

**MODIFICATION OF RUBBERWOOD FIBRE BY GRAFT  
COPOLYMERISATION AND ITS APPLICATION AS ION EXCHANGER  
AND FILLER IN POLYPROPYLENE COMPOSITE**

BY

**FARAJ AHMAD ABU-ILAIWI**

**Thesis Submitted to the School of the Graduate Studies, Universiti  
Putra Malaysia in Fulfilment of the Requirements for the Degree of  
Philosophy**

**November 2004**

**DEDICATION**

*TO ALL MUSLIMS IN THE WORLD*

Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree Doctor of Philosophy

**MODIFICATION OF RUBBERWOOD FIBRE BY GRAFT  
COPOLYMERISATION AND ITS APPLICATION AS ION EXCHANGER  
AND FILLER IN POLYPROPYLENE COMPOSITE**

By

**FARAJ AHMAD ABU-ILAIWI**

**May 2004**

**Chairman: Associate Professor Mansor Ahmad, Ph.D.**

**Faculty: Science and environmental studies**

Graft copolymerisation of vinyl monomers such as poly(methyl acrylate) (PMA), poly(methyl methacrylate) (PMMA), and polyacrylonitrile (PAN) onto rubberwood fibre (RWF) was carried out by a free radical initiation. Hydrogen peroxide and ferrous ion were used as an initiator system to graft PMA and PMMA onto RWF, while potassium permanganate was used to graft PAN onto RWF. Effect of the reaction parameters (reaction temperature and reaction period, as well as hydrogen peroxide, ferrous ammonium sulphate, potassium permanganate, and monomer concentrations) on the percentage of grafting was investigated. The grafting percentage showed dependence on the initiator,  $\text{Fe}^{2+}$  and monomer concentrations, as well as the reaction temperature and reaction period. High percentage of grafting was achieved when the optimum reaction conditions were used. Optimum temperature of the

polymerisation of PMA onto rubberwood fibre was found to be about 55 °C for the reaction period 120 minutes. They were 60 °C and 60 minutes for PMMA while they were 70 °C and 180 minutes for PAN. Optimum concentration of H<sub>2</sub>O<sub>2</sub> was 0.02 M and the amounts of Fe<sup>2+</sup> 0.26 mmol when 0.05 mole of MA were used. When  $2.36 \times 10^{-2}$  moles of MMA were used, the concentration of H<sub>2</sub>O<sub>2</sub> and amount of Fe<sup>2+</sup> were 0.03 M and 0.26 mmol, respectively. PMA and PMMA homopolymers were removed from the graft copolymers by Soxhlet extraction using acetone. Optimum reaction conditions for grafting of PAN onto RWF were as follows: monomer amount; 0.18 mole, initiator amount; 4.0 mmol, nitric acid concentration; 0.2 M. PAN as homopolymer was removed from grafted product by DMF. The presence of PMA, PMMA and PAN on the fibre was confirmed by FTIR spectroscopy and gravimetric analysis.

PAN grafted rubberwood fibre was converted to poly(amidoxime) ion exchange resin in order to remove heavy metal ions from aqueous solutions. The cation-exchange resin exists predominantly in the syn-hydroxyamino form. The water uptake by the resin was about 31 g/g dry resin, and hydrogen capacity was 3.6 mmol/g. The absorption capacity for different metal ions from wastewater was determined at different pH's from 1 to 6. The prepared chelating ion exchanger gives highest adsorption capacity for Cu<sup>2+</sup>, which was 3.83 mmol/g, followed by Cd<sup>2+</sup>, Fe<sup>3+</sup>, Pb<sup>2+</sup>, Ni<sup>2+</sup> and Co<sup>2+</sup>, respectively. Poly(amidoxime) ion exchanger

resin was also used to separate cobalt and nickel ions from copper ion by using column technique. PMA grafted RWF was converted to poly(hydroxamic acid) ion exchanger resin. Adsorption of metal ions was studied at different pH's and it showed that the highest adsorption was for lead ion. The results showed that the absorption capacity depended on the solution pH. FTIR spectroscopy was used to confirm the conversion of grafted fibre to ion exchanger resin.

The percentage of moisture absorbed by the grafted products decreased depending on the grafted polymer. The moisture content in the fibre was reduced from 6 to less than 1% when PMA was grafted onto RWF.

