



UNIVERSITI PUTRA MALAYSIA

RHEOLOGY OF POMELO JUICE

SAMANEH KESHANI

FK 2009 30

RHEOLOGY OF POMELO JUICE

SAMANEH KESHANI

**MASTER OF SCIENCE
UNIVERSITI PUTRA MALAYSIA**

APRIL 2009



RHEOLOGY OF POMELO JUICE

By

SAMANEH KESHANI

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfilment of the Requirements for the Degree of Master of Science**

April 2009



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment
of the requirement for the degree of Master of Science.

RHEOLOGY OF POMELO JUICE

By

SAMANEH KESHANI

April 2009

Chairman: **Luqman Chuah Abdullah, PhD**

Faculty: **Engineering**

The proximate analysis studies on pomelo juice provide useful data for its juice processing industry. The studies on the physico-chemical properties such as water activity, pH and density make better prediction of its behavior at the relevant processing concentration and temperature. The Pomelo juice is found to have high moisture content which contributes to high water activity and specific heat of capacity. Also, the PH of pomelo juice obtained indicates that pomelo juice is acidic fluid. The predictions of density yielded high regression coefficient. In general, the density increases with increasing concentration but decreases with temperature increase. The density of pomelo juice also presented a stronger dependence on concentration than the temperature.

Rheology is the science of deformation and flow behavior of fluid. Knowledge of rheological properties of fluid and their variation with temperature and concentration have been globally important for industrialization of food technology for quality, understanding the texture, process engineering application, correlation with



sensory evaluation, designing of transport system , equipment design (heat exchanger and evaporator), deciding pump capacity and power requirement for mixing.

The aim of this study was to determine the rheological behavior of pomelo juice at different concentrations ($20\text{-}60.4^{\circ}\text{Brix}$) and temperatures ($23\text{-}60^{\circ}\text{C}$) by using a rotational Haake Rheostress 600 rheometer and a cone and plate geometry sensor (35mm diameter, 2° cone angle), with a gap distance of 0.105 mm. Shear rates employed ranged from $0\text{-}1000\text{ s}^{-1}$.

Pomelo juice was found to exhibit both Newtonian and Non-Newtonian behavior. For lower concentration the Newtonian behavior is observed while at higher concentration Non-Newtonian behavior was observed. Standard error (SE) method was selected on the basis to carry out the error analysis due to the best fit model. For the four models the values of SE show that the Herschel-Bulkley and Power Law models perform better than the Bingham and Casson models but Herschel-Bulkley model is true at higher concentration.

The experimental data of apparent viscosity versus shear rate were successfully described by the Power-law model. The model parameters; flow behavior index, n varied in the range of 0.65-0.89, whereas the values for the consistency coefficient, K , were in the range of $0.004\text{-}0.3\text{ (Pa.s}^n\text{)}$. Apparent viscosity and consistency coefficient of pomelo juice increased with increasing concentration and decreasing temperature. The effect of temperature on n was found to be significant but did not



follow any descriptive trend.

The effect of temperature on pomelo juice is best described by the applicability of the Arrhenius model related to apparent viscosity, η at a constant shear rate of 100s^{-1} or the consistency index of the power law model. Activation energies (E_a) of the pomelo juice appeared in the range of $9103.83\text{-}23391.43 \text{ J/mol}$ as the pomelo juice concentration was increased from 20 to 60.4°Brix .

The relationship between η or K of pomelo juice concentration was explained by exponential and power functions while pomelo juice observed that at higher concentration will have higher viscosity.

The master-curve was investigated for comparing data from different products at a reference temperature of 40°C . Multiple regression analysis indicated Master-Curve presents good agreement for pomelo juice at all concentrations studied with $R^2 > 0.8$.

Response surface methodology (RSM) was applied to investigate the effects of the three main independent parameters; rotation, temperature and time on the pomelo juice concentration and optimizing the operating conditions of the treatment process. Values of parameters required to obtain efficiency of concentration pomelo juice are suggested based on rotation speed (60-120 rpm), temperature ($40\text{-}60^\circ \text{C}$) and time (5-60 min).

It is clear from the result that the concentration increased when the temperature,

rotation speed and time increased. It can be inferred that any parameters, individually, had positive effect on increase of concentration. The main effects of parameters are in following order: Main effect of time > temperature > rotation speed.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains.

