



UNIVERSITI PUTRA MALAYSIA

***SYNTHESIS AND CHARACTERIZATION OF PALM FATTY HYDRAZIDE
DERIVATIVES***

NORASHIKIN BINTI AHMAD

FS 2012 86



**SYNTHESIS AND CHARACTERIZATION OF PALM FATTY HYDRAZIDE
DERIVATIVES**

By

NORASHIKIN BINTI AHMAD

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfilment of the Requirements for the Degree of Master Science**

December 2012

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright© Universiti Putra Malaysia



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment
of the requirement for the degree of Master of Science

SYNTHESIS AND CHARACTERIZATION OF PALM FATTY HYDRAZIDE DERIVATIVES

By

NORASHIKIN BINTI AHMAD

December 2012

Chairman : Mansor Hj Ahmad @ Ayob, PhD

Faculty : Science

Hydrazide derivatives have been synthesized from methyl esters, hydrazones and vegetable oils. They are important due to their diverse applications in pharmaceutical products, detergents as well as in oil and gas industries. The chemical synthesis of fatty hydrazides is well-established; however, few publications described the synthesis of fatty hydrazide derivatives, particularly, when produced from refined, bleached and deodorized palm olein. Here, the synthesis and characterization of fatty hydrazide derivatives are reported. Besides, palmityl hydrazide derivative synthesized from palmityl hydrazide was used as reference. The fatty hydrazides derivatives were characterized using a Fourier transform infrared (FTIR), gas chromatography (GC) and nuclear magnetic resonance (NMR) spectroscopy. The synthesis of *N,N*-dimethyl fatty hydrazides was carried out at 100°C in dioxane for 6 hours with the molar ratio between hydrazide, dimethyl sulfate and potassium hydroxide was 1:1:1. The proton NMR confirmed the product obtained was *N,N*-dimethyl fatty hydrazides. Other fatty hydrazide derivatives were cationic surfactants

prepared using *N,N*-dimethyl fatty hydrazides. The surface tensions of these cationic surfactants vary according to the types of acyl chlorides. Lower surface tension was recorded for cationic surfactants prepared with acyl chloride with phenyl group. In addition, cationic surfactant can also be prepared from reaction between acyl chlorides and fatty hydrazides. Hence, fatty hydrazide derivatives from palm olein were successfully prepared and characterized.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk ijazah Master Sains

SINTESIS DAN PENCIRIAN TERBITAN HIDRAZIDA LEMAK SAWIT

Oleh

NORASHIKIN BINTI AHMAD

Disember 2012

Pengerusi : Mansor Hj Ahmad @ Ayob, PhD

Fakulti : Sains

Terbitan hidrazida telah disintesis daripada metil ester, hidrazon dan minyak sayur-sayuran. Ianya penting disebabkan oleh kepelbagaian penggunaan dalam produk farmaseutikal, bahan pencuci dan juga dalam industri minyak dan gas. Sintesis kimia lemak hidrazida telah lama dibangunkan; namun hanya terdapat beberapa penerbitan sahaja yang menerangkan sintesis terbitan lemak hidrazida khususnya menggunakan olein sawit yang dinyahbau, luntur dan ditapis. Di sini, melaporkan sintesis dan pencirian terbitan lemak hidrazida. Selain itu, terbitan palmitil hidrazida yang disintesis daripada palmitil hidrazida telah digunakan sebagai rujukan. Pencirian terbitan lemak hidrazida telah dijalankan dengan menggunakan spektroskopi fourier transform inframerah (FTIR), kromatografi gas (GC) dan spektroskopi resonans magnetik nuklear (NMR). Sintesis *N,N*-dimetil lemak hidrazida telah dijalankan pada suhu 100°C di dalam dioksana selama 6 jam pada nisbah molar 1:1:1 bagi hidrazida, dimetil sulfat dan kalium hidroksida. NMR proton mengesahkan bahawa produk

yang diperolehi adalah *N,N*-dimetil lemak hidrazida. Terbitan lemak hidrazida yang lain adalah surfaktan kationik yang dihasilkan menggunakan *N,N*-dimetil lemak hidrazida. Ketegangan permukaan untuk surfaktan kationik ini berbeza-beza bergantung kepada jenis asid klorida. Ketegangan permukaan yang lebih rendah telah dicatatkan untuk surfaktan kationik yang dihasilkan dengan asid klorida yang mempunyai kumpulan fenil. Di samping itu, surfaktan kationik juga boleh dihasilkan daripada tindak balas antara asid klorida dan lemak hidrazida. Oleh itu, penghasilan dan pencirian terbitan lemak hidrazida daripada olein sawit telah berjaya.