Activation energies ( $E_a$ ) of RWF and its grafted copolymers were analysed by thermogravimetric analysis (TGA) and dynamic derivatives of the thermogravimetric (DTG). It was found that RWF and RWF-g-PMMA were degraded by one-step decomposition and their  $E_a$  are 91 and 96 KJ/mole respectively. On the other hand, RWF-g-PMA degraded by two steps while RWF-PAN degraded by three steps. The highest  $E_a$  of RWF-g-PMA was 199 KJ/mole of the second step but it was 200 KJ/mole for RWF-g-PAN at the first step. Temperature of first steps degradation of RWF and its grafted copolymers were as follows: RWF; 352 °C, RWF-g-PMA; 362 °C, RWF-g-PMMA; 370 °C, and RWF-g-PAN; 354 °C.

As an application of RWF-g-PMMA copolymer, it was used with RWF to prepare fibre/polypropylene composite. It was found that the tensile strength decreased with increasing of the fibre loading. Tensile strength and modulus were improved when 20% of grafted fibre was used with RWF in preparation of composite when it was radiated by electron beam radiation. However, increasing the fibre loading increased tensile and flexural moduli. On the other hand, toughness of the composite was improved when grafted RWF was used as a filler of PP composite.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**PENGUBAHSUIAN FIBER KAYU GETAH MELALUI  
PENGKOPOLIMERAN CANGKUK DAN APLIKASINYA SEBAGAI  
PENUKAR ION DAN PENGISI DALAM KOMPOSIT POLIPROPILENA**

**Oleh**

**FARAJ AHMAD ABU-ILAIWI**

**Mei 2004**

**Pengerusi: Profesor Madya Mansor Ahmad, Ph.D.**

**Faculty: Sains dan Pengajian Alam Sekitar**

Pengkopolimeran cangkuk oleh monomer vinil seperti poli(metil akrilit) (PMA), poli(metil metakrilit) (PMMA), dan poliakrilonitril (PAN) ke atas fiber kayu getah (RWF) dilakukan melalui radikal bebas sebagai bahan permula. Hidrogen peroksida dan ion ferus telah digunakan sebagai sistem permula bagi cangkukan PMA dan PMMA ke atas RWF, sementara itu kalium permanganat telah digunakan untuk cangkukan PAN ke atas RWF. Kesan parameter tindak balas (seperti suhu, masa tindak balas, kepekatan hidrogen peroksida, ferus ammonia sulfat, kalium permanganat dan monomer) ke atas peratus cangkukan telah dikaji. Peratus cangkukan didapati bergantung kepada kepekatan bahan permula,  $\text{Fe}^{2+}$  dan monomer, serta suhu dan masa tindak balas. Peratus cangkukan yang tinggi dicapai apabila keadaan tindak balas optimum digunakan. Suhu optimum pempolimeran PMA ke atas fiber

kayu getah telah didapati pada 55 °C dan masa tindak balas selama 120 minit. Manakala suhu dan masa untuk PMMA adalah pada 60 °C dan 60 minit, sementara untuk PAN adalah pada 70 °C dan 180 minit. Kepekatan optimum bagi H<sub>2</sub>O<sub>2</sub> adalah 0.02 M dan amaun Fe<sup>2+</sup> 0.26 mmol, manakala 0.05 mol MA telah digunakan. Apabila  $2.36 \times 10^{-2}$  mol MMA digunakan, kepekatan bagi H<sub>2</sub>O<sub>2</sub> dan amaun Fe<sup>2+</sup> masing-masing adalah 0.03 M dan 0.26 mmol. Homopolimer PMA dan PMMA telah dikeluarkan daripada kopolimer cangkuk melalui pengekstrak Soxhlet menggunakan aseton. Keadaan optimum tindak balas untuk cangkukan PAN ke atas RWF adalah seperti berikut: Amaun monomer, 0.18 mol; amaun pemula, 4.0 mmol; kepekatan asid nitric, 0.2 M. Homopolimer PAN telah dikeluarkan daripada hasil cangkukan oleh DMF. Kehadiran PMA, PMMA dan PAN ke atas fiber telah disahkan oleh spektroskopi FTIR dan analisis gravimetri.