SIFAT REOLOGI JUS LIMAU BALI PEKAT

Oleh

SAMANEH KESHANI

April 2009

Pengerusi : Luqman Chuah Abdullah, PhD

Fakulti: Kejuruteraan

Kajian analisis anggaran terhadap jus limau tamun menyediakan data berguna kepada industri pemprosesan jus sementara itu kajian terhadap sifat fizikokimia seperti kegiatan air, pH, dan ketumpatan membolehkan ramalan bagi tingkah laku nya pada kepekatan dan suhu proses relevan. Jus limau tamun didapati mempunyai tinggi kandungan lembapan yang menyumbang kepada kegiatan air tinggi dan haba spesifik keupayaan. Selain itu, pH jus limau tamun didapati menunjukkan yang jus limau tamun adalah bendalir berasid. Ramalan-ramalan bagi ketumpatan menghasilkan pekali regresi yang tinggi. Secara umum, ketumpatan meningkat dengan bertambahnya kepekatan tetapi berkurangan dengan peningkatan suhu. Ketumpatan jus limau tamun juga menunjukkan pergantungan keatas kepekatan dibandingkan dengan suhu.

Reologi adalah sains penyahbentukan dan prestasi aliran bendalir. Pengetahuan sifat-sifat reologi tentang makanan bendalir dan variasi mereka dengan suhu dan kepekatan adalah penting secara global untuk pengindustrian teknologi makanan untuk kualiti, pemahaman tekstur, pemohonan kejuruteraan proses, pembetulan dengan penilaian deria, penciptaan



sistem pengangkutan, reka bentuk alatan (penukar haba dan penyejat), penentuan kapasiti pam dan keperluan kuasa untuk pencampuran.

Tujuan kajian ini adalah untuk menentukan sifat-sifat reologi jus limau tamun pada kepekatan (20-60.4%) dan suhu (23-60°C) yang berlainan dengan menggunakan berputar Haake Rheostress 600 reometer dan sebuah kon dan pengesan plat geometri (35mm garis pusat, 2° sudut kon), dengan jarak jurang 0.105 mm. Kadar-kadar rincih digunakan dalam 0-1000 s-1.

Jus pomelo didapati memperlihatkan kedua-dua ciri Newtonian dan Bukan-Newtonian. Pada kepekatan yang rendah, jus memperlihatkan ciri Newtonian manakala pada kepekatan yang lebih tinggi ciri Bukan-Newtonian dapat dilihat. Kaedah Ralat Piawai (SE) telah dipilih sebagai asas untuk menganalisa ralat, berdasarkan kepada model yang paling sesuai. Bagi keempat-empat model tersebut, didapati nilai ralat piawai model Herschel-Buckley dan model Hukum Kuasa memberikan keputusan yang lebih baik berbanding model Bingham Dan Casson, walaubagaimanapun pada kepekatan lebih tinggi ciri-ciri model Herschel-Buckley diperhatikan.

Data eksperimen kelikatan ketara lawan kadar ricihan telah berjaya diterangkan oleh model Power Law. Parameter model; indek sifat-sifat aliran, n berubah dalam julat 0.65-0.89, manakala nilai-nilai untuk pekali kekonsistenan, K , berada dalam julat 0.004-0.3 (Pa.sn). Kelikatan ketara dan pekali kekonsistenan jus limau tamun meningkat dengan bertambah ketumpatan dan pengurangan suhu. Kesan suhu pada n didapati adalah penting tetapi tidak mengikut mana-mana pola yang boleh diterang.

Kesan suhu pada jus limau tamun adalah terbaik digambarkan dengan kebolehgunaan model Arrhenius berkaitan dengan kelikatan ketara, pada kadar ricih malar 100s atau indek kekonsistenan model Power Law. Tenaga-tenaga pengaktifan (E_a) jus limau tamun muncul dalam julat 9103.83-23391.43 J/mol dengan ketumpatan jus limau tamun meningkat dari 20 hingga 60.4 Brix.

Hubungan antara η atau K kepekatan jus limau tamun telah dijelaskan dengan eksponen dan kuasa fungsi sementara jus limau tamun diperhatikan yang pada ketumpatan lebih tinggi akan mendapat kelikatan yang lebih tinggi.

Induk lengkung telah disiasat untuk membandingkan data dari produk berbeza pada suhu rujukan 40°C . Analisis regresi berbilang menunjukkan Master-Curve membentangkan perjanjian baik untuk jus limau tamun dalam semua kajian kepekatan dengan $R^2 > 0.8$.