ACKNOWLEDGEMENTS

I would like to take this opportunity to express my gratitude to my supervisors Prof. Dato' Dr Wan Md Zin Wan Yunus, Dr Mansor Hj Ahmad @ Ayob, Dr Nor Azowa Ibrahim as well as my co-supervisors Dr Hazimah Abu Hassan and Dr Yeong Shoot Kian from Advanced Oleochemical Technology Division (AOTD), Malaysian Palm Oil Board (MPOB) for their valuable advises, guidance, encouragement and supervision during this study.

I also would like to thank my families especially my parents Tuan Haji Ahmad Saad and Puan Hajjah Rokiah Mohammad for their supportive, understanding and not to forget my colleagues in AOTD which help me a lot in completion of this thesis especially Tuan Nor Maznee Tuan Ismail, Yusrabbil Amiyati Yusof, Zafarizal Aldrin Azizul Hasan, Fadzlina Abdullah, Dr Nadzrinahamin Ahmad Nazir, Dr Zainab Idris, Bahriah Bilal, Mohd Nor Mamat@ Jusoh, Siti Hajar Bilal and Aishah Mohd Tahir.

Last but not least, I would like to thank Dr Azwani Mat Lazim from Universiti Kebangsaan Malaysia (UKM) and also to all my colleagues and friends for their priceless effort and help. I would not gain this valuable experience without the support and encouragement from all of you.

I certify that a Thesis Examination Committee has met on 28 December 2012 to conduct the final examination of Norashikin Binti Ahmad on her thesis entitled “Synthesis and characterization of palm fatty hydrazide derivatives” in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Mohd Aspollah Hj Md Sukari, PhD

Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Kamaliah Sirat, PhD

Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Internal Examiner)

Siti Mariam Mohd Nor, PhD

Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Internal Examiner)

Mohd Ambar Yarmo, PhD

Professor
Faculty of Science and Technology
Universiti Kebangsaan Malaysia
Malaysia
(External Examiner)

NORITAH OMAR, PhD

Associate Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 26 June 2013

This thesis was submitted to the Senate of Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Mansor Hj Ahmad @ Ayob, PhD

Associate Professor
Faculty of Science
Universiti Putra Malaysia
(Chairman)

Nor Azowa Ibrahim, PhD

Senior Lecturer
Faculty of science
Universiti Putra Malaysia
(Member)

Hazimah Abu Hassan, PhD

Director of Advanced Oleochemical Technology Division (AOTD)
Malaysian Palm Oil Board
(Member)

Yeong Shoot Kian, PhD

Head of Synthesis Product Development Unit
Advanced Oleochemical Technology Division (AOTD)
Malaysian Palm Oil Board
(Member)

Wan Md Zin Wan Yunus, PhD

Professor
Chemistry Department
Centre for Defence Foundation Studies
National Defence University of Malaysia

BUJANG BIN KIM HUAT, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

DECLARATION

I declare that the thesis is my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.



NORASHIKIN BINTI AHMAD

Date: 28 December 2012

LIST OF TABLES

Table		Page
2.1	Fatty acid composition of palm oils, palm kernel oils versus tallow and coconut oil (Basiron, 1996)	10
2.2	Fatty acid composition of palm oil and its products (Ahmad, 1994)	12
2.3	Examples of alkylating agents (Lamoureux <i>et al.</i> , 2009)	19
4.1	The elemental analysis data of palmityl hydrazide and Compounds 9, 10, 11 and 12	89

LIST OF FIGURES

Figure		Page
1.1	Reaction to produce cationic gemini surfactant from <i>N,N</i> -dimethyl acyl hydrazides.	2
1.2	Reaction to synthesize fatty hydrazides from triglycerides.	4
2.1	Reaction of ester with hydrazine.	6
2.2	Mechanism of reduction and acylation of hydrazones (Perdicchia <i>et al.</i> , 2003).	7
2.3	Possible reaction for the biosynthesis of fatty hydrazides in the presence of water (Mohamad, 2008).	8
2.4	Synthesis of fatty hydrazides derivatives containing pyridazine, oxapyrazole and phthalazine compounds. Above reactions were redrawn from Eissa, (2002).	14
2.5	Synthesis of <i>N'</i> -dodecanoyl- <i>N,N,N</i> -trimethylhydrazinium iodide. Above reaction was redrawn from Badr <i>et al.</i> (2010).	15
2.6	Name and molecular structures of volatile corrosion inhibitors.	16
2.7	Synthesis of DHBTPH. Above reaction was redrawn from Saliyan <i>et al.</i> (2008).	16
2.8	Chemical structures of corrosion inhibitors.	17
2.9	Chemical structure of 5-alkyl-4 phenyl -4H-[1,2,4]triazole-3-thiol (triazole derivatives).	17
2.10	Reaction of hydrazide and acrylonitrile.	20
2.11	Position of N(1) and N(2) atoms in hydrazides.	20
2.12	A typical schematic representation of a monomer surfactant.	22
2.13	Schematic representation of gemini surfactant.	23
4.1	Synthesis of <i>N,N</i> -dimethyl fatty hydrazides.	37

