



UNIVERSITI PUTRA MALAYSIA

***FORMULATION OF NANOEMULSIONS ENCAPSULATED WITH
POTENTIAL ANTICANCER DRUG, BETULINIC ACID***

NUR NADIAH BINTI ABDUL RASHID

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BERILMU BERBAKTI

**FORMULATION OF NANOEMULSIONS ENCAPSULATED WITH
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By

NUR NADIAH BINTI ABDUL RASHID

**Thesis submitted to the School of Graduate Studies, Universiti Putra Malaysia, in
Fulfilment of the Requirements for the Degree of Master of Science**

January 2014

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Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Master of Science

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January 2014

Chairman : Assoc. Prof. Intan Safinar Ismail, PhD

Faculty : Science

The betulinic acid provided was recrystallized in order to obtain high purity compound and was confirmed by spectroscopic analysis. Betulinic acid was incorporated in the oil phase prior to the construction of ternary phase diagram. Phase behaviours of soybean oil and non-ionic surfactants were determined through the construction of ternary phase diagrams. The phase behaviours were affected by hydrophilic-lipophilic balance (HLB) value of surfactants. Higher HLB values produced larger one-phase regions: homogenous and isotropic, in ternary phase diagrams of soybean oil/non-ionic surfactant/deionized water and soybean oil/non-ionic surfactant-co-surfactant/deionized water. The largest one-phase regions were formed by soybean oil/Cremophor EL-Span 20/deionized water formulation.

A few compositions with 70% water content were selected on the ternary phase diagram of soybean oil/Cremophor EL/deionized water system as the formulation of emulsions. The selected compositions were 15:15:70, 18:12:70, 21:9:70 and 24:6:70. The first set of emulsions was prepared via low-energy emulsification method, while the other set was formulated via high-energy emulsification method using a high-pressure homogenizer with homogenizing cycle of 2, 4, 6 and 8. Characteristics of emulsions were studied. The average particle size of low-energy formulated emulsions was larger than 130 nm at week 1 and the size increased rapidly throughout 12-weeks of study while for emulsions formulated via 8 homogenizing cycles, the average particle size was below 57 nm at week 1 and remained below 100 nm after 12-weeks. Formulation of 24:6:70 produced the smallest average size which was 59 nm.

The surface charge values for all formulations with betulinic acid were more negative than -26.7 mV which indicates moderate stability of the emulsions. The stability of

emulsions was also studied via visual observation for 6 months. All high-energy formulated emulsions were still in one phase without any separation of layers observed. The pH values were between 3.9 to 4.1 for all formulations. Betulinic acid can still be detected by HPLC-RI detector in the selected 24:6:70 formulation even after 6 months of storage.



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**PENGHASILAN NANOEMULSI YANG MENGANDUNGI UBAT ANTI
KANSER YANG BERPOTENSI, ASID BETULINIK**

Oleh

NUR NADIAH BINTI ABDUL RASHID

Januari 2014

Pengerusi : Prof. Madya Intan Safinar Ismail, PhD

Fakulti : Sains

Asid betulitik yang disediakan telah melalui proses pengkristalan semula bagi mendapatkan asid betulitik yang lebih tulen. Ia dianalisis menggunakan FTIR dan NMR bagi memastikan ianya adalah asid betulitik. Asid betulitik dilarutkan ke dalam fasa minyak sebelum penghasilan rajah tiga fasa. Ciri-ciri fasa minyak kacang soya dan surfaktan tidak berion dikaji dan ditentukan dengan menggunakan rajah tiga fasa. Ciri-ciri ini dipengaruhi oleh nilai keseimbangan komponen hidrofilik dan lipofilik (HLB) surfaktan. Nilai HLB yang lebih tinggi menghasilkan kawasan satu fasa iaitu homogen dan isotropik yang lebih besar pada rajah tiga fasa formulasi minyak kacang soya/surfaktan tidak berion/air dinyah ion. Rajah tiga fasa formulasi minyak kacang soya/Cremophor EL-Span20/air dinyah ion menghasilkan fasa homogen dan isotropik terbesar.