Response surface methodology (RSM) diaplikasikan untuk menyiasat kesan tiga parameter mendiri yang utama; putaran, suhu dan masa terhadap kepekatan jus limau tamun dan mengoptimakan syarat-syarat beroperasi proses rawatan. Nilai parameter yang diperlukan untuk mendapat kepekatan jus pomelo yang lebih effisen adalah berdasarkan kelajuan putaran (60-120 rpm), suhu ($40\text{-}50^{\circ}\text{C}$) dan masa (5-60 min).

Ia adalah jelas daripada hasil bahawa kepekatan bertambah apabila suhu, kelajuan putaran dan masa bertambah. Ia boleh disimpulkan bahawa mana-mana parameter, secara individu, mempunyai kesan positif kepada peningkatan kepekatan. Kesan-kesan utama parameter adalah dalam aturan berikut: Kesan utama masa > Suhu > kelajuan putaran.

ACKNOWLEDGEMENTS

Alhamdulillah, at last this thesis has finally completed. I wish to express my deepest appreciation to my supervisor Dr. Luqman Chuah Abdullah and co-supervisors Professor Dr. Jamilah Bakar and Professor Dr. Russly Abdul Rahman for their guidance, insightful comments and patient supervision of this work. I also owe a special debt of thanks to all technicians in the Department of Food Science & Technology and Chemical Engineering and my fellow colleagues for their helping hand. The helpful collaboration of KD Technology Sdn. Bhd. is highly appreciated. I would like to thank my beloved husband for his understanding and support during this study. I am also very grateful to my parents and friends for their encouragement and sharing time throughout this two years journey.



I certify that an Examination Committee met on 29 April 2009 to conduct the final examination of Samaneh Keshani on her Master of Science thesis entitled “Rheological behavior of Pomelo Juice Concentrates” in accordance with Universiti Pertanian Malaysia (Higher Degree) Act 1980 and Universiti Pertanian Malaysia (Higher Degree) Regulations 1981. The Committee recommends that the candidate be awarded the relevant degree.

Members of the Examination Committee are as follows:

Thomas Choong Shean Yaw, PhD

Professor

Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Robiah Yunus, PhD

Professor

Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Siti Mazlina, PhD

Lecturer

Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Ir.Hj.Wan Ramli, PhD

Professor

Faculty of Engineering
Universiti Kebangsaan Malaysia
(External Examiner)

BUJANG KIM HUAT, PhD

Professor/Deputy Dean

School of Graduate Studies
Universiti Putra Malaysia

Date:



This thesis submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee are as follows:

Luqman Chuah Abdullah, PhD

Associate Professor

Faculty of Engineering

Universiti Putra Malaysia

(Chairman)

Jamilah Bakar, PhD

Professor

Faculty of Food Science & Technology

Universiti Putra Malaysia

(Member)

Russly Abdul Rahman , PhD

Professor

Faculty of Food Science & Technology

Universiti Putra Malaysia

(Member)

HASANAH MOHD GHAZALI, PhD

Professor and Dean

School of Graduate Studies

Universiti Putra Malaysia

Date: 9 July 2009

DECLARATION

I hereby declare that the thesis is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at UPM or other institutions.

SAMANEH KESHANI

Date: 22 June 2009



LIST OF TABLES

Table		Page
1.1	Reological models of some fruit juice	1.5
2.1	A summary of studies on pomelo fruit	2.2
2.2	Advantages and disadvantages of the current processes of concentrating fruit juices	2.7
2.3	Rheological Models for Time Independent Viscous Foods	2.16
2.4	Effects of temperature and concentration on rheological parameters	2.23
3.1	Experimental design matrix and response value.	3.10
4.1	Central composite design: independent (X_i) and response variables (Y_j).	4.6
4.2	Regression coefficients, R^2 , adjusted R^2 probability values and lack of fit for five variables.	4.7
4.3	ANOVA and regression coefficients of the first- and second-order polynomial regression models.	4.7
4.4	Minimum value of parameters required for obtaining the efficiency of concentration of pomelo juice (95 to 100%).	4.12
5.1	Percentage of different types of proximate analysis compositions in Fresh pomelo juice	5.1
5.2	Linear regression of density of pomelo juice concentrates using single Parameter, concentration, C, measured as total soluble solids content in °Brix.	5.6
5.3	Linear regression of density of pomelo juice concentrates using single Parameter, temperature, T, measured in °C.	5.6
5.4	Parameters of Newtonian Model	5.12
5.5	Parameters of Bingham Plastic Model	5.16