PAN cangkukan fiber kayu getah telah ditukarkan kepada resin penukar ion poli(amidoksim) untuk mengeluarkan ion logam berat daripada larutan akueus. Resin penukar-kation wujud separa dominan dalam bentuk sin-hidroksiamino. Penyerapan air oleh resin adalah 31 g/g resin kering, dan kapasiti hidrogen adalah 3.6 mmol/g. Kapasiti serapan ion logam yang berlainan daripada air buangan diukur pada pH yang berlainan, iaitu daripada 1 hingga 6. Penukar ion pengelat yang disediakan memberikan kapasiti serapan yang tinggi untuk Cu<sup>2+</sup>, iaitu

3.83 mmol/g, diikuti oleh Cd<sup>2+</sup>, Fe<sup>3+</sup>, Pb<sup>2+</sup>, Ni<sup>2+</sup> dan Co<sup>2+</sup>. Resin penukar ion poli(amidoksim) telah juga digunakan untuk memisahkan ion kobalt dan ion nikel daripada ion kuprum menggunakan teknik kolumn. PMA cangkukan RWF telah ditukarkan kepada resin penukar ion asid poli(hidroksimik). Serapan oleh ion logam telah dikaji pada pH yang berlainan dan serapan yang tinggi adalah bagi ion plumbum. Keputusan menunjukkan kapasiti serapan bergantung kepada larutan pH. Spektroskopi FTIR telah mengesahkan penukaran fiber cangkukan kepada resin penukar ion.

Peratus penyerapan lembapan oleh hasil cangkukan menurun, bergantung kepada polimer cangkukan. Kandungan lembapan dalam fiber telah dikurangkan daripada 6% kepada kurang dari 1% apabila PMA dicangkukkan ke atas RWF.

Tenaga pengaktifan ( $E_a$ ) bagi RWF dan kopolimer cangkukan telah dikaji menggunakan analisis termogravimetri (TGA) dan termogravimetri kamilan (DTG). Didapati RWF dan RWF-g-PMMA telah mengalami degradasi melalui penguraian satu langkah dan  $E_a$  masing-masing adalah 91 dan 96 KJ/mol. Selain itu, RWF-g-PMA mengalami degradasi dua langkah sementara RWF-g-PAN mengalami degradasi tiga langkah. Nilai  $E_a$  yang tertinggi oleh RWF-g-PMA adalah 199 KJ/mol daripada

langkah kedua, tetapi adalah 200 KJ/mol untuk RWF-g-PAN pada langkah pertama. Suhu degradasi langkah pertama oleh RWF dan kopolimer cangkukannya seperti berikut: RWF, 352 °C; RWF-g-PMA, 362 °C; RWF-g-PMMA, 370 °C; dan RWF-g-PAN, 354 °C.

Sebagai aplikasi bagi kopolimer RWF-g-PMMA, ia telah digunakan dengan RWF untuk menyediakan komposit fiber/polipropilena. Penyediaan komposit dengan menggunakan 20% cangkukan fiber dan RWF memberi kekuatan tensil dan modulus yang tinggi apabila telah kenakan iradiasi oleh pancaran alur elektron. Walau bagaimana pun, kekuatan modulus tensil dan fleksural meningkat dengan pertambahan fiber. Selain itu, kekuatan komposit telah meningkat apabila RWF cangkukan telah digunakan sebagai bahan tambah bagi komposit polipropilena.

## **ACKNOWLEDGEMENT**

IN THE NAME OF ALLAH, THE BENEFICENT, THE MERCIFUL

Praise be to Allah, Lord of the worlds, the one, the only and the Indivisible Creator and Sustainer of the world. To Him, we belong and to Him, we will return. I wish to thank Him for all that He has gifted us with, although, He can never be praised or thanked enough and for giving me the strength and the patient to let this work be finished.

I would like to take this opportunity to thank my advisor, Assoc. Prof. Mansor Bin Ahmad, for his continuing support and excellent mentorship, Prof. Wan Md. Zin Wan Yunus, Assoc. Prof. Mohamad Zaki Ab Rahman and Dr. Khairul Zaman Mohd Dahlan for their cooperation to serve as my co-advisors. I am also grateful to them for their assistance and guidance in completing this research. Acknowledgement is also due to Mr. Wan Ali and Mr. Zahid from MINT, Ms. Azowa and her husband, and Ms. Chuo for helping on the ICP machine and Ms. Rosnani from IR room in UPM for their tireless help. Finally, this work could not have been completed without the love and support of my family. I thank my beloved mother, father, brothers and sisters, for their endless support. Special thanks are also extended to my friends in Malaysia, Dr. Mohammad

Saleh and his wife Sara, Mohammad Alkhawaldeh, my roommate Ala' Albalooty, Mohammad Manna, Moaiad al-Saifi, Isam Qudsieh, and all other Jordanian friends for their help and support by one way or another.