Formulasi emulsi berkomposisi 15:15:70, 18:12:70, 21:9:70 dan 24:6:70 telah dipilih pada rajah tiga fasa minyak kacang soya/Cremophor EL/air dinyah ion bagi kajian lanjutan. Emulsi disediakan melalui dua kaedah: pengemulsian bertenaga rendah dan pengemulsian bertenaga tinggi yang menggunakan instrumen penghomogenan bertekanan tinggi dengan kitaran penghomogenan sebanyak 2, 4, 6 dan 8. Purata saiz partikel bagi emulsi yang disediakan dengan menggunakan kaedah pengemulsian bertenaga rendah adalah lebih besar dari 130 nm pada minggu pertama dan meningkat secara mendadak sepanjang kajian selama 12 minggu. Bagi emulsi yang dihasilkan melalui kaedah bertenaga tinggi, purata saiz partikel adalah lebih kecil dari 57 nm pada minggu pertama dan kekal di bawah saiz 100 nm selepas 12 minggu kajian. Formulasi 24:6:70 dengan 8 kitaran penghomogenan menghasilkan purata saiz partikel paling rendah iaitu 59 nm selepas 12 minggu.

Bacaan cas permukaan bagi semua formulasi adalah lebih negatif dari -26.7 mV, di mana ia membuktikan kestabilan formulasi yang sederhana. Kestabilan emulsi juga dikaji melalui pemerhatian fizikal selama 6 bulan. Kesemua emulsi yang dihasilkan melalui kaedah bertenaga tinggi masih berada dalam satu fasa selepas 6 bulan dan tiada pengasingan lapisan. Nilai pH bagi kesemua formulasi adalah di antara 3.9 hingga 4.1. Asid betulitik di dalam emulsi 24:6:70 masih dapat dikesan dengan menggunakan HPLC-RI selepas penyimpanan selama 6 bulan.



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I certify that a Thesis Examination Committee has met on 28 January 2014 to conduct the final examination of Nur Nadiah binti Abdul Rashid on her thesis entitled "Formulation of Nanoemulsions Encapsulated with Potential Anticancer Drug Betulinic Acid" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Mohd Aspollah bin Hj Md Sukari, PhD

- Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Mohd Zobir bin Hussein, PhD

Professor
Institute of Advance Technology
Universiti Putra Malaysia
(Internal Examiner)

Thahira Begum, PhD

Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Internal Examiner)

Sivakumar Manickam, PhD

Professor
University of Nottingham (Malaysia Campus)
Malaysia
(External Examiner)



NORITAH OMAR, PhD

Associate Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 21 April 2014

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement of the degree of Master of Science. The members of supervisory committee were as follows:-

Intan Safinar Ismail, PhD

Associate Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Siti Salwa Abdul Gani, PhD

Lecturer
Faculty of Science
Universiti Putra Malaysia
(Member)

Latifah Saiful Yazan, PhD

Associate Professor
Faculty of Medicine and Health Sciences
Universiti Putra Malaysia
(Member)

BUJANG BIN KIM HUAT, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

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Signature: _____

Name of Chairman of

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Supervisory Committee: Latifah Saiful Yazan, PhD

Signature: _____

Name of Member of

Supervisory Committee: Siti Salwa Abdul Gani, PhD

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LIST OF ABBREVIATIONS

BA	betulinic acid
DLS	dynamic light scattering
FTIR	Fourier transform infrared
HIV	human immunodeficiency virus
HLB	hydrophilic-lipophilic balance
IR	infrared
MOPI	Malaysian Organization of Pharmaceutical Industries
o/w	oil-in-water
o/w/o	oil-in-water-in-oil
PCCS	Photon cross correlation spectroscopy
PIT	phase inversion temperature
RI	refractive index
Span 20	sorbitan mono-laurate
Tween 80	polyoxyethylene(20) sorbitan mono-oleate
w/o	water-in-oil
w/o/w	water-in-oil-in-water