5.6	Parameters of Power-law Model	5.20
5.7	Parameters of Herschel-Bulkley Model	5.26
5.8	Parameters of Casson Model	5.31
5.9	Selection of flow models based on standard errors	5.42
5.10	Arrhenius model parameters for the apparent viscosity, η at 100s^{-1} and the consistency index, K of different concentration pomelo juice samples	5.44
5.11	Magnitudes of parameters in equations 2.12 and 2.13 for the apparent viscosity, at 100s^{-1} and the consistency index of different concentration	5.48

LIST OF FIGURES

Figure		Page
1.1	Outline of the study.	1.9
2.1	Rheological classification of food materials.	2.9
2.2	Imaginary representation of a Newtonian fluid flowing between a stationary plate and moving plate (F =force acting on the plate, v =velocity of moving fluid).	2.10
2.3	Rheograms for two different Newtonian fluids (lines A and B) and a non-Newtonian fluid (line C).	2.13
2.4	Types of non-Newtonian behavior.	2.14
2.5	Flow curves for typical time-independent fluids. (A), Bingham Plastic; (B), Power-law (Shear thinning); (C), Power-law (Shear thickening); (D), Herschel-Bulkley.	2.14
2.6	Log- log plot of τ versus $\dot{\gamma}$ and τ versus $\dot{\gamma}$	2.17
2.7	Apparent viscosity of Power Law fluids	2.18
2.8	Concentric cylinder rheometer	2.28
2.9	Cone and Plate rheometer	2.29
3.1	The Pomelo fruit used in this experiment.	3.1
3.2	Rotary Evaporator (HEIDOLPH)	3.2
3.3	Haake RheoStress 600 (RS 600)	3.11
4.1	Predicted versus Actual data for concentration of pomelo juice.	4.5
4.2	Response surface plots for concentration of pomelo juice as a function of (a) Rotation and Temperature, (b) Rotation and time and (c) Temperature and Time.	4.6
5.1	Effect of Concentration on pH of pomelo juice concentrates.	5.2

5.2	Effect of Concentration on water activity of pomelo juice concentrates.	5.3
5.3	Effect of Concentration on density of pomelo juice concentrate at temperatures ranging from 10 to 60°C.	5.5
5.4	Effect of temperature on density of pomelo juice Concentrate at Concentrations ranging from 10 to 50 °Brix.	5.5
5.5	Rheogram of pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 23°C.	5.8
5.6	Rheogram of pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 30°C.	5.9
5.7	Rheogram of pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 40°C	5.9
5.8	Rheogram of pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 50°C.	5.10
5.9	Rheogram of pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 60°C.	5.10
5.10	Mathematical Newtonian Model for pomelo juice at different temperature at 20 ° Brix.	5.12
5.11	Mathematical Newtonian Model for pomelo juice at different temperature at 30.4 ° Brix.	5.13
5.12	Mathematical Newtonian Model for pomelo juice at different temperature at 40.4 ° Brix.	5.13
5.13	Mathematical Newtonian Model for pomelo juice at different temperature at 53.4 ° Brix.	5.14
5.14	Mathematical Newtonian Model for pomelo juice at different temperature at 60.4 ° Brix	5.14
5.15	Mathematical Bingham Plastic Model for pomelo juice at concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 23°C.	5.17
5.16	Mathematical Bingham Plastic Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 30°C.	5.17

5.17	Mathematical Bingham Plastic Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 40°C.	5.18
5.18	Mathematical Bingham Plastic Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 50°C.	5.18
5.19	Mathematical Bingham Plastic Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 60°C.	5.19
5.20	Mathematical Power-law Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 23°C.	5.21
5.21	Mathematical Power-law Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 30°C.	5.21
5.22	Mathematical Power-law Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 40°C.	5.22
5.23	Mathematical Power-law Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 50°C.	5.22
5.24	Mathematical Power-law Model for pomelo juice at concentration of 20, 30.4, 40.4, 53.4, 60.4 ° Brix at 60°C.	5.23
5.25	Structure of pectin	5.25
5.26	Mathematical Herschel-Bulkley Model for pomelo juice at Concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 23°C.	5.27
5.27	Mathematical Herschel-Bulkley Model for pomelo juice at Concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 30°C.	5.27
5.28	Mathematical Herschel-Bulkley Model for pomelo juice at Concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 40°C.	5.28
5.29	Mathematical Herschel-Bulkley Model for pomelo juice at Concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 50°C.	5.28
5.30	Mathematical Herschel-Bulkley Model for pomelo juice at Concentration of 21, 30.4, 40.4, 53.4, 60.4 ° Brix at 60°C.	5.29
5.31	Mathematical Casson Model for pomelo juice at different concentration at 23°C	5.32