I certify that an Examination Committee met on 31 September 2004 to conduct the final examination of Faraj Ahmad Abu-Ilaiwi on his Doctor of Philosophy thesis entitled "Modification of Rubberwood Fibre by Graft Copolymerisation and its Application as Ion Exchanger and Filler in polypropylene Composite" in according with Universiti Pertanian Malaysia (Higher Degree) Act 1980 and Universiti Pertanian Malaysia (Higher Degree) Regulations 1981. The Committee recommends that the candidate be awarded the relevant degree. Members of the Examination Committee are as follows:

**Anuar Kasim, Ph.D.**

Professor  
Faculty of Graduate Studies  
Universiti Putra Malaysia  
(Chairman)

**Zulkarnain Zainal, Ph.D.**

Professor  
Faculty of Graduate Studies  
Universiti Putra Malaysia  
(Member)

**Sedik Silong, Ph.D.**

Professor  
Faculty of Graduate Studies  
Universiti Putra Malaysia  
(Member)

**Hanafi Ismail, Ph.D.**

Professor  
Faculty of Polymer Engineering  
Universiti Sains Malaysia  
(Independent Examiner)

---

**GULAM RUSUL RAHMAT ALI, Ph.D.**

Professor/Deputy Dean  
School of Graduate Studies  
Universiti Putra Malaysia

Date:

This thesis submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment for the degree of Doctor Philosophy. The members of the Supervisory Committee are as follows:

**Mansor Ahmad, Ph.D.**

Associate Professor  
Faculty of Science and Environmental Studies  
Universiti Putra Malaysia  
(Chairman)

**Wan Zin Wan Md. Yunus, Ph.D.**

Professor  
Faculty of Science and Environmental Studies  
Universiti Putra Malaysia  
(Member)

**Mohamad Zaki Ab. Rahman, Ph.D.**

Associate Professor  
Faculty of Science and Environmental Studies  
Universiti Putra Malaysia  
(Member)

**Khairul Zaman Mohd Dahlan, Ph.D.**

Director  
Radiation Processing Technology Division  
Malaysian Institute for Nuclear Technology Research (MINT)  
(Member)

---

**AINI IDERIS, Ph.D.**

Professor/Dean  
School of Graduate Studies  
Universiti Putra Malaysia

Date:

## **DECLARATION**

I hereby declare that the thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at UPM or other institutions.

---

**FARAJ AHMAD ABU-ILAIWI**

Date:

## TABLE OF CONTENTS

	<b>Page</b>
<b>DEDICATION</b>	<b>1</b>
<b>ABSTRACT</b>	<b>3</b>
<b>ABSTRAK</b>	<b>7</b>
<b>ACKNOWLEDGEMENT</b>	<b>11</b>
<b>APPROVAL</b>	<b>xii</b>
<b>DECLARATION</b>	<b>xiv</b>
<b>LIST OF TABLES</b>	<b>xix</b>
<b>LIST OF FIGURES</b>	<b>xx</b>
<b>ABBREVIATIONS</b>	<b>xxviii</b>
 <b>CHAPTER</b>	
<b>1 INTRODUCTION</b>	<b>1.1</b>
1.1 Background of the Study	1.1
1.2 Aims of the Project	1.6
 <b>2 LITERATURE REVIEW</b>	<b>2.1</b>
2.1 Chemical Composition of Cellulosic Fibres	2.1
2.1.1 Holocellulose	2.1
2.1.2 Cellulose	2.2
2.1.3 Hemicellulose	2.5
2.1.4 Lignin	2.5
2.2 Graft Copolymerization	2.6
2.2.1 Jute Fibre	2.7
2.2.2 Cellulose and Its Derivatives	2.11
2.2.3 Kenaf Fibre	2.27
2.2.4 Sisal Fibre	2.28
2.2.5 Other Lignocellulosic Fibres	2.29
2.3 Ion Exchangers	2.32
2.3.1 Resins Containing Amidoxime Groups	2.33
2.3.2 Resins Containing Hydroxamic Acid Groups	2.37
2.3.3 Other Compounds Used as Ion Exchangers	2.39
2.4 Fibre Composite	2.43
 <b>3 MATERIALS AND METHODS</b>	<b>3.1</b>
3.1 Materials	3.1
3.2 Methods	3.3