CHAPTER 1

INTRODUCTION

Pharmaceutical products or more commonly known as medicines or drugs are a fundamental components of both modern and traditional medicines. It is essential that such products are safe, effective, of good quality, and are prescribed and used rationally. The worldwide pharmaceuticals market growth is accelerating in this 21st century as the number of demand from consumer increases. This is due to increment of number of patients for all sorts of illnesses including cancer. Cancer is currently a fast recurring illness among men and women. According to World Health Organization (WHO), the global cancer rates could increase by 50% to 15 million by 2020. In many countries, more than a quarter of deaths are attributable to cancer.

One defining feature of cancer is the rapid creation of abnormal cells that grow beyond their usual boundaries, and which can then invade adjoining parts of the body and spread to other organs. Statistically, there were 7.6 million people worldwide died because of cancer. Approximately 70% of cancer deaths occur in low and middle income countries (World Health Organization, 2011). World Cancer Report provides clear evidence that action on smoking, diet and infections can prevent one third of cancers and another one third can be cured by the modern treatments.

Betulinic acid has attracted the interests of researchers due to its variety of biological and pharmacological activities. It can be easily extracted from barks of huge trees. Betulinic acid is a naturally occurring pentacyclic triterpenoid which exhibits the anticancer, anti-HIV, antibacterial, antimalarial and anthelmintic activities. In addition, it is also reported to exhibit analgesic and anti-inflammatory properties (Fulda & Debatin, 2000; Yogeewari & Sriram, 2005).

In these modern days, pharmaceutical products in the form of emulsions have been increasing in numbers. The main concern about emulsions is regarding its stability. According to Tadros in 2005, emulsions are thermodynamically stable. Emulsions with small particle size, generally below 500 nm are called as nanoemulsions. The idea of nanoemulsions formations has caught the attentions of industries due to its small average particle size. This small particle size property contributes to the improvement of drug stability and absorption in human's body. Nanoemulsions are commonly used as drug carrier for active ingredients. It has been suggested that the encapsulation of poor-water soluble agents such as betulinic acid in nanoemulsions can improve the solubility.

Nanoemulsions, which have an average droplet size of 20 to 200 nm, have the ability to penetrate the membranes and have higher chances of reaching the targeted areas and

improve absorption of the active ingredients. The physical appearance of nanoemulsions is translucent but it depends on the materials used in the formulation. This property is due to the fact that light waves are scattered by the droplets.

The basic compositions of nanoemulsions formation are water, surfactant and oil or ester. The purpose of surfactant is to lower the surface tension of a liquid or the interfacial tension between two different liquids. Nanoemulsions can be successfully formed through high-energy emulsification method. In this research, the high-energy emulsification method used is high-pressure homogenization. Before the formulation undergoes high-pressure homogenization process, the emulsions are initially formulated through low-energy emulsification method which involves the stepwise addition of water to oil-surfactant mixture or stepwise addition of oil to water-surfactant mixture and mixed vigorously using vortex mixer.

Problem Statements

Betulinic acid has been discovered as an anticancer agent for more than a decade. The main disadvantage of betulinic acid is the poor water-solubility property. Human's body consist of more than 55 % of water, which relates to the lower efficiency of betulinic acid in one's body. In contrast, betulinic acid has higher solubility property in oil and lipid phase. To combat solubility problem, betulinic acid is solubilised in oil-phase which is soybean oil, prior to the formulation of emulsion. Emulsions with large particle size are often related to low stability. In order to form small particle size emulsions with high stability, alternative preparation methods were used.