4.2	FTIR spectra of (a) fatty hydrazides and (bi-bv) <i>N,N</i> -dimethyl fatty hydrazides produced using different amounts of KOH.	39
4.3	GC chromatograms of (a) fatty hydrazides, (bi-biii) mixtures of fatty hydrazides and <i>N,N</i> -dimethyl fatty hydrazides and (biv-bv) <i>N,N</i> -dimethyl fatty hydrazides	41
4.4	FTIR spectra of (a) fatty hydrazides and (bi-biii) <i>N,N</i> -dimethyl fatty hydrazides produced at different reaction temperatures.	42
4.5	GC chromatograms of (a) fatty hydrazides, (bi-bii) mixtures of fatty hydrazides and <i>N,N</i> -dimethyl fatty hydrazides and (biii) <i>N,N</i> -dimethyl fatty hydrazides produced at different reaction temperatures.	43
4.6	FTIR spectra of (a) fatty hydrazides and (bi-biv) <i>N,N</i> -dimethyl fatty hydrazides produced using various reaction periods.	44
4.7	GC chromatograms of (a) fatty hydrazides, (bi-biii) mixtures of fatty hydrazides and <i>N,N</i> -dimethyl fatty hydrazides and (biv) <i>N,N</i> -dimethyl fatty hydrazides produced using various reaction periods.	45
4.8	FTIR spectra of (a) fatty hydrazides and (bi-biii) <i>N,N</i> -dimethyl fatty hydrazides produced in various solvents.	47
4.9	GC chromatograms of (a) fatty hydrazides and (bi-biii) <i>N,N</i> -dimethyl fatty hydrazides produced in various solvents.	48
4.10	FTIR spectra of (a) fatty hydrazides and (b) palmityl hydrazide.	49
4.11	GC chromatograms of (a) standard fatty acids (C8 to C18:0 and C18:1), (b) palmityl hydrazide and (c) fatty hydrazides.	50
4.12	GC/MS chromatogram of palmityl hydrazide.	52
4.13	Proposed fragment ions at retention time 8.50 min.	53
4.14	Proposed fragment ions at retention time 11.83 min.	53
4.15	GC/MS chromatogram of fatty hydrazides.	54
4.16	Proposed fragment ions at retention time 8.50 min.	55
4.17	Proposed fragment ions at retention time 11.10 min.	55
4.18	Proposed fragment ions at retention time 11.95 min.	55
4.19	Proposed fragment ions at retention time 14.45 min.	55

4.20	Proposed fragment ions at retention time 16.77 min.	55
4.21	Proton chemical shifts of (a) palmityl hydrazide and (b) fatty hydrazides.	57
4.22	Carbon chemical resonance of (a) palmityl hydrazide and (b) fatty hydrazides.	58
4.23	FTIR spectra of (a) fatty hydrazides, (b) <i>N,N</i> -dimethyl fatty hydrazides (c) palmityl hydrazide and (d) <i>N,N</i> -dimethyl palmityl hydrazide.	59
4.24	GC chromatograms of (a) fatty hydrazides, (b) <i>N,N</i> -dimethyl fatty hydrazides, (c) palmityl hydrazide and (d) <i>N,N</i> -dimethyl palmityl hydrazide.	60
4.25	Proton chemical shift of <i>N,N</i> -dimethyl palmityl hydrazide.	61
4.26	Proton chemical shift of <i>N,N</i> -dimethyl fatty hydrazides.	62
4.27	Reaction of <i>N,N</i> -dimethyl fatty hydrazides with adipoyl dichloride.	63
4.28	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 1 prepared in various solvents at 1 hour.	64
4.29	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 1 prepared in THF at various reaction periods.	66
4.30	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 1 prepared in MeCN at various reaction periods.	66
4.31	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 1 prepared in CHCl ₃ at various reaction periods.	67
4.32	Reaction of <i>N,N</i> -dimethyl fatty hydrazides with terephthaloyl dichloride.	67
4.33	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 2 prepared in various solvents at 1 hour.	69
4.34	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 2 prepared in THF at various reaction periods.	70
4.35	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 2 prepared in MeCN at various reaction periods.	70
4.36	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 2 prepared in CHCl ₃ at various reaction periods.	71
4.37	Reaction of <i>N,N</i> -dimethyl fatty hydrazides with hexanoyl chloride.	71