3.2.1	Graft Copolymerization of Polyacrylonitrile onto Rubberwood Fibre	3.3
3.2.2	Graft Copolymerisation of Poly(Methyl Methacrylate) and Poly(Methyl Acrylate) onto Rubberwood Fibre	3.4
3.2.3	Removal of Homopolymer and Determination of Grafting Level	3.4
3.2.4	Preparation of Chelating Resin Containing Amidoxime Groups	3.5
3.2.5	Preparation of Chelating Resin Containing Hydroxamic Acid Groups	3.6
3.2.6	Swelling Study	3.6
3.2.7	Cation Exchange Capacity	3.7
3.2.8	Sorption of Metal Ions by Batch Technique by Polyamidoxime Ion Exchanger	3.8
3.2.9	Sorption of Metal Ions by Batch Technique by Poly(Hydroxamic Acid) Ion Exchanger	3.8
3.2.10	Separation of Metal Ions by Polyamidoxime Ion Exchanger	3.9
3.2.11	FTIR Analysis	3.9
3.2.12	Thermal Analysis	3.10
3.2.13	Preparation of the Rubberwood Fober / PP Composite	3.10
3.2.14	Electron Beam Radiation	3.11
3.2.15	Determination of Mechanical Properties	3.12

<b>4</b>	<b>GRAFT COPOLYMERIZATION</b>	<b>4.13</b>
4.1	Graft Copolymerization of Acrylonitrile onto Rubberwood Fibre	4.13
4.1.1	Reaction Mechanism	4.13
4.1.2	Effect of Reaction Temperature	4.16
4.1.3	Effect of Reaction Period	4.17
4.1.4	Effect of Monomer Amount	4.18
4.1.5	Effect of Acid Concentration	4.19
4.1.6	Effect of Initiator Amount	4.21
4.1.7	FTIR Spectroscopy	4.22
4.1.8	Conclusion	4.23
4.2	Graft Copolymerization of Methyl Acrylate onto Rubberwood Fibre	4.24
4.2.1	Effect of Reaction Temperature	4.24
4.2.2	Effect of Reaction Period	4.25
4.2.3	Effect of Monomer Amount	4.26
4.2.4	Effect of Ferrous Ion Amount	4.27
4.2.5	Effect of Initiator Concentration	4.29

4.2.6	Conclusion	4.30
4.3	Graft Copolymerisation of Methyl Methacrylate onto Rubberwood Fibre	4.31
4.3.1	Effect of Reaction Temperature	4.31
4.3.2	Effect of Reaction Period	4.32
4.3.3	Effect of Monomer Amount	4.33
4.3.4	Effect of Ferrous Ion Amount	4.34
4.3.5	Effect of Initiator Concentration	4.36
4.3.6	Conclusion	4.36
4.4	Reaction Mechanism	4.37
4.5	FTIR Spectroscopy	4.40

## **5 CONVERSION OF GRAFTED RUBBERWOOD FIBRE INTO ION EXCHANGERS**

5.1	Ion Exchanger Containing Amidoxime Group	5.1
5.1.1	Swelling Behaviour	5.3
5.1.2	Sorption of Metal Ions	5.3
5.1.3	Separation of Metal Ions	5.5
5.1.4	FTIR Spectroscopy	5.7
5.1.5	Conclusion	5.9
5.2	Ion Exchanger Containing Hydroxamic Acid Group	5.10
5.2.1	Sorption of Metal Ions	5.11
5.2.2	FTIR Spectroscopy	5.13
5.2.3	Conclusion	5.14

## **6 THERMAL DECOMPOSITION KINETICS OF RUBBERWOOD FIBRE AND GRAFTED RUBBERWOOD FIBRE**

6.1	Introduction	6.1
6.1.1	Thermogravimetric Analysis (TGA)	6.2
6.1.2	Calculation of Activation Energy	6.9
6.1.3	Effect of Heating Rate	6.16
6.1.4	Conclusion	6.17

## **7 MECHANICAL PROPERTIES OF RUBBERWOOD FIBRE REINFORCED POLYPROPYLENE**

7.1	Introduction	7.1
7.2	Effect of Fibre Percentage Loading	7.1
7.3	Effect of RWF-g-PMMA Fraction	7.6
7.4	Effect of Electron Beam Radiation	7.13
7.5	Conclusion	7.22

<b>8 CONCLUSION AND RECOMMENDATIONS</b>	<b>8.1</b>
8.1 Conclusion	8.1
8.2 Further Studies	8.4
<b>REFERENCES</b>	<b>R.1</b>
<b>BIODATA OF THE AUTHOR</b>	<b>B.1</b>