Objectives

- i. To construct the ternary phase diagram of soybean oil/non-ionic surfactant/deionized water and soybean oil/non-ionic surfactant-co-surfactant/ deionized water.
- ii. To study the phase behaviour of the constructed phase diagrams and select the compositions based from the ternary phase diagram for formulation of nanoemulsion.
- iii. To formulate nanoemulsion as drug carrier with encapsulation of betulinic acid based on soybean oil.
- iv. To characterize the formulations through the stability study, particle size, zeta potential, pH value and drug analysis.

REFERENCES

- Acosta, E. (2009). Bioavailability of nanoparticles in nutrient and nutraceutical delivery. *Current Opinion in Colloid and Interface Science*. 14(1): 3-15.
- Ahmad, F. B. H., Hassan, V. U., Zakaria, R. and Ali, J.H. (1996). Betulinic acid and ursolic acid from the seed of *Melaleuca cajuputi* (Myrtaceae). *Oriental Journal of Chemistry*. 13: 231-232.
- Aliani, M. and Farmer, L. J. (2002). Postcolumn derivation method for determination of reducing and phosphorylated sugars in chicken by high performance liquid chromatography. *Journal of Agricultural and Food Chemistry*. 50(10): 2760-2766.
- Aviram, S. and Abraham, A. (2006). Microemulsion as carriers for drugs and nutraceuticals. *Advances in Colloid and Interface Science*. 128-130.
- Azeem, A., Rizwam, M., Ahmad, F. J., Iqbal, Z., Khar, R. K., Aqil, M. and Talegaonkar, S. (2009). Nanoemulsion components screening and selection: a technical note. *AAPS PharmSciTech*. 10(1): 69-76.
- Bali, V., Ali, M. and Ali, J. (2010). Study of surfactant combinations and development of a novel nanoemulsion for minimising variations in bioavailability of ezetimibe. *Colloids and Surfaces B: Biointerfaces*. 76: 410-420.
- Cater, B. R., Butterworth, K. R., Gaunt, I. F., Hooson, J., Grasso, P. and Gangolli, S. D. (1978). Short-term toxicity of sorbitan monolaurate (Span 20) in rats. *Food and Cosmetics Toxicology*. 16(6): 519-526.
- Chakraborty, S., Shukla, D., Mishra, B. and Singh, S. (2009). Lipid- An emerging platform for oral delivery of drugs with poor bioavailability, *European Journal of Pharmaceutics and Biopharmaceutics*. 73(1): 1-15.
- Daher, C. F., Baroody, G. M. and Howland, R. J. (2003). Effect of a surfactant, Tween 80, on the formation and secretion of chylomicrons in the rat. *Food Chemistry Toxicology*. 41(4): 575-582.
- Epstein, H. and Simion, F. A. (2001). Emulsion-based skincare products: formulating and measuring their moisturizing benefit. In A. O. Barel, M. Paye, H.I. Maibach. *Handbook of Cosmetic Science and Technology* (pp. 521-530). Marcel-Dekker, Inc. 270 Madison Avenue, New York.
- Fulda, S. (2008). Betulinic acid for cancer treatment and prevention. *International Journal of Molecular Sciences*. 9: 1096-1107.
- Fulda, S. and Debatin, K. (2000). Betulinic acid induces apoptosis through a direct effect on mitochondria in neuroectodermal tumors. *Medicinal and Pediatric Oncology*. 35: 616-618.