4.38	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 3 prepared in various solvents at 1 hour.	72
4.39	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 3 prepared in THF at various reaction periods.	74
4.40	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 3 prepared in MeCN at various reaction periods.	74
4.41	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 3 prepared in CHCl ₃ at various reaction periods.	75
4.42	Reaction of <i>N,N</i> -dimethyl fatty hydrazides with octanoyl chloride.	75
4.43	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 4 prepared in various solvents at 1 hour.	76
4.44	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 4 prepared in THF at various reaction periods.	78
4.45	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 4 prepared in MeCN at various reaction periods.	78
4.46	FTIR spectra of (a) <i>N,N</i> -dimethyl fatty hydrazides and (bi-biii) Compound 4 prepared in CHCl ₃ at various reaction periods.	79
4.47	Average surface tension of Tween 20 and its mixture with Compounds 1, 2, 3 and 4, respectively at different concentrations.	80
4.48	Reaction mechanism for acylation of amine by acyl chloride.	81
4.49	Possible reaction of fatty hydrazides or palmityl hydrazide with (a) adipoyl dichloride, (b) terephthaloyl dichloride, (c) hexanoyl chloride and (d) octanoyl chloride.	82
4.50	FTIR spectra of fatty hydrazides, (a) Compound 5, (b) Compound 6, (c) Compound 7 and (d) Compound 8.	84
4.51	FTIR spectra of palmityl hydrazide, (a) Compound 9, (b) Compound 10, (c) Compound 11 and (d) Compound 12.	85
4.52	¹³ C Carbon NMR spectra of (a) Compound 9 and (b) Compound 10.	87
4.53	¹³ C Carbon NMR spectra of (a) Compound 11 and (b) Compound 12.	88

LIST OF ABBREVIATIONS

RBDPOo	: Refined, Bleached and Deodorized Palm olein
RBD	: Refined, Bleached and Deodorized
MPOB	: Malaysian Palm Oil Board
KOH	: Potassium hydroxide
CHCl ₃	: Chloroform
THF	: Tetrahydrofuran
CMC	: Critical Micelle Concentration
MeCN	: Acetonitrile
HPLC	: High Performance Liquid Chromatography
DMSO- <i>d</i> ₆	: Dimethyl sulfoxide- <i>d</i> ₆
Tween 20	: Polyoxyethylene (20) sorbitan monolaurate
BSTFA	: <i>N,O</i> -bis(trimethylsilyl) trifluoroacetamide with 1% trimethylsilyl chloride
FTIR	: Fourier Transform Infrared
GC	: Gas Chromatography
GC/MS	: Gas Chromatography/Mass Spectrometry
NMR	: Nuclear Magnetic Resonance

TMS : Trimethylsilyl



© COPYRIGHT UPM

TABLE OF CONTENTS

ABSTRACT	Page
	ii
ABSTRAK	iv
ACKNOWLEDGEMENTS	vi
APPROVAL	vii
DECLARATION	ix
LIST OF TABLES	x
LIST OF FIGURES	xi
LIST OF ABBREVIATIONS	xv

CHAPTER

1	INTRODUCTION		
	1.0	Background of study	1
	1.1	Fatty hydrazides and fatty hydrazide derivatives	1
	1.2	Objectives of the study	5
2	LITERATURE REVIEW		6
	2.1	Synthesis of fatty hydrazides from esters and hydrazones	6
	2.2	Synthesis of fatty hydrazides from oils	7
	2.3	Palm oil and palm kernel oil	9
	2.3.1	Chemical composition of palm oil and palm kernel oil	10
	2.4	Oleochemicals	12
	2.5	Fatty hydrazides derivatives	13
	2.5.1	Application of fatty hydrazide derivatives in detergent products	13
	2.5.2	Application of fatty hydrazide derivatives in oil and gas industries	15
	2.5.3	Application of fatty hydrazide derivatives in medical fields	17
	2.6	Alkylation	18
	2.7	Acyl chloride and its application	20
	2.8	Surfactants	22
3	MATERIALS AND METHODS		26
	3.1	Materials, chemicals and solvents for the synthesis of hydrazide derivatives	26
	3.2	Synthesis of <i>N,N</i> -dimethyl fatty hydrazides	27
	3.2.1	Procedure for synthesis of <i>N,N</i> -dimethyl fatty Hydrazides	27
	3.3	Parameter study for synthesis of <i>N,N</i> -dimethyl fatty hydrazides	27
	3.3.1	Effect of catalyst loading	28
	3.3.2	Effect of temperature	28