, D. (2013). Chemical nature of carbonaceous materials from biomass by hydrothermal carbonization and low temperature conversion. *Journal of Biobased Materials and Bioenergy*, 7(3), 367–375.
- Wiedner, K., Rumpel, C., Steiner, C., Pozzi, A., Maas, R., & Glaser, B. (2013). Chemical evaluation of chars produced by thermochemical conversion (gasification, pyrolysis and hydrothermal carbonization) of agro-industrial biomass on a commercial scale. *Biomass and Bioenergy*, 59, 264–278.
- Wong, W. Y., Lim, S., Pang, Y. L., Shuit, S. H., Chen, W. H., & Lee, K. T. (2020). Synthesis of renewable heterogeneous acid catalyst from oil palm empty fruit bunch for glycerol-free biodiesel production. *Science of the Total Environment*, 727, 138534.
- Wu, F. C., Tseng, R. L., & Juang, R. S. (2005). Preparation of highly microporous carbons from fir wood by KOH activation for adsorption of dyes and phenols from water. *Separation and Purification Technology*, 47(1–2), 10–19.

- Yacob, M. R., Kabir, I., & Radam, A. (2015). Households willingness to accept collection and recycling of waste cooking oil for biodiesel input in Petaling District, Selangor, Malaysia. *Procedia Environmental Sciences*, 30, 332–337.
- Yadav, M., Singh, V., & Sharma, Y. C. (2017). Methyl transesterification of waste cooking oil using a laboratory synthesized reusable heterogeneous base catalyst: Process optimization and homogeneity study of catalyst. *Energy Conversion and Management*, 148, 1438–1452.
- Yahaya, M., Ramli, I., Muhamad, E. N., Ishak, N. S., Idris Nda-Umar, U., & Taufiq-Yap, Y. H. (2020). K2O doped dolomite as heterogeneous catalyst for fatty acid methyl ester production from palm oil. *Catalysts*, 10(7), 791.
- Yang, J., & Qiu, K. (2010). Preparation of activated carbons from walnut shells via vacuum chemical activation and their application for methylene blue removal. *Chemical Engineering Journal*, 165(1), 209–217.
- Yashim, M. M., Razali, N., Saadon, N., & Rahman, N. A. (2016). Effect of activation temperature on properties of activated carbon prepared from oil palm kernel shell (OPKS). ARPN Journal of Engineering and Applied Sciences, 11(10), 6389–6392. www.arpnjournals.com
- Yatish, K. V., Lalithamba, H. S., Suresh, R., & Latha, H. K. E. (2020). Ochrocarpus longifolius assisted green synthesis of CaTiO3 nanoparticle for biodiesel production and its kinetic study. *Renewable Energy*, 147, 310–321.
- Yihunu, E. W., Minale, M., Abebe, S., & Limin, M. (2019). Preparation, characterization and cost analysis of activated biochar and hydrochar derived from agricultural waste: A comparative study. SN Applied Sciences, 1(8).
- Yongphet, P., Wang, J., Wang, D., Mulbah, C., Fan, Z., Zhang, W., & Amaral, P. C. S. (2020). Optimization of operation conditions for biodiesel preparation from soybean oil using an electric field. *Biomass Conversion and Biorefinery*.
- Yue, M., Jiang, X., Zhang, S., Li, Y., & Xu, W. (2019). *N*-doped carbons accelerate the reducing decomposition of copper nitrate and construct bifunctional adsorbents for adsorption desulfurization.
- Yusuff, A. S., & Owolabi, J. O. (2019). Synthesis and characterization of alumina supported coconut chaff catalyst for biodiesel production from waste frying oil. *South African Journal of Chemical Engineering*, 30(August), 42–49.
- Zhao, B., Wang, J., Zhu, D., Song, G., Yang, H., Chen, L., Sun, L., Yang, S., Guan, H.,
 & Xie, X. (2019). Adsorption characteristics of gas molecules (H2O, CO2, CO, CH4, and H2) on CaO-based catalysts during biomass thermal conversion with in situ CO2 capture. *Catalysts*, 9(9).
- Zhao, C., Lv, P., Yang, L., Xing, S., Luo, W., & Wang, Z. (2018). Biodiesel synthesis over biochar-based catalyst from biomass waste pomelo peel. *Energy Conversion* and Management, 160(January), 477–485.