5.32	Mathematical Casson Model for pomelo juice at different concentration at 30°C.	5.32
5.33	Mathematical Casson Model for pomelo juice at different concentration at 40°C.	5.33
5.34	Mathematical Casson Model for pomelo juice at different concentration at 50°C.	5.33
5.35	Mathematical Casson Model for pomelo juice at different concentration at 60°C.	5.34
5.36	Apparent viscosity of pomelo juice for concentrate at 20°Brix at different temperatures.	5.35
5.37	Apparent viscosity of pomelo juice for concentrate at 30.4°Brix at different temperatures.	5.36
5.38	Apparent viscosity of pomelo juice for concentrate at 40.4°Brix at different temperatures.	5.36
5.39	Apparent viscosity of pomelo juice for concentrate at 53.4°Brix at different temperatures.	5.37
5.40	Apparent viscosity of pomelo juice for concentrate at 60.4°Brix at different temperatures.	5.37
5.41	Variation in Apparent viscosity of pomelo juice with shear rate for 23°C at different concentrate.	5.38
5.42	Variation in Apparent viscosity of pomelo juice with shear rate for 30°C at different concentrate.	5.38
5.43	Variation in Apparent viscosity of pomelo juice with shear rate for 40°C at different concentrate.	5.39
5.44	Variation in Apparent viscosity of pomelo juice with shear rate for 50°C at different concentrate.	5.39
5.45	Variation in Apparent viscosity of pomelo juice with shear rate for 60°C at different concentrate.	5.40
5.46	Applicability of the Arrhenius model to the apparent viscosity, η versus temperature data on pomelo juice at different concentration.	5.44

5.47	The influence of concentration on Apparent Viscosity at shear rate 100s^{-1} of pomelo juice at different temperature.	5.48
5.48	The influence of concentration on consistency index of pomelo juice at different temperature.	5.49
5.49	Plot of shear stress versus $(\frac{\dot{\gamma}}{a_T})$ providing a master-curve of pomelo juice (20° Brix) having a reference temperature of 40°C .	5.50
5.50	Plot of shear stress versus $(\frac{\dot{\gamma}}{a_T})$ providing a master-curve of pomelo juice (30.4° Brix) having a reference temperature of 40°C .	5.51
5.51	Plot of shear stress versus $(\frac{\dot{\gamma}}{a_T})$ providing a master-curve of pomelo juice (40.4° Brix) having a reference temperature of 40°C .	5.51
5.52	Plot of shear stress versus $(\frac{\dot{\gamma}}{a_T})$ providing a master-curve of pomelo juice (53.4° Brix) having a reference temperature of 40°C .	5.52
5.53	Plot of shear stress versus $(\frac{\dot{\gamma}}{a_T})$ providing a master-curve of pomelo juice (60.4° Brix) having a reference temperature of 40°C .	5.52

LIST OF NOTATIONS / SYMBOLS

Notations / symbols

τ_0	Shear stress	Pa
κ'	Bingham Plastic viscosity	
η	Apparent viscosity	Ns/m ² , pa s
k	Consistency index	Ns ⁿ /m ²
$\dot{\gamma}$	Shear rate	1/s
n	Flow behavior index	
K	Consistency coefficient-(power-low, Herschel-Bulkley, Arrhenius-type)	Pa. s ⁿ
n	Flow behavior index-(power-low)	
τ	Shear stress (Newton's law when the flow is laminar)	Pa
μ	Newtonian viscosity	
γ	Shear rate (Newton's law when the flow is laminar)	1/s
η_0	Experimental Constants (Arrhenius-type equation) at constant shear rate 100 s^{-1}	
Ea	Activation energy	J/mol
R	Universal gas law constant, Arrhenius-type equation	8.314 J/g.mol.K
T	Absolute temperature (Arrhenius-type equation)	°K
$\tau^{0.5}$	Shear Stress (Casson model)	Pa ^{0.5}
RO	Reverse osmosis	
$\tau_{oc}^{0.5}$	Casson yield	Pa ^{0.5}
k_c	Casson constant	Pa s ^{0.5}
η_{ca}	Casson plastic viscosity	
k_0	Experimental constants (Arrhenius-type equation)	
A'	Frequency factor at constant shear rate 100 s^{-1}	
B	Constant for equation (2.12)	
B'	Constant for equation (2.13)	