- Gani, S. S. A., Basri, M., Rahman, M. B. A., Kassim, A., Salleh, A.B., Abdul Rahman, R. N. Z. R. and Ismail, Z. (2009). Phase behavior of engkabang fat with nonionic surfactants. *Tenside, Surfactants, Detergents*. 46: 1-4.
- Gao, F., Zhang, Z., Bu, H., Huang, Y., Gao, Z., Shen, J., Zhao, C. and Li, Y. (2011). Nanoemulsion improves the oral absorption of candesartan cilexetil in rats: Performance and mechanism. *Journal of Controlled Release*. 149: 168-174.
- Geers, H. and Witt, W. (2008). Direct calculation of the volume based particle size distribution from PCS or PCCS measurements. *Particle Systems Analysis*, Stratford-upon-Avon, United Kingdom.
- Geers, H., Witt, W. and Babick, F. Stability analysis of emulsions and suspensions with photon cross-correlation spectroscopy. *Proceedings of the International Conference for Particle Technology, PARTEC*, 2007.
- Gosh, P. K., Majithiya Rita, J., Manish, L. U. and Rayasa S. R. M. (2006). Design and development of microemulsion drug delivery system of Acyclovir for improvement of oral bioavailability. *AAPS PharmSciTech*. 7(3): Article 77 E1-E6.
- Greenwood, R. and Kendall, K. (1999). Selection of suitable dispersants for aqueous suspensions of zirconia and titania powders using acoustophoresis. *Journal of the European Ceramic Society*. 19(4): 479-488.
- Gueglielmini, G. (2005). Novel emulsifier system for nano size droplets. *Proceedings of International Conference World Wide Wellness*. Florence, Italy. September 19-21, 2005.
- He, C. X., He, Z. G. and Gao, J. Q. (2010). Microemulsions as drug delivery systems to improve the solubility and bioavailability of poorly water-soluble drugs. *Expert Opinion on Drug Delivery* 7(4): 445-460.
- Innocente, N., Biasutti, M., Venir, E., Spaziani, M. and Marchesini, G. (2009). Effect of high-pressure homogenization on droplet size distribution and rheological properties of ice cream mixes. *Journal of Dairy Science*. 92(5): 1864-1875.
- Ismail, A. Z. B., Rashid, N. N. A. and Ahmad, F. B. H. (2011). Part 2: Solubility of betulinic acid in microemulsion system. *Oriental Journal of Chemistry*.
- Li, Y., Zheng, J., Xiao, H. and McClements, D.J. (2012). Nanoemulsion-based delivery systems for poorly water-soluble bioactive compounds: Influence of formulation parameters on polymethoxyflavone crystallization. *Food Hydrocolloids* 27: 517-528.
- Liu, W., Sun, D., Li, C., Liu, Q. and Xu, J. (2006). Formation and stability of paraffin oil-in-water nano-emulsions prepared by the emulsion inversion point method. *Journal of Colloid and Interface Science*. 303: 557-563.

- McClements, D. J. (2005). *Food Emulsions: Principles, Practice, and Techniques*. Second edition. CRC Press, Boca Rotan, FL.
- McClements, D. J. (2010). Emulsion design to improve the delivery of functional lipophilic components. *In Annual Review of Food Science and Technology*. 1: 241-269.
- McClements, D. J. and Li, Y. (2010). Structured emulsion-based delivery systems: Controlling the digestion and release of lipophilic food components. *Advances in Colloid and Interface Science*. 159(2): 213-228.
- Miller, R., Kragel, J., Fainerman, V. B., Makievski, A. V., Grigoriev, D.O., Ravera, F., Liggieri, L., Kwok, D. Y. and Neumann, A. W. (2001). Characterization of water/oil interfaces. In J. Sjöblom. *Encyclopedia Handbook of Emulsion Technology* (pp. 35-39). Marcel Dekker, Inc.
- Mitchell, L. and Schlossman, B. A. (2000). *The Chemistry and Manufacture of Cosmetics. Volume 2- Formulating* (pp. 7-8). Allured Publishing Corporation.
- Narsimhan, G. and Goel, P. (2001). Drop coalescence during emulsion formation in a high pressure homogenizer for tetradecane-in-water emulsion stabilized by sodium dodecyl sulfate. *Journal of Colloid and Interface Science*. 238: 420-432.
- Noudeh, G. D., Khazaeli, P., Mirzaei, S., Sharififar, F. and Nasrollahosaiani. S. (2009). Determination of the toxicity effect of the sorbitan esters surfactants group on biological membrane. *Journal of Biological Sciences*. 9(5): 423-430.
- Ophardt, C. E. (2003). *Anti-cancer drugs I* in Virtual Chembook, Elmhurst College, Illinois.
- Porter, M. R. (1994). *Use of Surfactant Theory. Handbook of Surfactants* (pp. 26-93). Blackie Academic & Professional. United Kingdom.
- Ragelle, H., Crauste-Manciet, S., Seguin, J., Brossard, D., Scherman, D., Arnaud, P. and Chabot, G. G. (2012). Nanoemulsion formulation of fisetin improves bioavailability and antitumor activity in mice. *International Journal of Pharmaceutics*. 427: 452-459.
- Sadurni, N., Solans, C., Azemar, N. and García-Celma, M. J. (2005). Studies on the formation of O/W nano-emulsions, by low-energy emulsification methods, suitable for pharmaceutical applications. *European Journal of Pharmaceutical Sciences*. 26: 438-445.
- Salager, J-L., Forgiarini, A., Márquez, L., Peña, A., Pizzino, A., Rodriguez, M. P. and Rondón-Gonzalez, M. (2004). Using emulsion inversion in industrial processes. *Advances in Colloid and Interface Science*. 108-109: 259-272.
- Shafiq, S., Faiyaz, S., Sushma, T., Ahmad, F. J., Khar, R. K. and Ali, M. (2007). Development and bioavailability assessment of ramipril nanoemulsion