C	Concentration	°Brix
a_w	Water Activity	Dimensionless
a_T	Shift factor	Dimensionless
ρ	Density	Kg/m ³
m_c	Mass fraction of carbohydrate	
m_p	Mass fraction of protein	
m_f	Mass fraction of fat	
m_a	Mass fraction of ash	
m_m	Mass fraction of moisture	
C_p	Specific heat	KJ/Kg.°C
F	Force	N
A	Area	m ²
v	Velocity of moving fluid	m/s

TABLE OF CONTENTS

	Page
ABSTRACT	ii
ABSTRAK	vi
ACKNOWLEDGEMENTS	x
APPROVAL	xi
DECLARATION	xiii
LIST OF TABLES	xiv
LIST OF FIGURES	xvi
LIST OF NOTATIONS / SYMBOLS	xxi
 CHAPTER	
1 INTRODUCTION	1.1
1.1 General	1.1
1.2 Problem Statement And Objectives Of The Study	1.4
1.3 Scope of Study	1.9
2 LITERATURE REVIEW	2.1
2.1 Anatomy of citrus fruit	2.1
2.1.1 Taxonomy of Pomelo	2.1
2.1.2 Pomelo	2.2
2.2 Postharvest physiology of fresh fruit	2.2
2.3 Concentration of juices	2.3
2.4 Conventional Methods for Concentrating Foods	2.4
2.4.1 Thermal Evaporation	2.4
2.4.2 Freeze Drying	2.4
2.4.3 Freeze-concentration	2.5
2.4.4 Reverse osmosis	2.6
2.5 Introduction to Food Rheology	2.7
2.5.1 Viscosity	2.9
2.6 Flow Models for Rheological Properties of Fluids	2.11
2.6.1 Newtonian Fluids	2.11
2.6.2 Non-Newtonian Fluids	2.12
A Time Independent Fluids	2.15
A.1 Bingham Model	2.16
A.2 Power-law Model	2.16
A.2.1 Apparent viscosity	2.16
A.2.2 Shear thinning Fluids	2.18
A.2.3 Shear thickening Fluids	2.19
A.3 Herschel-Bulkley Model	2.20
A.4 Casson Model	2.20
B Time-Dependent Fluids	2.21
B.1 Thixotropic Fluids	2.21

	B.2 Rheopectic Fluids	2.22
2.7	Effective Factors on the rheological properties of food	2.22
2.7.1	Effect of Temperature	2.23
2.7.2	Effect of Concentration	2.24
2.8	Rheological Properties Mesurments	2.25
2.8.1	Rotational Viscometers	2.26
2.8.2	Concentric Cylinder rheometer	2.26
2.8.3	Cone and Plate rheometer	2.29
2.9	Response Surfaces Methodology (RSM)	2.30
3	MATERIALS AND METHODS	3.1
3.1	Preparations of samples	3.1
3.2	Proximate Analysis	3.3
3.2.1	Determination of moisture content	3.3
3.2.2	Determination of ash content	3.3
3.2.3	Determination of crude protein by Kjeldahl method	3.4
3.2.4	Determination of crude fiber by Weende method	3.4
3.2.5	Determination of crude fat by Soxhlet method	3.5
3.2.6	Determination of carbohydrates	3.5
3.3	Physico-Chemical Properties	3.6
3.3.1	PH	3.6
3.3.2	Water Activity	3.6
3.4	Thermophysical Properties	3.6
3.4.1	Specific Heat Capacity	3.6
3.4.2	Density	3.7
3.5	Preparation of samples of different concentration using the Response Surface Methodology	3.7
3.5.1	Introduction	3.7
3.5.2	Experiment Design	3.9
3.6	Rheological Properties analysis	3.11
3.6.1	Rheological Measurment Equipment	3.11
3.6.2	Experiment	3.12
3.6.3	Data analysis	3.12
3.6.4	Error analysis	3.12
4	OPTIMIZATION OF CONCENTRATION PROCESS ON POMELO JUICE USING RESPONSE SURFACE METHODOLOGY	4.1
4.1	Response surface analysis	4.1
4.1.1	Concentration	4.4
5	CHARACTERIZATION AND RHEOLOGICAL ON POMELO JUICE	5.1
5.1	Proximate Analysis	5.1
5.2	Physico-Chemical Properties	5.2
5.2.1	PH	5.2
5.2.2	Water Activity	5.2