- formulation. *European Journal of Pharmaceutics and Biopharmaceutics*. 66: 227-243.
- Solans, C., Izquierdo, P., Nolla, J., Azamer, N. and Garcia-Celma, M.J. (2005). Nano-emulsion. *Journal of Current Opinion in Colloid and Interfaces Science*. 10: 102-110.
- Swarnalatha, S., Selvi, P. K., Ganesh Kumar, A. and Sekaran, G. (2008). Nanoemulsion drug delivery by ketene based polyester synthesized using electron rich carbon/silica composite surface. *Colloids and Surfaces B: Biointerfaces*. 65: 292-299.
- Tadros, T. F. and Vincent, B. (1983). *Encyclopedia of Emulsion Technology*. Paul Becher Edition, Marcel Dekker, New York.
- Tadros, T. F. (2005). *Applied Surfactant, Principle and Applications* (pp. 285-307). WILEY_VCH Verlag GmbH & Co. KGaA, Weinheim.
- Tadros, T. F., Izquierdo, P., Esquena, J. and Solans, C. (2003). Formation and stability of nano-emulsions. *Journal of Advances in Colloid and Interface Science*. 108-109: 303-318.
- Tang, S. Y., Sivakumar, M., Ng, A. M-H. and Shridharan, P. (2012). Anti-inflammatory and analgesic activity of novel oral aspirin-loaded nanoemulsion and nano multiple emulsion formulations generated using ultrasound cavitation. *International Journal of Pharmaceutics*. 430: 299-306.
- Troncoso, E., Aguilera, J. M. and McClements, D. J. (2011). Development of nanoemulsions by an emulsification-evaporation technique. *11th International Congress on Engineering and Food*. Athens, Greece.
- Uson, N., Garcia, M. J. and Solans, C. (2004). Formation of water-in-oil (W/O) nano-emulsions in a water/mixed non-ionic surfactant/oil systems prepared by a low-energy emulsification method. *Colloids and Surfaces A: Physicochemical Engineering Aspects*. 250: 415-421.
- Witt, W., Geers, H. and Aberle, L. (2004). Measurement of particle size and stability of nanoparticles in opaque suspensions and emulsions with photon cross correlation spectroscopy (PCCS). *International Conference for Particle Technology, PARTEC 2004*.
- Yogeeswari, P. and Sriram, D. (2005). Betulinic acid and its derivatives: A review on their biological properties. *Current Medicinal Chemistry*. 12: 657-666.
- Yurgec, M. J. (2009). A rheological of shear on a model emulsion system. Master of Science Thesis, North Carolina State University, Raleigh, United States.